



Cite this: *RSC Adv.*, 2019, 9, 24963

A comprehensive study of eriocitrin metabolism *in vivo* and *in vitro* based on an efficient UHPLC-Q-TOF-MS/MS strategy

Luya Li,^a Xue Feng,^a Yuting Chen,^a Shenghao Li,^b Yupeng Sun^a and Lantong Zhang^{*a}

Eriocitrin, a main flavonoid in lemons, possesses strong antioxidant, lipid-lowering and anticancer activities and has long been used in food, beverages and wine. However, its metabolism *in vivo* and *in vitro* is still unclear. In this study, an efficient strategy was developed to detect and identify metabolites of eriocitrin by using ultra-high-performance liquid chromatography coupled with hybrid triple quadrupole time-of-flight mass spectrometry (UHPLC-Q-TOF-MS) based on online data acquisition and multiple data processing techniques. A total of 32 metabolites *in vivo* and 27 metabolites *in vitro* were obtained based on the above method. Furthermore, the main metabolic pathways of eriocitrin included reduction, hydrogenation, *N*-acetylation, ketone formation, oxidation, methylation, sulfate conjugation, glutamine conjugation, glycine conjugation, desaturation and demethylation to carboxylic acid. This study will lay a foundation for further studies on the metabolic mechanisms of eriocitrin.

Received 24th April 2019
 Accepted 1st August 2019

DOI: 10.1039/c9ra03037a

rsc.li/rsc-advances

1. Introduction

Eriocitrin (eriodictyol 7-*O*-beta-rutinoside), belonging to the dihydroflavonoid compound class, is widely found in citrus fruits (lemon, citrus, grapefruit), vegetables, processed products (drinks, wine) and so on.^{1–3} Modern pharmacological studies show that eriocitrin has strong antioxidant, lipid-lowering and anticancer activities.⁴ It plays an important role in effectively preventing and improving oxidative stress, hyperlipidaemia, cardiovascular and cerebrovascular diseases as well as cancer.^{5–9}

As found in the literature, three metabolites of eriocitrin were mentioned in plasma and renal-excreted urine, detected through HPLC and LC-MS analyses.¹⁰ However, until now, no structural information about the metabolites in bile, faeces, intestinal flora and liver microsomes of rats has been reported.

It is commonly known that drugs can have four pharmacological effects through biotransformation: (1) conversion into inactive substances; (2) transformation of the previously inactive drugs into active metabolites; (3) an alteration of the types of drug pharmacological action; (4) the production of toxic substances.^{11–13} Thus, it is crucial to study the metabolism of drugs *in vivo* to ensure safety of use. In addition, as the main metabolic organ of the human body, the liver is rich in enzymes,

especially cytochrome P450 enzymes.¹⁴ In addition, the gastrointestinal tract is also an important place for drug metabolism, and its intestinal flora has a significant impact on drug absorption, metabolism and toxicology.^{15,16} Therefore, in this paper, mass spectrometry was used to investigate the metabolism of eriocitrin in rats, liver microsomes and intestinal flora in order to identify the metabolites and structural information of the products, which will lay a foundation for further studies on the toxicity and activity of metabolites and will provide greater possibilities for the development of new drugs.

With the development of technology, quadrupole time-of-flight mass spectrometry has been widely used as a reliable analytical technique to detect metabolites due to its advantages of high resolution, high sensitivity, high-efficiency separation and accurate quality measurement.^{17,18} In this study, ultra-high-performance liquid chromatography coupled with hybrid triple quadrupole time-of-flight mass spectrometry (UHPLC-Q-TOF-MS/MS) technology was used. The electrospray ionization (ESI) source was operated in negative ion mode, and full scan combined with multiple mass loss (MMDF) and dynamic background subtraction (DBS) was used to collect data online. MetabolitePilot 2.0.4 and PeakView 1.2 data loading software were adopted to obtain the precise mass number of metabolites, secondary mass spectrometry and decomposition rules of eriocitrin to infer the possible metabolites. Based on the above methods, 32 metabolites *in vivo* and 27 metabolites *in vitro* were finally observed. In addition, the metabolic pathways of eriocitrin were explored and summarized for the first time, which is an important part of drug discovery and development and can also provide a basis for further pharmacological research.

^aDepartment of Pharmaceutical Analysis, School of Pharmacy, Hebei Medical University, Shijiazhuang 050017, P. R. China. E-mail: zhanglantong@263.net; Fax: +86-311-86266419; Tel: +86-311-86266419

^bDepartment of Pathobiology and Immunology, Hebei University of Chinese Medicine, Shijiazhuang 050000, P. R. China



2. Materials and methods

2.1. Chemicals and materials

Eriocitrin (13463-28-0, purity > 98.5%) was purchased from Chengdu Desite Co., Ltd. (Chengdu, China). β -Nicotinamide adenine dinucleotide phosphate (NADPH) was purchased from Sigma Chemical (St. Louis, MO, USA). Alamethicin and uridine 5'-diphosphoglucuronic acid trisodium salt (UDPGA) were purchased from BD Biosciences (Woburn, MA, USA). Phosphate buffer saline (PBS) was purchased from Sangon Biotech (Shanghai) Co., Ltd. Acetonitrile, methanol and formic acid were all HPLC grade and were purchased from Fisher Scientific (Waltham, MA, USA). Dilute hydrochloric acid (HCl) was purchased from Shijiazhuang Reagent Factory. Purified water was purchased from Wahaha (Hangzhou Wahaha Group Co., Ltd.). L-Ascorbic acid, L-cysteine, erythrol, tryptone and nutrient agar were purchased from Beijing AoBoXing Bio-tech (Beijing) Co., Ltd. Sodium carboxymethyl cellulose (CMC-Na), sodium carbonate (Na_2CO_3), magnesium chloride (MgCl_2), potassium dihydrogen phosphate (KH_2PO_4), dipotassium phosphate (K_2HPO_4), calcium chloride (CaCl_2), ammonium sulfate ($(\text{NH}_4)_2\text{SO}_4$), sodium chloride (NaCl) and magnesium sulfate (MgSO_4) were obtained from Tianjin Guangfu Technology Development Co., Ltd. (Tianjin, China).

2.2. Instruments and conditions

UHPLC-Q-TOF-MS/MS analysis was performed on a Nexera-X2 UHPLC system (Shimadzu Corp., Kyoto, Japan), which was combined with a triple TOF 5600⁺ MS/MS system (AB SCIEX, Concord, Ontario, Canada). The chromatographic separation was achieved on a Poroshell 120 EC-C₁₈ column (2.1 × 100 mm, 2.7 μm) equipped with a Poroshell 120 EC-C₁₈ UHPLC guard column (Agilent Corp, Santa Clara, CA, USA).

The mobile phase was composed of 0.1% aqueous formic acid (eluent A) and acetonitrile (eluent B). The gradient elution programme was as follows: 5–17% B from 0 to 10 min, 17–35% B from 10 to 15 min, 35–95% B from 15 to 20 min, and 95% B held from 20 to 25 min. The injection volume and the flow rate were set at 3 μL and 0.3 mL min^{-1} , respectively. Equilibration was performed for 3 min before the next injection. The column temperature remained at 25 °C, and the autosampler tray temperature was maintained at 4 °C.¹⁹

Mass spectrometric detection was carried out by a Triple TOFTM 5600 system equipped with Duo-SprayTM ion sources in the negative electrospray ionization (ESI) mode. The following mass spectrometry parameter settings were used: ion spray voltage (IS), –4.5 kV; turbo spray temperature, 550 °C; optimized declustering potential (DP), –60 V; collision energy (CE), –10 eV; collision energy spread (CES), 15 eV; nebulizing gas (GAS1), 55 psi; heating gas (GAS2), 55 psi; and curtain gas, 35 psi.

2.3. Metabolism *in vivo*

2.3.1. Animals and drug administration. Eighteen male Sprague-Dawley rats (certificate no. 1804195, weighing 220 ± 20 g, 12–14 weeks old) were purchased from the Experimental

Animal Research Center of Hebei Medical University (SCXK 2018-004). All the protocols and procedures for animal handling were carried out following the guidelines of the Hebei committee for care and use of laboratory animals, and were approved by the Animal Experimentation Ethics Committee of the Hebei Medical University (Hebei, China). The conditions of temperature (22–25 °C), humidity (55–60%) and light (12 h light/dark cycle) were standard for the 8 days prior to use. All rats were fasted but allowed water for 12 h before the experiments. These rats were randomly divided into six groups with three rats per group. Groups 1, 3 and 5 were the control groups for blank blood, bile, and urine and faeces, respectively. Groups 2, 4 and 6 were the drug groups for blood, bile, and urine and faeces, respectively. Rats in groups 2, 4, and 6 were given eriocitrin by gavage, which was dissolved in a 0.5% CMC-Na solution at a dose of 50 mg kg^{-1} . However, the rats in groups 1, 3, and 5 were given the same dose of 0.5% CMC-Na solution with no eriocitrin. All rat experiments were conducted in accordance with the committee's guidelines on the Care and Use of Laboratory Animals.

2.3.2. Bio-sample collection. The plasma sample collection was completed as follows: approximately 300–500 μL for each blood sample was collected from the eye canthus of rats into 1.5 mL heparinized tubes at 0.083, 0.167, 0.25, 0.5, 1, 2, 3, 6, 9, 12 and 24 h after gavage. Every blood sample was centrifuged immediately at 1920 × *g* for 5 min to obtain the plasma. After that, all collected plasma samples were consolidated and stored at –80 °C.

Bile collection. To anaesthetize the rats, each rat was intraperitoneally injected with 1–2 mL of a 20% urethane solution after gavage. Then, the rats were subjected to bile duct cannulation, and the bile samples were collected during 0–1 h, 1–3 h, 3–5 h, 5–8 h, 8–12 h, 12–20 h and 20–24 h after gavage with PE-10 tubes.^{20,21} Finally, all bile samples were consolidated and frozen at –80 °C.

Urine and faeces collection. The rats were placed in separated metabolic cages with free access to purified water, and urine and faeces samples were collected over a 0–72 h period after gavage.^{22,23} Finally, all urine and faecal samples were separately combined and stored at –80 °C before pretreatment was conducted.

2.3.3. Bio-sample pretreatment. All biological samples were treated with two methods: protein precipitation with methanol and liquid–liquid extraction with ethyl acetate. An aliquot of 2 mL of mixed plasma, bile or urine was taken, and three-fold methanol or ethyl acetate was added to precipitate proteins or extract, respectively. Then, the mixture was vortexed for 5 min and centrifuged at 21 380 × *g* for 10 min at 4 °C to obtain the supernatant, which was collected and dried under nitrogen flow at room temperature.

Dried and powdered faecal samples (2.0 g) were added to 3-fold methanol or ethyl acetate and were ultrasonically extracted for 45 min. Next, samples were centrifuged at 21 380 × *g* for 10 min, and the supernatant was dried under nitrogen gas.

150 μL of acetonitrile was added to the residua above and subjected to ultrasonic treatment for 10 min and centrifugation at 21 380 × *g* for 10 min to obtain the supernatant that was



passed through a 0.22 μm millipore filter before injection into the UHPLC-Q-TOF-MS/MS system for analysis. Samples from the control and drug groups were treated the same.

2.4. Metabolism *in vitro* by rat liver microsomes

2.4.1. Phase I metabolism. The representative incubation mixture was assembled in PBS buffer (pH 7.4) with a final volume of 200 μL and contained liver microsomal protein (1.0 mg mL^{-1}), eriocitrin (100 $\mu\text{mol L}^{-1}$), MgCl_2 (3.3 mmol L^{-1}), and $\beta\text{-NADPH}$ (1.3 mmol L^{-1}).²⁴ Preincubation was conducted at 37 $^\circ\text{C}$ for 5 min, and NADPH was subsequently added to start the reaction. After incubation at 37 $^\circ\text{C}$ for an additional 90 min, the reaction was stopped by adding 1 mL of ethyl acetate. Next, samples were vortexed and centrifuged for 5 and 10 min, respectively, and the organic phase was collected and evaporated under nitrogen gas. After reconstitution in 100 μL of acetonitrile, samples were passed through 0.22 μm millipore filters and stored at -20 $^\circ\text{C}$ until analysis. Blank groups underwent incubation without the addition of eriocitrin, the control groups were incubated without the addition of NADPH, and the sample groups, which were carried out in triplicate, underwent the full treatment described above.²⁵

2.4.2. Phase II metabolism. The representative incubation mixture was implemented in PBS buffer (pH 7.4) with a final volume of 200 μL and contained liver microsomal protein (1.0 mg mL^{-1}), eriocitrin (100 $\mu\text{mol L}^{-1}$), MgCl_2 (3.3 mmol L^{-1}), and UDPGA (2 mmol L^{-1}). Preincubation was performed at 37 $^\circ\text{C}$ for 20 min; subsequently, UDPGA was added to start the reaction. After incubation at 37 $^\circ\text{C}$ for an additional 1 h, the reaction was stopped by adding 200 μL of acetonitrile. Next, samples were vortexed and centrifuged for 5 and 10 min, respectively. Finally, the supernatant was passed through a 0.22 μm millipore filter before injection into the UHPLC-Q-TOF-MS/MS system for analysis. Blank groups were incubated without the addition of eriocitrin, the control groups were incubated without the addition of UDPGA, and the sample groups, which were carried out in triplicate, underwent the treatment described above.

2.5. Metabolism *in vitro* by rat intestinal flora

2.5.1. Preparation of anaerobic culture medium. Solution A: K_2HPO_4 (0.78%), 37.5 mL; solution B: KH_2PO_4 (0.47%), NaCl (1.18%), $(\text{NH}_4)_2\text{SO}_4$ (1.2%), CaCl_2 (0.12%) and MgSO_4 (0.25%), 37.5 mL; solution C: Na_2CO_3 (8%), 50 mL; solution D: L-ascorbic acid (25%) 2 mL together with L-cysteine 0.5 g, eurythrol 1 g, tryptone 1 g and nutrient agar 1 g, brought to 1 L with ultrapure water. HCl (1 mol L^{-1}) was used to adjust the pH of the solution to 7.5–8.0.

