

# Coordination polymer particles as potential drug delivery systems†

Inhar Imaz,<sup>a</sup> Marta Rubio-Martínez,<sup>a</sup> Lorena García-Fernández,<sup>a</sup> Francisca García,<sup>b</sup> Daniel Ruiz-Molina,<sup>a</sup> Jordi Hernando,<sup>c</sup> Victor Puntès<sup>a</sup> and Daniel Maspoch<sup>\*a</sup>

Received 12th February 2010, Accepted 5th May 2010

First published as an Advance Article on the web 20th May 2010

DOI: 10.1039/c003084h

Micro- and nanoscale coordination polymer particles can be used for encapsulating and delivering drugs. *In vitro* cancer cell cytotoxicity assays showed that these capsules readily release doxorubicin, which shows anticancer efficacy. The results from this work open up new avenues for metal–organic capsules to be used as potential drug delivery systems.

Micro- and nanomaterials able to encapsulate pharmaceutical agents have been actively explored as carriers for therapy to achieve a prolonged and better controlled drug administration.<sup>1</sup> To date, the vast majority of carriers are based on dendrimers,<sup>2</sup> liposomes,<sup>3</sup> organic polymeric<sup>4</sup> and inorganic particles.<sup>5</sup> In this field, we have recently described a general method for encapsulating desired species into micro- and nanoscale coordination polymer particles<sup>6</sup> *via* a coordination polymerization followed by a fast precipitation,<sup>7</sup> and have demonstrated its utility by entrapping organic dyes, quantum dots and magnetic nanoparticles into blue fluorescent spheres (hereafter referred to as Zn(bix)) created by connecting Zn<sup>2+</sup> metal ions through 1,4-bis(imidazol-1-ylmethyl)benzene (bix) organic ligands. Owing to their small size and ability for entrapping a wide variety of substances, we believe that coordination polymer nanospheres also show promise for encapsulating drugs and thus being used as novel functional carriers for drug delivery.

Thus far, some advances have been made in using metal–organic frameworks (MOFs) as drug delivery systems.<sup>8–10</sup> Férey's group first described promising adsorption and release properties of ibuprofen on bulk hybrid inorganic–organic solids,<sup>8</sup> whereas Lin *et al.* have structured a Pt-based drug at the nanoscale by using it as one building block for creating the framework of a coordination polymer.<sup>9</sup> Very recently, Horcajada, Gref and co-workers have also shown that porous crystalline nano-MOFs can adsorb and release several drugs, thus acting as potential non-toxic drug nanocarriers.<sup>10</sup> Herein we wish to report an alternative general methodology for *in situ* encapsulating unmodified drugs into metal–organic frameworks in the form of micro- and nanoscale spherical particles (Fig. 1a). Doxorubicin (DOX), SN-38, camptothecin (CPT) and daunomycin (DAU) were chosen as archetypical



**Fig. 1** (a) Schematic illustration describing the encapsulation of drugs into metal–organic spheres created by the connection of metal ions, such as Zn<sup>2+</sup>, through multitopic organic ligands, such as bix. (b) SEM and (c) TEM images of a representative colloidal solution of DOX/Zn(bix) spheres.

drugs because of their current use for cancer therapy and fluorescence properties that facilitate their monitorization.<sup>11</sup> The encapsulation of all these drugs, their release and initial *in vitro* cellular studies serve as excellent tests to preliminary illustrate the use of coordination polymer spheres as drug delivery systems.

