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Structure and mechanical behavior of *in situ* developed Mg_2Si phase in magnesium and aluminum alloys – a review

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The demand for lightweight, high specific strength alloys has drastically increased in the last two decades. Magnesium and aluminum alloys are very suitable candidate materials. Research on alloys with an Mg_2Si phase started about three decades ago. The current scenario is that magnesium and aluminum alloys containing an Mg_2Si phase are very popular in the scientific community and extensively used in the automotive and aerospace industries. Mg_2Si is a very stable phase and exhibits excellent mechanical, thermal, electrochemical and tribological properties. This paper presents a brief review of Mg–Si binary alloys, and Mg–Si–Al and Al–Si–Mg alloys. Grain refinement methods and mechanical properties have been reported on lightweight alloys containing Mg and Si. The available results show that silicon reacts with magnesium and forms an intermetallic compound with the stoichiometric formula Mg_2Si . There is *in situ* formation of an Mg_2Si phase in Mg–Si, Mg–Si–Al and Al–Mg–Si alloys by the diffusion or precipitation process. The morphology and size of an *in situ* developed Mg_2Si phase depend on the synthesis route and base metal or matrix. In the liquid metallurgy process the precipitation sequence depends on the cooling

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rate. The morphology of the Mg_2Si phase depends on the precipitation sequence and the mechanical properties depend on the morphology and size of the Mg_2Si phase within the alloy matrix.

1. Introduction

During the past quarter century, the demand for lightweight materials has drastically increased, especially for magnesium alloys and its composites. The element of magnesium is the 6th most abundant element in the earth's crust.¹ The density of magnesium (1.739 g/cc) is a minimum among different structural materials. The density of magnesium is approximately equal to two thirds that of aluminum, one fourth that of zinc, and one fifth that of iron.² Magnesium and its alloys possess a very high specific strength among engineering metals. Additionally, magnesium alloys show high damping capacity, excellent castability and superior machinability.³ Currently, researchers have developed a large number of magnesium alloys. Among all these alloy systems, alloys with Mg_2Si particles are very useful. Magnesium-based Mg–Si and Mg–Si–Al alloys are primary alloys systems where Mg_2Si -phase particles are present. The Mg_2Si phase exhibits high hardness ($4.5 \times 10^9 \text{ Nm}^{-2}$), a high elastic modulus (120 GPa), a low density (1.99 g/cc) and a high melting temperature (1085 °C).⁴ It has been reported that aluminum alloys containing an Mg_2Si phase also show better mechanical, electrochemical, and tribological properties.^{5,6} Therefore, Al–Mg–Si alloys are also of great interest to the scientific community. Silicon and aluminum are the 2nd and 3rd most abundant elements in the earth's crust, respectively,⁷ i.e. magnesium, aluminum and silicon are easily available in the earth's crust. Therefore, the cost of raw materials for Mg–Si, Mg–Si–Al and Al–Mg–Si alloys is relatively low. At ambient temperatures, these alloys can substitute for steel

and copper alloys because of their better properties and lower manufacturing costs. This is the primary reason for the automobile and aerospace industry substituting denser materials with magnesium (Mg) and aluminum (Al) alloys as well as their composites.^{8–10}

The mechanical properties of magnesium and aluminum alloys containing an Mg_2Si phase are primarily dependent on the average grain size, morphology, and distribution pattern of the Mg_2Si phase. The intermetallic compound Mg_2Si phase forms *via* precipitation during solidification of molten Mg–Si. After solidification, the precipitates appear in three different patterns and sizes. The patterns are needles, Chinese script and polyhedral or star-like crystals.¹¹ During and after solidification the Mg_2Si phase is found in two forms: (1) coarse dendritic primary Mg_2Si and (2) Chinese script eutectic Mg_2Si . The morphologies of these Mg_2Si phases seriously worsen the mechanical properties. The primary Mg_2Si phase shows a coarse dendritic shape with sharp edges and eutectic Mg_2Si has high stress concentration at the pointed tips. Therefore, for enhancement of mechanical properties it is necessary to modify the structure of the primary and eutectic Mg_2Si phases.^{11–14} Several methods have been reported for modification of the structure of an Mg_2Si phase. Superheating, directional solidification, carbon inoculation and the addition of modifiers are popular methods.^{14–16} Recently Seth *et al.*¹⁷ reported the formation of a fine Mg_2Si phase in Mg–1Si, Mg–4Si and Mg–6Si alloys *via* solid-state sintering using high-energy ball milling. Each specimen of alloy was sintered at 500 °C for 30 minutes. The microstructure of the Mg_2Si phase is completely different from that in the alloy made by the casting process. The Mg_2Si phase is irregular in shape and uniformly distributed. Kondoh *et al.*¹⁸ synthesized Mg_2Si dispersoids *via* solid-state synthesis using a powder metallurgy process. They used fine particles of Si and the specimens were annealed at 580 °C for 240 s in a nitrogen gas atmosphere. The properties of the alloys and composites depend strongly on the physical and chemical characteristics and the microstructures of the dispersoids, *viz.* particle size, volume fraction of the dispersoids and the bonding strength between the matrix and the dispersoids. In another report Kondoh *et al.*¹⁹ reported the *in situ* formation of an Mg_2Si phase from a mixture of magnesium and silicon powder (Mg–33.33 mol% Si) by a repeated plastic working (RPW) process. During the RPW process, the *in situ* formation of an Mg_2Si phase by the reaction of magnesium and silicon is similar to the mechanical alloying process. X-ray diffraction shows that up to 150 cycles of RPW no peak of Mg_2Si was observed. On increasing the number of cycles from 150 to 600, the peak intensity of the Mg_2Si phase was enhanced. Limited literature is available for the *in situ* formation of an Mg_2Si phase *via* powder metallurgy and secondary processes. Most reports are based on the liquid metallurgy process.

The present study summarizes the results of the *in situ* formation of an Mg_2Si phase in magnesium and aluminum



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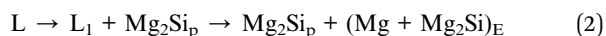
2. Mg₂Si phase and its *in situ* development in different alloys

$$2\text{Mg} + \text{Si} \rightarrow \text{Mg}_2\text{Si} \quad (1)$$

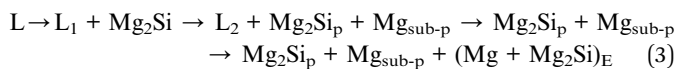
2.1 Magnesium (Mg-Si) alloys

The Mg–Si alloy system is a primary alloy system which forms an *in situ* Mg₂Si phase or particles. The maximum solid solubility of silicon in magnesium is only 0.003 at%.¹⁷ Therefore, Si atoms react with Mg atoms, leading to the formation of the intermetallic compound Mg₂Si. Silicon improves fluidity, which allows the strengthening of Mg by precipitation hardening.²⁰ Therefore, the mechanical properties of Mg–Si alloys are improved. As shown in the binary phase diagram of Mg–Si in an inert atmosphere, the eutectic phase forms above 649 °C containing 1.38 at% silicon²¹ (Fig. 1). Zhou *et al.*²¹ synthesized the as-cast hypoeutectic Mg–1Si alloy, eutectic Mg–1.38Si alloy, and hypereutectic Mg–2Si, Mg–3Si and Mg–4Si alloys and reported optical micrographs of these alloys. The micrographs of the hypoeutectic alloy show it contains α -Mg and the eutectic Mg₂Si phase. Moreover α -Mg, the eutectic Mg₂Si phase and the beginning of the formation of a primary Mg₂Si phase appeared in the eutectic composition Mg–1.38Si. On further increasing the amount of silicon, the primary Mg₂Si phase coarsened gradually in the hypereutectic composition.

Pan *et al.*²² studied the microstructural behavior and phase analysis of hypereutectic Mg-8Si alloy using electron probe microanalysis (EPMA) and reported that the solidification path depends on the cooling conditions. At a very low cooling rate, the solidification process is almost as per the equilibrium state. At equilibrium the solidification path is shown in eqn (2), where p represents the primary Mg₂Si particles and E represents the Mg₂Si eutectic crystals.



During solidification of the liquid phase, the primary Mg_2Si precipitate initially forms in the melt. When the temperature decreases and reaches the eutectic temperature, the eutectic reaction progresses and the rest of the melt solidifies as eutectic $\text{Mg-Mg}_2\text{Si}$. When the solidification rate is fast, such as in the case of ingot metallurgy, then solidification is a non-equilibrium process and a different path is followed, as shown in eqn (3).



