

Cite this: *RSC Adv.*, 2018, 8, 15961

High performance planar p-i-n perovskite solar cells based on a thin Alq₃ cathode buffer layer†

Lijia Chen,^{‡a} Gang Wang,^{‡bc} Lianbin Niu,^a Yanqing Yao,^{bc} Yunxia Guan,^a Yuting Cui^a and Qunliang Song^{id*bc}

As a thin cathode buffer layer (CBL) tris-(8-hydroxyquinoline), aluminum (Alq₃) is successfully introduced into the planar p-i-n perovskite solar cells (PSC) between the PCBM layer and cathode with a device structure of ITO/PEDOT:PSS/CH₃NH₃PbI₃(Cl)/PCBM/Alq₃/Ag. Due to the as-introduced thin Alq₃ CBL, a high performance planar PSC has been achieved with a fill factor (FF) of 72% and maximum power conversion efficiency (PCE) of 14.22%. The PCE value is approximately 29% higher than that of the reference device without Alq₃ CBL. Concerning the results of AC impedance spectra and transient photocurrent measurements, such a remarkable improvement of PCE is mainly attributed to the Alq₃-caused better charge-extraction at the cathode, which is induced by reducing charge accumulation between PCBM and Ag.

Received 24th February 2018
Accepted 23rd April 2018

DOI: 10.1039/c8ra01633j

rsc.li/rsc-advances

1. Introduction

Recently, organic-inorganic hybrid perovskite materials have attracted considerable attention due to their strong absorption across the visible spectrum, long carrier diffusion length (100–1000 nm) and low cost of fabrication.^{1–4} The power conversion efficiency (PCE) of the organometal halide-based perovskite solar cells (PSC) has skyrocketed from 3.8% to 22.1% during the past eight years.^{5–8} The dramatic improvement of the performance is mainly attributed to the precise control of the fabrication process, the choice of materials and the design of the device architecture.^{4,9} Up to now, two types of planar architectures are adopted in PSC community: (1) p-i-n structure with transparent conducting oxide (TCO) substrate/hole transport layer (HTL) (p)/perovskite (i)/electron transport layer (ETL) (n);^{10,11} (2) n-i-p structure with TCO substrate/ETL (n)/perovskite (i)/HTL (p).^{12,13} Among the two architectures, the n-i-p structure suffers a large hysteresis in the current density-voltage (*J*-*V*) characteristics since it consists of an n-type ETL (TiO₂, ZnO, *etc.*).^{14,15} From this point of view, the p-i-n structure is superior to the n-i-p structure due to its inherent small hysteresis.^{10,16} In general, solution processable poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS) and 6,6-phenyl C61-butyric acid methyl ester (PCBM) are two common

used HTL and ETL materials in p-i-n architecture (indium tin oxide (ITO)/PEDOT:PSS/perovskite/PCBM/cathode).^{17–20} However, due to the band bending and surface states, the direct contact of cathode, *e.g.* Ag, with PCBM is not an ideal choice for electron extraction. Lately, much attention have been paid on increasing the electron extraction ability at the PCBM/cathode interface to further enhance the performance of PSC.^{21–23} For example, Xue *et al.*²⁴ showed that inserting amino-functionalized polymer PN₄N between PCBM and cathode could improve the performance of PSC due to the reduction of the contact resistance by suppressing interface charge recombination. BCP and rhodamine have also been served as cathode buffer layers to achieve high PCE in PSC.^{25–27} Here, the insertion of a BCP or rhodamine CBL could create downward band bending toward the cathode contact to reduce the interface charge recombination, which is helpful for obtaining good ohmic contact. In addition, LiF^{16,18} and PEIE²⁸ are also commonly used CBL materials, which can create dipole moment across the interfaces to reduce energy barrier between the PCBM and cathode. Apparently, all above mentioned methods highlight the important role of the interface engineering on electron extraction ability at the PCBM/cathode interface to further enhance the performance of PSC. However, some of these CBLs show relatively unstable properties, such as the crystallization of BCP in air at room temperature.²⁹ Although heavily n-doped (n⁺) TiO_x was recently developed as stable CBL to extract electrons efficiently,³⁰ developing new stable CBLs with easy and precise controllability has been taken into account. Alternatively, tris-(8-hydroxyquinoline) aluminum (Alq₃) is another small molecular organic material, which can be thermal deposited to form a more stable and uniform thin layer.²⁹ Thus, it is probably to use Alq₃ as a potential CBL in PSC.

