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Dimensional interpolation for metallic hydrogen

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We employ a simple and mostly accurate dimensional interpolation formula using dimensional limits $D = 1$ and $D = \infty$ to obtain $D = 3$ ground-state energy of metallic hydrogen. We also present results describing the phase transitions for different symmetries of three-dimensional structure lattices. The interpolation formula not only predicts fairly accurate energies but also predicts a correct functional form of the energy as a function of the lattice parameters. That allows us to calculate different physical quantities such as the bulk modulus, Debye temperature, and critical transition temperature, from the gradient and the curvature of the energy curve as a function of the lattice parameters. These theoretical calculations suggest that metallic hydrogen is a likely candidate for high temperature superconductivity. The dimensional interpolation formula is robust and might be useful to obtain the energies of complex many-body systems.

1 Introduction

In 1935, Eugene Wigner and H.B. Huntington predicted the metallization of hydrogen,¹ a phase of hydrogen that behaves like an electrical conductor. Ever since this has been a major quest for condensed matter physics. In pursuing metallic hydrogen (MH), we have admired many papers, but cite a few dealing with extreme high-pressure experiments.^{2–7} Moreover, MH is a candidate for phase transitions from superconductivity to superfluidity and *vice versa* under the influence of a magnetic field.^{8–10}

Dimensional scaling offers simple solutions for $D = 1$ and $D \rightarrow \infty$ limits, then often interpolates to obtain useful results for $D = 3$, with accuracy adequate for many areas of chemical physics.^{11–21} Already, the $D \rightarrow \infty$ limit for MH was treated in 1992 by John Loeser.²² He employed a Hartree–Fock Hamiltonian that localizes the electrons in a lattice of hydrogen atoms with clamped nuclei for rigid three-dimensional simple cubic (SC), body-centered (BCC), face-centered (FCC) cubic proton-lattices. We shall tune up the $D \rightarrow \infty$ limit and develop the $D = 1$ limit for MH and interpolate to obtain $D = 3$. Recently, we used dimensional interpolation to apply ground-state energies for two, three, and four electron atoms and for ground-state H₂ molecule over a wide range of the internuclear distance R . The results compare well with the standard exact energies obtained from the Full Configuration Interaction.²³

Here we investigate the phase transition for metallic hydrogen using dimensional scaling with the three-dimensional proton lattices: SC, BCC, and FCC. Section 2 describes an interpolation formula for extended systems. Sections 3 and 4 develop a one-electron Hamiltonian for quantum theory of MH in SC, BCC, and FCC lattice with $D \rightarrow \infty$ and $D = 1$ respectively. Section 5 implements the dimensional interpolation to obtain $D = 3$, observes MH and its phase transition. Section 6 uses extrapolation and curve fitting to obtain a functional form of the ground state energy of MH in SC, BCC, FCC lattice as a function of the lattice parameter, R , and investigate the physical properties like bulk modulus, Debye temperature, melting curve, and superconductivity. Section 7 has some prospects.

2 Dimensional interpolation formula for extended systems

For dimensional scaling of atoms and molecules the energy erupts to infinity as $D \rightarrow 1$ and vanishes as $D \rightarrow \infty$. Hence, we adopt scaled units (with Hartree atomic units) whereby $E_D = (Z/\beta)^2 \varepsilon_D$ and $\beta = \frac{1}{2}(D - 1)$, so the reduced energy ε_D remains finite in both limits. When expressed in a $1/Z$ perturbation expansion, the reduced energy is given by

$$\varepsilon_D = -1 + \varepsilon_D^{(1)}\lambda + \varepsilon_D^{(2)}\lambda^2 + \dots \quad (1)$$

with $\lambda = 1/Z$, where Z is the total nuclear charge of the corresponding atom.

The interpolation for atoms, developed in ref. 20, weights the dimensional limits by $\delta = 1/D$, providing $\delta\varepsilon_1$ and $(1 - \delta)\varepsilon_\infty$ in a simple analytic formula

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$$\varepsilon_D = \delta\varepsilon_1 + (1 - \delta)\varepsilon_\infty + [\varepsilon_D^{(1)} - \delta\varepsilon_1^{(1)} - (1 - \delta)\varepsilon_\infty^{(1)}]\lambda. \quad (2)$$

For a diatomic molecule, a different scaling scheme is used and illustrated. The rescaling of distances is:

$$R \rightarrow \delta R' \text{ for } D \rightarrow 1; R \rightarrow (1 - \delta)R' \text{ for } D \rightarrow \infty. \quad (3)$$

An approximation for $D = 3$ (where $R = R'$) emerges:

$$\varepsilon_3(R') = \frac{1}{3}\varepsilon_1\left(\frac{1}{3}R'\right) + \frac{2}{3}\varepsilon_\infty\left(\frac{2}{3}R'\right), \quad (4)$$

interpolating linearly between the dimensional limits.^{24–27}

We keep the structure of the formula same as above for extended systems like metallic hydrogen. We assume the cubic symmetry for the metallic hydrogen in the large D limit and a linear chain of atoms in one dimension. The rescaled distances in different dimension is given by:

$$\text{In } D = 1: r \rightarrow r'/3 \text{ and } R \rightarrow R'/3; \quad (5a)$$

$$\text{In } D \rightarrow \infty: \rho \rightarrow 2\rho'/3 \text{ and } R \rightarrow 2R'/3. \quad (5b)$$

The coordinates r and ρ are the electronic coordinates, and the parameter R is the spacing between the atomic nuclei (or lattice parameters) in $D = 1$ and $D = \infty$ respectively.

