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REVIEW

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1. Introduction

Glycomimetics, also known as pseudosugars, are chemical entities that are structurally similar to sugars. As sugars play important roles in various structural and physiological aspects of living organisms, many glycomimetics have significant biological activities.¹ Consequently, some of them have been developed as drugs to treat diseases in humans, animals, and plants. Six-membered ring pseudosugar compounds such as the aminoglycoside antibiotics (e.g., streptomycin (1), gentamicin, and neomycin) have been used for more than a half century for the treatment of various infectious diseases.² The pseudo-**EXAMPLE WE THE Chemistry and biology of natural**
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The chemistry and biology of natural ribomimetics and related compounds

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Natural ribomimetics represent an important group of specialized metabolites with significant biological activities. Many of the activities, e.g., inhibition of seryl-tRNA synthetases, glycosidases, or ribosomes, are manifestations of their structural resemblance to ribose or related sugars, which play roles in the structural, physiological, and/or reproductive functions of living organisms. Recent studies on the biosynthesis and biological activities of some natural ribomimetics have expanded our understanding on how they are made in nature and why they have great potential as pharmaceutically relevant products. This review article highlights the discovery, biological activities, biosynthesis, and development of this intriguing class of natural products.

> oligosaccharide acarbose (2) as well as the iminosugar miglitol (3) (a synthetic derivative of deoxynojirimycin) have been used to treat patients with type 2 diabetes (Fig. 1).^{3,4} Two other synthetic iminosugars, miglustat (4) and migalastat (5), are used to treat Gaucher's disease type 1 and Fabry disease, respectively.^{5,6} Furthermore, the pseudo-oligosaccharide validamycin A (6) is used as a crop protectant against fungal infections.⁷

> In addition to the six-membered pseudosugars, many natural and synthetic compounds contain five-membered pseudosugar units, known as ribomimetics. Structurally, they resemble ribose, deoxyribose, or arabinose, or contain a highly modified ribose, which contributes to their various biological activities. A number of synthetic ribomimetics have also made their way to clinical use, e.g., the anti-influenza drug peramivir (7), the anti-hepatitis B drug entecavir (8), and the HIV nucleoside reverse

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mimetics. He is also interested in understanding how those compounds impact the producing organisms and/or other organisms in nature.

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Fig. 1 Chemical structures of six-membered ring pseudosugars.

transcriptase inhibitors lamivudine (9) and abacavir (10) (Fig. 2). In addition, many natural ribomimetics have attracted significant attention from scientists due to their promising biological activities.

Naturally occurring ribomimetics and related compounds may be classified into two groups. The first group consists of ribose-like compounds in which the ring oxygen of ribose, deoxyribose, arabinose and related five-membered ring sugars has been replaced by a sulphur, a nitrogen, or a carbon. Accordingly, they are classified as S-, N-, and C-ribomimetics. Members of this family of natural products include the albomycins (11–13), 1,4-dideoxy-1,4-iminoarabinitol (DAB-1, 14), aristeromycin (15) and pactamycin (16) (Fig. 3). The second group of ribomimetics consists of nucleotide/nucleoside antibiotics that still contain a ribose backbone, but with a highly modified C-5. This group of ribomimetics includes nikkomycin (17), caprazamycin B (18), and tunicamycin (19) (Fig. 3). A number of outstanding review articles describing the latter compounds (the nucleotide/nucleoside antibiotics) have been published recently; $8-10$ therefore, they are

excluded from this article, which focuses on the discovery, biological activities, biosynthesis, and development of the S-, N-, and C-ribomimetics and related compounds.

2. S-Ribomimetics and related compounds

Natural S-ribomimetics, also known as thiofuranoses, have been found in bacteria and plants (Table 1). On the basis of the known natural thiofuranoses identified so far, they may be classified into two structural groups, the albomycins (11–13) and the salacinols (20–27) (Fig. 4). The albomycins are a group of peptidyl nucleoside analogues containing a peptide-derived siderophore unit and a 6'-amino-4'-thioheptose nucleoside moiety connected by an amide bond (Fig. 3).¹¹ The salacinols, including kotalanol (22), ponkoranol (24), salaprinol (26), and other related compounds, are a group of plant-derived thiosugar sulfonium compounds, some of which contain sulfonium sulfate inner salt structures.

2.1. Discovery and biological activities of S-ribomimetics

The peptidyl nucleoside analogue albomycins have been isolated from several actinomycetes. $12,13$ They showed antibiotic activity against Gram-positive and Gram-negative bacteria, $14-16$ most likely by inhibiting bacterial seryl-tRNA synthetases.17 The thioheptose unit of the compounds is believed to be important

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for this activity, whereas the siderophore unit is important for the compounds to get into bacterial cells by utilizing the bacterial iron uptake system. After getting into the bacterial cells, the amide bond is hydrolysed by endogenous peptidases releasing the active thioheptose nucleoside unit (Fig. 5). Due to this unique mechanism, the albomycins have been branded as ''trojan horse'' antibiotics.18–20

The plant-derived S-ribomimetic compound salacinol (20) was first isolated from the roots and stems of Salacia reticulata,

which is traditionally used in Sri Lanka for the treatment of diabetes. 21 Consistent with the empirical use of the plant as an anti-diabetic, the active compound salacinol does have strong α -glucosidase inhibitory activity.²¹ In addition to salacinol, several of its analogues, e.g., neosalacinol $(21),^{22,23}$ kotalanol $(22),^{24,25}$ neokotalanol $(23),^{26}$ ponkoranol $(24),^{27}$ neoponkoranol (25) ,²⁸ salaprinol (26) ,²⁹ and neosalaprinol (27) ,²⁸ were also isolated from S. reticulata or a related species, Salacia oblonga (Fig. 4). All of them share the same S-ribomimetic core structure, but with different polyalcohol side chains with various chain lengths (C3–C7), either sulfonated or non-sulfonated. The antidiabetic effects of Salacia plants were reviewed very recently, describing the basic chemical and pharmacological research of the salacinols, as well as their applications as functional food ingredients.³⁰

2.2. Biosynthesis of S-ribomimetics

Despite their interesting chemical structures, no biosynthetic studies of the salacinols have been reported in the literature. On the other hand, studies on albomycin biosynthesis have been conducted by several groups. The biosynthetic gene cluster (BGC) of albomycin (the abm cluster) was first reported from Streptomyces sp. ATCC 700974, along with a related gene