2.5.2. Preparation of intestinal flora culture solution. Fresh intestinal contents (3 g) taken from SD rats were mixed with anaerobic culture medium (30 mL) immediately. After stirring with a glass rod, the samples were filtered with gauze to obtain the intestinal bacterial liquid.

2.5.3. Sample preparation. Eriocitrin (1 mg mL^{-1} , 100 μL) was added to 1 mL of intestinal flora culture medium, which was saturated with nitrogen to remove oxygen. After incubation

for 6 h, the reactions were terminated by adding 3 volumes of methanol. Then, the mixtures were vortexed for 5 min and centrifuged at $21\,380 \times g$ for 10 min. The organic phase was collected and evaporated under nitrogen gas, reconstituted in 100 μL of acetonitrile, and vortexed and centrifuged again for 5 and 10 min, respectively. The supernatant was passed through a 0.22 μm millipore filter before analysis. Blank groups were incubated without eriocitrin, while the control groups were incubated not in intestinal flora culture solution but in anaerobic culture medium, and sample groups were treated as described above.

3. Results and discussion

3.1. Analytical strategy

In this study, an efficient UHPLC-Q-TOF-MS/MS strategy was adopted to systematically identify the metabolites of eriocitrin *in vivo* and *in vitro*. The strategy was segmented into three steps: first, online full scan data acquisition was performed utilizing MMDF and DBS settings and the MS/MS spectrum of eriocitrin metabolites to collect data. Next, multiple data processing techniques were adopted by using PeakView 1.2 and MetabolitePilot 2.0.4 software, which contained many data-processing tools, such as XIC, MDF, PIF and NIF, and provided accurate MS/MS information to infer and identify the metabolites of eriocitrin. Finally, numerous metabolites were identified on the basis of accurate mass datasets, specific secondary mass spectrometry information and so on. With regard to the isomers of metabolites, clog *P* values calculated by ChemDraw 14.0 were used to further distinguish them. In general, the larger the clog *P* value, the longer the retention time will be in the reversed-phase chromatography system.^{26–30}

According to the above method, a total of 32 metabolites *in vivo* and 27 metabolites *in vitro* (12 metabolites in rat liver microsomes and 20 metabolites in rat intestinal flora) were identified and are shown in Tables 1, 2 and 3, respectively. In addition, the structures of all metabolites are presented in Fig. 2. Their XICs are shown in Fig. 3.

3.2. Mass fragmentation behaviour of eriocitrin

To identify the metabolites of eriocitrin, it is important to understand the pyrolysis of the parent drug (M0). Eriocitrin ($\text{C}_{27}\text{H}_{32}\text{O}_{15}$) eluted at 12.46 min and yielded the $[\text{M} - \text{H}]^-$ ion at 595.1669, and the mass deviation was 0.2 ppm under the chromatographic conditions. The main characteristic fragment ions of M0, seen at m/z 459.1128, 287.0552, 193.0125, 163.0474, 151.0028, 135.0440 and 125.0233, were detected according to the MS/MS spectrum. The typical fragment ions generated by eriocitrin at m/z 459.1128 and 287.0552 were attributed to the loss of $\text{C}_7\text{H}_4\text{O}_3$ and $\text{C}_{12}\text{H}_{20}\text{O}_9$, respectively. The ion at m/z 287.0552 produced a series of representative fragment ions at m/z 193.0125, 163.0474 and 125.0233 by the loss of CH_2O_5 , $\text{C}_6\text{H}_4\text{O}_3$ and $\text{C}_9\text{H}_6\text{O}_3$, respectively. In addition, the product ions at m/z 151.0028 and 135.0440 were formed because of the RDA reaction in ring C of the flavonoid. The MS/MS spectrum and the fragmentation pathways of eriocitrin are shown in Fig. 1.



Table 1 Summary of metabolites of eriocitrin in rat plasma, bile, urine and faeces samples^a

Metabolites ID	Composition	Formula	m/z	Error (ppm)	R.T. (min)	Score (%)	MS/MS fragments	clog P	Plasma	Bile	Urine	Faeces
M1	Loss of C ₁₂ H ₂₀ O ₉ [M - H] ⁻	C ₁₅ H ₁₂ O ₆	287.0560	-0.3	15.30	75.0	269.0458, 241.0489, 151.0032, 135.0447, 125.0242	1.84785	-	-	+ ^b	+ ^{a,b}
M2	Loss of C ₁₂ H ₂₀ O ₁₀ [M - H] ⁻	C ₁₅ H ₁₂ O ₅	271.0615	1.3	16.99	75.0	253.0507, 243.0682, 227.0715, 135.0468	2.39694	-	-	+ ^{a,b}	+ ^{a,b}
M3	Loss of O and C ₁₂ H ₂₀ O ₁₀ [M - H] ⁻	C ₁₅ H ₁₂ O ₄	255.0665	1.0	15.41	79.9	237.2257, 135.0064, 119.0480, 109.0795	2.2109	-	-	+ ^a	-
M4	Loss of O and C ₁₂ H ₂₀ O ₁₀ [M - H] ⁻	C ₁₅ H ₁₂ O ₄	255.0656	-2.8	18.43	75.7	239.0307, 135.0082, 119.0495, 93.0350	2.99394	-	-	+ ^a	-
M5	Loss of C ₁₂ H ₂₀ O ₉ + loss of CO [M - H] ⁻	C ₁₄ H ₁₂ O ₅	259.0610	-0.6	11.92	75.0	241.0487, 213.0566, 195.1086, 183.0853, 135.0081	1.59026	-	-	+ ^b	-
M6	Loss of C ₁₂ H ₂₀ O ₁₀ + loss of water [M - H] ⁻	C ₁₅ H ₁₀ O ₄	253.0507	0.3	18.18	75.0	225.0568, 209.0609, 181.0649, 119.0499	3.73694	+ ^a	+ ^b	+ ^a	+ ^b
M7	Loss of C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M - H] ⁻	C ₁₅ H ₁₄ O ₅	273.0772	1.3	16.57	75.0	245.0475, 227.1310, 209.1184, 165.1286, 135.0453	0.9752	-	-	-	+ ^a
M8	Loss of O and C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M - H] ⁻	C ₁₅ H ₁₄ O ₄	257.0817	-1.0	14.82	75.0	239.1684, 213.0584, 135.0454, 121.0226, 119.0503	1.462	-	-	+ ^{a,b}	-
M9	Loss of O and C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M - H] ⁻	C ₁₅ H ₁₄ O ₄	257.0829	3.7	18.59	70.7	241.0750, 213.0224, 137.0357, 119.0472	1.5722	-	-	+ ^{a,b}	-
M10	Loss of C ₁₂ H ₂₀ O ₁₀ + ketone formation [M - H] ⁻	C ₁₅ H ₁₀ O ₆	285.0401	-1.2	15.78	75.0	267.0283, 257.0428, 223.0290, 149.0245, 133.0294	1.48794	-	-	+ ^b	-
M11	Loss of C ₆ H ₁₀ O ₄ + N-acetylation [M - H] ⁻	C ₂₃ H ₂₄ O ₁₂	491.1167	-5.8	15.07	65.5	473.0057, 367.1394, 287.1502, 125.0238	0.53447	-	+ ^a	+ ^b	+ ^{a,b}
M12	Loss of C ₆ H ₁₀ O ₅ + N-acetylation [M - H] ⁻	C ₂₃ H ₂₄ O ₁₁	475.1245	-0.3	15.35	62.5	429.1966, 329.0920, 270.9431, 135.0438, 119.0481	0.0924672	-	+ ^a	-	-
M13	Loss of O and C ₁₂ H ₂₀ O ₁₀ + di-acetylation of amines [M - H] ⁻	C ₁₉ H ₁₆ O ₆	339.0869	-0.7	11.63	75.3	310.9990, 254.9868, 218.9874, 119.0523	1.9929	-	-	-	+ ^b
M14	Loss of O and C ₁₂ H ₂₀ O ₁₀ + di-acetylation of amines [M - H] ⁻	C ₁₉ H ₁₆ O ₆	339.0871	-1.1	11.85	75.5	311.0684, 254.9799, 177.2543, 161.0261	2.28159	-	-	-	+ ^b
M15	Loss of C ₁₂ H ₂₀ O ₉ + oxidation [M - H] ⁻	C ₁₅ H ₁₂ O ₇	303.0508	-0.8	15.73	75.0	285.0379, 257.1777, 166.9987, 151.0021, 135.0440	1.13085	-	-	+ ^a	-
M16	Loss of C ₆ H ₁₀ O ₄ + di-oxidation [M - H] ⁻	C ₂₁ H ₂₂ O ₁₃	481.1023	7.3	15.44	49.3	463.2680, 435.2738, 345.2393, 319.0931	-1.41361	-	+ ^a	-	-
M17	Loss of C ₁₂ H ₂₀ O ₉ + di-oxidation [M - H] ⁻	C ₁₅ H ₁₂ O ₈	319.0431	-8.8	15.31	45.4	303.1599, 300.9846, 291.1541, 182.9905, 167.0013, 151.0379, 135.0470	0.533851	-	-	-	+ ^a
M18	Methylation [M - H] ⁻	C ₂₈ H ₃₄ O ₁₅	609.1825	0.0	13.20	75.0	301.0714, 286.0456, 254.5695, 165.0182, 151.0027	-0.914082	-	+ ^a	-	-
M19	Methylation [M - H] ⁻	C ₂₈ H ₃₄ O ₁₅	609.1816	-1.4	13.51	75.0	301.0708, 286.0455, 254.0735, 151.0023	-0.290878	-	+ ^a	-	-
M20	Loss of C ₆ H ₁₀ O ₅ + methylation [M - H] ⁻	C ₂₂ H ₂₄ O ₁₀	447.1292	-1.0	23.83	75.0	417.1150, 287.0628, 259.1729, 149.0006, 135.0779	0.662468	-	+ ^{a,b}	+ ^{a,b}	-
M21	Loss of C ₁₂ H ₂₀ O ₉ + methylation [M - H] ⁻	C ₁₆ H ₁₄ O ₆	301.0718	0.2	16.81	75.0	287.0544, 285.0411, 165.0185, 135.0448	1.91062	+ ^{a,b}	+ ^a	+ ^{a,b}	+ ^{a,b}
M22	Loss of C ₁₂ H ₂₀ O ₉ + methylation [M - H] ⁻	C ₁₆ H ₁₄ O ₆	301.0722	1.3	16.99	75.0	287.0544, 285.0411, 165.0185, 135.0448	2.37062	+ ^{a,b}	+ ^a	+ ^{a,b}	+ ^{a,b}
M23	Loss of C ₁₂ H ₂₀ O ₁₀ + methylation [M - H] ⁻	C ₁₆ H ₁₄ O ₅	285.0763	-2.0	14.06	75.0	266.9834, 239.0535, 149.0046, 135.0018	2.23859	-	+ ^a	-	-
M24	Loss of O and C ₆ H ₁₀ O ₅ + methylation [M - H] ⁻	C ₂₂ H ₂₄ O ₉	431.1382	7.9	23.43	60.3	398.9752, 351.1731, 297.1256, 134.9860, 133.0041, 119.0479	2.02247	-	+ ^b	+ ^{a,b}	+ ^b
M25	Loss of O and C ₁₂ H ₂₀ O ₁₀ + methylation [M - H] ⁻	C ₁₆ H ₁₄ O ₄	269.0828	3.1	16.37	59.7	253.0508, 241.1208, 225.0570, 223.1369, 179.1056, 135.0772, 132.9712	2.6571	-	-	+ ^a	-



Table 1 (Contd.)

Metabolites ID	Composition	Formula	<i>m/z</i>	Error (ppm)	R.T. (min)	Score (%)	MS/MS fragments	<i>clog P</i>	Plasma	Bile	Urine	Faeces
M26	Loss of C ₁₂ H ₂₀ O ₉ + oxidation and methylation [M – H] [–]	C ₁₆ H ₁₄ O ₇	317.0670	1.1	16.81	75.0	302.0435, 299.0569, 271.0616, 181.0141, 165.0187, 151.0036, 135.0447	1.30533	–	–	–	+ ^{a,b}
M27	Loss of C ₁₂ H ₂₀ O ₉ + sulfate conjugation [M – H] [–]	C ₁₅ H ₁₂ O ₉ S	367.0125	–1.0	13.47	50.0	349.2043, 321.1743, 287.0567, 269.0479, 231.1580, 151.0033, 135.0446	–0.235379	+ ^a	–	–	+ ^{a,b}
M28	Loss of C ₁₂ H ₂₀ O ₉ + sulfate conjugation [M – H] [–]	C ₁₅ H ₁₂ O ₉ S	367.0125	–1.2	14.37	50.0	349.2022, 321.1694, 287.0567, 269.0451, 231.0159, 151.0032, 135.0446	0.224621	+ ^a	–	–	+ ^{a,b}
M29	Loss of C ₁₂ H ₂₀ O ₉ + bis-sulfate conjugation [M – H] [–]	C ₁₅ H ₁₂ O ₁₂ S ₂	446.9695	–0.5	12.68	40.8	402.8724, 367.0135, 311.2528, 287.0566, 151.0041, 135.0434	–3.72615	+ ^a	–	–	–
M30	Loss of C ₁₂ H ₂₀ O ₁₀ + bis-sulfate conjugation [M – H] [–]	C ₁₅ H ₁₂ O ₁₁ S ₂	430.9746	–0.6	18.38	50.0	351.0581, 270.9703, 294.9822, 134.9922, 118.9923	–0.62306	–	+ ^b	–	–
M31	Loss of C ₁₂ H ₂₀ O ₁₀ + glutamine conjugation [M – H] [–]	C ₂₀ H ₂₀ N ₂ O ₇	399.1164	–8.4	13.97	59.1	353.1116, 335.0996, 291.1097, 263.0735	0.68033	–	+ ^a	–	–
M32	Loss of O and C ₁₂ H ₂₀ O ₁₀ + glycine conjugation [M – H] [–]	C ₁₇ H ₁₅ NO ₅	312.0889	3.8	14.54	70.6	268.1001, 239.1068, 175.9600	2.62494	–	–	–	+ ^a

^a + detected; – undetected. ^b Obtained by methanol precipitation protein; ^c obtained by ethyl acetate extraction; ^d obtained by methanol precipitation protein and ethyl acetate extraction.

