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# A novel method for the highly efficient biotransformation of genistein from genistin using a high-speed counter-current chromatography bioreactor

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Genistein, an important soybean isoflavone compound, has gained attention for its significant properties. Compared with the glycone form of genistin, low content of genistein limits the use in food and pharmaceutical fields. In this study, a novel bioreactor with high-speed counter-current chromatography (HSCCC) was built for the highly efficient biotransformation of genistein from genistin. The solvent system for the bioreactor was selected according to the  $K_D$  values. The selected solvent system was evaluated by the enzyme activity of  $\beta$ -glucosidase. An ethyl acetate/buffer solution was selected as the preferred solvent system for the HSCCC bioreactor. Optimum reactor parameters were selected according to the retention of the stationary phase. The HSCCC bioreactor was operated using different flow rates, and 2.0 mL min<sup>-1</sup> was chosen as the optimal flow rate with a conversion rate of over 90% within 24 h; the novel bioreactor easily immobilized and recycled the enzyme and could be applied in the preparation of genistein.

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## 1. Introduction

Genistein (4',5,7-trihydroxyisoflavone), an important isoflavone compound, is widely distributed in leguminous plants such as soybeans, *Pueraria lobata* (Willd.) Ohwi and *Sophora japonica* L. Genistein is structurally similar to estrogen and has a number of biological properties that can prove beneficial in alleviating a variety of health conditions including kidney disease, menopausal symptoms, neuronal loss, Alzheimer's disease, and cancer.<sup>1-7</sup> Because of this broad range of biological properties, genistein has gained attention as a food additive and dietary supplement. Currently, there is an enormous demand for genistein in health and food products.

Genistein is a glycone form of genistin (4',5,7-trihydroxyisoflavone- $\beta$ -D-glucopyranoside) with a higher potential for intestinal absorption than genistin.<sup>8,9</sup> Genistein has also shown

stronger antioxidant potency in DPPH radical-scavenging and LDL oxidation assays than genistin.<sup>10</sup> However, genistein is mainly present as a glycoside form of genistin in soybeans. Therefore, the conversion of genistein from genistin is very important. There are many methods for the transformation of genistin to genistein, which include alkaline/acidic hydrolysis and microbial and enzymatic transformations, as shown in Fig. 1.<sup>11-13</sup> Alkaline or acidic hydrolysis always results in a strong reaction and causes environmental pollution. On the other hand, biotransformation using enzymes or microorganisms has many advantages including simplicity of procedure, mild conditions, low cost, and less pollution of the environment. To provide efficient biotransformation conditions, many types of reactors need to be built. These include a membrane, a packed bed, and an immobilized enzyme reactor.<sup>14,15</sup> There are also many obstacles that need to be overcome such as physical damage to the membrane, membrane pollution, cleaning of the immobilized enzyme, and enzyme recycling.

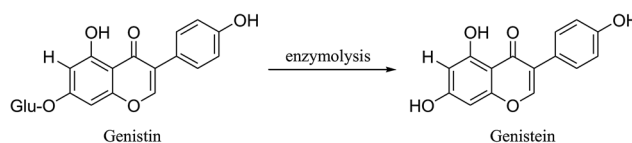


Fig. 1 Chemical structures of the compounds and a schematic of enzymolysis.

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High-speed counter-current chromatography (HSCCC) is a liquid–liquid partition chromatographic technique that eliminates the possibility of irreversible sample adsorption on a solid support. Furthermore, HSCCC provides high recovery and rapid and effective separation, is easily scaled-up and simple to operate, and has low solvent consumption. It has recently become a useful tool for the preparative isolation and purification of various natural products.<sup>16–19</sup> The HSCCC column is a hollow polytetrafluoroethylene coil twining around the HSCCC column holder. The two-phase solvent system is set at a high distribution rate of over 13 times per second (a rotation speed of 800 rpm) (Fig. 2). When the enzyme or microorganism remains in the stationary phase of the HSCCC solvent system and the substrate continues to be eluted in the mobile phase, a novel biotransformation reactor is constructed. There are many advantages of this novel reactor such as a decrease in substrate inhibition, contact between the enzyme and the substrate, and high purity of products outflowing in the mobile phase.

In this study, a novel HSCCC bioreactor was built for the highly efficient biotransformation of genistin to genistein. The bioreactor parameters, the type of enzyme used, and the flow rate of the mobile phase were optimized. To the best of our knowledge, to date, an HSCCC bioreactor has not been used for the biotransformation of genistein. In this study, an attempt was made to perform a biotransformation of genistin to genistein using HSCCC. This method could be used to transform other compounds and enriches the types of bioreactors.

