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Polydimethylsiloxane/aluminum oxide composites prepared by spatial confining forced network assembly for heat conduction and dissipation

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Constructing a compacted network in polymer matrices is an important method to improve the thermal conductivity (TC) of polymer composites. In this paper, a compacted network was built using the Spatial Confining Forced Network Assembly (SCFNA) method. The homogeneous compound of polymer and fillers, prepared using a conical twin-screw mixer, was placed in a compression mold with confining space to carry out two-stage compression, free compression and spatial confining compression. Aluminum oxide (Al₂O₃) was studied as filler in a polydimethylsiloxane (PDMS) matrix to illustrate the applicability of the SCFNA method. The polymer composites with an Al₂O₃ filler ranging from 10 to 80 wt% were prepared. When the filler content was 80 wt%, the TC of the PDMS/Al₂O₃ composites prepared using the SCFNA method increased by 16.35 times in comparison to the TC of pure PDMS. Observing the SEM of PDMS/Al₂O₃ composites with various thicknesses, the gap between fillers decreased with a decrease in thickness. The composite with TC up to 2.566 W (mK)⁻¹ obtained at 80 wt% filler was further employed as a heat spreader, causing a decrease of about 8.23 °C in the set-point compared with the temperature of the heat source.

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Introduction

Due to the demands of denser and faster circuits in electronic packaging, thermally conductive and electrically insulating composites have attracted great attention in the thermal management field.^{1–4} Heat accumulated in electronic devices can lead to low service life and operation efficiency of electronic systems because the reliability of a device depends on the operating temperature of the junction.⁵ Polymers have often been used as electronic packaging and substrate materials due to their low dielectric constant and excellent processing ability.^{2,6} However, most polymers are thermal insulators, and introducing ceramic particles with high thermal conductivity (TC) into polymer matrices is an efficient method to improve the TC of polymer composites.

Up to now, ceramic fillers, such as aluminum oxide (Al₂O₃),^{7,8} silicon nitride (Si₃N₄),^{9,10} aluminum nitride (AlN)¹¹ and boron

nitride (BN),^{12–15} were used frequently as the thermal conductive materials embedded in a polymer matrix. For example, Al₂O₃, Si₃N₄, BN and AlN were employed into epoxy resin to improve TC, and the TCs of composites were up to ~3.6, 3.89, 4.42 and 3.39 W (mK)⁻¹, respectively.^{7,9,16,17} Although the TCs of polymer composite has been improved greatly, much higher than the TC of epoxy (0.23 W (mK)⁻¹), TCs of polymer composites are still much lower than that of bulk ceramics, such as Al₂O₃ (30 W (mK)⁻¹), Si₃N₄ (16.7 W (mK)⁻¹), BN (600 W (mK)⁻¹) and AlN (320 W (mK)⁻¹). One reason is that the ceramic particles are isolated in the polymer matrix, causing no network formed. In order to further enhance the TC, researchers have tried to use ceramic fillers by dispersing high loading because the fillers can contact with each other to build a thermally conductive network (TCN). However, the mechanical properties of composite with a high filler content will be deteriorated, such as susceptible to thermal cracking and challenging to process.⁷ Other reason is due to the interface thermal resistance among fillers. The TC of interface resistance formed between neighboring fillers is much lower than that of bulk particles because of the phonon scattering at filler–filler contacts within the network. Until to now, the covalent and noncovalent functionalization methods have been developed to modify the fillers to improve the TC of polymer composite *via* suppressing the interfacial thermal resistance.¹⁸ For example, an enhancement by 15.5% (from 1.037 to 1.198 W (mK)⁻¹) in TC of h-BN/epoxy resin composite

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by surface modification of h-BN particle was reported in.¹⁹ Meanwhile, Zhou's group²⁰ also reported an enhancement by 32% (from 1.03 to 1.36 W (mK)⁻¹) in TC of aluminum/epoxy resin composite because of surface modification of aluminum particle. The improvement of TC after interface treatment seems acceptable, but the final result of TC is not large enough when comparing with that of fillers themselves. Besides, there are some disadvantages in both two methods. For example, the disadvantage of the covalent functionalization method is that it destroys part of the intrinsic structure of fillers and reduces the intrinsic thermal transport properties of fillers, and the disadvantage of the noncovalent functionalization method is the necessity to introduce other components, like surfactants, on fillers.²¹ There is other factor, such as the gap among fillers, which has been verified to be important in improving TC and electrical conductivity of polymer composites *via* 1D filler (*i.e.*, short carbon fiber) by Spatial Confining Forced Network Assembly (SCFNA) method because the polymer was extruded out after compression in a confining space.^{22–24}

