

Cite this: *Chem. Sci.*, 2020, 11, 1912

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 10th September 2019  
Accepted 22nd December 2019

DOI: 10.1039/c9sc04569d

rsc.li/chemical-science

# From 1,2-difunctionalisation to cyanide-transfer cascades – Pd-catalysed cyanosulfenylation of internal (oligo)alkynes†

Marcel Bürger, <sup>a</sup> Maximilian N. Loch, <sup>a</sup> Peter G. Jones<sup>b</sup> and Daniel B. Werz <sup>\*a</sup>

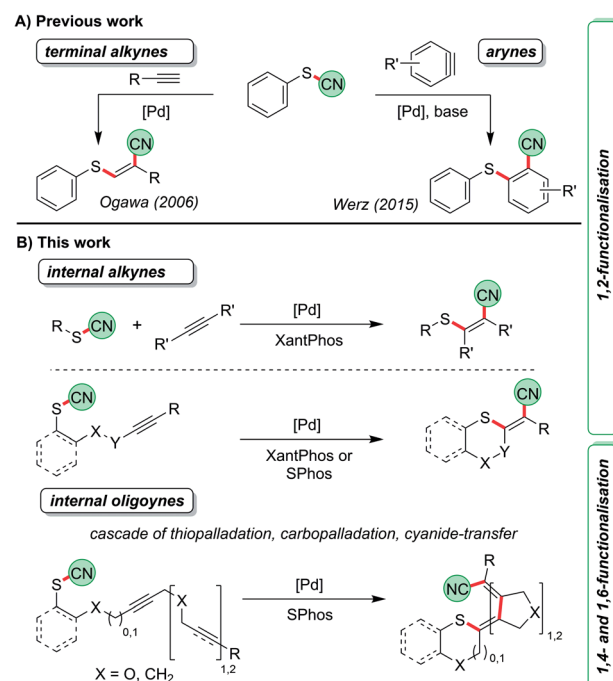
Internal alkynes substituted by aliphatic or aromatic moieties or by heteroatoms were converted into sulphur-substituted acrylonitrile derivatives. Key is the use of Pd catalysis, which allows the addition of aromatic and aliphatic thiocyanates in an intra- and intermolecular manner. Substrates with several alkyne units underwent further carbopalladation steps after the initial thiopalladation step, thus generating in a cascade-like fashion an oligoene unit with sulphur at one terminus and the cyano group at the other.

## Introduction

An important method of accessing tetrasubstituted double bonds is the dual functionalisation of internal alkynes.<sup>1</sup> In its ideal and atom-economic way a substrate X–Y is attached to the C–C triple bond in a completely regio- and diastereoselective fashion. Depending on the type of substituents, either specially designed reagents are required or transition-metal catalysts are employed.<sup>2–4</sup> As a prime example for the first scenario a *syn*-chlorocyanation using imidazolium thiocyanate in combination with BCl<sub>3</sub> was recently reported by Alcarazo *et al.*<sup>2a</sup> Morandi developed a Pd-catalysed intermolecular *syn*-aryliodination of internal alkynes<sup>5</sup> while Lautens and co-workers disclosed a series of intramolecular reactions leading to *syn*- and *anti*-carbohalogenations of triple bonds.<sup>6</sup> The key step of the latter reaction is a carbopalladation of the C–C triple bond followed by reductive elimination of the Pd to generate the carbon-halogen bond. Such a step is rare and can only be triggered by specific ligands.<sup>5,6</sup> The more common scenario is further reaction in a carbopalladation cascade<sup>7</sup> when other alkyne or alkene units are present.<sup>8</sup> In most cases a final elimination or cross-coupling reaction is employed to terminate the process. By such protocols, structures of astonishing complexity<sup>9</sup> such as fenestranes,<sup>10</sup> cyclooctatetraenes,<sup>11</sup> helicenes,<sup>12</sup> and several natural products<sup>13</sup> have been prepared from appropriately designed starting materials in just one synthetic step. The

groups of Lautens, Cook and others have precisely designed intramolecular iodine-transfer reactions along these lines.<sup>14</sup> With other moieties such transfer reactions over several carbon atoms have only rarely been investigated.

Cyanosulfenylation reactions of terminal alkynes have been developed to access sulphur-substituted acrylonitrile derivatives starting from thiocyanates (Scheme 1A, left).<sup>15</sup> In these transformations the S–CN bond is broken and added across the C–C triple bond, which is comparable to reactions where O–CN



Scheme 1 Inter- and intramolecular cyanosulfenylation reactions.

