

REVIEW

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Pyridyl disulfide-based thiol–disulfide exchange reaction: shaping the design of redox-responsive polymeric materials

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Recent years have witnessed an increase in the utilization of stimuli-responsive polymers in various areas of materials and biological sciences. In particular, disulfide linkage containing redox-responsive polymers have attracted interest in various biomedical applications ranging from fabrication of drug delivery vehicles to diagnostic interfaces. Cleavage of the disulfide linkage in the presence of an endogenous reducing agent, namely, glutathione, also found in increased amounts in diseased tissues has led to the incorporation of this particular linkage into several therapeutic platforms. Among the various methods available for introducing the redox-sensitive disulfide unit, the pyridyl disulfide (PDS) moiety has been one of the most widely employed building blocks. The rapid thiol–disulfide exchange reaction of the PDS group with thiol functional groups has been exploited from the reversible conjugation of therapeutic agents to the fabrication of redox-responsive crosslinked materials such as hydrogels and nanogels. This review provides an overview of various synthetic approaches utilized to incorporate this particular thiol-reactive motif into different types of polymeric materials and briefly highlights its utilization to obtain functional materials.

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1. Introduction

Conjugation of bioactive agents ranging from small molecules such as drugs, imaging agents, and receptor-specific targeting groups, to large bio-macromolecules such as proteins, oligonucleotides, and growth factors onto polymeric platforms is actively pursued in many areas of biomedical sciences such as

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materials ranging from nanofibers to nanogels for various biomedical applications such as combating bacterial infection to targeted drug delivery to cancer cells.



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ery and biomolecular immobilization. He holds interest in developing new condensation and multi-component polymerization methods by adapting versatile synthetic organic transformations into polymer chemistry.

diagnostics, therapeutics, and tissue engineering.¹ In all of these applications, the nature of the specific chemical linkages employed to conjugate the biologically relevant motif to the polymeric material plays a critical role in their eventual performance. Though seemingly trivial, the chemical attachment of biomolecules to polymeric materials has always been a challenge. The high efficiency of such conjugations under mild conditions, as well as their execution with high specificity, without the formation of by-products, is important. Enhanced steric hindrance posed by the large structure of polymers, as well as several biological molecules, makes such conjugations demanding. Furthermore, the multitude of several functional groups present on most biomolecules can lead to several undesirable reactions leading to eventual crosslinking or non-specific functionalization of these materials. For several decades, a lot of effort has been dedicated to developing versatile, effective, and easy to implement conjugation strategies. In recent years, click chemistry-based methodologies have expanded the conjugation tool-box and opened new avenues in the design, synthesis, and applications of functional polymeric materials.^{2–12} While efficient conjugation of (bio) molecules of therapeutic interest to polymeric materials can be a boon, robust conjugation to macromolecules may also compromise their performance. Thus conjugation of the therapeutic agent through a reversible linkage, which would release it from the polymeric component, preferably, at the site of action, would be highly desirable. While many reversible linkages such as hydrazones,¹³ acetals,¹⁴ carbamates,^{15,16} and Diels–Alder cyclo-adducts¹⁷ have been investigated to date, interest in the utilization of a disulfide-based linker has been ever increasing. The presence of an increased amount of glutathione, a thiol-containing tripeptide, in inter and intra-tumoral environments as an endogenous disulfide cleaving agent has propelled the

incorporation of the disulfide group into therapeutically relevant polymer conjugates.^{18–21} Examples in subsequent sections will highlight that one of the most effective approaches to attain disulfide linkage based conjugation is mediated through the utilization of the pyridyl-disulfide (PDS) moiety.

Another actively pursued area with emerging utilization of the disulfide linkage is fabrication of redox-responsive cross-linked polymeric materials.^{22,23} Crosslinked materials from nanogels, microgels to bulk hydrogels find widespread use in various applications related to diagnostics and therapy. In many of these applications, degradation of the crosslinked structure can be beneficial, e.g. promoting release of therapeutic agents when exposed to thiol-containing compounds, present either endogenously or exogenously. Facile fabrication of such disulfide linkage containing polymeric materials has been often undertaken by employing the thiol–disulfide exchange reaction. In particular, as discussed in sections below, the PDS based motif has been employed as the disulfide component due to its fast reaction with thiol-containing polymers or crosslinkers.

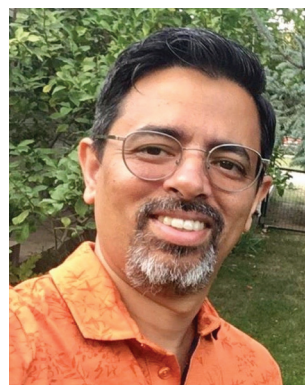
In this review, we aim to provide an overview of the vital role played by the thiol-reactive PDS group in the fabrication of various redox-responsive functional polymeric materials (Scheme 1). The range of polymeric materials where the PDS functional group has been employed spans from soluble polymers to crosslinked ones like nanoparticles, gels and surface coatings. This review highlights the synthetic approaches used for incorporating the PDS group, as well as discusses briefly the potential application of these PDS-containing polymeric materials. In particular, various approaches used over the past two decades are summarized. While earlier approaches involved introduction of the PDS-moiety through post-polymerization modifications, recent years have witnessed increased



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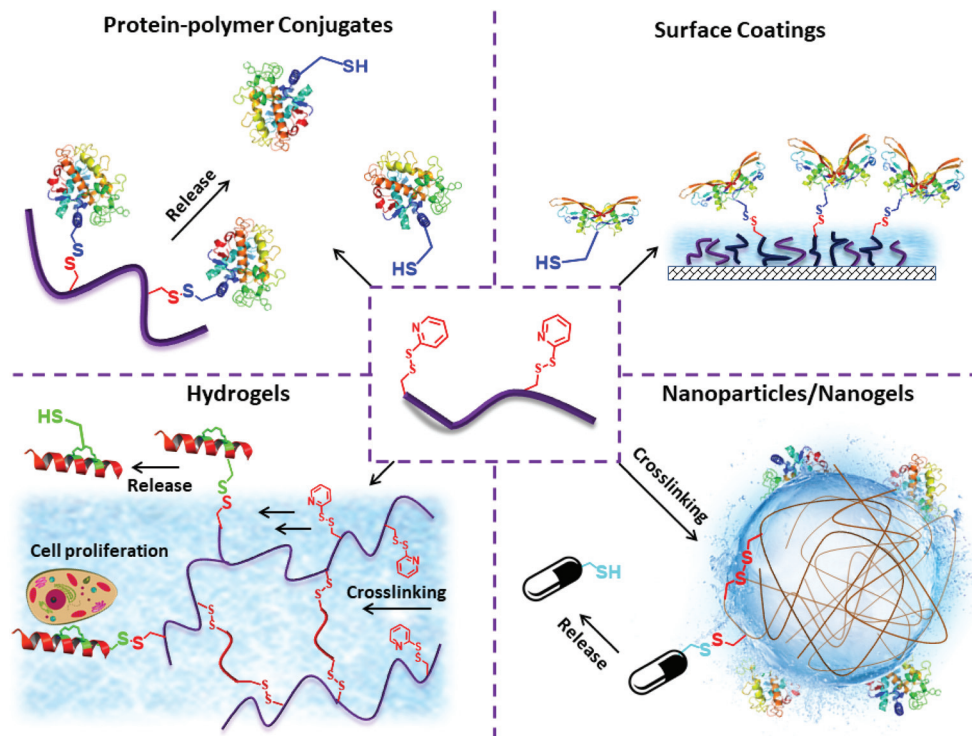
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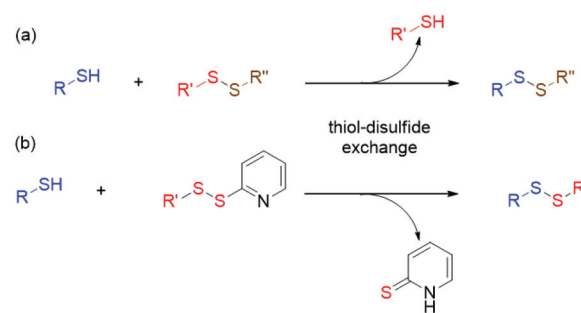
Scheme 1 Schematic representation of common applications of the PDS functional group based exchange chemistry in fabrication and functionalization of polymeric materials.

utilization of PDS-containing initiators, chain transfer agents (CTAs) and monomers.

2. Thiol-based conjugations: the thiol–pyridyl disulfide exchange reaction

The thiol functional group has been exploited for decades in bioconjugation processes ranging from modification of proteins with dyes to immobilization of ligands and proteins on solid supports for chromatographic purposes. In recent years, interest in thiol-based modifications has intensified due to high efficiency of the thiol-based click reactions. The thiol functional group is known to undergo efficient reactions²⁴ with a variety of functional groups such as electron-rich alkenes,²⁵ electron-deficient alkynes,^{26,27} *para*-fluoro benzenes,^{28,29} epoxides,^{30,31} thioesters^{32–34} and disulfides.^{35–38} In particular, the nucleophilic thiol–maleimide and radical thiol–ene reactions have been widely employed in fabrication and functionalization of polymeric materials.^{39–41} In particular, the nucleophilic Michael addition reaction between cysteine-containing proteins or peptides and highly electron-poor double bond containing maleimide functionalized polymers has been widely studied. While these thiol–ene reactions are quite efficient, other than the retro Michael-type reversal of phenylthio-maleimide adducts,⁴² these linkages are in general

not easily reversible. While not so common in the polymer community until recently, another reaction that the thiol function group undergoes with high efficiency is the thiol–disulfide exchange reaction. Biologists and biochemists have long used the thiol–disulfide exchange reactions for cleavage of disulfide bridges in proteins using dithiothreitol (DTT) as a reducing agent,⁴³ as well as for quantification of thiol groups using Ellman's assay.⁴⁴ The thiol–disulfide exchange is a S_N2-type nucleophilic substitution where a thiol-bearing molecule reacts with either a symmetric or mixed disulfide group to form a new thiol compound and a mixed disulfide (Scheme 2). Additionally, this exchange reaction is quite pragmatic since it can be generally implemented under mild conditions, at ambient temperature and physiological pH. The exchange



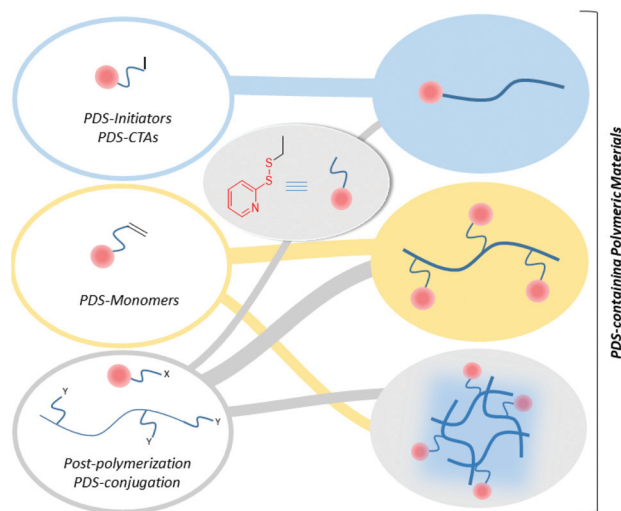
Scheme 2 General representations of (a) thiol–disulfide, and (b) thiol–pyridyl disulfide exchange reactions.

reaction can take place reversibly in various solvents including water, and can proceed in a broad range of reaction temperatures and pH.⁴⁵ Usually, thiol–disulfide exchange is promoted at high pH due to the significant formation of thiolate ions which enhances the active nucleophilic attacks to the disulfide bond. Additionally, the rate constant of the thiol-exchange reaction increases with increasing pK_a of the thiol group due to the increased nucleophilicity.⁴⁶

The PDS moiety has some characteristic attributes, which makes it a functional group of choice for the thiol-based exchange chemistry. In particular, the exchange reaction of the thiol group with a pyridyl disulfide functional group proceeds with very high efficiency. The reaction is driven to the right due to the formation of the more stable tautomeric pyridothione fragment from the pyridylthiol released upon exchange, which also precludes any thiol–disulfide exchange from the released thiol. Furthermore, the unique chemical structure of the PDS functional group *i.e.* the presence of a basic pyridine group *ortho* to the disulfide sulfur atoms efficiently facilitates the activation of disulfide groups in acidic medium through protonation of the pyridine nitrogen. Compared to other disulfide containing molecules, such as 5,5'-dithio-bis-(2-nitrobenzoic acid) (Ellman's reagent) which has applications in quantifying free thiol groups in a 7.8–10.4 pH range, the PDS functional group can react with free thiols under relatively low pH conditions (3.4–8.1) due to the favoured release of the 2-thiopyridone group.⁴⁷ Of pragmatic importance, the efficiency of the thiol-PDS exchange can be quantitatively followed using UV-Vis spectroscopy by determining the amount of released 2-thiopyridone through its characteristic absorbance *e.g.* near 343 nm in Krebs–Ringer phosphate buffer (pH 7.2).⁴⁸ The evolution of such an absorbance provides indirect evidence of the thiol-PDS exchange based functionalization, which is quite important for cases where it is either difficult or impossible to analyse the realization of reaction since the amounts of materials are too low for routine analytical techniques or in cases of functionalization on crosslinked materials and surfaces. The commercial availability of the commonly utilized starting material 2,2'-dithiodipyridine (2,2'-DTDP) has also facilitated the widespread adaptation of the PDS group for the fabrication of redox-responsive polymeric materials.

