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Silver-catalyzed tandem cycloisomerization/hydroarylation reactions and mechanistic investigations for an efficient access to 1,2-dihydroisoquinolines†‡

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An efficient silver-catalyzed tandem reaction for the formation of 1,2-dihydroisoquinoline derivatives is herein reported. Highly functionalized multiheterocyclic scaffolds are accessible in a straightforward manner using readily accessible starting materials under mild conditions. This methodology offers an attractive route for the synthesis and development of a biologically relevant new heterocyclic pharmacophore, merging the biological activities of isoquinolines with those of various nitrogen-containing heterocycles (indoles, pyrroles) incorporated during the tandem reaction. Mechanistic investigations were also conducted along with a large scope and limitation study, modifying various sites of this pharmacophore.

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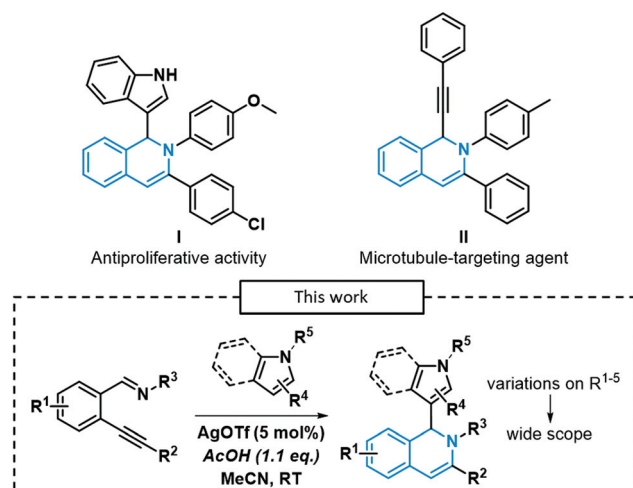
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Introduction

One of the most interesting subclasses of nitrogen-containing heterocycles is certainly the isoquinoline core and its derivatives due to its prevalence in many biologically active compounds, including marketed drugs. Indeed, isoquinolines have various therapeutic applications such as antipruritic (quinisocaine)¹ and antispasmodic (papaverine).² Recently, some 1,2-dihydroisoquinoline derivatives exhibited an interesting activity towards cancer cell lines (**I** and **II**, Scheme 1).^{3,4} Traditionally, isoquinolines were synthesized using the Bischler–Napieralski,⁵ the Pomeranz–Fritsch^{6,7} and the Pictet–Spengler⁸ processes, developed in the late 19th/early 20th century. However, harsh conditions were often required to perform such reactions (high operating temperatures, use of highly toxic and corrosive PCl₅ or POCl₃). Thereby, many new synthetic methodologies were developed in the last decades, relying on transition-metal catalysis. Larock's team reported the first example of a Pd-catalyzed synthesis of isoquinolines,⁹ followed by other examples derived from their methodology.^{10–12} Wu's team pioneered the develop-

ment of numerous examples of 1-substituted isoquinolines *via* Ag-catalyzed 6-*endo*-dig processes using silver as the catalyst, in the presence of many pro-nucleophiles, such as α,β -unsaturated ketones,¹³ phosphonates,¹⁴ imidazoles,¹⁵ ketones,¹⁶ indoles¹⁷ and even trifluoromethyl groups.¹⁸ These methods allow rapid access to highly functionalized scaffolds in very few steps. Moreover, the addition of heterocyclic nitrogen-containing pro-nucleophiles (indoles/pyrroles) at position 1 of isoquinolines



Scheme 1 Our approach for the synthesis of 1,2-dihydroisoquinoline derivatives.

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† This paper is dedicated to professor Jean Lhomme.

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gives rise to new potential pharmacophores since indole¹⁹ or pyrrole²⁰ derivatives also exhibit key biological activities.

Tandem reactions affording these complex scaffolds have been reported in the literature using several modes of activation and a myriad of catalysts, Lewis acids, and electrophiles.^{3,21} More recently, Han and Lu's team reported a Pd-catalyzed reaction for the synthesis of 1,2-dihydroisoquinolines with the concomitant insertion at position 1 of 3-indoles.²²

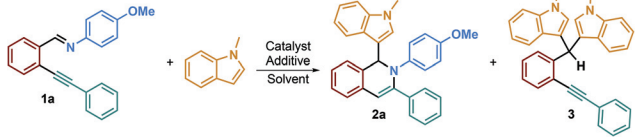
Such scaffolds can also be obtained using cobalt catalysts, as reported by Tandon *et al.*²³ There is only one report with few examples described in the presence of a silver catalyst.¹⁷ Following our interest in Ag-catalyzed cycloisomerization reactions to form furoquinoline/pyranoquinoline,^{24,25} isochromene²⁶ or pyranoquinoline²⁷ derivatives as well as isobenzofurane or isoindoline derivatives,²⁸ we envisioned to establish efficient reaction conditions for the formation of a wider variety of isoquinoline derivatives compared to previous literature reports.¹⁷ We propose herein to study the cyclization of *ortho*-alkynylarylaldimine derivatives using Ag catalysts (Scheme 1), with the concomitant addition of several types of heterocyclic nucleophiles in order to access highly functionalized 1,2-dihydroisoquinolines in a 3-step pathway.

Results and discussion

To test our hypothesis, we began with the synthesis of the starting materials (Scheme 2) using a Sonogashira cross-coupling reaction performed on 2-bromoarylaldehyde derivatives reacting with terminal alkynes.²⁶ The resulting desired *ortho*-alkynylarylaldehydes were then combined with aniline derivatives to form the imino starting materials (Scheme 2). A library of *ortho*-alkynyl(hetero)arylaldimines was obtained bearing various R¹, R² and R³ substituents.

These starting materials were then used for the optimization of the reaction conditions. Based on the conditions previously reported by our team,^{26,27} we investigated a plausible 6-*endo-dig*²⁹ ring-closing reaction on model substrate **1a** (entries 1 and 2, Table 1). Unfortunately, these conditions were not efficient on that kind of starting material and led only to the slow hydrolysis of the imino group. Since imine hydrolysis can be promoted by traces of water, we added 3 Å M.S. in the reaction medium in order to avoid such side reactions; regrettably, no improvement was observed even after two days of reaction (entry 3). A report from Zhang, Wu *et al.*³⁰ described the formation of 1-(trifluoromethyl)-1,2-dihydroisoquinolines by incorporating a trifluoromethyl group on 2-alkynylarylaldimine derivatives in the presence of acetic acid. They con-

Table 1 Optimization of the reaction conditions

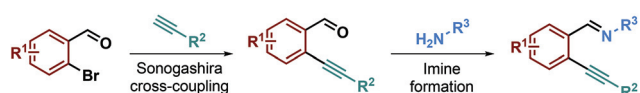


Entry	Catalyst (mol%)	Solvent	Additive	T (°C)	Time (h)	Yield ^a 2a/3
1	AgOTf (10)	DMF ^c	—	80	16	—
2	AgOTf (10)	DCE ^d	—	80	16	—
3	AgOTf (10)	DCE ^d	3 Å M.S.	80	48	—
4	AgOTf (5)	CH ₃ CN ^e	AcOH ^f	rt	16	79/0 ^b
5	AgOTf (5)	CH ₃ CN ^e	—	rt	16	—
6	—	CH ₃ CN ^e	AcOH ^f	rt	16	—
7	AgSbF ₆ (5)	CH ₃ CN ^e	AcOH ^f	rt	16	53/0
8	AgOTf (5)	DCE ^e	AcOH ^f	rt	16	78/0
9	AgOTf (5)	DMF ^e	AcOH ^f	rt	16	Traces
10	AgOTf (5)	CH ₃ CN ^e	AcOH ^f	rt	48	57/0
11	AgOTf (5)	CH ₃ CN ^e	PTSA ^f	rt	16	0/43
12	AgOTf (5)	CH ₃ CN ^e	CSA ^f	rt	16	0/46
13	AgOTf (5)	CH ₃ CN ^e	TFA ^f	rt	16	0/56
14	AgOTf (1 eq.)	CH ₃ CN ^e	—	rt	16	—

