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1 Introduction

DNA, famed in biology as the carrier of the genetic code, has attracted enormous attention from physics communities in recent years. A better insight into the physical behavior of these mysterious building blocks would provide more knowledge about their biological function and thus more possibilities for medical manipulation. Moreover, studies on DNA behavior offer hope for future molecular electronics by designing highspeed miniaturized circuits with self-organizing components.¹–⁶ DNA may be handled as a wire, transistor, switch or rectifier, depending on its electronic properties, $7,8$ and therefore, a comprehensive understanding of the electronic structure of these creations is vital. This need is felt particularly strongly while determining physical quantities such as the thermodynamic and transport properties, most of which depend on the electronic structure. However, various intrinsic and extrinsic complexities have rendered the complete achievement of this objective impossible. This contradiction has also persevered in the experimental domain. Natural DNA in solution has highly supported charge transfer rates,⁹⁻¹² but dried single DNA molecules have shown diverse insulating,13,14 semiconducting, $15,16$ and metallic^{12,17,18} behavior.

Further to this background, no comprehensive theory relating to the electronic structure and transport mechanisms

Sequence dependency of the thermodynamic properties of long DNA double-strands

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Using the Green's function technique and tight-binding (TB) Hamiltonian method for π -electrons within a ladder model of long DNA double-strands acting as semiconducting nanowires, the temperature and sequence dependency of the Pauli paramagnetic susceptibility (PMS) and electronic heat capacity (EHC) of some selected configurations are investigated. The results show the explicit reliance of the two quantities on the arrangement of the DNA base-pairs (BPs). The notable result of this study is the appearance of three temperature points for which the EHC (PMS) curves of the ordered systems converge (diverge) at the lowest and highest points, while they diverge (converge) at the middle point. Disordered systems exhibit a deviation from the trend of the ordered ones and this is more significant around the higher temperature points. Over a wide range of temperatures, the responses of the poly(GC) and poly(AT) double-helices limit the behavior of the random DNA chains. Our finding could inspire the design of experiments to assess the order of randomness in BP sequencing and also to better set the TB parameters of the model. PAPER

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through DNA strands has been proposed so far. Nonetheless, tremendous effort is still being made in this area of research.

The thermodynamic properties of DNA and relevant structures have long been under consideration.¹⁹⁻²⁷ For example, the thermodynamic parameters of DNA sequences with dangling ends and the contribution of these terminations to the duplex stability were inspected by Bommarito et al. via enthalpy and entropy changes extracted from UV melting curves of the synthesized oligonucleotides on solid supports.¹⁹ Using finitedifference Poisson–Boltzmann methods, Gallagher and Sharp determined the contribution of electrostatics to the changes of heat capacity of DNA binding reactions involving the ligands DAPI, netropsin and lexitropsin, and the λ repressor binding domain.²⁰ Based on the enthalpy and heat capacity changes, Holbrook et al. investigated the thermodynamics of the selfassembly of a DNA duplex with 14 BPs from complementary strands, using titration and differential scanning calorimetry along with van't Hoff analysis of the UV thermal scans.²¹ Using the same method, Bergqvist et al. probed the local effects of water molecules and ions on the heat capacity changes of a protein–DNA interaction.²² They found that although changes in the local ion binding capacity influenced the enthalpic and entropic contributions to the free energy of the interaction, there was no overall effect of ionic strength on the heat capacity changes. However, remarkable impacts were observed which originated from two particular symmetry-related mutations. An interesting case of the TB approach to be pointed out is the study conducted by Roche and Maciá who considered the genomic sequences of the chromosome 22, λ -bacteriophage and D1s80 genes of humans and pygmy chimpanzees and

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compared these with both periodic and quasiperiodic sequences of nucleotides.²³

The magnetic properties of DNA have also been under consideration and reported in the literature, particularly during the last two decades. Yan et al. developed a near-field magnetictweezer for micromanipulation of single DNA molecules consisting of a small permanent magnet which can move to as close as $10 \mu m$ from the particle and generate forces much larger than those generated by previous instruments.²⁴ By means of molecular dynamics simulations and quantum chemistry methods, Berashevich and Chakraborty investigated the effect of humidity on the transport and magnetic properties of DNA. They concluded that the interaction of water molecules with nucleobases leads to the breaking of some π bonds and the appearance of unbound π electrons which could then participate considerably, up to 103 times more, in the conductivity at room temperature, while at low temperature the efficiency of charge transfer is determined by the spin interaction of two unbound electrons on the intrastrand nucleobases.²⁵ Paper Were compared ones with both corrections are compared on 24 October 2017. Also the set of the standard paper of the standard ones are t

Scanning the literature for experimental and computational studies on the thermodynamic properties of DNA during the last few decades shows that the majority of them are carried out by chemists and biologists, but those by physicists are far fewer. In this study, we try to draw a theoretical picture of the EHC and PMS of the DNA double-strands with respect to the temperature and BP sequencing, using the Green's function technique and the TB Hamiltonian model.

