

Cite this: *Chem. Sci.*, 2017, 8, 4511

Received 9th March 2017

Accepted 19th April 2017

DOI: 10.1039/c7sc01093a

rsc.li/chemical-science

Enantioselective γ -borylation of unsaturated amides and stereoretentive Suzuki–Miyaura cross-coupling†

Gia L. Hoang and James M. Takacs *

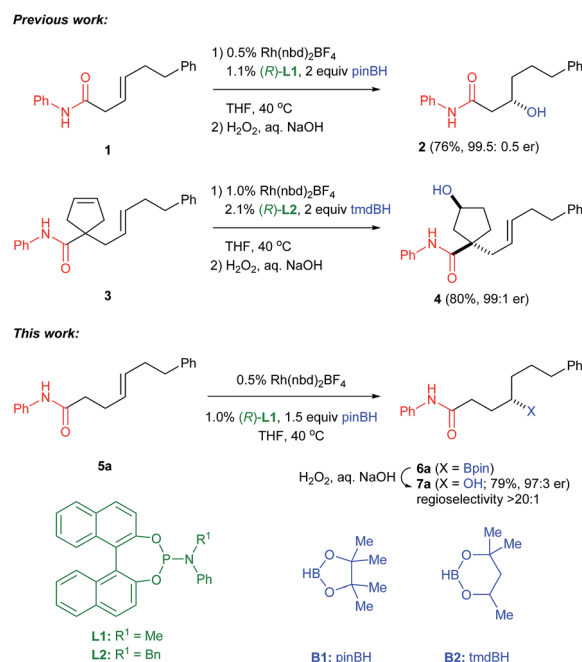
The rhodium-catalyzed, directed catalytic asymmetric hydroboration of γ,δ -unsaturated amides affords a direct route to chiral acyclic secondary γ -borylated carbonyl derivatives in high enantiomeric purity. In contrast to a similar β -borylated amide derivative, the γ -borylated amide undergoes Suzuki–Miyaura cross-coupling with stereoretention. The utility of the boronic ester products is further illustrated by other stereospecific C–B bond transformations leading to γ -amino acid derivatives, 1,4-amino alcohols, and 5-substituted- γ -lactone and γ -lactam ring systems.

Introduction

Chiral boronic esters are valuable synthetic intermediates for a variety of stereospecific transformations.^{1–3} Consequently, an assortment of enantioselective routes for their preparation are under active development.^{4–19} We have focused on the rhodium-catalyzed catalytic asymmetric hydroboration (CAHB) of β,γ -unsaturated amides,²⁰ esters,^{20a} and oxime ethers²¹ for the preparation of functionalized chiral boronic esters. For example, disubstituted alkenes such as (*E*)-**1** undergo highly enantioselective β -borylation by pinacolborane (pinBH) when catalyzed by the combination of (*R*)-**L1** with $[\text{Rh}(\text{nbd})_2\text{BF}_4]$ (*i.e.*, 2 : 1 **L1** : Rh). CAHB followed by oxidation of the C–B bond affords β -hydroxy amide (*S*)-**2** in an enantiomeric ratio (er) greater than 99 : 1 (Fig. 1).^{20e}

In an effort to expand the scope of CAHB, γ,δ -unsaturated amides such as **3** were explored.²² Such substrates differ from **1** in a number of important ways including, (i) the alkene is more remote than in the β,γ -unsaturated amides, (ii) substrate **3** contains two distinctly different alkene moieties, although each is ostensibly positioned γ,δ - with respect to the carbonyl directing group, and (iii) enantioselective borylation of the endocyclic alkene requires controlling the stereochemical course by desymmetrization rather than π -facial discrimination, a fundamentally different set of requirements.²³ In the event, substrate **3** undergoes CAHB/oxidation to give the monounsaturated γ -hydroxy amide (1*S*,3*S*)-**4** in high yield (80%), regioselectivity (>20 : 1) and enantioselectivity (99 : 1 er). Only the endocyclic double bond in **3** undergoes reaction; the

pendant acyclic alkene is untouched. We now report the efficient, regioselective γ -borylation of a similarly disposed alkene in γ,δ -unsaturated amide **5a** (Fig. 1). CAHB by pinBH (0.5% $[\text{Rh}(\text{nbd})_2\text{BF}_4/2$ (*R*)-**L1**]) followed by oxidation affords chiral γ -hydroxy amide (*S*)-**7a** in 79% overall yield (97 : 3 er). While conjugate addition¹¹ and C–H activation^{10a} methodologies provide efficient alternatives to CAHB for enantioselective β -borylation of carbonyl compounds, direct γ -borylation is unique to CAHB.

Fig. 1 CAHB/oxidation of β,γ - versus γ,δ -unsaturated amides.