3 Metallic hydrogen at the large- D -limit

With appropriate scaling, energies will be in units of $4/(D - 1)^2$ Hartrees, and distances in units of $D(D - 1)/6$ Bohr radii. Loeser applied with simplifications the Hartree–Fock one-electron Hamiltonian in the $D \rightarrow \infty$ limit in a lattice of hydrogen atoms with clamped nuclei:²²

$$\mathcal{H} = \frac{9}{8\rho^2} - \frac{3}{2\rho} + W(\rho, R), \quad (6)$$

where

$$W(\rho, R) = \frac{3}{4} \sum_{\substack{l,m,n \\ \in \mathcal{L}'}} \frac{1}{\sqrt{\sigma^2 R^2}} - \frac{2}{\sqrt{\sigma^2 R^2 + \rho^2}} + \frac{1}{\sqrt{\sigma^2 R^2 + 2\rho^2}}, \quad (7)$$

with

$$\sigma^2 = l^2 + m^2 + n^2, \quad (8)$$

for some set of integer triples l, m, n . In particular, the set of all integer triples gives the SC lattice, the set with l, m, n all even or all odd gives the BCC lattice, and the set with even $l + m + n$ gives the FCC lattice.

For any specified lattice type and scaled lattice constant R , the minimum of eqn (6) with respect to ρ gives the energy per electron. The whole lattice is three-dimensional, noted \mathcal{L}' minus the one site (0,0,0). The single variable ρ is the orbit radius and R is the lattice spacing. The quantity (ρ/R) is used to characterize the different density regimes.

We began with the simple cubic lattice. In Hartree–Fock approximation of MH the triple sum $W(\rho, R)$ is a kind of Madelung sum. Loeser evaluated $W(\rho, R)$ for the two limiting

cases. For, the low-density regime ($\rho \ll R$), the sums over integers are taken from the tables of Born and Misra²⁸ and the Hamiltonian of eqn (6) becomes:

$$\lim_{\rho \ll R} \mathcal{H} = \frac{9}{8\rho^2} - \frac{3}{2\rho} + 3.89157 \left(\frac{3\rho^4}{2R^5} \right). \quad (9)$$

For the high-density regime ($\rho \gg R$), the sum can be replaced by an integral and the Hamiltonian of eqn (6) becomes:

$$\lim_{\rho \gg R} \mathcal{H} = \frac{9}{8\rho^2} - \frac{3}{2\rho} + 2.17759 \left(\frac{3\rho^2}{2R^3} \right). \quad (10)$$

For the criticality of the metallic hydrogen, we have to find the critical density for which the above Hamiltonian (6) attends the minimum. For calculation-convenience we fix the parameter $R = 1$ and the lattice size to be $(400 \times 400 \times 400)$. Although ideally the number lattice points should be infinitely large, but for calculation our lattice size is sufficient to observe the critical phenomenon. We calculate numerically and display in Fig. 1(a) of the above Hamiltonian for a wide range of ρ , from 0.01 to 65, *i.e.* the density (ρ/R) ranges from 0.01 to 65. From Fig. 1(b), we find that, for $R = 1$, at density $\rho/R = 0.799$ Hamiltonian (6) attends the minimum.

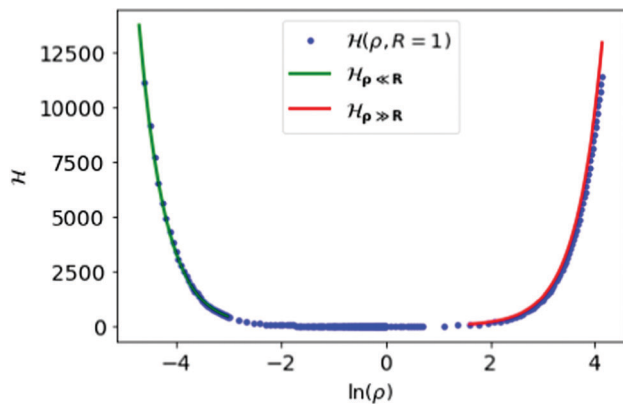
Then, we choose different values of R and calculate the minimum of the Hamiltonian $\mathcal{H}(\rho^*, R)$ by varying the parameter ρ for each R , which gives the electronic ground state energy per atom.

At $R = 1.28$, we find $\rho^* = 0.932$ at the minimum and see in Fig. 2(a), the ground state energy becomes positive to negative around $R = 1.28$. Physically this means for $R < 1.28$, the ground state energy becomes positive therefore makes the system is unstable. Therefore, at the point $R = 1.28$ and $\rho^* = 0.932$, the system goes through a phase transition. This is an elemental model at $D \rightarrow \infty$ limit with Hartree–Fock approximation and hydrogen atoms are in a simple cubic lattice, but we observe the criticality and find a critical density for the corresponding critical phenomenon.

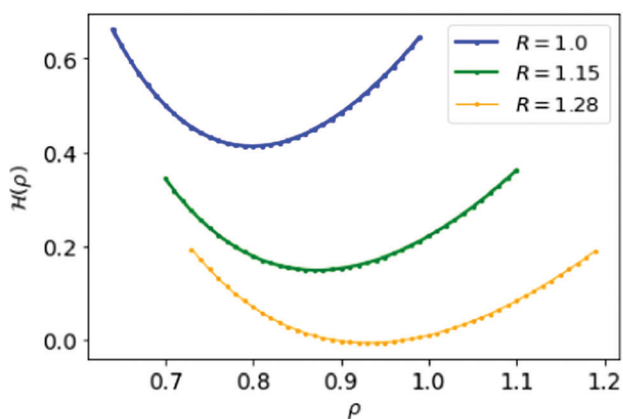
Remarkably, Loeser added to the Hartree–Fock approximation in the $D \rightarrow \infty$ limit by introducing inter-electronic correlation.²² That was essentially by opening the dihedral angles in the simple cubic lattice.²² As shown in Fig. 2(a), the energy with correlation is nearly the same as without the correlation. Further, in Fig. 2(b), we plot the difference between the correlation energy and the Hartree–Fock energy.

