


 Cite this: *Chem. Commun.*, 2023, 59, 11736

 Received 20th July 2023,  
 Accepted 7th September 2023

DOI: 10.1039/d3cc03484d

[rsc.li/chemcomm](https://rsc.li/chemcomm)

## Enhanced electrochemiluminescence imaging of single cell membrane proteins based on Co<sub>3</sub>O<sub>4</sub> nanozyme catalysis†

 Jingjing Zhang,<sup>a</sup> Lin Hao,<sup>\*b</sup> Jie Chao,<sup>a</sup> Lianhui Wang<sup>id</sup><sup>c</sup> and Shao Su<sup>id</sup><sup>\*c</sup>

The development of enhanced strategies with excellent biocompatibility is critical for electrochemiluminescence (ECL) imaging of single cells. Here, we report an ECL imaging technique for a single cell membrane protein based on a Co<sub>3</sub>O<sub>4</sub> nanozyme catalytic enhancement strategy. Due to the remarkable catalytic performance of Co<sub>3</sub>O<sub>4</sub> nanozymes, H<sub>2</sub>O<sub>2</sub> can be efficiently decomposed into reactive oxygen radicals, and the reaction with L012 was enhanced, resulting in stronger ECL emission. The anti-carcinoembryonic antigen (CEA) was coupled with nanozyme particles to construct a probe that specifically recognized the overexpressed CEA on the MCF-7 cell membrane. According to the locally enhanced visualized luminescence, the rapid ECL imaging of a single cell membrane protein was eventually realized. Accordingly, Co<sub>3</sub>O<sub>4</sub> nanozymes with highly efficient activity will provide new insights into ECL imaging analysis of more biological small molecules and proteins.

Electrochemiluminescence (ECL) is a kind of optical radiation phenomenon produced by chemical molecules undergoing electrochemical reaction.<sup>1–3</sup> ECL does not require an additional excitation light source, since the reaction process depends on the applied voltage to drive the luminescence process, which effectively avoids the interference of light scattering and improves the detection sensitivity. It has become a powerful analytical technique, widely used in biochemical analysis,<sup>4</sup> pharmaceutical analysis,<sup>5</sup> environmental analysis<sup>6</sup> and other fields. In view of the urgent need for high analytical throughput and spatial resolution, ECL imaging was created by coupling

ECL technology with optical microscopy. This new imaging method can not only achieve high-throughput and visualization of biomolecules,<sup>7</sup> but also be applied to the research of spatiotemporal resolution imaging of single entities such as particles, cells and bacteria.<sup>8–12</sup> It can provide more spatial detail information, such as identification of cellular contents and structures. Small molecules in cells or released from cells, (sub-)cellular structures including cell membranes,<sup>13,14</sup> proteins,<sup>10,15</sup> mitochondria<sup>16</sup> and intracellular hierarchical structures such as the nucleolus, nucleus and endoplasmic reticulum<sup>17</sup> have been visualized by ECL imaging.<sup>18</sup> With the enhancement effects of silica nanochannels on ECL intensity, Su *et al.* have explored the dynamic variation of cell–matrix adhesions during collective migration using the negative imaging mode.<sup>19–21</sup> Despite the significant achievements in ECL imaging, it has the technical limitation of insufficient sensitivity and time resolution due to the weak luminescence signal generated by the luminophore. It is a challenge to further develop a novel biocompatible approach of ECL imaging, and enhance the ECL intensity, thereby improving the sensitivity and temporal resolution.

