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Highly crystalline Nb-doped TiO₂ nanospindles as superior electron transporting materials for high-performance planar structured perovskite solar cells†

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Planar-structured perovskite solar cells (PSCs) have received tremendous attention due to their high power conversion efficiency (PCE), simple process and low-cost fabrication. A compact thin film of electron transport materials (ETMs) plays a key role in these PSCs. However, the traditional ETMs of PSCs, TiO₂ nanoparticulate films, suffer from low conductivity and high trap state density. Herein, we exploited TiO₂ nanospindles as a compact ETM in planar PSCs for the first time, and achieved an efficient device with a PCE of 19.1%. By optimization with Nb doping into the TiO₂ nanospindles, the PCE of the PSC was further improved up to 20.8%. The carrier transfer and collection efficiency were significantly improved after Nb⁵⁺ doping, revealed by Mott–Schottky (MS) analysis, space charge limited current (SCLC), photoluminescence (PL), time-resolved photoluminescence (TRPL) spectra, electrochemical impedance spectra (EIS) and so forth. Moreover, the hysteresis behavior was effectively inhibited and the stability was significantly enhanced. This work may provide a new avenue towards the rational design of efficient ETMs for perovskite solar cells.

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Introduction

Over the past few years, organic–inorganic lead halide perovskite solar cells (PSCs) have attracted enormous interest because of their excellent power-conversion efficiencies (PCEs), along with facile fabrication procedures and low material cost.^{1–3} Within 9 years, the certified PCE of PSCs has been rapidly promoted from 3.8% to 22.7%,^{4,5} demonstrating enormous potential for future applications. Generally, depending on whether a mesoporous charge transport layer is present or not, the device architectures of PSCs are classified as mesoscopic and planar type. To date, most of the certified high-efficiency PSCs are based on mesoscopic structures, in which a layer of mesoporous metal oxide (such as TiO₂, Al₂O₃, ZrO₂, *etc.*) can act

as a scaffold for perovskite to provide a larger interface contact area. However, due to the easier process and lower cost for large-scale fabrication, the planar PSCs received increasing attention and witnessed a rapid development recently.^{6,7}

Without a mesoporous scaffold layer, a compact layer of electron transport materials (ETMs) plays a key role in planar structured PSCs to achieve high PV performance.⁸ The ETMs can extract and transport the photo-generated electrons from perovskite to FTO substrate, and they also can suppress the front-surface recombination. Thus, the crystal phase, morphology, microstructure, conduction band and conductivity of ETM compact layer can significantly affect the PCE of resultant PSCs. To date, various n-typed metal oxide semiconductors, such as TiO₂, SnO₂, Nb₂O₅, and ZnO,^{9–15} have been used as ETMs of PSCs. Among them, TiO₂ is the most widely used one.

However, TiO₂ film has numerous trap states and suffers from low conductivity and carrier accumulation.¹⁶ Further investigation indicates that, metal doping can effectively increase the electrical conductivity of TiO₂, and modify the band structure and trap states. For instance, by using yttrium¹⁷ doped TiO₂ as compact layer, the charge extraction process was improved and the planar PSC device achieved a high PCE of 19.3%. Up to now, magnesium,¹⁸ lithium or sodium,^{19–22} samarium,²³ lanthanum,²⁴ tantalum,²⁵ fluorine,²⁶ thallium,²⁷ and other metal^{28–30} doped TiO₂ have been attempted in perovskite solar cells. Niobium-doped TiO₂ (Nb:TiO₂) has been

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tremendous used in transparent conductive oxides (TCOs),³¹ photocatalysis,^{32,33} lithium and sodium ion batteries,^{34,35} dye-sensitized solar cells (DSSCs).^{36–42} However, limited work on Nb:TiO₂ has been performed in planar PSCs to date. In 2014, Yang *et al.*⁴³ synthesized Nb:TiO₂ nanorods by adding NbCl₅ in the growth process. The Nb:TiO₂ nanorods based PSCs had lower series resistance and higher recombination resistance than the undoped ones, and showed a 50% increase in the efficiency (from 4.9% to 7.5%). In 2016, Chen *et al.* spin-coated a colloid solution containing NbCl₅ and titanium(IV) butoxide (TBT) to prepare a 2% Nb:TiO₂ film for planar PSCs,⁴⁴ yielding PCEs up to 16.3%, as a result of the positive shift of flat band potential and the increased conductivity. Similarly, Yin and co-workers reported⁴⁵ 1% Nb:TiO₂ thin films *via* a chemical bath deposition process in a solution containing NbCl₅ and titanium chloride, yielding a high PCE of 19.2% under reverse scan and improved stability. However, this kind of amorphous Nb:TiO₂ thin film based PSCs showed large *I*–*V* hysteresis, for which the PCE under forward scan is only 16.21%. The large *I*–*V* hysteresis in the above studies makes it difficult to correctly evaluate the photovoltaics performance of the PSCs.

Herein, we exploited TiO₂ nanospindles as the compact ETM in planar PSCs for the first time. Nb⁵⁺ was doped in TiO₂ nanospindles with a mild and facile one-step solvothermal method. The doping of Nb⁵⁺ in TiO₂ was confirmed by X-ray powder diffraction (XRD), and X-ray photoelectron spectroscopy (XPS). By optimizing the concentration of Nb⁵⁺ ions in the precursor solution, the conductivity and carrier density of the resultant TiO₂ compact films can be improved, thus the performance of PSCs can be enhanced. Specifically, the 3% Nb:TiO₂ based PSCs can obtain a highest PCE of 20.8% with a slight hysteresis, which is the highest one for Nb:TiO₂ based planar PSCs. Further characterizations such as Mott–Schottky (MS) analysis, space charge limited current (SCLC), photoluminescence (PL), time-resolved photoluminescence (TRPL) spectra, electrochemical impedance spectra (EIS) were carried out to reveal the mechanism behind the improved PV performance caused by Nb⁵⁺ doping in detail.

Experimental

Chemicals

The main text of the article should appear here with headings as appropriate. All of the solvents and reagents were used without post-treatment. Acetic acid (Sinopharm Chemical Reagent CO., Ltd), tetra-*n*-butyl titanate (Kermel), niobium ethoxide (thermofischer scientific, 99.95%) were used as the precursors to synthesis pristine and Nb:TiO₂ and terpineol (Aladdin, 95%) and ethanol (Aladdin, 99.8%) were employed to disperse them. FAI, MABr, MACl, and spiro-MeOTAD are all purchased from Xi'an Polymer Light Technology Corp.