^a Isolated yields. ^b Standard condition: Reactions were performed with **1a** (0.16 mmol), AgOTf (5 mol%), *N*-methylindole (5 eq.) and acetic acid (1.1 eq.) in 2 mL of solvent and the mixture was agitated at room temperature for 16 h. ^c 0.5 M. ^d 0.2 M. ^e 0.08 M. ^f 1.1 eq.

cluded that acetic acid, as an additive, was necessary to reach high product yields since it acts as a Brønsted acid for the activation of the imino group. Therefore, inspired by their results, we decided to add 1.1 eq. of acetic acid to the reaction medium using MeCN as the solvent and were pleased to obtain the desired product in a good 79% yield (entry 4). A control experiment, without acetic acid (entry 5), confirmed that the reaction could not proceed without this additive and also that switching the solvent from DCE to CH₃CN has no impact on the reaction course. In the same way, no product formation was observed without the silver catalyst, confirming the crucial role of both acetic acid and AgOTf in the formation of the desired product (entries 5 and 6). Several silver catalysts have been tested such as AgSbF₆ but no better yields were obtained (**2a** 53%, entry 7). Solvent change, at the same concentration (0.08 M), to DCE led to the formation of the desired product in a good 78% yield (entry 8), while switching to DMF afforded only traces of isoquinoline **2a** (entry 9). We also tried to perform the reaction with a catalytic amount of acetic acid since it should be regenerated during the reaction (see the mechanistic proposal). However, after using 0.1 eq. of acetic acid, we were surprised to obtain a lower 57% yield of **2a** after 2.5 days (entry 10), indicating that acetic acid present in slight excess allows faster formation of the desired isoquinoline, preventing the side product formation or hydrolysis reaction.

We then investigated the influence of the nature of the acid additive on the reaction by using other common acids. *para*-Toluenesulfonic acid (PTSA), camphorsulfonic acid (CSA) and trifluoroacetic acid (TFA) did not allow the formation of the



Scheme 2 Two-step preparation of the starting materials.



desired product; however, when using these acids, we were able to isolate product **3**, resulting in double addition of the nucleophile onto the starting material, in yields of 43%, 46% and 56%, respectively (entries 11, 12 and 13).³¹ The use of stronger acids than acetic acid will be discussed later on, but we may already propose that the iminyl group will be protonated, thereby promoting its hydrolysis and consequently the formation of side product **3** instead of the intramolecular addition of the free iminyl group on the alkynyl bond. Finally, we ran the reaction using 1 eq. of AgOTf in order to study whether the reaction could proceed, in this case, without acetic acid (entry 14). Unfortunately, no product was formed, indicating that both the silver catalyst and acetic acid additive were required to obtain the desired product **2a**. Thus, we decided that our reaction conditions are as follows: AgOTf (5 mol%), imino derivative (0.16 mmol), nucleophile (5 eq.), acetic acid (1.1 eq.) in MeCN (0.08 M) at rt for 16 h.

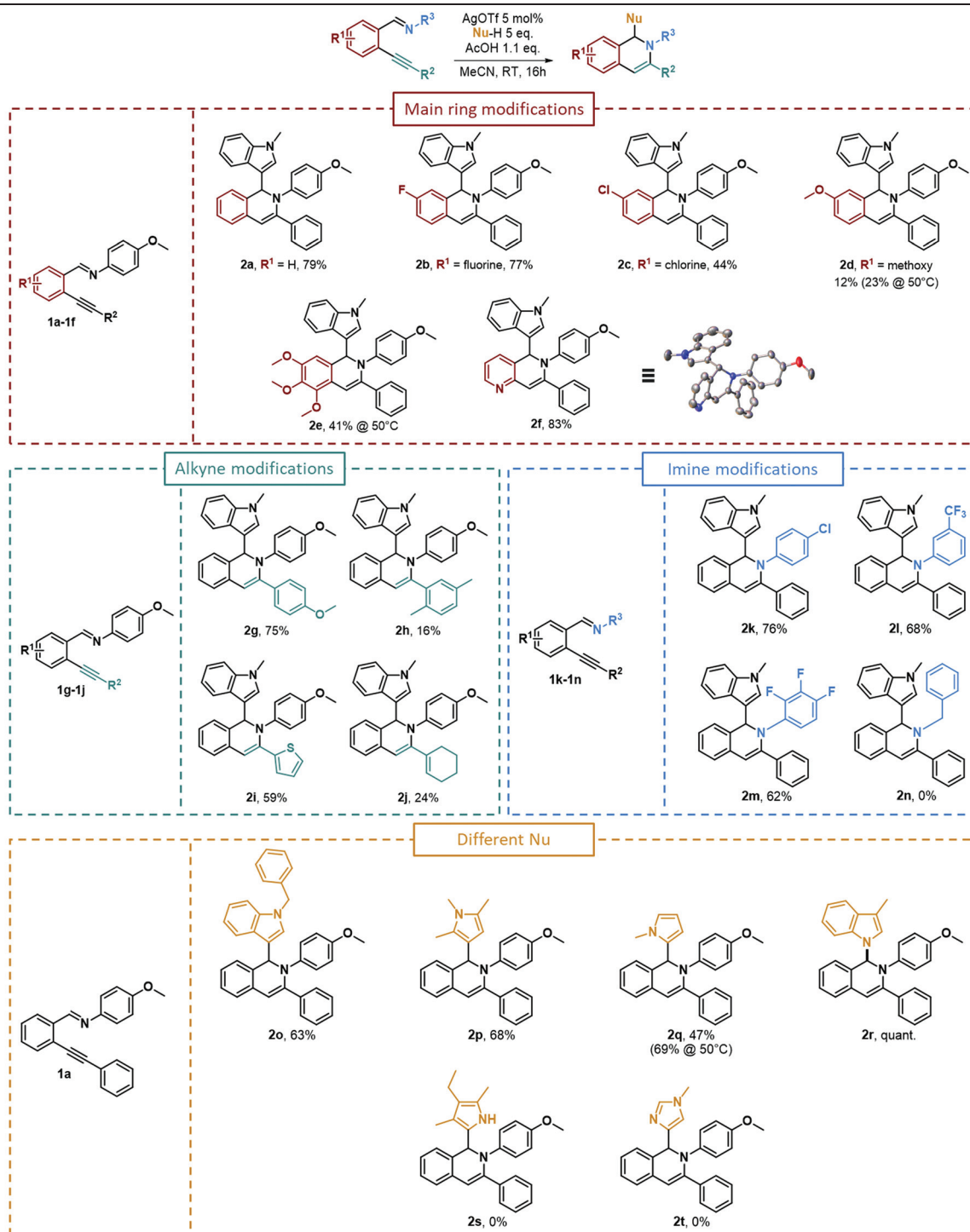
Using these optimized reaction conditions, we then investigated the functional group compatibility of the reaction, modifying the substituents present on various parts of imino derivatives **1**. As presented in Table 2, the tandem cycloisomerization/hydroarylation reaction of the model starting material **1a** led to the formation of 79% of the corresponding isoquinoline (**2a**). We then modified the substituents on the main ring, on the alkynyl part, on the imine substituent and finally the nucleophile. A fluorine substituent on the main ring (R^1) was not detrimental to the reaction, as the product was isolated in a good 77% yield (**2b**). However, when another halogen substituent such as chlorine (**2c**) was present, the isolated yield decreases to 44%. The yield dropped much further when a methoxy substituent was present on the same carbon (**2d**), leading to a low 12% isolated yield that could be slightly increased to 23% upon heating the reaction up to 50 °C. A trimethoxy substituent afforded the desired isoquinoline (**2e**) in a higher 41% yield upon heating at 50 °C. It is important to note that the reaction was compatible with the presence of a heterocycle on the main ring such as a pyridine (**2f**), leading to a high 83% yield. It appeared that the electron-donating groups on the main ring were detrimental to the reaction, while the electron-withdrawing groups led to better isolated yields. As for the alkyne modifications (R^2 , Table 2), 4-methoxyphenyl substitution on the alkyne of the starting material was well tolerated and led to a good 75% yield (**2g**), whereas a 2,5-dimethylphenyl group led to a low 16% yield (**2h**). This result could be explained by the steric hindrance on the *ortho* position of the alkyne that can interfere with the attack of the imino group on the alkynyl unit. A heterocycle, 2-thiophenyl, substituent led to a fair 59% yield (**2i**). Finally, an alkenyl-substituted alkyne (1-cyclohexenyl substitution) could also react and afforded the desired isoquinoline (**2j**) but in a low 24% yield. We then changed the nature of the substituents on the aniline (R^3 , Table 2), leading to the formation of the imine starting materials. The reaction could proceed normally when electron-withdrawing substituents were present on the aryl-imine unit such as 4-chlorophenyl (**2k**), 3-trifluoromethylphenyl (**2l**) and 2,3,4-trifluorophenyl (**2m**) substituents, which