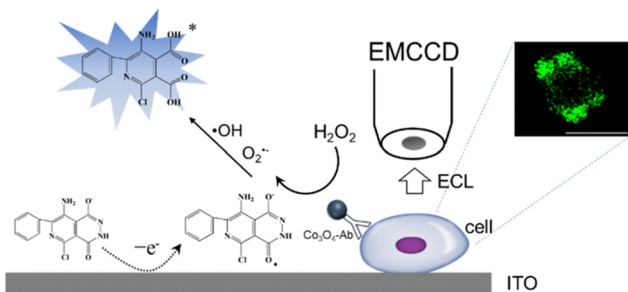
Nanozymes have sparked widespread research due to their unique properties of nanomaterials and catalytic activity.<sup>22–25</sup> They can efficiently catalyze the substrates of enzymes under mild conditions, showing catalytic efficiency and enzymatic reaction kinetics similar to natural enzymes. In view of their advantages such as adjustable catalytic activity, easy large-scale synthesis, higher physiochemical stability, higher durability and lower costs,<sup>26</sup> nanozymes have attracted considerable interest and represent great potential applications in biomedicine, disease diagnosis, the environment and biological detection. Numerous studies have indicated that many nanomaterials have been discovered to possess nanozyme properties, such as Fe<sub>3</sub>O<sub>4</sub>,<sup>26</sup> AuNPs,<sup>27,28</sup> carbon dots,<sup>29</sup> and so on. Among them, Co<sub>3</sub>O<sub>4</sub> nanoparticles, with catalase-like activity, can be adopted as accelerators of H<sub>2</sub>O<sub>2</sub> for enhancing the ECL emission of the luminol-H<sub>2</sub>O<sub>2</sub> system, because the conversion of the Co<sup>2+</sup>/Co<sup>3+</sup> redox pair could efficiently excite the generation of electron holes, and then react with the molecular orbital

<sup>a</sup> School of Geographic and Biologic Information, Nanjing University of Posts and Telecommunications, Nanjing 210023, China

<sup>b</sup> Department of Urology, Xuzhou Central Hospital, Xuzhou 221009, Jiangsu, China. E-mail: haolinxuzhou@163.com

<sup>c</sup> Key Laboratory for Organic Electronics and Information Displays, Jiangsu Key Laboratory for Biosensors, Institute of Advanced Materials (IAM), Jiangsu National Synergetic Innovation Center for Advanced Materials (SICAM), Nanjing University of Posts and Telecommunications, Nanjing 210023, China. E-mail: iamssu@njupt.edu.cn

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3cc03484d>



Scheme 1 ECL mechanism based on the  $\text{Co}_3\text{O}_4$  nanozyme for CEA antigen imaging at the cellular membrane.

of  $\text{H}_2\text{O}_2$  for promoting its decomposition.<sup>30</sup> Based on this advantage, the  $\text{Co}_3\text{O}_4$  nanozyme can be combined with the luminol- $\text{H}_2\text{O}_2$  ECL system, and it is expected to significantly improve the ECL intensity, thereby further improving the sensitivity and time resolution of the imaging technology and achieving rapid single-cell ECL imaging analysis.

Herein, a  $\text{Co}_3\text{O}_4$  nanozyme was used as a catalyst of the luminol- $\text{H}_2\text{O}_2$  system for single cell ECL imaging.  $\text{H}_2\text{O}_2$  produces reactive oxygen species by the catalysis of  $\text{Co}_3\text{O}_4$ . Under the condition of applying a suitable voltage, L012 (a luminol analog with high ECL efficiency) undergoes an electrochemical reaction to produce intermediates, which in turn react with reactive oxygen species to produce an ECL signal (Scheme 1). The ECL intensity of the L012- $\text{H}_2\text{O}_2$  system was significantly enhanced and achieved ultra-sensitive visualization of  $\text{H}_2\text{O}_2$ . Subsequently, a  $\text{Co}_3\text{O}_4$  nanozyme-labeled CEA antibody was used as a probe to achieve sensitive and rapid imaging of CEA at the cellular membrane. This work provides a new approach for high-resolution ECL imaging.

The diameter of the  $\text{Co}_3\text{O}_4$  nanoparticles used in this study is  $\sim 30$  nm, as characterized in the transmission electron microscopy (TEM) image (Fig. S1, ESI†). To investigate the enhancement effects of the  $\text{Co}_3\text{O}_4$  nanozyme, the electrochemical and ECL curves of the L012- $\text{H}_2\text{O}_2$  system on  $\text{Co}_3\text{O}_4$  coated ITO electrodes were measured in 10 mM PBS containing 200  $\mu\text{M}$  L012 and 20  $\mu\text{M}$   $\text{H}_2\text{O}_2$ . From Fig. 1A, in the presence of  $\text{Co}_3\text{O}_4$ , the ECL intensity from L012 and  $\text{H}_2\text{O}_2$  improved to approx. 3.0 fold higher than that of the bare ITO electrode. This enhancement should be ascribed to the excellent catalytic