In Section 2 we review a semiconducting ladder model with the corresponding Hamiltonian for long DNA double-strands, using the Green's function approach^{28,29} and related formulas for the physical quantities. Section 3 includes the calculations, results and the related discussions. Finally, Section 4 presents the summary and conclusion of all the results.

2 Model and approach

DNA (deoxyribonucleic acid) macro-molecules are twisted double-helix structures composed of repeated stacks of either adenine–thymine (A–T) or guanine–cytosine (G–C) BPs with an intimate sugar–phosphate (S–P) backbone, being coupled with hydrogen bonds (the Watson-Crick model).³⁰ Fig. 1 shows a schematic view of the DNA structure with the simplications used: firstly to untwist it to achieve a planar shape and secondly not to include the internal details of the constituents with respect to scale, spacing and position.

In general, the theoretical studies on DNA can be classified into two main groups of ab initio calculations³¹⁻³⁷ and the model-based Hamiltonian approach.³⁸⁻⁴⁹ Although the former method operates in a more fundamental way and provides more detail, it is very time-consuming for systems that are not small. The latter technique is generally favored when dealing with large systems wherein optimizing a few parameters can reproduce experimental results and reliable data. In this light, various TB models with proper on-site and hopping parameters have already been introduced with the aim of giving results compatible with experimental or ab initio reports.⁴⁷⁻⁵² These models have mostly evolved around the important finding that

overlapping of π orbitals, particularly those of the stacked BPs, can create a π pathway for the migration of electrons.^{2,10,11,24}

In this work, we proceed by means of a disordered ladder model proposed by Yamada,⁴² constructed by repetition of the S–P sites with interchain hoppings at the S sites which come from the attached BPs. This model is shown in Fig. 2; in fact it is an oversimplified representation of Fig. 1 obtained by ignoring the migration of electrons along successive BPs. As marked in the figure, t_1 and t_2 indicate the hopping amplitudes of a π -electron from a P site to the left and right attached S sites, respectively, while W_{GC} and W_{AT} indicate the same quantity but between the BPs on complementary strands. It is assumed that no hopping exists between the P sites. Additionally, the BPs with permuted nucleotides are equivalent, namely $W_{CG} = W_{GC}$ and $W_{TA} = W_{AT}.$

The second quantized Hamiltonian of this model reads as:⁴⁶

$$
\widehat{\mathcal{H}} = \sum_{i=1}^{N_{\rm c}} \sum_{\alpha=1}^{2N_{\rm b}} \left[\frac{1}{2} \sum_{\mu=1,2} \left(\varepsilon_{i,\mu}{}^{\alpha} \hat{c}_{i,\mu}{}^{\alpha\dagger} \hat{c}_{i,\mu}{}^{\alpha} - t_{i,\mu}{}^{\alpha\alpha+1} \hat{c}_{i,\mu}{}^{\alpha\dagger} \hat{c}_{i,\mu}{}^{\alpha+1} \right. \right. \\ \left. - t_{i,\mu}{}^{\alpha\alpha-1} \hat{c}_{i,\mu}{}^{\alpha\dagger} \hat{c}_{i,\mu}{}^{\alpha-1} \right) - W_{\alpha} \hat{c}_{i,1}{}^{\alpha\dagger} \hat{c}_{i,2}{}^{\alpha} \right] + \text{h.c.,}
$$
 (1)

where N_c is the number of Bravais lattice unit cells, each comprised of two strands labelled by μ , every one of which includes N_b , the number of P sites with a nucleotide in between, and the strands are coupled at the nucleotides (Fig. 2). For an electron at position α on strand μ of unit cell *i*, $\varepsilon_{i,\mu}^{\alpha}$ and $\hat{c}_{i,\mu}^{\alpha}$ $(\hat{c}_{i,\mu}^{\ \alpha})$ indicate the on-site energy and creation (annihilation) operator, respectively. In complete accordance, $t_{i\mu}^{a\alpha \pm 1} \in \{t_1, t_2\}$ and $W_\alpha \in \{W_{\text{GC}}, W_{\text{AT}}\}$ stand for the longitudinal and transverse hoppings between adjacent and opposite neighbors at site α . Evidently, since each unit cell includes two strands with $2N_b$ sites on each one, and assigned to each site is a single electron, there are in total $N_a = 4N_b$ points embedded and thus the Hamiltonian matrix would be of the order $N_a \times N_a$.

