ORGANIC CHEMISTRY

FRONTIERS

RESEARCH ARTICLE



View Article Online View Journal | View Issue

Cite this: Org. Chem. Front., 2014, **1**, 79

Benzoate dioxygenase from *Ralstonia eutropha* B9 – unusual regiochemistry of dihydroxylation permits rapid access to novel chirons†

Julia A. Griffen,^a Sarah J. Kenwright,^a Sarah Abou-Shehada,^a Scott Wharry,^b Thomas S. Moody*^b and Simon E. Lewis*^a

Received 26th November 2013, Accepted 20th December 2013 DOI: 10.1039/c3qo00057e Oxidation of benzoic acid by a microorganism expressing benzoate dioxygenase leads to the formation of an unusual *ipso, ortho* arene *cis*-diol in sufficient quantities to be useful for synthesis. This homochiral diol possesses an array of differentiated functionality which can be exploited to access diverse highly oxygenated structures by concise synthetic sequences.

Introduction

rsc.li/frontiers-organic

Of the various oxygenases that have found use in biocatalysis, it is arguably the arene dioxygenases which add the most value in terms of the synthetic versatility of the products they produce.¹ The direct transformation of an aromatic ring into a dearomatised cyclohexadiene diol (Scheme 1) is a reaction that has very little precedent in organic chemistry² and is therefore appealing to access rapidly uncharted chemical space. In wildtype organisms, these arene *cis*-diols are usually fleeting metabolic intermediates.³ However, mutants in which the subsequent enzyme in the metabolic pathway is blocked are able to accumulate these diols and they can be isolated in synthetically useful quantities.

The densely-packed, diverse functionality in these chirons finds ready application in different areas such as synthesis of



Scheme 1 Regio- and stereoselectivity of dioxygenases.

natural products,⁴ pharmaceuticals,⁵ carbohydrates,⁶ polymers⁷ and dyes.⁸ To date, in excess of 400 arene *cis*-diol products have been reported. The majority of these are produced by organisms expressing toluene dioxygenase (TDO), naphthalene dioxygenase (NDO) and biphenyl dioxygenase (BPDO) enzymes, which are Rieske type non-heme iron oxygenases.^{1e,9} These metabolise substituted aromatic substrates in a regioand stereoselective fashion. A robust predictive model has been developed for such transformations,¹⁰ with the sense of enantioinduction being consistent across organisms and substrates (Scheme 1a, ortho, meta oxygenation). However, organisms expressing benzoate dioxygenase (BZDO) enzymes dihydroxylate benzoic acids in a process that proceeds with both different regioselectivity and also the opposite absolute sense of enantioinduction. For example, Ralstonia eutropha B9¹¹ (formerly known as Alcaligenes eutrophus B9), Pseudomonas putida U103¹² and Pseudomonas putida KTSY01 (pSYM01)¹³ oxidise benzoic acid to benzoate 1,2-cis dihydrodiol 4 (Scheme 1b, ipso, ortho oxygenation).

Diol acid 4 is a highly versatile chiral pool starting material and many transformations of this building block can be envisaged (Fig. 1). Despite this, 4 has been comparatively underutilised to date in synthesis, in comparison with arene *cis*-diols of



Fig. 1 Possible transformations of diol acid 4.

^aCentre for Sustainable Chemical Technologies, University of Bath, Claverton Down, Bath BA2 7AY, UK. E-mail: S.E.Lewis@bath.ac.uk

^bDepartment of Biocatalysis and Isotope Chemistry, Almac Group Ltd, 20 Seagoe Industrial Estate, Craigavon, BT63 5QD, UK. E-mail: tom.moody@almacgroup.com †Electronic supplementary information (ESI) available: ¹H and ¹³C NMR spectra for all novel compounds, as well as selected 2D NMR spectra. See DOI: 10.1039/ c3q000057e

type 2. $^{4i,q,t,5h,i,6a,e,o,14-17}$ In the current work, we describe the synthesis of a library of cyclohexyl chirons from 4, both minimally and more extensively functionalised. This serves to showcase further the versatility of 4 and we anticipate these new building blocks will find diverse applications in synthesis and catalysis. With regards to handling and storage, it should be noted that although 4 is prone to exothermic decomposition by rearomatisation, as are all arene *cis*-diols, it may be stored in pure form in excess of a year at -78 °C without appreciable decomposition occurring. Additionally, storage of 4 as its mixed sodium/potassium salt has been described and reportedly leads to enhanced stability.^{16b} Production of 4 on a multihundred gram scale is possible without recourse to specialised equipment.¹⁵ The absolute configuration and enantiopurity of 4 have been demonstrated through chemical correlation and by X-ray crystallographic analysis of a derivative.14,15

Results and discussion

Ring-saturated derivatives

The diene in 4 readily undergoes hydrogenation over palladium on carbon to give saturated cyclohexane diol acid 5 (Scheme 2). Perhaps surprisingly, this compound has not been reported previously, although the diastereoisomeric trans-diol is known.¹⁸ Saturation of *ortho,meta* arene *cis*-diols of type 2 to give 3-substituted cyclohexane-1,2-diols and applications of these in catalysis have been reported;¹⁹ the analogous approach has not previously been applied to ipso, ortho- arene cis-diols of type 4, however. We wished to target derivatives of 5 with the diol protected, but direct acetonide introduction was surprisingly unsuccessful. To circumvent this, esterification of 5 was carried out prior to ketalisation. Acetonide protection of 6 was successful, albeit with traces of 8 being formed through competing transesterification. Subsequent hydrolysis of ester 7 did indeed give desired acetonide acid 9, but appreciable acetonide migration and deprotection were also observed.



Other chiral acid building blocks were also targeted; to this end, the diol in benzyl ester **11** was protected as dioxasilole **13** prior to hydrogenation to give **14** (Scheme 4). This last step proved capricious, however, with undesired desilylation also occurring to a varying extent. A bis(ether) derivative was also accessed by permethylation of **5**. With a slight excess of alkylating agent incomplete ether formation led to an inseparable mixture of bis(ether) **15** and monoether **16**, which required silylation to allow separation to be effected. However, a greater excess of alkylating agent led to clean formation of **15** in good yield and hydrolysis gave bis(ether) acid **18**.

Diol protection as a benzylidene acetal was explored and a moderate (3:2) diastereoselectivity was observed for formation of **19** over **20**. Structures were assigned on the basis of NOESY correlations (see ESI†); careful chromatography allowed for isolation of both diastereoisomers in pure form. We next sought to access novel ketones bearing an adjacent quaternary centre. One such target, **21**, was available simply by oxidation of



Scheme 3 Benzyl ester route to 9.



Scheme 2 Acetonide acid derivatives of 4.



Scheme 4 Other protected acid building blocks.

OTBDMS

29

OTBDMS

31

ó

O₂, TPF

hv, CCl₄, 12 h

10 °C, 90%

Ref 4

MOMO

MOMCI

[/]Pr₂NEt

CH₂Cl₂, rt

16 h. 66%

1.) H₃O[⊕]

2.) Ac₂O, pv

OTBDMS

20

OTBDMS

30

Ό

OTBDMS

QН

но

MOM

thiourea

CH₂Cl₂ / MeOH

rt, 90 min, 99%

Ref. 4

OsO4, NMO

acetone / H₂O

rt. 72 h. 67%



byproduct **10**. A second such ketone was accessed by a multistep procedure involving silyl protection of the secondary alcohol in **4**. Thus, ester **6** was silylated and reduced to give monoprotected triol **23**. Choice of reductant proved crucial, since with LiAlH₄, yields of **23** were low, with appreciable silyl migration observed, giving rise to **24**. In contrast, NaBH₄ cleanly effected reduction to **23** only, in good yield. Ketalisation, desilylation and oxidation then gave target ketone **27**, a reduced analogue of **21** (Scheme 5).

These comparatively minimally functionalised cyclohexyl chirons that all bear a quaternary stereocentre are synthetically valuable insofar as it is difficult to conceive of other means of accessing them as easily, in enantiopure form. We anticipate their finding diverse uses in synthesis and catalysis.

Highly oxygenated derivatives

The structures of arene *cis*-diols such as 2 and 4 are highly suggestive of applications in the synthesis of cyclitols such as inositols.²² A particular advantage of their use for synthesis of novel inositol derivatives is that they provide ready access to *C*-substituted derivatives.^{6d,23} In contrast, use of natural inositols or other carbohydrates as starting materials lends itself to synthesis of *O*-substituted derivatives, but *C*-substituted derivatives are accessible only by means of more involved synthetic sequences.

The *ortho,meta* diols of type **2** have been extensively exploited in this context^{6i,j,p,r,7a,24-27} but no inositol derivatives have been synthesised to date from **4**. To access rapidly such a species from **4**, we opted to introduce oxygenation into the diene by means of a singlet oxygen photocycloaddition. Use of singlet oxygen to access novel cyclitols from starting materials other than arene *cis*-diols has previously proven to be a very successful strategy.²⁸ Thus, known silyl ether **28** was transformed to endoperoxide **29** and reduced to protected pentaol



and ammonolysis to give 34, a C-hydroxymethyl-muco-inositol. C-Hydroxymethyl derivatives of other isomeric inositols are extremely rare³⁰ and those of muco-inositol are wholly unknown (Scheme 6). Endoperoxides derived from diene-¹O₂ photocycloaddition are versatile intermediates and in addition to reductive cleavage, they are capable of undergoing several other synthetically useful transformations. For example, they undergo basemediated Kornblum-DeLaMare fragmentation to afford γ -hydroxyenones; we have previously demonstrated such transformations for endoperoxides derived from 4.4i In addition, they may be isomerised to the corresponding bis(epoxides) upon treatment with cobalt-tetraphenylporphine complex.³¹ This latter transformation has not previously been applied to an endoperoxide derived from 4. To this end, known alcohol 35 was acetylated and subjected to the photocycloaddition. In addition to the expected endoperoxide 37, epoxide 38 was also isolated in small amounts. The structure of 38 was assigned on the basis of its spectroscopic data in comparison with those for the previously reported 39.15 As regards the mechanism of formation of 38, analogous epoxide byproducts of singlet oxygen photocycloaddition have been described previously for other substrates and are believed to arise via a radical pathway.³² Upon treatment with CoTPP, 37 underwent facile isomerisation to bis(epoxide) 40. These epoxides proved resistant to opening with ammonia, with ammonolysis instead

cleanly removing the acetate to give 41 (Scheme 7).



Scheme 7 Synthesis of a bis(epoxide).

Conclusions

We have described synthetic sequences which allow for the functionalization of every position on the cyclohexadiene ring of **4**. As the stereocentres in **4** are in close proximity to the diene, it has proven possible to introduce additional stereocentres in a highly selective fashion under substrate control. We anticipate that the novel chirons described here may find use in the synthesis of more complex targets.

Experimental

General procedures

Reactions were carried out under an atmosphere of nitrogen. In most cases, solvents were obtained by passing through anhydrous alumina columns using an Innovative Technology Inc. PS-400-7 solvent purification system. All other solvents were purchased as "anhydrous" grade from Fisher Scientific. "Petrol" refers to petroleum spirit b.pt 40-60 °C. TLC was performed using aluminium backed plates precoated with Alugram® SIL G/UV 254 nm. Visualization was accomplished by UV light and/or KMnO₄ followed by gentle warming. Organic layers were routinely dried with anhydrous MgSO4 and evaporated using a Büchi rotary evaporator. When necessary, further drying was facilitated by high vacuum. Flash column chromatography was carried out using Davisil LC 60 Å silica gel (35-70 micron) purchased from Fisher Scientific. IR spectra were recorded on a Perkin-Elmer 1600 FT IR spectrometer with only selected absorbances quoted as ν in cm⁻¹. NMR spectra were run on Bruker Avance 250, 300, 400 or 500 MHz instruments at 298 K. A micrOTOF electrospray timeof-flight (ESI-TOF) mass spectrometer (Bruker Daltonik GmbH, Bremen, Germany) was used; this was coupled to an Agilent 1200 LC system (Agilent Technologies, Waldbronn, Germany). The LC system was used as an autosampler only. 10 µL of sample was injected into a 30:70 flow of water-acetonitrile at 0.6 mL min⁻¹ to the mass spectrometer. For each acquisition 10 µL of calibrant of 5 mM sodium formate was injected after the sample. The observed mass and isotope

pattern matched the corresponding theoretical values as calculated from the expected elemental formula.