Synthesis of TiO₂ and Nb:TiO₂ colloid solution precursors

The main text of the article should appear here with headings as appropriate. The Nb:TiO₂ colloid solution was prepared according to our previous report.⁴⁶ In a typical synthesis, 5 mL

tetra-*n*-butyl titanate was added into 25 mL acetic acid under continuous stirring, with adding proper content niobium ethoxide into the solution according to different molar ratio (0%, 1%, 3%, 5%). Pouring into a 50 mL Teflon-lined stainless steel autoclave after the mixture solution became transparent. The solvothermal reaction was conducted at 200 °C for 24 h. The pristine and Nb:TiO₂ colloids were obtained by centrifuged and washed extensively of the precipitate in the autoclave, and stored in absolute ethanol for further use.

Fabrication of perovskite solar cells

The main text of the article should appear here with headings as appropriate. FTO glasses were cleaned in ultrasonic bath of water, ethanol, acetone and 2-propanol in sequence, and treated in an O₂-plasma cleaner. To prepare TiO₂ compact layers (bl-TiO₂), as-made pristine and Nb⁵⁺-doped TiO₂ colloids was mixed with terpineol, and deposited on the bl-TiO₂ by spin-coating at 6000 rpm, followed by annealing at 520 °C for 1 h. The thickness of the compact layers could be adjusted by the volume of added terpineol and the spin-parameters such as spin and accelerate speed. After the post-treatment with 40 mM aqueous TiCl₄ solution at 70 °C for 40 min and sintering at 520 °C for 30 min, the perovskite films were deposited on the treated TiO₂ samples by a two-step spin-coating process. First, 1.3 M of PbI₂ in (DMF : DMSO = 9.5 : 0.5 v/v) was spin coated onto TiO₂ at 1500 rpm for 30 s, and then annealed at 70 °C for 1 min, after PbI₂ had cooled down to room temperature, the mixture solution of FAI : MABr : MACl (60 mg : 6 mg : 6 mg in 1 mL IPA) was spin coated onto the PbI₂ at 1350 rpm for 30 s, and a thermal annealing of 150 °C for 15 min in ambient air condition (30–40% humidity) was processed. The HTM was then deposited by spin coating at 5000 rpm for 20 s. The spin-coating formulation was prepared by dissolving 72.3 mg spiro-MeOTAD, 28.8 mL 4-*tert*-butylpyridine, 17.5 mL of a stock solution of 520 mg mL^{−1} Li-TFSI in acetonitrile and 29 mL of a stock solution of 300 mg mL^{−1} cobalt(III) salt (FK209) in acetonitrile in 1 mL chlorobenzene (CB). Finally, 80 nm of gold top electrode was thermally evaporated under high vacuum.

Characterizations

The crystalline of synthesized pristine and Nb:TiO₂ nanoparticles were examined by XRD measurement performed on a D8 X-ray diffractometer (X' pert Pro-1), employing Cu K_α as incident radiation. The crystal structure and morphology of the products were characterized by transmission electron microscope (TEM; JEOL, JEM-2100, 200 kV) equipped with an EDS spectrometer (Oxford). Surface and sectional morphologies were obtained by scanning electron microscopy, operated at an accelerating voltage of 5 kV (Hitachi S5200). UV-Vis absorption spectra of all samples were carried out on an Evolution™ 201 spectrophotometer (Thermo fisher scientific Corporation). Photovoltaic performance was measured using a Keithley digital source meter (Model 2400), under illumination of a simulated sunlight (AM 1.5, 100 mW cm^{−2}) provided by a solar simulator (Enlitech) with an AM 1.5 filter. A metal aperture of 0.09 cm² was used during the measurement to define the active



area of the device and avoid light scattering through the sides. The incident photon-to-current efficiency (IPCE or EQE) was measured in AC mode on a QE-R3011 testing system (Enlitech) with a tungsten-halogen lamp as light source. The monochromatic light intensity for EQE efficiency was calibrated with a reference silicon photodiode. The PL measurement was conducted with time-correlated single photon counting (TCSPC) with a 508 nm laser (DD-510L, Deltaflex, Horiba). Tafel and J-V plots of the electron-only devices were characterized by Keithley 2400 Source in dark. The SCLC, EIS and Mott-Schottky plot were obtained by using a multi-channel potentiostat (VMP3, Bio-logic) under dark conditions. For the Mott-Schottky analysis, the electrodes were submerged in a 0.1 M KCl aqueous solution, with a Pt counter electrode and Ag/AgCl reference electrode. The values were recorded at frequency of 80 mHz. EIS datas were recorded at 0 and 0.8 V in the frequency range from 1 MHz to 100 mHz with an AC amplitude of 30 mV. The Mott-Schottky datas were recorded at the frequency of 80 mHz in the applied voltage range from -1.0 V to 0 V with an AC amplitude of 30 mV.

Results and discussion

Fig. 1a shows the X-ray diffraction (XRD) patterns of the pristine and Nb:TiO₂ with different doping levels. All diffraction peaks of the samples can be assigned to the anatase phase (PDF: 01-084-1286). Fig. S1† presents the Raman spectra of the pristine and 3% Nb:TiO₂, which are almost the same. The peaks at 144, 397, 515 and 640 cm⁻¹ are clearly observed, matching well with the modes of anatase TiO₂, which further confirms that the anatase nanocrystalline structure is retained after Nb⁵⁺ doping. In addition, the (101) anatase peaks (Fig. 1b) show a slight shift to a lower diffraction angle with increasing niobium concentration because of the larger radius of Nb⁵⁺ (0.64 Å) compared to that of Ti⁴⁺ (0.61 Å). Furthermore, a gradually and slightly broadened FWHM (full width at half maximum) of XRD in Fig. 1b suggest that the crystallite sizes decrease with increased

the niobium contents, which can be inferred from the Debye-Scherrer equation.⁴⁰

To confirm the inclusion of Nb element in TiO₂ nano-spindles, the energy-dispersive spectroscopy (EDS) and X-ray photoelectron spectroscopy (XPS) analysis were carried out. The quantitative analysis of Nb⁵⁺ contents by EDS is almost in agreement with the theoretical values, as shown in Fig. S2.† The surface chemical composition and the chemical states of the sample with 3% Nb:TiO₂ were analyzed by XPS as presented in Fig. 1c and d. The characteristic peaks of Ti⁴⁺ 3d_{5/2} and Ti⁴⁺ 3d_{3/2} (464.4 eV and 458.7 eV) were similar in pristine and 3% Nb:TiO₂ without any shift, indicating that no Ti³⁺ and Ti²⁺ existed in both samples. Two peaks with binding energies of 209.8 and 207.0 eV are assigned to the 3d_{5/2} and 3d_{3/2} of Nb⁵⁺. The atomic concentration of Nb in the doped sample was estimated to be 2.1 atom% by XPS, which is a little lower than that from the EDS. This may be due to the fact that XPS is a surface-sensitive analysis tool and can only detect the informations in 1–2 nm of the surface. Combined with the results of EDS and XPS, Nb element was successfully doped into the TiO₂ nanostructures.

