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An efficient synthesis of 4,5-diaryl-3,4 dihydropyrimidin-2(1H)-one via a cesium carbonate-promoted direct condensation of 1 aryl-2-propanone with 1,1'-(arylmethylene)diurea†

Yi-Cong Guo[,](http://orcid.org/0000-0002-6088-7278)^a Xuan-Di Song,^a Wei Deng,^a Weidong Rao, D^b Haiyan Xu^{*c} a[n](http://orcid.org/0000-0002-3365-9288)d Zhi-Liang Shen **D**^{*a}

An efficient method for the synthesis of 4,5-diaryl-3,4-dihydropyrimidin-2(1H)-one by using 1,1'-(arylmethylene)diurea and 1-aryl-2-propanone as substrates was developed. The reactions proceeded efficiently in the presence of $Cs₂CO₃$ to give the desired products in moderate to good yields with wide substrate scope and good functional group tolerance, serving as an attractive alternative or complement to the previously reported methods for the facile assembly of biologically and pharmaceutically active 3,4-dihydropyrimidin-2(1H)-ones.

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1. Introduction

3,4-Dihydropyrimidin-2(1H)-ones (DHPMs) are pivotal structural skeletons which widely exist in a variety of natural products¹ and bioactive molecules.² These compounds exhibit a wide range of biologically and pharmaceutically relevant activities, such as antibacterial, antitumour, antiviral, antimalarial, antidiabetic, and antiepileptic activities.² Normally, 3,4-dihydropyrimidin-2(1H)-one could be accessed via the Biginelli reaction of aldehyde with ethyl acetoacetate and urea.3,4 With more than one century of development, tremendous achievements have been made in the Biginelli reaction, especially in the field of organic synthesis, medicinal chemistry, polymer chemistry, and material sciences.^{4,5} In addition, the variants of the Biginelli reaction by using other active methylene compounds (or their equivalents), such as 1,3 cyclohexanedione,⁶ acetophenone,⁷ 1-tetralone,⁸ β -oxodithioesters,⁹ enaminone,¹⁰ cyclopentanone,¹¹ alkyl aldehydes,¹² and alkynol,¹³ to replace ethyl acetoacetate for accessing diverse 3,4-dihydropyrimidin-2(1H)-ones have also been reported. Normally, the Biginelli or Biginelli-like reaction should be performed under acidic conditions in the presence of protic or Lewis acid.³⁻⁵ In 2010, Ji and co-workers reported an efficient method for the synthesis of 4,5,6-triaryl-3,4-

dihydropyrimidin-2(1H)-ones via a three-component Biginellitype condensation of aldehyde with 2-phenylacetophenone and urea/thiourea under basic conditions in the presence of t-BuOK.¹⁴ In the studies, it was observed that the reaction using urea as substrate might proceed *via* the formation of $1,1$ ['] (arylmethylene)diurea as reaction intermediate. Apart from this study, the direct use of pre-prepared $1,1'$ -(arylmethylene) diurea for the synthesis of 3,4-dihydropyrimidin-2(1H)-one has been rarely investigated. In continuation of our interests in the synthesis of heterocycles, herein we report a cesium carbonate-promoted¹⁵ direct condensation of 1-aryl-2-PAPER
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Table 1 Optimization of reaction conditions by using different bases

^{*a*} Isolated yield. ^{*b*} Using 0.2 equivalents of Cs₂CO₃.

a Institute of Advanced Synthesis, School of Chemistry and Molecular Engineering, Nanjing Tech University, Jiangsu 211816, China. E-mail: ias_zlshen@njtech.edu.cn b Jiangsu Provincial Key Lab for the Chemistry and Utilization of Agro-Forest Biomass, College of Chemical Engineering, Nanjing Forestry University, Nanjing 210037, China c School of Environmental and Chemical Engineering, Jiangsu University of Science and Technology, Zhenjiang, Jiangsu 212003, China. E-mail: xuhaiyanjurong@163.com

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Table 2 (Contd.)

 a Isolated yield. b The reaction was performed in t-BuOH instead of EtOH at $100\text{ }^{\circ}\text{C}$.

propanone with pre-synthesized 1,1'-(arylmethylene)diurea, leading to the corresponding 4,5-diaryl-3,4-dihydropyrimidin- $2(1H)$ -one in moderate to good yields.

2. Results and discussion

To begin with, we chose 1,1′-((4-chlorophenyl)methylene)diurea (1a) and 1-(4-chlorophenyl)propan-2-one (2a) as model substrates to optimize the reaction conditions by performing the reaction in ethanol at 70 \degree C for 24 h in the presence of different bases. As shown in Table 1, no desired 4,5-diaryl-3,4 dihydropyrimidin-2(1H)-one 3a was produced when the reaction was performed in the presence of organic bases (entries 1– 3). In sharp contrast, the use of inorganic bases (entries 4–7) as reaction promoters were found to promote the condensation with varying performance, with the highest yield (75% yield) being obtained by using Cs_2CO_3 as reaction promoter (entry 7). However, when only a catalytic amount of Cs_2CO_3 (0.2 equiv.) was employed in the reaction, the product yield decreased considerably (entry 8). It should be mentioned that, when Dcamphorsulfonic acid (CSA) was introduced into the reaction as a promoter, the corresponding product 3a could also be obtained, albeit in a relatively poor yield of 32%. In addition, it should be noted that an alternative method for the synthesis of product 3a via the three-component reaction involving 4 chlorobenzaldehyde (0.5 mmol, 1 equiv.), 1-(4-chlorophenyl) propan-2-one (2a, 1.2 equiv.), and urea (2 equiv.) proceeded sluggishly in the presence of Cs_2CO_3 (2 equiv.) in ethanol (70 $\rm{^{\circ}C},$ 24 h), leading to the corresponding product 3a only in 3% yield.