Fig. 5 Albomycin enters a bacterial cell using the bacterial iron uptake system. The siderophore unit is important for the compound to get into bacterial cells by utilizing the bacterial iron uptake system. First, the ferric siderophore passes the outer membrane through the ferrichrome outer membrane transporter FhuA. In the periplasm, the ferrichrome binds FhuD, which is then transported into the cytoplasm through the membrane channel FhuBC. In the cytoplasm, the amide bond is hydrolysed by the endogenous peptidase PepN releasing the active thioheptose nucleoside unit, which inhibits seryl-tRNA synthetase (SerRS). OM, outer membrane; IM, inner membrane.

cluster (the ctj cluster) postulated for an albomycin analogue from another strain of *Streptomyces* (Fig. 6).³¹

While the formation of the thiofuranose core unit has not been fully determined, it has been proposed that the key reaction that incorporates sulfur into the sugar moiety may adopt a radical-mediated mechanism analogous to the biotin synthase BioB.³² In fact, in the albomycin biosynthetic gene cluster, the gene abmM encodes a radical SAM enzyme containing both [4Fe–4S] and [2Fe–2S] cluster binding motifs, indicating that AbmM may be involved in the formation of the thiofuranose core unit in albomycin biosynthesis. However, no biochemical data are available to support this notion.

On the other hand, gene inactivation and in vitro biochemical assays of several other enzymes from the albomycin pathway did shed some light on the overall biosynthesis of the antibiotics (Fig. 7). The PLP-dependent transaldolase homologue AbmH was found to be responsible for the incorporation of L-threonine into the thionucleoside core unit, and the PLPdependent epimerase AbmD catalyses the epimerization at $C-6$ ['] (Fig. 7).^{33,34} Inactivation of the *abmE* and *abmI* genes revealed the catalytic functions of AbmE and AbmI to be a N-carbamoyltransferase and a SAM-dependent methyltransferase that catalyse the N-carbamoylation and the methylation of the

nucleobase, respectively. AbmI converted cytidine to 3-Nmethylcytidine in an in vitro assay, indicating that AbmI is indeed involved in the modification of the cytidine moiety of albomycin.³¹ However, the order of these biotransformations is still unclear.

The hypothetical protein AbmF, which is similar to aminoacyltRNA synthetases, has been shown to catalyse the extension of the AbmE/AbmI product with L-serine, whereas the serine adenylase AbmK has been shown to play a role in supplying the amino acid for this reaction. The product (compound 32) is then epimerized at C-3' by the radical S-adenosyl-L-methionine (SAM) enzyme AbmJ to give SB-217452 (33). This sequence of reactions was biochemically confirmed, as the recombinant AbmF/AbmK proteins incubated with L-Ser and ATP can only use the AbmI/AbmE product (31), but not its $3'$ epimer,³⁴ whereas AbmJ can only

Fig. 7 Biosynthetic pathway to albomycins.

epimerize the product of AbmF (32), but not the product of AbmI/ AbmE (31).

The formation of the siderophore (ferrichrome) moiety and the latter steps of albomycin biosynthesis have also been investigated.³⁴ The ferrichrome building block N^5 -acetyl- N^5 hydroxy-L-ornithine (35) is biosynthesized from L-ornithine (34) catalysed by the N-acetyltransferase AbmA and the flavindependent monooxygenase AbmB (Fig. 7), whereas the assembly of the tripeptide moiety is catalysed by the NRPS protein AbmQ. Interestingly, AbmQ, which contains two condensation (C) domains, lacks a C-terminal thioesterase (TE) domain, suggesting that no free siderophore is released from AbmQ. It was also hypothesized that AbmC, which shows low similarity to ATP-dependent proteases/ligases, is responsible for off-loading the tripeptide and directly transferring it to the corresponding intermediate SB-217452 (33). Inactivation of the abmC gene in the albomycin producing strain indeed abolished the production of albomycin in this mutant, and 33 accumulated in the culture, indicating that AbmC is responsible for the ligation of the ferrichrome moiety to SB-217452 (33). RSC Chemical Biology

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2.3. Development of S-ribomimetics

Besides being used as anti-diabetic medications in Ayurvedic medicine, not many S-ribomimetics are used as clinically approved medications. Nevertheless, lamivudine (9), which contains both oxygen and sulfur atoms within the fivemembered ring, has been widely used as an anti-HIV and anti-hepatitis B medication. Several salacinol analogues have been synthesized and tested for their human maltase and glucoamylase inhibitory activity and anti-infective activity.³⁵ In addition, a number of thiofuranoses have been developed as nucleic acid-based therapeutics.³⁶ The replacement of ribose with S-ribose has been shown to increase serum stability and enhance silencing longevity of siRNAs.

Many deoxynucleoside analogues have also been developed as antineoplastic/anticancer drugs, as they interfere with DNA replication in cells to ultimately cause DNA damage and trigger apoptosis.³⁷ Among them is thiarabine (4'-thio-ß-D-arabinofuranosylcytosine or T-araC) (37), an analogue of the anticancer drug cytarabine (ara-C, 36) that contains thiofuranose instead of arabinose (Fig. 8).^{38,39}

Despite the similarity of their structures and mechanisms of action, T-Ara-C (37) shows dramatically higher antitumour activity than Ara-C (36) when tested against human tumour cells in mice.40 This may be due to quantitative differences in biochemical or metabolic activity of T-Ara-C (37) versus Ara-C (36).40 T-Ara-CTP (a triphosphate version of thiarabine) has \sim 10 times longer half-life in solid tumour cells compared to Ara-C (36) ,⁴¹ demonstrating how a simple ring heteroatom replacement of a sugar moiety can alter biological activity.

Other thiofuranoses that are currently in the clinical development are 4'-thio-2'-deoxycytidine (T-dCyd, 38) and 5-aza-TdCyd (39) (Fig. 8). 42 T-dCyd (38) depletes DNA methyltransferase 1 (DNMT1), both in vitro and in vivo in tumour cells. In nonsmall lung NCI-H23 xenograft studies, T-dCyd (38) inhibited tumour growth with concomitant depletion of DNMT1 at well

Fig. 8 Chemical structures of cytarabine and its thiofuranose analogues

tolerated doses.⁴² Unfortunately, the therapeutic index of 38 is relatively narrow, leading to the suspension of its clinical study. However, its analogue, 5-aza-T-dCyd (39), appears to be more promising, as it is 3- to 4-fold more potent than 38 in various human tumour cells with a larger therapeutic index in mice bearing CEM xenografts. Results from an on-going phase I clinical trial suggest that 5-aza-T-dCyd (39) is safe and well tolerated with a toxicity profile similar to currently approved hypomethylating agents.⁴³ It is important to note that the therapeutic index of 39 is much larger than those of the currently approved drugs 5-aza-dCyd and 5-aza-Cyd, which are normally administered for short periods followed by a long recovery period. This suggests that substitution of oxygen for sulfur in the furanose moiety of nucleosides and deoxynucleosides is beneficial for improving their pharmacological properties.

