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Enantioselective synthesis of quaternary 3,4-dihydroisoquinolinones *via* Heck carbonylation reactions: development and application to the synthesis of Minalrestat analogues†

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Minalrestat and its analogues represent structurally novel aldose reductase inhibitors, and the asymmetric synthesis of such pharmaceutically privileged molecules has not been reported yet. We have developed a palladium-catalyzed enantioselective intramolecular carbonylative Heck reaction by using formate esters as the source of CO, which represents the first enantioselective synthesis of quaternary 3,4-dihydroisoquinolines. The reaction provides a facile and efficient method for the synthesis of enantiopure nitrogen-containing heterocyclic compounds bearing an all-carbon quaternary stereocenter. The reaction has been successfully applied to the first asymmetric synthesis of Minalrestat analogues.

Introduction

Diabetes mellitus is a major health concern and affects millions of people worldwide.¹ Aldose reductase inhibitors (ARIs) are attractive therapeutic targets for designing drugs to prevent or slow the progression of diabetic complications.² Although various ARIs have been discovered, almost all of them have been withdrawn due to adverse side effects or low efficacy.³ Minalrestat is an important ARI that shows appreciable activity and safety profiles and is a promising drug candidate.⁴ Minalrestat is a 3,4-dihydroisoquinolinone derivative bearing a spiro-succinimide moiety at the 4-position (Fig. 1). Actually, the isoquinolinone backbone in Minalrestat represents a structurally novel framework for designing potent ARIs, and a range of derivatives derived from the backbone exhibit intrinsic activity and good oral potency.⁴ Notably, these bioactive isoquinolinone derivatives usually contain a quaternary carbon stereocenter, and the stereocenter plays crucial roles in the bioactivities and oral potency. Unfortunately, the stereocenters tend to racemize *via* enolization.^{4a} The asymmetric synthesis of this isoquinolinone skeleton is not available,⁵ and the enantiopure compounds were obtained by the resolution of racemates.^{4a} Therefore, it is a formidable task to develop enantioselective reactions for the construction of such six-membered isoquinolinones containing an all-carbon quaternary stereocenter,

which would allow us to not only obtain enantiopure compounds for drug discovery but also modify the structures to prevent racemization.

3,4-Dihydroisoquinolinone skeletons are widely found in natural products and are pivotal structural motifs in drug molecules,⁶ and many of the bioactive 3,4-dihydroisoquinolinone derivatives contain an all-carbon quaternary stereocenter at the 4-position⁷ (Fig. 1). As such, asymmetric reactions for the construction of such isoquinolinone skeletons would find wide applications in organic synthesis. Although

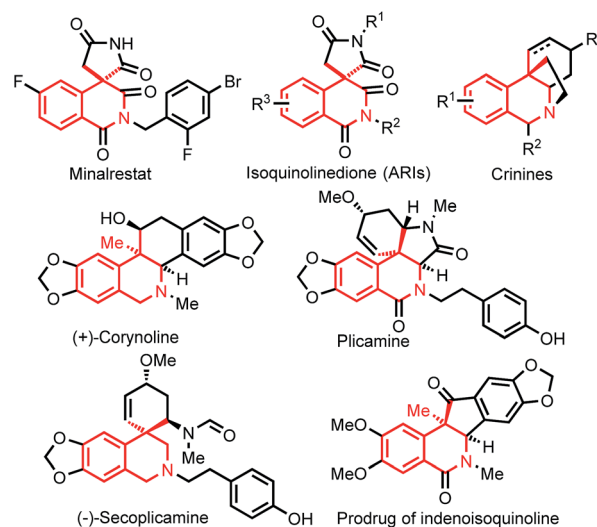


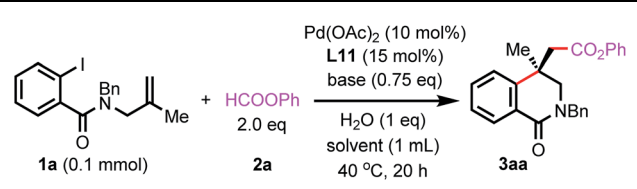
Fig. 1 Bioactive 3,4-dihydroisoquinolinone derivatives containing all-carbon quaternary stereocenters at the 4-position.

