RSC Advances



Check for updates

Cite this: RSC Adv., 2019, 9, 11922

Received 5th February 2019 Accepted 30th March 2019

DOI: 10.1039/c9ra00956f rsc.li/rsc-advances

Introduction

For actuator applications, electrostriction is an important property which describes how the material deforms when polarization develops inside.¹ Electrostriction occurs in materials with either centro or noncentrosymmetric crystal structures, and is caused by a slight displacement of ions (or complexes) in the crystal lattice when the material is subjected to an applied field.² In the case of electrostriction, the field-induced strain (*S*) can be simply related to the polarization (*P*) by: $S = Q_{33}P^2$, where Q_{33} represents the electrostrictive coefficient.^{2–5} A large Q_{33} value of ~0.020 m⁴ C⁻² with a good hysteresis-free electrostrictive strain (~0.1%) can be observed in some lead-based relaxor ferroelectrics, such as Pb(Mg_{1/3}Nb_{2/3}) O₃ (PMN),³ which is known as an important electrostrictive material.⁵ However, because of the concern for environmental problems that lead-based materials pose, the development of

Pharatree Jaita,^{ab} Ratabongkot Sanjoom,^c Narumon Lertcumfu^a and Gobwute Rujijanagul[®]*^{ab}

In this research, the effects of Ba(Fe_{0.5}Ta_{0.5})O₃ (BFT) additive on the phase evolution, the dielectric, ferroelectric, piezoelectric, electric field-induced strain responses, and energy storage density of the Bi_{0.5}(Na_{0.80}K_{0.20})_{0.5}TiO₃-0.03(Ba_{0.70}Sr_{0.03})TiO₃ (BNKT-0.03BSrT) ceramics have been systematically investigated. The ceramics have been prepared by a solid-state reaction method accompanied by two calcination steps. X-ray diffraction indicates that all ceramics coexist between rhombohedral and tetragonal phases, where the tetragonal phase becomes dominant at higher BFT contents. The addition of BFT also promotes the diffuse phase transition in this system. A significant enhancement of electric field-induced strain response ($S_{max} = 0.42\%$ and $d_{33}^* = 840$ pm V⁻¹) is noted for the x = 0.01 ceramic. Furthermore, the giant electrostrictive coefficient ($Q_{33} = 0.0404$ m⁴ C⁻²) with a giant normalized electrostrictive coefficient ($Q_{33}/E = 8.08 \times 10^{-9}$ m⁵ C⁻² V⁻¹) are also observed for this composition (x = 0.01). In addition, the x = 0.03 ceramic shows good energy storage properties, *i.e.* it has a high energy storage density (W = 0.65 J cm⁻³ @ 120 °C) with very high normalized storage energy density ($W/E = 0.13 \ \mu$ C mm⁻²), and good energy storage efficiency ($\eta = 90.4\%$ @ 120 °C). Overall, these results indicate that these ceramics are one of the promising candidate piezoelectric materials for further development for actuator and high electric power pulse energy storage applications.

lead-free electrostrictive piezoelectric materials has been urgent in recent years.^{3,5,6}

Among many lead-free piezoelectric materials, (Bi_{0.5}Na_{0.5}) TiO₃-based (BNT) ceramic is an interesting material due to its relatively excellent electrical properties, especially for ferroelectric and piezoelectric properties7 *i.e.* high electric energy densities⁸ and ultra-high strain under a high electric field (\geq 50 kV cm⁻¹).⁹ However, the large coercive electric field ($E_c = 7.3$ kV mm^{-1}) and high electrical conductivity made it difficult to be polarized during the poling process, which often leads to undesirable piezoelectricity ($d_{33} = 73-95 \text{ pC N}^{-1}$) and limited its applications.¹⁰ In order to reduce the E_c and/or improve the electrical properties, researches on composition-modified BNT systems have been conducted such as BNT-BT,11,12 BNT-ST,13 BNT-BZT-SBT,14 and BNT-BKT.15,16 Ullah et al.17 studied the effect of (Ba_{0.70}Sr_{0.30})TiO₃ (BST) on the structure and electrical properties of the Bi_{0.5}(Na_{0.40}K_{0.10})TiO₃ (BNKT) ceramic. They reported that this ceramic exhibited good electrical properties *i.e.* the ceramics presented high piezoelectric coefficient $(d_{33} =$ 223 pC N⁻¹) with dielectric loss (tan δ) of 3% at 1 kHz. They also found that the La additive interrupted the P-E hysteresis loops of the BNKT-BST ceramics, which lead to a reduction in P_r and $E_{\rm c}$ values.¹⁸ However, the destabilization of the ferroelectric order corresponded to a significant increase of the unipolar strain which showed the highest value of $\sim 0.39\%$ and



View Article Online

View Journal | View Issue

^aDepartment of Physics and Materials Science, Faculty of Science, Chiang Mai University, Chiang Mai 50200, Thailand. E-mail: rujijanagul@yahoo.com

^bScience and Technology Research Institute, Chiang Mai University, Chiang Mai 50200, Thailand

Department of Applied Science and Biotechnology, Faculty of Agro-Industrial Technology, Rajamangala University of Technology Tawan-ok Chanthaburi Campus, Chanthaburi 22210, Thailand

Improvement of electric field-induced strain and energy storage density properties in lead-free BNKT-based ceramics modified by BFT doping

corresponding a normalized strain $(d_{33}^* = S_{\text{max}}/E_{\text{max}})$ of 650 pm V^{-1} .¹⁸ Therefore, research on modified BNKT–BST ceramics is an important topic in order to develop lead-free ceramics with high electrical properties.

An interesting work for electrostrictive materials was performed by Zhang et al.¹⁹ who provided a research direction to develop the $(1 - x)[0.94Bi_{0.5}Na_{0.5}TiO_3 - 0.06BaTiO_3] - xK_{0.5}Na_{0.5}$ NbO₃ system with a large electrostrictive coefficient ($Q_{33} \sim$ 0.021–0.027 m⁴ C⁻²) and the maximum strain ($S_{\rm max} \sim 0.12$ – 0.30%) values simultaneously. Beside Zhang et al.'s work, a large electrostrictive effect with $Q_{11} = 0.0297 \text{ m}^4 \text{ C}^{-2}$ was also obtained by Shi et al.²⁰ in the $(0.94 - x)(Bi_{0.5}Na_{0.5})TiO_3$ -0.06BaTiO₃- $x(Sr_{0.7}Bi_{0.2}\Box_{0.1})TiO_3$ system, in which its Q_{11} exhibits a high temperature stability between ambient temperature to 180 °C.²⁰ Moreover, a high strain ($S_{\rm max} \sim 0.45\%$) was also found in the (Bi_{0.5}Na_{0.5})TiO₃-BaTiO₃-(K_{0.5}Na_{0.5})NbO₃ or BNT-BT-KNN ceramic by Zhang et al.,²¹ which is even higher than the strain value obtained in the lead ferroelectric Pb(Zr,Ti) O3 ceramic. However, due to the phase evolution in BNT-based materials, large strain hysteresis larger than 60% can be observed inherently, which restricts the practicability use as electrostrictive materials.4 Thus research on the reduction of the strain hysteresis for the lead-free materials such as BNKT-BST based materials is an interesting issue for the lead-free actuators.

