RSC Advances



View Article Online

View Journal | View Issue

PAPER

Check for updates

Cite this: RSC Adv., 2020, 10, 26686

Received 22nd May 2020 Accepted 10th July 2020

DOI: 10.1039/d0ra04541a

rsc.li/rsc-advances

1. Introduction

Topological materials such as a topological insulators, Dirac and Weyl semimetals, and axion insulators have attracted significant attention from physicists and chemists owing to their exciting and fascinating physical properties.¹⁻²⁹ Gapless surface states (Dirac-like linear dispersions) in a topological insulator are protected through a time reversal symmetry, which show a spin polarization to lock the spin along a direction perpendicular to the momentum (spin-momentum locking).¹⁻⁵ Here, it is noted that the bulk of a topological insulator possesses a band gap, as in a traditional insulator. The representative topological insulators are $Bi_{1-x}Sb_x$, Bi_2Se_3 , Bi_2Te_3 , and Sb_2Te_3 ,⁶⁻¹⁰ with Bi_2Se_3 being the most popular threedimensional (3D) topological insulator, having a single Dirac cone inside a band gap at the center of the Brillouin Zone (BZ).^{7,8}

Moreover, Dirac and Weyl semimetals have also elicited attention from physicists and chemists owing to their unique electronic structures, in which linear band crossings are in

A new protocol for the preparation of superconducting KBi₂

Huan Li,^a Yanan Wang,^a Yutaro Aoki,^a Saki Nishiyama,^a Xiaofan Yang,^a Tomoya Taguchi,^a Akari Miura,^a Ai Suzuki,^a Lei Zhi,^a Hidenori Goto,^a Ritsuko Eguchi,^a Takashi Kambe,^b Yen-Fa Liao,^c Hirofumi Ishii^c and Yoshihiro Kubozono ⁽⁾*^a

A superconducting KBi₂ sample was successfully prepared using a liquid ammonia (NH₃) technique. The temperature dependence of the magnetic susceptibility (M/H) showed a superconducting transition temperature (T_c) as high as 3.6 K. In addition, the shielding fraction at 2.0 K was evaluated to be 87%, *i.e.*, a bulk superconductor was realized using the above method. The T_c value was the same as that reported for the KBi₂ sample prepared using a high-temperature annealing method. An X-ray diffraction pattern measured based on the synchrotron X-ray radiation was analyzed using the Rietveld method, with a lattice constant, a, of 9.5010(1) Å under the space group of $Fd\overline{3}m$ (face-centered cubic, no. 227). The lattice constant and space group found for the KBi₂ sample using a liquid NH₃ technique were the same as those reported for KBi2 through a high-temperature annealing method. Thus, the superconducting behavior and crystal structure of the KBi₂ sample obtained in this study are almost the same as those for the KBi₂ sample reported previously. Strictly speaking, the magnetic behavior of the superconductivity was different from that of a KBi₂ sample reported previously, *i.e.*, the KBi₂ sample prepared using a liquid NH₃ technique was a type-II like superconductor, contrary to that prepared using a high-temperature annealing method, the reason for which is fully discussed. These results indicate that the liquid NH_3 technique is effective and simple for the preparation of a superconducting KBi2. In addition, the topological nature of the superconductivity for KBi₂ was not confirmed.

> fourfold/twofold degenerated points.11-15 A Dirac semimetal appears when both time-reversal symmetry and special inversion symmetry are maintained, whereas a Weyl semimetal appears when either the time-reversal symmetry or special inversion symmetry is broken. Namely, by breaking either the time-reversal symmetry or the special inversion symmetry, one Dirac fermion will transform into two Weyl fermions with opposite chirality in the BZ. The above semimetals are categorized as type-I or type-II depending on whether the Lorentz invariance is preserved. With type-I Dirac/Weyl semimetals, massless Dirac fermions (or Weyl fermions) should emerge at the Dirac and Weyl points,15-19 whereas in type-II Dirac/Weyl semimetals, they emerge at the topologically protected crossing points of the electron and hole pockets.17,20-22 Currently, many researchers are working on the condensed matter physics of Dirac/Weyl semimetals, owing to not only the above unique electronic properties, but also interesting physical properties such as a negative magnetoresistance,^{23,24} chiral magnetic effects,25 and quantum anomalous Hall (QAH) effect.26 Furthermore, recent interest in the complex interplay between topology and magnetism has led to a topological quantum state, which is indicated as an axion insulator.27,28 The attempt to realize an axion insulator began in the heterostructures of QAH

^aResearch Institute for Interdisciplinary Science, Okayama University, Okayama 700-8530, Japan. E-mail: kubozono@cc.okayama-u.ac.jp