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Table 1 Optimization of the reaction conditions

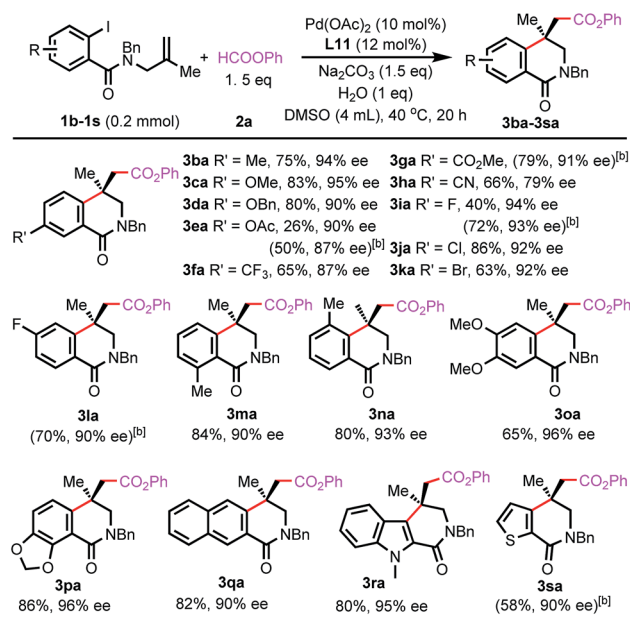


Entry	Base	Solvent	Yield (%) ^a	ee (%) ^b
1	CS ₂ CO ₃	DMSO	42 ^{c,d}	81
2	K ₂ CO ₃	DMSO	75 ^{c,d}	84
3	Na ₂ CO ₃	DMSO	78 ^{c,d}	86
4	Li ₂ CO ₃	DMSO	81 ^{c,d}	83
5	Na ₂ CO ₃	DMSO	50 ^d	93
6	Na ₂ CO ₃	DMSO	75	92
7	Na ₂ CO ₃	DMSO	75 ^e	92
8	Na ₂ CO ₃	DMSO	77 ^{e,f}	92
9	Na ₂ CO ₃	DMSO	82 ^{e,f,g}	92
10	Na ₂ CO ₃	DMSO	88 ^{e,f,g,h}	92
11	Na ₂ CO ₃	NMP	68 ^e	93
12	Na ₂ CO ₃	DMF	56 ^{e,f,g,h}	87
13	Na ₂ CO ₃	CH ₃ CN	0 ^{e,f,g,h}	—
14	Na ₂ CO ₃	THF	0 ^{e,f,g,h}	—
15	Na ₂ CO ₃	Toluene	0 ^{e,f,g,h}	—
16	Na ₂ CO ₃	DMSO	70 ^{f,g,h,i}	92
17	Na ₂ CO ₃	DMSO	40 ^{f,g,h,j}	92

^a Isolated yields. ^b Determined by HPLC analysis. ^c 110 °C. ^d Without 1 equiv. of H₂O. ^e 12 mol% of L11. ^f 1.5 equiv. of 2a. ^g 1.5 equiv. of Na₂CO₃. ^h 2 mL solvent. ⁱ 5 mol% of Pd(OAc)₂ and 6 mol% of L11. ^j 2 mol% of Pd(OAc)₂ and 2.4 mol% of L11.

ligand to 12 mol% failed to increase the yield, and the ee value remained unchanged (entry 7). The use of 1.5 equivalents of 2a gave a similar yield (entry 8). Gratefully, the yield was improved to 82% by increasing the quantity of Na₂CO₃ (entry 9) and further enhanced to 88% by running the reaction in 2 mL DMSO (entry 10). High enantioselectivity was also achieved when the reaction was carried out in NMP or DMF, albeit in a lower yield (entries 11 and 12). However, the reaction failed to form 3aa in other solvents. In these reactions, 1a just remained intact (entries 13–15). Although reducing catalyst and ligand loading led to lower yields, the ee values remained unchanged (entries 16 and 17).

