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# Total syntheses of *ent*-hypocoprin A and *ent*-hypocoprin B<sup>+</sup>

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This study reports the stereoselective total syntheses of the antipodes of the unique 3/10 bicyclic skeletal sesquiterpenoids, namely, hypocoprin A and hypocoprin B. The synthesis involved conjugate addition accelerated by trimethylsilyl chloride, construction of the ten-membered ring *via* the intramolecular  $S_N2$  reaction promoted by 1,8-diazabicyclo[5.4.0]undec-7-ene, and osmium-mediated  $\pi$ -facial selective dihydroxylation to functionalize the 1,1-disubstituted alkene.

## Introduction

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Hypocoprin A and hypocoprin B, isolated from the coprophilous fungus Hypocopra rostrata on horse dung, are sesquiterpenoids with an unprecedented 3/10 bicyclic ring system (Fig. 1).<sup>1</sup> The 3/10 bicyclic scaffold of these sesquiterpenoids is thought to arise biosynthetically by 8,11-cyclization of the transhumulyl cation derived from farnesyl diphosphate. The only other examples with the same scaffold are the marine diterpenoid palmatol<sup>2</sup> (from the Mediterranean octocoral Alcyonium palmatum) and pacificins<sup>3,4</sup> (from the Formosan soft coral Nephthea sp.). Consequently, the biosynthetic origins of the bicyclic system in hypocoprins are largely unknown. Hypocoprin A moderately inhibits the growth of the Gram-positive bacteria Staphylococcus aureus and Bacillus subtilis but is ineffective against Candida albicans or the Gram-negative bacterium Escherichia coli.1 To explore the unreported biological properties of hypocoprins A and B with novel bicyclic scaffolds, we describe novel total syntheses of ent-hypocoprin A (1) and enthypocoprin B (2) starting from the optically active acetonide 3.



Fig. 1 Structures of natural hypocoprin A and hypocoprin B.

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### **Results and discussion**

Our retrosynthetic analysis of hypocoprin A is depicted in Scheme 1. In the initial planning stage, we recognized that the C-7 tertiary alcohol moiety can be constructed in hypocoprin A *via* diastereoselective addition of the C<sub>1</sub>-unit to bicyclic ketone **A**. We thus planned the synthesis of bicyclic ketone **A** with a core ten-membered ring from  $\beta$ -ketosulfone **B** *via* an intramolecular S<sub>N</sub>2 reaction followed by reductive desulfonylation. We assumed that  $\beta$ -ketosulfone **B** would arise from the chemoselective addition of a vinyllithium species to ketoaldehyde **C** and the introduction of a halogen (X) atom. The  $\beta$ , $\beta$ -



Scheme 1 Retrosynthetic analysis of *ent*-hypocoprin A (1) and *ent*-hypocoprin B (2).

<sup>†</sup> Electronic supplementary information (ESI) available. CCDC 2150062. For ESI and crystallographic data in CIF or other electronic format see https://doi.org/10.1039/d2ra02891c

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disubstituted aldehyde moiety in  $\beta$ -ketosulfone **C** can be constructed by the Horner–Wadsworth–Emmons homologation of cyclopropyl ketone **D** followed by conjugate addition with an organocopper reagent. Cyclopropyl ketone **D** can be obtained from acetonide **E**, which possesses C-8 and C-10 stereochemistry established by substrate-controlled stereoselective cyclopropanation. We established a synthetic route for the antipodes of hypocoprins A and B from acetonide 3, which corresponds to the enantiomer of acetonide **E** synthesized from *D*-mannitol, an inexpensive and ideal chiral pool for total synthesis. The total syntheses of *ent*-hypocoprin A (1) and *ent*-hypocoprin B (2) were initiated from acetonide 3 *via* a retrosynthetic analysis.

