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#### 1. Introduction

Bi<sub>2</sub>Te<sub>3</sub>-based alloys are the only commercialized thermoelectric materials for solid-state cooling and low-mid temperature (473–873 K) power generation, and their ingot-type materials are widely used due to a high thermoelectric figure of merit ( $zT = S^2 \sigma T/\kappa_{tot}$ , where *S*,  $\sigma$ ,  $\kappa_{tot}$ , and *T* are the Seebeck coefficient, electrical conductivity, total thermal conductivity, and the absolute temperature, respectively) of about 1.0 near room-temperature.<sup>1</sup> However, ingots of Bi<sub>2</sub>Te<sub>3</sub>-based alloys have a poor mechanical reliability (fracture strength of ~10 MPa) because of 00*l*-oriented structure weakly bonded by van der Waals forces,<sup>2</sup> which limits their wider applications such as in automobile thermoelectric generator (ATEG). To address this,

# Improvement in the thermoelectric performance of highly reproducible n-type (Bi,Sb)<sub>2</sub>Se<sub>3</sub> alloys by Cl-doping<sup>+</sup>

Nadra Nasir,‡<sup>a</sup> Kyu Hyoung Lee,‡<sup>b</sup> Sang-il Kim, <sup>b</sup><sup>c</sup> Hyun-Sik Kim,<sup>d</sup> Jae-Hong Lim,<sup>e</sup> Liangwei Fu<sup>\*a</sup> and Sung Wng Kim<sup>b</sup>\*<sup>af</sup>

 $(Bi,Sb)_2Se_3$  alloys are promising alternatives to commercial n-type  $Bi_2(Te,Se)_3$  ingots for low-mid temperature thermoelectric power generation due to their high thermoelectric conversion efficiency at elevated temperatures. Herein, we report the enhanced high-temperature thermoelectric performance of the polycrystalline Cl-doped  $Bi_{2-x}Sb_xSe_3$  (x = 0.8, 1.0) bulks and their sustainable thermal stability. Significant role of Cl substitution, characterized to enhance the power factor and reduce the thermal conductivity synergetically, is clearly elucidated. Cl-doping at Se-site of both  $Bi_{1,2}Sb_{0,8}Se_3$  and  $BiSbSe_3$  results in a high power factor by carrier generation and Hall mobility improvement while maintaining converged electronic band valleys. Furthermore, point defect phonon scattering originated from mass fluctuations formed at Cl-substituted Se-sites reduces the lattice thermal conductivity. Most importantly, spark plasma sintered Cl-doped  $Bi_{2-x}Sb_xSe_3$  bulks are thermally stable up to 700 K, and show a reproducible maximum thermoelectric figure of merit, *zT*, of 0.68 at 700 K.

polycrystalline bulk form materials have been intensively studied, and an improved mechanical strength (~80 MPa) with a higher  $zT \sim 1.1$  at 300 K has been obtained in micro-grained ptype  $Bi_{2-x}Sb_{x}Te_{3}$  prepared by ball milling (BM) and spark plasma sintering (SPS).<sup>2</sup> Its n-type counterpart with a comparable mechanical strength and zT is required to construct thermoelectric module with improved mechanical reliability as well as high performance, however, no marked improvement in zT from n-type micro-grained materials was achieved in BMed and SPSed Bi<sub>2</sub>Te<sub>2.7</sub>Se<sub>0.3</sub> (~0.63 at 300 K).<sup>3</sup> Furthermore, a severe reproducibility problem was also found in this polycrystalline sample owing to the uncontrollable defect structures such as vacancies (Te- or Se-site) and antisite defects. Polycrystalline bulk of n-type  $Bi_2Te_{2.7}Se_{0.3}$  with high zT (~0.98 at 300 K) and improved reproducibility has been demonstrated by combining a cold deformation and a hot extrusion benefitting from precise control of point defects,4 however, a simpler and yet easily scalable approach is always sought after.