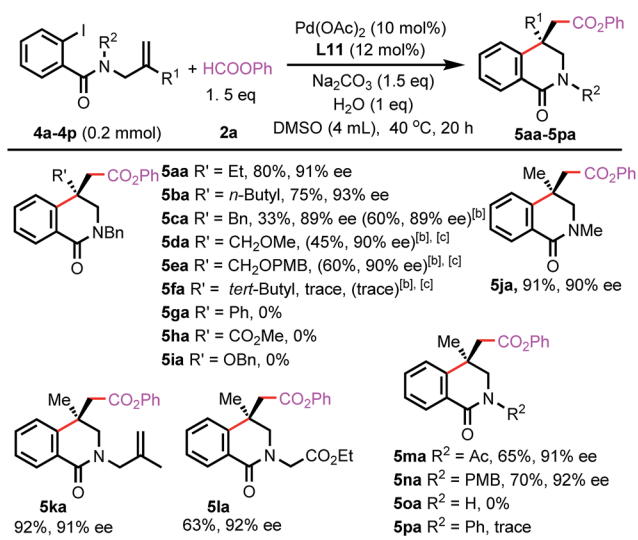
Next, we investigated the substrate scope of the enantioselective intramolecular carbonylative Heck reaction. The compatibility of various functional groups on the benzene rings of the benzamides was first examined. As shown in Scheme 2, a range of electron-donating groups were compatible, and good or excellent ee values were obtained (3ba–3ea). In the reaction of 1e, the yield was low, and the major side-product was the compound resulting from the capture of the alkylpalladium species by a hydride. The yield was improved to 50% by running the reaction at 60 °C with a slight decrease of ee value. A slightly lower ee value and yield were observed for the benzamides bearing an electron-withdrawing group (3fa–3ha). Fluoro, chloro, and even bromo groups were well-tolerated, and the corresponding isoquinolinones were formed in high ee values (3ia–3ka). A fluoro group *para* to the amide group was



Scheme 2 Scope of benzamides with respect to substituents on the arene rings.^[a] Isolated yields.^[b] 60 °C.

compatible, and the corresponding product was obtained in 70% yield and 90% ee (3la). The substrates bearing a substituent at the *ortho*-position (3ma and 3na) or two substituents (3pa–3qa) were also suitable. Notably, heteroarene-derived amides also underwent the asymmetric reaction with high enantioselectivities (3ra and 3sa). It should be mentioned that the resulting structures are the core motifs in ubiquitous bioactive compounds, such as MAPKAP-K2 inhibitors^{23a} and MCH receptor antagonists.^{23b}

The reactions of benzamides bearing different substituents on the allyl groups were also investigated (Scheme 3). The ethyl, *n*-butyl, and benzyl groups were compatible (5aa–5ca), and even



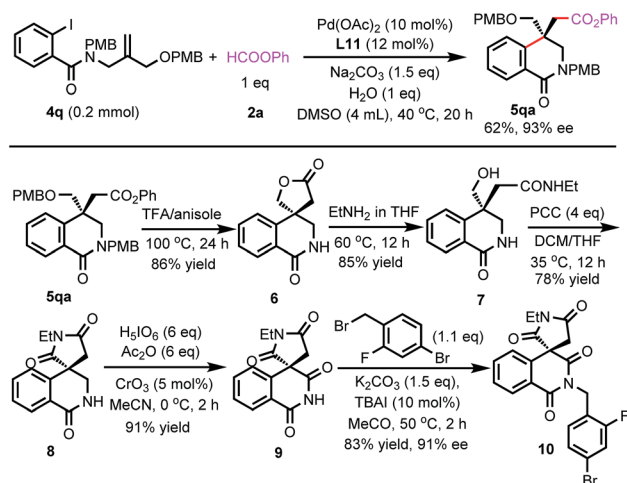
Scheme 3 Scope of benzamides with respect to substituents on the allyl and amide groups.^[a] Isolated yield.^[b] 60 °C.^[c] 1 equiv. of HCOOPh.