With $Cs₂CO₃$ being recognized as the optimum base for the condensation, subsequently we investigated the substrate scope of the reaction by using a range of diureas $1a-j$ as starting materials. As summarized in Table 2, a variety of aryl substituted diureas could efficiently undergo the intermolecular cyclization under the optimized conditions to afford the expected 4,5-diaryl-3,4-dihydropyrimidin-2(1H)-one 3a–j in 50–86% yields. In addition to phenyl-substituted diureas Table 3 Substrate scope study by using various ketones

1a–h, diureas 1i–j containing naphthenyl and thienyl substituents also efficiently participated in the organic transformation, leading to the anticipated products 3i and 3j in 64% and 85% yields, respectively. Especially noteworthy is that, the mild reaction conditions also allowed the reaction to proceed with the tolerance to various functional groups or substituents (e.g., halogen, nitro, cyano, methyl, and methoxy) in the phenyl ring, which could potentially be utilized at a late stage.

Encouraged by the above results, we continued to study the generality of the reaction by utilizing an array of phenylacetones 2b–i as substrates (Table 3). In all cases, the reactions proceeded efficiently under well-established conditions to deliver the expected products 4b–i in moderate to good yields. Apart from 1-aryl-2-propanones 2b– e bearing electron-withdrawing groups which well reacted with substrate 1a (entries 1–4), substrates 2f–h possessing electron-donating groups in the aryl ring of 1-aryl-2 propanones were also proven to be suitable candidates for the transformation, giving rise to the desired products 4f–h in excellent yields (85–90% yields; entries 5–7). In a same manner, 1-phenylbutan-2-one (2i) could also be applied in the protocol, furnishing the corresponding product 4i in 65% yield (entry 8). Moreover, the method could also be applied to the use of ethyl acetoacetate (2j) as substrate, albeit in a relatively low yield of 26% (entry 9). Analogously, the reaction also showed good compatibility to different functional groups or substituents, including trifluoromethyl, halogen, and methoxy group.

Mechanistically, the reaction might proceed via the initial formation of an imine-type intermediate 5, which subsequently reacts with 1-aryl-2-propanone 2 in the presence of Cs_2CO_3 leading to compound 6 (Scheme 1).¹⁴ Finally, an ensuing intramolecular cyclization followed by elimination of a hydroxyl group under basic conditions produces the desired product 3 and 4.

3. Conclusions

In summary, we have developed an efficient method for the synthesis of 4,5-diaryl-3,4-dihydropyrimidin-2(1H)-one by using 1,1⁰ -(arylmethylene)diurea and 1-aryl-2-propanone as substrates. The reactions proceeded efficiently in the presence of $Cs₂CO₃$, leading to the desired products in moderate to good yields. The transformation exhibited wide substrate scope and good tolerance to various important functional groups or substituents. We believed that the present protocol employing readily available and pre-synthesized diurea as substrate should serve as an attractive alternative or complement to the existing methods for the facile generation of biologically and pharmaceutically active 3,4-dihydropyrimidin- $2(1H)$ -ones. Paper
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4. Experimental

4.1 General information

1-Aryl-2-propanone, cesium carbonate, and ethanol were purchased from chemical companies and used directly without further purification (without the need of precautions to exclude air and moisture unless otherwise noted). Starting materials 1a– j were prepared according to reported methods.¹⁶ Analytical thin layer chromatography (TLC) was performed using silica gel plate (0.2 mm thickness). Subsequent to elution, plates were visualized using UV radiation (254 nm). Flash chromatography was performed using Merck silica gel (200–300 mesh) for column chromatography with freshly distilled solvents. Columns were typically packed as slurry and equilibrated with the appropriate solvent system prior to use. IR spectra were recorded on a FT-IR spectrophotometer using KBr optics. ¹H, 19 F, and 13 C NMR spectra were recorded in d⁶-DMSO on Jeol 400 MHz spectrometer. Tetramethylsilane (TMS) served as internal standard for ${}^{1}H$, ${}^{19}F$, and ${}^{13}C$ NMR analysis.

4.2 Experimental procedure

General procedure for the preparation of $1,1'$ -(arylmethylene)diurea:.¹⁶ To a 250 mL round-bottomed flask was sequentially added aryl aldehyde (30 mmol), urea (90 mmol), a catalytic amount of p-toluenesulfonic acid (3 mmol) and toluene (100 mL). The reaction mixture was refluxed for overnight. After reaction, the precipitate was collected by filtration, washed with saturated NaHCO₃ (10 mL), pure water (20 mL),

and Et_2O (20 mL). It was further dried in oven at 120 \degree C for half an hour to give the desired product of diurea.

Typical procedure for the condensation of $1,1'$ -(arylmethylene)diurea with 1-aryl-2-propanone. To a mixture of diurea (0.5 mmol) and 1-aryl-2-propanone (0.6 mmol) was added absolute ethanol (3 mL), and it was stirred at room temperature for 5 minutes. Then Cs_2CO_3 (1 mmol) was added and the reaction mixture was stirred vigorously at 70 $^{\circ} \mathrm{C}$ for 24 h. After the completion of the reaction, solvent was removed under vacuum. The residue was purified by silica gel column chromatography by using EtOAc/petroleum ether or $CH_2Cl_2/MeOH$ as eluant to afford the desired product of 4,5-biaryl-3,4-dihydropyrimidin-2(1H)-one.

4.3 Characterization data of products

 $1,1'$ -((4-Chlorophenyl)methylene)diurea (1a). 6.6 g. Yield $=$ 91%. White solid. Mp: 190.5-191.2 $^{\circ}$ C. ¹H NMR (400 MHz, DMSO- d_6): δ 7.40 (d, J = 8.5 Hz, 2H), 7.32 (d, J = 8.5 Hz, 2H), 6.79 $(d, J = 8.3 \text{ Hz}, 2\text{H})$, 6.08 $(t, J = 8.3 \text{ Hz}, 1\text{H})$, 5.72 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 157.8, 141.9, 131.6, 128.1, 128.0, 58.6 ppm. FTIR (KBr, neat): ν 3453, 3312, 1667, 1608, 866, 817 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₉H₁₂ClN₄O₂: 243.0643, found: 243.0643.