Bi<sub>2</sub>Se<sub>3</sub> is a narrow-bandgap layered semiconductor (space group  $R\bar{3}m$ - $D_{3d}^5$ ) with tetradymite structure and it has singly degenerate conduction band. The conduction band minimum (CBM) is observed at the center of the Brillouin zone ( $\Gamma$ -point) and the second conduction band is located 150–250 meV (*Z*-point) above the CBM,<sup>5-7</sup> thus the *zT* of pristine Bi<sub>2</sub>Se<sub>3</sub> is very low (<0.1 at 300 K) mainly due to low *S* (~-40 µV K<sup>-1</sup> at 300 K).<sup>8</sup> High  $\kappa_{tot} \sim 2.4$  W m<sup>-1</sup> K<sup>-1</sup> at 300 K is another reason for the low *zT* of Bi<sub>2</sub>Se<sub>3</sub>. Very recently, Te-free (Bi,Sb)<sub>2</sub>Se<sub>3</sub>-based alloys have been received attention as promising alternatives to Bi<sub>2</sub>(Te,Se)<sub>3</sub>-based alloys especially for low-mid temperature power

<sup>&</sup>lt;sup>a</sup>Department of Energy Science, Sungkyunkwan University, Suwon 16419, South Korea. E-mail: fulw@skku.edu; kimsungwng@skku.edu

<sup>&</sup>lt;sup>b</sup>Department of Materials Science and Engineering, Yonsei University, Seoul 03722, South Korea

<sup>&</sup>lt;sup>c</sup>Department of Materials Science and Engineering, University of Seoul, Seoul 02504, South Korea

<sup>&</sup>lt;sup>d</sup>Department of Materials Science and Engineering, Hongik University, Seoul 04066, South Korea

<sup>&</sup>lt;sup>e</sup>Department of Materials Science and Engineering, Gachon University, Seongnam 13120, South Korea

<sup>&</sup>lt;sup>f</sup>Center for Integrated Nanostructure Physics, Institute of Basic Science, Suwon 16419, South Korea

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<sup>‡</sup> These authors contributed equally to this study.

generation applications. High *zT* values of  $\sim$ 1.0 and  $\sim$ 1.4 were obtained at 800 K in micro-grained I- and Br-doped BiSbSe<sub>3</sub>, respectively.<sup>5,9</sup> One main origin of the high thermoelectric performance of these compounds was the simultaneous improvement of electronic (enlarged S) and thermal transport properties (reduced lattice thermal conductivity ( $\kappa_{lat}$ )) due to a structural transition.5 A phase transition triggered the convergence of conduction band resulting in largely increased density of states (DOS) effective mass  $(m_d^*)$ . Additionally,  $\kappa_{lat}$  was reduced due to the phonon softening and substantial lattice anharmonicity benefitting from weakened interchain interaction in orthorhombic phase. The high zTs were obtained by enhanced power factor  $(S^2\sigma)$  of I- and Br-doped BiSbSe<sub>3</sub> (by an order of magnitude), resulted from the increase in carrier concentration  $(n_c)$ . However, intrinsic drawbacks of I- and Brdoped BiSbSe<sub>3</sub>, which included solubility limit of I ( $\sim$ 3 at%) and Hall mobility  $(\mu_{\rm H})$  deterioration with Br doping, should be resolved to enhance the low  $\sigma$  value (<300 S cm<sup>-1</sup> at 300 K) in order to increase the efficiency of the thermoelectric power generation module. Moreover, the thermal stability and reproducibility of high-temperature thermoelectric performance was still elusive, demanding a clear criterion for the temperature limit of the module.

