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Theoretical prediction of electronic properties and contact barriers in a metal/semiconductor NbS₂/Janus MoSSe van der Waals heterostructure†

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The emergence of van der Waals (vdW) heterostructures, which consist of vertically stacked two-dimensional (2D) materials held together by weak vdW interactions, has introduced an innovative avenue for tailoring nanoelectronic devices. In this study, we have theoretically designed a metal/semiconductor heterostructure composed of NbS₂ and Janus MoSSe, and conducted a thorough investigation of its electronic properties and the formation of contact barriers through first-principles calculations. The effects of stacking configurations and the influence of external electric fields in enhancing the tunability of the NbS₂/Janus MoSSe heterostructure are also explored. Our findings demonstrate that the NbS₂/MoSSe heterostructure is not only structurally and thermally stable but also exfoliable, making it a promising candidate for experimental realization. In its ground state, this heterostructure exhibits p-type Schottky contacts characterized by small Schottky barriers and low tunneling barrier resistance, showing its considerable potential for utilization in electronic devices. Additionally, our findings reveal that the electronic properties, contact barriers and contact types of the NbS₂/MoSSe heterostructure can be tuned by applying electric fields. A negative electric field leads to a conversion from a p-type Schottky contact to an n-type Schottky contact, whereas a positive electric field gives rise to a transformation from a Schottky into an ohmic contact. These insights offer valuable theoretical guidance for the practical utilization of the NbS₂/MoSSe heterostructure in the development of next-generation electronic and optoelectronic devices.

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1 Introduction

In recent years, significant attention has been directed towards two-dimensional (2D) materials due to their remarkable attributes and diverse potential applications.^{1–3} The discovery of graphene⁴ nearly two decades ago marked the inception of a transformative era in materials science and nanotechnology, fueling extensive investigations into the distinctive characteristics and promising uses of 2D materials. Over this period, researchers have identified and synthesized numerous 2D materials, each possessing unique properties and potential applications, including transition metal dichalcogenides

(TMDs),^{5,6} MXenes^{7–10} and the MA₂Z₄ family (M = Mo, W; A = Si, Ge; Z = N, P).^{11–13} As research on 2D materials continues to advance, it becomes increasingly evident that their versatility and intriguing properties have the potential to reshape various fields, presenting both exciting challenges and opportunities for innovative technologies.^{14–17} Among the myriad of 2D materials, the family of 2D TMDs has emerged as a particularly captivating and extensively studied category due to its unique attributes and broad range of applications, including electronics,¹⁸ optoelectronics,¹⁹ energy storage²⁰ and photocatalysis.⁶ 2D TMDs can exhibit either metallic or semiconductor behavior. For example, monolayers of MoS₂ and MoSe₂ are known as semiconductors, while NbS₂ and NbSe₂ monolayers exhibit metallic characteristics. These 2D materials have been successfully synthesized through various methods such as mechanical exfoliation²¹ or chemical vapor deposition (CVD).^{22,23}

Recently, a new category of 2D materials has emerged, namely Janus structures.²⁴ Specifically, Janus MoSSe has been successfully synthesized using the chemical vapor deposition (CVD) method.^{25,26} Janus MoSSe can be obtained by either replacing the top-layer sulfur (S) atoms in a MoS₂ monolayer with selenium (Se) atoms²⁵ or by sulfurization of the top layer of

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selenium atoms in a MoSe₂ monolayer.²⁶ The creation of Janus MoSSe monolayers from MoS₂ and MoSe₂ monolayers disrupts their out-of-plane mirror symmetry, resulting in the emergence of unique properties that distinguish them from traditional TMD monolayers, such as Rashba splitting and strong out-of-plane piezoelectricity.^{27,28} Interestingly, the electronic and transport properties of a Janus MoSSe monolayer can be controlled by using the layer thickness,²⁹ vacancies^{30,31} and strain engineering.³² Hence, Janus MoSSe monolayers hold promise for diverse applications, including gas sensing,³³ water splitting^{34,35} and Li-ion batteries.³⁶

Notably, the development of van der Waals (vdW) heterostructures, composed of distinct 2D materials vertically stacked and held together by weak vdW interactions, has introduced a novel approach to tailor nanoelectronic devices.^{37,38} This stacking arrangement allows for the preservation of the intrinsic properties of the constituent materials to a significant extent. Furthermore, the combination of different 2D materials often leads to unexpected and novel behaviors, opening up exciting possibilities in nanoelectronics. In recent times, the majority of investigations have centered on pairing two distinct 2D semiconductors. However, the creation of metal/semiconductor heterostructures through the combination of 2D metals and 2D semiconductors plays a pivotal role in advancing electronic device development.^{39–41} Currently, there is growing interest in the combination of metallic NbS₂ with other 2D semiconductors, as they can form metal/semiconductor heterostructures with either low Schottky barriers or ohmic contacts, depending on the stacking orientations. Hence, in this study, we construct a novel heterostructure by vertically stacking metallic NbS₂ and Janus MoSSe monolayers and investigate the electronic properties and interfacial characteristics using first-principles predictions. The stacking effects and effects of external electric fields are also explored to enhance the tunability of these heterostructures. We find that the NbS₂/MoSSe heterostructure exhibits p-type Schottky contacts with small Schottky barriers and low tunneling barrier resistance, highlighting its significant potential for use in electronic devices. Furthermore, we demonstrate that the electronic properties and contact characteristics of the NbS₂/MoSSe heterostructure can be finely tuned through the application of electric fields. These insights provide valuable theoretical guidance for the practical application of the NbS₂/MoSSe heterostructure in the development of next-generation electronic and optoelectronic devices.

2 Computational model and methods

In this study, we conducted first-principles calculations using the Vienna *ab initio* simulation package (VASP)^{42,43} to investigate the geometric optimization and electronic properties of all considered materials, including NbS₂, Janus MoSSe monolayers and their combined heterostructure. All calculations were performed within the framework of the generalized gradient approximation (GGA)⁴⁴ for the exchange-correlation energy, utilizing the Perdew–Burke–Ernzerhof (PBE) functional.⁴⁵ To describe the electron–ion interactions, we employed projector augmented wave (PAW)

pseudopotentials.⁴⁶ The energy cut-off was set to 510 eV, and we employed a Monkhorst–Pack *k*-point mesh of $9 \times 9 \times 1$. During the structural optimization process, all atomic structures were relaxed until the convergence criteria for force and energy reached $0.01 \text{ eV } \text{\AA}^{-1}$ and 10^{-6} eV , respectively. To eliminate spurious interactions between adjacent layers, a vacuum thickness of 25 \AA was introduced along the *z*-direction. We accounted for the weak van der Waals (vdW) interactions inherent in layered structures using DFT-D3 and DFT-D2 methods, which were proposed by Grimme.^{47,48} Furthermore, to rectify the underestimation of the band gap associated with the PBE functional, we employed the Heyd–Scuseria–Ernzerhof (HSE06) functional.^{49,50} Dipole corrections were also incorporated into all calculations. The spin–orbit coupling (SOC) effect has also been added to the calculations.