## 2. Materials and methods

### 2.1 Reagents and materials

*n*-Hexane, methanol, citric acid, sodium citrate, ethyl acetate, *n*-butanol, acetic acid and dichloromethane were analytical grade (Sinopharm Chemical Reagent Co., Ltd, Shanghai, China). *p*-Nitrophenyl  $\beta$ -D-glucopyranoside (*p*-NPG) and  $\beta$ -glucosidase were biochemical grade (Sangon Biotech Co., Ltd. Shanghai,

China). HPLC-grade methanol was purchased from the Fisher Company (Fairlawn, NJ, USA). Water used was deionized by an osmosis Milli-Q system (Millipore, Bedford, MA, USA). Reverse osmosis Milli-Q water (Millipore, USA) was used in this study. The purities of genistin and genistein were over 98% (HPLC, Desite Biotech Co., Ltd. Chengdu, China).

### 2.2 Apparatus

The HSCCC equipment was TBE-300A (Shanghai, Tauto Biotech, China) with three 300 mL multi-layer coil separation columns (diameter of the PTFE tube was 1.6 mm) as well as a 20 mL manual sample loop. The  $\beta$ -values of the HSCCC column ranged from 0.47 (internal) to 0.73 (external) ( $\beta = r/R$ , where  $r$  is the rotation radius or the distance from the coil to the holder shaft, and  $R$  ( $R = 7.5$  cm) is the revolution radius or the distance between the holder axis and the central axis of the centrifuge). The HSCCC apparatus was equipped with four other instrument modules, *i.e.* a 8823A-UV monitor at 280 nm (Beijing Emilion Technology, Beijing, China), a Model 3057 portable recorder (Yokogawa, Sichuan Instrument Factory, Sichuan, China), a TBP-5002 constant-flow pump and a DC-0506 low constant temperature bath (Tauto Biotechnique, Shanghai, China), to maintain the temperature at 40 °C. The spectrophotometer used was the Genesys 10S UV-Vis scanning spectrophotometer (Thermo Fisher Scientific, Waltham, MA, USA). HPLC separation was performed using the Agilent 1260 system (Agilent Technologies, USA) consisting of a quaternary solvent delivery system, vacuum degasser, diode array detector (DAD), auto-sampler and column oven connected to the Agilent ChemStation software. The HPLC column used was the Waters Symmetry C<sub>18</sub> column (250 mm  $\times$  4.6 mm, i.d. 5  $\mu$ m, USA).

### 2.3 Determination of the partition coefficient

The partition coefficient ( $K_D$ ) of the substrate and the product in different solvent systems was determined by HPLC. The substrate (genistin) and the product (genistein) at appropriate

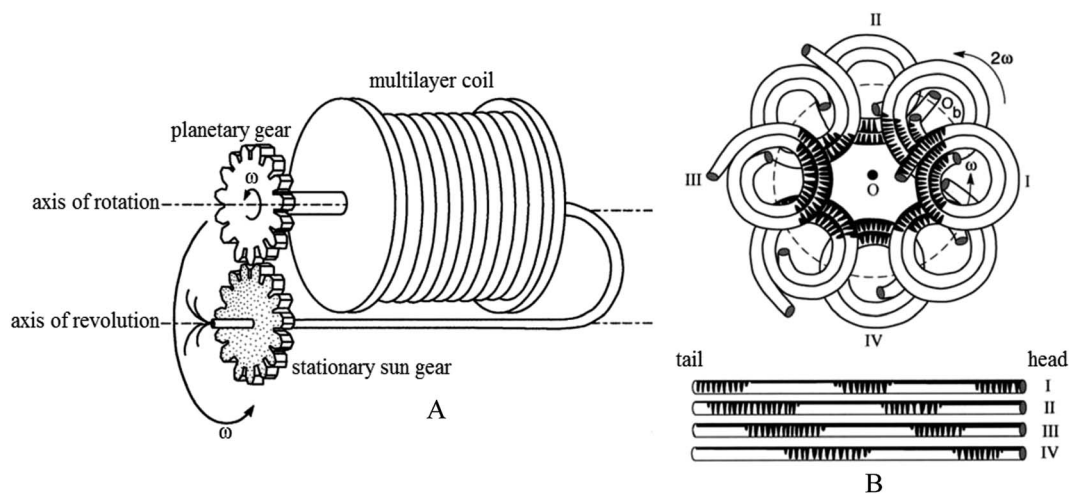


Fig. 2 The instrument construction of the HSCCC reactor. (A) Type-J planetary motion of the HSCCC multi-layer coil column. (B) Schematic of the distribution and motion of two phases in the HSCCC column.



amounts were placed in 5 mL test tubes, dissolved with the stationary phase of the two-phase solvent system, and detected by HPLC. The peak areas  $A_{S1}$  and  $A_{P1}$  were obtained. An equal volume of the mobile phase solution was added to the lower phase solution followed by vigorous shaking to ensure complete mixing. After the equilibrium of the partition was reached, the stationary phase was analyzed by HPLC to obtain the peak areas  $A_{S2}$  and  $A_{P2}$ . The  $K_D$  value was defined as the peak area of the target compound in the stationary phase divided by the peak area of the target compound in the mobile phase. The partition coefficients of the substrate and product were calculated as follows:  $K_{SD} = (A_{S1} - A_{S2})/A_{S2}$ ,  $K_{PD} = (A_{P1} - A_{P2})/A_{P2}$ .