3. Incorporation of the pyridyl disulfide group into soluble polymers

In general, three different approaches have been employed for the synthesis of PDS group containing polymeric materials. These are categorized in the subsequent sections according to the chemical strategies used to introduce the PDS functionality. The widely employed methods are: (a) polymerizations with PDS-containing initiators and CTAs, (b) polymerization of PDS-containing monomers, and (c) post-polymerization modification with PDS-containing small molecules (Scheme 3). The sections below outline the synthetic details to highlight the

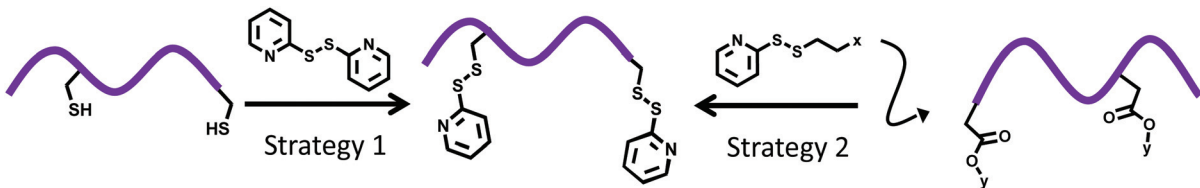


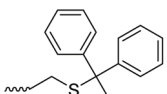
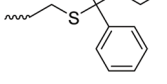
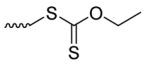
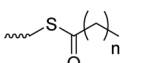
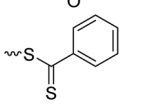
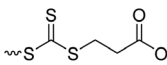
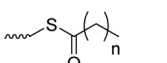
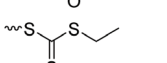
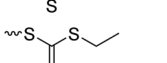
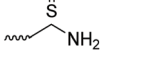
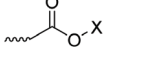
Scheme 3 General approaches for incorporating the PDS functional group into polymeric materials.

facile accessibility of PDS group bearing polymeric materials, along with a brief description of their possible applications.

3.1. Installation of the pyridyl disulfide group through post-polymerization modification

Modification of macromolecular constructs is challenging as a result of the encountered steric hindrance. Over the past decade, several studies demonstrate that employment of 'click' chemistry based transformations addresses such issues with post-polymerization functionalization. Efficient functionalization of polymeric materials can be achieved using chemical transformations that proceed with fast kinetics under mild conditions.⁴⁹ Along similar lines, the high efficiency and fast kinetics of the PDS based thiol–disulfide exchange reaction make it an attractive method for achieving a disulfide based post-polymerization functionalization of polymers appended with the PDS group. The most commonly used method to prepare PDS-functionalized polymers is to modify the polymers with a PDS-containing small molecule. This method requires appropriate reactive moieties as terminal end groups or side chains of polymers. In many of the early examples, the PDS group was used as a common moiety in heterobifunctional polymers because of its selective and orthogonal reactivity with thiol groups in the presence of other reactive groups.⁵⁰ With the advent of polymerization methods such as controlled living radical polymerization and ring opening polymerization, the incorporation of reactive functionalities into polymers to afford subsequent PDS attachment could be modularly achieved. The reaction conditions such as the type of polymer, temperature, pH, solvent and reaction stoichiometry between the polymer reacting moiety and PDS-containing molecule obviously have utmost importance to attain a high degree of functionalization. One of the widely used approaches for incorporating the PDS functional group onto polymers involves treatment of the thiol group on the polymer with 2,2'-DTDP

Table 1 General overview of post-polymerization PDS-group modification methodologies of various functional polymers


Polymer	Reacting moiety	Strategy	Modification method	Application/significance	Ref.
PEG		1	Deprotection of the trityl group, followed by exchange with pyridylsulfenyl chloride	Synthesis of heterobifunctional PEG derivatives	50
PLA		1	Deprotection of the trityl group, followed by exchange with 2,2'-DTDP	Synthesis of side chain thiol reactive polyesters	60
PEG		1	Aminolysis, followed by exchange with 2,2'-DTDP	Synthesis of heterobifunctional PEG derivatives	51
PEG		1	Aminolysis, followed by exchange with 2,2'-DTDP	Synthesis of heterobifunctional PEG derivatives	52
PMMA, PHPMA, PNIPAAm, POEGA, PS		1	Aminolysis, followed by exchange with 2,2'-DTDP	Synthesis of end chain thiol reactive RAFT polymers	54
PHPMA		1	Aminolysis, followed by exchange with 2,2'-DTDP	Preparation of polymeric carbohydrate scaffolds for the attachment of biomolecules	55
PNIPAAm		1	Aminolysis, followed by exchange with 2,2'-DTDP	Synthesis of thiol end chain reactive RAFT polymers	56
PEOz		1	Aminolysis, followed by exchange with 2,2'-DTDP	Preparation of diblock copolymers as nonviral gene carriers	57
PNIPAAm		1	Aminolysis, followed by exchange with 2,2'-DTDP	Preparation of thermally-responsive protein nanocapsules	58
PolyPOEGA Stars		1	Aminolysis, followed by exchange with 2,2'-DTDP	Synthesis of biodegradable star polymers with β -cyclodextrin complexes	59
PNIPAAm		2	Amidation by NHS functional PDS molecules	Preparation of thermo-responsive polymer-protein conjugates through disulfide bonding	61
PGA		2	Amidation by amine functional PDS molecule	Preparation of redox-responsive polymer-protein conjugates	62
PF-DBT-PEG		2	Amidation by amine functional PDS molecules	Preparation of conducting polymers with solubility-induced fluorescence conversion for detection of thiols	63
Hyaluronic acid		2	Amidation by amine functional PDS molecules	Preparation of redox-responsive shell-crosslinked nanoparticles for tumor-targeted drug delivery	64
PCL		2	Amidation by amine functional PDS molecules	Side chain functionalization of PCL with thiol containing peptides and resulting self-assembly	67

Abbreviations: PLA: poly(lactic acid); PMMA: poly(methyl methacrylate); PHPMA: poly(hydroxypropylacrylamide); POEGA: poly(oligoethylene glycol acrylate); PS: polystyrene; PGA: poly(glutamic acid); PF-DBT-PEG: poly(1,4-dithienyl benzothiadiazole)-poly(ethylene oxide).

(Table 1). Although this method proceeds with high efficiency by virtue of the facile nature of this particular thiol-disulfide exchange reaction, due to the facile oxidative coupling of the thiol group, the thiol-based precursors should be generated either *in situ* or just before the exchange reaction. As an alternative approach, the PDS group is incorporated into non-thiolated polymers through post-polymerization modification with a PDS-containing fragment (Table 1).

One of the very first examples of trapping of an *in situ* generated thiol group as a PDS group was reported by Kataoka and coworkers.⁵¹ In this study, acid sensitive initiator potass-

ium 4-(diethoxymethyl)benzylalkoxide was used in ring-opening polymerization of ethylene oxide to synthesize a linear polyethylene glycol (PEG) polymer, followed by addition of methanesulfonyl chloride to terminate the polymerization. The methylsulfonyl end group was then converted to a dithiocarbonate group with addition of potassium *O*-ethylthiocarbonate. Thereafter, the ethylthiocarbonate end group was converted to the PDS group in a one pot procedure involving reduction to the thiol group by aminolysis with addition of *n*-propylamine, followed by treatment with 2,2'-DTDP. Based on ¹H NMR analysis, 99% of ω -end conver-

sion to the 2-pyridyldithio group was confirmed. As an alternative, aminolysis of thioesters also generates reactive thiol groups on the polymer chain. Thioester terminated polymers can be readily obtained by subjecting an alkene terminated polymer to the thiol-ene reaction with thioacetic acid. For example, Nagasaki and coworkers synthesized linear PEG from ethylene oxide using allyl alcohol as an initiator, and terminated it with succinic anhydride to obtain a carboxylic acid and alkene terminated hetero-telechelic polymer.⁵² To generate the free thiol group on the polymer, the double bond was first functionalized with thioacetic acid through the thiol-ene reaction under UV irradiation, and then the thioester group was converted to the PDS group in the presence of 2,2'-DTDP and *n*-propylamine to give pyridyl disulfide and acid terminated hetero-bifunctional linear PEG. These studies also demonstrated orthogonal reactivity of the PDS groups in the presence of other functional groups. More recently, Wandrey and coworkers reported synthetic pathways for obtaining azide/PDS and amine/PDS terminated linear PEGs through modification of commercially available hydroxyl group terminated linear PEG (Fig. 1a).⁵³ In one approach, a PDS group was introduced to one end of the polymer chain by converting one of the hydroxyl groups to a tosylate leaving group, followed by its replacement with a thioester functional group. At this point, the other hydroxyl group was transformed to a mesylate leaving group. Aminolysis was used to convert the mesylate group to an amine moiety, while reducing the thioester group to a thiol functionality that was reacted with 2,2'-DTDP to install the PDS end group. Using a similar approach a hetero-telechelic PEG containing an azide and PDS group was also obtained.

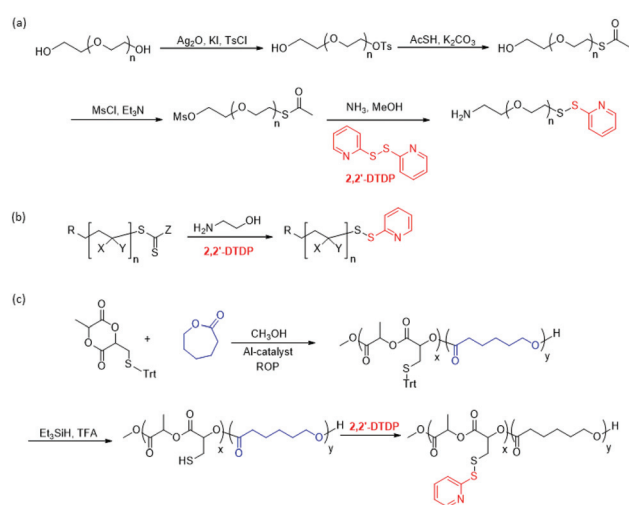


Fig. 1 Examples of synthetic approaches to obtain PDS-containing polymers using 2,2'-DTDP: (a) from commercial PEG polymers,⁵³ (b) from thioester terminated polymers obtained using RAFT polymerization,⁵⁶ and (c) from polymers containing a protected-thiol group side chain.⁶⁰ Adapted from Wandrey et al.⁵³ with permission. Adapted from Davis et al.⁵⁶ with permission. Copyright © 2009 by Wiley-VCH Verlag GmbH & Co. Adapted from Pappalardo et al.⁶⁰ with permission. Copyright © 2016 by American Chemical Society.

Over the past two decades, polymers synthesized through reversible addition fragmentation chain transfer (RAFT) polymerization have been widely used with PDS chemistry. As mentioned earlier, trithiocarbonate and thioester groups from RAFT chain transfer agents can be reduced to a free thiol group in the presence of primary or secondary amines, and the freshly generated thiol group can be converted to a PDS functionality in the presence of 2,2'-DTDP through the thiol-disulfide exchange reaction (Fig. 1b). Bulmus, Davis and coworkers have made several vital contributions in the field of post-modification of polymers with pyridyl disulfide.^{54,55} As one of the early examples, they reported introduction of a PDS end group on a polymer synthesized using RAFT polymerization.⁵⁶ They synthesized poly(*N*-isopropylacrylamide) (PNIPAAm) using RAFT polymerization for subsequent conjugation with a biomolecule. 2-Aminoethanol was used to reduce the trithiocarbonate group on the RAFT agent to a thiol, which upon treatment with 2,2'-DTDP furnished a PDS group containing polymer, which was subsequently functionalized with a hexapeptide through the thiol-PDS exchange reaction. This approach of installation of the PDS end group was also utilized for obtaining diblock copolymers consisting of a disulfide group at the block junction. Hsiue and coworkers reported a diblock copolymer system by coupling of poly(2-ethyl-2-oxazoline) (PEOz) and linear polyethylenimine (LPEI) through the thiol exchange reaction.⁵⁷ Briefly, thioacetate terminated PEOz was reacted with 2,2'-DTDP in the presence of ammonia to yield a PDS group terminated PEOz. This polymer is converted to LPEI in acidic solution at high temperature. Thereafter, thus obtained PDS-terminated LPEI was reacted with a thioacetate terminated PEOz using aminolysis to yield a disulfide junction containing diblock copolymer PEOz-*b*-LPEI. These diblocks were utilized to condense DNA into stable polyplexes. While the transfection ability of PEOz-*b*-LPEI was as high as LPEI, it exhibited lower toxicity toward cells, presumably due to lower cationic charge density. A similar approach was utilized by Maynard and coworkers for obtaining temperature-responsive protein cages using PNIPAAm synthesized *via* RAFT polymerization.⁵⁸ PDS-terminated PNIPAAm was obtained through post-polymerization aminolysis followed by treatment with 2,2'-DTDP. Conjugation of PDS-terminated PNIPAAm with a thiol-containing protein cage yielded thermally-responsive protein nanocapsules. A similar strategy for installation of the PDS group at the chain end in the three-arm star polymer was used by Davis and coworkers.⁵⁹ Apart from post-polymerization modification of end groups of polymers, side chains of polymers can also be modified to incorporate the PDS group. For example, Pappalardo and coworkers utilized a trityl (trt) protected thiol-containing lactide-based monomer, which they copolymerized with caprolactone (CL) and L-lactide to obtain degradable copolymers (Fig. 1c).⁶⁰ Removal of the protecting groups using triethylsilane and trifluoroacetic acid generated free thiol groups, which upon treatment with 2,2'-DTDP installed PDS groups along the side chain.

