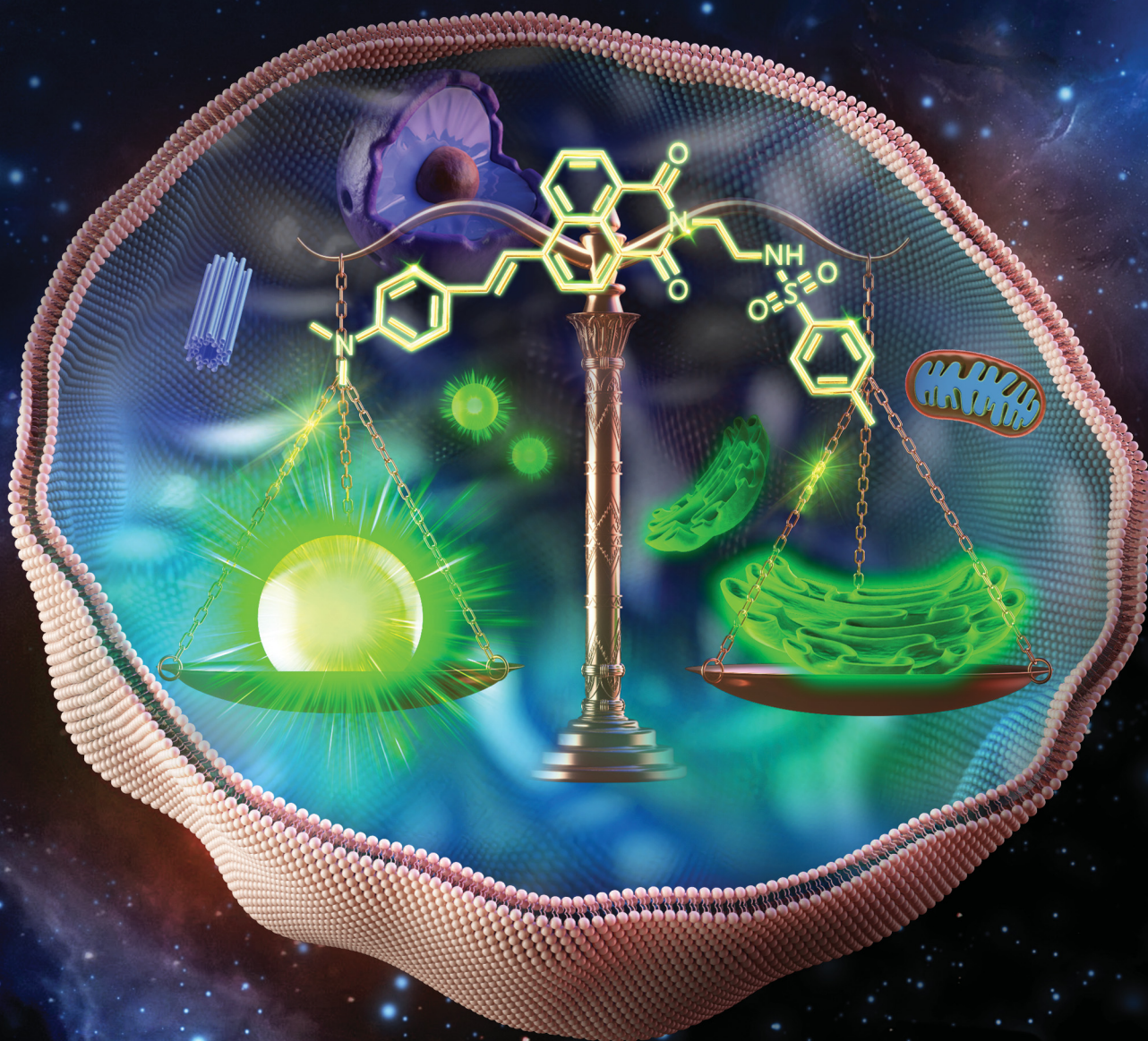


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**PAPER**

Peng Gao, Xiaoqiang Yu *et al.*

A dual-targeting fluorescent probe for simultaneous and discriminative visualization of lipid droplets and endoplasmic reticulum



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## A dual-targeting fluorescent probe for simultaneous and discriminative visualization of lipid droplets and endoplasmic reticulum†

Fangfang Meng,<sup>a,c</sup> Junyi He,<sup>a</sup> Jie Niu,<sup>b</sup> Yawen Li,<sup>a</sup> Peng Gao<sup>\*ac</sup> and Xiaoqiang Yu<sup>ib\*</sup>

A single fluorescent probe (SF-probe) that can simultaneously and discriminatively visualize two organelles is a powerful tool to investigate their interaction in cellular processes. However, it is still challenging to develop this unique type of fluorescent probe due to the lack of a feasible design strategy. Herein, we proposed a dual-targeting group strategy to construct SF-probes by integrating two different organelle-targeting groups into the same fluorophore. A versatile fluorophore and two nonintrusive organelle-targeting groups are elements of this strategy. In view of only a few SF-probes having been developed for the simultaneous and discriminative imaging of lipid droplets (LDs) and endoplasmic reticulum (ER), as a proof of concept, a SF-probe, **LDER**, was designed and synthesized by introducing an LD targeting group and an ER targeting group onto the 1,8-naphthalimide fluorophore. Owing to the specific structure of the fluorophore, both targeting groups at two terminals of 1,8-naphthalimide can fully play their respective roles without mutual interference. Furthermore, the ability of the two groups to target their respective targets is comparable, enabling **LDER** to bind LDs and ER evenly. Meanwhile, **LDER** is very susceptible to polarity, which is advantageous for the discriminative imaging of LDs and ER. In addition, the interaction between LDs and ER was investigated.

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### Introduction

Organelles, as fundamental subunits in eukaryotic cells, play indispensable roles in complex biological processes.<sup>1–3</sup> They not only have unique functionalities but also cooperate with each other to form organelle interaction networks.<sup>4–6</sup> For example, lipid droplets (LDs) are actively involved in regulating lipid metabolism and cellular homeostasis.<sup>7</sup> Endoplasmic reticulum (ER) has significant roles in protein synthesis, secretion, and signal transduction.<sup>8</sup> Lysosomes are responsible for endocytic, autophagic, and phagocytic pathways.<sup>9</sup> Furthermore, LDs and ER cooperate with each other to complete complicated biological procedures.<sup>10,11</sup> Particularly, it is widely believed that

LDs derive from ER.<sup>12,13</sup> At the same time, LDs are degraded through lysosomes, which is known as lipophagy.<sup>14,15</sup> To visualize two organelles and their associated interaction, a strategy of utilizing two individual organelle trackers has often been employed. However, the unsatisfying spectral cross-talk, increased cytotoxicity and tedious operating procedures limit their application in living cells.<sup>16</sup> Therefore, it is of great significance to develop single fluorescent probes (SF-probes) to simultaneously and discriminatively visualize two organelles and their associated interaction. However, such a fluorescent probe or feasible design strategy has rarely been reported.

