



CrossMark
 click for updates

Cite this: *RSC Adv.*, 2017, 7, 1724

Electrospray biodegradable microcapsules loaded with curcumin for drug delivery systems with high bioactivity†

Zhuoxian Mai,^{‡a} Jiali Chen,^{‡b} Ting He,^a Yang Hu,^a Xianming Dong,^a Hongwu Zhang,^{*b} Wenhua Huang,^b Frank Ko^{*c} and Wuyi Zhou^{*a}

In the present work, polylactic acid (PLA) microcapsules as novel drug delivery systems were successfully fabricated by one-step processing using an electrospray technique. Curcumin (Cur) was chosen as model drug with satisfactory loading capacity (LC%) and entrapment efficiency (EE%) greater than 95%. By judiciously adjusting spinning solvents, flow rates, polymer and drug concentrations, the monodisperse and spherical structures of Cur/PLA microcapsules observed using scanning electron microscopy (SEM) and optical microscopy were successfully generated with diameter distribution ranging from 3.8 μm to 4.4 μm . The physical-chemical characterization including, FTIR, XRD, TG, and DTA, are explored and *in vitro* release profiles described by Ritger–Peppas models were also investigated, showing sustained release for 200 h after a burst release in the initial 12 h. The drug-loaded microcapsules showed excellent anti-bacterial activities towards *Escherichia coli* (*E. coli*) and *Staphylococcus aureus* (*S. aureus*) by using both disk diffusion and minimal inhibitory concentration methods. The anti-oxidant performance was also evaluated by using DPPH assays. *In vitro* cell tests, including Cell Counting Kit-8, hemolysis experiments, and cell adhesion revealed that PLA-based microcapsules had significant biocompatibility and low cytotoxicity. The study showed that the PLA-based electrospray strategy combined with spherical microcapsules has the potential for a broad range of applications in medical fields, especially in drug delivery.

Received 16th October 2016
 Accepted 13th December 2016

DOI: 10.1039/c6ra25314h

www.rsc.org/advances

Introduction

Nowadays, more and more substances such as mesoporous inorganic materials,^{1–4} microspheres,^{5–8} fibrous membranes,^{9–12} gelatin,^{13,14} and polymeric scaffolds^{15–19} have been extensively studied as carriers in drug delivery systems (DDSs) to encapsulate drugs and regulate release behavior, aiming to offer long-term durability and targeted therapy, as well as improving drug stability and bioavailability.^{20–22} Accordingly, therapeutic efficacy could be not only influenced by the nature of drug itself, but also the methods and materials used to construct the DDSs. Among these, compared to the non-biodegradable polymers, biopolymer has attracted more attention due to their excellent biocompatibility and biodegradation, improved bioavailability

of drugs.^{23,24} Moreover, the employment of biodegradable materials can immensely reduce the side effect and the risks of undesired toxicity inside the body.^{8,30} In recent years, polyethylene glycol (PEG),²⁵ poly(lactic-co-glycolic) acid (PLGA),²⁶ polycaprolactone (PCL),²⁷ polyethylene oxide (PEO),²⁸ polyvinyl alcohol (PVA)²⁹ and chitosan (CS)³⁰ have been widely employed in biomedical application owing to their excellent biodegradation properties inside the bodies. Specifically, polylactic acid (PLA) has been widely used as bioactive and biocompatible materials for different biomedical applications, such as drug delivery,³¹ wound dressing,³² artificial skin³³ and bone and tissue engineering.³⁴ CO₂ and H₂O are supposed to be the final products of PLA *in vivo*, which are nontoxic and friendly to human and environment. Additionally, PLA is more preferable due to its low-cost and availability from renewable sources. Furthermore, its properties such as molecular weight and degradation rates can be tuned to modify the controlled-release profiles of encapsulated drugs.³¹

In particular, PLA-based microcapsules have been largely investigated and proven to be one of the most efficient ways to encapsulate the therapeutic agents, providing high drug loading rate and sustain-release profiles. The conventional techniques of constructing microcapsules involve complex coacervation,^{5–7} solvent evaporation/extraction,³⁵ spray drying,³⁶

^aInstitute of Biomaterial, College of Materials and Energy, South China Agricultural University, Guangzhou, 510642, China. E-mail: zhouwuyi@scau.edu.cn

^bDepartment of Anatomy, Guangdong Provincial Key Laboratory of Construction and Detection in Tissue Engineering, Southern Medical University, Guangzhou 510515, China. E-mail: huangwenhua2009@139.com

^cDepartment of Materials Engineering, The University of British Columbia, Vancouver, BC, Canada V6T 1Z4. E-mail: frank.ko@ubc.ca

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c6ra25314h

‡ Both authors contributed equally to this work.



precipitation,³⁷ *in situ* polymerization,³⁸ *etc.* However, limitation of the above-mentioned methods was thoroughly investigated, demonstrating that all these techniques required complicate process and harsh conditions. Moreover, the key restriction is the difficulty to control the morphology and size distribution, which has been reported to be vital to tailor the drug release profiles and circulation inside the body. Monodispersity and appropriate shape of the microcapsules are more preferred since various uniformity and size distribution had different impact on the mechanisms of drug delivery in comprehensive ways.^{8,30,35}

Electrospray (ES) has emerged to be one of the most cost-effective and versatile technique to prepare polymeric fibrous membranes and micro/nanoparticles with diameters ranging from tens of nanometer to a few micrometer.^{8–12,29–31} The electro-spray process depends on external high voltage electrical field and columbic repulsion, which induces the solvent of the charged droplets evaporate quickly, leaving micro/nano-sized solid directly reach the receptor. In the course of electro-spraying, the solvent evaporates from the droplet and the drug remains entrapped within the polymer structure, ideally randomly distributed. The ES process have many unique advantages, *i.e.*, high-effectiveness, rapidness and ease of operation.^{29–31} Additionally, both morphology and particle size, which influences the cellular uptake and tissue uptake, can be facily controlled by adjusting the operating parameters, *i.e.*, flow rate, external voltage and polymer concentration, *etc.* Furthermore, either hydrophobic or hydrophilic drugs could be encapsulated into electro-spray microcapsules with entrapment efficiency up to 100%. As a result, electro-spray micro/nano capsules were extensively investigated as novel drug vehicles owing to the above-mentioned advantages.

Curcumin (Cur), which was obtained from rhizomes of the *Curcuma longa* and used as food additives in lots of Asian countries, is becoming more and more popular for its diverse therapy properties and beneficial bioactivities. It has various biological applications such as antioxidant, anti-cancer, anti-microbial, antiphlogosis and antilipidemic effect.^{3,4} However, curcumin molecules were found to be unstable to a variety of physical and chemical environments, *e.g.*, heat, pH, light or alkali medium. Furthermore, its poor water solubility results in lower bioavailability, restricting its practical clinical trials. Therefore, in order to overcome these drawbacks, researchers have devoted to improve a series of drug delivery systems with enhanced remedial properties, including biocompatibility, hemocompatibility, cytotoxicity, cell adhesion and proliferation.^{25–27} A number of different techniques have been developed to construct such bioactive platforms. Compared to other manufacturing process, ES has been proven to be the most straightforward method to fabricate drug delivery systems.