 $1,1'$ -((4-Bromophenyl)methylene)diurea (1b). 4.4 g. Yield $=$ 51%. White solid. Mp: 206.6-207.5 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 7.53 (d, J = 8.4 Hz, 2H), 7.26 (d, J = 8.4 Hz, 2H), 6.79 $(d, J = 8.3 \text{ Hz}, 2\text{H})$, 6.06 $(t, J = 8.3 \text{ Hz}, 1\text{H})$, 5.72 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 158.2, 142.8, 131.4, 128.8, 120.5, 59.1 ppm. FTIR (KBr, neat): ν 3453, 3312, 1667, 1608, 866, 813 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₉H₁₂BrN₄O₂: 287.0138, found: 287.0137.

 $1,1'$ -((2-Bromophenyl)methylene)diurea (1c). 6.3 g. Yield $=$ 73%. White solid. Mp: 199.2-199.7 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 7.60 (d, J = 7.8 Hz, 1H), 7.52 (d, J = 6.7 Hz, 1H), 7.40 $(t, J = 7.4 \text{ Hz}, 1H)$, 7.23 $(t, J = 7.0 \text{ Hz}, 1H)$, 6.71 $(d, J = 6.9 \text{ Hz},$ 2H), 6.25 (t, $J = 7.5$ Hz, 1H), 5.63 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 157.4, 141.2, 132.7, 129.4, 128.0, 127.5, 122.4, 59.6 ppm. FTIR (KBr, neat): ν 3430, 3331, 1668, 1612, 750, 719 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₉H₁₂BrN₄O₂: 287.0138, found: 287.0140.

 $1,1'$ -((4-Nitrophenyl)methylene)diurea (1d). 6.7 g. Yield $=$ 88%. White solid. Mp: 209.8-210.4 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.21 (d, J = 8.8 Hz, 2H), 7.56 (d, J = 8.6 Hz, 2H), 6.98 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 6.17 \text{ (t, } J = 8.2 \text{ Hz}, 1\text{H}), 5.82 \text{ (s, 4H)} \text{ ppm}.$ ¹³C NMR (100 MHz, DMSO-d₆): δ 157.8, 150.8, 146.5, 127.4, 123.4, 58.8 ppm. FTIR (KBr, neat): v 3459, 3312, 1682, 1516, 1351, 856, 835 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₉H₁₂N₅O₄: 254.0884, found: 254.0881.

 $1,1'$ -((4-Cyanophenyl)methylene)diurea (1e). 6.3 g. Yield $=$ 90%. White solid. Mp: 222.6-222.9 °C. ¹H NMR (400 MHz, DMSO- d_6 : δ 7.82 (d, $J = 8.1$ Hz, 2H), 7.49 (d, $J = 8.3$ Hz, 2H), 6.90 $(d, J = 8.2 \text{ Hz}, 2\text{H})$, 6.12 $(t, J = 8.1 \text{ Hz}, 1\text{H})$, 5.76 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 158.2, 149.1, 132.6, 127.6, 119.5, 110.1, 59.3 ppm. FTIR (KBr, neat): ν 3352, 3300, 2234, 1656, 862, 825 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₀H₁₂N₅O₂: 234.0986, found: 234.0987.

 $1,1'$ -(Phenylmethylene)diurea (1f). 4.0 g. Yield $=64\%$. White solid. Mp: 190.6–191.2 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 7.34– 7.31 (m, 4H), 7.27-7.23 (m, 1H), 6.72 (d, $J = 8.3$ Hz, 2H), 6.12 (t, J $= 8.3$ Hz, 1H), 5.68 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): d 158.2, 143.2, 128.6, 127.6, 126.5, 59.6 ppm. FTIR (KBr, neat): n 3420, 3312, 1667, 1535, 747, 699 cm $^{-1}$. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₉H₁₃N₄O₂: 209.1033, found: 209.1033.

1,1'-(p-Tolylmethylene)diurea (1g). 3.7 g. Yield $=$ 55%. White solid. Mp: 196.2–196.8 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 7.20 $(d, J = 8.0 \text{ Hz}, 2\text{H}), 7.14 (d, J = 8.1 \text{ Hz}, 2\text{H}), 6.67 (d, J = 8.1 \text{ Hz},$ 2H), 6.08 (t, $J = 8.3$ Hz, 1H), 5.67 (s, 4H), 2.28 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): δ 158.2, 140.2, 136.6, 129.1, 126.4, 59.4, 21.2 ppm. FTIR (KBr, neat): ν 3458, 3312, 1678, 1532, 864, 814 cm⁻¹. HRMS (ESI, *m*/z): $[M + H]^+$, calcd for C₁₀H₁₅N₄O₂: 223.1190, found: 223.1191.

1,1'-((4-Methoxyphenyl)methylene)diurea (1h). 5.2 g. Yield $=$ 73%. White solid. Mp: 186.6–187.3 $^{\circ}$ C. ¹H NMR (400 MHz, DMSO- d_6): δ 7.24 (d, J = 8.6 Hz, 2H), 6.90 (d, J = 8.7 Hz, 2H), 6.70 $(d, J = 8.0 \text{ Hz}, 2\text{H})$, 6.09 $(t, J = 8.1 \text{ Hz}, 1\text{H})$, 5.70 $(s, 4\text{H})$, 3.73 $(s,$ 3H) ppm. 13 C NMR (100 MHz, DMSO- d_6): δ 158.9, 158.3, 135.1, 127.7, 114.0, 59.2, 55.6 ppm. FTIR (KBr, neat): ν 3428, 3308, 1671, 1515, 1261, 1176, 865, 836 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₀H₁₅N₄O₃: 239.1139, found: 239.1134.

 $1,1'$ -(Naphthalen-1-ylmethylene)diurea (1i). 6.8 g. Yield $=$ 88%. White solid. Mp: 221.5-222.7 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.02 (d, J = 8.0 Hz, 1H), 7.96 (d, J = 7.4 Hz, 1H), 7.87 $(d, J = 8.1$ Hz, 1H), 7.64–7.47 (m, 4H), 6.91 (t, $J = 7.8$ Hz, 1H), 6.82 (d, $J = 7.7$ Hz, 2H), 5.66 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO-d6): d 157.4, 138.1, 133.5, 130.3, 128.6, 128.0, 126.3, 125.8, 125.2, 123.4, 122.6, 56.5 ppm. FTIR (KBr, neat): ν 3418, 3297, 1667, 1608, 882, 770 cm⁻¹. HRMS (ESI, *m*/z): $[M + H]^+$, calcd for $C_{13}H_{15}N_4O_2$: 259.1190, found: 259.1189.

