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Simple, rapid, sensitive, selective and label-free lincomycin detection by using HAuCl_4 and NaOH †

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A simple, rapid, sensitive, selective and label-free method is presented for the colorimetric determination of lincomycin (Lin) by using HAuCl_4 and NaOH . Upon the addition of Lin, the mixture of HAuCl_4 and NaOH shows a color change from colorless to blue (or dark blue). The limit of colorimetric detection is as low as $1 \mu\text{M}$, observed both in Milli-Q water and real samples. The selectivity of Lin detection is excellent compared with 9 other common antibiotics. On the basis of the "three-color" principle of Thomas Young, we extracted the red, green and blue (RGB) alterations of the sensor in the absence and presence of different concentrations of Lin. The color changes are quantitatively illustrated by the total Euclidean distances ($\text{EDs} = [\Delta\text{R}^2 + \Delta\text{G}^2 + \Delta\text{B}^2]^{1/2}$). The good linear relationship between the EDs and Lin concentration is used for the quantitative assay of Lin. The developed method demonstrates great potential for the detection of Lin in environmental water and milk.

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1. Introduction

The pattern for raising herds of food animals appeared in the 1960s, with routine mass use of veterinary drugs (e.g. antibiotics) to prevent animal diseases and to improve growth.¹ The total antibiotic usage in China in 2013 was reported to be $\sim 162\,000$ tons, and this number is further increasing with the development of industry and the increase of the population.^{2,3} Lincomycin (Lin) is an antibiotic against bacteria, and is widely used in the feed and animal production industry.⁴ Some studies have suggested that a high concentration of Lin in animal-derived food can cause health problems.⁵ Hence, it is crucial to detect and monitor Lin residues in food products as well as in the environment.

Various methods have been described in the literature for the analysis of Lin, such as microbiological assay,⁶ high-performance liquid chromatography (HPLC),⁷ liquid chromatography-mass spectrometry (LC-MS),⁸ thin layer chromatography (TLC),⁹ electrochemiluminescence (ECL),¹⁰ electrochemical method,^{11,12} capillary electrophoresis (CE),¹³ atomic absorption spectrometry¹⁴ and spectrophotometry (SP).¹⁵ The microbiological assays were originally used, but they need long-term culture and are non-specific. HPLC is efficient and fast, but

its sensitivity is limited by the UV absorption of Lin. LC-MS methods are quite sensitive, but the expensive equipment needs to be handled by trained staff, limiting the application for small farms or in-field tests. TLC methods have strong separation capability and only need small amount of sample, but suffer from poor reproducibility. CE and SP methods are not sensitive enough for the detection of trace Lin. Hence, it is of significance to develop a simple, sensitive, selective and cost-effective method for detecting Lin.

In the present, we described a method that we have developed for the colorimetric determination of Lin by using HAuCl_4 and NaOH . The method is based on the different coordination interaction of Au and ligands. With the increase of Lin concentrations, the color of HAuCl_4 and NaOH mixture gradually changes from colorless to wathet blue, then blue and finally dark blue. The limit of detection (LOD) by the naked eyes is as low as $1 \mu\text{M}$ observed in Milli-Q water and real sample. The sensor shows low response to 9 other common antibiotics comparing with Lin. The applicability of our detection system is also verified by analysis of Lin in real sample. The method is characterized by rather simple, high speed, reproducibility, selective and sensitivity.

2. Experimental section

2.1. Materials and instruments

Chloroauric acid tetrahydrate ($\text{HAuCl}_4 \cdot 4 \text{H}_2\text{O}$) and NaOH were obtained from Sinopharm Chemical Reagent Co., Ltd (Beijing, China). Amoxicillin (Am), erythromycin (Er), spiramycin (Sp), gentamycin sulfate (Ge), chlortetracycline hydrochloride (Ch), ceftriaxone sodium (Ce), enrofloxacin (En), ciprofloxacin (Ci),

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sulfadiazine (Su) and lincomycin hydrochloride (Lin) were purchased from Aladdin-Reagent Co. Ltd (Shanghai, China). 96-well plates (Corning 3632) were obtained from Genetimes Technology. All chemicals were used as received without further purification. Millipore-Q water was used throughout the experiment.

To eliminate the influence of environmental light, the color images were taken with a scanner in 96-well polystyrene plates.

2.2. Preparation of sensor

The HAuCl_4 (1 mL, 1 mM) and NaOH (1 mL, 1 mM) were mixed to produce the colorless solution. The colorless solution (sensor) was harvested and stored at 4 °C for further use.