3 Results and discussion

We first examine the atomic and electronic properties of the perfect NbS₂ and Janus MoSSe monolayers, as depicted in Fig. 1. Both the NbS₂ and Janus MoSSe monolayers show the same hexagonal crystal structure. The lattice constants of NbS₂ and Janus MoSSe monolayers are calculated to be 3.32 and 3.28 \AA , respectively. These values are in good agreement with the experimental measurements,^{25,26,51} confirming the reliability of our calculations. The electronic band structures of NbS₂ and Janus MoSSe monolayers are illustrated in Fig. 1(b and d). The NbS₂ monolayer exhibits metallic behavior, featuring a band that crosses the Fermi level, while Janus MoSSe is a semiconductor with an indirect band gap of $1.67/2.15 \text{ eV}$ obtained by using the PBE/HSE functional. Both the maxima of the valence bands and the minima of the conduction bands are located at the *K* point. It should be noted that the PBE functional underestimates the band gap of 2D materials, whereas the HSE functional provides more accurate results for the band gap.

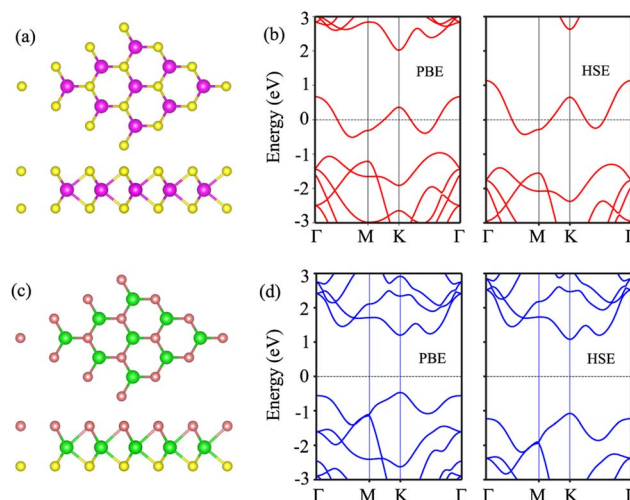


Fig. 1 Atomic structures of (a) NbS₂ and (c) Janus MoSSe monolayers. Yellow, purple, green and brown balls represent the S, Nb, Mo and Se atoms, respectively. Band structures of (b) NbS₂ and (d) Janus MoSSe monolayers given by using PBE and HSE functionals.



However, due to its low computational cost, the PBE functional is the choice for all calculations.

We now build the atomic structure of the NbS₂/Janus MoS₂ heterostructure for different stacking configurations, as illustrated in Fig. 2. Owing to the small difference in the lattice parameters of NbS₂ and Janus MoS₂ monolayers, the atomic structure of the NbS₂/Janus MoS₂ heterostructure consists of a (1 × 1) unit cell of NbS₂ and (1 × 1) unit cell of the Janus MoS₂ monolayer. The lattice mismatch in the NbS₂/Janus MoS₂ heterostructure is as small as 0.6%, and it does not significantly affect the main characteristics of the heterostructure. Furthermore, due to the lack of symmetry in the two faces of the Janus MoS₂ monolayer, the atomic structures of the NbS₂/Janus MoS₂ heterostructure split into two main configurations, namely NbS₂/SMoS₂ and NbS₂/SeMoS₂ heterostructures. Each NbS₂/SMoS₂ or NbS₂/SeMoS₂ heterostructure consists of four different stacking configurations, as presented in Fig. 2. After the geometric optimization, the interlayer distance between the NbS₂ and Janus MoS₂ layers for each stacking configuration of the NbS₂/Janus MoS₂ heterostructure is obtained, as listed in Table 1. The AB3 stacking configuration of NbS₂/Janus MoS₂ has the shortest interlayer distance, while the AA1 stacking configuration shows the largest *d*. Furthermore, to evaluate the stability of the NbS₂/Janus MoS₂ heterostructure, we calculate the binding energy (exfoliation energy)⁵² as follows:

$$E_b = \frac{E_{\text{NbS}_2/\text{Janus MoS}_2} - E_{\text{NbS}_2} - E_{\text{Janus MoS}_2}}{A} \quad (1)$$

Here, $E_{\text{NbS}_2/\text{Janus MoS}_2}$, E_{NbS_2} and $E_{\text{Janus MoS}_2}$ are the total energies of the NbS₂/Janus MoS₂ heterostructure, isolated NbS₂ and Janus MoS₂ monolayers, respectively. *A* stands for the surface area of the NbS₂/Janus MoS₂ heterostructure. The calculated binding energy values obtained from the DFT-D3 method of the NbS₂/Janus MoS₂ heterostructure for all eight stacking configurations are in the range from −32 to 20 meV Å^{−2}, which are listed in Table 1. Our results show that the binding energy values are of a similar magnitude to those observed in typical van der Waals (vdW) heterostructures, including HfSeX (X = S, Se)/graphene,⁵³ NbS₂/MoSi₂P₄,⁵⁴ MoS₂/MoS₂,³⁵ MoS₂/g-GeC⁵⁵ and GaN/BP.⁵⁶ This finding suggests that all stacking configurations of the NbS₂/Janus MoS₂ heterostructure exhibit typical vdW interactions. These weak vdW interactions are responsible for maintaining the stability of the heterostructure, which can be synthesized in experiments using

various techniques, including chemical vapor deposition (CVD)³⁹ and epitaxial growth.⁵⁷ Additionally, the negative values of the binding energies of the NbS₂/Janus MoS₂ heterostructure for all stacking configurations confirm that they are energetically stable. Notably, the AB3 stacking configuration exhibits the lowest binding energy, establishing it as the most energetically stable structure. Additionally, we also used the DFT-D2 method to calculate the binding energy of the NbS₂/MoS₂ heterostructure for comparison. One can observe that the DFT-D2 binding energy is larger than that obtained by the DFT-D3 method. However, both DFT-D2 and DFT-D3 consistently predict that the AB3 stacking configuration exhibits the lowest binding energy. Furthermore, to consider the dynamic stability of the AB3 stacking configuration, we further measure its phonon spectrum, as illustrated in Fig. S1 of the ESI.† We observe that there are no negative frequencies in the phonon spectrum of the AB3 stacking configuration of the NbS₂/Janus MoS₂ heterostructure, indicating that such a configuration is dynamically stable. Hence, we focus on this stacking configuration for all the subsequent investigations.