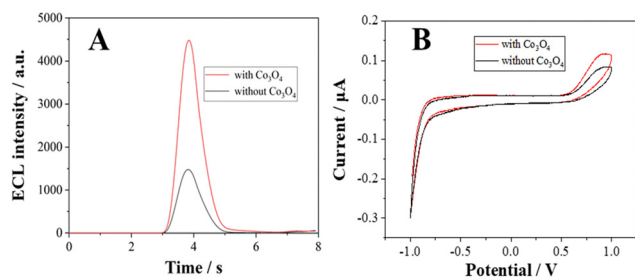
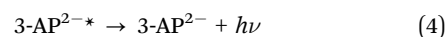
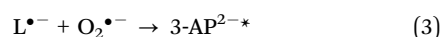
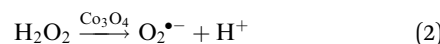
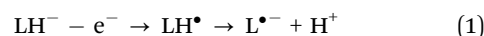


Fig. 1 (A) ECL profiles and (B) CV curves of the L012- $\text{H}_2\text{O}_2$  system at the ITO surface with and without  $\text{Co}_3\text{O}_4$ . Working solution: 10 mM PBS containing 200  $\mu\text{M}$  L012. Scan range:  $-1.0$  to  $1.0$  V, scan rate:  $0.1$  V  $\text{s}^{-1}$ , PMT voltage: 600 V.

performance of  $\text{Co}_3\text{O}_4$ , which accelerates the decomposition of  $\text{H}_2\text{O}_2$  to produce reactive oxygen radicals. The corresponding CV curves show that the peak current of the  $\text{Co}_3\text{O}_4$ -coated ITO electrode is significantly enhanced compared with the bare ITO electrode (Fig. 1B), indicating that the presence of  $\text{Co}_3\text{O}_4$  could promote electron transport. The enhancements of both the ECL and peak current demonstrate that the introduction of  $\text{Co}_3\text{O}_4$  served as an accelerator promoting the decomposition of coreactant  $\text{H}_2\text{O}_2$  and enhancing the ECL intensity of the L012- $\text{H}_2\text{O}_2$  system. A possible ECL mechanism is proposed (eqn (1)–(4)). L012 anion  $\text{LH}^-$  is oxidized to radical  $\text{L}^{\bullet-}$  during the electrochemical scanning. Meanwhile, the  $\text{Co}_3\text{O}_4$  catalyzes  $\text{H}_2\text{O}_2$  to produce abundant oxygen radicals  $\text{O}_2^{\bullet-}$  and  $\text{OH}^\bullet$  on the ITO surface. In addition,  $\text{Co}_3\text{O}_4$  can facilitate the reaction of  $\text{L}^{\bullet-}$  and the oxygen radical to generate activated intermediate  $3\text{-AP}_2^{\bullet-}$ , which goes back to the ground state, generating the ECL emission.



The diluted  $\text{Co}_3\text{O}_4$  was coated on the ITO slide, which was then placed in 10 mM PBS containing 200  $\mu\text{M}$  L012 and 20  $\mu\text{M}$   $\text{H}_2\text{O}_2$ , and the ECL signal was generated by applying a conversion mode voltage between 1.0 V (2 s) and  $-1.0$  V (0.5 s) using a voltage transmitter. The ECL images were obtained using EMCCD with an exposure time of 1 s by applying 1.0 V voltage. Fig. S2 (ESI†) shows the bright-field and ECL image of  $\text{Co}_3\text{O}_4$  on the ITO slide. The light spots generated in the ECL image can coincide with the particles in the bright field image, indicating that the presence of  $\text{Co}_3\text{O}_4$  produces locally enhanced light spots. In addition, the voltage from 0.5 V to 1.2 V at the ITO slide was applied to induce ECL. The distinguished luminescence spots at  $\text{Co}_3\text{O}_4$  nanoparticles can be clearly seen under a voltage of 0.7 V from Fig. 2, while no luminescence is recorded from the ITO slide, which is attributed to the fact that  $\text{Co}_3\text{O}_4$  can catalyze the L012- $\text{H}_2\text{O}_2$  system to produce locally enhanced signals. When the applied voltage reaches 1.0 V, the luminescence spots can be seen from all the  $\text{Co}_3\text{O}_4$  nanoparticles.