Chlorine (Cl) is a commonly used doping element especially at Se-site of various selenides such as In<sub>4</sub>Se<sub>3-x</sub>, PbSe, AgPb<sub>18</sub>-SbSe<sub>20</sub>, and SnSe<sub>2</sub> to increase  $n_{\rm c}$ .<sup>10-13</sup> Moreover, a large difference in atomic mass between Se ( $M_{\rm Se} = 78.971$ ) and Cl ( $M_{\rm Cl} =$ 35.45) is advantageous to further reduce  $\kappa_{lat}$  by mass-defect phonon scattering. In this work, we fabricated the polycrystalline bulks of Cl-doped  $Bi_{2-x}Sb_xSe_3$  (x = 0.8, 1.0) and evaluated their electronic and thermal transport properties in an effort to develop  $(Bi,Sb)_2Se_3$ -based alloys with high  $\sigma$  and zT, at the same time. We found that Cl was an effective doping element to facilitate the carrier transport in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>, thus high  $\mu_{\rm H}$  of ~27.3 cm<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup> was observed even in highly Cldoped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub> with high  $n_c$  (~9.0 × 10<sup>19</sup> cm<sup>-3</sup>). Compared to I-doped  $Bi_{1,2}Sb_{0,8}Se_{2,91}I_{0,09}$  ( $\sigma \sim 80$  S cm<sup>-1</sup> and zT~0.53 at 700 K),<sup>5</sup> higher  $\sigma$  (~165 S cm<sup>-1</sup> at 700 K) and higher zT(~0.67 at 700 K) were obtained in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub>. Moreover, we confirmed the thermal stability of the sample by a cyclic measurement.

#### 2. Experimental methods

High purity elements of bismuth (Bi shot, 99.99%, 5N Plus), antimony (Sb shot, 99.99%, 5N Plus), selenium (Se shot, 99.99%, iTASCO), and antimony trichloride (SbCl<sub>3</sub> crystalline, 99.99%, Sigma Aldrich) were used as starting raw materials. Stoichiometric (BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0, 0.12, 0.18, 0.24) and Bi<sub>1.2</sub>-Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0, 0.12, 0.18, 0.24, 0.3)) amount of these materials was weighted and sealed into a quartz tube under vacuum ( $\sim 10^{-3}$  Pa), and the mixtures were melted at 1173 K for 12 h. After melting, the quartz tubes were water quenched and annealed at 673 K for 48 h. The acquired ingots were pulverized into powders *via* BM, and compacted bulks (13 mm in diameter and 11 mm in thickness) were fabricated by using SPS for 2 min at 773 K under a pressure of 40 MPa.

Phase formation behavior in SPSed bulks was analyzed by Xray diffraction (XRD, SmartLab (9 kW), Rigaku, Japan) with CuK $\alpha$  radiation ( $\lambda = 1.5418$  Å). The microstructures of the fractured surface of the SPSed bulks were investigated by scanning electron microscopy (SEM, JSM-7600F, JEOL, Japan). The temperature dependences of S and  $\sigma$  were measured using a commercial measurement system (ZEM-3, Ulvac-Riko, Japan) from 300–700 K under a He atmosphere. The  $\kappa_{tot}$  (= $D \times C_p \times \rho$ , where D,  $C_{\rm p}$ , and  $\rho$  are the thermal diffusivity, specific heat capacity, and the density, respectively) was calculated from the separate measurement of D and  $\rho$ . Temperature-dependent D was measured by laser flash method (TC-1200RH, Ulvac-Riko, Japan) from 300–700 K under a vacuum and  $\rho$  was measured by the Archimedes principle (MD-300S, Alfa Miracle, Japan). Temperature dependence of  $C_{\rm p}$  was estimated from the reported values.<sup>5</sup> The rectangular bar-type sample (10 mm  $\times$ 3 mm  $\times$  3 mm) and square plate-type sample (10 mm  $\times$  10 mm  $\times$  1 mm) were cut in a plane perpendicular and parallel to the SPS press direction, respectively. In this manner, electronic (S,  $\sigma$ ) and thermal (D) transport properties can be measured in the same direction. The Hall coefficient  $(R_{\rm H})$  was measured by the van der Pauw method via a commercial Hall effect measurement system (8400 Series, LakeShore, USA) at room temperature. The  $n_c$  and  $\mu_H$  were calculated by  $n_c = e^{-1}R_H^{-1}$  and  $\mu_H =$  $\sigma R_{\rm H}$ .