Another widely used approach to introduce the PDS group involves treatment of polymers containing reactive functional

To collectively demonstrate the preferred post-polymerization modification strategies, Table 1 summarizes a list of studies that utilize either end-chain or side chain functional polymers for installation of the PDS group.

3.2. Utilization of PDS-functionalized initiators and chain transfer agents

Over the past several years, Maynard and coworkers have utilized the PDS-thiol exchange chemistry to design several polymer-biomolecule conjugates.^{68,69} In one of their seminal contributions, a PDS-containing ATRP initiator was prepared by reaction of a PDS-containing alcohol with 2-bromo-2-methylpropionic acid through *N,N'*-dicyclohexylcarbodiimide (DCC) coupling to yield a PDS group bearing ATRP initiator (Fig. 2a).⁶⁹ A 2-hydroxyethyl methacrylate (HEMA) based hydrophilic polymer was synthesized, and utilized for conjugation with bovine serum albumin (BSA) through a thiol-PDS exchange reaction. Few years later, the same group polymerized a glycomonomer using the same PDS-containing initiator for conjugation of biomolecules and patterning on gold surfaces. Obtained glycopolymers were used for conjugation of small interfering RNA (siRNA) through the disulfide exchange reaction.⁷⁰

Due to concerns with residual amounts of metal salts that pose challenges in their complete removal and thus may possibly interfere with the activity of biomolecules or pose cellular toxicity, metal-free polymerization techniques such as RAFT have gained popularity.^{71,73} In the following years, several PDS group containing CTAs for utilization in RAFT polymerization were synthesized through esterification or amidation reaction between a RAFT CTA and with an amine or alcohol group bearing PDS moiety containing molecules.⁷⁴ A great deal of contributions with several innovative examples of PDS containing RAFT agents have been reported by Davis and coworkers. Using a ‘graft from’ approach Davis, Bulmus and coworkers reported a clever approach to obtain a hetero-bifunctional polymer–protein conjugate.⁷⁵ A bifunctional RAFT agent con-

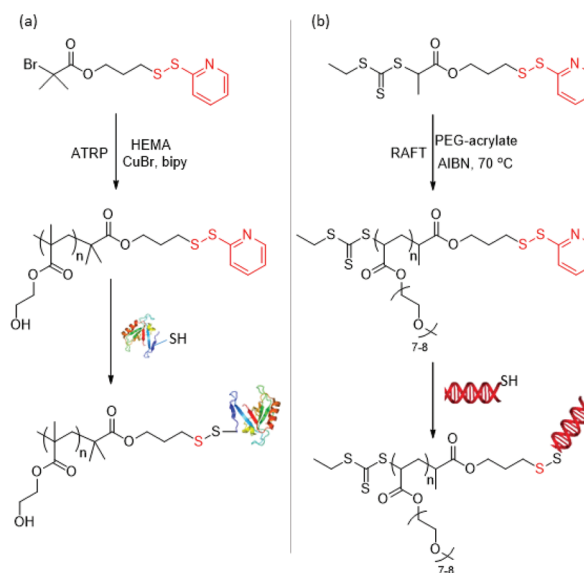
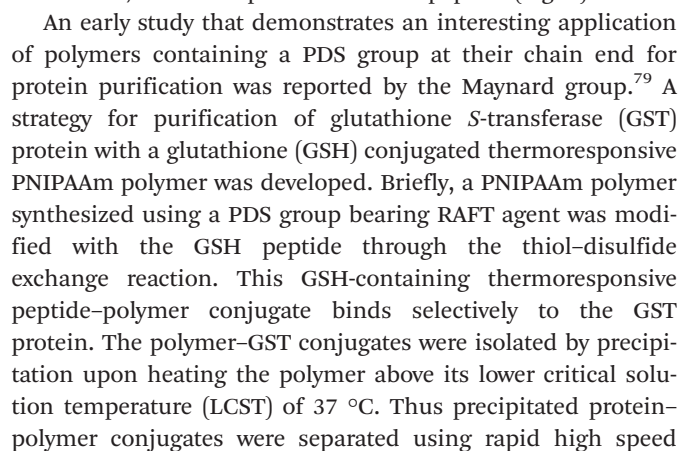


Fig. 2 (a) Synthesis of pHEMA with PDS containing the ATRP initiator,⁶⁹ and (b) synthesis of reversible siRNA–polymer conjugates⁷² mediated by thiol–pyridyl disulfide exchange reactions. Adapted from Maynard *et al.*⁶⁹ with permission. Copyright © 2018 by American Chemical Society. Adapted from Bulmus, Maynard *et al.*⁷² with permission. Copyright © 2008 by Royal Society of Chemistry.



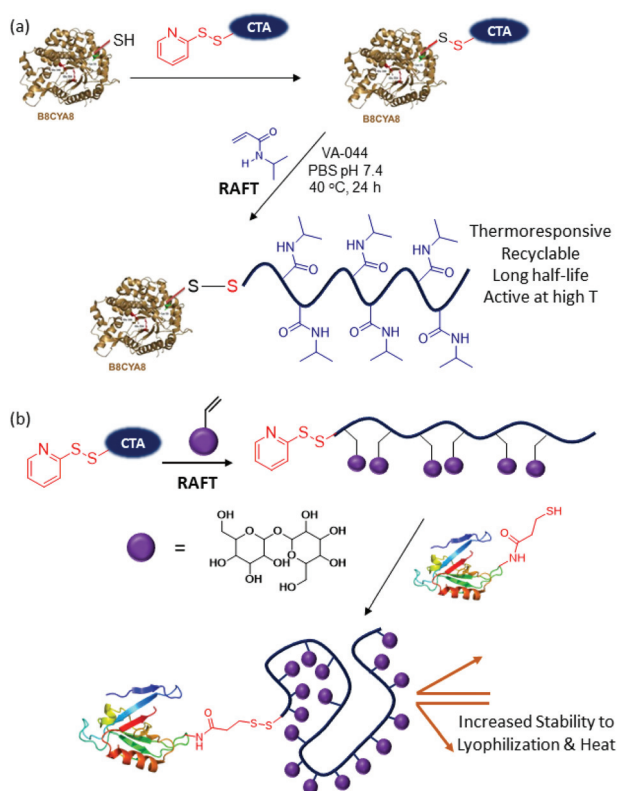


Fig. 4 (a) 'Graft-from' strategy for the synthesis of the thermo-responsive enzyme-polymer conjugate,⁸² and (b) 'graft-to' strategy for the synthesis of the trehalose based protein-polymer conjugate for protein stabilization.⁸³ Adapted from De, Datta *et al.*⁸² with permission. Copyright © 2018 by American Chemical Society. Adapted from Maynard *et al.*⁸³ with permission. Copyright © 2012 by American Chemical Society.

Geest and coworkers.⁸⁴ In this extensive study, authors compared the efficiency and suitability of polymers containing different reactive groups at the chain end, including the PDS group, to obtain protein-polymer conjugates using the 'graft-to' approach. *N*-Hydroxysuccinimide (NHS), pentafluorophenyl (PFP), PDS and maleimide (MAL) modified RAFT agents were prepared, and utilized to obtain hydrophilic pHEMA polymers. Obtained results suggested that the PDS group acts as one of the most efficient groups that enables efficient thiol-disulfide exchange based coupling of the investigated polymers with proteins like BSA and ovalbumin (OVA).

The Maynard group has also demonstrated that polymers containing the PDS group at the chain end can be conjugated to vault nanoparticles, which are natural non-immunogenic ribonucleoprotein particles present in most eukaryotes.⁸⁵ Authors fabricated protein nanoparticles responsive to both temperature and pH by using polymers composed of NIPAAm and acrylic acid (AA) monomers. Thiol-reactive polymers were conjugated with the human major vault protein through the thiol exchange reaction with thiol groups on the surface of the vault protein nanoparticles. In this design the temperature responsiveness of nanoparticles was dependent on pH of the

media. While there was no response below 60 °C at pH 7.0, nanoparticles started to respond reversibly to the temperature below 60 °C at pH 6 and started to aggregate.

Hetero-telechelic polymers containing the PDS group at the chain end have been reported by employing appropriately designed initiators or chain transfer agents. Bulmus, Davis and coworkers reported an elegant strategy for direct access of α,ω -hetero-telechelic copolymers bearing an azide and a PDS group at chain ends.⁸⁶ A trithiocarbonate RAFT CTA containing an azide and a PDS group was used for polymerization of various monomers such as styrene, NIPAAm, HPMA, OEGA and methyl methacrylate (MMA). Efficient hetero-bifunctionalization was demonstrated with the α -azide, ω -PDS containing PNIPAAm polymer through functionalization with an alkyne-containing biotin ligand and thiol-containing tripeptide GSH, among other biomolecules.

In recent years, a set of elegant reports by Ghosh and coworkers have demonstrated that the versatile chemistry of the thiol-PDS exchange reaction can be extended to functionalization of biodegradable polymers. To obtain PDS end group containing polyesters, hydroxyethyl pyridyl disulfide was used to initiate ROP of lactide.⁸⁷ Obtained polymers could be easily modified with thiol-containing bioactive ligands like carbohydrates, and chromophores like naphthalene-diimide. Apart from using the thiol-disulfide exchange based conjugations, the free thiol groups generated by addition of DTT to the PDS terminated polylactide allow subsequent functionalization using radical and nucleophilic thiol-ene conjugations (Fig. 5).

While most of the reports employ PDS-group containing initiators to install this thiol-reactive group at the chain end of the linear polymers, the use of appropriately designed

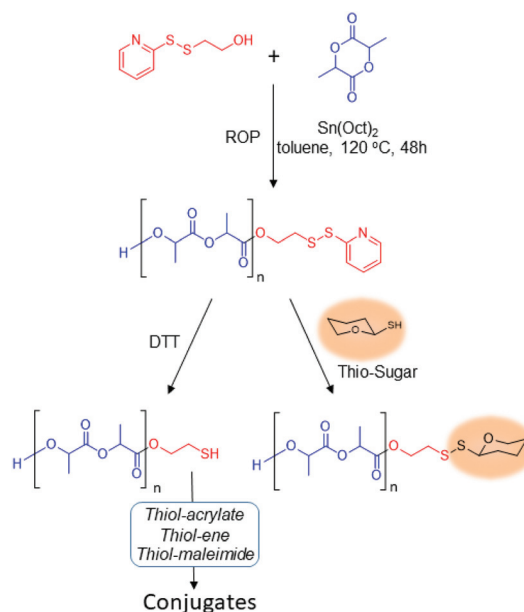


Fig. 5 Synthesis of thiol-reactive PDS-containing biodegradable polymers.⁸⁷ Adapted from Ghosh *et al.*⁸⁷ with permission. Copyright © 2012 by American Chemical Society.

initiators can yield multiarm polymers with the thiol-reactive group at their core. As one of the examples, Davis and co-workers reported a branched PHPMA polymer containing the PDS group at the mid-chain location.⁸⁸ A PDS group containing CTA bearing two thiocarbonate moieties was synthesized to obtain polymers using RAFT polymerization. The thiol exchangeable disulfide group was thus located at the midpoint of the obtained polymers. Successful conjugation of the obtained polymer with the thiol-bearing protein BSA was demonstrated.

Another example of installing the PDS group at the core of multiarm polymers was reported by Sanyal and coworkers.⁸⁹ A tetra-arm dendron based initiator was designed to contain a PDS group at the focal point and radical initiators at its periphery. The PDS-modified core allows for redox-responsive conjugation of thiol containing small molecules or peptides to obtain multiarm star polymers. The hydroxyl groups at the periphery of dendrons were converted to bromine-containing ATRP initiators to grow PEGMA-based multiarm star polymers. Water soluble multi-arm star polymers bearing the PDS at the core were efficiently modified with thiol-containing dyes and bioactive ligands. In particular, a thiol bearing KLAK-sequence containing apoptosis inducing peptides was conjugated and enhanced cytotoxicity of the polymer-peptide conjugate compared to the naked peptide was demonstrated. Instead of incorporation of the PDS group at the core of the multi-arm star polymers, recent work by Kent, Davis and coworkers shows that it is also possible to fabricate star copolymers where the PDS groups are located at the chain termini of the arms emanating from a crosslinked core.⁹⁰ First, polymers containing PDS groups at chain ends were synthesized using RAFT polymerization. The hetero-telechelic arms were core-cross-linked in the presence of an amine-reactive monomer (to subsequently attach a fluorescent dye) and a bis-acrylamide cross-linker. Enhanced association with particular subsets of white blood cells was observed for the thiol-reactive nanoparticles. The authors postulated that such thiol-reactive PDS-containing nanoparticles may provide a viable strategy for targeting subsets of primary human immune cells.