According to eqn (3) after the formation of primary Mg_2Si particles, magnesium will subsequently precipitate as sub-primary particles ($\alpha\text{-Mg}$) before the final eutectic reactions in the rest of the melt. Therefore, primary Mg_2Si particles, $\alpha\text{-Mg}$ and $\text{Mg-Mg}_2\text{Si}$ eutectic crystals are present in the optical micrograph (Fig. 2(a and b)). The primary Mg_2Si dendrites show a polygonal shape with an average grain size of $30\text{ }\mu\text{m}$.²² Jaing *et al.*¹² reported the microstructure of the Mg-5Si alloy with an X-ray diffraction pattern. The X-ray diffraction pattern confirms the formation of the Mg_2Si phase and SEM micrographs clearly show the morphology pattern of primary Mg_2Si dendritic crystals, $\alpha\text{-Mg}$ halos, and a eutectic Mg_2Si structure (Fig. 2(c and d)).

Chen *et al.*²³ reported the FESEM micrographs of the primary Mg₂Si phase formed from Mg–4Si alloy. The Mg₂Si phase was an

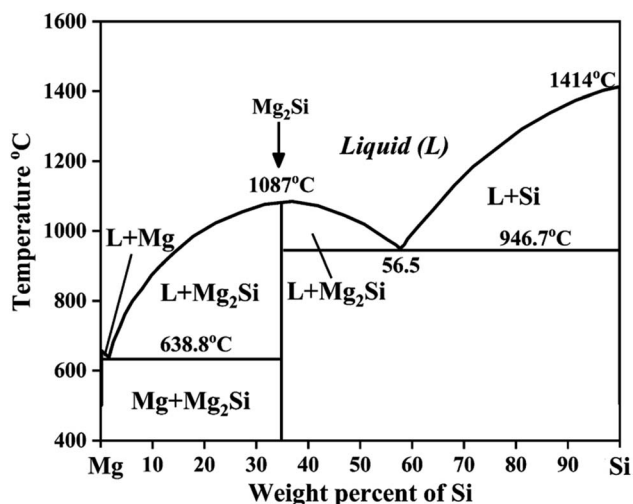


Fig. 1 Phase diagram of Mg–Si alloys. This figure has been reproduced from ref. 11 with permission from Elsevier, copyright 2008, after redrawing.

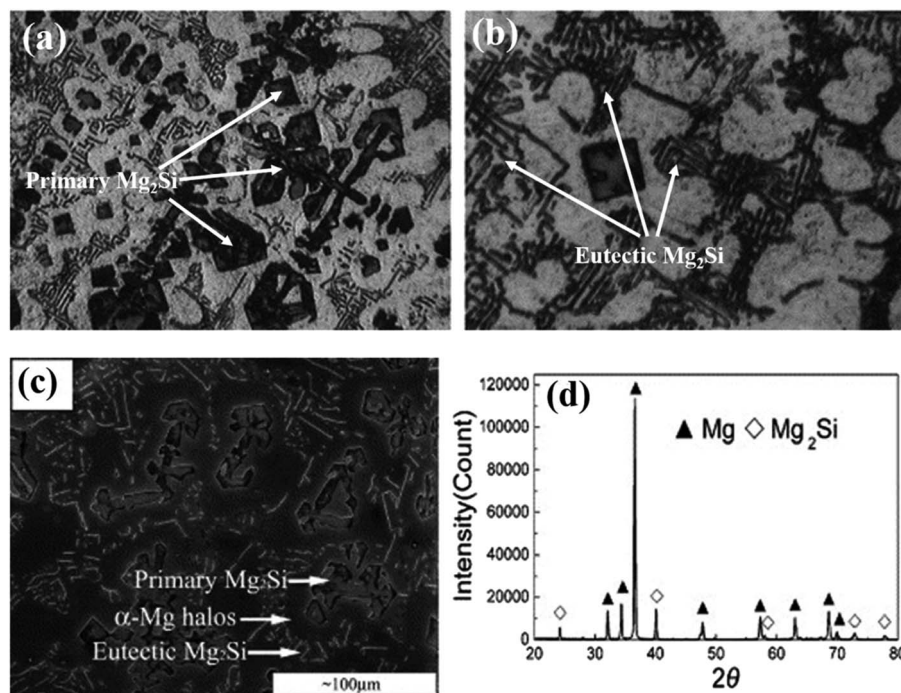


Fig. 2 (a and b) Optical micrographs of Mg–8Si alloy. These figures have been reproduced from ref. 22 with permission from Elsevier, copyright 2005. (c and d) SEM micrograph and XRD pattern of as-cast Mg–5Si alloy, respectively. These figures have been reproduced from ref. 12 with permission from Elsevier, copyright 2005.

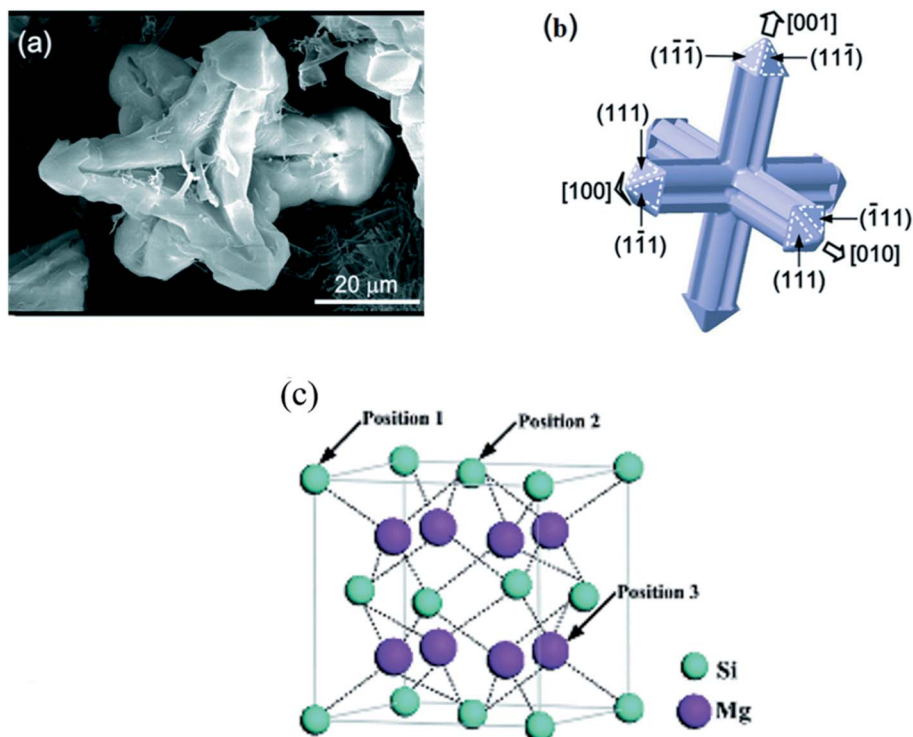


Fig. 3 (a and b) Structure and planes of the primary Mg₂Si phase. (c) Molecular structure of the Mg₂Si phase. This figure has been reproduced from ref. 23 with permission from Royal Society of Chemistry, copyright 2014.



The formation of the primary Mg_2Si phase is dependent on the solidification conditions.²² The solidification begins with nucleation as well as the growth of Mg_2Si primary crystals. After nucleation, the growth will occur *via* the kinetics of atom attachment to the solid-liquid interface, heat and mass transfer, and capillarity. The growth rate of Mg_2Si dendrites is mainly solute-diffusion-controlled. Mg_2Si phase dendrites continuously grow into liquid under a high degree of super cooling, because atoms migrate rapidly towards the cooling interface. Diffusion of atoms, however, is relatively slow in the liquid. Therefore, Mg_2Si crystals show anisotropic properties, such as the interface energy, which leads to growth of dendrite arms in specific crystallographic directions, resulting in a complex dendritic morphology. The mechanical properties of these alloys are inversely proportional to the square root of the average grain size of the Mg_2Si phase. The average grain size of the Mg_2Si phase formed *via* a liquid metallurgy process (casting method) is $\sim 100 \mu\text{m}$.^{12,24}

Magnesium (AS series) alloys containing aluminum and silicon are also very attractive for the automotive and aerospace industries due to their high potential as heat-resistant, creep-resistant and low-density materials.²⁵ The Mg–Si–Al (AS21) alloy was primarily developed by Volkswagen around 1970 for its stress-resistant and creep-resistant capabilities at temperatures up to 150 °C.²⁶ Grains of Mg–Si–Al alloys were composed of an Mg-rich solid solution (α -phase). The grain boundaries were composed of intermetallic compounds Mg₁₇Al₁₂ (β -phase) and a silicon-rich Mg₂Si phase.²⁷ The presence of the intermetallic compounds Mg₂Si and Mg₁₇Al₁₂ is responsible for these advantageous properties. In Mg–Si–Al alloys, Mg₂Si acts as an excellent heat-resistant strengthening phase because the Mg₂Si phase has a high melting point and a low thermal expansion coefficient.²⁵ During the last two decades numerous studies have been done on AS series alloys to understand their machinability and creep mechanisms at elevated temperatures.²⁷ AS series alloys have high creep resistance and corrosion resistance. Therefore, AS series alloys are used in automotive engine components. Zhang²⁸ investigated the creep behavior of the die-cast AS21 alloy at temperatures up to 150 °C under compression. Initially the creep rate (primary creep) decreases and the stress exponent n is in the range of 13–19. This is because the die-cast AS21 alloy was hardened by the stable rod-shaped Mg₂Si precipitates. Furthermore, the creep rate goes to a minimum followed by a gradual increase corresponding to