The partition coefficient of  $\beta$ -glucosidase was determined in a two-phase solvent system and analyzed as follows: 2 mg of  $\beta$ -glucosidase was fully dissolved in 6 mL of the stationary phase solution of the two-phase solvent system to obtain solution A. Then, 3 mg of *p*-NPG was fully dissolved in 1 mL of the stationary phase solution to obtain solution B. Furthermore, 2 mL of solution A and 1 mL of solution B were heated in a 45 °C water bath for 30 min, and 1 mL of 1 mol L<sup>-1</sup> Na<sub>2</sub>CO<sub>3</sub> solution was added to terminate the reaction. The absorbance  $E_1$  was measured by a spectrophotometer at 400 nm. Then, 4 mL of solution A was mixed with an equal volume of the mobile phase solution of the two-phase solvent system followed by vigorous shaking to ensure complete mixing. After the equilibrium of the partition was reached, 2 mL of the lower phase solution was heated with 1 mL of solution B in a 45 °C water bath for 30 min. The reaction was terminated by adding 1 mL of 1 mol L<sup>-1</sup> Na<sub>2</sub>CO<sub>3</sub> solution, and the absorbance  $E_2$  was measured by a spectrophotometer at 400 nm. The reference solution was prepared by heating an inactivated  $\beta$ -glucosidase solution A (2 mg of inactivated  $\beta$ -glucosidase in 6 mL of the stationary phase solution) with 1 mL of solution B in a 45 °C water bath for 30 min. The reaction was terminated by adding 1 mL of 1 mol L<sup>-1</sup> Na<sub>2</sub>CO<sub>3</sub> solution, and the absorbance  $E_0$  was measured by a spectrophotometer at 400 nm. The partition coefficient of  $\beta$ -glucosidase in the two phases was calculated using the following formula:  $K_D = (E_1 - E_2)/(E_2 - E_0)$ .

#### 2.4 Determination of the maximum solubility and enzyme activity

After 2.0 mg of  $\beta$ -glucosidase was accurately weighed, the stationary phase of the two-phase solvent system was gradually added under sonication. When the solution became clear, its volume was determined, and the maximum solubility of  $\beta$ -glucosidase was determined to be 750 mg L<sup>-1</sup>. Then, 2.0 mg of genistin was accurately weighed, and the mobile phase of the two-phase solvent system was gradually added under sonication. When a clear solution was obtained, its volume was determined, and the maximum solubility of genistin was determined to be 62.5 mg L<sup>-1</sup>.

Aqueous solutions of citric acid and sodium citrate at 0.10 mol L<sup>-1</sup> were separately prepared, mixed, and filtered, and a citric acid/sodium citrate blank buffer solution (pH = 5) was obtained. Then, 600 mL of a two-phase solvent system consisting of equal volumes of ethyl acetate and buffer solution was placed

in a separatory funnel, to which 120 mg of enzyme was added. The funnel was placed on a shaker for uninterrupted mixing. Every 4 h, 6 mL of the upper and lower phase solutions were sampled. The organic phase was discarded, and the aqueous phase was divided into three equal portions. Each portion was mixed with 2 mL of 0.8 mg mL<sup>-1</sup> *p*-NPG solution and heated in a 45 °C water bath for 30 min. Next, 2 mL of 1.0 mol L<sup>-1</sup> Na<sub>2</sub>CO<sub>3</sub> aqueous solution was added to terminate the reaction, and changes in the color of the solution were observed. The absorbance values  $A_{Y1}$ ,  $A_{Y2}$ , and  $A_{Y3}$  were measured at 400 nm, and the average absorbance value  $\bar{A}_Y$  was calculated.

The blank and reference solution were prepared by mixing 2 mL of 0.4 mg mL<sup>-1</sup> initial and inactivated  $\beta$ -glucosidase buffer solution with 2 mL of 0.8 mg mL<sup>-1</sup> *p*-NPG solution, which were then heated in a water bath at 45 °C for 30 min. The reactions were terminated by the addition of 2 mL of 1 mol L<sup>-1</sup> Na<sub>2</sub>CO<sub>3</sub> solution. Moreover, the absorbances  $A_M$  (blank) and  $A_0$  (reference) at 400 nm were measured by a spectrophotometer. Each measurement was averaged three times, and the average enzyme activity was calculated using the following formula: enzyme activity/% =  $(\bar{A}_Y - A_0) \times 100\% / (A_M - A_0)$ .