The projected band structures of the NbS₂/Janus MoS₂ heterostructure for all eight stacking configurations are illustrated in Fig. 3. One can find that the electronic band structures of such configurations appear to be a combination of those of the constituent NbS₂ and Janus MoS₂ monolayers. The nature of such a preservation arises from the dominance of weak van der Waals (vdW) interactions between the NbS₂ and Janus MoS₂ layers in their corresponding heterostructure. All the stacking configurations of the NbS₂/Janus MoS₂ heterostructure exhibit metallic behavior with a rough band crossing the Fermi level. By analyzing the projected band structures in Fig. 3, one can observe that a rough band crossing the Fermi level comes from the metallic NbS₂ layer. More interestingly, upon the formation of the metal/semiconductor NbS₂/Janus MoS₂ heterostructure, either an ohmic contact (OhC) or Schottky contact (ShC) is generated, depending on the energy position of the band edges of Janus MoS₂ relative to the Fermi level of the NbS₂ layer. Hence, comprehending the characteristics of both OhC and ShC contacts plays a crucial role in enabling charge carrier injection and extraction in the metal/semiconductor NbS₂/Janus MoS₂ heterostructure. From Fig. 3, we can find that all the stacking configurations form a ShC contact because the Fermi level of the metallic NbS₂ layer lies between two band edges of semiconductor Janus MoS₂. Based on the Schottky–Mott rule⁵⁸ the Schottky barriers for n-type and p-type ShC contacts can be established as follows:

$$\Phi_n = E_C - E_F \quad (2)$$

and

$$\Phi_p = E_V - E_F \quad (3)$$

Here, E_C , E_V and E_F are the CBM and VBM band edges of semiconductor Janus MoS₂ and the Fermi level of the metallic NbS₂ layer, respectively. The obtained Schottky barriers for all eight stacking configurations of the NbS₂/Janus MoS₂

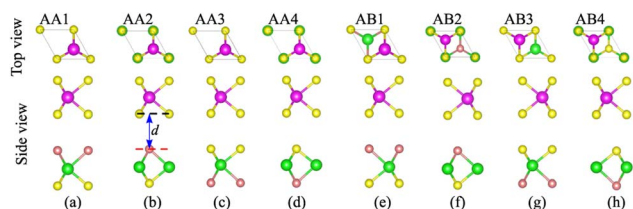


Fig. 2 Top and side views of the atomic structures of the NbS₂/Janus MoS₂ heterostructure for different stacking patterns of (a) AA1, (b) AA2, (c) AA3, (d) AA4, (e) AB1, (f) AB2, (g) AB3 and (h) AB4.



Table 1 Calculated interlayer distance (d , Å), binding energy calculated by DFT-D3 and DFT-D2 methods (E_b , meV Å⁻²), Schottky barrier heights (eV), work functions (W , eV) and contact types in the NbS₂/Janus MoS₂ heterostructure for the different stacking configurations

| Materials | Stacking types | d | E_b | | Φ_n | Φ_p | W | Schottky type |
|-------------------------|----------------|------|--------|--------|----------|----------|------|---------------|
| | | | DFT-D3 | DFT-D2 | | | | |
| NbS ₂ /SMoSe | AA1 | 3.42 | −21.50 | −16.85 | 1.14 | 0.26 | 6.05 | p-type ShC |
| | AA2 | 2.91 | −20.33 | −16.84 | 0.78 | 0.51 | 6.11 | p-type ShC |
| | AB1 | 3.34 | −31.06 | −28.32 | 1.18 | 0.18 | 6.03 | p-type ShC |
| | AB2 | 2.88 | −28.15 | −25.36 | 0.95 | 0.28 | 6.06 | p-type ShC |
| NbS ₂ /SMoSe | AA3 | 3.39 | −21.99 | −17.03 | 1.14 | 0.25 | 6.07 | p-type ShC |
| | AA4 | 3.31 | −20.74 | −17.33 | 0.80 | 0.49 | 6.07 | p-type ShC |
| | AB3 | 2.83 | −32.74 | −30.84 | 1.16 | 0.19 | 6.06 | p-type ShC |
| | AB4 | 2.78 | −30.16 | −26.54 | 0.89 | 0.31 | 6.08 | p-type ShC |

heterostructure are listed in Table 1. One can observe that all the stacking configurations of the NbS₂/Janus MoS₂ heterostructure exhibit p-type ShC contact with the Schottky barriers ranging from 0.18 to 0.51 eV. The AB3 stacking configuration has the narrowest Schottky barrier compared to other configurations.

Furthermore, to determine whether the formation of the metal/semiconductor NbS₂/Janus MoS₂ heterostructure leads to a narrower Schottky barrier compared to the metal/semiconductor NbS₂/MoS₂ and NbS₂/MoSe₂ heterostructures, we next calculate the electronic properties of the latter, as depicted in Fig. 4. One can find that all the NbS₂/MoS₂, NbS₂/MoSe₂ and NbS₂/MoSe₂ heterostructures form a p-type ShC contact. The Φ_p Schottky barriers for the NbS₂/MoS₂, NbS₂/MoSe₂ and NbS₂/MoSe₂ heterostructures, respectively, are

calculated to be 0.32/0.37, 0.18/0.25 and 0.23/0.32 eV by using the PBE/HSE functional. It is clear that the NbS₂/Janus MoS₂ heterostructure exhibits a narrower Schottky barrier compared to the NbS₂/MoS₂ and NbS₂/MoSe₂ heterostructures for both the PBE and HSE prediction. A narrower Schottky barrier results in improved device performance for those based on the metal/semiconductor NbS₂/Janus MoS₂ heterostructure. These findings illustrate that the Janus MoS₂ monolayer can serve as an ideal channel when combined with the 2D metallic NbS₂ monolayer, resulting in the formation of a metal/semiconductor NbS₂/Janus MoS₂ heterostructure that enhances charge carrier injection. Besides, for a more comprehensive understanding of the contributions of each layer in the band structure of the NbS₂/Janus MoS₂ heterostructure, we plot the partial density states (PDOS) of all atoms in the NbS₂/Janus MoS₂