Department of Chemistry, University of Nebraska-Lincoln, Lincoln, NE 68588-0304, USA. E-mail: jtakacs1@unl.edu

† Electronic supplementary information (ESI) available: Catalyst optimization data; experimental procedures; compound characterization data; spectra. See DOI: 10.1039/c7sc01093a



Results and discussion

The γ,δ -unsaturated amides (*E*)-**5b–n** shown in Fig. 2, along with several related structures, were treated with the catalyst system for γ -borylation (*i.e.*, 2 : 1 combinations of **L1** or **L2** with $[\text{Rh}(\text{nbd})_2\text{BF}_4]$, 0.5 mol% unless otherwise noted). In addition to phenyl amide **5a** described above, the corresponding Weinreb amide **5b**, the morpholine amide²⁴ **5c** and benzyl amide **5d** undergo CAHB to afford the intermediate γ -borylated amides **6b–d** with high levels of enantioselectivity. Within this series of amides, the secondary amides (*i.e.*, *N*-phenyl and *N*-benzyl) give the highest γ -selectivity (>20 : 1). Oxidations of **6b–d** afford the corresponding chiral γ -hydroxy amides **7b–d** ($\geq 94 : 6$ er). CAHBs of **5c** and **5d** were carried out on gram scale giving chiral boronic esters **6c** (82%, 94.5 : 5.5) and **6d** (81%, 96.5 : 3.5 er) in good yield and without loss of enantioselectivity. CAHB/oxidation of an isomeric (*Z*)-alkene, benzyl amide (*Z*)-**5d**, affords results similar to those obtained with the (*E*)-isomer.

In contrast to amides **5a–d**, the analogous ester **8** is largely recovered unchanged upon attempted CAHB; only trace amounts of borylated products are identified along with some evidence for alkene isomerization. δ,ϵ -Unsaturated amide **9**, a one-carbon homologue of benzyl amide (*E*)-**5d**, is found to be

considerably less reactive and less selective. Complete consumption of starting material requires 2% catalyst loading to afford 56% yield of a mixture of borylated products.²⁵

Benzyl amides **5e–g** containing heteroaromatic ring systems are nonetheless good substrates under the standard CAHB/oxidation conditions yielding **7e–g**, respectively. Certain branched alkyl substituents (*i.e.*, **5h–j**) are also well tolerated. In particular, the chiral substrate (*E*)-**5j** demonstrates that (i) the proximal disubstituted alkene with respect to the amide directing group undergoes CAHB while the more distal trisubstituted alkene is untouched and (ii) the stereochemical course of γ -borylation is efficiently catalyst controlled. CAHB/oxidation with (*R*)-**L2** affords (4*S*,7*S*)-**7j**; (*S*)-**L2** affords (4*R*,7*S*)-**7j**. The silyl ether moiety in **5k** is tolerated and affords **7k** (78%, 97 : 3 er). Chiral acetal **5l** again undergoes catalyst controlled γ -borylation with high diastereoselectivity; (*R*)-**L1** affords (*S,S*)-**7l** (72%, 95 : 5 dr); (*S*)-**L1** affords (*R,S*)-**7l** in the same yield and diastereomer ratio. However, substrate **5m**, in which the chiral acetal moiety is in closer proximity to the site of hydroboration, shows a strong matched/mismatched effect. While (*R*)-**L1** affords (*R,S*)-**7m** (70%, 92 : 8 dr), the catalyst employing (*S*)-**L1** gives rise to a complex mixture of regioisomers **10**. Substrate **5n** ($\text{R}^1 = \text{Me}$) also exhibits only modest regioselectivity (3 : 1), perhaps due to