Generally, an organelle-targeting fluorescent probe contains two components: a fluorophore and a targeting group.<sup>17</sup> It should be noted that many organelles have a special targeting group<sup>18,19</sup> (Scheme 1A). For example, LDs contain a hydrophobic lipid core covered by a phospholipid-monolayer membrane, thus probes with lipophilic groups such as alkylamino have been used to image LDs.<sup>20</sup> ER is the largest membrane organelle in eukaryotic cells, and is where the ATP-sensitive K<sup>+</sup> channel is mainly localized, thus methyl sulphonamide binding to the sulfonylurea receptor in the K<sup>+</sup> channel can be used to construct ER probes.<sup>21</sup> Mitochondria are organelles with a negative membrane potential of −180 mV, therefore Triphenylphosphonium (TPP<sup>+</sup>) with a cationic unit has been applied to

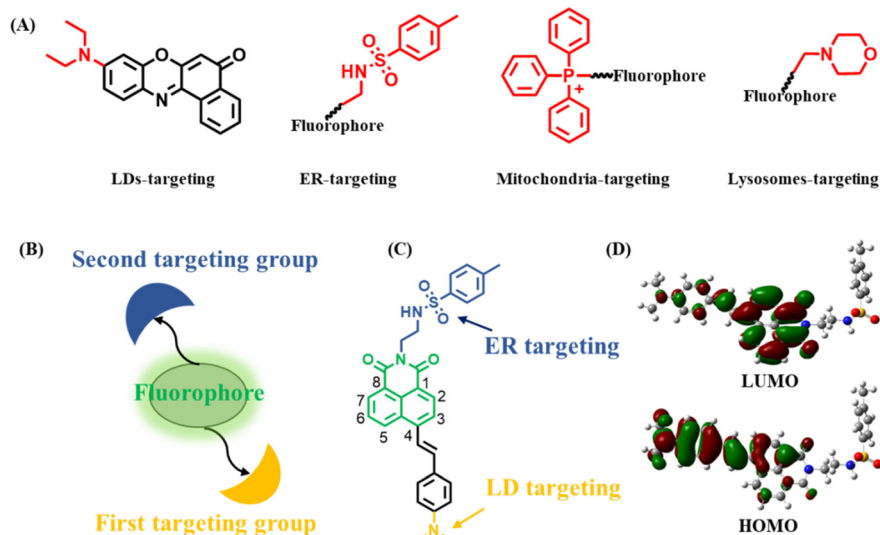
<sup>a</sup> Department of Pathology, Qilu Hospital, Cheeloo College of Medicine, Shandong University, Jinan, Shandong, China. E-mail: gaopeng@sdu.edu.cn

<sup>b</sup> Center of Bio & Micro/Nano Functional Materials, State Key Laboratory of Crystal Materials, Shandong University, Jinan, Shandong, China.  
E-mail: yuxq@sdu.edu.cn

<sup>c</sup> Key Laboratory for Experimental Teratology of Ministry of Education, Department of Pathology, School of Basic Medical Sciences, Shandong University, Jinan, Shandong, China

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**Scheme 1** (A) Various targeting units used to design organelle-targeting fluorescent probes; (B) potential dual-targeting model; (C) structure of the probe **LDER**; (D) the frontier orbitals of **LDER**.

target mitochondria.<sup>22</sup> Lysosomes have an acidic environment, thus morpholine being weakly alkaline has often been used to design lysosomal probes.<sup>23</sup> These key facts provide the possibility that a SF-probe for simultaneous and discriminative visualization of two organelles can be developed by exquisitely incorporating two different organelle-targeting groups into the same fluorophore (Scheme 1B). Obviously, the elements of this strategy should include: (1) a fluorophore ring that can be substituted at two terminals by two organelle-targeting groups. This will decrease the steric hindrance of the two organelle-targeting groups, making full use of their respective roles without mutual interference. (2) Organelle-targeting groups that have balanced targeting abilities towards two organelles. In this way, the probe can be evenly distributed among these two organelles. (3) High specificity of the probe. The constructed probe should be of high specificity towards these two organelles without interacting with others.

It is well known that LDs and ER are two organelles with similar polarity but different functions and there are multiple interactions between them,<sup>24</sup> so their discriminative visualization is extremely valuable yet challenging. Also, as mentioned above, SF-probes capable of discriminative imaging of LDs and ER have rarely been reported.<sup>16,25,26</sup> Therefore, as a proof-of-concept, we firstly applied this strategy to construct a SF-probe for simultaneous and discriminative visualization of LDs and ER. In view of the versatility, easy availability and substitution potential at the 3, 4, 5, 6 or N-imide position of 1,8-naphthalimide, it was selected as the basic building block.<sup>27–29</sup> Dimethylamino and methyl sulphonamide moieties served as the LD- and ER-targeting groups owing to their similar targeting ability towards LDs and ER.<sup>30,31</sup> By incorporating dimethylamino and methyl sulphonamide moieties at the 4-position and N-imide position of the 1,8-naphthalimide fluorophore, the probe **LDER** was developed (Scheme 1C). On the one hand, the two targeting groups were placed at opposite ends of 1,8-naphthalimide, which ensured that

there was no interference caused by each other. On the other hand, these groups have the same ability to target their respective organelles, thus the probe can target LDs and ER evenly. Delightfully, as a strong electron donor, the dimethylamino group enhanced the intramolecular charge transfer (ICT) character of the fluorophore, resulting in a polarity sensitivity of this probe. The experimental results showed that the probe **LDER** could precisely and evenly target LDs and ER simultaneously and discriminatively visualize them by their morphological differences. Meanwhile, **LDER** displayed high sensitivity to slight variations in polarity, enabling discriminative visualization of LDs and ER with different fluorescence. In addition, the dynamics of LDs and ER and their interaction in living cells were also investigated using **LDER**. The results demonstrated that this strategy has great potential in the design of SF-probes for simultaneous and discriminative visualization of two organelles.

