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# PAPER



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### 1. Introduction

Saxitoxin (STX) is one of the most important marine toxins, which has a broad group of natural neurotoxic alkaloids, commonly known as the paralytic shellfish toxins (PSTs).<sup>1,2</sup> STX is produced by Dinoflagellattatte genus, Gonyaulax catenella, Protogonyaulax tamarensis, Alexandrium catenella, and Alexandrium minutum, presenting a considerably high concentration in domestic water especially during harmful algal bloom (HAB) events.3-5 Excessive STX causes water pollution and has mortal impact on other animals, plants, and microorganism. Even though, it does not have effect on the growth of some animals, for instance, fish or shellfish etc., but it is englobed by them and accumulated in their body. Intoxication with STX may result in the severe and occasionally fatal illness and there is no antidote for STX with artificial respiration and fluid therapy available currently.<sup>6</sup> In drinking water, a guidance value of 3 ng mL<sup>-1</sup> for saxitoxin concentration (toxicity equivalents) is currently implemented in Australia, Brazil and New Zealand.7 It is demanding rapid and sensitive detection of STX for real-time monitoring water environmental pollution, marine aquaculture pollution, and seafood safety.

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## A rapid and ultrasensitive colorimetric biosensor based on aptamer functionalized Au nanoparticles for detection of saxitoxin

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Saxitoxin (STX) is one of the most important marine toxins which affects the safety of domestic water. Rapid, sensitive and selective recognition of STX is crucial in environment monitoring. Here, we demonstrate a facile and ultrasensitive colorimetric sensor based on gold nanoparticles (Au NPs) and aptamer (Au NPs-aptamer biosensor) for specific and quantitative detection of STX. The aptamer reacts specifically with STX, resulting in the aggregation of Au NPs and the color change of the Au NP solution. The lowest detection concentration of the colorimetric sensor is 10 fM (3 fg mL<sup>-1</sup>), and a good linear relationship ( $R^2 = 0.9852$ ) between the absorbance ratio and STX concentrations (10 fM to 0.1  $\mu$ M) indicates that our Au NPs-aptamer biosensor can be used for quantitative sensing of STX. The detection time of STX is 30 minutes, and the sensor is successfully applied in the specific detection of STX in seawater. The Au NP-aptamer biosensor shows great potential in practical applications to monitor environmental pollution, marine aquaculture pollution, and seafood safety.

The traditional analytical techniques for STX include high performance liquid chromatography (HPLC),<sup>8</sup> liquid chromatography-tandem mass spectrometry (LC-MS),<sup>9</sup> and enzyme-linked immunosorbent assay (ELISA).<sup>10</sup> HPLC and LC-MS need sophisticated equipment and complex pre-treatment, which limit the detection to the laboratory.<sup>11,12</sup> It involves tedious lab work and non-specificity during ELISA methods,<sup>13</sup> so it is hard to conduct real-time detection. As a result, to realize real-time detection, it desires new approaches to develop a facile, rapid, and sensitive platform to identify and monitor STX.

Au NPs have been employed as colorimetric sensors due to their excellent optical properties14 and their ability to bond biomolecules.15 The colorimetric sensor based on Au NPs has the advantages of high sensitivity and simplicity, and is capable of realizing real-time detection without large-scale instruments and equipment. It has been used to detect proteins,16-18 DNAs,19-21 metal ions,22,23 and small molecules.24-27 To construct Au NPs based colorimetric sensor, a specific molecule is needed to bind with Au NPs, which can specifically react with target molecules, and then gets desorbed from Au NPs resulting in the aggregation of Au NPs to induce color change. Here we report an Au NPs-aptamer based sensor, and the aptamer functionalized on Au NPs specifically reacts with STX, resulting in the aggregation of Au NPs and the change of the surface plasmon resonance absorption peak, which can be used for the quantitative analysis of STX. Our Au NPs-aptamer based colorimetric sensor achieves the lowest detection concentration of 10 fM ( $\sim$ 3 fg mL<sup>-1</sup>), and presents a good linear relationship ( $R^2 = 0.9852$ )

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between the absorbance ratio and STX concentrations (10 fM to 0.1  $\mu$ M). The detection time of STX is 30 minutes, and the sensor is successfully applied in the specific detection of STX in seawater. This work opens new applications of colorimetric sensor in marine science and technology.

