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Pd-catalyzed dearomative arylborylation of indoles*

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A palladium-catalyzed dearomative arylborylation of indoles is reported, which provides straightforward access to structurally diverse indolines bearing vicinal tetrasubstituted and borylated trisubstituted stereocenters in moderate to good yields with excellent diastereoselectivities. By using a BINOL-based chiral phosphoramidite ligand and an $sp^2 - sp^3$ mixed-boron reagent, an enantioselective dearomative arylborylation was achieved and chiral boron-containing products were accessed in up to 94% ee. Synthetic tranformations of the resulting organoborons were conducted to afford a number of unique indoline derivatives

reductive-Heck reaction,9 which results in the dearomative

mono-functionalization of indole derivatives. Furthermore, the dearomative difunctionalization of indoles was realized

through the trapping of the benzyl-Pd intermediate of arylpalladation using a series of trapping agents, such as

cyanides,10 boroxines,11 terminal alkynes,12 propiolic acid,13 and

heteroarenes,14 efficiently delivering 2,3-disubstituted indo-

lines. While the formation of the C-H and C-C bonds is docu-

mented in the aforementioned reports, there are no examples

reported for C-B bond formation to afford chiral organoboron

compounds. We envisioned a dearomative Heck-borylation

domino reaction of indole to provide benzylic-boron indo-

lines; the protocol mainly relies on the capture of the in situ

generated benzyl-Pd species with diboron compounds.

Herein, we report this dearomative arylborylation reaction

using bis(pinacolato)diboron (B₂pin₂), and its enantioselective

variant with a pre-activated sp²-sp³ mixed-boron reagent and

Boron containing molecules serve as important building blocks due to their capability of participating in carbon-carbon and carbon-heteroatom bond forming reactions and thus are ubiquitous intermediates in the synthesis of various natural products and bioactive compounds.1 The palladium-catalyzed 1,2-difunctionalization of olefins involving Heck/anioniccapture sequences has been intensely studied and proceeds *via* a carbopalladation followed by the capture of the alkyl-Pd species with a variety of nucleophiles.2 Domino Heckborylation reactions employing boron reagents as nucleophiles have enabled an efficient method to synthesize C_{sp³}based organoboron compounds (Scheme 1a).^{3,4} It is worthwhile to extend this methodology to the synthesis of structurally diverse, chiral organoboron heterocycles.

The transition-metal-catalyzed dearomative functionalization of aromatics has recently emerged as a useful approach to the synthesis of unique aliphatic cyclic molecules.⁵ In this context, dearomative functionalization of indoles has rendered the synthesis of indolines, a frequently occurring key substructure of natural products and alkaloids, extremely straightforward and efficient.6 Documented reports include the intramolecular C3-arylation of an indolic enolate,7 the Heck arylation of N-tethered 2,3-disubstituted indoles,8 and the

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C-C/C-B bonds formation good to excellent enantioselectivity

Scheme 1 Palladium-catalyzed arylborylation of C=C bonds.

to 94% ee

up to 92% yield



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The reaction of 1a with B_2pin_2 2 was chosen as the starting condition for optimization. An initial test using Pd(OAc)₂ (5 mol%), PPh₃ (10 mol%), and ^tBuOLi (2.0 equiv.) in CH₂Cl₂ (0.2 M) at 100 °C led to the desired arylborylation product 3a in 25% yield with >20 : 1 dr (Table 1, entry 1). To improve the yield, some bases were screened (entries 2-5). A poor yield was observed when using ^tBuOK (Table 1, entry 2), while K_3PO_4 , K_2CO_3 , and Na_2CO_3 significantly improve the yield with 3a isolated in 66% in the case of K₂CO₃ (Table 1, entries 3-5). Other commercially available ligands, such as P(p-tolyl)₃, P^tBu₃, and PCy₃, were then tested, none of which increased the yield (Table 1, entries 6-8). Moreover, poor yields were also observed for bidentate phosphine ligands, e.g. dppe and xantphos. A lower yield was observed when changing the catalyst from Pd(OAc)₂ to $Pd(dba)_2$ (Table 1, entry 9). Higher yields could be obtained by lowering the temperature and 3a was isolated in 83% yield when the reaction was run at 60 °C (Table 1, entries 10 and 11). Finally, the solvent effect was examined (Table 1, entries 12-15). Comparable yields were observed in toluene and CH₃CN, while the best yield of 3a was achieved in DCE solvent (Table 1, entry 15).