Similarly, aluminum alloys containing Mg and Si are heavily used in engineering applications³⁰ for critical structural shapes because these alloys have super ductility with high strength.³¹ The 6XXX series³² of alloys are mainly Al-Mg-Si alloys. Moreover, the 3XX.0 series, the 4XX.0 series of cast alloys and the 4XXX series of wrought alloys may also be Al-Mg-Si alloys. The characteristics of the 6XXX series of alloys are heat treatability, high corrosion resistance and excellent extrudability. The mechanical properties of these alloys can be easily improved by conventional thermo-mechanical processing, including quenching, cold drawing, and artificial aging.³³ The required level of mechanical strength can be achieved through dislocation strengthening and precipitation hardening. Enhanced electrical conductivity is attained by a decrease in the content of solute Mg and Si atoms in the Al matrix. In the 3XX.0, 4XX.0 and 4XXX series of alloys, magnesium is intentionally added to induce precipitation hardening *via* the formation of an Mg_2Si phase, metastable phases, or Guinier-Preston zones.³⁴ During the last two decades a lot of research has been done on Al-Si-Mg alloys. Generally, the precipitation behavior of Al-Mg-Si alloys is considered to follow the sequence shown in eqn (4).

The addition of an excess amount of silicon in Al-Mg-Si alloys significantly promotes the precipitation of silicon, forming fine precipitates with an unknown structure. The excess Si is tied up with the clusters/zones and β'' particles. The β'' particles are rapidly dissolved during high-temperature heating or aging and Si and other precipitates are formed.³⁶ Mandal *et al.*³⁷ reported that hypereutectic Al-Si-Mg alloys with more than 2 wt% Mg exhibit a microstructure in which the Si particles are inherently refined and uniformly distributed. On further increasing the Mg content, both the amount and size of the Mg_2Si phase increase. It has been reported that during isothermal aging (in the temperature range 150 °C to 200 °C), the strength of Mg-Si-Al alloys increases rapidly.³⁸ With the high rate of cooling of Mg-Si-Al

alloys, supersaturated solids are formed. With the help of isothermal aging or the age hardening process, the supersaturated solid solutions are transformed and very fine coherent or semi-coherent precipitates nucleate and grow.³⁸

For a better understanding of magnesium and aluminum alloy systems containing an Mg_2Si phase, it is necessary to study the refinement modification method. Because of refinement, the morphology and grain size are primarily responsible for the enhancement in mechanical properties.

3. Synthesis routes of the formation of *in situ* Mg_2Si phase

The synthesis processes of magnesium and aluminum alloys containing an Mg_2Si phase are approximately similar. Generally, liquid metallurgy, powder metallurgy and mechanical alloying methods are used for the *in situ* fabrication of alloys containing an Mg_2Si phase. Each synthesis process uses a controlled atmosphere. The formation of the Mg_2Si phase occurs *via* the precipitation or diffusion of silicon into magnesium. Apart from liquid metallurgy, powder metallurgy and mechanical alloying, a few novel methods have also been reported by researchers: *e.g.* mechanically recitative extrusion process and spark plasma sintering.

3.1 Liquid metallurgy process

In the liquid metallurgy process, the alloying elements (powder or ingot forms) and base metal (magnesium or aluminum) are heated at high temperature, in an inert gas atmosphere. When the metal has melted, the liquid metal is poured into a mold, within which it solidifies into the required shape. At high temperature magnesium has high affinity towards oxygen. Therefore, the fabrication of magnesium alloys and aluminum alloys containing magnesium is very challenging. To overcome the limitations of the conventional casting process, various new techniques have been developed for the casting of magnesium-based alloys. High-pressure die casting, gravity casting

(permanent mold casting and gravity sand casting) and other relevant casting processes, *e.g.* squeeze casting, thixo-casting and thixo-molding, are used to produce lightweight high-performance aerospace, automotive and defense equipment components.^{39,40} Kumar *et al.*⁵ synthesized hypo (Mg–0.5, 0.7, 1.15 wt% Si) and hyper (Mg–2, 4, 6, 8 and 10 wt% Si) eutectic Mg–Si alloys using the gravity casting method and reported that the wear rate of Mg–Si alloys decreases with an increase in Si addition under normal loads. Dargusch *et al.*⁴¹ synthesized Mg–Si–Al alloys *via* high-pressure die casting and reported that the morphology of the Mg_2Si phase is a function of silicon content. Similarly Dwivedi *et al.*⁴² reported the influence of silicon content (varying from 4 to 20 wt%) on the mechanical properties of cast Al–Si–Mg alloys. Ductility reduced monotonically and hardness increased with increasing silicon content over the range 4–20%. Generally, casting processes are economical fabrication techniques.^{43,44}

3.2 Powder metallurgy process

In general, powder metallurgy includes the mixing of powders, pressing, degassing and sintering under a controlled atmosphere or in a vacuum. Generally, powders have particle sizes less than 1000 nm. Powder particles have a high ratio of surface area to volume. This helps the use of metal powders as catalysts in various metallurgical and chemical reactions.⁴⁵ The metal powders are consolidated *via* three basic approaches: (1) pressure-based densification, (2) sinter-based densification and (3) hybrid densification. Pressure-based densification establishes density *via* the compaction pressure, the surface energy of the powder particles and sintering parameters. This process is mainly used by researchers and industry. The powder particles have irregularities, good flow characteristics and must be compressible. The compacted specimens of these particles show high green density (density before sintering). Irregular particles have a high surface energy and the powder has the ability to withstand higher compressibility during compaction. These particles are highly plastically deformed and easily interlocked. Seth *et al.*¹⁷ reported the SEM micrographs of pure

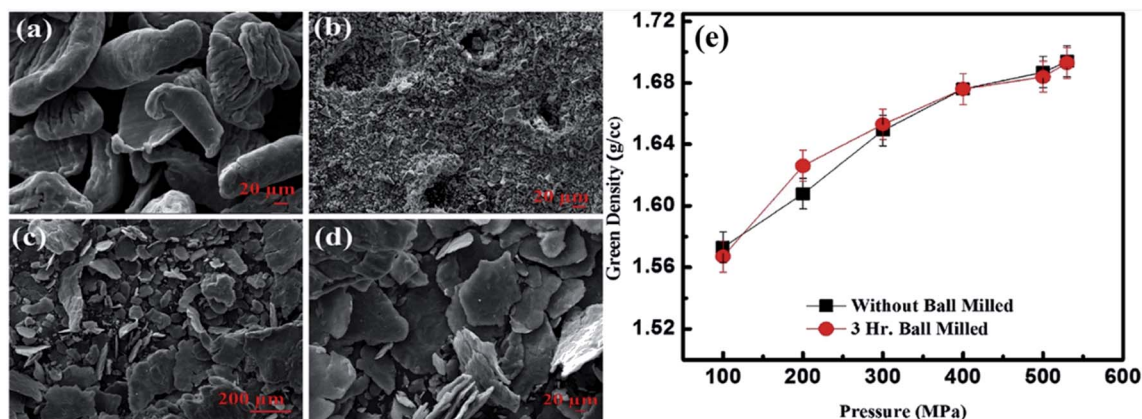


Fig. 4 SEM micrographs of (a) pure magnesium powders, (b) silicon powders, (c and d) Mg–Si alloy powders at two different magnifications and (e) the green density of pure magnesium powders with and without ball milling and compacted at different pressures. This figure has been reproduced from ref. 17 with permission from Elsevier, copyright 2020.



magnesium, silicon (as received) and magnesium–silicon alloy powder after three hours of ball milling and reported the effect of compaction pressure on the green density of magnesium powder with and without ball milling at different compaction pressures.

Fig. 4(a–d) show the SEM micrographs of magnesium, silicon and magnesium–silicon alloy powders at two different magnifications. The morphologies of magnesium and silicon are irregular semi-elliptical three-dimensional structures and irregular angular fine as well as coarse crystal shapes, respectively (Fig. 4(a and b)). The morphologies of the alloy powders after three hours of ball milling are shown in Fig. 4(c and d). Magnesium particles were converted into irregular thin sheets like lamellae. The size of the silicon particles was reduced without any change in the morphology.

Fig. 4(e) shows the green density of as-received and 3 hours-ball-milled powders. The green density of both as-received and ball-milled powders increased rapidly with an increase in the compaction pressure up to 300 MPa. However, the rate of increase of green density diminishes with increasing load after a load of 300 MPa in both cases. Moreover, the green density of as-received powder and 3 hours-ball-milled powders are same at a higher compaction pressure (400 MPa to 530 MPa). The green density at a pressure of 530 MPa is >95% of the theoretical density of pure magnesium. Therefore, researchers generally use a pressure-based densification process for the fabrication of Mg–Si alloys and composites.