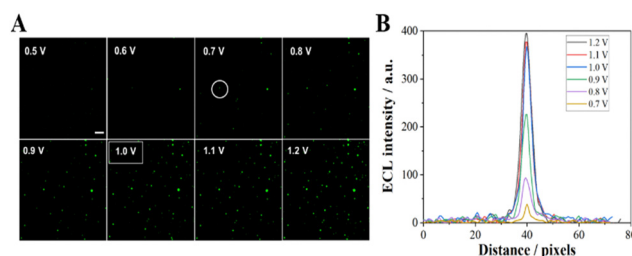


Fig. 2 (A) ECL image of  $\text{Co}_3\text{O}_4$  at the ITO slides with different applied potentials; the exposure time was 1 s. (B) The luminescence intensity across the particle circle in image A. Scale bar: 5  $\mu\text{m}$ .

A continuous increase in the voltage from 0.5 to 1.2 V results in an increase in the ECL intensity from the  $\text{Co}_3\text{O}_4$  particle (Fig. 2B), which is consistent with our previous observation. When the voltage exceeds 1.0 V, the increase in ECL intensity is not significant, therefore 1 V was selected as the ECL imaging voltage.

The effect of exposure time on ECL imaging was also explored. Under the condition of applied voltage at 1 V, when the exposure time is 500 ms, all the particles at the ITO interface produce clearly distinguishable luminescence spots (Fig. 3). Compared with the previous ECL imaging analysis,<sup>12,31,32</sup> the presence of the  $\text{Co}_3\text{O}_4$  nanozyme can shorten the exposure time obviously, indicating that the catalytic activity of the  $\text{Co}_3\text{O}_4$  nanozyme can significantly enhance the ECL luminescence of the L012- $\text{H}_2\text{O}_2$  system, endowing it with higher temporal resolution. In addition, the stability of ECL luminescence using the  $\text{Co}_3\text{O}_4$  nanozyme as a catalyst was also investigated. As shown in Fig. 4, the intensity maintains good stability (the relative standard deviation is less than 4.4%) during 30 consecutive images, which is extremely important in single-cell ECL imaging.

The quantitative visualization of  $\text{H}_2\text{O}_2$  at individual  $\text{Co}_3\text{O}_4$  nanoparticles can further validate the catalytic performance of the  $\text{Co}_3\text{O}_4$  nanozyme. Thus, we investigated the ECL visualization of different concentrations of  $\text{H}_2\text{O}_2$  (50 pM–20  $\mu\text{M}$ ) at  $\text{Co}_3\text{O}_4$  nanoparticles in 10 mM PBS containing 200  $\mu\text{M}$  L012. As shown in Fig. 5 and Fig. S3 (ESI<sup>†</sup>), as the concentration of  $\text{H}_2\text{O}_2$  increases, the ECL intensity of the light spot gradually increases. The spot intensity of three randomly selected nanoparticles shows a good positive correlation with  $\text{H}_2\text{O}_2$  concentration. The visualization detection limit is as low as 50 pM, which is much less than the previously reported detection limit (2  $\mu\text{M}$ ) under a voltage of 1.0 V.<sup>32</sup> The results demonstrate that the  $\text{Co}_3\text{O}_4$  nanozyme can effectively catalyze the L012- $\text{H}_2\text{O}_2$  system to generate strong luminescence under a low  $\text{H}_2\text{O}_2$  concentration, endowing this strategy with extraordinary visualization sensitivity and ensuring the possibility of further visualization of single cell membrane proteins.

