RSC Advances

PAPER

Cite this: RSC Adv., 2017, 7, 27243

Received 7th March 2017 Accepted 18th May 2017 DOI: 10.1039/c7ra02758c

rsc.li/rsc-advances

Highly efficient synthesis of unsymmetrical 1,3 diynes from organoalane reagents and alkynyl bromides mediated by a nickel catalyst†

Song Mo, Xue-Be[i](http://orcid.org/0000-0002-8463-4910) Shao, Gang Zhang and Qing-Han Li

Highly efficient and simple cross-coupling reactions of alkynylbromides with organoalane reagents for the synthesis of unsymmetrical 1,3-diynes derivatives using Ni(OAc)₂ (2–5 mol%)/(o-furyl)₃P (4–10 mol%) as a catalyst are reported. Excellent yields (up to 94%) were obtained for a wide range of substrates at rt or 60 °C for 2–3 h in Et₂O or toluene.

Diynes have been widely used in organic synthesis due to their significance as building blocks.¹ Amongst them, conjugated 1,3diynes have significant applications in the preparation of natural products,² pharmaceuticals,³ π -conjugated acetylenic polymers,⁴ advanced materials⁵ and heterocyclic compounds,⁶ as well as in electronic and optical materials,⁷ and elsewhere.⁸ In addition, 1,3-diynes are common structural motifs found in pharmaceutically and biologically active compounds, which are known to have antifungal,⁹ antibacterial,¹⁰ anti-inflammatory,¹¹ anti-HIV, 12 and anticancer properties.¹³ Thus, their synthesis and applications has attracted a great amount of attention over the past few decades. Since the discovery of the Glaser homocoupling reaction of terminal alkynes in 1869 ,¹⁴ a great amount of attention has been given to the development of new and efficient procedures for the synthesis of conjugated 1,3 diyne derivatives. Recently, a variety of processes have been developed for the preparation of conjugated 1,3-diynes, typical synthetic protocols including the cross-coupling reactions of 1 haloalkynes with terminal alkynes,¹⁵ cross-coupling reactions of terminal alkynes,¹⁶ cross-coupling reaction of terminal alkynes with 1,2-dihaloalkenes, 17 cross-coupling of alkynylsilanes with haloalkynes,¹⁸ conjugated diynones decarbonylation reaction,¹⁹ cross-coupling reactions of aryl iodides and alkynyl acid,²⁰ oxidative coupling of alkynyl Grignard reagents,²¹ oxidative decarboxylative homo-coupling of alkynyl acids,²² oxidative homo-coupling reactions of terminal alkynes with the oxidants,²³ homo-coupling of alkynyltellurides,²⁴ haloalkynes,²⁵ and alkynylsilanes.²⁶ PAPER
 **EXERCT AND ACTION CONTROL CONTROL CONTROVIDENT SUPPLIES OF UNISYMMENTICAL 1.5-

Creative ratios, 2017, 27245

Signal Creative ratios are considered by a nickel catalysty

Signal Creative ratios and the set of the**

Despite that, these efforts have provided alternative methods for the synthesis of conjugated 1,3-dyines. However, these reactions still suffer from excess oxidants, excess bases,

YAL SOCIETY
CHEMISTRY

In recent times, metal-catalyzed cross-coupling reactions of electrophiles with alkynylmetallic reagents have provided an alternative route for the preparation of alkyne compounds.²⁸ Previous studies show that organoalanes reagents are a highly efficient nucleophiles for cross-coupling reactions with aromatic halides²⁹ or benzylic halides,³⁰ and investigations have demonstrated that nickel is a good catalytic metal.³¹ To continue our efforts in developing coupling reactions using reactive organometallic reagents,^{29,30,32} we report a novel nick $el(n)$ -catalyzed cross-coupling of alkynylhalides with alkynylaluminum reagents at ambient temperature or 60 \degree C in a short time with good isolated yield for the synthesis of unsymmetrical conjugated 1,3-diynes (Scheme 1).