In this work, a thin Alq₃ CBL has been inserted into the PCBM/cathode interface with a planar configuration of ITO/

^aCollege of Physics and Electronics Engineering, Chongqing Normal University, Chongqing 401331, P. R. China

^bInstitute for Clean Energy and Advanced Materials, Faculty of Materials and Energy, Southwest University, Chongqing 400715, P. R. China. E-mail: qsong@swu.edu.cn

^cChongqing Key Laboratory for Advanced Materials and Technologies of Clean Energy, Chongqing 400715, P. R. China

† Electronic supplementary information (ESI) available: XRD, SEM of perovskite films and stability of solar cells were showed in ESI. See DOI: 10.1039/c8ra01633j

‡ L. Chen and G. Wang contributed equally to this work.



PEDOT:PSS/CH₃NH₃PbI₃(Cl)/PCBM/Alq₃(1.5 nm)/Ag. A maximum PCE (14.22%) and FF (72%) were achieved in the device with Alq₃ CBL, which is ~29% and ~20% higher than that of PSC without Alq₃ CBL (PCE of 11.04% and FF of 60%). Such improvements can be attributed to the high efficiency charge-extraction, which was solidly confirmed by the reduction of charge transfer resistance from AC impedance spectrum measurement and fast charge-extraction from transient photocurrent measurement.

2. Experimental section

2.1 Materials

PEDOT:PSS (Clevis PVP Al 4083), methylammonium iodide (MAI, 99.5%), PbCl₂ (99.99%), PbI₂ (99.99%), PCBM (99%) were purchased from Xi'an Polymer Light Technology Corp (China), while *N,N*-dimethylformamide (DMF, 99.8%), dimethyl sulfoxide (DMSO, 99.9%), chlorobenzene (CB, 99.8%) and Alq₃ (99.995%) were purchased from Sigma-Aldrich. A solution of 0.14 M PbCl₂, 1.26 M PbI₂ and 1.4 M MAI in a cosolvent of DMSO and DMF (vol. ratio = 1 : 9) was prepared in a glove box and stirred overnight under room temperature. 15 mg PCBM was dissolved in 1 mL CB and stirred overnight under room temperature.

2.2 Device fabrication

ITO was cleaned by an ultrasonic cleaning machine (KQ3200DV) with 4% detergent (Decon 90) sequentially in deionized water and ethanol at 60 °C. PEDOT:PSS was spin-coated on ITO at 6000 rpm for 60 s in air and dried on hotplate at 120 °C for 20 min. Then the perovskite precursor solution was spin-coated on the PEDOT:PSS at 1000 rpm for 3 s and 4000 rpm for 20 s. When the perovskite precursor was roll out for 10 s, chlorobenzene (200 μL) was dropwise added onto the 4000 rpm spinning film. The film was heated at 60 °C for 2 min and then 85 °C for 25 min on a hotplate. The PCBM was spin-coated on the perovskite film from chlorobenzene solution (15 mg mL⁻¹) with a speed of 4000 rpm for 40 s. All the above procedures were conducted inside a N₂-filled glovebox with oxygen and moisture levels <3 ppm except for casting the PEDOT:PSS in ambient air. Finally, 1.5 nm Alq₃ and 100 nm Ag were deposited under high-vacuum (~5 × 10⁻⁵ Pa) through

a shadow mask, resulting in an active area of 0.09 cm². The film thickness was monitored *in situ* by a quartz crystal.