The morphologies of the pristine TiO₂ and the 3% Nb:TiO₂ TiO₂ samples are shown in Fig. 2(a–d). All samples exhibited uniform and narrow size distribution (Fig. S4†). Additionally, the particle size decreases with increasing the amount of Nb precursor, which is consistent with the slightly broader FWHM of XRD in Fig. 1b. The pristine TiO₂ is spindle-shaped (Fig. 1a and c), and doping ions with larger radius (Nb⁵⁺) changes the aspect ratio by reduction along the *c*-axis.⁴⁷ The mean sizes and shapes of the Nb:TiO₂ spindles changed slightly in comparison with that of the undoped one when the doping concentration varied from 1% (Fig. S5a and b†) to 3% (Fig. 2b and d). While for the one, a significant morphological change happens for 5% Nb:TiO₂ system (Fig. S5c and d†). HRTEM in Fig. 2c and d demonstrate the highly crystalline nature for both samples.

A thin layer of 35 nm Nb:TiO₂ film was used as ETLs for the n-i-p PSCs with a planar structure (FTO/bl-Nb:TiO₂/FAPbI₃)_x(MAPbBr₃)_{1-x}/spiro-OMeTAD/Au) (Fig. 3a). Fig. S6†

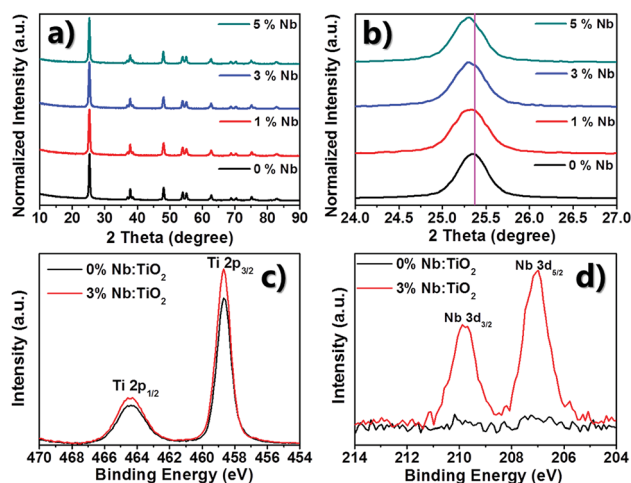


Fig. 1 (a) XRD patterns of the pristine TiO₂ and Nb:TiO₂; (b) the magnified XRD patterns for the (101) plane; (c and d) XPS spectra of TiO₂ and 3% Nb⁵⁺-doped for Ti and Nb states respectively.

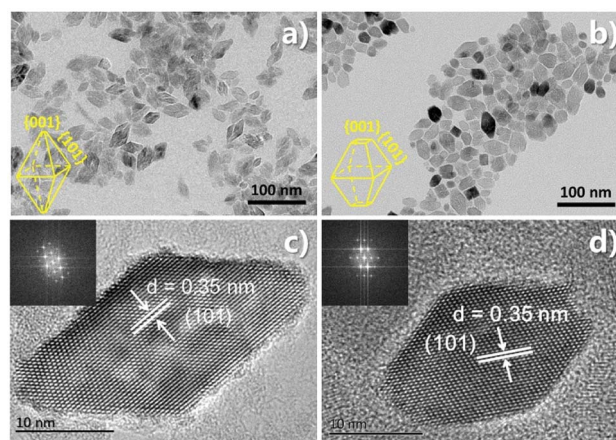


Fig. 2 TEM (a and b) images and HRTEM (c and d) images of the pristine TiO₂ nanoparticles (left) and 3% Nb:TiO₂ nanoparticles (right), respectively.



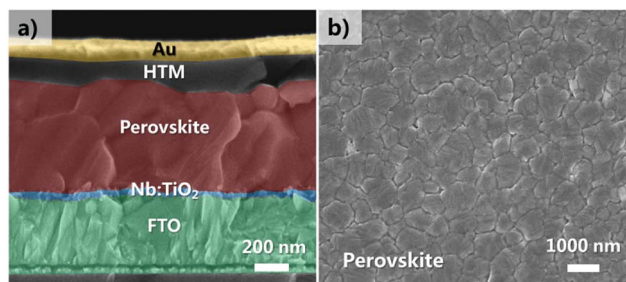


Fig. 3 (a) Cross-sectional SEM image for the Nb:TiO₂ based PSC, (b) plane-view SEM image for the (FAPbI₃)_x(MAPbBr₃)_{1-x} perovskite film.

shows the atomic force microscope (AFM) images for the bare FTO, pristine and 3% Nb:TiO₂ thin films deposited on the FTO substrate. Both of them show dense and compact grains in all the thin films without significant voids. The root mean square (RMS) roughness value of FTO is 10.6 nm, which gradually decreases to 8.53 and 8.23 nm after depositing pristine and Nb:TiO₂ nanocrystals, respectively. With coating a layer of TiO₂ film, the substrate become much more uniform and smoother, indicating that the TiO₂ layers have been successfully coated on the FTO, which is in favor for the deposition of uniform perovskite films. A two-step spin coating method was used to deposit the perovskite layer according to a reported process incorporated with modification.⁴⁸ The obtained perovskite film was showed in Fig. 3b. The grain size ranging from 300 to 1500 nm was observed, together with dense film morphology and a full surface coverage. The film thickness of the perovskite is ~600 nm (Fig. 3a), which was comparable to the average grain size from Fig. 3b. As a result, most of the photoinduced electrons could be transported across the perovskites layer toward the electrode without crossing the grain boundaries, which is favorable for reducing the recombination loss.