<sup>a</sup>Technische Universität Braunschweig, Institute of Organic Chemistry, Hagenring 30, 38106 Braunschweig, Germany. E-mail: d.werz@tu-braunschweig.de

<sup>b</sup>Technische Universität Braunschweig, Institute of Inorganic and Analytical Chemistry, Hagenring 30, 38106 Braunschweig, Germany

† Electronic supplementary information (ESI) available: Experimental details and characterisation data. CCDC 1948156–1948162 and 1969696–1969698. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c9sc04569d



or N–CN-bonds are activated to react with alkenes.<sup>16</sup> Later, we found that a similar Pd-catalysed process allows the reaction with arynes, which contain highly reactive formal triple bonds, leading to *o*-thiobenzonitriles (Scheme 1A, right).<sup>17</sup>

Thus, we were first interested in seeking conditions to transform internal non-activated triple bonds to generate tetrasubstituted alkenes with sulphur and cyano in 1,2-position (Scheme 1B), and to determine whether this reaction might be also the starting point for a longer cascade involving – besides the thiopalladation step – carbopalladation steps of additional C–C triple bonds. The termination of the cascade was envisioned as a reductive elimination of the Pd catalyst to form the C–CN bond. Such an approach would lead to formal 1,4- and 1,6-cyanosulfonylation reactions (Scheme 1B).

## Results and discussions

### Optimisation of the reaction conditions

Initially, we explored the Pd-catalysed reaction of the aromatic thiocyanate unit across the C–C triple bond in **1a**. PdCl<sub>2</sub>(PhCN)<sub>2</sub>, tris(*t*-butyl)phosphine as ligand (derived from Fu's salt<sup>18</sup>) and triethylamine in DMF at 100 °C generated the benzoxathiin **2a** in 4 h with 35% yield. Various other ligands gave no better results. A breakthrough was achieved using XantPhos, which led to an excellent yield of 93% (for full optimisation details, see ESI†).

### Scope of aromatic thiocyanates

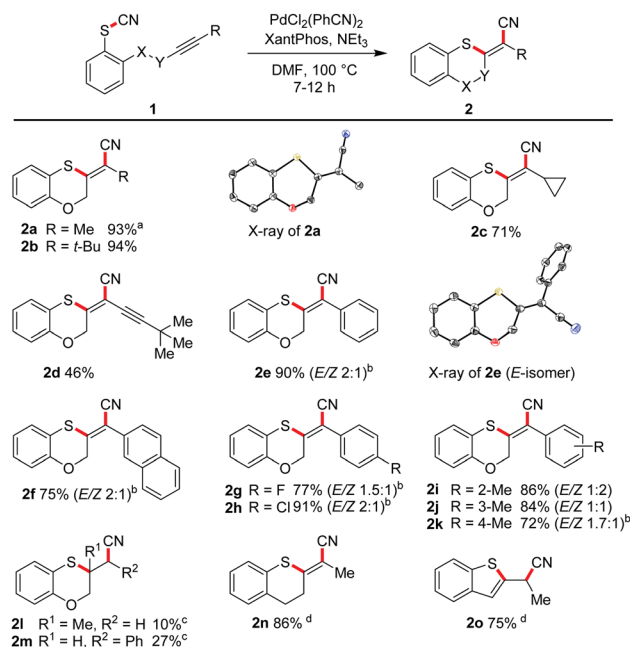
With these conditions in hand, we explored the scope and limitations of this transformation (Scheme 2).

Substrates with aliphatic termini were smoothly transformed and yielded products **2a–2c** in 71–94% yield. A 1,3-diyne unit reacted to enyne **2d** (46%). Differently substituted aromatic termini were tolerated and furnished products **2e–2k** in up to 91% yield. However, mixtures of *E/Z*-isomers were found because of the highly polarized character of the emerging tetrasubstituted olefin moiety. In these cases, the thioacrylonitrile moiety, as a strong push–pull-system, easily stabilizes a zwitterionic structure.<sup>19</sup> The positive charge is well stabilized at the sulphur and the negative charge next to the aryl substituent. This formal single bond character allows rotation leading to isomeric mixtures.<sup>20</sup>

Furthermore, **2l** (10%) and **2m** (27%) were synthesised from two alkenes as starting materials to demonstrate that also other  $\pi$ -systems can be employed in our protocol. To address smaller ring sizes the tether between the aromatic ring system and the C–C triple bond was shortened to afford substituted benzothiophene **2o** as the result of a subsequent isomerisation to the more stabilized aromatic system. A further shortening led not to an isolable four-membered ring benzothiete, but to a 3-cyano-substituted benzothiophene (**2s**, see ESI†), as proved by X-ray crystallography.<sup>21</sup>