#### 2.5 Optimization of the HSCCC reactor parameters

During the process of counter-current chromatography, there were two modes of rotation direction for the main unit: forward rotation (FWD) and reverse rotation (REV). The input port could serve as the inlet (IN) or the outlet (OUT) of the main unit. Therefore, there were four elution modes for the counter-current chromatography: REV-IN, FWD-OUT, FWD-IN, and REV-OUT. In this experiment, we tested the retention of the stationary phase using an ethyl acetate/buffer solution (1 : 1, v/v) solvent system, four elution modes of the counter-current chromatography (REV-IN, FWD-OUT, FWD-IN, and REV-OUT), the lower phase as the stationary phase, and the upper phase as the mobile phase. At first, the entire length of the Teflon tubing of the high-speed counter-current chromatography system was filled with the stationary phase solution at the flow rate of 30 mL min<sup>-1</sup>; once the main unit was initiated, the rotation speed was slowly adjusted to 800 rpm, and the temperature was maintained at 40 °C. After the rotation speed stabilized, the mobile phase was pumped in at 2 mL min<sup>-1</sup> until equilibrium was reached. After the rotation was stopped, the two-phase solvent system in the separation column was transferred under nitrogen to a graduated cylinder to calculate the retention of the stationary phase ( $S_f$ ) such that  $S_f$  (%) =  $(V_S/V_T) \times 100\%$ , where  $V_S$  is the volume of the stationary phase and  $V_T$  is the total volume of the HSCCC column.

In this experiment,  $S_f$  was measured in parallel at 550, 600, 650, 700, 750, and 800 rpm. The measurement method was the same as that used for different elution modes, and each measurement was averaged three times.

#### 2.6 Operation of the HSCCC bioreactor

**2.6.1 Preparation of the two-phase solvent and sample solutions.** The ethyl acetate/buffer solution (1 : 1, v/v) solvent system was placed in a separatory funnel, fully shaken, and allowed to stand for phase separation. After the two phases



reached equilibrium, the upper and lower phases were separately sampled and degassed by ultrasonication. According to the requirements of the counter-current chromatography bioreactor, the following three solutions were prepared: (1) a portion of the upper phase solution was used as the mobile phase of the counter-current chromatography system, to which genistin was added and allowed to fully dissolve under ultrasonication to obtain a substrate solution of  $62.5 \text{ mg L}^{-1}$ ; (2) the lower phase was the stationary phase, to which  $\beta$ -glucosidase was added and fully mixed to prepare a  $750 \text{ mg L}^{-1}$  bioreactor enzyme solution; and (3) a portion of the upper phase solution was stored as the blank mobile phase. The abovementioned three solutions were kept in a  $40 \text{ }^\circ\text{C}$  water bath before use.

**2.6.2 Operation of the HSCCC bioreactor.** At first, an enzyme solution of the bioreactor was pumped into the entire HSCCC column at the flow rate of  $30 \text{ mL min}^{-1}$ . After the loading was completed, the main unit was started. Using the FWD-OUT mode (rotating in a clockwise direction), the mobile phase was pumped in from the end of the HSCCC column. The rotation speed was adjusted to 800 rpm, and the blank mobile phase was introduced at the end of the counter-current chromatography separation column at a certain flow rate. After the two-phase solvent system reached hydrodynamic equilibrium, the substrate mobile phase solution containing genistin was pumped in. The detector was turned on, the detector wavelength was set at 254 nm, and the recorder was started to initiate data acquisition. The eluted solution was obtained in 10 mL test tubes.

After biotransformation of genistin by the HSCCC bioreactor, 1 mL of the eluted solution was periodically obtained, dried with nitrogen, re-dissolved with methanol, and filtered with a  $0.45 \mu\text{m}$  needle filter. The conversion rate of the target peak was analyzed by HPLC. After the conversion dropped below a certain value, the run was stopped, the liquid in the separation column was extracted with nitrogen, and  $S_f$  was calculated.

## 2.7 HPLC analysis condition and structural identification

The mobile phase comprised methanol/water (70 : 30, v/v), with isocratic elution at the flow rate of  $1.0 \text{ mL min}^{-1}$  and a column temperature of  $30 \text{ }^\circ\text{C}$ . The detection wavelength was set at 254 nm, and the injection volume was  $10 \mu\text{L}$ .

The identification of the converted product was performed *via* mass spectrometry (ESI-MS) with electrospray ionization (ESI)/quadrupole (Q) ESI-MS data (quadrupole LC/MS 6120, Agilent Technologies, Palo Alto, CA) in the direct injection mode, and the MS conditions were as follows: mass range: 100–1000 *m/z*, capillary temperature:  $300 \text{ }^\circ\text{C}$ , spray voltage: 4 kV, skimmer voltage: 40 V, auxiliary gas (nitrogen) flow:  $6 \text{ L min}^{-1}$ , and positive mode. NMR spectra were obtained using the Bruker AV-400 spectrometer (Bruker BioSpin, Rheinstetten, Germany) with TMS as the internal standard.