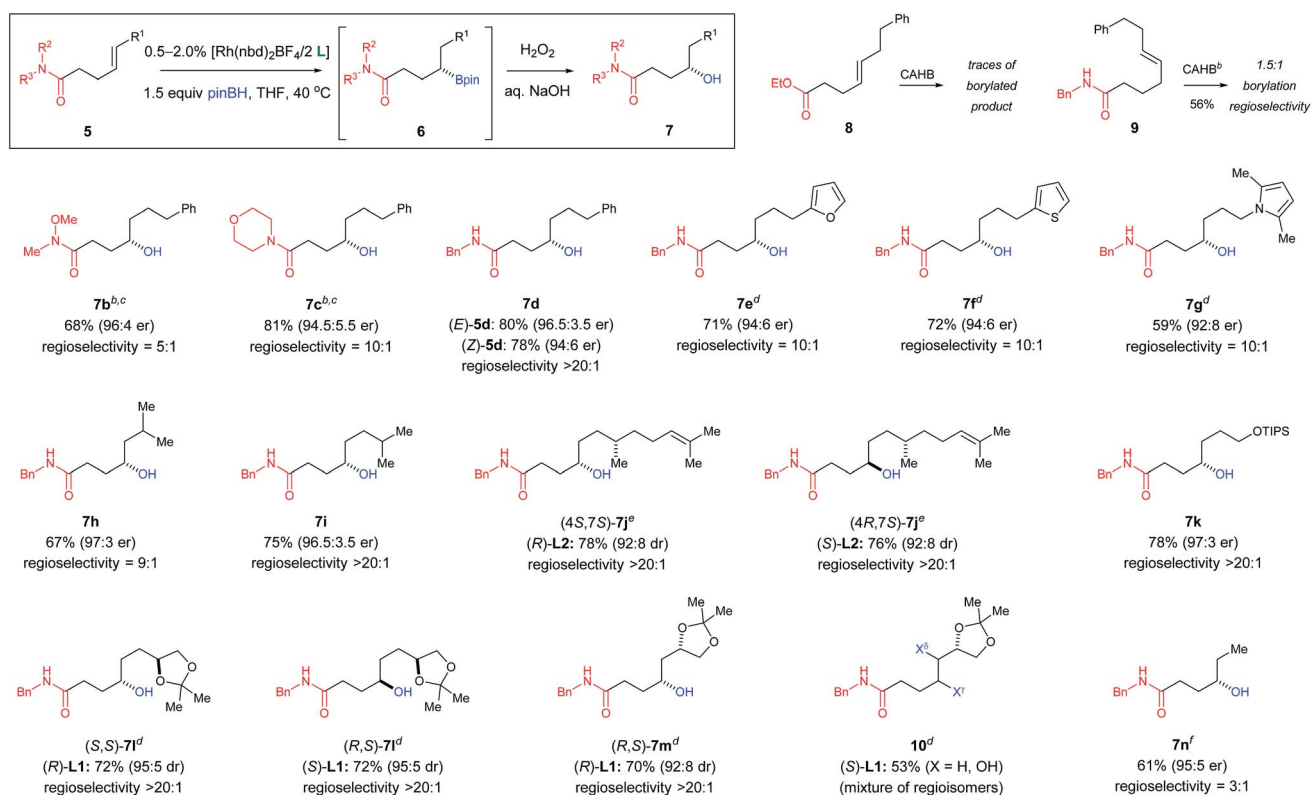


Fig. 2 Substrate scope for CAHB of γ,δ -unsaturated amides. ^aUnless otherwise noted all reaction use 0.5% $[\text{Rh}(\text{nbd})_2\text{BF}_4/2 \text{ (R)-L1}]$, 1.5 equiv. pinBH, THF, 40 °C (12 h) followed by oxidation using $\text{H}_2\text{O}_2/\text{aq. NaOH}$. Unless otherwise noted, the isolated yield is that of the major regioisomer and reflects the average of three experiments generally exhibiting a spread of $\pm 2\%$; regioselectivity is determined from the crude ^1H NMR of **7**. Enantiomer ratios (er) are determined by chiral HPLC analysis; diastereomer ratios (dr) are determined for the purified mixture of diastereomers by integrating major and minor ^{13}C NMR resonances. ^b2.0% $[\text{Rh}(\text{nbd})_2\text{BF}_4/2 \text{ (R)-L1}]$. ^cOxidation conditions: $\text{NaBO}_3/\text{H}_2\text{O}$. ^d1.0% $[\text{Rh}(\text{nbd})_2\text{BF}_4/2 \text{ L1}]$; (*R*)-**L1** is used unless noted otherwise in the figure. ^e1.0% $[\text{Rh}(\text{nbd})_2\text{BF}_4/2 \text{ L2}]$. ^fer is determined by ^{19}F NMR of the corresponding Mosher ester (see ESI for details†).

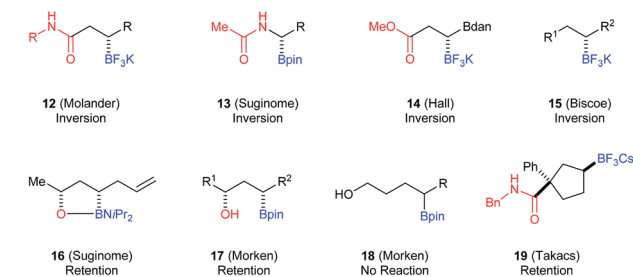


the size of the vinyl substituent compared to other derivatives described above; however, CAHB proceeds in good yield and high enantioselectivity (61%, 95 : 5 er).