3. N-Ribomimetics and related compounds

3.1. Discovery and bioactivities of N-ribomimetics

N-Ribomimetics are defined as ribose-like compounds in which the ring oxygen atom has been replaced by a nitrogen. Here, we define N-ribomimetics as nitrogen-containing five-membered rings that bear one or more hydroxy groups – to discriminate them from other pyrrolidine alkaloids. Compared to Sribomimetics, N-ribomimetics appear to be more common in nature. Many N-ribomimetics have been isolated from a variety of sources including bacteria, fungi, and plants (Table 2). Among them are the broussonetines (40–47) (Fig. 9), which were isolated from the moraceous plant Broussonetia kazinoki.⁴⁴⁻⁴⁸ Broussonetines have been shown to have inhibitory activity against various glycosidases.^{49,50}

Some other plants also produce N-ribomimetics. For example, the Asian bell flower plant Codonopsis clematidea produces codonopsinine (48), codonopsine (49), and codonopsinol (50) ,^{51–53} all of which showed good glycosidase inhibitory Table 2 Selected natural N-ribomimetics

activity.⁵⁴ Codonopsinine (48) and codonopsine (49) also have antibacterial and anti-hypertensive activities.^{52,54} Radicamines (51–52), analogues of codonopsinine, are produced by the traditional Chinese medicinal plant Lobelia chinensis.^{55,56} Similar to codonopsinine, the radicamines also show inhibitory activity against glycosidases.⁵⁴ A related metabolite, tinctormine (53), is produced by the asteraceous plant Carthamus tinctorius, which is also known as safflower. Tinctormine showed Ca^{2+} channel antagonistic activity.⁵⁷

The broad-spectrum glycosidase inhibitor 1,4-dideoxy-1,4 iminoarabinitol (DAB-1, 14) is a widespread plant secondary metabolite that inhibits several glycosidases including α -glucosidases, α -D-arabinosidase, β -glucosidases and α -mannosidases.⁵⁸ This simple N-ribomimetic natural product has been isolated from both tropical and temperate plants.⁵⁸ More recently, DAB-1 was also identified in the bacterium Chitinophaga pinensis DSM 2588. In addition, several DAB-1 analogues, e.g., 2-epi-DAB-1 (54), 2-deoxy-DAB-1 (CYB3, 55), and N-hydroxyethyl-2-deoxy-DAB-1 (56) (Fig. 10), have been isolated from Morus spp.⁵⁹ and Castanospermum australe.^{60,61} FR900483 (57), an unsaturated-DAB (Fig. 10), was isolated from the fungus Nectria lucida F-4490.⁶² The latter compound, a.k.a. nectrisine, showed immunomodulator activity

in proliferatively suppressed lymphocytes and immunosuppressed mice.⁶² It also showed potent α -glucosidase inhibitory activity.⁶³

Crystal structures of Bombyx mori a-glucosidase (BmSUH) in complex with DAB-1 or sucrose, as well as those of α -Larabinofuranosidase in complex with DAB-1 or arabinose showed that DAB-1 binds to the active site of both enzymes and functions as a competitive inhibitor of the enzymes (Fig. 11).^{64,65}

Some plants and bacteria also produce 2,5-dideoxy-2,5 imino-p-mannitol (DMDP, 58) and related compounds, e.g., 6deoxy-DMAP $(2,5\text{-imino-1},2,5\text{-trideoxy-p-mannitol}, 59)$ ⁶⁶ homo-DMAP (2,5-dideoxy-2,5-imino-DL-glycero-D-manno-heptitol, 60),⁶⁷ 3,6-deoxy-homo-DMDP (61) , 68,69 and 2,5-imino-2,5,6-trideoxy-pgulo-heptitol (62) (Fig. 12).^{58,70,71} This class of compounds is widely distributed in nature as they have been isolated from unrelated families of both tropical and temperate plants as well as from strains of *Streptomyces* soil bacteria.⁵⁸ Consistent with their chemical structures, they showed inhibitory activity against

various glycosidases including glucosidases, mannosidases, galactosidases, and fucosidases.58 More recently, DMDP N-acetic acid (63) was isolated from Stevia rebaudiana, a commercially important plant – a source of the natural sweetener steviol. Although such derivatives with acetic and propionic acids have been detected by GC-MS in a broad range of plants, it was the first isolated natural amino acid derived from an iminosugar.⁷² Compared to DMDP (58), which showed strong inhibitory activity against β -galactosidases, DMDP N-acetic acid (63) is more active against α -galactosidases.⁷²

Besides plants, fungi have also been known as producers of N-ribomimetics. The mycosporine compound collemin A (64), which possesses sunscreen activity, was isolated from the lichenized ascomycete Collema cristatum (Fig. 13).⁷³ Collemin A (64) absorbs UV-B and prevents UV-B induced cell destruction and erythema when applied to the skin prior to UV-B irradiation.⁷³ Other fungal-derived N-ribomimetics include the pramanicins (65 and 66) and the virgaricins (67 and 68) (Fig. 13). Although their core structure is effectively a γ -lactam ring, due to a high degree of hydroxylation, their structures fall into our definition of N-ribomimetics. Thus, they are included in this review.

The pramanicins, which contain a long alkyl or alkenyl side chain, were isolated from the fungus Stagonospora sp. ATCC 74235. They showed antifungal activity toward Candida albicans KF1 and Cryptococcus neoformans.^{74,75} An analogue of pramanicin,

Fig. 11 Crystal structures of glycosidases and DAB-1. (A) binding model of sucrose (Suc) with BmSUH (left; PDB code, 6LGF) and DAB-1 (DAB) with BmSUH (right; PDB code, 6LGD); (B) binding of arabinose (Ara) with α -Larabinofuranosidase (left; PDB code, 6zpy) and DAB-1 with α -Larabinofuranosidase (right; PDB code, 6zq1).

kiliense Grutz TC 1703, as part of a screening effort to find interleukin-4 inhibitors.⁷⁶ Virgaricin (67) , another analogue of pramanicin, was isolated from Virgaria sp. FKI-4860. This compound showed modest antibacterial activity toward Bacillus subtilis KB 211.⁷⁷ Virgaricin B (68) , which was isolated from Virgaria boninensis FKI-4958, also showed antibacterial activity against E. coli KB 213 and B. subtilis KB $211.^{78}$

The bacterial-derived N-ribomimetic anisomycin (70) was isolated from several Streptomycetes including Streptomyces roseochromogenes, S. griseolus, and S. hygrospinosus (Fig. 14).⁷⁹ Anisomycin (70) showed various biological activities including toxicity toward pathogenic protozoa, fungi, and human tumour

Fig. 13 Chemical structures of fungal-derived N-ribomimetics.

