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Programmed serial stereochemical relay and its application in the synthesis of morphinans[†]

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Herein we report a rationally designed, serial point-to-axial and axial-to-point stereoinduction and its integration into multi-step and target-oriented organic synthesis. In this proof-of-concept study, the configurational stability of several carefully designed atropisomeric intermediates and the fidelity of their unconventional stereoinductions were systematically investigated. The highly functionalized prepared synthetic intermediate was further applied in a novel chemical method to access the morphinans and it is potentially applicable to other structurally related alkaloids.

Introduction

Thoughtful orchestration of substrate and reagent controlled stereochemical induction is essential in any stereoselective synthesis.1 To date, while the fundamental principles behind stereochemical induction are well-grounded, some forms of stereoinduction are much less conventional than others. For example, if we consider "point" and "axial" as two of the most common forms of stereochemical elements, one can appreciate that while the intermolecular point-to-point (for example, oxazaborolidine-mediated reduction)² and axial-to-point (for example, Noyori BINAP hydrogenation)3 stereoinductions are routinely practiced, intramolecular point-to-axial, axial-topoint, and axial-to-axial stereoinductions are considerably more rare.⁴ In the asymmetric synthesis of longithorone A^{5a} and rhazinilam,5b unconventional yet highly effective stereoinductions were elegantly illustrated by the Shair and Zakarian groups, respectively (Scheme 1a). Furthermore, we pondered if a series of these unconventional forms of stereoinduction could be logically sequenced as part of a multi-step synthetic scheme, and in doing so demonstrate their utility in the context of targetoriented synthesis (Scheme 1b).

In preparation for the proposed studies, it is important to recognize that while a "point" stereochemical element can be readily generated and/or identified, an "axial" stereochemical element is only observable if the system can attain a sufficiently high rotational energy barrier. In line with this requirement, we began our proof-of-concept studies by leveraging the wellestablished biaryl system as the source of the "axial" stereochemical element (Scheme 1b). Furthermore, the proposed biaryl system harbors additional functional group (FG) handles which allow for the subsequent programmed-stereoinduction events.

Results and discussion

We first investigated if a "point" stereochemistry that resides in close proximity to the biaryl axis can render, and hence induce, an axial stereochemical property *i.e.* a point-to-axial stereo-induction. In this context, the synthesis of biaryl aldehyde **4** was



Scheme 1 (a) Selected examples of unconventional stereochemical induction in target-oriented organic synthesis; (b) proposed point-to-axial-to-point serial stereochemical relay and its application in the synthesis of morphinans. TBS = tert-butyldimethylsilyl.



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Scheme 2 Synthesis of biaryl benzylic alcohols 5a/5a' to 5d/5d' exhibiting atropisomeric properties. Reagents and conditions: (a) Pd(OAc)₂ (0.04 equiv.), K₂CO₃ (4.0 equiv.), TBAB (1.0 equiv.), DMF, 120 °C, 16 h, 86%; (b) DDQ (1.0 equiv.), CH₂Cl₂/pH 9.2 buffer (10 : 1), 0 °C, 1.5 h, 76%; (c) for 5a/5a': methylmagnesium bromide (3.0 M in Et₂O, 3.3 equiv.), THF, -78 to 23 °C, 16 h, 92% ($5a : 5a' \sim 1 : 1$); for 5b'. 5b': phenylmagnesium bromide (1.0 M in THF, 4.7 equiv.), THF, -78 to 23 °C, 16 h, 85% ($5b : 5b' \sim 6 : 1$); for 5c/5c': t-butyllithium (1.7 M in pentane, 3.5 equiv.), THF, -78 to 23 °C, 16 h, 92% ($5c : 5c' \sim 4 : 1$); for 5d/5d': allylmagnesium bromide (1.0 M in THF, 2.6 equiv.), THF, -78 to 23 °C, 16 h, 97% ($5d : 5d' \sim 3 : 1$). DDQ = 2,3-dichloro-5,6-dicyano-1,4-benzoquinone; DMF = *N*,*N*'-dimethylformamide; OAc = acetate; TBAB = tetra-*n*-butylammonium bromide.