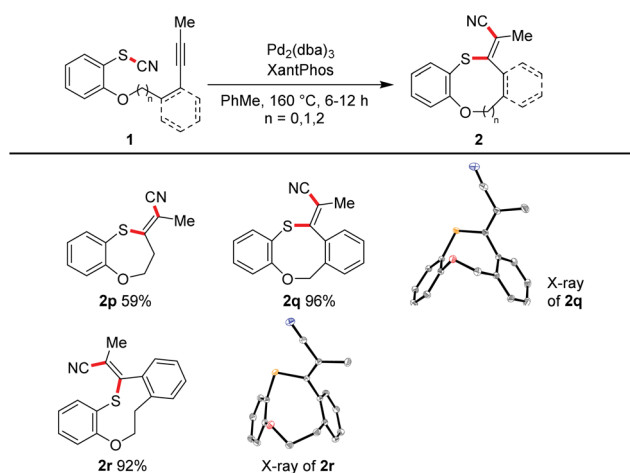
### Seven- to nine-membered ring systems

We varied the ring size also in the opposite direction by increasing the tether length. In moderate to excellent yields the



**Scheme 2** Intramolecular cyanosulfonylation with aromatic thiocyanates. Reaction conditions: **1** (1.00 equiv.), PdCl<sub>2</sub>(PhCN)<sub>2</sub> (10 mol%), XantPhos (20 mol%), NEt<sub>3</sub> (5.00 equiv.), DMF (20 mM), 100 °C, 7 h. <sup>a</sup>Large scale (1.5 mmol, 93%), Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 160 °C, 3 h, PhMe (20 mM). <sup>b</sup>*Z*-isomer is shown even if it is the minor isolated product because that is consistent with the plausible mechanism; *E*-isomer arises as result of a strong push–pull-system.<sup>19</sup> <sup>c</sup>Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 160 °C, 6 h, DMF (20 mM). <sup>d</sup>Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 110 °C, 3 h, DMF (20 mM). X = O, (CH<sub>2</sub>)<sub>m</sub>; Y = (CH<sub>2</sub>)<sub>n</sub>; m, n = 0, 1, 2.

seven-membered ring **2p** (59%), the eight-membered oxathiocin **2q** (96%) and even the nine-membered oxathionin derivative **2r** (92%) were obtained. The structures of the latter two compounds were proved by X-ray crystallography (Scheme 3). The formation of a ten- or a twelve-membered ring system was, however, not achieved.<sup>22</sup>



**Scheme 3** Intramolecular cyanosulfonylation for large ring synthesis. Reaction conditions: **1** (1.00 equiv.), Pd<sub>2</sub>(dba)<sub>3</sub> (10 mol%), XantPhos (20 mol%), PhMe (20 mM), 160 °C, 6–12 h.



## Scope of aliphatic thiocyanates

Next, we focused on the scope with respect to aliphatic thiocyanates (Scheme 4). We realized that higher temperatures are required to enable the transformation. This observation might be traced back to a stronger S–CN bond (because of its non-conjugated nature) and/or the lack of preorganisation. Hence, a complete reoptimisation of the reaction conditions was necessary (see ESI†). Finally, Pd<sub>2</sub>(dba)<sub>3</sub> or Pd(PPh<sub>3</sub>)<sub>4</sub> and the use of toluene as solvent and a reaction temperature of 160 °C proved to be the optimal choice. Under these conditions **4a** was obtained in 89% yield. The successful synthesis of products **4b–4d** shows that the reaction tolerates aliphatic termini, as well as a conjugated triple bond as exemplified in isothiochromene **4d**. The motif of a molecular switch was achieved with the synthesis of **4e** (64%). The generation of five-membered exocyclic thioenol ethers also proceeded smoothly. Electron-poor (**4g**), electron-rich (**4h**), sterically encumbered (**4i**), aromatic and hetero-aromatic (**4j**) termini also permit the transformation. Compound **4f** was prepared to investigate whether an acceleration of the reaction by a Thorpe–Ingold effect can be observed; the effect was small.<sup>23</sup>

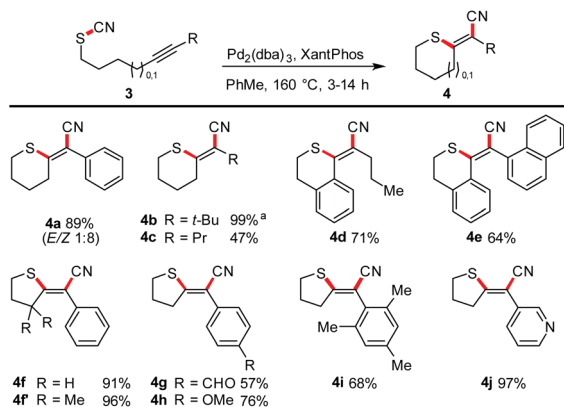
## Multiple heteroatom-substitution

Because we have a special interest in multiply heterosubstituted C–C double bonds, we subjected several  $\alpha$ - and  $\alpha,\alpha'$ -(di) substituted triple bonds to the reaction conditions (Scheme 5).