3.3. Identification of metabolites *in vivo*

Metabolite M1 (C₁₅H₁₂O₆) eluted at 15.30 min and had a deprotonated molecular ion [M – H][–] at *m/z* 287.0560, 308 Da (C₁₂H₂₀O₉) lower than that of the parent drug. The crucial fragment ions of M1 at *m/z* 269.0458 ([M–H₂O–H][–]), *m/z* 241.0489 ([M–H₂O–CO–H][–]), *m/z* 151.0032 (RDA reaction) and *m/z* 135.0447 (RDA reaction) were detected according to the secondary mass spectrum of M1.

Metabolite M2 (C₁₅H₁₂O₅) was identified with a peak at *m/z* 271.0615 in the UPLC system, which eluted at 16.99 min and was 324 Da (C₁₂H₂₀O₁₀) lower than that of M0. Based on the MS/MS information, the typical fragment ions of M2 at *m/z* 253.0507 ([M–H₂O–H][–]), *m/z* 243.0682 ([M–CO–H][–]), *m/z* 227.0715 ([M–CO–O–H][–]) and *m/z* 135.0447 (RDA reaction) were observed.

Metabolites M3 and M4 (C₁₅H₁₂O₄) are isomers with the deprotonated molecular ions [M – H][–] at *m/z* 255.0665 and 255.0656, which were 16 Da (O) lower than that of M2. They eluted at 15.41 min and 18.43 min, respectively. Based on the secondary fragment ions of [M – H][–] at *m/z* 237.2257 ([M–H₂O–H][–]) and 109.0795 ([M–C₉H₆O₂–H][–]) generated by M3, the B-ring remained the adjacent hydroxyl group, and the loss of O occurred at C-5 in ring A. Moreover, the *clog P* values of M3 and M4 were 2.2109 and 2.99394, respectively, aiding in their verification.

Metabolite M5 (C₁₄H₁₂O₅) eluted at 11.92 min and displayed a deprotonated molecular ion [M – H][–] at *m/z* 259.0610, 28 Da (CO) lower than that of M1. In its secondary mass spectrum, the diagnostic fragment ions produced by M5 at *m/z* 241.0487 ([M–H₂O–H][–]), *m/z* 213.0566 ([M–H₂O–CO–H][–]), *m/z* 195.1086 ([M–4O–H][–]), *m/z* 183.0853 ([M–3O–CO–H][–]) and *m/z* 135.0081 (RDA reaction) were observed.

Metabolite M6 (C₁₅H₁₀O₄) had a deprotonated molecular ion [M – H][–] at *m/z* 253.0507 and observed at the retention time of 18.18 min, which was 18 Da (H₂O) lower than that of M2. M6 contained representative fragment ions at *m/z* 225.0568 ([M–CO–H][–]), *m/z* 209.0609 ([M–CO₂–H][–]) and *m/z* 181.0649 ([M–CO₂–CO–H][–]) in its secondary mass spectrum. Moreover, the noteworthy fragment ion at *m/z* 119.0499 was created by the loss of O through the RDA reaction of M6.

Metabolite M7 (C₁₅H₁₄O₅) eluted at 16.57 min with the deprotonated molecular ion [M – H][–] at *m/z* 273.0772, which was 2 Da (2H) higher than M2. Crucial fragment ions at *m/z* 245.0475, 227.1310, 209.1184, 165.1286 and 135.0453 were produced by the loss of CO, CH₂O₂, 4O, C₆H₄O₂ and the RDA reaction, respectively.

Metabolites M8 and M9 (C₁₅H₁₄O₄) eluted at 14.82 min and 18.59 min, respectively, with deprotonated molecular ions [M – H][–] at *m/z* 257.0817 and 257.0829, which were 2 Da (2H) higher than M3 and M4. The representative fragment ions at *m/z* 239.1684, 135.0454 and 121.0226 were generated by M8, implying that the loss of O occurred in ring A. Similarly, the prominent fragment ions of metabolite M9 at *m/z* 137.0357 and 119.0472 indicated that the loss of O occurred in the B ring. Furthermore, the *clog P* values of M8 and M9 were 1.462 and 1.5722, respectively, so their structural identities were confirmed.

Table 2 Summary of metabolites of eriocitrin in rat liver microsomes^a

Metabolites ID	Composition	Formula	<i>m/z</i>	Error (ppm)	R.T. (min)	Score (%)	MS/MS fragments	<i>clog P</i>	Blanks	Controls	Samples
N1	Loss of C ₆ H ₁₀ O ₅ [M – H] [–]	C ₂₁ H ₂₂ O ₁₀	433.1159	4.4	15.87	85.9	417.0309, 353.1632, 271.0645, 135.2507	0.83947	–	–	+
N2	Loss of C ₁₂ H ₂₀ O ₉ [M – H] [–]	C ₁₅ H ₁₂ O ₆	287.0562	0.4	15.59	96.3	269.0391, 241.2241, 151.0039, 135.0451, 125.0236	1.84785	–	–	+
N3	Desaturation [M – H] [–]	C ₂₇ H ₃₀ O ₁₅	593.1508	–0.6	12.59	95.4	549.1892, 447.0645, 285.0406, 151.0032	–0.21048	–	–	+
N4	Loss of C ₆ H ₁₀ O ₄ + loss of water [M – H] [–]	C ₂₁ H ₂₀ O ₁₀	431.0976	–1.7	14.49	76.6	399.0170, 387.0883, 313.0631, 269.0528	1.62817	–	–	+
N5	Loss of C ₆ H ₁₀ O ₃ + hydrogenation [M – H] [–]	C ₂₁ H ₂₄ O ₁₀	435.1320	5.4	15.66	67.4	389.1652, 298.9101, 273.0560, 153.0836, 135.0666	–0.52288	–	–	+
N6	Loss of C ₆ H ₁₀ O ₄ + <i>N</i> -acetylation [M – H] [–]	C ₂₃ H ₂₄ O ₁₂	491.1161	–7.0	15.42	64.0	473.2817, 445.1889, 287.1221, 125.0191	0.53447	–	–	+
N7	Loss of C ₆ H ₁₀ O ₃ + demethylation to carboxylic acid [M – H] [–]	C ₂₁ H ₂₀ O ₁₂	463.0868	–3.0	12.86	83.0	417.1839, 271.0245, 258.9180, 151.0005	–0.19068	–	–	+
N8	Loss of C ₆ H ₁₀ O ₃ + ketone formation [M – H] [–]	C ₂₁ H ₂₀ O ₁₁	447.0925	–1.9	13.67	80.6	429.3176, 301.0338, 285.0364, 151.0020	–0.06953	–	–	+
N9	Loss of C ₁₂ H ₂₀ O ₁₀ + ketone formation [M – H] [–]	C ₁₅ H ₁₀ O ₆	285.0406	0.4	15.85	84.6	267.1811, 257.0448, 239.0672, 223.0274, 119.0614	1.48794	–	–	+
N10	Oxidation [M – H] [–]	C ₂₇ H ₃₂ O ₁₆	611.1615	–0.3	12.39	78.9	303.0505, 287.0565, 227.9309, 151.0036, 135.0468	–1.45408	–	–	+
N11	Loss of C ₆ H ₁₀ O ₃ + methylation [M – H] [–]	C ₂₂ H ₂₄ O ₁₀	447.1328	6.9	23.89	65.2	417.1216, 285.1388, 148.9971, 135.0752	0.662468	–	–	+
N12	Loss of C ₁₂ H ₂₀ O ₉ + oxidation and methylation [M – H] [–]	C ₁₆ H ₁₄ O ₇	317.0694	8.6	16.61	70.4	299.0555, 273.0489, 180.9719, 165.0828	1.30533	–	–	+

^a + detected; – undetected.

Table 3 Summary of metabolites of eriocitrin in rat intestinal flora^a

Metabolites ID	Composition	Formula	<i>m/z</i>	Error (ppm)	R.T. (min)	Score (%)	MS/MS fragments	<i>clog P</i>	Blanks	Controls	Samples
Q1	Loss of C ₆ H ₁₀ O ₃ [M – H] [–]	C ₂₁ H ₂₂ O ₁₀	433.1166	6.0	15.96	65.0	417.2409, 387.2026, 271.0645, 151.3214	0.83947	–	–	+
Q2	Loss of C ₁₂ H ₂₀ O ₉ [M – H] [–]	C ₁₃ H ₁₂ O ₆	287.0561	0.1	15.74	89.3	269.0485, 241.2146, 151.0038, 135.0451, 125.0239	1.84785	–	–	+
Q3	Loss of C ₁₂ H ₂₀ O ₁₀ [M – H] [–]	C ₁₃ H ₁₂ O ₅	271.0611	–0.3	17.02	82.1	253.1802, 243.0655, 227.0713, 135.0105	2.39694	–	–	+
Q4	Loss of O and C ₁₂ H ₂₀ O ₁₀ [M – H] [–]	C ₁₃ H ₁₂ O ₄	255.0675	1.0	15.52	79.9	237.2196, 135.0113, 119.0398, 109.0724	2.2109	–	–	+
Q5	Loss of O and C ₁₂ H ₂₀ O ₁₀ [M – H] [–]	C ₁₃ H ₁₂ O ₄	255.0659	–2.8	18.49	75.7	239.0385, 135.0107, 119.0420, 93.0267	2.99394	–	–	+
Q6	Loss of C ₁₂ H ₂₀ O ₉ + loss of CO [M – H] [–]	C ₁₄ H ₁₂ O ₅	259.0608	–1.5	11.79	75.0	241.0505, 213.0568, 195.1018, 183.0887, 135.0011	1.59026	–	–	+
Q7	Loss of C ₁₂ H ₂₀ O ₉ + loss of water [M – H] [–]	C ₁₃ H ₁₀ O ₅	269.0456	0.3	16.98	84.3	241.0545, 225.0605, 197.0598, 181.0638, 151.0020	3.18785	–	–	+
Q8	Loss of C ₁₂ H ₂₀ O ₁₀ + loss of water [M – H] [–]	C ₁₃ H ₁₀ O ₄	253.0505	–0.4	18.33	75.3	225.0217, 209.0610, 181.0660, 119.0488	3.73694	–	–	+
Q9	Loss of C ₆ H ₁₀ O ₃ + hydrogenation [M – H] [–]	C ₂₁ H ₂₄ O ₁₀	435.1323	6.0	15.32	65.6	391.6900, 389.2463, 273.0863, 135.0793	–0.5229	–	–	+
Q10	Loss of C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M – H] [–]	C ₁₃ H ₁₄ O ₅	273.0766	–1.0	16.33	75.0	245.0473, 227.1979, 209.1302, 165.0140, 135.0435	0.9752	–	–	+
Q11	Loss of O and C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M – H] [–]	C ₁₃ H ₁₄ O ₄	257.0818	–0.5	14.90	75.6	239.0718, 213.0933, 135.0455, 121.0304, 119.0492	1.462	–	–	+
Q12	Loss of O and C ₁₂ H ₂₀ O ₁₀ + hydrogenation [M – H] [–]	C ₁₃ H ₁₄ O ₄	257.0820	0.3	18.61	75.1	241.0505, 213.0933, 137.0228, 119.0498	1.5722	–	–	+
Q13	Loss of C ₁₂ H ₂₀ O ₁₀ + ketone formation [M – H] [–]	C ₁₃ H ₁₀ O ₆	285.0403	–0.6	15.86	75.8	267.0287, 257.0427, 223.0309, 149.0227, 133.0291	1.48794	–	–	+
Q14	Loss of O and C ₁₂ H ₂₀ O ₁₀ + N-acetylation [M – H] [–]	C ₁₇ H ₁₄ O ₅	297.0762	–2.2	14.81	74.4	281.0060, 264.9847, 253.0871, 160.9785, 135.0547	3.00994	–	–	+
Q15	Loss of O and C ₁₂ H ₂₀ O ₁₀ + di-acetylation of amines [M – H] [–]	C ₁₉ H ₁₆ O ₆	339.0872	–0.6	11.52	75.3	311.0629, 297.1756, 254.9424, 219.0413, 119.0621	1.9929	–	–	+
Q16	Loss of O and C ₁₂ H ₂₀ O ₁₀ + di-acetylation of amines [M – H] [–]	C ₁₉ H ₁₆ O ₆	339.0872	–0.8	11.70	75.5	311.1732, 297.2141, 254.9866, 177.8865, 161.0382	2.28159	–	–	+
Q17	Loss of C ₁₂ H ₂₀ O ₉ + oxidation [M – H] [–]	C ₁₃ H ₁₂ O ₇	303.0521	3.5	15.49	71.2	285.1301, 257.2180, 151.0760, 135.0594	1.13085	–	–	+
Q18	Loss of C ₁₂ H ₂₀ O ₁₀ + methylation [M – H] [–]	C ₁₆ H ₁₄ O ₅	285.0764	–1.6	14.09	78.8	269.0475, 267.2314, 239.0588, 149.0227, 135.0007	2.23859	–	–	+
Q19	Loss of O and C ₁₂ H ₂₀ O ₁₀ + methylation [M – H] [–]	C ₁₆ H ₁₄ O ₄	269.0822	1.0	16.31	75.0	253.0590, 223.1247, 179.0093, 135.0695, 132.9781	2.6571	–	–	+
Q20	Loss of C ₁₂ H ₂₀ O ₉ + oxidation and methylation [M – H] [–]	C ₁₆ H ₁₄ O ₇	317.0671	1.3	16.70	75.6	302.0746, 299.0547, 271.0600, 180.9713, 165.0804, 151.0016, 135.0452	1.30533	–	–	+

^a + detected; – undetected.