#### 3. Results and discussion

In the present study, we selected two different matrixes; (1)  $BiSbSe_3$  with an orthorhombic phase and (2)  $Bi_{1,2}Sb_{0,8}Se_3$  with orthorhombic and rhombohedral phases.5 Fig. 1a shows the XRD patterns for  $BiSbSe_{3-\nu}Cl_{\nu}$  samples. All the peaks can be indexed as a pure orthorhombic structure of Sb<sub>2</sub>Se<sub>3</sub> without any secondary phases, suggesting the Cl substitution at Se-site. Structure factors including lattice parameters of BiSbSe<sub>3</sub>based compounds with orthorhombic phase (Fig. S1<sup>†</sup>) and those of Bi1.2Sb0.8Se3-based compounds with mixed (orthorhombic and rhombohedral) phases (Fig. S3<sup>†</sup>) were extracted by the Rietveld refinement (GSAS II suite) after refinements with different structural models at the condition of convergence with the best pattern match. The slight decrease in lattice parameters (a and c) of  $BiSbSe_3$  after the Cl doping is another evidence for Cl substitution due to the smaller ionic radius of Cl- (167 pm) when compared to that of  $Se^{2-}$  (184 pm) (Fig. S1<sup>†</sup>). On the other hand, as reported in the previous report,<sup>14</sup> both orthorhombic and rhombohedral phases are clearly detected in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> samples as shown in Fig. 1b. Peaks for the rhombohedral structure of  $Bi_2Se_3$  were observed at  $2\theta \sim 18.58^\circ$ and  $\sim 29.35^{\circ}$ . The strongest intensity of (402) indicates the preferred crystal orientation generated during the SPS process. Oriented grain structure is also found in SEM images for the fractured surface of SPSed Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (Fig. 1c and S2<sup>+</sup>). The mole fraction of rhombohedral phase estimated by Rietveld refinement is about 0.74, and this value does not show significant change with Cl doping contents (see the Table S1<sup>†</sup>). And slight decrease in cell volume by Cl-doping is observed in orthorhombic phase (Fig. S3<sup>†</sup>). Cl-related impurity phase

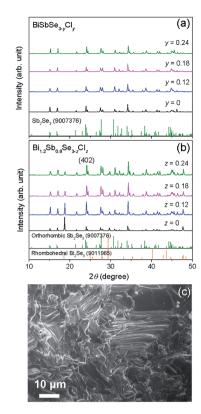


Fig.1 XRD patterns for the SPSed bulks of (a)  $BiSbSe_{3-y}Cl_y$  (y = 0, 0.12, 0.18, 0.24) and (b)  $Bi_{1.2}Sb_{0.8}Se_{3-z}Cl_z$  (z = 0, 0.12, 0.18, 0.24). (c) SEM image for the fractured surface of  $Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$ .

(Bi<sub>3</sub>Se<sub>4</sub>Cl) was observed in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.7</sub>Cl<sub>0.3</sub>, suggesting that the solubility limit of Cl for Se-site is about 8 at% in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> (Fig. S4<sup> $\dagger$ </sup>).