# 3. Results and discussion

## 3.1 Selection of the solvent system

The selection of an appropriate two-phase solvent system is one of the key steps in the application of HSCCC as a bioreactor for

processing the active ingredients in natural medicines. A suitable two-phase solvent system requires that (1) the enzymes are mainly distributed in the stationary phase since distribution in the mobile phase results in a loss of enzymes; (2) the product is mainly distributed in the mobile phase and can be rapidly eluted with the mobile phase; and (3) the substrate has good solubility in the mobile phase. Based on the principles of enzymatic reactions, the larger contact area between the substrate and the enzyme could enhance the catalytic efficiency. However, HSCCC is a partition chromatography in which the substrate undergoes vigorous mixing and partitioning inside the column. Therefore, if the substrate is adequately distributed in the stationary phase of the selected two-phase solvent system, the HSCCC bioreactors can be implemented.

As shown in Table 1, the distribution of the substrate and product was analyzed in the *n*-hexane/buffer solution (1 : 1, v/v), ethyl acetate/buffer solution (1 : 1, v/v), dichloromethane/buffer solution (1 : 1, v/v), and *n*-butanol/buffer (1 : 1, v/v) solution. The results showed that the  $K_D$  values in the *n*-hexane/buffer solution solvent system were below 0.1, indicating that both the substrate and product were mainly distributed in the stationary phase, and the product could not be continuously eluted with the mobile phase after enzymatic hydrolysis. Therefore, the *n*-hexane/buffer solution was not suitable as a solvent system for the HSCCC reactors. The remaining solvent systems provided satisfactory  $K_D$  values for the substrate and product and could be further tested for enzymatic activity as suitable two-phase solvent systems.

## 3.2 Effects of the solvent system on the enzyme activity

Fig. 3 shows activity changes for  $\beta$ -glucosidase in three solvent systems: ethyl acetate/buffer (1 : 1, v/v), dichloromethane/buffer (1 : 1, v/v), and *n*-butanol/buffer (1 : 1, v/v) solutions. The results showed that enzyme activity in the dichloromethane/buffer solution solvent system decreased to less than 1% at 4 h and to 0% at 8 h. We suspect that dichloromethane readily denatures  $\beta$ -glucosidase and thus destabilizes the enzyme structure. Therefore, dichloromethane is not suitable as a solvent for the HSCCC bioreactor. The ethyl acetate/buffer and *n*-butanol/buffer solution solvent systems provided higher stability of the enzyme activity during the test. The enzyme activity was above 99% at 48 h, indicating that the ethyl acetate and *n*-

Table 1 The  $K_D$ -values for resveratrol in a HSCCC bioreactor with different solvent systems<sup>a</sup>

Solvent systems	$K_D$ -values	
	Genistin	Genistein
<i>n</i> -Hexane/buffer solution (1 : 1, v/v) <sup>b</sup>	<0.1	<0.1
Ethyl acetate/buffer solution (1 : 1, v/v) <sup>b</sup>	0.66	>10
<i>n</i> -Butanol/buffer solution (1 : 1, v/v) <sup>b</sup>	3.05	>10
Dichloromethane/buffer solution (1 : 1, v/v) <sup>c</sup>	>10	<0.1

<sup>a</sup> The buffer solution contained citric acid (0.10 M) and a sodium citrate buffer solution (pH 5). <sup>b</sup> Lower phase as the stationary phase. <sup>c</sup> Upper phase as the stationary phase.





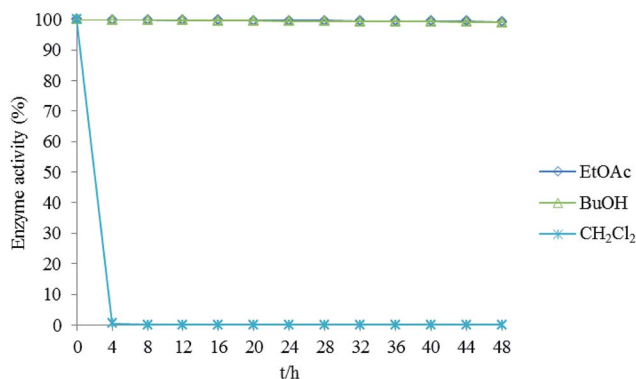


Fig. 3 Effect of the solvent system on enzyme activity. EtOAc: ethyl acetate/buffer solution (1 : 1, v/v), BuOH: *n*-butanol/buffer solution (1 : 1, v/v), and CH<sub>2</sub>Cl<sub>2</sub>: dichloromethane/buffer solution (1 : 1, v/v).

butanol systems did not readily cause enzyme denaturation. In the *n*-butanol/buffer solution solvent system, the *n*-butanol phase was saturated with a large amount of water. This caused  $\beta$ -glucosidase to elute with the mobile phase and lowered the enzyme concentration during the operation of the bioreactor. Eventually, the ethyl acetate/buffer solution was selected as the preferred two-phase solvent system for the HSCCC bioreactor.