In a typical experiment, metal–organic Zn(bix) spheres with encapsulated DOX [DOX/Zn(bix)], SN-38 [SN-38/Zn(bix)], CPT [CPT/Zn(bix)] and DAU [DAU/Zn(bix)] were prepared by addition of an aqueous solution of Zn(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O to an ethanolic solution of bix containing the drug (*c* ~ 3.3 × 10<sup>-3</sup> M) under stirring at room temperature. The resulting spheres were then purified by centrifugation and washed several times with ethanol. This process was repeated until no fluorescence signal from free, non-encapsulated drug was detected in the supernatant solution. Fig. 1b and c show representative scanning (SEM) and transmission (TEM) electron microscopy images of the resulting coordination polymer spherical capsules, whose diameter was controlled from 100 to 1500 nm by adjusting the initial concentrations of the reactants.<sup>6</sup>

The entrapment of DOX, CPT, SN-38 and DAU into these spheres was confirmed by both absorption and fluorescence measurements, and encapsulation efficiencies up to 21% of the initial drug concentration were measured. In all cases, the fluorescence emission spectrum of the colloids matches that of

<sup>a</sup> Centro de Investigación en Nanociencia y Nanotecnología (ICN-CSIC) Campus UAB, 08193 Bellaterra, Spain.  
E-mail: daniel.maspoch.icn@uab.es

<sup>b</sup> Departament de Química, Universitat Autònoma de Barcelona, Campus UAB, 08193-Bellaterra, Spain

<sup>c</sup> Institut de biotecnologia i de biomedicina, Campus UAB, 08193-Bellaterra, Spain

† Electronic supplementary information (ESI) available: Detailed synthetic and experimental procedures, SEM, TEM, fluorescence optical and DLS images, and *in vitro* cytotoxicity assay curves. See DOI: 10.1039/c003084h

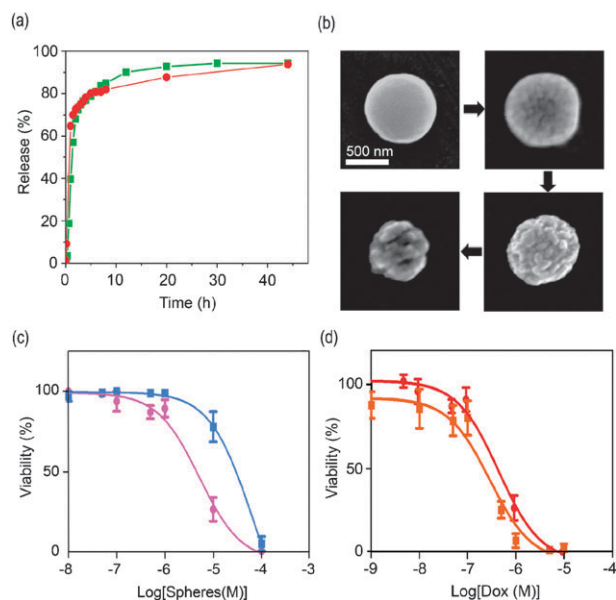


**Fig. 2** (a–d) Fluorescence optical microscope images of (a) CPT/Zn(bix), (b) SN-38/Zn(bix), (c) DOX/Zn(bix), and (d) DAU/Zn(bix) spheres. (e–h) Fluorescence emission spectra of (e) free CPT and CPT/Zn(bix) (collected at  $\lambda_{\text{exc}} = 355$  nm), (f) free SN-38 and SN-38/Zn(bix) (collected at  $\lambda_{\text{exc}} = 450$  and  $355$  nm), (g) free DOX and DOX/Zn(bix) (collected at  $\lambda_{\text{exc}} = 490$  and  $355$  nm), and (h) free DAU and DAU/Zn(bix) (collected at  $\lambda_{\text{exc}} = 490$  and  $355$  nm).

free DOX, DAU, SN-38 and CPT (Fig. 2e–h), which results in blue [CPT/Zn(bix)], green [SN-38/Zn(bix)] and red [DOX/Zn(bix) and DAU/Zn(bix)] emitting spheres upon selective excitation of the drug (Fig. 2a–d). Furthermore, since bare Zn(bix) particles already show blue luminescence when excited at  $355$  nm, broadband fluorescence spectra are measured upon UV excitation, which show contributions from both the drug and Zn(bix) spheres emission (Fig. 2e–h).