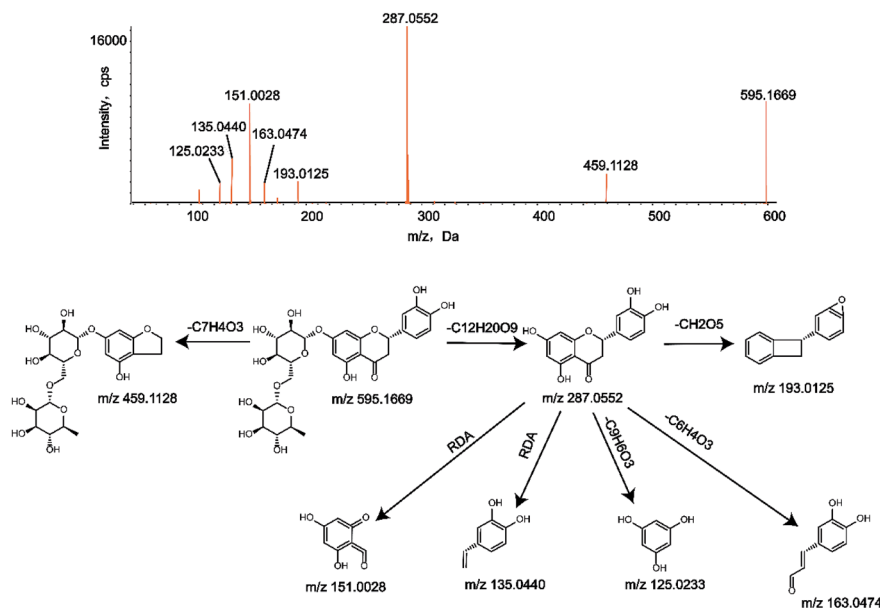


Fig. 1 MS/MS spectrum of eriocitrin and its predominant fragmentation pathways.

Metabolite M10 ($C_{15}H_{10}O_6$) eluted at 15.78 min and exhibited the deprotonated molecular ion $[M - H]^-$ at m/z 285.0401, which was 14 Da higher than that of M2. The characteristic fragment ions of M10 at m/z 267.0283 ($[M - H_2O - H]^-$), 257.0428 ($[M - CO - H]^-$), 223.0290 ($[M - H_2O - CO - O - H]^-$) and 149.0245 (RDA reaction) were obtained according to the MS/MS information.

Metabolite M11 ($C_{23}H_{24}O_{12}$) eluted at 15.07 min and displayed the deprotonated molecular ion $[M - H]^-$ at m/z 491.1167, 104 Da lower than that of the parent drug. M11 had typical fragment ions at m/z 473.0057 ($[M - H_2O - H]^-$), m/z 367.1394 ($[M - C_6H_4O_3 - H]^-$), m/z 287.1502 ($[M - C_8H_{12}O_6 - H]^-$), and m/z 125.0238 ($[M - C_{17}H_{18}O_9 - H]^-$), as seen in the secondary mass spectrum.

Metabolite M12 ($C_{23}H_{24}O_{11}$) was detected at a retention time of 15.35 min with the deprotonated molecular ion $[M - H]^-$ at m/z 475.1245, 120 Da lower than that of eriocitrin, which exhibited the characteristic fragment ions at m/z 429.1966 ($[M - CO - H_2O - H]^-$), m/z 329.0920 ($[M - C_6H_{10}O_4 - H]^-$), m/z 270.9431 ($[M - C_6H_4O_2 - C_2H_2O_2 - H]^-$) and m/z 135.0438 (RDA reaction) in the MS/MS information.

Metabolites M13 and M14 ($C_{19}H_{16}O_6$) eluted at 11.63 min and 11.85 min, respectively, with the deprotonated molecular ions $[M - H]^-$ at m/z 339.0869 and 339.0871, which were 84 Da higher than those of M3 and M4, implying that the diacetylation reaction occurred based on the loss of O and $C_{12}H_{20}O_{10}$. According to the fragment ions at m/z 218.9874 and 119.0523 acquired after the RDA reaction, the loss of O occurred in the A ring, while the diacetylation occurred in the B ring. Similarly, the fragment ions at m/z 177.0911 and 161.0237 indicated that the loss of O occurred in ring B and diacetylation occurred in both rings A and B. Therefore, the structures of metabolites M13 and M14, which are isomers of each other, were determined. In addition, they were also validated with $\log P$ values of M13 and M14, which were 1.9929 and 2.28159, respectively.

Metabolite M15 ($C_{15}H_{12}O_7$) was detected at 15.73 min and showed the deprotonated molecular ion $[M - H]^-$ at m/z 303.0508, 16 Da (O) higher than that of M1, which contained the characteristic fragment ions at m/z 285.0379 ($[M - H_2O - H]^-$), m/z 257.1777 ($[M - CO - H_2O - H]^-$), m/z 151.0021 (RDA reaction) and m/z 135.0440 (RDA reaction) according to the secondary mass spectrum.

Metabolite M16 ($C_{21}H_{22}O_{13}$) exhibited an $[M - H]^-$ ion of m/z 481.1023 that eluted at 15.44 min and was 114 Da lower than that of M0. According to its secondary mass spectrum, the characteristic fragment ions produced at m/z 463.2680 ($[M - H_2O - H]^-$), m/z 435.2738 ($[M - CO - H_2O - H]^-$), m/z 345.2393 (RDA reaction) and m/z 319.0931 ($[M - C_6H_{10}O_5 - H]^-$) were observed and identified.

Metabolite M17 ($C_{15}H_{12}O_8$) eluted at a retention time of 15.31 min. The MS/MS spectrum showed the deprotonated molecular ion $[M - H]^-$ at m/z 319.0431, which was 32 Da (2O) higher than that of M1. The metabolite M17 generated characteristic fragment ions at m/z 303.1599, 300.9846 and 291.1541 through losing O, H_2O , and CO, respectively, and the possible structures of the metabolite M17 were predicted by the characteristic fragment ions at m/z 182.9905, 167.0013, 151.0379 and 135.0470 generated by RDA cleavage.

Metabolites M18 and M19 ($C_{28}H_{34}O_{15}$) eluted at 13.20 min and 13.51 min, respectively, showing the deprotonated molecular ions $[M - H]^-$ at m/z 609.1825 and 609.1816, which were 14 Da (CH_2) higher than that of eriocitrin. Metabolites successively lost $C_{12}H_{20}O_9$, CH_3 , and 2O and underwent RDA reactions, resulting in crucial product ions at m/z 301.0714, 286.0456, 254.5695 and 151.0027. $\log P$ values of metabolites M18 and M19 were -0.914082 and -0.290878 , respectively, so they were verified.

Metabolite M20 ($C_{22}H_{24}O_{10}$) eluted at 23.83 min and had a deprotonated molecular ion $[M - H]^-$ at m/z 447.1292, 148 Da lower than that of M0. The characteristic fragment ions of M20



at m/z 417.1150 ($[M-CH_2O-H]^-$), m/z 287.0628 ($[M-C_6H_{10}O_4-CH_2-H]^-$), m/z 259.1729 ($[M-C_6H_{10}O_4-CH_2-CO-H]^-$) and m/z 149.0006 (RDA reaction) were detected according to its secondary mass spectrum. After RDA cleavage of the metabolite M20, a methoxy group was lost to obtain a secondary fragment ion at m/z 135.0779. Hence, the possible metabolites of it were inferred.

Metabolites M21 and M22 ($C_{16}H_{14}O_6$) eluted at 16.81 min and 16.99 min, respectively. They had deprotonated molecular ions $[M-H]^-$ at m/z 301.0718 and 301.0722, which were 14 Da (CH_2) higher than that of M1. Characteristic ions at m/z 287.0544 and

285.0411 were formed through the loss of CH_2 and O, respectively. According to the fragment ions at m/z 165.0185 and 135.0448 produced by RDA reaction, methylation occurred in the A ring. In addition, they were validated with the clog P values of M21 and M22, which were 1.91062 and 2.37062, respectively.

Metabolite M23 ($C_{16}H_{14}O_5$) eluted at 14.06 min and displayed a deprotonated molecular ion $[M-H]^-$ at m/z 285.0763, 14 Da (CH_2) higher than that of M2. The noteworthy fragment ions at m/z 149.0046 and 135.0018 were yielded by RDA cleavage, indicating that methylation occurred at position 5 in ring A, so the structure was determined.

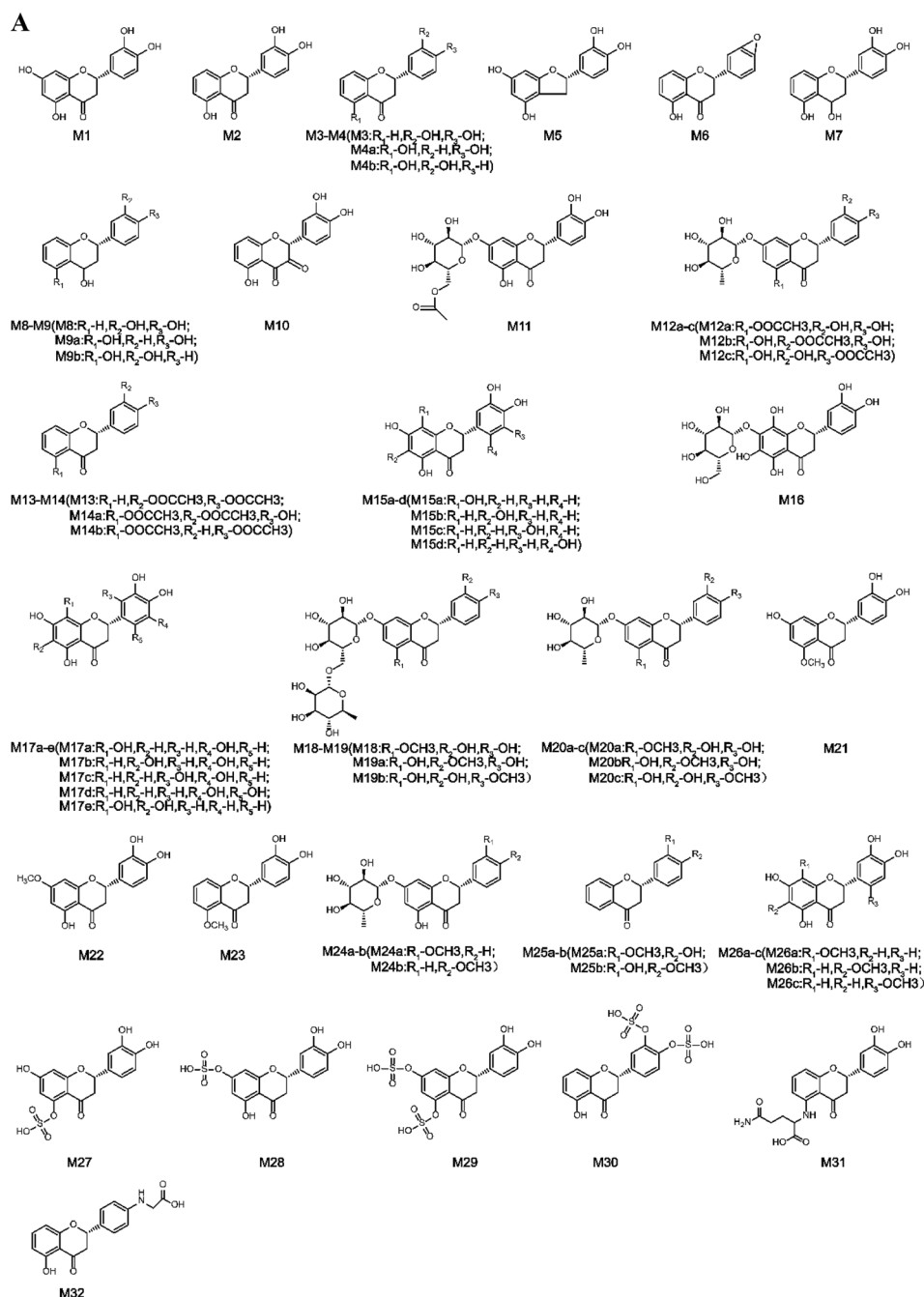


Fig. 2 Structures of all metabolites of eriocitrin *in vivo* and *in vitro* ((A) *in vivo*, (B) in rat liver microsomes, (C) in rat intestinal flora. (a–d) Possible chemical structure).