While a lot of focus has been given toward utilization of the PDS group containing polymers at the chain end for the fabrication of polymer-biomolecule conjugation, this particular thiol-disulfide exchange reaction has been also demonstrated as a powerful methodology to furnish redox-responsive diblock and multiblock copolymers. Thayumanavan and coworkers exploited the high efficiency of this coupling to obtain cleavable diblock copolymers (Fig. 6).⁹¹

Polymers with the PDS group located at one of the chain ends were synthesized using ATRP. The PDS end group in these polymers could be reduced to yield thiol-terminated polymers. Coupling of polymers bearing the PDS and thiol groups proceeded under mild conditions in an efficient manner to provide diblock copolymers with a redox-responsive linkage at the junction point. Thayumanavan and coworker utilized this strategy in a subsequent study to obtain block copolymer assemblies that were sensitive to changes in three

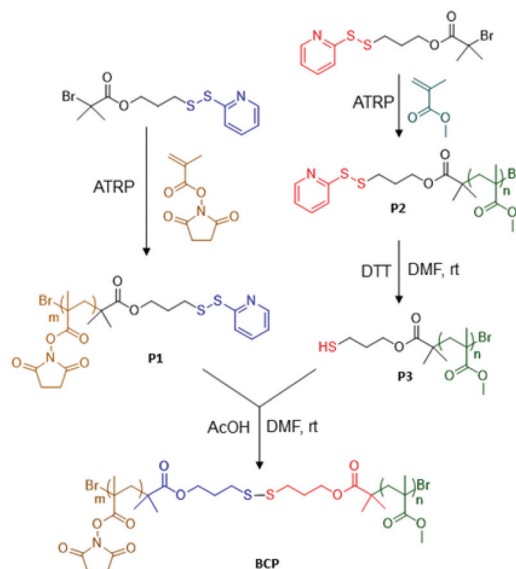


Fig. 6 Synthesis of a redox-responsive diblock copolymer obtained using thiol-disulfide mediated polymer-polymer coupling.⁹¹ Adapted from Thayumanavan et al.⁹¹ Copyright © 2007 by American Chemical Society.

different stimuli: temperature, pH and redox potential sensitive.⁹² The amphiphilic diblock copolymer with a redox-sensitive disulfide junction was obtained through combination of the thiol-end group containing tetrahydropyran-protected HEMA polymer as the acid-sensitive hydrophobic block and the PDS end group containing PNIPAAm polymer as the thermo-sensitive hydrophilic block. Authors note that such multi-responsive systems not only allow fine-tuning of release kinetics but also would provide delivery of therapeutics in a location-specific manner.

In an elegant study, Gibson and coworkers employed the reversible nature of the disulfide linkage to modulate the solubility of hydrophilic polymers. They demonstrated that the LCST of the polymer bearing the PDS group at the chain end changes upon treatment with glutathione. In this study, PEGMA and NIPAM were polymerized using a PDS-bearing RAFT agent.⁹³ Interestingly, it was observed that in reducing environments, the LCST of these polymers changes because of the exchange of the 2-mercaptopyridine end group with a thiol group. Thus a coil-to-globule transition could be attained for the polymer under 'isothermal conditions' *i.e.* without changing the temperature. Using these types of polymers, another interesting approach to redox-responsive polymers was reported by Gibson and coworkers. They polymerized a series of acrylamides, methacrylates and acrylates with a PDS-bearing RAFT agent.^{94,95}

Aminolysis of the sulphur containing functional group of the RAFT CTA fragment at the chain end of these polymers induced a poly(condensation) reaction (Fig. 7). The thiol group containing polymers reacted *in situ* with each other through thiol-PDS exchange reaction to result in step-growth polymerization. Using this approach, dual responsive polymers consist-

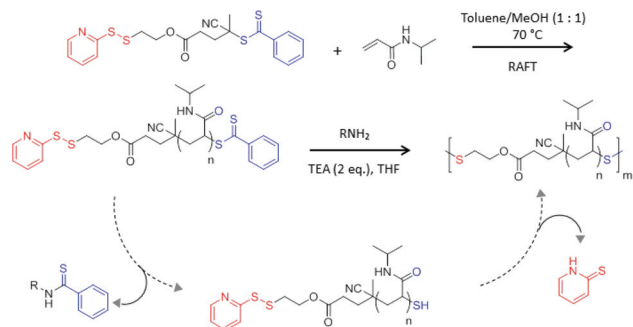


Fig. 7 Synthesis of redox-responsive thermoresponsive polymers through *in situ* polycondensation.⁹⁵ Adapted from Gibson *et al.*⁹⁵ with permission. Copyright © 2012 by Royal Society of Chemistry.

ing of thermoresponsive PNIPAAm based domains connected to each other through redox-responsive disulfide bonds formed during polycondensation were obtained.

A variety of examples that utilize appropriately designed PDS containing initiators and chain transfer agents in preparation of functional polymeric materials are summarized in Table 2.

The above examples demonstrate that a wide variety of semi-telechelic or hetero-telechelic polymers bearing the PDS group can be obtained through clever design of initiators and CTAs. While most examples above demonstrate the usage of the PDS group bearing polymer for obtaining a polymer–biomolecule conjugate using the ‘graft-to’ approach, the PDS group has also been employed to facilitate the synthesis of polymer–biomolecule conjugates using the ‘graft-from’ approach. For example, Davis, Bulmus and coworkers utilized a PDS group bearing short PEG based water-soluble RAFT agent for undertaking polymerization of NIPAAm from BSA.⁹⁶ The PDS group containing a hydrophilic RAFT agent was conjugated to the protein using the thiol-PDS exchange reaction with the available free cysteines on BSA. The obtained BSA-RAFT agent was used to grow PNIPAAm, which rendered the protein with ability to undergo thermo-responsive aggregation. Likewise, in another study, Maynard and coworkers demonstrated that the ‘grafting from’ strategy can be successfully used to obtain polymer–siRNA conjugates, where the siRNA conjugation with an ATRP initiator was mediated through exchange between the thiol group on the oligonucleotide and a PDS group bearing the ATRP initiator.⁹⁷ Overall, the wide range of applications of the PDS end group containing polymers as discussed above highlights the versatile nature of this reactive group.

3.3. Installation through utilization of PDS-functionalized monomers

The thiol-reactive PDS functional group can be directly incorporated into polymers as side chains using PDS-containing monomers. In recent years, polymers containing multiple PDS groups as chain units have attracted attention since they allow conjugation of several therapeutically active molecules which can be subsequently cleaved under reducing conditions, and thus provide a selective pathway for release.⁹⁸ Also, as mentioned

above, due to the orthogonal reactivity of the PDS group to many other reactive monomers, copolymers bearing the PDS group along with other reactive groups can undergo multi-functionalization in a selective manner.⁹⁹ In this section, we review approaches taken to incorporate the PDS functional group as side chain moieties into different classes of polymers, obtained by using PDS-containing monomeric building blocks.

An early example of PDS-containing monomer was reported by Stayton, Hoffman and coworkers.¹⁰⁰ A novel pyridyl disulfide acrylate (PDSA) monomer was synthesized through reaction between hydroxypropyl pyridyl disulfide and acryloyl chloride in the presence of pyridine. Later, this monomer was polymerized with methacrylic acid and butyl acrylate using AIBN-initiated free-radical polymerization to yield a redox- and pH-sensitive amphiphilic copolymer. Peptides containing cysteine residues were conjugated to this terpolymer using the thiol–disulfide exchange reaction. Antisense oligonucleotides are ionically complexed with hexalysine conjugated polymer and cellular uptake of the thus-obtained GSH sensitive complex was investigated. Shortly after, Thayumanavan and coworkers reported the PDS group bearing methacrylic monomer (PDSMA), which was polymerized using ATRP to yield homopolymers and copolymers, the latter upon polymerization in the presence of *N*-hydroxysuccinimide methacrylate.¹⁰¹ The PDSMA monomer was obtained in two steps: 2,2'-DTDP was reacted with 2-mercaptoethanol to yield a PDS group bearing alcohol, which was subsequently treated with methacryloyl chloride in the presence of a base. The orthogonally reactive copolymers could be functionalized in a quantitative manner using thiol and amine containing molecules either in a one-step or sequential reaction. Such orthogonal functionalization provides an attractive method to obtain multi-functional systems given the presence of amine and cysteine groups in various peptides and proteins. The versatility of the PDS group containing polymers was also demonstrated by Bulmus and coworkers, who synthesized polymers using RAFT polymerization of the PDS containing methacrylate monomer.¹⁰² They showed that thiol-reactive homopolymers could undergo simultaneous multi-functionalization in a highly efficient manner upon treatment with mixtures of thiol-containing molecules. They also observed that the PDS-containing hydrophobic homopolymers upon treatment with a hydrophilic peptide like GSH yield amphiphilic copolymers that assemble into spherical aggregates in aqueous media, as suggested through dynamic light scattering (DLS) and atomic force microscopy (AFM) studies. The same group exploited the PDS group containing homopolymers obtained using RAFT polymerization as a macroinitiator to obtain diblock copolymers for fabrication of crosslinked micellar drug delivery systems,¹⁰³ as discussed in the subsequent section. Along with these contributions, synthesis of other multiblock copolymers using the PDSMA monomer has also been reported.¹⁰⁴ As expected, amphiphilic multi-block copolymers assemble to provide micellar structures that can act as delivery vehicles for therapeutic agents. Stayton and coworkers designed a diblock copolymer with HPMa and PDSMA containing hydrophilic blocks and pH-responsive PAA,

Table 2 General overview of functional initiators and chain transfer agents utilized in the synthesis of PDS end group containing polymeric materials

Monomer	Initiator (ATRP and ROP)/chain transfer agent (RAFT)	Polymerization method	Application/significance	Ref.
NIPAAm		ATRP	Modification of reduced BSA for polymer grafting	68
HEMA			Cysteine-reactive polymers for conjugation to BSA	69
Glycomonomer			Reactive polymers for conjugation to siRNA and patterning on gold surface	70
NHSMA, MMA		RAFT	Synthesis of disulfide bridged block copolymers	91
NIPAAm, HEMA			Synthesis of multi stimuli-responsive block copolymers	92
PEGA, BA			Synthesis of end group pyridyl disulfide-functionalized homo- and amphiphilic block copolymers	76
NIPAAm, PEGA		RAFT	Mechanistic study to elucidate the stability of PDS group under radical polymerization conditions	74
NIPAAm, HEA		RAFT	Modification of reduced BSA for polymer grafting	96
PEGA		RAFT	Synthesis of heterotelechelic protein-polymer-biomolecule conjugates	75
HPMA, AA, BMA, DMAEMA		RAFT	Preparation of peptide block copolymer conjugates	78
Glycomonomer			Synthesis of protein-trehalose glycopolymer conjugates	83
PEGA			Synthesis of siRNA-polymer conjugates	72
NIPAAm			Synthesis of glutathione-polymer conjugates for glutathione; S-transferase protein purification	79
PEGA			Synthesis of siRNA-polymer conjugates	77
NIPAAm, DEGMA, OEGMA			Synthesis of chain termini thiol reactive polymers displays glutathione triggered LCST change	93
PEGMA, fluoromonomer			Synthesis of fluoruous copolymers for protein bioconjugation	80
OEGMA			Synthesis of thiol-reactive star polymers for biocompatible polymeric nanoparticle fabrication	90
HEA, HPMA			Synthesis of different end group reactive polymers and comparative study to assign their protein conjugation efficiencies	84
NIPAAm, AA			Preparation of pH- and temperature-responsive vault protein nanoparticles	85
NIPAM, DMA, OEGMA, MMA, DEGMA, OEGA, tBuA			Synthesis of biodegradable poly(disulfide)s	94
NIPAAm		RAFT	Synthesis of recyclable thermoresponsive polymer-β-glucosidase conjugate	82
			Synthesis of thermoresponsive polymers displaying redox-responsive LCST behaviour	95
Lactide		ROP	Synthesis of end group functional polylactide	87

Table 2 (Contd.)

Monomer	Initiator (ATRP and ROP)/chain transfer agent (RAFT)	Polymerization method	Application/significance	Ref.
HPMA		RAFT	Synthesis of branched polymer-protein conjugates	88
DEGMEMA		ATRP	Synthesis of dendritic multi-arm polymers with a thiol-reactive PDS core	89
HPMA, MMA, styrene, OEGA, NIPAAm		RAFT	Synthesis of heterobifunctional polymers	86

Abbreviations: NHSMMA: *N*-hydroxysuccinimide methacrylate; PEGA: poly(ethylene glycol) acrylate, HEA: 2-hydroxyethyl acrylate; HPMA: *N*-(2-hydroxypropyl) methacrylamide; DMAEMA: 2-dimethylaminoethyl methacrylate; DEGMA: diethylene glycol methacrylate; OEGMA: oligoethylene glycol methacrylate; DMA: *N,N*-dimethyl acrylamide; *t*BuA: *tert*-butyl acrylate; DEGMEMA: diethylene glycol methyl ether methacrylate.

DMAEMA and BMA containing hydrophobic blocks.¹⁰⁵ The obtained pH-responsive polymeric micellar carrier with PDS side chains was used to conjugate ovalbumin through the thiol-PDS exchange reaction (Fig. 8). Compared to free ovalbumin, the micelle-ovalbumin conjugate showed higher antigen cross-penetration during *in vitro* experiments. Moreover, 90% of micelle-ovalbumin conjugates were internalized in cells after 1.5 h, compared to 40% internalization of free albumin.