The second process of powder metallurgy was sintering-based densification. In sintering-based densification the molding process is used to give the shape of the components. Generally, metal powders are mixed with binder (wax and plasticizers) and the mixture is poured into a molding machine (a metal injection molding machine). Green specimens are received from the molding machine. Before sintering, the binders are removed by a debinding process. The densification is enhanced by using finer powder particles and a higher sintering temperature. Scharrer *et al.*⁴⁶ fabricated the AZ, AJ and AS series of alloys *via* the magnesium injection molding process and reported the yield strength of the AS41 alloy.

The third process of powder metallurgy is hybrid densification. In the hybrid-densification process, pressure and temperature are applied at the same time. Spark sintering and hot pressing (often in an inert atmosphere or vacuum) are hybrid-densification processes. Sun *et al.*⁴⁷ synthesized Mg–Mg₂Si composites *via* the solid-state reaction of magnesium and silicon by using spark plasma sintering (SPS) techniques. The reaction occurs by diffusion of magnesium and silicon, leading to the formation of an Mg₂Si phase. The activation energy of formation of the Mg₂Si phase was found to be 376.0 kJ mol^{−1}.

3.3 Mechanical alloying

Mechanical alloying (MA) is the other powder metallurgy process used to develop intermetallic compounds in metals powders. MA is a solid-state synthesis technique that involves repeated fracturing and re-welding of powder particles in a high-energy ball mill. This technique is also used to synthesize supersaturated

solid solutions, intermediate phases, quasi-crystalline phases, amorphous alloys, and high-entropy alloys.⁴⁸ This technique is also used to produce very fine crystallite sizes down to the nano-scale.⁴⁹ There are very few published reports concerning the formation of an Mg₂Si phase by mechanical alloying at present. Niu *et al.*⁴⁸ formed the Mg₂Si phase *via* a mechanical alloying method using a high-energy planetary ball mill. Stoichiometric compositions of magnesium and silicon powders are milled at 250 RPM with a ball to powder ratio of 20 : 1 for 30 hours. X-ray diffraction patterns are taken after 0, 6, 10, 20, 30 hours of ball milling. X-ray diffraction patterns of the ball-milled powders show that the peaks of the Mg₂Si phase are initially seen in powders ball milled for 10 hours. The relative intensity of the Mg₂Si phase peaks increased with an increase in the ball milling time. This indicates that a longer milling time leads to the formation of more Mg₂Si phase. The activation energy for the formation of the Mg₂Si phase is found to be 215 kJ mol^{−1} based on the Kissinger approach. Similarly, Jung *et al.*⁵⁰ successfully synthesized thermoelectric Mg₂Si by a mechanical alloying method for 24 hours of ball milling and examined the electronic transport properties (Hall coefficient, carrier concentration and mobility) and the thermoelectric properties (Seebeck coefficient, electrical conductivity and thermal conductivity). Mg₂Si shows n-type conduction behavior. This indicates that the electrons are responsible for the electrical conduction.

3.4 Other process

Liquid metallurgy, powder metallurgy and mechanical alloy processes are suitable and highly acceptable methods for the *in situ* development of Mg₂Si for the scientific community. Some reports show that an *in situ* Mg₂Si phase is developed *via* a repetitive extrusion or mechanical working process. The mechanism of development of the Mg₂Si phase *via* mechanical working and extrusion processes can be diffusion controlled similar to powder metallurgy and mechanical alloying. Aizawa *et al.*⁵¹ reported a mechanically induced reaction for the solid-state synthesis of Mg₂Si. In general, the mechanically induced reaction occurs *via* two processes: a nucleation controlled

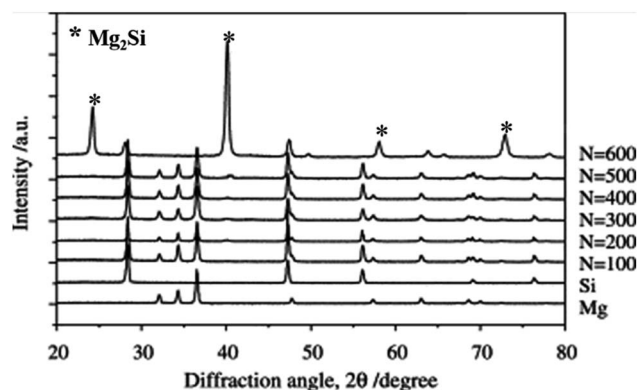


Fig. 5 Variation in XRD profiles with increasing number of cycles in the bulk mechanical alloying for the Mg–Si system. This figure has been reproduced from ref. 51 with permission from Elsevier, copyright 2006.

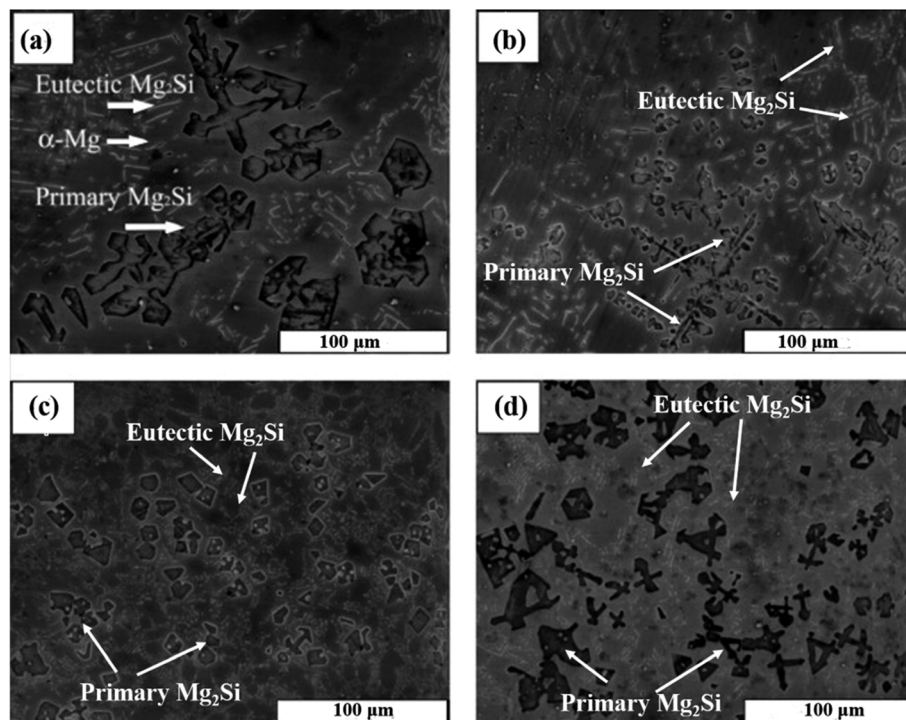


Fig. 7 SEM micrographs of Mg–5 wt% Si alloys (a) without modifier, (b) 5 wt% K_2TiF_6 , (c) 5 wt% KBF_4 and (d) 5 wt% $KBF_4 + K_2TiF_6$ (mixtures with weight ratio of 4 : 1). This figure has been reproduced from ref. 24 with permission from Elsevier, copyright 2005.

Table 1 Effect of K_2TiF_6 , KBF_4 and $KBF_4 + K_2TiF_6$ in Mg–5Si alloys.²⁴

Modifiers	Effect of modifier	Morphology
K_2TiF_6 (5 wt%)	The size of primary and eutectic Mg_2Si phase particles are slightly reduced	Primary Mg_2Si – dendritic shape Eutectic Mg_2Si – Chinese script
KBF_4 (5 wt%)	Average grain size of primary Mg_2Si phase significantly decreased from 100 μm to $\sim 20 \mu m$	Primary Mg_2Si – polyhedral shape Eutectic Mg_2Si – fine fibers
$KBF_4 + K_2TiF_6$ (5 wt%) weight ratio 4 : 1	Primary Mg_2Si became coarser, and eutectic Mg_2Si became finer	Primary Mg_2Si – irregular shape Eutectic Mg_2Si – modified morphology with a finer size

Jiang *et al.*¹² modified the Mg–5 wt% Si alloy by the addition of 0.1, 0.4, 0.8 and 1.2 wt% of yttrium (Y). At an early stage of solidification, the addition of a small amount of yttrium forms

Y-containing compounds in magnesium–silicon melts, because Y has a relatively high solubility in magnesium alloys. Y-containing compounds act as heterogeneous nucleation sites

Table 2 Effect of modifiers on average grain size of the Mg_2Si phase in magnesium and aluminum alloys

