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1. Introduction

The design of organic–inorganic hybrid materials, based on trivalent metal halides, has attracted scientific attention not only due to their interesting structural diversity but also to their unique physical properties as well as possible novel applications in optoelectronics, photovoltaic cells, semiconductor devices, electrochemical transistors, and electroluminescent diodes.^{1–4}

Crystals of halogeno-bismuthates(m) and antimonates(m) with tetralkylammonium cations with the general formula $[(C_nH_{2n+1})_4N]_3M_2X_9$ (where $[(C_nH_{2n+1})_4N] =$ organic cations with n = 1, 2, 3 or 4; M = Bi, Sb and X = Cl, Br, I) are frequently characterized by their phase transitions.⁵⁻⁸ It was found that the complexity of the phase transitions increases with increasing the "*n*" number of carbon atoms in the alkylammonium chain up to four carbon atoms.^{9,10}

The phase transitions are usually connected with an appearance of interesting physical properties such as ferroelectricity and ferroelasticity, *etc.*¹¹⁻¹³ Thus, the paraferroelectric phase transitions are characterized by an "order– disorder" mechanism related to the dynamical reorientation of the alkyl cation's chains.¹⁴⁻¹⁷ This motion is substantially diminished on lowering temperature yielding ordered phases.¹⁸

Ferroelectric properties and Raman spectroscopy of the $[(C_4H_9)_4N]_3Bi_2Cl_9$ compound⁺

W. Trigui 🕩 * and F. Hlel

The exploration of ferroelectric hybrid materials is highly appealing due to their great technological significance. They have the potential to conserve power and amazing applications in information technology. In line with this, we herein report the development of a $[(C_4H_9)_4N]_3Bi_2Cl_9$ tetra-alkyl hybrid compound that exhibits ferroelectric properties. The phase purity was confirmed by Rietveld refinement of the X-ray powder diffraction pattern. It crystallizes, at room temperature, in the monoclinic system with the $P2_1/n$ space group. The outcome of temperature dependence of the dielectric constant proved that this compound is ferroelectric below approximately 238 K. The dielectric constants have been fitted using the modified Curie–Weiss law and the estimated γ values are close to 1. This confirms classical ferroelectric behavior. Raman spectroscopy is efficiently utilized to manifest the origin of the anions. Moreover, the analysis of the wavenumbers and the half-width for $\delta_s(Cl-Bi-Cl)$ and $\omega(CH_2)$ modes, based on the order–disorder model, allowed us to obtain the thermal coefficient and activation energy near the para-ferroelectric phase transition.

The anionic $[M_2X_9]^{3-}$ sublattices are built of basic distorted MX₆ octahedral units sharing corners, edges or faces to form anionic discrete bioctahedra (0D),¹⁹ chains (1D)²⁰ or layers (2D).²¹ It is worthwhile to mention that a special interest has been focused on compounds with the inorganic halide M₂X₉ since the discovery of their ferroelectric properties.^{22,23}

In this work we intend to spot on the nature of paraferroelectric phase transition at 238 K in $[(C_4H_9)_4N]_3Bi_2Cl_9$ compound. The crystal structure of this hybrid compound, at room temperature, was previously determined and reported.²⁴

In this contribution, we report the results of the X-ray powder diffraction, dielectric and Raman studies. The temperature dependence of Raman spectra have enabled us to establish the key role played by the tetra-butylammonium cations and the halogeno-bismuthate anions, in the para-ferroelectric transition.

2. Experiments and methods

Phase purity, space group and lattice parameters were checked by X-ray powder diffraction analysis. XRD data was collected, at room temperature, using a Philips PANalytical X'pert Pro X-ray powder diffractometer operating with copper radiation ($\lambda =$ 1.540598 Å) over a wide range of Bragg angles ($6^{\circ} \le 2\theta \le 52^{\circ}$). Refinements were carried out using FULLPROF program based on Rietveld method.²⁵

The impedance analyzer (Agilent 4294A) was used to characterize the dielectric properties in the frequency range, between 50 Hz and 6 MHz, in the temperature range within

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Faculty of Sciences, Laboratory of Spectroscopic and Optical Characterization of Materials, University of Sfax, BP1171, 3018 Sfax, Tunisia

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195–260 K. The polycrystalline material was used in a form of a pressed pellet with a diameter equal to 8 mm and a thickness of 1.3 mm.