The PDS moiety has been also integrated as side chains in biodegradable constructs, as reported by Zhong and co-workers.¹⁰⁶ In their study, a PDS group containing a cyclic carbonate monomer and poly(ϵ -caprolactone) were copolymerized using ROP. Thereafter, the PDS groups were exchanged with PEG-thiol to install hydrophilic side chains. DOX-loaded micelles were fabricated using this amphiphilic copolymer and redox-triggered release was demonstrated. As expected, DOX release was enhanced in the presence of DTT (10 mM), in contrast to release in a non-reductive environment. Another interesting approach to incorporate the PDS group as side chains into poly(β -amino ester)s was reported by Langer and coworkers. Poly(β -amino ester)s can be readily synthesized by the reaction of primary amines and diacrylates. In this study, 2-(pyridyldithio)-ethylamine was used as an amine-bearing monomer and polymerized with a series of diacrylates to yield poly(β -amino ester)s bearing PDS based side chains. Obtained polymers were functionalized with thiol-containing peptide

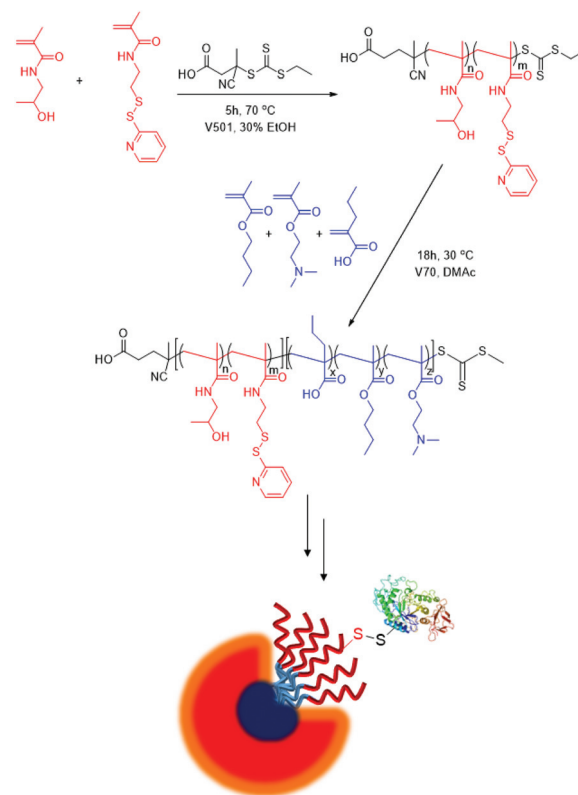


Fig. 8 Antigen conjugation on PDS-bearing pH-responsive polymers.¹⁰⁵ Adapted from Stayton *et al.*¹⁰⁵ with permission. Copyright © 2014 by Elsevier.

RGDC and mercaptoethylamine through the thiol–disulfide exchange reaction. Thus obtained polymers showed DNA-binding ability, which decreases upon fragmentation when exposed to intercellular glutathione concentration, and thus can be anticipated to facilitate release of DNA inside the cell.¹⁰⁷

In an interesting approach, the thiol–disulfide exchange reaction was used in an elegant manner by Ghosh and co-workers to obtain telechelic poly(disulfide) polymers bearing the PDS group at chain termini.¹⁰⁸ The polycondensation reaction of aldrithiol and 1,6-hexanedithiol provides access to polymers with disulfide groups along the backbone and PDS group at chain ends. Importantly, the selective thiol–PDS exchange reaction allows modification of the chain ends to install desired groups. For example, hydroxyl groups were obtained at chain ends by treatment with 2-mercaptoethanol, followed by ROP of cyclic lactide to yield triblock copolymers. In another study, the authors demonstrated that the end groups could be capped with a variety of functional groups using PDS-containing capping reagents during the polymerization step.¹⁰⁹ In a subsequent study, the authors utilized this strategy to obtain a redox-responsive micellar construct with tunable degradation.¹¹⁰ The triblock copolymers composed of a middle block possessing different hydrophobicity (attained by using 1,6-hexanedithiol or 2,2'-(ethylenedioxy)diethane thiol), and outer blocks based on poly(triethylene glycol monomethyl ether) methacrylate (PTEGMA) were used (Fig. 9). The outer blocks were grown using ATRP, after installation of the required bromine initiator on the telechelic hydroxyl group

containing polymer obtained using their previous protocol. Triblock copolymers thus obtained were assembled into micelles in aqueous media and were loaded with a hydrophobic dye, namely, Nile Red. It was demonstrated that the overall hydrophobic character of the inner block has a noteworthy effect on release of the cargo. An increased hydrophobicity of the micelle interior leads to slower disintegration and release of cargo in the presence of GSH, presumably due to the slow diffusion of the reducing agent into the non-polar interior. In a recent study, authors extended the thiol–PDS exchange based condensation polymerization to the synthesis of hyperbranched polydisulfides by employing an A2 + B3 type condensation approach of complementary monomers.¹¹¹ In another related study, PDS incorporation *via* termination of polydisulfide synthesis *via* 2,2'-DTDP was later employed by Lu and coworkers to obtain redox-responsive amphiphilic triblock mPEG-*b*-PDS-*b*-mPEG copolymers.¹¹²

Another elegant and mechanistically interesting approach to synthesize redox-degradable polyamides was recently reported by Thayumanavan and coworkers by employing cascaded step-growth polymerization methodology based on thiolactone ring opening and disulfide–thiol exchange reactions. A set of backbone disulfide containing polymers with variable side chain groups were obtained by polymerization of an amine containing PDS monomer along with thiolactone monomers carrying diverse functionalities.¹¹³

An overview of the preparation and applications of a variety of PDS-containing polymeric materials obtained using PDS functional monomers is shown in Table 3.

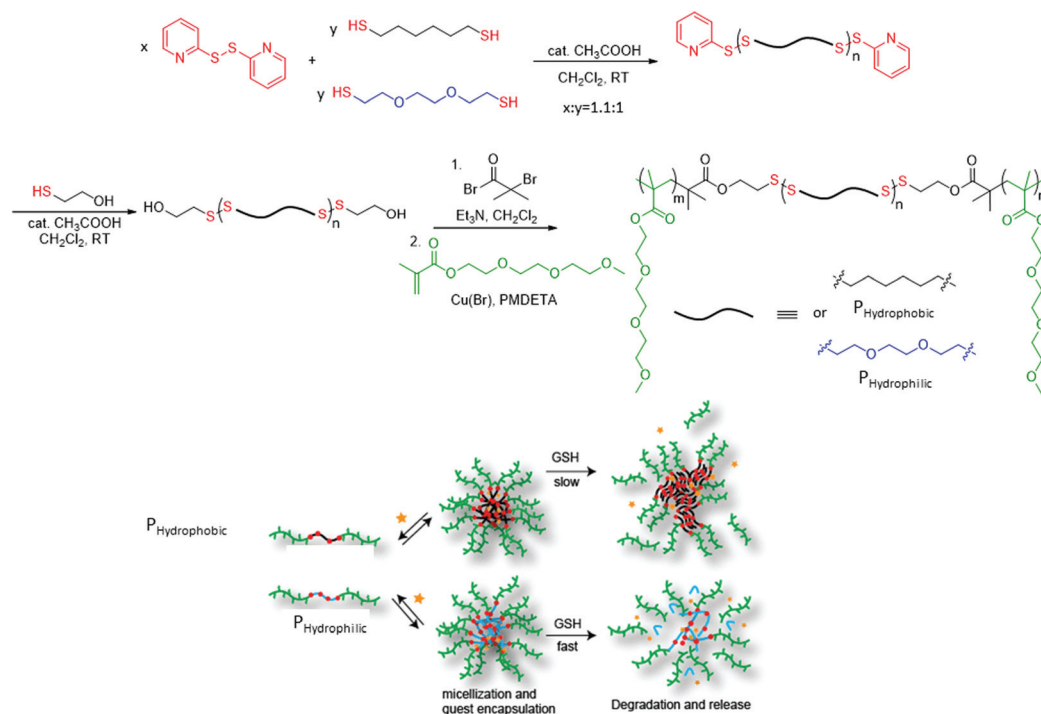
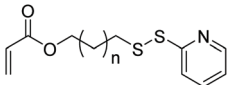
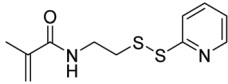
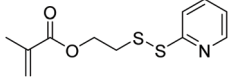
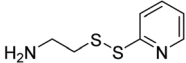
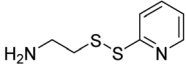
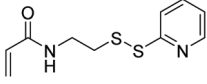
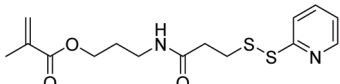


Fig. 9 Synthesis of poly(disulfide) polymers using the PDS-thiol exchange reaction.¹¹⁰ Adapted from Ghosh et al.¹¹⁰ with permission. Copyright © 2020 by Royal Society of Chemistry.

Table 3 Examples of functional monomers, comonomers and types of polymerization employed for preparation of PDS-containing polymeric materials

Comonomer	PDS monomer	Polymerization method	Application/significance	Ref.
—		FRP	Synthesis of pH and redox responsive nanoparticles for drug delivery	98
AA, BA		FRP	Synthesis of pH and glutathione responsive polymers	100
OEMA, PEGA		RAFT	Preparation of graphene/copolymer composites for pH and biodegradation controlled drug delivery	104
HPMA, DMAEMA, AzEMAm		FRP	Preparation of peptide-functionalized mRNA polyplexes	99
HPMA, AA, DMAEMA, BMA		RAFT	Synthesis of pH-responsive polymer micelle carriers	105
NHSMA		ATRP	Synthesis of orthogonally functionalizable copolymers	101
—		RAFT	Synthesis of side chain thiol reactive polymers	102
HPMA		RAFT	Preparation of doxorubicin conjugated acid- and redox sensitive crosslinked micelles	103
OEGMA		RAFT	Synthesis of surface-functionalizable polymer nanogels	122
OEGMA		RAFT	Synthesis of polymer nanogels for concurrent binding and delivery of proteins and lipophilic small molecules	123
OEGMA		RAFT	Synthesis of self-crosslinked polymer nanogels for drug delivery	124
OEGMA		RAFT	Synthesis of self-crosslinked polymer nanogels	125
OEGMA		RAFT	Synthesis of polymer nanogels for intracellular drug delivery	126
OEGMA, DEGMA		RAFT	Synthesis of enzyme-polymer hybrid nanogels	127
TrMA		FRP	Synthesis of trehalose glycopolymer nanogels for glucagon stabilization	128
HPMA		RAFT	Fabrication of gelatin based dynamic hydrogels	134
PEGMA, DEGMA		FRP	Fabrication of bulk and patterned hydrogels for reversible protein capture and cell attachment	137
VP, VPI, VCL		RAFT	Synthesis of <i>N</i> -vinyl lactam based reactive copolymers	138
Diacrylates		MAP	Synthesis of cationic polymers with thiol-reactive side chains for DNA delivery	107
Thiolactones		ROP/disulfide exchange	Synthesis of polyamides with backbone disulfide units	113
EAMFB, TBCPAM, DMA		RAFT	Synthesis of pH and redox responsive polymer nanoparticles	114
CBMA		RAFT	Fabrication of cell viable degradable hydrogels	135

Abbreviations: OEMA: oligoethyl acrylate; AzEMAm: 2-azidoethylmethacrylamide; TrMA: trehalose methacrylate; VP: *N*-vinylpyrrolidone; VPI: *N*-vinylpiperidone; VCL: *N*-vinylcaprolactam; EAMFB: *N*-ethylacrylamide-2-(4-formylbenzamide); TBCPAM: *N*-(*tert*-butoxycarbonyl)-propylaminoacrylamide; MAP: Michael addition polymerization.

4. Pyridyl disulfide chemistry in fabrication and functionalization of crosslinked materials: polymer nanoparticles, nanogels and hydrogels

Apart from obtaining redox-responsive attachment of therapeutic agents to soluble polymers, the thiol-PDS exchange reaction has also been exploited for reversible crosslinking or attachment *i.e.* for fabrication and/or functionalization of crosslinked materials such as micellar nanoparticles, nano-

gels, hydrogels, and surface coatings. In particular, the incorporation of redox-sensitive crosslinks into bulk or particle based drug delivery platforms is quite important since it enhances the release of the therapeutic agent under enhanced GSH environments specific to cancerous tumors and cells. The breakdown of the crosslinked material enhances release of encapsulated therapeutic agents, as compared to release pathways such as diffusion or erosion of the matrix. In this part of the review, we highlight different types of chemically cross-linked materials that have been obtained using the PDS group mediated thiol–disulfide exchange reactions.

4.1. PDS-functional group-based fabrication and functionalization of polymer nanoparticles

One of the earliest examples of a redox-sensitive crosslinked micellar nanoparticle system was reported by Bulmus and co-workers.¹⁰³ An amphiphilic diblock copolymer was synthesized using RAFT polymerization of HPMa and PDS-containing methacrylate monomers. Treatment of this diblock copolymer with tris(2-carboxyethyl)phosphine (TCEP), in the presence of a maleimide-containing hydrazine based doxorubicin derivative, resulted in the *in situ* generation of drug conjugated disulfide crosslinked micellar nanoparticles with an average diameter of 60 nm in aqueous media. Redox-responsive degradation of nanoparticles was observed upon treatment with an excess amount of TCEP. The clever design of this drug delivery system benefits from the pH-dependent drug release due to the presence of the hydrazone linkage between the drug and the carrier, as well as from redox-responsive degradation of the carrier to warrant eventual clearance. Another example of incorporation of the PDS group onto crosslinked polymeric nanoparticles was reported by Hubbell and coworkers.¹¹⁴ They reported the synthesis of PDS-containing nanoparticles for reversible conjugation of thiol-containing small molecules, peptides and proteins. Using emulsion polymerization in the presence of a carboxylate-pluronic, followed by treatment with a cysteamine-PDS derivative, they obtained crosslinked polymeric nanoparticles bearing the thiol-reactive PDS groups on their surface. Reversible conjugation of these nanoparticles with a thiol-containing biotin, antigenic peptides and ovalbumin was also demonstrated. An example of polymeric nanoparticles responsive to both acidic pH and reducing environments was reported by Fulton and coworkers.¹¹⁵ A set of copolymers both bearing the PDS group along with either aldehyde or amine functional groups were synthesized using a PDS-containing acrylamide monomer. These copolymers were crosslinked through imine- and disulfide-linkages in a sequential manner to afford dual-crosslinked polymeric nanoparticles. Complete dissociation of nanoparticles was obtained only in the presence of both low acidic pH and reducing environments obtained using TCEP. While examples of such redox-responsive nanoparticles are limited, a surge of recent interest in nanogels has led to many reports where the PDS group has played an important role, as discussed in the following section.