^bDepartment of Physics, Okayama University, Okayama 700-8530, Japan

National Synchrotron Radiation Research Center, Hsinchu 30076, Taiwan

effect films,²⁷ and is currently realized in a stoichiometric material.²⁸

Alkali-Bi based compounds $(ABi_3 (A = Na, K, or Rb))$ have recently attracted significant attention because crystalline ABi₃ has been proposed as a Dirac semimetal in which bulk 3D Dirac points are protected by a crystal symmetry.^{15,29} Moreover, these materials were predicted to possess a nontrivial Fermi arc on the surface. In addition, various novel physical properties (giant diamagnetism, linear quantum magnetoresistance, and the QAH effect) were expected for ABi₃.²⁹ Since the above reports, the physical properties of a family of materials (alkaline and alkaline-earth metal-Bi compounds) have been investigated owing to an interest in the coupling of a Dirac semimetal and the superconductivity, *i.e.*, under the topological nature of the superconductivity, the particular superconducting properties of BaBi₃ were fully investigated.³⁰ In 2016, the superconductivity of the KBi₂ sample, which was a traditional superconductor,³¹ was fully studied.³² The detailed data of the superconductivity in KBi₂ were collected in this study. The motivation of this study is the interplay of the topological nature and the superconductivity. In addition, the strong spin orbit coupling (SOC) may affect the superconductivity in a metal-Bi compound because Bi is a heavy atom (atomic no. 83). All metal-Bi compounds were prepared using a high-temperature annealing method,30-35 for instance, a stoichiometric amount of Bi and K were mixed and sealed in a quartz tube. The tube was heated at more than 500 °C for several hours and slowly cooled. Finally, the crystals of KBi₂ were isolated using a centrifuge. Thus, the heating of metal and Bi is indispensable for the preparation of metal-Bi compounds, and isolation using a centrifuge is frequently required.

In this study, we report a new protocol for the preparation of KBi₂ in which a sample heating is not employed, *i.e.*, the liquid ammonia (NH₃) technique was applied to the sample preparation of a superconducting KBi2. The KBi2 sample obtained using liquid NH₃ showed a bulk superconductivity, and a clear crystalline powder was prepared. To characterize its superconductivity and crystal structure, the magnetic susceptibility (M/H)and X-ray diffraction (XRD) pattern of the obtained sample were measured; M and H refer to the magnetization and applied magnetic field, respectively. The superconducting transition temperature, T_c , and crystal structure of the KBi₂ sample prepared using liquid NH₃ were consistent with those of the sample prepared through a high-temperature annealing method. However, the magnetic behavior of the superconductivity was significantly different from that of the KBi₂ sample reported previously,32 which is fully discussed herein.

2. Experimental

Stoichiometric amounts of K and Bi were inserted into a glass tube in an Ar-filled glove box, and the space inside the glass tube was pumped to 10^{-2} torr. NH₃ gas was then collected in a glass tube and frozen. The K was dissolved in liquid NH₃ at less than -50 °C, and K atoms were introduced into the Bi crystallites, forming KBi₂. Finally, NH₃ was removed through vacuum distillation, and the removal of NH₃ was then carried out

through pumping under 10^{-2} torr at room temperature. Because the KBi₂ sample is quite sensitive to air, the handling of the sample was achieved without air exposure. Details of the reaction process are described later.

The temperature dependence of M/H of the obtained KBi_2 sample was measured using a SQUID magnetometer (Quantum Design, MPMS2 or MPMS–SQUID–VSM). The powder XRD pattern of the sample was measured at room temperature, using synchrotron radiation at BL12B2 of SPring-8, and the wavelength λ of the X-ray beam was 0.6859610 Å. The XRD pattern was analyzed using a Rietveld refinement.

3. Results and discussion

Fig. 1 shows the process of the sample preparation of KBi_2 . A photograph of the initial stage of the reaction between K and Bi is shown in Fig. 1(a), and K is dissolved in liquid NH₃. The color of the K-dissolved NH₃ solution is dark blue. After 3 days, the solution became transparent and black precipitate was obtained, as shown in Fig. 1(b). After removal of NH₃, a black powder sample was obtained, as shown in Fig. 1(c). The sample was then introduced into a quartz cell and glass capillary for *M*/*H* – *T* and XRD measurements, respectively, in an Ar-filled glove box. Here, it should be stressed that the most important point in the sample preparation of KBi₂ is to handle the sample without exposure to air to avoid a degradation of the sample.

Fig. 2(a) shows M/H - T plots of the KBi₂ sample obtained using the liquid NH₃ technique in zero-field cooling (ZFC) and field-cooling (FC) measurement modes; the value of H was 10 Oe. As shown in the inset of Fig. 2(a), the superconducting transition temperature, T_c , was determined to be 3.56 K from the crossing point between the dropped M/H - T plot and that under a normal state. In addition, the onset superconducting transition temperature, T_c^{onset} , was 3.6 K (Fig. 2(a)). The values of T_c and T_c^{onset} determined from the M/H - T plot in ZFC mode were the same as those from the M/H - T plot in FC mode. The values of T_c and T_c^{onset} were the same as those of the KBi₂ sample prepared using a high-temperature heating method.³² The shielding fraction at 2.0 K determined from the M/H - T plot in



Fig. 1 Process of KBi₂ sample formation. Photographs of (a) the first stage where K metal dissolves in liquid NH₃, and (b) the final stage where KBi₂ is formed. In the first stage, the color of the liquid NH₃ solution is dark blue, and in the final stage, the color of the solution is transparent. (c) Photograph of KBi₂ sample obtained after the removal of NH₃.