Synthesis of (1S,2R)-1,2-dihydroxycyclohexane-1-carboxylic acid (5). A stirred solution of 4 (857 mg, 5.48 mmol) and Pd/C (50 mg, matrix activated carbon support) in MeOH (20 mL) was exposed to a hydrogen atmosphere (balloon) at room temperature. After 24 h the solution was filtered through a plug of celite and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (60:35:2.5:2.5 EtOAc-petrol-H₂O-AcOH) to vield pure 5 (625 mg, 71%) as a white crystalline solid. m.pt = 112-115 °C; $R_{\rm f} = 0.10$ (60:35:2.5:2.5 EtOAc-petrol-H₂O-AcOH); $[\alpha]_{\rm D}^{25}$ +13.3 (c 0.3, H₂O); ¹H NMR (300 MHz, D₂O) δ = 3.77 (1H, dd, J = 11.5, 3.5 Hz, CH(OH)), 1.60-1.56 (4H, m, CH), 1.40-1.08 (4H, m, CH); ¹³C NMR (125 MHz, MeOH- d_4) δ = 180.1 (C=O), 79.5 (C(OH)CO₂H), 73.7 (CH(OH)), 35.0, 30.8, 25.3, 21.1; v_{max} (film) 3445, 3086, 2924, 2858, 1736, 1439, 1199, 1138, 1023, 948, 821 cm⁻¹; HRMS (ESI-) m/z calcd for $(C_7H_{11}O_4)^-$, 159.0663; found 159.0672.

Synthesis of methyl (1S,2R)-1,2-dihydroxycyclohexane-1-carboxylate (6). To a stirred solution of 5 (47 mg, 0.29 mmol) dissolved in MeOH-C₆H₆ (6 mL, 1:1), was added dropwise TMS-CHN₂ (0.200 mL, 2.0 M solution in THF, 0.400 mmol, 1.4 equiv.) until gas evolution ceased and a yellow colour persisted. The reaction mixture was concentrated under reduced pressure (caution: TMS-CHN₂ toxic by inhalation). Pure 6 was obtained as a yellow oil (52 mg, 100%). Rf = 0.05 (30% EtOAcpetrol); $[\alpha]_D$ –10.0 (c 0.3, CHCl₃); ¹H NMR (250 MHz, CDCl₃) δ = 3.82–3.74 (1H, m, CH(OH)), 3.77 (3H, s, CH₃), 2.86 (2H, br s, OH), 1.83–1.22 (8H, m, CH); ¹³C NMR (75 MHz, CDCl₃) δ = 176.6 (C=O), 76.8 (C(OH)(CO₂CH₃)), 72.2 (CH(OH)), 52.9 (OCH₃), 34.2, 30.1, 24.0, 19.8; v_{max} (film) 3453, 2938, 2861, 1729, 1438, 1272, 1238, 1207, 1149, 1079, 998, 923 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_8H_{15}O_4)^+$, 175.0965; found 175.0981. NMR Data are in agreement with those previously reported.33

Synthesis of methyl (3aS,7aR)-2,2-dimethyltetrahydrobenzo-[d][1,3]dioxole-3a(4H)-carboxylate (7) and (1R,2S)-2-hydroxy-2-(methoxycarbonyl)cyclohexyl (3aS,7aR)-2,2-dimethyltetrahydrobenzo[d][1,3]dioxole-3a(4H)-carboxylate (8). To a stirred solution of 6 (504 mg, 2.89 mmol) dissolved in acetone (20 mL, freshly distilled), was added 2,2-dimethoxypropane (10 mL) and p-toluenesulfonic acid (25 mg, 0.145 mmol, 5 mol%). The solution was stirred at room temperature for 20 h, then diluted with EtOAc (10 mL) followed by the addition of $NaHCO_{3(aq)}$ (1.0 M, 20 mL). The biphasic system was extracted with EtOAc $(4 \times 10 \text{ mL})$ and the organic layers combined and washed with saturated brine, dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography ($0 \rightarrow 20\%$ EtOAc-petrol) to give 7 as a colourless oil (392 mg, 63%) and 8 as a colourless oil (20 mg, 2%). Data for 7: $R_{\rm f} = 0.70$ (30% EtOAc-petrol); $[\alpha]_{\rm D}^{25} - 37.8$ $(c \ 0.65, \text{CHCl}_3)$; ¹H NMR (300 MHz, CDCl₃) $\delta = 4.35$ (1H, t, J = 3.5 Hz, CH(OC(CH₃)₂), 3.76 (3H, s, OCH₃), 2.06-1.93 (2H, m, CH), 1.85–1.23 (6H, m, CH); 1.50 (3H, d, J = 0.5 Hz, CH₃) 1.35 (3H, d, J = 0.5 Hz, CH₃); ¹³C NMR (75 MHz, CDCl₃) $\delta = 173.2$

(C=O), 109.0 (*C*(CH₃)₂), 81.0 *C*(OC(CH₃)₂)(CO₂CH₃), 75.0 (*C*H(OC(CH₃)₂)), 52.5 (OCH₃), 32.3, 28.0, 26.0, 25.9, 20.5, 18.7; $\nu_{\rm max}$ (film) 2997, 2941, 2873, 1733, 1450, 1383, 1217, 1160, 1054, 1025, 905, 725 cm⁻¹; HRMS (ESI+) *m/z* calcd for (C₁₁H₁₈NaO₄)⁺, 237.1097; found 237.1079. Data for **8**: *R*_f = 0.85 (50% EtOAc-petrol); $[\alpha]_{\rm D}^{25}$ -21 (*c* 1.6, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ = 5.08 (1H, dd, *J* = 10.0, 6.0 Hz, CH-(OC=O)), 4.26 (1H, t, *J* = 3.2 Hz, CH(O)), 3.73 (3H, s, OCH₃), 3.24 (1H, s, OH), 2.06–1.24 (16H, m, CH), 1.50 (3H, s, CH₃), 1.35 (3H, s, CH₃); ¹³C NMR (125 MHz, CDCl₃) δ = 175.3 (C=O), 171.8 (C=O), 108.9 (*C*(CH₃)₃), 80.7, 75.6, 75.2, 74.9, 53.0 (OCH₃), 34.1, 32.2, 27.8, 26.2, 25.7, 25.6, 23.6, 20.2, 19.7, 18.3; $\nu_{\rm max}$ (film) 3523, 2987, 2938, 2865, 1733, 1449, 1381, 1370, 1243, 1216, 1152, 1124, 1046, 1003, 874, 735 cm⁻¹; HRMS (ESI+) *m/z* calcd for (C₁₈H₂₉O₇)⁺, 357.1908; found 357.1929.

Synthesis of (3aS,7aR)-2,2-dimethylhexahydrobenzo[d][1,3]dioxole-3a-carboxylic acid (9) and (5S,6R)-6-hydroxy-2,2dimethyl-1,3-dioxaspiro[4.5]decan-4-one (10) from ester 7. Ester 7 (253 mg, 1.18 mmol) was dissolved in THF (10 mL) and NaOH (6.0 mL, 2.0 M in H₂O, 12 mmol, 10 equiv.) and refluxed for 14 h. The resulting solution was washed with EtOAc $(3 \times 10 \text{ mL})$ to remove any unreacted starting material. The aqueous layer was acidified to pH 2.0 with HCl and extracted with EtOAc (3 × 10 mL) and combined organic layers were washed with saturated brine and dried over MgSO₄. The resulting oil was purified via flash column chromatography (50% EtOAc-petrol) to give desired acid 9 (20 mg, 8%), acetonide migration product 10 (30 mg, 13%) and deprotected diol acid 5 (40 mg, 21%). Data for 9: $R_{\rm f}$ = 0.40 (50% EtOAc-petrol); $[\alpha]_{D}^{25}$ -40.0 (c 0.70, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ = 4.31 $(1H, t, J = 3.5 Hz, CH(OC(CH_3)_2), 2.12-1.97 (2H, m, CH),$ 1.90-1.78 (1H, m, CH), 1.71-1.51 (4H, m, CH), 1.54 (3H, s, CH₃), 1.43 (3H, s, CH₃), 1.28–1.21 (1H, m, CH); ¹³C NMR (75 MHz, $CDCl_3$) δ = 175.6 (C=O), 109.6 ($C(CH_3)_2$), 80.7 $C(OC(CH_3)_2)(CO_2H)$, 74.9 ($CH(OC(CH_3)_2)$), 32.2, 28.0, 26.0, 25.7, 20.7, 18.5; v_{max} (film) 2994, 2939, 2873, 2714, 1450, 1371, 1383, 1217, 1174, 905, 725 cm⁻¹; HRMS (ESI-) m/z calcd for $(C_{10}H_{15}O_4)^-$, 199.0976; found 199.0975. Data for 10: $R_f = 0.83$ (50% EtOAc-petrol); $[\alpha]_{D}^{25}$ +2.0 (c 1.5, CHCl₃); ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) \delta = 3.75 (1\text{H}, \text{dd}, J = 11.5, 4.6 \text{ Hz}, \text{CH}(\text{OH})),$ 1.95-1.93 (1H, m, CH), 1.92-1.89 (1H, m, CH), 1.88 (1H, br s, OH), 1.77-1.80 (1H, m, CH), 1.73 (1H, td, J = 14.0, 4.5 Hz, CH), 1.66 (3H, s, CH₃), 1.63 (3H, s, CH₃), 1.49-1.55 (1H, m, CH), 1.44–1.48 (1H, m, CH), 1.38 (1H, tt, J = 13.0, 3.5 Hz, CH); ¹³C NMR (125 MHz, CDCl₃) δ = 174.0 (C=O), 110.1 (C(CH₃)₂), 83.2 (C(O)(C=O), 71.0 (C(OH)H), 33.7 (CH₂), 30.6 (CH₂), 29.1 $(C(CH_3)_2)$, 27.9 $(C(CH_3)_2)$, 23.6 (CH_2) , 20.1 (CH_2) ; ν_{max} (film) 3484, 2991, 2939, 2863, 2774, 1448, 1385, 1291, 1262, 1060, 1036, 908, 860, 626 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{10}H_{17}O_4)^+$, 201.1121; found 201.1113.

Synthesis of benzyl (1*S*,6*R*)-1,6-dihydroxycyclohexa-2,4diene-1-carboxylate (11). Benzyl bromide (0.92 mL, 7.60 mmol, 1.1 equiv.) was dissolved in CH_2Cl_2 (50 mL) containing triethylamine (1.12 mL, 8.29 mmol, 1.2 equiv.) to which was added dropwise a suspension of diol acid 4 (1.08 g, 6.91 mmol, 1.0 equiv.) in CH_2Cl_2 (50 mL). The resulting

solution was stirred for 20 h at room temperature, then diluted with H_2O (50 mL), and extracted with EtOAc (4 × 50 mL). The organic layer was dried over MgSO4 and concentrated under reduced pressure. Purification by flash column chromatography (10 \rightarrow 80% EtOAc-petrol) gave 11 as a colourless oil. (1.30 g, 76%). $R_{\rm f} = 0.29$ (40% EtOAc-petrol); $[\alpha]_{\rm D}^{25} - 135.6$ (c 1.6, CH_2Cl_2 ; ¹H NMR (300 MHz, $CDCl_3$) δ = 7.36 (5H, br s, Bn), 6.13 (1H, dddd, J = 9.5, 4.0, 1.0, 0.5 Hz, C=CH), 5.94 (1H, dddd, J = 9.5, 5.0, 2.5, 0.5 Hz, C=CH), 5.82 (1H, ddd, J = 9.5, 2.0, 1.0 Hz, C=CH), 5.76 (1H, ddd, J = 9.5, 2.0, 1.0 Hz, C=CH), 5.29 (2H, s, CO_2 -CH₂-Ar), 4.87 (1H, m, C(O)H); ¹³C NMR (75.5 MHz, $CDCl_3$) δ = 175.2 (C=O), 135.1, 132.1, 128.8, 128.7, 128.2, 127.0, 124.8, 122.9, 74.1 (C(OH)C=O), 71.1 (C(OH)-H), 68.5 (O-CH₂); ν_{max} (film) 3451, 3038, 1731, 1660, 1455, 1378, 1234, 1168, 1077, 1020, 909, 753, 695 cm⁻¹; HRMS (+ve ESI-TOF) m/z calculated for $(C_{14}H_{14}NaO_4)^+$, 269.0784; found 269.0863.