### 3.3 Optimization of the HSCCC reactor parameters

Fig. 4A shows the  $S_f$  of the ethyl acetate/buffer solution solvent system in four HSCCC elution modes (FWD-OUT, REV-IN, FWD-IN, and REV-OUT). The results showed that the  $S_f$  was higher in the FWD-OUT (64.5%) and REV-IN (64.6%) modes, which were

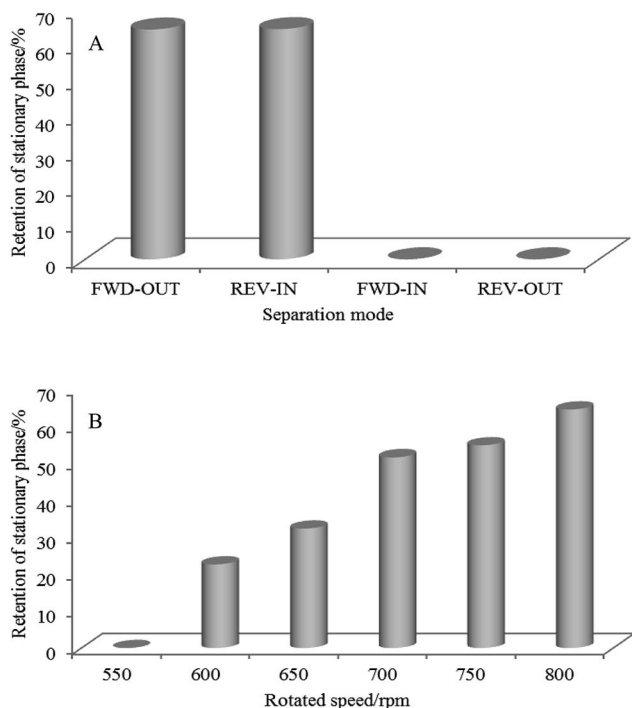


Fig. 4 The results of the optimized HSCCC bioreactor parameters. (A) different separation modes. (B) Different rotated speed. Solvent system: ethyl acetate/buffer solution (1 : 1, v/v). Rotation speed: 800 rpm. Flow rate: 2.0 mL min<sup>-1</sup>. Temperature: 40 °C.

suitable elution modes for the HSCCC reactors. The FWD-IN and REV-OUT modes, under which the  $S_f$  was zero, were not suitable for the HSCCC reactors.

Fig. 4B shows the effect of different rotation speeds (*i.e.* 550, 600, 650, 700, 750, and 800 rpm) on the  $S_f$  in the FWD-OUT mode. As shown, the  $S_f$  was 0 when the rotation speed was 550 rpm, and it significantly increased with the increasing rotation speed. The stationary phase retention rate increased to as high as 64.5% as the rotation speed reached 800 rpm. Therefore, the FWD-OUT mode and 800 rpm rotation speed were chosen as the optimal parameters for the HSCCC bioreactor.

### 3.4 Optimization of the flow rate for the HSCCC bioreactor

Fig. 5 shows the HSCCC chromatograms and purity results for the biotransformation of genistin to genistein in the novel bioreactor at the flow rates of 4.0, 3.0, 2.0, and 1.0 mL min<sup>-1</sup>. When the flow rate was 3.0 mL min<sup>-1</sup> (Fig. 5A), the  $S_f$  was 42%, and the conversion rate significantly declined within 3 h. It remained stable at 60–65% during the 3–12 h time period. When the flow rate continued to decrease to 2.5 mL min<sup>-1</sup> (Fig. 5B), the retention of the stationary phase was 51%, the absorption curve showed significant fluctuations, and the conversion rate of genistin was initially higher and then decreased over time. The conversion rate of genistin began to rapidly drop after 8 h and was eventually reduced to 70% at 14 h. When the flow rate was 2.0 mL min<sup>-1</sup> (Fig. 5C), the  $S_f$  was 62%, and the conversion rate of genistin was initially higher and then decreased over time. The conversion rate was over 90% within 24 h. When the flow rate was 1.0 mL min<sup>-1</sup> (Fig. 5D), the  $S_f$  was 75%. The absorption value increased in the time period of 1.6–2.4 h, and the conversion rate for genistin reached 100%, followed by a slow decline. This may be attributed to the fact that during the initial introduction of the substrate, the reaction time for the substrate is longer, allowing the substrate to be fully converted in the HSCCC reactor. Within 24 h of the reaction time, genistin was efficiently processed by  $\beta$ -glucosidase with over a 99% conversion rate (Fig. 6).