The aim of this study is to make a comparison between the EHC and PMS of long DNA semiconducting nanowires with respect to their dependence on the temperature and BP sequencing. We start with three specific ordered chains, namely $poly(GC)$, $poly(AT)$ and $poly(GC-AT)$ double-strands, defined as an infinite arrangement of the GC, AT and alternating GC–AT pairs. We then move towards more realistic states of randomly arranged chains by considering periodic one-dimensional systems composed of big unit cells with many BPs arranged randomly inside. It is obvious from Fig. 2 that the length of such a unit cell with *n* BPs is $a = 2na_0$, where a_0 is the distance between a P site and a proximate S site. From our previous work,⁴⁶ larger unit cells lead to a better resemblance of natural DNA and even medium-sized cells (for example those with a couple of hundred BPs) bring about acceptable outcomes.

It is convenient to continue in a system of units in which most of the physical constants, such as h and k_B , are equal to one. Recalling from the Green's function approach,⁵³⁻⁵⁶ the matrix elements are generally expressed by $G_{\mu\nu}^{\alpha\beta}(i,j;\tau)$ = $-\langle T\hat{c}_{i,\mu}{}^{\alpha}(\tau)\hat{c}_{j,\nu}{}^{\beta\dagger}(0)\rangle$, where $\tau = it$ is the imaginary time, T is the time ordering operator, and $\langle \ldots \rangle$ is a symbol for ensemble

Fig. 1 A planar view of the chemical composition of DNA with four bases (G (light red), C (dark red), A (light blue) and T (dark blue)) connecting the backbones of phosphorylated sugar (brown). P (pink) stands for the phosphate group.

averaging over the ground state of the system. Calculation of the dynamical equations for the operators, $d\hat{c}_{i,\mu}^{\ \alpha}(\tau)/d\tau =$ $\left[\hat{\mathcal{H}}(\tau), \hat{c}_{i,\mu}^{\ \alpha}(\tau)\right]$, results in:

$$
\sum_{\ell} \left[\delta_{i\ell} \left(- \boldsymbol{I} \frac{\partial}{\partial \tau} + \varepsilon_i \right) + t_{i\ell} \right] G(\ell, j; \tau) = \delta(\tau) \delta_{ij} \boldsymbol{I}, \tag{2}
$$

where I is the $N_a \times N_a$ identity matrix and $\delta(\tau)$ stands for the Dirac δ -function. Applying the following imaginary time Fourier transform equations:

$$
\boldsymbol{G}(\ell,j;\tau) = \frac{1}{\beta} \sum_{n} e^{-i\omega_{n}\tau} \boldsymbol{G}(\ell,j;i\omega_{n}), \qquad (3a)
$$

$$
\frac{1}{\beta} \int_0^\beta d\tau e^{i(\omega_m - \omega_n)\tau} = \delta_{mn}, \qquad (3b)
$$

leads to:

$$
\sum_{\ell} [(\mathrm{i}\omega_n \boldsymbol{I} + \varepsilon_i)\delta_{i\ell} + \boldsymbol{t}_{i\ell}] \boldsymbol{G}(\ell, j; \mathrm{i}\omega_n) = \delta_{ij}\boldsymbol{I}, \qquad (4)
$$

where β is the inverse of temperature, $\omega_n = \pi(2n + 1)/\beta$ represents the fermionic Matsubara frequencies and $\{m,n\}$ are both integer numbers. With the transformation of $i\omega_n \to E = \varepsilon + i0^+,$ the analytical continuation of eqn (4) takes the form of:

$$
\sum_{\ell} [(EI + \varepsilon_i)\delta_{i\ell} + t_{i\ell}] G(\ell, j; E) = \delta_{ij}I.
$$
 (5)

In the last step, applying the k -space Fourier transform is helpful:

$$
G(\ell, j; E) = \frac{1}{N_c} \sum_{k \in \text{FBZ}} e^{i k.R_{ij}} G(k; E), \tag{6a}
$$

$$
\delta_{ij} = \frac{1}{N_c} \sum_{\mathbf{k} \in \text{FBZ}} e^{i\mathbf{k} \cdot \mathbf{R}_{ij}}, \tag{6b}
$$

in which R_i is the position of the unit cell *i* and $R_{ij} = R_i - R_j$, while $k = ke_x$ signifies a wave vector that is along the unit vector e_x in the x-direction and runs over the entire first Brillouin zone (FBZ). All of these are concluded by

$$
G(k;E) = (EI - \epsilon_k)^{-1}
$$
 (7)

where ϵ_k denotes the Fourier transform of t_{ij} ,

$$
\epsilon_{k} = \frac{1}{N_{\rm c}} \sum_{i,j=1}^{N_{\rm c}} e^{-{\rm i}k.R_{ij}} t_{ij}.
$$
 (8)

Regarding the relationship between the total DOS and the imaginary part of the Green's function, $D(\varepsilon) = -\text{Im} \text{Tr} \mathbf{G}(E)$ / π ,⁵⁴ by summing over the quantum numbers which label the Hamiltonian, the DOS of the system takes the form of:

Fig. 2 Simplified model of a two-channel periodic DNA with the appropriate parameter sets to attain semiconductivity.

$$
f_{\rm{max}}
$$

 $D(\varepsilon) = -\frac{1}{\pi N_c}$ $\pi N_{\rm a}N_{\rm c}$ $\sum_{\alpha}^{N_{\rm a}}$ $\alpha=1$ \sum $\sum_{k \in \text{FBZ}} \text{Im} G^{\alpha \alpha}(\mathbf{k}; E),$ (9)

in which the diagonal elements, $G^{\alpha\alpha}$ ($k;E$), are adapted from eqn (7).