Alloy phase composition	Modifiers	Results	
		Average grain size of primary Mg_2Si grains in unmodified alloys	Minimum size of primary Mg_2Si grain in modified alloys
Mg + 5Si/Mg–Si ¹²	Yttrium	$\sim 100 \mu m$	$\sim 30 \mu m$ at (0.8 wt% of yttrium)
Mg + 5Si/Mg–Si ¹⁴	Lanthanum	$\sim 100 \mu m$	$\sim 25 \mu m$ at (0.5 wt% of lanthanum)
Mg + 5Si/Mg–Si ¹¹	Bismuth	$\sim 100 \mu m$	$\sim 15 \mu m$ at (0.5 wt% of bismuth)
Mg + 5Si/Mg–Si ¹³	KBF_4	$\sim 90 \mu m$	$\sim 20 \mu m$ at (2–10 wt% of KBF_4)
$Mg_2Si/Al-Si$ ⁵⁴	K_2TiF_6	$\sim 100 \mu m$	$\sim 30 \mu m$ at (3 wt% of K_2TiF_6)
Mg + 5Si/Mg–Si ²⁴	$KBF_4 + K_2TiF_6$ ratio (4 : 1)	$\sim 100 \mu m$	$\sim 20 \mu m$ at (5 wt% of $KBF_4 + K_2TiF_6$)
$Mg_2Si/Al-Si-Cu$ ⁵⁵	Cerium	$\sim 100 \mu m$	$\sim 15 \mu m$ at (0.4 wt% Ce)
$Al-Mg_2Si$ ⁵⁶	Lithium	$\sim 30 \mu m$	$\sim 15 \mu m$ at (0.3 wt% of lithium)
Mg–4%Si alloy ⁵⁷	Strontium	$\sim 46 \mu m$	$\sim 16 \mu m$ at 3% Al–10% Sr



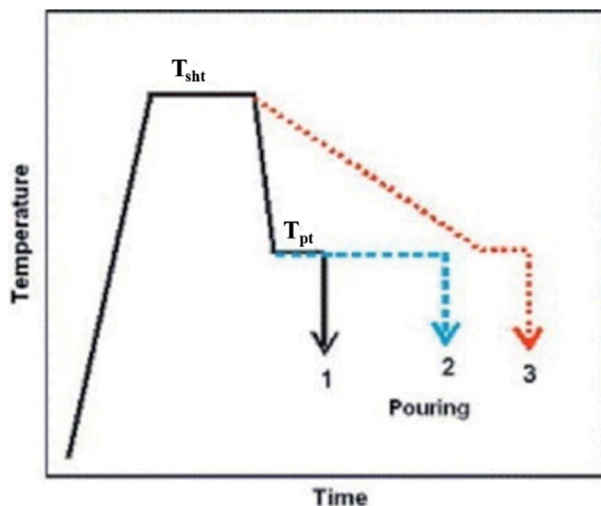


Fig. 8 Schematic of typical superheating process with three different temperature profiles. This figure has been reproduced from ref. 58 with permission from Elsevier, copyright 2007.

for the primary Mg_2Si phase particles. Therefore, the growth of Mg_2Si particles is significantly influenced. On the addition of yttrium (0.1 or 0.4 wt%), the average grain size of primary and eutectic Mg_2Si phase particles is not significantly reduced. However, the addition of 0.8 wt% of yttrium reduces the average grain size of primary Mg_2Si phase particles from $100\text{ }\mu\text{m}$ to $\sim 30\text{ }\mu\text{m}$. The average grain size of the primary Mg_2Si phase increases with the increase in wt% of yttrium (Fig. 6).

Wang *et al.*²⁴ modified the Mg–5 wt% Si alloy with K_2TiF_6 , KBF_4 and $\text{KBF}_4 + \text{K}_2\text{TiF}_6$. The unmodified microstructure of Mg–5 wt% Si alloy consists of coarse primary Mg_2Si crystals, Chinese script eutectic Mg_2Si phase particles, and $\alpha\text{-Mg}$. The average grain size of the Mg_2Si phase in the unmodified microstructure is $100\text{ }\mu\text{m}$ (Fig. 7). Table 1 shows the effect of different modifiers on the primary Mg_2Si phase, eutectic Mg_2Si phases and the grain size of the Mg_2Si phase.

Several reports are available on the modification of Mg_2Si phase particles in magnesium and aluminum alloys. A critical review of important articles concluded that bismuth and lithium show good Mg_2Si phase refinement in magnesium and aluminum alloys, respectively. Based on these reviews, Table 2 shows that the Mg_2Si phase refinement results in magnesium and aluminum alloys.

4.2 Refinement via superheating method

In the superheating process, the refinement of grains by applying high-temperature heating (more than $150\text{ }^\circ\text{C}$ to $300\text{ }^\circ\text{C}$ from the equilibrium liquidus temperature of the alloy) and holding it for a short period of time followed by rapid cooling near to the pouring temperature of the alloy melt and again holding it for a short period of time before casting. Primarily this method was reported in a British patent granted in 1931 (Fig. 8).⁵⁸

Cao *et al.*⁵⁸ reported three different cooling conditions for Mg–Al alloys, which are shown in Fig. 8: (1) rapid cooling from T_{sht} (super heating temperature) to T_{pt} (pouring temperature)

with a short holding time before casting; (2) rapid cooling from T_{sht} to T_{pt} , but with a lengthy holding time at T_{pt} before casting; and (3) slow cooling from T_{sht} to T_{pt} . In Mg–Al alloys, a high level of grain refinement requires rapid cooling of the alloy melt from T_{sht} to the pouring temperature T_{pt} and pouring it immediately (Case 1 in Fig. 8). The desirable grain refinement is achieved by the appropriate superheating temperature and time. The range of specific superheating temperatures (T_{sht}) for desirable grain refinement depends on the alloy composition.

During the last few decades extensive research has been reported and a large number of hypothesis have been proposed for various applications. The superheating process is generally used to refine the Mg_2Si phase in magnesium and aluminum alloys. Min *et al.*⁵⁹ investigated the melt superheating treatment effect on the grain refinement of Mg–3.5Si–1Al alloy as unmodified specimens or specimens modified with 0.2% Sr–Sb (mass fraction). The unmodified Mg–3.5Si–1Al alloy was refined progressively with an increase in the temperature from $750\text{ }^\circ\text{C}$ to $900\text{ }^\circ\text{C}$. However, alloys modified by Sr–Sb could be refined up to a melt superheating temperature of $850\text{ }^\circ\text{C}$. A further increase in the temperature from $850\text{ }^\circ\text{C}$ to $900\text{ }^\circ\text{C}$ leads to a slight increase in the grain size. Gu *et al.*⁶⁰ reported that on increasing the temperature from $750\text{ }^\circ\text{C}$ to $900\text{ }^\circ\text{C}$, in Mg–1.5Si–1Zn alloy, the grain size of the primary Mg_2Si phase decreased from $18\text{ }\mu\text{m}$ to $9\text{ }\mu\text{m}$. The polyhedral morphology remains unchanged. Moreover, the grains of the eutectic Mg_2Si phase were refined without a change in morphology.

4.3 Refinement via rapid solidification

In the rapid solidification process the alloy melts are solidified at a high cooling rate. The cooling rate is enhanced by using substrate quenching and atomization.⁶¹ The substrate quenching techniques include the thermal spray method, planar flow casting, the melt-spinning technique, twin rolling copper, and mold casting.^{61–64} In the atomization technique the molten alloy stream of the liquid is broken into small spheres using a high-pressure jet of gas. This process includes high pressure and centrifugal gas atomization.⁶⁵ Gensing *et al.*⁶⁶ reported the microstructures of rapidly solidified magnesium–silicon alloys. Three different microstructures were observed in the rapidly solidified Mg–Si alloys: a cellular (dendritic) structure based on magnesium, a coupled eutectic structure, and a dendritic structure based on Mg_2Si . Matsuda *et al.*⁶⁷ reported that with the rapid solidification process, it is possible to produce an alloy of homogeneous composition, a fine and uniform distribution of the intermetallic phase and a highly fine-grained matrix. Mabuchi *et al.*⁶⁸ reported the tensile behavior of Mg– Mg_2Si composites processed under rapid solidification. At room temperature, the Mg– Mg_2Si composite shows an ultimate tensile strength of 506 MPa , a yield strength of 455 MPa and a percentage of elongation of 1.6% up to failure.