In order to optimize the reaction conditions, effects of phosphine ligand, nickel source, reaction time, solvent, the amount of organoalanes reagents, and the molar ratio of metal to ligand on the cross-coupling reaction were investigated using alkynylaluminum and alkynyl bromide as a model system. The reaction of diethyl(2-phenylethynyl)aluminum $(C_6H_5C\equiv CAIEt_2)$ (1a) with 1-(2-bromo-ethynyl)-4-methylbenzene $(p$ -MeC₆H₅-C \equiv CBr) (2a) using a nickel catalyst of phosphine (eqn (1)) was first examined. Under the optimized conditions of 2 mol% $Ni(OAc)_2$ and 4 mol% P(o-furyl)₃ conducting in diethyl ether at room temperature over 2 h, producing 3aa in 87% isolated yield (ESI: Tables S1 and S2†).

Scheme 1 Metal-catalyzed coupling reactions of 1-haloalkynes derivatives with organoalanes nucleophiles.

College of Chemistry and Environmental Protection Engineering, Southwest University for Nationalities, Chengdu 610041, P. R. China. E-mail: lqhchem@163.com; Fax: +86- 28-85524382

[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/c7ra02758c

Generalities of the catalytic system were examined with various 1-bromoalkynes derivatives (eqn (2)), and the results were summarized in Table 1. As shown in Table 1, various 1 bromoalkynes derivatives bearing electron-withdrawing or electron-donating substituents were smoothly coupled to give the corresponding unsymmetrical conjugated 1,3-diynes 3 in 64–94% isolated yield (Table 1, entries 1–14). The crosscoupling reaction with 1-bromoalkynes bearing an electronwithdrawing group, such as fluoro, chloro, bromo and trifluoromethyl groups, or electron-donating group, such as methyl, ethyl, propyl and tertbutyl groups, produced the conjugated products 1,3-diyne in good to excellent yield (Table 1, entries 1–12). In particular, the cross-coupling of 1-bromo-4- (2-bromoethynyl) benzene (2g) with diethyl-(2-phenylethynyl) aluminum (1a) could also be selectively obtained the unsymmetrical conjugated 1,3-diynes 3ag in 74% isolated yield (Table 1, entry 7). However, the cross-coupling reaction of diethyl-(2 phenylethynyl)aluminum (1a) with 1-(2-bromoethynyl)-4 methoxybenzene (4-MeOC₆H₅C \equiv CBr) (2f) only gave products of unsymmetrical conjugated 1,3-diyne 3af in 64% isolated yield (Table 1, entry 6) potentially due to the 1-(2-bromoethynyl)-4- **EXALUATION ACCESS ARTICLE ARTICLE CONSULTATION ACCESS ARTICLE CONSULTATION ACCESS ARTICLE CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION CONSULTATION**

Table 1 Nickel-catalyzed cross-coupling reactions of $C_6H_5C\equiv CAEE_2$ (1a) with various 1-bromoalkynes $(2)^{a}$

Entry	1-Bromoalkynes 2	Product 3	Yield b (%)
$\mathbf{1}$	4 -Me-Ph $(2a)$	3aa	87
$\overline{2}$	4 -Et-Ph $(2b)$	3ab	85
3	4-Propyl-Ph $(2c)$	3ac	86
$\overline{4}$	4 -'Bu-Ph $(2d)$	3ad	84
5	$3-Me-Ph(2e)$	3ae	81
6	$4-MeO-Ph(2f)$	3af	64
7	$4-Ph(2g)$	3ag	74
8	4 -Cl-Ph $(2h)$	3ah	82
9	$4-F-Ph(2i)$	3ai	90
10	$3-F-Ph(2i)$	3aj	93
11	$2-F-Ph(2k)$	3ak	86
12	4 -CF ₃ -Ph $(2I)$	3al	92
13	2-Pyridinyl (2m)	3am	94
14	2-Thienyl $(2n)$	3an	83
15	4 -CN-Ph $(2o)$	3ao	$< 5\%$
16	4 -CH ₃ OOC-Ph $(2p)$	3ap	$\mathbf{0}$