Photovoltaic properties of undoped and doped TiO₂-based PSCs (with a 0.09 cm² aperture) were measured under AM 1.5G conditions (100 mW cm⁻²). Fig. S7† shows the device performance optimized as a function of Nb doping level; Table S1† displays the corresponding average photovoltaic parameters. It's noteworthy that the PCE increases from 18.6% to 20.4% when the Nb content is changed from 0 to 3%. But the PCE drops to 18.8% when the doping content of Nb in TiO₂ is further increased to 5%. It is apparent that there is an optimum Nb content for the device performance at 3%. Fig. 4a shows the *J*-*V* curves of the champion devices based on both pristine TiO₂ and 3% Nb:TiO₂ ETMs. The undoped TiO₂ based cells exhibited reverse scanning PCEs of 19.1% (with *J*_{SC} = 23.2 mA cm⁻², *V*_{OC} = 1.07 V, and FF = 77.0%), and forward scanning PCE of 18.2% (with *J*_{SC} = 23.2 mA cm⁻², *V*_{OC} = 1.04 V, and FF = 75.3%), showing an obvious hysteresis behavior. In contrast, the best device based on the 3% Nb:TiO₂ compact layer present a reverse scanning PCE of 20.8% (with *J*_{SC} = 23.6 mA cm⁻², *V*_{OC} = 1.13 V, and FF = 78.4%), and a forward scanning PCE of 20.2% (with *J*_{SC} = 23.5 mA cm⁻², *V*_{OC} = 1.10 V, and FF = 77.8%). The efficiencies are much higher than the pristine counterparts under both

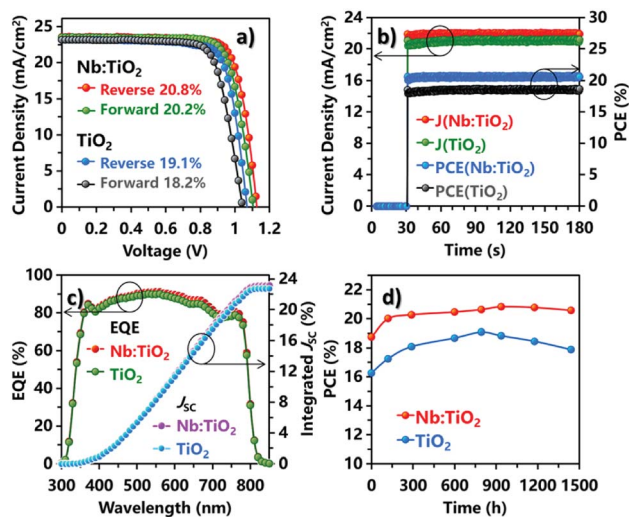


Fig. 4 Photovoltaic performance for the PSCs: (a) *J*-*V* curves, (b) stabilized current densities and PCEs at maximum power points, (c) external quantum efficiencies (EQEs) (d) stability of unencapsulated PSCs devices with TiO₂ and 3% Nb-doped TiO₂ as ETMs under air exposure (humidity: 35%, stored in dark) for different time periods.

scan directions. Moreover, the hysteresis was apparently suppressed due to the Nb⁵⁺ doping. Similar results have been reported previously,⁴⁵ and will be discussed later.

Additionally, to study the power conversion efficiency under real-world working conditions, we have measured the steady output power for both devices under constant voltage bias at the maximum power-points (0.94 V for Nb:TiO₂ and 0.88 V for the undoped TiO₂) as plotted in Fig. 4b. It can be seen that the output current density was around 21.9 mA cm⁻² for the Nb:TiO₂ spindles-based PSCs at steady state after ~150 s, resulting in a steady output efficiency of 20.6%, which were much better than that of the undoped TiO₂ based device (with output current density of 21.2 mA cm⁻² and efficiency of 18.7%). To the best of our knowledge, the PCE of 20.6% represents the highest efficiency among the Nb:TiO₂ based PSCs reported so far. More data about the optimization of Nb⁵⁺ content from 0% to 5% were shown in details in Fig. S7 and S8.† Furthermore, we have also optimized the thicknesses of the Nb:TiO₂ films by changing the concentration or the spinning coating parameters. Too thin Nb:TiO₂ films (≤35 nm) shows low surface coverage, and is not suitable for use as ETMs of PSCs. The cross-sectional SEM images of the 50 and 80 nm Nb:TiO₂ films based PSCs were shown in Fig. S9a and b.† With the increasing of the film thickness, the *V*_{OC} and FF decreased obviously as a result of the enhanced series resistance and recombination, leading to a decrease of the PCEs as shown in Fig. S9c and d.†

The typical external quantum efficiencies (EQEs) of the devices made from pristine and Nb⁵⁺ doped TiO₂ compact layers are comparatively shown in Fig. 4c. The photo response edges for both devices locate at ~820 nm with almost the same shape. In the visible region, the EQEs can reach as high as 91.1% and 90.0% at 560 nm wavelength for the Nb:TiO₂ based device and the undoped TiO₂ based devices, respectively. The



integrated short-circuit current of 23.2 mA cm^{-2} and 22.8 mA cm^{-2} matched with the corresponding J_{SC} obtained from the J - V results (in Fig. 4a). The higher EQE and larger J_{SC} of Nb:TiO₂ based-device should be ascribed to the higher conductivity caused by Nb⁵⁺ doping considering the same UV-vis absorption spectra for the pristine and Nb:TiO₂ films (in Fig. S10†), as well as the basically unchanged band gaps in both cases.