In order to verify the viability of SCFNA method further, the 3D filler (*i.e.*, Al₂O₃) was employed as a thermally conductive and electrical insulating filler and PDMS was employed to fabricate a thermally conductive composite due to its good flexibility and excellent electrical insulation properties by SCFNA method. The results showed that TC of sample with filler content of 80 wt% by SCFNA method was 16.35 times

higher than that of pure PDMS and 8.03 times higher than that of sample by traditional compounding method. A testing device was set up to analyze the ability of heat conduction and dissipation of PDMS/Al₂O₃ composites.

The theory of SCFNA method

There are three steps in SCFNA method. Firstly, a homogeneous compound was obtained by melt compounding or solution mixing. Secondly, the homogeneous compound was compressed in a confining space to initiate and finish the network by self-assembly mechanism. Thirdly, the samples with self-assembly network were further compressed to a thickness less than characteristic thickness in the confining space and samples with compacted network were achieved. The basic theory of SCFNA method is to introduce a mechanical delivered additional interaction to carry out a forced assembly network after a self-assembled network was built.²³ The technological pathway of SCFNA method was shown in Fig. 1.

For a given TCN, much heat flux flows along thermally conductive path in the whole thermally conductive network, as shown in Fig. 2a. Heat flux in any conductive path, consisting of two fillers, can be calculated with the following equation:

$$q_i = \frac{\Delta T}{2(\delta_f/\lambda_f) + \delta_p/\lambda_p} = \frac{\Delta T}{R} \quad (1)$$

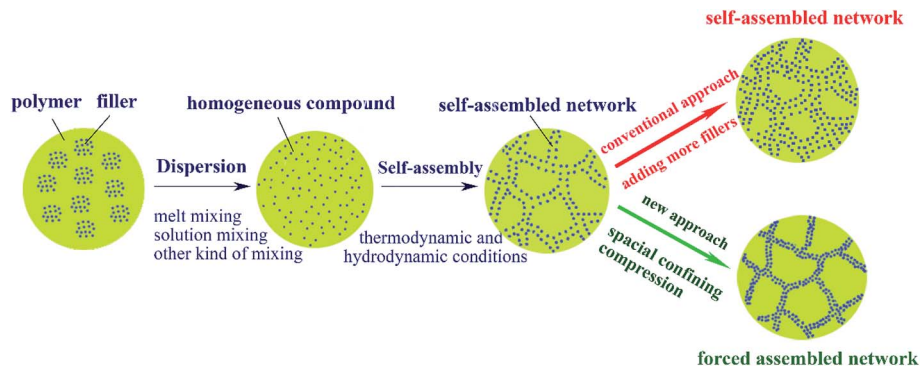


Fig. 1 Scheme of technological pathway of SCFNA method and conventional compounding.^{22,23}

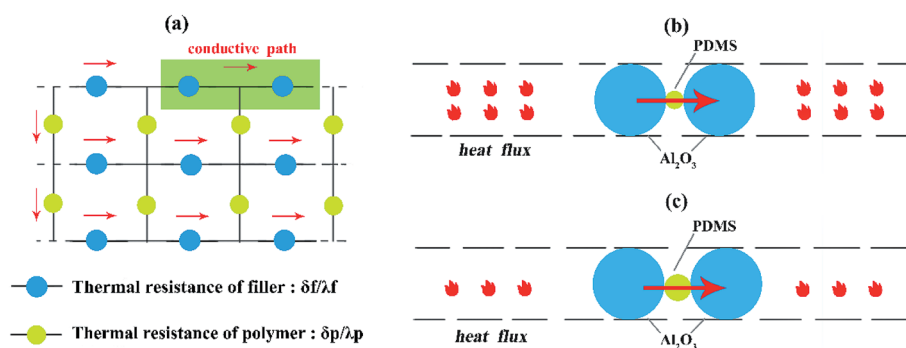


Fig. 2 The thermally conductive principle of polymeric composites. (a) thermally conductive network; (b) conductive path constructed by SCFNA method; (c) conductive path constructed by conventional compounding.



where q_i is the heat flux per unit area, ΔT is the temperature difference between any two fillers, δ_f and λ_f are the length and TC of filler, δ_p and λ_p are the gap between two fillers and TC of polymer, and \bar{R} is the average thermal resistance.