Now we are sufficiently equipped to extract the desired physical quantities. The electronic heat capacity could be calculated by:⁵⁵⁻⁵⁷

$$
C(T) = \int_{-\infty}^{+\infty} d\varepsilon \varepsilon D(\varepsilon) \left[\frac{\partial f(\varepsilon, T)}{\partial T} \right],
$$
 (10)

where $f(\varepsilon,T)$ denotes the Fermi–Dirac distribution function,

$$
f(\varepsilon, T) = \frac{1}{1 + \exp(\varepsilon/T)}.
$$
 (11)

Likewise, recalling the definition of static susceptibility,⁵⁸

$$
\chi = \frac{1}{\Omega_{\rm c}} \left(\frac{\partial M}{\partial B} \right)_{T, B \to 0}, \tag{12}
$$

where M indicates the dimensionless magnetization, it is feasible to exploit the PM susceptibility as:⁵⁸

$$
\chi(T) = \int_{-\infty}^{+\infty} d\varepsilon D(\varepsilon) \bigg[-\frac{\partial f(\varepsilon, T)}{\partial \varepsilon} \bigg]. \tag{13}
$$

Finally, from the combination of eqn (9) – (13) , the following forms for the EHC and PMS of our systems are derived:

\n**Paper**
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_c} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_0} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n
$$
D(\varepsilon) = -\frac{1}{\pi N_0 N_0} \sum_{\alpha=1}^{N_0} \sum_{k \in \mathbb{N}^2} \text{Im} G^{\alpha\alpha}(k; E),
$$
\n $$

$$
\chi(T) = -\frac{1}{\pi N_a N_c T} \sum_{\alpha=1}^{N_a} \exp\left\{-\frac{\exp(\varepsilon/T)}{[\ln(\varepsilon)]^2} G^{\alpha\alpha}(\mathbf{k}; E)\right\}.
$$
 (15)

3 Calculations and results

By rescaling all energies to that of t_1 for further simplification, we adapt the subsequent TB parameters from the model given by Yamada. The parameters were extracted while investigating the long-range correlation effects on the localization property of the one-electron states in such disordered models via the DOS and Lyapunov exponents of the wave functions:⁴²

$$
\varepsilon_{i,\mu}{}^{\alpha} = 0, \quad t_1 = 1.0, \quad t_2 = \frac{1}{2}t_1 = 0.5, \quad W_{GC} = 1.0, \quad W_{AT} = 2W_{GC} = 2.0.
$$
 (16)

Perhaps these parameters, particularly the fixing of all the onsite energies to zero, seem unsound and different to some other works, $35-38,50-52$ however, this simple model with the above set of parameters has succeeded in reproducing the semiconductivity of DNA double-strands.42,43,46

For the simple case of the poly(GC) duplex, as a practical example, the orthogonal basis kets of the Hilbert space corresponding to four sites within the unit cell (two coupled strands of G–P and C–P) are as follows:

$$
\{|\Phi_1^{\text{G}}\rangle, |\Phi_1^{\text{P}}\rangle, |\Phi_2^{\text{C}}\rangle, |\Phi_2^{\text{P}}\rangle\},\tag{17}
$$

and based on this set, by compact symbolization of $G_{\mu\nu}^{\alpha\beta} \equiv$ $G_{\mu\nu}^{\alpha\beta}(i,j;E)$, the Hamiltonian has the matrix form of:

$$
G(i,j;E) = \begin{pmatrix} G_{11}^{GG} & G_{11}^{GP} & G_{12}^{GG} & G_{12}^{GP} \\ G_{11}^{PG} & G_{11}^{PP} & G_{12}^{PC} & G_{12}^{PP} \\ G_{21}^{CG} & G_{21}^{CP} & G_{22}^{CC} & G_{22}^{CP} \\ G_{21}^{PG} & G_{21}^{PP} & G_{22}^{PC} & G_{22}^{PP} \end{pmatrix}.
$$
 (18)