Imaging analysis of low-abundance proteins on the cell surface is crucial for elucidating the molecular mechanisms and related research on disease progression.<sup>10,33,34</sup> Since the CEA antibody enables the selective recognition of cells that express CEA, MCF-7 cells with overexpressed CEA and HeLa

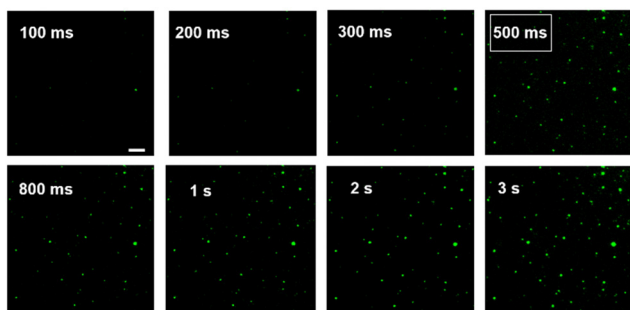


Fig. 3 ECL image of  $\text{Co}_3\text{O}_4$  at ITO slides with different exposure times; the applied potential was 1 V. Scale bar: 5  $\mu\text{m}$ .

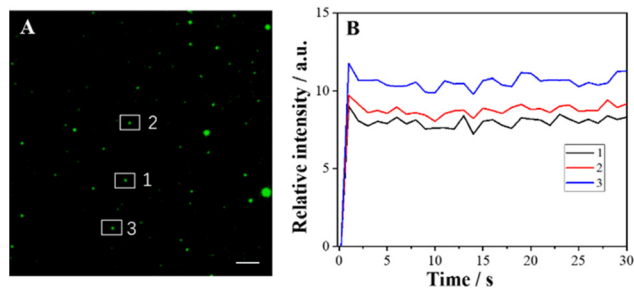


Fig. 4 (A) ECL image of  $\text{Co}_3\text{O}_4$  at ITO slides; (B) the fluctuation of ECL intensity at three particles labeled as 1, 2 and 3 in (A). The exposure time was 500 ms; the applied potential was 1 V. Scale bar: 5  $\mu\text{m}$ .

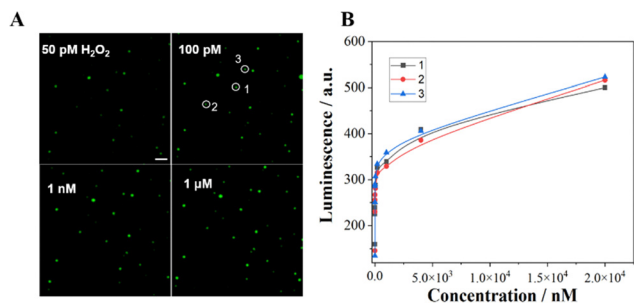
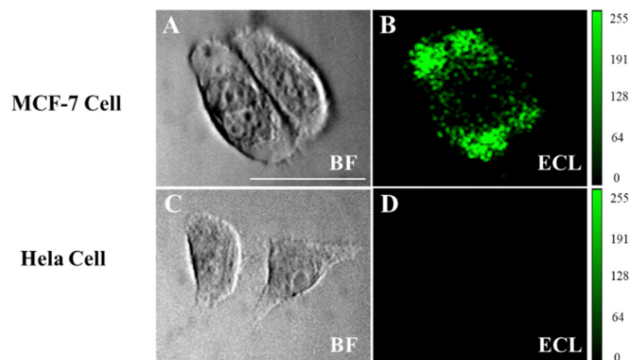


Fig. 5 (A) ECL image of  $\text{Co}_3\text{O}_4$  at ITO slides with different concentrations of  $\text{H}_2\text{O}_2$ ; the exposure time was 500 ms, and the applied potential was 1 V. (B) Correlation between the luminescence of three particles and the concentration of  $\text{H}_2\text{O}_2$ . Scale bar: 5  $\mu\text{m}$ .

cells were chosen as the experimental and negative control group, respectively. MTT assay confirms that  $\text{Co}_3\text{O}_4$  has a low cytotoxicity to MCF-7 cells (Fig. S4, ESI<sup>†</sup>). The cells were cultured on the ITO slides to the adherent state and fixed with 4% paraformaldehyde to ensure the interaction between the antigen and the antibody.<sup>10</sup> MCF-7 cells show obvious luminescence signals after modifying the functionalized probe  $\text{Co}_3\text{O}_4$ -antibody, while no obvious signal was generated in the HeLa cell region (Fig. 6). These results confirm that the luminescence observed in MCF-7 cells is attributed to the connection of the functionalized probe to the CEA antigen on the cell membrane, so that L012 and  $\text{H}_2\text{O}_2$  in the solution only produced an enhanced ECL signal on the cell surface. We consider that the  $\text{Co}_3\text{O}_4$  nanozyme has great potential in imaging other biomarkers on cell membranes owing to its favorable enzyme catalytic activity and biocompatibility.