In general, δ_f/λ_f is much less than δ_p/λ_p (*i.e.*, the thermal resistance of polymer is much greater than that of fillers by more than 2 orders of magnitude). The heat flux of conductive path mainly depends on the thermal resistance of polymer. As a result, eqn (1) can be changed into eqn (2), as shown below.

$$q_i = \frac{\Delta T}{\delta_p/\lambda_p} \quad (2)$$

It is obvious that the average thermal resistance mainly depends on the gap (δ_p) between two fillers. When the gap (δ_p) decreases, the thermal resistance decreases, then the heat flux increases. Fig. 2b shows that the gap (δ_p) between two fillers is decreased by SCFNA method because the polymer is extruded out within the confining space after forced compression. However, the gap between two fillers cannot be decreased sharply only by adding more fillers, as shown in Fig. 2c.

Experimental

Materials and equipment

PDMS (Sylgard 184 Silicone Elastomer, Dow Corning Corp, USA) was purchased as two-part liquid component kits, *i.e.*, based polymer (Kit A) and curing agent (Kit B). Kit A and Kit B were mixed at a weight ratio of 10 : 1. Its density is 1.030 g cm^{-3} (cured), and its curing temperature is ranging from 25 to $150 \text{ }^\circ\text{C}$. Al_2O_3 of 3.8 g cm^{-3} in density was provided by Guangzhou Dinghua nano-material Co. Ltd, China. The morphology and particle size analysis of Al_2O_3 spheres are shown in Fig. 3 and the average diameter is about $5.5 \text{ }\mu\text{m}$. A stirrer (provided by Shanghai Hc Mechanical Equipment Co. Ltd) was used to prepare PDMS/ Al_2O_3 compound. The experimental hot embossing device^{22–26} was used to prepare composites.

Preparation of composite samples

There are three steps for preparing PDMS/ Al_2O_3 composites.

(1) PDMS and Al_2O_3 of 10, 30, 50, 70 and 80 in mass fraction were stirred at the rotational speed of 50 rpm for 30 min at $25 \text{ }^\circ\text{C}$

by stirrer. Then, the homogeneously dispersed mixture was put into the vacuum pump for 30 min to remove the air bubbles from mixture.

(2) In the first stage, the homogeneously dispersed mixture was put into a confining space of mold and compressed to a thickness of 2 mm. It stayed for 60 seconds to complete self-assembled network before the second stage.

(3) In the second stage, the mixture with 10, 50 and 80 wt% Al_2O_3 was further compressed from a thickness of 2 mm to a final thickness of 0.15, 0.2, 0.3 and 0.4 mm, respectively. The mixture with 30 and 70 wt% Al_2O_3 was compressed from a thickness of 2 mm to a final thickness of 0.05, 0.15, 0.2, 0.3, 0.4, 0.6, 0.8, 1.0 and 1.4 mm, respectively. Finally, the mixture was heated to cure at $120 \text{ }^\circ\text{C}$ for 30 min to obtain specimens for testing.

Material characterizations

A Hitachi JSM-7800F scanning electron microscopy (SEM) and a XJ-55C trinocular transfective metallographic microscope (provided by Shanghai Puzhe Photoelectric Instrument Co. Ltd) were used to study the dispersion and location of Al_2O_3 in PDMS.

The TC of composite was calculated by the eqn (3).

$$\lambda = \alpha C_p \rho \quad (3)$$

where α is the thermal diffusivity ($\text{mm}^2 \text{ s}^{-1}$) in-plane, measured based on laser flash technology by a Netzsch system (LFA 467, German). C_p is the specific heat (J (g K)^{-1}), measured by DSC Pyris 1 (Perkin Elmer, America). ρ is the bulk density (g cm^{-3}), calculated by geometry and weight. All the TC's measurements above were finished in $25 \text{ }^\circ\text{C}$.