Synthesis of benzyl (3aS,7aR)-2,2-dimethylbenzo[d][1,3]dioxole-3a(7aH)-carboxylate (12). To a stirred solution of 11 (335 mg, 1.37 mmol) dissolved in acetone (10 mL, freshly distilled), was added 2,2-dimethoxypropane (2.0 mL, 16.46 mmol, 12 equiv.) and para-toluenesulfonic acid (26 mg, 0.03 mmol, 10 mol%). The solution was stirred at room temperature for 20 h, then diluted with EtOAc (10 mL) followed by the addition of water (20 mL). The biphasic system was extracted with EtOAc (4 \times 10 mL) and the organic layers combined and washed with saturated brine, dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography ($10 \rightarrow 20\%$ EtOAc-petrol) to give 12 a colourless oil (358 mg, 92%). R_f = 0.60 (30% EtOAcpetrol); ¹H NMR (250 MHz, CDCl₃) δ = 7.31 (5H, br s, Bn), 6.08-5.94 (3H, m, C=CH), 5.86-5.79 (1H, m, C=CH), 5.18 (2H, s, CH₂), 4.97 (1H, d, J = 4.0 Hz, CH(O)), 1.38 (3H, s, CH₃), 1.37 (3H, s, CH_3). Data in agreement with those previously reported.6e

Synthesis of (3aS,7aR)-2,2-dimethylhexahydrobenzo[d][1,3]dioxole-3a-carboxylic acid (9) from ester 12. A stirred solution of benzyl ester 12 (32 mg, 0.11 mmol) and Pd/C (10 mg, matrix activated carbon support) in MeOH (20 mL) was exposed to a hydrogen atmosphere (balloon) at room temperature. After 24 h the solution was filtered through a plug of celite and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (0 \rightarrow 50% EtOAc-petrol) to yield pure 9 (14 mg, 63%) as a colourless oil; data as above.

Synthesis of benzyl (3aS,7aR)-2,2-di-*tert*-butylbenzo[*d*][1,3,2]dioxasilole-3a(7a*H*)-carboxylate (13). To a stirred solution of 11 (48 mg, 0.195 mmol) dissolved in CH₂Cl₂ (2 mL) was added triethylamine (65 μ L, 0.468 mmol, 2.4 equiv.) and di-*tert*-butylsilanediyl bis(trifluoromethanesulfonate) (70 μ L, 0.215 mmol, 1.1 equiv.). The solution was stirred at room temperature for 9 h. The resulting solution was diluted with EtOAc (10 mL) followed by the addition of water (20 mL). The biphasic system was extracted with EtOAc (4 × 10 mL) and the organic layers combined and washed with saturated brine, dried over MgSO₄ and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (0 \rightarrow 5% EtOAc-petrol) to give **13** as a colourless oil (61 mg, 81%). $R_{\rm f}$ = 0.65 (10% EtOAc-petrol); $[\alpha]_{\rm D}^{25}$ -327.0 (*c* 3.31, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ = 7.35-7.31 (5H, m, Ar-H), 6.03-5.94 (3H, m, C=CH), 5.75-5.71 (1H, m, C=CH), 5.25 (1H, d, *J* = 12.0 Hz, -CHH-), 5.18 (1H, d, *J* = 12.0 Hz, -CHH-), 4.97 (1H, d, *J* = 1.8 Hz, CH(OSi)), 0.99 (9H, s, CH₃) 0.98 (9H, s, CH₃); ¹³C NMR (75 MHz, CDCl₃) δ = 172.2 (C=O), 135.6, 128.6, 128.4, 128.3, 126.3, 125.8, 123.9, 122.3, 78.6 (*C*(OSi)(C=O)), 71.5 (*C*H(OSi)), 67.4 (CH₂), 27.0 (CH₃), 26.9 (CH₃), 21.2 (*C*(CH₃)₃), 20.2 (*C*(CH₃)₃); $\nu_{\rm max}$ (film) 3045, 2966, 2934, 2891, 2859, 1733, 1473, 1229, 1091, 1031, 1012, 1000, 875, 825, 696 cm⁻¹; HRMS (ESI+) *m*/z calcd for (C₂₂H₃₁O₄Si)⁺, 387.1986; found 387.1993.

Synthesis of (3aS,7aR)-2,2-di-tert-butyltetrahydrobenzo[d]-[1,3,2]dioxasilole-3a(4H)-carboxylic acid (14). A stirred solution of 13 (61 mg, 0.16 mmol) and Pd/C (10 mg, matrix activated carbon support) in MeOH (20 mL) was exposed to a hydrogen atmosphere (balloon) at room temperature. After 24 h the solution was filtered through a plug of celite and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (10 -> 70% EtOAcpetrol) to give pure 14 (25 mg, 53%) as a colourless oil. $R_{\rm f}$ = 0.30 (30% EtOAc-petrol); $[\alpha]_{D}^{25}$ -17.0 (*c* 0.18, CHCl₃); ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta = 4.36 (1\text{H}, \text{dd}, J = 11.0, 4.5 \text{ Hz}, \text{CH}(\text{OSi})),$ 1.99–1.20 (8H, m, CH), 1.02 (9H, s, CH₃), 0.98 (9H, s, CH₃); ¹³C NMR (75 MHz, $CDCl_3$) δ = 179.1 (C=O), 77.9 (C(OSi)(C=O)), 73.9 (CH(OSi)), 33.0, 30.1, 27.7 (CH₃), 27.5 (CH₃), 23.7, 20.5, 20.3, 19.6; v_{max} (film) 3070, 2935, 2894, 2859, 1717, 1448, 1472, 1094, 827 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{15}H_{29}O_4Si)^+$, 301.1830; found 301.1830.

Synthesis of methyl (1S,2R)-1,2-dimethoxycyclohexane-1-carboxylate (15) and methyl (1S,2R)-2-methoxy-1-((trimethylsilyl)oxy)cyclohexane-1-carboxylate (17). Diol acid 5 (133 mg, 0.83 mmol) in DMF (1.00 mL), was added dropwise to a suspension of NaH (109 mg, 2.74 mmol, 3.3 equiv., 60% in mineral oil) in DMF (2.00 mL) at -22 °C. Iodomethane (0.170 mL, 2.83 mmol, 3.4 equiv.) was added dropwise over 5 min. The resulting solution was stirred at -22 °C and allowed to warm to room temperature over 19 h. The solution was cooled to -22 °C, NH₄Cl_(aq) (10 mL) was added to the solution dropwise, then the reaction mixture was transferred into a separating funnel. EtOAc (20 mL) was added to the solution and the organic layers were washed with $LiCl_{(aq)}$ (3 × 10 mL), dried over MgSO₄ and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (5 \rightarrow 20% EtOAc-petrol) to yield an inseparable mixture of 15 and 16 as a colourless oil (99 mg). The inseparable mixture was dissolved in CH₂Cl₂ (3.00 mL), to which was added triethylamine (40 µL, 0.32 mmol) followed by trimethylsilyl trifluoromethanesulfonate (50 µL, 0.29 mmol). The reaction mixture was stirred at room temperature for 30 h, then diluted with EtOAc (20 mL), washed with saturated brine (20 mL), dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (1 \rightarrow 20% EtOAc-petrol) to give 15 (48 mg, 25%) and 17 (20 mg, 9%) as colourless oils. Data for 15: $R_{\rm f}$ = 0.20 (20% EtOAc-petrol); $[\alpha]_{D}^{25}$ -26.9 (c 2.23, CHCl₃); ¹H NMR

(250 MHz, CDCl₃) δ = 3.73 (3H, s, CO₂CH₃), 3.44 (1H, dd, *J* = 11.0, 4.0 Hz, C*H*(OCH₃)), 3.36 (3H, s, OCH₃), 3.26 (3H, s, OCH₃), 2.04–0.80 (8H, m, CH); ¹³C NMR (75 MHz, CDCl₃) δ = 174.1 (C=O), 82.9, 81.7, 57.0, 52.1, 51.8, 29.7, 24.8, 23.5, 20.3; ν_{max} (film) 2938, 2861, 1730, 1446, 1373, 1275, 1227, 1099, 1061, 997 cm⁻¹; HRMS (ESI+) *m/z* calcd for (C₁₀H₁₉O₄)⁺, 203.1278; found 203.1289. Data for 17: $R_{\rm f}$ = 0.85 (20% EtOAcpetrol); [α]_D²⁵ –1.4 (*c* 0.73, CHCl₃); ¹H NMR (250 MHz, CDCl₃) δ = 3.73 (3H, s, CO₂CH₃), 3.47 (1H, dd, *J* = 11.0, 4.5 Hz, CH(OCH₃)), 3.29 (3H, s, OCH₃), 1.85–0.84 (8H, m, CH), 0.15 (9H, s, Si(CH₃)₃); ¹³C NMR (75 MHz, CDCl₃) δ = 175.6 (C=O), 82.2, 76.1, 56.8, 51.8, 36.2, 25.1, 24.0, 20.3, 2.2; $\nu_{\rm max}$ (film) 2949, 2860, 1753, 1446, 1366, 1279, 1244, 1156, 1062, 1041, 837, 760, 734 cm⁻¹; HRMS (ESI+) *m/z* calcd for (C₁₂H₂₅O₄Si)⁺, 261.1517; found 261.1515.

Synthesis of (1S,2R)-1,2-dimethoxycyclohexane-1-carboxylic acid (18). Ester 15 (155 mg, 0.76 mmol) was dissolved in THF (1.00 mL) and NaOH (4.00 mL, 2.0 M in H₂O, 0.020 mol, 60 equiv.) and refluxed for 24 h. The resulting solution was extracted with EtOAc (3 × 20 mL) to remove any unreacted starting material. The remaining aqueous layer was acidified to pH 2.0 and extracted with EtOAc (3 × 20 mL). Combined organic layers were washed with saturated brine and dried over MgSO₄, then concentrated under reduced pressure. The resulting oil was purified via flash column chromatography $(50:45:2.5:2.5 \text{ EtOAc-petrol-AcOH-H}_2O)$ to give 18 as a colourless oil (108 mg, 75%). Rf = 0.25 (50:45:2.5:2.5 EtOAcpetrol-AcOH-H₂O); $[\alpha]_{D}^{25}$ -20.5 (*c* 0.73, CHCl₃); ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) \delta = 3.40 - 3.37 (1H, m, CH(\text{OCH}_3)), 3.38 (3H, M)$ s, OCH₃), 3.35 (3H, s, OCH₃), 2.18 (1H, dd, J = 15.0, 2.0 Hz, CH), 1.96 (1H, m, CH), 1.82-1.81 (1H, m, CH), 1.68-1.53 (3H, m, CH), 1.37–1.25 (2H, m, CH); 13 C NMR (125 MHz, CDCl₃) δ = 175.3 (C=O), 82.7, 81.2, 57.1, 52.0, 28.1, 24.8, 23.7, 20.0; v_{max} (film) 3143, 2940, 2860, 2838, 1721, 1464, 1448, 1308, 1196, 1099, 1077, 973, 939 cm⁻¹; HRMS (ESI-) m/z calcd for $(C_9H_{15}O_4)^-$, 187.0976; found 187.0972.