## Results and discussion

### Synthesis and characterization of **LDER**

The designed fluorescent probe **LDER** was synthesized according to the synthesis route shown in Scheme S1 (ESI<sup>†</sup>), and the synthetic procedures are included in the ESI.<sup>†</sup> The chemical structure of **LDER** was characterized using <sup>1</sup>H NMR, <sup>13</sup>C NMR and HRMS spectra and they are presented in the ESI.<sup>†</sup>

To confirm the ICT property of the probe, the frontier orbitals of **LDER** were calculated by Gaussian 09 software (Scheme 1D). The highest occupied molecular orbital (HOMO) of **LDER** is mainly distributed on the dimethylamino moiety, while the lowest unoccupied molecular orbital (LUMO) is mainly distributed on the 1,8-naphthalimide moiety. This result demonstrated that **LDER** may undergo a charge transfer process from dimethylamino to the 1,8-naphthalimide moiety

owing to its typical D- $\pi$ -A structural feature. Probes with such D- $\pi$ -A structures are sensitive to solvent polarity, whereby their photophysical properties, particularly their emission behaviors, change upon varying the polarity. Thus, we think that the probe **LDER** has the potential to show polarity-sensitive emission behavior.

### Optical properties of LDER

The absorption and emission spectra of **LDER** in various solvents were first recorded. As displayed in Fig. 1A, the absorption of **LDER** was in the range of 400–550 nm, and peaked at around 470 nm in the solvents with different polarity. In addition, the absorption spectra showed a slight red shift as the solvent polarity increased (Table S1, ESI<sup>†</sup>). On the other hand, under excitation at 470 nm, the fluorescence spectra of **LDER** displayed an obvious solvation effect (Fig. 1B and Table S1, ESI<sup>†</sup>). **LDER** showed an emission peak at 586 nm in toluene with low polarity. In contrast, in a high-polarity solvent, such as DMSO, **LDER** displayed a red-shifted emission peak at 646 nm. Simultaneously, a pronounced decrease in fluorescence intensity and fluorescence quantum yield with solvent polarity enhancement was observed. These results demonstrated that the emission of **LDER** was sensitive to solvent polarity.

To evaluate the polarity-sensitive emission behavior of **LDER**, the change in maximum emission wavelength with the solvent polarity parameter  $E_T(30)$  value, a solvent polarity parameter that indicates solute-solvent interactions, was plotted.<sup>32</sup> As  $E_T(30)$  increased, the maximum emission wavelength of **LDER** increased linearly (Fig. 1C); such a good relationship indicated that **LDER** has a twisted ICT (TICT) character.<sup>33–35</sup> The fluorescence intensity of **LDER** with dielectric constant ( $\epsilon$ ) of various media was also correlated, as shown in Fig. 1D. The fluorescence intensity varies inversely with the solvent dielectric constant ( $\epsilon$ ), which implies the polarity sensitivity of **LDER**.

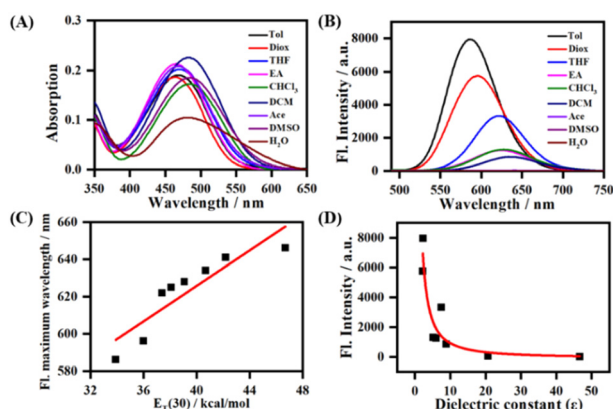


Fig. 1 The absorption (A) and fluorescence (B) spectra of **LDER** in different polarity solvents; (C) plot of the maximum wavelength of **LDER** with  $E_T(30)$ ; (D) fluorescence intensity change of **LDER** versus dielectric constant ( $\epsilon$ ). Concentration = 10  $\mu$ M. Solvents: Tol: toluene; Diox: 1,4-dioxane; THF: tetrahydrofuran; EA: ethyl acetate;  $\text{CHCl}_3$ : trichloromethane; DCM: dichloromethane; Ace: acetone; DMSO: dimethyl sulfoxide;  $\text{H}_2\text{O}$ : water.

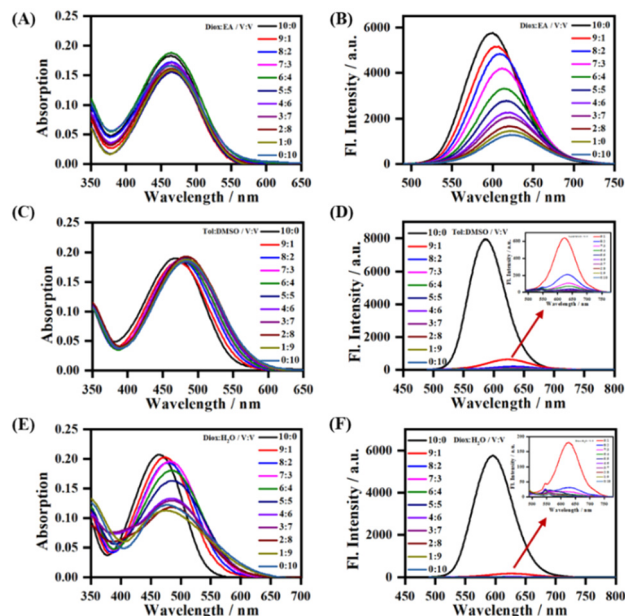


Fig. 2 The absorption (A, C and E) and fluorescence (B, D and F) spectra of **LDER** in 1,4-dioxane-EA, toluene-DMSO and 1,4-dioxane- $\text{H}_2\text{O}$  mixtures with different volume ratios; inset: the enlarged part indicated by the red arrow. Concentration = 10  $\mu$ M.