Our synthesis commenced from acetonide 3 (>97% ee),<sup>5</sup> a known enantiopure compound prepared from p-mannitol (Scheme 2).6 After sequential acidic deprotection of the isopropylidene group of 3 and oxidative cleavage with periodic acid,<sup>7</sup> the aldehyde was methylated with MeMgBr and the resulting secondary alcohol was oxidized with 2-iodoxybenzoic acid (IBX) to afford cyclopropyl ketone 4.8 C2-Homologation of cyclopropyl ketone 4 by the Horner-Wadsworth-Emmons reaction afforded  $\alpha$ , $\beta$ -unsaturated esters 5 and 6 in 95% yield (E:Z = 9:1). We next examined various conditions for the conjugate addition of 5 and 6, but unexpectedly, the starting material was recovered with only trace amounts of the desired adducts, presumably because the conjugate addition of organocopper reagents was inhibited by steric hindrance of the adjacent cyclopropyl group. To improve the reactivity, we extensively screened the solvents, temperatures, and addition of Lewis acid. The conjugate addition proceeded efficiently under a combined lithium dimethylcuprate-chlorotrimethylsilane reagent<sup>9</sup> in CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O followed by desilylation, delivering primary alcohol 7 in high yields (85% overall yield over two steps). IBX oxidation converted the primary alcohol 7 into aldehyde 8 (88% yield). Aldehyde 8 was subsequently treated with the  $\alpha$ -sulforvl carbanion derived from methyl phenyl sulfone and butyllithium (BuLi), providing secondary alcohols as a diastereomeric mixture. The resulting alcohols were

reduced with diisobutylaluminium hydride (DIBAH) followed by oxidation with IBX to give  $\beta$ -ketosulfone **9** (77% overall yield over three steps).

Having obtained  $\beta$ -ketosulfone 9, which corresponds to the retrosynthetic intermediate C shown in Scheme 1, we introduced a trisubstituted alkene moiety followed by cyclization of the ten-membered ring (Scheme 3). The trisubstituted alkene moiety was introduced to  $\beta$ -ketosulfone **9** by aldehyde-selective addition of the vinyllithium species derived from (E)-tertbutyl((3-iodobut-2-en-1-yl)oxy)diphenylsilane10 and <sup>t</sup>BuLi. forming the secondary alcohol 10 as a pair of inseparable diastereomers (79% yield, dr = 1:1). After acetylating the secondary alcohol 10, primary alcohol was selectively unmasked to allylic alcohol 11 (89% overall yield over two steps), which was converted to allylic bromide with CBr<sub>4</sub> and Ph<sub>3</sub>P. The resulting allylic bromide was an important precursor for cyclizing the tenmembered ring, but was unstable. Therefore, after confirming the disappearance of the starting material by thin-layer chromatography, we immediately added 1,8-diazabicyclo[5.4.0] undec-7-ene (DBU). To our delight, the intramolecular  $S_N 2$ alkylation proceeded smoothly, followed by SmI2-mediated reductive desulfonylation<sup>11</sup> to give a pair of ten-membered cyclic ketones (12 and 13) as an inseparable diastereomeric mixture (76% overall yield over two steps).

The ten-membered cyclic ketones **12** and **13** were deacetylated and separated as secondary alcohols **14** and **15**, respectively, and then re-acetylated to give ketones **12** and **13** as single stereoisomers (Scheme 4). The ten-membered cyclic ketones **12** and **13** were exposed to Wittig methylenation, providing the *exo*olefins **2** and **16** in 92% and 90% yield, respectively. Comparing the <sup>1</sup>H and <sup>13</sup>C NMR spectral data of the *exo*-olefins **2** and **16** with the reported data, the NMR spectrum of **2** was found to be consistent with that of natural hypocoprin B.<sup>1</sup> However, whereas the specific rotation should be opposite to that of natural hypocoprin B ( $[\alpha]_D^{20}$  +13 (*c* 0.58 in MeOH)),<sup>1</sup> it was instead consistent with that of synthesized **2** ( $[\alpha]_D^{20}$  +55.3 (*c* 1.29 in MeOH)). In the previous isolation paper,<sup>1</sup> the absolute configuration of natural hypocoprin B was determined by conversion





Scheme 2 Synthesis of aldehyde 9. Reagents and conditions (a)  $(EtO)_2POCH_2CO_2Et$ , NaH, toluene, reflux, 95% (E:Z=9:1); (b) MeLi, Cul, TMSCl, CH<sub>2</sub>Cl<sub>2</sub>, Et<sub>2</sub>O, 0 °C; (c) TBAF, THF, r.t., 85% (2 steps); (d) IBX, DMSO/THF, r.t., 88%; (e) MeSO<sub>2</sub>Ph, BuLi, THF, -78 °C; (f) DIBAH, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C; (g) IBX, DMSO/THF, r.t., 77% (3 steps).