Results and discussions

Table 1 shows the TC of the composites with 10, 30, 50, 70 and 80 wt% Al_2O_3 by SCFNA method. The TC of provided neat PDMS was 0.27 W (mK)^{-1} . In order to make a comparison of the effect between self-assembly network and forced assembly network on TC of composites, the samples with a thickness of 2.0 mm finished in the first stage were taken as the polymeric composites prepared by traditional compounding method.

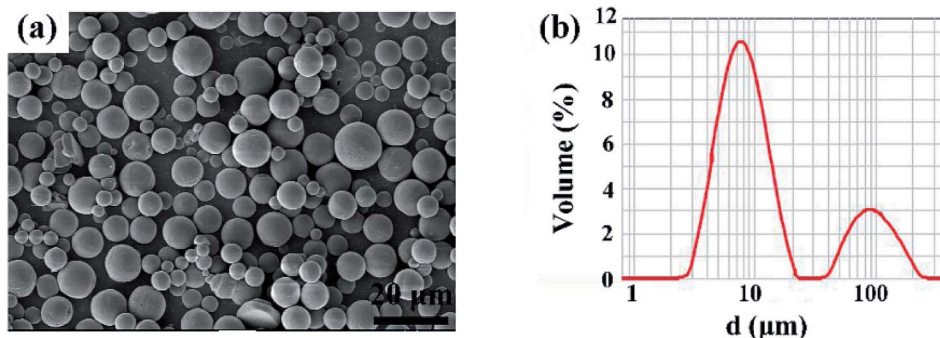


Fig. 3 (a) SEM image; (b) particle size analysis of spherical Al_2O_3 powder.



Table 1 The comparison of TC of PDMS/Al₂O₃ composites at different filler content and at different thickness

Sample		Thermal conductivities		
Filler contents (wt%)	Thickness (mm)	λ_s (W (mK) ⁻¹)	λ_c (W (mK) ⁻¹)	λ_s/λ_c
10	0.15	2.005	0.234	8.568
	0.20	1.518		6.487
	0.30	1.109		4.739
	0.40	0.826		3.530
30	0.05	3.425	0.252	13.595
	0.15	2.017		8.004
	0.20	1.632		6.476
	0.30	1.214		4.817
	0.40	0.853		3.385
	0.60	0.589		2.337
	0.80	0.435		1.726
	1.00	0.383		1.520
	1.40	0.288		1.143
50	0.15	2.018	0.328	6.152
	0.20	1.761		5.369
	0.30	1.260		3.841
	0.40	0.877		2.674
	0.05	3.700		9.069
70	0.15	2.351	0.408	5.762
	0.20	2.030		4.975
	0.30	1.609		3.944
	0.40	1.261		3.091
	0.60	0.988		2.422
	0.80	0.846		2.074
	1.00	0.668		1.637
	1.40	0.509		1.248
	0.05	4.414		8.025
80	0.15	2.566	0.550	4.665
	0.20	2.197		3.995
	0.30	1.942		3.531
	0.40	1.577		2.867

Fig. 4 showed the relationship between TC and thickness of samples when the filler content was 30 and 70 wt%. As shown in Fig. 4, the TCs of composites with filler content of 30 and 70 wt% increased slowly when the further compression

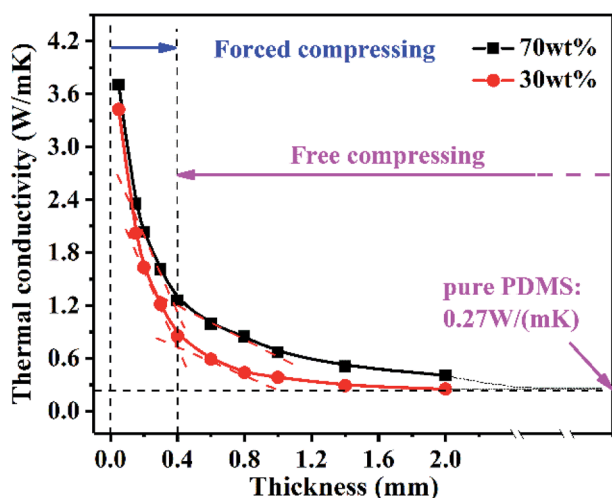


Fig. 4 Plot of TC versus thickness in different filler content.