In the present work, we firstly developed ES-based method to fabricate PLA microcapsules with uniform spherical shape and regular size distribution. Different operating parameters such as flow rate, chosen solvent, polymer/drug concentration were also investigated to further explore the mechanism of the fabrication process. Cur was chosen as the model drug molecular and successfully entrapped into the PLA-based

microcapsules through mixing drugs with the polymer solution during electro-spraying process. The obtained products were analyzed by Fourier Transform Infrared Spectroscopy (FTIR), X-Ray Diffraction (XRD), thermogravimetry (TG), and Differential Thermal Analysis (DTA). Morphology and size distribution were also explored by scanning electron microscope (SEM) and optical microscope (OM). The biocompatibility and cytotoxicity were further evaluated by cell proliferation and adhesion assays, as well as the hemolysis test (HT) and Cell Counting Kit-8 (CCK-8). The *in vitro* sustained release of Cur was also tested in 5 wt% SDS phosphate solution. Moreover, several bioactivities, *i.e.*, antioxidant and antibacterial were also demonstrated by using 1,1-diphenyl-2-picrylhydrazyl (DPPH) assays and disk diffusion methods. Minimal inhibitory concentration method was also employed to quantitatively investigate the antibacterial ability. To the best of our knowledge, this is the first time to combine ES-based PLA microcapsules with Cur drugs, which gives an insight into how electro-spray can provide a new method to fabricate polymeric novel drug carrying vehicles.

Experimental methods

Materials

PLA ($M_w = 80\,000\text{--}100\,000$) was supplied by Jiejin Chemical Co., Ltd. (Guangdong, China). Curcumin (Cur) and 1,1-diphenyl-2-picrylhydrazyl (DPPH) were purchased from Sigma-Aldrich. Acetone (AC), dichloromethane (DCM), tetrahydrofuran (THF) and trichloromethane (TCM) were provided by Sinopharm Chemical Reagent Co., Ltd. Human dermal fibroblast (HDF), Human embryonic kidney (HEK) 293T cells and rat pheochromocytoma PC12 cells were obtained from American Type Culture Collection (ATCC), USA. Fetal Bovine Serum (FBS) was supplied by GIBCO, Invitrogen, USA. 5 wt% SDS (sodium dodecyl sulfate) solution was prepared as medium for drug release *in vitro*. All other materials and reagents used in this work were analytical graded unless otherwise noted.

Experiment setup

The schematic diagram of ES process is shown in Fig. 1. Spinning solutions were prepared in advance by dissolving different

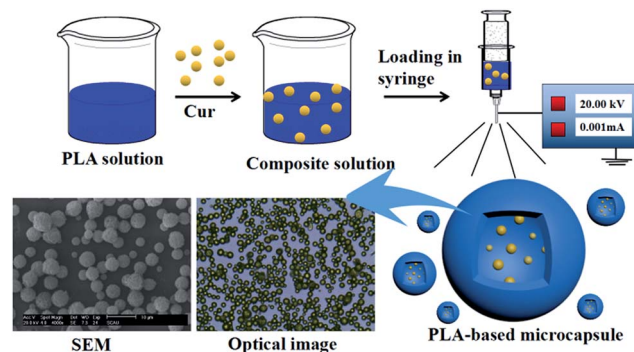


Fig. 1 Schematic description for fabrication of PLA-based microcapsules loaded with curcumin through electro-spray process.



Table 1 Summary of operating parameters setup for the ES fabrication process

Spinning solvent	PLA concentration	Feeding rate	Drug concentration
Acetone	1%	0.1 mL h ⁻¹	0%
Dichloromethane	3%	0.3 mL h ⁻¹	5%
Tetrahydrofuran	5%	0.5 mL h ⁻¹	10%
Trichloromethane	7%	0.7 mL h ⁻¹	15%

amount of PLA in various chosen solvent to achieve polymer mass fraction ranging from 1% to 7%. Also, different amount of Cur (ranging from 0% to 15%) was added to the polymeric solution. Each solution was kept under stirring overnight to form homogeneous system. All solutions were degassed and fed into 10 mL syringe, combined with a stainless steel capillary. The distance between the needle and grounded receptor was kept at 15 cm and applied voltage was maintained at 20 kV. The flow rates were kept at different values by a DT-200 precision syringe pump (Dalian Dingtong Technology Co., Ltd., China) to investigate the influence on morphology of the microcapsules. The obtained powders on grounded aluminum foil were collected and dried at 50 °C to remove the residual solvent. The powder was denoted as Cur@PLA (or CPLA) and details on different setup parameters were listed in Table 1. In particular, drug delivery systems with various drug concentrations (0%, 5%, 10%, 15%) were denoted as C₀PLA, C₅PLA, C₁₀PLA and C₁₅PLA, respectively.

Characterization

The morphology and shape of the obtained microcapsules under different operating parameters were observed *via* scanning electron microscope (SEM, Hitachi S-4800) and optical microscope (OM, Olympus CX23). The average particle sizes and distributions were determined by analyzing the SEM micrographs using the software nano-measurer ($n = 200$). Crystal structures of both PLA and Cur were determined by using XRD (D8 Advance, BRUKER). FTIR (Nicolet-Avatar 360) was used to evaluate the chemical structure and stability of drugs entrapped in carriers. The thermal degradation properties of different samples were examined using TG and DTA (DTG-60, SHIMADZU) under a Nitrogen flow rate of 20 mL min⁻¹. The drug release behavior was studied by using UV-vis spectrophotometer (UV-2550, Shimadzu).

Loading capacity of cur-loaded microcapsules

The loading capacity (LC%) and entrapment efficiency (EE%) of Cur-loaded samples were measured by the following method. In brief, 5 mg of sample was placed into 5 mL ethanol under sonication for 30 min, in order to promote the drugs release thoroughly. After that, 45 mL PBS was mixed with Cur ethanol solution, which was then kept under vigorously shaking for 2 h. The mixture was then centrifuged, and the supernatant was analyzed by UV-vis spectroscopy at the 425 wavenumber. The LC and EE were calculated by eqn (1) and (2).

$$LC\% = M_t/M_s \times 100\% \quad (1)$$

$$EE\% = M_t/M_0 \times 100\% \quad (2)$$

where M_t , M_s and M_0 are the weight of Cur entrapped inside the microcapsules, the weight of drug delivery systems and the initial weight of Cur, respectively.

In vitro release properties

To measure the drug release behaviors of Cur@PLA microcapsules, 10 mg of various samples were sealed in dialysis bags (8000–14 000 Da, Sigma-Aldrich), which were further placed in 500 mL conical flasks filled PBS (0.2 M, pH = 7.4) containing 0.5 wt% SDS. The content was kept under gentle stirred in the dark at 37 °C. At given intervals, 2 mL of drug release solution was withdrawn and then equivalent original PBS solution was added promptly, so as to keep the volume constant. The release solution was then measured by UV-vis spectrophotometer at the 425 wavenumber. The cumulative release percentage of Cur was calculated by using the following eqn (3).

$$M_{t\text{corr}} = M_t + (v/V) \sum M_t \quad (3)$$

$$\text{cumulative release rate}\% = M_{t\text{corr}}/M_{\text{total}} \times 100\%$$

where M_t and $M_{t\text{corr}}$ are apparent weight and corrected weight at time t . v is the volume of sample taken and V is the total volume of release medium. M_{total} is the total weight of Cur entrapped inside the microcapsules.

Antioxidant activity test

The antioxidant activity of Cur-loaded samples was estimated by using the DPPH assays. Briefly, various amounts of Cur-loaded samples were dispersed in ethanol to achieve different drug concentrations (80 mg mL⁻¹, 120 mg mL⁻¹, 160 mg mL⁻¹, 200 mg mL⁻¹, 240 mg mL⁻¹). After that, 5 mL 0.03 g L⁻¹ DPPH ethanol solution was mixed with 1 mL the obtained suspension. Subsequently, the reaction mixture was incubated in the dark at 25 °C for 30 min. The purple DPPH solution faded and then absorbance at 517 nm of the supernatant was examined by UV-vis spectroscopy. The antioxidant activity of each sample was assessed by using the eqn (4) below:

$$\text{DPPH}\% = \frac{A_0 - A_1}{A_0} \times 100\% \quad (4)$$

where A_1 is the absorbance of DPPH solution mixture of C₅PLA, C₁₀PLA and C₁₅PLA and A_0 is the absorbance of control group using C₀PLA.