Having developed an efficient method for the γ -borylation of γ,δ -unsaturated amides, Suzuki–Miyaura cross-coupling of **11** was examined (Fig. 3). Stereochemical aspects of the palladium-catalyzed cross-coupling of chiral secondary organoboron derivatives have recently attracted a great deal of attention. Molander,²⁶ Suginome,²⁷ and Hall²⁸ reported that β -borylated carbonyl derivatives **12–14**, whether as the boronic ester or the trifluoroborate salt, undergo cross-coupling with stereo-inversion. The stereochemical course is rationalized by intramolecular coordination between the carbonyl oxygen and the boron atom of the boronic ester or the partially hydrolyzed trifluoroborate. The intramolecular coordination promotes invertive transmetalation resulting in overall stereoinversion for cross-coupling. Biscoe^{19a,29} also found stereoinversion for simple substrates lacking functionality needed for coordination to boron during the course of transmetalation (*e.g.*, **15**). On the other hand, Suginome³⁰ reported that boracyclic intermediate **16** undergoes cross-coupling with stereoretention. Similarly, Morken³¹ reported that **17** undergoes hydroxyl-directed, inner-sphere, retentive transmetalation and overall cross-coupling with stereoretention. However, when the hydroxyl is one-carbon further removed, **18** fails to undergo cross-coupling under the otherwise same conditions. We have previously shown that **19** undergoes cross-coupling with overall stereoretention.²²

Chiral boronic ester **6c** (*i.e.*, the morpholine amide) was converted to its corresponding trifluoroborate salt **11c**³² and subjected to palladium-catalyzed cross-coupling using the Buchwald cataCXium® A Pd G3 (**20**) precatalyst.^{22,33} Cross-coupling with chlorobenzene yields amide (*S*)-**21c** (63%); 4-chloroanisole yields (*S*)-**22c** (52%). The products are obtained with essentially complete overall stereoretention.³⁴ We find that

Literature: Coordination-directed stereocontrol in Pd-catalyzed cross-coupling



This work: Stereoretention

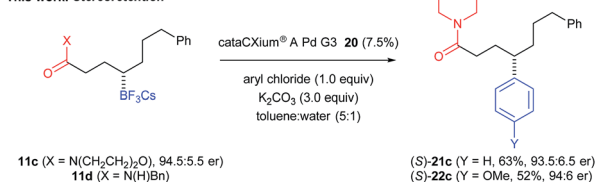


Fig. 3 Stereochemical course of Suzuki–Miyaura cross-coupling with chiral secondary boronic esters or trifluoroborate salts.

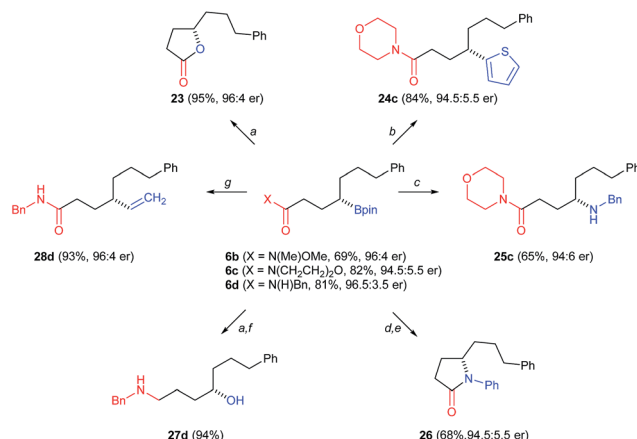


Fig. 4 Stereospecific transformations of organoboranes **6b–d**. Conditions: (a) NaOH/H₂O₂; (b) (i) *n*-BuLi, thiophene, (ii) NBS; (c) BCl₃, BnN₃; (d) BCl₃, PhN₃; (e) 6 M HCl; (f) LiAlH₄; (g) (i) vinylMgBr, (ii) I₂/NaOMe.

the nature of the amide is important to the success of the cross-coupling. In contrast to the tertiary morpholine amide, the analogous secondary amide **11d** does not undergo cross-coupling under the conditions employed for **11c**. Hall *et al.*^{28a} reported that β -boronic esters of secondary amides failed to cross-couple in cases where the corresponding tertiary amide coupled smoothly.

Chiral organoboronates are useful for a variety of other stereospecific transformations. Fig. 4 illustrates several examples starting from chiral boronic esters **6b–d**; the latter are isolated in 69–82% yield from the corresponding alkenes. Treating **6b** with H₂O₂/aq. NaOH affords the known chiral 5-substituted- γ -lactone **23** (95%).³⁵ As an alternative to palladium-catalyzed cross-coupling, the morpholine amide derivative **6c** undergoes stereoretentive cross-coupling with 2-lithiothiofuran under the conditions developed by Aggarwal^{2e} to give **24c** (84%). Compound **6c** also undergoes BCl₃-assisted amination with benzyl azide under the conditions reported by Knochel³⁶ to form the γ -amino acid derivative **25c** (65%). Phenyl azide also serves as a good nucleophile in such amination reactions, and **6c** is converted to the corresponding *N*-phenyl γ -amino acid en route to the 5-substituted- γ -lactam **26** (68%) by acid catalyzed cyclization. Benzyl amide derivative **6d** is efficiently converted to 1,4-aminoalcohol **27d** after oxidation of the C–B bond followed by amide reduction with LAH (94%). While the secondary *N*-benzyl amide **11d** failed in the attempted palladium-catalyzed cross-coupling described above, **6d** undergoes efficient vinylation in a sequence initiated by treatment with excess vinyl Grignard;^{2h} amide **28d** is formed in high yield (93%).