It should be noted that the PDS functional group has been also used to modify inorganic surfaces. For example, thiol-modified surfaces of silica nanoparticles can be converted to PDS group containing surfaces which can be further functionalized with thiol-containing polymers and proteins.^{116,117} Alternatively, using a slightly different approach, thiol-modified surfaces of nanoparticles can be directly functionalized with PDS-bearing polymers through the thiol-PDS exchange reaction. Surface modification using the latter strategy was reported by Oupicky, Brock and coworkers, who reported PNIPAM coated mesoporous silica nanoparticles for temperature-depending uptake and release of small molecules.¹¹⁸ In

particular, thiol-functionalized silica nanoparticles were coated with PNIPAM containing a PDS group at the chain end. Results showed that polymer-coated nanoparticles exhibited a low level of leakage at 38 °C, a temperature above the LCST of the polymer.

4.2. PDS-functional group-based fabrication and functionalization of redox-responsive nanogels

In recent years, PDS-containing nanogels have been used for protein/peptide delivery applications since these biologically active agents can be attached in a reversible fashion, and thus can be transported and released at the site of interest.^{119,120} Naturally occurring or engineered cysteine containing proteins and peptides can be conjugated on PDS-containing nanogels under mild conditions using the thiol-PDS exchange reaction. Besides protein/peptide conjugation, the PDS functional group has also been used to introduce crosslinking to obtain stable nanogels that are degradable under a reducing environment. Two different approaches for crosslinking have been investigated. One of the methods involves *in situ* generation of thiols by cleaving few of the PDS groups on the polymer, which leads to crosslinking by reaction with the residual PDS units on polymers. The alternative method involves formation of self-assembled nanoaggregates from PDS-containing polymers, followed by crosslinking with dithiol-containing molecules through the thiol exchange reaction.

Using the first approach, Thayumanavan and coworkers have made seminal contributions to the fabrication of redox-responsive nanogels using PDS-containing copolymers.^{121–123} In one of their work, PDSMA and PEGMA were randomly copolymerized using RAFT polymerization.¹²⁴ These PEGMA-based polymers have relatively low LCST and give nanoaggregates above their LCST. After formation of nanoaggregates, DTT is added to the solution to induce crosslinking to yield nanogels (Fig. 10). *In situ* encapsulation of hydrophobic molecules such as drugs can be achieved during formation of the nanogels, which can be released when exposed to biological reducing

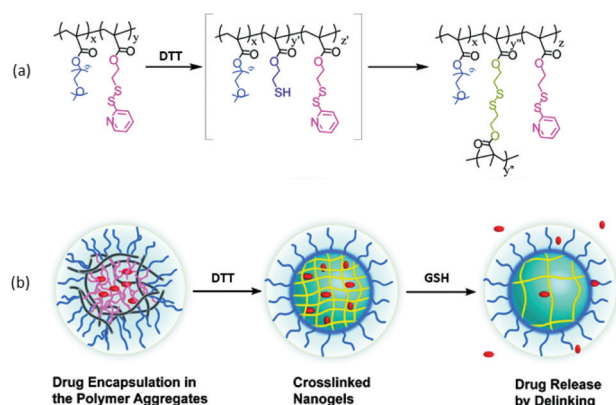


Fig. 10 (a) Synthetic approach, and (b) crosslinking strategy of PDS bearing polymers to fabricate a redox-responsive nanogel based drug delivery system.¹²⁴ Adapted from Thayumanavan et al.¹²⁴ with permission. Copyright © 2010 by American Chemical Society.

agents such as GSH. Facile fabrication of the drug encapsulated delivery vehicle, where crosslinking can be disrupted under biological conditions found in cancerous tissues and cells, makes this nano-drug delivery system a promising platform for addressing challenges faced in diseases such as cancer.

In another study, Thayumanavan, Maynard and coworkers used a similar strategy to fabricate dye encapsulated nanogels that were conjugated with the thiol-containing protein BSA.¹²⁵ Lipophilic small dye molecules were loaded into nanoaggregates and were encapsulated during nanogel formation. Thereafter, unreacted PDS groups on the nanogels were utilized for conjugation of thiol-modified BSA through the thiol-PDS exchange reaction. Recently, in an elegant approach, Thayumanavan and Liu demonstrated that self-assembled nanogels can be obtained through simultaneous crosslinking and functionalization of nano-aggregates.¹²⁶ A copolymer containing two reactive components was synthesized using PDSMA, PEGMEMA and a thio-lactone based monomer. Reaction with an amine-containing molecule leads to the opening of the thio-lactone ring which results in the formation of a thiol-functional group. This newly formed thiol group attacks the PDS group on the polymer and thus leads to crosslinking of the aggregate. This three-component reaction based fabrication of nanogels does not utilize DTT as in previous cases and leads to desired functionalization in the same step.

Using the alternative approach, which involves crosslinking of PDS-containing copolymers with a di-thiol based crosslinker, Zhao, Liu and coworkers prepared nanoparticles using the LCST behaviour of the PDS-containing PEG based polymer, in the presence of PDS modified porcine pancreatic lipase (PPL).¹²⁷ Assembled polymers and PDS-PPL were crosslinked using *meso*-2,3-dimercaptosuccinic acid while they are in the nanoaggregate form. The resulting PPL containing nanogels showed enhanced heat resistance and reusability compared to native PPL, which could have been a promising candidate for biocatalyst applications. As another example of nanogel based biomolecular stabilization, Maynard and coworkers reported a PDS-containing glycopolymers based nanogel for delivery of glucagon (Fig. 11). Glucagon is a peptide hormone which hypoglycemia patients used to regulate the blood sugar level. However, instability and poor solubility of glucagon limit the use of drugs. In particular, PDSMA and trehalose based monomers were polymerized and crosslinked by PEG-dithiol to form a nanogel. The residual PDS groups in the nanogels were used for conjugation of thiolated glucagon. The glucagon-nanogel conjugate showed high stability in solution compared to non-encapsulated glucagon.¹²⁸ A recent study by Zhao and coworkers reported that thiolated proteins can be employed as crosslinkers to obtain nanogels using PDSMA-containing copolymers.¹²⁹ A P(NIPAM-*co*-PDSMA) copolymer was crosslinked using thiolated lipase to obtain protein containing nanogels. It was demonstrated that subsequent treatment of nanogels with reduced BSA, followed by bio-mineralization of CaCO₃, increased the stability of encapsulated proteins against trypsin.

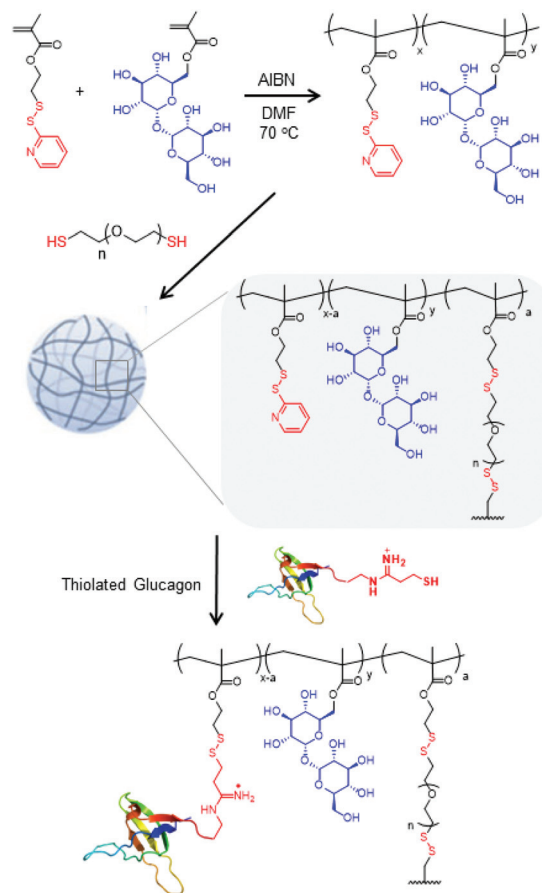


Fig. 11 Glucagon conjugation on a nanogel obtained using a thiol-disulfide exchange reaction.¹²⁸ Adapted from Maynard et al.¹²⁸ with permission. Copyright © 2011 by Wiley-VCH Verlag GmbH & Co.

4.3. PDS-functional group in fabrication and functionalization of hydrogels

Hydrogels are three dimensional structures of crosslinked hydrophilic polymers that have applications in a wide range of biomedical applications. While their ability to undergo facile functionalization is important, incorporation of degradable junctions augments their performance in many applications. The dynamic nature of the disulfide linkage can impart hydrogels with either or both of these characteristics.^{130,131} Hence, PDS groups have been used to install crosslinking junctions in polymeric networks. For example, Wang and coworkers reported an injectable hydrogel system which undergoes fast gelation through the thiol-PDS exchange reaction.¹³² In their study, hyaluronic acid (HA) was first modified with 1,6-diaminohexane to introduce free amine groups on the polymer, which were reacted with succinimidyl 3-(2-pyridyldithio)propionate (SPDP) to generate PDS groups. Hydrogels were fabricated upon mixing of the PDS-containing HA and PEG-dithiol at 37 °C in PBS. A cysteine-containing chemokine and stromal cell-derived factor 1R α were encapsulated inside the hydrogel and cell viability tests were performed with various cell lines. Results showed that cell viability in the hydrogels was sus-

tained up to 7 days. Likewise, Ossipov, Redl and coworkers utilized a PDS-conjugated HA derivative to obtain a HA-fibrin interpenetrating double network hydrogel through fibrogenesis and disulfide-linkage formation.¹³³

The PDS group is often used as a masked thiol group; in particular, for cases where multi-thiol systems are needed. Generation of free thiol groups is achieved quite rapidly under mild reducing conditions. For example, Ayres and coworkers synthesized the PDS bearing polymer *via* RAFT polymerization of PDSMA and HPMa.¹³⁴ Reactive thiol groups were generated after cleaving the PDS groups with TCEP. The thiolated polymer was crosslinked with norbornene functionalized gelatin using the radical mediated thiol–norbornene click reaction in the presence of UV irradiation to afford gelatin based hydrogels. As another example, poly(carboxybetaine) based hydrogels were reported by Jiang, Tsai and coworkers using a copolymer obtained from carboxybetaine methacrylate and a PDS-containing monomer, namely, *N*-(3-(2-(pyridyldithio)propanamido)ethyl methacrylate (PDPMA), using RAFT polymerization.¹³⁵ A telechelic dithio-terminated poly(carboxybetaine) polymer synthesized using RAFT polymerization was used as the crosslinker. Hydrogels were formed upon mixing the polymer and crosslinker within a minute. Enhancement in cell proliferation was observed for hydrogels containing RGD-based peptide crosslinkers, also integrated within the hydrogels using the thiol–disulfide exchange reaction.

Apart from utilization of hydrogels as bulk materials, these soft materials find applications in the form of surface coatings, in areas ranging from diagnostics to implant surface modifications for better bio-assimilation. The reversibility of the disulfide bond affords a dynamic linkage for fabrication of ‘catch and release’ interfaces. Such systems can be used to isolate certain biomolecules upon modification of PDS-containing hydrogels with specific ligands which can capture those biomolecules from a mixture, and subsequently release them for further analysis. For example, Iwasaki and coworkers immobilized antibody fragments onto polymeric brushes.¹³⁶ In this study, 2-methacryloyloxyethyl phosphorylcholine (MPC) and glycidyl methacrylate (GMA) were polymerized *via* ATRP. Epoxy groups on the polymer brushes were modified with PDS groups which were modified with Fab’ fragments through the thiol–PDS exchange reaction, and higher affinity of the thus modified surface against FITC-conjugated mouse IgG was demonstrated. In a recent example, Sanyal and coworkers reported PDS-containing hydrogels for capture and release of proteins.¹³⁷ A surface coating was fabricated using photopolymerization of the PDSMA monomer to afford a soft interface that could be modified with a thiol-containing protein binding ligand, namely, a thiolated biotin. Upon exposure to a mixture of extravidin and BSA, only extravidin was captured on the surface through specific biotin–avidin interaction. Subsequently, the bound protein was released through treatment with a solution containing DTT and the purity of the isolated protein was analysed using gel electrophoresis. Furthermore, upon using a cell-adhesive peptide decorated hydrogel, cells adhered and proliferated on the surface, and

could be released upon treatment with DTT, while preserving their viability (Fig. 12).