^a 1a/2 = 1.0/0.5 mmol; 2 mL Et₂O; room temperature; 2 h. ^b Isolated yield.

methoxybenzene was being decreased by methoxyl group. More importantly, 2-(2-bromoethynyl)pyridine and 2-(2-bromoethynyl)thiophene reagents were also effective substrates for this reaction, giving the corresponding coupling products of unsymmetrical conjugated 1,3-diynes 3am and 3an in 83% and 94% isolated yield, respectively (Table 1, entries 13 and 14). Unfortunately, cross-coupling reaction of 4-(2-bromoethynyl) benzonitrile (2o) $(p\text{-CNC}_6H_5C\equiv CBr)$ and diethyl-(2phenylethynyl)-aluminum (1a) $(C_6H_5C\equiv CAIEt_2)$ only gave products of unsymmetrical conjugated 1,3-diyne 3ao in 5% yield (Table 1, entries 15). While, coupling of methyl 4-(2-bromoethynyl)benzoate (2p) (p -CH₃OOCC₆H₅C \equiv CBr) could not produce the target compound (Table 1, entries 16).

We subsequently investigated cross-coupling reactions of substituted aryl alkynylaluminum reagents with various 1-bromoalkynes. However, the cross-coupling reaction of diethyl(2- (4-methoxy phenyl)-ethynyl)aluminum (4-MeOC \equiv CAlEt₂) (1b) with 1-(2-bromo-ethynyl)-benzene (2o) employing the catalyst of 2 mol% $Ni(OAc)_2$ and 4 mol% $P(o$ -furyl)₃ yielded coupling products of unsymmetrical conjugated 1,3-diyne 3bo with only a 62% isolated yield. Therefore, the reaction conditions were retuned, and the best performed catalyst was found to be 5 mol% $Ni(OAc)_2$ and 10 mol% P(o-furyl)₃ while the reaction was conducting in toluene at 60 °C for 3 h (ESI Table S3†), furnishing 3bo with 75% isolated yield.

The reaction scope was further explored on substrates of $R'C \equiv CAIEt_2 (R' = 4-MeOPh (1b)$ or 4-FPh $(1c)$, 4-MePh $(1d)$) and various 1-bromoalkynes using 5 mol% $Ni(OAc)_{2}$ and 10 mol% $P(o$ -furyl)₃ conducting in toluene at 60 °C for 3 h (eqn (3)), and results are summarized in Table 2. Satisfactory application scope was demonstrated by this section of experiment. The cross-coupling reaction can be applied to $C(sp)-C(sp)$ bond formations, affording the coupling products unsymmetrical conjugated 1,3-diynes in 51–80% isolated yield (Table 2, 1–15). Cross-coupling reactions of substituted aryl alkynylaluminum reagents 1b or 1c or 1d with 1-bromoalkyne reagents containing different functional groups such as alkyl, halides (2a, 2g, 2h, 2i, 2j, 2k, 2l) could potentially couple, giving the corresponding unsymmetrical conjugated 1,3-diynes moderate to good isolated yields (Table 2, entries 1–4, 7, 8, 10–14). Furthermore, 2-(2 bromoethynyl)thiophene (2n) reacted successfully with diethyl(2-(4-methoxyphenyl)ethynyl)aluminum (4- MeOPhC \equiv C–AlEt₂) (1b) to provide unsymmetrical conjugated 1,3-diyne 3bn with 74% isolated yield (Table 2, entry 5). The 2-(4 p-tolylbuta-1,3-diynyl)pyridine (3dm) resulting from 2-(2 bromo-ethynyl)pyridine (2m) and diethyl(2-p-tolyl-ethynyl) aluminum (4-MePhC \equiv CAlEt₂) (1d) was obtained with 72% isolated yield (Table 2, entry 15). While, the cross-coupling reaction of diethyl(2-(4-fluorophenyl)-ethynyl) aluminium (4-FPhC \equiv CAlEt₂) (1c) with 1-bromo-hept-1-yne (2p) gave the corresponding unsymmetrical conjugated 1,3-diyne 3cp with a 51% isolated yield (Table 2, entry 9). Unfortunately, crosscoupling of 1-(2-bromoethynyl)-4-methylbenzene (2a) $(p\text{-CH}_3)$ - $C_6H_5C\equiv CBr$ and diethyl(2-(4-methylbenzoate)ethynyl) aluminium (4-MeOOCPhC \equiv CAlEt₂) (1e) could not produce the coupling product (Table 2, entries 16).