Table 1 Configurational stability study of biaryl benzylic alcohols 5a/ 5a' to 5d/5d'. Diastereoisomeric pairs: 5a/5a': R = Me; 5b/5b': R = Ph; 5c/5c': R = t-Bu; 5d/5d': $R = allyl^a$

	Temp. °C					
Commed	40 (¹ 11 NI	70	90	100	110	120
compa.	(FINING analysis and I in at each temperature up to 110 C)					
5a	1:0	1:0.10	1:0.43	1:0.88	0.79:1	0.69:1
5a′	0:1	0.07:1	0.16:1	0.39:1	0.60:1	0.70:1
5b	1:0	1:0.13	1:0.43	1:0.98	0.68:1	0.63:1
5 b ′	0:1	0.06:1	0.23:1	0.44:1	0.58:1	0.65:1
5 c	1:0	1:0	1:0	1:0.09	1:0.35	0.53:1
5 c ′	0:1	0:1	0:1	0.06:1	0.18:1	0.52:1
5 d	1:0	1:0	1:0.36	0.93:1	0.71:1	0.71:1
5 d ′	0:1	0:1	0.53:1	0.70:1	0.70:1	0.70:1
a Ratio determined by $^1\mathrm{H}$ NMR integration. For details, see the ESI.†						

originally envisioned based on the desymmetrization of the readily accessible (and potentially optically active) dialdehyde **1** (Scheme 2).⁶ However, a more direct approach was later realized through one of the earliest illustrations of CH-activation chemistry pioneered by Dyker,⁷ and the as-obtained biaryl system **3** underwent smooth benzylic oxidation (DDQ) to afford the targeted aldehyde **4** together with its equilibrating hemiacetal 4' in 65% yield over the two steps. The phenolic aldehyde **4** (and 4') was next subjected to a selection of organometallic

reagents, and gratifyingly, each organometallic addition reaction afforded a separable mixture of two stereoisomers in high yield (5a/5a'-5d/5d', 85-97%). Recognizing that the biaryl aldehyde 4 does not have atropisomeric properties at ambient temperature,⁸ this finding indicated that the newly formed biaryls 5a/5a'-5d/5d' exhibit restricted rotation and constitute an illustration of point-induced atropisomerism.

Before pressing onto our next objective, namely the "axial-topoint" stereochemical induction, two crucial criteria had to be considered. First, the substrate must exhibit sufficient atropisomeric configurational stability during the axial-to-point stereoinduction event. Second, any preparatory transformations leading to the axial-to-point stereoinduction event must preserve the atropisomeric property within the substrate. In order to assess the configurational stability of the organometallic addition products (5a/5a'-5d/5d'), chromatographically separated and atropisomerically pure biaryl benzylic alcohols were subjected to thermal conditions and their atropisomerization was monitored by ¹H NMR analysis (Table 1). Our qualitative analysis revealed that all of the substrates exhibited good configurational stability from low to moderately elevated temperatures (up to 70 $^{\circ}$ C), with the *t*-butyl substrates 5c and 5c' demonstrating extended stability up to 100 °C (for details, see the ESI[†]).

Next, we envisaged that the proposed axial-to-point stereoinduction would take place via a dearomatization process, in which the dearomatized product or its chemically transformed derivative would possess new "point" stereochemical element(s).9 After contemplating a variety of well-documented dearomatization protocols, hypervalent iodine mediated oxidative dearomatization appeared the most appealing owing to its operational ease, substrate scope and product versatility.10 Accordingly, benzylic alcohols 5a/5a'-5d/5d' (except 5c/5c', which remained as benzylic alcohols) were protected as their corresponding TBS ethers to prevent any unwanted side reactions and to enhance their atropisomeric stabilities, followed by treatment with PIFA in the presence of methanol (5d/5d' in Scheme 3a, for 5a/5a'-5d/5d' see the ESI[†]). Much to our surprise, each of the isomerically pure phenols 6d and 6d' underwent oxidative dearomatization to afford an identical mixture of two dienone products 7d and 7d' ($7d: 7d' \sim 1:1$). The fact that 7d and 7d' were observed as two distinct diastereoisomers based on our ¹H NMR analysis suggested that the epimerization of the biaryl axis must take place during the course of the reaction. However, a postulated transition state (6-PIFA-TS, Scheme 4a), formed through an associative mechanism by invoking a coordinated phenol-aryliodide complex, should increase the rotational energy barrier and thus render higher configurational stability. While seeking a more concrete explanation, we also became aware of a recent report from the Pappo laboratory describing that several optically pure BINOL substrates underwent metal-catalyzed racemization under single-electron-transfer (SET) conditions (Scheme 4b), although the precise mechanistic origin behind this racemization process remained unclear.11 Very recently, NMR and DFT studies reported by the Koltunov group also suggested an acidmediated atropisomerization of optically pure BINOL via