Recently, many authors have suggested that the iron oxide such as Fe₂O₃ additive can improve the piezoelectric and magnetic properties of some lead-free piezoelectric ceramics.^{22,23} In the case of piezoelectric properties, Jaita et al.²³ reported that the Fe₂O₃ additive can enhance the electric fieldinduced strain of BNKT-based ceramics. However, the electrostriction and electric field-induced strain properties of lead-free piezoelectric materials such as BNT-based ceramics doped with complex perovskite materials containing Fe have not been widely investigated. For the Fe-containing complex perovskite material, *i.e.* barium iron tantalate (BaFe_{0.5}Ta_{0.5}O₃; BFT) is interesting because it exhibits a giant dielectric material which has attracted great attention due to its very high dielectric constant ($\varepsilon_r \sim 1.9 \times 10^5$ at 550 °C and 1 kHz).²⁴ Many works have reported that BFT, which was first synthesized by Galasso et al.,²⁵ presents a cubic perovskite-type structure with space group $Pm\bar{3}m$ (221)²⁴⁻²⁶ and the lattice parameter a = 4.056 Å.^{24,27} BFT can be used to modify other materials such as BiFeO₃²⁴ and $Ba(Zr_{0.05}Ti_{0.95})O_3$ (BZT),²⁷ and the properties of the modified ceramics have been improved. For example, the dielectric constants of the BZT-BFT solid solution were enhanced by BFT.28

In the present research, new lead-free of $Bi_{0.5}(Na_{0.80}-K_{0.20})_{0.5}TiO_3-0.03(Ba_{0.70}Sr_{0.03})TiO_3$ doped with $Ba(Fe_{0.5}Ta_{0.5})O_3$ ceramics were synthesized with the aim of improving their electrical properties. The role of BFT content on the ceramic properties including phase formation, microstructure, electrical properties (*i.e.* dielectric, ferroelectric and piezoelectric, and the electric field-induced strain behavior) of the BNKT-0.03BSrT ceramic were investigated. Furthermore, since many recent works have focused on energy storage materials due to

global energy problems, thus the storage energy density behavior of the presented ceramics was also investigated.

Experimental

The conventional mixed oxide technique was used to synthesize the $(1 - x)[Bi_{0.5}(Na_{0.80}K_{0.20})_{0.5}TiO_3 - 0.03(Ba_{0.70}Sr_{0.03})TiO_3]$ $xBa(Fe_{0.5}Ta_{0.5})O_{3}$ or (1 - x)[BNKT-0.03BSrT]-xBFT powders. The analytical grade reagents of metal oxide powders, including Na₂CO₃, Bi₂O₃, K₂CO₃, TiO₂, Fe₂O₃, SrCO₃, BaCO₃, and Ta₂O₅ were used as starting materials. All carbonate powders were first dried at 120 °C for 24 h in order to eliminate any moisture. The raw materials of BNKT-0.03BSrT and BFT were stoichiometrically weighed, ball milled for 24 h in an ethanol solution, and then dried in an oven. Since there is a large difference in calcination temperatures between BNKT-0.03BSrT (900 °C) and BFT (1200 °C) powders, two calcination steps were employed *i.e.* the BNKT-0.03BSrT and BFT powders were synthesized separately. The resulting powders were weighed, mixed and then dried to obtained a powder of (1 - x)[BNKT-0.03BSrT]-xBFT with x = 0, 0.0, 0.02 and 0.03 mol fraction. A few drops of polyvinyl alcohol (PVA) binders (4 wt%) were added to the obtained powders before being uniaxially pressed into discs 10 mm in diameter. The green pellets were sintered at 1125 °C for 2 h by using a heating and cooling rate of 5 $^{\circ}$ C min⁻¹.

Bulk density was determined by the Archimedes' method. A scanning electron microscope (SEM, JEOL JSM-6335F) was used to study microstructural features of the ceramics. An X-ray diffractometer (XRD-Phillip, X-pert) was used to study the phase evolution of all ceramics. Grain size of the ceramics was carried out by using linear intercept method (ASTM no. E112-88). Before investigating the electrical measurements, all samples were polished into a parallel surface with 1 mm thickness. Silver paste was subjected onto both sides of the sample. Then the samples were heated at 700 °C for 30 min to form electrodes. Dielectric properties as a function of temperature (25–400 °C) were carried out using a LCR-meter (HP model 4192A) at frequencies ranging from 1 to 1000 kHz. The ferroelectric properties were carried out by a Radiant Precision ferroelectric tester both at room temperature (RT) and high temperatures (HT) of 25–150 °C. A maximum electric field of 50 kV cm⁻¹ at a frequency of 1 Hz was applied to each studied sample. Remanent polarization (P_r) , maximum polarization (P_{max}) , and coercive field (E_c) values were determined from the hysteresis loops or P-E loops. By using data from ferroelectric properties, the energy storage density (W) and energy storage coefficient (η) values were also calculated. Strain–electric field (S–E) data at RT were carried out using an optical displacement sensor (Fotonic Sensor model MTI-2100) injunction with a Radiant ferroelectric test system. A maximum electric field of 50 kV cm⁻¹ and a frequency of 0.1 Hz were used to measure the bipolar and unipolar strain curves. The maximum strain (S_{max}) and the negative strain (S_{neg}) values were carried out from the bipolar curve. The normalized strain coefficient d_{33}^* was also determined following: $d_{33}^* = S_{\text{max}}/E_{\text{max}}$. The electrostrictive coefficient (Q_{33}) was determined from *S*-*P* curves using equation: *S* = QP^2 . For piezoelectric measurement, all samples were poled at RT with an applied DC electric field of 5 kV mm⁻¹ (for 15 min) and using silicone oil as the media. The low-field piezoelectric coefficient (d_{33}) was determined by a d_{33} -meter.

Results and discussion

Phase formation and microstructure

Fig. 1(a) shows the X-ray diffraction patterns with different doping contents of the (1 - x)[BNKT-0.03BSrT]-xBFT ceramics. Within the resolution limit of XRD, all ceramics exhibit a single phase of perovskite structure with no secondary phase, confirming that BFT has been incorporated into the BNKT-0.03BSrT lattice to form solid solutions of the end compounds. For analyzing the phase transition process, the XRD patterns for selected narrow angular ranges of $2\theta = 39-41^{\circ}$ and $2\theta = 44-48^{\circ}$ are presented in Fig. 1(b) and (c), respectively. The x = 0 ceramic presents a mixed phase of rhombohedral and tetragonal structure, which is demonstrated by a slight splitting of the $(111)_{\rm R}/(1\overline{1}1)_{\rm R}$ at $2\theta \sim 40^{\circ}$ and splitting of the $(200)_{\rm T}/(002)_{\rm T}$ peaks at $2\theta \sim 46^{\circ}$. This result also agrees with the result observed by Ge et al.⁶ However, the coexistence of the mixed rhombohedral and tetragonal phases transforms into a tetragonal-rich phase at higher BFT contents. This can be evidenced by, the existence of the splitting $(200)_T/(002)_T$ peaks, and the merger of the $(111)_{\rm R}/(1\overline{1}1)_{\rm R}$ peak into a single $(111)_{\rm PC}$ peak, thus indicating that the amount of tetragonal phase is higher than that of the rhombohedral phase, with increasing BFT content. To check the appearance of the phase transformation in more detail, the tetragonality ratio (c/a) was determined and its values are listed in Table 1. The c/a value increases from 1.0105 for the x =





0 ceramic to 1.0128 for the x = 0.03 ceramic, as expected. Furthermore, the diffraction peaks gradually shift to a lower angle with increasing BFT content. This can be caused by the differences in the ionic radii between Na⁺, Bi³⁺, K⁺, Sr²⁺, and Ba²⁺ at the A-site and Ti⁴⁺, Fe³⁺, and Ta⁵⁺ at the B-site²⁹ which, as a result, can induce a structural distortion such as an enlargement of unit cell size, as shown in Fig. 2. A similar peak slightly shifted by partial substitution of Ba²⁺ for [Bi_{0.5}(Na_{0.80}K_{0.20})]²⁺ and Zr⁴⁺ for Ti⁴⁺ was reported by Chen *et al.*³⁰ in the BNKT-BZT system. They also found that the (110) peak shifted to lower 2 θ angles could produce an increase in the lattice constant and the unit cell dimension with increasing BZT fraction.