Table 2 Cross-coupling reactions of substituted aryl organoalanes reagents ArC \equiv CAlEt $_2$ (1) with various 1-bromoalkynes (2)^a

	\equiv AIEt ₂ $+$ -Br \sim 2	5 mol% $Ni(OAc)_2$ 10 mol% $P(o$ -furyl) ₃ toluene, 60°C, 3h	R' = R (3)	
Entry	1 R'	2 R	Product	Yield b (%)
1	4 -MeOPh $(1b)$	4 -MePh $(2a)$	3ba	62

^a $1/2 = 1.0/0.5$ mmol; 2 mL toluene; 60 °C; 3 h. ^b Isolated yield.

The cross-coupling reactions of aliphatic alkynylaluminum reagents with various 1-bromoalkynes was also investigated under the same catalytic system. The unsymmetrical conjugated 1,3-diyne 3fo was obtained with 52% isolated yield when 1-(2-bromoethynyl)-benzene (2o) coupled with the $TMSC \equiv CAIEt_2$ (1f) nucleophile. However, the cross-coupling reaction of 1-(2-bromoethynyl)-benzene (2o) coupled with the $TMSC = CAIEt₂$ (1f) favoured a formation of the unsymmetrical conjugated 1,3-diyne 3fo with 80% isolated yield in DME at room temperature for 3 h (ESI Table S4†). So, the reaction scope was further explored on substrates of $TMSC \equiv CAIEt_2$ (1f) and various 1-bromoalkynes using 5 mol% $Ni(OAc)_2$ and 10 mol% $P(o$ -furyl)₃ conducting in DME at room temperature for 3 h (eqn (4)), and results are summarized in Table 3. As shown in Table 3, various 1-bromoalkynes derivatives bearing electronwithdrawing or electron-donating substituents were smoothly coupled to give the corresponding unsymmetrical conjugated 1,3-diynes 3 in 67–80% isolated yield (Table 3, entries 1–6). The cross-coupling reaction with 1-bromoalkynes bearing an electron-withdrawing group, such as fluoro and trifluoromethyl groups, or electron-donating group, such as methoxyl and tertbutyl groups, produced the conjugated products 1,3-diyne in good isolated yield (Table 3, entries 2–5). In particular, the cross-coupling of 2-(2-bromoethynyl)naphthalene (2q) with TMSC \equiv CAlEt₂ (1f) could also be obtained the conjugated 1,3diynes 3fq in 74% isolated yield (Table 3, entry 6).

The effect of the alkylhalides and dummy ligand of organoalane reagents on the reaction were investigated, and the results were summarized in Table 4. As shown in Table 4, cross-

Table 3 Cross-coupling reactions of organoalanes reagents TMSC \equiv CAlEt₂ (1f) with various 1-bromoalkynes (2)^a

$TMS \rightarrow AIEt_2$ 1f	5 mol% $Ni(OAc)2$ 10 mol% $P(o$ -furyl) ₃ DME,rt, 3h	\rightarrow TMS	$\equiv -\equiv -R(4)$
R \equiv Br			

^a $1/2 = 1.0/0.5$ mmol; 2 mL DME; room temperature; 3 h. ^b Isolated yield.