2.3 Characterization

The crystal structures of the perovskite were investigated by an X-ray diffraction system (Shimadzu XRD-7000). The grain size of the perovskite structure was characterized by field-emission scanning electron microscopy (SEM, JSM-6700F). UV-vis absorption spectra measurements were carried out using Shimadzu UV-2550 spectrometer. The *J-V* characteristics were measured by a Keithley 2400 source meter with 100 mW cm⁻² (AM 1.5G) simulated light from a Newport simulator (model 94043A). During the measurement, the scan direction was reverse (from +1.2 to -1.2 V) or forward (from -1.2 to +1.2 V) with a voltage step of 10 mV and 0 ms delay time by Keithley 2400. The external quantum efficiency (EQE) of devices was calculated from photocurrent generated from the modulated monochromatic light which was measured by a lock-in amplifier (SR-830). The *J-V* and EQE characteristics were measured in nitrogen filled glove box with less than 3 ppm of H₂O and O₂. The alternating-current (AC) impedance spectrum measurement was performed on an electrochemical workstation (model 660D, Shanghai Chenhua Instrument Co., Ltd., China) in dark, with an applied bias of *V*_{oc} (1.01 V). An AC signal with an amplitude of 10 mV over the frequency range of 1 MHz to 1 Hz was adopted. The impedance parameters were determined by fitting the impedance spectra using Z-view software. In transient photocurrent (TPC) characteristics, the PSC were connected directly in series with an oscilloscope (Agilent DSO-X3102A) and the input impedance of the oscilloscope is switched to 50 ohms. All the voltages were relative to the Ag electrode.

3. Results and discussion

3.1 Photovoltaic performance of PSC

Schematic structure of the planar p-i-n PSCs with Alq₃ CBL and its corresponding energy level^{2,31} are shown in Fig. 1(a) and (b), respectively.

The *J-V* characteristics of the devices with different thickness (0, 0.5, 1.5, 2.5, 3.5 nm) of Alq₃ CBL are shown in Fig. 2(a), and their corresponding photovoltaic parameters are listed in Table 1. It is found that the PCE of PSC increases with the increase of

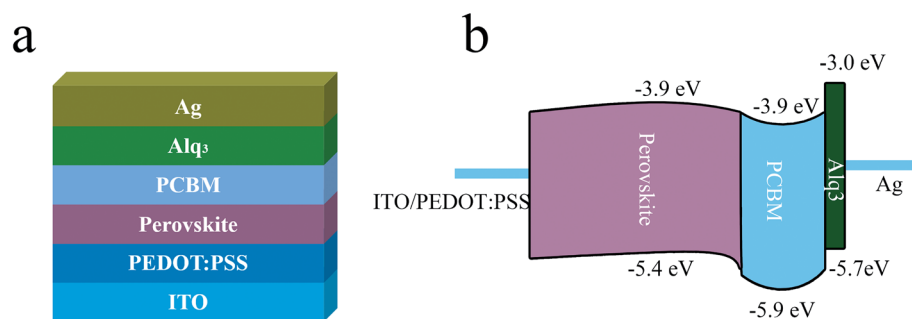


Fig. 1 (a) Schematic structure of the PSC with Alq₃ CBL. (b) The corresponding energy level diagram of the PSC.