all led to satisfactory yields of 76%, 68% and 62%, respectively. However, an aliphatic imine formed using benzylamine could not afford the corresponding product **2n**. Finally, we investigated the reactivity of various nucleophiles towards the starting material **1a** and their ability to perform a hydroarylation reaction. We used various nitrogen-containing heterocycles such as a *N*-methylindole and *N*-benzylindole that gave the desired products in good yields of 79% (**2a**) and 63% (**2o**), respectively. We also used differently substituted pyrroles such as 1,2,5-trimethylpyrrole that led to a good 68% yield (**2p**) (with an attack on position 3) and a *N*-methylpyrrole that afforded the desired product **2q** in a moderate 47% yield, increasing up to 69% upon heating at 50 °C (with an attack on position 2). The use of 3-methylindole led to the quantitative formation of the hydroamination product **2r**. We purposely used this indole derivative to prevent any attack on position 3 and instead to favor the attack on indole's position 2. But, we did not obtain any attack from position 2, which may be because of the steric hindrance of the methyl substitution at position 3 and the only attack observed was from the nitrogen atom. Other nucleophiles such as 3-ethyl-2,4-dimethylpyrrole or 1-methylimidazole could not react, and no traces of the corresponding products **2s** and **2t** were detected.

It is important to note that imines formed using acetophenone or benzophenone derivatives instead of aldehyde derivatives could not lead to the formation of the expected dihydroisoquinoline products. Moreover, we employed numerous heterocyclic nucleophiles in our methodology, from nitrogen to oxygen and sulfur-containing ones. Unfortunately, we did not observe any attack from the latter two (2-methylfuran and 2-methylthiophene for example). This lack of reactivity could be explained by their nucleophilicity. Indeed, using the work of Mayr's team on the establishment of a reactivity scale for various molecules (both for their electrophilicity and nucleophilicity), we can clearly notice a strong difference in their nucleophilicity parameter (N),^{32–36} with, for instance, values of N for 2-methylfuran and 2-methylthiophene being respectively 3.61 and 1.35. For a comparison, the “weakest” nucleophile that led to a product in this methodology was *N*-methylindole with a N parameter of 5.75. Another point that should be noted is that the opposite reasoning is not necessarily applicable. Some tested nucleophiles having a high N value did not lead to the desired final compound, such as 3-ethyl-2,4-dimethylpyrrole or 1-methylimidazole, which present N values of 11.63 and 9.91, respectively. Such results might be due to the impossibility to perform a reaction with the carbon atom (C-nucleophile) due to steric hindrance around the nucleophilic atom or even the presence of an unprotected N–H bond. Moreover, during the optimization work (Table 1), we tested acid additives having pK_a values much lower than that of acetic acid (4.76). Indeed, PTSA has a pK_a value of –6.5, CSA has a pK_a value of 1.2 and TFA has a pK_a value of 0.52 (in aqueous medium).³⁷ On the other hand, for our iminyl starting material, the pK_a prediction is around a value of 2.57.³⁸ Therefore, the acetic acid additive is crucial for the protodemetalation step (Scheme 5), leading to dihydroquinoline for-



Table 2 Scope of the reaction

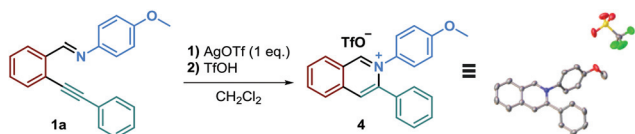


mation. In contrast, stronger acids may facilitate iminyl group hydrolysis of the aldimine starting materials, thereby leading to the parent carbonyl derivatives that may undergo the well-known double indole addition, leading to compound **3**.³⁹

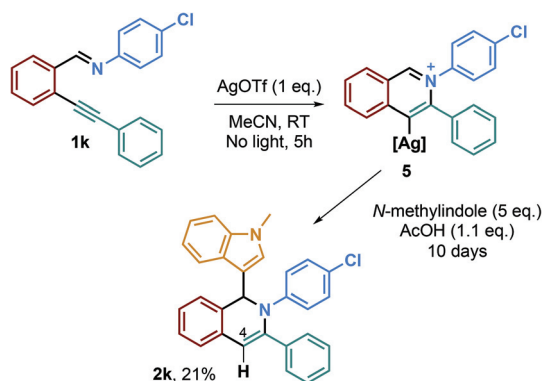
Finally, in order to gain more insight into the reaction mechanism, we performed some mechanistic experiments,

starting with deuteration experiments with *N*-methylindole (deutered at position 3) or AcOD. Indeed, we wondered if the hydrogen atom incorporated at position 4 of the isoquinoline scaffold (e.g. compound **2k**, Scheme 4) originated from the *N*-methylindole or from the acid additive. We never observed any deuteration after workup at position 4 of **2k**. We believe





Scheme 3 Isolation of the isoquinolinium salt and XRD analysis.



Scheme 4 Sequential formation and reaction of the hypothetical alkenyl silver intermediate 5.

that the absence of deuteration is due to the very reactive and labile H at position 4 (Scheme 4). An NMR experiment confirmed this hypothesis since the proton on position 4 instantly disappeared after the addition of an excess of AcOD, leading to the isolated deuterated isoquinoline **2k-D** (see ESI†).

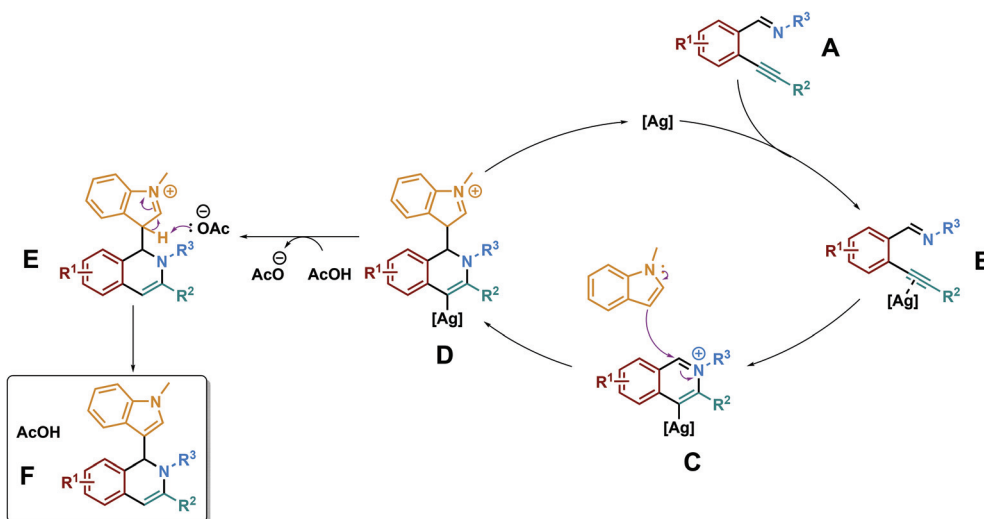
Moreover, in order to deepen our understanding of the reaction mechanism, we tried to isolate a reaction intermediate.⁴⁰ Precipitation and recrystallization of the isoquinolinium intermediate **4** were possible by running the reaction in dichloromethane with 1 eq. of AgOTf, without adding a nucleo-

phile, followed by the addition of triflic acid. Thanks to these conditions, we were able to obtain a single crystal of **4** suitable for an X-ray diffraction (Scheme 3).

We then used this isoquinolinium salt in the presence of *N*-methylindole, with and without AcOH, in acetonitrile. To our surprise, no product formation was observed. These results questioned our understanding of this reaction's mechanism since we were expecting a straightforward reaction between the nucleophile and the isoquinolinium ion. We therefore concluded that this isolated salt was probably not a reaction intermediate and the addition of the nucleophile might occur onto the isoquinolinium that still bears the alkenylsilver unit.