coupling reaction of dimethyl-(2-phenylethynyl)-aluminum $(1g)$ $(C_6H_5C\equiv CAlMe_2)$ with p-CH₃C₆H₅C=CBr (2a) gave product of unsymmetrical conjugated 1,3-diyne 3aa in 68% isolated yield (Table 4, entry 1). The coupling reaction of propargyl iodide (2q) with diethyl-(2-phenylethynyl)-aluminum (1a) $(C_6H_5C\equiv CAIEt_2)$ gave product of unsymmetrical conjugated 1,3-diyne 3aa in a 75% isolated yield (Table 4, entry 2). While, coupling of propargyl chloride (2r) gave coupling product 3aa in only 48% isolated yield (Table 4, entry 3). Gratifyingly, 70% isolated yield of product 3aa was obtained using tri(2 phenylethynyl)-aluminum (1h) as nucleophiles (Table 4, entry 4). The results showed that the different alkylhalides and dummy ligand of organoalane reagents are compatible with this procedure. Paper

Take 2 Converticular precision of substituted any organisation $\frac{1}{2}$ May 2017. This article is licensed under the common access Article is licensed under the article is licensed under the common and the common

While a precise understanding of the reaction mechanism awaits further study, a plausible catalytic pathway is proposed in Scheme 2. The first step is the oxidative addition of 1-bromoalkyne (1) to Ni(0) phosphine complex (4) (which in turn from $\text{Ni}(\text{OAc})_2$ and $\text{R}'\text{C}\equiv\text{CAlet}_2$ (2) reagents) that forms the

Table 4 The effect of the alkylhalides and dummy ligand of organoalane reagents on the reaction⁶

^a $1/2 = 1.0/0.5$ mmol; 2 mL Et₂O; room temperature; 3 h. ^b Isolated yield.

Scheme 2 The proposed catalytic cycle for the formation of unsymmetrical conjugated 1,3-diyne 3.

organonickel $\left(\text{II} \right)$ bromide intermediate (5). Transmetalation of R^\prime C \equiv CAlEt $_2$ (2) with 5 gives alkyne-alkynenickelium($\scriptstyle\rm II$) intermediate (6) and Et₂AlBr. Finally, complex 6 under goes reductive elimination to afford the desired cross-coupling product 3 and regenerate the active Ni(0) species for the next catalytic cycle.

Conclusions

In conclusion, a novel nickel-catalyzed cross-coupling reaction of 1-bromoalkynes with alkynylaluminum reagents is reported. A variety of 1-bromoalkynes with aryl, heterocycloyl, or alkyl group can be efficiently coupled to give the corresponding coupling products unsymmetrical conjugated 1,3-diynes in good yield and with good functional group compatibility. Moreover, the ready availability of the starting materials, the mild reaction conditions, and the simplicity of the operations involved are additional features making the methodology could therefore serve as a complementary option in the field of conjugated 1,3-diyne synthesis. A possible catalytic cycle was proposed to explain the formation of the products conjugated 1,3-diynes. Further application of these conjugated 1,3-diynes in organic synthesis are underway.

Acknowledgements

The authors are grateful to the Sichuan Provincial Department of science and technology support program (No. 2015NZ0033), and the Postgraduate Degree Construction Project of Southwest University for Nationalities (No. CX2016SZ063) for financial support.

Notes and references

1 (a) Y. Matsuda, S. Naoe, S. Oishi, N. Fujii and H. Ohno, Chem.–Eur. J., 2015, 21, 1463; (b) Y. G. Zhu and Y. A. Shi, Org. Biomol. Chem., 2013, 11, 7451.