 $1,1'$ -(Thiophen-2-ylmethylene)diurea (1j). 4.2 g. Yield $= 65\%$. White solid. Mp: 197.4–198.9 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 7.41–7.35 (m, 1H), 6.96 (dd, $J = 5.0$, 3.5 Hz, 1H), 6.91–6.88 (m, 1H), 6.86 (d, $J = 8.5$ Hz, 2H), 6.32 (t, $J = 8.4$ Hz, 1H), 5.75 (s, 4H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): δ 157.5, 147.7, 126.8, 124.9, 123.7, 56.2 ppm. FTIR (KBr, neat): ν 3467, 3284, 1671, 1589, 1282, 1123 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for $C_7H_{11}N_4O_2S: 215.0597$, found: 215.0602.

4,5-bis(4-Chlorophenyl)-6-methyl-3,4-dihydropyrimidin-

 $2(1H)$ -one (3a). 124.6 mg. Yield = 75%. Yellow solid. Mp: 214.7– 215.8 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.50 (s, 1H), 7.38 (s, 1H), 7.35–7.31 (m, 2H), 7.30–7.26 (m, 2H), 7.18–7.14 (m, 2H), 7.11–7.06 (m, 2H), 5.13 (s, 1H), 1.73 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 152.8, 143.0, 136.7, 131.8, 131.2, 130.9, 130.8, 128.7, 128.4, 128.1, 106.8, 58.5, 16.2 ppm. FTIR (KBr, neat): v 3265, 1693, 1491, 1241, 1091, 758 cm^{-1} . HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₇H₁₅Cl₂N₂O: 333.0556, found: 333.0558.

4-(4-Bromophenyl)-5-(4-chlorophenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-one (3b). 161.6 mg. Yield = 86%. Brown solid. Mp: 206.8–208.2 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.51 $(s, 1H)$, 7.46 $(d, J = 8.3 \text{ Hz}, 2H)$, 7.39 $(s, 1H)$, 7.30–7.24 $(m, 2H)$, 7.14–7.05 (m, 4H), 5.12 (s, 1H), 1.73 (s, 3H) ppm. 13 C NMR (100) MHz, DMSO- d_6): δ 152.8, 143.4, 136.7, 131.3, 131.2, 130.9, 130.8, 129.1, 128.1, 120.4, 106.7, 58.6, 16.2 ppm. FTIR (KBr, neat): v

3268, 1693, 1489, 1241, 757 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for $C_{17}H_{15}BrClN_2O$: 377.0051, found: 377.0051.

4-(2-Bromophenyl)-5-(4-chlorophenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-one (3c). 126.8 mg. Yield = 67% . Yellow solid. Mp: 189.8–191.3 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.56 $(s, 1H)$, 7.49 $(d, J = 7.6 \text{ Hz}, 1H)$, 7.43 $(d, J = 8.0 \text{ Hz}, 1H)$, 7.40– 7.33 (m, 2H), 7.24 (d, $J = 8.2$ Hz, 2H), 7.13 (t, $J = 7.2$ Hz, 1H), 7.01 $(d, J = 8.3$ Hz, 2H), 5.58 (s, 1H), 1.69 (s, 3H) ppm. ¹³C NMR $(100 \text{ MHz}, \text{ DMSO-}d_6)$: δ 152.5, 142.6, 136.2, 132.6, 131.3, 131.1, 131.0, 129.9, 129.5, 128.5, 128.1, 121.9, 106.3, 58.6, 16.1 ppm. FTIR (KBr, neat): ν 3220, 1694, 1493, 1253, 1095, 737 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₇H₁₅BrClN₂O: 377.0051, found: 377.0050. **PSC** Advances Articles. Published on 14 August 2020. The Common Co

5-(4-Chlorophenyl)-6-methyl-4-(4-nitrophenyl)-3,4-dihy-

dropyrimidin-2(1H)-one (3d). 88.8 mg. Yield $= 50\%$. Brown solid. Mp: 200.2–201.7 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.51 $(s, 1H)$, 7.44–7.37 (m, 3H), 7.35–7.30 (m, 2H), 7.16 (d, $J = 8.3$ Hz, 2H), 7.02 (d, $J = 8.3$ Hz, 2H), 5.13 (s, 1H), 1.73 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): δ 152.8, 148.5, 136.6, 131.1, 131.0, 130.8, 128.1, 126.6, 125.3, 124.4, 107.4, 54.2, 16.3 ppm. FTIR (KBr, neat): ν 2926, 1721, 1491, 1262, 1093, 1014, 831 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₇H₁₅ClN₃O₃: 344.0796, found: 344.0799.

4-(5-(4-Chlorophenyl)-6-methyl-2-oxo-1,2,3,4-

tetrahydropyrimidin-4-yl)benzonitrile (3e). 80.3 mg. Yield $=$ 50%. Yellow solid. Mp: 249.3-250.6 °C. ¹H NMR (400 MHz, DMSO-d6): d 8.57 (s, 1H), 7.78–7.72 (m, 2H), 7.47 (s, 1H), 7.33– 7.30 (m, 2H), 7.30–7.26 (m, 2H), 7.11–7.06 (m, 2H), 5.24 (s, 1H), 1.73 (s, 3H) ppm. 13 C NMR (100 MHz, DMSO- d_6): δ 152.7, 149.3, 136.4, 132.6, 131.3, 131.2, 130.9, 128.2, 127.8, 118.8, 110.1, 106.2, 58.8, 16.2 ppm. FTIR (KBr, neat): ν 3397, 2226, 1679, 1475, 1242, 759 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for $C_{18}H_{15}C/N_3O: 324.0898$, found: 324.0904.