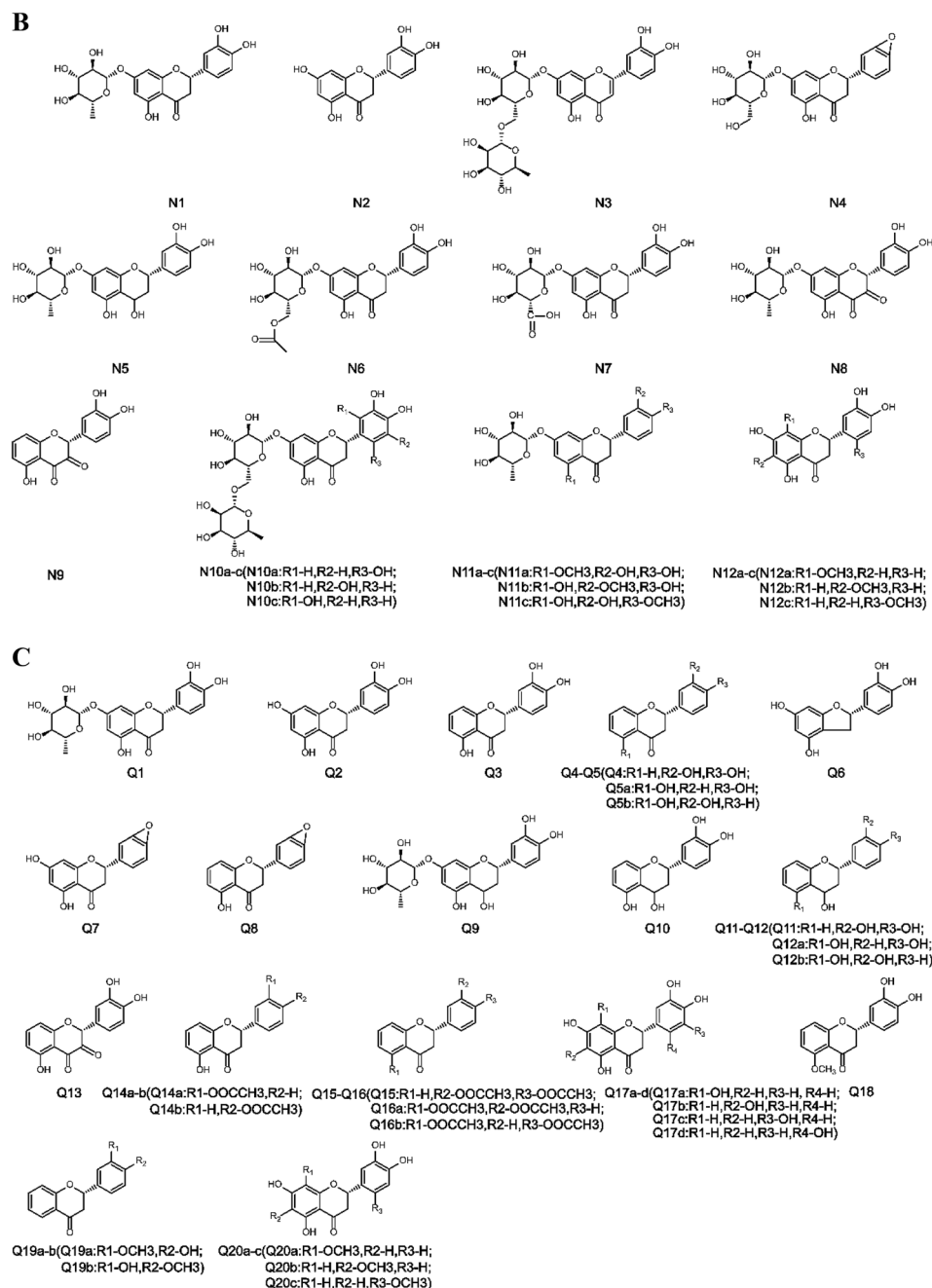


Fig. 2 (contd.)

Metabolite M24 ($C_{22}H_{24}O_9$), with a deprotonated molecular ion of m/z 431.1382 and retention time of 23.43 min, was 164 Da lower than that of M0. The characteristic fragment ions at m/z 398.9752, 351.1731, 297.1256 and 133.0041 were formed by the loss of O, 5O and RDA cleavage, while the fragment ions at m/z 297.1256 and 133.0041 created the ions at m/z 134.9860 and 119.0479 via $C_6H_{10}O_5$ and CH_2 loss, respectively. Thus, M24 was illustrated according to the above information.

Metabolite M25 ($C_{16}H_{14}O_4$) with the $[M - H]^-$ ion of m/z 269.0828, eluted at 16.37 min and had a mass 14 Da (CH_2) higher than that of M3. Secondary fragment ions at m/z 253.0508, 223.1369 and 179.1056 were generated by the successive loss of O, CH_2O and CO_2 , while fragment ions at m/z

241.1208 and 225.0570 were produced by losing CO and O successively. According to the dominant fragment ions at m/z 132.9712 and 135.0772 gained by the RDA reaction, both oxygen loss and methylation occurred in the B ring.

Metabolite M26 ($C_{16}H_{14}O_7$) was detected at 16.81 min and displayed a deprotonated molecular ion $[M - H]^-$ at m/z 317.0670, 30 Da (CH_2O) higher than that of M1. Based on the MS/MS spectrum, M26 contained characteristic fragment ions at m/z 302.0435 ($[M - CH_3 - H]^-$), 299.0569 ($[M - H_2O - H]^-$), and 271.0616 ($[M - CH_2O - O - H]^-$), and the possible structures of M26 could be inferred by combining the evaluation with the fragment ions at m/z 181.0141, 165.0187, 151.0036 and 135.0447 caused by RDA reaction.



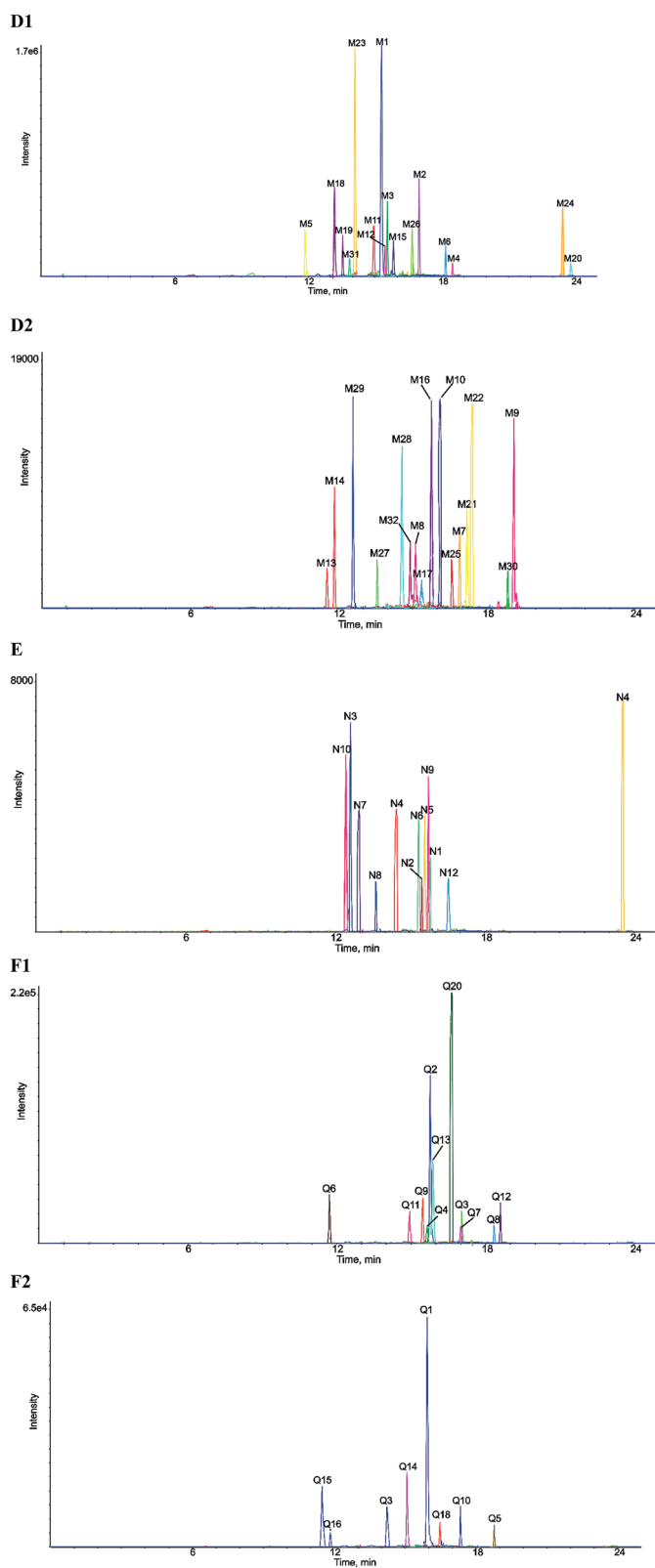


Fig. 3 Extracted ion chromatograms of all metabolites of eriocitrin *in vivo* and *in vitro* (D1) and (D2) *in vivo*, (E) in rat liver microsomes, (F1) and (F2) in rat intestinal flora.

Metabolites M27 and M28 ($C_{15}H_{12}O_9S$) were detected in the negative-ion full scan mass spectra as deprotonated $[M - H]^-$ ions at m/z 367.0125, which eluted at 13.47 min and 14.37 min, respectively, and were 80 Da (SO_3) higher than that of M1. According to the fragment ion at m/z 349.2043 generated by water loss and the fragment ion at m/z 231.1580 produced after the RDA reaction, sulfate conjugation took place in the A ring. Other characteristic fragment ions at m/z 321.1743 ($[M - H_2O - CO - H]^-$), 287.0567 ($[M - SO_3 - H]^-$), 269.0479 ($[M - SO_3 - H_2O - H]^-$), 231.1580 (RDA reaction) and 135.0446 (RDA reaction) were also created, and the fragment ion at m/z 151.0033 was obtained by losing SO_3 from the fragment ion at m/z 231.1580. In addition, the *clog P* values of metabolites M27 and M28 were -0.235379 and 0.224621 , respectively. Hence, they were distinguished.

Metabolite M29 ($C_{15}H_{12}O_{12}S_2$) eluted at 12.68 min and showed the deprotonated molecular ion $[M - H]^-$ at m/z 446.9695, 160 Da ($2SO_3$) higher than that of M1, and generated characteristic product ions at m/z 402.8724, 367.0135, 287.0566, 311.2528 and 151.0041 by the loss of CO_2 , SO_3 , $2SO_3$ and RDA reaction, respectively. The fragment ion at m/z 311.2528 produced by the RDA reaction generated the fragment ion at m/z 151.0041 by loss of $2SO_3$.

Metabolite M30 ($C_{15}H_{12}O_{11}S_2$) had an $[M - H]^-$ ion of m/z 430.9746 that eluted at 18.38 min and was 160 Da ($2SO_3$) higher than that of M2. Secondary fragment ions at m/z 351.0581, 270.9703, 294.9822, and 134.9922 were generated by M30 through the loss of SO_3 , $2SO_3$ and RDA cleavage, respectively, which implied that bis-sulfate conjugation occurred in the B ring.

Metabolite M31 ($C_{20}H_{20}N_2O_7$) was detected at 13.97 min with the deprotonated molecular ion $[M - H]^-$ at m/z 399.1164, 196 Da lower than that of the parent drug. The distinctive fragment ion at m/z 271.0683 was observed, corresponding to the loss of glutamine.^{31–33} Furthermore, a strong ion at m/z 263.0735 appeared in the secondary mass spectrum after the RDA reaction, and other characteristic fragment ions at m/z 353.1116 ($[M - CO - H_2O - H]^-$), m/z 335.0996 ($[M - 4O - H]^-$) and m/z 291.1097 ($[M - C_6H_4O_2 - H]^-$) were observed according to the MS/MS spectrum.

Metabolite M32 ($C_{17}H_{15}NO_5$) showed the $[M - H]^-$ ion at m/z 312.0889 that eluted at 14.54 min and was 57 Da higher than that of M3 and M4, indicating that it may experience the loss of O and $C_{12}H_{20}O_{10}$ followed by glycine conjugation.^{31,32} According to the secondary fragment ion at m/z 175.9600 produced by RDA cleavage, the structure of M32 could be inferred. In addition, the characteristic fragment ions at m/z 268.1001 and 239.1068 were produced by M32 through the loss of CO_2 and $C_2H_3NO_2$.

3.4. Identification of metabolites *in vitro*

3.4.1. Identification of metabolites in rat liver microsomes.

Metabolite N1 ($C_{21}H_{22}O_{10}$) was detected in the negative-ion full scan mass spectrum as the deprotonated molecular ion $[M - H]^-$ at m/z 433.1159, which eluted at 15.87 min, was 162 Da lower than that of eriocitrin. The typical fragment ions at m/z 417.0309, 353.1632, 271.0645 and 135.2507 were observed



in the MS/MS spectrum through the loss of O, 5O, C₆H₁₀O₅ and RDA cleavage, respectively.

Metabolite N2 (C₁₅H₁₂O₆) eluted at 15.59 min and had the deprotonated molecular ion [M – H][–] at *m/z* 287.0562, a loss of C₁₂H₂₀O₉ when compared with eriocitrin. The prominent fragment ions at *m/z* 269.0391, 241.2241, 151.0039, and 135.0451 were consistent with M1; hence, its structure was identified.

Metabolite N3 (C₂₇H₃₀O₁₅) eluted at 12.59 min having a deprotonated molecular ion [M – H][–] at *m/z* 593.1508, 2 Da (2H) lower than that of eriocitrin, which implied that desaturation occurred. According to the fragment ion at *m/z* 285.0406, C₁₂H₂₀O₉ was lost from the parent ion, suggesting that desaturation occurred in the flavonoid skeleton. Other typical fragment ions of N3 at *m/z* 549.1892 ([M–CO₂–H][–]), *m/z* 447.0645 ([M–C₆H₁₀O₄–H][–]), and *m/z* 151.0039 (RDA reaction) were also observed, which helped to confirm the structure.

Metabolite N4 (C₂₁H₂₀O₁₀) was detected at 14.49 min with the deprotonated molecular ion [M – H][–] at *m/z* 431.0976, 164 Da lower than that of eriocitrin. Metabolite N4 generated fragment ions at *m/z* 399.0170, 387.0883, 269.0528 and 313.0631 by loss of 2O, CO₂, C₆H₁₀O₅ and RDA cleavage, respectively, so N4 could be determined.