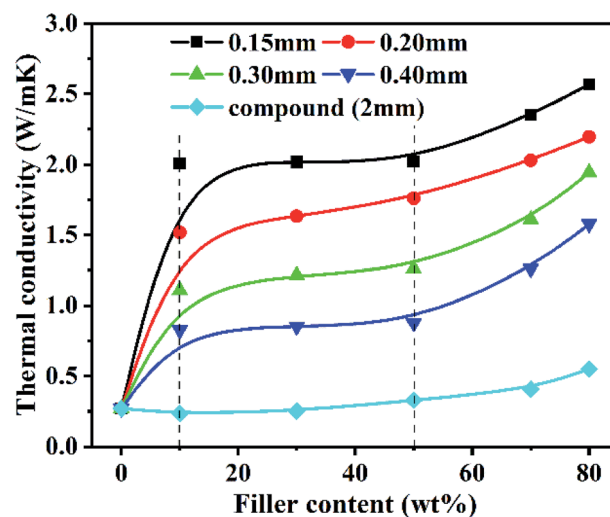


Fig. 5 Comparison of thermal conductivities of PDMS/Al₂O₃ composites by SCFNA and compounding method.

thickness was higher than ~ 0.4 mm, while it increased sharply when the further compression thickness was less than ~ 0.4 mm. This result was almost like the result in the paper and the thickness of ~ 0.4 mm can be still called the characteristic thickness.²⁴

The relationship between the filler content and TC of samples prepared by SCFNA was shown in Fig. 5. It showed that TC of the samples increased sharply when filler content increased from 0 to 10 wt%, and then TC of samples increased steadily when filler content was increasing from 10 to 50 wt%. After that, the TC of samples increased sharply again when filler content increased from 50 to 80 wt%. For example, for the sample with thickness of 0.15 mm, the TC of sample with the content of 10 wt% increased about 7.43 times compared with pure PDMS, and then TC increased steadily only from 2.005 to 2.019 W (mK)⁻¹. After that, TC increased from 2.019 to 2.566 W (mK)⁻¹, causing an enhancement of 1.27 times. TCs of samples with other thickness have the similar pattern like that of samples with thickness of 0.15 mm. However, the TC of samples prepared by compounding method increasing slowly, only from 0.27 to 0.55 W (mK)⁻¹. The effect of thickness on the TC of PDMS/Al₂O₃ composites by SCFNA method was shown in Fig. 6. The TC of samples increase with the decreasing in thickness. For example, the TC with filler content of 30 wt% was enhanced by 2.36 times when thickness was decreased from 0.4 to 0.15 mm.

Table 2 presents the obtained TC of the composites filled with Al₂O₃ obtained from other literature. Compared the TC in this paper with TC of other studies, the fabricated composites in our works display a relatively higher TC at the same filler content, and display almost the same TC with a lower filler content

Heat conduction and dissipation of PDMS/Al₂O₃ composites in heat spreaders was examined by recording the temperature variations in the center of heat source. The heat dissipation performance was evaluated on all samples with diameter of 40 mm, heated in a heat source with 5 mm in diameter. The testing



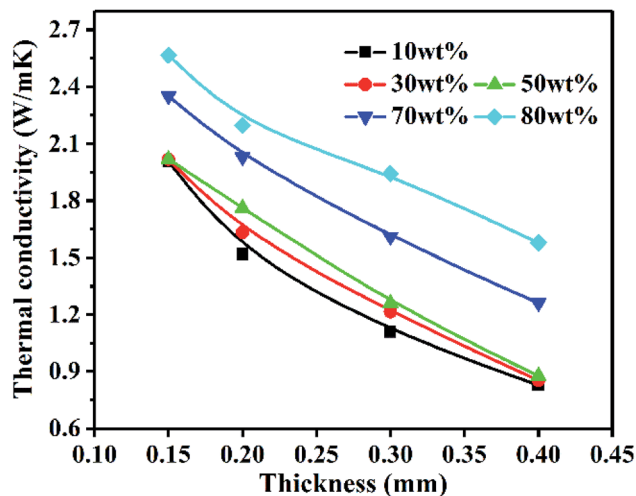


Fig. 6 The effect of thickness on the thermal conductivity of PDMS/ Al_2O_3 composites by SCFNA method.