To further study the sensitivity of **LDER** towards polarity, the absorption and emission spectra of the probe in mixed solvents with different polarity were recorded (Fig. 2). In mixtures of 1,4-dioxane and ethyl acetate (EA) with similarly low polarity, **LDER** exhibited tiny variations in absorption (Fig. 2A). As the EA content increased, the maximum emission wavelength of **LDER** red-shifted. Simultaneously, the fluorescence intensity decreased, showing a clear function of the solvent ratio (Fig. 2B). Such spectral changes are consistent with the TICT character. Furthermore, the absorption and emission of **LDER** were also recorded in a binary solvent system of toluene-DMSO and 1,4-dioxane- $\text{H}_2\text{O}$ . As the polar solvent (DMSO or  $\text{H}_2\text{O}$ ) content increased, a pronounced decrease in the fluorescence intensity was observed (Fig. 2D and F). These results indicated that the emission of the probe **LDER** was highly sensitive to polarity, which provided the potential for *in situ* imaging of organelles with polarity differences in living cells.

In addition to polarity, the effects of different pH and viscosity on the probe **LDER** were studied. As shown in Fig. S1A (ESI<sup>†</sup>), in a wide pH range from 2 to 12, the fluorescence intensity did not change. Meanwhile, the fluorescence intensity remained stable at various viscosity levels (Fig. S1B, ESI<sup>†</sup>). These results indicated that the probe **LDER** was only sensitive to polarity.

### Precise subcellular localization of LDER in living cells

Before exploring the application of **LDER** in living cell imaging, the cytotoxicity of the probe was evaluated. As shown in Fig. S2 (ESI<sup>†</sup>), **LDER** displayed no significant toxicity within the concentration range of 0–20  $\mu$ M. This result indicated that **LDER** was favorable for bioimaging applications.

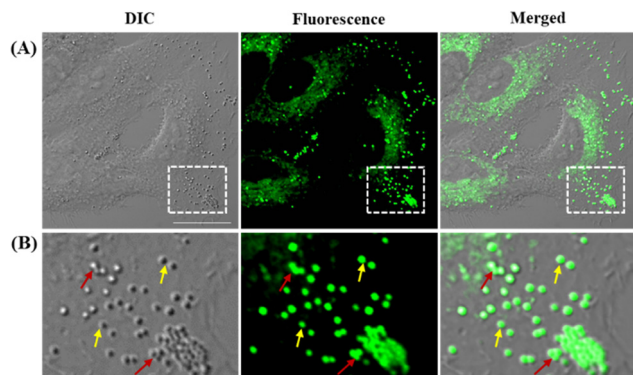


Fig. 3 (A) Fluorescence images of living SiHa cells stained with 5  $\mu$ M LDER; (B) enlarged fluorescence images of the regions of interest (ROIs) indicated by the white rectangles in (A).  $\lambda_{\text{ex}} = 473$  nm,  $\lambda_{\text{em}} = 550$ –650 nm; Scale bar = 20  $\mu$ m.

Based on our design strategy, the probe **LDER** could simultaneously image LDs and ER in living cells. To verify the cellular localization of **LDER**, fluorescence imaging in living SiHa cells was carried out. As shown in Fig. 3A, spherical morphologies with bright green fluorescence and reticulate morphologies with weak green fluorescence were detected in the cytoplasm. Moreover, the two signals were not located in the same area, which indicated that **LDER** can stain two subcellular structures. We expected that the bright fluorescence was derived from LDs. There are two reasons for this: first, from the magnified images (Fig. 3B), the bright fluorescence from **LDER** appeared as spherical “droplets”. Some droplets are visible as individuals (yellow arrows), whereas some aggregate into large droplets (red arrows). Although the distribution, morphology and size of these droplets stained by **LDER** were different, they all well overlapped with the globular organelles in the DIC image. It is generally accepted that LDs present as globular spots in DIC owing to their high refractive index in relation to other subcellular structures.<sup>16,36</sup> Thus, the bright dots shown in Fig. 3 should be LDs. Second, BODIPY 493/503, a commercially available probe for LD targeting, was used to co-stain with **LDER**. As shown in Fig. S3 (ESI<sup>†</sup>), the strong green fluorescence from **LDER** overlapped well with the red fluorescence from BODIPY 493/503, and the co-localization coefficient was around 0.86. This result demonstrated that the bright spherical dots stained by **LDER** were indeed LDs. The above results indicated that the probe **LDER** could stain both LDs and another subcellular structure.

To reveal the specific localization of **LDER** in the other subcellular region, co-localization experiments were carried out. A co-localization experiment was first performed with a commercially available ER probe, ER-Tracker Blue. As displayed in Fig. 4, we could see that the weak green fluorescence from **LDER** overlapped well with the blue fluorescence of ER-Tracker Blue. To further analyse the imaging results, we randomly magnified three small parts of the merged image. As shown in the enlarged images in Fig. 4 and Fig. S4 (ESI<sup>†</sup>), the co-localization coefficients were determined to be 0.89, 0.87 and 0.90, respectively. This result indicated that the reticulate area stained by **LDER** was ER.

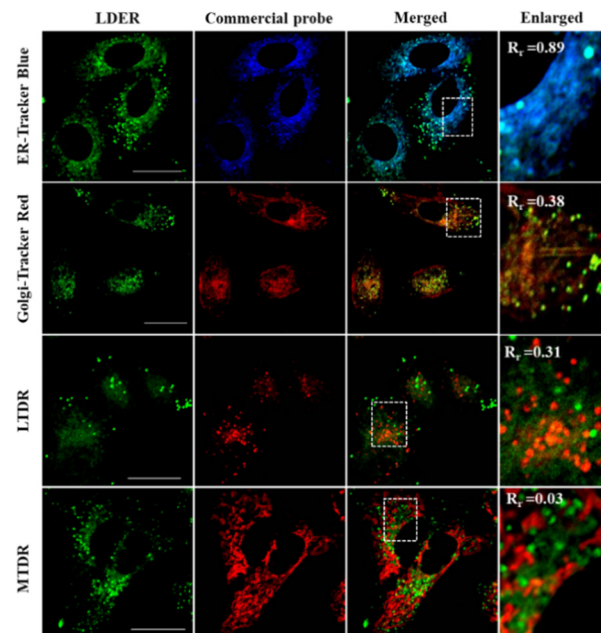


Fig. 4 Fluorescence images of **LDER** co-stained with various commercial probes (ER-Tracker Blue, Golgi-Tracker Red, LTDR and MTDR) in living SiHa cells. **LDER**:  $\lambda_{\text{ex}} = 473$  nm,  $\lambda_{\text{em}} = 560$ –650 nm; ER-Tracker Blue:  $\lambda_{\text{ex}} = 405$  nm,  $\lambda_{\text{em}} = 450$ –500 nm; Golgi-Tracker Red:  $\lambda_{\text{ex}} = 543$  nm,  $\lambda_{\text{em}} = 600$ –700 nm; LTDR:  $\lambda_{\text{ex}} = 633$  nm,  $\lambda_{\text{em}} = 650$ –750 nm; MTDR:  $\lambda_{\text{ex}} = 633$  nm,  $\lambda_{\text{em}} = 650$ –750 nm. Scale bar = 20  $\mu$ m.