SEM micrographs and their corresponding histogram of grain size distribution of the (1 - x)[BNKT-0.03BSrT]-xBFT ceramics are shown in Fig. 3. The average grain size values are also summarized in Table 1. The SEM result confirms that all samples are dense with relatively high densities $(5.80-5.86 \text{ g cm}^{-3})$. Furthermore, the porosity levels evident in the micrographs are noted to be consistent with the trend of measured density value, where the density increases with increasing BFT content. Most grains of all samples show a clear grain boundary with round and cubic shapes. The addition of BFT seems to have a slight influence on the microstructure as well as the average grain size of the BNKT-BSrT ceramics. The average grain size values of all samples are rather similar (0.56-0.61 µm) and the grain size distribution shows a monomodal normal distribution. However, a slight narrow grain size distribution is noted for a ceramic containing a higher amount of BFT. This suggests that the microstructure can be improved by a BFT additive.



Fig. 2 Plots of unit cell volume and the tetragonality (c/a) as functions of BFT content of the (1 - x)[BNKT-0.03BSrT]-xBFT ceramics.

Table 1	1 Phase, physical, microstructure and dielectric properties of the $(1 - x)[BNKT - 0.03BSrT] - xBFT$ ceramics	

x	Density (g cm ⁻³)	Relative density (%)	c/a	Unit cell volume (Å ³)	Grain size (μm)	$T_{\rm m}$ (°C)	ε _{max} (@1 kHz)	$\delta_{\gamma}{}^{a}\left(\mathrm{K}\right)$	$\delta_{\gamma}^{\ b}(\mathrm{K})$
0	5.80	98	1.0105	60.18	0.61	316	5156	83	73
0.01	5.83	98	1.0117	60.28	0.59	303	5151	91	94
0.02	5.85	99	1.0121	60.43	0.58	292	4814	94	117
0.03	5.86	99	1.0128	60.46	0.56	284	4716	105	123

^a Poled sample. ^b Unpoled sample.



Fig. 3 SEM micrographs and their corresponding histograms of grain size distribution of the (1 - x)[BNKT - 0.03BSrT] - xBFT ceramics where (a) x = 0, (b) x = 0.01, (c) x = 0.02, and (d) x = 0.03.

Dielectric properties

Temperature dependence of the dielectric properties, including constant (ε_r) and dielectric loss (tan δ) of the poled (1 – *x*) [BNKT-0.03BSrT]-*x*BFT ceramics measured at different frequencies from 1–1000 kHz are shown in Fig. 4, and the related values are also listed in Table 1. It can be seen that the ε_r *versus* temperature (ε_r -*T*) curves of all samples are rather similar. The ε_r -*T* curves for all ceramics exhibit two dielectric anomaly peaks at ~100 °C and 300 °C.³¹⁻³³ Normally, the high temperature anomaly peak is called *T*_m, where the dielectric constant reaches its maximum value.³⁴⁻³⁶ The lower anomaly peak is located near *T*_{F-R}, which is known as the ferroelectric (FE) to ergodic relaxor (ER) phase transition temperature.⁶ Normally, the *T*_{F-R} can be usually carried out from the peak of tan δ -*T* curve of a poled sample. However, the *T*_{F-R} could not be detected from the tan δ -*T* curves for this work. The *T*_m and ε_{max}



Fig. 4 Temperature dependence of the dielectric constant (ε_r) and dielectric loss (tan δ) of the poled (1 - x)[BNKT-0.03BSrT]-xBFT ceramics measured at various frequencies from 1–1000 kHz where x = 0-0.03.

values of the x = 0 ceramic in this study are observed from Fig. 4 to be 316 and 5156, respectively. The ε_{max} decreases with increasing BFT content. Furthermore, the T_{m} also shifts from 316 °C for the x = 0 ceramic to 284 °C the x = 0.03 ceramic. This result also agrees with the results of previous report for other BNKT-based ceramics.³⁴

To study the diffuse phase transition in this system, the diffuseness of the phase transition was calculated by a modified Curie–Weiss law eqn (1) which is given by ref. 37 and 38.

$$\frac{\varepsilon_{\max}}{\varepsilon_{\rm r}} = 1 + \frac{(T - T_{\rm m})^2}{2\delta_{\gamma}^2},\tag{1}$$

where δ_{γ} is the diffuseness parameter. The δ_{γ} value can be carried out from the curve of $\ln(\varepsilon_{\max}/\varepsilon_{\rm r} - 1) \nu s. \ln(T - T_{\rm m})^2$, where the curve should be linear.³⁷ The δ_{γ} is used to compare the degree of diffuseness of the phase transition for many materials. The δ_{γ} values for poled and unpoled ceramics are summarized in Table 1. Based on Table 1, it can be seen that the addition of BFT has an effect on the degree of diffuseness, where the δ_{γ} value increases with increasing BFT content. This result corresponds to the previous work,³⁹ which indicates that the addition of BFT produces a higher degree of diffuseness for the phase transition in the modified BNKT ceramics.

Polarization and strain behaviors

The *P–E* hysteresis and bipolar strain–electric field (*S–E*) loops measured at room temperature (RT) and electric field of 50 kV cm⁻¹ are shown in Fig. 5 and 6, respectively. The related values are also listed in Tables 2 and 3. The x = 0 ceramic displays a small pinched loop with the maximum values of $P_{\text{max}} = 37.70$ μ C cm⁻², $P_r = 21.74 \ \mu$ C cm⁻² and $E_c = 12.87 \ \text{kV cm}^{-1}$. The *S–E* curve for this sample also exhibits a butterfly-shaped loop. The S_{max} and d_{33}^* of this ceramic are 0.33% and 650 pm V⁻¹, respectively. The ferroelectric loop indicates a ferroelectric domain switching at E_c , where the largest S_{neg} is -0.11%. However, this ceramic sample should contain a mixed phase



Fig. 5 Polarization–electric field (*P*–*E*) hysteresis loops and current– electric field (*I*–*E*) loops of the (1 - x)[BNKT-0.03BSrT]-xBFT ceramics where x = 0-0.03, measured at RT under an electric field of 50 kV cm⁻¹ and a frequency of 1 Hz.