temperature is approx. 28°C . The samples with filler content of 10, 50 and 80 wt% were separately connected to the same heater for 300 s, which guaranteed the set-point temperature reached a saturation state for comparison. The set-point position was in the center of the heater because this was typically the highest temperature and the crucial point that affected the performance and life time of electric device, like LED. As shown in Fig. 7a, the set-point temperature of composites with higher filler content exhibited a lower temperature over time. The set-point temperature was approx. 188.28, 184.65 and 181.3°C in the composite with filler content of 10, 50 and 80 wt% after heating

for 300 s, respectively. These temperatures decreased by 1.25, 4.88 and 8.23°C comparing with temperature of heat source (189.53°C), were shown in Fig. 7b.

As shown in Fig. 4–6, there are two factors to influence the TC of PDMS composites: the filler content and the thickness of samples prepared by SCFNA method. The TC of composites increases with the filler content due to the better formation of conductive network with higher filler content. The TC of samples increases when decreasing in thickness due to a compacted network was built. Morphology evolution of Al_2O_3 in-plane and through-plane of composite was observed to explain the effect of compacted network on TC of composite further.

Fig. 8 displays the morphology evolution of Al_2O_3 in-plane of composite with different thickness when filler content is 30 wt%. As shown in Fig. 8, Al_2O_3 formed continuous network in PDMS matrix when thickness decreasing from 0.4 mm to 0.05 mm. However, a more compacted network was formed only when the thickness of samples decreased. For example, the network formed when thickness is 0.05 mm is more compacted than the network formed when thickness is 0.4 mm. The morphology evolution of Al_2O_3 through-plane of composites with different thickness when filler content was 30 wt% was observed to further explain that a compact network could be formed when the thickness decreasing.

Fig. 9 showed the morphology evolution of Al_2O_3 through-plane of composites with different thickness when filler content was 30 wt% (10.4 vol%). As shown in Fig. 9a–e, Al_2O_3 aggregated step by step after compressed by SCFNA method, and the sample with a thickness of 0.05 mm showed the most compacted TCN than others. The average gap between Al_2O_3

Table 2 Comparison of the enhanced TC for different polymeric composites

Polymer composites	Filler content vol%	TC of composite $\text{W}(\text{mK})^{-1}$	Ref.
$\text{Al}_2\text{O}_3/\text{PDMS}$	38.7	3.7	This work
	52	4.414	
$\text{Al}_2\text{O}_3/\text{Epoxy}$	58.4	2.1	2
$\text{Al}_2\text{O}_3/\text{Epoxy}$	55	3.6	7
$\text{Al}_2\text{O}_3/\text{Epoxy}$	60	4.3	27
$\text{Al}_2\text{O}_3/\text{PVB}$	40	~ 2.7 (10 μm) ~ 3.5 (20 μm)	28

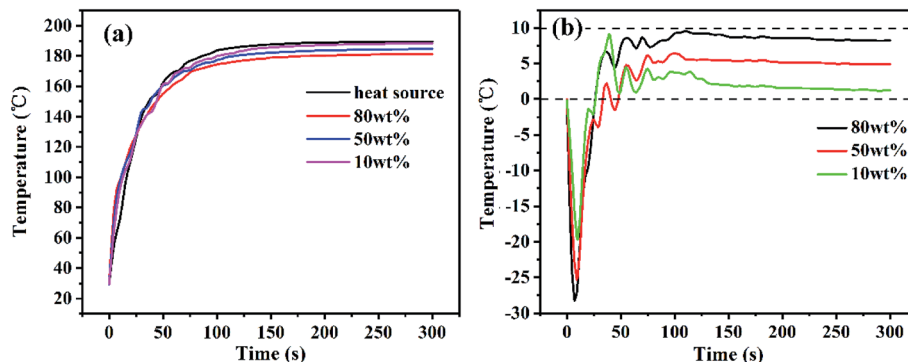


Fig. 7 (a) The center temperature of samples with different filler content at room temperature of 28°C ; (b) center temperature difference of composites with different filler content.