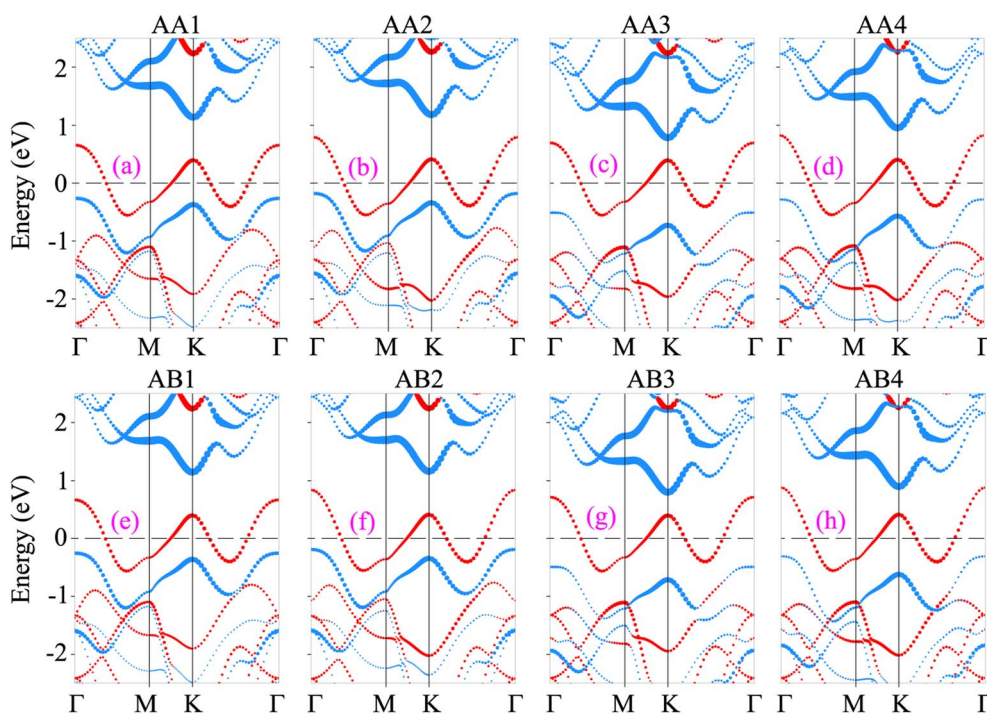


Fig. 3 Projected band structures of the NbS₂/Janus MoS₂ heterostructure for different stacking patterns of (a) AA1, (b) AA2, (c) AA3, (d) AA4, (e) AB1, (f) AB2, (g) AB3 and (h) AB4. Red and cyan circles represent the contributions of metallic NbS₂ and semiconductor MoS₂ layers, respectively.



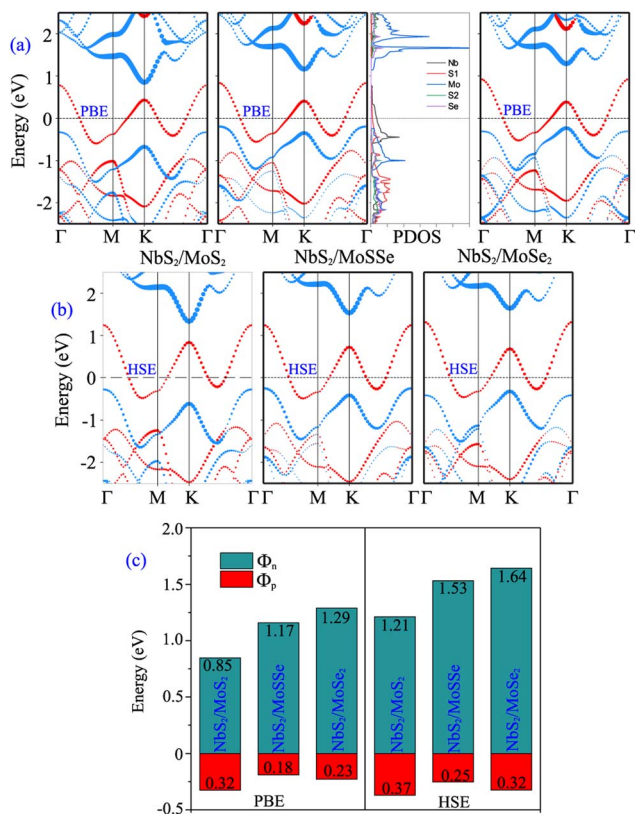


Fig. 4 Projected band structures of NbS₂/MoS₂, NbS₂/MoSSe and NbS₂/MoSe₂ heterostructures obtained using (a) PBE and (b) HSE functionals. (c) Schottky barriers of NbS₂/MoS₂, NbS₂/MoSSe and NbS₂/MoSe₂ heterostructures for the most energetically stable AB3 configuration. The inset presents the partial density of states (PDOS) of all atoms in the NbS₂/MoSSe heterostructure for the AB3 stacking configuration.

heterostructure, as shown in the inset of Fig. 4(a). We can observe that the band crossing the Fermi level is mainly contributed by the Nb-orbitals of the NbS₂ layer. Meanwhile, the Mo-orbital states contribute mainly to the CBM of the semi-conducting Janus MoSSe layer and its VBM comes from the hybridization between Mo and S atoms in the MoSSe layer. In addition, to evaluate the SOC effect on the electronic properties of the NbS₂/MoSSe heterostructure, we further plot its projected band structure with the inclusion of the SOC effect, as depicted in Fig. S2 of the ESI.† It is evident that SOC introduces a split in the valence band of the Janus MoSSe layer at the *K* point, as illustrated in Fig. S2(b).† However, it is crucial to highlight that despite the split observed in the valence band of the Janus MoSSe layer, this does not lead to a change in the contact type or barrier heights of the NbS₂/MoSSe heterostructure because the valence band of the Janus MoSSe layer is located at the *Γ* point. Hence, the SOC effect has a negligible impact on the electronic properties of the heterostructure.

Furthermore, to investigate the charge redistribution at the interface of the NbS₂/Janus MoSSe heterostructure, we calculate the charge density difference (CDD) as follows:

$$\Delta\rho = \rho_{\text{NbS}_2/\text{MoSSe}} - \rho_{\text{NbS}_2} - \rho_{\text{MoSSe}} \quad (4)$$