We first choose the parameter $R = 1$ in the electronic Hamiltonian per atom and calculate the energies with respect to the variable ρ ; see Fig. 3(a). Like the SC case we find a minimum for the Hamiltonian, which is $\rho_{\text{BCC}} = 0.88 = \rho_{\text{FCC}}$.

Again we calculate the minimum of the Hamiltonian $\mathcal{H}(\rho^*, R)$ for the different values of R by varying the parameter ρ , which gives the electronic ground state energy per atom. From Fig. 3(b), we see that the ground state energy becomes positive to negative around $R = 1.15$ for both FCC and BCC lattices. Hence, the ground state energy becomes positive so makes the lattice is unstable or the system loses the crystalline structure. Therefore, around the point $R_{\text{BCC}} = 1.15 = R_{\text{FCC}}$ and the system goes through a phase transition. Since the parameters (ρ, R) at the transition point are same for BCC and FCC



(a)



(b)

Fig. 1 At the top panel (a), we plot the electronic energy per atom $H(\rho, R)$, for MH in SC lattice, as a function of $\ln(\rho)$ keeping the parameter $R = 1$. We also plot the energy for $\rho \gg R$ obtained from eqn (10) in red, and the energy for $\rho \ll R$ obtained from eqn (9) in green. At the bottom (b), we plot the energy for each atom (ρ) as a function of ρ for $R = 1.0, 1.5$ and 1.28 respectively.

lattices, this could be coming from some symmetry working akin to their reciprocal spaces.

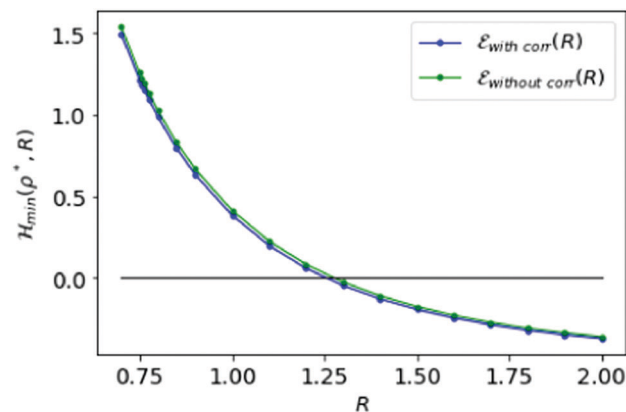
4 Metallic hydrogen in $D = 1$

A lonely, single hydrogen molecule in one-dimension is described by I. R. Lapidus.^{29,30} However, to investigate the quantum theory of MH in one dimension, we need to develop a long chain of N individual hydrogen atoms, with the nuclei sitting on the lattice sites located at $r = 0, R, 2R, \dots, (N-1)R$. Hence one-electron Hamiltonian in $D = 1$ using atomic units can be written as:

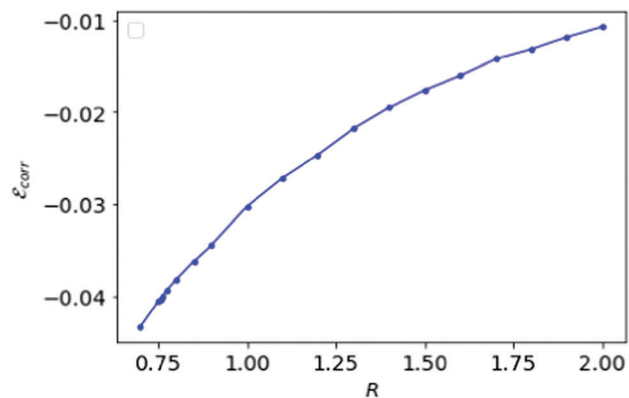
$$\mathcal{H} = -\frac{1}{2} \frac{\partial^2}{\partial r^2} - \sum_{i=0}^{N-1} \delta(r + iR) \quad (11)$$

+ electron – electron interaction part.

We choose the linear combination of atomic orbitals (LCAO) representation and construct the state for each electron in the



(a)



(b)

Fig. 2 At the top panel (a), we plot the energy obtained at $D \rightarrow \infty$ with inter-electronic correlation in blue, compared with the HF energy as a function of R in green. At the bottom (b), the difference between the correlation energy and the HF energy ($\mathcal{E}_{\text{corr}}$) is plotted as a function of R .

metallic hydrogen chain in one dimension as follows:

$$\psi(r) = \frac{1}{\mathcal{N}} [\phi_0(r) + \phi_1(r) + \phi_2(r) + \dots], \quad (12)$$

with the of the normalization constant

$$\mathcal{N} = \left[\sum_{ij} e^{-|i-j|R} (1 + |i-j|R) \right]^{1/2}, \quad (13)$$

and the individual normalized wave functions

$$\phi_i(r) = e^{-|r+iR|}.$$

The kinetic and potential energy part of the above Hamiltonian is calculated as:

$$\begin{aligned} E_{\text{KE+PE}} &= \langle \psi(r) | -\frac{1}{2} \frac{\partial^2}{\partial r^2} - \sum_{k=0}^{N-1} \delta(r+kR) | \psi(r) \rangle \\ &= \frac{1}{\mathcal{N}^2} \left[-\frac{1}{2} \sum_{ij} (-1 + |i-j|R) e^{-|i-j|R} - \sum_{ij,k} e^{-|i-k|R} e^{-|j-k|R} \right]. \end{aligned} \quad (14)$$



Fig. 3 At the top panel (a), we plot the energy per atom $H(\rho, R)$, for MH in BCC and FCC lattices, as a function of the parameter ρ fixing $R = 1$ and 1.15 respectively. At the bottom (b), we plot the ground state energy $H_{\min}(\rho^*, R)$ as a function of R .