Scheme 3 Closure of the ten-membered ring *via* the intramolecular  $S_N2$  reaction. Reagents and conditions (a) (*E*)-*tert*-butyl((3-iodobut-2-en-1-yl)oxy)diphenylsilane, <sup>t</sup>BuLi, THF/Et<sub>2</sub>O, -78 °C, 79% (dr = 1 : 1); (b) Ac<sub>2</sub>O, pyridine, r.t.; (c) TBAF, THF, r.t., 89% (2 steps); (d) CBr<sub>4</sub>, Ph<sub>3</sub>P, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C then DBU, benzene, 0 °C to r.t.; (e) Sml<sub>2</sub>, THF, 0 °C, 76% (2 steps).



Scheme 4 Synthesis of *ent*-hypocoprin B (2) and 7-*epi-ent*-hypocoprin A (17). Reagents and conditions (a)  $K_2CO_3$ , MeOH, r.t., 14 48%, 15 46%; (b) Ac<sub>2</sub>O, pyridine, r.t., 96%; (c) Ph<sub>3</sub>PCH<sub>3</sub>I, BuLi, THF, r.t., 99%; (d) Ac<sub>2</sub>O, pyridine, r.t., 95%; (e) Ph<sub>3</sub>PCH<sub>3</sub>I, BuLi, THF, r.t., 90%; (f) MeLi, Et<sub>2</sub>O/THF, 0 °C, 95%.

from hypocoprin A, whose absolute configuration was determined by the Mosher method. Therefore, withholding our doubts on stereochemistry at this stage, we continued with the synthesis of *ent*-hypocoprin A (1). Toward the completion of the total synthesis of *ent*-hypocoprin A (1), we first attempted a direct nucleophilic addition of MeLi to the intermediate tenmembered cyclic ketone 12, obtaining the tertiary alcohol 17 as a single diastereomer in 98% yield. Unfortunately, the <sup>1</sup>H and <sup>13</sup>C NMR spectra of 17 were inconsistent with those of natural hypocoprin A.<sup>1</sup> Although the relative configuration of 17 could not be determined by nuclear Overhauser effect (NOE) spectroscopy, most of the substrate was converted to *ent*-hypocoprin B (2) by dehydration during three-weeks' storage of 17 in CDCl<sub>3</sub>. Consequently, compound 17 was inferred as the C-7 epimer of *ent*-hypocoprin A (1).

The exclusive diastereoselectivity of alkylation was attributable to steric congestion inside the ten-membered ring. We then estimated the conformation of *ent*-hypocoprin B (2) based on the observed NOE correlations (Fig. 2a). The results suggested that the C-3/C-13 single bond and the C-7/C-12 double bond are approximately perpendicular to the average plane of the ten-membered ring in the same direction, whereas the C-3/ C-4 double bond and the C-8/C-10 single bond preferentially face each other in parallel across the ten-membered ring. That is, the C-3/C-4 and C-7/C-12 double bonds are clearly distinguished inside and outside the ten-membered ring, suggesting the feasibility of a  $\pi$ -facial selective approach to alkene moieties.

Diene **18** obtained by acetylation of *ent*-hypocoprin B (**2**) was subjected to osmium-mediated dihydroxylation using AD-mix-



Fig. 2 Plausible mechanism of osmium-mediated  $\pi$ -facial selective dihydroxylation. (a) Selected NOE correlations for *ent*-hypocoprin B (2). (b) Top view of *ent*-hypocoprin B (2).