- 2 (a) A. L. K. Shi Shun and R. R. Tykwinski, Angew. Chem., Int. Ed., 2006, 45, 1034; (b) H. Yun, T. C. Chou, H. J. Dong, Y. Tian, Y. M. Li and S. J. Danishefsky, J. Org. Chem., 2005, 70, 10375.
- 3 (a) M. L. Lerch, M. K. Harper and D. J. Faulkner, J. Nat. Prod., 2003, 66, 667; (b) K. C. Nicolau, P. G. Bulger and D. Sarlah, Angew. Chem., Int. Ed., 2005, 44, 4490.
- 4 (a) B. J. Morgan, X. Xie, P. W. Phuan and M. C. Kozlowski, J. Org. Chem., 2007, 72, 6171; (b) F. Babudri, D. Colangiuli, P. A. Di Lorenzo, G. M. Farinola, O. H. Omar and F. Naso, Chem. Commun., 2003, 130.
- 5 (a) Acetylene Chemistry: Chemistry, Biology and Material Science, ed. F. Diederich, P. J. Stang and R. R. Tykwinski, Wiley-VCH GmbH KGaA, Weinheim, Germany, 2005; (b) M. Gholami and R. R. Tykwinski, Chem. Rev., 2006, 106, 4997.
- 6 (a) V. Lavallo, G. D. Frey, B. Donnadieu, M. Soleilhavoup and G. Bertrand, Angew. Chem., Int. Ed., 2008, 47, 5224; (b) T. Matsuda, S. Kadowaki and M. Murakami, Chem. Commun., 2007, 2627.
- 7 (a) J. A. Marsden and M. M. Haley, J. Org. Chem., 2005, 70, 10213; (b) S. Eisler, A. D. Slepkov, E. Elliott, T. Luu, R. McDonald, F. A. Hegmann and R. R. Tykwinski, J. Am. Chem. Soc., 2005, 127, 2666.
- 8 (a) M. B. Nielsen and F. Diederich, Chem. Rev., 2005, 105, 1837; (b) J. Z. Liu, J. W. Y. Lam and B. Z. Tang, Chem. Rev., 2009, 109, 5799.
- 9 A. Stütz, Angew. Chem., Int. Ed., 1987, 26, 320.
- 10 D. Lechner, M. Stavri, M. Oluwatuyi, R. Perda-Miranda and S. Gibbons, Phytochemistry, 2004, 65, 331.
- 11 (a) R. Schmidt, R. Thorwirth, T. Szuppa, A. Stolle, B. Ondruschka and H. Hopf, Chem.–Eur. J., 2011, 17, 8129; (b) R. A. Xiao, R. Y. Yao and M. Z. Cai, Eur. J. Org. Chem., 2012, 22, 4178; (c) A. Narani, R. K. Marella, P. Ramudu, K. S. R. Rao and D. R. Burri, RSC Adv., 2014, 4, 3774.
- 12 M. L. Lerch, M. K. Harper and D. J. Faulkner, J. Nat. Prod., 2003, 66, 667.
- 13 (a) S. Morandi, F. Pellati, S. Benvenuti and F. Prati, Tetrahedron, 2008, 64, 6324; (b) S. F. Mayer, A. Steinreiber, R. V. A. Orru and K. Faber, J. Org. Chem., 2002, 67, 9115.
- 14 C. Glaser, Ber. Dtsch. Chem. Ges., 1869, 2, 422.
- 15 (a) H. L. Li, L. L. Wang, M. Yang and Y. X. Qi, Catal. Commun., 2012, 17, 179; (b) D. Saha, T. Chatterjee, M. Mukherjee and B. C. Ranu, J. Org. Chem., 2012, 77, 9379; (c) W. Shi, Y. D. Luo, X. C. Luo, L. Chao, H. Zhang, J. Wang and A. W. Lei, J. Am. Chem. Soc., 2008, 130, 14713; (d) Z. W. Chen, H. F. Jiang, A. Z. Wang and S. R. Yang, J. Org. Chem., 2010, 75, 6700; (e) S. Ahammed, D. Kundu and B. C. Ranu, J. Org. Chem., 2014, 79, 7391.