With the optimal conditions in hand, we then examined the scope of the reaction by varying the substituents on the halobenzene and indole rings. As shown in Scheme 2, substituent effect on the benzene ring of the 2-bromobenzoyl moiety was first investigated. Moderate to excellent yields of products **3a–3h** were achieved for indoles bearing substituents (methyl, methoxyl, and chloride) at the C3–C5 position of the benzene ring. Higher yields were generally obtained for substrates having an electron-donating group than for those bearing an



11 K₂CO₃ PPh_3 60 CH₂Cl₂ 83 12 PPh₃ 60 THF 68 K_2CO_3 13 K₂CO₃ PPh₃ 60 toluene 84 14 K₂CO₃ PPh₃ 60 MeCN 83 15 K₂CO₃ PPh₃ 60 DCE 87 ^a Reaction conditions: 1a (0.2 mmol), 2 (2 eq.), 5 mol% Pd(OAc)₂,

10 mol% ligand, and 2 eq. base in solvent (2 mL) at 100 °C; isolated yield, dr > 20 : 1; DCE = 1,2-dichloroethane. ^{*b*} 5 mol% Pd(dba)₂ was used.



electron-withdrawing substituent (**3e** and **3g** *vs.* **3b–3d** and **3f**). Product **3h** having two methoxyl groups was isolated in 92% yield. Of note, **3b**, having a sterically congested methyl group at the C3 position of the bromobenzoyl moiety, was obtained in 78% yield. Next, the substituent effect on the indole ring was examined. A range of C2-alkylated and C2-arylated indoles were subjected to the reaction at 70 °C, which led to the arylborylated products **3i–3n** in moderate yields. In contrast, the yields of these borylated indolines were lower than those achieved for 2methyl products **3a–3h**. It is noteworthy that the reaction of a 2furyl indole substrate successfully delivered **3n** in 53% yield. Moreover, the substituent at the C5-position of 2-substituted indoles was examined and the reactions of substrates having MeO, iPr, and Me groups afforded **3o–3q** in moderate yields.

To demonstrate the synthetic utility of this reaction, a gramscale reaction (4.0 mmol) was carried out under the optimal conditions and afforded **3a** in 77% yield (Scheme 3). Synthetic transformations of **3a** were then conducted. Oxidation of **3a** using NaBO₃·4H₂O in THF/H₂O led to alcohol **4** as a single isomer in almost quantitative yield.¹⁵ Compound **4** was further



Scheme 3 Gram-scale synthesis of 3a and transformations of 3a. PCC: pyridinium chlorochromate.

oxidized to ketone **5** in 85% yield with PCC as an oxidant at room temperature. Subjecting **3a** to KHF₂ in THF/H₂O led to the corresponding potassium trifluoroborate **6** in 91% yield. Compound **6** was further converted to amide **7** in MeCN with 64% yield, as a single isomer through a Cu(OAc)₂-promoted oxidative nucleophilic substitution.¹⁶ The relative configurations of alcohol **4** and amide **7** were determined from their 2D-NOESY spectra.

An enantioselective Heck/borylation reaction of **1a** was then investigated using phosphoramidite **L1** as the chiral ligand^{12b} (Table 2, for more details see the ESI[†]). Early on, we observed that the benzylic boron was susceptible to inorganic-base promoted proto-deborylation at the high temperatures necessary for this metal–ligand system (*vide infra*). In order to avoid the use of the inorganic base necessary to activate B₂Pin₂, we utilized an sp²–sp³ mixed boron reagent first reported by Santos for copper-catalyzed hydroboration reactions.¹⁷ To the best of our knowledge, this reagent has not been used in palladiumcatalyzed borylations. It was necessary to change the solvent from DCE to MTBE since the former was not efficient in the enantioselective variant (Table 2, entry 1 and 2). Although the bromo- and iodo- substrates **1a** and **1a**" delivered product in higher yields than the aryl-chloride, it was evident that the smaller halide improved the enantioselectivities (Table 2, entries 2–4). The absolute stereochemistry of **3a** was assigned by single-crystal X-ray analysis.¹⁸

In the case of aryl-chloride, there was no reaction using B_2Pin_2 (Table 2, entry 5). We employed an organic base to neutralize the by-product of the boron-reagent and the conversion was improved to 50% while maintaining the ee (Table 2, entry 6). Further increasing the steric bulk of the ligand improved the yield of the reaction (Table 2, entry 7). Other amines (Table 2, entry 8) or lowering the temperature (Table 2, entry 9) were not effective. By increasing ligand and reagent loading the product was delivered in 74% yield and 94% ee (Table 2, entry 10 and 11). With respect to the ligand, the nitro variant L3 was not an effective ligand for the transformation (Table 2, entry 12). The 3,3'-orthoanisole ligand L4 also did not improve the yield (Table 2, entry 13). The importance of the 3.3'rings was evident as the simple BINOL-derived phosphoramidite L5 did not produce the product (Table 2, entry 14). Contrary to the dearomative reductive-Heck,9a BINAP was not effective in catalyzing the reaction (Table 2, entry 15).