Conclusions

γ,δ -Unsaturated secondary (*i.e.*, *N*-phenyl and *N*-benzyl) and tertiary (*i.e.*, Weinreb and morpholine) amides undergo efficient rhodium-catalyzed CAHB to afford γ -borylated derivatives in good yield and with high levels of asymmetric induction; enantioselectivity as high as 97 : 3 er is observed. While two



good alternative methods are available to prepare chiral secondary β -borylated carbonyl compounds, the present method of directed-CAHB provides to our knowledge the first direct route to chiral acyclic secondary γ -borylated carbonyl compounds with high regio- and enantioselectivity.

A previous study found β - and γ -borylation of related substrates differ in the sense of stereoinduction, *i.e.*, π -facial discrimination.^{20a} However, it is not the case in the present study; β -borylation of β,γ -unsaturated amide **1** and γ -borylation of the one-carbon homologue γ,δ -unsaturated amide **5** add to the same face of the alkene. In the present study, CAHB of a substrate bearing both di- and trisubstituted alkene moieties (*i.e.*, **5j**) occurs only on the disubstituted double bond proximal to the carbonyl group. Chiral substrates **5j** and **5l** undergo highly diastereoselective CAHB with catalyst control; however, substrate **5m**, in which the resident oxygen-bearing stereocenter resides adjacent to the alkene, exhibits a strong matched and mismatched effect with enantiomeric catalysts.

The γ -borylated products are found to undergo stereoretentive palladium-catalyzed Suzuki-Miyaura cross-coupling, presumably *via* amide-directed inner-sphere stereoretentive transmetallation, as well as stereoretentive C–B to C–C transformations using main group organometallic reagents (*e.g.*, lithium and magnesium). In addition, a variety of other stereospecific transformations are highlighted by the conversions of chiral, secondary γ -boronic esters **6b–d** to 1,4-amino alcohols, γ -amino acid derivatives, and 5-substituted- γ -lactone and γ -lactam ring systems. Further studies are in progress.

Acknowledgements

Funding from the NIH National Institutes of General Medical Sciences (R01-GM100101) is gratefully acknowledged. We thank S. Zhang (UNL) for some preliminary experiments.