To investigate the drug release from coordination polymer particles, we first prepared colloidal solutions of  $300 \pm 23$  nm in diameter DOX/Zn(bix) and SN-38/Zn(bix) spheres in phosphate buffered saline solution (PBS) at  $\text{pH} = 7.4$ , and the resulting colloids were placed in a dialysis bag (cut-off molecular weight:  $3500$ ) at  $37^\circ\text{C}$ . Both dispersions were dialyzed against  $100$  mL of PBS for  $48$  hours. The DOX and SN-38 release profiles measured by fluorescence spectroscopy are depicted in Fig. 3a. At  $37^\circ\text{C}$ , Zn(bix) spheres containing those drugs showed a fast release of  $\sim 80\%$  at  $8$  hours, followed by an additional release of  $\sim 15\%$  over the next  $2$  days. The DOX release from Zn(bix) spheres was also perceptible with naked eyes because the initial violet-pink colour of the colloid gradually changed to a white colour characteristic of bare Zn(bix) spheres.

A number of factors can contribute to the release of DOX and SN-38 from coordination polymer particles, including desorption of drug adsorbed on the sphere surface, diffusion of drug through the coordination polymeric sphere, and erosion of the sphere. Similar to other encapsulating systems, the fast release can be attributed to both desorption and diffusion of drug as well as to the gradual erosion of Zn(bix) spheres in PBS for subsequent diffusion from the surface or through developing pores.<sup>12</sup> To confirm the erosion of Zn(bix) spheres at  $37^\circ\text{C}$ , a colloidal PBS solution of DOX/Zn(bix) spheres was maintained at this temperature and sequentially investigated by SEM and dynamic light scattering (DLS). Fig. 3b shows representative SEM images showing the time-dependent degradation of such spheres. As can be seen there, the surface of Zn(bix) spheres becomes gradually rougher and



**Fig. 3** (a) *In vitro* release profile of DOX and SN-38 from DOX/Zn(bix) (dot, red) and SN-38/Zn(bix) (square, green) spheres incubated in  $\text{pH} 7.2$  PBS at  $37^\circ\text{C}$ . (b) SEM micrographs of DOX/Zn(bix) spheres taken at  $1, 4, 8,$  and  $24$  hours, showing representative degradation in  $\text{pH} 7.4$  PBS at  $37^\circ\text{C}$ . (c and d) *In vitro* cytotoxicity assay curves after  $24$  h for HL60 cells obtained by plotting the cell viability percentage against the (c) Zn(bix) (square, blue) and DOX/Zn(bix) (dot, pink) concentration and (d) the DOX release from DOX/Zn(bix) spheres (dot, red) and DOX (square, orange) concentration.

more cracked, reducing their overall volume with time, as also confirmed by DLS. Noticeably, such process was found to slow down at room temperature.

Our abilities to synthesize coordination polymer spheres for encapsulating drugs and the capacity to release them prompted us to preliminary evaluate their anticancer efficacies. We performed *in vitro* cytotoxicity assays on HL60 (Human promyelocytic leukemia cells) cell line with  $24$  h and  $48$  h of

incubation (Fig. 3c). Treatment of HL60 with Zn(bix) and DOX/Zn(bix) spheres did not lead to any appreciable cell death after 24 h and 48 h of incubation at concentrations ranging from 0.01  $\mu\text{M}$  to 0.5  $\mu\text{M}$ . However, between 0.5  $\mu\text{M}$  and 10  $\mu\text{M}$ , some differences were observed between both types of spheres. In this range, the DOX from DOX/Zn(bix) spheres acts on the cells notably reducing the cell-viability down to 25% at 10  $\mu\text{M}$ , whereas the Zn(bix) spheres gave a cell-viability close to 80%. Overall, DOX/Zn(bix) spheres gave a half maximal inhibitory concentration ( $\text{IC}_{50}$ ) of 5.2  $\mu\text{M}$  and 4.5  $\mu\text{M}$  after 24 h and 48 h of incubation, respectively, whereas the non-encapsulated Zn(bix) spheres had respective  $\text{IC}_{50}$  values of only 62.5  $\mu\text{M}$  and 99.9  $\mu\text{M}$ . These results demonstrate that at this range the Zn(bix) matrix has a low contribution on the cytotoxic effects, and confirms that delivered DOX from DOX/Zn(bix) spheres is the major responsible of their cytotoxic activity against HL60.