Metabolite N5 (C₂₁H₂₄O₁₀) was detected at 15.66 min and displayed a deprotonated molecular ion [M – H][–] at *m/z* 435.1320, 2 Da (2H) higher than that of N1. The typical fragment ions at *m/z* 389.1652 ([M–CO–H₂O–H][–]), 273.0560 ([M–C₆H₁₀O₅–H][–]), 298.9101 (RDA reaction) and 135.0666 (RDA reaction) were observed in its secondary mass spectrum, and the possible structure of metabolite N5 could be inferred by combining the evaluation with the fragment ion at *m/z* 153.0836, which was gained from the fragment ion at *m/z* 298.9101 by loss of C₆H₁₀O₄.

Metabolite N6 (C₂₃H₂₄O₁₂) eluted at 15.42 min. The MS/MS spectrum of N6 showed the deprotonated molecular ion [M – H][–] at *m/z* 491.1161, which was 104 Da lower than that of eriocitrin. Based on the dominant fragment ions at *m/z* 473.2817 ([M–H₂O–H][–]), *m/z* 445.1889 ([M–H₂O–CO–H][–]), *m/z* 287.1221 ([M–C₆H₄O₃–H][–]) and *m/z* 125.0191 ([M–C₁₇H₁₈O₉–H][–]), its structure was confirmed.

Metabolite N7 (C₂₁H₂₀O₁₂) eluted at 12.86 min and displayed a deprotonated molecular ion [M – H][–] at *m/z* 463.0868, 30 Da larger than that of N1. The characteristic fragment ions at *m/z* 417.1839 ([M–CO–H₂O–H][–]), *m/z* 271.0245 ([M–C₆H₈O₇–H][–]), *m/z* 258.9180 ([M–C₅H₁₀O₃–CO–H][–]) and *m/z* 151.0005 (RDA reaction) were observed in the MS/MS spectrum.

Metabolite N8 (C₂₁H₂₀O₁₁) was observed with a peak at *m/z* 447.0925 in the chromatogram, which eluted at 13.67 min and was 14 Da higher than that of N1. According to the characteristic secondary fragment ion at *m/z* 301.0338 attributed to the loss of C₆H₁₀O₄, the structure of N8 could be inferred, and other diagnostic secondary fragment ions at *m/z* 429.3176 ([M–H₂O–H][–]), *m/z* 285.0364 ([M–C₆H₁₀O₄–H][–]) and *m/z* 151.0020 (RDA reaction) were observed.

Metabolite N9 (C₁₅H₁₀O₆), with an [M – H][–] ion of *m/z* 285.0406, eluted at 15.85 min and exhibited a mass 310 Da lower than that of the parent drug. The typical fragment ions of N9 at *m/z* 267.1811 ([M–H₂O–H][–]), *m/z* 239.0672 ([M–H₂O–CO–

H][–]), *m/z* 223.0274 ([M–H₂O–CO–O–H][–]) and *m/z* 257.0448 ([M–CO–H][–]) were detected in the MS/MS spectrum, and the fragment ion at *m/z* 119.0614 was formed by the loss of O after RDA cleavage.

Metabolite N10 (C₂₇H₃₂O₆) displayed the deprotonated molecular ion [M – H][–] at *m/z* 611.1615 with a retention time of 12.39 min and a mass 16 Da (O) higher than that of eriocitrin. N10 created characteristic product ions at *m/z* 303.0505 ([M–C₁₂H₂₀O₉–H][–]), *m/z* 287.0565 ([M–C₁₂H₂₀O₁₀–H][–]), *m/z* 227.9309 ([M–C₁₂H₂₀O₉–3O–CO–H][–]), *m/z* 151.0036 (RDA reaction) and *m/z* 135.0468 (RDA reaction), indicating that the oxidation reaction occurred in the B ring.

Metabolite N11 (C₂₂H₂₄O₁₀) eluted at 23.89 min and had the deprotonated molecular ion at *m/z* 447.1328 [M – H][–], which was 14 Da (CH₂) higher than that of N1. Characteristic product ions at *m/z* 417.1216, 148.9971 and 135.0752 were in line with M20. Additionally, the distinctive fragment ion at 285.1388 ([M–C₆H₁₀O₅–H][–]) was yielded, so the possible structures of N11 could be determined.

Metabolite N12 (C₁₆H₁₄O₇) had a deprotonated molecular ion [M – H][–] at *m/z* 317.0694, a retention time of 16.61 min and a mass that was 30 Da (CH₂O) higher than that of N2. The dominant product ions at *m/z* 299.0555, 180.9719 and 165.0828 were consistent with those seen for M26; thus, its possible structures could be inferred.

3.4.2. Identification of metabolites in rat intestinal flora.

Metabolite Q1 (C₂₁H₂₂O₁₀) eluted at 15.96 min, possessing the deprotonated molecular ion [M – H][–] at *m/z* 433.1166, which was 162 Da (C₆H₁₀O₅) lower than that of eriocitrin. The typical fragment ions at *m/z* 417.2409, 387.2026, 271.0645 and 151.3214 were created through the loss of O, CO and H₂O, C₆H₁₀O₅ and RDA cleavage, respectively.

Metabolite Q2 (C₁₅H₁₂O₆) was observed with a precursor ion at *m/z* 287.0561 which eluted at 15.74 min and was 308 Da (C₁₂H₂₀O₉) lower than that of eriocitrin. It was recognized through the conspicuous fragment ions at *m/z* 269.0485, 241.2146, 151.0038 and 135.0451 that were consistent with M1 and N2.

Metabolite Q3 (C₁₅H₁₂O₅) eluted at 17.02 min and had the deprotonated molecular ion [M – H][–] at *m/z* 271.0611, 324 Da (C₁₂H₂₀O₁₀) lower than that of Q0. The distinctive fragment ions at *m/z* 253.1802, 243.0655, 227.0713 and 135.0105 yielded by Q3 were in accordance with M2, so it was verified.

Metabolites Q4 and Q5 (C₁₅H₁₂O₄) were isomeric metabolites with the deprotonated [M – H][–] ions at *m/z* 255.0675 and 255.0659, respectively, masses that were 16 Da (O) lower than that of Q3. Q4 and Q5 eluted at 15.52 min and 18.49 min, respectively. The crucial fragment ions of Q4 at *m/z* 237.2196 and 109.0724 were similar to those of M3, which confirmed that the reaction occurred in ring A. In addition, Q4 and Q5 were also verified based on their clog *P* values of 2.2109 and 2.99394, respectively.

Metabolite Q6 (C₁₄H₁₂O₅) exhibited the [M – H][–] ion of *m/z* 259.0608, which eluted at 11.79 min and was 28 Da (CO) lower than that of Q2. The dominant product ions at *m/z* 241.0505, 213.0568, 195.1018, 183.0887 and 135.0011 were consistent with M5; thus, the structure of Q6 could be inferred.



Metabolite Q7 ($C_{15}H_{10}O_5$) eluted at 16.98 min and displayed a deprotonated molecular ion $[M - H]^-$ at m/z 269.0456, 18 Da (H_2O) lower than that of Q2. Q7 generated fragment ions at m/z 241.0545, 225.0605, 197.0598, 181.0638 and 151.0020 by successively losing CO, O, CO, and O and undergoing RDA cleavage.

Metabolite Q8 ($C_{15}H_{10}O_4$) eluted at 18.33 min and exhibited the deprotonated molecular ion $[M - H]^-$ at m/z 253.0505, which was 18 Da (H_2O) lower than that of Q3. It was confirmed

by the characteristic fragment ions at m/z 225.0217, 209.0610, 181.0660 and 119.0488, which were in line with M6.

Metabolite Q9 ($C_{21}H_{24}O_{10}$) exhibited the deprotonated molecular ion $[M - H]^-$ of m/z 435.1323, was observed at 15.32 min and was 2 Da ($2H$) higher than that of Q1. The possible structure of Q9 could be inferred by the characteristic fragment ions at m/z 391.6900 ($[M - CO_2 - H]^-$), m/z 389.2463 ($[M - CO - H_2O - H]^-$), m/z 273.0863 ($[M - C_6H_{10}O_5 - H]^-$) and m/z 135.0793 (RDA reaction).

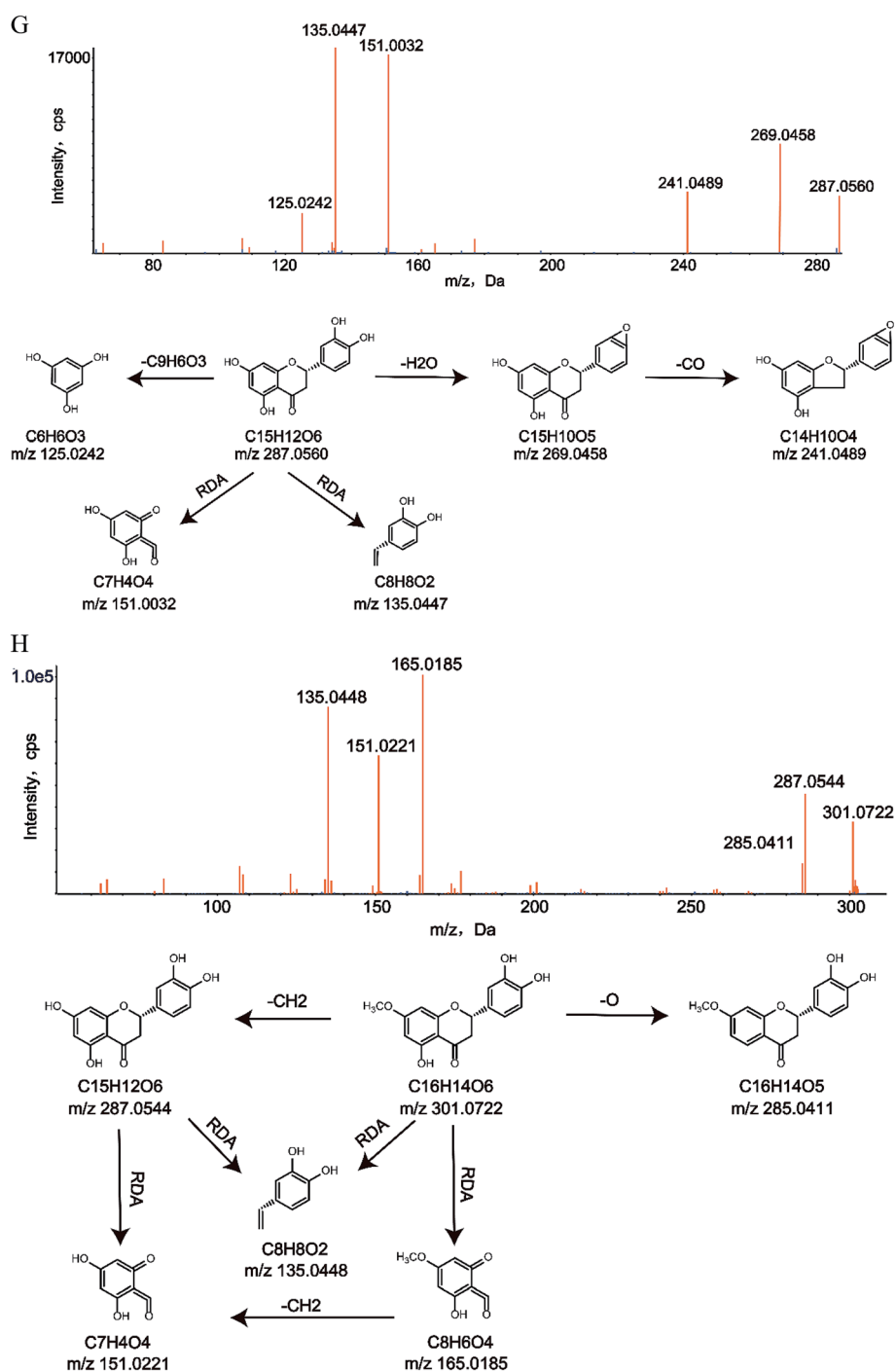


Fig. 4 MS/MS spectra and major proposed fragmentation patterns of M1 (G), M22 (H) and M28 (I).



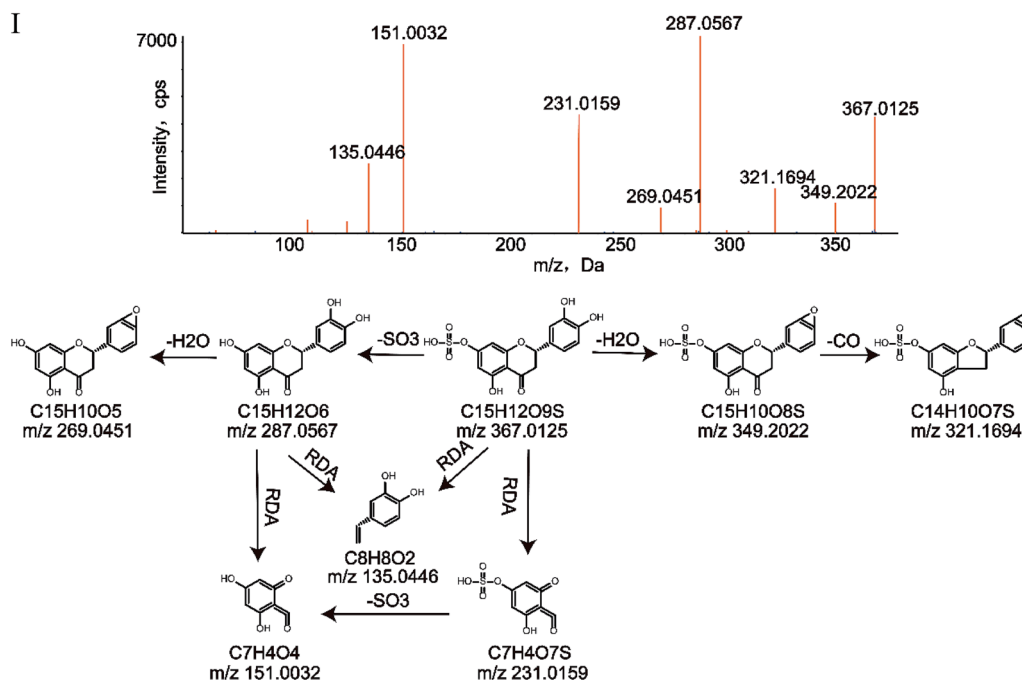


Fig. 4 (contd.)