Scheme 3 (a) Oxidative dearomatization of atropisomerically pure biaryl phenols 6d and 6d'; (b) oxidative dearomatization and configurational stability studies of biaryl phenols 6e and 6e'; and (c) oxidative dearomatization and configurational stability studies of biaryl phenols 6f and 6f'. PCC = pyridinium chlorochromate; PIDA = [bis(acetoxy)iodo]benzene; PIFA = [bis(trifluoroacetoxy)iodo]benzene; and TBSCl = tert-butyldimethylsilyl chloride.

a. MeO CF₃CO₂ CE-CO OTBS HO OTBS PIFA or PIDA MeO MeC MeOH MeO : increased sterics b [Fe] or [Cu] (10 mol%) DCE/HFIP or CH₂Cl₂ οн OH 'nн он он (SET racemization) (R)-BINOL (R)-BINOL (S)-BINOL c. ОН H⁴ OH ОН OH (R)-BINOL BINOL-H⁺ (protonation racemization) (S)-BINOL

Scheme 4 (a) Oxidative dearomatization of **6** *via* an associative mechanism; (b) racemization of (R)-BINOL under SET conditions; and (c) acid promoted racemization of (R)-BINOL. BINOL = 1,1'-bi-2-naphthol; DCE = 1,2-dichloroethane; and HFIP = hexafluoroisopropanol.

several enone intermediates (*e.g.* **BINOL-H**⁺) that structurally closely resemble those generated during the hypervalent iodine and metal–salt mediated phenol oxidations (Scheme 4c).¹²

Continuing our studies, several related phenolic biaryl substrates were conceived in order to understand and to preserve the atropisomeric information (Scheme 3b and c). In this context, the phenolic substrates 6e and 6e' with the phenol and methoxy positions switched compared to those in 6d and 6d' demonstrated much improved thermal stability, however, when separately treated with PIDA in the presence of methanol, also afforded an identical mixture of dienones 7e and 7e' $(\sim 1:1)$. Gratifyingly, phenols 6f and 6f' with an additional methyl substituent were found to retain their atropisomeric purity upon PIDA-mediated oxidative dearomatization, and the atropisomerically pure dienones 7f and 7f' faithfully underwent intramolecular Diels-Alder reactions to afford tetracycles 8f and 8f' in 70% and 61% yield, respectively, with their stereochemical relationship confirmed upon oxidation with PCC. An analogous intramolecular Diels-Alder reaction could also be realized with a 1:1 mixture of dienones 7d and 7d' to afford a near 1:1 mixture of tetracycles 8 and 8' (Scheme 3a). Considering that dienones 7d and 7d' are likely to be configurationally labile at elevated temperatures (see Table 1), this latter result implied that the benzylic OTBS stereocenter offered essentially no stereoinduction during the intramolecular Diels-Alder process. Furthermore, the selective formation of the Diels-Alder product 8f from 7f (and 8f' from 7f') strongly suggested that the stereoinduction arose exclusively from the atropisomeric property



Scheme 5 Atropisomerism dictated stereoinduction leading to the stereocontrolled formation of Diels–Alder products 8f and 8f'.



Scheme 6 Oxidative dearomatization and intramolecular Diels–Alder reaction of biaryl phenols 6g and 6h.

of **7f** (and **7f**') (Scheme 5). In retrospect, the benzylic "point" stereochemical directing element was positioned in proximity to the biaryl-axis for the first "point-to-axial" stereoinduction, while it was distant from the second "axial-to-point" stereo-induction event to suppress its stereo-directing ability. From a design perspective, a single stereo-directing element at each stereochemistry inducing step is synthetically more attractive to avoid any complicated synergistic stereo-directing phenomena.