Raman spectra of $[(C_4H_9)_4N]_3Bi_2Cl_9$ monocrystalline sample were collected using a T-64000 Raman (Horiba-Jobin-Yvon) with cooled CCD detector. The excitation wave-length radiation was 647.1 nm, using an argon-krypton laser (coherent spectrum) with a 50 mW power. The spectra were registered with 1800 per mm grating from 170 to 3300 cm⁻¹. The resolution of the instrument was set at 0.2 cm⁻¹. The measurements were carried out under microscope in the backscattering geometry using an X-50 LF objective, on the (001) natural face of single crystal in Z(XX)Z polarization, where X stands for a birefringence direction. The low temperature measurements were performed with a cold liquid nitrogen cryostat. Temperature stability around ±1 K was achieved. The error range for the determination of the peak position was ±0.3 cm⁻¹.

3. Results and discussion

3.1. X-ray powder diffraction analysis

The X-ray diffraction pattern at 293 K for the tritetrabutylammonium nona chlorobibismuthate (TBA₃Bi₂Cl₉) compound is plotted in Fig. 1. The circles symbolize the experimental data and the line corresponds to the simulated pattern. The green bars indicate good matching of all the peak positions. The purity of the studied compound was proved and all reflection peaks are successfully refined in the monoclinic system with $P2_1/n$ space group, using Rietveld method. The corresponding lattice parameters were found to be: a = 11.356 Å, b = 22.342 Å, c = 28.587 Å, $\beta = 96.59^{\circ}$ and the cell volume was V = 7155.32 Å³. Those parameters are in good agreement with the published results.²⁴ The quality of the refinement was evaluated through the goodness of fit $\chi^2 = 1.20$ and the reliability factors obtained are $R_{\omega p} = 11.90$, $R_p = 11.80$ and $R_{exp} = 10.85$.

3.2. Dielectric studies

The temperature dependence of the real part ε' , imaginary part ε'' of the dielectric permittivity and the dielectric loss $T_g(\delta)$, of different frequencies, are plotted in Fig. 2. From these curves, we can observe two dielectric relaxations around 214 and 235 K in the ε' variation but only one relaxation in the ε'' and $T_g(\delta)$ plots ($T_c = 235$ K). This prominent dielectric peak at T_c may be associated to a para-ferroelectric phase transition in the title compound. Its maximum does not change while the magnitude decreases with the increase in frequency. In fact, this result suggests that this sample does not present any type of dielectric relaxation in the investigated frequency range.^{26,27}

The dielectric data in the paraelectric state obey quite well the Curie–Weiss law, described by the following relation:

$$\varepsilon' = \frac{C}{T - T_0} \tag{1}$$



Fig. 1 Plot of the Rietveld refinement at room temperature of $[(C_4H_9)_4N]_3Bi_2Cl_9$ compound. The red dots indicate the experimental data and the black continuous line represent the calculated pattern. The green vertical bars describe the Bragg position.





Fig. 2 Temperature dependence at various frequencies of the real part ε' of the complex permittivity (a), the imaginary part ε'' of the complex permittivity (b) and the dielectric loss $T_{q}(\delta)$ (c).

where C and T_0 are the Curie–Weiss constant and temperature, respectively.