To test this hypothesis, we performed new experiments, in a one-pot two-step fashion (Scheme 4). The first step was the reaction of the imine derivative **1k** with 1 eq. of AgOTf in order to form the hypothetical isoquinolinium alkenylsilver specie **5**, and then, the second step was the addition of 1.1 eq. of AcOH to 5 eq. of *N*-methylindole. After 10 days of reaction, we were able to isolate the desired isoquinoline **2k** in a low 21% yield, which may indicate a plausible alkenylsilver intermediate.

Thanks to all these data, we may propose the following reaction mechanism (Scheme 5). The first step of the reaction is expected to be the coordination between the silver catalyst and the alkynyl group of the starting material **A**, leading to the activated alkynyl compound **B**. The imine group of **B** would attack intramolecularly the electron-deficient silver-activated alkynyl unit, leading to the formation of an alkenylsilver intermediate **C**. We believe that this intermediate would then be attacked by the external nucleophile (*N*-methylindole here), yielding the intermediate **D**. Then, after a protodesilveration step, intermediate **E** would be obtained and would yield the final desired product **F** upon rearomatization of the indole core.



Scheme 5 Proposed mechanism.



Conclusions

We herein reported an efficient tandem cycloisomerization/hydroarylation silver-catalyzed reaction leading to the formation of a broad range of substituted isoquinoline structures under mild conditions. Thanks to the synergistic association of AgOTf as a catalyst with acetic acid, the access to a wide library of heterocyclic derivatives, bearing various substitutions, was possible, thanks to modifications of the starting material's main ring, alkynyl and imino parts, as well as the choice of different types of nitrogen-containing heterocycles as nucleophiles. Biological evaluation of these new scaffolds and the study of the central chirality obtained at the end of the tandem reaction are both currently underway.

Experimental section

General informations

^1H NMR, ^{13}C NMR, ^{31}P NMR and ^{19}F NMR spectra were recorded on a Bruker® Avance™ 300, 400 or 600 MHz spectrometer in CDCl_3 , $\text{DMSO}-d_6$, acetone- d_6 , acetonitrile- d_3 or methanol- d_4 solution with the internal solvent signal as a reference. NMR data are reported as follows: chemical shift (ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, dd = doublet of doublets, ddd = doublet of doublet of doublets, td = triplet of doublets, qd = quartet of doublets, m = multiplet, br. s. = broad singlet), coupling constants (Hz) and the number of protons (for ^1H NMR). All reactions were monitored by thin-layer chromatography using Merck silica gel plates 60 F254. All the TLC plates used in order to check the completion of the imines formation or cycloisomerization reactions were previously neutralized in a solution of cyclohexane containing 1% triethylamine. Visualization was accomplished with short-wavelength UV light (254 and 365 nm) and/or staining with appropriate stains (anisaldehyde, *ortho*-phosphomolybdic acid). Purification was performed either *via* recrystallization, preparative TLC, standard flash chromatography (using silica gel of particle size 40–63 μm) or *via* a flash purification system (Biotage Isolera One) unless specified otherwise. IR spectra were recorded on a PerkinElmer Spectrum 65 FT-IR spectrometer with the absorption frequencies reported in wavenumber (cm^{-1}). Melting points were measured using capillaries in a melting point apparatus. Silver catalysts were purchased from Strem, Aldrich or Alfa Aesar. All other commercially available reagents and solvents were used without further purification. High-resolution mass spectrometry (HRMS) measurements were recorded either on a Bruker MicroTOF-Q II™ or Q-TOF maXis™ spectrometer using electrospray ionization (ESI) or atmospheric-pressure chemical ionization (APCI).

General procedure for the formation of 1,2-dihydroisoquinolines

An oven-dried re-sealable tube was charged with a mixture of *ortho*-alkynylbenzalimine (1 eq.), nucleophile (5 eq.), AgOTf

(5 mol%) and acetic acid (1.1 eq.), in acetonitrile (0.08 M). The tube was capped with a rubber septum, evacuated and back-filled with nitrogen. The reaction mixture was stirred at room temperature overnight. After completion of the reaction checked by TLC, the crude reaction mixture was immediately filtered on a short Celite column to eliminate the catalyst. The filtrate was concentrated, and purification of the residue by silica gel column chromatography eluting with cyclohexane/ethyl acetate mixture gave the desired product.

2-(4-Methoxyphenyl)-1-(1-methyl-1*H*-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2a). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), *N*-methylindole (63 mg, 0.48 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 56 mg (79% yield); R_f = 0.34 (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 8.01 (d, J = 7.3 Hz, 1H), 7.47 (d, J = 7.1 Hz, 2H), 7.36 (d, J = 7.6 Hz, 2H), 7.30–7.07 (m, 8H), 7.09 (d, J = 7.2 Hz, 2H), 6.74 (d, J = 7.2 Hz, 2H), 6.72 (s, 1H), 6.61 (s, 1H), 6.34 (s, 1H), 3.62 (s, 6H); ^{13}C NMR (101 MHz, $\text{DMSO}-d_6$) δ 154.4, 141.5, 140.6, 137.6, 136.9, 131.8, 131.6, 128.3, 127.8, 127.7, 127.2, 126.4, 126.0, 125.7, 124.4, 123.1, 121.2, 119.4, 118.9, 116.5, 114.5, 114.2, 111.8, 109.9, 61.3, 55.1, 32.3; IR (neat): 3048, 2922, 2853, 1712, 1607, 1558, 1505, 1488, 1463, 1452, 1372, 1330, 1291, 1242 cm^{-1} ; mp: 159–161 °C; HR-MS (ESI): m/z calculated for $\text{C}_{31}\text{H}_{26}\text{N}_2\text{ONa}$ 465.1943 obtained 465.1931.

7-Fluoro-2-(4-methoxyphenyl)-1-(1-methyl-1*H*-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2b). The general protocol for the cycloisomerization reaction was applied on the following quantities: 1-(5-fluoro-2-(phenylethynyl)phenyl)-*N*-(4-methoxyphenyl)methanimine (53 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 57 mg (77% yield); R_f = 0.39 (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (400 MHz, chloroform- d) δ 8.16–8.08 (m, 1H), 7.54–7.47 (m, 2H), 7.25 (h, J = 4.7, 3.8 Hz, 4H), 7.16 (dd, J = 9.8, 7.0 Hz, 3H), 7.09–7.01 (m, 2H), 7.00–6.86 (m, 2H), 6.73–6.64 (m, 2H), 6.61 (d, J = 8.3 Hz, 2H), 6.27 (s, 1H), 3.67 (s, 3H), 3.58 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 161.7 (d, J = 245.4 Hz), 155.1, 141.6 (d, J = 2.3 Hz), 141.1, 138.0, 137.3, 133.9 (d, J = 6.4 Hz), 128.5 (d, J = 2.8 Hz), 128.4, 127.9, 127.9, 127.8, 126.1, 125.8 (d, J = 7.9 Hz), 123.8, 121.8, 119.5 (d, J = 11.6 Hz), 116.8, 114.3, 114.2, 114.1, 113.0 (d, J = 21.8 Hz), 110.4, 109.7, 61.9, 55.5, 32.8; ^{19}F NMR (376 MHz, chloroform- d) δ –115.62; IR (neat): 3052, 2925, 2953, 1712, 1660, 1607, 1566, 1539, 1506, 1492, 1464, 1447, 1423, 1371, 1331, 1273, 1242 cm^{-1} ; mp: 158–160 °C; HR-MS (ESI): m/z calculated for $\text{C}_{31}\text{H}_{25}\text{N}_2\text{OFNa}$ 483.1849 obtained 483.1850.