Fig. 2 (a) M/H - T plots recorded in ZFC and FC modes (H = 10 Oe), (b) M/H - T plots in ZFC modes at H = 10-250 Oe, (c) M - H loop recorded at 2.0 K and (d) $4\pi M - H$ plots at 2.0–3.6 K. The measured sample is KBi₂ prepared using a liquid NH₃ technique. The definition of T_c is shown in the inset of (a). The method for determining H_{c1} is shown in the inset of (d).

ZFC mode was evaluated to be 87%, indicating a bulk superconductivity.

The M/H - T plots at H = 10-250 Oe in ZFC mode are shown in Fig. 2(b), exhibiting a clear suppression of the superconducting transition by applying *H*. The T_{c}^{onset} value against each H is employed for depicting the H - T phase diagram (Fig. 3(c)), as described later. Fig. 2(c) shows the M - H loop of a KBi₂ sample prepared using a liquid NH₃ technique, recorded at 2.0 K, showing a clear hysteresis loop. This behavior indicates a type-II superconductor, which is different from that (type-I superconductor) reported previously for a KBi₂ crystal,32 which was prepared using a high-temperature annealing method. However, the M - H loop of the KBi₂ sample prepared using a liquid NH_3 technique (Fig. 2(c)) is similar to that of a α -PdBi₂ compound (type-II superconductor).36 As described later, the values of the lower critical field, H_{c1} , and the upper critical field, H_{c2} , were too low, similar to those of KBi₂.³²

Fig. 2(d) shows $4\pi M - H$ plots of a KBi₂ sample prepared using a liquid NH₃ technique, and recorded at 2.0–3.6 K. As shown in the inset of Fig. 2(d), the lower critical field, H_{c1} , was evaluated to be 24 Oe at 2.0 K, whereas the H_{c2} value was determined to be >500 Oe at 2.0 K; the value of H_{c1} was determined from the deviation from the linear line with a slope of -1.0 (as shown in the inset of Fig. 2(d)). A previous report concluded that the KBi₂ prepared using an annealing method was presumably a type-I superconductor.³² Actually, the M - Hloop with a small hysteresis for the KBi₂ sample³² was similar to other type-I superconductors such as YbSb₂ and LaRhSi₃.³⁷⁻³⁹ However, the M - H loop of KBi₂ prepared in this study (Fig. 2(c)) seems to be different from those of YbSb₂ and LaRhSi₃.

Herein, we should discuss why a type-II like superconducting behavior can be seen in the KBi₂ sample prepared using a liquid NH₃ technique. The first scenario is based on a type of impurity effect, and the second is based on KBi₂ prepared using liquid NH₃, creating an intrinsically bulk type-II superconductor. In a type-II superconductor, $\kappa \left(= \frac{\lambda}{\xi_{\rm GL}} \right)$ is larger than 1.0 $\left(\text{or } \kappa > \frac{1}{\sqrt{2}} \right)$, where $\xi_{\rm GL}(T) = 0.74 \frac{\xi_0}{(1-t)^{0.5}}$ for a clean limit, and $\xi_{\rm GL}(T) = 0.855 \frac{(\xi_0 l)^{0.5}}{(1-t)^{0.5}}$ for a dirty limit, at near $T_{\rm c}$; in addition, $t = T/T_{\rm c}$. Here, $\xi_{\rm GL}$, λ , ξ_0 , and l refer to the Ginzburg-Landau coherence length, magnetic penetration depth, Pippard coherence length, and mean-free path of a conduction electron,

concrete length, and mean-free path of a conduction electron, respectively.⁴⁰ Moreover, the magnetic penetration depth under a dirty limit, λ_{eff} (or λ), is expressed as $\lambda_{\text{eff}}(t) = \lambda_{\text{L}}(t) \left(\frac{\xi_0}{1.33l}\right)^{0.5}$;

in addition, $\lambda_{\rm L}(t)$ is London's penetration depth. Therefore, under a dirty limit, a smaller *l* must lead to a smaller $\xi_{\rm GL}$ and greater λ , yielding a greater κ , *i.e.*, a type-II superconductor. Actually, *l* is governed by the impurity, as indicated by the fact that the conversion of a type-I superconductor into a type-II superconductor is found in the In doping of Pb.⁴¹ To summarize, the polycrystalline powder sample of KBi₂ obtained using a liquid NH₃ technique may result in a small *l* because of the impurity effect. In addition, more detailed experiments such as specific heat measurements at H = 0 and $H \neq 0$ will be significant for determining either a type-I or type-II superconductor, as was achieved for a single crystal of KBi₂.³² In addition, the experiment results will simultaneously indicate whether the KBi₂ prepared using liquid NH₃ is a true bulk type-II superconductor, or the result of an impurity effect.