Fig. 14 Chemical structures of bacterial N-ribomimetics

cell lines, as well as activation of the mitogen-activated protein kinase signalling pathways. Another bacterial N-ribomimetic, gualamycin (71), was isolated from the broth of Streptomyces sp. NK1168. This gulosamine- and galactose-containing glycoside showed potent acaricidal activity with no phytotoxicity to kidney bean (Fig. 14).^{80,81}

3.2. Biosynthesis of N-ribomimetics

Known natural N-ribomimetics have been shown to derive from different precursors or pathways, e.g., the acetate pathway, the pentose phosphate pathway, or the glycolytic pathway. The nitrogen atoms generally originate from amino acids, either incorporated intact or functioning as a source of nitrogen through the catalysis of an aminotransferase.

3.2.1. Biosynthesis of broussonetines and pramanicin. The biosynthesis of the broussonetines was studied by supplementing the cell culture of the plant *Broussonetia kazinoki* with $[1 - 13C]$ glucose (72) and the secondary metabolite broussonetine J (46) was isolated and analyzed.⁸²

Incorporation of 13 C atoms into the 1, 4, 6, 8, 10, 12, 14, 16 and 18 positions, as determined by NMR, indicated that the C-4–C-16 of broussonetine J originates from palmitoyl-CoA (77), which may be synthesized from glucose via the acetatemalonate pathway. The result also suggested that the C-1–C-3 unit of broussonetine J is derived from serine (76), as C-3 of serine normally originates from C-1 of glucose (Fig. 15). However, no incubation experiments using isotopically labelled serine have been performed to confirm this finding.

However, in the biosynthetic studies of pramanicin (65) in Stagonospora sp. ATCC 74235, both $[1,2^{-13}C_2]$ acetate (80) and $DL-[1¹³C]$ serine were tested, and good enrichment of ^{13}C was observed when either one of them was added to the culture of the producing fungus.⁸³ Consistent with the labelling observed in broussonetine biosynthesis, supplementation with $[1,2^{-13}C_2]$ acetate resulted in the incorporation of ¹³C into the 2–3, 7–8, 9–10, 11–12, 13–14, 15–16, 17–18 and 19–20 positions, whereas incubation with $PL-[1^{-13}C]$ serine showed ^{13}C incorporation into the C-4 position. These results indicated that the alkenyl side chain is biosynthesized by seven acetate units and the ribomimetic moiety is formed from acetate and serine (Fig. 16).

3.2.2. Biosynthesis of DAB-1. While the pseudosugar moieties in broussonetines and pramanicin are derived from acetate and serine, DAB-1 (14) was predicted to be derived from

xylulose 5-phosphate (81). Genes that are involved in the biosynthesis of DAB-1 were first identified in the bacterium Chitinophaga pinensis DSM 2588 through a genome mining study using the gabT1, yktC1 and gutB1 genes from the sixmembered iminosugar deoxynojirimycin biosynthetic pathway as queries.⁸⁴ The DAB-1 cluster contains genes homologous to gabT1 (encodes an aminotransferase) and gutB1 (encodes an oxidoreductase), but lacks the $yktC1$ (encodes a phosphatase) homologue. Instead, it contains a HAD family hydrolase, which may function as a phosphatase. Heterologous expression of the three C. pinensis genes in E. coli resulted in the production of DAB-1 (14).

While no biochemical studies of the biosynthetic enzymes have been conducted, the biosynthetic pathway to DAB-1 (14) is believed to resemble that of deoxynojirimycin.^{85,86} Xylulose 5-phosphate (81), which is an intermediate of the pentose phosphate pathway, may be converted to 4-amino-4-deoxyarabinitol (83) by the GabT1 orthologue and the phosphatase (Fig. 17). Subsequently, the compound is converted to 4-deoxy-4-imino-D-

Fig. 16 Biosynthetic origin of pramanicin.

arabinitol (85) by the GutB1 orthologue and reduced to DAB-1 by an unknown enzyme/mechanism. It is important to note that while bacteria produce DAB-1 from xylulose 5-phosphate, plants may use a different mechanism to produce it. Thus, how this compound is made in plants remains an open question.

3.2.3. Biosynthesis of anisomycin. The biosynthesis of the antifungal agent anisomycin (70) was first investigated by labelling experiments with isotopically labelled amino acids in the producing organism S. griseolus. As expected from its chemical structure, DL-tyrosine was found to be efficiently incorporated into anisomycin. 87 The biosynthetic gene cluster of anisomycin was later identified in S. hygrospinosus var. beijingensis, another producer of anisomycin, using a bioactivity-guided genomic library screening approach.⁸⁸ This somewhat unusual approach used an anti-yeast assay to screen the genomic library of S. hygrospinosus heterologously expressed in Streptomyces lividans TK24. Four S. lividans exconjugants were found to have significant anti-yeast activity and all of them indeed produced anisomycin. DNA sequencing and comparative analysis of the clones led to the identification of the anisomycin BGC (the *ani* cluster).

On the basis of gene inactivation, labelling experiments, and/or biochemical studies, the biosynthetic pathway to anisomycin (70) is proposed to initiate by deamination of L -tyrosine to 4-hydroxyphenylpyruvate (HPP, 86) catalysed by the aminotransferase AniQ (Fig. 18).⁸⁸ This is followed by a coupling reaction of the product with glyceraldehyde 3-phosphate catalysed by the transketolase AniP. The product (DHPP, 87) is then glucosylated by the glycosyltransferase AniO and converted to the glycosylated hydroxyphenyl-aminopentitol (Glc-HPAP, 89) by AniQ. As aniQ is the only aminotransferase gene in the anisomycin BGC, the gene product is hypothesized to work twice in the pathway, the deamination of L-tyrosine and the amination of the glycosylated ketone Glc-DHPP (88). In fact, in the presence of PLP and a-ketoglutarate the recombinant AniQ can convert 89 back to 88.