Synthesis of methyl (2R,3aS,7aR)-2-phenyltetrahydrobenzo-[d][1,3]dioxole-3a(4H)-carboxylate (19) and methyl (2S,3aS,7aR)-2-phenyltetrahydrobenzo[d][1,3]dioxole-3a(4H)-carboxylate (20). Diol 6 (30 mg, 0.27 mmol) was dissolved in toluene (15 mL). Benzaldehyde (0.030 mL, 0.34 mmol, 2.0 equiv.) and para-toluenesulfonic acid (4.0 mg, 0.02 mmol, 10 mol%) were added. The resulting solution was refluxed for 24 h. After cooling, the reaction mixture was diluted with NaHCO3 (15 mL) and extracted with EtOAc (3 \times 10 mL). Combined organic layers were dried over MgSO4, then concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (5 \rightarrow 10% EtOAc-petrol) to yield 19 (15 mg, 35%) and 20 (10 mg, 24%). In addition a further 17 mg (40%) of material was isolated, shown to be a 1:1 mixture of **19** and **20** by NMR. Data for **19**: $R_f = 0.49 (10\%)$ EtOAc-petrol); $[\alpha]_{D}^{25}$ -31 (*c* 0.7, CHCl₃); ¹H NMR (300 MHz, $CDCl_3$) $\delta = 7.56-7.52$ (2H, m, Ar-H), 7.40-7.37 (3H, m, Ar-H), 5.97 (1H, s, CH(O)(O)), 4.42 (1H, t, J = 4.0 Hz, CH(O)), 3.82 (3H, s, OCH₃), 2.13–1.81 (4H, m, CH), 1.76–1.45 (4H, m, CH); ¹³C NMR (75 MHz, CDCl₃) δ = 173.4 (C=O), 137.2, 129.3, 128.3,

126.7 (Ar–C), 103.3 (CH(O)(O)(Ar)), 81.5 (C(O)(C=O)), 77.3 (C(O)H), 52.5 (OCH₃), 31.1, 26.1, 19.4, 18.5; ν_{max} (film) 2951, 2869, 1734, 1451, 1247, 1163, 1088, 1024, 697 cm⁻¹; HRMS (ESI+) m/z calcd for ($C_{15}H_{19}O_4$)⁺, 263.1278; found 263.1267. Data for **20**: $R_f = 0.53$ (10% EtOAc–petrol); $[\alpha]_D^{25}$ –1.6 (c 0.08, CHCl₃); ¹H NMR (300 MHz, CDCl₃) $\delta = 7.50$ –7.46 (2H, m, Ar–H), 7.39–7.32 (3H, m, Ar–H), 6.21 (1H, s, CH(O)(O)), 4.53 (1H, t, J = 5.5 Hz, CH(O)), 3.70 (3H, s, OCH₃), 2.03–1.89 (4H, m, CH), 1.75–1.39 (4H, m, CH); ¹³C NMR (75 MHz, CDCl₃) $\delta = 173.0$ (C=O), 138.9, 129.0, 128.3, 126.4 (Ar–C), 102.8 (CH(O)-(O)(Ar)), 81.9 (C(O)(C=O)), 75.9 (C(O)H), 52.3 (OCH₃), 30.7, 26.2, 20.9, 20.4; ν_{max} (film) 2938, 2863, 1735, 1451, 1247, 1162, 1093, 698 cm⁻¹; HRMS (ESI+) m/z calcd for ($C_{15}H_{19}O_4$)⁺, 263.1278; found 263.1268.

Synthesis of (S)-2,2-dimethyl-1,3-dioxaspiro[4.5]decane-4,6dione (21). To a stirred solution of 10 (25 mg, 0.12 mmol) in CH₂Cl₂ (10 mL) was added Dess-Martin periodinane (131 mg, 0.31 mmol, 2.5 equiv.). The resulting solution was heated to reflux for 24 h. To the cooled solution was added NaHCO_{3(aq)} (20 mL). The biphasic system was extracted with EtOAc (4 \times 10 mL) and the organic layers combined and washed with saturated brine, dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (0 \rightarrow 20% EtOAc-petrol) to yield 21 as a colourless oil (15 mg, 60%). $R_{\rm f}$ = 0.25 (10% EtOAc-petrol); $[\alpha]_{D}^{25}$ -50.0 (c 2.2, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ = 2.81 (1H, ddd, J = 13.9, 10.9, 5.5 Hz), 2.63 (1H, dt, J = 13.9, 5.5 Hz), 2.32-2.17 (2H, m), 2.08-1.97 (2H, m), 1.86-1.73 (2H, m), 1.62 (3H, s, CH₃), 1.57 (3H, s, CH₃); 13 C NMR (125 MHz, CDCl₃) δ = 203.1 (C=O), 169.3 (C=O), 111.4 (C(CH₃)₂), 85.2, 39.5, 38.7, 28.4, 27.5, 26.6, 21.0; v_{max} (film) 2996, 2941, 2921, 2872, 2851, 1784, 1732, 1394, 1380, 1283, 1254, 1130, 1077, 1045, 930 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{10}H_{14}NaO_4)^+$, 221.0784; found 221.0787.

Synthesis of methyl (1S,2R)-2-((tert-butyldimethylsilyl)oxy)-1-hydroxycyclohexane-1-carboxylate (22). Diol ester 6 (370 mg, 2.12 mmol, 1.0 equiv.) was dissolved in CH₂Cl₂ (20 mL) and cooled to -78 °C. Triethylamine (355 µL, 257 mg, 2.54 mmol, 1.2 equiv.) was added dropwise, then tert-butyldimethylsilyl trifluoromethanesulfonate (487 µL, 560 mg, 2.12 mmol, 1.0 equiv.) was added dropwise. The reaction mixture was stirred at -78 °C for 1 h, then quenched by addition of H₂O (20 mL). Phases were separated and the organic phase was washed further with NaCl_(aq) (satd, 20 mL). The organic phase was dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (10% EtOAc-petrol) to give 22 (465 mg, 76%) as a colourless oil. $R_{\rm f} = 0.60$ (20% EtOAc-petrol); $\left[\alpha\right]_{\rm D}^{25}$ -7.0 $(c 1.3, CHCl_3)$; ¹H NMR (250 MHz, CDCl₃) δ = 3.94 (1H, dd, J = 10.5, 5.0 Hz), 3.72 (3H, s), 3.05 (1H, d, J = 1.5 Hz), 1.91-1.14 (8H, m), 0.83 (9H, s) 0.04 (3H, s), -0.02 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ = 176.1 (C=O), 77.5 (C(OH)), 73.8 (CH-(OSi)), 52.2, 33.1, 30.1 (CH), 25.6 (SiC(CH₃)₃), 23.8, 19.6, 17.8 (SiC(CH₃)₃), -4.1, -5.2; v_{max} (film) 3555, 3027, 2928, 1705, 1494, 1453, 1281, 908, 732, 699 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{14}H_{29}O_4Si)^+$, 289.1830; found 289.1831.

Synthesis of (1R,2R)-2-((*tert*-butyldimethylsilyl)oxy)-1-(hydroxymethyl)cyclohexan-1-ol (23) and (1R,2R)-1-(((*tert*-butyldimethylsilyl)oxy)methyl)cyclohexane-1,2-diol (24). To a stirred solution of 22 (375 mg, 1.3 mmol) in THF at -78 °C was added dropwise LiAlH₄ (0.54 mL, 2.4 M in THF, 1.3 mmol, 1.0 equiv.) over 20 min. The resulting solution was left to warm to room temperature for 12 h. The reaction was quenched with addition of H₂O (0.05 mL) followed by NaOH (10% aq. sol., 0.09 mL) followed by H₂O (0.14 mL). The solution was filtered through MgSO₄ and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (5–20% EtOAc–petrol) to give 23 (45 mg, 13%) as a colourless oil and 24 (19 mg, 5%) as a colourless oil.

Alternative procedure. To a stirred solution of 22 (465 mg, 1.6 mmol) in THF at -78 °C was added dropwise LiBH₄ (0.80 mL, 4.0 M in THF, 3.22 mmol, 2.0 equiv.) over 20 min. The resulting solution was left to warm to room temperature for 12 h. The solution was quenched with addition of Et₂O (10 mL) followed by H₂O (10 mL). The mixture was separated, and the aqueous layer extracted with EtOAc (4×10 mL); the organic layers were combined, dried with MgSO4 and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (5 \rightarrow 20% EtOAc-petrol) to yield 23 (400 mg, 91%) as a colourless oil. Data for 23: $R_f =$ 0.45 (20% EtOAc-petrol); $[\alpha]_{D}^{25}$ -18 (c 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ = 3.65 (1H, br s), 3.56 (1H, d, *J* = 10.0 Hz), 3.37 (1H, d, J = 10.0 Hz), 2.51 (2H, br s, OH), 1.82 (1H, d, J = 13.0 Hz), 1.63 (3H, m), 1.52 (1H, m), 1.40 (1H, m), 1.29 (1H, m), 1.19-1.14 (1H, m), 0.89 (9H, s, C(CH₃)₃), 0.08 (6H, s, SiCH₃); ¹³C NMR (125 MHz, CDCl₃) 73.2, 72.9, 68.3 (OCH₂), 31.7, 30.7, 25.8 (SiC(CH₃)₃), 23.2, 20.7, 18.0 (SiC(CH₃)₃), -4.0 (SiCH₃), -5.0 (SiCH₃); v_{max} (film) 3443, 2930, 2857, 1463, 1389, 1261, 1252, 1079, 835, 777 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{13}H_{29}O_3Si)^+$, 261.1881; found 261.1886. Data for 24: $R_f = 0.55$ (20% EtOAc-petrol); $[\alpha]_{D}^{25}$ +10 (*c* 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃) 3.66 (1H, d, J = 10.0 Hz), 3.64 (1H, dd, J = 10.0, 4.5 Hz), 3.52 (1H, d, J = 10.0 Hz), 1.73–1.63 (3H, m), 1.60-1.50 (2H, m), 1.45-1.39 (1H, m), 1.22-1.08 (2H, m), 0.90 (9H, s), 0.09 (3H, s), 0.08 (3H, s); ¹³C NMR (125 MHz, CDCl₃) $\delta = 73.8, 71.9, 71.8, 31.8, 29.5, 25.8$ (SiC(CH₃)₃), 23.5, 20.3, 18.1 $(SiC(CH_3)_3)$, -5.5 $(SiCH_3)$, -5.6 $(SiCH_3)$; ν_{max} (film) 3555, 3027, 2929, 1494, 1453, 1218, 908, 732, 699 cm⁻¹; HRMS (ESI+) m/zcalcd for $(C_{13}H_{29}O_3Si)^+$, 261.1881; found 261.1858.

Synthesis of *tert*-butyl(((5R,6R)-2,2-dimethyl-1,3-dioxaspiro-[4.5]decan-6-yl)oxy)dimethylsilane (25). To a stirred solution of 23 (1.81 g, 6.59 mmol) dissolved in acetone (10 mL, freshly distilled), was added 2,2-dimethoxypropane (12 mL, 100 mmol) and *para*-toluenesulfonic acid (13 mg, 0.06 mmol, 10 mol%). The solution was stirred at room temperature for 20 h, then diluted with EtOAc (10 mL) followed by the addition of water (20 mL). The biphasic system was extracted with EtOAc (4 × 10 mL) and the organic layers were combined and washed with saturated brine, dried over MgSO₄ and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (10 \rightarrow 20% EtOAc-petrol) to give 25 as a colourless oil (1.72 g, 86%). $R_{\rm f} = 0.80$ (10% EtOAc-

View Article Online

petrol); $[\alpha]_{\rm D}^{25}$ –20 (*c* 0.3, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ = 3.88 (1H, d, *J* = 8.0 Hz), 3.70 (1H, d, *J* = 8.0 Hz), 3.53 (1H, dd, *J* = 7.0, 3.0 Hz), 1.95–1.86 (1H, m), 1.75–1.56 (3H, m), 1.51–1.18 (4H, m) 1.39 (6H, s, O-C(CH₃)₂), 0.90 (9H, s SiC(CH₃)₃), 0.07 (3H, s, SiCH₃), 0.05 (3H, s, SiCH₃); ¹³C NMR (75 MHz, CDCl₃) δ = 109.2, 83.4, 72.9, 70.9, 34.0, 32.2, 27.7, 26.9, 25.9, 22.6, 21.6, 18.2, -4.5, -4.6; $\nu_{\rm max}$ (film) 2988, 2933, 2894, 2856, 1472, 1462, 1377, 1368, 1251, 1212, 1141, 1094, 1056, 988, 898, 773 cm⁻¹; HRMS (ESI+) *m*/*z* calcd for (C₁₆H₃₂NaO₃Si)⁺, 323.2013; found 323.2015.

Synthesis of (5R,6R)-2,2-dimethyl-1,3-dioxaspiro[4.5]decan-6ol (26). To a stirred solution of 25 (1.72 g, 5.72 mmol) in THF (50 mL) at -78 °C, was added dropwise tetrabutylammonium fluoride (1.0 M solution in THF, 5.72 mL, 2.0 equiv.) over 5 min. The resulting solution was allowed to warm to room temperature over 16 h, then was quenched with Et₂O (10 mL) and NH₄Cl_(aq) (10 mL). The reaction mixture was extracted with EtOAc $(4 \times 10 \text{ mL})$ and the organic layers were combined and washed with saturated brine, dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (5% EtOAc-petrol) to yield 26 as a colourless oil (591 mg, 55%). $R_{\rm f} = 0.30$ (20%) EtOAc-petrol); $[\alpha]_{D}^{25}$ -4.0 (c 0.9, CHCl₃); ¹H NMR (250 MHz, $CDCl_3$) $\delta = 3.99$ (1H, d, J = 8.5 Hz), 3.71 (1H, d, J = 8.5 Hz), 3.49 (1H, m), 2.11 (1H, br s, OH), 1.99-1.23 (8H, m), 1.40 (3H, s), 1.39 (3H, s); ¹³C NMR (100 MHz, CDCl₃) δ = 109.4, 83.2, 71.5, 71.3, 33.7, 31.2, 29.7, 27.2, 27.1, 22.6; v_{max} (film) 3204, 2930, 2855, 1563, 1406, 1371, 1184, 903, 729 cm⁻¹; HRMS (ESI+) m/zcalcd for $(C_{10}H_{18}NaO_3)^+$, 209.1148; found 209.1104.