Fig. 3(a) shows the value of upper critical filed, H_{c2} , *versus T* plot in which H_{c2} value at each temperature was defined as *H* value providing 5% of the absolute value of minimum *M*, as indicated in inset of Fig. 3(a). In addition, the experimental M - H plots for the KBi₂ sample prepared in this study using high-temperature annealing substantially followed a linear line,³² and the value of H_c (~160 Oe)³² for the KBi₂ sample at 2.0 K is close to the critical field, H'_c (= 210 Oe), for KBi₂ at 2.0 K (see the inset of Fig. 3(a)).

Fig. 3(b) shows the normalized temperature $(t = T/T_c)$ dependence of the reduced critical field, h^* , where $h^*(T)$ at any temperature (T) is given by the equation, $[H_{c2}(T)]$

$$h^*(T) = \frac{\left\lfloor \overline{T_c} \right\rfloor}{\left\lceil -\frac{\mathrm{d}H_{c2}(T)}{\mathrm{d}T} \right\rceil_{T=T_c}} = \frac{H_{c2}(T)}{T_c \times \left\lceil -\frac{\mathrm{d}H_{c2}(T)}{\mathrm{d}T} \right\rceil_{T=T_c}} = \frac{H_{c2}(T)}{H_{c2}^{\mathrm{L}}(0)}.$$
 The

value of $H_{c2}(T)$ corresponds to that shown in Fig. 3(a), and the value of 1040 Oe was utilized for $H_{c2}^{L}(0)$; here, $H_{c2}^{L}(0)$ corresponds to $T_{\rm c} \times \left[-\frac{\mathrm{d}H_{c2}(T)}{\mathrm{d}T} \right]_{T=T_{\rm c}}$. It should be noted that the dirty limit superconductivity of an *s*-wave corresponds to the Werthamer-Helfand-

Hohenberg (WHH) model ($h^*(0) = 0.69$).^{42,43} In addition, the $h^*(0)$ value should reach 0.80–0.85 in the case of *p*-wave pairing.^{44–49} As shown in Fig. 3(b), it is difficult to uniquely conclude which model follows the $h^* - t$ plot between the *s*-wave and *p*-wave. The H_{c2} data at a temperature of lower than 2.0 K are required to achieve a unique

conclusion. However, the *p*-wave Cooper pair coupling of KBi₂ seems to lack a positive support. Thus, the superconductivity of KBi₂ may not comprise a topologically non-trivial nature because the Cooper pair symmetry will be a *p*-wave (with an odd parity) in the case of a topologically non-trivial superconductor.⁴⁶⁻⁴⁹ A more detailed study, including a theoretical approach, may be indispensable for pursuing its topological nature. In this study, we further discuss the H_{c2} value of the KBi₂ sample within the framework of an *p*-wave dirty limit model.^{42,43} The H_{c2} value was determined to be 720 Oe based on the WHH model,

$$H_{c2}(0) = -0.69 \ T_{c} \times \left[\frac{dH_{c2}}{dT} \right]_{T=T_{c}} = 0.69 \times H_{c2}^{L}(0), \quad \text{using} \quad \text{an}$$
$$H^{L}(0) \text{ where of } 1040 \text{ Opc} \text{ (Fig. 2(a))}$$

 $H_{c2}^{L}(0)$ value of 1040 Oe (Fig. 3(a)).

Moreover, although we tried to determine the value of H_{c2} at each temperature, $H_{c2}(T)$, from T_c^{onset} at each H (see Fig. 2(b)), and to depict an $h^* - t$ plot, it was difficult to evaluate T_c^{onset} at high H. Therefore, an $h^* - t$ plot was not created. Based on the WHH model^{42,43} $\left(H_{c2}(0) = -0.69 \ T_c \times \left[\frac{dH_{c2}}{dT}\right]_{T=T_c} = 0.69 \times H_{c2}^{L}(0)\right)$, the $H_{c2}(0)$ value was determined to be 620 Oe from a linear fitting

the $H_{c2}(0)$ value was determined to be 620 Oe from a linear fitting at near T_c for the $H_{c2} - T$ plot ($H_{c2}^L(0) = 900$ Oe in Fig. 3(c)). An $H_{c2}(0)$ value of 620 Oe may be reasonable owing to an agreement with 720 Oe, as evaluated from the $H_{c2} - T$ plot shown in Fig. 3(a). As shown in Fig. 3(d), the lower critical field ($H_{c1}(0)$) was 37 Oe, as determined from the fitting of the conventional formula, $H_{c1}(T) = H_{c1}(0) \left[1 - \left(\frac{T}{T_c}\right)^2\right]^{\gamma}$, in which γ is ~1.0. This empirical formula was employed for an evaluation of H_{c1} of various



Fig. 3 (a) $H_{c2} - T$, (b) $h^* - t$, (c) $H_{c2} - T$, and (d) $H_{c1} - T$ plots of KBi₂ sample prepared using liquid NH₃ technique. Definition of all parameters in (a)–(d) are fully shown in the text. The method for determining the H_{c2} value at each T is shown in the inset of (a). Inset of (c), the M/H - T plot at 10 Oe shown in Fig. 2(b) is provided to show the T_c^{onset} value, as an example. The value of T_c^{onset} was utilized to depict the $H_{c2} - T$ plot in (c).