2.3. Colorimetric detection of Lin

2.3.1. Sensitivity. To evaluate the detectable minimum concentration of Lin, the typical colorimetric analysis was carried out by mixing the sensing solution and different concentrations of Lin. With the increase of Lin concentrations, the color of sensing solution gradually changes from colorless to wathet blue, then blue and finally dark blue. Color change was rapid and observed with the naked eye.

The sensing solution in the absence and presence of different concentrations of Lin were loaded into a 96-well polystyrene plate, and the images were recorded using a scanner. Their color alterations (ΔRGB) were obtained by using the Photoshop software. Four replicates were tested for the different concentrations of Lin. The color changes were quantitatively illustrated by the total Euclidean distances (EDs = $[\Delta R^2 + \Delta G^2 + \Delta B^2]^{1/2}$).

2.3.2. Selectivity. The selectivity for Lin was investigated as follows: 10^{-2} M of Lin (2 μL) and other common antibiotics (Ci, En, Su, Er, Ce, Ch, Sp, Am and Ge) were added into sensor solution (1 mL), respectively. The final concentrations of antibiotics are 20 μM .

The sensor solution in the absence and presence of antibiotics (20 μM) were loaded into a 96-well polystyrene plate, and the images were recorded by using a scanner. Their color alterations (ΔRGB) were obtained by using the Photoshop software. Four replicates were tested for the different concentrations of Lin. The color changes were quantitatively illustrated by the total Euclidean distances (EDs = $[\Delta R^2 + \Delta G^2 + \Delta B^2]^{1/2}$).

2.3.3. Practical application. In order to assess the validity of the developed procedure, the method was applied to detect Lin in real sample such as environmental water and milk. 20 μL of Lin spiked in environmental water was added to 2 mL of the sensor solution. The mixture was loaded into a 96-well polystyrene plate, and the image was recorded by using a scanner. The color change was quantitatively illustrated by EDs. According to the linear relationship between the EDs and concentrations of Lin, we obtained the content of Lin contained in environmental water. The result was compared with a laboratory grade standard of HPLC-mass spectrometry (HPLC-MS).

Due to the complex elements contained in milk, it was usually selected as the test real sample for antibiotics detection.^{16,17} The selected milk is made up of carbohydrates, proteins, fats, citric acid, Na^+ , *etc.* According to the reported method,¹⁶ direct 30-fold dilution was performed. The spiked samples were obtained by adding Lin with different concentration to the diluted milk. The detection procedure of Lin in the real sample is similar with that in pure water.

3. Results and discussion

3.1. Colorimetric detection of Lin

3.1.1. Sensitivity. The HAuCl_4 and NaOH were mixed to produce the colorless solution (as sensor). In the presence of Lin and upon addition of increasing concentrations of Lin, the color of sensing solution gradually changes from initially colorless to wathet blue, then blue and finally dark blue. The limit of colorimetric detection is 1 μM (see Fig. 1, panel A). For semiquantitative analysis of Lin, the total EDs were then applied to the concentration-dependent color response. The EDs *versus* different concentrations of Lin are shown in Fig. 1B. The linear relationship ($R^2 = 0.996$) can be inferred, manifesting the dynamic ranges from 1 to 25 μM and limit of detection (LOD) being 0.97 μM (see Table S1 and the calculation method in ESI†). This result demonstrates that this method possesses great potential for Lin detection. As shown in Table 1, the detection results from the proposed method were compared with the results obtained from previous work. Although the linear range of our method is not wider than the reported ranges, the detection limit is comparable to some of theirs. It is remarkable that almost all of the reported methods require using expensive instrument while our method does