To further evaluate the cytotoxic effect of the DOX released from DOX/Zn(bix) spheres, HL60 cells were also treated with free DOX. Fig. 3d shows the dose response of free DOX and DOX from DOX/Zn(bix) spheres after 24 h of incubation.<sup>13</sup> In both cases, an appreciable cell death was observed at concentrations ranging from 0.1  $\mu\text{M}$  and 10  $\mu\text{M}$ . Indeed, both free DOX and DOX from DOX/Zn(bix) spheres gave similar  $\text{IC}_{50}$  values of 0.3  $\mu\text{M}$  and 0.4  $\mu\text{M}$ , respectively. These results suggest that DOX/Zn(bix) spheres have similar cytotoxic effects against HL60 than free DOX. Once encapsulated, the DOX released from the metal–organic DOX/Zn(bix) spheres can induce the cell death in cancer cells.

In summary, the presented results show that coordination polymer micro- and nanospheres constitute a novel and promising type of materials to be used as functional matrices for encapsulating a large panel of drugs. We first demonstrate the release of drugs from coordination polymer capsules, and their potential anticancer efficacies *in vitro*. As coordination polymers with a broad range of structures and functionalities can be prepared, this approach could be generalized for obtaining metal–organic delivery systems with novel compositions and functionalities. Future work includes the *in vivo* studies of drug delivery and targeting of coordination polymer particles as well as the development of metal–organic capsules with novel properties and functionalities.

This work was supported by projects VALTEC08-2-003, MAT2009-13977-C03 and CTQ2009-07469. D. M. and I. I. thank the Ministerio de Ciencia y Tecnología for respective RyC and JdC contracts. M. R.-M. and L. G.-F. thank the Institut Català de Nanotecnologia for research fellowships. The authors thank the Servei de Microscopia of the UAB.