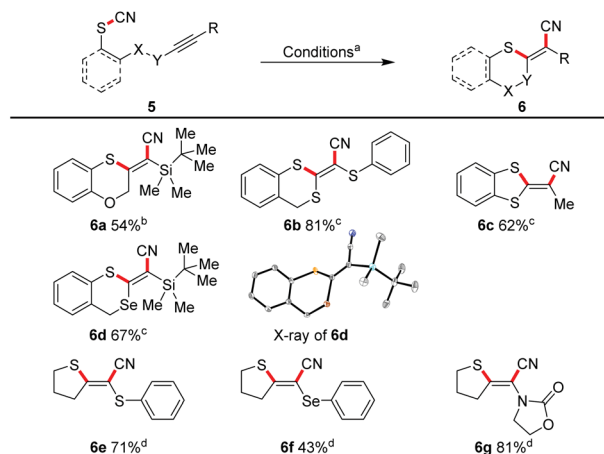
This allowed a diastereoselective access to the corresponding doubly or triply heterosubstituted olefin moieties in moderate to good yields. TBS-substituted triple bonds (e.g. **6a**, **6d**), and also the very electron-rich mono- or bis-substituted triple bonds with sulphur (**6b**, **6c**, **6e**) and selenium (**6d**, **6f**) were successfully transformed. An ynamide furnished the sulphur/nitrogen-substituted (*E*)-alkene **6g** in 81%.

## Intermolecular cyanosulfonylation

All the transformations described so far proceeded in an intramolecular manner. With an excess of alkyne (2.0 equiv.)



**Scheme 4** Intramolecular cyanosulfonylation with aliphatic thiocyanates. Reaction conditions: **3** (1.00 equiv.), Pd<sub>2</sub>(dba)<sub>3</sub> (10 mol%), XantPhos (20 mol%), PhMe (20 mM), 160 °C, 3–14 h. <sup>a</sup>Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 160 °C, 3 h, PhMe (20 mM).

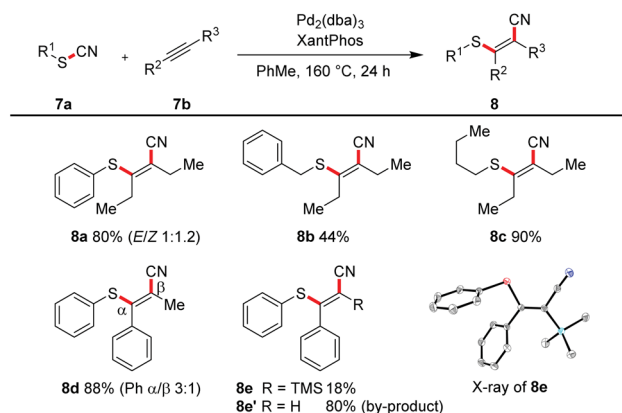


**Scheme 5** Cyanosulfonylation of heterosubstituted triple bonds. <sup>a</sup>Reaction conditions: **5** (1.00 equiv.). <sup>b</sup>Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 110 °C, 3 h, DMF (20 mM). <sup>c</sup>PdCl<sub>2</sub>(PhCN)<sub>2</sub> (10 mol%), XantPhos (20 mol%), NEt<sub>3</sub> (5.00 equiv.), DMF (20 mM), 100 °C, 3–7 h. <sup>d</sup>Pd<sub>2</sub>(dba)<sub>3</sub> (10 mol%), XantPhos (20 mol%), 160 °C, 3 h, PhMe (20 mM).

and the catalytic system of Pd<sub>2</sub>(dba)<sub>3</sub> and Xantphos in toluene at 160 °C we were able to realize an intermolecular variant (Scheme 6). 3-Hexyne was converted in yields of 44–90% into the tetrasubstituted alkenes **8a–8c**. The reaction with methylphenyl acetylene afforded in 88% yield a regioisomeric mixture (3 : 1); the steric and electronic differentiation of the two acetylenic carbons seems to be not high enough to lead to a clear preference of the thiopalladation step. In contrast, with phenyl trimethylsilyl acetylene only one regioisomeric product **8e** was formed (compare also X-ray crystal structure); however, as the major product desilylated trisubstituted olefin **8e'** was found.