Besides this, co-localization experiments of **LDER** and other organelle markers (Golgi-Tracker Red, lysosomal deep red fluorescence marker (LTDR) and mitochondrial deep red fluorescence marker (MTDR)) were also carried out. As shown in Fig. 4 and Fig. S4 (ESI<sup>†</sup>), the overlap of **LDER** and Golgi-Tracker Red, LTDR and MTDR was not high, and the co-localization coefficients were 0.43, 0.47 and 0.03 on average, respectively. Taken together, the results show that the probe **LDER** can simultaneously image LDs and ER, which was consistent with our design.

#### Discriminative imaging of LDs and ER in living cells

As described above, the probe **LDER** achieved the simultaneous visualization of LDs and ER in living SiHa cells, and could discriminate between them by morphological differences. Since **LDER** exhibited polarity-sensitive emission behavior and LDs and ER have different polarity, we suspected that the probe **LDER** could also discriminate between LDs and ER with different fluorescence.

As shown in Fig. 5A, LDs and ER could be clearly seen in the **LDER**-stained SiHa cells, and these zones exhibited very different fluorescence intensities. According to the *in situ* fluorescence spectra, the LD- and ER-like regions were different (Fig. 5B). The emissive peak of **LDER** in the LDs was located at 580 nm, while it was at 600 nm in ER. Relative to ER, the LDs constitute more hydrophobic microenvironments, and these zones displayed high fluorescence intensity that was similar to those of **LDER** in solvents with low polarity. In contrast, the less hydrophobic microenvironments of ER translated into a weak fluorescence intensity for **LDER**, which was similar to its



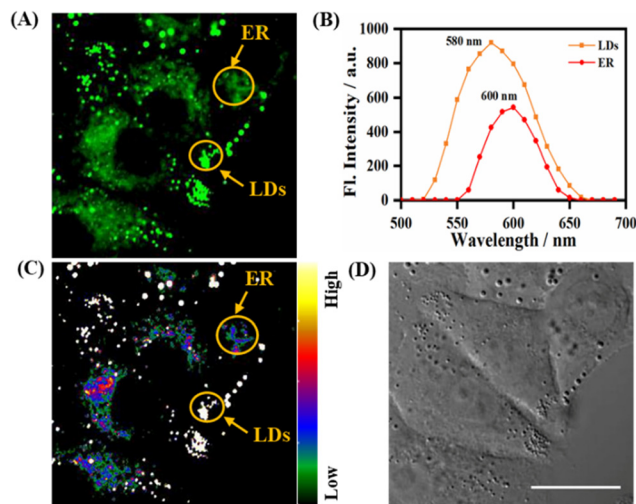


Fig. 5 (A) Fluorescence images of LDER in living SiHa cells; (B) *in situ* emission spectra of LDs and ER in (A); (C) the relative fluorescence image of (A), with a rainbow of color representing the fluorescence intensity; (D) DIC image. Scale bar = 20  $\mu\text{m}$ .

fluorescence in high polarity solvents. This result was in good accordance with the photophysical data recorded in solution (Fig. 1B).

In order to intuitively discriminate between LDs and ER, the conversion of different fluorescence intensities to different colors was achieved by automatic calculation by the microscopy software. With the aid of a rainbow of color, we could easily discern LDs and ER, as the LDs exhibited a higher emission intensity than the ER (Fig. 5C). The high fluorescence intensity of the LDs was mainly due to their low polarity, while the low fluorescence intensity of the ER was due to its high polarity. These differences demonstrated that the polarity-sensitive probe **LDER** may be used to discriminate between LDs and ER by fluorescence intensity.

To verify the universality of the staining pattern, fluorescence images were recorded in Human Cervical Carcinoma cells (HeLa cells) and Human Breast Cancer cells (MCF-7 cells) (Fig. S5, ESI<sup>†</sup>). It can be easily seen that **LDER** emitted different fluorescence in LDs and ER in these cell lines. The above results confirmed that the staining pattern of **LDER** is universal across different cell lines, and it can be used as a universal tool for the simultaneous and discriminative imaging of LDs and ER.

#### Visualization of the changes of LDs and ER during starvation and oleic acid treatment

Since **LDER** can simultaneously and discriminatively visualize LDs and ER, the changes in LDs and ER in reaction to different stimuli were studied. As dynamic lipid storage depots, LDs are highly responsive to the cellular metabolic state. During starvation, LD-associated neutral lipid lipases mediate the breakdown of LDs, leading to a decrease of LDs.<sup>37</sup> Living SiHa cells were starved in nutrient-free medium for different times and then incubated with **LDER** to visualize the dynamic changes of LDs and ER. As shown in Fig. 6A, in the untreated cells, **LDER** emitted strong fluorescence in LDs and weak fluorescence in