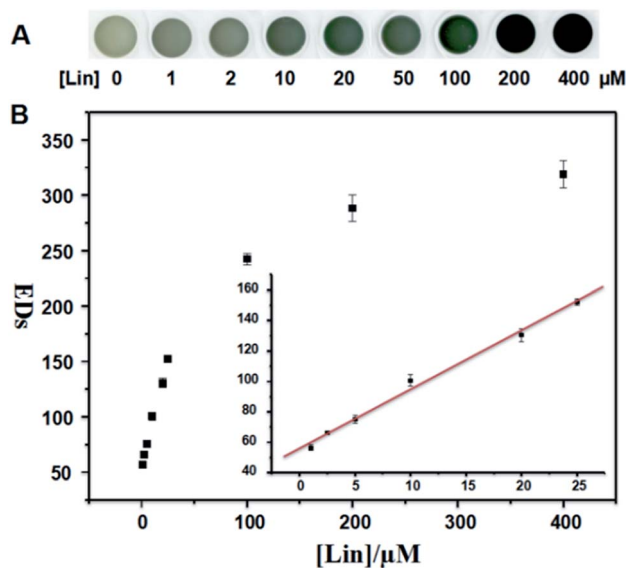


Fig. 1 (A) Color images of the sensing solution exposure to different concentrations of Lin. (B) The total EDs *versus* different concentrations of Lin. (The concentrations of Lin are 0, 1, 2, 10, 20, 50, 100, 200 and 400 μM).



Table 1 The comparison of different methods for the detection of Lin

Method	Linear range (μM)	Detection limit (μM)	Ref.
HPLC-UV	7–2260	2	18
CE	23–339	2	19
CE	38–1896	5.6	20
ECL	5–95	2.9	10
Fluorescence	2–590	0.65	21
Colorimetric	1–25	0.97	Present method

not need. Thus, it can be derived the suggested method is fast, sensitive and convenient.

3.1.2. Selectivity. In order to evaluate the selectivity of Lin detection, control experiments were conducted with common antibiotics, such as Ci, En, Su, Er, Ce, Ch, Sp, Am and Ge. The concentrations of analytes are 20 μM . The corresponding photo image and EDs data are shown in Fig. 2. Only the addition of Lin can induce a dramatic color change, whereas little changes are observed in the presence of other common antibiotics. It is clear that the sensing solution has very high specificity toward Lin (see Fig. 2).

3.1.3. Detection principle. Fig. 3 shows the probable mechanism for the detection of Lin. The light yellow HAuCl_4 (1 mL, 1 mM) was injected into NaOH (1 mL, 1 mM) to produce the colorless solution of NaAuO_2 (*i.e.*, $\text{HAuCl}_4 + 5\text{NaOH} \rightarrow \text{NaAuO}_2 + \text{NaCl} + 3\text{H}_2\text{O}$). The AuO_2^- may form a coordination complex with available donor groups (*e.g.*, sulfur) in Lin. Comparing the structures of 10 common antibiotics (see Fig. S1, ESI[†]), only Lin possesses the sulfur which can coordinate with Au in solution. Moreover, the comparative experiments were performed, as shown in Fig. S2 (ESI[†]). There was no color change of the HAuCl_4 and NaOH solution exposure to different

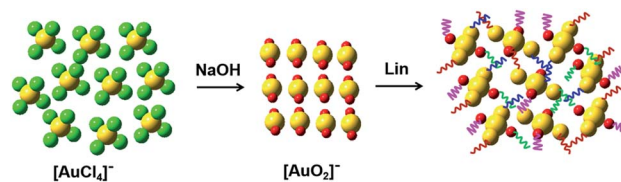


Fig. 3 Schematic representation of the sensing procedure for colorimetric detection of Lin based on the different coordination interaction of Au and ligands.

concentrations of Lin, respectively. This phenomenon indicates that there exists the interaction between Au and Lin. In the presence of Lin, Au may form coordination complexes with sulfur through the $\text{Au}\cdots\text{S}$ covalent interactions.^{22,23} As shown in Fig. 3 and S3 (ESI[†]), the chelation interaction may happen between Lin and Au in solution. This coordination reaction may change the sensing solution from colorless to blue. The calculation to prove the interaction of Lin and Au will be performed in future research.

3.1.4. Practical application. Based on the linear relationship (Fig. 1B inset) and the EDs data (136.78 ± 5.13) obtained from the real sample, we find that the concentration of Lin contained in environmental water is $20.91 \pm 1.32 \mu\text{M}$ (see Table 2). The detection result from our sensor was compared with the result obtained from HPLC-MS analysis using the same sample. The result obtained by the as-developed sensing system was in good agreement with that acquired by HPLC-MS (see Table 2), which strongly prove the high reliability of our developed sensor.

The sensing solution was applied in detection of Lin in real sample (*e.g.* milk). The spiked samples were obtained by adding concentrations of Lin to the diluted milk. The corresponding photo images are shown in Fig. 4A. The color of the sensing solution changes from colorless to wathet blue, blue and dark blue with the increase of Lin concentrations (see Fig. 4A). Although the complex elements in the spiked samples slightly affect the color changes (compared with Fig. 1, panel A), the limit of colorimetric detection is up to 1 μM . The EDs as a function of Lin concentration is shown in Fig. 4B. The linear relationships can be inferred, manifesting the dynamic ranges from 0 to 10 μM and the LOD of 0.94 μM (see Table S1 and the calculation method in ESI[†]). Therefore, the proposed sensor has a great promise to be applied as a powerful tool for the detection of Lin in environmental water and food (*e.g.* milk).