5-(4-Chlorophenyl)-6-methyl-4-phenyl-3,4-dihydropyrimidin- $2(1H)$ -one (3f). 97.4 mg. Yield = 65%. Light yellow solid. Mp: 179.6–180.1 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.45 (s, 1H), 7.34 $(s, 1H)$, 7.29–7.23 (m, 4H), 7.22–7.18 (m, 1H), 7.16 (d, $J = 7.9$ Hz, 2H), 7.10–7.05 (m, 2H), 5.08 (s, 1H), 1.74 (s, 3H) ppm. 13C NMR $(100 \text{ MHz}, \text{ DMSO-}d_6)$: δ 153.0, 144.0, 137.0, 131.1, 130.7, 130.7, 128.5, 128.1, 127.3, 126.8, 107.1, 59.2, 16.3 ppm. FTIR (KBr, neat): ν 3055, 1690, 1493, 1014, 744, 698 cm $^{-1}$. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₇H₁₆ClN₂O: 299.0946, found: 299.0947.

5-(4-Chlorophenyl)-6-methyl-4-(p-tolyl)-3,4-dihydropyrimidin-2(1H)-one (3g). 118.2 mg. Yield = 76%. Light yellow solid. Mp: 215.6–216.8 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.44 (s, 1H), 7.30–7.24 (m, 3H), 7.10–7.03 (m, 6H), 5.04 (s, 1H), 2.23 (s, 3H), 1.74 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): δ 152.9, 141.1, 137.1, 136.4, 131.1, 130.6, 130.5, 129.0, 128.1, 126.8, 107.2, 58.9, 20.7, 16.3 ppm. FTIR (KBr, neat): ν 3271, 1694, 1491, 1241, 832, cm^{-1} . HRMS (ESI, m/z): $[M + H]^+$, calcd for $C_{18}H_{18}C/N_2O: 313.1102$, found: 313.1102.

5-(4-Chlorophenyl)-4-(4-methoxyphenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-one (3h). 120.0 mg. Yield $= 73\%$. Yellow solid. Mp: 161.8–163.2 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.42 (s, 1H), 7.29–7.24 (m, 3H), 7.10–7.06 (m, 4H), 6.85–6.80 (m, 2H), 5.02 (s, 1H), 3.69 (s, 3H), 1.74 (s, 3H) ppm. 13C NMR (100 MHz, DMSO-d₆): δ 158.5, 152.9, 137.1, 136.2, 131.1, 130.6, 130.5, 128.1, 128.0, 113.8, 107.3, 58.6, 55.0, 16.3 ppm. FTIR (KBr, neat): ν 3238, 1679, 1510, 1248, 1174, 834 cm $^{-1}$. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₈H₁₈ClN₂O₂: 329.1051, found: 329.1054.

5-(4-Chlorophenyl)-6-methyl-4-(naphthalen-1-yl)-3,4-dihydropyrimidin-2(1H)-one (3i). 112.3 mg. Yield = 64%. Light yellow solid. Mp: 250.4–251.6 °C. 1 H NMR (400 MHz, DMSO- d_{6}): d 8.58 (s, 1H), 8.29–8.19 (m, 1H), 7.92–7.86 (m, 1H), 7.82–7.75 $(m, 1H)$, 7.53–7.40 $(m, 4H)$, 7.36 (s, 1H), 7.15 (d, $J = 7.8$ Hz, 2H), 7.07 (d, $J = 8.0$ Hz, 2H), 5.99 (s, 1H), 1.81 (s, 3H) ppm. ¹³C NMR $(100 \text{ MHz}, \text{ DMSO-}d_6): \delta$ 152.7, 148.0, 144.0, 138.7, 136.9, 133.7, 131.1, 131.0, 130.6, 130.4, 128.6, 128.1, 127.9, 126.0, 125.9, 125.6, 123.6, 106.8, 16.3 ppm. FTIR (KBr, neat): v 3230, 1703, 1493, 1253, 776 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₂₁H₁₈ClN₂O: 349.1102, found: 349.1103.

5-(4-Chlorophenyl)-6-methyl-4-(thiophen-2-yl)-3,4-dihy-

dropyrimidin-2(1H)-one (3j). 128.8 mg. Yield = 85%. Light yellow solid. Mp: 193.2–194.4 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.63 (s, 1H), 7.55 (s, 1H), 7.35 (dd, $J = 5.0$, 1.1 Hz, 1H), 7.33– 7.28 (m, 2H), 7.21–7.14 (m, 2H), 6.87 (dd, $J = 5.0$, 3.5 Hz, 1H), 6.82–6.77 (m, 1H), 5.39 (s, 1H), 1.76 (s, 3H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 152.9, 148.5, 136.6, 131.1, 131.1, 130.8, 128.1, 126.6, 125.3, 124.4, 107.5, 54.2, 16.3 ppm. FTIR (KBr, neat): v 3244, 1686, 1490, 1389, 1247, 1091, 839, 698 cm^{-1} . HRMS (ESI, m/z : $[M + H]^+$, calcd for C₁₅H₁₄ClN₂OS: 305.0510, found: 305.0509.

4-(4-Chlorophenyl)-6-methyl-5-(3-(triuoromethyl)phenyl)-

3,4-dihydropyrimidin-2(1H)-one (4b). 132.6 mg. Yield = 72%. Yellow solid. Mp: 109.8–110.4 $^{\circ}$ C. 1 H NMR (400 MHz, DMSO- d_{6}): d 8.60 (s, 1H), 7.51–7.42 (m, 3H), 7.40–7.31 (m, 4H), 7.23–7.14 (m, 2H), 5.22 (s, 1H), 1.74 (s, 3H) ppm. 13C NMR (100 MHz, DMSO-d6): d 152.7, 142.9, 139.0, 133.6, 131.9, 131.7, 129.2, 129.0 $(q, J = 31.3 \text{ Hz})$, 128.8, 128.5, 125.7 $(q, J = 3.8 \text{ Hz})$, 124.2 $(q, J = 3.8 \text{ Hz})$ 271.0 Hz), 123.0 (q, $J = 4.1$ Hz), 106.6, 58.4, 16.2 ppm. ¹⁹F NMR (376 MHz, DMSO- d_6): δ -60.98 ppm. FTIR (KBr, neat): ν 3238, 1683, 1490, 1339, 1126, 804, 704 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₈H₁₅ClF₃N₂O: 367.0820, found: 367.0822.