The long-term stability of the Nb:TiO₂ based PSCs without encapsulation was further investigated, as shown in Fig. 4d for the testing under ambient conditions without encapsulation. At the beginning, PCEs of undoped TiO₂ formed devices increased steadily with time and reached a peak value at about 800 h storage, then started to decline obviously with prolonging the storage time. In contrast, the PCE of the Nb:TiO₂ based PSCs only increased drastically in the first 100 h, and then saturated and kept basically unchanged for the rest of the testing time (more than 1400 h). It is reported that a critical instability of PSCs arose from the UV light-induced desorption of surface-adsorbed oxygen from TiO₂, which may introduce more deep surface traps.^{49,50} The enhanced stability of the present Nb:TiO₂ devices may be due to the reduced trap state density and improved charge transferring at Nb:TiO₂/perovskite interface, thereby suppressing the charge accumulation and recombination within the devices.⁴⁵

To further understand the origin of the improvement in the device performance in the present experiments, Mott-Schottky analysis, Tafel curves, space-charge-limited current (SCLC) model analysis, EIS and PL spectra were conducted. Mott-Schottky analysis was employed to reveal the changes of electronic properties of TiO₂ compact layers in this work. As shown in Fig. 5a, both curves have positive slopes, indicating that both TiO₂ and Nb:TiO₂ are n-type.⁵¹ The slope of the fitting linear part (blue line) in the Mott-Schottky curves can be used to deduce the carrier density (N_e)^{51,52} using eqn (1):⁵³

$$\frac{1}{C^2} = \frac{2}{\epsilon_0 \epsilon A^2 e N_e} \left(V - V_{\text{fb}} - \frac{k_B}{e} T \right) \quad (1)$$

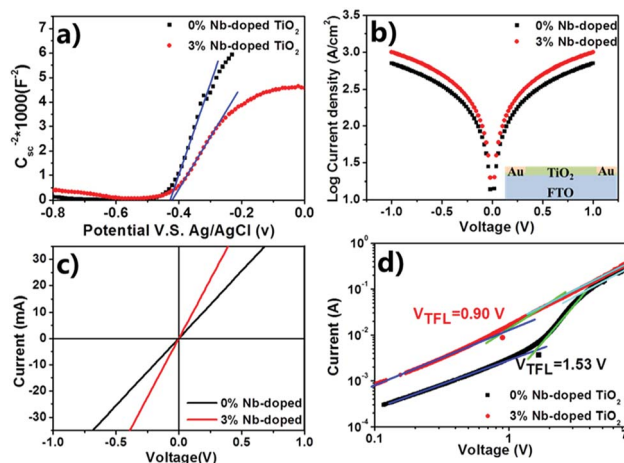


Fig. 5 (a) Mott-Schottky plots of TiO₂ and Nb:TiO₂; (b) Tafel plot of the FTO/TiO₂(Nb:TiO₂)/Au devices (inset) and (c) the corresponding dark current-voltage (I - V) characteristics; (d) I - V plots for TiO₂ and Nb:TiO₂ films using the space-charge-limited current (SCLC) model with device structure of Ag/TiO₂(Nb:TiO₂)/Ag.

$$\text{Slop} = \frac{2}{\epsilon_0 \epsilon A^2 e N_e} \quad (2)$$

where C is the interfacial capacitance, ϵ_0 is vacuum permittivity ($8.85 \times 10^{-12} \text{ F m}^{-1}$), ϵ is the relative dielectric constant of anatase TiO₂ ($\epsilon = 55$ (ref. 19)), A is the area of interface or the electrode (0.2826 cm^2), e is the elementary charge ($e = 1.6 \times 10^{-19} \text{ C}$), V is the applied voltage (V), k_B is Boltzmann's constant, and T is the absolute temperature (K), and N_e is the carrier density. The resulting N_e value of Nb:TiO₂ is $3.8 \times 10^{20} \text{ cm}^{-3}$, which is about twice more than that of the undoped one ($1.9 \times 10^{20} \text{ cm}^{-3}$). The conductivity of the metal oxide (σ) is given by eqn (3):

$$\sigma = N_e \mu_e \quad (3)$$

where μ_e is the mobility of the electrons. When the mobility keeps unchanged, an increase of N_e will result in a higher conductivity. Therefore, larger N_e should have a positive effect on J_{SC} of the resultant devices.

As well-known, the band alignment,⁵⁴ or the difference between the LUMO of perovskite and the quasi-Fermi level (E_{Fermi}) of the electron transport layer (ETL),⁵⁵ have great impact on the V_{OC} of the resultant perovskite solar cells. The quasi-Fermi level (E_{Fermi}) of ETL is equal to the voltage in the flat-band state V_{fb} , which is the external voltage when no band bending occurs and can be deduced from the intersection of the fitting linear part of $M - S$ plot.⁵¹ A more positive V_{fb} means a lower E_{Fermi} of TiO₂, which is benefit for the injection of photoelectrons between the FTO and ETL. Fig. 4a shows that upon Nb⁵⁺ doping, V_{fb} of TiO₂ shifted from -466.3 mV to -446.3 mV . The positive shift has also been found in Nb⁵⁺-doped rutile nanorod.⁵⁶ The increase of 20 mV of V_{fb} may contribute to the increased V_{OC} in the devices.^{52,56} Moreover, in the $M - S$ analysis, the depletion zone can be regarded as a bulk resistance from the FTO to the films, and expressed with depletion zone width (W).^{53,57,58} The narrower the W is, the less the contact resistance is. Similar scenario occurs for the charge extraction at adjacent interfaces of ETMs. The value of W can be obtained from the eqn (4):

$$W = \left(\frac{2\epsilon\epsilon_0}{eN_e} V - V_{\text{fb}} \right)^{1/2} \quad (4)$$

Table S2† indicated that Nb:TiO₂ has the narrower W of only 2.67 nm . Therefore, doping of Nb⁵⁺ not only have positive effect on the conductivity, but also help improve the charge extraction ability from perovskite layer to TiO₂ film.

Meanwhile, the improvement in electrical conductivity of TiO₂ could be observed from Tafel and the corresponding J - V plot^{19,44,59,60} (shown in Fig. 5b and c). The direct conductivity could be estimated from the eqn (5):

$$I = \sigma_0 A L^{-1} V \quad (5)$$

where A is the area (0.09 cm^2) of the devices and L is the thickness ($\sim 35 \text{ nm}$) of film; the obtained values of the



conductivity are summarized in Table S2.† It can be found that the conductivity of the 3% Nb:TiO₂ is threefold than that of the pristine TiO₂, in agreement with the increase of carrier density (N_e). Hence, Nb⁵⁺ doping in TiO₂ is an effective way to improve the conductivity of TiO₂ film, which have a positive effect on improving J_{SC} of the resulting PSCs.