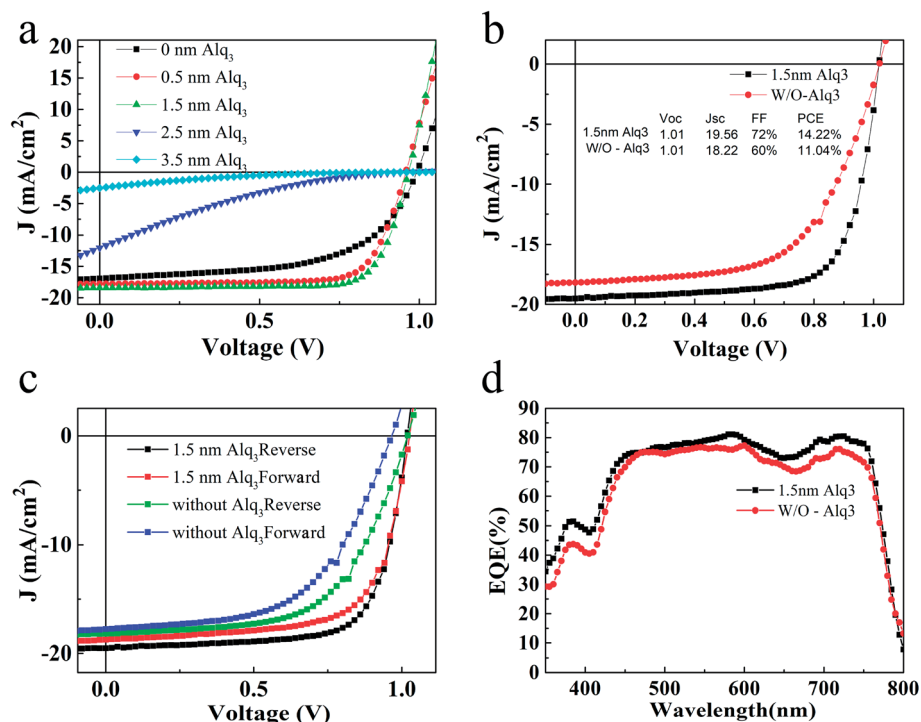


Fig. 2 (a) J - V curves of PSCs with different thickness (0, 0.5, 1.5, 2.5, 3.5 nm) of Alq₃ CBL under an illumination of AM 1.5G, 100 mW cm⁻². (b) J - V curves of best devices with 1.5 nm Alq₃ and without Alq₃ CBL. (c) The forward and reverse J - V curves under an illumination of AM 1.5G, 100 mW cm⁻² for PSC devices. (d) EQE spectra for PSC with 1.5 nm Alq₃ and without Alq₃ CBL.

Alq₃ thickness until 1.5 nm and then decreases with further increasing Alq₃ thickness. The J - V characteristics of the best devices are shown in Fig. 2(b). The PSC without Alq₃ CBL exhibits an open circuit voltage (V_{oc}), a short current density (J_{sc}), FF and PCE of 1.01 V, 18.22 mA cm⁻², 60% and 11.04%, respectively. After inserting 1.5 nm Alq₃ CBL between PCBM and Ag interface, the highest FF and J_{sc} increase to 72% and 19.56 mA cm⁻², respectively. As a consequence, the PCE increases to 14.22%. To confirm the accuracy of J_{sc} , the EQE spectra of the devices were measured and display in Fig. 2(d). The integrated current densities of the devices with 1.5 nm Alq₃ and without Alq₃ CBL are 18.97 mA cm⁻² and 17.97 mA cm⁻², respectively, agreeing with the previous J - V curves results. In addition, the similar shape of the EQE spectra indicate the Alq₃ was too thin to alter the spatial distribution of the optical field in the device, which are consistent with the similar absorbance

of perovskite films with and without Alq₃ (Fig. S3†). As the existence of Alq₃ varies the two devices, the characteristics of device interface should be the reason of a different performance in the Alq₃ CBL PSC: a reduced interface resistance may improve the performance. Thereafter, the series resistance (R_s) was calculated from the slope of J - V curves at 0 mA cm⁻², as listed in Table 1. Compared with the R_s extracted from the PSC without Alq₃ CBL (7.24 Ω cm²), the notable reduction of R_s with 1.5 nm Alq₃ CBL (2.52 Ω cm²) can reasonably explain the improvement of J_{sc} and FF. Fig. 2(c) shows the forward (from -1.2 to +1.2 V) and reverse (from +1.2 to -1.2 V) J - V characteristics of PSCs with 1.5 nm Alq₃ CBL and without Alq₃ CBL, respectively. The hysteresis is obviously reduced in the device with 1.5 nm Alq₃ CBL, suggesting less surface traps formation at the Alq₃/Ag interface and then highlights the role of the Alq₃ CBL. And the J - V hysteresis curves under an illumination of AM 1.5G, 100 mW cm⁻² at different delay time are presented in ESI Fig. S4.†

The most obvious effect of adding Alq₃ CBL on the device performance is the enhancement of FF and J_{sc} in the p-i-n PSC, similar to the results by using other CBLs.^{16,26} This effect thus can be similarly explained as the result of suppressing upward band bending of PCBM, reducing the interfacial charge recombination to achieve a better electron extraction from PCBM. Fig. 3 exhibits schematic band diagrams of the cells under light illumination. Owing to the mismatch of the Fermi level of PCBM and the work function of the metal cathode, a Schottky barrier and then a reverse electrical field in PCBM are created in the device without Alq₃ CBL. Because of the Schottky barrier, electrons from the perovskite accumulate in PCBM,

Table 1 Photovoltaic performance of p-i-n PSC with different thickness Alq₃ and without Alq₃ CBL