Synthesis of (R)-2,2-dimethyl-1,3-dioxaspiro[4.5]decan-6-one (27). To alcohol 26 (591 mg, 3.17 mmol) dissolved in CH₂Cl₂ (50 mL) was added Dess-Martin periodinane (2.69 g, 6.3 mmol, 2.0 equiv.) and left to stir at room temperature for 72 h. The reaction was quenched with the addition of water (20 mL) and EtOAc (50 mL). The aqueous layer was extracted with EtOAc (5 \times 10 mL) and the organic layers were combined and washed with saturated brine, dried over MgSO4 and concentrated under reduced pressure. The resulting oil was purified via flash column chromatography (10% EtOAc-petrol) to yield 27 as a colourless oil (517 mg, 86%). R_f = 0.50 (20% EtOAc-petrol); $[\alpha]_{D}^{25}$ -6.0 (c 1.6, CHCl₃); ¹H NMR (400 MHz, $CDCl_3$) $\delta = 4.42$ (1H, d, J = 8.5 Hz), 3.65 (1H, d, J = 8.5 Hz), 2.78 (1H, ddd, J = 15.0, 9.0, 5.0 Hz), 2.34-2.27 (1H, m), 2.04-1.86 (3H, m), 1.80–1.59 (3H, m), 1.40 (3H, s), 1.33 (3H, s); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta = 209.1, 110.5, 85.5, 69.1, 39.7, 38.5, 27.4,$ 26.9, 26.1, 22.3; v_{max} (film) 2986, 2937, 2865, 1723, 1452, 1431, 1380, 1371, 1050, 858, 811 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{10}H_{16}NaO_3)^+$, 207.0992; found 207.0995.

Synthesis of ((((3aR,4S,7R,7aS)-3a,4,7,7a-tetrahydro-4,7-bis-(methoxymethoxy)-2,2-dimethylbenzo[d][1,3]dioxol-7a-yl)methoxy)-(*tert*-butyl)dimethylsilane (31). To a solution of known⁴ⁱ diol 30 (304 mg, 0.92 mmol, 1.0 equiv.) in CH₂Cl₂ (10 mL) was added *N*-ethyldiisopropylamine (1.61 mL, 9.20 mmol, 10.0 equiv.), followed dropwise by methoxymethyl chloride (0.42 mL, 5.52 mmol, 6.0 equiv.). The resulting solution was left to stir at room temperature for 16 h. The reaction mixture was extracted with EtOAc $(3 \times 30 \text{ mL})$ and the combined organic layers were washed with water (30 mL) and brine (30 mL). The resulting solution was dried over MgSO₄. Removal of the solvent under reduced pressure and purification by flash column chromatography (5 \rightarrow 30% EtOAcpetrol) gave 31 as a colourless oil (254 mg, 66% yield). $R_{\rm f}$ = 0.64 (20% EtOAc-petrol); $[\alpha]_{D}^{25}$ -22.2 (c 0.09, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ = 5.74 (2H, s, C=CH), 4.86 (1H, d, J = 6.5 Hz, OCH₂O), 4.85 (1H, d, J = 6.5 Hz, OCH₂O), 4.72 (1H, d, J = 6.5 Hz, OCH₂O), 4.70 (1H, d, J = 6.5 Hz, OCH₂O), 4.40 (1H, d, J = 3.0 Hz, CHO), 4.29 (1H, br s, CHOMOM), 4.22 (1H, br s, CHOMOM), 3.96 (1H, d, *J* = 11.0 Hz, CHHOTBDMS), 3.49 (1H, d, J = 11.0 Hz, CHHOTBDMS), 3.41 (3H, s, CH₂OCH₃), 3.39 (3H, s, CH₂OCH₃), 1.47 (3H, s, O-C-CH₃), 1.43 (3H, s, O-C-CH₃), 0.89 (9H, s, SiC(CH₃)₃), 0.04 (6H, s, Si(CH₃)₂); ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3) \delta = 130.3 (\text{HC}=\text{CH}), 127.9 (\text{HC}=\text{CH}), 108.9$ (C(CH₃)₂), 96.3 (OCH₂O), 95.5 (OCH₂O), 85.3 (4°CCH₂O), 78.4 (CHO), 76.8 (CHO), 75.0 (CHO), 61.9 (CH₂OTBDMS), 55.5 (OCH₃), 55.4 (OCH₃), 28.4 (CCH₃), 27.0 (CCH₃), 25.9 (SiC- $(CH_3)_3$, 18.3 $(SiC(CH_3)_3)$, -5.3 $(SiCH_3)$, -5.7 $(SiCH_3)$; ν_{max} (film) 2954, 1252, 1043, 907, 837, 732, 649 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{20}H_{38}NaO_7Si)^+$, 441.2279; found 441.2285.

Synthesis of (3aS,4R,5S,6R,7S,7aR)-3a-((tert-butyldimethylsilyl)oxy)methyl-4,7-bis(methoxymethoxy)-2,2-dimethylhexahydrobenzo[d][1,3]dioxole-5,6-diol (32). To a solution of 31 (75.0 mg, 0.185 mmol, 1.0 equiv.) in acetone-H₂O (4:1, 8.0 mL) was added NMO (43.4 mg, 0.370 mmol, 2.0 equiv.) followed dropwise by OsO4 (40 µL, 2.5% w/v in tert-BuOH, 3.7 µmol). The resulting solution was stirred at room temperature for 72 h. The reaction mixture was diluted with EtOAc (30 mL) and extracted with saturated $Na_2S_2O_{3(aq)}$ (2 × 30 mL). The organic phase was washed with brine (30 mL) and dried over MgSO₄. Removal of the solvent under reduced pressure and purification by flash column chromatography (50% EtOAc-petrol) gave 32 as a colourless oil (56 mg, 67%). $R_{\rm f}$ = 0.43 (50% EtOAc-petrol); $[\alpha]_{D}^{25}$ -10.0 (c 0.1, CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ = 4.82 (2H, s, OCH₂O), 4.74 (1H, d, J = 6.5 Hz, OCHHO), 4.67 (1H, d, J = 6.5 Hz, OCHHO), 4.47 (1H, d, J = 2.5 Hz, CHO), 4.18 (1H, m, CHO), 4.06 (1H, br s, CHO), 3.93 (1H, d, J = 11.0 Hz, CHHOTBDMS), 3.94-3.89 (1H, m, CHO) 3.79 (1H, dd, J = 10.0, 4.0 Hz, CHO) 3.64 (1H, d, J = 11.0 Hz, CHHOTBDMS), 3.45 (3H, s, CH₂OCH₃), 3.40 (3H, s, CH₂OCH₃), 3.05 (1H, br s, OH), 1.51 (3H, s, C(CH₃)₂), 1.37 (3H, s, C(CH₃)₂), 0.89 (9H, s, SiC(CH₃)₃), 0.07 (3H, s, SiCH₃), 0.06 (3H, s, SiCH₃); ¹³C NMR (75 MHz, CDCl₃) δ = 108.6 (C(CH₃)₂), 98.1 (OCH₂OCH₃), 95.9 (OCH₂OCH₃), 85.3 (4°CCH₂O), 83.0 (CHO), 76.4 (CHO), 73.4 (CHO), 70.9 (CHO), 69.5 (CHO), 61.2 (CH₂OTBDMS), 55.9 (CH₂OCH₃), 55.8 (CH₂OCH₃), 28.2 (C(CH₃)₂), 26.3 (C(CH₃)₂), 25.9 (SiC(CH₃)₃), 18.3 (SiC(CH₃)₃), -5.4 (SiCH₃), -5.7 (SiCH₃); v_{max} (film) 3455, 2930, 2857, 1463, 1370, 1252, 1217, 1152, 1101, 1033, 999, 948, 919, 859, 837, 814, 777, 675 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{20}H_{40}NaO_9Si)^+$, 475.2334; found 475.2344.

Synthesis of (1R,2S,3R,4S,5R,6S)-6-(acetoxymethyl)-6-hydroxycyclohexane-1,2,3,4,5-pentayl pentaacetate (33). To a solution of 32 (53.0 mg, 0.17 mmol) in Et₂O (5 mL) was added HCl_(aq) (1.0 M, 5 mL). The solution was stirred vigorously at room temperature for 24 h, then diluted with Et₂O (10 mL) and extracted with H_2O (3 × 10 mL). The combined aqueous layers were concentrated under reduced pressure to give heptaol 34. ¹H NMR showed this to be impure, so crude 34 was dissolved in pyridine (0.7 mL), to which was added acetic anhydride (1.0 mL). The reaction mixture was stirred at room temperature for a further 24 h, then diluted with EtOAc (10 mL). HCl (1.0 M, 10 mL) was added dropwise and the reaction mixture transferred to a separating funnel. The organic phase was washed with NaHCO_{3(aq)} (satd, 3×10 mL) and H₂O $(3 \times 10 \text{ mL})$, then dried over Na₂SO₄ and concentrated under reduced pressure. Purification via flash column chromatography (50% EtOAc-petrol) gave 33 as a colourless oil (22 mg, 37%). $R_{\rm f} = 0.29$ (50% EtOAc-petrol); $[\alpha]_{\rm D}^{25} - 5.5$ (c 0.36, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃) δ = 5.66 (1H, t, J = 10.0 Hz), 5.34 (1H, t, J = 3.5 Hz), 5.29 (1H, d, J = 9.5 Hz), 5.25 (1H, dd, J = 10.0, 3.5 Hz), 5.24 (1H, d, J = 3.5 Hz), 4.21 (1H, d, J = 12.0 Hz, -CHHOAc), 3.95 (1H, d, J = 12.0 Hz, -CHHOAc), 2.15 (3H, s, -OAc), 2.14 (3H, s, -OAc), 2.11 (3H, s, -OAc), 2.05 (3H, s, -OAc), 2.02 (3H, s, -OAc), 1.98 (3H, s, -OAc); ¹³C NMR (75 MHz, CDCl₃) δ = 170.7 (C=O), 169.7 (2 × C=O), 169.6 (C=O), 169.0 (C=O), 168.3 (C=O), 75.0, 71.4, 68.9, 68.8, 68.1, 68.0, 65.0, 20.8 (CH₃), 20.7 (CH₃), 20.7 (CH₃), 20.6 (CH₃), 20.5 (CH₃), 20.5 (CH₃); v_{max} (film) 3479, 2965, 1745, 1431, 1369, 1218, 1039, 899, 821, 731 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{19}H_{26}NaO_{13})^+$, 485.1266; found 485.1318.

Synthesis of (1*S*,2*R*,3*S*,4*R*,5*S*,6*R*)-1-(hydroxymethyl)cyclohexane-1,2,3,4,5,6-hexaol (34). Hexaacetate 33 (22 mg, 0.0476 mmol) was dissolved in MeOH (5 mL) at room temperature. NH_{3(g)} was slowly bubbled through the reaction mixture for 3 d, then the reaction mixture was concentrated under reduced pressure. The crude product was then dried under high vacuum (flask heated to 60 °C to drive off acetamide) to give pure 34 (9 mg, 99%) as a colourless gum. [α]₂₅²⁵ +2.78 (*c* 3.3, H₂O); ¹H NMR (500 MHz, D₂O) δ = 4.07 (2H, d, *J* = 1.3 Hz), 3.88 (1H, t, *J* = 9.8 Hz), 3.81–3.78 (1H, m), 3.76 (2H, d, *J* = 0.8 Hz), 3.56 (1H, d, *J* = 9.5 Hz); ¹³C NMR (75 MHz, D₂O) δ = 77.8, 73.6, 72.5, 71.5, 71.2, 69.3, 63.8; ν_{max} (film) 3331, 2956, 2922, 2854, 1667, 1540, 1455, 1205, 1151, 1020, 742 cm⁻¹; HRMS (ESI+) *m/z* calcd for (C₇H₁₄NaO₇)⁺, 233.0632; found 233.0620.