7-Chloro-2-(4-methoxyphenyl)-1-(1-methyl-1*H*-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2c). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(5-chloro-2-(phenylethynyl)benzylidene)-4-methoxyaniline (55.3 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 33.5 mg (44% yield); R_f = 0.27 (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (300 MHz, chloroform- d) δ 8.19–8.10 (m, 1H),



7.57–7.48 (m, 2H), 7.35–7.16 (m, 9H), 7.07 (d, $J = 8.2$ Hz, 2H), 6.72 (d, $J = 8.2$ Hz, 2H), 6.66 (s, 1H), 6.61 (s, 1H), 6.30 (s, 1H), 3.73 (s, 3H), 3.66 (s, 3H); ^{13}C NMR (75 MHz, DMSO- d_6) δ 154.8, 141.8, 140.5, 137.4, 137.0, 133.4, 130.9, 130.3, 128.5, 128.1, 127.9, 127.4, 127.2, 126.1, 125.9, 125.7, 123.4, 121.5, 119.5, 119.2, 116.1, 114.3, 110.8, 110.1, 60.8, 55.2, 32.5; IR (neat): 3059, 2925, 2853, 1720, 1609, 1552, 1506, 1479, 1464, 1446, 1371, 1292, 1242 cm^{-1} ; mp: 163–165 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{31}\text{H}_{26}\text{ClN}_2\text{O}$ 477.1728 obtained 477.1718.

7-Methoxy-2-(4-methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2d). The general protocol for the cycloisomerization reaction was applied on the following quantities at 50 $^{\circ}\text{C}$: 4-methoxy-*N*-(5-methoxy-2-(phenylethynyl)benzylidene)aniline (54.6 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 17.3 mg (23% yield); $R_f = 0.5$ (cyclohexane : EA, 7 : 3); yellow solid; ^1H NMR (300 MHz, chloroform- d) δ 8.17 (s, 1H), 7.51 (s, 2H), 7.31–7.12 (m, 7H), 7.07 (d, $J = 8.4$ Hz, 2H), 6.84 (d, $J = 9.0$ Hz, 1H), 6.78–6.66 (m, 3H), 6.63 (s, 1H), 6.58 (s, 1H), 6.26 (s, 1H), 3.77 (s, 3H), 3.69 (s, 3H), 3.60 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 158.7, 154.8, 141.5, 140.1, 138.4, 137.3, 133.8, 128.3, 128.1, 127.6, 127.5, 126.4, 125.7, 125.6, 123.5, 121.6, 119.5, 119.4, 117.1, 114.2, 113.2, 111.6, 111.5, 109.6, 62.2, 55.5, 55.4, 32.8; IR (neat): 3054, 2925, 2852, 1712, 1607, 1559, 1506, 1463, 1446, 1423, 1371, 1319, 1285, 1243 cm^{-1} ; mp: 164–166 $^{\circ}\text{C}$; HR-MS (ESI): m/z calculated for $\text{C}_{32}\text{H}_{28}\text{N}_2\text{O}_2\text{Na}$ 495.2048 obtained 495.2028.

5,6,7-Trimethoxy-2-(4-methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2e). The general protocol for the cycloisomerization reaction was applied on the following quantities at 50 $^{\circ}\text{C}$: 4-methoxy-*N*-(3,4,5-trimethoxy-2-(phenylethynyl)benzylidene)aniline (64.2 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 34.9 mg (41% yield); $R_f = 0.46$ (cyclohexane : EA, 7 : 3); ochre solid; ^1H NMR (300 MHz, chloroform- d) δ 8.13 (d, $J = 5.4$ Hz, 1H), 7.51 (d, $J = 7.3$ Hz, 2H), 7.26 (s, 3H), 7.16 (q, $J = 8.0, 7.3$ Hz, 3H), 7.06 (d, $J = 8.5$ Hz, 2H), 6.83 (s, 1H), 6.70 (d, $J = 8.5$ Hz, 2H), 6.62 (s, 1H), 6.54 (s, 1H), 6.21 (s, 1H), 4.04 (s, 3H), 3.91 (s, 3H), 3.80 (s, 3H), 3.69 (s, 3H), 3.62 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 154.8, 152.4, 148.5, 141.4, 141.4, 141.1, 140.1, 138.4, 137.3, 128.3, 128.2, 128.1, 127.6, 127.5, 126.3, 123.4, 121.6, 119.4, 119.4, 117.1, 114.2, 109.7, 106.6, 105.5, 61.8, 61.6, 61.1, 56.1, 55.5, 32.9; IR (neat): 3053, 2926, 2853, 1729, 1712, 1595, 1561, 1505, 1486, 1459, 1409, 1370, 1328, 1299 cm^{-1} ; mp: 68–70 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{34}\text{H}_{33}\text{N}_2\text{O}_4$ 533.2435 obtained 533.2430.

6-(4-Methoxyphenyl)-5-(1-methyl-1H-indol-3-yl)-7-phenyl-5,6-dihydro-1,6-naphthyridine (2f). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)pyridin-3-yl)methanimine (61.7 mg, 0.2 mmol), *N*-methylindole (131.7 mg, 0.8 mmol), AgOTf (2.6 mg, 5 mol%), AcOH (10 μL). Product: 60 mg (83% yield); $R_f = 0.16$ (cyclohexane : EA, 7 : 3); yellow solid; ^1H NMR (400 MHz, DMSO- d_6) δ 8.39 (d, $J = 4.0$ Hz, 1H), 7.99 (d, $J = 7.8$ Hz, 1H),

7.66 (d, $J = 7.3$ Hz, 1H), 7.49 (d, $J = 6.6$ Hz, 2H), 7.39 (d, $J = 8.1$ Hz, 1H), 7.29–7.07 (m, 6H), 6.97 (d, $J = 8.8$ Hz, 2H), 6.87 (s, 1H), 6.74 (d, $J = 8.8$ Hz, 2H), 6.64 (s, 1H), 6.44 (s, 1H), 3.65 (s, 3H), 3.62 (s, 3H); ^{13}C NMR (101 MHz, DMSO- d_6) δ 154.9, 150.6, 148.1, 145.8, 140.2, 137.1, 136.9, 133.3, 128.4, 127.7, 127.4, 126.2, 125.3, 123.9, 121.4, 121.1, 119.3, 119.1, 116.3, 114.1, 110.9, 110.0, 79.2, 61.2, 55.1, 32.4; IR (neat): 3055, 2925, 2852, 1711, 1602, 1575, 1557, 1507, 1461, 1437, 1424, 1372, 1243 cm^{-1} ; mp: 168–170 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{30}\text{H}_{26}\text{N}_3\text{O}$ 444.2070 obtained 444.2072.

2,3-Bis(4-methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-1,2-dihydroisoquinoline (2g). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-((4-methoxyphenyl)ethynyl)phenyl)methanimine (54.6 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 57 mg (75% yield); $R_f = 0.2$ (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (300 MHz, acetone- d_6) δ 8.18–8.12 (m, 1H), 7.47 (d, $J = 8.8$ Hz, 2H), 7.35–7.32 (m, 1H), 7.31 (t, $J = 1.4$ Hz, 1H), 7.25 (dq, $J = 5.9, 1.5$ Hz, 2H), 7.19–7.13 (m, 3H), 7.08 (d, $J = 9.0$ Hz, 2H), 6.72 (dd, $J = 8.9, 4.4$ Hz, 4H), 6.66 (s, 1H), 6.61 (d, $J = 0.9$ Hz, 1H), 6.35 (s, 1H), 3.70 (s, 3H), 3.66 (s, 3H), 3.63 (s, 3H); ^{13}C NMR (75 MHz, acetone- d_6) δ 160.4, 155.9, 142.6, 142.2, 138.3, 133.6, 132.7, 131.4, 129.7, 128.8, 128.0, 127.3, 126.8, 126.7, 125.0, 124.5, 122.1, 120.4, 119.8, 117.9, 114.8, 114.4, 111.3, 110.5, 63.0, 55.5, 55.4, 32.7; IR (neat): 3050, 2954, 2925, 2854, 1714, 1606, 1577, 1559, 1505, 1483, 1463, 1451, 1441, 1424, 1371, 1328, 1289, 1242 cm^{-1} ; mp: 44–46 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{32}\text{H}_{29}\text{N}_2\text{O}_2$ 473.2224 obtained 473.2214.