In summary, we successfully developed a novel  $\text{Co}_3\text{O}_4$  nanozyme-based ECL imaging approach for the analysis of single cell membrane proteins for the first time. The excellent catalytic performance of  $\text{Co}_3\text{O}_4$  nanozymes can effectively accelerate the decomposition of  $\text{H}_2\text{O}_2$  into reactive oxygen species and promote the electrochemical reaction of the L012- $\text{H}_2\text{O}_2$  system, inducing an enhanced ECL signal in a very short period. As a result, our ECL imaging system achieves rapid and sensitive imaging of single cell membrane proteins. Furthermore, our proposed  $\text{Co}_3\text{O}_4$  nanozyme-enhanced ECL



**Fig. 6** Bright field (A) and ECL image (B) of MCF-7 cells; bright field (C) and ECL image (D) of HeLa cells on ITO slides. Both MCF-7 and HeLa cells were incubated with  $\text{Co}_3\text{O}_4$ -CEA antibodies and then washed three times. Working solution: 10 mM PBS containing 200  $\mu\text{M}$  L012 and 100  $\mu\text{M}$   $\text{H}_2\text{O}_2$ . The exposure time was 500 ms, the applied potential was 1 V. Scale bar: 40  $\mu\text{m}$ .

imaging exhibits good versatility, which sheds new light on the ECL imaging of biological molecules.

This work was supported by the National Nature Science Foundation of China (62101281, 62235008, 21922408 and 22274081), Natural Science Foundation of Jiangsu Province (BK20210584), the Belt and Road Innovation Cooperation Project of Jiangsu (BZ2022011), Natural Science Research Start-up Foundation of Recruiting Talents of Nanjing University of Posts and Telecommunications (Grant No. NY221032), and State Key Laboratory of Analytical Chemistry for Life Science (SKLACLS2205).

## Conflicts of interest

There are no conflicts to declare.