The temperature dependence of the dielectric reciprocal (1/ ε'), at 500 Hz, is illustrated in Fig. 3a. This plot gives a straight line with the *X*-axis intercept at $T_0 = 227$ K. For the classical ferroelectrics, the order of the para-ferroelectric phase transition can be determined from the value of T_0 . In case of $T_0 \neq T_C$, the phase transition is of a first order type. If $T_0 = T_C$, this transition is of a second order type.²⁸ According to our measurements, the value of T_0 is different from $T_C = 235$ K which suggests that this transition is of the first order type^{29,30} and confirms the drawing results from the differential scanning calorimetry analysis.²⁴

In order to determine the nature of classic or relaxor ferroelectrics, Uchino and Nomura³¹ have proposed a more general expression of the Curie–Weiss law by introducing the exhibitor criticizes γ :

$$\frac{1}{\varepsilon'} - \frac{1}{\varepsilon'_{\max}} = \frac{(T - T_{\rm C})^{\gamma}}{C}$$
(2)

where *C* is the Curie–Weiss constant, $\varepsilon'_{\text{max}}$ is the real dielectric constant at $T = T_{\text{C}}$ and $\gamma (1 < \gamma < 2)$ corresponds to the degree of relaxation.

The factor γ translates the diffuse character of the transition. However, for ideal relaxor ferroelectrics, the γ value should amount to 2, and for classical ferroelectrics it is near to 1.^{32,33}

Fig. 3b shows the logarithmic plots related to the eqn (2) at various frequencies. The obtained γ values are 1.152, 1.286 and 1.008 corresponding to the frequencies 100 Hz, 2.5 kHz and 10 kHz, respectively. This result proves that the $[(C_4H_9)_4N]_3Bi_2Cl_9$ is a classical ferroelectric compound.



Fig. 3 (a) Temperature variation of $1/\epsilon'$ at some frequency, (b) $\ln(1/\epsilon' - 1/\epsilon'_{max})$ as a function of $\ln(T - T_c)$ for different frequency (the solid red lines denote the fitting curve to the modified Curie–Weiss law).

3.3. Temperature evolution of the Raman spectra

The assignment of the Raman spectrum at room temperature has been performed and presented in table S. Raman spectra in the frequency range $170-3300 \text{ cm}^{-1}$, have been collected for several temperatures (from 193 to 258 K). Fig. 4 shows the temperature evolution of Raman spectra obtained in three spectral ranges.

The positions of most analyzed bands are constant even in the close vicinity of both phase transitions. However, one can distinguish several bands showing changes in their positions and width at half maximum, by the order of a few cm⁻¹ when crystal undergoes the phase transitions at T_1 and T_2 .

In order to quantitatively analyze the evolution of Raman bands *versus* temperature, each spectrum were analyzed by means Labspec software on the basis of Gaussian and Lorentzian shapes. Fig. S1 and S2† reports position and half-width temperature dependence of selected lines, respectively.

The lines issued from the internal modes of Bi_2Cl_9 anions do not show significant changes where the wavenumbers and lines width exhibit a slight change. However, the Cl–Bi–Cl bending vibration at 184.4 cm⁻¹, presents a remarkable shift toward high frequencies (of 3 cm⁻¹) and a line broadening (of 5 cm⁻¹) on heating above the phase transitions.

We may say that the increase of dynamical motion of the alkyl chains of tetrabutylammonium cations has evoked the distortion of the bi-octahedral anions.^{5,34}



Fig. 4 Temperature evolution of the Raman spectra for selected spectral ranges

The band related to the symmetric stretching vibration $\nu_{\rm s}$ (C–C) is located at 797.4 cm⁻¹. Its maximum shifts toward lower wavenumbers by 3 cm⁻¹ around T_1 and towards higher wavenumbers around T_2 . A similar behavior was observed for the band located at 769.8 cm⁻¹ which is assigned to ν_4 (NC₄) and $\rho_{\rm r}$ (CH₂). The symmetric stretching vibration $\nu_{\rm s}$ of (CH₂) and (CH₃) groups shows a variation in its position to the high frequencies by 3 cm⁻¹ and by 5 cm⁻¹ in its half-width. The band at 1319.8 cm⁻¹ assigned to the wagging mode of (CH₂) group increase by 3 cm⁻¹, while the variation of the symmetric bending vibration $\delta_{\rm s}$ (C–C–C) and $\delta_{\rm s}$ (C–N–C) observed at 906.9 cm⁻¹ is by 2 cm⁻¹ for both wavenumber and half-width.