3-(2,5-Dimethylphenyl)-2-(4-methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-1,2-dihydroisoquinoline (2h). The general protocol for the cycloisomerization reaction was applied on the following quantities: 1-(2-((2,5-dimethylphenyl)ethynyl)phenyl)-*N*-(4-methoxyphenyl)methanimine (54 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 11.7 mg (16% yield); $R_f = 0.49$ (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (300 MHz, chloroform- d) δ 8.01 (d, $J = 7.8$ Hz, 1H), 7.67 (s, 2H), 7.41–7.19 (m, 4H), 7.13 (s, 1H), 7.05 (s, 1H), 6.98 (d, $J = 8.8$ Hz, 2H), 6.91 (s, 1H), 6.85 (s, 2H), 6.62 (d, $J = 8.9$ Hz, 2H), 6.26 (s, 1H), 6.14 (s, 1H), 3.67 (s, 3H), 3.65 (s, 3H), 2.15 (s, 6H). We observed a rapid degradation of this compound so no ^{13}C NMR could be undertaken.

2-(4-Methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-3-(thiophen-2-yl)-1,2-dihydroisoquinoline (2i). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(thiophen-2-ylethynyl)phenyl)methanimine (50.8 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 42.4 mg (59% yield); $R_f = 0.3$ (cyclohexane : EA, 9 : 1); yellow solid; ^1H NMR (400 MHz, DMSO- d_6) δ 8.03 (d, $J = 7.8$ Hz, 1H), 7.35 (d, $J = 7.8$ Hz, 2H), 7.31 (d, $J = 5.2$ Hz, 1H), 7.29–7.24 (m, 2H), 7.20 (d, $J = 7.1$ Hz, 1H), 7.18–7.12 (m, 2H), 7.07 (d, $J = 9.1$ Hz, 3H), 6.89–6.86 (m, 1H), 6.79 (d, $J = 7.2$ Hz, 3H), 6.47 (s, 1H), 6.24 (s, 1H), 3.65 (s, 3H), 3.59 (s, 3H); ^{13}C NMR (101 MHz,



DMSO- d_6) δ 154.8, 142.2, 140.6, 137.0, 135.5, 131.7, 131.3, 127.8, 127.4, 127.3, 126.4, 126.2, 126.2, 125.9, 125.4, 124.4, 123.2, 121.2, 119.7, 118.7, 115.9, 114.3, 111.3, 109.8, 61.9, 55.1, 32.3; IR (neat): 3065, 2926, 2854, 1713, 1604, 1560, 1530, 1505, 1482, 1464, 1440, 1425, 1372, 1351, 1331, 1289, 1243 cm^{-1} ; mp: 39–41 $^{\circ}\text{C}$; HR-MS (ESI): m/z calculated for $\text{C}_{29}\text{H}_{24}\text{N}_2\text{OSNa}$ 471.1507 obtained 471.1500.

3-(Cyclohex-1-en-1-yl)-2-(4-methoxyphenyl)-1-(1-methyl-1H-indol-3-yl)-1,2-dihydroisoquinoline (2j). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(2-(cyclohex-1-en-1-ylethynyl)benzylidene)-4-methoxyaniline (48.2 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 17.3 mg (24% yield); R_f = 0.4 (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (400 MHz, chloroform- d) δ 8.19 (d, J = 7.9 Hz, 1H), 7.42–7.29 (m, 5H), 7.25 (m, 4H), 6.93 (d, J = 8.9 Hz, 2H), 6.64 (s, 1H), 6.49 (s, 1H), 6.32 (s, 1H), 6.28 (brs, 1H), 3.91 (s, 3H), 3.75 (s, 3H), 2.35–2.25 (m, 1H), 2.23–2.07 (m, 3H), 1.71–1.55 (m, 4H); ^{13}C NMR (101 MHz, chloroform- d) δ 154.8, 144.3, 142.3, 137.4, 134.1, 132.6, 131.7, 129.5, 128.1, 127.1, 126.5, 126.1, 125.8, 124.3, 123.0, 121.5, 119.9, 119.1, 117.4, 114.0, 109.5, 109.1, 62.2, 55.6, 32.8, 26.7, 25.9, 22.8, 22.2; IR (neat): 2928, 2832, 1711, 1601, 1557, 1505, 1483, 1464, 1371, 1241 cm^{-1} ; mp: 91–93 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{31}\text{H}_{31}\text{N}_2\text{O}$ 447.2431 obtained 447.2425.

2-(4-Chlorophenyl)-1-(1-methyl-1H-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline (2k). The general protocol for the cycloisomerization reaction was applied on the following quantities: 4-chloro-*N*-(2-(phenylethynyl)benzylidene)aniline (50 mg, 0.158 mmol), *N*-methylindole (103.8 mg, 0.791 mmol), AgOTf (2 mg, 5 mol%), AcOH (10 μL). Product: 53 mg (76% yield); R_f = 0.5 (petroleum ether : EA, 9 : 1); ochre solid; ^1H NMR (300 MHz, chloroform- d) δ 8.14–8.07 (m, 1H), 7.50 (dd, J = 7.7, 2.0 Hz, 2H), 7.36–7.18 (m, 10H), 7.13–7.03 (m, 4H), 6.70 (s, 1H), 6.59 (s, 1H), 6.40 (s, 1H), 3.59 (s, 3H); ^{13}C NMR (75 MHz, chloroform- d) δ 146.0, 141.1, 137.7, 137.3, 132.6, 131.9, 128.8, 128.5, 128.1, 128.0, 127.6, 127.5, 126.8, 126.7, 126.1, 126.0, 124.8, 123.0, 121.8, 119.5, 119.3, 116.7, 113.1, 109.7, 61.5, 32.8; IR (neat): 3057, 2923, 2852, 1660, 1625, 1613, 1557, 1532, 1488, 1453, 1419, 1392, 1371, 1341, 1318 cm^{-1} ; mp: 65–67 $^{\circ}\text{C}$; HR-MS (ESI): m/z calculated for $\text{C}_{30}\text{H}_{23}\text{N}_2\text{ClNa}$ 469.1447 obtained 469.1429.

1-(1-Methyl-1H-indol-3-yl)-3-phenyl-2-(3-(trifluoromethyl)phenyl)-1,2-dihydroisoquinoline (2l). The general protocol for the cycloisomerization reaction was applied on the following quantities: 1-(2-(phenylethynyl)phenyl)-*N*-(3-(trifluoromethyl)phenyl)methanimine (56 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 53 mg (68% yield); R_f = 0.31 (cyclohexane : EA, 9 : 1); brown solid; ^1H NMR (300 MHz, chloroform- d) δ 8.11–8.05 (m, 1H), 7.50–7.43 (m, 2H), 7.38–7.32 (m, 2H), 7.29–7.25 (m, 4H), 7.24–7.17 (m, 7H), 7.07 (d, J = 6.2 Hz, 1H), 6.72 (s, 1H), 6.59 (s, 1H), 6.45 (s, 1H), 3.61 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 147.7, 140.8, 137.5, 137.3, 132.9, 131.7, 131.1 (q, J = 32.0 Hz), 129.2, 128.6, 128.2, 128.1, 127.6, 127.5, 126.9, 126.1, 126.0, 125.0, 124.5,

124.1 (q, J = 272.0 Hz), 121.9, 119.6, 119.2, 118.1 (q, J = 3.8 Hz), 117.9 (q, J = 3.7 Hz), 116.5, 113.8, 109.8, 61.1, 32.9; ^{19}F NMR (376 MHz, chloroform- d) δ –62.77; IR (neat): 2953, 2923, 2854, 1735, 1654, 1607, 1492, 1453, 1372, 1329, 1248 cm^{-1} ; mp: 184–186 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{31}\text{H}_{24}\text{F}_3\text{N}_2$ 481.1886 obtained 481.1889.