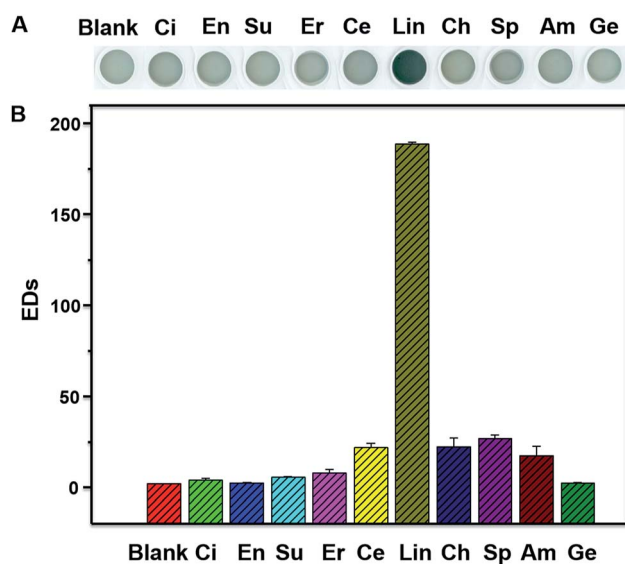


Fig. 2 (A) Color photographs and (B) EDs data of the sensing solution with representative antibiotics (Ci, En, Su, Er, Ce, Lin, Ch, Sp, Am and Ge). The concentrations of the antibiotics are 20 μM .

Table 2 Determination of Lin in environmental water by using the as-developed sensor

Sample	[Lin]/ μM		HPLC-MS ^a
	Spiked	Found ^a	
Environmental water	0	0	0
	20	20.91 ± 1.32	19.75 ± 1.18

^a Mean \pm standard deviation, $n = 3$.



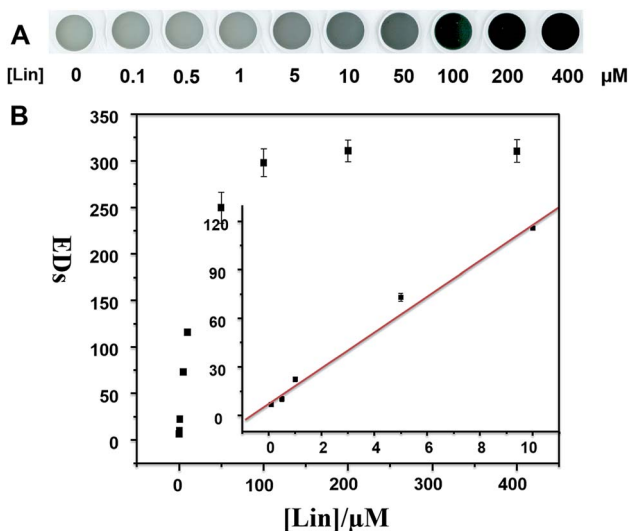


Fig. 4 (A) Color images of the sensing solution exposure to different concentrations of Lin in real sample. (B) The corresponding EDs versus different concentrations of Lin. (The concentrations of Lin are 0, 0.1, 0.5, 1, 5, 10, 50, 100, 200 and 400 μM).

4. Conclusions

In summary, we have developed a simple, rapid, sensitive, selective and label-free method for the colorimetric detection of Lin. In the presence of increasing concentrations of Lin, the color of H₂AuCl₄ and NaOH mixture gradually changes from initially colorless to wathet blue, then blue and finally dark blue. This different coordination interaction may change the color of AuCl₄⁻ solution from light yellow to colorless and blue. The selectivity of the Lin detection is excellent comparing with other 9 common antibiotics. The limits of colorimetric detection have been determined to be 1 μM in Milli-Q water and real sample. The good linear relationship between EDs and Lin concentration is used for the quantitative assay of Lin. The LODs have been calculated to be 0.97 μM (Milli-Q water) and 0.94 μM (real sample), respectively. These results indicate that the developed detection system is applicable for rapid colorimetric detection of Lin in environmental water and food (*e.g.* milk).

Conflicts of interest

There are no conflicts to declare.

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