Unlike the nuclei, which are localized at single points, the superpositioned electron-clouds are smeared around the whole lattice. So, to calculate the inter-electronic repulsion part we consider an electron density (or a negative charge density) over an infinitesimally small line element dr at a location r which interacts with another electron density sitting over an infinitesimally small line element dr' at a location r' . The repulsion energy for these two infinitesimally small electron clouds is given by $dE_{ee} = \psi(r)\psi(r')\delta(r - r')\psi(r)\psi(r')$. Therefore, the total electron-electron interaction part is given by:

$$\begin{aligned}
 E_{ee} &= \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} dr dr' \psi(r)\psi(r')\delta(r - r')\psi(r)\psi(r') \\
 &= \frac{1}{\mathcal{N}^4} \int_{-\infty}^{\infty} dr \sum_{i,j,k,l} \phi_i(r)\phi_j(r)\phi_k(r)\phi_l(r) \\
 &= \frac{1}{\mathcal{N}^4} \int_{-\infty}^{\infty} dr \sum_{i,j,k,l} e^{-|r-iR|} e^{-|r-jR|} e^{-|r-kR|} e^{-|r-lR|}.
 \end{aligned} \quad (15)$$

We break the above expression (15) into many smaller terms and calculate all them individually as follows:

$$\begin{aligned}
 \int_{-\infty}^{\infty} dr \phi^4(r) &= \frac{1}{2}, \\
 \int_{-\infty}^{\infty} dr \phi_i^2(r)\phi_j^2(r) &= e^{-2|i-j|R} \left(\frac{1}{2} + |i-j|R \right), \text{ for } i \neq j, \\
 \int_{-\infty}^{\infty} dr \phi_i(r)\phi_j^3(r) &= \frac{3}{4} e^{-|i-j|R} - \frac{1}{4} e^{-3|i-j|R}, \text{ for } i \neq j, \\
 \int_{-\infty}^{\infty} dr \phi_i(r)\phi_j^2(r)\phi_k(r) &= \frac{1}{2} e^{(k-i)R} - \frac{1}{4} e^{(3k-i-2j)R} \\
 &\quad - \frac{1}{4} e^{(2j+k-3i)R}, \text{ for } i > j > k, \\
 \int_{-\infty}^{\infty} dr \phi_i(r)\phi_j^2(r)\phi_k(r) &= e^{(i+k-2j)R} \left[\frac{3}{4} + (j-i)R \right] \\
 &\quad - \frac{1}{4} e^{(3k-i-2j)R}, \text{ for } j > i > k, \\
 \int_{-\infty}^{\infty} dr \phi_i(r)\phi_j^2(r)\phi_k(r) &= e^{(2j-k-i)R} \left[\frac{3}{4} + (k-j)R \right] \\
 &\quad - \frac{1}{4} e^{(2j+k-3i)R}, \text{ for } i > k > j, \\
 \int_{-\infty}^{\infty} dr \phi_i(r)\phi_j(r)\phi_k(r)\phi_l(r) &= e^{(k+l-i-j)R} [1 + (k+j)R] \\
 &\quad - \frac{1}{4} e^{(j+k+l-3i)R} - \frac{1}{4} e^{(3l-i-j-k)R}, \\
 &\quad \text{for } i > k > j > l,
 \end{aligned}$$

and we use the multinomial theorem:

$$(x_1 + x_2 + \dots + x_m)^n = \sum_{k_1+k_2+\dots+k_m=n} \frac{n!}{k_1!k_2!\dots k_m!} \prod_{t=1}^m (x_t)^{k_t}$$

to calculate the interaction part in eqn (15).

We set the number of lattice points equal to $i, j, k, l = 100$ and calculate all the above quantities numerically and obtain the final energy as a function of the inter-atomic distance R .

For FCC or BCC lattices the calculations in $D = 1$ is slightly different from SC lattice in one dimension. For *e.g.* in case of BCC lattice in $D = \infty$ the Hamiltonian (6), the lattice parameters l, m, n are either even or odd integers.²² We choose the same conversion for the calculation in one dimension. In one dimension the FCC and BCC lattices are the same.

For MH in $D = 1$ with inter-electronic correlation there is no way to define dihedral angles between electrons. Therefore, we keep the Hamiltonian same as defined in eqn (11). In Fig. 4, we plot the energy per electron as a function of the inter-atomic distance R for simple cubic and FCC/BCC lattices.

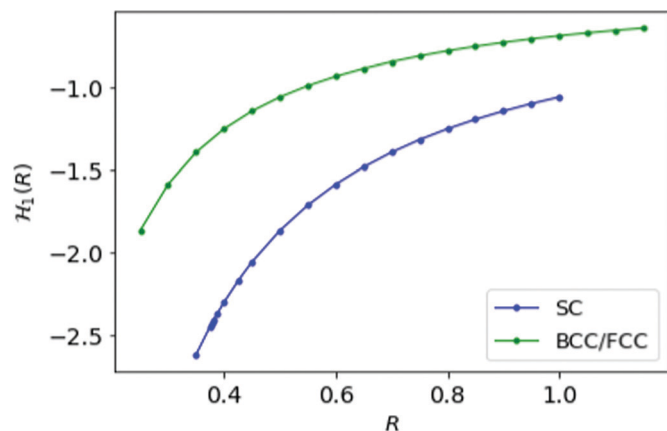


Fig. 4 We plot the energy $\mathcal{H}_1(R)$ for MH at $D = 1$ as a function of R ; blue curve is for SC lattice and green for FCC or BCC lattice.