- 16 (a) J. R. Suárez, D. Collado-Sanz, D. J. Cárdenas and J. L. Chiara, J. Org. Chem., 2015, 80, 1098; (b) M. Nasr-Esfahani, I. Mohammadpoor-Baltork, A. R. Khosropour, M. Moghadam, V. Mirkhani, S. Tangestaninejad, V. Agabekov and H. A. Rudbaria, RSC Adv., 2014, 4, 14291; (c) Z. Y. Ma, X. Y. Wang, S. Y. Wei, H. L. Yang, F. W. Zhang, P. Wang, M. Xie and J. T. Ma, Catal. Commun., 2013, 39, 24.
- 17 M. L. N. Rao, P. Dasgupta, B. S. Ramakrishna and V. N. Murty, Tetrahedron Lett., 2014, 55, 3529.
- 18 Y. Nishihara, K. Ikegashira, A. Mori and T. Hiyama, Tetrahedron Lett., 1998, 39, 4075.
- 19 A. Dermenci, R. E. Whittaker and G. Dong, Org. Lett., 2013, 15, 2242.
- 20 (a) Y. Kim, A. Park, K. Park and S. Lee, Tetrahedron Lett., 2011, 52, 1766; (b) J. Moon, M. Jeong, H. Nam, J. Ju, J. H. Moon, H. M. Jung and S. Lee, Org. Lett., 2008, 10, 945; (c) J. Moon, M. Jang and S. Lee, J. Org. Chem., 2009, 74, 1403; (d) K. Park, G. Bae, J. Moon, J. Choe, K. H. Song and S. Lee, J. Org. Chem., 2010, 75, 6244; (e) K. Pak, G. Bae, A. Park, Y. Kim, J. Choe, K. H. Song and S. Lee, Tetrahedron Lett., 2011, 52, 576.
- 21 (a) Y. G. Zhu, T. Xiong, W. Y. Han and Y. A. Shi, Org. Lett., 2014, 16, 6144; (b) G. Cahiez, C. Chaboche, F. Mahuteau-Betzer and M. Ahr, Org. Lett., 2005, 7, 1943; (c) A. Krasovskiy, A. Tishkov, V. del Amo, H. Mayr and P. Knochel, Angew. Chem., Int. Ed., 2006, 45, 5010.
- 22 D. X. Liu, F. L. Li, H. X. Li, J. Gao and J. P. Lang, Tetrahedron, 2014, 70, 2416.
- 23 For selected recent reports on the oxidative homocoupling reactions of terminal alkynes with the oxidants, see: (a) H. L. Li, M. Yang, X. Zhang, L. Yan, J. Li and Y. X. Qi, New J. Chem., 2013, 37, 1343; (b) F. Alonso and M. Yus, ACS Catal., 2012, 2, 1441; (c) K. Yin, C. J. Li, J. Li and X. S. Jia, Green Chem., 2011, 13, 591; (d) R. A. Xiao, R. Y. Yao and M. Z. Cai, Eur. J. Org. Chem., 2012, 4178; (e) S. L. Zhang, X. Y. Liu and T. Q. Wang, Adv. Synth. Catal., 2011, 353, 1463; (f) W. Y. Yin, C. He, M. Chen, H. Zhang and A. W. Lei, Org. Lett., 2009, 11, 709; (g) X. Meng, C. B. Li, B. C. Han, T. S. Wang and B. H. Chen, Tetrahedron, 2010, 66, 4029; (h) Y. G. Zhu and Y. A. Shi, Org. Biomol. Chem., 2013, 11, 7451; (i) J. R. Suárez, D. Collado-Sanz, D. J. Cárdenas and J. L. Chiara, *J. Org. Chem.*, 2015, 80, 1098; (j) K. Fang, M. H. Xie, Z. N. Zhang, P. Ning and G. Y. Shu, Tetrahedron Lett., 2013, 54, 3819. Open Access Article. Published on 23 May 2017. Downloaded on 5/1/2025 8:58:03 PM. This article is licensed under a [Creative Commons Attribution-NonCommercial 3.0 Unported Licence.](http://creativecommons.org/licenses/by-nc/3.0/) **[View Article Online](https://doi.org/10.1039/c7ra02758c)**