Notes and references

- For selected recent reviews, see: (a) C.-Y. Wang, J. Derosa and M. R. Biscoe, *Chem. Sci.*, 2015, **6**, 5105; (b) L. Xu, S. Zhang and P. Li, *Chem. Soc. Rev.*, 2015, **44**, 8848; (c) A. H. Cherney, N. T. Kadunce and S. E. Reisman, *Chem. Rev.*, 2015, **115**, 9587; (d) D. Leonori and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2015, **54**, 1082.
- (a) Y. Wang, A. Noble, E. L. Myers and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2016, **55**, 4270; (b) A. Noble, S. Roesner and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2016, **55**, 15920; (c) A. Fawcett, D. Nitsch, M. Ali, J. M. Bateman, E. L. Myers and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2016, **55**, 14663; (d) A. Millán, J. R. Smith, J. L.-Y. Chen and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2016, **55**, 2498; (e) M. Odachowski, A. Bonet, S. Essafi, P. Conti-Ramsden, J. N. Harvey, D. Leonori and V. K. Aggarwal, *J. Am. Chem. Soc.*, 2016, **138**, 9521; (f) J. Lloveria, D. Leonori and V. K. Aggarwal, *J. Am. Chem. Soc.*, 2015, **137**, 10958; (g) C. Sandford, R. Rasappan and V. K. Aggarwal, *J. Am. Chem. Soc.*, 2015, **137**, 10100; (h) R. P. Sonawane, V. Jheengut, C. Rabalakos, R. Larouche-Gauthier, H. K. Scott and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2011, **50**, 3760.
- (a) J. Schmidt, J. Choi, A. T. Liu, M. Slusarczyk and G. C. Fu, *Science*, 2016, **354**, 1265; (b) B. Potter, E. K. Edelstein and J. P. Morken, *Org. Lett.*, 2016, **18**, 3286; (c) T. Rybak and D. G. Hall, *Org. Lett.*, 2015, **17**, 4156; (d) S. C. Matthew, B. W. Glasspoole, P. Eisenberger and C. M. Crudden, *J. Am. Chem. Soc.*, 2014, **136**, 5828; (e) A. W. Buesking and J. A. Ellman, *Chem. Sci.*, 2014, **5**, 1983; (f) D. Imao, B. W. Glasspoole, V. S. Laberge and C. M. Crudden, *J. Am. Chem. Soc.*, 2009, **131**, 5024.
- Stoichiometric reagents: J. M. Brown and B. N. Nguyen, *Stereoselective hydroboration and diboration of carbon-carbon double bonds*, Georg Thieme Verlag, 2011, vol. 1, p. 295.
- For a review on recent advances in the preparation and application of allylboron species, see: C. Diner and K. J. Szabó, *J. Am. Chem. Soc.*, 2017, **139**, 2.
- Enantioselective allyl- and arylborylation: (a) H. M. Nelson, B. D. Williams, J. Miró and F. D. Toste, *J. Am. Chem. Soc.*, 2015, **137**, 3213; (b) T. Jia, P. Cao, B. Wang, Y. Lou, X. Yin, M. Wang and J. Liao, *J. Am. Chem. Soc.*, 2015, **137**, 13760.
- Enantioselective allylic substitution with diborylmethane: (a) Y. Shi and A. H. Hoveyda, *Angew. Chem., Int. Ed.*, 2016, **55**, 3455; (b) M. Zhan, R.-Z. Li, Z.-D. Mou, C.-G. Cao, J. Liu, Y.-W. Chen and D. Niu, *ACS Catal.*, 2016, **6**, 3381.
- Enantioselective aminoborylation: (a) D. Nishikawa, K. Hirano and M. Miura, *Org. Lett.*, 2016, **18**, 4856; (b) N. Matsuda, K. Hirano, T. Satoh and M. Miura, *J. Am. Chem. Soc.*, 2013, **135**, 4934; (c) K. Hong and J. P. Morken, *J. Am. Chem. Soc.*, 2013, **135**, 9252.
- Enantioselective B–H bond insertion: (a) D. Chen, X. Zhang, W.-Y. Qi, B. Xu and M.-H. Xu, *J. Am. Chem. Soc.*, 2015, **137**, 5268; (b) Q.-Q. Cheng, S.-F. Zhu, Y.-Z. Zhang, X.-L. Xie and Q.-L. Zhou, *J. Am. Chem. Soc.*, 2013, **135**, 14094.
- Enantioselective C–H borylation: (a) J. He, Q. Shao, Q. Wu and J.-Q. Yu, *J. Am. Chem. Soc.*, 2017, **139**, 3344; (b) L.-L. Li, Z.-L. Tao, Z.-Y. Han and L.-Z. Gong, *Org. Lett.*, 2017, **19**, 102.
- Enantioselective conjugate borylation: (a) Q. Jiang, T. Guo and Z. Yu, *J. Org. Chem.*, 2017, **82**, 1951; (b) C. Jarava-Barrera, A. Parra, A. López, F. Cruz-Acosta, D. Collado-Sanz, D. J. Cárdenas and M. Tortosa, *ACS Catal.*, 2016, **6**, 442; (c) J.-B. Xie, S. Lin, S. Qiao and G. Li, *Org. Lett.*, 2016, **18**, 3926; (d) Z. Niu, J. Chen, Z. Chen, M. Ma, C. Song and Y. Ma, *J. Org. Chem.*, 2015, **80**, 602; (e) H. Wu, J. M. Garcia, F. Haeffner, S. Radomkit, A. R. Zhugralin and A. H. Hoveyda, *J. Am. Chem. Soc.*, 2015, **137**, 10585; (f) S. Radomkit and A. H. Hoveyda, *Angew. Chem., Int. Ed.*, 2014, **53**, 3387; (g) Y. Luo, I. D. Roy, A. G. E. Madec and H. W. Lam, *Angew. Chem., Int. Ed.*, 2014, **53**, 4186.
- Enantioselective conjunctive cross-coupling: (a) G. J. Lovinger, M. D. Aparece and J. P. Morken, *J. Am. Chem. Soc.*, 2017, **139**, 3153; (b) L. Zhang, G. J. Lovinger, E. K. Edelstein, A. A. Szymaniak, M. P. Chierchia and J. P. Morken, *Science*, 2016, **351**, 70.
- Enantioselective diboration: (a) L. Fang, L. Yan, F. Haeffner and J. P. Morken, *J. Am. Chem. Soc.*, 2016, **138**, 2508; (b) Z. Yu,