Synthesis of ((3aR,7aR)-2,2-dimethylbenzo[d][1,3]dioxol-3a(7aH)-yl)methyl acetate (36). To a stirred solution of known⁴ⁱ alcohol 35 (637 mg, 3.50 mmol, 1 equiv.) in CH₂Cl₂ (20 mL) was added triethylamine (0.48 mL, 3.49 mmol, 1 equiv.), DMAP (42 mg, 0.35 mmol, 10 mol%) and Ac₂O (0.33 mL, 3.50 mmol, 1 equiv.). The reaction mixture was stirred for 30 min, then H₂O (20 mL) was added. The reaction mixture was then extracted with EtOAc (4 × 20 mL). The organic layers were combined and dried over MgSO₄ and concentrated under reduced pressure. The resulting oil was purified *via* flash column chromatography (15% EtOAc-petrol) to give **36** (600 mg, 77%) as a yellow oil. $R_{\rm f} = 0.45$ (15% EtOAc-petrol); $[\alpha]_{\rm D}^{25} - 81.6$ (*c* 1.20, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃) $\delta =$ 6.13–6.08 (1H, m, C=CH), 6.03–5.98 (2H, m, C=CH), 5.72 (1H, d, *J* = 10.0 Hz, C=CH), 4.41 (1H, d, *J* = 4.5 Hz, CHOC- $\begin{array}{l} ({\rm CH}_3)_2), \, 4.15 \, (1{\rm H}, \, {\rm d}, J=11.5 \, {\rm Hz}, \, -{\rm C}H{\rm H}-), \, 3.94 \, (1{\rm H}, \, {\rm d}, J=11.5 \, {\rm Hz}, -{\rm C}H{\rm H}-), \, 2.07 \, (3{\rm H}, \, {\rm s}, \, {\rm OAc}), \, 1.44 \, (3{\rm H}, \, {\rm s}, \, {\rm CH}_3), \, 1.37 \, (3{\rm H}, \, {\rm s}, \, {\rm CH}_3); \, ^{13}{\rm C} \, {\rm NMR} \, (75 \, {\rm MHz}, \, {\rm CDCl}_3) \, \delta=170.6 \, ({\rm C=\!-}{\rm O}), \, 128.1, \, 125.4, \, 124.3, \, 123.1 \, ({\rm C=\!-}{\rm C}), \, 106.6 \, (C({\rm CH}_3)_2), \, 78.3 \, (CO({\rm CH}_2))), \, 71.8 \, (C({\rm H}){\rm OC}({\rm CH}_3)_2), \, 66.1 \, ({\rm CH}_2), \, 27.1, \, 26.4 \, ({\rm C}({\rm CH}_3)_2), \, 20.8 \, ({\rm COCH}_3); \, \nu_{\rm max} \, ({\rm film}) \, 2991, \, 2937, \, 1741, \, 1415, \, 1372, \, 1239, \, 1172, \, 1043, \, 906, \, 728, \, 648 \, {\rm cm}^{-1}; \, {\rm HRMS} \, ({\rm ESI+}) \, m/z \, {\rm calcd} \, {\rm for} \, ({\rm C}_{12}{\rm H}_{16}{\rm NaO_4})^+, \, 247.0941; \, {\rm found} \, 247.0931. \, \end{array}$

Synthesis of ((3aR,4R,7S,7aR)-2,2-dimethyl-7,7a-dihydro-4,7epidioxybenzo[d][1,3]dioxol-3a(4H)-yl)methyl acetate (37) and ((3aR,5aS,6aS,6bR)-2,2-dimethyl-6a,6b-dihydrooxireno[2',3':3,4]benzo[1,2-d][1,3]dioxol-3a(5aH)-yl)methyl acetate (38). To a stirred solution of 36 (94 mg, 0.42 mmol) in CH₂Cl₂ (20 mL) was added a solution of 5,10,15,20-tetraphenyl-21H,23H-porphine (10 mg, 0.016 mmol, 4 mol%) in CH₂Cl₂ (50 mL) dropwise over a period of 18 h, while the solution was irradiated with 150 W halogen lamps, and simultaneously sparged with oxygen. After 18 h no more conversion was taking place; the solution was concentrated under reduced pressure and purified via flash column chromatography (15% EtOAc-petrol) to give 37 (52 mg, 48%) as a colourless oil, byproduct 38 as a pale pink oil (10 mg, 9%) and recovered starting material 36 (11 mg, 12%). Data for 37: $R_{\rm f} = 0.35$ (15% EtOAc-petrol); $[\alpha]_{\rm D}^{25}$ -16.9 (c 0.83, CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ = 6.65 (1H, t, J = 8.0 Hz, C=CH), 6.55 (1H, t, J = 8.0 Hz, C=CH), 4.88-4.91 (2H, m, C(H)O), 4.55 (1H, d, J = 12.0 Hz -CHH-), 4.31 (1H, d, I = 12.0 Hz -CHH-), 4.25 (1H, d, I = 4.8 Hz, C(H)OC(CH₂)₃), 2.13 (3H, s, COCH₃), 1.39 (3H, s, CH₃), 1.30 (3H, s, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ = 170.4 (C=O), 131.5 (C=C), 130.3 (C=C), 112.3 (C(CH₃)₂), 79.9 (CO(CH₂), 74.6 (C(H)OC(CH₃)₂), 72.1 (CHO-O), 71.8 (CHO-O), 65.9 (CH₂), 27.8 (C(CH₃)₂), 26.9 (C(CH₃)₂), 20.8 (COCH₃); v_{max} (film) 2995, 2988, 2928, 1736, 1348, 1372, 1450, 1244, 1204, 1144, 1039, 919, 743, 712, 644 cm⁻¹; HRMS (ESI+) m/z calcd for $(C_{12}H_{16}NaO_6)^+$, 279.0839; found 279.0829. Data for 38: $R_f =$ 0.40 (15% EtOAc-petrol); $[\alpha]_{D}^{25}$ -14.3 (c 0.28, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃) δ = 6.09 (1H, dd, J = 10.0, 4.0 Hz, HC=CHCO), 5.70 (1H, dt, J = 10.0 1.5 Hz, HC=CHCO), 4.65 (1H, t, *J* = 1.0 Hz, CHOC(CH₃)₂), 4.33 (1H, *J* = 11.0 Hz, -CHH-), 3.83 (1H, J = 11.0 Hz, -CHH-), 3.64 (1H, dd, J = 3.5, 2.5 Hz, CH(O)), 3.37 (1H, m, CH(O)), 2.09 (3H, s), 1.41 (3H, s), 1.37 (3H, s); ¹³C NMR (75 MHz, CDCl₃) 170.4 (C=O), 132.5 (C=CC- $(O)CH_2$, 124.1 (C=CCH(O)), 110.5 $(C(CH_3)_2)$, 79.0 $(CO(CH_2))$, 71.3 (C(H)OC(CH₃)₂), 66.9 (CH₂), 50.3 (CH(O_{epox})CH(O)), 46.5 (CH(O_{epox})CH=CH), 27.8 (C-CH₃), 26.6 (C-CH₃), 20.9 (COCH₃); *v*_{max} (film) 2989, 2943, 1745, 1455, 1379, 1236, 1181, 1162, 1089, 1060, 1042, 989, 829, 721 cm⁻¹; HRMS (ESI+) m/zcalcd for $(C_{12}H_{16}NaO_5)^+$, 263.0890; found 263.0903.

Synthesis of ((1a*R*,1b*R*,2a*R*,2b*S*,5a*R*,5b*S*)-4,4-dimethyltetrahydrobis(oxireno)[2',3':3,4;2'',3'':5,6]benzo[1,2-d][1,3]dioxol-2b-(1a*H*)-yl)methyl acetate (40). To a stirred solution of 37 (755 mg, 2.94 mmol) in CH_2Cl_2 (50 mL) was added 5,10,15,20tetraphenyl-21*H*,23*H*-porphine cobalt(π) (11 mg, 0.01 mmol, 6 mol%). The solution was stirred for 30 min at room temperature, then concentrated under reduced pressure and purified *via* flash column chromatography (15% EtOAc-petrol) to give **40** (724 mg, 96%) as a colourless oil. $R_{\rm f} = 0.15$ (15% EtOAcpetrol); $[a]_{\rm D}^{25} - 33.8$ (c 0.80, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃) $\delta = 4.35$ (1H, d, J = 11.0 Hz, -CHH-), 4.32 (1H, d, J = 2.0 Hz, C(H)OC(CH₂)₃), 4.02 (1H, d, J = 11.0 Hz, -CHH-), 3.58 (1H, t, J = 3.0 Hz), 3.52 (1H, t, J = 3.0 Hz), 3.39 (1H, dd, J = 2.0, 3.0 Hz), 3.04 (1H, d, J = 3.5 Hz), 2.11 (3H, s, COCH₃), 1.43 (3H, s, CH₃), 1.42 (3H, s, CH₃); ¹³C NMR (75 MHz, CDCl₃) $\delta = 170.4$ (C=O), 110.4 (C(CH₃)₂), 78.2 (C(O)CH₂OAc), 71.9 (C(H)OC-(CH₃)₂), 65.5 (CH₂), 51.5 (C(H)O), 50.9 (C(H)O), 47.6 (C(H)O), 47.6 (C(H)O), 28.0 (C(CH₃)₂), 26.2 (C(CH₃)₂), 20.8 (COCH₃); $\nu_{\rm max}$ (film) 2991, 2938, 1742, 1455, 1435, 1380, 1231, 1175, 1063, 1043, 992, 964, 803, 630 cm⁻¹; HRMS (ESI+) m/z calcd for (C₁₂H₁₆NaO₆)⁺, 279.0839; found 279.0861.

Synthesis of ((1aR,1bR,2aR,2bS,5aR,5bS)-4,4-dimethyltetrahydrobis(oxireno)[2',3':3,4;2",3":5,6]benzo[1,2-d][1,3]dioxol-2b-(1aH)-yl)methanol (41). Acetate 40 (23 mg, 0.089 mmol) was dissolved in MeOH (10 mL) at room temperature, $NH_{3(g)}$ was slowly bubbled through the reaction mixture for 10 h, then the reaction mixture was concentrated under reduced pressure and purified via flash column chromatography (50% EtOAcpetrol) to give 41 (18 mg, 95%) as a colourless oil. $R_{\rm f}$ = 0.20 (50% EtOAc-petrol); $[\alpha]_{D}^{25}$ -46 (*c* 1.02, CH₂Cl₂); ¹H NMR $(250 \text{ MHz}, \text{CDCl}_3) \delta = 4.41 (1\text{H}, \text{d}, J = 2.0 \text{ Hz}, \text{C}(H)\text{OC}(\text{CH}_2)_3),$ 3.77 (1H, d, J = 11.0 Hz, -CHH-), 3.61 (1H, d, J = 11.0 Hz, -CHH-), 3.57 (1H, t, J = 3.0 Hz C(H)O), 3.53 (1H, t, J = 3.5 Hz, C(H)O), 3.41 (1H, dd, J = 3.5, 2.0 Hz, C(H)O), 3.04 (1H, dd, J = 3.5, 2.0 Hz, C(H)O), 2.09 (3H, br s, OH), 1.44 (6H, s, CH₃); ¹³C NMR (100 MHz, CDCl₃) 110.0 (C(CH₃)₂), 79.4 (CO(CH₂), 71.5 (C(H)OC(CH₃)₂), 64.4 (CH₂), 51.7 (C(H)O), 51.4 (C(H)O), 47.6 (C(H)O), 47.5 (C(H)O), 28.1 (CH₃), 26.5 (CH₃); v_{max} (film) 3491, 2982, 2253, 1457, 1383, 1247, 1219, 1080, 1063, 907, 726, 647 cm⁻¹; HRMS (ESI+) m/z calcd for (C₁₀H₁₄NaO₅)⁺, 237.0733; found 237.0792.

Acknowledgements

We thank EPSRC for funding (Studentship to JAG, Grant EP/G03768X/1; Impact Acceleration Account award to SEL).