## References

- W. Miao, *Chem. Rev.*, 2008, **108**, 2506–2553.
- M. Richter, *Chem. Rev.*, 2004, **104**, 3003–3036.
- J. Zhang, R. Jin, D. Fang and D. Jiang, *Chem. Res. Chin. Univ.*, 2020, **41**, 2421–2425.
- M. Zhao, W. Zeng, Y. Chai, R. Yuan and Y. Zhuo, *Anal. Chem.*, 2020, **92**, 11044–11052.
- Y. Zhang, R. Zhang, X. Yang, H. Qi and C. Zhang, *J. Pharm. Anal.*, 2019, **9**, 9–19.
- H. Chu, W. Guo, J. Di, Y. Wu and Y. Tu, *Electroanalysis*, 2009, **21**, 1630–1635.
- N. Wang, Z. Wang, L. Chen, W. Chen, Y. Quan, Y. Cheng and H. Ju, *Chem. Sci.*, 2019, **10**, 6815–6820.
- Y. Liu, H. Zhang, B. Li, J. Liu, D. Jiang, B. Liu and N. Sojic, *J. Am. Chem. Soc.*, 2021, **143**, 17910–17914.
- B. Li, X. Huang, Y. Lu, Z. Fan, B. Li, D. Jiang, N. Sojic and B. Liu, *Adv. Sci.*, 2022, **9**, 2204715.
- J. Zhang, R. Jin, D. Jiang and H.-Y. Chen, *J. Am. Chem. Soc.*, 2019, **141**, 10294–10299.
- Y. Wang, D. Jiang and H.-Y. Chen, *CCS Chem.*, 2022, **4**, 2221–2227.
- Y. Lu, X. Huang, S. Wang, B. Li and B. Liu, *ACS Nano*, 2023, **17**, 3809–3817.
- S. Voci, B. Goudeau, G. Valenti, A. Lesch, M. Jovic', S. Rapino, F. Paolucci, S. Arbault and N. Sojic, *J. Am. Chem. Soc.*, 2018, **140**, 14753–14760.
- C. Ma, M. Wang, H. Wei, S. Wu, J. Zhang, J. Zhu and Z. Chen, *Chem. Commun.*, 2021, **57**, 2168–2171.
- G. Valenti, S. Scarabino, B. Goudeau, A. Lesch, M. Jovic', E. Villani, M. Sentic, S. Rapino, S. Arbault, F. Paolucci and N. Sojic, *J. Am. Chem. Soc.*, 2017, **139**, 16830–16837.
- S. Li, X. Ma, C. Pang, M. Wang, G. Yin, Z. Xu, J. Li and J. Luo, *Biosens. Bioelectron.*, 2021, **176**, 112944.
- Y. Ma, C. Colin, J. Descamps, S. Arbault and N. Sojic, *Angew. Chem., Int. Ed.*, 2021, **60**, 18742–18749.
- P. Zhou, L. Ding, Y. Yan, Y. Wang and B. Su, *Chem. Commun.*, 2023, **59**, 2341–2351.
- W. Guo, P. Zhou, L. Sun, H. Ding and B. Su, *Angew. Chem., Int. Ed.*, 2021, **60**, 2089–2093.
- H. Ding, W. Guo and B. Su, *Angew. Chem., Int. Ed.*, 2020, **59**, 449–456.
- H. Ding, P. Zhou, W. Fu, L. Ding, W. Guo and B. Su, *Angew. Chem., Int. Ed.*, 2021, **60**, 11769–11773.
- J. Wu, X. Wang, Q. Wang, Z. Lou, S. Li, Y. Zhu, L. Qin and H. Wei, *Chem. Soc. Rev.*, 2019, **48**, 1004–1076.
- S. M. Khoshfetrat, P. Hashemi, A. Afkhami, A. Hajian and H. Bagheri, *Sens. Actuators, B*, 2021, **348**, 130658.
- L. Jiao, H. Y. Yan, Y. Wu, W. L. Gu, C. Z. Zhu, D. Du and Y. H. Lin, *Angew. Chem., Int. Ed.*, 2020, **59**, 2565–2576.
- X. Wang, Y. Xu, N. Cheng, X. Wang, K. Huang and Y. Luo, *Catalysts*, 2021, **11**, 638.
- S. W. Wu, D. Z. Guo, X. C. Xu, J. M. Pan and X. H. Niu, *Sens. Actuators, B*, 2020, **303**, 127225.
- Y. Chen, X. Gou, C. Ma, D. Jiang and J. Zhu, *Anal. Chem.*, 2021, **93**, 7682–7689.
- J. Zhang, Z. Huang, Y. Xie and X. Jiang, *Chem. Sci.*, 2022, **13**, 1080–1087.
- W. Gao, J. He, L. Chen, X. Meng, Y. Ma, L. Cheng, K. Tu, X. Gao, C. Liu, M. Zhang, K. Fan, D. Pang and X. Yan, *Nat. Commun.*, 2023, **14**, 160.
- J. Dong, L. Song, J. Yin, W. He, Y. Wu, N. Gu and Y. Zhang, *ACS Appl. Mater. Interfaces*, 2014, **6**, 1959–1970.
- G. Valenti, S. Scarabino, B. Goudeau, A. Lesch, M. Jovic, E. Villani, M. Sentic, S. Rapino, S. Arbault, F. Paolucci and N. Sojic, *J. Am. Chem. Soc.*, 2017, **139**, 16830–16837.
- J. Zhang, R. Jin, Y. Chen, D. Fang and D. C. Jiang, *Sens. Actuators, B*, 2021, **329**, 129208.
- T. Gao, B. Wang, L. Shi, X. L. Zhu, Y. Xiang, J.-I. Anzai and G. Li, *Anal. Chem.*, 2017, **89**, 10776–10782.
- Y. Liu, H. Zhang, B. Li, J. Liu, D. Jiang, B. Liu and N. Sojic, *J. Am. Chem. Soc.*, 2021, **143**, 17910–17914.