Here, $\rho_{\text{NbS}_2/\text{MoSSe}}$, ρ_{NbS_2} and ρ_{MoSSe} are the charge densities of the NbS₂/MoSSe heterostructure, isolated NbS₂ and Janus MoSSe monolayers, respectively. The CDD at the interface of the NbS₂/MoSSe heterostructure is depicted in Fig. 5(a). One can find that the electrons are mainly depleted on the side of the NbS₂ layer, while they are mainly accumulated on the side of the Janus MoSSe layer. This finding demonstrates that the electrons are mainly transferred from the NbS₂ to the Janus MoSSe layer. Such charge movement indicates that the metallic NbS₂ layer can be used as contact electrodes for efficient electron injection, with the Janus semiconductor MoSSe layer serving as the channel in field-effect transistors (FETs) based on the NbS₂/MoSSe heterostructure. Bader charge analysis shows that there are only 0.012 electrons, migrating from the NbS₂ to the Janus MoSSe layer. The electrostatic potential of the NbS₂/MoSSe heterostructure is depicted in Fig. 5(b). It is evident that the NbS₂ layer has a higher potential than the Janus MoSSe layer in the NbS₂/MoSSe heterostructure, confirming that the electrons migrated from the NbS₂ to the MoSSe layer. Furthermore, the charge transfer at the interface of the heterostructure could lead to the formation of an interface dipole, which can change the barrier heights of such a heterostructure. The interface dipole can be obtained as follows:

$$\Delta V = W_{\text{NbS}_2} - W_{\text{NbS}_2/\text{MoSSe}} \quad (5)$$

where W_{NbS_2} and $W_{\text{NbS}_2/\text{MoSSe}}$ are the work functions of the isolated NbS₂ monolayer and NbS₂/MoSSe heterostructure, respectively. From Fig. 5(b), the interface dipole at the interface of the NbS₂/MoSSe heterostructure is obtained to be 0.08 eV. In

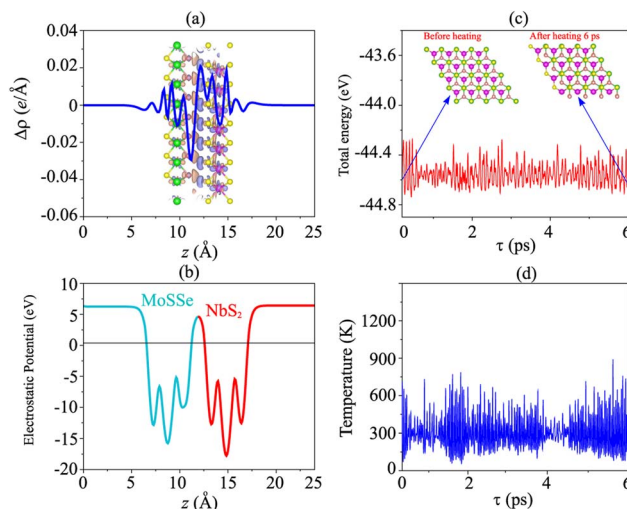


Fig. 5 (a) Planar averaged charge density difference and (b) electrostatic potential of the NbS₂/MoSSe heterostructure for the most energetically stable AB3 configuration. Dark purple and dark pink represent the positive and negative charges, respectively. AIMD simulations of the fluctuation of (c) total energy and (d) temperature as a function of time steps. The insets present the atomic structures of the NbS₂/MoSSe heterostructure before and after heating for 6 ps.

the presence of an interface dipole, the barrier heights of the NbS₂/MoSSe heterostructure can be rewritten as:

$$\Phi_n = E_C - W_{\text{NbS}_2} \quad (6)$$

and

$$\Phi_p = W_{\text{NbS}_2} - E_V \quad (7)$$

We observe that the barrier heights of the NbS₂/MoSSe heterostructure remain nearly unchanged in the presence of an interface dipole compared to those without it. Such a phenomenon can be attributed to the small magnitude of the interface dipole and the minimal amount of charge transfer.

To evaluate the charge injection efficiency of the NbS₂/MoSSe heterostructure, we further calculate its tunneling probability (T) and the tunneling-specific resistivity (ρ_t) as follows:

$$T = \exp\left(\frac{-2w_{\text{TB}}\sqrt{2m_e\hbar w_{\text{TB}}}}{\hbar}\right) \quad (8)$$

and

$$\rho_t = \frac{4\pi^2\hbar w_{\text{TB}}^2}{e^2} \times \frac{\exp\left[\frac{2(2m_e)^{1/2}}{\hbar}w_{\text{TB}}h_{\text{TB}}^{1/2}\right]}{\frac{(2m_e)^{1/2}}{\hbar}w_{\text{TB}}h_{\text{TB}}^{1/2} - 1} = \frac{8\pi^2\hbar^2 w_{\text{TB}}}{3e^2\sqrt{2m_e\hbar w_{\text{TB}}}} \times \mathcal{J}^{-1} \quad (9)$$

Here, h_{TB} and w_{TB} are the tunneling barrier height and width, respectively. \hbar is the reduced Planck's constant. e and m_e are the electron magnitude and mass of a free electron, respectively. From the electrostatic potential of the NbS₂/MoSSe heterostructure, the h_{TB} and w_{TB} are calculated to be 4.61 eV and 1.40 Å, respectively. Using eqn (8) and (9), the tunneling probability and tunneling-specific resistivity of the NbS₂/MoSSe heterostructure are 3.23% and $1.89 \times 10^{-9} \Omega \text{ cm}^2$, respectively. Interestingly, the tunneling-specific resistivity of the NbS₂/MoSSe heterostructure is comparable to that observed in other heterostructures, including Bi/MoS₂,⁵⁹ Bi, Sb semimetals/TMDs,⁶⁰ metal/MSi₂N₄ ($M = \text{Mo}, \text{W}$)⁶¹ and 2D (3D) metals/GeSe,⁶² suggesting that such a heterostructure could serve as an efficient contact for electronic devices.

Furthermore, to check the thermal stability of the NbS₂/MoSSe heterostructure, we perform the *ab initio* molecular dynamics (AIMD) simulation. The fluctuations of total energy and temperature of the NbS₂/MoSSe heterostructure for all stacking configurations as a function of time steps are displayed in Fig. 5(c, d) and S3 of the ESI.† It is evident that the fluctuations of total energy in Fig. 5(c) and S3 of the ESI† are small, and the atomic structures of the NbS₂/MoSSe heterostructure for all stacking configurations after heating for 6 ps show no distortion or bond breaking, implying that the NbS₂/MoSSe heterostructure is thermally stable at room temperature.