Metabolite Q10 ($C_{15}H_{14}O_5$) was detected at 16.33 min with the deprotonated molecular ion $[M - H]^-$ at m/z 273.0766, 2 Da (2H) higher than Q3, which created fragment ions at m/z 245.0473, 227.1979, 209.1302, 165.0140 and 135.0435 by dropping CO, CH_2O_2 and O, 4O, $C_6H_4O_2$ and undergoing RDA cleavage, respectively.

Metabolites Q11 and Q12 ($C_{15}H_{14}O_4$) arose as deprotonated molecules $[M - H]^-$ at m/z 257.0818 and 257.0820, with retention times of 14.90 min and 18.61 min, respectively. Their masses were 2 Da (2H) higher than those of Q4 and Q5. According to the product ions seen at m/z 241.0505 ($[M - O - H]^-$), 239.0718 ($[M - H_2O - H]^-$), 213.0933 ($[M - CO_2 - H]^-$), 137.0228 (RDA reaction), 135.0793 (RDA reaction), 121.0304 (RDA reaction) and 119.0492 (RDA reaction) in the secondary mass spectra, the structures of metabolites Q11 and Q12 could be predicted. Additionally, Q11 and Q12 were also verified based on their clog P values of 1.462 and 1.5722, respectively.

Metabolite Q13 ($C_{15}H_{10}O_6$) eluted at 15.86 min and showed the deprotonated molecular ion $[M - H]^-$ at m/z 285.0403, which was 14 Da higher than that of Q3. It contained the product ions at m/z 267.0287, 257.0427, 223.0309 and 149.0227, which were consistent with M10. In addition, the product ion at m/z 133.0291 was attributed to loss of O after the RDA cleavage.

Metabolite Q14 ($C_{17}H_{14}O_5$) was obtained with a precursor ion at m/z 297.0762, which eluted at 14.81 min and was 42 Da larger than that of Q5, indicating that it may undergo the loss of O and $C_{12}H_{20}O_{10}$ followed by acetylation. Q14 generated fragment ions at m/z 281.0060, 264.9847 and 253.0871 through loss of O, 2O and CO_2 , respectively. It could be determined from the key fragment ions at m/z 160.9785 and 135.0547 produced by the RDA reaction that both oxygen loss and acetylation occurred in the B ring.

Metabolites Q15 and Q16 ($C_{19}H_{16}O_6$) were observed in the mass spectrum at m/z 339.0872, with retention times of 11.52 min and 11.70 min, respectively. The masses which were 84 Da higher than those of Q4 and Q5. Secondary fragment ions at m/z 311.0629, 297.1756 and 254.9424 were generated through the loss of CO, C_2H_2O and $2C_2H_2O$, respectively. Moreover, the fragment ions at m/z 219.0413, 119.0621 and 177.8865, 161.0382 produced after RDA reactions of Q15 and Q16, respectively, were similar to M13 and M14, so the structures of metabolites Q15 and Q16 could be inferred. In addition, the clog P values of Q15 and Q16 were 1.9929 and 2.28159, respectively, which also supports the confirmation of the structures.

Metabolite Q17 ($C_{15}H_{12}O_7$) eluted at 15.49 min, which presented a deprotonated molecular ion $[M - H]^-$ at m/z 303.0521, 16 Da (O) higher than that of Q2. The secondary fragment ions formed at m/z 285.1301, 257.2180, 151.0760 and 135.0594 through the loss of H_2O , CO and H_2O and undergoing RDA cleavage, respectively, were consistent with M15, so the structure of Q17 could be ascertained.

Metabolite Q18 ($C_{16}H_{14}O_5$) was detected at 14.09 min and displayed the deprotonated molecular ion $[M - H]^-$ at m/z 285.0764, 14 Da (CH_2) higher than that of Q3. Based on the fragment ions at m/z 149.0227 and 135.0007 that were produced by the RDA reaction, methylation occurred in ring A. Other key fragment ions at m/z 269.0475 ($[M - O - H]^-$), m/z 267.2314 ($[M - H_2O - H]^-$) and m/z 239.0588 ($[M - H_2O - CO - H]^-$) were also observed to help confirm the structure of Q18.

Metabolite Q19 ($C_{16}H_{14}O_4$) eluted at 16.31 min with the deprotonated molecular ion $[M - H]^-$ at m/z 269.0822, which was 14 Da (CH_2) higher than that of Q4. The key fragment ions at m/z 132.9781 and 135.0695 produced by RDA reaction were in accordance with M25, so the structure of Q19 could be confirmed.



Metabolite Q20 ($C_{16}H_{14}O_7$) displayed a peak at 16.70 min as well as a deprotonated molecular ion $[M - H]^-$ at m/z 317.0671, 30 Da (CH_2O) higher than that of Q2. It was noted that the characteristic ions at m/z 302.0746, 299.0547, 271.0600, 180.9713, 165.0804, 151.0016 and 135.0452 were similar to those of M26, so the possible structures of Q20 could be inferred.

In this study, a total of 41 metabolites were identified: 32 metabolites were detected *in vivo*, including 6 metabolites in plasma, 14 metabolites in bile, 19 metabolites in urine and 13 metabolites in faeces. Meanwhile, 27 metabolites were observed *in vitro*, including 12 metabolites in liver microsomes and 20 metabolites in intestinal flora. Representative MS/MS spectra are shown in Fig. 4, and the proposed metabolic pathways of eriocitrin *in vivo*, in rat liver microsomes and in rat intestinal flora are shown in Fig. 5. It is worth mentioning that the loss of $C_6H_{10}O_5$, $C_{12}H_{20}O_9$,

and $C_{12}H_{20}O_{10}$ was the primary metabolic step that produced further reactions such as the loss of CO, the loss of water, hydrogenation, *N*-acetylation, ketone formation, oxidation and methylation. Nevertheless, no sulfate conjugation, glutamine conjugation and glycine conjugation occurred *in vitro*, and desaturation and demethylation to carboxylic acid metabolites were not discovered *in vivo* but were found in rat liver microsomes.

Although glutamine conjugation and glycine conjugation did not seem to be very common in metabolism, they can occur.^{31–33} In this study, due to the loss of a powerful sugar group at C-7 in M31 and the 3'-hydroxyl in M32, the steric hindrance would decrease, and glutamine conjugation and glycine conjugation could occur at the 5 and 4' sites, respectively.

Protein precipitation with methanol and liquid-liquid extraction with ethyl acetate were used in this research to

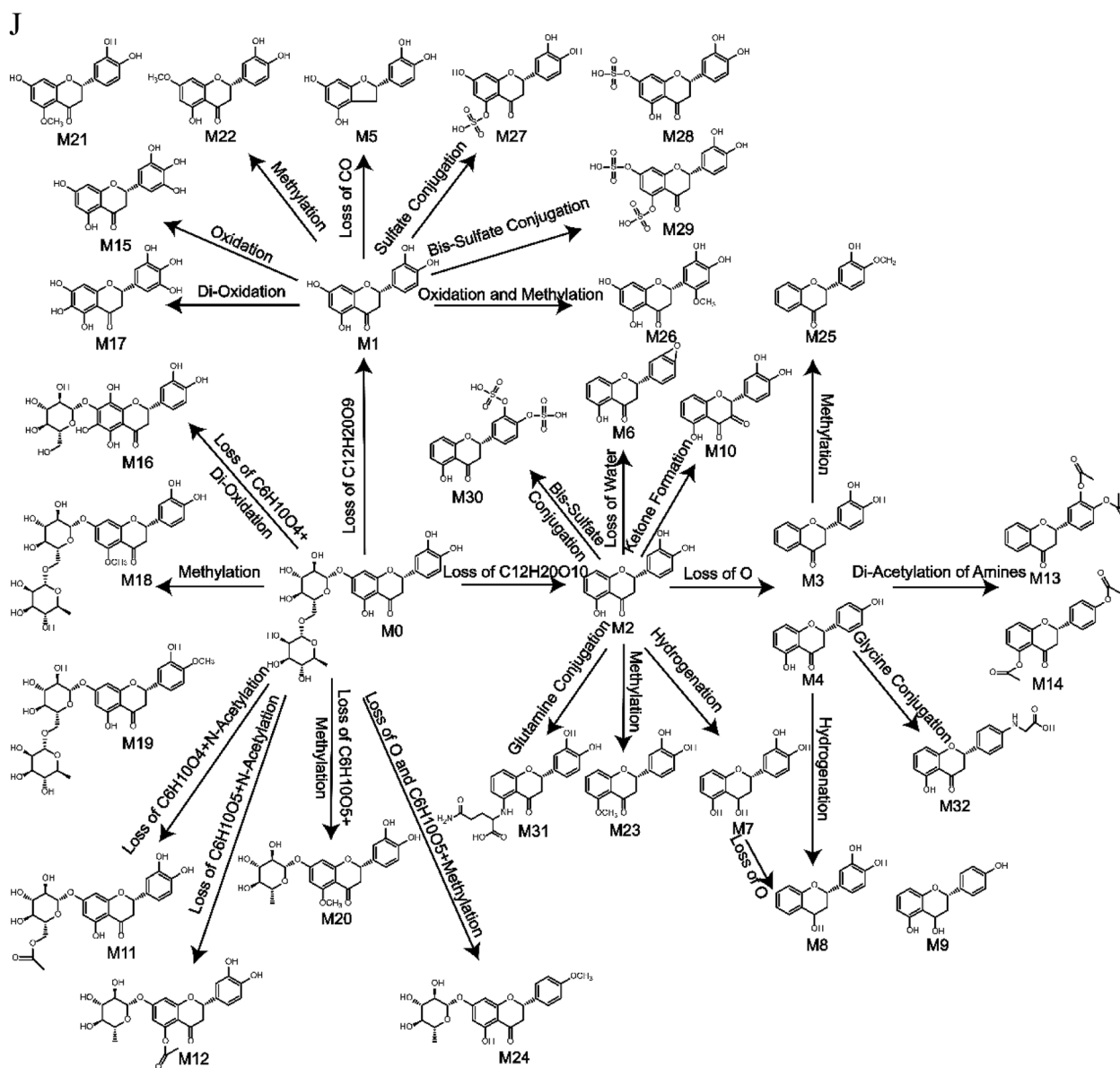
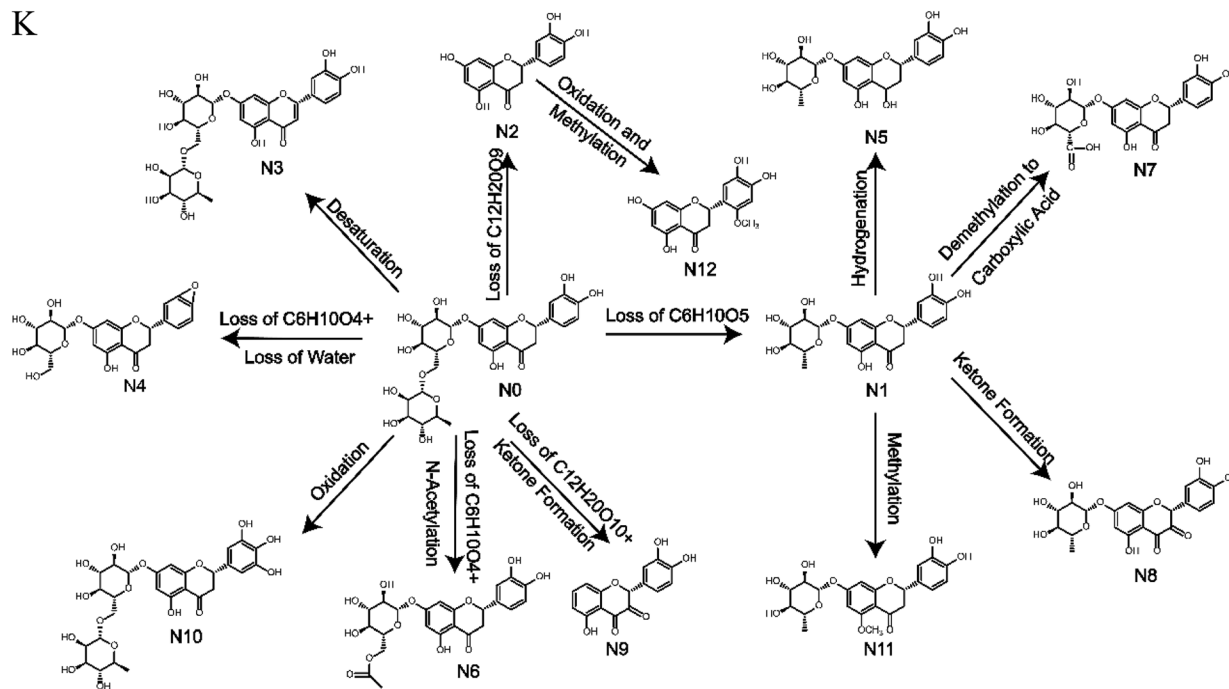


Fig. 5 Metabolic profile and proposed metabolic pathways of eriocitrin *in vivo* and *in vitro* ((J) *in vivo*, (K) in rat liver microsomes, (L) in rat intestinal flora).