This article is licensed under a Creative Commons Attribution 3.0 Unported Licence.



Fig. 4 XRD patterns of the KBi₂ sample prepared using a liquid NH₃ technique, together with the XRD pattern calculated through the Rietveld refinement. The experimental XRD pattern (black crosses) is shown in each graph, together with the pattern (red solid line) calculated. The red and blue sticks refer to the positions of the predicted Bragg reflections for KBi₂ (major phase) and Bi (minor phase), respectively. In this graph, the solid blue line corresponds to the difference between the experimental and calculated XRD patterns.

superconducting materials including $BaBi_3,\ CeRu_2,\ CaIr_2,\ and\ SrIr_2.^{30,50-52}$

In addition, we evaluated $\xi_{\rm GL}$ of KBi₂ to have a value of 73 nm using the expression,⁴¹ $H_{\rm c2}(0) = \frac{\Phi_0}{2\pi\xi_{\rm GL}^2}$ and the real $H_{\rm c2}(0)$ to be 620 Oe, where Φ_0 is 2.0678 × 10⁻⁷ G cm². By contrast, the λ value was evaluated as 220 nm using the formula, $H_{\rm c1}(0) = \frac{\Phi_0}{4\pi\lambda^2} \ln\left(\frac{\lambda}{\xi_{\rm GL}}\right)$,⁴⁰ and a $H_{\rm c1}(0)$ value of 37.0 Oe. The value of $\kappa\left(=\frac{\lambda}{\xi_{\rm GL}}\right)$ was ~3, which is much larger than $\frac{1}{\sqrt{2}}$, indicating a type-II superconductor, as described above.

Fig. 4 shows an XRD pattern of the KBi₂ sample prepared using a liquid NH₃ technique, along with the XRD pattern reproduced through a Rietveld refinement with a space group of $Fd\bar{3}m$ (face-centered cubic, no. 227). The lattice constant, a, was determined to be 9.5010(1) Å, which is consistent with that (9.5017(1) Å) from a Le Bail analysis (not shown) and a =9.5233(2) Å of KBi₂ prepared using a high-temperature annealing method.³² This phase can be assigned to KBi₂. Table 1 lists the atomic coordinates of KBi2 as determined through a Rietveld refinement. The actual stoichiometry of KBi₂ determined by a Rietveld refinement was K_{0.97(3)}Bi₂, which was evaluated from the occupancy of K and Bi atoms (see Table 1). The actual stoichiometry is the same as the nominal stoichiometry. In a Rietveld refinement, an additional minor phase that can be assigned to Bi ($R\bar{3}m$ (rhombohedral, no. 166) and a = 4.53440(8)Å and c = 11.8310(5) Å in a hexagonal structure) was included.

Table 1 Crystal structure of KBi₂ determined through Rietveld refinement using the space group, $Fd\overline{3}m$ (no. 227, choice 2). Lattice constant a = 9.5010(1) Å

	Occupancy	x	у	z	$B(\text{\AA}^2)$
16c Bi	1.0	0	0	0	0.72(5)
8b K	0.97(3)	0.375	0.375	0.375	1.5(6)



Fig. 5 Schematic representation of crystal structure of KBi_{2} , which was determined using a Rietveld refinement (Fig. 4(b)).

The lattice constants obtained are the same as those of the Bi metal.⁵³ In the Rietveld refinement, the experimental and calculated XRD patterns (Fig. 4) showed a good fit. The values of $R_{\rm p}$ and $wR_{\rm p}$ were 1.55% and 2.87%, respectively.

Herein, we discuss whether NH₃ is included in the KBi₂ sample prepared using a liquid NH₃ technique. The *a* value of KBi₂ prepared in this study is consistent with that of the KBi₂ sample (a = 9.5233(2) Å) prepared using a high-temperature annealing method.³² For instance, the *c* value of ammoniated K-doped FeSe ((NH₃)_yK_xFeSe) was 14.84–16.2 Å,^{54,55} which is larger than that of K_xFeSe, *i.e.*, 14.0367(7) Å, prepared using a high-temperature annealing method.³⁶ As a consequence, the KBi₂ sample prepared using the liquid NH₃ technique is exactly KBi₂ without NH₃. It can be concluded that the liquid NH₃ technique can exactly yield a KBi₂ sample in the same manner as a high-temperature annealing method.