5 Metallic hydrogen in $D = 3$ from dimensional interpolation

The dimensional interpolation formula described in Section 2 combines the $D = 1$ and $D = \infty$ limits and obtains the $D = 3$ reduced energy from

$$\varepsilon_3(R) = \frac{1}{3}\varepsilon_1\left(\frac{1}{3}R\right) + \frac{2}{3}\varepsilon_\infty\left(\frac{2}{3}R\right). \quad (16)$$

In Fig. 5, we plot the electronic energies involved for the metallic hydrogen in simple cubic lattice (SC). We compare our interpolation $\varepsilon_3(R)$ curve (blue) with points (red) that come from density functional theory³¹ (there Table III). The agreement is very good. As noted in Fig. 2, the inter-electronic correlation examined in the $D \rightarrow \infty$ limit turns quite minor. From Fig. 5 we see that the ground state energy at $D = 3$ becomes positive to negative around $R = 1.14$. Physically this means for $R < 1.14$ the ground state energy becomes positive

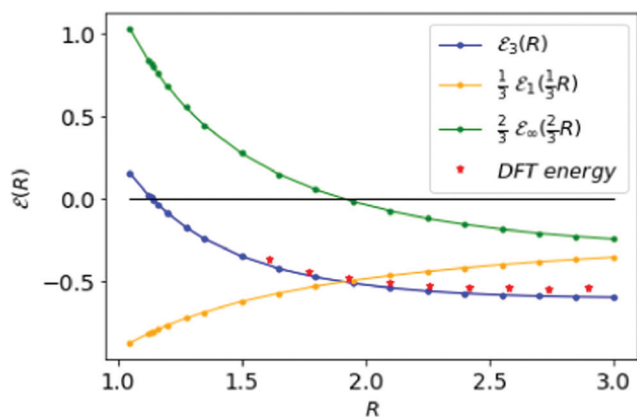


Fig. 5 The orange and green curves describe the results from $D = 1$ and $D \rightarrow \infty$ respectively. The blue curve energy $\varepsilon_3(R)$ represented the interpolation result at $D = 3$ for MH, in the SC lattice. The red points represent the results obtained from density functional theory in ref. 31. The black line shows the zero-energy line, where the blue curve goes through on $R = 1.14$ at a phase transition.

therefore makes the lattice is unstable or the system loses the crystalline structure. Therefore, at the point $R = 1.14$, the system goes through a phase transition.

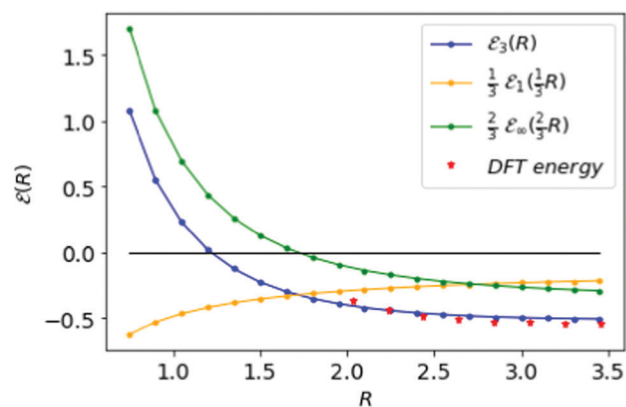
Fig. 6 displays the MH energies from the BCC and FCC lattices, similar to the SC lattice in Fig. 5, interpolating to $D = 3$ from eqn (16). The corresponding transition point for BCC and FCC lattice, where the ground state energy changes sign, is given by $R_{\text{FCC}} = 1.21 = R_{\text{BCC}}$.

6 Physical properties of metallic hydrogen

The scaled lattice constant R is related to r_s , the standard solid state parameter defined as the radius of a sphere (in a_0 Bohr units) in which contains on average one electron. For the SC lattice,

$$\frac{4}{3}\pi r_s^3 = R^3. \quad (17)$$

Thus, $r_s = 0.71$ corresponds to the transition point $R = 1.14$, for the $D \rightarrow \infty$ limit. Many studies for $D = 3$ have obtained



(a)



(b)

Fig. 6 The top panel (a) is for the BCC lattice, the bottom panel (b) for the FCC lattice. Compare results in Fig. 5 for the SC lattice. Both of the blue curves meet the black line at zero-energy on $R = 1.14$ at the same phase transition for all three lattices.

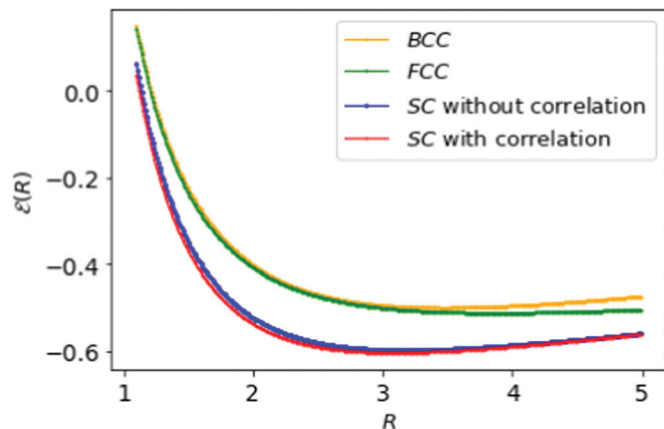


Fig. 7 The simple formula, eqn (18), fits neatly the energies of MH in Table 1: SC with HF (in blue) and with correlation (in red); BCC (in orange), and FCC (in green) lattice structures respectively.

$r_s = 0.8$ for the existence of crystalline phase of metallic hydrogen.³² However, we have simply used R when the interpolation formula provides appropriate energies of metallic hydrogen for different lattice symmetries and parameters.

In this section, we are allowed to calculate different physical quantities such as the bulk modulus, Debye temperature, and critical transition temperature, from the gradient and the curvature of the energy curve as a function of the lattice parameters. Thus, the numerical results of the interpolation formula can be fitted to the following functional form as a function of the lattice parameter R ,^{33,34} driven in a following table and Fig. 7.

$$\mathcal{E}(R) = \frac{A}{R^2} + \frac{B}{R} + C - D \ln(R), \quad (18)$$

with the parameters given in the following Table 1:

We also calculate the \mathcal{E}_{\min} and R_{\min} from eqn (18) and in Table 2 compare them for the lattice structures:

The minimum energies of the MH lattices are evident in Fig. 7 and the energy differences are $\Delta_{\text{FCC-SC}} = 0.091$, $\Delta_{\text{BCC-SC}} = 0.104$, $\Delta_{\text{FCC-BCC}} = 0.013$ respectively. The energy differences between FCC and BCC are very modest compared to SC. That may lead to possibility for a phase transition from BCC to FCC structure or *vice versa*.