Anti-bacterial assessment

The anti-bacterial properties against *Escherichia coli* (*E. coli*) and *Staphylococcus aureus* (*S. aureus*) were evaluated by a disk diffusion method. Briefly, *E. coli* and *S. aureus* were incubated by suspending bacterial in sterilized broth culture overnight at 37 °C. Afterwards, the inoculation bacterial suspension was further diluted with sterilized water to 10⁵ (CFU) mL⁻¹. 50 μL of the diluted bacterial suspension was then spread throughout



the nutrient agar plates. A filter paper was cut into disc with diameter of 6 mm and put on the surface of the agar plates. Next, 500 μL of Cur@PLA suspension with concentration of 10 g mL^{-1} was placed on the filter paper, which was used as support to absorb the micro samples. After incubating at 37 $^{\circ}\text{C}$ for 24 h, the diameters of inhibition zones around the paper discs were measured. In order to further investigate the antibacterial activity quantitatively, minimal inhibitory concentration method was also applied. Briefly, Cur@PLA suspended in 5 mL PBS solution was incorporated with 5 mL nutrient broth to obtain a concentration of 20 mg mL^{-1} . Then, the suspension was treated with serial two-fold method and each gradient concentration was diluted to 10 mg mL^{-1} , 5 mg mL^{-1} , 2.5 mg mL^{-1} , 1.25 mg mL^{-1} , 0.625 mg mL^{-1} , 0.3125 mg mL^{-1} and 0.1563 mg mL^{-1} , respectively. After regular culturing for 24 h, 0.1 mL of the diluted suspension was plated on agar plates. Afterwards, the lowest concentration of capsules that showed visible antibacterial effect was defined as minimal inhibitory concentration (MIC). Specifically, suspension without drug-loaded samples was used as control group.

Cytotoxicity assay and hemolysis test

In order to estimate the biocompatibility of the PLA-based microcapsules, both 293T cells and PC12 cells were incubated to assess the cytotoxicity through Cell Counting Kit-8 (CCK-8) and Hoechst 33342 strains experiments as observed by an upright fluorescence microscope (Olympus, BX50). Additionally, hemolytic test was also performed by using the blood drawn from SD rats to further assess the toxicity of the PLA-based samples. This research was conducted in accordance with the Declaration of Helsinki and with the Guide for Care and Use of Laboratory Animals as adopted and promulgated by the United National Institutes of Health. All experimental protocols were approved by the Review Committee for the Use of Human or Animal Subjects of Southern Medical University.

Cell proliferation and adhesion assay

The biocompatibility of C₀PLA and Cur@PLA was also assessed by using the human dermal fibroblast (HDF) as cell models. In brief, HDF cells were maintained firstly in DMEM, followed by adding 10% FBS and 1% anti-mycotic and anti-biotic at 37 $^{\circ}\text{C}$. The sterilized samples were immersed in the cell culture medium overnight to promote cell attachment. Typically, an empty culture plate was used as the control group. Afterwards, HDF cells was incubated with the samples at a density of 10⁴ cells per well. The number of HDF cells on different samples were counted using Beckman Vi-Cell XR Cell Viability Analyzer (Beckman Coulter, Inc., USA) after 1, 3 and 5 days. In addition, the morphology of HDF cells was also observed through SEM at specific times of cell culturing.

Statistical analysis

All data were expressed as mean standard deviation in this work. Statistical analysis was performed using *t*-test and statistical significance between mean values was regulated at $p < 0.10$.

Results and discussion

Morphology of microcapsules fabricated under different spinning parameters

Electrospray parameters, *e.g.*, flow rate, polymeric concentration and solution composition were found to be most vital factors to be the most vital factors to influence the spinning products as previous discussed.^{8–12,29–31} The relation between the morphology of as-spun PLA-based products and above-mentioned spinning parameters was firstly explored.

Effect of spinning solvent

The effect of organic solution used in ES process on the morphology of PLA-based microcapsules was firstly explored by selecting AC, DCM, THF and TCM as the spinning solvent. As is observed in Fig. 2, it is obvious that different solvents have significant impact on the morphology of the capsules. Apart from selecting TCM as solvent, other solvents were undesired because irregular and cracked spheres (Fig. 2a–c) were obtained when polymer concentration ranged from 1% to 7%. Irregular spheres were obtained when using AC as solvent during ES process, which is mainly due to the poor dissolvability of PLA in such hydrophilic organics (Fig. 2a). Other undesired structures, *i.e.*, cracked and sunken spheres, were also observed in Fig. 2b and c, which can be attributed to the volatility of DCM and THF. In particular, uniform and spherical PLA capsules can be obtained by selecting TCM as solvent, which resulted from its proper volatility and excellent solubility to PLA.

Effect of flow rate

Fig. 3 reveals the influence of flow rate on the morphology of PLA capsules during ES process. According to the above-mentioned research, TCM was chosen as the solvent to form the spinning solution. In order to investigate the influence of flow rate more precisely, we collected the samples by placing a glass slide on the grounded receptor as soon as steady polymer

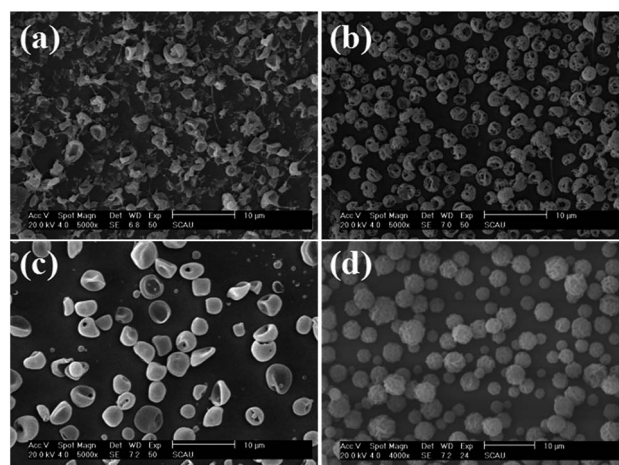


Fig. 2 SEM images of PLA microcapsules using different spinning solvent ((a) acetone, (b) dichloromethane, (c) tetrahydrofuran, (d) trichloromethane).



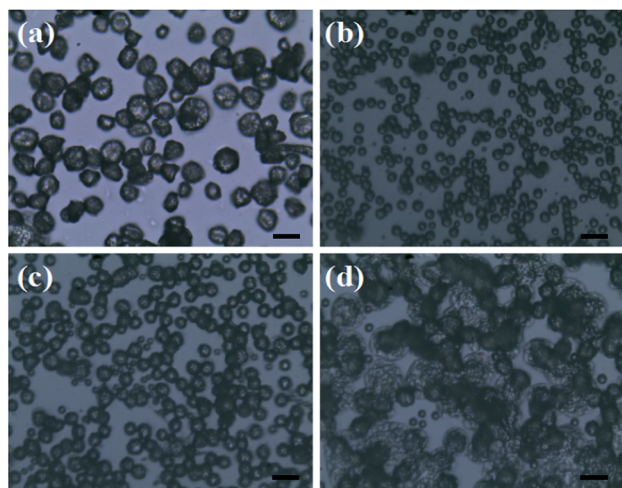


Fig. 3 Optical images of PLA microcapsules under different flow rate ((a) 0.1 mL h^{-1} , (b) 0.3 mL h^{-1} , (c) 0.5 mL h^{-1} , (d) 0.7 mL h^{-1}) scale bar = $20 \mu\text{m}$.

flow was generated. As can be seen in optical micrographs shown in Fig. 3b, 0.3 mL h^{-1} can be the optimal flow rate to fabricate uniform and monodisperse PLA spheres. By contrast, at flow rates of either higher or lower than 0.3 mL h^{-1} it would be difficult to control the monodispersity and structure, as well as the size distribution (Fig. 3a, c and d).