methoxy and *p*-methoxybenzyl groups were tolerated (**5da** and **5ea**), which allows for further manipulation of the resulting products. In the presence of a bulky *tert*-butyl group, only a trace amount of desired product **5fa** was formed. A phenyl or ester group suppressed the carbonylative Heck reaction (**5ga** and **5ha**). In these reactions, the arylpalladium intermediates were directly carbonylated to form phenyl benzoates. For the reaction of the substrate bearing a benzyloxy group, carbopalladation occurred. However, the resulting alkylpalladium species underwent β -O-elimination before it reacted with CO (**5ia**). Furthermore, a range of functionalities on the amide groups were suitable, and the reactions were highly enantioselective (**5ja**–**5na**). However, the unprotected benzamide failed to form the desired product (**5oa**) and remained intact in the reaction. For phenyl-protected benzamide, only a trace amount of product **5pa** was observed, and several unidentified side-products were formed.

The reaction scope with respect to aryl formates was also evaluated (Scheme 4). Aryl formates bearing an electron-donating methoxy group or electron-withdrawing ester group were efficient carbonylating reagents (**3ab** and **3ac**). The chloro group was well-tolerated and naphthalen-2-yl formate was also suitable (**3ad** and **3ae**). Reactions using alkyl formates were also investigated. The desired carbonylated isoquinolinone products were not observed for methyl or benzyl formate.

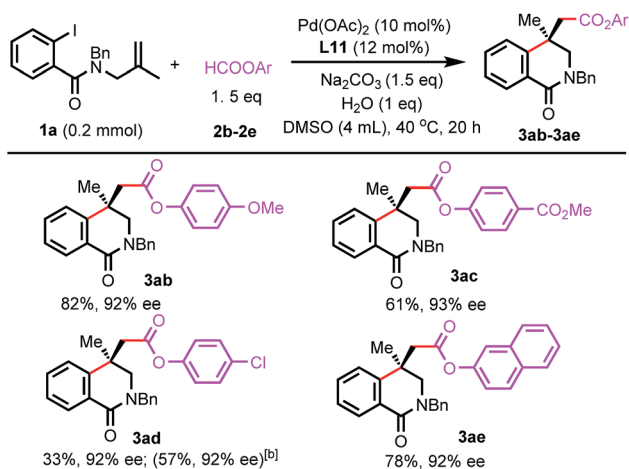
Next, we tried to synthesize Minalrestat with our enantioselective carbonylative Heck reaction. The quaternary carbon stereocenter in Minalrestat tends to racemize. However, the introduction of an *N'*-alkyl group could prevent the racemization, and it did not change the oral AR inhibitory potency of the *N'*-unsubstituted parent compounds. Furthermore, the analogues without the 6-fluoro group racemize more slowly and still have potent activity.^{4a} Therefore, we chose Minalrestat analogue **10** as the target molecule (Scheme 5). On the other hand, a modular synthetic approach is highly desirable for drug discovery, because it would allow easy access to a number of analogues. To this end, we designed benzamide **4q** as the substrate. **4q** contains two PMB groups that can be readily



Scheme 5 Synthesis of the Minalrestat analogue.