### 2. Materials and methods

#### 2.1 Materials

Sodium chloride (NaCl) was purchased from Solarbio Science and Technology Corporation, Beijing, China. Saxitoxin toxin was purchased from National Research Council-Conseil National de Recherché Canada (NRC-CNRC, Canada). 96 well microplates were polystyrene, flat-bottomed, tissue culture treated (Corning, cat. no. 3595).

#### 2.2 Au NPs synthesis and characterization

Gold colloids were prepared by citrate thermal reduction method.<sup>28</sup> Typically, in the process of thermal reduction, a gold sol was prepared by mixing 1 mL of 1 wt% HAuCl<sub>4</sub> aqueous solution and 1.5 mL of 38.8 mM sodium citrate aqueous solution with 90 mL boiling water. The solution turned purple red within 30 s, and was cooled quickly in the ice bath.<sup>29</sup> This resulted in a stable dispersion of gold particles with an average diameter of around 14 nm. The microstructure of Au NPs was characterized using a FEI-Nova NanoSEM 450 scanning electron microscope (SEM), and the applied acceleration voltage was 10 kV with a working distance of 5.0 mm. The high-resolution transmission electron microscope (HRTEM) measurement of Au NPs was performed by a FEI Tecnai G2 F20 electron microscope operated at 200 kV. The particle size of Au NPs was calculated by Nano Measurer software.

#### 2.3 STX aptamer synthesis

The sequence of STX aptamer was from the literature<sup>30</sup> and synthesized from Sangon Biotech Corporation, Shanghai, China, and listed in Table 1.

#### 2.4 General procedure of STX detection

First, 10  $\mu$ L of 0.36 M NaCl solution was added into 80  $\mu$ L Au NPs solution. Then 10  $\mu$ L of 1  $\mu$ M STX aptamer solution was added and incubated with Au NPs for 14 hours. Next, 10  $\mu$ L of tenfold dilutions STX by f/2 medium<sup>31</sup> was added into the STX aptamer-Au NPs complex and reacted with aptamer for 30 min. The UV-vis absorption spectra of samples were scanned from

Table 1         Sequences of STX aptamer					
Name	Sequence (from the 5' to 3'-end)	NT			
STX aptamer	GGTATTGAGGGTCGCATCCC GTGGAAACATGTTCATTG GGCGCACTCCGCTTTCTG TAGATGGCTCTA ACTCTCCTCT	78			



**Fig. 1** (a) The detection scheme of colorimetric STX sensor based on Au NPs. (I) Dispersed Au NPs in the initial reaction system; (II) Au NPs aggregated with NaCl input; (III) Au NPs back to dispersion status after binding with aptamer; (IV) Au NPs aggregated after STX loading. (b) Structure of STX aptamer and STX. (c) UV-vis spectra of Au NPs under different conditions.

400 to 800 nm using a BioTek microplate reader (Synergy<sup>TM</sup> H1) with pure water as the control in 96 well microplates. A working curve was drawn in which *x* and *y*-coordinates were wavelength and absorption spectra, respectively. The final volume of all samples was 110  $\mu$ L. Each detection was performed at least three times in parallel and GraphPad Prism 8 software (https://www.graphpad.com/scientific-software/prism/) was used for data analysis.

### 3. Results and discussion

# 3.1 Colorimetric detection mechanism of STX based on aptamer functionalized Au NPs

The colorimetric detection mechanism of STX is based on the aggregation of Au NPs functionalised with aptamer as shown in Fig. 1. Dispersed Au NPs aggregate with appropriate NaCl concentration, and get back to dispersion status after the binding with aptamer. Once STX is loaded, the aptamer reacts with STX and expose Au NPs to get aggregated. The absorbance peaks changes to form different  $A_{680}/A_{520}$  ratios with different STX concentrations, which can be used for the quantitative analysis of STX.

#### 3.2 Characterization of Au NPs

Here we employed Au NPs which were prepared by citrate thermal reduction method, and the material characteristics were shown in Fig. 2. SEM image of Au NPs is shown in Fig. 2a, and Fig. 2b presents the statistical size of Au NPs is around 14 nm. The charge balance of Au NPs and recovery after aggregation are the most crucial factors for Au NPs based sensor, which are determined by Au NPs material properties and reaction system of sensing assay, including Au NPs size, optical characteristics, and composites in aqueous reaction. HRTEM images in Fig. 2c and d are Au NPs morphology in



Fig. 2 (a) SEM image of Au NPs. (b) Particle size analysis of Au NPs. HRTEM images of Au NPs (c) in dispersion and (d) in aggregation.

dispersed and aggregated status. Fig. 2c presents good dispersion of Au NPs, most Au NPs are spherical and while few are triangle shape. Dispersed Au NPs are triggered by NaCl or other conditions, and aggregate to form large size particles, as shown in Fig. 2d viewed under HRTEM.