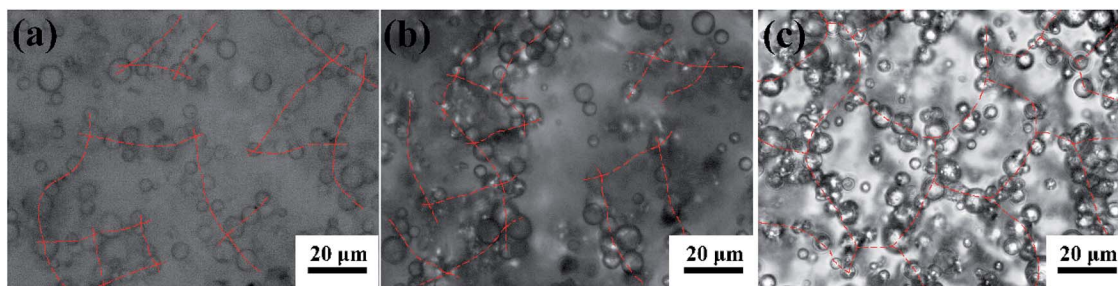


Fig. 8 Morphology evolution in-plane of PDMS/Al₂O₃ (30 wt%) composites versus thickness: (a) thickness = 0.40 mm; (b) thickness = 0.15 mm; (c) thickness = 0.05 mm.

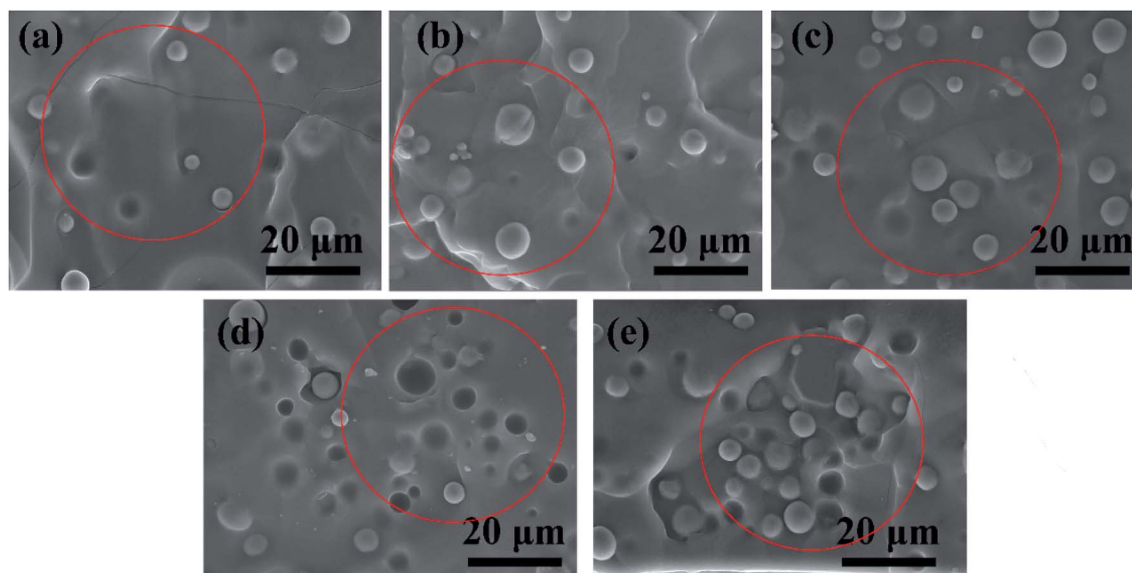


Fig. 9 Morphology evolution through-plane of PDMS/Al₂O₃ (30 wt%) composites versus thickness: (a) thickness = 0.40 mm; (b) thickness = 0.30 mm; (c) thickness = 0.20 mm; (d) thickness = 0.15 mm; (e) thickness = 0.05 mm.

was 13.82 μm ($\sigma = 4.87 \mu\text{m}$), 8.72 μm ($\sigma = 3.64 \mu\text{m}$), 5.65 μm ($\sigma = 3.09 \mu\text{m}$), 3.35 μm ($\sigma = 2.09 \mu\text{m}$) and 1.06 μm ($\sigma = 0.47 \mu\text{m}$) when the thickness was 0.4, 0.3, 0.2, 0.15 and 0.05 mm, respectively. The average gap among Al₂O₃ and standard deviation were calculated according to eqn (4) and (5).

$$\bar{x} = \frac{\sum_i^m x_i}{m} \quad (4)$$

$$\sigma = \sqrt{\frac{\sum_i^m (x_i - \bar{x})^2}{m - 1}} \quad (5)$$

where \bar{x} is the average gap among Al₂O₃, x_i is the distance between any two Al₂O₃ in the red circle region, σ is the standard deviation.