5-(4-Bromophenyl)-4-(4-chlorophenyl)-6-methyl-3,4-dihy-

dropyrimidin-2(1H)-one (4c). 149.6 mg. Yield = 79%. Light yellow solid. Mp: 209.5–209.8 °C. 1 H NMR (400 MHz, DMSO- d_{6}): d 8.51 (s, 1H), 7.42–7.37 (m, 3H), 7.35–7.30 (m, 2H), 7.18–7.14 (m, 2H), 7.04–7.00 (m, 2H), 5.12 (s, 1H), 1.73 (s, 3H) ppm. 13C NMR (100 MHz, DMSO-d₆): δ 152.8, 143.0, 137.1, 131.8, 131.5, 131.0, 130.9, 128.7, 128.4, 119.3, 106.8, 58.4, 16.2 ppm. FTIR (KBr, neat): ν 3269, 1692, 1490, 1240, 1099, 844, 778 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₇H₁₅BrClN₂O: 377.0051, found: 377.0050.

4-(4-Chlorophenyl)-5-(4-fluorophenyl)-6-methyl-3,4-dihy-

dropyrimidin-2(1H)-one (4d). 139.5 mg. Yield = 88%. Yellow solid. Mp: 189.0–190.2 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.46 (s, 1H), 7.35 (s, 1H), 7.34–7.31 (m, 2H), 7.18–7.14 (m, 2H), 7.11– 7.02 (m, 4H), 5.10 (s, 1H), 1.71 (s, 3H) ppm. 13C NMR (100 MHz, DMSO- d_6): δ 160.6 (d, J = 241.7 Hz), 152.8, 143.0, 134.1 (d, J = 3.3 Hz), 131.7, 131.3 (d, $J = 8.0$ Hz), 130.4, 128.7, 128.4, 115.0 (d, $J = 21.1$ Hz), 107.0, 58.8, 16.2 ppm. ¹⁹F NMR (376 MHz, DMSO d_6 : δ -115.91 ppm. FTIR (KBr, neat): v 3259, 1694, 1491, 1242, 1094, 1014, 845, 780 cm⁻¹. HRMS (ESI, *m*/z): [M + H]⁺, calcd for C₁₇H₁₅ClFN₂O: 317.0851, found: 317.0851.

5-(3-Chlorophenyl)-4-(4-chlorophenyl)-6-methyl-3,4-dihy-

dropyrimidin-2(1H)-one (4e). 155.9 mg. Yield = 94%. Yellow solid. Mp: 93.8–95.3 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.59 (s, 1H), 7.45 (s, 1H), 7.32–7.29 (m, 2H), 7.22–7.15 (m, 4H), 7.11 (s, 1H), 7.04–6.98 (m, 1H), 5.17 (s, 1H), 1.75 (s, 3H) ppm. 13C NMR $(100 \text{ MHz}, \text{ DMSO-}d_6)$: δ 153.1, 142.9, 140.2, 133.0, 132.1, 131.5, 130.1, 129.1, 128.9, 128.6, 128.3, 126.4, 107.0, 58.6, 16.4 ppm. FTIR (KBr, neat): ν 3236, 1682, 1489, 1239, 1091, 785 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₇H₁₅Cl₂N₂O: 333.0556, found: 333.0556.

4-(4-Chlorophenyl)-5-(4-methoxyphenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-one (4f). 140.5 mg. Yield = 85%. Yellow solid. Mp: 184.6–186.2 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.42 (s, 1H), 7.35–7.29 (m, 3H), 7.19–7.14 (m, 2H), 6.99–6.94 (m, 2H), 6.82–6.76 (m, 2H), 5.07 (s, 1H), 3.68 (s, 3H), 1.71 (s, 3H) ppm. 13 C NMR (100 MHz, DMSO- d_6): δ 157.6, 153.1, 143.3, 131.7, 130.5, 129.8, 129.6, 128.7, 128.3, 113.6, 107.7, 58.9, 54.9, 16.2 ppm. FTIR (KBr, neat): ν 3259, 2924, 1693, 1489, 1249, 842 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₈H₁₈ClN₂O₂: 329.1051, found: 329.1051.

4-(4-Chlorophenyl)-5-(3,4-dimethoxyphenyl)-6-methyl-3,4 dihydropyrimidin-2(1H)-one (4g). 160.4 mg. Yield = 89%. Yellow solid. Mp: 151.6–152.7 °C. 1 H NMR (400 MHz, DMSO- d_{6}): δ 8.36 (s, 1H), 7.34–7.29 (m, 3H), 7.19–7.15 (m, 2H), 6.78 (d, $J =$ 8.3 Hz, 1H), 6.63 (d, $J = 2.0$ Hz, 1H), 6.50 (dd, $J = 8.2$, 2.0 Hz, 1H), 5.10 (s, 1H), 3.67 (s, 3H), 3.64 (s, 3H), 1.73 (s, 3H) ppm. 13C NMR (100 MHz, DMSO-d₆): δ 153.0, 148.2, 147.2, 143.4, 131.6, 130.2, 129.6, 128.8, 128.3, 121.9, 113.0, 111.4, 108.0, 58.9, 55.4, 55.3, 16.3 ppm. FTIR (KBr, neat): ν 3343, 1698, 1515, 1457, 1251, 1135, 1011, 766 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for $C_{19}H_{20}C/N_2O_3$: 359.1157, found: 359.1159. Paper

126.1, 13.6, 13.6, 13.6, 15.6, 15.6, 15.6, ppm. FTR [Kin, mail: $\pm \frac{1}{2}$ (Schempheng) 4 (4.4 burrying) on entity 3.4 dispersed under a Creative Commons Article is licensed under a Creative Commons Article is lic

4-(4-Chlorophenyl)-5-(4-ethylphenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-one (4h). 146.5 mg. Yield = 90%. Yellow solid. Mp: 229.8–231.2 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.40 $(s, 1H)$, 7.35–7.29 (m, 3H), 7.19–7.14 (m, 2H), 7.06 (d, $J = 8.2$ Hz, 2H), 6.97 (d, $J = 8.2$ Hz, 2H), 5.07 (s, 1H), 2.52 (q, $J = 7.6$ Hz, 2H), 1.73 (s, 3H), 1.12 (t, $J = 7.6$ Hz, 3H) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 153.0, 143.2, 141.6, 135.1, 131.7, 130.1, 129.2, 128.8, 128.4, 127.5, 107.8, 58.7, 27.7, 16.3, 15.3 ppm. FTIR (KBr, neat): ν 3223, 2928, 1686, 1241, 1088, 842 cm $^{-1}$. HRMS (ESI, m/ z): $[M + H]^+$, calcd for C₁₉H₂₀ClN₂O: 327.1259, found: 327.1260.