1-(1-Methyl-1H-indol-3-yl)-3-phenyl-2-(2,3,4-trifluorophenyl)-1,2-dihydroisoquinoline (2m). The general protocol for the cycloisomerization reaction was applied on the following quantities: 1-(2-(phenylethynyl)phenyl)-*N*-(2,3,4-trifluorophenyl)methanimine (54 mg, 0.16 mmol), *N*-methylindole (105 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 46 mg (62% yield); R_f = 0.33 (cyclohexane : EA, 9 : 1); ochre solid; ^1H NMR (400 MHz, chloroform- d) δ 8.08–8.01 (m, 1H), 7.61–7.53 (m, 2H), 7.38–7.28 (m, 2H), 7.28–7.15 (m, 8H), 6.80 (s, 1H), 6.68 (s, 1H), 6.64–6.56 (m, 2H), 6.16 (s, 1H), 3.61 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 147.64 (ddd, J = 247.0, 10.5, 2.3 Hz), 146.06 (ddd, J = 251.0, 10.9, 3.6 Hz), 141.76, 139.42 (dd, J = 16.0, 14.5 Hz), 137.14, 136.96, 132.74 (dd, J = 7.7, 3.4 Hz), 132.53, 131.98, 128.52, 128.42, 128.05, 127.60, 127.49, 127.10, 126.38, 126.14, 125.04, 121.60, 119.97, 119.91, 119.38, 116.25, 113.42, 110.97 (dd, J = 17.7, 3.9 Hz), 109.44, 61.33, 32.80; ^{19}F NMR (282 MHz, chloroform- d) δ –139.96 (dd, J = 21.0, 3.6 Hz), –141.36 (dd, J = 20.2, 3.7 Hz), –158.82 (t, J = 20.6 Hz); IR (neat): 3057, 2923, 2853, 1708, 1606, 1562, 1504, 1491, 1422, 1361, 1264, 1220 cm^{-1} ; m.p.: 65–67 $^{\circ}\text{C}$; HR-MS (APCI): m/z calculated for $\text{C}_{30}\text{H}_{22}\text{F}_3\text{N}_2$ 467.1730 obtained 467.1710.

1-(1-Benzyl-1H-indol-3-yl)-2-(4-methoxyphenyl)-3-phenyl-1,2-dihydroisoquinoline (2o). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), *N*-benzylindole (165.8 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 52 mg (63% yield); R_f = 0.45 (petroleum ether : EA, 9 : 1); dark oil; ^1H NMR (400 MHz, chloroform- d) δ 8.14–8.06 (m, 1H), 7.53–7.40 (m, 2H), 7.32–7.09 (m, 13H), 7.08–6.98 (m, 2H), 6.95 (dd, J = 6.6, 2.9 Hz, 2H), 6.78 (s, 1H), 6.71–6.62 (m, 2H), 6.58 (s, 1H), 6.32 (s, 1H), 5.15 (s, 2H), 3.68 (s, 3H); ^{13}C NMR (101 MHz, chloroform- d) δ 154.9, 142.1, 141.2, 138.3, 137.6, 136.8, 132.2, 131.9, 128.7, 128.3, 127.8, 127.8, 127.6, 127.5, 127.3, 126.6, 126.5, 126.4, 126.1, 124.5, 123.8, 121.8, 119.7, 119.6, 118.1, 114.1, 111.4, 110.2, 62.1, 55.5, 50.1; IR (neat): 3058, 2923, 2853, 1726, 1671, 1623, 1606, 1559, 1506, 1464, 1453, 1384, 1357, 1336, 1301, 1244 cm^{-1} ; HR-MS (APCI): m/z calculated for $\text{C}_{37}\text{H}_{31}\text{N}_2\text{O}$ 519.2431 obtained 519.2428.

2-(4-Methoxyphenyl)-3-phenyl-1-(1,2,5-trimethyl-1H-pyrrol-3-yl)-1,2-dihydroisoquinoline (2p). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), 1,2,5-trimethylpyrrole (69.9 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 46 mg (68% yield); R_f = 0.38 (cyclohexane : EA, 9 : 1); brown solid; ^1H NMR (400 MHz, chloroform- d) δ 7.50 (d, J = 6.6 Hz, 2H), 7.18 (m, 5H), 7.07 (t, J = 7.9 Hz, 1H), 6.99 (d, J = 7.4 Hz, 1H), 6.86 (d, J = 8.9 Hz, 2H), 6.61 (d, J = 8.9 Hz, 2H), 6.42 (s,



1H), 5.85 (s, 1H), 5.81 (s, 1H), 3.66 (s, 3H), 3.35 (s, 3H), 2.41 (s, 3H), 2.05 (s, 3H); ¹³C NMR (101 MHz, chloroform-*d*) δ 154.7, 142.6, 141.9, 138.7, 133.6, 131.8, 128.2, 128.0, 127.5, 127.3, 126.8, 126.3, 125.6, 124.2, 123.8, 123.0, 121.4, 113.8, 110.0, 104.6, 62.3, 55.5, 30.2, 12.7, 11.0; IR (neat): 3060, 2923, 2854, 1710, 1652, 1600, 1558, 1506, 1492, 1452, 1398, 1361, 1243 cm⁻¹; mp: 78–80 °C; HR-MS (APCI): *m/z* calculated for C₂₉H₂₉N₂O 421.2274 obtained 421.2267.

2-(4-Methoxyphenyl)-1-(1-methyl-1H-pyrrol-2-yl)-3-phenyl-1,2-dihydroisoquinoline (2q). The general protocol for the cycloisomerization reaction was applied on the following quantities at 50 °C: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), *N*-methylpyrrole (64.9 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 44 mg (69% yield); *R*_f = 0.24 (petroleum ether:EA, 9:1); ochre solid; ¹H NMR (400 MHz, chloroform-*d*) δ 7.52 (d, *J* = 7.9 Hz, 2H), 7.19 (m, 6H), 7.05 (d, *J* = 7.4 Hz, 1H), 6.97 (d, *J* = 8.9 Hz, 2H), 6.70–6.62 (m, 3H), 6.61–6.56 (m, 1H), 5.98 (s, 1H), 5.93–5.87 (m, 1H), 5.65–5.57 (m, 1H), 3.99 (s, 3H), 3.67 (s, 3H); ¹³C NMR (101 MHz, chloroform-*d*) δ 155.5, 142.1, 141.2, 137.9, 133.4, 132.8, 130.4, 128.4, 127.9, 127.6, 127.4, 126.6, 126.4, 124.6, 124.3, 123.4, 114.2, 112.0, 110.0, 106.5, 62.4, 55.4, 35.3; IR (neat): 3059, 2927, 1710, 1609, 1560, 1507, 1491, 1452, 1405, 1366, 1301, 1244 cm⁻¹; mp: 62–64 °C; HR-MS (APCI): *m/z* calculated for C₂₇H₂₅N₂O 393.1961 obtained 393.1956.

2-(4-Methoxyphenyl)-1-(3-methyl-1H-indol-2-yl)-3-phenyl-1,2-dihydroisoquinoline (2r). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), 3-methylindole (104.9 mg, 0.8 mmol), AgOTf (2.1 mg, 5 mol%), AcOH (10 μL). Product: 71 mg (quant.); *R*_f = 0.15 (cyclohexane:EA, 9:1); brown solid; ¹H NMR (400 MHz, chloroform-*d*) δ 7.78 (d, *J* = 8.2 Hz, 1H), 7.61 (d, *J* = 7.8 Hz, 1H), 7.44–7.29 (m, 5H), 7.20 (td, *J* = 8.0, 6.8, 3.5 Hz, 6H), 7.11 (td, *J* = 7.4, 1.5 Hz, 1H), 6.98–6.94 (m, 1H), 6.87–6.81 (m, 2H), 6.62 (d, *J* = 8.9 Hz, 2H), 6.48 (s, 1H), 3.67 (s, 3H), 2.28–2.17 (m, 3H); ¹³C NMR (101 MHz, chloroform-*d*) δ 156.2, 142.6, 140.5, 137.8, 134.5, 131.4, 129.1, 128.6, 128.4, 128.2, 128, 127.9, 126.6, 126.3, 125.8, 124.8, 123.3, 123.3, 122.1, 119.4, 119.2, 114.1, 112.6, 109.5, 107.4, 72.9, 55.4, 9.9; IR (neat): 3055, 2924, 2854, 1709, 1682, 1605, 1561, 1508, 1492, 1454, 1387, 1347, 1296, 1247 cm⁻¹; mp: 52–54 °C; HR-MS (APCI): *m/z* calculated for C₃₁H₂₇N₂O 443.2118 obtained 443.2104.

