## Materials Advances

## CORRECTION



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## Correction: DFT investigation of the oxygen reduction reaction over nitrogen (N) doped graphdiyne as an electrocatalyst: the importance of pre-adsorbed OH\* and the solvation effect

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The authors regret that on the 5th page of the article, in the three paragraphs from "Consequently, upon applying a limiting potential of 0.22 V..." to "...cannot be used as an ORR electrocatalyst.", all instances of 'sp-N1GDY(OH)/GDY' and 'sp-N2GDY(OH)/GDY' were mislabelled. They should be read as 'sp-N1GDY(OH)/G' and 'sp-N2GDY(OH)/G', respectively.

In Fig. 4, section (a), 'sp-N1GDY/G (OH)' and 'sp-N2GDY/G (OH)' were mislabelled. They should be read as 'sp-N1GDY(OH)/G' and 'sp-N2GDY(OH)/G', respectively. The correctly labelled Fig. 4 is given below.

In the section titled "The electronic structure of the active site relates to  $O_2$  activation", the  $O_2$  dissociation barrier values for sp-N1GDY(OH)/G and Pyri-NGDY/G are given incorrectly. The correct text should be "...and  $O_2$  dissociation barrier is high (1.86 eV and 1.52 eV)".

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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Fig. 4 (a) PDOS of C 2p of the active site in sp-N1GDY/G, sp-N1GDY/G, sp-N2GDY/G, sp-N2GDY/G, sp-N2GDY/G, and Pyri-NGDY/G. (b) The relationship between PDOS height value at the Fermi level and the adsorption energy of  $O_2$ .

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