In recent years, thin polymeric coatings have been extensively used to obtain functional interfaces that can be used for attachment or detection of biomolecules. While most reports utilize the amine-activated ester or thiol–maleimide conjugations, a recent example by Pich and coworkers reported the PDS group mediated exchange reaction as a viable tool.¹³⁸ They synthesized various poly(*N*-vinyl lactam)-based copolymers with PDS groups as side chains using RAFT polymerization. Contact angle measurement results and a fluorescence microscopy study indicated that the reactive films based on the PDS-functionalized copolymers allowed facile, direct, and environmentally friendly modification of surfaces from aqueous solution suggesting potential application in surface decoration of tissue-engineering scaffolds and medical implants. Another interesting application of the disulfide based surface obtained using thiol–disulfide exchange was reported by Akimoto and coworkers who modified cell-culture glass surfaces with disulfide linked polyethyleneimine (PEI).¹³⁹ In this study, PEI, a transfection reagent, was modified with the PDS functional group through amidation with a PDS-con-

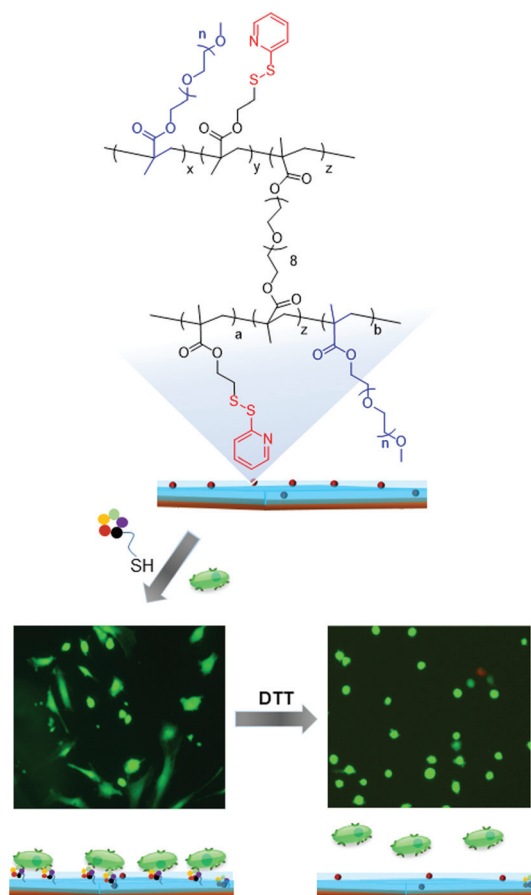


Fig. 12 Cell attachment and release using thiol–disulfide exchange reaction on a hydrogel interface.¹³⁷ Adapted from Sanyal et al.¹³⁷ Copyright © 2018, American Chemical Society.

taining activated carboxylic acid, namely, *N*-succinimidyl-3-(2-pyridyldithio)propionate. Thus obtained PDS-conjugated PEI was coated onto the glass surface which was previously modified with thiol- and amine groups using appropriate trialkoxysilane reagents. *In vitro* studies showed that attached PEI was released from the surface upon exposure to a cysteine containing buffer solution (1 mM), which was not toxic against 293T cells. Authors note that such modified redox-responsive cell culture surfaces can find applications in reverse transfection.

5. Conclusions

The diverse examples of disulfide bearing redox-responsive polymeric materials fabricated using the thiol-PDS exchange reaction surveyed in this review clearly highlight the powerful role that this transformation has played in shaping the current state of the art of such materials. Utilization of PDS-based initiators and monomers allows one to incorporate this exchangeable unit at precise locations such as chain ends and side chains, which in turn facilitates subsequent programmed installation of the reversible disulfide linkage within polymeric materials. As noted in various examples, such predetermined incorporation of the disulfide unit can either lead to dissociations that may result in the release of a single therapeutic agent like drugs or biomolecules from the polymer or result in complete dissociation of the polymeric material to release the encapsulated materials. While the advances in contemporary polymer synthesis have played a critical role in the fabrication of PDS-containing materials, it is the stable and compatible nature of the PDS functional group toward various polymerization techniques that have enabled its facile incorporation. Perhaps the most important reason that is responsible for its widespread utilization is the high specificity and fast kinetics of the exchange reaction with thiols, under extremely mild conditions. Given the increasing sophistication of emerging therapeutic platforms, one can only expect that the PDS-based exchange reaction will continue to play a crucial role in the design and fabrication of future redox-responsive smart materials.

Conflicts of interest

There are no conflicts to declare.

References

- 1 E. M. Pelegri-Oday, E. W. Lin and H. D. Maynard, *J. Am. Chem. Soc.*, 2014, **136**, 14323–14332.
- 2 E. Blasco, M. B. Sims, A. S. Goldmann, B. S. Sumerlin and C. Barner-Kowollik, *Macromolecules*, 2017, **50**, 5215–5252.
- 3 W. H. Binder and R. Sachsenhofer, *Macromol. Rapid Commun.*, 2007, **28**, 15–54.
- 4 R. M. Arnold, N. E. Huddleston and J. Locklin, *J. Mater. Chem.*, 2012, **22**, 19357.
- 5 J.-F. Lutz, *Angew. Chem., Int. Ed.*, 2007, **46**, 1018–1025.
- 6 Y. Zou, L. Zhang, L. Yang, F. Zhu, M. Ding, F. Lin, Z. Wang and Y. Li, *J. Controlled Release*, 2018, **273**, 160–179.
- 7 M. Arslan, G. Acik and M. A. Tasdelen, *Polym. Chem.*, 2019, **10**, 3806–3821.
- 8 N. Cengiz, T. N. Gevrek, R. Sanyal and A. Sanyal, *Chem. Commun.*, 2017, **53**, 8894–8897.
- 9 Y. N. Yuksekdog, T. N. Gevrek and A. Sanyal, *ACS Macro Lett.*, 2017, **6**, 415–420.
- 10 M. A. Gauthier, M. I. Gibson and H. A. Klok, *Angew. Chem., Int. Ed.*, 2009, **48**, 48–58.
- 11 K. S. Anseth and H. A. Klok, *Biomacromolecules*, 2016, **17**, 1–3.
- 12 Z. P. Tolstyka, W. Richardson, E. Bat, C. J. Stevens, D. P. Parra, J. K. Dozier, M. D. Distefano, B. Dunn and H. D. Maynard, *ChemBioChem*, 2013, **14**, 2464–2471.
- 13 V. N. Rao, S. Mane, A. Kishore, J. Das Sarma and R. Shunmugam, *Biomacromolecules*, 2012, **13**, 221–230.
- 14 E. R. Gillies, A. P. Goodwin and J. M. J. Fréchet, *Bioconjugate Chem.*, 2004, **15**, 1254–1263.
- 15 D. Aydin, M. Arslan, A. Sanyal and R. Sanyal, *Bioconjugate Chem.*, 2017, **28**, 1443–1451.
- 16 L. Chambre, A. Degirmenci, R. Sanyal and A. Sanyal, *Bioconjugate Chem.*, 2018, **29**, 1885–1896.
- 17 A. Castonguay, E. Wilson, N. Al-Hajaj, L. Petitjean, J. Paoletti, D. Maysinger and A. Kakkar, *Chem. Commun.*, 2011, **47**, 12146–12148.
- 18 B. Sui, C. Cheng and P. Xu, *Adv. Ther.*, 2019, **2**, 1900062.
- 19 J. K. Oh, *Polym. Chem.*, 2019, **10**, 1554–1568.
- 20 B. Saha, S. Bhattacharyya, S. Mete, A. Mukherjee and P. De, *ACS Appl. Polym. Mater.*, 2019, **1**, 2503–2515.
- 21 Z. Deng, J. Hu and S. Liu, *Macromol. Rapid Commun.*, 2020, **41**, 1900531.
- 22 M. Ejaz, H. Yu, Y. Yan, D. A. Blake, R. S. Ayyala and S. M. Grayson, *Polymer*, 2011, **52**, 5262–5270.
- 23 Z. Gao, B. Golland, G. Tronci and P. D. Thornton, *J. Mater. Chem. B*, 2019, **7**, 7494–7501.
- 24 L. T. T. Nguyen, M. T. Gokmen and F. E. Du Prez, *Polym. Chem.*, 2013, **4**, 5527–5536.
- 25 A. B. Lowe, *Polym. Chem.*, 2010, **1**, 17–36.
- 26 O. Daglar, U. S. Gunay, G. Hizal, U. Tunca and H. Durmaz, *Macromolecules*, 2019, **52**, 3558–3572.
- 27 U. S. Gunay, M. Cetin, O. Daglar, G. Hizal, U. Tunca and H. Durmaz, *Polym. Chem.*, 2018, **9**, 3037–3054.
- 28 S. Agar, E. Baysak, G. Hizal, U. Tunca and H. Durmaz, *J. Polym. Sci., Part A: Polym. Chem.*, 2018, **56**, 1181–1198.
- 29 G. Delaittre and L. Barner, *Polym. Chem.*, 2018, **9**, 2679–2684.
- 30 S. De and A. Khan, *Chem. Commun.*, 2012, **48**, 3130–3132.
- 31 N. Cengiz, J. Rao, A. Sanyal and A. Khan, *Chem. Commun.*, 2013, **49**, 11191–11193.
- 32 B. T. Worrell, S. Mavila, C. Wang, T. M. Kontour, C. H. Lim, M. K. McBride, C. B. Musgrave, R. Shoemaker and C. N. Bowman, *Polym. Chem.*, 2018, **9**, 4523–4534.
- 33 C. Ghobril, K. Charoen, E. K. Rodriguez, A. Nazarian and M. W. Grinstaff, *Angew. Chem., Int. Ed.*, 2013, **52**, 14070–14074.