Notes and references

 For reviews, see: (a) D. J.-Y. D. Bon, B. Lee, M. G. Banwell and I. A. Cade, Chim. Oggi, 2012, 30, 22; (b) T. Hudlický, Pure Appl. Chem., 2010, 82, 1785; (c) T. Hudlický and J. W. Reed, Synlett, 2009, 685; (d) K. A. B. Austin, M. Matveenko, T. A. Reekie and M. G. Banwell, Chem. Aust., 2008, 75, 3; (e) D. R. Boyd and T. D. H. Bugg, Org. Biomol. Chem., 2006, 4, 181; (f) R. A. Johnson, Org. React., 2004, 63, 117; (g) M. G. Banwell, A. J. Edwards, G. J. Harfoot, K. A. Jolliffe, M. D. McLeod, K. J. McRae, S. G. Stewart and M. Vögtle, Pure Appl. Chem., 2003, 75, 223; (h) T. Hudlický, D. Gonzales and D. T. Gibson, Aldrichimica Acta, 1999, 32, 35; (i) D. A. Widdowson, D. W. Ribbons and S. D. Thomas, Janssen Chim. Acta, 1990, 8, 3.

- 2 (a) J. M. Wallis and J. K. Kochi, J. Am. Chem. Soc., 1988,
 110, 8207; (b) W. B. Motherwell and A. S. Williams, Angew. Chem., Int. Ed. Engl., 1995, 34, 2031; (c) P. M. J. Jung,
 W. B. Motherwell and A. S. Williams, Chem. Commun.,
 1997, 1283; (d) Y. Feng, C.-Y. Ke, G. Xue and L. Que Jr., Chem. Commun., 2009, 50.
- 3 D. T. Gibson, J. R. Koch, C. L. Schuld and R. E. Kallio, *Biochem.*, 1968, 7, 3795.
- 4 For recent (2011 onwards) examples, see: (a) D. K. Winter, M. A. Endoma-Arias, T. Hudlický, J. A. Beutler and J. A. Porco Jr., J. Org. Chem., 2013, 78, 7617; (b) V. Varghese and T. Hudlický, Synlett, 2013, 369; (c) X. Ma, M. G. Banwell and A. C. Willis, J. Nat. Prod., 2013, 76, 1514; (d) B. D. Schwartz, E. Matoušová, R. White, M. G. Banwell and A. C. Willis, Org. Lett., 2013, 15, 1934; (e) D. J.-Y. D. Bon, M. G. Banwell, J. S. Ward and A. C. Willis, Tetrahedron, 2013, 69, 1363; (f) D. R. Boyd, N. D. Sharma, J. G. Carroll, P. L. Loke, C. R. O'Dowd and C. C. R. Allen, RSC Adv., 2013, 3, 10944; (g) J. C. Ramos, M. Brovetto and G. A. Seoane, Org. Lett., 2013, 15, 1982; (h) M. G. Banwell, N. Y. Gao, X. Ma, L. Petit, L. V. White, B. D. Schwartz, A. C. Willis and I. A. Cade, Pure Appl. Chem., 2012, 84, 1329; (i) M. J. Palframan, G. Kociok-Köhn and S. E. Lewis, Chem.-Eur. J., 2012, 18, 4766; (*j*) D. R. Boyd, N. D. Sharma, M. Kaik, P. B. A. McIntyre, P. J. Stevenson and C. C. R. Allen, Org. Biomol. Chem., 2012, 10, 2774; (k) M. K. Sharma, M. G. Banwell, A. C. Willis and A. D. Rae, Chem. Asian J., 2012, 7, 676; (1) M. G. Banwell, N. Y. Gao, B. D. Schwartz and L. V. White, Top. Curr. Chem., 2012, 309, 163; (m) D. J.-Y. D. Bon, M. G. Banwell, I. A. Cade and A. C. Willis, Tetrahedron, 2011, 67, 8348; (n) M. A. A. Endoma-Arias and T. Hudlický, Tetrahedron Lett., 2011, 52, 6632; (o) B. D. Schwartz, M. G. Banwell and I. A. Cade, Tetrahedron Lett., 2011, 52, 4526; (p) L. V. White, B. D. Schwartz, M. G. Banwell and A. C. Willis, J. Org. Chem., 2011, 76, 6250; (q) M. J. Palframan, G. Kociok-Köhn and S. E. Lewis, Org. Lett., 2011, 13, 3150; (r) J. Duchek, T. G. Piercy, J. Gilmet and T. Hudlický, Can. J. Chem., 2011, 89, 709; (s) M. G. Banwell, A. L. Lehmann, R. S. Menon and A. C. Willis, Pure Appl. Chem., 2011, 83, 411; (t) D. R. Adams, C. Aichinger, J. Collins, U. Rinner and T. Hudlický, Synlett, 2011, 725; (u) M. Labora, V. Schapiro and E. Pandolfi, Tetrahedron: Asymmetry, 2011, 22, 1705; (v) X. Ma, J. C. Jury and M. G. Banwell, Tetrahedron Lett., 2011, 52, 2192.
- 5 For selected examples, see: (a) L. Werner, A. Machara, B. Sullivan, I. Carrera, M. Moser, D. R. Adams and T. Hudlický, J. Org. Chem., 2011, 76, 10050; (b) L. Werner, A. Machara and T. Hudlický, Adv. Synth. Catal., 2010, 352, 195; (c) F. Fabris, J. Collins, B. Sullivan, H. Leisch and T. Hudlický, Org. Biomol. Chem., 2009, 7, 2619; (d) B. Sullivan, I. Carrera, M. Drouin and T. Hudlický, Angew. Chem., Int. Ed., 2009, 48, 4229; (e) M. Matveenko, A. C. Willis and M. G. Banwell, Tetrahedron Lett., 2008, 49, 7018; erratum, 2009, 50, 2982; (f) J.-J. Shie, J.-M. Fang and C.-H. Wong, Angew. Chem., Int. Ed., 2008, 47, 5788;

(g) A. T. Omori, K. J. Finn, H. Leisch, R. J. Carroll and T. Hudlický, Synlett, 2007, 2859; (h) M. G. Charest, C. D. Lerner, J. D. Brubaker, D. R. Siegel and A. G. Myers, Science, 2005, 308, 395; (i) M. G. Charest, D. R. Siegel and A. G. Myers, J. Am. Chem. Soc., 2005, 127, 8292;
(j) I. W. Davies and P. J. Reider, Chem. Ind., 1996, 11, 412;
(k) I. W. Davies, C. H. Senanayake, L. Castonguay, R. D. Larsen, T. R. Verhoeven and P. J. Reider, Tetrahedron Lett., 1995, 36, 7619.

6 For selected examples, see: (a) J. A. Griffen, J. C. White, G. Kociok-Köhn, M. D. Lloyd, A. Wells, T. C. Arnot and Ε. Lewis, Tetrahedron, 2013, 69, 5989; S. (b) R. M. B. Carrilho, V. Heguaburu, V. Schapiro, E. Pandolfi, L. Kollár and M. M. Pereira, Tetrahedron, 2012, 68, 6935; (c) V. de la Sovera, A. Bellomo, J. M. Pena, D. Gonzalez and H. A. Stefani, Mol. Diversity, 2011, 15, 163; (d) J. Duchek, D. R. Adams and T. Hudlický, Chem. Rev., 2011, 111, 4223; (e) S. Pilgrim, G. Kociok-Köhn, M. D. Lloyd and S. E. Lewis, Chem. Commun., 2011, 47, 4799; (f) J. C. Ramos, P. Bracco, M. Mazzini, J. R. Fernández, D. Gamenara and G. A. Seoane, Tetrahedron: Asymmetry, 2010, 21, 969; (g) D. R. Boyd, N. D. Sharma, N. I. Bowers, G. B. Coen, J. F. Malone, C. R. O'Dowd, P. J. Stevenson and C. C. R. Allen, Org. Biomol. Chem., 2010, 8, 1415; (h) A. Bellomo, A. Bertucci, H. Stefanic, A. Vázquez and D. Gonzalez, Tetrahedron: Asymmetry, 2009, 20, 2673; (i) A. Bellomo, J. B. Bonilla, J. López-Prados, M. Martín-Lomas and D. Gonzalez, Tetrahedron: Asymmetry, 2009, 20, 2061; (j) A. Bellomo, S. Camarano, C. Rossini and D. Gonzalez, Carbohydr. Res., 2009, 344, 44; (k) A. Bellomo, D. Gonzalez and H. A. Stefani, J. Organomet. Chem., 2008, 693, 1136; (l) A. Bellomo and D. Gonzalez, Tetrahedron Lett., 2007, 48, 3047; (m) G. Fonseca and G. A. Seoane, Tetrahedron: Asymmetry, 2005, 16, 1393; (n) D. R. Boyd, N. D. Sharma, N. M. Llamas, J. F. Malone, C. R. O'Dowd and C. C. R. Allen, Org. Biomol. Chem., 2005, 3, 1953; (o) M. H. Parker, B. E. Maryanoff and A. B. Reitz, Synlett, 2004, 2095; (p) C. Vitelio, A. Bellomo, M. Brovetto, G. Seoane and D. Gonzalez, Carbohydr. Res., 2004, 339, 1773; (q) M. G. Banwell, X. Ma, N. Asano, K. Ikeda and J. N. Lambert, Org. Biomol. Chem., 2003, 1, 2035; (r) B. J. Paul, J. Willis, T. Martinot, I. Ghiviriga, K. A. Abboud and T. Hudlický, J. Am. Chem. Soc., 2002, 124, 10416; (s) M. G. Banwell, A. M. Bray and D. J. Wong, New J. Chem., 2001, 25, 1351; (t) S. H. Lee and C. S. Cheong, Bull. Korean Chem. Soc., 2000, 21, 1061; (u) M. G. Banwell, C. DeSavi, D. C. R. Hockless, S. Pallich and K. G. Watson, Synlett, 1999, S1, 885-888; (v) M. Banwell, C. DeSavi and K. Watson, J. Chem. Soc., Perkin Trans. 1, 1998, 2251; (w) M. Banwell, C. DeSavi and K. Watson, Chem. Commun., 1998, 1189; (x) M. Desjardins, L. E. Brammer Jr. and T. Hudlický, Carbohydr. Res., 1997, 304, 39; (y) T. Hudlický, K. A. Abboud, D. A. Entwistle, R. Fan, R. Maurya, A. J. Thorpe, J. Bolonick and B. Myers, Synthesis, 1996, 897. 7 (a) V. P. Bui and T. Hudlický, Tetrahedron, 2004, 60, 641;

(b) M. G. Williams, P. E. Olson, K. J. Tautvydas,

R. M. Bitner, R. A. Mader and L. P. Wackett, *Appl. Microbiol. Biotechnol.*, 1990, **34**, 316; (c) D. G. H. Ballard, A. Courtis, I. M. Shirley and S. C. Taylor, *Macromolecules*, 1988, **21**, 294; (d) D. G. H. Ballard, A. Courtis, I. M. Shirley and S. C. Taylor, *J. Chem. Soc., Chem. Commun.*, 1983, 954.