Fig. 6 Bipolar strain–electric field (S-E) loops of the (1 - x)[BNKT– 0.03BSrT]–xBFT ceramics where x = 0-0.03, measured at RT under an electric field of 50 kV cm⁻¹ and a frequency of 0.1 Hz.

between FE and ER, since the *P*–*E* loop presents a small pinched shape. For the x = 0.01 ceramic, however, the *P*–*E* hysteresis loop exhibits a more pinched loop with a decrease of P_r from 21.74 µC cm⁻² to 8.15 µC cm⁻² and E_c from 12.87 kV cm⁻¹ to 7.34 kV cm⁻¹. The hysteresis loop shows a clear pinching-type character with further increasing BFT content, similarly observed in previous works.^{6,35} The pinched *P*–*E* loops corresponded with the two current peaks (denoted as "1" and "2") in the *I*–*E* curves (Fig. 5), where it has been suggested that this evidence indicates a formation of ER phase.^{14,36}

The more pinched *P*–*E* loops result also corresponds to a large electric field-induced strain response with S_{max} of 0.42% and d_{33}^* of 840 pm V⁻¹ along with a drastic reduction in S_{neg} from -0.11% to 0% for the x = 0.01 ceramic in bipolar *S*–*E* loops. This enhancement is also confirmed by the unipolar *S*–*E*



0.00 0.01 0.02 0.03

0.02

0

0.03

xBFT

Fig. 7 Plots of (a) unipolar strain–electric field (*S*–*E*) loops of the (1 – *x*) [BNKT–0.03BSrT]–*x*BFT ceramics where x = 0–0.03, measured at RT under an electric field of 50 kV cm⁻¹ and a frequency of 0.1 Hz and (b) plots of the maximum strain (S_{max}) and normalized strain coefficient ($d_{33}^* = S_{max}/E_{max}$) as a function of BFT content (inset: Q_{33} value as a function of BFT content).

xBFT

0.01

0.15

0.00

loops, as shown in Fig. 7(a). It is seen that the unipolar *S*–*E* loops for all compositions show a similar trend as compared to that of the bipolar *S*–*E* loops, where the maximum strain value of the unipolar strain is 0.42% for the x = 0.01 ceramic (Fig. 7(b)). The pinching of the *P*–*E* hysteresis loops with rapid reductions of P_r and E_c can be correlated with the increase of S_{max} along with a drastic reduction in S_{neg} . This may be related

Table 2Ferroelectric and energy storage density properties of the (1 - x)[BNKT - 0.03BSrT] - xBFT ceramics

x	$P_{\rm max} \left(\mu {\rm C} \ {\rm cm}^{-2} \right)$	$P_{\rm r} \left(\mu {\rm C} ~{\rm cm}^{-2} \right)$	$E_{\rm c} (\rm kV cm^{-1})$	W^a (J cm ⁻³)	W^b (J cm ⁻³)	W^{c} (J cm ⁻³)	η^{a} (%)	η^{b} (%)	η^{c} (%)
0	37.70	21.74	12.87	0.20	0.54	0.60	20.0	69.5	81.2
0.01	35.65	8.15	7.34	0.37	0.61	0.61	38.6	80.0	87.4
0.02	32.78	4.26	5.66	0.45	0.63	0.62	51.7	87.9	92.0
0.03	29.18	3.15	4.77	0.49	0.65	0.63	60.7	90.4	92.1

^{*a*} Ferroelectric data obtained at high temperature of 25 $^{\circ}$ C and a frequency of 1 Hz. ^{*b*} Ferroelectric data obtained at high temperature of 120 $^{\circ}$ C and a frequency of 1 Hz. ^{*c*} Ferroelectric data obtained at high temperature of 150 $^{\circ}$ C and a frequency of 1 Hz.

Table 3 Piezoelectric properties of the (1 - x)[BNKT - 0.03BSrT] - xBFT ceramics

x	S_{\max} (%)	$S_{ m neg}$ (%)	$d_{33}^* (\mathrm{pm}\mathrm{V}^{-1})$	Н (%)	$Q_{33} (m^4 C^{-2})$	$d_{33} ({ m pC} { m N}^{-1})$
0	0.33	-0.11	650	78.9	0.0376	172
0.01	0.42	0	840	55.0	0.0404	38
0.02	0.37	0	722	44.6	0.0383	29
0.03	0.32	0	654	39.3	0.0368	28

with a phase transition from a ferroelectric (FE) to an ergodic relaxor (ER) phase, accompanied by the disruption of longrange ferroelectric order with increasing BFT content.6 Normally, the free energy of FE and ER phases is comparable under zero field, thus it can be easily induced by an external electric field to become saturated.³⁶ Therefore, ER phase can be transformed reversibly into FE phase by an external electric field. This behaviour consists with the work done by Dong et al.⁴⁰ who also reported that the $(1 - x)(0.8Bi_{1/2}Na_{1/2}TiO_3 - 0.2Bi_{1/2}K_{1/2}K_{1/2})$ 2TiO3)-xBiMg2/3Nb1/3O3 or (BNT-BKT)-BMN system produced a large strain response when a small amount of BMN was added into the BNT-BKT ceramic. The addition of 2 mol% BMN into the BNT-BKT ceramic shows a pronounced ergodic relaxor (ER) characteristic, which is confirmed by the absence of S_{neg} , the pinched P-E loop and the double current peaks of I-E loop. However, in the present work when x > 0.01, the strain is reduced gradually. Therefore, the small amount of BFT additive has effects on the piezoelectric properties of the BNKT-0.03BSrT ceramics. It should be noted that the S_{max} in this work (x = 0.01) is considered high when compared with other leadfree piezoelectric ceramics (see Table 4).18,41-48 Furthermore, the bipolar strain loops for the present work exhibit a low antisymmetry in their shape compared to many previous works for the giant strain response which showed high anti-symmetry.42,45 In the present work, an average maximum strain (\bar{S}_{max}) was calculated to be used as a parameter to check for overall averaged maximum strain of the bipolar strain loop, where \bar{S}_{max}

averaged from maximum strain between the right and left hand sides of the bipolar strain loop. The result is also shown in Table 4. The obtained \bar{S}_{max} for the present work is considered high for the lead-free piezoelectric ceramics.

The Q_{33} was determined from the slope of *S* vs. P^2 curve.^{49,50} The plotting of Q_{33} value as a function of BFT content is shown in inset of Fig. 7(b). The highest Q_{33} of 0.040 m⁴ C⁻² is noted for the x = 0.01 ceramic. It is evident that the currently presented Q_{33} value is highly competitive to many materials.⁵¹⁻⁵⁸ In addition, a comparison of the room temperature Q_{33} and normalized Q_{33} (Q_{33}/E) values with other lead-based and lead-free BNTbased electrostrictive materials are summarized in Table 5.⁵¹⁻⁵⁸ It should be noted that the Q_{33}/E value for the present work is very high for both lead-based and lead-free BNT-based piezoelectric materials. Therefore, the (1 - x)[BNKT-0.03BSrT]-xBFTcompound may be a promising candidate for new electromechanical devices due to its giant Q_{33} value.