Now we examine briefly consequential properties that involve from the interpolated formula (eqn (16)) for MH energy.

Table 1 Parameters describing $\mathcal{E}(R)$ in different lattice structures

Lattice	A	B	C	D
SC in HF limit	1.3236	-0.0975	-0.9650	-0.2311
SC with correlation ^a	1.3458	-0.1689	-0.9457	-0.2264
BCC	0.7889	0.7273	-1.1974	-0.3397
FCC	1.1628	-0.1070	-0.7329	-0.1265

^a See Fig. 2 has added SC with inter-electronic correlation. Others with Hartree-Fock (HF).

Table 2 \mathcal{E}_{\min} and R_{\min} in different lattice structures

Lattice structure	R_{\min}	\mathcal{E}_{\min}
SC in HF limit	3.18	-0.597
SC with correlation	3.09	-0.603
BCC	3.48	-0.499
FCC	3.88	-0.512

First is pressure:

$$P = -\frac{d\mathcal{E}}{dV} = -\frac{\eta}{3R^2} \frac{d\mathcal{E}}{dR}, \quad (19)$$

with η = number of atoms in a unit cell. And we see from the graph that around $R = 3.18$, where $\mathcal{E} = \mathcal{E}_{\min}$, the pressure changes sign; the corresponding $r_s = 1.9$. This transition has a physical significance. Although at high densities the crystalline phase is preferred for metallic hydrogen, however at low densities ($r_s > 1.6$) the metallic hydrogen behaves like a fluid or liquid metal.^{32,35,36} This particular point $R = 3.18$ and $\mathcal{E} = -0.597$ possibly signifies that crystalline-fluid transition point for MH in simple cubic lattice (Fig. 8).

From the curvature of the energy curves in Fig. 7, we can calculate the elastic modulus of MH for different structures as the bulk modulus B of a cubic metallic lattice is given by

$$B = -V \frac{dP}{dV}, \quad (20)$$

with pressure, P , and volume of the unit cell $V = R^3$. The shear modulus G can be calculated from the same equation.³⁷ Another important quantity that is available from the MH energy *via* pressure is the Debye temperature Θ ; the formula^{38,39} involves quite a few items:

$$\Theta = \frac{h}{k} \left(\frac{3N_A \rho}{4\pi M} \right)^{1/3} v_m, \quad (21)$$

where h is the Planck constant, k is the Boltzmann constant, N_A is the Avogadro constant, M is the atomic weight, and ρ is the density, and includes B , the bulk modulus as well G , the shear

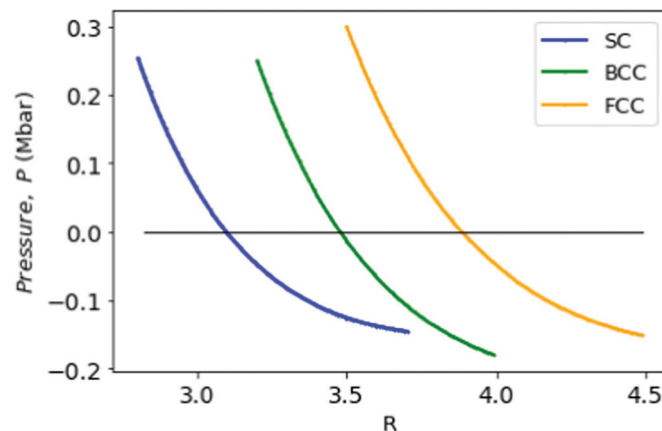


Fig. 8 We plot the pressure P in Mbar for MH in SC (in blue), BCC (in green), and FCC (in orange) lattices as a function of R .

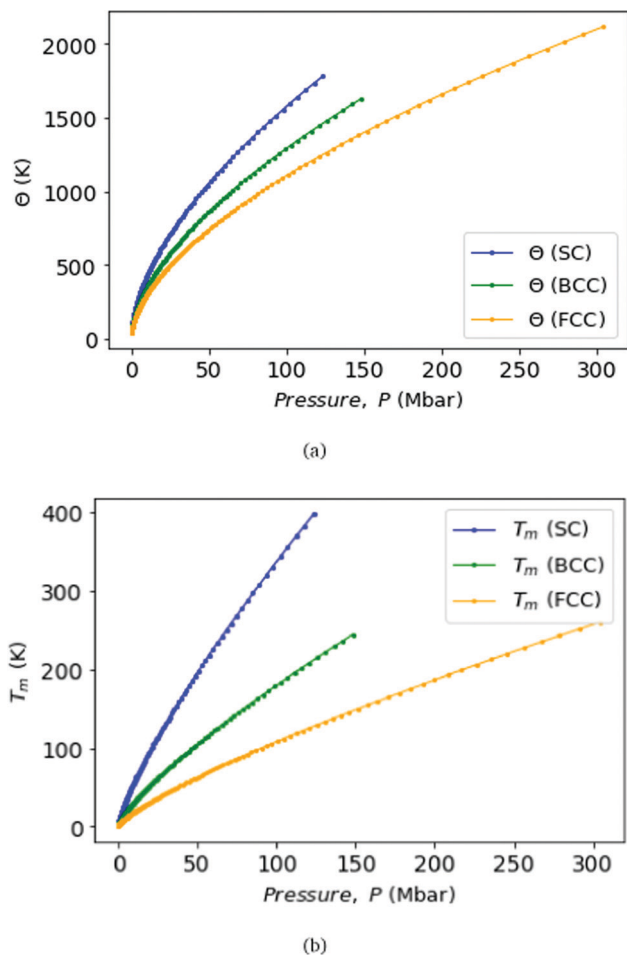


Fig. 9 We plot the Debye temperatures Θ (top panel) and melting points T_m (bottom panel) of MH in SC, BCC, and FCC lattice structure with the correlation with respect to the pressure, P .