## Notes and references

- (a) R. Langer, *Nature*, 1998, **392**, 5; (b) T. M. Allen and P. R. Cullis, *Science*, 2004, **303**, 1818.
- (a) J. M. J. Fréchet, *J. Polym. Sci., Part A: Polym. Chem.*, 2003, **43**, 3713; (b) S. Svenson and D. A. Tomalia, *Adv. Drug Delivery Rev.*, 2005, **57**, 2106.
- (a) F. M. Muggia, J. D. Hainsworth, S. Jeffers, P. Miller, S. Groshen, M. Tan, L. Roman, B. Uziely, L. Muderspach, A. Garcia, A. Burnett, F. A. Greco, C. P. Morrow, L. J. Paradiso and L.-J. Liang, *J. Clin. Oncol.*, 1997, **15**, 987; (b) D. D. Lasic and D. Papahadjopoulos, *Science*, 1995, **267**, 1275.
- (a) J. Panyam and V. Labhasetwar, *Adv. Drug Delivery Rev.*, 2003, **55**, 329; (b) B. J. Nehilla, P. G. Allen and T. A. Desai, *ACS Nano*, 2008, **2**, 538.
- (a) H. Otsuka, Y. Nagasaki and K. Kataoka, *Adv. Drug Delivery Rev.*, 2003, **55**, 403; (b) I. Brigger, C. Dubernet and P. Couvreur, *Adv. Drug Delivery Rev.*, 2002, **54**, 631.
- (a) I. Imaz, J. Hernandez, D. Ruiz-Molina and D. Maspoch, *Angew. Chem., Int. Ed.*, 2009, **48**, 2325.
- (a) M. Oh and C. A. Mirkin, *Nature*, 2005, **438**, 651; (b) X. Sun, S. Dong and E. Wang, *J. Am. Chem. Soc.*, 2005, **127**, 13102; (c) H. Maeda, M. Hasegawa, T. Hashimoto, T. Kakimoto, S. Nishio and T. Nakanishi, *J. Am. Chem. Soc.*, 2006, **128**, 10024; (d) K. H. Park, K. Jang, S. U. Son and D. A. Sweigart, *J. Am. Chem. Soc.*, 2006, **128**, 8740; (e) H. Wei, B. Li, Y. Du, S. Dong and E. Wang, *Chem. Mater.*, 2007, **19**, 2987; (f) S. Jung and M. Oh, *Angew. Chem., Int. Ed.*, 2008, **47**, 2049; (g) M. Oh and C. A. Mirkin, *Angew. Chem., Int. Ed.*, 2007, **45**, 5492; (h) Y.-M. Jeon, G. S. Armatas, J. Heo, M. G. Kanatzidis and C. A. Mirkin, *Adv. Mater.*, 2008, **20**, 2105; (i) I. Imaz, D. Maspoch, C. Rodríguez-Blanco, J.-M. Pérez-Falcón, J. Campo and D. Ruiz-Molina, *Angew. Chem., Int. Ed.*, 2008, **47**, 1860.
- (a) P. Horcajada, C. Serre, M. Vallet-Regí, M. Sebban, F. Taulelle and G. Férey, *Angew. Chem., Int. Ed.*, 2006, **45**, 5974; (b) P. Horcajada, C. Serre, G. Maurin, N. A. Ramsahye, F. Balas, M. Vallet-Regí, M. Sebban, F. Taulelle and G. Férey, *J. Am. Chem. Soc.*, 2008, **130**, 6774.
- W. J. Rieter, K. M. Pott, K. M. L. Taylor and W. Lin, *J. Am. Chem. Soc.*, 2008, **130**, 11584.
- P. Horcajada, T. Chalati, C. Serre, B. Gillet, C. Sebrie, T. Baati, J. F. Eubank, D. Heurtaux, P. Clayette, C. Kreuz, J.-S. Chang, Y. K. Hwang, V. Marsaud, P.-N. Bories, L. Cynober, S. Gil, G. Férey, P. Couvreur and R. Gref, *Nat. Mater.*, 2010, **9**, 172.
- (a) M. Muggia, J. D. Hainsworth, S. Jeffers, P. Miller, S. Groshen, M. Tan, L. Roman, B. Uziely, L. Muderspach, A. Garcia, A. Burnett, F. A. Greco, C. P. Morrow, L. J. Paradiso and L.-J. Liang, *J. Clin. Oncol.*, 1997, **15**, 987; (b) H. L. Wong, R. Bendayan, A. M. Rauth, Y. Li and X. Y. Wu, *Adv. Drug Delivery Rev.*, 2007, **59**, 491.
- (a) R. P. Batycky, J. Hanes, R. Langer and D. A. Edwards, *J. Pharm. Sci.*, 1997, **86**, 1464; (b) N. Faisant, J. Siepmann and J. P. Benoit, *Eur. J. Pharm. Sci.*, 2002, **15**, 355; (c) J. Wang, B. M. Wang and S. P. Schwendeman, *J. Controlled Release*, 2002, **82**, 289; (d) C. Berkland, K. Kim and D. W. Pack, *Pharm. Res.*, 2003, **20**, 1055.
- The concentration of delivered DOX from DOX/Zn(bix) spheres was calculated according to a maximum encapsulation efficiency of 21% with respect to the initial DOX concentration used for the encapsulation process, and a release of 88% (for 24 h) of the encapsulated DOX, according to the *in vitro* delivery studies.