Normally for the actuator applications, a large strain with low hysteresis is necessary. The degree of strain hysteresis (*H*) can be determined from the following equation:^{38,59}

$$H = \frac{\Delta S_{\underline{E_{\text{max}}}}}{S_{\text{max}}} \times 100 \%, \qquad (2)$$

where the $\Delta S_{E_{max}/2}$ is the strain deviation during the application and removal of the field which is carried out at half of the maximum electric field. The *H* values of all ceramics are summarized in Table 3. It was found that the *H* decreases with

Systems	S_{\max} (%)	\bar{S}_{\max} (%)	$d_{33}^* \text{ (pm V}^{-1}\text{)}$	$E_{\rm max}$ (kV cm ⁻¹)	Ref.
0.99[BNKT-0.03BSrT]-0.01BFT	0.42	0.42	827	50	This worl
BNKT-BST-La _x	0.39	0.39	650	60	18
$Bi_{0.5}(Na_{0.78}K_{0.22})_{0.5}(Ti_{1-x}Z_x)O_3, x = 0.03$	0.43	0.43	614	70	41
$Bi_{1/2}(Na_{0.8}K_{0.2})_{1/2}](Ti_{1-x}Ta_x)O_3, x = 0.015$	0.62	0.42	1240	50	42
BNKT-BST-Nb _x , $x = 0.02$	0.38	0.38	634	60	43
Bi _{0.5} (Na _{0.78} K _{0.22}) _{0.5} TiO ₃ -0.01(Bi _{0.5} La _{0.5})AlO ₃	0.40	0.40	579	70	44
$((Bi_{1/2}(Na_{0.84}K_{0.16})_{1/2})_{0.96}Sr_{0.04})(Ti_{1-x}Nb_x)O_3, x = 0.025$	0.70	0.39	1400	50	45
Bi _{0.5} (Na _{0.78} K _{0.22}) _{0.5} TiO ₃ -0.03BiFeO ₃	0.37	0.36	676	50	46
Bi _{0.5} (Na _{0.80} K _{0.20}) _{0.5} TiO ₃ -0.05Ba(Ti _{0.90} Sn _{0.10})O ₃	0.36	0.36	649	55	47
[0.93(Bi _{0.5} Na _{0.5})TiO ₃ -0.07BaTiO ₃]-Pr	0.43	0.42	770	50	48

Table 4 Comparison of S_{max} and d_{33}^* values of the (1 - x)[BNKT - 0.03BSrT] - xBFT (x = 0.01) with other lead-free ceramics

Systems	S_{\max} (%)	$Q_{33} (m^4 C^{-2})$	$Q_{33}/E_{ m max} \left({ m m}^5~{ m C}^{-2}~{ m V} ight) imes 10^{-9}$	E (kV cm ⁻¹)	Ref.
0.99[BNKT-0.03BSrT]-0.01BFT	0.42	0.0404	8.08	50	This work
$Pb(Mg_{1/2}Nb_{1/2})O_3$	_	0.023	_	_	51
PLZT 8/65/35	0.16	0.015	5.00	30	52
0.96BNKT-0.04BNiT	0.38	0.0250	4.17	60	53
0.95BNKT-0.05BNiT	0.32	0.0300	5.00	60	53
N44B48T94-6BT [or $(Na_y, Bi_z)Ti_{1-x}O_3(1 - x)-xBaTiO_3$]	0.48	0.0260	3.71	70	54
8% Sn-doped Bi _{1/2} (Na _{0.82} K _{0.18}) _{1/2} TiO ₃	0.14	0.0230	3.83	60	55
BNKT100y-xKNN ($y = 0.20, x = 0.16$)	0.10	0.0250	3.12	80	56
(0.94 - x)BNT-0.06BT-xKNN, $x = 0.20$	0.09	0.0260	3.25	80	57
$Bi_{0.5-x}La_xNa_{0.40}K_{0.10}Ti_{0.98}Zr_{0.02}O_3$, $x = 0.02$	0.42	0.0360	6.00	60	58

Table 6 Comparisons of the degree of strain hysteresis (*H*) value of the (1 - x)[BNKT - 0.03BSrT] - xBFT (x = 0.01 and x = 0.03) with other lead-free ceramics

Systems	Н (%)	S_{\max} (%)	$d_{33}^* \; (\mathrm{pm} \; \mathrm{V}^{-1})$	Ref.
0.99[BNKT-0.03BSrT]-0.01BFT	55.0	0.42	840	This worl
0.97[BNKT-0.03BSrT]-0.03BFT	39.3	0.32	654	This worl
$Bi_{1/2}(Na_{0.82}K_{0.12})_{1/2}Ti_{1-x\%}(Fe_{0.5}Nb_{0.5})_{x\%}O_3, x = 5\%$	57.0	0.46	660	32
BNKT20-1.5SZ	51.6	0.39	488	60
BNT–BKT–5La	60.3	0.38	857	61
BNT-25ST	45.8	0.29	650	62
BNKT–3 mol% CZ	49.0	0.37	603	63
BNKT-2 mol% BCZ	25.0	0.30	549	63
(Ho,Sb)-modified (Bi _{0.5} Na _{0.5}) _{0.945} Ba _{0.065} TiO ₃	39.0	0.37	463	64
BNBT6.5–100 <i>x</i> ES, $x = 0.50\%$	39.0	0.40	500	65
BNT-BKT- x BZT, $x = 0.02$	40.0	0.32	503	66

 Open Access Article. Published on 16 April 2019. Downloaded on 7/16/2025 11:57:22 PM.

 Y
 This article is licensed under a Creative Commons Attribution 3.0 Unported Licence.

increasing BFT concentration. The lowest hysteresis of 39.3% is obtained at x = 0.03 (a) 50 kV cm⁻¹, which reveals that BFT doping obviously reduces the strain hysteresis. A comparison of the *H* value for selected ceramic samples with other lead-free ceramics is shown in Table 6.^{32,60-66} The result clearly demonstrates that the *H* value of the present work is low when compared to that of many previous works (for the lead-free piezoceramics with d_{33}^* value > 450 pm V⁻¹). This suggests that a BFT modifier can improve the actuating performance.³⁸

(ER) state at zero electric field.^{67–69} On the other hand, the modified ceramics ($x \ge 0.01$), exhibit an ER characteristic at RT. With further increasing temperature, the drastic decrease in both P_r and E_c values may also be related to the onset of strong ergodicity.⁶⁸ Thus, both chemical modification and temperature can disrupt the FE long-range order leading to a decrease in the polarization states.⁶⁸ This phenomenon agrees with that observed in many lead-free piezoelectric materials.^{40,69} Dong

Polarization-temperature analysis

Temperature dependence of polarization–electric field (*P–E*) hysteresis loops of the (1 - x)[BNKT–0.03BSrT]–*x*BFT ceramics where x = 0–0.03, measured under an electric field of 50 kV cm⁻¹ and a frequency of 1 Hz is shown in Fig. 8. It can be seen that the *P–E* hysteresis loop for the x = 0 ceramic shows slightly pinched loops at RT (25 °C). When the temperature increases to 120–150 °C, the *P–E* hysteresis loops become more pinched, confirming that the ferroelectric to relaxor phase transition is induced by thermal activation, thus leaving an ergodic relaxor



Fig. 8 Temperature dependence on polarization–electric field (*P*–*E*) hysteresis loops of the (1 - x)[BNKT-0.03BSrT]-xBFT ceramics where x = 0-0.03, measured under an electric field of 50 kV cm⁻¹ and a frequency of 1 Hz.



Fig. 9 Plots of (a) energy storage density (*W*) and (b) energy storage coefficient (η) as a function of temperature of the (1 - x)[BNKT - 0.03BSrT] - xBFT ceramics.