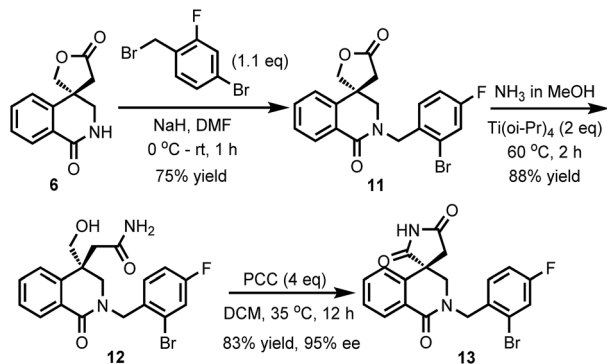
removed. Gratefully, **4q** underwent the carbonylative Heck reaction under the standard conditions to afford product **5qa** in 93% ee. The two PMB groups could be removed using trifluoroacetic acid simultaneously, and the subsequent lactonization yielded **6**. Lactone **6** was transformed into amide **7** with EtNH₂. The free hydroxy group of **7** was oxidized with PCC, which led to the formation of succinimide **8**. The α -methylene group of **8** was then oxidized to the carbonyl group, yielding the isoquinolinone skeleton in 91% yield. The free amino group can be derivatized with various substituents for bioactivity studies. The 4-bromo-2-fluorobenzyl moiety could enhance the oral potency of aldose reductase inhibitors effectively,²⁴ so it was introduced to the amide group of **9**. The ee value of the final product **10** was 91%, which indicates that its quaternary carbon stereocenter did not racemize during the entire reaction process. Actually, the ee value of **10** remained unchanged even after one week.

Although the introduction of an *N'*-alkyl group can prevent the racemization of Minalrestat and its analogues, the acidity of the succinimides is very critical to the intrinsic activity of the parent molecules. The reason for the *N'*-alkylated analogues having equal oral AR inhibitory potency should be that the alkyl groups could be removed through biotransformation.^{4a} The regenerated (N–H) compounds can still racemize, which results in the reduction of potency or might even cause unexpected side-effects. Therefore, it is highly desirable to develop new structures as ARIs that do not racemize. The racemization of Minalrestat is caused primarily by the presence of the 3-carbonyl group. Our asymmetric carbonylative Heck reaction provides a facile method for the synthesis of isoquinolinones. Therefore, we synthesized a decarbonylated analogue of the isoquinolinones. As shown in Scheme 6, compound **6** was first allowed to react with 4-bromo-2-fluorobenzyl bromide to give **11**. The subsequent aminolysis with NH₃ and oxidation with PCC afforded isoquinolinone **13**. The ee value of **13** was 95%, which indicates the high stability of the quaternary carbon stereocenter.



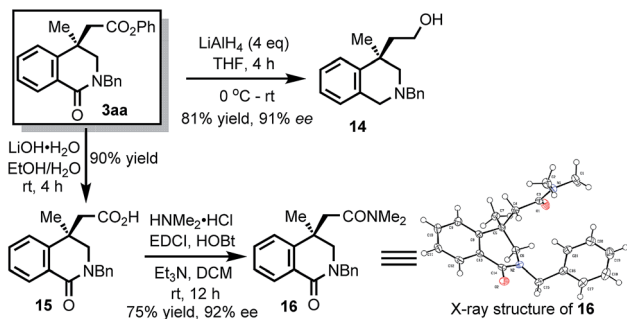
Scheme 4 Scope of formate esters.^[a] Isolated yield.^[b] 60 °C.



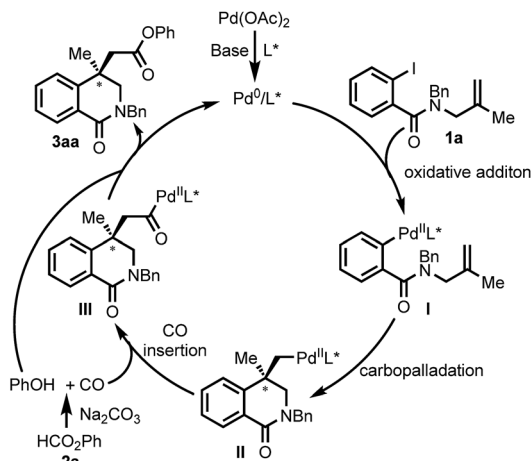


Scheme 6 The synthesis of the non-racemizable derivative of Minalrestat.