## Thiopalladation–carbopalladation cascade with cyanide transfer

Our ultimate goal was to extend the sequence that starts with the thiopalladation by further carbopalladation steps. The reductive elimination of the Pd should then occur as the final



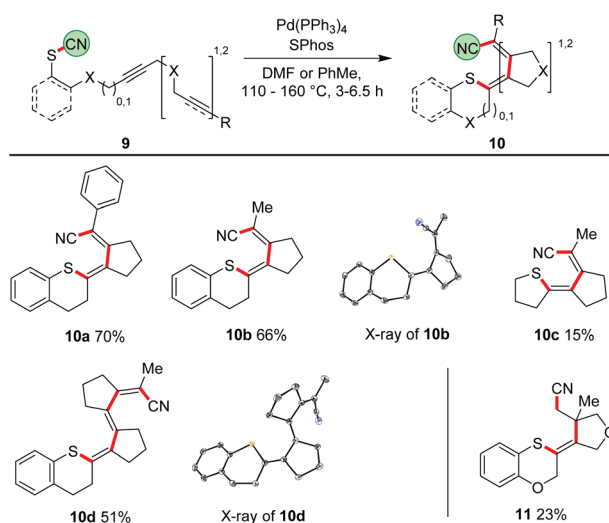
**Scheme 6** Intermolecular cyanosulfonylation with different thiocyanates. Reaction conditions: **7a** (1.00 equiv.), **7b** (2.00 equiv.), Pd<sub>2</sub>(dba)<sub>3</sub> (10 mol%), XantPhos (20 mol%), PhMe (20 mM), 160 °C, 24 h.



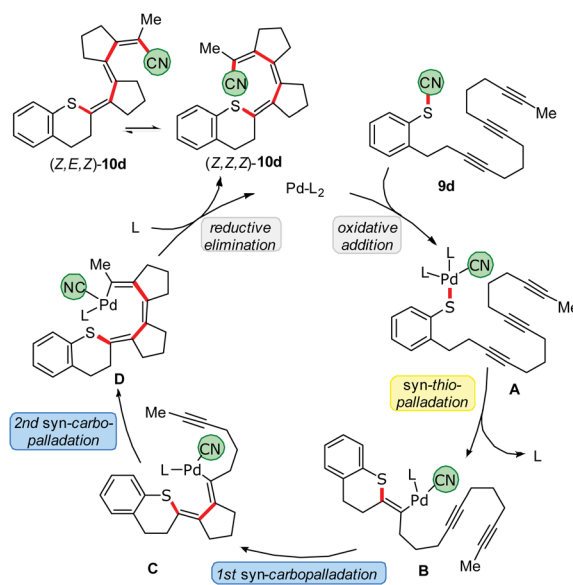
step, generating a C–CN bond. Such an approach using oligoalkynes would lead to a formal 1,4- or 1,6-cyanosulfonylation with a conjugated double bond system in between (Scheme 7). To perform such a cascade successfully, the carbopalladation step must be faster than the reductive elimination of Pd forming the C–CN bond. Initial experiments using model compound **9a** and trimethylphosphine as ligand for Pd showed that a thiopalladation–carbopalladation sequence is possible. However, as final step a protodepalladation occurred and thus the cyano moiety was not found in the product (for X-ray analysis of this undesired product **10a'**, see ESI†). Reoptimisation studies revealed that the best results are obtained with SPhos and Pd(PPh<sub>3</sub>)<sub>4</sub> as Pd source. The 1,4-cyanosulfonylation products **10a** and **10b** were obtained in yields of 66–70%. Because the conditions had been optimised for aryl thiocyanates, we expected a much lower yield for cascades starting with alkyl thiocyanates. Indeed, compound **10c** was obtained in only 15% yield. The thiopalladation–carbopalladation–carbopalladation cascade, generating one C–S and three C–C bonds in one step, yielded push–pull-substituted triene **10d** in 51% yield. X-ray crystallography unequivocally showed the translocation of the cyano group; the central double bond has isomerized to avoid a helical arrangement.<sup>24</sup> First unoptimised experiments with an alkyne-alkene system, furnishing **11**, demonstrated that, after formation of the tetrasubstituted double bond, the subsequent carbopalladation also allows the generation of a C(sp<sup>3</sup>)-CN bond.<sup>25</sup>

### Proposed mechanism

A plausible mechanism of the thiopalladation/carbopalladation/carbopalladation/cyanide transfer cascade to **10d**, based on our additional experimental results<sup>26</sup> (such as protodepalladation) and previously reported carbopalladation cascades,<sup>10–12</sup> is



**Scheme 7** Thiopalladation/carbopalladation/(carbopalladation)/cyanide transfer cascades with translocation of the CN group over four or six atoms. Product **9** uses an enyne as starting material. Conditions: **7** (1.00 equiv.), Pd(PPh<sub>3</sub>)<sub>4</sub> (5 mol%), SPhos (20 mol%), 110–160 °C, 3–7 h, DMF or PhMe (20 mM).