Table 7	Comparison of W and W/E of the (1	 – x)[BNKT–0.03BSrT]–xBF1 	ceramics ($x = 0.03$) with other lead-free ceramics
---------	---------------------------------------	--	---

Systems	W (J cm ⁻³)	E (kV mm ⁻¹)	W/E (µC mm ⁻²)	η (%)	Ref.
·		_			
0.97[BNKT–0.03BSrT]–0.03BFT @ 120 °C	0.65	5	0.13	90.4	This work
0.96BNKT-0.04MN	0.65	7	0.09	34.2	38
0.92BNKT-0.08AN	1.41	10.5	0.13	—	70
Nb-modified 0.96(Bi _{0.5} Na _{0.84} K _{0.16} TiO ₃)-0.04SrTiO ₃	1.00	6	0.17	>70	72
0.94Bi _{0.47} Na _{0.47} Ba _{0.06} TiO ₃ -0.06KNbO ₃	0.89	10	0.09	—	73
$(0.9 - x)[0.92Bi_{0.5}Na_{0.5}TiO_3 - 0.08BaTiO_3] - xSrZrO_3 - 0.10NaNbO_3, x = 0.08$	0.95	11	0.09	66.0	74
0.92(0.65BaTiO ₃ -0.35Bi _{0.5} Na _{0.5} TiO ₃)-0.08Na _{0.73} Bi _{0.09} NbO ₃	1.70	17.2	0.10	82.0	75
BNTBT-0.20NBN	1.36	13.6	0.10	73.9	76
(Bi _{0.5} Na _{0.5})TiO ₃ -0.10KNbO ₃	1.17	10.4	0.11	—	77
0.7SrTiO ₃ -0.3(0.65BaTiO ₃ -0.35Bi _{0.5} Na _{0.5} TiO ₃)	1.40	19.6	0.07	90.0	78
0.95(BNTBT)–0.05KN	0.64	6	0.11	84	79
0.99[0.7BNT-0.3BST]-0.01NN	1.03	8.5	0.12	85.8	80
(BNT-6BT)-NBN	1.40	14.2	0.11	66.3	81

*et al.*⁴⁰ also studied the temperature dependence of *P*–*E* hysteresis loops of the lead-free $(1 - x)(0.8Bi_{1/2}Na_{1/2}TiO_3-0.2Bi_{1/2}K_{1/2}TiO_3)-xBiMg_{2/3}Nb_{1/3}O_3$ ceramics. With increasing temperature, the *P*_r and *E*_c values decrease gradually while *P*_{max} slight changed and the *P*–*E* hysteresis loop showed severe pinches. This indicated the presence of an ergodic relaxor (ER) state, *i.e.* there exists a temperature-caused ferroelectric-to-relaxor phase transformation in their work.

Energy storage density

The polarization and dielectric breakdown strength are two important parameters related to the energy storage performance. As is known to all, the BNT–BKT ceramics with high $P_{\rm r}$ of 34 µC cm⁻² and low breakdown field ($E_{\rm b} \sim 50$ kV cm⁻¹) are difficult to a satisfactory characteristic for high energy storage applications.⁷⁰ Therefore, the design of BNT–BKT energy storage materials with low $P_{\rm r}$, large $P_{\rm max}$ as well as the improvement of $E_{\rm b}$ become two critical issues.

To evaluate the capability of the studied ceramics for energy storage applications, the energy storage density (*W*) was calculated from the *P*–*E* hysteresis loop as shown in the following equation:^{70,71}

$$W = \int_{P_{\rm r}}^{P_{\rm max}} E {\rm d}P, \tag{3}$$

where *E* is the electric field, *P* the polarization, P_{max} the maximum polarization, and P_{r} is the remnant polarization. For practical applications, higher energy storage efficiencies (η) are required. The η value is calculated by using the following formula:^{68,72}

$$\eta(\%) = \frac{W}{W + W_{\rm loss}} \times 100, \tag{4}$$

where energy loss density is denoted as W_{loss} . In this work, the energy storage density properties are summarized in Table 2. At RT, the *W* and η of the x = 0 ceramic are 0.20 J cm⁻³ and 20.0%, respectively. In this composition, the high E_c and low $P_{\text{max}} - P_r$ values are observed, which led to a low energy storage density, high energy loss, and low efficiency.⁷² The x = 0.01

ceramic with a pinched P-E hysteresis loop shows relatively high W = 0.37 J cm⁻³ and $\eta = 38.6\%$ values when compared with those observed for the x = 0 ceramic. This improvement can be related with the decreases in $P_{\rm r}$, and slight invariant P_{max} with the degradation of the long range FE order and a relatively easy reorientation of the micro-domains.72 The W and η values increase with increasing BFT content and exhibit the maximum values of 0.49 J cm⁻³ and 60.7% at the electric field of 5 kV mm⁻¹ for the x = 0.03 ceramic. These values are also higher than the value obtained for the x = 0 sample by \sim 145% and 203%, respectively. In addition, the W and η values as a function of temperature of the (1 - x)[BNKT-0.03BSrT]xBFT ceramics are presented in Fig. 9. It can also be seen that the W and η increase with increasing temperature for all compositions and reach the maximum value of ~ 0.65 J cm⁻³ and 90.4% at 120 °C for the x = 0.03 ceramic. These obtained W values are close to the value obtained in Bi0.48La0.02NKTZ ceramics ($W \sim 0.63$ J cm $^{-3}$, at RT) as reported by Butnoi et al.58

A comparison of normalized energy storage density (*W/E*) with the previous reports is presented in Table 7.^{38,70,72-81} The comparison of *W/E* value with previous results clearly shows the superiority of this composition (x = 0.03) in terms of high *W/E* = 0.13 µC mm⁻² and $\eta = 90.4\%$. The obtained *W/E* value is high for Bi-based piezoelectric ceramics. Based on the obtained results, it is suggested that these synthesized ceramics are promising materials for actuator, high electric power and energy storage applications.^{72,77}

Conclusions

In this work, the BNKT–0.03BSrT doped with BFT ceramics have been successfully synthesized by a solid-state mixed oxide method. The coexistence of rhombohedral and tetragonal phases is present throughout the entire compositional range, while a dominant tetragonal-rich phase is observed at a higher BFT content. The x = 0.01 ceramic shows the high S_{max} of 0.42%, d_{33}^* of 840 pm V⁻¹, and high Q_{33} of 0.0404 m⁴ C⁻² (at RT) with a very high Q_{33}/E (8.08 × 10⁻⁹ m⁵ C² V⁻¹). The maximum

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research work was supported by the Thailand Research Fund (TRF, BRG6080002) and partially supported by Chiang Mai University, Materials Science Research Center. Department of Physics and Materials Science, Faculty of Science, Science and Technology Research Institute, Chiang Mai University is also acknowledged.