#### 3.3 Optimization of experimental conditions

The NaCl concentration is a critical parameter to induce the aggregation of Au NPs, and detailed study is conducted to investigate the NaCl concentration threshold for Au NPs aggregation. Here, the used reaction system has Au NPs density of 47.8  $\mu$ g mL<sup>-1</sup>. Fig. 3a shows the representative UV-vis adsorption spectra of Au NPs solution mixed with different concentrations of NaCl from 0 M to 0.4 M, and the 520 nm peak decreases with increased NaCl concentration. It shows a sudden



**Fig. 3** (a)The alteration of Au NPs aggregation in different NaCl concentrations including color and absorption spectra. (b) Function of Au NPs and concentrations of aggregated NaCl. The red dot is the threshold we chose. (c) The alteration of color and absorption spectra with STX aptamer incubation time. (d) The absorption spectra and alteration of color with reaction time after loading STX samples.



Fig. 4 Sensitivity of STX sensing. (a) Absorption spectrum and (b)  $A_{680}/A_{520}$  responses to different concentration of STX.

drop at 520 nm and color change from red to blue when the concentration increases from 0.2 M to 0.4 M, which indicates the threshold lies in the range of 0.2–0.4 M. The detailed absorption spectra of NaCl concentration with fine steps shows the concentration threshold of 0.36 M. As a result, we use 0.36 M NaCl to aggregate Au NPs in the subsequent detection.

Actually, NaCl concentration threshold is dependent on Au NPs density in the reaction system. A higher Au NPs density needs higher NaCl concentration threshold, as shown in Fig. 3b, and NaCl concentration threshold presents a linear relationship with Au NPs concentration. In our reaction system, the concentration of Au NPs is 47.8  $\mu$ g mL<sup>-1</sup> and 0.36 M NaCl is used to aggregate Au NPs.

During the sensing process, well-dispersed Au NPs aggregate after the addition of NaCl, and return to the dispersion status after bonding with aptamers, because aptamers protect the charge balance of Au NPs from aggregation.<sup>32,33</sup> The aptamer immobilization efficiency depends on incubation time with Au NPs. As shown in Fig. 3c, the color gradually changes from bluepurple back to red and the specific peaks recover from 680 nm to 520 nm with increasing incubation time, indicating that Au NPs get back to dispersed status after the bonding of aptamers. When the incubation time is more than 14 hours, the bonding of aptamers reaches saturation. Therefore, we chose 14 hours to immobilize aptamers with Au NPs.



Fig. 5  $A_{\rm 680}/A_{\rm 520}$  ratio of STX and other toxins with different concentrations.

 Table 2
 Comparison of STX detection with different approaches<sup>a</sup>

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Method	Detector	Detection limit	Preparation time	Detection time	Detection cost	References
MBA	Mouse	250 pg	>1 hour	4–6 hours	High	34 and 35
HPLC	No need	$0-27.6 \text{ nmol g}^{-1}$	>20 hours	>2 hours	High	8 and 12
LC-MS	No need	16 pmol $\mu L^{-1}$	>20 hours	>2 hours	Very high	9 and 36
Immuno-chromatography	Antibody	$5.2 \text{ ng mL}^{-1}$	>10 hours	>1 hour	High	37
Immuno-assay	Antibody	$0.8-66.6 \text{ ng mL}^{-1}$	>16 hours	36 min	High	38
ELISA	Antibody	$0.41-1.9 \text{ ng g}^{-1}$	>8 hours	>100 min	High	10 and 39
Colorimetric	Antibody	$1-50 \text{ ng mL}^{-1}$	>8 hours	$\sim$ 1 hour	High	40
Bioelectricity	Na <sup>+</sup>	5-1000 fg	>15 hours	>1 hour	High	41
Colorimetric	Aptamer	10 fM (3 fg mL <sup><math>-1</math></sup> )	14.5 hours	30 min	Low	This work

<sup>*a*</sup> MBA is short for mouse bioassay. HPLC is short for high performance liquid chromatography. LC-MS is short for liquid chromatography-tandem mass spectrometry. ELISA is short for enzyme-linked immunosorbent assay.