- R. J. Ely and J. P. Morken, *Angew. Chem., Int. Ed.*, 2014, **53**, 9632; (c) J. R. Coombs, F. Haefner, L. T. Kliman and J. P. Morken, *J. Am. Chem. Soc.*, 2013, **135**, 11222; (d) G. E. Ferris, K. Hong, I. A. Roundtree and J. P. Morken, *J. Am. Chem. Soc.*, 2013, **135**, 2501.
- 14 Stereospecific catalytic Miyaura borylation: (a) Q. Zhou, H. D. Srinivas, S. Zhang and M. P. Watson, *J. Am. Chem. Soc.*, 2016, **138**, 11989; (b) C. H. Basch, K. M. Cobb and M. P. Watson, *Org. Lett.*, 2016, **18**, 136.
- 15 Enantioselective nucleophilic addition to aldehyde: (a) M. V. Joannou, B. S. Moyer and S. J. Meek, *J. Am. Chem. Soc.*, 2015, **137**, 6176; (b) K. Kubota, E. Yamamoto and H. Ito, *J. Am. Chem. Soc.*, 2015, **137**, 420.
- 16 Group-selective cross-coupling of diboranes: (a) S. N. Mlynarski, C. H. Schuster and J. P. Morken, *Nature*, 2014, **505**, 386; (b) C. Sun, B. Potter and J. P. Morken, *J. Am. Chem. Soc.*, 2014, **136**, 6534; (c) R. D. Dewhurst and T. B. Marder, *Nat. Chem.*, 2014, **6**, 279.
- 17 Enantioselective homologation of organoboron *via* lithiation–borylation: (a) A. Varela, L. K. B. Garve, D. Leonori and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2017, **56**, 2127; (b) D. J. Blair, D. Tanini, J. M. Bateman, H. K. Scott, E. L. Myers and V. K. Aggarwal, *Chem. Sci.*, 2017, **8**, 2898; (c) For a review on lithiation–borylation methodology and its application in synthesis, see: D. Leonori and V. K. Aggarwal, *Acc. Chem. Res.*, 2014, **47**, 3174; (d) For related computational studies, see: S. Essafi, S. Tomasi, V. K. Aggarwal and J. N. Harvey, *J. Org. Chem.*, 2014, **79**, 12148.
- 18 From vinylboronates: (a) J. T. Han, W. J. Jang, N. Kim and J. Yun, *J. Am. Chem. Soc.*, 2016, **138**, 15146; (b) D. Nishikawa, K. Hirano and M. Miura, *J. Am. Chem. Soc.*, 2015, **137**, 15620; (c) J. J. Verendel, O. Pamies, M. Dieguez and P. G. Andersson, *Chem. Rev.*, 2014, **114**, 2130.
- 19 Examples of CAHB by copper catalysts (*a–f*), cobalt catalysts (*k–m*), iron catalysts (*n*), iridium catalysts (*o–p*), and rhodium catalysts (*q–r*): (a) M. G. Ceinos, A. Parra, V. M. Heras and M. Tortosa, *Angew. Chem., Int. Ed.*, 2016, **55**, 6969; (b) Y. Xi and J. F. Hartwig, *J. Am. Chem. Soc.*, 2016, **138**, 6703; (c) D. Wang, P. Cao, B. Wang, T. Jia, Y. Lou, M. Wang and J. Liao, *Org. Lett.*, 2015, **17**, 2420; (d) A. Parra, L. Amenós, M. Guisán-Ceinos, A. López, J. L. G. Ruano and M. Tortosa, *J. Am. Chem. Soc.*, 2014, **136**, 15833; (e) Z.-T. He, Y.-S. Zhao, P. Tian, C.-C. Wang, H.-Q. Dong and G.-Q. Lin, *Org. Lett.*, 2014, **16**, 1426; (f) X. Feng, H. Jeon and J. Yun, *Angew. Chem., Int. Ed.*, 2013, **52**, 3989; (g) D. Noh, S. K. Yoon, J. Won, J. Y. Lee and J. Yun, *Chem.–Asian J.*, 2011, **6**, 1967; (h) Y. Sasaki, C. Zhong, M. Sawamura and H. Ito, *J. Am. Chem. Soc.*, 2010, **132**, 1226; (i) D. Noh, H. Chea, J. Ju and J. Yun, *Angew. Chem., Int. Ed.*, 2009, **48**, 6062; (j) R. Corberán, N. W. Mszar and A. H. Hoveyda, *Angew. Chem., Int. Ed.*, 2011, **50**, 7079; (k) H. Zhang and Z. Lu, *ACS Catal.*, 2016, **6**, 6596; (l) L. Zhang, Z. Zuo, X. Wan and Z. Huang, *J. Am. Chem. Soc.*, 2014, **136**, 15501; (m) J. Chen, T. Xi, X. Ren, B. Cheng, J. Guo and Z. Lu, *Org. Chem. Front.*, 2014, **1**, 1306; (n) J. Chen, T. Xi and Z. Lu, *Org. Lett.*, 2014, **16**, 6452; (o) M. Magre, M. Biosca, O. Pàmies and M. Diéguez, *ChemCatChem*, 2015, **7**, 114; (p) C. Mazet and D. Gérard, *Chem. Commun.*, 2011, **47**, 298; (q) N. Hu, G. Zhao, Y. Zhang, X. Liu, G. Li and W. Tang, *J. Am. Chem. Soc.*, 2015, **137**, 6746; (r) S. A. Moteki, K. Toyama, Z. Liu, J. Ma, A. E. Holmes and J. M. Takacs, *Chem. Commun.*, 2012, **48**, 263.