4.4 Refinement of Mg_2Si phase and enhancement of mechanical properties via mechanical working process

Mechanical working processes are the most promising approaches ensuring property enhancement in alloys and



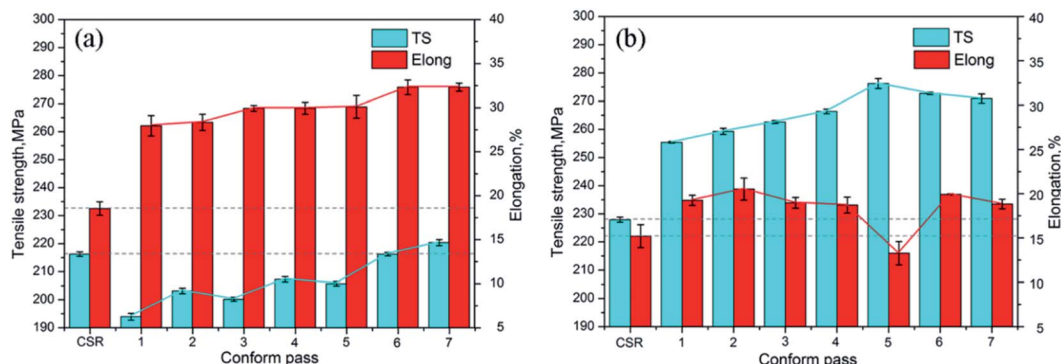


Fig. 9 The tensile strength (TS) and elongation (strain %) of the Al–Mg–Si alloy versus Conform pass: (a) after 1 to 7 Conform processes (N) and (b) after 1 to 7 Conform processes followed by aging at 175 °C for 8 h. This figure has been reproduced from ref. 73 with permission from Elsevier, copyright 2017.

composites.^{69,70} Mechanical working processes have high potential to refine the grain and the Mg_2Si phase present in aluminum and magnesium alloys. Mabuchi *et al.*⁷¹ synthesized Al–Mg–Si– Si_3N_4 composites with hot extrusion at 500 °C with a reduction ratio of 100 : 1. During hot extrusion, recrystallization and dynamic precipitation occurred. Therefore, very small grains, of less than 3 μm were produced. Ke *et al.*⁷² reported that during the hot extrusion, dynamic recrystallization of the aluminum matrix and dynamic precipitation of deep sub-micron silicon and Mg_2Si particles and deformation-induced twinning of silicon particles occurred in the Al–12.0%Si–0.2% Mg alloy. By the hot extrusion process, a good combination of tensile strength and elongation (256.3 MPa and 15%, respectively) are obtained after aging for 12 hours of ESA (hot-extruded, solutionized and aged) specimens. Hu *et al.*⁷³ prepared Al–Mg–Si alloy with a diameter of 9.5 mm by a continuous casting and rolling process. The alloy rod was continuously deformed through the R-Conform process (repetitive continuous extrusion forming) and followed by aging at 175 °C for 8 h.

The variation in tensile properties and % of elongation (strain) of Al–Mg–Si alloys after different Conform passes with and without corresponding aging (hardening) treatment is shown in Fig. 9. The result revealed that after a single R-Conform pass, the tensile strength decreases and the percentage of elongation increases. With a further increase in the number of passes, the values of tensile strength and percentage of elongation both increase and after seven passes, the value of tensile strength is more than that of initial continuous casting and rolling (CSR) materials (from 216.2 to 220.3 MPa). After artificial aging (hardening) at 175 °C for 8 h, the tensile strength of the aged Al–Mg–Si alloys increases significantly up to a value of 276.2 MPa after artificial aging treatment after the fifth Conform pass. Moreover, on increasing the number of passes the Mg_2Si phase gradually dissolves back in the Al matrix during the initial four Conform passes, and is no longer present after the 4th Conform pass. Gazizov *et al.*⁷⁴ studied the effect of equal-channel angular pressing and aging on the microstructure and mechanical properties of Al–Cu–Mg–Si alloys. Ingots were homogenized at 500 °C for 24 h and

subsequently cooled to room temperature in a furnace. Then tensile tests were performed under different conditions of equal-channel angular pressing (ECAP) and aging. The intermediate ECAP is an efficient method that allows one to increase the strength of AA 2014 alloy. After treatment (12 h aging) the yield stress ($\sigma_{0.2}$) and ultimate tensile strength (σ_{UTS}) were ~415 and ~450 MPa, respectively.

4.5 Other methods

Some other methods are also available for the refinement of magnesium and aluminum alloys, such as the physical grain refinement method,⁶⁵ the agitation method,⁵² the Elfinal process (FeCl_3),⁶⁵ carbon inoculation^{52,65} and the powder metallurgy¹⁷ process. However, in these methods insufficient reports are available for refinement of the Mg_2Si phase in magnesium and aluminum alloys. In the carbon inoculation and Elfinal processes, carbon and FeCl_3 are introduced into molten magnesium and aluminum alloys. The addition of carbon and FeCl_3 enhance heterogeneous nucleation by *in situ* precipitation in the alloy melt, forming another compound. The carbon inoculation grain refinement method is only possible in aluminum-containing magnesium alloys (normally with >2% Al).⁷⁵ Qian *et al.* reported that grain refinement of Mg–Al alloys by carbon inoculation stems from the interaction between carbon and aluminium.⁷⁵ The formation of Al_4C_3 particles provides effective nucleants for the carbon inoculation treatment. Carbon inoculation also offers many advantages, such as low operating temperature and less fading with long holding times for application in mass production.⁶⁵ Recent research indicates that the Al, C and Al_2CO_3 particles act as effective nucleants for grain refinement of carbon inoculation. The Elfinal process involves plunging anhydrous FeCl_3 into melts between 740 and 780 °C. This process is effective with and without aluminum-containing magnesium alloys. This process was invented by a German magnesium company “I. G. Farbenindustrie” based on the hypothesis that iron particles could act as nucleation sites for magnesium grains. This also has a lower operating temperature than the superheating method. It provides some economic advantages and it also allows the melt to be held at the pouring temperature for at least an hour without any loss in grain



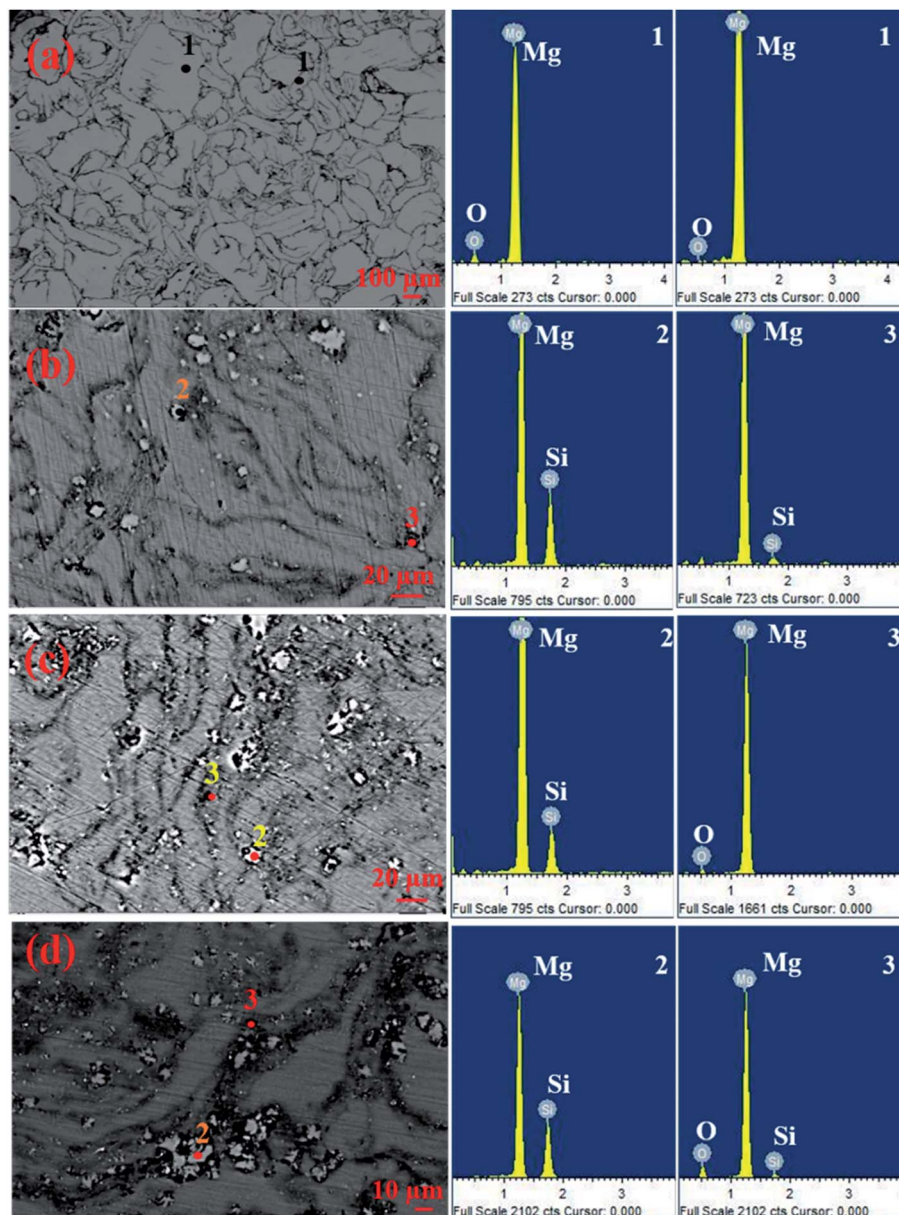


Fig. 10 (a–d) SEM micrographs with EDS spectra: (a) pure Mg, (b) Mg–1Si, (c) Mg–4Si and (d) Mg–6Si. This figure has been reproduced from ref. 17 with permission from Elsevier, copyright 2020.

refinement efficiency.¹⁶ This method can produce a grain refining effect similar to that obtained by superheating.