References

- 1 Z. Jiang, R. Zhang, F. Li, L. Jin, N. Zhang, D. Wang and C. L. Jia, *AIP Adv.*, 2016, **6**, 065122.
- 2 J. Hao, W. Li, J. Zhai and H. Chen, *Mater. Sci. Eng., R*, 2019, 135, 1–57.
- 3 X. Liu, F. Li, J. Zhai, B. Shen, P. Li, Y. Zhang and B. Liu, *Mater. Res. Bull.*, 2018, **97**, 215–221.
- 4 W. Bai, L. Wang, P. Zheng, F. Wen, Y. Yuan, M. Ding, D. Chen, J. Zhai and Z. Ji, *Ceram. Int.*, 2018, 44, 8628–8634.
- 5 W. Bai, D. Chen, P. Zheng, J. Zhang, B. Shen, J. Zhai and Z. Ji, *Ceram. Int.*, 2017, **43**, 3339–3345.
- 6 R. F. Ge, Z. H. Zhao, S. F. Duan, X. Y. Kang, Y. K. Lv, D. S. Yin and Y. Dai, *J. Alloys Compd.*, 2017, 724, 1000–1006.
- 7 N. Zhao, H. Fan, X. Ren, S. Gao, J. Ma and Y. Shi, *Ceram. Int.*, 2018, 44, 571–579.
- 8 F. Li, Y. Liu, Y. Lyu, Y. Qi, Z. Yu and C. Lu, *Ceram. Int.*, 2017, 43, 106–110.
- 9 P. Jaita, A. Watcharapasorn, N. Kumar, S. Jiansirisomboon and D. P. Cann, *J. Am. Ceram. Soc.*, 2016, **99**, 1615–1624.
- 10 T. Zheng, J. Wu, D. Xiao and J. Zhu, *Prog. Mater. Sci.*, 2018, 98, 552–624.
- 11 T. Takenaka, K. I. Maruyama and K. Sakata, *Jpn. J. Appl. Phys.*, 1991, **30**, 2236–2239.
- 12 W. Jo, J. E. Daniels, J. L. Jones, X. Tan, P. A. Thomas, D. Damjanovic and J. Rödel, *J. Appl. Phys.*, 2011, **109**, 014110.
- 13 S. Kim, H. Choi, S. Han, J. S. Park, M. H. Lee, T. K. Song, M. H. Kim, D. Do and W. J. Kim, *J. Eur. Ceram. Soc.*, 2017, 37, 1379–1386.
- 14 X. Liu, S. Xue, J. Ma, J. Zhai, B. Shen, F. Wang, X. Zhao and H. Yan, *J. Eur. Ceram. Soc.*, 2018, **38**, 4631–4639.
- 15 A. Sasaki, T. Chiba, Y. Mamiya and E. O. Tsuki, *Jpn. J. Appl. Phys.*, 1999, **38**, 5564–5567.
- 16 Y. Hiruma, K. Yoshii, H. Nagata and T. Takenaka, *J. Appl. Phys.*, 2008, **103**, 084121.

- 17 A. Ullah, C. W. Ahn, R. A. Malik, J. S. Lee and I. W. Kim, *J. Electroceram.*, 2014, **33**, 187–194.
- 18 A. Ullah, A. Ullah, I. W. Kim, D. S. Lee, S. J. Jeong and C. W. Ahn, J. Am. Ceram. Soc., 2014, 97, 2471–2478.
- 19 S. T. Zhang, A. B. Kounga, W. Jo, C. Jamin, K. Seifert, T. Granzow, J. Rödel and D. Damjanovic, *Adv. Mater.*, 2009, 21, 4716–4720.
- 20 J. Shi, H. Fan, X. Liu and A. J. Bell, *J. Am. Ceram. Soc.*, 2014, 97, 848–853.
- 21 S. T. Zhang, A. B. Kounga, E. Aulbach, H. Ehrenberg and J. Rödel, *Appl. Phys. Lett.*, 2007, **91**, 112906.
- 22 Y. Hong, J. Li, W. Wu, Y. Wu, H. Bai, K. Shi, Q. Meng, Z. Zhou and D. Jia, *Ceram. Int.*, 2018, 44, 16069–16075.
- 23 P. Jaita, P. Butnoi, R. Sanjoom, C. Randorn, R. Yimnirun and G. Rujijanagul, *Ceram. Int.*, 2017, **43**, S2–S9.
- 24 S. Manotham, S. Eitssayeam, K. Pengpat, T. Tunkasiri, D. Sweatman and G. Rujijanakul, *Ferroelectrics*, 2015, 487, 149–155.
- 25 F. Galasso, L. Katz and R. Ward, *J. Am. Chem. Soc.*, 1959, **81**, 820–823.
- 26 Z. Wang, X. M. Chen, L. Ni, Y. Y. Liu and X. Q. Liu, *Appl. Phys. Lett.*, 2007, **90**, 102905.
- 27 C. Kruea-In and G. Rujijanagul, J. Nanosci. Nanotechnol., 2017, 17, 8603–8607.
- 28 C. Kruea-In, S. Eitssayeam, K. Pengpat and G. Rujijanagul, Mater. Res. Bull., 2012, 47, 2859–2862.
- 29 R. D. Shannon, Acta Crystallogr., Sect. A: Cryst. Phys., Diffr., Theor. Gen. Crystallogr., 1976, 32, 751–767.
- 30 Z. W. Chen and J. Q. Hu, Adv. Appl. Ceram., 2008, 107, 222– 226.
- 31 R. A. Malik, A. Hussain, A. Zaman, A. Maqbool,
 J. U. Rahman, T. K. Song, W. J. Kim and M. H. Kim, *RSC Adv.*, 2015, 5, 96953–96964.
- 32 P. Fan, Y. Zhang, B. Xie, Y. Zhu, W. Ma, C. Wang, B. Yang, J. Xu, J. Xiao and H. Zhang, *Ceram. Int.*, 2018, 44, 3211–3217.
- 33 W. Bai, F. Liu, P. Li, B. Shen, J. Zhai and H. Chen, *J. Eur. Ceram. Soc.*, 2015, **35**, 3457–3466.
- 34 Y. K. Lv, S. F. Duan, Z. H. Zhao, X. Y. Kang, R. F. Ge, H. Wang and Y. Dai, *J. Mater. Sci.*, 2018, 53, 8059–8066.
- 35 R. A. Malik, A. Hussain, A. Maqbool, A. Zaman, C. W. Ahn, J. U. Rahman, T. K. Song, W. J. Kim and M. H. Kim, *J. Am. Ceram. Soc.*, 2015, **98**, 3842–3848.
- 36 A. Maqbool, A. Hussain, R. A. Malik, J. U. Rahman, A. Zaman, T. K. Song, W. J. Kim and M. H. Kim, *Mater. Sci. Eng.*, *B*, 2015, **199**, 105–112.
- 37 A. A. Bokov and Z. G. Ye, Solid State Commun., 2000, 116, 105–108.
- 38 H. Xie, Y. Zhao, J. Xu, L. Yang, C. Zhou, H. Zhang, X. Zhang,
 W. Qiu and H. Wang, *J. Alloys Compd.*, 2018, 743, 73–82.
- 39 X. Wei, Y. Feng, X. Wan and X. Yao, *Ceram. Int.*, 2004, **30**, 1397–1400.
- 40 G. Dong, H. Fan, J. Shi and Q. Li, J. Am. Ceram. Soc., 2018, 101, 3947–3955.
- 41 A. Hussain, C. W. Ahn, J. S. Lee, A. Ullah and I. W. Kim, *Sens. Actuators, A*, 2010, **158**, 84–89.
- 42 X. Liu and X. Tan, J. Appl. Phys., 2016, 120, 034102.