Besides, the volume fraction of Al₂O₃ in the TCN of PDMS/Al₂O₃ composites was calculated after compressed in confining space according to the SEM. The volume fraction of Al₂O₃ was calculated by eqn (6).

$$v_i = \frac{S_i}{S} \quad (6)$$

where v_i is the volume fraction of Al₂O₃. S_i is the whole area of Al₂O₃, calculated by product of numbers and diameter of Al₂O₃. S is the whole area of SEM, calculated by product of the width and length of SEM.

The calculated volume fraction of Al₂O₃ was 10.6%, 11.66%, 18.02%, 20.14% and 23.32%, and almost all the calculated volume fraction was higher than the original volume fraction 10.4 vol% of Al₂O₃. That means, the Al₂O₃ agglomerates to form a more and more compacted network with decrease of thickness of samples.

The compacted network built can be explained below. When the thickness of composites is larger than characteristic thickness, which is called free compression, it could only cause the self-assembly network wiggling in polymer matrix without densification of the network threads, and the gap between fillers cannot be decreased largely, as shown in Fig. 10a. While when the thickness is less than characteristic thickness, a noteworthy densification of the network threads could come into by spatial confining compression. Due to the restriction of



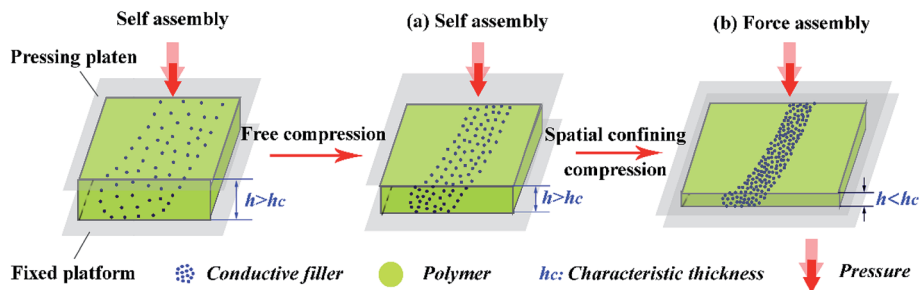


Fig. 10 Model of SCFNA: (a) free compression and (b) spatial confining compression.

wiggling freedom of the network and the prevention of aggregate force between granules from falling apart, the filler at the threads will tend to get closer and extra polymer between the filler granules will be squeezed out in spatial confining compression process. The smaller thickness of samples means more extra polymer will be squeezed out, the gap between fillers becomes smaller, as shown in Fig. 10b. The mechanism of forced assembly can be explained as follows. In spatial confining compressing process, characterized by a thickness of sample below characteristic thickness, the compression force could be transmitted to the granules to break the balance of self-assembly network. The granules under compression tend to get much closer until a new balance was established. In this way, the self-assembly network was converted to a forced-assembly network with enhanced TC. This method can improve the polymer composite with high thermal conductivity, which provides the possibility to replace thermally conductive materials such as metal with polymeric composites.

Conclusions

In this paper, polymeric composites with high thermal conductivity were prepared by a SCFNA method. The main mechanism of SCFNA method can be explained that the external load applied on the composites with self-assembly network in a confining space induces the extra polymers extruded between fillers to form a compact thermally conductive network. The thermal conductivity of composites was improved because the densification network decreased the gap among fillers. The SEMs showed that a compact thermally conductive network could be obtained by SCFNA method. PDMS/Al₂O₃ composites with 10 to 80 wt% Al₂O₃ were prepared by SCFNA method. When the filler content was 80 wt%, the thermal conductivity obtained by SCFNA method was improved by 16.35 times than pure PDMS. Produced PDMS/Al₂O₃ composite with filler content of 80 wt% was used for heat spreader for heat source, resulting in a significant decrease by about 8.23 °C in the set-point and the saturation temperatures and showing sufficient heat dissipation efficiency.

Conflicts of interest

There are no conflicts to declare.

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