CBL	V_{oc} (V)	J_{sc} (mA cm ⁻²)	FF (%)	PCE (%)	R_s
0 nm Alq ₃	0.99	16.08	58	9.23	9.73
0.5 nm Alq ₃	0.99	16.88	76	12.15	8.77
1.5 nm Alq ₃	0.97	17.05	77	13.06	6.08
2.5 nm Alq ₃	0.95	10.15	17	1.60	292.26
3.5 nm Alq ₃	0.95	5.39	14	0.71	1854.21
1.5 nm Alq ₃ (best)	1.01	19.56	72	14.22	2.52
Without Alq ₃ (best)	1.01	18.22	60	11.04	7.24



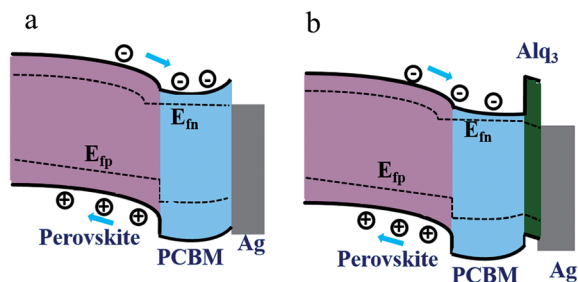


Fig. 3 Schematic band diagrams under light illumination for the PSCs without and with Alq₃ CBL. (a) perovskite/PCBM/Ag and (b) perovskite/PCBM/Alq₃/Ag.

leading to charge recombination with holes in perovskite, as shown in Fig. 3(a). The insertion of a thin Alq₃ CBL shifts down the band bending to the cathode contact in the PCBM layer (Fig. 3(b)), which is beneficial to electron transportation. Meanwhile, the inserted Alq₃ CBL is thin enough for photo-induced electrons tunneling across the PCBM/cathode interface. Therefore, charge accumulation seldomly emerges at the PCBM/Alq₃/Ag interface. The reduction of R_s after adding a thin Alq₃ layer are observed in Table 1. When the thickness of Alq₃ was further increased over 1.5 nm, it would block the electron transportation due to the rapidly increased R_s .

3.2 Alternating-current impedance spectrum measurement

To further explore the effect of adding Alq₃ CBL on electrical properties, the AC impedance spectrum measurement was

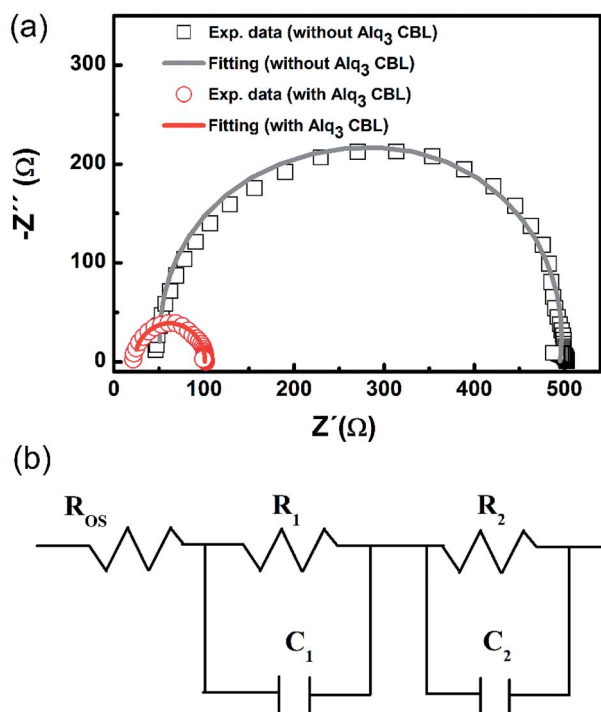


Fig. 4 (a) Nyquist plots and fitting curves (solid line) of PSCs without (black square) and with (red circle) 1.5 nm Alq₃ CBL. AC impedance was measured in the dark at near the V_{oc} . (b) Equivalent electrical circuit used for fitting the impedance data.