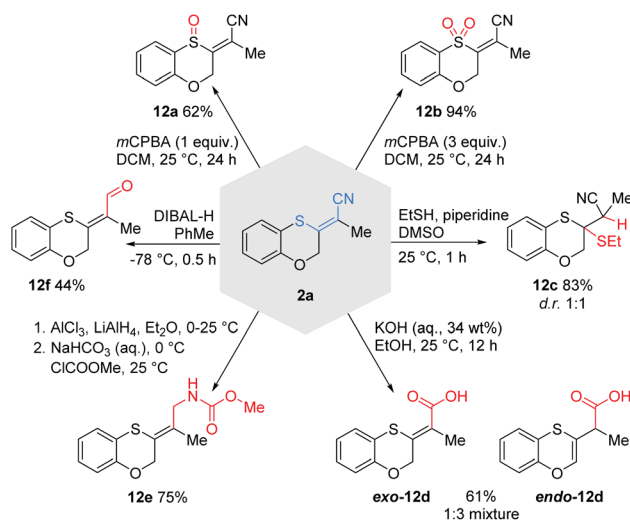


**Scheme 8** Proposed mechanism of the thiopalladation/carbopalladation/carbopalladation/cyanide transfer cascade.

depicted in Scheme 8. After the oxidative addition of the Pd catalyst into the S–CN bond of **9d**, a thiopalladation occurs (**A** → **B**). Instead of delivering the CN group *via* reductive elimination of the Pd to the emerging double bond, the sequence continues by a first *syn*-carbopalladation to yield **C**. A repetition with another triple bond affords **D**. As terminating step, the C(sp<sup>2</sup>)-CN bond is formed by reductive elimination of the Pd to furnish **E**. Because a strongly push–pull-substituted oligoene is formed, the double bonds are much weaker than in non-push–pull-substituted oligoenes; thus, isomerisation from (Z,Z,Z)-**10d** to (Z,E,Z)-**10d** might happen relatively easily.

### Follow-up chemistry

To demonstrate the versatility of the thioacrylonitrile moiety, some postsynthetic functionalisations with **2a** were performed



**Scheme 9** Follow-up chemistry.





(Scheme 9). Depending on the amount of *m*CPBA, we were able to prepare either sulfoxide **12a** or sulfone **12b** in good to excellent yields, thus transforming the strongly polarized olefin into very electron-poor olefins. A Michael addition with ethanethiol led to a diastereomeric mixture of **12c** in 83% yield. Furthermore, the hydrolysis of the nitrile with aqueous KOH gave access to the carboxylic acids *exo*-**12d** and *endo*-**12d**. The partial reduction of the Michael system trapped by methyl chloroformate led to carbamate **12e** in a good yield of 75%. Reaction of **2a** with DIBAL-H furnished the aldehyde **12f** in a moderate yield.

## Conclusions

In conclusion, we have developed a Pd-catalysed *syn*-1,2-cyano-sulfenylation of internal alkynes to access tetra-substituted double bonds with sulphur and cyano in adjacent positions. Both aromatic and aliphatic thiocyanates undergo the reaction. Various substitution patterns of the C–C triple bond are tolerated, such as aliphatic and aromatic residues, but also heteroatoms such as sulphur, selenium, silicon and nitrogen. Our methodology facilitates access to tetrasubstituted olefins with four different elements as substituents of the double bond. The reaction works in an intramolecular manner paving the way to five-, six-, seven-, eight, and nine-membered ring systems, but also in an intermolecular way leading to acyclic compounds. By offering further alkyne moieties the transformation was extended to a thiopalladation/carbopalladation/(carbopalladation) cascade with translocation of the cyano group over four or six carbon atoms, generating up to four new bonds in one step.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

We are grateful to the TU Braunschweig for funding. Marcel Bürger thanks Anna Lehmann, Theresa Schitter and Adrian Bauschke for their support.