A schematic representation of the crystal structure of KBi_2 depicted with the atomic coordinates determined from the Rietveld refinement for an XRD pattern (Fig. 4) is shown in Fig. 5. As shown in the projection from the [111] direction, the Bi atoms form a Kagomé net, which is the same as a previously reported crystal structure.³² The atoms of K are linked to the Bi atoms in a 3D network, as shown in Fig. 5. The crystal structure obtained is the same as that reported previously.³² As a consequence, throughout this study, it was demonstrated that the liquid NH₃ technique can easily produce a crystalline powder sample of KBi₂ in the same manner as used in a high-temperature annealing method. The value of T_c and the crystal structure are consistent with those previously reported.³²

4. Conclusions

A new protocol for the preparation of superconducting KBi_2 was completely verified through this study, *i.e.*, a liquid NH₃ technique is available for the preparation of a crystalline powder of superconducting KBi₂. The T_c value of KBi₂ prepared in this study was consistent with that reported previously.³² The crystal structure was also the same as that described elsewhere.³² Regardless of the preparation when using liquid NH₃, the inclusion of NH₃ or amide was not confirmed in the KBi₂ sample. Actually, the magnetic properties of a polycrystalline KBi₂ sample prepared using a liquid NH₃ technique were different from those of the single-crystal KBi₂ prepared through

View Article Online RSC Advances

a high-temperature annealing method.³² The reason why the former is a type-II superconductor and the latter is type-I remains questionable, although pursuing it may be extremely significant from the chemistry viewpoint of material synthesis as well as the physics perspective of a superconductor.

Because the liquid NH₃ technique is extremely effective and simple in terms of the preparation of KBi₂, various metal–Bi compounds may easily be fabricated using the technique, leading to an elucidation of their superconducting behavior and quantum/topological properties. Admittedly, this type of material is interesting owing to the suggested topological nature such as a Dirac semimetal.^{29,30} At the present stage, the topological nature of the superconducting KBi₂ sample was definitely not demonstrated. The pursuit of a topological nature in metal–Bi compounds must be continued; our approach regarding the preparation of a Bi-rich compound using a liquid NH₃ technique is currently being used to produce various alkali–Bi compounds such as LiBi, NaBi, RbBi₂, and CsBi₂ as superconductors.⁵⁷

To summarize, the development of an effective, simple, and exact preparation of metal–Bi compounds as described in this paper is significant for a detailed study on the superconducting behaviors of their compounds. The next target is to synthesize new superconducting metal–Bi compounds that have yet to be prepared using a high-temperature annealing method.

Author contributions

Huan Li, Yanan Wang, and Yutaro Aoki prepared the KBi₂ sample using a liquid NH₃ technique. In addition, Huan Li, Yanan Wang, and Yutaro Aoki carefully measured the magnetic behavior of the KBi₂ sample, under cooperation with Takashi Kambe. Huan Li, Yanan Wang, Xiaofan Yang, Tomoya Taguchi, Akari Miura, Ai Suzuki, Lei Zhi, and R. Eguchi measured the Xray diffraction patterns (XRDs) of the KBi₂ sample and analyzed the data using Le Bail and Rietveld analysis methods. The XRD measurement system using synchrotron radiation was designed by Yen-Fa Liao and Hirofumi Ishii, which allowed the suitable results of this study to be achieved. Yoshihiro Kubozono supervised all of the measurements and data analyses conducted in this study and prepared the manuscript under discussions with Hidenori Goto and Huan Li. Saki Nishiyama suggested the idea of this study, and the preliminary experiment was conducted by Yoshihiro Kubozono. All coauthors finally reviewed the manuscript.

Conflicts of interest

The authors declare no competing financial interests.

Acknowledgements

This study was partly supported by a Grant-in-Aid (17K05500, 18K04940, 18K18736, and 19H02676) from MEXT, and by the Program for Promoting the Enhancement of Research Universities. The XRD measurements at SPring-8 were supported by 2018B4140 and 2019A4131.