It is chiefly caused by adjusting the evaporating rate of TCM during the ES process when exposed to different flow rates.

Effect of PLA concentration

Polymer concentration is a recognized important factor in controlling the morphology of as-spun products. As shown in Fig. 4, polymer structures varied from monodispersed spheres

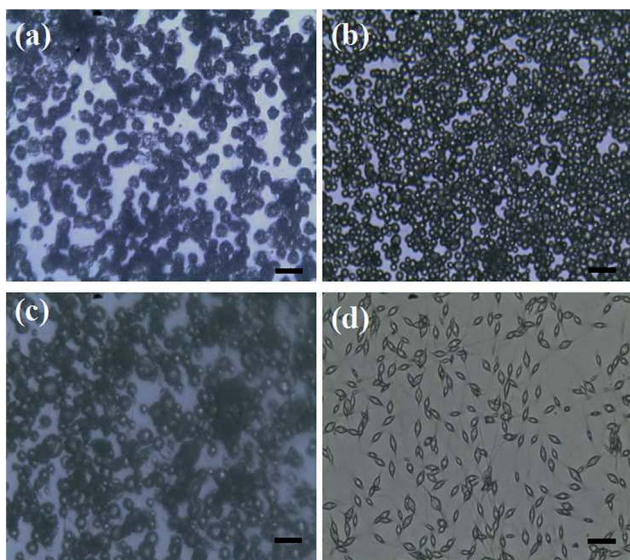


Fig. 4 Optical images of PLA microcapsules using different PLA concentration ((a) 1%, (b) 3%, (c) 5%, (d) 7%). Scale bar = $20 \mu\text{m}$.

to beaded fibers, which was supposed to be relevant to the polymer concentration used in ES process. In other words, when polymer concentration is low (1 wt%), microspheres are more likely to be non-uniform (Fig. 4a). When polymer concentration reaches 5%, agglomerated spherical hollow particles are produced (Fig. 4c). Furthermore, as the polymer concentration became higher (Fig. 4, 7 wt%), beaded nanofibers tend to be generated, which can be due to the effect of electrical conductivity and the surface tension of the spinning solutions with different polymer concentrations.^{29–31} Thus, too high a polymer concentration resulted in the formation of agglomeration. In general, as shown in Fig. 4a, PLA concentration of 3 wt% is most suitable to produce uniform and microspheres with smooth surface and less defect structures.

Effect of drug concentration

At a given flow rate (0.3 mL h^{-1}) and polymer concentration (3 wt%), Fig. 5 displays the Cur@PLA microcapsules with different drug contents of 0%, 5%, 10% and 15 wt% with all other spinning parameters fixed, along with the particle size distribution graphs. As seen from the SEM images, the particle distribution is slightly influenced after adding different amounts of drug molecular to the spinning solution. The mean particle diameters for $C_0\text{PLA}$, $C_5\text{PLA}$, $C_{10}\text{PLA}$, $C_{15}\text{PLA}$ were $3.8 \pm 0.3 \mu\text{m}$, $4.0 \pm 0.7 \mu\text{m}$, $4.2 \pm 0.6 \mu\text{m}$ and $4.4 \pm 0.8 \mu\text{m}$, respectively. The slight increase of spherical size is evident of the successful loading of drug molecules, which could be further interpreted as reduction of the conductivity with adding more Cur drugs into the spinning solution, resulting in increases of the diameters of PLA microparticles. In other words, there was no obvious drug crystals observed on the surface of the microparticles, revealing the consummate drug entrapment process. Moreover, Fig. 6 shows the optical images of $C_0\text{PLA}$, $C_5\text{PLA}$, $C_{10}\text{PLA}$ and $C_{15}\text{PLA}$. It is clearly observed that the yellow Cur molecular was encapsulated inside the core of the PLA-based

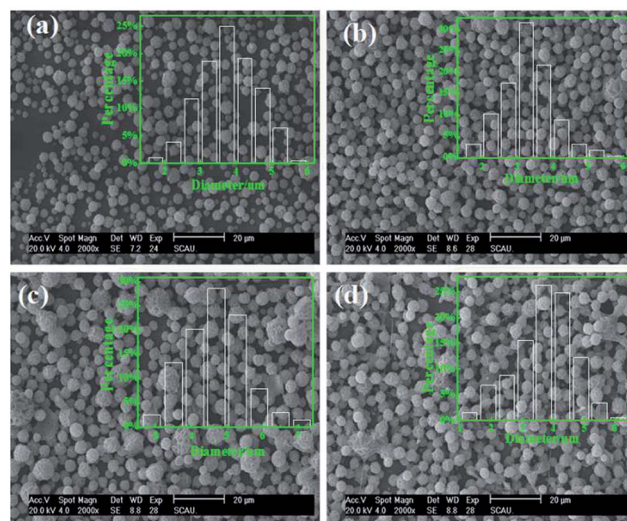


Fig. 5 SEM images of PLA microcapsules using different drug concentration ((a) 0%, (b) 5%, (c) 10%, (d) 15%).



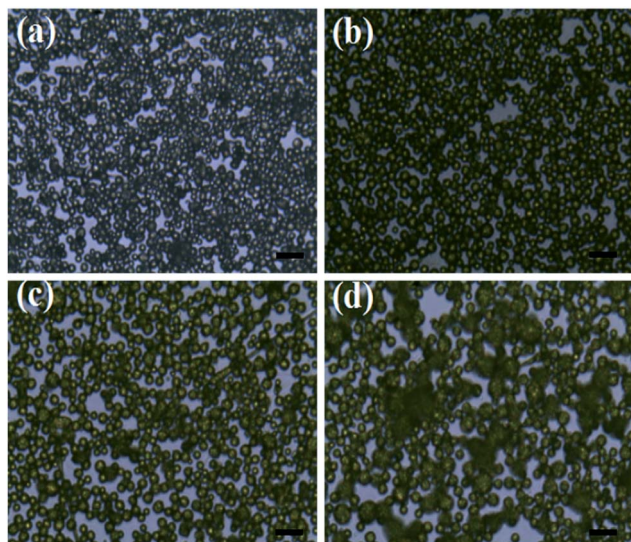


Fig. 6 Optical images of PLA microcapsules using different drug concentration ((a) 0%, (b) 5%, (c) 10%, (d) 15%) scale bar = 20 μm.

carrier vehicles. In addition, bright green fluorescent signals were observed through fluorescence microscope (Fig. S2†), which further indicated the presence of Cur in the core of the microcapsules. Meanwhile, the LC (%) and EE (%) for different samples produced through the ES process are listed in Fig. S3.† As can be seen, the drug entrapment efficiencies of the microcapsules at optimal parameters (*i.e.* flow rate = 0.3 mL h⁻¹, PLA concentration = 3% and TCM as the solvent) are found to be in the range of 95–97%, which can be considered to be no drug loss during the fabrication process. This shows the unique advantage of the ES process compared to the conventional methods.^{35–38} Taking into consideration of toxic side effect and bioavailability of the targeted drugs, C₁₀PLA is chosen as the best candidate and 10% of drug content was supposed to be reasonable in this work.