α or AD-mix-β (Scheme 5).<sup>12,13</sup> In dihydroxylation using AD-mixβ, diene **18** was converted to completely unrecoverable highpolarity compounds within six hours. However, when AD-mixα was attempted, the starting material **18** disappeared after around 24 hours, yielding the desired stereoisomer **19** with complete selectivity for C-7 as a mixture with MeSO<sub>2</sub>NH<sub>2</sub>. The stereochemistry of C-7 after derivatization to *ent*-hypocoprin A (**1**) was determined by X-ray crystallography. Interestingly, the C-3/C-4 trisubstituted alkene moiety was unaffected during this reaction. To understand this ideal π-facial selectivity, we employed a mnemonic device for predicting the stereoselectivity of dihydroxylation using AD-mix. The two alkene moieties of **18** were considered individually because they belong to different alkene-substitution classes. Considering the trisubstituted alkene (C-3/C-4) moiety, AD-mix-β and AD-mix-



Scheme 5 Completion of the total synthesis of *ent*-hypocoprin A (1). Reagents and conditions (a)  $Ac_2O$ , pyridine, r.t.; (b) AD-mix- $\alpha$ ,  $MeSO_2NH_2$ ,  ${}^{t}BuOH/H_2O$ , 0 °C; (c) *p*TsCl, pyridine, 30 °C; (d) LiEt<sub>3</sub>BH, THF, r.t., 94% (4 steps).

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 $\alpha$  are expected to preferentially proceed from outside and inside the ten-membered ring, respectively (Fig. 2b). As mentioned in Scheme 4, the inside is sterically congested, so dihydroxylation was unlikely to proceed in the presence of AD-mix-α. In contrast, 1,1-disubstituted alkene (C-7/C-12) moiety and ADmix-ß are undoubtedly a "matched" pair for dihydroxylation proceeding from outside the ring, as evidenced by the short reaction time. Nevertheless, AD-mix-a satisfactorily mediated the stereospecific dihydroxylation despite being mismatched for dihydroxylation proceeding from the outside. As is well known, the stereoselectivity of dihydroxylation is lower for 1,1disubstituted alkenes than for trisubstituted alkenes.14 We thus hypothesized that the present regio- and diastereoselective dihydroxylations resulted from a combination of strict recognition of trisubstituted alkenes and permissive recognition of 1,1-disubstituted alkenes. Subsequently, monotosylation of diol 19, followed by reductive hydrogenation of the resulting tosylate with lithium triethylborohydride,15 led to ent-hypocoprin A (1). Compound 1 presented the same <sup>1</sup>H NMR, <sup>13</sup>C NMR, and high-resolution mass spectra as natural hypocoprin A but its specific rotation was opposite in sign  $[\alpha]_{\rm D}^{20}$  –69.8 (c 0.65 in MeOH) to that of natural hypocoprin A (ref. 1  $\left[\alpha\right]_{D}^{20}$  +18 (c 2.31 in MeOH). Furthermore, the relative configuration of synthetic compound 1 was confirmed by single-crystal X-ray diffraction (Fig. 3).16

As expected, the specific rotation of the synthesized enthypocoprin A (1) was opposite in sign to that of natural hypocoprin A but the values of the two compounds were quite different. Many signals in the reported NMR spectrum of natural hypocoprin A were derived from impurities (mainly hypocoprin B), so the accuracy of the reported specific rotation was doubtful. In addition, as the sign of the synthesized ent-hypocoprin B (2) matched that of the reported specific rotation of hypocoprin B, one or both of the reported specific rotations might be incorrect. Therefore, we should not conclude that the synthesized ent-hypocoprin A (1) is a natural antipode only because the sign of its specific rotation is reversed. To confirm the correctness of the reported absolute configuration of hypocoprin A, we compared the NMR spectra of α-methoxy-α-trifluoromethylphenylacetic acid (MTPA) ester derivatives. In a previous study, the absolute configuration of natural hypocoprin A was determined by assigning the stereocenter of C-2 by the Mosher method.<sup>1,17,18</sup>



**Fig. 3** Oak Ridge Thermal Ellipsoid Plot (ORTEP) of *ent*-hypocoprin A (1).

Therefore, *ent*-hypocoprin A (1) was converted into (R)-MTPA and (S)-MTPA esters and their NMR spectra were obtained. The spectra were consistent with the (S)-MTPA and (R)-MTPA esters of natural hypocoprin A described in the isolation paper. Therefore, one may reasonably conclude that the absolute configuration of the natural product reported in the isolation paper is correct.