Oxidation of Glc-HPAP by the short-chain dehydrogenase AniN would give an aldehyde intermediate (90), which then

Fig. 18 Biosynthesis of anisomycin.

undergoes Schiff base formation and reduction to give the Nribomimetic-containing product Glc-HPMP (91). This step is also confirmed by gene inactivation, as the ΔaniN mutant only produces the glycosylated amino intermediate. Glc-HPMP (91) is hypothesized to undergo deglycosylation by the putative glucosidase AniG before being modified by the methyltransferase AniK and the Oacetyltransferase AniI to give anisomycin (70). The attachment of glucose to the AniP product 87 and the removal of the sugar moiety later in the pathway by AniG is somewhat unusual and puzzling, as in principle the biosynthesis may proceed without these glucosylation–deglucosylation steps. However, the presence of both genes in the anisomycin BGC and the identification of the glucosylated intermediates in the AniG mutant strongly suggest that the glucosylation–deglucosylation steps are involved. The timing of the glycosylation step in the pathway and the selectivity of the glycosyltransferase enzyme were confirmed by in vitro enzymatic assay where the recombinant AniO could only glycosylate the AniP product DHPP (87) and not any other intermediates. In addition, the transamination reaction by AniQ took place more efficiently with a glucosylated substrate. While there is no biochemical study on the glucosidase AniG, deglycosylation must precede the Omethylation on the phenol moiety. Interestingly, the methyltransferase AniK can methylate both HPMP (92) and its acetylated derivative. Therefore, while disruption of the putative Oacetyltransferase gene *aniI* resulted in a strain that accumulated only deacetylanisomycin, it is still possible that acetylation takes place before O-methylation in the pathway.

3.3. Development of N-ribomimetics

Similar to the S-ribomimetics, the development of Nribomimetics has been done mostly by chemical synthesis.

One significant example of this effort is the development of purine nucleoside analogue (PNA)-based immunosuppressive agents that can inhibit purine nucleoside phosphorylase (PNP), leading to the inhibition of human lymphocyte proliferation.⁸⁹ Among the PNAs developed, forodesine (BCX1777 or immucillin H) (94) (Fig. 19) was found to have human PNP inhibitory activity. Subsequent preclinical and clinical evaluation of 94 showed its efficacy in treating patients with B-cell acute lymphoblastic leukemia.⁹⁰

Another useful nucleoside analogue structurally similar to forodesine is galidesivir (BCX4430, 95) (Fig. 19). This compound is being developed as an antiviral drug. Galidesivir (95) is active in vitro against several pathogenic RNA viruses, such as MERS-CoV, SARS-CoV, and other lethal viruses (Ebola virus, Marburg virus, and Yellow Fever virus).^{91,92} Similar to the mechanism of action of other nucleoside-based antiviral drugs, 95 is phosphorylated by cellular kinases to form a triphosphate version of the compound that resembles ATP. This ATP mimic is then used by the viral RNA polymerase as one of the building blocks, incorporating galidesivir monophosphate into the growing RNA chain, leading to premature chain termination. The fact that forodesine (94) and galidesivir (95) are structurally very similar to each other but their biological activities are very different is remarkable. Therefore, it is expected that many more analogues of N-ribomimetics will be developed, and more exciting biological activities will emerge from this class of compounds.

4. C-Ribomimetics and related compounds

4.1. Discovery and bioactivities of C-ribomimetics

The C-ribomimetics (carbafuranoses) are among naturally occurring sugar mimetics that have been studied more extensively, both for their biological activities and biosynthesis. Most, if not all, of the known C-ribomimetics are of bacterial origin (Table 3). Among them is bacteriohopanetetrol (BHT) cyclitol ether (96) (Fig. 20), which was first identified as a component of glycolipids in the Gram-negative bacterium Zymomonas mobilis.

Other C-ribomimetics, aristeromycin (15) (Fig. 3) and neplanocin A (97) (Fig. 20), were isolated from the cultures of Streptomyces citricolor and Ampullariella regularis A11079, respectively.^{94,95} These compounds exhibited significant antitumour and antibacterial activities, as well as inhibition of vaccinia virus multiplication in monolayer cultures of mouse

Table 3 Selected natural C-Ribomimetics

BHT cyclitol ether (96) Aristeromycin (15) Neplanocin A (97)	Zymomonas mobilis S. citricolor		
		Component of glycolipids in Zymomonas mobilis Antitumour and antibacterial activities (S-adenosyl-L-	145 94-99, 151-159, 161 and 162
	Ampullariella regularis A11079	homocysteine hydrolase inhibitor) Antitumour and antibacterial activities (S-adenosyl-L- homocysteine hydrolase inhibitor)	94-98, 151-158 and 160-162
Adecypenol (98)	Streptomyces sp. OM-3223	Calf intestinal adenosine deaminase inhibitor	100
Pactamycin (16)	S. pactum	Antibiotic, antiplasmodial, and antitumour activity	105-109, 114, 146-150 and 163-168
7-Deoxypactamycin (99)	S. pactum	More active than pactamycin (16)	$110 - 113$
Pactamycate (100)	S. pactum	Less active than pactamycin (16)	110–112, 114 and 115
Pactalactam (101)	S. pactum	Less active than pactamycin (16)	110–112, 114 and 115
Jogyamycin (102)	S. pactum	Antitrypanosomal activity	112
Allosamidin (103)	<i>Streptomyces sp. no.</i> 1713	Chitinase inhibitor, stress tolerance enhancement, inhibition of moulting, inhibition of yeast cell separation, inhibition of malarial ookinete transmis- sion, inhibition of encystment of Entamoeba invadens, promotion of chitinase production in Streptomyces, suppression of airway inflammation in mammals	116-129
Trehazolin (105)	Micromonospora coriacea (SANK 62192)	Trehalase inhibitor	131, 132 and 140
Caryose (107) Calditol (108)	Pseudomonas caryophylli Sulfolobus acidocaldarius	Lipid membrane constituents of P. caryophylli Thermoacidophilic archaebacterial lipid membrane constituent	133 and 134 135, 136 and 141-144
Queuosine (115)	Certain bacteria and eukaryotes	Enhances translation fidelity	137-139
Epoxyqueuosine (114)	Certain bacteria and eukaryotes	Enhances translation fidelity	137-139

Crystal structures of SAHH in complex with 3'-ketoaristeromycin (3KA) indicates that 3KA binds to SAHH at the same position as adenosine, which is one of the substrates for SAHH in the formation of S-adenosyl-L-homocysteine (SAH) (Fig. 21). 99 Docking studies of SAHH with aristeromycin (15) or 3KA showed that the distance between the cofactors (NAD⁺ or NADH) and the 3'-position of the corresponding substrate (aristeromycin or 3KA, respectively) changed from 3.2 Å in NAD⁺ -aristeromycin to 3.6 Å in NADH-3KA, suggesting that the oxidation of aristeromycin by NAD⁺ occurs more readily than the reduction of 3KA back to aristeromycin by NADH. Therefore, 3KA appears to be more thermodynamically stable than aristeromycin in SAHH, making it a better inhibitor for the enzyme. Thus, it is postulated that after competing with the natural substrate adenosine for SAHH, aristeromycin is subsequently oxidized to 3KA.