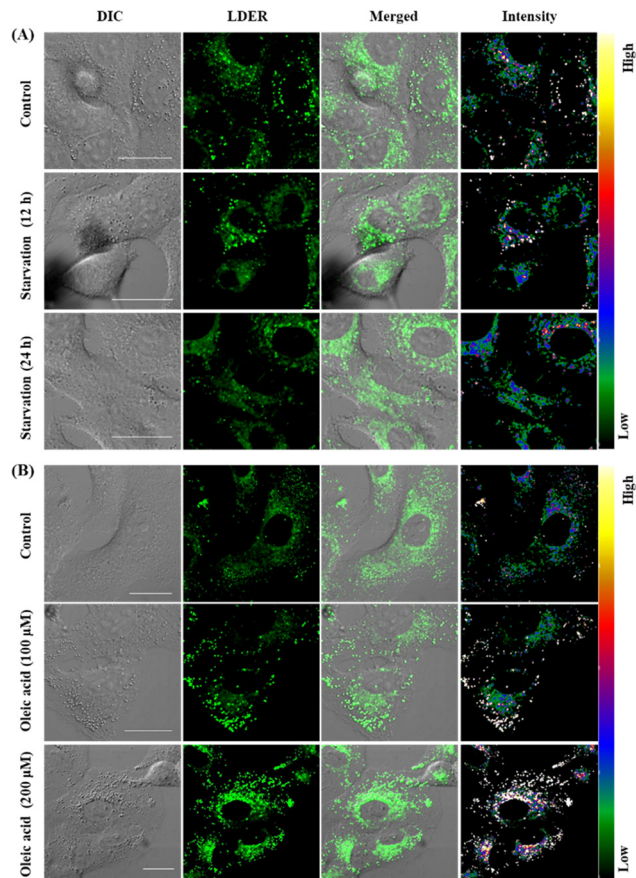


Fig. 6 (A) Fluorescence images of SiHa cells incubated in nutrient-free medium for different times, followed by staining with **LDER** (5  $\mu\text{M}$ ); (B) fluorescence images of SiHa cells stained with **LDER** (5  $\mu\text{M}$ ) after the cells were pretreated with different concentrations of oleic acid for 4 h. The rainbow of color represents the fluorescence intensity. Scale bar = 20  $\mu\text{m}$ .

ER. After the cells were starved, the fluorescence of **LDER** in LDs decreased obviously, indicating the consumption of LDs during starvation. Meanwhile, no obvious changes of the fluorescence in ER were observed. This result suggested that the number of LDs decreased sharply with starvation time.

Oleic acid (OA) is known to stimulate the production of LDs in cells and thus we examined the dynamic changes of LDs and ER in SiHa cells before and after addition of OA.<sup>38</sup> After staining with **LDER**, the LDs and ER in the control group could be seen with strong and weak fluorescence, respectively (Fig. 6B). The LDs were comparatively small and dispersed in the cytoplasm. Upon treatment with OA, the emission of the LDs enhanced evidently, and the number and size of the LDs increased as well. This result indicated that **LDER** could monitor OA-stimulated LD production. These results verified that **LDER** can be used for imaging the dynamic changes of LDs and ER during various biological procedures within living cells.

#### Visualization of the interaction between LDs and ER

LDs and ER are dynamic organelles and their interaction is closely related to cellular metabolism.<sup>39–41</sup> To visualize their interaction, the following experiments were carried out.

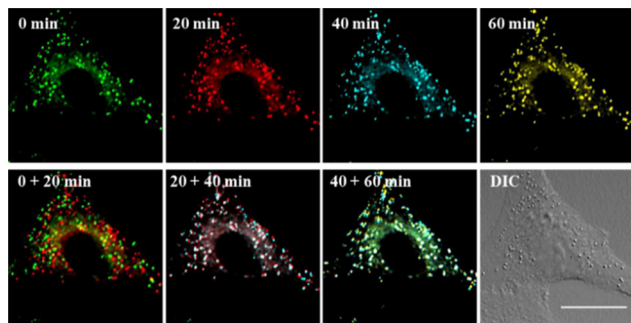


Fig. 7 Fluorescence images of living SiHa cells stained with **LDER**. Different pseudocolors are used to represent the fluorescence images at times of 0, 20, 40, and 60 min. Merged images at two different times, DIC image obtained at 0 min. Scale bar = 20  $\mu$ m.

Photostability is an important factor for a probe used for real-time tracking of the dynamics of targets. Therefore, the photostability of **LDER** was investigated. As shown in Fig. S6 (ESI<sup>†</sup>), under continuous laser irradiation at 473 nm, the fluorescence of **LDER** in cells showed almost no loss (Fig. S6A, ESI<sup>†</sup>). Compared to its original fluorescence intensity, it still retained a fluorescence intensity of 92% in LDs and 74% in ER (Fig. S6B, ESI<sup>†</sup>). This implied that **LDER** has high stability and can be utilized for real-time monitoring of LD- and ER-related dynamics.

Encouraged by this, the concurrent movement of LDs and ER was tracked in real-time with **LDER**. The position, distribution and dynamics of LDs and ER are clearly shown in Fig. 7. It demonstrates that both the LDs and ER have a characteristic distribution and dispersion pattern in the cells, and the dynamic LDs and ER move distinctly over the course of a long observation period (60 min). To display the movements of the LDs and ER at different times more intuitively, four different pseudocolors were adopted. From the combined images at different time points, we could see the dynamics of the LDs and ER. In addition, the fluorescence intensity of the LDs and ER in the observed area did not show any obvious decrease during the imaging time. Specifically, the LDs revealed near spatial and temporal contacts with ER, suggesting interaction between LDs and ER.

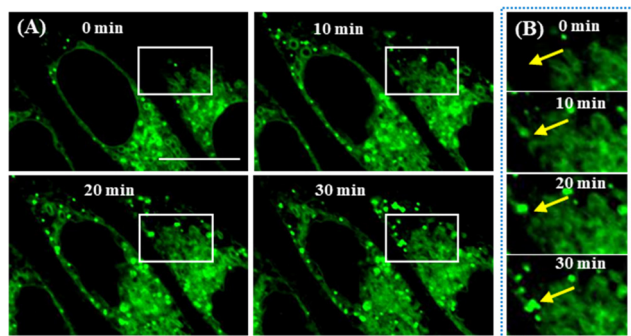


Fig. 8 (A) *In situ* monitoring of LD formation in OA-treated SiHa cells stained with **LDER**. (B) Enlarged fluorescence images of the ROIs indicated by the white rectangles in (A). Scale bar = 20  $\mu$ m.

## Visualization of the generation of LDs from ER

It is widely thought that the spatiotemporal generation of LDs occurs from ER. Here, we used **LDER** to intuitively visualize the emergence of LDs from the ER. First, SiHa cells were starved to force them to consume LDs, incubated with **LDER**, and then treated with OA to stimulate LD formation. Subsequently, the dynamics of the LDs and ER were observed on the same area of the dish. As illustrated in Fig. 8A, several globular LDs and large amounts of reticular ER were seen in the OA-untreated cells (0 min). After treatment with OA, more globular LDs appeared in the ER regions, indicating the formation of LDs in ER. With prolonged OA incubation time, the number and size of LDs increased, and they aggregated to form enlarged LDs. From the magnified images of the ROIs of the white rectangles (Fig. 8B), no green LDs (the position of the yellow arrow) could be seen in the beginning. As time went on, a green LD dot appeared near the ER region, and grew during the first 30 min. The above results demonstrated that **LDER** can monitor LD production from the ER.