Based on the abovementioned results, the  $S_f$  of HSCCC decreased as the flow rate increased. On the one hand, the increase in the flow rate correlated with the increase in the amount of the substrate entering the reactor. If the substrate was completely converted, the product yield produced per unit of time increased. On the other hand, a decrease in  $S_f$  decreased  $\beta$ -glucosidase, resulting in a decreased amount of the enzyme and an incomplete conversion of the substrate. Consequently, in the operation of the HSCCC bioreactor, a certain balance between the flow rate and  $S_f$  was required to achieve a sufficiently high product concentration at the outlet. Based on this study, a flow rate of 2.0 mL min<sup>-1</sup> was chosen as the optimal flow rate, which provided a conversion rate of over 90% within 24 h of the reaction. Moreover, the purity of the product appeared to gradually decrease as the reaction time was extended. A possible explanation for this effect is that after genistin is hydrolyzed, the detached glucose remains in the



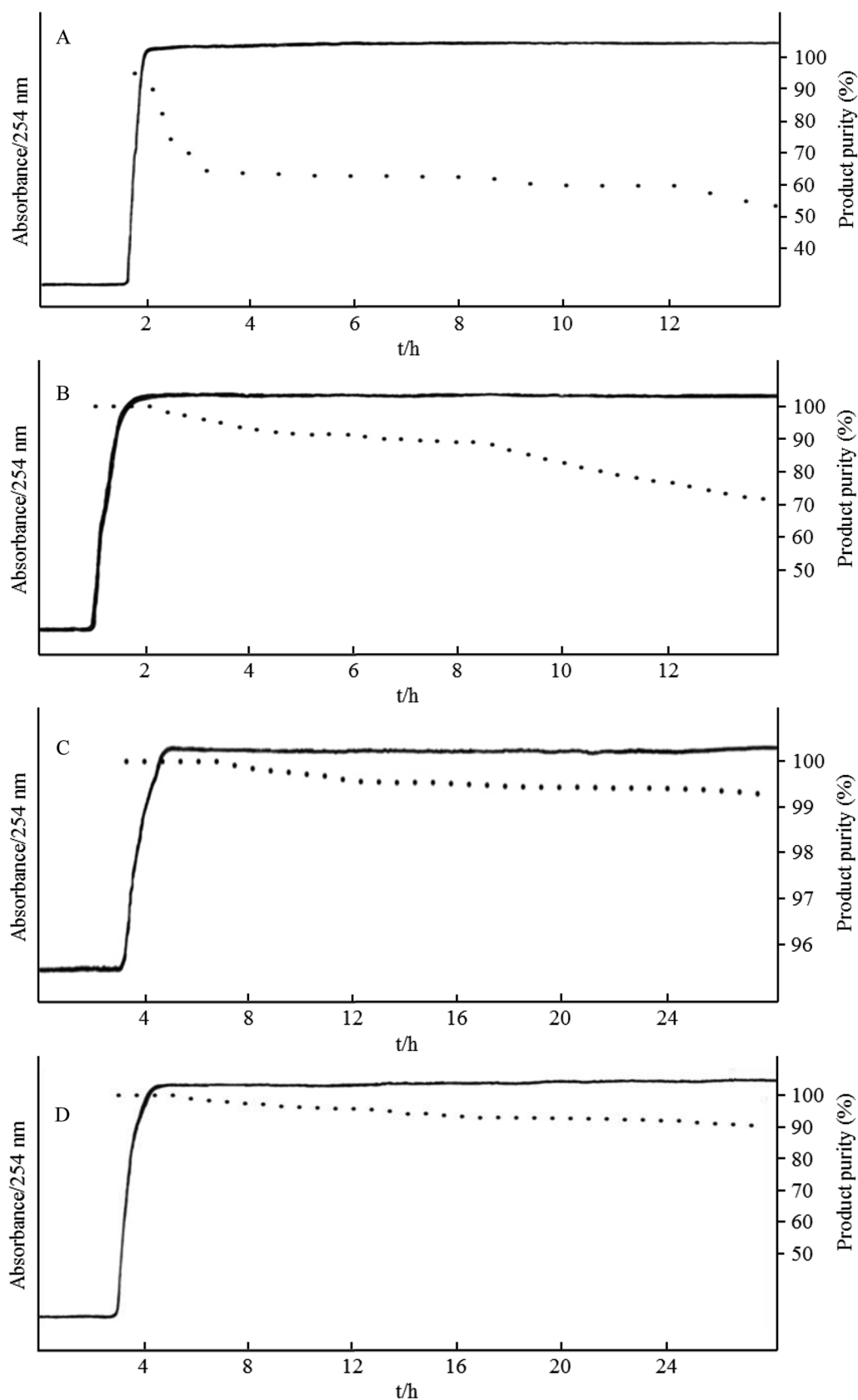


Fig. 5 Effect of the HSCCC bioreactor at different flow rates. (A) 3.0 mL min<sup>-1</sup>, (B) 2.5 mL min<sup>-1</sup>, (C) 2.0 mL min<sup>-1</sup>, and (D) 1.0 mL min<sup>-1</sup>. Solvent system: ethyl acetate/buffer solution (1 : 1, v/v). Enzyme: 750 mg L<sup>-1</sup> of  $\beta$ -glucosidase. Substrate: 62.5 mg L<sup>-1</sup> of genistin. Separation mode: FWD-OUT. Rotation speed: 800 rpm. Detection wavelength: 254 nm. Biotransformation temperature: 40 °C.



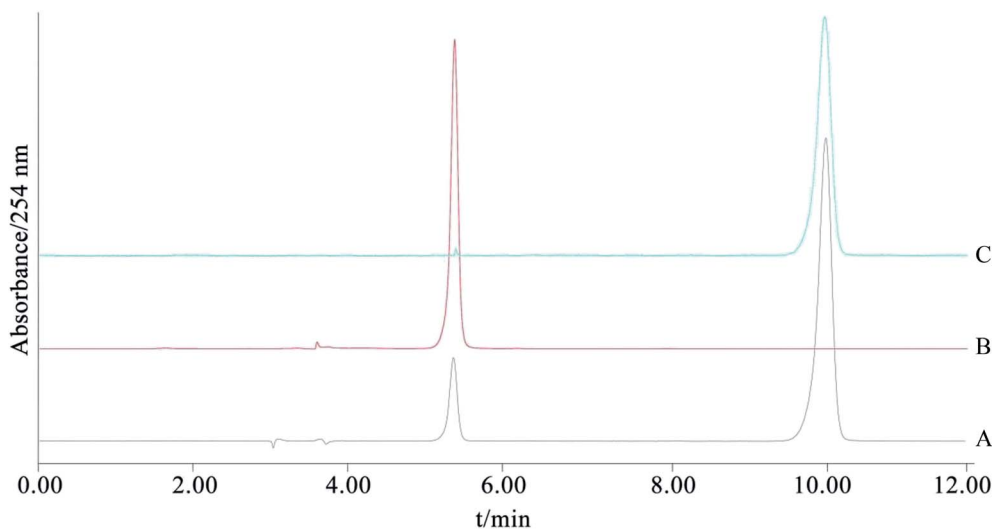


Fig. 6 HPLC chromatograms of the genistin and genistein. (A) Mixture of genistin and genistein, (B) genistin, and (C) genistein. Experimental conditions: Waters Symmetry  $C_{18}$  column (5  $\mu\text{m}$ , 4.6 mm  $\times$  250 mm, i.d.), mobile phase: methanol/water (70 : 30, v/v), flow-rate: 1.0 mL  $\text{min}^{-1}$ , detection wavelength: 254 nm, and injection volume: 10  $\mu\text{L}$ .