References

- 1 C. L. Kane and E. J. Mele, Phys. Rev. Lett., 2005, 95, 146802.
- 2 C. L. Kane and E. J. Mele, Phys. Rev. Lett., 2005, 95, 226801.
- 3 J. E. Moore and L. Balents, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, 75, 121306.
- 4 L. Fu and C. L. Kane, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2007, **76**, 045302.
- 5 M. Z. Hasan and C. L. Kane, Rev. Mod. Phys., 2010, 82, 3045.
- 6 A. A. Schafgans, K. W. Post, A. A. Taskin, Y. Ando, X.-L. Qi,
 B. C. Chapler and D. N. Basov, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, 85, 195440.
- 7 H. Zhang, C.-X. Liu, X.-L. Qi, X. Dai, Z. Fang and S.-C. Zhang, *Nat. Phys.*, 2009, 5, 438.
- 8 Y. Xia, D. Qian, D. Hsieh, L. Wray, A. Pal, H. Lin, A. Bansil, D. Grauer, Y. S. Hor, R. J. Cava and M. Z. Hasan, *Nat. Phys.*, 2009, 5, 398.
- 9 Y. L. Chen, J. G. Analytis, J.-H. Chu, Z. K. Liu, S.-K. Mo, X. L. Qi, H. J. Zhang, D. H. Lu, X. Dai, Z. Fang, S. C. Zhang, I. R. Fisher, Z. Hussain and Z.-X. Shen, *Science*, 2009, 325, 178.
- 10 D. Hsieh, Y. Xia, D. Qian, L. Wray, F. Meier, J. H. Dil, J. Osterwalder, L. Patthey, A. V. Fedorov, H. Lin, A. Bansil, D. Grauer, Y. S. Hor, R. J. Cava and M. Z. Hasan, *Phys. Rev. Lett.*, 2009, **103**, 146401.
- 11 X. Wan, A. M. Turner, A. Vishwanath and S. Y. Savrasov, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 205101.
- 12 A. A. Burkov, M. D. Hook and L. Balents, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 235126.
- 13 H. Weng, X. Dai and Z. Fang, J. Phys.: Condens. Matter, 2016, 28, 303001.
- 14 A. Bansil, H. Lin and T. Das, *Rev. Mod. Phys.*, 2016, 88, 021004.
- 15 Z. K. Liu, B. Zhou, Y. Zhang, Z. J. Wang, H. M. Weng, D. Prabhakaran, S.-K. Mo, Z. X. Shen, Z. Fang, X. Dai, Z. Hussain and Y. L. Chen, *Science*, 2014, 343, 864.
- M. Neupane, S.-Y. Xu, R. Sankar, N. Alidoust, G. Bian, C. Liu,
 I. Belopolski, T.-R. Chang, H.-T. Jeng, H. Liu, A. Bansil,
 F. Chou and M. Z. Hasan, *Nat. Commun.*, 2014, 5, 3786.
- 17 R. C. Xiao, P. L. Gong, Q. S. Wu, W. J. Lu, M. J. Wei, J. Y. Li, H. Y. Lv, X. Luo, P. Tong, X. B. Zhu and Y. P. Sun, *Phys. Rev. B*, 2017, **96**, 075101.
- 18 H. Weng, C. Fang, Z. Fang, B. A. Bernevig and X. Dai, *Phys. Rev. X*, 2015, 5, 011029.
- S.-Y. Xu, I. Belopolski, N. Alidoust, M. Neupane, G. Bian, C. Zhang, R. Sankar, G. Chang, Z. Yuan, C.-C. Lee, S.-M. Huang, H. Zheng, J. Ma, D. S. Sanchez, B. Wang, A. Bansil, F. Chou, P. P. Shibayev, H. Lin, S. Jia and M. Z. Hasan, *Science*, 2015, 349, 613.
- 20 A. A. Soluyanov, D. Gresch, Z. Wang, Q. Wu, M. Troyer, X. Dai and B. A. Bernevig, *Nature*, 2015, **527**, 495.
- 21 K. Deng, G. Wan, P. Deng, K. Zhang, S. Ding, E. Wang, M. Yan, H. Huang, H. Zhang, Z. Xu, J. Denlinger, A. Fedorov, H. Yang, W. Duan, H. Yao, Y. Wu, S. Fan, H. Zhang, X. Chen and S. Zhou, *Nat. Phys.*, 2016, 12, 1105.

- 22 K. Zhang, M. Yan, H. Zhang, H. Huang, M. Arita, Z. Sun, W. Duan, Y. Wu and S. Zhou, *Phys. Rev. B*, 2017, 96, 125102.
- 23 J. Xiong, S. K. Kushwaha, T. Liang, J. W. Krizan, M. Hirschberger, W. Wang, R. J. Cava and N. P. Ong, *Science*, 2015, 350, 413.
- 24 T. Liang, Q. Gibson, M. N. Ali, M. Liu, R. J. Cava and N. P. Ong, *Nat. Mater.*, 2014, 14, 280.
- 25 A. A. Zyuzin and A. A. Burkov, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **86**, 115133.
- 26 C.-X. Liu, P. Ye and X.-L. Qi, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **87**, 235306.
- 27 M. Mogi, M. Kawamura, R. Yoshimi, A. Tsukazaki, Y. Kozuka, N. Shirakawa, K. S. Takahashi, M. Kawasaki and Y. Tokura, *Nat. Mater.*, 2017, **16**, 516.
- 28 C. Liu, Y. Wang, H. Li, Y. Wu, Y. li, J. Li, K. He, Y. Xu, J. Zhang and Y. Wang, *Nat. Mater.*, 2020, **19**, 522–527.
- 29 Z. Wang, Y. Sun, X.-Q. Chen, C. Franchini, G. Xu, H. Weng, X. Dai and Z. Fang, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 195320.
- 30 N. Haldolaarachchige, S. K. Kushwaha, Q. Gibson and R. J. Cava, *Supercond. Sci. Technol.*, 2014, 27, 105001.
- 31 J. M. Reynolds and C. T. Lane, Phys. Rev., 1950, 79, 405.
- 32 S. Sun, K. Liu and H. Lei, *J. Phys.: Condens. Matter*, 2016, 28, 085701.
- 33 B. T. Mattias and J. K. Hulm, Phys. Rev., 1952, 87, 799.
- 34 K. Nishimura, T. Yasukawa and K. Mori, *Phys. B*, 2003, 329– 333, 1399–1400.
- 35 K. Górnicka, S. Gutowska, M. J. Winiarski, B. Wiendlocha, W. Xie, R. J. Cava and T. Klimczuk, *Chem. Mater.*, 2020, 32, 3150–3159.
- 36 Y. Zhou, X. Chen, C. An, Y. Zhou, L. Ling, J. Yang, C. Chen, L. Zhang, M. Tian, Z. Zhang and Z. Yang, *Phys. Rev. B*, 2019, **99**, 054501.
- 37 L. L. Zhao, S. Lausberg, H. Kim, M. A. Tanatar, M. Brando, R. Prozorov and E. Morosan, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, 85, 214526.
- 38 Y. Yamaguchi, S. Waki and K. Mitsugi, J. Phys. Soc. Jpn., 1987, 56, 419–420.
- 39 V. K. Anand, A. D. Hillier and D. T. Adroja, *Phys. Rev. B:* Condens. Matter Mater. Phys., 2011, 83, 064522.
- 40 M. Thinkham, *Introduction to Superconductivity*, McGraw-Hill Inc., New York, 1975.
- 41 C. Kittel, *Introduction to Solid State Physics*, John Willey & Sons, Inc., New York, 2005.