- 8 (a) J. L. Royo, E. Moreno-Ruiz, A. Cebolla and E. Santero, J. Biotechnol., 2005, 116, 113; (b) J. Y. Kim, K. Lee, Y. Kim, C.-K. Kim and K. Lee, Lett. Appl. Microbiol., 2003, 36, 343; (c) A. Berry, T. C. Dodge, M. Pepsin and W. Weyler, J. Ind. Microbiol. Biotechnol., 2002, 28, 127; (d) B. D. Ensley, B. J. Ratzkin, T. D. Osslund, M. J. Simon, L. P. Wackett and D. T. Gibson, Science, 1983, 222, 167.
- 9 (a) T. D. H. Bugg, *Tetrahedron*, 2003, 59, 7075;
 (b) D. T. Gibson and R. E. Parales, *Curr. Opin. Biotechnol.*, 2000, 11, 236.
- 10 D. R. Boyd, N. D. Sharma, M. V. Hand, M. R. Groocock, N. A. Kerley, H. Dalton, J. Chima and G. N. Sheldrake, *J. Chem. Soc., Chem. Commun.*, 1993, 974.
- 11 A. M. Reiner and G. D. Hegeman, *Biochemistry*, 1971, 10, 2530.
- 12 A. E. G. Cass, D. W. Ribbons, J. T. Rossiter and S. R. Williams, *Biochem. Soc. Trans.*, 1986, 14, 1268.
- 13 S.-Y. Sun, X. Zhang, Q. Zhou, J.-C. Chen and G.-Q. Chen, *Appl. Microbiol. Biotechnol.*, 2008, **80**, 977.
- 14 G. N. Jenkins, D. W. Ribbons, D. A. Widdowson, A. M. Z. Slawin and D. J. Williams, *J. Chem. Soc., Perkin Trans.* 1, 1995, 2647.
- 15 A. G. Myers, D. R. Siegel, D. J. Buzard and M. G. Charest, *Org. Lett.*, 2001, **3**, 2923.
- 16 (a) M. D. Mihovilovic, H. G. Leisch and K. Mereiter, *Tetrahedron Lett.*, 2004, 45, 7087; (b) T. C. M. Fischer, H. G. Leisch and M. D. Mihovilovic, *Monatsh. Chem.*, 2010, 141, 699.
- 17 (a) M. Ali Khan, M. F. Mahon, J. P. Lowe, A. J. W. Stewart and S. E. Lewis, *Chem.-Eur. J.*, 2012, **18**, 13480; (b) D. van der Waals, T. Pugh, M. Ali Khan, A. J. W. Stewart, A. L. Johnson and S. E. Lewis, *Chem. Central J.*, 2011, 5, 80; (c) J. A. Griffen, A. M. Le Coz, G. Kociok-Köhn, M. Ali Khan, A. J. W. Stewart and S. E. Lewis, *Org. Biomol. Chem.*, 2011, **9**, 3920; (d) M. Ali Khan, J. P. Lowe, A. L. Johnson, A. J. W. Stewart and S. E. Lewis, *Chem. Commun.*, 2011, **47**, 215; (e) M. Ali Khan, M. F. Mahon, A. J. W. Stewart and S. E. Lewis, *Organometallics*, 2010, **29**, 199.
- 18 (a) J. P. Guthrie, J. Cossar and J. Lu, Can. J. Chem., 1991, 69, 1904; (b) G. Hesse and G. Krehbiel, Liebigs Ann., 1955, 593, 42.
- 19 D. R. Boyd, N. D. Sharma, M. V. Berberian, K. S. Dunne, C. Hardacre, M. Kaik, B. Kelly, J. F. Malone, S. T. McGregor and P. J. Stevenson, *Adv. Synth. Catal.*, 2010, 352, 855.
- 20 J. G. Buchanan, A. R. Edgar, D. I. Rawson, P. Shahidi and R. H. Wightman, *Carbohydr. Res.*, 1982, **100**, 75.
- 21 (a) J. T. Sharp, I. Gosney and A. G. Rowley, in *Practical Organic Chemistry*, Chapman and Hall, New York, 1989;
 (b) J. M. Bohen, M. M. Joullié, F. A. Kaplan and B. Loev, *J. Chem. Educ.*, 1973, **50**, 367.

- 22 For reviews on inositols, see: (a) B. Kilbas and M. Balci, *Tetrahedron*, 2011, 67, 2355; (b) S. J. Conway and G. J. Miller, *Nat. Prod. Rep.*, 2007, 24, 687; (c) B. V. L. Potter and D. Lampe, *Angew. Chem., Int. Ed. Engl.*, 1995, 34, 1933.
- 23 T. Hudlický, D. A. Entwistle, K. K. Pitzer and A. J. Thorpe, *Chem. Rev.*, 1996, **96**, 1195.
- 24 (a) H. A. J. Carless and Y. Dove, Tetrahedron: Asymmetry, 1996, 7, 649; (b) H. A. J. Carless and S. S. Malik, J. Chem. Soc., Chem. Commun., 1995, 2447; (c) H. A. J. Carless, K. Busia, Y. Dove and S. S. Malik, J. Chem. Soc., Perkin Trans. 1, 1993, 2505; (d) H. A. J. Carless and K. Busia, Carbohydr. Res., 1992, 234, 207; (e) H. A. J. Carless, Tetrahedron Lett., 1992, 33, 6379; (f) H. A. J. Carless, K. Busia and O. Z. Oak, Synlett, 1993, 672; (g) H. A. J. Carless and S. S. Malik, Tetrahedron: Asymmetry, 1992, 3, 1135; (h) H. A. J. Carless, J. Chem. Soc., Chem. Commun., 1992, 234; (i) H. A. J. Carless and O. Z. Oak, Tetrahedron Lett., 1991, 32, 1671; (j) H. A. J. Carless and O. Z. Oak, J. Chem. Soc., Chem. Commun., 1991, 61; (k) H. A. J. Carless and 1990, Κ. Busia, Tetrahedron Lett., 31, 3449; (l) H. A. J. Carless and K. Busia, Tetrahedron Lett., 1990, 31, 1617; (m) H. A. J. Carless, J. A. Billinge and O. Z. Oak, Tetrahedron Lett., 1989, 30, 3113; (n) H. A. J. Carless and O. Z. Oak, Tetrahedron Lett., 1989, 30, 1719.
- 25 (a) S. Freeman and T. Hudlický, Bioorg. Med. Chem. Lett., 2004, 14, 1209; (b) B. J. Paul, E. Hobbs, P. Buccino and T. Hudlický, Tetrahedron Lett., 2001, 42, 6433; (c) B. J. Paul, T. A. Martinot, J. Willis and T. Hudlický, Synthesis, 2001, 952; (d) T. Hudlický, N. Restrepo-Sanchez, P. D. Kary and L. M. Jaramillo-Gomez, Carbohydr. Res., 2000, 324, 200; (e) T. Hudlický, C. D. Claeboe, L. E. Brammer Jr., L. Koroniak, G. Butora and I. Ghiviriga, J. Org. Chem., 1999, 64, 4909; (f) K. A. Oppong, T. Hudlický, F. Yan, C. York and Tetrahedron, V. Nguyen, 1999, 55, B. 2875: (g) L. E. Brammer Jr. and T. Hudlický, Tetrahedron: Asymme*try*, 1998, 9, 2011; (*h*) M. Desjardins, L. E. Brammer Jr. and T. Hudlický, Carbohydr. Res., 1997, 304, 39; (i) B. V. Nguyen, C. York and T. Hudlický, Tetrahedron, 1997, 53, 8807; (j) T. Hudlický, K. A. Abboud, J. Bolonick, R. Maurya, M. L. Stanton and A. J. Thorpe, Chem. Commun., 1996, 1717; (k) T. Hudlický, K. A. Abboud, D. A. Entwistle, R. Fan, R. Maurya, A. J. Thorpe, J. Bolonick and B. Myers, Synthesis, 1996, 897; (l) T. Hudlický and A. J. Thorpe, Synlett, 1994, 899; (m) T. Hudlický, M. Mandel, J. Rouden, R. S. Lee, B. Bachmann, T. Dudding, K. J. Yost and J. S. Merola, J. Chem. Soc., Perkin Trans. 1, 1994, 1553; (n) M. Mandel and T. Hudlický, J. Chem. Soc., Perkin Trans. 1, 1993, 741; erratum, p. 1537; (o) M. Mandel, T. Hudlický, L. D. Kwart and G. M. Whited, J. Org. Chem., 1993, 58, 2331.
- 26 T. Tschamber, F. Backenstrass, H. Fritz and J. Streith, *Helv. Chim. Acta*, 1992, **75**, 1052.
- 27 (a) S. V. Ley, A. J. Redgrave and L. L. Yeung, ACS Symp. Ser., 1991, 463, 132; (b) S. V. Ley, Pure Appl. Chem., 1990, 62, 2031; (c) S. V. Ley, M. Parra, A. J. Redgrave and F. Sternfeld, Tetrahedron, 1990, 46, 4995; (d) S. V. Ley and F. Sternfeld, Tetrahedron, 1989, 45, 3463; (e) S. V. Ley, M. Parra,

A. J. Redgrave, F. Sternfeld and A. Vidal, *Tetrahedron Lett.*, 1989, **30**, 3557; (*f*) S. V. Ley and F. Sternfeld, *Tetrahedron Lett.*, 1988, **29**, 5305; (*g*) S. V. Ley, F. Sternfeld and S. Taylor, *Tetrahedron Lett.*, 1987, **28**, 225.

- 28 For examples, see: (a) A. Baran, G. Aydin, T. Savran, E. Sahin and M. Balci, Org. Lett., 2013, 15, 4350; (b) Z. Ekmekci and M. Balci, Eur. J. Org. Chem., 2012, 4988; (c) A. Baran, S. Cambul, M. Nebioglu and M. Balci, J. Org. Chem., 2012, 77, 5086; (d) N. I. Kurbanoğlu, M. Celik, H. Kilic, C. Alp, E. Sahin and M. Balci, Tetrahedron, 2010, 66, 3485; (e) S. Cantekin, A. Baran, R. Calişkan and M. Balci, Carbohydr. Res., 2009, 344, 426; (f) A. Baran and M. Balci, J. Org. Chem., 2009, 74, 88; (g) A. Coşkun, M. Güney, A. Dastan and M. Balci, Tetrahedron, 2007, 63, 4944; (h) M. Celik and M. Balci, ARKIVOC, 2007, 150; (i) E. Salamci, B. Bayram and M. Balci, Turk. J. Chem., 2006, 30, 441; (j) S. D. Yardımcı, N. Kaya and M. Balci, Tetrahedron, 2006, 62, 10633; (k) L. Kelebekli, M. Çelik, E. Şahin, Y. Kara and M. Balci, Tetrahedron Lett., 2006, 47, 7031; (l) L. Kelebekli, Y. Kara and M. Balci, Carbohydr. Res., 2005, 340, 1940; (m) M. S. Gültekin, M. Çelik, E. Turkut, C. Tanyeli and M. Balci, Tetrahedron: Asymmetry, 2004, 15, 453; erratum, p. 1959; (n) M. S. Gültekin, E. Salamci and M. Balci, Carbohydr. Res., 2003, 338, 1615; (o) Y. Kara and M. Balci, Tetrahedron, 2003, 59, 2063.
- (a) G. Mehta and S. Sen, *Tetrahedron Lett.*, 2010, 51, 503;
 (b) G. Luchetti, K. Ding, M. d'Alarcao and A. Kornienko, *Synthesis*, 2008, 3142;
 (c) Y. J. Lee, K. Lee, S. I. Jung, H. B. Jeon and K. S. Kim, *Tetrahedron*, 2005, 61, 1987;
 (d) G. Mehta, R. S. Senaiar and S. Sen, *Proc. Indian Acad. Sci. A*, 2005, 71, 121;
 (e) G. Mehta, R. S. Senaiar and M. K. Bera, *Chem.-Eur. J.*, 2003, 9, 2264;
 (f) A. Kornienko and M. d'Alarcao, *Tetrahedron: Asymmetry*, 1999, 10, 827;
 (g) K. K. Reddy, J. R. Falck and J. Capdevila, *Tetrahedron Lett.*, 1993, 34, 7869.
- 30 (a) H. Paulsen and F. R. Heiker, Liebigs Ann., 1981, 2180;
 (b) H. Paulsen and F. R. Heiker, Angew. Chem., 1980, 92, 930;
 (c) S. J. Angyal, J. E. Klavins and J. A. Mills, Aust. J. Chem., 1974, 27, 1075;
 (d) T. Posternak, D. Janjic, E. A. C. Lucken and A. Szente, Helv. Chim. Acta, 1967, 50, 1027;
 (e) T. Posternak, Proc. 2nd Meet. Fed. Eur. Biochem. Soc., 1966, p. 31;
 (f) T. Posternak and J.-G. Falbriard, Helv. Chim. Acta, 1960, 43, 2142;
 (g) T. Posternak and D. Reymond, Helv. Chim. Acta, 1953, 36, 1370;
 (h) T. Posternak, Helv. Chim. Acta, 1950, 33, 1594;
 (i) T. Posternak, Helv. Chim. Acta, 1944, 27, 457.
- 31 (a) J. D. Boyd, C. S. Foote and D. K. Imagawa, J. Am. Chem. Soc., 1980, 102, 3641; (b) Y. Sütbeyaz, H. Seçen and M. Balci, J. Org. Chem., 1988, 53, 2312; (c) M. Balci and Y. Sütbeyaz, Tetrahedron Lett., 1983, 24, 311.
- 32 (a) C. W. Jefford, *Chem. Soc. Rev.*, 1993, 22, 59;
 (b) A. A. Frimer, *Chem. Rev.*, 1979, 79, 359; (c) C. W. Jefford and C. G. Rimbault, *J. Org. Chem.*, 1978, 43, 1908.
- 33 (a) T. Nagasawa, N. Shimada, M. Torihata and S. Kuwahara, *Tetrahedron*, 2010, **66**, 4965; (b) J. Huang and E. J. Corey, *Org. Lett.*, 2003, **5**, 3455.