FTIR

The results of FTIR analysis of C₀PLA, C₁₀PLA, pure Cur and physical mixture of Cur and C₀PLA (mass ratio of Cur to C₀PLA: 1/10) are shown in Fig. 7a. In order to verify the successful fabrication procedure, different peaks of each sample were examined. As for C₀PLA, peaks at 1766 cm⁻¹ and 1092 cm⁻¹ can be attributed to the stretching vibration of C=O bonds and C–O bonds, respectively. In addition, asymmetrical vibrations of –CH₃ bonds at 1455 cm⁻¹ and 1390 cm⁻¹ are also observed. Similarly, C₁₀PLA kept the characteristic peaks at featured wave numbers, which indicated that the polymer remained stable during ES process. The spectrum of pure Cur exhibited the characteristic peaks at 1618 cm⁻¹, 1516 cm⁻¹, 1270 cm⁻¹ and 879 cm⁻¹, which can be ascribed to the stretching vibration of C=O bonds, C=C bonds, C–O bonds and C–O–C bonds, respectively. Based on the FTIR analysis, no additional peaks were found, which was the same as the FTIR spectra of physical mixture of drug and polymer, proving that no chemical reaction occurred among the raw materials during the ES process. The

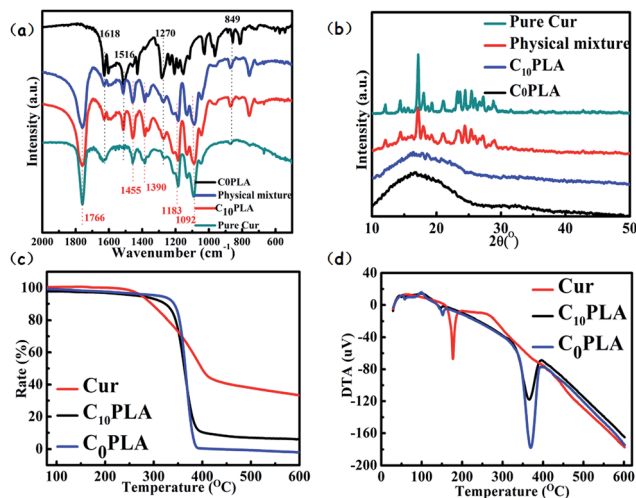


Fig. 7 (a) FTIR (b) XRD (c) TGA (d) DTA curves of Cur, C₁₀PLA and C₀PLA, respectively.

main peaks of Cur were also seen in C₁₀PLA, which could be concluded as the successful immobilization of cargoes inside the PLA-based vehicles.^{3,4}

XRD

X-ray diffraction experiments were performed to assess the crystallinity of each samples, *i.e.*, C₀PLA, pure Cur and C₁₀PLA. To investigate the crystalline nature during the ES process thoroughly, physical mixture of Cur and C₀PLA (mass ratio of Cur to C₀PLA: 1/10) was also tested. All the XRD results were displayed in Fig. 7b. Typically, C₀PLA did not show obvious peaks in the wide-angle range from 10° to 50° corresponding to any crystal phase, implying the amorphous nature of PLA microcapsules fabricated by ES techniques. Besides, it was seen that pure Cur showed main peaks at 2θ = 17.5°, and other characteristic diffraction peaks ranging from 20° to 30° were also detected with strong intensity, indicating that pure Cur drugs existed in crystalline form. Similar peaks corresponding to the Cur crystal phase are also observed in the XRD pattern of the physical mixture. However, C₁₀PLA showed no diffraction peaks, illuminating the transform of Cur crystals to amorphous phase during the ES process. This could probably be caused by the fact that most drug molecules were trapped into the core of the microcapsules, which is consistent with the SEM observation mentioned above.

Thermal properties

The thermal features of C₀PLA and C₁₀PLA were analyzed based on TG and DTA curves. As observed in Fig. 7c, Cur began to lose weight at about 250 °C due to the thermal decomposition, gradually resulting in 40% of its weigh under 600 °C in N₂ atmosphere. Apart from pure Cur, both C₀PLA and C₁₀PLA underwent similar decomposition path except for the starting temperature of losing weight. Typically, both of them started losing weight at about 350 °C, which was supposed to be higher than that of pure Cur. It was notable that the gap of weight-loss



rates between them was estimated to stay at 9.96%, which can be considered as the loading capacity of Cur (10% theoretically), as well as another evidence to prove the high entrapment efficiency of ES fabrication. The DTA curves gave detailed description of the decomposition occurred during the temperature-rise period as shown in Fig. 7d. As described above, endothermic peaks were observed for pure Cur at 200 °C due to the organic decomposition under N₂. However, endothermic peaks for C₀PLA and C₁₀PLA were found at 350 °C, which was consistent with the TG analysis. In particular, no weight loss for C₁₀PLA below 200 °C were observed, which can be concluded as the improved thermostability of drugs provided by the outer shell materials.

In vitro release

The *in vitro* drug release properties of various Cur-loaded samples were analyzed by using PBS (0.2 M, pH = 7.4) containing 0.5 wt% SDS. Fig. 8 show the release profiles of microcapsules with different drug loading capacities. It can be seen that all of them exhibited similar Cur sustained release behaviors during the whole period, including an initial burst release phase and an apparently slow release stage. In detail, the cumulative release rates of the microcapsules reached 51.23 ± 2.3%, 58.5 ± 3.5% and 67.6 ± 2.2% after 24 h for C₅PLA, C₁₀PLA and C₁₅PLA, respectively. All of them displayed initial burst effect during the first 24 h, followed by a sustained release for 200 h. This phenomenon in the first stages can be attributed to fast release of Cur absorbed on the outer surface or near the interior of the microcapsules.^{3,5} These external drugs dissolved in SDS solution quickly at the early stage of the release process. Subsequently, the cumulative rates of Cur became gradually steady as time passing. It can be explained by the fact that Cur molecules dissolved and migrated to the surroundings through permeation and effusion from the voids of microcapsules. In other words, diffusion path of Cur was increased and differential concentration was decreased between the nanosystems and surroundings.^{5,6} Meanwhile, the cumulative release amounts

of Cur increased slowly after 24 h and the drug release would be fostered by the gradual degradation of PLA at this stage. Finally, it can be seen that after 200 h, the cumulative release amounts of C₅PLA, C₁₀PLA and C₁₅PLA are 94.22 ± 1.1%, 95.37 ± 1.5% and 96.77 ± 1.2%, respectively. This suggests that by using ES method, high loading capacity and long-acting drug delivery nanosystems can be prepared.

In order to further investigate the drug release mechanism of Cur from the PLA-based microcapsules fabricated by ES process, four pharmacokinetic models (zero-order, first-order, Higuchi and Ritger–Peppas) were explored to fit to the drug release profiles.

Zero-order model:

$$Q = Kt \quad (5)$$

First order model:

$$Q = 1 - \exp(-Kt) \quad (6)$$

Higuchi model:

$$Q = Kt^{1/2} \quad (7)$$

Ritger–Peppas model:

$$Q = Kt^n \quad (8)$$

where K , Q , n and t are the kinetics constant, cumulative release ratio, diffusion exponent and release time, respectively. Accordingly, the diffusion exponent (n) is used to indicate various release mechanism, or to examine whether the release mechanisms followed the Fickian diffusion or not.⁹

From the fitted results, as shown in Table 2, zero-order model does not suitably describe the release mechanism of the microcapsules, with correlation coefficient (R^2) of 0.81368, 0.84413 and 0.74724 for C₅PLA, C₁₀PLA and C₁₅PLA, respectively. However, the release results showed preference towards Ritger–Peppas model with a correction coefficient R^2 higher than 0.95 for each sample, compared to the first-order model ($R^2 = 0.90$ – 0.92) and Higuchi model ($R^2 = 0.91$ – 0.93). In addition, diffusion exponent (n) was found to be in the range from 0.43 to 0.85, which indicate that release of Cur-loaded microcapsules followed a non Fickian diffusion.⁵ This also reveals the fact that the Cur release from the microcapsules is controlled by both diffusion and matrix degradation, when the diffusion exponent was larger than 0.45.⁶ The degradation is probably a result of the hydrolysis of PLA on the surface.⁹