The temperature dependences of  $\sigma$  for both BiSbSe<sub>3-v</sub>Cl<sub>v</sub> (y = 0.12, 0.18, 0.24 and  $Bi_{1,2}Sb_{0,8}Se_{3-z}Cl_z$  (z = 0.12, 0.18, 0.24) samples are plotted in Fig. 2a. All thermoelectric transport properties ( $\sigma$ , S, and  $\kappa$ ) are measured perpendicular to SPS pressing direction since the electrical transport is dominant along the in-plane direction. The  $\sigma$  values of BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> are effectively increased by Cl-doping. Interestingly, the  $\sigma$  values of Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> are higher than those of Cl-doped BiSbSe3 in the whole measured temperature range. The  $\sigma$  values of the BiSbSe<sub>2.76</sub>Cl<sub>0.24</sub> are 132 S cm<sup>-1</sup> and 61.2 S cm<sup>-1</sup> at 300 K and 700 K, respectively, while those of  $Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$  are 397 S cm<sup>-1</sup> and 159 S cm<sup>-1</sup> at 300 K and 700 K, respectively. To clarify this, we estimated the  $n_{\rm c}$  and  $\mu_{\rm H}$  of both Cl-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> at 300 K (Fig. 2b). The improvement in  $\sigma$  by Cl-doping is resulted from the increase of  $\mu_{\rm H}$  as well as  $n_{\rm c}$  both in BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>. It is noted that  $\mu_{\rm H}$  values of Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> are much higher than those of Cl-doped BiSbSe<sub>3</sub>. The  $\mu_{\rm H}$  values of BiSbSe<sub>3-v</sub>Cl<sub>v</sub> (y = 0.12, 0.18, 0.24) at 300 K is ranged from 8.44 to 9.01 cm<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup>, whereas that of  $Bi_{1,2}Sb_{0.8}Se_{2.88}Cl_{0.12}$  is ~50.4 cm<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup>. Moreover, the  $\mu_{\rm H}$  value of highly Cl-doped BiSbSe<sub>2.76</sub>Cl<sub>0.24</sub> is retained in value about 27.3 cm<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup> despite of the high  $n_c$  $\sim$ 9.0  $\times$  10<sup>19</sup> cm<sup>-3</sup>. This high  $\mu_{\rm H}$  has been also reported in Idoped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>.5

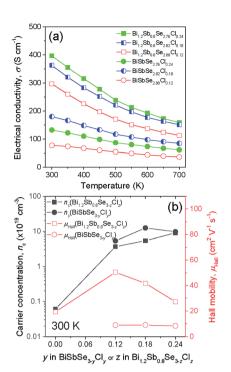
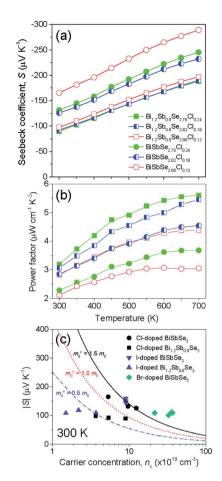


Fig. 2 (a) Temperature dependence of electrical conductivity and (b) carrier concentration and Hall mobility for  $BiSbSe_{3-y}Cl_y$  (y = 0.12, 0.18, 0.24) and  $Bi_{1.2}Sb_{0.8}Se_{3-z}Cl_z$  (z = 0.12, 0.18, 0.24).

Unexpected difference between electronic transport properties of Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> and those of I-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> is observed in *S*. Fig. 3a depicts the temperature dependences of *S* for both BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0.12, 0.18, 0.24) and Bi<sub>1.2</sub>Sb<sub>0.8</sub>-Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0.12, 0.18, 0.24) samples. The *S* values of the all samples are negative in the whole measured temperature range, indicating n-type semiconductors. The large |S| values of Cldoped BiSbSe<sub>3</sub> samples due to the convergence of conduction band by phase transition are well demonstrated both in I-doped and Br-doped BiSbSe<sub>3</sub>.<sup>5,9</sup> To investigate the change in band structure by Cl-doping especially in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>, we calculate the  $m_d^*$  by using measured *S* and  $n_c$  at 300 K based on the following eqn (1):<sup>1</sup>

$$S = \frac{8\pi^2 k_{\rm B}^2 T}{3eh^2} \left(\frac{\pi}{3n_{\rm c}}\right)^{2/3} m_{\rm d}^*, \tag{1}$$