modulus, and

$$v_m = \left[\frac{1}{3} \left(\frac{2}{v_s^3} + \frac{1}{v_p^3} \right) \right]^{-1/3}; v_s = \sqrt{\frac{G}{\rho}}; v_p = \sqrt{\frac{B + 4G/3}{\rho}}. \quad (22)$$

Fig. 9 displays Debye temperature Θ (top) and melting temperature T_m (bottom) as functions of pressure for MH lattices, SC, BCC, and FCC. As well known, MH will climb to a very high Debye temperature, and could become a quantum liquid. Melting curves of metallic hydrogen as a function of pressure *via* the Lindemann melting law.^{40,41} Some samples: for SC lattice at $R = 3.1$, $\Theta = 57$ K, and at $R = 1.12$ obtains $\Theta = 1779$ K at $P = 118$ Mbar. For BCC lattice at $R = 3.48$, $\Theta = 52$ K, and at $R = 1.21$ obtains $\Theta = 1627$ K at $P = 147$ Mbar. For FCC lattice at $R = 3.88$, $\Theta = 41$ K, whereas at $R = 1.21$ obtains $\Theta = 2114$ K at $P = 294$ Mbar.

Finally, as many authors predicted that metallic hydrogen is a superconductor below some critical temperature and others even argued that it might be a superconductor up to a room temperature.^{42,43} Following Koblishka, *et al.*,⁴⁴ we have used our MH energy curves *via* interpolation from eqn (16) and

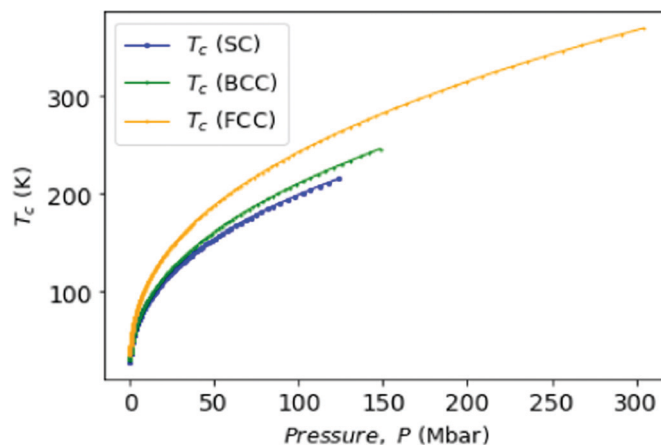


Fig. 10 We plot the transition temperature T_c as a function of pressure, P .

formulate the transition temperature T_c for a MH superconductor:

$$T_c = \frac{h^2}{(2x)^2 2M_L n^{-2/3} \pi k}. \quad (23)$$

Here again h is the Planck constant; k is the Boltzmann constant; x is the atomic distance. A correction factor n is usually taken to be 1 for metals, and M_L is taken to be equal to the mass of a proton m_p .

In Fig. 10, we plot the possible values of transition temperature T_c calculated from eqn (23) as a function of pressure for MH in SC, BCC, and FCC lattices, within a range where the energy is negative and pressure is positive. For the atomic distances, $x = R$ for SC; $x = (\sqrt{3}/2)R$ for BCC; $x = R/\sqrt{2}$ for FCC. Samples: For SC lattice at $R = 3.1$, $T_c = 28$ K whereas for $R = 1.12$, $T_c = 215$ K and $P = 118$ Mbar. For BCC lattice at $R = 3.48$, $T_c = 30$ K whereas at $R = 1.21$, $T_c = 246$ K, and $P = 147$ Mbar. For FCC lattice at $R = 3.88$, $T_c = 36$ K whereas for $R = 1.21$, $T_c = 369$ K and $P = 294$ Mbar.

Although the experimental verification of superconductivity in MH is yet to be confirmed, but from the above theoretical calculations we see that the metallic hydrogen is a very good candidate for a high temperature superconductor.

7 Conclusion

The simplicity of the $D \rightarrow \infty$ limit causes the disappearance of derivatives from a Hamiltonian as well as $h \rightarrow 0$, so is a true classical limit.²⁷ It is different from a semiclassical approximation such as WKB theory for small h . The simplicity of the $D = 1$ limit keeps derivatives in a Hamiltonian and is a true hyperquantum limit. Combining these extreme partner limits delivers the dimensional interpolation formula. It was tried out with two-electron atoms²⁰ and generalized out with few electron atoms and simple diatomic molecules.²³ Here we find the interpolation approach is appropriate for metallic hydrogen. With beginning by Loeser,²² we find the interpolation not only provided adequate energies but also the correct function forms of symmetry and lattice parameters. From the gradient and the

curvature of the energy curves as a function of the lattice parameter R , we were able to calculate some important physical quantities. Among are the bulk and shear moduli, and three temperatures governed by pressure: the Debye temperature, the Lindemann melting temperature, and the critical transition temperature for superconductivity.

It is relatively easy to calculate the $D \rightarrow \infty$ and $D = 1$ limits, so the interpolation formula can predict results for the physical dimension, $D = 3$. Therefore, D -scaling might approach the electronic structure of strongly correlated systems, where often traditional approaches are faced with computational difficulties!

Conflicts of interest

There are no conflicts to declare.