Table 2	Optimization of the ena	antioselecitive arylborylat	tion ^a		
$\begin{array}{c} & & & \\ & & & & \\ & & & &$					
	C C C C C C C C C C C C C C C C C C C	Cy L2 CME	$\begin{array}{c} & & & & & \\ & & & & \\ &$	$ \begin{array}{ccc} & & & & \\ & & & & \\ & & & & \\ & & & &$	
Entry	Х	Additive	Changes to condition	Yield (%)	ee (%)
1	Br	None	None	73	64
2	Br	None	DCE as solvent	34	20
3	Ι	None	None	40	50
4	Cl	None	None	15	88
5	Cl	K_2CO_3 (2 eq.)	$B_2Pin_2 \cdot (2 \text{ eq.})$	n.r.	_
6	Cl	NEt_3 (3 eq.)	None	50	88
7	Cl	NEt_3 (3 eq.)	using L2	65	88
Entries	8–11 using L2				
8	Cl	^{<i>i</i>} Pr ₂ NEt (3 eq.)	None	27	—
9	Cl	NEt_3 (3 eq.)	80 °C	n.r.	—
10	Cl	NEt_3 (3 eq.)	10 mol% L2	68	91
11	Cl	NEt_3 (5 eq.)	3 eq. mixed-boron reagent	74	94
Entries	12-15 using conditions in	n entry 11			
12	Cl	NEt_3 (5 eq.)	L3	17	78
13	Cl	NEt_3 (5 eq.)	L4	70	91
14	Cl	NEt_3 (5 eq.)	L5	Trace	_
15	Cl	NEt_3 (5 eq.)	L6	Trace	—

^{*a*} Standard conditions: **1** (0.2 mmol), mixed-boron reagent (2 eq.), 5 mol% Pd(dba)₂, 6 mol% L1, and additive in MTBE (2 mL) at 100 °C for 18 h; isolated yield; ee was determined by chiral HPLC; TMG = tetramethyl guanidine; n.r. = no reaction.





Scheme 4 Scope of Heck/borylation of indoles. Reaction conditions: 1a' (0.2 mmol), (mixed-boron reagent) (0.6 mmol), Pd(dba)₂ (5 mol%), L2 (10 mol%), NEt₃ (5 eq.), 100 °C, 18 h. [a] Reaction of 4a (0.5 mmol) with NaBO₃ (5 eq.), THF : H₂O (1 : 1), rt. [b] Reaction of 5 (0.2 mmol) with PCC (3 eq.), CHCl₃, 35 °C.

We then examined the scope of the Heck/borylation on various aryl-chloride substrates (Scheme 4). The steric influence at the halide (**3r**), as well as functionalities *para* to the amide tether (**3d** and **3s**) provided products in diminished enantioselectivities. In contrast, the substitution *para* to chloride (**3t**) or on the indole moiety yielded **3t-3v** in moderate yields and excellent enantioselectivities. The aryl functionality at R^2 resulted in **3j** and **3w** in moderate and good yields with excellent enantioselectivities. Finally, heterocycle containing scaffold **3x** was accessed in good yield albeit in a diminished enantioselectivity. Oxidation of compound **3a** provided the chiral alcohol (+)-**4** and ketone (+)-**5** with no loss in enantiomeric excess.

Conclusions

In conclusion, we have developed a dearomative difunctionalization of indoles through a palladium-catalyzed intramolecular arylborylation reaction. Indolines possessing vicinal borylated trisubstituted and tetrasubstituted stereocenters were accessed in good yields and excellent diastereoselectivities. The asymmetric variant of this reaction was explored with a new BINOLbased chiral phosphoramidite ligand and moderate to excellent enantioselectivities were obtained (up to 94% ee). Transformations of the benzylic boron to alcohol and amide were presented to show the synthetic utilities of this reaction.

Conflicts of interest

There are no conflicts to declare.

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