Recently Seth *et al.* reported that the average grain sizes of Mg_2Si particles in Mg–1Si, Mg–4Si and Mg–6Si alloys synthesized *via* solid-state sintering using high-energy ball milling are 10 μm , 8 μm and 6 μm , respectively.¹⁷ On increasing the amount of silicon, the average grain size of the Mg_2Si phase decreases.

Fig. 10(a–d) show the SEM and EDS spectra at different regions present in the SEM micrographs of pure Mg, Mg–1Si, Mg–4Si, and Mg–6Si alloys synthesized *via* the powder metallurgy route.¹⁷ Each micrograph consists of three different regions: (1) matrix regions, (2) gray particles, (3) dark regions. These regions are indicated by points 1, 2 and 3, respectively. The reported results show that in each micrograph all the gray

particles are present in the grain boundary and are surrounded by dark regions. The results of EDS spectral analysis at point 2 show that the Mg and Si elements are present in the gray particles. The ratio of the intensity of the spectral lines due to magnesium and silicon is equivalent to the stoichiometric ratio of magnesium and silicon in Mg_2Si . This indicates that the gray particles are the Mg_2Si phase. These results show that the formation of the Mg_2Si phase is possible using the powder metallurgy method with a finer grain size.

5. Mechanical properties

The mechanical properties of magnesium and aluminum alloys containing Mg–Si and Mg_2Si are highly satisfactory. These alloys are suitable candidates for different applications. The



Table 3 Tensile properties of Mg–Si alloy with different alloying elements

S. No	Alloy compositions	UTS (MPa)	YS (MPa)	Elongation (%)	Reference no.
1	Mg–0.6Si	166.2 ± 8	60.11 ± 2.27	6.62 ± 0.88	77
2	Mg–0.6Si–0.2Ca	154.4 ± 5.4	50.05 ± 1.13	6.62 ± 0.58	
3	Mg–0.6Si–0.4Ca	156.8 ± 4.9	56.85 ± 0.98	6.22 ± 0.24	
4	Mg–0.6Si–1.5Zn	182.8 ± 6.3	53.72 ± 2.61	14.28 ± 0.64	
5	Mg–0.6Si	164 ± 4	60 ± 2	6 ± 0.5	78
6	Mg–0.6Si–0.5Zn	175 ± 3	60 ± 2	13 ± 1	
7	Mg–0.6Si–1.3Zn	180 ± 2	63 ± 2	14 ± 0.5	
8	Mg–0.6Si–1.5Zn	190 ± 2	62 ± 2	15 ± 0.6	
9	Mg–0.6Si–1.5Zn	181 ± 2	54 ± 2	14.2 ± 1	79 * (T + CC: cyclic closed-die forging at 450 °C)
10	Mg	87 ± 2	—	—	
11	Mg ± 1.5Si	126 ± 1	—	—	
12	Mg ± 3.3 Si	85 ± 1	—	—	
13	Mg (T + CC)	126 ± 1	—	—	
14	Mg ± 1.5Si (T + CC)	132 ± 1	—	—	
15	Mg ± 3.3 Si (T + CC)	142 ± 1	—	—	

Table 4 Tensile properties of Al–Mg–Si alloys with different alloying elements

Alloy compositions	Yield strength (MPa)	Elongation (%)	Hardness (BHN)	Grain size (μm)	Reference no.
Al–0.35Mg–7Si	174.31	2.67	72.3	200–300	80
Al–0.35Mg–7Si–0.6Ba	178.51	3.76	72.3	80–150	
Al–0.35Mg–7Si–1Ba	198.04	4.06	73.5	30–80	
7Al–0.35Mg–7Si–1.4Ba	171.74	2.92	71.0	100–200	
7Al–0.35Mg–7Si–1.8Ba	167.55	2.07	70.2	—	
Al–Mg–Si longitudinal	102 ± 1	—	—	—	81
Al–Mg–Si transverse	99 ± 4	—	—	—	

Table 5 Tensile properties of Mg–Al–Si (AS series) of alloys under different conditions

Alloy compositions	Yield strength (MPa)	Elongation (%)	UTS (MPa)	Hardness (BHN)	Grain size (μm)	Reference no.
AS52 (ambient)	58 ± 3	13.75 ± 0.3	165 ± 3	57 ± 1	—	82
AS52 (at 150 °C)	50 ± 3	15 ± 0.3	150 ± 2		—	
AS31 (ambient)	58.9	6.4	95.6		—	83
AS31 (at 150 °C)	65.5	6.8	87.3			

specific strength and specific stiffness of materials are important for the design of lightweight components. At present, applications of high-strength lightweight components are increasing enormously in automotive and aerospace industries to save weight. Weight saving is particularly important for automotive bodies, components and other products, where energy consumption and power limitations are major concerns.⁷⁶ There are large numbers of reports available on the mechanical properties of aluminum and magnesium alloys containing the Mg₂Si phase. In general, mechanical properties are evaluated by tensile, compression and hardness tests. Some significant tensile properties of Mg–Si alloys and higher alloys containing the Mg₂Si phase are given in Tables 3–5.

5.1 Tensile properties

The yield strength of Mg–Si alloys and those with other alloying elements vary from 50 ± 1 MPa to 60 ± 1 MPa. The ultimate

tensile strength (UTS), however, varies over a wider range; the Mg–0.6Si–1.5Zn alloy shows higher tensile properties.^{77,78} The UTS of Mg–0.6Si, Mg–1.5Si, and Mg–3.3Si alloys are 166.2 ± 8.0 MPa, 126 ± 1 MPa and 85 ± 1 MPa, respectively. The UTS of Mg–Si alloys decreases with an increase in the silicon content. However, after secondary process (cyclic closed-die forging at 450 °C), the UTS of Mg–1.5Si and Mg–3.3Si alloys are 132 ± 1 MPa and 142 ± 1 MPa, respectively. The enhancement in UTS of Mg–1.5Si and Mg–3.3Si alloys is due to the refinement of the Mg₂Si phase during the secondary processes (Table 3).

Al–Mg–Si based aluminum alloys show better tensile properties than the Mg–Al–Si (AS series) of alloys (Table 4 and 5). The yield strength of Al–0.35Mg–7Si alloys increases on addition of alloying elements. The addition of up to 1 wt% of barium enhances the yield strength of Al–0.35Mg–7Si, as barium works as a grain modifier up to 1 wt%. The tensile properties of Al–Mg–Si also depend on the rolling direction. Test specimens cut



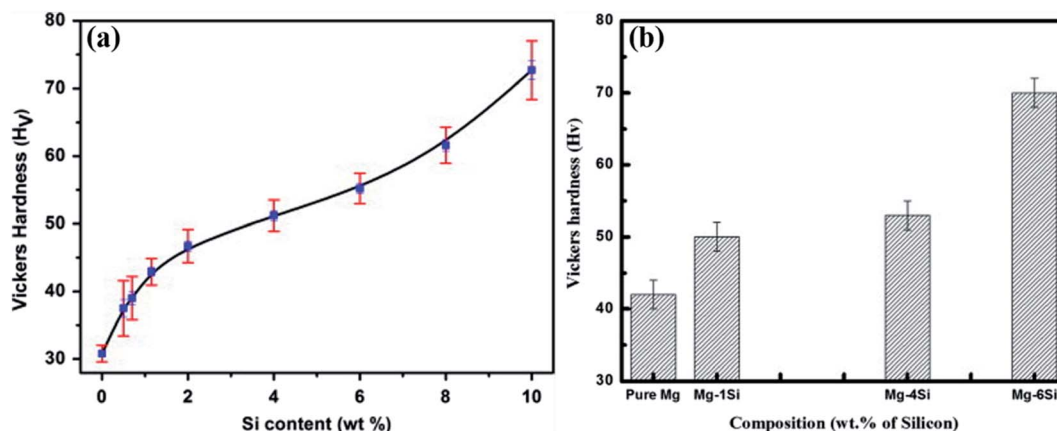


Fig. 11 Hardness behavior of Mg–Si alloys. The specimens are synthesized by: (a) casting process. This figure has been reproduced from ref. 5 with permission from Elsevier, copyright 2013. (b) Powder metallurgy process. This figure has been reproduced from ref. 17 with permission from Elsevier, copyright 2020.