where  $k_{\rm B}$ , e, and h are the Boltzmann constant, elementary charge, and Planck constant, respectively. The  $m_{\rm d}^*$  values are listed in Table 1 together with those for I-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> samples, which are estimated from eqn (1) by using previously reported data.<sup>5</sup> Large  $m_{\rm d}^*$  values of  $1.55m_0$  and  $1.67m_0$ are obtained both in BiSbSe<sub>2.82</sub>Cl<sub>0.18</sub> and BiSbSe<sub>2.91</sub>I<sub>0.09</sub>, respectively, with pure orthorhombic phase benefiting from converged electronic band valleys. However, the  $m_{\rm d}^*$  values of Cldoped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> and those of I-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> are smaller than those of Cl- and I-doped BiSbSe<sub>3</sub> mainly due to the large mole fraction of rhombohedral phase (Table S1<sup>†</sup>) with singly-degenerate conduction band. It is noted that  $m_{\rm d}^*$  value of Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub> reaches value about 0.90m<sub>0</sub>, which results



**Fig. 3** Temperature dependence of (a) Seebeck coefficient and (b) power factor for  $BiSbSe_{3-y}Cl_y$  (y = 0.12, 0.18, 0.24) and  $Bi_{1.2}Sb_{0.8}$ - $Se_{3-z}Cl_z$  (z = 0.12, 0.18, 0.24). (c) Pisarenko plot for Cl-, I-, and Br-doped BiSbSe<sub>3</sub> and Cl- and I-doped  $Bi_{1.2}Sb_{0.8}Se_3$  at 300 K.

in a large *S* even in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> systems. Fig. 3c shows the Pisarenko plots  $(n_c - |S|)$  for both BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0.12, 0.18, 0.24) and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0.12, 0.18, 0.24) samples at 300 K. Those for I-doped BiSbSe<sub>3</sub>, I-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>, and Br-doped BiSbSe<sub>3</sub> samples are also shown for comparison.<sup>5,9</sup>

As clearly shown in Fig. 3c, similar value of *S* is obtained in  $Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$  despite of the large increase in  $n_c$  when

**Table 1** The density-of-states effective mass  $(m_d^*)$  values of Cl-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>. Those of I-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>, which are estimated from the reported data ref. 5, are also shown for comparison

Compositions (nominal)	$m_{\rm d}^*$ $(m_0)$	Compositions (nominal)	$m_{\rm d}^*$ $(m_0)$
BiSbSe <sub>2.88</sub> Cl <sub>0.12</sub> BiSbSe <sub>2.82</sub> Cl <sub>0.18</sub>	1.17 1.55	BiSbSe <sub>2.97</sub> I <sub>0.03</sub> BiSbSe <sub>2.94</sub> I <sub>0.06</sub>	1.50 1.63
BiSbSe <sub>2.76</sub> Cl <sub>0.24</sub>	1.38	BiSbSe <sub>2.91</sub> I <sub>0.09</sub>	1.67
Bi <sub>1.2</sub> Sb <sub>0.8</sub> Se <sub>2.88</sub> Cl <sub>0.12</sub>	0.54	Bi <sub>1.2</sub> Sb <sub>0.8</sub> Se <sub>2.97</sub> I <sub>0.03</sub>	0.33
Bi <sub>1.2</sub> Sb <sub>0.8</sub> Se <sub>2.82</sub> Cl <sub>0.18</sub>	0.66	Bi <sub>1.2</sub> Sb <sub>0.8</sub> Se <sub>2.94</sub> I <sub>0.06</sub>	0.65
$Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$	0.90	$Bi_{1.2}Sb_{0.8}Se_{2.91}I_{0.09}$	0.43

compared to that of I-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> samples. Resultantly, a maximum power factor values of ~3.19  $\mu$ W cm<sup>-1</sup> K<sup>-2</sup> and ~5.61  $\mu$ W cm<sup>-1</sup> K<sup>-2</sup> at 300 K and 700 K, respectively, are obtained in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub> (Fig. 3b), which ensures the enhanced *zT* especially at higher temperatures. This beneficial characteristic feature for the realization of highly-efficient lowmid temperature thermoelectric power generation system is only found in Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> among other (Bi,Sb)<sub>2</sub>Se<sub>3</sub>based alloys.