To conclude our studies in remote stereoinductions, we found that substrates **6g** and **6h** (racemic or optically active) with their OH/OTBS stereocenter relocated in proximity to the intramolecular Diels–Alder transition-state could render a significantly higher level of point-to-point stereoinduction than **6d/6d'** (Scheme 6), an observation that was consistent with our previous synthetic studies towards platencin.¹³

Synthetic applications

Finally, we turned our attention to the synthetic utility of the synthesized advanced intermediates. To this end, we recognized

that the highly functionalized phenanthrene-type tetracycle 8/8'bearing a quaternary stereocenter shared some similarities with the core structure of the morphinan family of natural products.14 Further structural analysis implied that the [2.2.2]bicyclic domain within 8 (or 8') had to be ruptured while retaining its quaternary stereocenter. As shown in Scheme 7, Diels-Alder product 8 (and 8') was readily converted to diketone 9 via a regioselective hydroboration and an oxidative work-up with PCC,15 followed by an acid-induced elimination. It is worth noting that, while tetracycles 8 and 8' could be used in combination in this racemic illustration, enantio- and diastereo-isomerically pure 8 (or 8') will be required to access optically pure intermediates. The more sterically accessible ketone in 9 was converted to its corresponding oxime and thermally equilibrated to a single geometric isomer 10, which was subsequently tosylated (TsCl, 87% yield from 9) to set the stage for the Beckmann rearrangement.¹⁶ Under optimized conditions, tosyloxime 11 underwent regioselective ring expansion under the influence of ZnCl₂ to afford lactam 12 in 73% vield. Further cleavage of the bridgehead C-N bond in 12 was realized through a Hoffmann elimination¹⁷ of its quaternary ammonium salt derivative 13 to furnish dimethylamine 14. Unmasking the dimethoxy acetal in 14 under acidic conditions, very surprisingly, afforded a 1,2-migratory product 15 (after treatment with ethyl chloroformate).18 This undesired bond migration was easily rectified upon partial reduction of diene 14,19 followed by acidic deketalization to yield hydroxy ketone 16 in 66% overall yield. a-Deoxygenation and N-demethylation were sequentially realized through the action of SmI2 and 1chloroethyl chloroformate and the as-obtained ketone 18 represents a valuable common intermediate that is also amenable for a synthetic method to access the hasubananes (bridgehead cyclization "a").²⁰ Advancing tricyclic ketone 18 next called for the formation of a five-membered oxacycle that resides in several flagship morphinans. To this end, phenolic ketone 18 was converted to its dioxolane derivative followed by dioxolane-directed phenolic demethylation.²² Dioxolane а removal followed by treatment of the resulting phenolic ketone (19) with pyridinium tribromide²³ and subsequent heating smoothly delivered an inconsequential mixture of oxa-tetracycle 20 and arylbromide 20a. Next, relocation of the bridgehead olefin²⁴ in 20/20a first converged both compounds through hydrogenation and dioxolane formation, and on further exposure to NBS under radical conditions afforded styrene 21. Reductive detosylation of 21 under Birch-type conditions took place with concomitant piperidine formation^{21s} and furnished synthetic dihydrocodeinone upon dioxolane removal (HCl, MeOH).²⁵ Dihydrocodeinone serves as a valuable intermediate to readily access a diverse array of naturally occurring and designed morphinans, including but not limited to dihydrocodeine, codeine, morphine, thebaine and oxycodone.26

Conclusions

In summary, we have demonstrated proof-of-concept, rationally designed serial point-to-axial-to-point stereoinductions and examined the stereochemical fidelity of these processes. During