## Conclusions

In summary, we proposed a dual-targeting group strategy for construction of a SF-probe to simultaneously and discriminatively visualize two types of organelles by incorporating two different organelle-targeting groups into the same fluorophore. As a proof of concept, a 1,8-naphthalimide-based fluorescent probe (**LDER**) was precisely designed. The probe **LDER** was capable of imaging LDs and ER simultaneously. We found that the fluorescence intensity of **LDER** was sensitive to polarity, which guaranteed its usability in the discrimination of LDs and ER. Additionally, **LDER** could visualize the dynamics of LDs and ER upon either starvation stress or nutrient-rich oleic acid exposure. Notably, the interaction between LDs and ER was investigated with **LDER**, which revealed that the LDs are formed in ER regions. We expect that this strategy can serve as a novel strategy to design SF-probes for imaging different organelles and uncovering interactions between them.

## Experimental section

### Materials and instrumentation

All of the chemicals were purchased from commercial suppliers. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded by 400 MHz Bruker Avance NMR spectrometers. The mass spectra were recorded by an Agilent Technologies 6510 Q-TOF LC/MS system. The absorbance and fluorescence spectra were recorded on a HITACHI U-2910 spectrophotometer and HITACHI F-2700 spectrofluorimeter, respectively. Confocal fluorescence imaging was performed using an Olympus FV-1200 confocal laser scanning microscope.

### Synthesis of **LDER**

The syntheses of compound **1**, compound **2** and compound **3** are described in the supporting information (Scheme S1, ESI<sup>†</sup>).

Compound **3** (1 mmol; 470 mg), *N,N*-dimethyl-4-vinylaniline (1.5 mmol; 220 mg), Pd(OAc)<sub>2</sub> (0.1 mmol; 25 mg) and (o-tolyl)<sub>3</sub>P (0.2 mmol; 60 mg) were dissolved in NEt<sub>3</sub>/DMF (3 mL/15 mL). The mixture was reacted at 100 °C for 48 h under N<sub>2</sub>; after cooling to room temperature, the product was poured into water. The resulting precipitate was purified using ethanol to yield **LDER** as a red powder (54%). <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ 8.98 (d, *J* = 8 Hz, 1H), 8.48 (d, *J* = 8 Hz, 1H), 8.39 (d, *J* = 8 Hz, 1H), 8.18 (d, *J* = 8 Hz, 1H), 7.93 (d, *J* = 16 Hz, 1H), 7.87 (t, *J* = 8 Hz, 1H), 7.76–7.70 (m, 3H), 7.60–7.53 (m, 3H), 7.21 (d, *J* = 8 Hz, 2H), 6.77 (d, *J* = 8 Hz, 2H), 4.12 (t, *J* = 8 Hz, 2H), 3.11 (t, *J* = 8 Hz, 2H), 3.00 (s, 6H), 2.25 (s, 3H). <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>) δ 164.04, 163.70, 151.17, 142.90, 142.41, 138.09, 136.16, 131.18, 131.15, 128.77, 124.84, 122.53, 117.88, 112.42, 40.40, 21.35. HRMS (ESI<sup>+</sup>): *m/z* calcd for C<sub>31</sub>H<sub>30</sub>N<sub>3</sub>O<sub>4</sub>S<sup>+</sup> 540.1952; found: [M + H]<sup>+</sup> 540.1952.

### Theoretical calculation

The geometrical structure and frontier orbitals of **LDER** were obtained using the Gaussian 09 program package. All geometry optimizations were carried out using the B3LYP functional basis set.

### Cell culture and fluorescence imaging

SiHa cells are Human Cervical Carcinoma cells, which are the commonly used cells in fluorescence imaging research. So, we picked SiHa cells as the cellular model. SiHa cells were incubated in H-DMEM (Dulbecco's Modified Eagle's Medium, High Glucose) supplemented with 10% FBS (fetal bovine serum) and 1% penicillin and streptomycin, and cultured in a 5% CO<sub>2</sub> incubator at 37 °C. The probe **LDER** was dissolved in DMSO at a concentration of 5 mM.

For living cell staining experiments, SiHa cells were incubated with 5 μM **LDER** for 30 min. After rinsing with PBS, SiHa cells were imaged using an FV-1200.

For co-localization experiments, SiHa cells were incubated with 5 μM **LDER** for 30 min, then incubated with various commercial probes (BODIPY 493/503, ER-Tracker Blue, Golgi-Tracker Red, Lyso-Tracker Deep Red (LTDR) and MitoTracker Deep Red (MTDR)) for another 10 min. Cells were rinsed before imaging with an FV-1200.

The calculation was performed using the software of FV10-ASW 4.2 Viewer based on the intensity analysis function of the confocal microscope. This function enables the conversion of fluorescence intensity to color. As a result, the fluorescence intensity change can be presented as a change in color. Based on an image consisting of different fluorescence intensities, a new image with different colors could be automatically obtained. The color legend represents the values of fluorescence intensity. With the aid of the rainbow of color, the fluorescence intensity in different areas can be easily visualized.

### Starvation-induced experiments

SiHa cells were incubated in culture medium without nutrition for different times, then stained with **LDER** (5 μM, 30 min). After washing with PBS, cells were imaged with an FV-1200.

### Oleic acid treatment experiments

SiHa cells were pretreated with different concentrations of oleic acid (0, 100, and 200 μM) for 4 h, and then washed with PBS. After staining with **LDER** (5 μM, 30 min), cells were imaged with an FV-1200.

### Dynamic monitoring of LDs and ER

SiHa cells were incubated with **LDER** (5 μM, 30 min) and then washed with PBS. Using an FV-1200, fluorescence images were taken every 20 min for a total of 60 min.

### Dynamic monitoring of LD formation

SiHa cells were starved for 12 h, then cultured in H-DMEM, stained with **LDER**, and treated with 400 μM oleic acid. Fluorescence images were obtained at different time points using an FV-1200.

## Conflicts of interest

There are no conflicts to declare.