- 34 M. D. Konieczynska, J. C. Villa-Camacho, C. Ghobril, M. Perez-Viloria, K. M. Tevis, W. A. Blessing, A. Nazarian, E. K. Rodriguez and M. W. Grinstaff, *Angew. Chem., Int. Ed.*, 2016, **55**, 9984–9987.
- 35 C. Boyer, A. H. Soeriyadi, P. J. Roth, M. R. Whittaker and T. P. Davis, *Chem. Commun.*, 2011, **47**, 1318–1320.
- 36 P. J. Roth, D. Kessler, R. Zentel and P. Theato, *J. Polym. Sci., Part A: Polym. Chem.*, 2009, **47**, 3118–3130.
- 37 M. Arslan, R. Sanyal and A. Sanyal, *Polym. Chem.*, 2020, **11**, 1763–1773.
- 38 M. Arslan, *Eur. Polym. J.*, 2020, **126**, 109543.
- 39 D. P. Nair, M. Podgórski, S. Chatani, T. Gong, W. Xi, C. R. Fenoli and C. N. Bowman, *Chem. Mater.*, 2014, **26**, 724–744.
- 40 C. E. Hoyle and C. N. Bowman, *Angew. Chem., Int. Ed.*, 2010, **49**, 1540–1573.
- 41 C. Resetco, B. Hendriks, N. Badi and F. Du Prez, *Mater. Horiz.*, 2017, **4**, 1041–1053.
- 42 A. D. Baldwin and K. L. Kiick, *Bioconjugate Chem.*, 2011, **22**, 1946–1953.
- 43 W. W. Cleland, *Biochemistry*, 1964, **3**, 480–482.
- 44 G. L. Ellman, *Arch. Biochem. Biophys.*, 1959, **82**, 70–77.
- 45 R. Singh and G. M. Whitesides, *J. Am. Chem. Soc.*, 1990, **112**, 1190–1197.
- 46 R. Singh and G. M. Whitesides, in *Sulphur-Containing Functional Groups (1993)*, John Wiley & Sons, Inc., Chichester, UK, 2010, pp. 633–658.
- 47 D. R. Grasseti and J. F. Murray, *Arch. Biochem. Biophys.*, 1967, **119**, 41–49.
- 48 N. J. Kavimandan, E. Losi, J. J. Wilson, J. S. Brodbelt and N. A. Peppas, *Bioconjugate Chem.*, 2006, **17**, 1376–1384.
- 49 R. K. Iha, K. L. Wooley, A. M. Nyström, D. J. Burke, M. J. Kade and C. J. Hawker, *Chem. Rev.*, 2009, **109**, 5620–5686.
- 50 J. D. Thomas and T. R. Burke, *Tetrahedron Lett.*, 2011, **52**, 4316–4319.
- 51 Y. Akiyama, Y. Nagasaki and K. Kataoka, *Bioconjugate Chem.*, 2004, **15**, 424–427.
- 52 T. Ishii, M. Yamada, T. Hirase and Y. Nagasaki, *Polym. J.*, 2005, **37**, 221–228.
- 53 R. Mahou and C. Wandrey, *Polymers*, 2012, **4**, 561–589.
- 54 C. Boyer, J. Liu, V. Bulmus and T. P. Davis, *Aust. J. Chem.*, 2009, **62**, 830.
- 55 J. Xu, C. Boyer, V. Bulmus and T. P. Davis, *J. Polym. Sci., Part A: Polym. Chem.*, 2009, **47**, 4302–4313.
- 56 C. Boyer, V. Bulmus and T. P. Davis, *Macromol. Rapid Commun.*, 2009, **30**, 493–497.
- 57 G. H. Hsiue, H. Z. Chiang, C. H. Wang and T. M. Juang, *Bioconjugate Chem.*, 2006, **17**, 781–786.
- 58 N. M. Matsumoto, P. Prabhakaran, L. H. Rome and H. D. Maynard, *ACS Nano*, 2013, **7**, 867–874.
- 59 E. Setijadi, L. Tao, J. Liu, Z. Jia, C. Boyer and T. P. Davis, *Biomacromolecules*, 2009, **10**, 2699–2707.
- 60 T. Fuoco, A. Finne-Wistrand and D. Pappalardo, *Biomacromolecules*, 2016, **17**, 1383–1394.
- 61 H. Tan, L. Zhao, W. Liu, L. Ren, S. Xu, L. Chen and W. Li, *RSC Adv.*, 2014, **4**, 60413–60420.
- 62 M. Talelli and M. J. Vicent, *Biomacromolecules*, 2014, **15**, 4168–4177.
- 63 J. Li, C. Tian, Y. Yuan, Z. Yang, C. Yin, R. Jiang, W. Song, X. Li, X. Lu, L. Zhang, Q. Fan and W. Huang, *Macromolecules*, 2015, **48**, 1017–1025.
- 64 H. S. Han, T. Thambi, K. Y. Choi, S. Son, H. Ko, M. C. Lee, D. G. Jo, Y. S. Chae, Y. M. Kang, J. Y. Lee and J. H. Park, *Biomacromolecules*, 2015, **16**, 447–456.
- 65 M. Andersson, K. Elihn, K. Fromell and K. D. Caldwell, *Colloids Surf., B*, 2004, **34**, 165–171.
- 66 J. T. Li, J. Carlsson, J. N. Lin and K. D. Caldwell, *Bioconjugate Chem.*, 1996, **7**, 592–599.
- 67 Y. Ju, M. Zhang and H. Zhao, *Polym. Chem.*, 2017, **8**, 5415–5426.
- 68 K. L. Heredia, D. Bontempo, T. Ly, J. T. Byers, S. Halstenberg and H. D. Maynard, *J. Am. Chem. Soc.*, 2005, **127**, 16955–16960.
- 69 D. Bontempo, K. L. Heredia, B. A. Fish and H. D. Maynard, *J. Am. Chem. Soc.*, 2004, **126**, 15372–15373.
- 70 V. Vázquez-Dorbatt, Z. P. Tolstyka, C. W. Chang and H. D. Maynard, *Biomacromolecules*, 2009, **10**, 2207–2212.
- 71 A. P. Daniselson, M. L. Dougherty, R. Falatach, T. A. Wright, E. E. Clark, A. Craig, I. D. Sahu, J. A. Berberich, R. C. Page, G. A. Lorigan and D. Konkolewicz, Biocatalytic Polymerization, Bioinspired Surfactants, and Bioconjugates Using RAFT Polymerization, in *ACS Symposium Series*, American Chemical Society, 2018, vol. 1285, pp. 219–232.
- 72 K. L. Heredia, T. H. Nguyen, C. W. Chang, V. Bulmus, T. P. Davis and H. D. Maynard, *Chem. Commun.*, 2008, **28**, 3245–3247.
- 73 C. Boyer, V. Bulmus, T. P. Davis, V. Ladmiral, J. Liu and S. Perrier, *Chem. Rev.*, 2009, **109**, 5402–5436.
- 74 C. Boyer, J. Liu, L. Wong, M. Tippet, V. Bulmus and T. P. Davis, *J. Polym. Sci., Part A: Polym. Chem.*, 2008, **46**, 7207–7224.
- 75 J. Liu, H. Liu, V. Bulmus, L. E. I. Tao, C. Boyer and T. P. Davis, *J. Polym. Sci., Part A: Polym. Chem.*, 2010, **48**, 1399–1405.
- 76 J. Liu, V. Bulmus, C. Barner-Kowollik, M. H. Stenzel and T. P. Davis, *Macromol. Rapid Commun.*, 2007, **28**, 305–314.
- 77 K. Gunasekaran, T. H. Nguyen, H. D. Maynard, T. P. Davis and V. Bulmus, *Macromol. Rapid Commun.*, 2011, **32**, 654–659.
- 78 C. L. Duvall, A. J. Convertine, D. S. W. Benoit, A. S. Hoffman and P. S. Stayton, *Mol. Pharm.*, 2010, **7**, 468–476.
- 79 C. W. Chang, T. H. Nguyen and H. D. Maynard, *Macromol. Rapid Commun.*, 2010, **31**, 1691–1695.
- 80 Y. Koda, T. Terashima, M. Sawamoto and H. D. Maynard, *Polym. Chem.*, 2015, **6**, 240–247.
- 81 T. A. Wright, R. C. Page and D. Konkolewicz, *Polym. Chem.*, 2019, **10**, 434–454.

- 82 I. Mukherjee, S. K. Sinha, S. Datta and P. De, *Biomacromolecules*, 2018, **19**, 2286–2293.
- 83 R. J. Mancini, J. Lee and H. D. Maynard, *J. Am. Chem. Soc.*, 2012, **134**, 8474–8479.
- 84 N. Vanparijs, S. Maji, B. Louage, L. Voorhaar, D. Laplace, Q. Zhang, Y. Shi, W. E. Hennink, R. Hoogenboom and B. G. De Geest, *Polym. Chem.*, 2015, **6**, 5602–5614.
- 85 N. M. Matsumoto, G. W. Buchman, L. H. Rome and H. D. Maynard, *Eur. Polym. J.*, 2015, **69**, 532–539.
- 86 C. Boyer, J. Liu, V. Bulmus, T. P. Davis, C. Barner-Kowollik and M. H. Stenzel, *Macromolecules*, 2008, **41**, 5641–5650.
- 87 M. R. Molla and S. Ghosh, *Macromolecules*, 2012, **45**, 8561–8570.
- 88 L. Tao, J. Liu and T. P. Davis, *Biomacromolecules*, 2009, **10**, 2847–2851.
- 89 O. Gok, P. Erturk, B. Sumer Bolu, T. N. Gevrek, R. Sanyal and A. Sanyal, *Biomacromolecules*, 2017, **18**, 2463–2477.
- 90 J. J. Glass, Y. Li, R. De Rose, A. P. R. Johnston, E. I. Czuba, S. Y. Khor, J. F. Quinn, M. R. Whittaker, T. P. Davis and S. J. Kent, *ACS Appl. Mater. Interfaces*, 2017, **9**, 12182–12194.
- 91 A. Klaikherd, S. Ghosh and S. Thayumanavan, *Macromolecules*, 2007, **40**, 8518–8520.
- 92 A. Klaikherd, C. Nagamani and S. Thayumanavan, *J. Am. Chem. Soc.*, 2009, **131**, 4830–4838.
- 93 M. J. Summers, D. J. Phillips and M. I. Gibson, *Chem. Commun.*, 2013, **49**, 4223–4225.
- 94 D. J. Phillips and M. I. Gibson, *Biomacromolecules*, 2012, **13**, 3200–3208.
- 95 D. J. Phillips and M. I. Gibson, *Chem. Commun.*, 2012, **48**, 1054–1056.
- 96 C. Boyer, V. Bulmus, J. Liu, T. P. Davis, M. H. Stenzel and C. Barner-Kowollik, *J. Am. Chem. Soc.*, 2007, **129**, 7145–7154.
- 97 E. W. Lin and H. D. Maynard, *Macromolecules*, 2015, **48**, 5640–5647.
- 98 B. Remant, B. Thapa and P. Xu, *Mol. Pharm.*, 2012, **9**, 2719–2729.
- 99 B. Lou, S. De Koker, C. Y. J. Lau, W. E. Hennink and E. Mastrobattista, *Bioconjugate Chem.*, 2019, **30**, 461–475.
- 100 M. E. H. El-Sayed, A. S. Hoffman and P. S. Stayton, *J. Controlled Release*, 2005, **101**, 47–58.
- 101 S. Ghosh, S. Basu and S. Thayumanavan, *Macromolecules*, 2006, **39**, 5595–5597.
- 102 L. Wong, C. Boyer, Z. Jia, H. M. Zareie, T. P. Davis and V. Bulmus, *Biomacromolecules*, 2008, **9**, 1934–1944.
- 103 Z. Jia, L. Wong, T. P. Davis and V. Bulmus, *Biomacromolecules*, 2008, **9**, 3106–3113.
- 104 Z. Song, Y. Xu, W. Yang, L. Cui, J. Zhang and J. Liu, *Eur. Polym. J.*, 2015, **69**, 559–572.
- 105 S. Keller, J. T. Wilson, G. I. Patilea, H. B. Kern, A. J. Convertine and P. S. Stayton, *J. Controlled Release*, 2014, **191**, 24–33.
- 106 W. Chen, Y. Zou, J. Jia, F. Meng, R. Cheng, C. Deng, J. Feijen and Z. Zhong, *Macromolecules*, 2013, **46**, 699–707.
- 107 G. T. Zugates, D. G. Anderson, S. R. Little, I. E. B. Lawhorn and R. Langer, *J. Am. Chem. Soc.*, 2006, **128**, 12726–12734.
- 108 D. Basak, R. Kumar and S. Ghosh, *Macromol. Rapid Commun.*, 2014, **35**, 1340–1344.
- 109 R. Bej, J. Sarkar and S. Ghosh, *J. Polym. Sci., Part A: Polym. Chem.*, 2018, **56**, 194–202.
- 110 D. Basak, R. Bej and S. Ghosh, *Polym. Chem.*, 2015, **6**, 6465–6474.
- 111 R. Bej, P. Rajdev, R. Barman and S. Ghosh, *Polym. Chem.*, 2020, **11**, 990–1000.
- 112 X. Liu, J. He, D. Hu, Y. Niu, X. Xiaa and Y. Lu, *RSC Adv.*, 2014, **4**, 48897–48900.
- 113 J. Zhuang, B. Zhao and S. Thayumanavan, *ACS Macro Lett.*, 2019, **8**, 245–249.
- 114 A. J. Van Der Vlies, C. P. Oneil, U. Hasegawa, N. Hammond and J. A. Hubbell, *Bioconjugate Chem.*, 2010, **21**, 653–662.
- 115 A. W. Jackson and D. A. Fulton, *Macromolecules*, 2012, **45**, 2699–2708.
- 116 S. Zhu, Z. W. Li and H. Zhao, *Langmuir*, 2015, **31**, 4129–4136.
- 117 X. Yang, D. Chen and H. Zhao, *Acta Biomater.*, 2016, **29**, 446–454.
- 118 Y. Z. You, K. K. Kalebaila, S. L. Brock and D. Oupicky, *Chem. Mater.*, 2008, **20**, 3354–3359.
- 119 Z. Jiang and S. Thayumanavan, *Isr. J. Chem.*, 2020, **60**, 132–139.
- 120 D. C. González-Toro and S. Thayumanavan, *Eur. Polym. J.*, 2013, **49**, 2906–2918.
- 121 J. H. Ryu, S. Jiwanich, R. Chacko, S. Bickerton and S. Thayumanavan, *J. Am. Chem. Soc.*, 2010, **132**, 8246–8247.
- 122 D. C. González-Toro, J. H. Ryu, R. T. Chacko, J. Zhuang and S. Thayumanavan, *J. Am. Chem. Soc.*, 2012, **134**, 6964–6967.
- 123 J. H. Ryu, S. Bickerton, J. Zhuang and S. Thayumanavan, *Biomacromolecules*, 2012, **13**, 1515–1522.
- 124 J. H. Ryu, R. T. Chacko, S. Jiwanich, S. Bickerton, R. P. Babu and S. Thayumanavan, *J. Am. Chem. Soc.*, 2010, **132**, 17227–17235.
- 125 N. M. Matsumoto, D. C. González-Toro, R. T. Chacko, H. D. Maynard and S. Thayumanavan, *Polym. Chem.*, 2013, **4**, 2464–2469.
- 126 B. Liu and S. Thayumanavan, *Chem*, 2019, **5**, 3166–3183.
- 127 X. Ji, J. Liu, L. Liu and H. Zhao, *Colloids Surf., B*, 2016, **148**, 41–48.
- 128 N. Boehnke, J. K. Kammeyer, R. Damoiseaux and H. D. Maynard, *Adv. Funct. Mater.*, 2018, **28**, 1–11.
- 129 Q. Yu, X. Ma, Y. Liu and H. Zhao, *Chem. – Eur. J.*, 2019, **25**, 16712–16717.
- 130 N. Cengiz, *Eur. Polym. J.*, 2020, **123**, 109441.
- 131 I. Altinbasak, R. Sanyal and A. Sanyal, *RSC Adv.*, 2016, **6**, 74757–74764.
- 132 S. Y. Choh, D. Cross and C. Wang, *Biomacromolecules*, 2011, **12**, 1126–1136.

- 133 Y. Zhang, P. Heher, J. Hilborn, H. Redl and D. A. Ossipov, *Acta Biomater.*, 2016, **38**, 23–32.
- 134 M. M. Perera and N. Ayres, *Polym. Chem.*, 2017, **8**, 6741–6749.
- 135 H. W. Chien, X. Xu, J. R. Ella-Menye, W. B. Tsai and S. Jiang, *Langmuir*, 2012, **28**, 17778–17784.
- 136 R. Iwata, R. Satoh, Y. Iwasaki and K. Akiyoshi, *Colloids Surf., B*, 2008, **62**, 288–298.
- 137 T. N. Gevrek, M. Cosar, D. Aydin, E. Kaga, M. Arslan, R. Sanyal and A. Sanyal, *ACS Appl. Mater. Interfaces*, 2018, **10**, 14399–14409.
- 138 H. Peng, K. Rübsam, X. Huang, F. Jakob, M. Karperien, U. Schwaneberg and A. Pich, *Macromolecules*, 2016, **49**, 7141–7154.
- 139 A. M. Akimoto, T. Takarada and M. Maeda, *Colloids Surf., B*, 2013, **103**, 360–365.