stationary phase, accumulates to high concentration, and results in the by-product inhibition of  $\beta$ -glucosidase.

### 3.5 Identification of the product

ESI-MS  $m/z$ : 271.2415  $[\text{M} + \text{H}]^+$ .  $^1\text{H-NMR}$  (DMSO- $d_6$ , 400 MHz)  $\delta$ : 12.93 (1H, s, OH-5), 10.83 (1H, s, OH-7), 9.55 (1H, s, OH-4'), 8.29 (1H, s, H-2), 7.35 (2H, d,  $J = 8.5$  Hz, H-2', 6'), 6.80 (2H, d,  $J = 8.5$  Hz, H-3', 5'), 6.36 (1H, d,  $J = 2.0$  Hz, H-8), 6.20 (1H, d,  $J = 2.0$  Hz, H-6). After comparing the data with the reported values,<sup>20</sup> the chemical structure was identified as genistein.

## 4. Conclusions

In the present study, a novel HSCCC bioreactor was established for the efficient biotransformation of genistein from genistin. A suitable solvent system for the HSCCC bioreactor was chosen according to the enzyme activity (*i.e.*  $K_D$  values for the substrate and product). The HSCCC bioreactor parameters were selected as FWD-OUT and 800 rpm with a retention of the stationary phase as the index. Using the optimal bioreactor parameters, different flow rates for the substrate solution were investigated, and 2.0 mL  $\text{min}^{-1}$  was chosen as the optimal flow rate with a conversion rate of over 90% within 24 h. The established bioreactor was a novel type of immobilized enzyme reactor with the advantages of simple operation as well as the ability to immobilize and recycle the enzyme; moreover, the established bioreactor can be applied in the preparation of genistein and other biotransformations of natural products.

## Conflicts of interest

All authors declare no conflicts of interest.

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