# Conclusions

In summary, we have accomplished a novel asymmetric total synthesis of *ent*-hypocoprin A (1) from optically active ketone 4 (19 steps, 13% overall yield). The key steps of our synthesis were TMSCI-accelerated conjugate addition, DBU-promoted cyclization to form the ten-membered ring, and osmium-mediated dihydroxylation under "mismatched" conditions to furnish the antipodes of the unique bicyclic sesquiterpenoids hypocoprin A and B. The biological activity of the two synthesized antipodes is currently under investigation. Moreover, we are applying our synthetic strategy to other unexplored 3/10 bicyclic sesquiterpenoids and diterpenoids.

# Conflicts of interest

There are no conflicts to declare.

# Notes and references

- 1 D. R. Jayanetti, Q. Yue, G. F. Bills and J. B. Gloer, *J. Nat. Prod.*, 2015, **78**, 396.
- 2 E. Zubia, A. Spinella, G. B. Guisto, A. Crispino and G. Cimino, *Tetrahedron Lett.*, 1994, **35**, 7069.
- 3 A. A. H. El-Gamal, S.-K. Wang, C.-F. Dai, I.-G. Chen and C.-Y. Duh, *J. Nat. Prod.*, 2005, **68**, 74.
- 4 A. A. H. El-Gamal, S.-K. Wang and C.-Y. Duh, *Chem. Pharm. Bull.*, 2007, 55, 890.
- 5 The specific rotation of acetonide 3 was  $[\alpha]_{D}^{27}$  -7.7 (*c* 1.07 in CHCl<sub>3</sub>) (ref. 6  $[\alpha]_{D}^{27}$  -7.9 (*c* 1.15 in CHCl<sub>3</sub>)).
- 6 T. Morikawa, H. Sasaki, R. Hanai, A. Shibuya and T. Taguchi, J. Org. Chem., 1994, **59**, 97.
- 7 J. B. Son, S. N. Kim, N. Y. Kim, M.-H. Hwang, W. Lee and D. H. Lee, *Bull. Korean Chem. Soc.*, 2010, 31, 653.
- 8 Y. Kazuta, H. Abe, T. Yamamoto, A. Matsuda and S. Shuto, *J. Org. Chem.*, 2003, **68**, 3511.
- 9 F. Romanov-Michailidis, M. Pupier, L. Guénée and A. Alexakis, *Chem. Commun.*, 2014, **50**, 13461.
- 10 K. Komine, Y. Nomura, J. Ishihara and S. Hatakeyama, Org. Lett., 2015, 17, 3918.
- 11 G. A. Molander and G. Hahn, J. Org. Chem., 1986, 51, 1135.
- 12 K. B. Sharpless, W. Amberg, Y. L. Bennani, G. A. Crispino, J. Hartung, K.-S. Jeong, H.-L. Kwong, K. Morikawa, Z.-M. Wang, D. Xu and X.-L. Zhang, *J. Org. Chem.*, 1992, 57, 2768.
- 13 H. C. Kolb, M. S. VanNieuwenhze and K. B. Sharpless, *Chem. Rev.*, 1994, 94, 2483.
- 14 D. Xu, G. A. Crispino and K. B. Sharpless, *J. Am. Chem. Soc.*, 1992, **114**, 7570.

- 15 H. Suemune, Y. Miyao and K. Sakai, Chem. Pharm. Bull., 1989, 37, 2523.
- data for ent-hypocoprin A (1).
- 17 I. Ohtani, T. Kusumi, Y. Kashman and H. Kakisawa, J. Am. Chem. Soc., 1991, 113, 4092.
- 16 CCDC 2150062 contains the supplementary crystallographic 18 J. A. Dale and H. S. Mosher, J. Am. Chem. Soc., 1973, 95, 512.