Antioxidant tests

The antioxidant properties of the Cur-loaded capsules were tested using the DPPH radicals, which was recognized as typical free radicals and widely used to evaluate the antioxidative effect due to its sensitivity to free-radical scavenger.^{39,40} Accordingly, a range of gradient concentrations of samples (80 mg mL⁻¹, 120 mg mL⁻¹, 160 mg mL⁻¹, 200 mg mL⁻¹, 240 mg mL⁻¹) were incubated with DPPH solution. After reaction for 30 min, the purple solution faded due to the clearance of DPPH radicals and

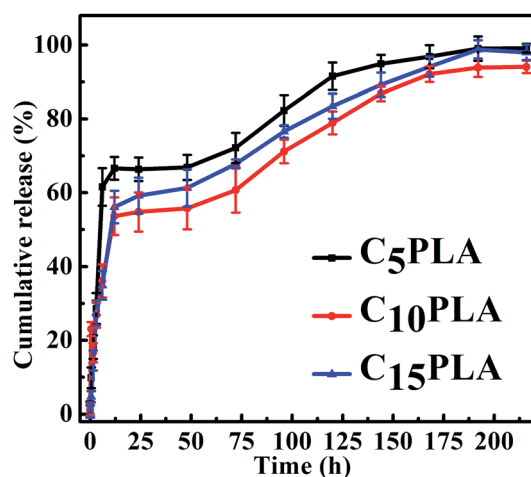


Fig. 8 Drug release profiles of C₅PLA, C₁₀PLA and C₁₅PLA in 5% SDS phosphate solution (pH = 7.4).



Table 2 Fitting parameters of Cur release behaviors to different kinetic models

Sample	Zero-order model	First order model	Higuchi model	Ritger-Peppas model
C ₅ PLA	$Q = 0.27877 + 0.00495t, R^2 = 0.81368$	$\ln(1 - Q) = -0.25790 - 0.01422t, R^2 = 0.92129$	$Q = 0.20088 + 0.06592t^{1/2}, R^2 = 0.93591$	$\ln Q = -2.13143 + 0.53694 \ln t, R^2 = 0.94307$
C ₁₀ PLA	$Q = 0.32571 + 0.00455t, R^2 = 0.84413$	$\ln(1 - Q) = -0.31940 - 0.01318t, R^2 = 0.93313$	$Q = 0.20291 + 0.06094t^{1/2}, R^2 = 0.94817$	$\ln Q = -1.67791 + 0.44071 \ln t, R^2 = 0.95361$
C ₁₅ PLA	$Q = 0.34484 + 0.00483t, R^2 = 0.74724$	$\ln(1 - Q) = -0.34072 - 0.01706t, R^2 = 0.90436$	$Q = 0.13812 + 0.06462t^{1/2}, R^2 = 0.92339$	$\ln Q = -1.74589 + 0.48167 \ln t, R^2 = 0.93726$

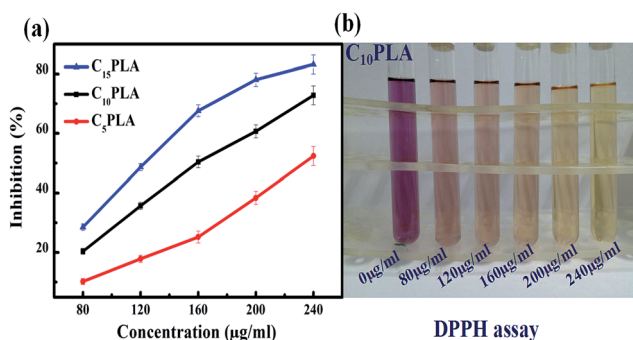


Fig. 9 (a) Antioxidant test results of C₅PLA, C₁₀PLA and C₁₅PLA towards DPPH free radicals. (b) DPPH series towards the sample of C₁₀PLA.

the inhibition rates were calculated by using the UV-vis spectroscopy. As shown in Fig. 9, the half maximal inhibitory concentrations (IC₅₀) of C₅PLA, C₁₀PLA and C₁₅PLA were 236.66 µg mL⁻¹, 163.11 µg mL⁻¹ and 122.26 µg mL⁻¹, respectively. This result may be associated with the content inside the microcapsules and notably Cur remained bioactive after ES fabrication, which is an important consideration when producing novel drug delivery systems.

Antibacterial tests

The antibacterial effects of C₁₀PLA suspension towards *E. coli* and *S. aureus* were assessed by using disc diffusion method, combined with plate counting method. The C₀PLA suspension showed no inhibition zone, declaring that pure PLA does not possess any antibacterial properties. In comparison, Fig. 10c and d revealed the inhibition zone of C₁₀PLA against *E. coli* and *S. aureus*, and the diameters of inhibition zones were measured at 2.90 ± 0.04 cm and 3.24 ± 0.04 cm, indicating the distinct antibacterial effect of C₁₀PLA on *E. coli* and *S. aureus*. Furthermore, in order to investigate the antibiotic ability of C₁₀PLA quantitatively, plate-counting method was also applied. As seen in Table 3, the MICs for C₁₀PLA against *E. coli* and *S. aureus* were found to be 1.25 mg mL⁻¹ and 0.625 mg mL⁻¹, revealing the high-efficient antibacterial activity of microcapsules fabricated through ES process even at such low concentration.

In vitro cell test

The drug delivery material, C₀PLA, was tested for hemolysis experiments and cell apoptosis with Hoechst 33 342 staining

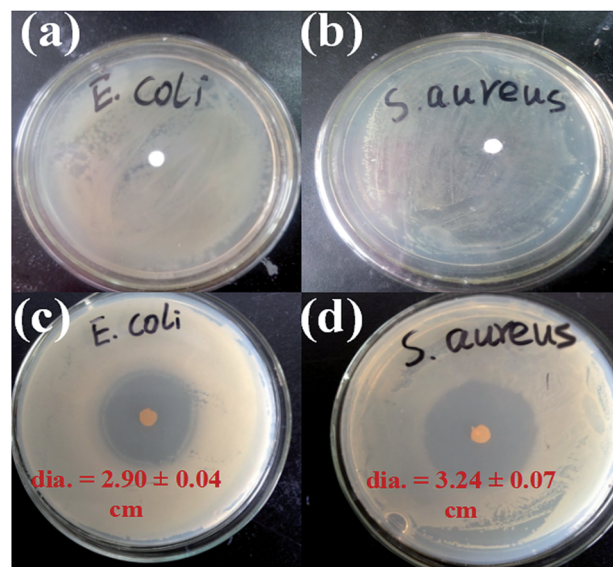


Fig. 10 Bactericidal activity of C₀PLA towards (a) *E. coli* and (b) *S. aureus* and antibacterial activities of C₁₀PLA showing inhibition zones towards (c) *E. coli* and (d) *S. aureus*.

experiment to determine the cytotoxicity and biocompatibility. In addition, cell adhesion and proliferation assays with HDF were also conducted to evaluate the cytocompatibility after culturing for 5 days.

Hemolytic assay assessment

To evaluate the biocompatibility of PLA-based microcapsules, a wide range of concentrations of C₀PLA (15.625–500 µg mL⁻¹) in PBS suspension were incubated with blood samples from SD rats. As shown in Fig. 11a, the hemolytic rate results revealed no significant hemolysis of red blood cells (RBCs) with concentrations of samples reaching up to 500 µg mL⁻¹. Typically, only almost 13% hemolytic activity was detected even at a high concentration of 500 µg mL⁻¹, indicating that the PLA capsule possessed good biocompatibility and low cytotoxicity.