Table 2 Parameters of equivalent circuit for the p-i-n PSCs with 1.5 nm Alq₃ CBL and without Alq₃ CBL

CBL	R_{os} (Ω)	R_1 (Ω)	C_1 (F)	R_2 (Ω)	C_2 (F)
1.5 nm Alq ₃	23	39	3.03×10^{-8}	39	3.03×10^{-8}
Without Alq ₃	48	174	8.33×10^{-9}	265	1.67×10^{-8}

performed on the two devices with 1.5 nm Alq₃ and without Alq₃ CBL in the dark at near the V_{oc} (1.01 V). The Nyquist plots of the two PSC are shown in Fig. 4(a). The semicircle of the PSC with 1.5 nm Alq₃ CBL shrinks to a smaller one compared with the device without Alq₃ CBL. These AC impedances can be well fitted by an equivalent electrical circuit, as shown in Fig. 4(b). The equivalent electrical circuit is composed of three serial connected parts:³² (1) an ohmic series resistance (R_{os}) related to the electrical contacts, and sheet resistance of collecting electrodes; (2) a RC circuit (R_1C_1) describing the property of interface between the active layer and electrodes; (3) another RC circuit (R_2C_2) describing the behavior of photoactive layer. The fitting parameters are summarized in Table 2. A little difference of the R_{os} between the two cells are observed, which could be attributed to the similar configuration. Especially, the value of R_1 extracted from the devices with and without Alq₃ CBL are respectively 39 and 174 Ω . The obvious reduction of R_1 in the Alq₃ CBL is consistent with the R_s obtained from $J-V$ curves and confirms a better charge transfer between the PCBM and Ag electrodes.

3.3 TPC measurements of PSC

To assess the practical benefits of the reduced interface resistance, we examined the charge-extraction time of the two PSC *via* transient photocurrent (TPC) analyses. The charge-extraction time can be extracted from the TPC response to accurately describe the efficiency of charge-extraction.³³ Fig. 5 displays the normalized TPC in the PSC with 1.5 nm Alq₃ and without Alq₃ CBL. The PSC with 1.5 nm Alq₃ CBL shows a shorter charge-extraction time (0.51 μ s), indicating that there is an efficient charge-extraction in the Alq₃ CBL device compared with that without Alq₃ CBL (0.67 μ s). Thus, the better contact interface (due to reduced interface resistance) and the reduced surface traps (reflects in reduced hysteresis) contribute together to gain the high efficient charge-extraction in the device with 1.5 nm Alq₃ CBL. These results have no doubt provided strong and unequivocal support to our speculation that Alq₃ CBL leads to high efficient charge-extraction at the cathode interface and then the high FF, and J_{sc} .

Compared with state of the art planar perovskite solar cells, the relative low performance in this study is due to relative low V_{oc} and J_{sc} . The higher V_{oc} can be achieved to choose better ETL/HTL. The low J_{sc} might be partially due to the unoptimized thickness of active perovskite film to generate charge carrier. The thickness of active perovskite film measured by step profiler in this study is about 350 nm, which means that some incident light is not absorbed by the perovskite film, as shown in Fig. S3.† Another reason for this low J_{sc} is the charge



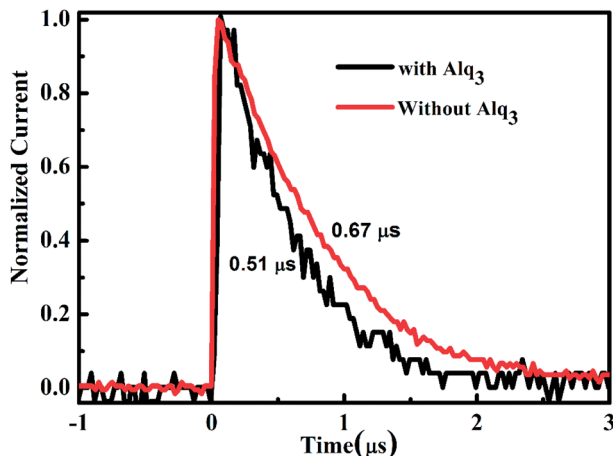


Fig. 5 TPC measurements of PSCs with and without Alq₃ CBL.

recombination in the perovskite film and at the perovskite/ETL(HTL) interface, which has been improved in this study. Thus, by further optimizing the perovskite film and choosing better pair of ETL/HTL, the performance of planar perovskite solar cells can be hopefully improved in the future work.