Adecypenol (98), a homologue of neplanocin A (97) in which the adenine moiety has been modified, was isolated from the culture broth of Streptomyces sp. OM-3223.¹⁰⁰ It contains a unique homopurine chromophore, 8(R)-3,6,7,8-tetrahydroimidazo[4,5-d]-[1,3]diazepin-8-ol, similar to that found in adechlorin, coformycin, and deoxycoformycin (pentostatin).¹⁰¹⁻¹⁰³ However, in contrast to the latter compounds, which have potent and tight-binding inhibitory activity against adenosine deaminase, adecypenol is a semi tight-binding type inhibitor

The antitumour antibiotic pactamycin (16) (Fig. 3) was identified in the culture medium of Streptomyces pactum.¹⁰⁵ Pactamycin inhibits protein synthesis by interacting with the 16S rRNA of the 30S ribosomal subunit of bacteria or the equivalent small ribosomal subunit in other organisms. Analysis of the crystal structures of the Thermus thermophilus 30S ribosomal subunit in complex with pactamycin and 16S rRNA and in vitro assays suggest that pactamycin interacts with the ribosomal E-site and prevents tRNA-mRNA complex relocation into the E site (Fig. 22).^{106–108} Despite its potent biological activity, the development of pactamycin as a clinically useful drug was hampered by its broad-spectrum cytotoxicity.¹⁰⁹

In addition to pactamycin, a number of pactamycin analogues, e.g., 7-deoxypactamycin (99), pactamycate (100), pactalactam (101), and jogyamycin (102), have been isolated from various strains of *S. pactum* (Fig. 20).^{110–112} 7-Deoxypactamycin (99), which lacks a hydroxy group at C-7, showed superior activities compared to pactamycin in a number of biological assays.¹¹³ On the other hand, pactamycate (100) and pactalactam (101) are significantly less active than pactamycin against bacteria, Plasmodium falci*parum*, and several cancer cell lines,^{114,115} indicating the importance of the free C-2 amino group and the dimethylurea moiety for their activities. On the other hand, jogyamycin has been shown to have better antitrypanosomal activity compared to pactamycin.¹¹²

Other interesting C-ribomimetic-containing natural products include the chitinase inhibitor allosamidin (103) and the trehalase inhibitor trehazolin (105) (Fig. 23). Allosamidin (103), from Streptomyces sp. no. 1713 , 116 is a pseudotrisaccharide consisting

Fig. 20 Chemical structures of C-ribomimetics.

Fig. 21 Structures of the mammalian SAHH complexed with NAD^+ and adenosine (Ado) (left; PDB code, 5AXA) and with NADH and 3KA (right; PDB code, 5AXC).

of a ribomimetic moiety (allosamizoline, 104), and two units of N-acetyl-D-allosamine. The carbafuranose structure of allosamizoline is linked with a dimethylaminooxazoline ring, resembling the oxazolium ion intermediate of chitin hydrolysis mediated by chitinases from family 18. Allosamidin has various biological activities including stress tolerance enhancement in plants,¹¹⁷ inhibition of moulting in insects, $116,118-120$ inhibition of yeast cell separation,¹²¹ inhibition of malarial ookinete transmission,^{122–124} inhibition of encystment of *Entamoeba invadens*,¹²⁵ promotion of chitinase production in Streptomyces,¹²⁶⁻¹²⁸ and suppression of airway inflammation in mammals.¹²⁹

Analogues of allosamidin include methylallosamidin, demethylallosamidin, glucoallosamidin A, glucoallosamidin B, methyl-N-

Fig. 22 Structure of the T. thermophilus ribosome with bound pactamycin (PDB code, 1HNX). The red spheres show pactamycin.

demethylallosamidin, and didemethylallosamidin.¹³⁰ All of them can also bind to family 18 chitinases and inhibit their catalytic activity. Trehazolin (105), from Micromonospora coriacea (SANK 62192), 131 is a pseudodisaccharide with an N-glycosidic linkage between the glucose residue and the C-ribomimetic unit trehalamine (106) (Fig. 23).¹³² Similar to allosamizoline, trehalamine consists of a C-ribomimetic structure fused with an aminooxazoline ring.

Although most natural ribomimetics are secondary metabolites, there are examples of C-ribomimetics that are essential for the survival of the producing microorganisms. Caryose $(4,8$ -cyclo-3,9-dideoxy-L-erythro-D-ido-nonose) (107) , a C₉ribomimetic (Fig. 24), was found alongside caryophyllose (3,6,10 trideoxy-4- C - $[(R)$ -1-hydroxyethyl]-p-erythro-p-gulo-decose) in the lipopolysaccharide (LPS) fractions of the phytopathogenic bacterium Pseudomonas caryophylli (Burkholderia caryophylli), which is responsible for the wilting of carnations. This interesting C-ribomimetic can be found as a standalone monosaccharide as well as a homopolymer, known as caryan, with a $(1 \rightarrow 7)$ glycosidic linkage.^{133,134}

Another important membrane-related C-ribomimetic is calditol (108) (Fig. 24), which was found in the membrane lipids of a subset of thermoacidophilic archaea, e.g., Sulfolobus acidocaldarius.¹³⁵ This C-ribomimetic unit is ether-linked directly to the glycerol backbone of glycerol dibiphytanyl glycerol tetraethers (GDGTs) or related compounds, such as in calditolglycerocaldarchaeol (109), to form the components of the membrane lipids. These lipids are thought to be essential for archaebacteria to preserve their membrane functions under extreme conditions, such as high temperatures and strong

Fig. 23 Structures of allosamidin, trehazolin and their ribomimetic moieties.