## Notes and references

- I. Marek and Y. Minko, in *Metal-Catalyzed Cross-Coupling Reactions and More*, ed. A. de Meijere, M. Oestreich and S. Bräse, Wiley-VCH, Weinheim, 2014.
- For examples of addition of special reagents see: (a) A. G. Barrado, A. Zieliński, R. Goddard and M. Alcarazo, *Angew. Chem., Int. Ed.*, 2017, **56**, 13401; (b) X. Wang and A. Studer, *J. Am. Chem. Soc.*, 2016, **138**, 2977.
- Selected examples for transition-metal catalysed functionalisations: (a) P. D. G. Greenwood, E. Grenet and J. Waser, *Chem.–Eur. J.*, 2019, **25**, 3010; (b) M. Ahmad, A.-C. Gaumont, M. Durandetti and J. Maddaluno, *Angew. Chem., Int. Ed.*, 2017, **56**, 2464; (c) N. Sakata, K. Sasakura, G. Matsushita, K. Okamoto and K. Ohe, *Org. Lett.*, 2017, **19**, 3422; (d) Y. Nakao, *Bull. Chem. Soc. Jpn.*, 2012, **85**, 731; (e) N. R. Rondla, S. M. Levi, J. M. Ryss, R. A. Vanden Berg and C. J. Douglas, *Org. Lett.*, 2011, **13**, 1940; (f) Y. Hirata, A. Yada, E. Morita, Y. Nakao, T. Hiyama, M. Ohashi and S. Ogoshi, *J. Am. Chem. Soc.*, 2010, **132**, 10070; (g) Y. Nakao, A. Yada, S. Ebata and T. Hiyama, *J. Am. Chem. Soc.*, 2007, **129**, 2428; (h) M. Suginome, A. Yamamoto and M. Murakami, *Angew. Chem., Int. Ed.*, 2005, **44**, 2380; (i) Y. Nakao, S. Oda and T. Hiyama, *J. Am. Chem. Soc.*, 2004, **126**, 13904; (j) M. Suginome, A. Yamamoto and M. Murakami, *J. Am. Chem. Soc.*, 2003, **125**, 6358.
- For miscellaneous examples see: (a) S. Higashimae, D. Kurata, S.-I. Kawaguchi, S. Kodama, M. Sonoda, A. Nomoto and A. Ogawa, *J. Org. Chem.*, 2018, **83**, 5267; (b) D. Xu, R. Rios, F. Ba, D. Ma, G. Gu, A. Ding, Y. Kuang and H. Guo, *Asian J. Org. Chem.*, 2016, **5**, 981; (c) M. Murai, R. Hatano, S. Kitabata and K. Ohe, *Chem. Commun.*, 2011, **47**, 2375.
- Y. H. Lee and B. Morandi, *Angew. Chem., Int. Ed.*, 2019, **58**, 6444.
- (a) T. Sperger, C. M. Le, M. Lautens and F. Schoenebeck, *Chem. Sci.*, 2017, **8**, 2914; (b) C. M. Le, P. J. C. Menzies, D. A. Petrone and M. Lautens, *Angew. Chem., Int. Ed.*, 2015, **54**, 254; (c) C. M. Le, X. Hou, T. Sperger, F. Schoenebeck and M. Lautens, *Angew. Chem., Int. Ed.*, 2015, **54**, 15897.
- (a) L.-F. Tietze, G. Brasche and K. M. Gericke, *Domino Reactions in Organic Synthesis*, Wiley-VCH, Weinheim, 2006; (b) L. F. Tietze, *Chem. Rev.*, 1996, **96**, 115.
- (a) Review: A. Düfert and D. B. Werz, *Chem.–Eur. J.*, 2016, **22**, 16718; (b) M. Pawliczek, T. F. Schneider, C. Maaß, D. Stalke and D. B. Werz, *Angew. Chem., Int. Ed.*, 2015, **54**, 4119; (c) M. Leibelng, M. Pawliczek, D. Kratzert, D. Stalke and D. B. Werz, *Org. Lett.*, 2012, **14**, 346.
- (a) L. Bai, Y. Yuan, J. Liu, J. Wu, L. Han, H. Wang, Y. Wang and X. Luan, *Angew. Chem., Int. Ed.*, 2016, **55**, 6946; (b) L. F. Tietze, B. Waldecker, D. Ganapathy, C. Eichhorst, T. Lenzer, K. Oum, S. O. Reichmann and D. Stalke, *Angew. Chem., Int. Ed.*, 2015, **54**, 10317; (c) J. Wallbaum, R. Neufeld, D. Stalke and D. B. Werz, *Angew. Chem., Int. Ed.*, 2013, **52**, 13243; (d) K. Parthasarathy, H. Han, C. Prakash and C.-H. Cheng, *Chem. Commun.*, 2012, **48**, 6580; (e) M. Leibelng, D. C. Koester, M. Pawliczek, S. C. Schild and D. B. Werz, *Nat. Chem. Biol.*, 2010, **6**, 199; (f) G. Blond, C. Bour, B. Salem and J. Suffert, *Org. Lett.*, 2008, **10**, 1075; (g) Y. Zhang, G. Z. Wu, G. Agnel and E. I. Negishi, *J. Am. Chem. Soc.*, 1990, **112**, 8590.
- (a) M. Charpenay, A. Boudhar, C. Hulot, G. Blond and J. Suffert, *Tetrahedron*, 2013, **69**, 7568; (b) M. Charpenay, A. Boudhar, G. Blond and J. Suffert, *Angew. Chem., Int. Ed.*, 2012, **51**, 4379.
- S. Blouin, V. Gandon, G. Blond and J. Suffert, *Angew. Chem., Int. Ed.*, 2016, **55**, 7208.
- (a) B. Milde, M. Leibelng, M. Pawliczek, J. Grunenberg, P. G. Jones and D. B. Werz, *Angew. Chem., Int. Ed.*, 2015, **54**, 1331; (b) B. Milde, M. Leibelng, A. Hecht, P. G. Jones, A. Visscher, D. Stalke, J. Grunenberg and D. B. Werz, *Chem.–Eur. J.*, 2015, **21**, 16136.