4. Conclusion

In summary, we demonstrated an enhancement of J_{sc} , FF and PCE in p-i-n PSCs by inserting a thin Alq₃ CBL between PCBM and Ag electrode. Due to the existence of 1.5 nm Alq₃ CBL, the PSC with a J_{sc} of 19.56 mA cm⁻², a FF of 72% and a PCE of 14.22% was achieved. In comparison, the PSC without Alq₃ CBL only show a J_{sc} of 18.22 mA cm⁻², a FF of 60% and a PCE of 11.04%. The remarkable improvement of performance is attributed to the efficient charge-extraction due to the reduction of charge accumulation and contact resistance at the cathode interface, which is confirmed by comparing AC impedance and TPC results between the two PSC. We believe that inserting a thin Alq₃ CBL provides a promising strategy to achieve high performance of p-i-n planar perovskite solar cells.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (Grant No. 11274256, 11774293, 11504036), Program for Innovation Team Building at Institutions of Higher Education in Chongqing (Grant No. CXTDX201601011), Scientific and Technological Research Program of Chongqing Municipal Education Commission (Grant No. KJ1500318 and KJ1600327), The Natural Science Foundation Project of CQ CSTC (Grant No. CSTC2017jcyjAX0333 and CSTC2015jcyjBX0032) and Fundamental Research Funds for the Central Universities (Grant No. XDJK2017A002).

References

- G. Xing, N. Mathews, S. Sun, S. S. Lim, Y. M. Lam, M. Grätzel, S. Mhaisalkar and T. C. Sum, *Science*, 2013, **34**, 344.
- G. Zhou, J. Wu, Y. Zhao, Y. Li, J. Shi, Y. Li, H. Wu, D. Li, Y. Luo and Q. Meng, *ACS Appl. Mater. Interfaces*, 2018, **10**, 9503.
- S. D. Stranks, G. E. Eperon, G. Grancini, C. Menelaou, M. J. Alcocer, T. Leijtens, L. M. Herz, A. Petrozza and H. J. Snaith, *Science*, 2013, **342**, 341.
- J. Zheng, M. Zhang, C. F. J. Lau, X. Deng, J. Kim, Q. Ma, C. Chen, M. A. Green, S. Huang and A. W. Y. Ho-Baillie, *Sol. Energy Mater. Sol. Cells*, 2017, **168**, 165.
- A. Kojima, K. Teshima, Y. Shirai and T. Miyasaka, *J. Am. Chem. Soc.*, 2009, **131**, 6050.
- M. Saliba, T. Matsui, K. Domanski, J. Y. Seo, A. Ummadisingu, S. M. Zakeeruddin, J. P. Correa-Baena, W. R. Tress, A. Abate and A. Hagfeldt, *Science*, 2016, **354**, 206.
- Y. Q. Yao, G. Wang, F. Wu, D. B. Liu, C. Y. Lin, X. Rao, R. Wu, G. D. Zhou and Q. L. Song, *RSC Adv.*, 2017, **7**, 42973.
- Z. Hu, J. Miao, T. Li, M. Liu, I. Murtaza and H. Meng, *Nano Energy*, 2018, **43**, 72.
- D. B. Liu, G. Wang, F. Wu, R. Wu, T. Chen, B. F. Ding and Q. L. Song, *Org. Electron.*, 2017, **43**, 189.
- D. Forgács, M. Sessolo and H. J. Bolink, *J. Mater. Chem. A*, 2015, **3**, 14121.
- J. Miao, Z. Hu, M. Liu, M. U. Ali, O. Goto, W. Lu, T. Yang, Y. Liang and H. Meng, *Org. Electron.*, 2018, **52**, 200.
- M. Liu, M. B. Johnston and H. J. Snaith, *Nature*, 2013, **501**, 395.
- X. Zheng, H. Lei, G. Yang, W. Ke, Z. Chen, C. Chen, J. Ma, Q. Guo, F. Yao, Q. Zhang, H. Xu and G. Fang, *Nano Energy*, 2017, **38**, 1.
- M. Shahiduzzaman, M. Karakawa, K. Yamamoto, T. Kusumi, K. Yonezawa, T. Kuwabara, K. Takahashi and T. Taima, *Sol. Energy Mater. Sol. Cells*, 2018, **178**, 1.
- J. Cao, B. Wu, R. Chen, Y. Wu, Y. Hui, B.-W. Mao and N. Zheng, *Adv. Mater.*, 2018, 30.
- X. Liu, H. Yu, L. Yan, Q. Dong, Q. Wan, Y. Zhou, B. Song and Y. Li, *ACS Appl. Mater. Interfaces*, 2015, **7**, 6230.
- S. Cong, H. Yang, Y. Lou, L. Han, Q. Yi, H. Wang, Y. Sun and G. Zou, *ACS Appl. Mater. Interfaces*, 2017, **9**, 2295.
- J. Seo, S. Park, Y. C. Kim, N. J. Jeon, J. H. Noh, S. C. Yoon and S. I. Seok, *Energy Environ. Sci.*, 2014, **7**, 2642.
- H. Yoon, S. M. Kang, J. K. Lee and M. Choi, *Energy Environ. Sci.*, 2016, **9**, 2262.
- W. K. Lin, S. H. Su, M. C. Yeh, C. Y. Chen and M. Yokoyama, *Vacuum*, 2017, **140**, 82.
- Q. Wang, Y. Shao, Q. Dong, Z. Xiao, Y. Yuan and J. Huang, *Energy Environ. Sci.*, 2014, **7**, 2359.
- G. Wang, D. B. Liu, J. Xiang, D. C. Zhou, K. Alameh, B. F. Ding and Q. L. Song, *RSC Adv.*, 2016, **6**, 43299.
- C. Li, F. Wang, J. Xu, J. Yao, B. Zhang, C. Zhang, M. Xiao, S. Dai, Y. Li and Z. Tan, *Nanoscale*, 2015, **7**, 9771.