Cell viabilities assessment

We used cell counting kit-8 (CCK-8) to further assess the cytotoxicity of PLA-based microcapsule with HEK 293T cells (2500 cells per well) or PC12 cells (2500 cells per well) incubated with different concentration of samples for 48 h, as



Table 3 Minimal inhibitory concentration towards *E. Coli* and *S. aureus* for C₁₀PLA ("–" there is no growth for bacteria, "+" there is growth for bacteria)

Sample (mg mL ⁻¹)	Bacteria	20	10	5	2.5	1.25	0.625	0.312	0.156
C ₁₀ PLA	<i>E. coli</i>	–	–	–	–	–	–	–	+
	<i>S. aureus</i>	–	–	–	–	–	–	–	+

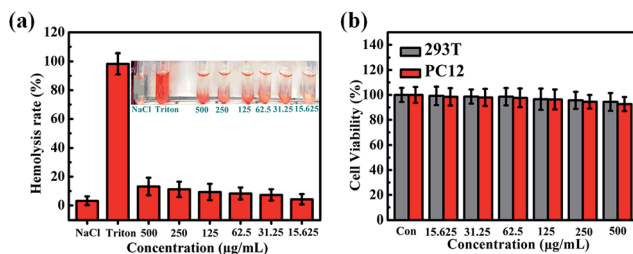


Fig. 11 Hemolysis assays for (a) C₀PLA at concentrations from 15.625 µg mL⁻¹ to 500 µg mL⁻¹, (b) cell viabilities assessments of PLA microcapsules (or C₀PLA).

shown in Fig. 11b. It is obvious that C₀PLA showed no significant effect on the cell viability for both HEK 293T cells and PC12 cells, implying the low cytotoxicity of the microcapsules. Specifically, it was noted that the cell viability was measured as over 80% even at a high concentration of 500 µg mL⁻¹. The immunofluorescence microscopy experiment of cell apoptosis in 293T cells and PC12 cells is also displayed in Fig. S4.† As can be seen, few deformed cell nuclei was detected, indicating that the microcapsules do not cause apoptosis at high concentration.^{41,42}

Cell adhesion and proliferation

In order to further investigate the biocompatibility of PLA-based microcapsules, the morphology of HDF cells incubated with different samples were also explored in depth. The SEM graphs of proliferated HDF cells on C₀PLA and C₁₀PLA at different times were displayed in Fig. 12a–f. As can be seen, the cells tended to grow and adhere to the gaps among microcapsules on the first day, thin cellular layers were obviously observed after proliferation for 3 days, indicating the excellent biocompatibility of both PLA-based samples. In particular, it is worth noting that cell morphology for C₀PLA and C₁₀PLA shows no difference on the fifth day, which meant that drug molecules had no negative effect on the cell adhesion and proliferation.

Fig. 12g indicated the numbers of HDF cells on different samples at given times. Notably, HDF cells grew more rapidly on the C₀PLA microcapsules and C₁₀PLA microcapsules than the control group, which can be simply attributed to the better biocompatibility of PLA. In particular, higher surface area provided by the micro-sized capsules was probably another important factor leading to the rapid cell adhesion and proliferation. Consistently, there was no significant difference between C₀PLA and C₁₀PLA on cell

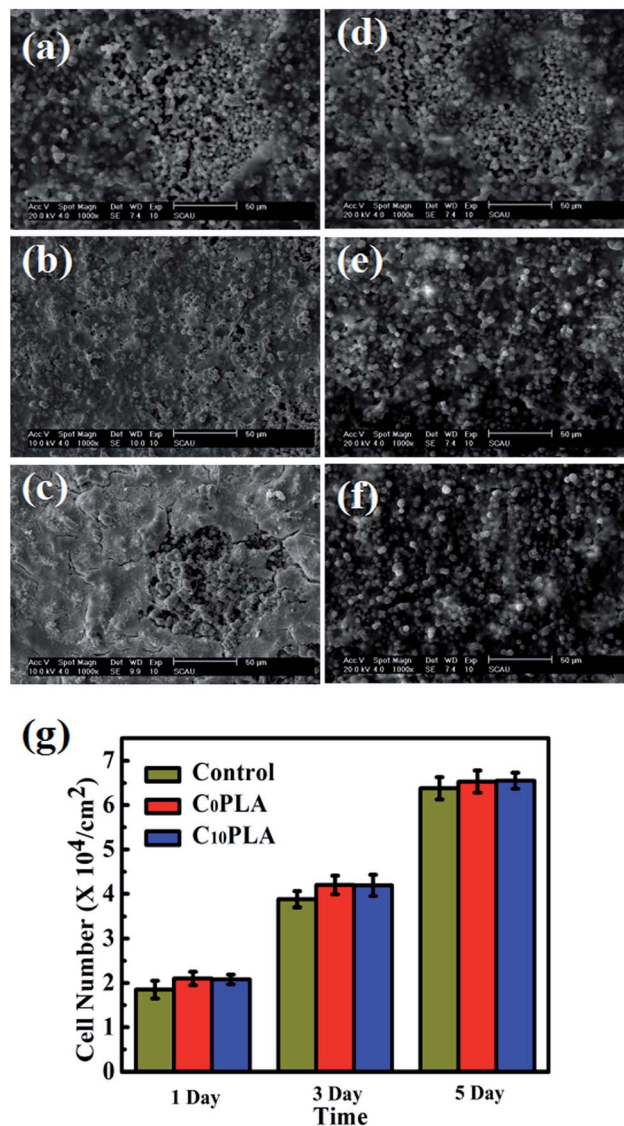


Fig. 12 (a)–(f) SEM images of cells growth on the surface of C₀PLA and C₁₀PLA for 1, 3, 5, 7 days, respectively; (g) proliferation of the HDF cells cultured on C₀PLA and C₁₀PLA and control group.

number, indicating that both of them possessed outstanding biocompatibility.⁴³

Conclusions

In summary, we successfully fabricated novel PLA-based microcapsules entrapped with Cur drugs through an electro-spray process. The morphology and particle sizes could be



optimized by adjusting spinning solvent, flow rates, polymer and drug concentrations. Cur could be encapsulated inside the PLA microcapsules with entrapment efficiency higher than 95%. A sustained release of Cur from the microcapsules can be reached for up to 200 h. Moreover, several bioactivities such as anti-microbial and antioxygenation were confirmed to be outstanding. Meanwhile, with low cytotoxicity and brilliant biocompatibility evaluated by CCK-8, hemolysis and cell adhesion tests, we believed that these novel electrospun PLA-based drug delivery systems have further potential in clinical medicine.

Acknowledgements

The work was supported by the Science and Technology Foundation of Guangdong Province (2016A050502044, 2015B020237009), Natural Science Foundation of Guangdong Province in China (2014A030310035; 2016A030313602), and Aid Program for Science and Technology Innovative Research Team in Higher Educational Institutions of Hunan Province.