- 43 A. Ullah, R. A. Malik, A. Ullah, D. S. Lee, S. J. Jeong, J. S. Lee, I. W. Kim and C. W. Ahn, *J. Eur. Ceram. Soc.*, 2014, 34, 29–35.
- 44 A. Ullah, C. W. Ahn, S. Y. Lee, J. S. Kim and I. W. Kim, *Ceram. Int.*, 2012, **385**, S363–S368.
- 45 X. Liu and X. Tan, Adv. Mater., 2016, 28, 574-578.
- 46 K. T. Lee, J. S. Park, J. H. Cho, Y. H. Jeong, J. H. Paik and J. S. Yun, *Ceram. Int.*, 2015, **41**, 10298–10303.
- 47 P. Jaita, A. Watcharapasorn, D. P. Cann and S. Jiansirisomboon, *J. Alloys Compd.*, 2014, **596**, 98–106.
- 48 Q. Yao, F. Wang, F. Xu, C. M. Leung, T. Wang, Y. Tang, X. Ye, Y. Xie, D. Sun and W. Shi, *ACS Appl. Mater. Interfaces*, 2015, 7, 5066–5075.
- 49 V. D. N. Tran, T. H. Dinh, H. S. Han, W. Jo and J. S. Lee, *Ceram. Int.*, 2013, **39**, S119–S124.
- 50 V. D. N. Tran, H. S. Han, C. H. Yoon, J. S. Lee, W. Jo and J. Rödel, *Mater. Lett.*, 2011, **65**, 2607–2609.
- 51 J. Kuwata, K. Uchino and S. Nomura, *Jpn. J. Appl. Phys.*, 1980, **19**, 2099–2103.
- 52 M. B. Rauls, W. Dong, J. E. Huber and C. S. Lynch, *Acta Mater.*, 2011, **59**, 2713–2722.
- 53 A. Ullah, H. B. Gul, A. Ullah, M. Sheeraz, J. S. Bae, W. Jo, C. W. Ahn, I. W. Kim and T. H. Kim, *APL Mater.*, 2018, 6, 016104.
- 54 Y. Guo, M. Gu, H. Luo, Y. Liu and R. L. Withers, *Phys. Rev. B*, 2011, **83**, 054118.
- 55 H. S. Han, W. Jo, J. K. Kang, C. W. Ahn, I. W. Kim, K. K. Ahn and J. S. Lee, *J. Appl. Phys.*, 2013, **113**, 154102.
- 56 J. Hao, Z. Xu, R. Chu, W. Li and J. Du, *J. Mater. Sci.*, 2015, **50**, 5328–5336.
- 57 S. T. Zhang, F. Yan, B. Yang and W. Cao, *Appl. Phys. Lett.*, 2010, **97**, 122901.
- 58 P. Butnoi, S. Manotham, P. Jaita, C. Randorn and G. Rujijanagul, *J. Eur. Ceram. Soc.*, 2018, **38**, 3822–3832.
- 59 H. Zhang, P. Xu, E. Patterson, J. Zang, S. Jiang and J. Rödel, *J. Eur. Ceram. Soc.*, 2015, **35**, 2501–2512.
- 60 J. Hao, Z. Xu, R. Chu, W. Li and J. Du, J. Alloys Compd., 2015, 647, 857–865.
- 61 T. H. Dinh, J. K. Kang, J. S. Lee, N. H. Khansur, J. Daniels, H. Y. Lee, F. Z. Yao, K. Wang, J. F. Li, H. S. Han and W. Jo, *J. Eur. Ceram. Soc.*, 2016, **36**, 3401–3407.
- 62 W. Krauss, D. Schütz, F. A. Mautner, A. Feteira and K. Reichmann, *J. Eur. Ceram. Soc.*, 2010, **30**, 1827–1832.

- 63 H. B. Lee, D. J. Heo, R. A. Malik, C. H. Yoon, H. S. Han and J. S. Lee, *Ceram. Int.*, 2013, **39**, S705–S708.
- 64 H. Shi, Y. Zhu, Y. Zhao, C. Liu, X. Ren, J. Hao and W. Li, *J. Alloys Compd.*, 2018, **767**, 666–674.
- 65 J. Hao, Z. Xu, R. Chu, S. Chu, W. Li, P. Fu, J. Du and C. Hu, *Mater. Lett.*, 2017, **193**, 138–141.
- 66 W. Bai, D. Chen, Y. Huang, P. Zheng, J. Zhong, M. Ding, Y. Yuan, B. Shen, J. Zhai and Z. Ji, *Ceram. Int.*, 2016, 42, 7669–7680.
- 67 N. Chen, W. Yao, C. Liang, S. Xiao, J. Hao, Z. Xu and R. Chu, *Ceram. Int.*, 2016, 42, 9660–9666.
- R. A. Malik, A. Hussain, A. Maqbool, A. Zaman, T. K. Song,
 W. J. Kim and M. H. Kim, *J. Alloys Compd.*, 2016, 682, 302–310.
- 69 K. Wang, A. Hussain, W. Jo and J. Rödel, *J. Am. Ceram. Soc.*, 2012, **95**, 2241–2247.
- 70 Y. Zhao, J. Xu, L. Yang, C. Zhou, X. Lu, C. Yuan, Q. Li,
 G. Chen and H. Wang, *J. Alloys Compd.*, 2016, 666, 209–216.
- 71 B. Hu, H. Fan, L. Ning, Y. Wen and C. Wang, Ceram. Int., 2018, 44, 15160–15166.
- 72 R. A. Malik, A. Hussain, M. Acosta, J. Daniels, H. S. Han,
 M. H. Kim and J. S. Lee, *J. Eur. Ceram. Soc.*, 2018, 38, 2511–2519.
- 73 B. Wang, L. Luo, X. Jiang, W. Li and H. Chen, *J. Alloys Compd.*, 2014, **585**, 14–18.
- 74 Z. Liu, P. Ren, C. Long, X. Wang, Y. Wan and G. Zhao, J. Alloys Compd., 2017, 721, 538–544.
- 75 H. Yang, F. Yan, Y. Lin, T. Wang, F. Wang, Y. Wang, L. Guo, W. Tai and H. Wei, *J. Eur. Ceram. Soc.*, 2017, 37, 3303–3311.
- 76 Q. Xu, H. Liu, Z. Song, X. Huang, A. Ullah, L. Zhang, J. Xie,
 H. Hao, M. Cao and Z. Yao, *J. Mater. Sci.: Mater. Electron.*,
 2016, 27, 322–329.
- 77 L. Luo, B. Wang, X. Jiang and W. Li, J. Mater. Sci., 2014, 49, 1659–1665.
- 78 H. Yang, F. Yan, Y. Lin, T. Wang, L. He and F. Wang, J. Alloys Compd., 2017, 710, 436–445.
- 79 Q. Xu, J. Xie, Z. He, L. Zhang, M. Cao, X. Huang, M. T. Lanagan, H. Hao, Z. Yao and H. Liu, *J. Eur. Ceram. Soc.*, 2017, **37**, 99–106.
- 80 Q. Li, Z. Yao, L. Ning, S. Gao, B. Hu, G. Dong and H. Fan, *Ceram. Int.*, 2018, 44, 2782–2788.
- 81 Q. Xu, M. T. Lanagan, X. Huang, J. Xie, L. Zhang, H. Hao and H. Liu, *Ceram. Int.*, 2016, **42**, 9728–9736.