Fig. 24 Structures of caryose and calditol found in lipopolysaccharides or archaeal lipids.

acidity.¹³⁶ Deletion of calditol synthase (Cds) in *S. acidocaldarius* resulted in mutants that are sensitive to extremely low pH, implying that the C-ribomimetic moiety is important for protecting archaeal cells from environmental stress.¹³⁶

A C-ribomimetic unit was also found in modified nucleosides, queuosine and epoxyqueuosine, from certain bacterial and eukaryotic tRNAs. Queuosine and epoxyqueuosine are conserved products of the post-translational processing of tRNA in bacteria and eukaryotes, specifically at position 34 in the anticodons of tRNAs coding for the amino acids aspartic acid, asparagine, histidine, and tyrosine.137 While bacteria are capable of producing these compounds de novo, eukaryotes may obtain queousine from their diet or intestinal flora.¹³⁷ In higher vertebrates, including humans, some queousine is glycosylated to form galactosyl- and mannosyl-queousine.¹³⁸ The biological significance of these hypermodified RNA nucleosides is still elusive.

4.2. Biosynthesis of C-ribomimetics

Studies on the biosynthesis of several C-ribomimetics revealed that they are derived from different precursors employing distinct catalytic mechanisms. The C-ribomimetic units in queuosine (115) and epoxyqueuosine (114) are derived from S-adenosylmethionine catalysed by a unique enzyme, S-adenosylmethionine: tRNA ribosyltransferase-isomerase (QueA) (Fig. 25), 137,139 whereas the C-ribomimetic units in BHT cyclitol ether (96), pactamycin (16), and allosamidin (103) are derived from N-acetylglucosamine. Interestingly, the C-ribomimetic unit of trehazolin (105), which is highly similar to that of allosamidin, is derived from glucose, instead of N-acetylglucosamine.140 In contrast to allosamidin, the two nitrogen atoms of trehazolin originate from L-arginine.¹⁴⁰

4.2.1. Biosynthesis of calditol. The archaeal C-ribomimetic calditol (108) is also derived from glucose, presumably via a pathway similar to that of trehazolin.¹⁴¹⁻¹⁴⁴ Unfortunately, only limited knowledge is currently available with regard to the mechanisms underlying the formation of allosamidin (103), trehazolin (105), and calditol (108), as their biosynthetic studies were mostly done by incorporation studies using isotopically labelled precursors. However, the identification of calditol synthase (Cds), which is a radical SAM enzyme, in S. acidocaldarius strongly suggests that the conversion of glucose to calditol takes place via a radical reaction.¹³⁶ This conversion is most likely to occur after the sugar is attached to the lipids (Fig. 26), as the Cds mutant, while lacking the ability to produce calditol-GDGTs (120), can still produce glycosylated GDGTs.¹³⁶

Fig. 25 Formation of the C-ribomimetics of epoxyqueuosine and queuosine in modified tRNA (oQ-tRNA and Q-tRNA, respectively).

On the other hand, the biosynthetic origin and the mode of formation of the pseudomonal LPS C_9 -ribomimetic caryose (107) are currently unknown.

4.2.2. Biosynthesis of BHT cyclitol ether and pactamycin. Similar to calditol, the formation of the C-ribomimetic moieties in BHT cyclitol ether (96) and pactamycin (16) is also postulated to be catalysed by radical SAM enzymes. Inactivation of the radical SAM gene hpnJ in Burkholderia cenocepacia, which produces BHT cyclitol ether, resulted in a mutant that only produces BHT-GlcN (121) (Fig. 27A).¹⁴⁵ However, there are no biochemical data to confirm the function of the enzyme. In pactamycin biosynthesis, the radical SAM enzyme PtmC (PctE) was proposed to catalyse a similar reaction.^{114,146} However, inactivation of the ptmC gene in S. pactum ATCC 27456 only resulted in a mutant that lacks the ability to produce any pactamycin intermediates.¹¹⁴ Therefore, the actual substrate for PtmC is still unknown. However, a recent study showed that in pactamycin biosynthesis, N-acetylglucosamine (the precursor of the *C*-ribomimetic unit of pactamycin) is coupled with a β ketoacyl intermediate (123) while it is still attached to an acyl carrier protein (PtmI) (Fig. 27B). 147 This highly unusual phenomenon, catalysed by the glycosyltransferase PtmJ, indicates that the formation of the C-ribomimetic unit takes place after the sugar precursor is attached to the 3-aminobenzoate/polyketide-derived intermediate (e.g., 3-aminoacetophenone).¹⁴⁷ Interestingly, free N-

acetylglucosaminyl-3-aminoacetophenone is not involved in pactamycin biosynthesis, suggesting that further modification(s) of the sugar moiety is necessary before the compound is released from the carrier protein. Unfortunately, except for the preliminary report on the in vitro activities of the oxidoreductase PctP and the aminotransferase PctC, which were characterized using GlcNAc-3AAP as a substrate, 148 no information is available on the conversion of the glycosylated b-ketoacyl-PtmI to the first known C-ribomimetic-containing intermediate TM-101 (125) .¹⁴⁹ Similar to *ptmC*, inactivation of the putative genes involved in this part of the pathway, e.g., ptmA, ptmB, ptmG, ptmL, ptmM, or ptmO, only gave mutants that lack the ability to produce pactamycin or its intermediates.114

TM-101 (125) and the subsequent downstream intermediates in the pactamycin pathway were identified through a series of gene inactivation experiments.149 Based on these experiments the radical SAM-dependent C-methyltransferase PtmH is believed to catalyse the conversion of TM-101 to TM-102 (126), which is then converted to jogyamycin (102) by the O-methyltransferase PtmD (Fig. 28).149 The KASIII-like protein PtmR has been shown to catalyse the transfer of 6-methylsalicylyl moiety from the iterative type I PKS PtmQ to the primary alcohol of jogyamycin.150 The product, 7-deoxypactamycin (99), is then hydroxylated to the final product pactamycin, most likely by the cytochrome P450 protein PtmY. 109

4.2.3. Biosynthesis of neplanocin A and aristeromycin. In contrast to the C-ribomimetics described above, the carbafuranose moieties of aristeromycin (15) and neplanocin A (97) originate from fructose 6-phosphate (128). The conversion of fructose 6-phosphate to a five-membered cyclitol is catalysed by a myo-inositol phosphate synthase (MIPS) orthologue, Ari2 (Fig. 29).151,152 Further studies on the reaction mechanism of Ari2 using $(6S)$ -D- $[6^{-2}H_1]$ - and $(6R)$ -D- $[6^{-2}H_1]$ -glucose 6phosphate, which were then converted to their corresponding