Notes and references

- X. J. Wang, D. H. Chen, L. Cao, Y. C. Li, B. J. Boyd and R. A. Caruso, *ACS Appl. Mater. Interfaces*, 2013, **5**, 10926–10932.
- F. Liu, J. N. Wang, P. L. Huang, Q. Zhang, J. T. Deng, Q. Y. Cao, J. L. Jia, J. H. Cheng, Y. P. Fang, D. Y. B. Deng and W. Y. Zhou, *J. Mater. Chem. B*, 2015, **3**, 2206–2214.
- P. L. Huang, B. Z. Zeng, Z. X. Mai, J. T. Deng, Y. P. Fang, W. H. Huang, H. W. Zhang, J. Y. Yuan, Y. Wei and W. Y. Zhou, *J. Mater. Chem. B*, 2016, **4**, 46–56.
- R. H. He, X. F. Hu, H. C. Tan, J. Feng, C. Steffi, K. Wang and W. Wang, *J. Mater. Chem. B*, 2015, **3**, 2137–2146.
- M. J. Chen, J. Y. Liu, Y. J. Liu, C. Guo, Z. H. Yang and H. Wu, *RSC Adv.*, 2015, **5**, 14222–14530.
- M. J. Chen, Y. Hu, J. Zhou, Y. R. Xie, H. Wu, T. Yuan and Z. H. Yang, *RSC Adv.*, 2016, **6**, 13032–13039.
- J. Y. Liu, C. H. Liu, Y. J. Liu, M. J. Chen, Y. Hu and Z. H. Yang, *Colloids Surf., B*, 2013, **109**, 103–108.
- S. D. Nath, S. Son, A. Sadiasa, Y. K. Min and B. T. Lee, *Int. J. Pharm.*, 2013, **443**, 87–94.
- T. He, J. N. Wang, P. L. Huang, B. Z. Zeng, H. H. Li, Q. Y. Cao, S. Y. Zhang, Z. Luo, D. Y. B. Deng, H. W. Zhang and W. Y. Zhou, *Colloids Surf., B*, 2015, **130**, 278–286.
- S. H. Chen, Y. Chang, K. R. Lee and J. Y. Lai, *J. Membr. Sci.*, 2014, **450**, 224–234.
- R. Zheng, H. C. Duan, J. X. Xue, Y. Liu, B. Feng, S. F. Zhao, Y. Q. Zhu, Y. Liu, A. J. He, W. J. Zhang, W. Liu, Y. L. Cao and G. D. Zhou, *Biomaterials*, 2014, **35**, 152–164.
- L. Y. Cheng, X. M. Sun, C. M. Hu, R. Jin, B. S. Sun, Y. M. Shi, L. Zhang, W. G. Cui and Y. G. Zhang, *Acta Biomater.*, 2013, **9**, 9461–9473.
- A. Mahor, S. K. Prajapati, A. Verma, R. Gupta, A. K. Iyer and P. Kesharwani, *J. Colloid Interface Sci.*, 2016, **483**, 132–138.
- C. D. Liu, Z. X. Zhang, X. Liu, X. P. Ni and J. Li, *RSC Adv.*, 2013, **3**, 25041–25049.
- H. L. Che, M. Huo, L. Peng, T. Fang, N. Liu, L. Feng, Y. Wei and J. Y. Yuan, *Angew. Chem.*, 2015, **127**, 9062–9066.
- J. F. Huang, J. Zhong, G. P. Chen, Z. T. Lin, Y. Q. Deng, Y. L. Liu, P. Y. Cao, B. W. Wang, Y. T. Wei, T. F. Wu, J. Yuan and G. B. Jiang, *ACS Nano*, 2016, **10**, 6464–6473.
- Y. Hu, X. Y. Gu, Y. Yang, J. Huang, M. Hu, W. K. Chen, Z. Tong and C. Y. Wang, *ACS Appl. Mater. Interfaces*, 2014, **6**, 17166–17175.
- A. Servant, V. Leon, D. Jasim, L. Methven, P. Limousin, E. V. F. Pacheco, M. Prato and K. Kostarelos, *Adv. Healthcare Mater.*, 2014, **3**, 1334–1343.
- H. W. Kim, J. C. Knowles and H. E. Kim, *Biomaterials*, 2004, **25**, 1279–1287.
- D. Q. Chen and J. F. Sun, *Polym. Chem.*, 2015, **6**, 998–1004.
- A. Baeza, M. Manzano, M. Colilla and M. V. Regí, *Biomater. Sci.*, 2016, **4**, 803–813.
- T. Pengpong, P. Sangvanich, K. Sirilertmukul and N. Muangsin, *Eur. J. Pharm. Biopharm.*, 2014, **86**, 487–497.
- M. Y. Baia and S. Z. Liu, *Colloids Surf., B*, 2014, **117**, 346–353.
- J. M. Yang, L. S. Zha, D. G. Yu and J. Y. Liu, *Colloids Surf., B*, 2013, **102**, 737–743.
- A. A. Samad, Y. Bakkour, C. Fanny, F. E. Omar, J. Coudane and B. Nottelet, *Polym. Chem.*, 2015, **6**, 5093–5102.
- D. Chopra, L. Ray, A. Dwivedi, S. K. Tiwari, J. Singh, K. P. Singh, H. N. Kushwaha, S. Jahan, A. Pandey, S. K. Gupta, R. K. Chaturvedi, A. B. Pant, R. S. Ray and K. C. Gupta, *Biomaterials*, 2016, **84**, 25–41.
- S. Y. Zhai, Y. H. Ma, Y. Y. Chen, D. Li, J. Cao, Y. J. Liu, M. T. Cai, X. X. Xie, Y. W. Chen and X. L. Luo, *Polym. Chem.*, 2014, **5**, 1285–1297.
- X. J. Zhao, S. Chen, Z. F. Lin and C. Du, *Carbohydr. Polym.*, 2016, **148**, 98–106.
- J. M. Yang, J. H. Yang, S. C. Tsou, C. H. Ding, C. C. Hsu, K. C. Yang, C. C. Yang, K. S. Chen, S. W. Chen and J. S. Wang, *Mater. Sci. Eng., C*, 2016, **66**, 170–177.
- L. G. G. Mascaraque, G. Sanchez and A. L. Rubio, *Carbohydr. Polym.*, 2016, **150**, 121–130.
- L. H. Cao, J. Luo, K. H. Tu, L. Q. Wang and H. L. Jiang, *Colloids Surf., B*, 2014, **115**, 212–218.
- K. Ito, A. Saito, T. Fujie, K. Nishiwaki, H. Miyazaki, M. Kinoshita, D. Saitoh, S. Ohtsubo and S. Takeoka, *Acta Biomater.*, 2015, **24**, 87–95.
- X. H. Wu, Y. Ma, G. Q. Zhang, Y. L. Chu, J. Du, Y. Zhang, Z. Li, Y. R. Duan, Z. Y. Fan and J. Huang, *Adv. Funct. Mater.*, 2015, **25**, 2138–2146.
- W. L. Shao, J. X. He, Q. M. Han, F. Sang, Q. Wang, L. Chen, S. Z. Cui and B. Ding, *Mater. Sci. Eng., C*, 2016, **67**, 599–610.
- H. Sawalha, K. Schroën and R. Boom, *Chem. Eng. J.*, 2011, **169**, 1–10.
- H. Z. He, Y. Hong, Z. B. Gu, G. D. Liu, L. Cheng and Z. F. Li, *Carbohydr. Polym.*, 2016, **147**, 243–250.
- X. T. Wang, Z. H. Li, Y. K. Yang, X. H. Gong, Y. G. Liao and X. L. Xie, *Langmuir*, 2015, **31**, 5456–5463.
- D. W. Sun, H. Zhang, X. Z. Tang and J. L. Yang, *Polymer*, 2016, **91**, 33–40.
- M. S. Su and J. L. Silva, *Food Chem.*, 2006, **97**, 447–451.



- 40 X. Kang, C. Zhao, L. S. Yan, R. G. Qi, X. B. Jing and Z. H. Wang, *Colloids Surf., B*, 2016, **145**, 812–819.
- 41 R. G. Qi, H. H. Xiao, S. H. Wu, Y. X. Li, Y. Zhang and X. B. Jing, *J. Mater. Chem. B*, 2015, **3**, 176–179.
- 42 H. Q. Song, R. Wang, H. H. Xiao, H. D. Cai, W. J. Zhang, Z. G. Xie, Y. B. Huang, X. B. Jing and T. J. Liu, *Eur. J. Pharm. Biopharm.*, 2013, **83**, 63–75.
- 43 X. Sun, L. Y. Cheng, W. K. Zhu, C. M. Hu, R. Jin, B. S. Sun, Y. M. Shi, Y. G. Zhang and W. G. Cui, *Colloids Surf., B*, 2014, **115**, 61–70.