The isoquinolinone products can be readily transformed into other quaternary stereocarbon-containing compounds (Scheme 7). For example, the amide group could be reduced to give **14**. It should be noted that the resulting 4,4-disubstituted 1,2,3,4-tetrahydroisoquinoline represents an essential skeleton widely found in bioactive compounds such as crinanes^{7d} and Secoplicamine.^{7c} Furthermore, the ester group could be hydrolyzed to give compound **15**. Notably, 4-carboxyl isoquinolinones also showed activities for aldose reductase



Scheme 7 Transformation of the isoquinolinone product.



Scheme 8 Plausible mechanism.

enzyme.^{4b} Finally, **15** was transformed to amide **16**. The structure of **16** was characterized by X-ray crystallography.

Based on the previous report,^{16a} a plausible mechanism for the Pd-catalyzed Heck carbonylation is depicted in Scheme 8. The oxidative addition of aryl iodide **1a** to a Pd⁰ species that is generated *in situ* forms Pd^{II} intermediate **I**. **I** undergoes intramolecular carbopalladation to afford σ -alkyl Pd^{II} complex **II**. Phenyl formate **2b** generates CO and phenol with Na₂CO₃ as the base.²⁵ The migratory insertion of CO into the σ -alkyl-Pd bond gives acylpalladium complex **III**. The nucleophilic attack of the phenol and subsequent reductive elimination give the final product **3aa** with regeneration of Pd⁰ catalyst.

Conclusions

In summary, we have developed a Pd-catalyzed enantioselective intramolecular carbonylative Heck reaction by using formate esters as the source of CO. A range of *N*-allyl benzamides were transformed into isoquinolinones bearing an all-carbon quaternary center in good yields and high enantioselectivities in the presence of (*R*)-SEGPHOS as the ligand. This reaction represents one of the rare examples of asymmetric six-membered cyclization reactions through a domino intramolecular Heck-nucleophilic capture sequence. The resulting isoquinolinone represents the core skeleton in ubiquitous bioactive compounds, and therefore the reaction should have great potential to be applied in organic synthesis. The reaction has been successfully applied to the first asymmetric construction of the isoquinolinone skeleton in Minalrestat and allows easy access to a range of Minalrestat analogues for developing new aldose reductase inhibitors. Synthesizing other Minalrestat analogues and testing their bioactivities are currently underway in our laboratory.

Conflicts of interest

The authors declare no conflict of interest.

Acknowledgements

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Notes and references

- 1 A. Pau, B. Asproni, G. Murineddu, G. Boatto, G. E. Grella, D. Rakowitz, L. Costantino and G. A. Pinna, *Med. Chem.*, 2006, **2**, 39.
- 2 M. Brownlee, *Diabetes*, 2005, **54**, 1615.
- 3 R. Maccari and R. Ottaná, *J. Med. Chem.*, 2015, **58**, 2047.
- 4 (a) M. S. Malamas, T. C. Hohman and J. Millen, *J. Med. Chem.*, 1994, **37**, 2043; (b) M. S. Malanias and T. C. Hohman, *J. Med. Chem.*, 1994, **37**, 2059.
- 5 The asymmetric reaction for the synthesis of Ranirestat is limited to pyrrole-based substrates: T. Mashiko, K. Hara, D. Tanaka, Y. Fujiwara, N. Kumagai and M. Shibasaki, *J. Am. Chem. Soc.*, 2007, **129**, 11342.