For the corresponding unit cell of length $a = 2a_0$ with the nearest neighbor vectors $R_{i,i+1} = -R_{i.i-1} = ae_x$, the FBZ interval would be $-\pi/a \leq k \leq \pi/a$, and hence by inserting the parameters of eqn (16) into eqn (8), we are led to:

$$
\epsilon_{k} = \begin{pmatrix} 0 & t_{1} + t_{2}e^{-ika} & W_{GC} & 0 \\ t_{1} + t_{2}e^{ika} & 0 & 0 & 0 \\ W_{GC} & 0 & 0 & t_{1} + t_{2}e^{-ika} \\ 0 & 0 & t_{1} + t_{2}e^{ika} & 0 \end{pmatrix}.
$$
 (19)

Diagonalizing the above matrix through some calculations determines the band structure (BS) of the system as below:

$$
\mathsf{E}(k) = \frac{1}{2} \bigg[\pm W_{\text{GC}} \pm \sqrt{W_{\text{GC}}^2 + 4[t_1^2 + t_2^2 + 2t_1t_2\cos(ka)]} \bigg]. \tag{20}
$$

The energy gap of the system can be extracted directly as the minimum difference between the mid-upper and mid-lower bands,

$$
\Delta_{\rm g} = \sqrt{W_{\rm GC}^2 + 4(t_1 - t_2)^2} - W_{\rm GC}.
$$
 (21)

Obviously these results hold true for poly(AT) DNA as well, however, W_{GC} should be replaced with W_{AT} . Substituting the numerical values of the parameters generates values of $0.414t_1$ and $0.236t_1$ for the energy gaps of the poly(GC) and poly(AT) chains, respectively. The results indicate the typical dependence of the band gap and electrical profile of such systems on their geometry and TB parameters.

For the case of the poly(GC–AT) duplex, which includes eight sites within its unit cell (two coupled strands of G–P–A–P and C–P–T–P), the orthogonal basis kets of the Hilbert space are:

$$
\{|\phi_1^G\rangle,|\phi_1^{P_1}\rangle,|\phi_1^A\rangle,|\phi_1^{P_2}\rangle,|\phi_2^C\rangle,|\phi_2^{P_1}\rangle,|\phi_2^T\rangle,|\phi_2^{P_2}\rangle\},\qquad(22)
$$

and concordantly,

Numerically solving the above matrix and reiterating for all k vectors of the FBZ determines the BS of this case. The acquired results have found $0.30t_1$ to be the value of the band gap. Furthermore, substitution of eqn (23) in eqn (7) and (9) specifies the profile and details of the DOS.

Similarly, larger calculations are carried out to extract the DOS of bigger duplexes. Fig. 3 shows the results for the three mentioned ordered systems. Panel (a) for poly(GC) DNA shows the greatest band gap at the Fermi level, $\varepsilon_F = 0$, and mini-gaps between the subbands on both sides. In contrast, in panel (b), poly(AT) DNA appears to have the smallest band gap at ε_F with widely separated subbands. Principally, van Hove singularities directly relate to the number of atoms in the unit cell and address the regions in which the eigenstates are larger. In both systems, van Hove singularities occur at the edges of the bands and subbands, implying that the eigenstates are denser in these vicinities. The emergence of central gaps and the observed discrepancies for different systems can be explained by symmetry breaking, that is asymmetry of the hopping terms from a typical site to either the left or right neighbor, which resembles the picture of a molecule located between two other

molecules of different types. If the neighbors on both sides of a typical molecule are different, they have different energy levels, leading to the opening of a gap in the energy band. The noticeable point in panel (c) for poly(GC–AT) is a band gap of medium size compared to the two other cases. Moreover, there are higher numbers of allowed bands and van Hove singularities due to its larger unit cell that has a larger number of atoms inside. Therefore, it might be speculated that poly(GC) and poly(AT) DNA chains can set the limits and confine the curves of all other double-strands. This could be supported by more inspection of the plots (a) – (c) of Fig. 4, representing the DOS of the three disordered chains with unit cells including 20, 100 and 200 randomly arranged BPs. Details of the sequencing have not been demonstrated except to consider only one constraint: that every case is composed of equal numbers of G–C and A–T pairs but in a random arrangement. It can be clearly seen that the different shapes of the DOS corresponding to unit cells of different sizes are in agreement with ref. 46 where this result is verified not only for this condition but also for different configurations of BPs within a fixed size unit cell. **PSC** Advances $\begin{pmatrix} |\Phi_1^{\text{S}}\rangle, |\Phi_1^{\text{S}}\rangle, |\Phi_2^{\text{S}}\rangle, |\Phi_2$

Illustrated in panels (a) and (b) of Fig. 5 are the EHC and PMS of the ordered chains with respect to the dimensionless temperature $(\epsilon = k_{\text{B}}T)/t_1$. The behavior of the curves in the limit of very high and very low temperatures can be explained because the thermal effects are dominant in the upper limit and overcome all other mechanisms, while in the lower extreme the thermal energy is not sufficient to excite the electrons to cross the energy gap. There are two mechanisms governing the systems: the temperature gradient as a driving agent and the scattering of electrons from the site potentials as a disturbing factor. For different systems both mechanisms are locally of different strengths and their competition dictates the overall behavior. The confluence points of the curves are quite salient. Concerning the EHC, there are three special temperatures of about $\varepsilon_1 = 0.20t_1$, $\varepsilon_2 = 0.53t_1$ and $\varepsilon_3 = 0.96t_1$. For the points ε_1

Fig. 3 The DOS of three ordered chains of (a) poly(GC), (b) poly(AT) and (c) poly(GC–AT) double-strands, based upon the parameter sets of eqn (16).