- 42 E. Helfand and N. R. Werthamer, Phys. Rev., 1966, 147, 288.
- 43 N. R. Werthamer, E. Helfand and P. C. Hohenberg, *Phys. Rev.*, 1966, **147**, 295.
- 44 K. Scharnberg and R. A. Klemm, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1980, **22**, 5233.
- 45 K. Maki, E. Puchkaryov, G.-F. Wang and H. Won, *Chin. J. Phys.*, 2000, **38**, 386.
- 46 K. Kirshenbaum, P. S. Syers, A. P. Hope, N. P. Butch, J. R. Jeffries, S. T. Weir, J. J. Hamlin, M. B. Maple, Y. K. Vohra and J. Paglione, *Phys. Rev. Lett.*, 2013, **111**, 087001.
- 47 Y. Zhou, X. Chen, R. Zhang, J. Shao, X. Wang, C. An, Y. Zhou, C. Park, W. Tong, L. Pi, Z. Yang, C. Zhang and Y. Zhang, *Phys. Rev. B*, 2016, **93**, 144514.
- 48 T. He, X. Yang, T. Terao, T. Uchiyama, T. Ueno, K. Kobayashi, J. Akimitsu, T. Miyazaki, T. Nishioka, K. Kimura, K. Hayashi, N. Happo, H. Yamaoka, H. Ishii, Y.-F. Liao, H. Ota, H. Goto and Y. Kubozono, *Phys. Rev. B*, 2018, **97**, 104503.
- 49 T. He, X. Yang, T. Taguchi, T. Ueno, K. Kobayashi,
 J. Akimitsu, H. Yamaoka, H. Ishii, Y.-F. Liao, H. Ota,
 H. Goto, R. Eguchi, K. Terashima, T. Yokoya,
 H. O. Jeschke, X. Wu and Y. Kubozono, *Phys. Rev. B*, 2019, 100, 094525.
- 50 M. Hedo, Y. Inada, E. Yamamoto, Y. Haga, Y. Onuki, Y. Aoki, T. D. Matsuda, H. Sato and S. Takahashi, *J. Phys. Soc. Jpn.*, 1998, **67**, 272–279.
- 51 N. Haldolaarachchige, Q. Gibson, L. M. Schoop, H. Luo and R. J. Cava, *J. Phys.: Condens. Matter*, 2015, **27**, 185701.
- 52 R. Horie, K. Horigane, S. Nishiyama, M. Akimitsu, K. Kobayashi, S. Onari, T. Kambe, Y. Kubozono and J. Akimitsu, *J. Phys.: Condens. Matter*, 2020, **32**, 175703.
- 53 B. P. Cucka and C. S. Barrett, Acta Crystallogr., 1962, 15, 865.
- 54 T. Ying, X. Chen, G. Wang, S. Jin, X. Lai, T. Zhou, H. Zhang, S. Shen and W. Wang, *J. Am. Chem. Soc.*, 2013, **135**, 2951– 2954.
- 55 L. Zheng, M. Izumi, Y. Sakai, R. Eguchi, H. Goto, Y. Takabayashi, T. Kambe, T. Onji, S. Araki, T. C. Kobayashi, J. Kim, A. Fujiwara and Y. Kubozono, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, 88, 094521.
- 56 J. Guo, S. Jin, G. Wang, S. Wang, K. Zhu, T. Zhou, M. He and X. Chen, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, 82, 180520.
- 57 H. Li, Y. Wang, A. Suzuki, T. Taguchi, H. Goto, R. Eguchi and Y. Kubozono, to be submitted.