	- 24 F. V. Singh, M. F. Z. J. Amaral and H. A. Stefani, Tetrahedron Lett., 2009, 50, 2636.
- 25 (a) K. Osowska, T. Lis and S. Szafert, Eur. J. Org. Chem., 2008, 27, 4598; (b) Z. Chen, H. Jiang, A. Wang and S. Yang, J. Org. Chem., 2010, 75, 6700.
- 26 Y. Nishihara, K. Ikegashira, K. Hirabayashi, J. Ando, A. Mori and T. Hiyama, J. Org. Chem., 2000, 65, 1780.
- 27 (*a*) B. Shrestha, S. Thapa, S. K. Gurung, R. A. S. Pike and R. Giri, *J. Org. Chem.*, 2016, 81, 787; (b) D. F. Crépin and J. P. A. Harrity, Org. Lett., 2013, 15, 4222; (c) O. V. Larionov and E. J. Corey, Org. Lett., 2010, 12, 300; (d) W. Dávila, W. Torres and J. A. Prieto, Tetrahedron, 2007, 63, 8218; (e) M. Sasaki, K. Tanino and M. Miyashita, Org. Lett., 2001, 3, 1765; (f) M. Sasaki, M. Miyazawa, K. Tanino and M. Miyashita, Tetrahedron Lett., 1999, 40, 9267.
- 28 (a) B. M. Wang, M. Bonin and L. Micouin, Org. Lett., 2004, 6, 3481; (b) R. Chinchilla and C. Nàjera, Chem. Rev., 2007, 107 , 874; (c) E.-I. Negishi and L. Anastasia, Chem. Rev., 2003, 103, 1979; (d) J. Kessabi, R. Beaudegnies, P. M Jung, B. Martin, F. Montel and S. Wendeborn, Synthesis, 2008, 4, 655; (e) H. Tanaka and Y. Shishido, Bioorg. Med. Chem. Lett., 2007, 17, 6079.
- 29 (a) D. B. Biradar and H. M. Gau, Chem. Commun., 2011, 47, 10467; (b) W. T. Shu, S. L. Zhou and H. M. Gau, Synthesis, 2009, 3, 4075; (c) S. L. Ku, X. P. Hui, C. A. Chen, Y. Y. Kuo and H. M. Gau, Chem. Commun., 2007, 3847.
- 30 D. B. Biradar and H. M. Gau, Org. Biomol. Chem., 2012, 10, 4243.
- 31 (a) J. Caeiro, J. P. Sestelo and L. A. S arandeses, Chem.–Eur. J., 2008, 14, 741; (b) I. Koyama, T. Kurahashi and S. Matsubara, J. Am. Chem. Soc., 2009, 131, 1350; (c) Y. Shi, J. Huang, Y. F. Yang, L. Y. Wu, Y. N. Niu, P. F. Huo, X. Y. Liu and Y. M. Liang, Adv. Synth. Catal., 2009, 351, 141; (d) J. Terao, F. Bando and N. Kambe, Chem. Commun., 2009, 7336; (e) S. Sako, T. Kurahashi and S. Matsubara, Chem. Commun., 2011, 47, 6150.
- 32 (a) Q. H. Li, J. Y. Jeng and H. M. Gau, Eur. J. Org. Chem., 2014, 31, 7531; (b) Q. H. Li, J. W. Liao, Y. L. Huang, R. T. Chiang and H. M. Gau, Org. Biomol. Chem., 2014, 12, 7634; (c) Q. H. Li and H. M. Gau, Synlett, 2012, 5, 747; (d) Q. H. Li, Y. Ding and X. J. Yang, Chin. Chem. Lett., 2014, 25, 1296.