Fig. 4 The DOS of three disordered double-strands with (a) 20, (b) 100, and (c) 200 randomly arranged BPs.

Fig. 5 The (a) EHC and (b) PMS of the three considered ordered double-strands.

and ε_3 , all the ordered systems are almost coincident, while for ε_2 they diverge significantly. In addition, the poly(GC) and poly(AT) curves cross, and thus which case is higher alternates, at ε_1 and ε_3 . Every system exhibits an unsteady increasing trend until it reaches its peak, after which the figures steadily decline. The poly(GC) DNA curve reaches its peak at about $0.56t_1$ with the least degree of fluctuation, while the poly(AT) DNA curve has the lowest peak among all the systems, reaching its peak at $0.93t₁$ with the most recognizable fluctuations among all the systems. The existence of another extreme at $0.27t_1$ in the latter case may provide enough evidence to consider a doublecrossover. As expected based on the above interpretation, the curve of the poly(GC–AT) DNA resides between those of the other two, with the exception of a meagre departure at about ε_1 . This is clearer than the peak at $0.7t_1$, in terms of both its position and height.

Similar description emerges with obvious discrepancies for the PMS of the systems. As panel (b) of Fig. 5 shows, the PMS of poly(AT) DNA reaches its peak abruptly at $0.2t_1$ and decreases much slowly after that, without any considerable fluctuation. However, the curve for poly(GC) DNA reaches its maximum at about 0.57 t_1 after a noticeable deviation from the initial trend. Again, the poly(GC–AT) DNA curve stays between the two other curves and has a peak at about $0.26t_1$. In contrast to those of the EHC, the curves of the PMS of the ordered chains meet at ε_2 and show a distinguishable divergence at ε_1 and ε_3 .

According to the results based on this model, perhaps it can be claimed that the physical properties of the poly(GC) and poly(AT) DNA nanowires delineate the boundaries which enclose the graphs of other double-strands. Of course, any extension must be inferred from more work on more systems of other types, including periodic, quasi-periodic, and natural DNA structures.⁵⁹–⁶³ This idea has been assessed further, within

Fig. 6 The (a) EHC and (b) PMS of a few disordered double-strands including different numbers of BPs with equal contributions of both types.

the scope of our results, by comparing the mentioned quantities calculated for such systems with different random DNA chains whose DOS is plotted in Fig. 4. The outcomes, as portrayed in Fig. 6, are moderately distributed around the poly(GC–AT) DNA, giving further evidence to support our first speculation. We observe that these random systems disobey the trend of the ordered chains, i.e. they diverge around the points where the ordered systems converge and vice versa. Furthermore, this

Fig. 7 The (a) EHC and (b) PMS of the three different configurations of disordered double-strands with 200 BPs, labelled by their GC numbers.

discrepancy is more noticeable around the higher coincidence temperatures. For instance, the EHC curves of the random chains deviate from being coincident at ε_1 and ε_3 , with more divergence around ε_3 . They also don't show a severe divergence at ε_2 , as opposed to what was previously observed for the ordered systems. Likewise, the PMS curves of the random chains deviate from divergence at ε_1 and ε_3 and coincidence at ε_2 . Moreover, this deviation from the behavior of the ordered systems is greater around the higher temperature points.

This behavior holds even for disordered systems of the same unit cell size but with different arrangements of BPs. Fig. 7 depicts the calculation for a few configurations of such a unit cell with 200 BPs which have been labelled by their respective number of GC pairs. The first two double-strands both contain 100 GCs but they are arranged in different ways, labelled I and II. Obviously, the mentioned discrepancy is still present but it is not as severe. The dependence of EHC and PMS of the DNA on temperature and sequencing, along with the particular temperatures at which the ordered chains exhibit exclusive behavior, can inspire the design of experiments for the investigation of BP arrangement and the order of randomness in random DNA double-strands. The outcomes of these investigations would be helpful for regulating and quantifying the TB parameters of the model applied here. **BSC Advances**

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4 Summary and conclusion

Using a proper TB Hamiltonian model and the Green's function method, the EHC and PMS of some selected ordered and disordered DNA double strands have been studied. The main finding from our calculations using this model is that the EHC and PMS of poly(GC) and poly(AT) DNA draw the boundaries which enclose these graphs for the other disordered chains. Additionally, for each of the mentioned quantities, there exist particular temperature points at which the EHC curves of the ordered double-strands coincide (diverge) while their PMS curves diverge (coincide). Interestingly, these are the points where disordered systems exhibit the most discrepancy with respect to the ordered systems, and the deviation is more significant for points of higher temperature. Therefore, the deviation from these points can be considered as a criterion for evaluating the degree of randomness in BP sequencing of random DNA chains. Furthermore, they allow the possibility to better quantify the TB parameters of the model.