- 6 (a) G. R. Pettit, S. Ducki, S. A. Eastham and N. Melody, *J. Nat. Prod.*, 2009, **72**, 1279; (b) Z. Jin, *Nat. Prod. Rep.*, 2013, **30**, 849.
- 7 (a) I. Ninomiya, O. Yamamoto and T. Naito, *J. Chem. Soc., Chem. Commun.*, 1976, 437; (b) X. Xiao, Z.-H. Miao, S. Antony, Y. Pommier and M. Cushman, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 2795; (c) B. V. S. Reddy, D. Medaboina and B. Sridhar, *J. Org. Chem.*, 2015, **80**, 653; (d) D. Hulcová, K. Breiterová, T. Siatka, K. Klímová, L. Davani, M. Šafratová, A. Hošťálková, A. D. Simone, V. Andrisano and L. Cahliková, *Molecules*, 2018, **23**, 719.
- 8 (a) M. Cushman and J.-K. Chen, *J. Org. Chem.*, 1987, **52**, 8; (b) D. A. Petrone, H. Yoon, H. Weinstabl and M. Lautens, *Angew. Chem., Int. Ed.*, 2014, **53**, 7908; (c) H. Yoon, D. A. Petrone and M. Lautens, *Org. Lett.*, 2014, **16**, 6420.
- 9 (a) Y. Sato, M. Sodeoka and M. Shibasaki, *J. Org. Chem.*, 1989, **54**, 4738; (b) N. E. Carpenter, D. J. Kucera and L. E. Overman, *J. Org. Chem.*, 1989, **54**, 5846.
- 10 Selected reviews: (a) A. B. Dounay and L. E. Overman, *Chem. Rev.*, 2003, **103**, 2945; (b) M. Shibasaki, E. M. Vogl and T. Ohshima, *Adv. Synth. Catal.*, 2004, **346**, 1533; (c) D. M. McCartney and P. Guiry, *Chem. Soc. Rev.*, 2011, **40**, 5122; (d) Y. Ping, Y. Li, J. Zhu and W. Kong, *Angew. Chem., Int. Ed.*, 2018, **57**, 2; (e) C.-X. Zhuo, C. Zheng and S.-L. You, *Acc. Chem. Res.*, 2014, **47**, 2558.
- 11 A. Pinto, Y. Jia, L. Neuville and J. Zhu, *Chem.–Eur. J.*, 2007, **13**, 961.
- 12 (a) H. Liu, C. Li, D. Qiu and X. Tong, *J. Am. Chem. Soc.*, 2011, **133**, 6187; (b) S. G. Newman, J. K. Howell, N. Nicolaus and M. Lautens, *J. Am. Chem. Soc.*, 2011, **133**, 14916; (c) Z.-M. Zhang, B. Xu, L. Wu, L. Zhou, D. Ji, Y. Liu, Z. Li and J. Zhang, *J. Am. Chem. Soc.*, 2019, **141**, 8110.
- 13 (a) W. Kong, Q. Wang and J. Zhu, *J. Am. Chem. Soc.*, 2015, **137**, 16028; (b) S. Tong, A. Limouni, Q. Wang, M. X. Wang and J. Zhu, *Angew. Chem., Int. Ed.*, 2017, **56**, 14192.
- 14 W. Kong, Q. Wang and J. Zhu, *Angew. Chem., Int. Ed.*, 2016, **55**, 9714.
- 15 R. Liu, Y. Wang, Y. Li, B. Huang, R. Liang and Y.-X. Jia, *Angew. Chem., Int. Ed.*, 2017, **56**, 7475.
- 16 (a) R. C. Carmona, O. D. Köster and C. R. D. Correia, *Angew. Chem., Int. Ed.*, 2018, **57**, 12067; (b) H. Hu, F. Teng, J. Liu, W. Hu, S. Luo and Q. Zhu, *Angew. Chem., Int. Ed.*, 2019, **58**, 9225.