- 13 (a) B. Milde, M. Pawliczek, P. G. Jones and D. B. Werz, *Org. Lett.*, 2017, **19**, 1914; (b) L. Li, Q. Yang, Y. Wang and Y. Jia, *Angew. Chem., Int. Ed.*, 2015, **54**, 6255; (c) S. S. Goh, G. Chaubet, B. Gockel, M.-C. A. Cordonnier, H. Baars, A. W. Phillips and E. A. Anderson, *Angew. Chem., Int. Ed.*, 2015, **54**, 12618; (d) L. F. Tietze, S.-C. Duefert, J. Clerc, M. Bischoff, C. Maaß and D. Stalke, *Angew. Chem., Int. Ed.*, 2013, **52**, 3191; (e) D. Shan, Y. Gao and Y. Jia, *Angew. Chem., Int. Ed.*, 2013, **52**, 4902.
- 14 (a) B. M. Monks and S. P. Cook, *Angew. Chem., Int. Ed.*, 2013, **52**, 14214; (b) X. Jia, D. A. Petrone and M. Lautens, *Angew. Chem., Int. Ed.*, 2012, **51**, 9870; (c) S. G. Newman and M. Lautens, *J. Am. Chem. Soc.*, 2011, **133**, 1778; (d) H. Liu, C. Li, D. Qiu and X. Tong, *J. Am. Chem. Soc.*, 2011, **133**, 6187.
- 15 (a) T. Ozaki, A. Nomoto, I. Kamiya, J.-i. Kawakami and A. Ogawa, *Bull. Chem. Soc. Jpn.*, 2011, **84**, 155; (b) W. Zheng, A. Ariafard and Z. Lin, *Organometallics*, 2008, **27**, 246; (c) M. Wang, L. Cheng and Z. Wu, *Dalton Trans.*, 2008, 3879; (d) Y. T. Lee, S. Y. Choi and Y. K. Chung, *Tetrahedron Lett.*, 2007, **48**, 5673; (e) I. Kamiya, J.-i. Kawakami, S. Yano, A. Nomoto and A. Ogawa, *Organometallics*, 2006, **25**, 3562.
- 16 (a) K. M. Korch and D. A. Watson, *Chem. Rev.*, 2019, **119**, 8192; (b) Z. Pan, S. Wang, J. T. Brethorst and C. J. Douglas, *J. Am. Chem. Soc.*, 2018, **140**, 3331; (c) S. V. C. Vummaleti, M. Al-Ghamdi, A. Poater, L. Falivene, J. Scaranto, D. J. Beetstra, J. G. Morton and L. Cavallo, *Organometallics*, 2015, **34**, 5549; (d) B. Rao and X. Zeng, *Org. Lett.*, 2014, **16**, 314; (e) Y. Miyazaki, N. Ohta, K. Semba and Y. Nakao, *J. Am. Chem. Soc.*, 2014, **136**, 3732; (f) D. C. Koester, M. Kobayashi, D. B. Werz and Y. Nakao, *J. Am. Chem. Soc.*, 2012, **134**, 6544; (g) A. Yada, S. Ebata, H. Idei, D. Zhang, Y. Nakao and T. Hiyama, *Bull. Chem. Soc. Jpn.*, 2010, **83**, 1170.
- 17 M. Pawliczek, L. K. B. Garve and D. B. Werz, *Org. Lett.*, 2015, **17**, 1716.
- 18 M. R. Netherton and G. C. Fu, *Org. Lett.*, 2001, **3**, 4295.
- 19
- 
- 20 The configurational stability of *E*- and *Z*-isomers of **2e** was investigated in additional experiments under the reaction conditions. The *E*-configured starting material shows the generation of the *Z*-configured counterpart and *vice versa*. For more information, see ESI.†
- 21 CCDC 1948156 (**2a**), 1948157 (**2e**), 1969697 (**2q**), 1969698 (**2r**), 1948158 (**6d**), 1996969 (**8e**), 1948159 (**10b**), 1948160 (**10d**), 1948161 (**2s**, ESI†) and 1948162 (**10a'**, ESI†) contain the supplementary crystallographic data for this paper.
- 22 For the synthesis of the starting material for larger rings, see ESI.†
- 23 For respective GC experiments, see ESI.†
- 24 An HPLC experiment directly after the reaction showed the existence of (*Z,Z,Z*)-**8d**, which slowly isomerizes to (*Z,E,Z*)-**8d**. For results of thermal and photochemical experiments, see ESI.†
- 25 The formation of a product from an alkene-alkene system was observed only in traces in GC-MS. For the synthesis of starting material, see ESI.†
- 26 Experiments in the presence of TEMPO did not significantly decrease the yield (77% of **2a**); thus, we rule out the involvement of any radical intermediates.