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## References

- 1 C. P. Satori, M. M. Henderson, E. A. Krautkramer, V. Kostal, M. D. Distefano and E. A. Arriaga, *Chem. Rev.*, 2013, **113**, 2733–2811.
- 2 H. Zhu, J. Fan, J. Du and X. Peng, *Acc. Chem. Res.*, 2016, **49**, 2115–2126.
- 3 W. Xu, Z. Zeng, J. H. Jiang, Y. T. Chang and L. Yuan, *Angew. Chem., Int. Ed.*, 2016, **55**, 13658–13699.
- 4 A. M. Valm, S. Cohen, W. R. Legant, J. Melunis, U. Hershberg, E. Wait, A. R. Cohen, M. W. Davidson, E. Betzig and J. Lippincott-Schwartz, *Nature*, 2017, **546**, 162–167.
- 5 Q. Ba and G. Yang, *Front. Biol.*, 2017, **12**, 7–18.
- 6 Y. Tu, L. Zhao, D. D. Billadeau and D. Jia, *Front. Cell Dev. Biol.*, 2020, **8**, 163.
- 7 R. V. Farese, Jr. and T. C. Walther, *Cell*, 2009, **139**, 855–860.
- 8 S. Chen, P. Novick and S. Ferro-Novick, *Curr. Opin. Cell Biol.*, 2013, **25**, 428–433.
- 9 B. Turk and V. Turk, *J. Biol. Chem.*, 2009, **284**, 21783–21787.
- 10 M. J. Phillips and G. K. Voeltz, *Nat. Rev. Mol. Cell Biol.*, 2016, **17**, 69–82.
- 11 F. Wilfling, J. T. Haas, T. C. Walther and R. V. Farese, Jr., *Curr. Opin. Cell Biol.*, 2014, **29**, 39–45.
- 12 D. Li, Y. G. Zhao, D. Li, H. Zhao, J. Huang, G. Miao, D. Feng, P. Liu, D. Li and H. Zhang, *Cell Rep.*, 2019, **27**, 343–358.



- 13 S. Martin and R. G. Parton, *Nat. Rev. Mol. Cell Biol.*, 2006, **7**, 373–378.
- 14 K. Liu and M. J. Czaja, *Cell Death Differ.*, 2013, **20**, 3–11.
- 15 H. Dong and M. J. Czaja, *Trends Endocrinol. Metab.*, 2011, **22**, 234–240.
- 16 L. Guo, M. Tian, Z. Zhang, Q. Lu, Z. Liu, G. Niu and X. Yu, *J. Am. Chem. Soc.*, 2021, **143**, 3169–3179.
- 17 I. Johnson, *Histochem. J.*, 1998, **30**, 123–140.
- 18 N.-E. Choi, J.-Y. Lee, E.-C. Park, J.-H. Lee and J. Lee, *Molecules*, 2021, **26**, 217.
- 19 P. Gao, W. Pan, N. Li and B. Tang, *Chem. Sci.*, 2019, **10**, 6035–6071.
- 20 T. K. Fam, A. S. Klymchenko and M. Collot, *Materials*, 2018, **11**, 1768.
- 21 D. Singh, D. Rajput and S. Kanvah, *Chem. Commun.*, 2022, **58**, 2413–2429.
- 22 R. A. Smith, R. C. Hartley and M. P. Murphy, *Antioxid. Redox Signaling*, 2011, **15**, 3021–3038.
- 23 Q. Wan, S. Chen, W. Shi, L. Li and H. Ma, *Angew. Chem., Int. Ed.*, 2014, **53**, 10916–10920.
- 24 D. I. Danylchuk, P. H. Jouard and A. S. Klymchenko, *J. Am. Chem. Soc.*, 2021, **143**, 912–924.
- 25 Z. Zhang, B. Dong and M. Tian, *Sens. Actuators B Chem.*, 2021, **343**, 130168.
- 26 Z. Zhang, Z. Gou, B. Dong and M. Tian, *Sens. Actuators B Chem.*, 2022, **355**, 131349.
- 27 R. M. Duke, E. B. Veale, F. M. Pfeffer, P. E. Kruger and T. Gunnlaugsson, *Chem. Soc. Rev.*, 2010, **39**, 3936–3953.
- 28 C. Geraghty, C. Wynne and R. B. P. Elmes, *Coord. Chem. Rev.*, 2021, **437**, 213713.
- 29 S. Banerjee, E. B. Veale, C. M. Phelan, S. A. Murphy, G. M. Tocci, L. J. Gillespie, D. O. Frimannsson, J. M. Kelly and T. Gunnlaugsson, *Chem. Soc. Rev.*, 2013, **42**, 1601–1618.
- 30 F. Meng, J. Niu, H. Zhang, R. Yang, Q. Lu, Y. Yu, Z. Liu, G. Niu and X. Yu, *Sens. Actuators B Chem.*, 2021, **329**, 129148.
- 31 H. Xiao, P. Li, X. Hu, X. Shi, W. Zhang and B. Tang, *Chem. Sci.*, 2016, **7**, 6153–6159.
- 32 C. Reichardt, *Chem. Rev.*, 1994, **94**, 2319–2358.
- 33 C. W. Song, U. Tamima, Y. J. Reo, M. Dai, S. Sarkar and K. H. Ahn, *Dyes Pigm.*, 2019, **171**, 107718.
- 34 M. Jiang, X. Gu, J. W. Y. Lam, Y. Zhang, R. T. K. Kwok, K. S. Wong and B. Z. Tang, *Chem. Sci.*, 2017, **8**, 5440–5446.
- 35 R. Miao, J. Li, C. Wang, X. Jiang, Y. Gao, X. Liu, D. Wang, X. Li, X. Liu and Y. Fang, *Adv. Sci.*, 2022, **9**, 2104609.
- 36 L. Guo, M. Tian, R. Feng, G. Zhang, R. Zhang, X. Li, Z. Liu, X. He, J. Z. Sun and X. Yu, *ACS Appl. Mater. Interfaces*, 2018, **10**, 10706–10717.
- 37 X. Zheng, W. Zhu, F. Ni, H. Ai and C. Yang, *Sens. Actuators, B*, 2018, **255**, 3148–3154.
- 38 Y. Guo, T. C. Walther, M. Rao, N. Stuurman, G. Goshima, K. Terayama, J. S. Wong, R. D. Vale, P. Walter and R. V. Farese, *Nature*, 2008, **453**, 657–661.
- 39 H. Xu, H. Zhang, G. Liu, L. Kong, X. Zhu, X. Tian, Z. Zhang, R. Zhang, Z. Wu, Y. Tian and H. Zhou, *Anal. Chem.*, 2019, **91**, 977–982.
- 40 E. Zhao, C. Kam, M.-Y. Wu, E. Wang, J. Wang, B. Z. Tang and S. Chen, *CCS Chem.*, 2022, **4**, 515–525.
- 41 S. Zang, X. Kong, J. Cui, S. Su, W. Shu, J. Jing and X. Zhang, *J. Mater. Chem. B*, 2020, **8**, 2660–2665.