- 17 (a) P. Diaz, F. Gendre, L. Stella and B. Charpentier, *Tetrahedron*, 1998, **54**, 4579; (b) M. Minatti, X. Zheng and S. L. Buchwald, *J. Org. Chem.*, 2007, **72**, 9253; (c) G. Yue, K. Lei, H. Hirao and J. Zhou, *Angew. Chem., Int. Ed.*, 2015, **54**, 6531; (d) C. Shen, R. Liu, R. Fan, Y. Li, T. Xu, J. Gao and Y. Jia, *J. Am. Chem. Soc.*, 2015, **137**, 4936; (e) W. Kong, Q. Wang and J. Zhu, *Angew. Chem., Int. Ed.*, 2017, **56**, 3987; (f) Z.-M. Zhang, B. Xu, Y. Qian, L. Wu, Y. Wu, L. Zhou, Y. Liu and J. Zhang, *Angew. Chem., Int. Ed.*, 2018, **57**, 10373.
- 18 (a) H. Cong and G. C. Fu, *J. Am. Chem. Soc.*, 2014, **136**, 3788; (b) W. You and M. K. Brown, *J. Am. Chem. Soc.*, 2015, **137**, 14578; (c) X. Qin, M. W. Y. Lee and J. S. Zhou, *Angew. Chem., Int. Ed.*, 2017, **56**, 12723; (d) K. Wang, Z. Ding, Z. Zhou and W. Kong, *J. Am. Chem. Soc.*, 2018, **140**, 12364; (e) Y. Jin and C. Wang, *Angew. Chem., Int. Ed.*, 2019, **58**, 6722.
- 19 For two asymmetric six-membered cyclization examples with low enantioselectivities, see: ref. 12a and 18e. Although an asymmetric reaction of vinyl iodides to form six-membered tetrahydropyridines has been reported, a mechanism involving the vinylborylation of alkenes is preferred. (a) Z. Jiang, L. Hou, C. Ni, J. Chen, D. Wang and X. Tong, *Chem. Commun.*, 2017, **53**, 4270. For a diastereoselective reaction, see: (b) H. Yoon, Y. J. Jang and M. Lautens, *Synthesis*, 2016, **48**, 1483.
- 20 For selected racemic six-membered cyclization reactions, see: (a) R. Grigg, S. Sukirthalingam and V. Sridharan, *Tetrahedron Lett.*, 1991, **32**, 2545; (b) R. Grigg, V. Santhakumar and V. Sridharan, *Tetrahedron Lett.*, 1993, **34**, 3163; (c) T. Yao, T. Liu and C. Zhang, *Chem. Commun.*, 2017, **53**, 2386.
- 21 D. Formenti, F. Ferretti and F. Ragaini, *ChemCatChem*, 2018, **10**, 148.
- 22 F. Ozawa, A. Kubo and T. Hayashi, *Chem. Lett.*, 1992, **21**, 2177.
- 23 (a) J.-P. Wu, J. Wang, A. Abeywardane, D. Andersen, M. Emmanuel, E. Gautschi, D. R. Goldberg, M. A. Kashem, S. Lukas, W. Mao, L. Martin, T. Morwick, N. Moss, C. Pargellis, U. R. Patel, L. Patnaude, G. W. Peet, D. Skow, R. J. Snow, Y. Ward, B. Werneburg and A. White, *Bioorg. Med. Chem. Lett.*, 2007, **17**, 4664; (b) M. Lindvall, C. McBride, M. McKenna, T. G. Gesner, A. Yabannavar, K. Wong, S. Lin, A. Walter and C. M. Shafer, *ACS Med. Chem. Lett.*, 2011, **2**, 720.
- 24 D. Stribling and D. R. Brittan, Structure Activity Relationships of Aldose Reductase Inhibitors and Their Possible Use in Diabetic Complications, in *Innovative Approaches in Drug Research*, ed. A. F. Hams, Elsevier Science Publishers B. V., Amsterdam, 1986, pp. 297–313.
- 25 T. Ueda, H. Konishi and K. Manabe, *Org. Lett.*, 2011, **14**, 3100.