- 24 Q. Xue, Z. Hu, J. Liu, J. Lin, C. Sun, Z. Chen, C. Duan, J. Wang, C. Liao and W. M. Lau, *J. Mater. Chem. A*, 2014, **2**, 19598.
- 25 C. Chen, S. Zhang, S. Wu, W. Zhang, H. Zhu, Z. Xiong, Y. Zhang and W. Chen, *RSC Adv.*, 2017, **7**, 35819.
- 26 J. Liu, G. Wang, K. Luo, X. He, Q. Ye, C. Liao and J. Mei, *ChemPhysChem*, 2017, **18**, 617–625.
- 27 J. Ciro, S. Mesa, J. I. Uribe, M. A. M. Escobar, D. Ramirez, J. F. Montoya, R. Betancur, H. S. Yoo, N. G. Park and F. Jaramillo, *Nanoscale*, 2017, **9**, 9440.
- 28 H. Zhang, H. Azimi, Y. Hou, T. Ameri, T. Przybilla, E. Spiecker, M. Kraft, U. Scherf and C. J. Brabec, *Chem. Mater.*, 2014, **26**, 5190.
- 29 Q. L. Song, F. Y. Li, H. Yang, H. R. Wu, X. Z. Wang, W. Zhou, J. M. Zhao, X. M. Ding, C. H. Huang and X. Y. Hou, *Chem. Phys. Lett.*, 2005, **416**, 42.
- 30 W. Chen, Y. Wu, Y. Yue, J. Liu, W. Zhang, X. Yang, H. Chen, E. Bi, I. Ashraful and M. Grätzel, *Science*, 2015, **350**, 944.
- 31 J. H. Huang, L. J. Chen, F. He, T. Zhang, G. X. Niu, J. Pan, Z. H. Xiong and Q. L. Song, *J. Phys. D: Appl. Phys.*, 2012, **45**, 195101.
- 32 X. S. Zhao, J. Xiang, D. B. Liu, D. C. Zhou, G. Wang, G. D. Zhou, K. Alameh, B. F. Ding and Q. L. Song, *Org. Electron.*, 2016, **37**, 358.
- 33 Q. Xue, Y. Bai, M. Liu, R. Xia, Z. Hu, Z. Chen, X. F. Jiang, F. Huang, S. Yang and Y. Matsuo, *Adv. Energy Mater.*, 2017, **7**, 1602333.