More importantly, the ability to adjust contact characteristics in the NbS₂/MoSSe heterostructure is crucial for its applications in next-generation electronic devices. Recently, applying electric

fields is considered as an effective way to control both the electronic properties and contact characteristics of 2D heterostructures, such as graphene-based heterostructures^{63–66} and TMD-based heterostructures.^{67–70} Therefore, we further investigate the ability to adjust the electronic properties and contact characteristics in the NbS₂/MoSSe heterostructure by applying electric fields. The schematic model of applying electric fields along the z direction of the NbS₂/MoSSe heterostructure is illustrated in Fig. 6(a). The positive electric field is defined as the direction pointing from the Janus MoSSe to the NbS₂ layers. The variation of the Schottky barriers of the NbS₂/MoSSe heterostructure under different electric fields is depicted in Fig. 6(b). It is evident that applying an electric field varies the Schottky barriers, resulting in a transformation of the contact types in the NbS₂/MoSSe heterostructure. The Φ_n of the NbS₂/MoSSe heterostructure increases as the strength of the electric field increases from -1.5 to $+1.5 \text{ V Å}^{-1}$, whereas the Φ_p had decreased accordingly. This finding implies that the Schottky barriers Φ_n and Φ_p change in two opposite directions under the application of electric fields. Under the influence of a negative electric field of -1.3 V Å^{-1} , we observe a reduction in the value of Φ_n , causing it to become narrower than the Φ_p of the NbS₂/MoSSe heterostructure. This change leads to a shift from p-type ShC to n-type ShC in the NbS₂/MoSSe heterostructure. Conversely, when a positive electric field is applied, Φ_n decreases, yet it remains narrower than the Φ_p of the heterostructure, maintaining its p-type ShC characteristics. Upon further application of a positive electric field, reaching $+1.5 \text{ V Å}^{-1}$, Φ_n continuously decreases until it reaches zero, signifying a transition from p-type ShC to p-type OhC in the NbS₂/MoSSe heterostructure.

In order to gain deeper insight into the mechanisms underlying the change in the contact barriers and contact types in the NbS₂/MoSSe heterostructure, we further plot its projected band structures under different strengths of both negative and positive electric fields, as illustrated in Fig. 7. The impact of a negative electric field is clearly observable on the band edges of the Janus MoSSe semiconductor. Specifically, the VBM of the Janus MoSSe semiconductor shifts to a lower binding energy, moving further away from the Fermi level. In contrast, the CBM

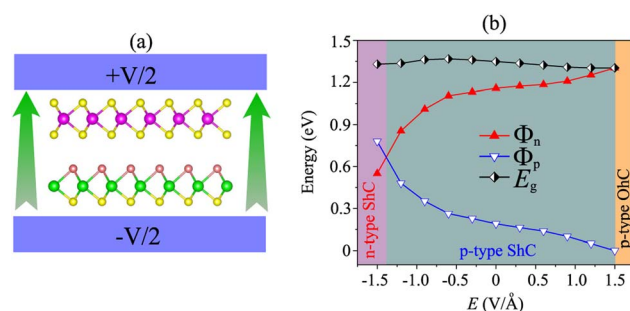


Fig. 6 (a) The schematic model of an applied electric field along the z direction of the NbS₂/MoSSe heterostructure for the most energetically stable AB3 configuration. (b) The variation of the Schottky barriers and contact types in the NbS₂/MoSSe heterostructure as a function of electric fields.



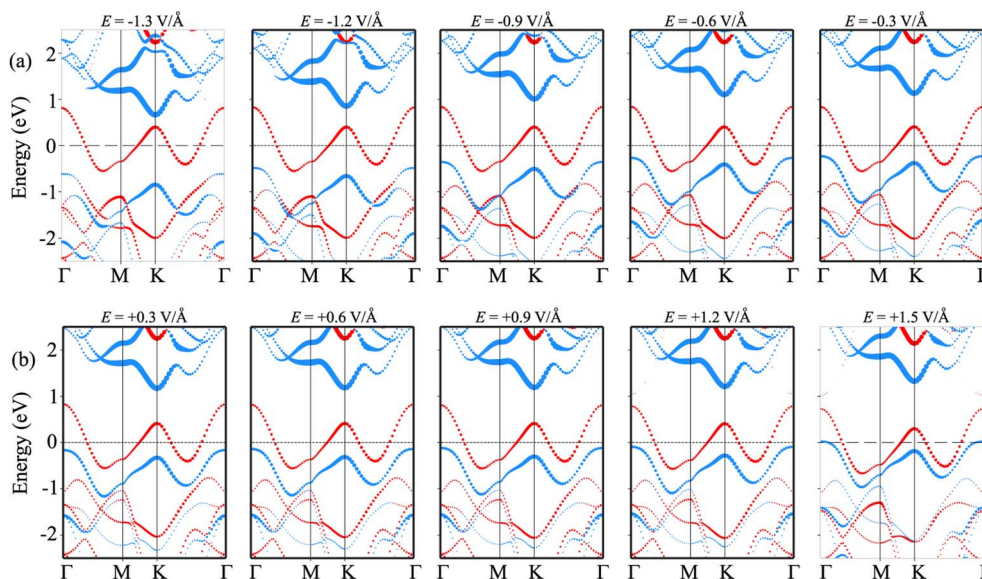


Fig. 7 Projected band structures of the metal/semiconductor NbS₂/MoSSe heterostructure for the most energetically stable AB3 configuration under different strengths of (a) negative and (b) positive electric fields. Red and blue lines represent the contributions of the metallic NbS₂ and semiconducting MoSSe layers, respectively.

shifts to a lower binding energy but moves closer to the Fermi level. Hence, the application of a negative electric field causes a decrease in the Φ_n and an increase in the Φ_p . Under a negative electric field of $-1.3 \text{ V } \text{\AA}^{-1}$, the CBM of the Janus MoSSe semiconductor is positioned closer to the Fermi level than its VBM, suggesting that the NbS₂/MoSSe heterostructure is transformed from p-type ShC into n-type ShC. Likewise, the application of a positive electric field induces shifts in the band edges of the Janus MoSSe monolayer, although in different directions compared to a negative electric field. The VBM of the semiconducting MoSSe layer shifts closer to the Fermi level, while its CBM moves farther away from the Fermi level. Hence, under the application of a positive electric field, the VBM of the Janus MoSSe semiconductor is always positioned closer to the Fermi level than the CBM, indicating that the NbS₂/MoSSe heterostructure maintains the p-type ShC characteristics. With a further increase in the positive electric field, the VBM of the Janus MoSSe layer continues to shift towards the Fermi level and eventually crosses the Fermi level at a positive electric field strength of $+1.5 \text{ V } \text{\AA}^{-1}$. This finding demonstrates that a positive electric field results in a transformation from ShC into OhC in the NbS₂/MoSSe heterostructure. Indeed, it's important to note that the transition from ShC to OhC in the metal/semiconductor heterostructure is a critical factor that plays a pivotal role in enhancing the performance of devices built based on the NbS₂/MoSSe heterostructure. For instance, with the application of a positive electric field of $+1.5 \text{ V } \text{\AA}^{-1}$, the tunneling-specific resistivity of the NbS₂/Janus MoSSe heterostructure reduces to $0.82 \times 10^{-9} \text{ } \Omega \text{ cm}^2$. Hence, this transformation can have a profound impact on device characteristics and is a key consideration for optimizing device functionality and efficiency.