4-(4-Chlorophenyl)-6-ethyl-5-phenyl-3,4-dihydropyrimidin- $2(1H)$ -one (4i). 101.8 mg. Yield = 65%. Yellow solid. Mp: 184.8– 185.7 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 8.41 (s, 1H), 7.34–7.30 $(m, 3H)$, 7.23 $(t, J = 7.4$ Hz, 2H), 7.17-7.12 $(m, 3H)$, 7.03-6.98 $(m,$ 2H), 5.05 (d, $J = 2.5$ Hz, 1H), 2.04-1.94 (m, 2H), 1.03 (t, $J =$ 7.4 Hz, 3H) ppm. 13 C NMR (100 MHz, DMSO- d_6): δ 153.2, 143.1, 137.9, 135.5, 131.7, 129.3, 128.7, 128.4, 128.2, 126.5, 107.7, 59.0, 22.7, 13.0 ppm. FTIR (KBr, neat): v 3235, 1679, 1489, 1227, 1091, 1014, 765, 701 cm⁻¹. HRMS (ESI, m/z): [M + H]⁺, calcd for C₁₈H₁₈ClN₂O: 313.1102, found: 313.1102.

Ethyl 4-(4-chlorophenyl)-6-methyl-2-oxo-1,2,3,4-tetrahydro pyrimidine-5-carboxylate (4j). 38.2 mg. Yield $= 26\%$. White solid. Mp: 212.3-213.6 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 9.25 $(s, 1H)$, 7.78 $(s, 1H)$, 7.39 $(d, J = 8.4 \text{ Hz}, 2H)$, 7.24 $(d, J = 8.4 \text{ Hz},$ 2H), 5.14 (d, $J = 3.2$ Hz, 1H), 3.98 (q, $J = 7.0$ Hz, 2H), 2.24 (s, 3H), 1.09 (t, $J = 7.1$ Hz, 3H) ppm. ¹³C NMR (100 MHz, DMSO- d_6): d 165.3, 152.0, 148.8, 143.8, 131.8, 128.5, 128.3, 98.9, 59.3, 53.5, 17.9, 14.1 ppm. The characterization data of this product is in accordance with the reported ones.¹⁷

Conflicts of interest

The authors declare no conflict of interest.