3,3'-((2-(Phenylethynyl)phenyl)methylene)bis(1-methyl-1H-indole) (3). The general protocol for the cycloisomerization reaction was applied on the following quantities: *N*-(4-methoxyphenyl)-1-(2-(phenylethynyl)phenyl)methanimine (50 mg, 0.16 mmol), *N*-methylindole (63 mg, 0.48 mmol), AgOTf (2.1 mg, 5 mol%), acid (1.1 eq.). *R*_f = 0.22 (cyclohexane:EA, 9:1); white solid; ¹H NMR (400 MHz, chloroform-*d*) δ 7.63–7.57 (m, 1H), 7.48 (d, *J* = 7.9 Hz, 2H), 7.34–7.28 (m, 3H), 7.26–7.18 (m, 9H), 7.01 (t, *J* = 8.0, 6.9, 1.1 Hz, 2H), 6.61–6.54 (m, 3H), 3.68 (s, 6H); ¹³C NMR (101 MHz, chloroform-*d*) δ 146.5, 137.6, 132.3, 131.7, 128.7, 128.5, 128.4, 128.3, 128.1, 127.8, 126.1, 123.5, 122.8, 121.5, 120.2, 118.8, 117.9, 109.1,

94.0, 88.5, 37.9, 32.8.; mp: 177–179 °C; HR-MS (ESI+): *m/z* calculated for C₃₃H₂₆N₂Na 473.1994 obtained 473.2004.

General procedure for the formation of 2k–D

To a solution of **2k** in CDCl₃ was added an excess of AcOD. An aliquot of this mixture was immediately taken for NMR analysis.

2-(4-Chlorophenyl)-1-(1-methyl-1H-indol-3-yl)-3-phenyl-1,2-dihydroisoquinoline-4-d (2k–D). ¹H NMR (300 MHz, chloroform-*d*) δ 8.07 (d, *J* = 7.1 Hz, 1H), 7.45 (d, *J* = 6.6 Hz, 2H), 7.33–7.14 (m, 10H), 7.03 (q, *J* = 8.7 Hz, 4H), 6.54 (s, 1H), 6.35 (s, 1H), 3.59 (s, 3H).

Conflicts of interest

There are no conflicts to declare.

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References

- 1 E. Bangha, H. I. Maibach and P. Elsner, *Skin Pharmacol. Physiol.*, 1996, **9**, 376–380.
- 2 N. F. Kassell, G. Helm, N. Simmons, C. D. Phillips and W. S. Cail, *J. Neurosurg.*, 1992, **77**, 848–852.
- 3 Y.-H. Zhao, Y. Yu, D. Hu, L. Zhao, W. Xie and Z. Zhou, *Asian J. Org. Chem.*, 2020, **9**, 953–960.
- 4 Y.-M. Ma, Y.-B. Zhou, C.-M. Xie, D.-M. Chen and J. Li, *Acta Pharmacol. Sin.*, 2012, **33**, 407–417.
- 5 A. Bischler and B. Napieralski, *Ber. Dtsch. Chem. Ges.*, 1893, **26**, 1903–1908.
- 6 C. Pomeranz, *Monatsh. Chem. Verw. Teile Anderer Wiss.*, 1893, **14**, 116–119.
- 7 P. Fritsch, *Ber. Dtsch. Chem. Ges.*, 1893, **26**, 419–422.
- 8 A. Pictet and T. Spengler, *Ber. Dtsch. Chem. Ges.*, 1911, **44**, 2030–2036.
- 9 K. R. Roesch and R. C. Larock, *J. Org. Chem.*, 1998, **63**, 5306–5307.
- 10 G. Dai and R. C. Larock, *Org. Lett.*, 2001, **3**, 4035–4038.
- 11 G. Dai and R. C. Larock, *Org. Lett.*, 2002, **4**, 193–196.
- 12 Q. Huang and R. C. Larock, *Tetrahedron Lett.*, 2002, **43**, 3557–3560.
- 13 S. Ye and J. Wu, *Tetrahedron Lett.*, 2009, **50**, 6273–6275.
- 14 W. Sun, Q. Ding, X. Sun, R. Fan and J. Wu, *J. Comb. Chem.*, 2007, **9**, 690–694.



- 15 H. Lou, S. Ye, J. Zhang and J. Wu, *Tetrahedron*, 2011, **67**, 2060–2065.
- 16 Q. Ding and J. Wu, *Org. Lett.*, 2007, **9**, 4959–4962.
- 17 X. Yu and J. Wu, *J. Comb. Chem.*, 2010, **12**, 238–244.
- 18 X. Wang, G. Qiu, L. Zhang and J. Wu, *Tetrahedron Lett.*, 2014, **55**, 962–964.
- 19 T. P. Singh and O. M. Singh, *Mini-Rev. Med. Chem.*, 2018, **18**, 9–25.
- 20 V. Bhardwaj, D. Gumber, V. Abbot, S. Dhiman and P. Sharma, *RSC Adv.*, 2015, **5**, 15233–15266.
- 21 Q. Huang, J. A. Hunter and R. C. Larock, *J. Org. Chem.*, 2002, **67**, 3437–3444.
- 22 H. Wang, X. Han and X. Lu, *Chin. J. Chem.*, 2011, **29**, 2611–2618.
- 23 U. Urvashi, G. K. Rastogi, S. K. Ginotra, A. Agarwal and V. Tandon, *Org. Biomol. Chem.*, 2015, **13**, 1000–1007.
- 24 T. Godet, C. Vaxelaire, C. Michel, A. Milet and P. Belmont, *Chem. – Eur. J.*, 2007, **13**, 5632–5641.
- 25 E. Parker, N. Leconte, T. Godet and P. Belmont, *Chem. Commun.*, 2011, **47**, 343–345.
- 26 G. Mariaule, G. Newsome, P. Y. Toullec, P. Belmont and V. Michelet, *Org. Lett.*, 2014, **16**, 4570–4573.
- 27 A. Bontemps, G. Mariaule, S. Desbène-Finck, P. Helissey, S. Giorgi-Renault, V. Michelet and P. Belmont, *Synthesis*, 2016, 2178–2190.
- 28 X. Bantreil, A. Bourderioux, P. Mateo, C. E. Hagerman, M. Selkti, E. Brachet and P. Belmont, *Org. Lett.*, 2016, **18**, 4814–4817.
- 29 J. E. Baldwin, *J. Chem. Soc., Chem. Commun.*, 1976, **18**, 734–736.
- 30 X. Wang, G. Qiu, L. Zhang and J. Wu, *Tetrahedron Lett.*, 2014, **55**, 962–964.
- 31 For recent selected examples: (a) I. Alonso, J. Esquivias, R. Gomez-Arrayas and J. C. Carretero, *J. Org. Chem.*, 2008, **73**, 6400–6404; (b) H. Lin and X.-W. Sun, *Tetrahedron Lett.*, 2008, **49**, 5343–5346; (c) D. Enders, A. Narine, F. Toulgoat and T. Bisschops, *Angew. Chem., Int. Ed.*, 2008, **47**, 5661–5665.
- 32 H. Mayr and A. R. Ofial, *Tetrahedron Lett.*, 1997, **38**, 3503–3506.
- 33 H. Mayr, T. Bug, M. F. Gotta, N. Hering, B. Irrgang, B. Janker, B. Kempf, R. Loos, A. R. Ofial, G. Remennikov and H. Schimmel, *J. Am. Chem. Soc.*, 2001, **123**, 9500–9512.
- 34 S. Lakhdar, M. Westermaier, F. Terrier, R. Goumont, T. Boubaker, A. R. Ofial and H. Mayr, *J. Org. Chem.*, 2006, **71**, 9088–9095.
- 35 T. A. Nigst, M. Westermaier, A. R. Ofial and H. Mayr, *Eur. J. Org. Chem.*, 2008, 2369–2374.
- 36 M. Baidya, F. Brotzel and H. Mayr, *Org. Biomol. Chem.*, 2010, **8**, 1929.
- 37 W. M. Haynes, *CRC Handbook of Chemistry and Physics*, 96th edn, CRC Press, 2015.
- 38 pK_a Plugin, *Marvin* 20.11.0, 2020, ChemAxon (<https://chemaxon.com>).
- 39 P. Hazarika, S. Das Sharma and D. Konwar, *Synth. Commun.*, 2008, **38**, 2870–2880.
- 40 N. Asao, S. S. Yudha, T. Nogami and Y. Yamamoto, *Angew. Chem., Int. Ed.*, 2005, **44**, 5526–5528; N. Asao, S. S. Yudha, T. Nogami and Y. Yamamoto, *Heterocycles*, 2008, **76**, 471–483.