In order to determine the influence of Nb⁵⁺ doping on the electron trap state density in the TiO₂ films, we have carried out the dark current–voltage (I – V) analysis using the space-charge-limited current (SCLC) model, as shown in Fig. 5d. At low bias voltage section, the linear relation (blue line) means an ohmic response of the device structure. When the bias voltage exceeded a kink point, the current quickly increased nonlinearly (green line), indicating that the trap-states were completely filled. The value of the kink point, named as trap-filled limit voltage (V_{TFL}), can be calculated according to eqn (6):^{61,62}

$$V_{TFL} = \frac{en_i L^2}{2\epsilon\epsilon_0} \quad (6)$$

where e is the elementary charge of the electron ($e = 1.6 \times 10^{-19}$ C), L is the thickness of the pristine TiO₂ and 3% Nb:TiO₂ film (~ 35 nm), ϵ is the relative dielectric constant of anatase TiO₂ ($\epsilon = 55$ F m⁻¹ (ref. 19)), ϵ_0 is the vacuum permittivity (8.85×10^{-12} F m⁻¹), and n_i is the trap-state density. The V_{TFL} of devices based on the pristine TiO₂ (1.53 V) is higher than that of 3% Nb:TiO₂ (0.9 V). Therefore, the electron traps density in the pristine TiO₂ film (7.60×10^{18} cm⁻³) is higher than that of 3% Nb:TiO₂ (4.47×10^{18} cm⁻³). Clearly, doping of Nb⁵⁺ effectively passivates electron traps, and improve the transport and extraction capability of photogenerated electrons, leading to higher electron mobility, a reduced J – V hysteresis and an enhanced efficiency.

Steady-state photoluminescence (PL) (Fig. 6a) showed a more significant PL quench when depositing the perovskite layer on Nb:TiO₂ films than on the undoped TiO₂ samples, which could be ascribed to the higher electron mobility of doped TiO₂, and lower electron barrier at the interface between TiO₂ and

perovskite. In addition, the time-resolved photoluminescence (TRPL) spectra of the perovskite films on glass, TiO₂ and Nb:TiO₂ was displayed in Fig. 6b. The PL decay time and amplitudes are modeled using a bi-exponential eqn (7):

$$f(t) = \sum_i A_i \exp(-t/\tau_i) + B \quad (7)$$

where A_i is the decay amplitude, τ_i is the decay time and B is a constant for the base-line offset. All above mentioned parameters obtained from the analysis are listed in Table S3.† The average PL decay times (τ_{ave}) were estimated using the τ_i and A_i values from the fitted curve data (Table S3†) according to eqn (8):

$$\tau_{ave} = \frac{\sum A_i \tau_i^2}{\sum A_i \tau_i} \quad (8)$$

The τ_{ave} of glass/perovskite is 453.7 ns. The average PL decay time for TiO₂/perovskite and Nb:TiO₂/perovskite were 347.5 and 266.7 ns respectively, indicating that charge transfer in Nb:TiO₂/perovskite system was the most efficient. The enhanced charge transfer in Nb:TiO₂/perovskite demonstrated that the energy barrier had been reduced, which could be caused by the high mobility and low trap density of the Nb:TiO₂ film, resulting in the superior PV performance for the Nb:TiO₂-formed PSCs, as discussed in above part (Fig. 2c and d).

To further gain insight into the charge transport process in the devices, electrochemical impedance spectroscopy (EIS) measurements were carried out. Fig. 6c shows an incomplete semicircle in high frequency range, which is assigned to the electron transfer process (R_{tr}), whereas the main semicircle in the low frequency range is attributed the recombination process (R_{rec}). The value of series resistance (R_s) is equal to the intercept on the real axis at high frequency range, which is 31.06 and 33.33 Ω for the device for the pristine and 3% Nb:TiO₂ ETMs, respectively. Meanwhile, the 3% Nb:TiO₂ device possesses a larger semicircle in the low frequency and a smaller semicircle in the high frequency region. The recombination resistance R_{rec} is associated with the interface charge transport, which increased from $2.08 \times 10^6 \Omega$ for the undoped device to $2.51 \times 10^6 \Omega$ for the 3% Nb-doped device. This clearly confirms the effect of Nb doping on reducing the recombination losses in the devices with Nb-doped TiO₂ film. These results confirm that the cell with 3% Nb:TiO₂ as ETMs exhibits faster electron transfer and slower recombination.

Conclusions

In summary, Nb⁵⁺-doped TiO₂ (Nb:TiO₂) nanocrystals synthesized *via* a mild and facile one-step solvothermal method have been successfully used as efficient ETMs in planar PSCs. Doping of Nb⁵⁺ increases the conductivity and quasi-Fermi level, eliminates the trap state density, and thus leads to an obvious increase in J_{SC} , V_{OC} and FF of the PSC device. Nb:TiO₂ based PSCs exhibited a stabilized PCE of 20.6%, superior to that of the control cell (18.7%), which is among the best level of the Nb:TiO₂ formed planar PSCs. Additionally, the Nb:TiO₂ based

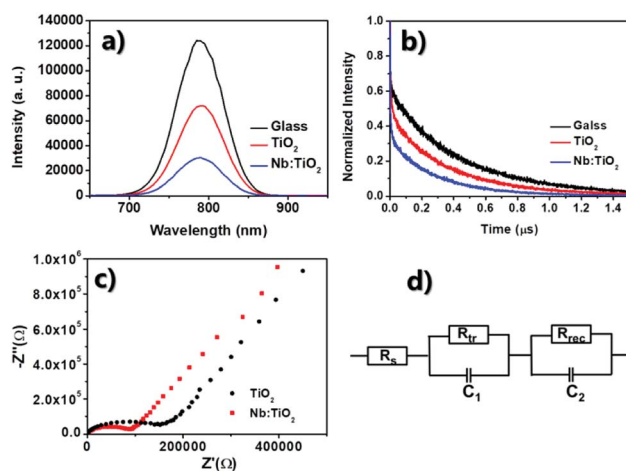


Fig. 6 (a) Steady-state photoluminescence spectrum and (b) time-resolved photoluminescence (TRPL) of perovskite deposited on glass (black), undoped TiO₂ (red) and Nb:TiO₂ (blue), respectively; (c) Nyquist plots of electrochemical impedance spectra (EIS) and (d) equivalent circuit for PSCs based on undoped TiO₂ and Nb:TiO₂.