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Notes and references

- 1 (a) Z. D. Aron and L. E. Overman, J. Am. Chem. Soc., 2005, 127, 3380; (b) M. A. Arnold, K. A. Day, S. G. Duron and D. Y. Gin, J. Am. Chem. Soc., 2006, 128, 13255.
- 2 (a) R. W. Lewis, J. Mabry, J. G. Polisar, K. P. Eagen, B. Ganem and G. P. Hess, Biochemistry, 2010, 49, 4841; (b) X. Zhu, G. Zhao, X. Zhou, X. Xu, G. Xia, Z. Zheng, L. Wang, X. Yang and S. Li, Bioorg. Med. Chem. Lett., 2010, 20, 299; (c) K. L. Dhumaskar, S. N. Meena, S. C. Ghadi and S. G. Tilve, Bioorg. Med. Chem. Lett., 2014, 24, 2897; (d) T. G. M. Treptow, F. Figueiró, E. H. F. Jandrey, A. M. O. Battastini, C. G. Salbego, J. B. Hoppe, P. S. Taborda, S. B. Rosa, L. A. Piovesan, C. R. M. D'Oca, D. Russowsky and M. G. M. D'Oca, Eur. J. Med. Chem., 2015, 95, 552; (e) R. Chikhale, S. Menghani, R. Babu, R. Bansode, G. Bhargavi, N. Karodia, M. V. Rajasekharan, A. Paradkar and P. Khedekar, Eur. J. Med. Chem., 2015, 96, 30; (f) U. Rashid, R. Sultana, N. Shaheen, S. F. Hassan, F. Yaqoob, M. J. Ahmad, F. Iftikhar, N. Sultana, S. Asghar, M. Yasinzai, F. L. Ansari and N. A. Qureshi, Eur. J. Med. Chem., 2016, 115, 230; (g) K. Singh and T. Kaur, RSC Med. Chem., 2016, 7, 749.
- 3 P. Biginelli, Gazz. Chim. Ital., 1893, 23, 360.
- 4 (a) C. O. Kappe, Tetrahedron, 1993, 49, 6937; (b) C. O. Kappe, Acc. Chem. Res., 2000, 33, 879; (c) C. Simon, T. Constantieux and J. Rodriguez, Eur. J. Org. Chem., 2004, 4957; (d) D. Dallinger, A. Stadler and C. O. Kappe, Pure Appl. Chem., 2004, 76, 1017; (e) L. Z. Gong, X. H. Chen and X. Y. Xu, Chem.–Eur. J., 2007, 13, 8920; (f) M. A. Kolosov and V. D. Orlov, Mol. Diversity, 2009, 13, 5; (g) H. Nagarajaiah, A. Mukhopadhyay and J. N. Moorthy, Tetrahedron Lett., 2016, 57, 5135; (h) R. V. Patil, J. U. Chavan, D. S. Dalal, V. S. Shinde and A. G. Beldar, ACS Comb. Sci., 2019, 21, 105; (i) A. Domling, W. Wang and K. Wang, Chem. Rev., 2012, 112, 3083; (j) Y. Zhao, H. Wu, Z. Wang, Y. Wei, Z. Wang and L. Tao, Sci. China: Chem., 2016, 59, 1541; (k) R. V. Patil, J. U. Chavan, D. S. Dalal, V. S. Shinde and A. G. Beldar, ACS Comb. Sci., 2019, 21, 105; (l) R. Afshari and A. Shaabani, ACS Comb. Sci., 2018, 20, 499.
- 5 (a) H. Xue, Y. Zhao, H. Wu, Z. Wang, B. Yang, Y. Wei, Z. Wang and L. Tao, J. Am. Chem. Soc., 2016, 138, 8690; (b) T. Mao, G. Liu, H. Wu, Y. Wei, Y. Gou, J. Wang and L. Tao, J. Am. Chem. Soc., 2018, 140, 6865; (c) N. Sahota, D. I. AbuSalim, M. L. Wang, C. J. Brown, Z. Zhang, T. J. El-Baba, S. P. Cook and D. E. Clemmer, Chem. Sci., 2019, 10, 4822; (d) W. Fan, Y. Queneau and F. Popowycz, Green Chem., 2018, 20, 485; (e) N. Li, X.-H. Chen, J. Song, S.-W. Luo, W. Fan and L.-Z. Gong, J. Am. Chem. Soc., 2009, 131, 15301; (f) H. G. O. Alvim, D. L. J. Pinheiro, V. H. Carvalho-Silva, M. Fioramonte, F. C. Gozzo, W. A. da Silva, G. W. Amarante and B. A. D. Neto, J. Org. Chem., 2018, 83, 12143; (g) F.-J. Meng, L. Shi, G.-S. Feng, L. Sun and Y.-G. Zhou, J. Org. Chem., 2019, 84, 4435; (h) I. L. Gonçalves, L. Davi, L. Rockenbach, G. M. das Neves, L. P. Kagami, R. F. S. Canto, F. Figueiró, A. M. O. Battastini and V. L. Eifler-Lima, Tetrahedron Lett., 2018, 59, 759; (i) P. Chen and M. Tu, Tetrahedron Lett., 2018, 59, 987; (j) R. Pathoor, T. Puthiyedath and D. Bahulayan, Tetrahedron Lett., 2019, 60, 191; (k) S. Zheng, Y. Jian, S. Xu, Y. Wu, H. Sun, G. Zhang, W. Zhang and Z. Gao, RSC Adv., 2018, 8, 8657; (l) E. F. Freitas, R. Y. Souza, S. T. A. Passos, J. A. Dias, S. C. L. Dias and B. A. D. Neto, RSC Adv., 2019, 9, 27125; (m) N. Li, X.-H. Chen, S.-M. Zhou, S.-W. Luo, J. Song, L. Ren and L.-Z. Gong, Angew. Chem., Int. Ed., 2010, 49, 6378. **PSC Articles. Articles. Published on 14 August 2020. Article is licensed under a Creative Commons Articles. Although August 2020. Al**
	- 6 B. Jauk, T. Pernat and C. O. Kappe, Molecules, 2000, 5, 227.
	- 7 (a) Z. T. Wang, L. W. Xu, C. G. Xia and H. Q. Wang, Tetrahedron Lett., 2004, 45, 7951; (b) B. Liang, X. T. Wang, J. X. Wang and Z. Y. Du, Tetrahedron, 2007, 63, 1981.
	- 8 M. M. Abelman, S. C. Smith and D. R. James, Tetrahedron Lett., 2003, 44, 4559.
	- 9 (a) O. M. Singh and N. S. Devi, J. Org. Chem., 2009, 74, 3141; (b) G. C. Nandi, S. Samai and M. S. Singh, J. Org. Chem., 2010, 75, 7785.
	- 10 J. P. Wan and Y. J. Pan, Chem. Commun., 2009, 2768.
	- 11 (a) H. H. Zhang, Z. Q. Zhou, Z. G. Yao, F. Xu and Q. Shen, Tetrahedron Lett., 2009, 50, 1622; (b) Z.-L. Zhou, P.-C. Wang and M. Lu, Chin. Chem. Lett., 2016, 27, 226.
	- 12 C. D. Bailey, C. E. Houlden, G. L. J. Bar, G. C. Lloyd-Jones and K. I. Booker-Milburn, Chem. Commun., 2007, 2932.
	- 13 S. Yu, J. Wu, H. Lan, L. Gao, H. Qian, K. Fan and Z. Yin, Org. Lett., 2020, 22, 102.
	- 14 Z.-L. Shen, X.-P. Xu and S.-J. Ji, J. Org. Chem., 2010, 75, 1162.
	- 15 (a) H. Liu, Y. Fang, S.-Y. Wang and S.-J. Ji, Org. Lett., 2018, 20, 930; (b) H. Liu, Y. Fang, L. Yin, S.-Y. Wang and S.-J. Ji, J. Org. Chem., 2017, 82, 10866; (c) W.-B. Cao, X.-P. Xu and S.-J. Ji, Org. Biomol. Chem., 2017, 15, 1651; (d) X. Liu, H. Zhu, S. B. Zhang, Y. Cheng, H. Y. Peng and Z. B. Dong, Tetrahedron Lett., 2018, 59, 3165; (e) X.-X. Feng, Z. Wu, Q.-D. Wang, B.-Z. Chen, W. Rao, J.-M. Yang and Z.-L. Shen, Appl. Organomet. Chem., 2019, 33, e5110; (f) T. Xie, G.-Q. Wang, Y.-W. Wang, W. Rao, H. Xu, S. Li, Z.-L. Shen and X.-Q. Chu, iScience, 2020, 23, 101259.
	- 16 C. O. Kappe, J. Org. Chem., 1997, 62, 7201.
	- 17 (a) J. Ma, L. Zhong, X. Peng and R. Sun, Green Chem., 2016, 18, 1738; (b) N. Li, Y. Wang, F. Liu, X. Zhao, X. Xu, Q. An and K. Yun, Appl. Organomet. Chem., 2020, 34, e5454.