Scheme 7 Synthesis of morphinans. Reagents and conditions: (a) BH₃·THF (5.0 equiv.), THF, -78 to 70 °C, 4 h; then PCC (5.0 equiv.), 23 °C, 5 h, 41%; (b) p-TsOH+H₂O (1.0 equiv.), toluene, 110 °C, 20 min, 95%; (c) H₂NOH+HCl (3.0 equiv.), NaOAc (3.0 equiv.), MeOH, 23 to 70 °C, 16 h (heat to neat), 90%; (d) TsCl (1.5 equiv.), NaOH (2.5 equiv.), acetone/H₂O (1 : 1), 23 °C, 2 h, 97%; (e) ZnCl₂ (2.0 equiv.), MeCN, 100 °C, 4 h; then ZnCl₂ (2.0 equiv.), 100 °C, 4 h, 73%; (f) LiAlH₄ (10 equiv.), THF, 0 to 70 °C, 8 h; (g) Mel (14 equiv.), CH₂Cl₂, 23 °C, 10 h, 42% for two steps; (h) KOH (20% in MeOH), 70 °C, 10 h, 80%; (i) HCl (10% aq.)/MeOH (4 : 1), 23 °C, 1 h, 67%; (j) EtOCOCl (5.0 equiv.), NaHCO3 (20 equiv.), 1,2-dichloroethane, 100 °C, 2 h, 93%; (k) Pd/CaCO₃ (1.0 equiv.), MeOH, H₂ (1 atm), 23 °C, 16 h, (1,4 : 1,2-hydrogenated compound ~ 5 : 1); then HCl (10% aq.)/MeOH (2 : 1), 23 °C, 1 h, 66% for two steps; (I) Ac₂O (10 equiv.), Et₃N (10 equiv.), DMAP (0.1 equiv.), CH₂Cl₂, 23 °C, 4 h; (m) Sml₂ (0.1 M in THF), THF/MeOH (1 : 1), -78 °C, 0.5 h, 78% for two steps; (n) 1-chloroethyl chloroformate (20 equiv.), NaHCO₃ (20 equiv.), 1,2-dichloroethane, 100 °C, 2 h; then MeOH, 60 °C, 1.5 h; then TsCl (2.0 equiv.), DMAP (0.4 equiv.), Et₃N (2.3 equiv.), CH₂Cl₂, 23 °C, 1.5 h, 64% for two steps; (o) TMSCl (2.0 equiv.), ethylene glycol/CH₂Cl₂ (1 : 1), 23 to 50 °C, 5 h, 92%; (p) L-selectride (1.0 M in THF, 5.0 equiv.), THF, 80 °C, 24 h; then HCl (4.0 N aq.)/MeOH (1 : 15), 23 to 45 °C, 3 h, 77% for two steps; (q) PyH·Br₃ (2.0 equiv.), CH₂Cl₂/AcOH (2 : 5), 23 °C, 30 min; then LiBr (5.0 equiv.), Et₃N (10.0 equiv.), MeCN, 60 °C, 20 min, 53%; (r) Pd/C (10% wt/wt, 2.0 equiv.), EtOAc/MeOH (1 : 3), H₂, 23 °C, 30 min, 90%; then TMSCl (excess), ethylene glycol/CH₂Cl₂ (1 : 1), 23 to 50 °C, 5 h, 80%; (s) NBS (1.05 equiv.), benzoyl peroxide (0.5 equiv.), CCl₄, 80 °C, 1 h; then Et₃N (7.8 equiv.), 80 °C, 15 min, 60%; (t) Li (excess), NH₃, -78 °C, 10 min, 69%; then HCl (4.0 N aq.)/MeOH (1 : 15), 70 °C, 6 h, 75%; and (u) LiAlH₄ (excess), THF, 0 °C, 45 min, 74%. DMAP = N,N'dimethylaminopyridine; EtOAc = ethyl acetate; L-selectride = lithium tri-sec-butylborohydride; NBS = N-bromosuccinimide; PyH·Br₃ = pyridinium tribromide; TMSCl = trimethylsilyl chloride; p-TsOH·H₂O = para-toluenesulfonic acid monohydrate; TsCl = para-toluenesulfonyl chloride; and Ts = para-toluenesulfonyl.

this investigation, an unexpected atropisomeric epimerization upon hypervalent iodine-mediated oxidative dearomatization of isomerically pure biaryl phenols was discovered and systematically investigated. This finding bears important ramifications in related oxidative generation¹¹ and transformations of designed and naturally occurring phenolic biaryl systems and their atropisomeric integrity. While the undesired atropisomeric epimerization could be suppressed by increasing the steric pressure about the biaryl axis of the substrate (*e.g.* **6f/6f'**), more in depth mechanistic studies are in progress to render a substrate-independent solution.²⁷ Further application of the developed synthetic strategy also enabled a novel synthetic method to access the morphinan family of natural products and potential access to other related alkaloid structures. Conceptually related programmed serial-stereoinductions, particularly those involving unconventional intramolecular axial-to-point and point-to-axial processes, are currently being designed and implemented in the context of target-oriented synthesis in our laboratory.

Conflicts of interest

There are no conflicts to declare.

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