PSCs showed less hysteresis and great long-term stability due to the enhanced electron transport, reduced charge accumulation and trap states. This study demonstrated that Nb:TiO₂ nanocrystals show promise for the practical use as superior ETMs of PSCs.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 H.-S. Kim, C.-R. Lee, J.-H. Im, K.-B. Lee, T. Moehl, A. Marchioro, S.-J. Moon, R. Humphry-Baker, J.-H. Yum, J. E. Moser, M. Graetzel and N.-G. Park, *Sci. Rep.*, 2012, **2**, 591.
- 2 C. Yang, Y. Wu, Q. Ma and W.-H. Zhang, *J. Energy Chem.*, 2018, **27**, 622.
- 3 Y. Wu, P. Wang, S. Wang, Z. Wang, B. Cai, X. Zheng, Y. Chen, N. Yuan, J. Ding and W.-H. Zhang, *ChemSusChem*, 2018, **11**, 837.
- 4 W. S. Yang, B.-W. Park, E. H. Jung, N. J. Jeon, Y. C. Kim, D. U. Lee, S. S. Shin, J. Seo, E. K. Kim, J. H. Noh and S. I. Seok, *Science*, 2017, **356**, 1376.
- 5 B. Cai, Y. Xing, Z. Yang, W.-H. Zhang and J. Qiu, *Energy Environ. Sci.*, 2013, **6**, 1480.
- 6 Z. M. Zhou, X. Li, M. L. Cai, F. X. Xie, Y. Z. Wu, Z. Lan, X. D. Yang, Y. H. Qiang, A. Islam and L. Y. Han, *Adv. Energy Mater.*, 2017, **7**, 1700763.
- 7 Y. Z. Wu, F. X. Xie, H. Chen, X. D. Yang, H. M. Su, M. L. Cai, Z. M. Zhou, T. Noda and L. Y. Han, *Adv. Mater.*, 2017, **29**, 1701073.
- 8 G. Yang, H. Tao, P. Qin, W. Ke and G. Fang, *J. Mater. Chem. A*, 2016, **4**, 3970.
- 9 M. H. Kumar, N. Yantara, S. Dharani, M. Graetzel, S. Mhaisalkar, P. P. Boix and N. Mathews, *Chem. Commun.*, 2013, **49**, 11089.
- 10 Z. Zhu, Y. Bai, X. Liu, C. C. Chueh, S. Yang and A. K. Jen, *Adv. Mater.*, 2016, **28**, 6478.
- 11 P. Zhang, J. Wu, T. Zhang, Y. Wang, D. Liu, H. Chen, L. Ji, C. Liu, W. Ahmad, Z. D. Chen and S. Li, *Adv. Mater.*, 2018, **30**.
- 12 X. Ling, J. Yuan, D. Liu, Y. Wang, Y. Zhang, S. Chen, H. Wu, F. Jin, F. Wu, G. Shi, X. Tang, J. Zheng, S. Liu, Z. Liu and W. Ma, *ACS Appl. Mater. Interfaces*, 2017, **9**, 23181.
- 13 J. Feng, Z. Yang, D. Yang, X. Ren, X. Zhu, Z. Jin, W. Zi, Q. Wei and S. Liu, *Nano Energy*, 2017, **36**, 1.
- 14 X. Ren, D. Yang, Z. Yang, J. Feng, X. Zhu, J. Niu, Y. Liu, W. Zhao and S. F. Liu, *ACS Appl. Mater. Interfaces*, 2017, **9**, 2421.
- 15 D. Yang, R. Yang, J. Zhang, Z. Yang, S. Liu and C. Li, *Energy Environ. Sci.*, 2015, **8**, 3208.
- 16 K. Wojciechowski, M. Saliba, T. Leijtens, A. Abate and H. J. Snaith, *Energy Environ. Sci.*, 2014, **7**, 1142.
- 17 H. Zhou, Q. Chen, G. Li, S. Luo, T.-b. Song, H.-S. Duan, Z. Hong, J. You, Y. Liu and Y. Yang, *Science*, 2014, **345**, 542.
- 18 H. Zhang, J. Shi, X. Xu, L. Zhu, Y. Luo, D. Li and Q. Meng, *J. Mater. Chem. A*, 2016, **4**, 15383.
- 19 J. H. Heo, M. S. You, M. H. Chang, W. Yin, T. K. Ahn, S.-J. Lee, S.-J. Sung, D. H. Kim and S. H. Im, *Nano Energy*, 2015, **15**, 530.
- 20 F. Giordano, A. Abate, J. P. C. Baena, M. Saliba, T. Matsui, S. H. Im, S. M. Zakeeruddin, M. K. Nazeeruddin, A. Hagfeldt and M. Graetzel, *Nat. Commun.*, 2016, **7**, 10379.
- 21 X. Li, J. Yang, Q. Jiang, W. Chu, D. Zhang, Z. Zhou and J. Xin, *ACS Appl. Mater. Interfaces*, 2017, **9**, 41354.
- 22 D. Liu, S. Li, P. Zhang, Y. Wang, R. Zhang, H. Sarvari, F. Wang, J. Wu, Z. Wang and Z. D. Chen, *Nano Energy*, 2017, **31**, 462.
- 23 Y. Xiang, Z. Ma, J. Zhuang, H. Lu, C. Jia, J. Luo, H. Li and X. Cheng, *J. Phys. Chem. C*, 2017, **121**, 20150.
- 24 H. Li, B. Zheng, Y. Xue, S. Liu, C. Gao and X. Liu, *Sol. Energy Mater. Sol. Cells*, 2017, **168**, 85.
- 25 R. Ranjan, A. Prakash, A. Singh, A. Singh, A. Garg and R. K. Gupta, *J. Mater. Chem. A*, 2018, **6**, 1037.
- 26 M. Che, Y. Fang, J. Yuan, Y. Zhu, Q. Liu and J. Song, *Int. J. Electrochem. Sci.*, 2017, **12**, 1064.
- 27 Q. Cui, X. Zhao, H. Lin, L. Yang, H. Chen, Y. Zhang and X. Li, *Nanoscale*, 2017, **9**, 18897.
- 28 C. Liang, P. Li, Y. Zhang, H. Gu, Q. Cai, X. Liu, J. Wang, H. Wen and G. Shao, *J. Power Sources*, 2017, **372**, 235.
- 29 W.-Q. Wu, D. Chen, R. A. Caruso and Y.-B. Cheng, *J. Mater. Chem. A*, 2017, **5**, 10092.
- 30 X. L. Gu, Y. F. Wang, T. Zhang, D. T. Liu, R. Zhang, P. Zhang, J. Wu, Z. D. Chen and S. B. Li, *J. Mater. Chem. C*, 2017, **5**, 10754.
- 31 X. Yang, M.-J. Zhang, Y. Min, M. Xu, Z. Mei, J. Liang, J. Hu, S. Yuan, S. Xiao, Y. Duan, F. Liu, H. Lin, Y. Lin and F. Pan, *ACS Appl. Mater. Interfaces*, 2017, **9**, 29021.
- 32 C. Das, P. Roy, M. Yang, H. Jha and P. Schmuki, *Nanoscale*, 2011, **3**, 3094.
- 33 L. Kong, C. Wang, H. Zheng, X. Zhang and Y. Liu, *J. Phys. Chem. C*, 2015, **119**, 16623.
- 34 D. Bauer, A. J. Roberts, N. Matsumi and J. A. Darr, *Nanotechnology*, 2017, **28**, 195403.
- 35 T. Lan, W. Zhang, N.-L. Wu and M. Wei, *Chem.-Eur. J.*, 2017, **23**, 5059.
- 36 S. Parthiban, K. S. Anuratha, S. Arunprabakaran, S. Abinash and N. Lakshminarasimhan, *Ceram. Int.*, 2015, **41**, 205.
- 37 M. Yang, D. Kim, H. Jha, K. Lee, J. Paul and P. Schmuki, *Chem. Commun.*, 2011, **47**, 2032.
- 38 S. Lee, J. H. Noh, H. S. Han, D. K. Yim, D. H. Kim, J.-K. Lee, J. Y. Kim, H. S. Jung and K. S. Hong, *J. Phys. Chem. C*, 2009, **113**, 6878.
- 39 S. Guo, Y. Horie, S. Imada, M. Z. Bin Mukhlis and T. Nomiyama, *J. Mater. Sci.: Mater. Electron.*, 2017, **28**, 13084.