K



L

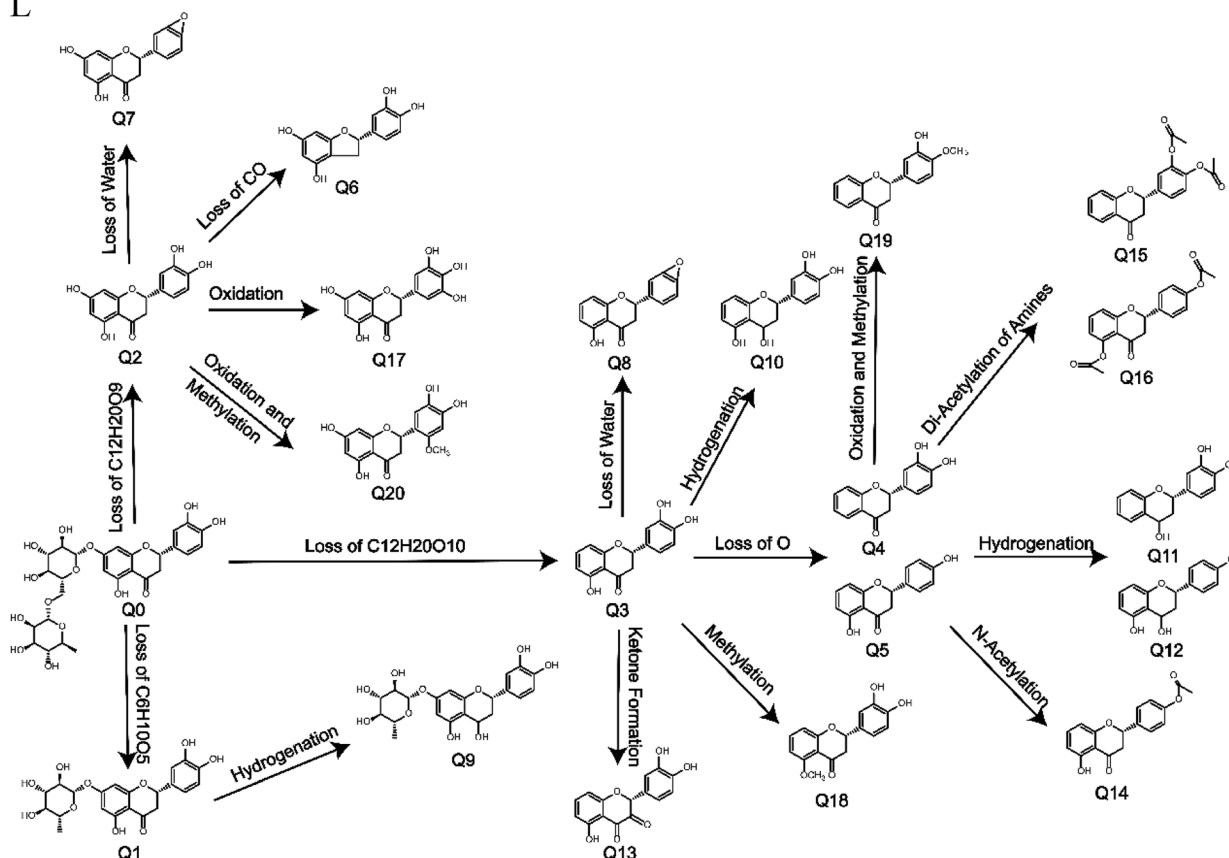


Fig. 5 (contd.)

acquire more types and quantities of metabolites. The results showed that 13 metabolites were extracted by methanol and ethyl acetate simultaneously, additionally, 14 metabolites can only be extracted from methanol and 5 metabolites can just be

extracted from ethyl acetate, which did increase the types and the number of metabolites of eriocitrin. Meanwhile, we can see that more metabolites were extracted from methanol than from



ethyl acetate, which may be related to their ability to form hydrogen bonds and polarity.³⁴

It has been reported in the literature that eriocitrin has strong antioxidant activity.⁴ In this paper, oxidation occurred both *in vivo* and *in vitro* and was found to be a vital metabolic reaction of eriocitrin, which may be related to its strong antioxidant activity. In addition, many of the metabolites of eriocitrin have been studied. For example, M1 (N2, Q2), namely, eriodictyol, a natural flavonoid compound present in citrus fruits, has been reported to have broad bioactivities such as antioxidant, anti-inflammatory, immunomodulatory and anti-diabetic activities.^{35–37} It is worth mentioning that eriodictyol was found to be one of the most potent insulin secretagogues among hundreds of compounds tested.³⁸ M19b, namely, hesperidin, a type of citrus bioflavonoid distributed in foods including grapefruits, oranges and lemons, has many pharmacological activities, such as antioxidant, anti-depression and antitumour activities.^{39–41} Overall, the identification of metabolites of eriocitrin provides a basis for new pharmacological studies, and these metabolites will be further explored in the future.

4. Conclusions

In conclusion, an efficient strategy for screening and identifying the metabolites of eriocitrin *in vivo* and *in vitro* was established first by UHPLC-Q-TOF-MS/MS using online data acquisition and multiple data processing techniques. The results showed that a total of 41 metabolites were identified: 32 metabolites were detected *in vivo* (6 metabolites in the plasma, 14 metabolites in the bile, 19 metabolites in the urine and 13 metabolites in the faeces), and 27 metabolites were detected *in vitro* (12 metabolites in the rat liver microsomes and 20 metabolites in rat intestinal flora) under the experimental conditions. In addition to identifying the above metabolites, we also elucidated the metabolic pathways of eriocitrin. Moreover, the incubation of liver microsomes and intestinal flora was applied to eriocitrin for the first time. It was also the first study to investigate the metabolic mechanisms of eriocitrin *in vivo* and *in vitro*, all of which provided reference and valuable evidence for further development of new pharmaceuticals and pharmacological mechanisms, laying a foundation for clinical examination and application.

Conflicts of interest

All the authors have declared no conflict of interest.

Acknowledgements

The project was financially supported by the National Natural Science Foundation of China (No. 81473180). Thanks to the Department of Pharmaceutical Analysis, School of Pharmacy, Hebei Medical University for the instrument support.

References

- M. Makni, R. Jemai, W. Kriaa, Y. Chtourou and H. Fetoui, *BioMed Res. Int.*, 2018, **2018**, 6251546.
- X. Dong, Y. Hu, Y. Li and Z. Zhou, *Sci. Hortic.*, 2019, **243**, 281–289.
- J. J. Peterson, G. R. Beecher, S. A. Bhagwat, J. T. Dwyer, S. E. Gebhardt, D. B. Haytowitz and J. M. Holden, *J. Food Compos. Anal.*, 2006, **19**, S74–S80.
- M. Hajimahmoodi, G. Moghaddam, S. M. Mousavi, N. Sadeghi, M. R. Oveisi and B. Jannat, *Trop. J. Pharm. Res.*, 2014, **13**, 951–956.
- P. S. Ferreira, L. C. Spolidorio, J. A. Manthey and T. B. Cesar, *Food Funct.*, 2016, **7**, 2675–2681.
- K.-i. Minato, Y. Miyake, S. Fukumoto, K. Yamamoto, Y. Kato, Y. Shimomura and T. Osawa, *Life Sci.*, 2003, **72**, 1609–1616.
- M. Hiramitsu, Y. Shimada, J. Kuroyanagi, T. Inoue, T. Katagiri, L. Zang, Y. Nishimura, N. Nishimura and T. Tanaka, *Sci. Rep.*, 2014, **4**, 3708.
- Y. Miyake, E. Suzuki, S. Ohya, S. Fukumoto, M. Hiramitsu, K. Sakaida, T. Osawa and Y. Furuichi, *J. Food Sci.*, 2006, **71**, S633–S637.
- Z. Wang, H. Zhang, J. Zhou, X. Zhang, L. Chen, K. Chen and Z. Huang, *Cancer Chemother. Pharmacol.*, 2016, **78**, 1143–1150.
- Y. Miyake, K. Shimoi, S. Kumazawa, K. Yamamoto, N. Kinai and T. Osawa, *J. Agric. Food Chem.*, 2000, **48**, 3217–3224.
- M. W. Attwa, A. A. Kadi, H. W. Darwish and H. Alrabiah, *Clin. Chim. Acta*, 2018, **482**, 84–94.
- A. A. Kadi, S. M. Amer, H. W. Darwish and M. W. Attwa, *RSC Adv.*, 2017, **7**, 36279–36287.
- M. W. Attwa, A. A. Kadi, H. Alrabiah and H. W. Darwish, *J. Pharm. Biomed. Anal.*, 2018, **160**, 19–30.
- O. A. Almazroo, M. K. Miah and R. Venkataramanan, *Clin. Liver Dis.*, 2017, **21**, 1–20.
- H. Li, J. He and W. Jia, *Expert Opin. Drug Metab. Toxicol.*, 2016, **12**, 31–40.
- K. Noh, Y. R. Kang, M. R. Nepal, R. Shakya, M. J. Kang, W. Kang, S. Lee, H. G. Jeong and T. C. Jeong, *Arch. Pharm. Sci. Res.*, 2017, **40**, 1345–1355.
- J. Y. Zhang, Z. J. Wang, Y. Li, Y. Liu, W. Cai, C. Li, J. Q. Lu and Y. J. Qiao, *Talanta*, 2016, **147**, 16–27.
- Z. Yisimayili, X. Guo, H. Liu, Z. Xu, R. Abdulla, H. Akber Aisa and C. Huang, *J. Pharm. Biomed. Anal.*, 2019, **165**, 251–260.
- M. Liao, X. Diao, X. Cheng, Y. Sun and L. Zhang, *RSC Adv.*, 2018, **8**, 14925–14935.
- C. Liang, X. Zhang, X. Diao, M. Liao, Y. Sun and L. Zhang, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2018, **1084**, 69–79.
- X. Diao, M. Liao, X. Cheng, C. Liang, Y. Sun, X. Zhang and L. Zhang, *Biomed. Chromatogr.*, 2018, e4263.
- T. Tian, Y. Jin, Y. Ma, W. Xie, H. Xu, K. Zhang, L. Zhang and Y. Du, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2015, **1006**, 80–92.
- L. Hou, Y. Jin, W. Sun, S. Guan, H. Xu, Q. Wang, L. Zhang and Y. Du, *Fitoterapia*, 2019, **133**, 85–95.



- 24 X. Zhang, C. Liang, J. Yin, Y. Sun and L. Zhang, *RSC Adv.*, 2018, **8**, 11813–11827.
- 25 P. Jia, Y. Zhang, Q. Zhang, Y. Sun, H. Yang, H. Shi, X. Zhang and L. Zhang, *Biomed. Chromatogr.*, 2016, **30**, 1498–1505.
- 26 L. Yuan, C. Liang, X. Diao, X. Cheng, M. Liao and L. Zhang, *Xenobiotica*, 2018, **48**, 332–341.
- 27 W. Xie, Y. Jin, L. Hou, Y. Ma, H. Xu, K. Zhang, L. Zhang and Y. Du, *J. Pharm. Biomed. Anal.*, 2017, **145**, 865–878.
- 28 Y. Ma, W. Xie, T. Tian, Y. Jin, H. Xu, K. Zhang and Y. Du, *Anal. Biochem.*, 2016, **511**, 61–73.
- 29 X. Zhang, J. Yin, C. Liang, Y. Sun and L. Zhang, *J. Agric. Food Chem.*, 2017, **65**, 10959–10972.
- 30 D. Yao, Y. Wang, C. Huo, Y. Wu, M. Zhang, L. Li, Q. Shi, H. Kiyota and X. Shi, *Food Chem.*, 2017, **214**, 328–338.
- 31 Y. Chen, X. Feng, L. Li, X. Zhang, K. Song, X. Diao, Y. Sun and L. Zhang, *J. Pharm. Biomed. Anal.*, 2019, **169**, 19–29.
- 32 X. Zhang, J. Yin, C. Liang, Y. Sun and L. Zhang, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2017, **1061–1062**, 193–208.
- 33 M. Liao, X. Cheng, X. Diao, Y. Sun and L. Zhang, *J. Chromatogr. B: Anal. Technol. Biomed. Life Sci.*, 2017, **1068–1069**, 297–312.
- 34 N. Zhang, Z. Shen, C. Chen, G. He and C. Hao, *J. Mol. Liq.*, 2015, **203**, 90–97.
- 35 Y. Wang, Y. Chen, Y. Chen, B. Zhou, X. Shan and G. Yang, *Biomed. Pharmacother.*, 2018, **107**, 1128–1134.
- 36 P. Lv, J. Yu, X. Xu, T. Lu and F. Xu, *J. Cell. Biochem.*, 2019, **120**, 5644–5651.
- 37 I. Mokdad-Bzeouich, N. Mustapha, A. Sassi, A. Bedoui, M. Ghoul, K. Ghedira and L. Chekir-Ghedira, *Cell Stress Chaperones*, 2016, **21**, 773–781.
- 38 A. Hameed, R. M. Hafizur, N. Hussain, S. A. Raza, M. Rehman, S. Ashraf, Z. Ul-Haq, F. Khan, G. Abbas and M. I. Choudhary, *Eur. J. Pharmacol.*, 2018, **820**, 245–255.
- 39 S. Estruel-Amades, M. Massot-Cladera, P. Garcia-Cerda, F. J. Perez-Cano, A. Franch, M. Castell and M. Camps-Bossacoma, *Nutrients*, 2019, **11**, 783.
- 40 H. Fu, L. Liu, Y. Tong, Y. Li, X. Zhang, X. Gao, J. Yong, J. Zhao, D. Xiao, K. Wen and H. Wang, *Eur. J. Pharmacol.*, 2019, **853**, 236–246.
- 41 J. M. P. Ferreira de Oliveira, C. Santos and E. Fernandes, *Phytomedicine*, 2019, 152887.