Fig. 4a shows the temperature dependence of  $\kappa_{tot}$  for BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0.12, 0.18, 0.24) and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0.12, 0.18, 0.24) samples. The  $\kappa_{tot}$  values of Cl-doped Bi<sub>1.2</sub>-Sb<sub>0.8</sub>Se<sub>3</sub> are higher than those of Cl-doped BiSbSe<sub>3</sub>. The room temperature  $\kappa_{tot}$  values of both Cl-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>-Sb<sub>0.8</sub>Se<sub>3</sub> are  $\sim 0.58-0.62$  W m<sup>-1</sup> K<sup>-1</sup> and  $\sim 0.72-0.83$  W m<sup>-1</sup> K<sup>-1</sup>, respectively. This is considered to be related to the increased electronic contribution ( $\kappa_{ele}$ ) originated from the higher  $\sigma$  of Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>. On the other hand, as shown in Fig. 4a, the  $\kappa_{tot}$  of all the samples gradually decrease with temperature, suggesting the small contribution of bipolar thermal conduction ( $\kappa_{bp}$ ). We estimated the  $\kappa_{lat}$  and  $\kappa_{ele}$  by using the relationship of  $\kappa_{tot} = \kappa_{ele} + \kappa_{lat}$ . Details for the calculation are described in Section 6 of ESI.<sup>†</sup>

Fig. 4b shows the temperature dependence of  $\kappa_{\text{lat}}$  for both Cldoped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>. The temperature dependence of  $\kappa_{\text{lat}}$  shows roughly  $\kappa_{\text{lat}} \propto T^{-0.5}$ , indicating the additional point defect phonon scattering from the mass difference between host Se ( $M_{\text{Se}} = 78.971$ ) and dopant Cl ( $M_{\text{Cl}} = 35.45$ ) by Cl substituted at Se-site. The  $\kappa_{\text{lat}}$  values of Cl-doped BiSbSe<sub>3</sub> are lower than those of Cl-doped Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> mainly due to the soft bonding in orthorhombic phase, however,  $\kappa_{\text{lat}}$  reduction effect

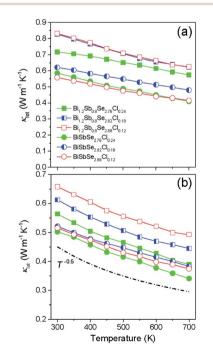


Fig. 4 Temperature dependence of (a) total thermal conductivity and (b) lattice thermal conductivity for BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0.12, 0.18, 0.24) and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0.12, 0.18, 0.24).

by Cl-doping is relatively small compared to that of Cl-doped  $Bi_{1.2}Sb_{0.8}Se_3$  due to cumulative phonon scattering by soft bonding and point defect. Thus the significantly reduced  $\kappa_{lat}$  (~0.56 W m<sup>-1</sup> K<sup>-1</sup> at 300 K and ~0.39 W m<sup>-1</sup> K<sup>-1</sup> at 700 K) is obtained in  $Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$  mainly due to the intensified mass-defect phonon scattering. The slightly higher  $\kappa_{lat}$  of  $BiSbSe_{2.82}Cl_{0.18}$  than that of  $BiSbSe_{2.88}Cl_{0.12}$  is considered to be related with the difference in preferred orientation (Fig. 1b).