4 Conclusions

In summary, we have designed a metal/semiconductor NbS₂/MoSSe heterostructure and systematically investigated its electronic properties and the formation of contact types using first-principles calculations. The NbS₂/MoSSe heterostructure exhibits exceptional structural and thermal stability, making it a compelling candidate for practical applications. The NbS₂/MoSSe heterostructure manifests p-type Schottky contacts characterized by small Schottky barriers and low tunneling resistance, signifying its substantial utility in electronic device integration. Furthermore, we found that the electronic properties, contact barriers, and contact types within this heterostructure can be controlled by manipulating the stacking configurations and applying external electric fields. Notably, negative electric fields facilitate a transition from p-type Schottky to n-type Schottky contacts, whereas positive electric fields induce a shift from Schottky to ohmic contacts. These findings offer valuable theoretical insights for exploiting the NbS₂/MoSSe heterostructure's versatility in advancing nano-electronic and optoelectronic device development.

Conflicts of interest

There are no conflicts to declare.

References

- 1 A. Zavabeti, A. Jannat, L. Zhong, A. A. Haidry, Z. Yao and J. Z. Ou, *Nano-Micro Lett.*, 2020, **12**, 1–34.
- 2 P. Miró, M. Audiffred and T. Heine, *Chem. Soc. Rev.*, 2014, **43**, 6537–6554.



- 3 S. Z. Butler, S. M. Hollen, L. Cao, Y. Cui, J. A. Gupta, H. R. Gutiérrez, T. F. Heinz, S. S. Hong, J. Huang, A. F. Ismach, *et al.*, *ACS Nano*, 2013, **7**, 2898–2926.
- 4 K. S. Novoselov, A. K. Geim, S. V. Morozov, D.-e. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *Science*, 2004, **306**, 666–669.
- 5 S. Manzeli, D. Ovchinnikov, D. Pasquier, O. V. Yazyev and A. Kis, *Nat. Rev. Mater.*, 2017, **2**, 1–15.
- 6 R. Yang, Y. Fan, Y. Zhang, L. Mei, R. Zhu, J. Qin, J. Hu, Z. Chen, Y. Hau Ng, D. Voiry, *et al.*, *Angew. Chem., Int. Ed.*, 2023, **62**, e202218016.
- 7 Y. Wei, P. Zhang, R. A. Soomro, Q. Zhu and B. Xu, *Adv. Mater.*, 2021, **33**, 2103148.
- 8 S. Abdolhosseinzadeh, X. Jiang, H. Zhang, J. Qiu and C. J. Zhang, *Mater. Today*, 2021, **48**, 214–240.
- 9 W. Meng, X. Liu, H. Song, Y. Xie, X. Shi, M. Dargusch, Z.-G. Chen, Z. Tang and S. Lu, *Nano Today*, 2021, **40**, 101273.
- 10 H. Shi, Y. Dong, S. Zheng, C. Dong and Z.-S. Wu, *Nanoscale Adv.*, 2020, **2**, 4212–4219.
- 11 Y.-L. Hong, Z. Liu, L. Wang, T. Zhou, W. Ma, C. Xu, S. Feng, L. Chen, M.-L. Chen, D.-M. Sun, *et al.*, *Science*, 2020, **369**, 670–674.
- 12 Y. Yin, Q. Gong, M. Yi and W. Guo, *Adv. Funct. Mater.*, 2023, **33**, 2214050.
- 13 Q. Wang, L. Cao, S.-J. Liang, W. Wu, G. Wang, C. H. Lee, W. L. Ong, H. Y. Yang, L. K. Ang, S. A. Yang, *et al.*, *npj 2D Mater. Appl.*, 2021, **5**, 71.
- 14 C. Liu, H. Chen, S. Wang, Q. Liu, Y.-G. Jiang, D. W. Zhang, M. Liu and P. Zhou, *Nat. Nanotechnol.*, 2020, **15**, 545–557.
- 15 L. Hu, B. J. Kim, S. Ji, J. Hong, A. K. Katiyar and J.-H. Ahn, *Appl. Phys. Rev.*, 2022, **9**, 041308.
- 16 S. Conti, G. Calabrese, K. Parvez, L. Pimpolari, F. Pieri, G. Iannaccone, C. Casiraghi and G. Fiori, *Nat. Rev. Mater.*, 2023, 1–17.
- 17 H. Xiang, Y.-C. Chien, Y. Shi and K.-W. Ang, *Small Struct.*, 2022, **3**, 2200060.
- 18 Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman and M. S. Strano, *Nat. Nanotechnol.*, 2012, **7**, 699–712.
- 19 M. Wu, Y. Xiao, Y. Zeng, Y. Zhou, X. Zeng, L. Zhang and W. Liao, *InfoMat*, 2021, **3**, 362–396.
- 20 X. Duan, J. Xu, Z. Wei, J. Ma, S. Guo, H. Liu and S. Dou, *Small Methods*, 2017, **1**, 1700156.
- 21 Z. Zhang, P. Yang, M. Hong, S. Jiang, G. Zhao, J. Shi, Q. Xie and Y. Zhang, *Nanotechnology*, 2019, **30**, 182002.
- 22 T. Kang, T. W. Tang, B. Pan, H. Liu, K. Zhang and Z. Luo, *ACS Mater. Au*, 2022, **2**, 665–685.
- 23 Q. Wang, Y. Lei, Y. Wang, Y. Liu, C. Song, J. Zeng, Y. Song, X. Duan, D. Wang and Y. Li, *Energy Environ. Sci.*, 2020, **13**, 1593–1616.
- 24 P. A. L. Sino, L.-Y. Feng, R. A. B. Villaos, H. N. Cruzado, Z.-Q. Huang, C.-H. Hsu and F.-C. Chuang, *Nanoscale Adv.*, 2021, **3**, 6608–6616.
- 25 A.-Y. Lu, H. Zhu, J. Xiao, C.-P. Chuu, Y. Han, M.-H. Chiu, C.-C. Cheng, C.-W. Yang, K.-H. Wei, Y. Yang, *et al.*, *Nat. Nanotechnol.*, 2017, **12**, 744–749.
- 26 J. Zhang, S. Jia, I. Kholmanov, L. Dong, D. Er, W. Chen, H. Guo, Z. Jin, V. B. Shenoy, L. Shi, *et al.*, *ACS Nano*, 2017, **11**, 8192–8198.
- 27 A. C. Riis-Jensen, M. Pandey and K. S. Thygesen, *J. Phys. Chem. C*, 2018, **122**, 24520–24526.
- 28 L. Dong, J. Lou and V. B. Shenoy, *ACS Nano*, 2017, **11**, 8242–8248.
- 29 W.-J. Yin, B. Wen, G.-Z. Nie, X.-L. Wei and L.-M. Liu, *J. Mater. Chem. C*, 2018, **6**, 1693–1700.
- 30 Y.-N. Wen, M.-G. Xia and S.-L. Zhang, *Comput. Mater. Sci.*, 2018, **152**, 20–27.
- 31 C. Long, Y. Dai and H. Jin, *Phys. Rev. B*, 2021, **104**, 125306.
- 32 J. Yuan, Y. Shan and T. Li, *J. Phys. D: Appl. Phys.*, 2020, **53**, 125502.
- 33 C. Jin, X. Tang, X. Tan, S. C. Smith, Y. Dai and L. Kou, *J. Mater. Chem. A*, 2019, **7**, 1099–1106.
- 34 X. Lei, C. Ouyang and K. Huang, *Appl. Surf. Sci.*, 2021, **537**, 147919.
- 35 A. Singh, M. Jain and S. Bhattacharya, *Nanoscale Adv.*, 2021, **3**, 2837–2845.
- 36 C. Shang, X. Lei, B. Hou, M. Wu, B. Xu, G. Liu and C. Ouyang, *J. Phys. Chem. C*, 2018, **122**, 23899–23909.
- 37 A. Castellanos-Gomez, X. Duan, Z. Fei, H. R. Gutierrez, Y. Huang, X. Huang, J. Quereda, Q. Qian, E. Sutter and P. Sutter, *Nat. Rev. Methods Primers*, 2022, **2**, 58.
- 38 S.-J. Liang, B. Cheng, X. Cui and F. Miao, *Adv. Mater.*, 2020, **32**, 1903800.
- 39 Q. Fu, X. Wang, J. Zhou, J. Xia, Q. Zeng, D. Lv, C. Zhu, X. Wang, Y. Shen, X. Li, *et al.*, *Chem. Mater.*, 2018, **30**, 4001–4007.
- 40 J. Tan, S. Li, B. Liu and H.-M. Cheng, *Small Struct.*, 2021, **2**, 2000093.
- 41 D. M. Seo, J.-H. Lee, S. Lee, J. Seo, C. Park, J. Nam, Y. Park, S. Jin, S. Srivastava, M. Kumar, *et al.*, *ACS Photonics*, 2019, **6**, 1379–1386.
- 42 J. Hafner, *J. Comput. Chem.*, 2008, **29**, 2044–2078.
- 43 G. Kresse and J. Furthmüller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 11169.
- 44 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 45 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1998, **80**, 891.
- 46 G. Kresse and D. Joubert, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **59**, 1758.
- 47 S. Grimme, *J. Comput. Chem.*, 2004, **25**, 1463–1473.
- 48 S. Grimme, *J. Comput. Chem.*, 2006, **27**, 1787–1799.
- 49 J. Heyd and G. E. Scuseria, *J. Chem. Phys.*, 2004, **121**, 1187–1192.
- 50 J. Heyd, J. E. Peralta, G. E. Scuseria and R. L. Martin, *J. Chem. Phys.*, 2005, **123**, 174101.
- 51 P. Zhang, C. Bian, J. Ye, N. Cheng, X. Wang, H. Jiang, Y. Wei, Y. Zhang, Y. Du, L. Bao, *et al.*, *Sci. China Mater.*, 2020, 1548–1559.
- 52 S. R. Tavares, P. I. R. Moraes, R. B. Capaz and A. A. Leitao, *Appl. Surf. Sci.*, 2019, **483**, 762–771.
- 53 Q. Luo, S. Yin, X. Sun, G. Guo and X. Dai, *Diamond Relat. Mater.*, 2022, **128**, 109223.