Over the phase I, the band corresponding to the antisymmetric bending mode $\delta_{as}(CH_3)$ weakly moves to high frequencies. To verify that the phase transitions are connected to a change in the dynamical state of the $[(C_4H_9)_4N]^+$ cations and to the distortion of the $[Bi_2Cl_9]^{3-}$ anions, we have focused our study on tow selected bands P2 and P29 associated to the bending vibration of (Cl–Bi–Cl) and to the wagging mode $\omega(CH_2)$ (Fig. 5), respectively.

3.3.1. Temperature dependence of the wavenumber. The temperature dependence of Raman wavenumber, of a phonon connected to an order–disorder mechanism, according to Andrade and Porto^{35,36} can be described by:

$$\nu^{2} = \nu_{0}^{2} \left[1 + \gamma \left(T - T_{j} \right) \right]$$
(3)

where γ is the thermal coefficient, T_j is the temperature of phase transition j, ν_0 is the hard-core wavenumber at transition temperature.

Generally, the values of γ are small so that we can approximate eqn (3) by:

$$\nu = \nu_0 \left[1 + \frac{\gamma}{2} \left(T - T_j \right) \right] \tag{4}$$

The experimental values of P_2 and P_{29} modes wavenumber, at various temperatures, were fitted to the equation above (Fig. 6a and b). A summary of the experimental thermal coefficient γ values are collected in Table 1. The "hard-core" frequency ν_0 values, for the two bands, are (182.7, 1316.5 cm⁻¹) at T_1 and (185.8, 1319.1 cm⁻¹) at T_2 , respectively.

The thermal coefficient is related to the variation of the wavenumber position and the volume of the crystal. It is expressed by the following equation:

$$\gamma_{i} = -\frac{\Delta \nu_{i}}{\nu_{i}} \times \frac{V}{\Delta V}$$
(5)

where ΔV is the volume, Δv_i is the variation of the wavenumber position, *V* symbolizes the original volume and v_i the band position of the i mode at room temperature.

According to Gruneisen,³⁷ the relative change of each vibrational frequency is directly proportional to the relative change in the volume. The increase of the thermal coefficient, related to the shift of the two peaks positions to high frequencies, indicates a decrease of the crystal volume.³⁸ Indeed, the cation geometry can be changed from cross symmetry to broken cross symmetry by reorientational motion of the alkylammonium chains going through the phase transitions.^{16,39} This behavior suggests that the observed phase transitions are related to the reorientational motion of the cations coupled to the distortion of the anions.

3.3.2. Temperature dependence of the full width at half maximum. We performed a detailed analysis of the bands shape for Cl–Bi–Cl bending and wagging mode of the (CH₂) group. It was shown that the temperature dependence of the full-width at the half-maximum (FWHM) fulfils the Carabatos-Nédelec and Becker model.⁴⁰ This model is based on the damping theory of the oscillation associated with an order–disorder mechanism in the solid state.⁴¹

According to Langevin^{42,43} the temperature dependence of FWHM can be expressed as:

$$\tau = a + bT + c \frac{\tau_{\rm R}}{1 + \omega^2 \tau_{\rm R}^2} \tag{6}$$



Fig. 5 Temperature evolution of the Raman spectra in the frequency region assigned to the bending vibration of (Cl-Bi-Cl) and to the wagging mode of (CH₂).



Fig. 6 (a and b) Raman wavenumbers versus temperature of the bands P_2 and P_{29} , (c and d) line width versus temperature of the bands P_2 and P_{29} (the red line is the fit to the Carabatos-Nédelec and Becker model).