Conflicts of interest

There are no conflicts to declare.

References

- 1 K. Keren, R. S. Berman, E. Buchstab, U. Sivan and E. Braun, Science, 2003, 302, 1380.
- 2 R. Gutiérrez, S. Mohapatra, H. Cohen, D. Porath and G. Cuniberti, Phys. Rev. B: Condens. Matter Mater. Phys., 2006, 74, 235105.
- 3 V. Bhalla, R. P. Bajpai and L. M. Bharadwaj, Eur. Mol. Biol. Organ. J., 2003, 4, 442.
- 4 M. Mertig, R. Kirsch, W. Pompe and H. Engelhardt, Eur. Phys. *J. D*, 1999, 9, 45.
- 5 D. B. Hall, R. E. Holmlin and J. K. Barton, Nature, 1996, 382, 731.
- 6 E. Meggers, M. E. Michel-Beyerle and B. Giese, J. Am. Chem. Soc., 1998, 120, 12950.
- 7 E. M. Conwell and S. V. Rakhmanova, Proc. Natl. Acad. Sci. U. S. A., 2000, 97, 4556.
- 8 G. Cuniberti, L. Craco, D. Porath and C. Dekker, Phys. Rev. B: Condens. Matter Mater. Phys., 2002, 65, 241314.
- 9 S. O. Kelley and J. K. Barton, Science, 1999, 283, 375.
- 10 C. J. Murphy, M. A. Arkin, Y. Jenkins, N. D. Ghatlia, S. Bossman, N. J. Turro and J. K. Barton, Science, 1993, 262, 1025.
- 11 C. R. Treadway, M. G. Hill and J. K. Barton, Chem. Phys., 2002, 281, 409.
- 12 E. Braun, Y. Eichen, U. Sivan and G. B. Yoseph, Nature, 1998, 391, 775.
- 13 A. J. Storm, J. van Noort, S. de Vries and C. Dekker, Appl. Phys. Lett., 2001, 79, 3881.
- 14 J. P. Lewis, T. E. Cheatham, E. B. Starikov, H. Wang and O. F. Sankey, J. Phys. Chem. B, 2003, 107, 2581.
- 15 D. Porath, A. Bezryadin, S. Vries and C. Dekker, Nature, 2000, 403, 635.
- 16 H. Cohen, C. Nogues, R. Naaman and D. Porath, Proc. Natl. Acad. Sci. U. S. A., 2005, 102, 11589.
- 17 H. W. Fink and C. Schonenberger, Nature, 1999, 398, 407.
- 18 B. Xu, P. Zhang, X. Li and N. Tao, Nano Lett., 2004, 4, 1105.
- 19 S. Bommarito, N. Peyret and J. SantaLucia Jr, Nucleic Acids Res., 2000, 28, 1929.
- 20 K. R. Gallagher and K. R. Sharp, Biophys. J., 1998, 75, 769.
- 21 J. A. Holbrook, M. W. Capp, R. M. Saecker and M. T. Record, Biochemistry, 1999, 38, 8409.
- 22 S. Bergqvist, M. A. Williams, R. O'Brien and J. E. Ladbury, J. Mol. Biol., 2004, 336, 829.
- 23 S. Roche and E. Maciá, Mod. Phys. Lett. B, 2004, 18, 847.
- 24 J. Yan, D. Skoko and J. F. Marko, Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys., 2004, 70, 011905.
- 25 J. Berashevich and T. Chakraborty, J. Phys. Chem. B, 2008, 112, 14083.
- 26 J. Zenk, D. Scalise, K. Wang, P. Dorsey, J. Fern, A. Cruz and R. Schulman, RSC Adv., 2017, 7, 18032.
- 27 C. G. Evans and E. Winfree, Chem. Soc. Rev., 2017, 46, 3808.
- 28 H. Bruus and K. Flensberg, Many-Body Quantum Theory in Condensed Matter Physics: An Introduction, 2nd edn, Oxford University Press, UK, 2004.
- 29 H. Mousavi and J. Khodadadi, Superlattices Microstruct., 2015, 88, 434.
- 30 J. D. Watson and F. H. C. Crick, Nature, 1953, 171, 737.
- 31 F. Bogar, A. Bende and J. Ladik, Phys. Lett. A, 2014, 378, 2157.
- 32 R. Di Felice, A. Calzolari and H. Zhang, Nanotechnology, 2004, 15, 1256.
- 33 Q. Cui and M. Elstner, Phys. Chem. Chem. Phys., 2014, 16, 14368.
- 34 M. Elstner and G. Seifert, Philos. Trans. R. Soc., A, 2014, 372, 20120483.