- 54 Z. Li, J. Han, S. Cao and Z. Zhang, *Appl. Surf. Sci.*, 2023, **157766**.
- 55 Y.-L. Liu, Y. Shi and C.-L. Yang, *Appl. Surf. Sci.*, 2021, **545**, 148952.
- 56 H. Shu, M. Zhao and M. Sun, *ACS Appl. Nano Mater.*, 2019, **2**, 6482–6491.
- 57 Y. Gong, J. Lin, X. Wang, G. Shi, S. Lei, Z. Lin, X. Zou, G. Ye, R. Vajtai, B. I. Yakobson, *et al.*, *Nat. Mater.*, 2014, **13**, 1135–1142.
- 58 J. Bardeen, *Phys. Rev.*, 1947, **71**, 717.
- 59 P.-C. Shen, C. Su, Y. Lin, A.-S. Chou, C.-C. Cheng, J.-H. Park, M.-H. Chiu, A.-Y. Lu, H.-L. Tang, M. M. Tavakoli, *et al.*, *Nature*, 2021, **593**, 211–217.
- 60 T. Su, Y. Li, Q. Wang, W. Zhao, L. Cao and Y. S. Ang, *J. Phys. D: Appl. Phys.*, 2023, **56**, 234001.
- 61 C. C. Tho, C. Yu, Q. Tang, Q. Wang, T. Su, Z. Feng, Q. Wu, C. Nguyen, W.-L. Ong, S.-J. Liang, *et al.*, *Adv. Mater. Interfaces*, 2023, **10**, 2370006.
- 62 L. Cao, X. Deng, Z. Tang, G. Zhou and Y. S. Ang, *Appl. Phys. Lett.*, 2022, **121**, 113104.
- 63 X. Gong, G. Yuan, X. Ma, H. Lv, F. Zhang, J. He and W. Zou, *Diamond Relat. Mater.*, 2022, **128**, 109206.
- 64 W. Xiong, C. Xia, X. Zhao, T. Wang and Y. Jia, *Carbon*, 2016, **109**, 737–746.
- 65 X. Ma, H. Bo, X. Gong, G. Yuan, Z. Peng, J. Lu and Q. Xie, *Appl. Surf. Sci.*, 2023, **615**, 156385.
- 66 X. Yan, I. S. Esqueda, J. Ma, J. Tice and H. Wang, *Appl. Phys. Lett.*, 2018, **112**, 032101.
- 67 Q. Peng, C. Si, J. Zhou and Z. Sun, *Appl. Surf. Sci.*, 2019, **480**, 199–204.
- 68 W. Li, T. Wang, X. Dai, Y. Ma and Y. Tang, *J. Alloys Compd.*, 2017, **705**, 486–491.
- 69 Y. Feng, H. Bai, M. An, Y. Wu and X. Wang, *ACS Appl. Nano Mater.*, 2023, **6**, 5795–15806.
- 70 X. Li, G. Jia, J. Du, X. Song, C. Xia, Z. Wei and J. Li, *J. Mater. Chem. C*, 2018, **6**, 10010–10019.