Table 1 Thermal coefficients of the wavenumber variation vs. temperature of the peaks P_2 and P_{29}

| Peak | Temperature range | Thermal coefficient, γ (K ⁻¹) |
|-----------------|-------------------|--|
| P ₂ | $T < T_1$ | $\gamma = -1.83 \times 10^{-4}$ |
| | $T_1 < T < T_2$ | $\gamma = 1.23 	imes 10^{-3}$ |
| | $T > T_2$ | $\gamma = 6.83 	imes 10^{-4}$ |
| P ₂₉ | $T \leq T_1$ | $\gamma=-2.48\times10^{-5}$ |
| | $T_1 < T < T_2$ | $\gamma = 2.49 \times 10^{-4}$ |
| | $T > T_2$ | $\gamma = 5.97 \times 10^{-5}$ |
| | | |

where $(\omega \tau_R)^2 \gg 1$, $\omega = 2\pi \nu$ is the phonon wavenumber and τ_R is the mean reorientational time of the atoms to jump from one potential well to another.

The mean reorientational time of the atoms τ_{R} is given by:

$$\tau_{\rm R} = \tau_{\infty} \, \exp\!\left(\frac{E_{\rm a}}{RT}\right) \tag{7}$$

where τ_{∞} is the relaxation time at infinite temperature, E_a is the activation energy for the mode connected to the order–disorder transition and R (=8.314472 J K⁻¹ mol⁻¹) is the perfect gas constant. Therefore the eqn (6) can be written as:⁴⁴

$$FWHM(T) = (a+bT) + c \exp\left(-\frac{E_a}{RT}\right)$$
(8)

where a, b, c and E_a are the fitting parameters. E_a is the activation energy of the reorientation process. The linear part of eqn (8) corresponds to the vibrational relaxation and the exponential term, related to the thermal molecular motions, corresponds to the reorientational relaxation.

The values of the full widths at half maximum are obtained using ultra-high resolution triple additive configuration. Fig. 6c and d show the FWHM experimental values of P_2 and P_{29} , at various temperatures, fitted according to eqn (8).

For the bending vibration of (Cl–Bi–Cl), the estimated activation energy values are $E_a(I) = 18.23 \text{ kJ mol}^{-1}$ for $T < T_1$, $E_a(II) =$

15.59 kJ mol⁻¹ for $T_1 < T < T_2$ and $E_a(III) = 11.46$ kJ mol⁻¹ for $T > T_2$. Whereas, the obtained values for the wagging mode $\omega(CH_2)$ are $E_a(I) = 40.60$ kJ mol⁻¹ for $T < T_1$, $E_a(II) = 35.92$ kJ mol⁻¹ for $T_1 < T < T_2$ and $E_a(III) = 30.96$ kJ mol⁻¹ for $T > T_2$. In fact, the decrease in activation energy values for these Raman bands is probably due to the decrease in the population involved in these vibrations, which may be related to the change of the $[(C_4H_9)_4N]^+$ cations conformation as well as the distortion of the $[Bi_2Cl_9]^{3-}$ anions.

4. Conclusion

The $[(C_4H_9)_4N]_3Bi_2Cl_9$ compound, with $P2_1/n$ space group of the monoclinic system was investigated as a function of temperature. The existence of a para-ferroelectric phase transition at 238 K has been established by the dielectric measurements. The empirical parameter γ indicates that this compound is a classical ferroelectric. The Raman spectroscopic study clearly shows spectacular temperature changes of the internal vibrations of cation and the anion, especially the deformation of (Cl–Bi–Cl) and the (CH₂) wagging mode. A detailed analysis of the wavenumber and the bandwidth was performed in the framework of order–disorder models. This study proves that the phase transitions are related to the rotational motion of the tetrabutylammonium cations and to the distortion of the $[Bi_2Cl_9]^{3-}$ anions.

Conflicts of interest

There are no conflicts to declare.

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