Fig. 5a and b show the temperature dependent zT of  $BiSbSe_{3-\nu}Cl_{\nu}$  ( $\nu = 0.12, 0.18, 0.24$ ) and that of  $Bi_{1,2}Sb_{0,8}Se_{3-2}Cl_{z}$ (z = 0.12, 0.18, 0.24), respectively. The Cl-doping effectively enhances the zT both in BiSbSe3 and Bi1.2Sb0.8Se3 due to improvement of electronic and thermal transport properties. High zT of Cl-doped BiSbSe<sub>3</sub> with pure orthorhombic phase is mainly due to the enlarged  $m_d^*$  benefitting from the increased valley degeneracy and flattened band, which results in a larger S. Reduced  $\kappa_{lat}$  by the bond softening in orthorhombic phase is another origin for high zT of Cl-doped BiSbSe<sub>3</sub>. On the other hand, higher *zT* found in Cl-doped Bi<sub>1,2</sub>Sb<sub>0,8</sub>Se<sub>3</sub> despite of high rhombohedral phase fraction ( $\sim 0.74$ ) is attributed to the simultaneous improvement of electronic (enlarged  $m_d^*$  and improved  $\mu_{\rm H}$ ) and thermal (reduced  $\kappa_{\rm lat}$ ) transport properties by Cl-doping. The highly-reproducible maximum zT reaches in value about 0.68  $\pm$  0.04 at 700 K for three different Bi<sub>1.2</sub>Sb<sub>0.8</sub>- $Se_{2.76}Cl_{0.24}$  samples. Moreover, high  $\sigma$  values of 397 S cm<sup>-1</sup> at 300 K and 159 S  $cm^{-1}$  at 700 K make this material a promising candidate for practical applications.

We verify the thermal stability of  $Bi_{1.2}Sb_{0.8}Se_{2.76}Cl_{0.24}$  via the cyclic measurement of zT within temperature range from 300 K to 700 K (Fig. 6a) and remeasurement of power factor after annealing at 800 K for 10 h (Fig. 6b). Fig. 6a and b indicate that the Cl-doped  $Bi_{1.2}Sb_{0.8}Se_3$  alloys are chemically stable up to 700 K.

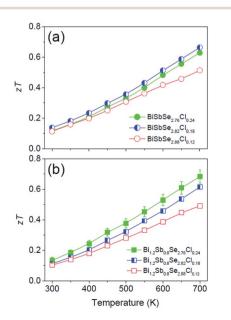


Fig. 5 Temperature dependence of dimensionless figure of merit zT for (a) BiSbSe<sub>3-y</sub>Cl<sub>y</sub> (y = 0.12, 0.18, 0.24) and (b) Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3-z</sub>Cl<sub>z</sub> (z = 0.12, 0.18, 0.24).

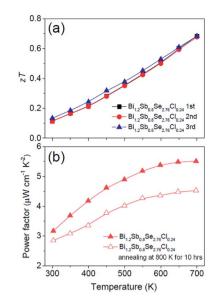


Fig. 6 (a) Cyclic measurement of temperature dependence of dimensionless figure of merit zT for Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub>. (b) Variation in temperature dependent power factor of Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>2.76</sub>Cl<sub>0.24</sub> after annealing at 800 K for 10 h.

#### 4. Conclusions

In summary, temperature dependent thermoelectric transport properties of Te-free Cl-doped BiSbSe<sub>3</sub> and Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> are systematically investigated. Improved Seebeck coefficient and electrical conductivity are simultaneously obtained due to the enlarged density-of-states effective mass by high content Cldoping (~8 at%), while maintaining intrinsic high mobility of Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub>-based alloys. This provides the optimized power factor of ~5.61  $\mu$ W cm<sup>-1</sup> K<sup>-2</sup> at 700 K. Additionally, despite of the weaker phonon scattering owing to decreased bond softening effect in Bi<sub>1.2</sub>Sb<sub>0.8</sub>Se<sub>3</sub> compared to that in BiSbSe<sub>3</sub>, lattice thermal conductivity is effectively reduced in value about 0.39 W m<sup>-1</sup> K<sup>-1</sup> at 700 K by 8 at% Cl-doping from the intensified mass-defect phonon scattering. This synergetic effect contributes to a high electrical conductivity of ~159 S cm<sup>-1</sup> and the high *zT* ~0.68 at 700 K.

#### Conflicts of interest

There are no conflicts to declare.

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