- 20 (a) S. M. Smith, G. L. Hoang, R. Pal, M. O. B. Khaled, L. S. W. Pelter, X. C. Zeng and J. M. Takacs, *Chem. Commun.*, 2012, **48**, 12180; (b) S. M. Smith, M. Uteuliyev and J. M. Takacs, *Chem. Commun.*, 2011, **47**, 7812; (c) S. M. Smith and J. M. Takacs, *Org. Lett.*, 2010, **12**, 4612; (d) S. M. Smith and J. M. Takacs, *J. Am. Chem. Soc.*, 2010, **132**, 1740; (e) S. M. Smith, N. C. Thacker and J. M. Takacs, *J. Am. Chem. Soc.*, 2008, **130**, 3734.
- 21 V. M. Shoba, N. C. Thacker, A. J. Bochat and J. M. Takacs, *Angew. Chem., Int. Ed.*, 2016, **55**, 1465.
- 22 G. L. Hoang, Z.-D. Yang, S. M. Smith, R. Pal, J. L. Miska, D. E. Pérez, L. S. W. Pelter, X. C. Zeng and J. M. Takacs, *Org. Lett.*, 2015, **17**, 940.
- 23 The symmetry in substrate **3** was also exploited in a DFT study of potential catalytic cycle for the reaction; see: Z.-D. Yang, R. Pal, G. L. Hoang, X. C. Zeng and J. M. Takacs, *ACS Catal.*, 2014, **4**, 763.
- 24 R. Peters, P. Waldmeier and A. Joncour, *Org. Process Res. Dev.*, 2005, **9**, 508.
- 25 When reacted separately substrate **9** requires higher catalyst loading for complete consumption as stated above (*i.e.*, 2.0% Rh for **9** versus 0.5% for **5d**). However, the direct competition of **9** and **5d** for limiting pinBH (equivalent amounts of each) affords only a modest excess of recovered **9** (74% recovered) relative to recovered **5d** (61% recovered). The results suggest that **5d** is consumed only slightly faster than **9** by active catalyst. A relatively slow conversion of the rhodium precatalyst to active catalyst by **9** in the absence of **5d** could account of the conflicting observations; see ref. 22 for a similar example.
- 26 D. L. Sandrock, L. Jean-Gerard, C. Chen, S. D. Dreher and G. A. Molander, *J. Am. Chem. Soc.*, 2010, **132**, 17108.
- 27 (a) T. Awano, T. Ohmura and M. Sugimoto, *J. Am. Chem. Soc.*, 2011, **133**, 20738; (b) T. Ohmura, T. Awano and M. Sugimoto, *J. Am. Chem. Soc.*, 2010, **132**, 13191.
- 28 (a) J. C. H. Lee, H.-Y. Sun and D. G. Hall, *J. Org. Chem.*, 2015, **80**, 7134; (b) J. C. Lee, R. McDonald and D. G. Hall, *Nat. Chem.*, 2011, **3**, 894.
- 29 L. Li, S. Zhao, A. Joshi-Pangu, M. Diane and M. R. Biscoe, *J. Am. Chem. Soc.*, 2014, **136**, 14027.
- 30 M. Daini and M. Sugimoto, *J. Am. Chem. Soc.*, 2011, **133**, 4758.
- 31 T. P. Blaisdell and J. P. Morken, *J. Am. Chem. Soc.*, 2015, **137**, 8712.
- 32 A. J. J. Lennox and G. Lloyd-Jones, *Angew. Chem., Int. Ed.*, 2012, **51**, 9385.
- 33 N. C. Bruno, M. T. Tudge and S. L. Buchwald, *Chem. Sci.*, 2013, **4**, 916.
- 34 Authentic (*R*)-**21c** was prepared *via* enolate allylation using the Evans chiral auxiliary: *in situ* half reduction to aldehyde followed by Wadsworth–Horner–Emmons



- olefination, Pd/C hydrogenation and transamidation with morpholine is described in the accompanying ESI.†
- 35 (a) C. H. Larsen, B. H. Ridway, J. T. Shaw, D. M. Smith and K. A. Woerpel, *J. Am. Chem. Soc.*, 2005, **127**, 10879; (b) M. Aquino, S. Cardani, G. Fronza, C. Fuganti, R. P. Fernandez and A. Tagliani, *Tetrahedron*, 1991, **47**, 7887.
- 36 (a) E. Hupe, I. Marek and P. Knochel, *Org. Lett.*, 2002, **4**, 2681, see also; (b) P. Y. Chavant, F. Lhermitte and M. Vaultier, *Synlett*, 1993, 519; (c) D. S. Matteson and G. Y. Kim, *Org. Lett.*, 2002, **4**, 2153; (d) V. Bagutski, T. G. Elford and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2011, **50**, 1080.