Fig. 27 Proposed formation of the C-ribomimetics of BHT cyclitol ether and pactamycin. ACP, acyl carrier protein.

fructose 6-phosphates by a glucose 6-phosphate isomerase,¹⁵² revealed that the reaction is initiated by the elimination of the pro-6R proton, which is similar to the reaction mechanism of MIPS (Fig. 30). While there are no other enzymes from the aristeromycin pathway biochemically characterized, experiments using blocked mutants of the producing organism S. citricolor and complementing them with isotopically labelled compounds suggested that the ketone and the alcohol are intermediates in the neplanocin A and aristeromycin pathway.¹⁵³ In addition, experiments with a partially purified cell-free extract of the producing organism suggest that aristeromycin (15) is derived from neplanocin A (97) .¹⁵⁴

4.3. Development of C-ribomimetics

As described earlier, ribomimetics are privileged chemical scaffolds, as structurally they resemble sugars that play critical roles in living organisms. A number of synthetic C-ribomimetics have been used to treat patients with influenza, hepatitis B, and HIV infections. In addition, analogues of natural C-ribomimetics, e.g., aristeromycin (15), neplanocin A (97), and pactamycin (16), possess great potential to be developed as clinically useful drugs. Aristeromycin and neplanocin A showed good antiviral activity, $97,155$ but their significant cytotoxicity against mammalian

Fig. 28 Proposed tailoring steps in pactamycin biosynthesis. KS, ketosynthase; AT, acyltransferase; DH, dehydratase; KR, ketoreductase; ACP, acyl carrier protein.

cells has hampered their clinical development as novel antiviral or anticancer drugs.156,157 A number of structure–activity relationship studies of aristeromycin and neplanocin A have been done, and several derivatives were found to have enhanced biological activities.158 Of particular interest are the C-6 fluorinated aristeromycin and neplanocin A, as they have increased antiviral or antitumour activities, respectively. 6,6-Difluoroaristeromycin showed greater activity against several viruses including MERS-CoV, SARS-CoV, ZIKA, and CHIKV, with 3–100 times more activity compared to aristeromycin.¹⁵⁹ Similarly, 6'-fluoroneplanocin A showed more potent antitumour activity than neplanocin A against several human tumour cell lines.¹⁶⁰ While the mechanism of action of neplanocin A for its antitumour activity is still unclear, it is suggested that the toxicity results from the induction of signalling pathways leading to apoptosis.^{161,162}

Analogues of pactamycin have also been produced by precursor-directed biosynthesis, mutasynthesis, genetic engineering, and by chemoenzymatic and chemical synthesis. Earlier incorporation studies by adding alternative precursors to the cultures of S. pactum resulted in fluorinated pactamycins, albeit in low yields. A mutasynthetic study with a $ptmT$ knockout mutant of S. pactum ATCC 27456 also resulted in a number of fluorinated pactamycins.¹⁶³ Analogues of pactamycin have also been produced by generating mutant strains of S. pactum ATCC 27456, in which one or more of the putative tailoring genes have

Fig. 29 Biosynthesis of neplanocin A and aristeromycin.

been inactivated.¹¹⁴ Some of the products, $e.g.,$ TM-025 and TM-026, showed better selectivity against the malarial parasite Plasmodium falciparum with significantly less cytotoxicity and antibacterial activity. 115 These new pactamycin analogues also induce p53 dependent cell-cycle arrest at S-phase in human head and neck cell carcinoma cells.¹⁶⁴ The highly promiscuous KASIIIlike protein PtmR has also been used to produce analogues of pactamycin with different C-6 ester moieties.¹⁵⁰ This enzyme recognizes various N-acetylcysteamine (NAC)-thioesters as acyldonors and TM-025 as an acyl acceptor.

The Structure–Activity Relationship (SAR) studies of pactamycin have also been done by chemical synthesis during the past decade, revealing the role of functional groups within the compound in its biological activities (Fig. 31).¹⁶⁵⁻¹⁶⁷ Modifications of the urea moiety of pactamycin generally led to a loss or reduction of antibacterial, cytotoxic, and antimalarial activities.166 Replacement of the 3-aminoacetophenone moiety

with several aniline derivatives also affects their biological activities, more prominently on cytotoxicity and antimalarial activity.166,167

The C-5 position appears to play a significant role in the cytotoxicity of the compound. Replacing the C-5 methyl with hydrogen resulted in an analogue that is more cytotoxic than the parent compound. This was proposed to be due to the change in lipophilicity.¹⁶⁷ On the other hand, replacement of the C-5 methyl with longer-chain alkanes, *i.e.*, ethyl or hexyl groups, resulted in a loss of cytotoxicity against several human tumour cell lines.¹⁶⁷

The C-6 ester moiety does not appear to be essential for the biological activity of pactamycin. Diversification of this side chain did not result in any significant differences in biological activity compared to pactamycin, implying the limited role of the C-6 ester moiety in the compound binding interactions with the 30S ribosomal subunit.^{166,168} A number of C-6 and C-7 bisacylated derivatives have also been synthesized.¹⁶⁷ The results showed a linear decrease in activity as the steric hindrance of the C-7 ester group enlarges, hinting that the hydroxy group at C-7 plays a more important role in the bioactivity compared to that at $C-6$.¹⁶⁷

5. Conclusions

Due to their high resemblance to ribose, deoxyribose, or arabinose, which play significant roles in the structural, physiological, and/or reproductive functions of living organisms, many ribomimetics and related compounds have potent biological activities. Consequently, they have a tremendous potential to be developed as drug leads. In fact, some of them have been used to treat viral infections. Natural ribomimetics are produced by many different organisms including plants, fungi, and bacteria. They include ribose-like compounds in which the ring oxygen atom of ribose or related five-member ring sugars has been replaced by a sulphur (S-ribomimetics), a nitrogen (N-ribomimetics), or a carbon (C-ribomimetics) atom. With the exception of some N-ribomimetics, most of them are derived from a simple sugar such as glucose, fructose, or glucosamine. However, little is known about why they are produced or what benefits they provide to the producing organisms. Therefore, future studies should not only focus on the potential use of natural ribomimetics as pharmaceutically useful compounds but also include investigations on their physiological and/or ecological functions in nature.

Author contributions

T. Tsunoda: writing – original draft. S. Tanoeyadi: writing – original draft. P. J. Proteau: writing – review and editing. T. Mahmud: writing – review and editing.

Conflicts of interest

There are no conflicts to declare.

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