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References

- 1 E. Wigner and H. B. Huntington, *J. Chem. Phys.*, 1935, **3**, 764–770.
- 2 R. J. Hemley and H. K. Mao, *Phys. Rev. Lett.*, 1988, **61**, 857.
- 3 R. J. Hemley, M. Hanfland and H. K. Mao, *Nature*, 1991, **350**, 488–491.
- 4 R. J. Hemley, H. k. Mao, A. F. Goncharov, M. Hanfland and V. Struzhkin, *Phys. Rev. Lett.*, 1996, **76**, 1667.
- 5 M. Zaghoo, A. Salamat and I. F. Silvera, *Phys. Rev. B*, 2016, **93**, 155128.
- 6 I. F. Silvera and R. Dias, *J. Phys.: Condens. Matter*, 2018, **30**, 254003.
- 7 P. Loubeyre, F. Occelli and P. Dumas, *Nature*, 2020, **577**, 631–635.
- 8 E. Babaev and N. W. Ashcroft, *Nat. Phys.*, 2007, **3**, 530–533.
- 9 E. Babaev, A. Sudbø and N. W. Ashcroft, *Nature*, 2004, **431**, 666–668.
- 10 E. Babaev, *Phys. Rev. Lett.*, 2002, **89**, 067001.
- 11 D. R. Herschbach, J. S. Avery and O. Goscinski, *Dimensional Scaling in Chemical Physics*, Springer Science & Business Media, 1993.
- 12 D. Z. Goodson, M. López-Cabrera, D. R. Herschbach and J. D. Morgan III, *J. Chem. Phys.*, 1992, **97**, 8481–8496.
- 13 Z. Zhen and J. Loeser, *Large-D Limit for N-Electron Atoms*, Springer, 1993, pp. 83–114.
- 14 S. Kais and D. R. Herschbach, *J. Chem. Phys.*, 1994, **100**, 4367–4376.
- 15 J. Rudnick and G. Gaspari, *Science*, 1987, **237**, 384–389.
- 16 J. G. Loeser, Z. Zhen, S. Kais and D. R. Herschbach, *J. Chem. Phys.*, 1991, **95**, 4525–4544.
- 17 S. Kais and D. R. Herschbach, *J. Chem. Phys.*, 1993, **98**, 3990–3998.
- 18 Q. Wei, S. Kais and D. Herschbach, *J. Chem. Phys.*, 2008, **129**, 214110.
- 19 Q. Wei, S. Kais and D. Herschbach, *J. Chem. Phys.*, 2007, **127**, 094301.
- 20 D. R. Herschbach, J. G. Loeser and W. L. Virgo, *J. Phys. Chem. A*, 2017, **121**, 6336–6340.
- 21 T. C. Germann and S. Kais, *J. Chem. Phys.*, 1993, **99**, 7739–7747.
- 22 J. G. Loeser, *Large-D Limit for Metallic Hydrogen, Dimensional Scaling in Chemical Physics*, Springer, 1993, pp. 389–427.
- 23 K. J. B. Ghosh, S. Kais and D. R. Herschbach, *Front. Phys.*, 2020, **8**, 331.
- 24 D. D. Frantz and D. R. Herschbach, *Chem. Phys.*, 1988, **126**, 59–71.
- 25 A. L. Tan and J. G. Loeser, in *Simple molecules and variant scalings*, ed. D. R. Herschbach, J. Avery and O. Goscinski, Springer Netherlands, Dordrecht, 1993, pp. 230–255.
- 26 M. López-Cabrera, A. L. Tan and J. G. Loeser, *J. Phys. Chem.*, 1993, **97**, 2467–2478.
- 27 J. G. Loeser, J. H. Summerfield, A. L. Tan and Z. Zheng, *J. Chem. Phys.*, 1994, **100**, 5036–5053.
- 28 M. Born and R. D. Misra, *Math. Proc. Cambridge Philos. Soc.*, 1940, 466–478.
- 29 I. R. Lapidus, *Am. J. Phys.*, 1975, **43**, 790–792.
- 30 I. R. Lapidus, *Am. J. Phys.*, 1982, **50**, 453–456.
- 31 B. I. Min, H. J. F. Jansen and A. J. Freeman, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1984, **30**, 5076–5083.
- 32 K. K. Mon, G. V. Chester and N. W. Ashcroft, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1980, **21**, 2641–2646.
- 33 P. Nozières and D. Pines, *Phys. Rev.*, 1958, **111**, 442–454.
- 34 E. E. Salpeter and H. S. Zapolsky, *Phys. Rev.*, 1967, **158**, 876–886.
- 35 P. M. Robitaille, *Prog. Phys.*, 2011, **7**, 4005.
- 36 S. A. Bonev, E. Schwegler, T. Ogitsu and G. Galli, *Nature*, 2004, **431**, 669–672.
- 37 K. K. Phani and D. Sanyal, *Mater. Sci. Eng., A*, 2008, **490**, 305–312.
- 38 O. L. Anderson, *J. Phys. Chem. Solids*, 1963, **24**, 909–917.
- 39 C. Li and Z. Wang, in *Advances in Science and Technology of Mn + 1AXn Phases*, ed. I. Low, Woodhead Publishing, 2012, pp. 197–222.
- 40 M. Ross and C. Shishkevish, *Molecular and metallic hydrogen*, Tech. Rep. (RAND CORP SANTA MONICA CA), 1977.
- 41 V. P. Trubitsyn, *Sov. Astron.*, 1971, **15**, 303.
- 42 N. W. Ashcroft, *Phys. Rev. Lett.*, 1968, **21**, 1748–1749.
- 43 B. Cheng, G. Mazzola, C. J. Pickard and M. Ceriotti, *Nature*, 2020, **585**, 217–220.
- 44 M. R. Koblischka, S. Roth, A. Koblischka-Veneva, T. Karwoth, A. Wiederhold, X. L. Zeng, S. Fasoulas and M. Murakami, *Metals*, 2020, **10**, 158.