STX aptamers bonding to Au NPs prevent them from aggregation even in the presence of NaCl. After sample loading, STX is specifically bonded with STX aptamer, and Au NPs lose the protection from aptamers and aggregate with the assistance of NaCl. Experiments show that STX reacts with aptamer in 10 minutes, and absorbance decreases at 520 nm while increases at 680 nm obviously, as in Fig. 3d. The solution color changes from red to blue-purple. Longer reaction time resulted to more adequate combination of STX aptamer and STX, and the reaction reaches saturation when it is more than 30 min, and the absorbance curve gets stable. As a result, 30 min is selected to permit the sufficient specific capture of STX by aptamer.

#### 3.4 Detection of STX in seawater

Sensitivity is one of the most parameter to evaluate biosensor and quantify the toxin concentration in test samples. In order to achieve the sensitivity, STX is mixed with seawater to form a series concentration from 0.1 fM to 1  $\mu$ M, and their absorbance spectra are tested, as shown in Fig. 4a. As the STX concentration increases,  $A_{520}$  peak drops while  $A_{680}$  ascends, and solution color changes from red to blue-purple. The ratio  $A_{680}/A_{520}$  is applied to quantify toxin concentration, and Fig. 4b shows its dependence on toxin concentration. In the range of 10 fM to 0.1  $\mu$ M,  $A_{680}/A_{520}$  ratio presents a linear relationship with STX concentration. The lowest STX detection concentration of 10 fM is achieved, and STX concentration higher than 0.1  $\mu$ M is limited. However, higher concentration of STX sensing can be achieved by expanding reaction system, since it's easier to get saturated in smaller reaction system.

To evaluate the specificity of STX detection by aptamer and Au NPs based biosensor, STX and other three kinds of toxin are mixed with seawater, for instance, cylindrospermopsin (CYN), okdaic acid (OA) and domoic acid (DA). Experiments indicate that STX aptamer is specific only for STX toxin, and the  $A_{630}/A_{520}$  ratio of other three toxins is similar even their concentrations vary from  $10^{-14}$  M to  $10^{-6}$  M, while it increases obviously with ascending STX concentration, as shown in Fig. 5. These results demonstrate the capability of aptamer and Au NPs based biosensor to conduct specific and sensitive STX detection. Our biosensor is constructed by specific aptamer and chemically-synthesized Au NPs, which realizes ultrasensitive, fast, and economic detection of STX. In comparison with other approaches, as listed in Table 2, our biosensor achieves the lowest sensing concentration of 10 fM ( $3 \text{ fg mL}^{-1}$ ) in detection time of 30 minutes. In addition, aptamer-Au NPs based biosensor is economic because it does not require expensive regents, complex process, and equipment during the whole process, which makes it promising in practical applications to monitor environmental pollution condition, marine aquaculture pollution, and seafood safety.

The detection sensitivity is strongly dependent on the sensing systems, including the binding affinity of STX, physical size of Au NPs, appropriate concentration match between Au NPs, STX aptamer and NaCl. The STX aptamer used in this work has much higher binding affinity than antibody.<sup>42,43</sup> In the range of 5–20 nm, the peak displacement of Au NPs shows a linear relationship with increasing particle size.<sup>44</sup> The average diameter of our Au NPs is 14.51 nm, which is located in the most significant range of peak changes. Finally, the optimized concentrations of reaction regents in the whole system also plays important role in performance improvement.

### 4. Conclusions

In summary, we demonstrate an ultrasensitive, specific, fast, and economic colorimetric biosensor for STX detection. After optimization of reaction system, aptamer-Au NPs based biosensor achieves the lowest STX detection concentration of 10 fM (3 fg mL<sup>-1</sup>) and sensing time of 30 minutes, presenting a linear relationship in the range of 10 fM to 0.1  $\mu$ M aptamer-Au NPs based biosensor is a universal sensing platform, and it is applicable to other toxins detection. Further work is desired to realize a high throughput chip-based platform for multiplex toxins detection. The simple manipulation and economic expense empower the colorimetric biosensor promising applications in monitor environmental pollution condition, marine aquaculture pollution, and seafood safety.

### Conflicts of interest

All the authors declare no conflicts of interest with this work.

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