- 40 Y. Liu, H. Ran, J. Fan, X. Zhang, J. Mao and G. Shao, *Appl. Surf. Sci.*, 2017, **410**, 241.
- 41 X. Lue, X. Mou, J. Wu, D. Zhang, L. Zhang, F. Huang, F. Xu and S. Huang, *Adv. Funct. Mater.*, 2010, **20**, 509.
- 42 W. Liu, H.-g. Wang, X. Wang, M. Zhang and M. Guo, *J. Mater. Chem. C*, 2016, **4**, 11118.
- 43 M. Yang, R. Guo, K. Kadel, Y. Liu, K. O'Shea, R. Bone, X. Wang, J. He and W. Li, *J. Mater. Chem. A*, 2014, **2**, 19616.
- 44 B.-X. Chen, H.-S. Rao, W.-G. Li, Y.-F. Xu, H.-Y. Chen, D.-B. Kuang and C. Su, *J. Mater. Chem. A*, 2016, **4**, 5647.
- 45 G. Yin, J. Ma, H. Jiang, J. Li, D. Yang, F. Gao, J. Zeng, Z. Liu and S. F. Liu, *ACS Appl. Mater. Interfaces*, 2017, **9**, 10752.
- 46 Y. Lv, B. Cai, Y. Wu, S. Wang, Q. Jiang, Q. Ma, J. Liu and W.-H. Zhang, *J. Energy Chem.*, 2018, **27**, 951.
- 47 L. De Trizio, R. Buonsanti, A. M. Schimpf, A. Llordes, D. R. Gamelin, R. Simonutti and D. J. Milliron, *Chem. Mater.*, 2013, **25**, 3383.
- 48 Q. Jiang, Z. Chu, P. Wang, X. Yang, H. Liu, Y. Wang, Z. Yin, J. Wu, X. Zhang and J. You, *Adv. Mater.*, 2017, **29**, 1703852.
- 49 T. Leijtens, G. E. Eperon, S. Pathak, A. Abate, M. M. Lee and H. J. Snaith, *Nat. Commun.*, 2013, **4**, 2885.
- 50 W. Ke, G. Fang, Q. Liu, L. Xiong, P. Qin, H. Tao, J. Wang, H. Lei, B. Li, J. Wan, G. Yang and Y. Yan, *J. Am. Chem. Soc.*, 2015, **137**, 6730.
- 51 L. L. K. Gelderman and S. W. Donne, *J. Chem. Educ.*, 2007, 685.
- 52 B. Roose, C. M. Johansen, K. Dupraz, T. Jaouen, P. Aebi, U. Steiner and A. Abate, *J. Mater. Chem. A*, 2018, **6**, 1850.
- 53 A. G. R. van de Krol and J. Schoonman, *J. Electrochem. Soc.*, 1997, **144**, 1723.
- 54 Q. Guo, Y. Xu, B. Xiao, B. Zhang, E. Zhou, F. Wang, Y. Bai, T. Hayat, A. Alsaedi and Z. a. Tan, *ACS Appl. Mater. Interfaces*, 2017, **9**, 10983.
- 55 J. P. C. Baena, L. Steier, W. Tress, M. Saliba, S. Neutzner, T. Matsui, F. Giordano, T. J. Jacobsson, A. R. S. Kandada, S. M. Zakeeruddin, A. Petrozza, A. Abate, M. K. Nazeeruddin, M. Graetzel and A. Hagfeldt, *Energy Environ. Sci.*, 2015, **8**, 2928.
- 56 M. Yang, B. Ding and J.-K. Lee, *J. Power Sources*, 2014, **245**, 301.
- 57 H. Yu, H. I. Yeom, J. W. Lee, K. Lee, D. Hwang, J. Yun, J. Ryu, J. Lee, S. Bae, S. K. Kim and J. Jang, *Adv. Mater.*, 2018, **30**, 1704825.
- 58 T.-S. Su, T.-Y. Hsieh and T.-C. Wei, *Sol. RRL*, 2018, **2**, 1700120.
- 59 G. Kakavelakis, T. Maksudov, D. Konios, I. Paradisanos, G. Kioseoglou, E. Stratakis and E. Kymakis, *Adv. Energy Mater.*, 2017, **7**, 1602120.
- 60 J. Obrzut and K. A. Page, *Phys. Rev. B*, 2009, **80**, 195211.
- 61 R. H. Bube, *J. Appl. Phys.*, 1962, **33**, 1733.
- 62 D. Yang, X. Zhou, R. Yang, Z. Yang, W. Yu, X. Wang, C. Li, S. Liu and R. P. H. Chang, *Energy Environ. Sci.*, 2016, **9**, 3071.