- 35 H. Mehrez and M. P. Anantram, Phys. Rev. B: Condens. Matter Mater. Phys., 2005, 71, 115405.
- 36 A. Migliore, S. Corni, D. Varsano, M. L. Klein and R. Di Felice, J. Phys. Chem. B, 2009, 113(28), 9402.
- 37 A. Ivanova, P. Shushkov and N. Rösch, J. Phys. Chem. A, 2008, 112(30), 7106.
- 38 L. G. D. Hawke, G. Kalosakas and C. Simserides, Eur. Phys. J. E, 2010, 32, 291.
- 39 C. J. Paez, P. A. Schulz, N. R. Wilson and R. A. Römer, New J. Phys., 2012, 14, 093049.
- 40 F. Palmero, J. F. R. Archilla, D. Hennig and F. R. Romero, New J. Phys., 2004, 6, 13.
- 41 O. R. Davies and J. E. Inglesfield, *Phys. Rev. B: Condens.* Matter Mater. Phys., 2004, 69, 195110.
- 42 H. Yamada, Int. J. Mod. Phys. B, 2004, 18, 1697.
- 43 H. Yamada and K. Iguchi, Adv. Condens. Matter Phys., 2010, 380710.
- 44 A. Kole and K. Radhakrishnan, RSC Adv., 2017, 7, 8474.
- 45 C. Simserides, Chem. Phys., 2014, 440, 31.
- 46 H. Mousavi, J. Khodadadi and M. Grabowski, Solid State Commun., 2015, 222, 42.
- 47 K. Iguchi, Int. J. Mod. Phys. B, 2003, 17, 2565.
- 48 K. Iguchi, Int. J. Mod. Phys. B, 2004, 18, 1845.
- 49 D. Klotsa, R. A. Römer and M. S. Turner, Biophys. J., 2005, 89, 2187.
- 50 T. Kubar, P. B. Woiczikowski, G. Cuniberti and M. Elstner, J. Phys. Chem. B, 2008, 112(26), 7937.
- 51 K. Senthilkumar, F. C. Grozema, C. F. Guerra, F. M. Bickelhaupt, F. D. Lewis, Y. A. Berlin, M. A. Ratner and L. D. Siebbeles, J. Am. Chem. Soc., 2005, 127 42, 14894.
- 52 A. A. Voityuk, J. Jortner, M. Bixon and N. Rösch, *J. Chem.* Phys., 2001, 114 13, 5614.
- 53 E. N. Economou, Green's Functions in Quantum Physics, 3rd edn, Springer-Verlag, Berlin-Heidelberg, 2006.
- 54 G. Grosso and G. P. Parravicini, Solid State Physics, 2nd edn, Academic Press, 2014.
- 55 H. Mousavi, M. Bagheri and J. Khodadadi, Phys. E, 2015, 74, 135.
- 56 H. Mousavi and J. Khodadadi, Appl. Phys. A, 2016, 122, 14.
- 57 C. Kittel, Introduction to SolidState Physics, 8th edn, Wiley, New York, 2004.
- 58 W. Nolthing and A. Ramakanth, Quantum Theory of Magnetism, Springer, New York, 2009.
- 59 K. Lambropoulos, M. Chatzieleftheriou, A. Morphis, K. Kaklamanis, M. Theodorakou and C. Simserides, Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys., 2015, 92, 032725.
- 60 K. Lambropoulos, M. Chatzieleftheriou, A. Morphis, K. Kaklamanis, R. Lopp, M. Theodorakou, M. Tassi and C. Simserides, Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys., 2016, 94, 062403. Paper Wewer Moter and C. Stilet, Publis Trans. R. St., A, 2014, 372, 53 K. Schellargen, F. D. Concerns. C. T. Concerns. Attribution and Access Article is licensed under a Creative Commons Attribution 3.15:53 PM. This artic
	- 61 E. L. Albuquerque, U. L. Fulco, V. N. Freire, E. W. S. Caetano, M. L. Lyra and F. A. B. F. de Moura, Phys. Rep., 2014, 535(4), 139.
	- 62 R. G. Sarmento, G. A. Mendes, E. L. Albuquerque, U. L. Fulco, M. S. Vasconcelos, O. Ujsághy, V. N. Freire and E. W. S. Caetano, Phys. Lett. A, 2012, 376, 2413.
	- 63 C. T. Shih, Y. Y. Cheng, S. A. Wells, C. L. Hsu and R. A. Römer, Comput. Phys. Commun., 2011, 182, 36.