Table 6 Hardness behavior of Mg–Al–Si alloys

Hardness (HV) of Mg–Al–Si (AS series) alloys ²⁹		
1	AS11	41 ± 1
2	AS21	48 ± 2
3	AS41	52 ± 1
4	AS61	54 ± 1
5	AS91	62 ± 1

in the longitudinal direction show better properties than specimens cut in the transverse direction in Al–Mg–Si alloys.

5.2 Hardness behavior

Fig. 11(a) and (b) show the bulk hardness behaviour of Mg–Si alloys, where the specimens are synthesised *via* the casting process (CP) and the powder metallurgy process (PM) using high-energy ball milling, respectively. The specimens prepared *via* the powder metallurgy process show higher hardness values for the same composition. The hardness values (Hv) of Mg–6Si alloys synthesized *via* the casting and powder metallurgy processes are 55 ± 2 and 72 ± 2, respectively. The specimens synthesized *via* the powder metallurgy process show smaller average grain size of the Mg₂Si phase.¹⁷ Therefore, pinning of dislocations is greater at Mg₂Si precipitate particles in specimens synthesized *via* the powder metallurgy process. Therefore,

specimens prepared by the powder metallurgy route show higher hardness values.

The addition of aluminum also improves the hardness of Mg–Si alloys. Aluminum reacts with magnesium and forms another intermetallic compound, Mg₁₇Al₁₂ (β -phase). The quantity of β -phase increases due to an increase in the amount of aluminum. The β -phase and Mg₂Si phase could be very effective in strengthening magnesium alloys at elevated and room temperatures.²⁹ Therefore, AS11 (Mg–1Al–1Si) shows lower hardness and AS91 (Mg–9Al–1Si) alloy shows higher hardness in the AS series of alloys (Table 6).

Al–Mg–Si alloys have good strength and formability. Therefore, numerous attempts have been made to improve the strength and ductility of these alloys while maintaining their formability. Artificial aging is the one of the best methods to improve the strength and ductility of these alloys.⁸⁴ Generally artificial aging was done on heat-treatable (Al–Mg–Si) alloys. During the artificial aging process, the supersaturated solid solution present in alloys decomposes into fine precipitates that strengthen the alloys by acting as obstacles to the movement of dislocations. The nano-scale dispersive precipitates increase both the strength and the ductility of the alloys.⁸⁵ Hu *et al.*⁸⁴ reported the Vickers microhardness of Al–Mg–Si alloys with nano-scale precipitates after repetitive continuous extrusion forming (R-Conform passes) and T8 tempering (aging at 120, 155 and 175 °C for various times up to 8 h). Table 7 shows the results of the maximum micro-hardness value with respect to

Table 7 Maximum Vickers micro-hardness values of Al–Mg–Si alloys with respect to aging time of four different mechanical working conditions.⁸⁴

Aging temperatures	Maximum micro-hardness with respect to aging time			
	Direct drawing	1R-Conform passes and drawing	4R-Conform passes and drawing	7R-Conform passes and drawing
RT	92.5 ± 0.2	91.5 ± 0.2	106 ± 0.2	109 ± 0.2
120 °C	95.2 ± 0.2 (2 h)	103 ± 1 (1 h)	108 ± 0.4 (3 h)	112 ± 0.5 (3 h)
155 °C	93 ± 0.4 (2 h)	96.5 ± 0.5 (1 h)	105 ± 0.2 (3 h)	107 ± 0.2 (2 h)
175 °C	85 ± 0.6 (1 h)	96 ± 0.3 (2 h)	96.5 ± 0.4 (1 h)	97 ± 0.5 (1 h)



aging time of four different conditions: direct drawing, 1R-Conform passes and drawing, 4R-Conform passes and drawing, and 7R-Conform passes and drawing. All specimens are aged at 120, 155 and 175 °C for various durations up to 8 hours (1 hour intervals).

The results of Table 7 reveal that the hardness of specimens with and without aging increased with increasing number of R-Conform passes. The specimens aged at 120 °C show higher hardness values under all conditions. The specimens with 7R-Conform passes and drawing and aged at 120 °C for 3 hours show maximum micro-hardness values among all the specimens.

The hardness behavior and other mechanical properties of Mg-Si, Mg-Si-Al and Al-Mg-Si alloys are due to the precipitation behavior and decomposition of a supersaturated solid solution of fine particles during the synthesis process, secondary mechanical working (severe plastic deformation, cold and hot extrusion) processes and artificial aging under different conditions (temperatures and holding time).

Mg-Si, Mg-Si-Al and Al-Mg-Si alloys show a wide range of mechanical properties. Therefore, they have the ability to be used in different applications according to the properties required. Generally Mg-Si and Mg-Si-Al alloys are used for manufacturing components which are used in high-temperature applications (higher creep resistance).⁸⁶ The formability of Al-Mg-Si alloys is high. Therefore, these alloys are used to form sheets and wires. The applications of these alloys are briefly explained in the next section.

6. Applications based on properties

Magnesium and aluminum alloys containing an Mg₂Si phase cover a wide area of applications. Generally, magnesium-based

alloys (Mg-Si and Mg-Si-Al) are used in automotive, aerospace, biomedical, and electronics applications. However, aluminum-based alloys (Al-Mg-Si) are used in automotive, aerospace and, electrical applications. These alloys have a variety of miscellaneous advantageous properties: *e.g.* light weight, high specific strength, good castability, low elastic modulus, recyclability, high corrosion resistance, thermal stability, low thermal conductivity, higher electrical conductivity, and low toxicity.⁸⁷ Based on these properties, Mg-Si and Mg-Si-Al alloys are heavily used in biomedical applications for transplantation into human bones and teeth. Moreover, these alloys are used in automotive and aerospace applications for body and engine components which are working at high temperature. These alloys are also used in electronic applications.⁸⁸ Al-Mg-Si alloys are used for making single-shape relatively complex architectural forms and body-cover parts of automobiles and aerospace vehicles. These alloys have the ability to withstand maximum tensile and compressive stresses. This characteristic is particularly important for structural components where stiffness is critically important. These alloys are also used in electrical applications for wire manufacturing.

6.1 Application of magnesium alloys containing an Mg₂Si phase

Mg-Si and Mg-Si-Al alloys have outstanding characteristics, including higher creep resistances, excellent damping performance, high stiffness-weight ratio and they are highly biocompatible for the human body. Due to their mechanical properties and biocompatibility, these alloys are heavily used in biomedical applications. Aluminum and silicon ions are also present in the human body and they are involved in many metabolic reactions and biological mechanisms. These alloys

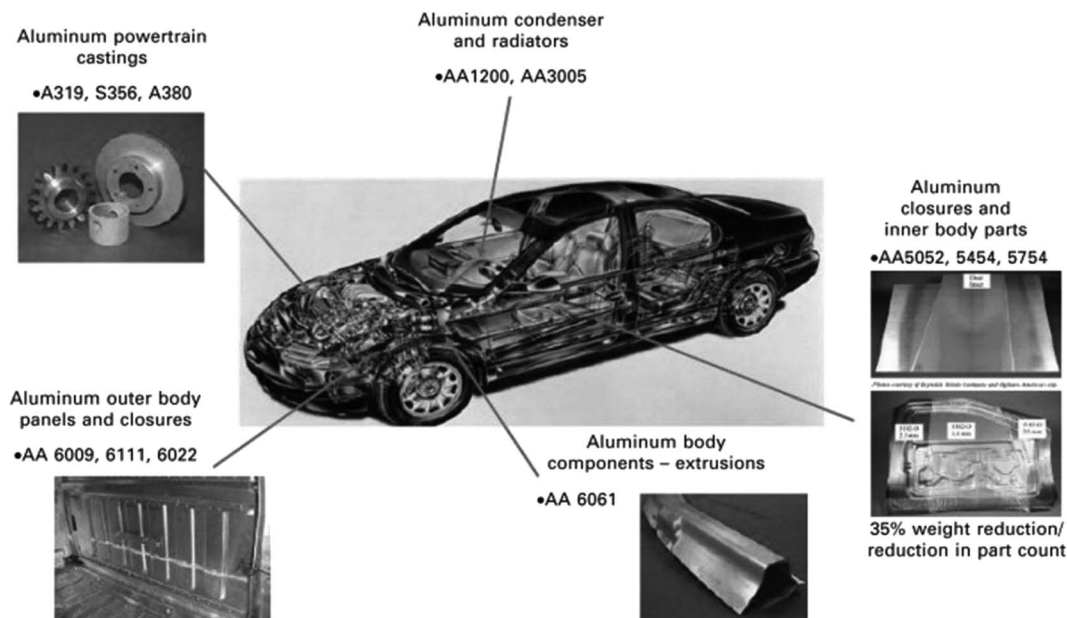


Fig. 12 Application of different series of aluminum alloys in automotive product forms. This figure has been reproduced from ref. 92 with permission from Elsevier, copyright 2010.

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