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1 Introduction

Oral cancer is one of the most common malignant tumors occurring in head and neck, among which more than 90% these cases is oral squamous cell carcinoma (OSCC).¹ Statistically, there are over 350 000 new cases of OSCC each year and over 150 000 deaths;² thus, this disease poses a serious threat to human health. At present, the main non-surgical therapeutic modalities of OSCC primarily focus on radiotherapy and systemic chemotherapy, and the enrichment of drugs is difficult to achieve in local tumor lesions.

Microneedle-based drug delivery systems have been proven to possess excellent drug loading and release capabilities to overcome the skin/mucosal barrier for transdermal therapy.³⁻⁵ With its advantages of a minimally invasive procedure and painlessness, it has been used to treat a variety of diseases, including melanoma⁶ and breast cancer,⁷ and in the therapy for other tumors.⁸⁻¹⁰ Unlike other deep tumors, OSCC often occurs in superficial sites such as the tongue and buccal mucosa, which are within the penetration capability of microneedles.^{11,12}

A tumor microenvironment-responsive microneedle patch for chemodynamic therapy of oral squamous cell carcinoma⁺

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Oral squamous cell carcinoma (OSCC) is one of the most common malignant tumors of the head and neck, and this disease has become a threat to public health due to its poor prognosis and high fatality rate. Chemodynamic therapy (CDT) is an emerging oncology treatment based on the Fenton reaction. However, the lack of endogenous hydrogen peroxide (H_2O_2) in tumor cells and the high concentration of glutathione (GSH) that depletes toxic hydroxyl radicals (\cdot OH) significantly impair the efficacy of CDT. Here, we developed a polyvinyl alcohol (PVA)-based soluble microneedle patch (denoted as Fe₃O₄ + VC-MN) loaded with Fe₃O₄ nanoparticles (NPs) and vitamin C (VC) for the effective treatment of OSCC. When Fe₃O₄ + VC-MNs are inserted into the OSCC tissue, the Fe₃O₄ NPs and VC loaded in the tip of the needle are released in a targeted manner. After VC is converted into oxidized vitamin C (DHA), it can consume GSH in tumor cells and generate sufficient intracellular H_2O_2 *in situ*. Moreover, by virtue of their peroxidase-like activity, Fe₃O₄ NPs can induce the generation of lethal \cdot OH through the Fenton reaction with the aforementioned H_2O_2 , leading to tumor cell ferroptosis and apoptosis, thus achieving CDT. Collectively, this functional microneedle patch provides a more efficient and minimally invasive targeted drug delivery solution for the treatment of OSCC.

> Chemodynamic therapy (CDT) is a new developing treatment that uses the endogenously overexpressed hydrogen peroxide (H_2O_2) in tumors to kill cancer cells by producing toxic hydroxyl radicals ('OH) through a metal-catalysed Fenton reaction.^{13,14} Nanozymes, such as Fe₃O₄ nanoparticles (NPs), possess peroxidase-like activity and can catalyse the production of 'OH from H₂O₂, a mechanism that has been applied for tumor therapy.¹⁵⁻¹⁷ However, insufficient H₂O₂ in the local tumor microenvironment (TME) and the depletion of ·OH by the glutathione (GSH) that is present in high concentrations severely limit the efficacy of CDT.

> Currently, the direct delivery of H₂O₂ or peroxide materials is a common strategy to compensate for H₂O₂ deficiency in the tumors. For example, Park W. H. et al. used exogenous H2O2 to inhibit the growth of lung cancer cells through blocking the G1 phase.18 Lin L. S. et al. proposed using copper peroxide nanodots as stimulative agents for enhanced CDT by sustaining the H₂O₂ supply.¹⁹ Although these methods have increased the concentration of H₂O₂ in tumors to some extent, they do not continuously supply H₂O₂. Recent studies have shown that vitamin C (VC), also known as ascorbic acid, is oxidized vitamin C (DHA) in the intercellular matrix. Afterwards, these compounds can selectively enter tumor cells through glucose transporters (GLUTs) on the cell membrane, consuming high concentration of intracellular GSH and producing large amounts of H₂O₂.^{20,21} Based on these properties, integrating VCmediated GSH depletion and H2O2 production and the Fe3O4-



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Paper



Fig. 1 Schematic illustration of the synthesis of Fe_3O_4 + VC-MNs and its transmucosal drug delivery in OSCC.

catalyzed Fenton reaction can significantly improve the effectiveness of CDT.

Therefore, in this study, we constructed a PVA-based microneedle patch loaded with VC and Fe_3O_4 NPs (Fe_3O_4 + VC-MNs) for OSCC treatment. Puncture of the OSCC tissue with Fe_3O_4 + VC-MNs allowed for targeted release of the drugs loaded in the tip of the needle, after which VC was converted to DHA to react with reduced GSH and produce large amounts of H_2O_2 in tumor cells. Then, the Fe_3O_4 NP-induced Fenton reaction generated lethal 'OH, leading to tumor cell ferroptosis and apoptosis, thus providing a more efficient and minimally invasive targeted drug delivery solution for the treatment of OSCC (Fig. 1).

2 Materials and methods

2.1 Materials

Polyvinyl alcohol (PVA) and Fe_3O_4 nanoparticles were purchased from Shanghai Aladdin Bio-Chem Technology Co., Ltd (Shanghai, China). Ascorbic acid (VC) and fluorescein isothiocyanate (FITC) were purchased from Sigma (USA). Agar was purchased from Biosharp (Anhui, China). The cell counting kit-8 (CCK-8) and cell death assay kit were purchased from Beyotime (Shanghai, China). DMEM and fetal bovine serum (FBS) were purchased from Gibco (USA). An apoptosis assay kit was purchased from Bestbio (Shanghai, China). MDA assay kits and GSH assay kits were purchased from Elabscience Biotechnology Co., Ltd (Wuhan, China).

2.2 Preparation of microneedle patches

7.5 g of PVA powder was added to 60 mL of pure water, and stirred at 95 °C until the powder was completely dissolved. Then, the mixture was centrifuged at 3000 rpm to remove the bubbles (PVA solution). 10 mg Fe₃O₄ and 10 mg VC were added to 100 mL PVA solution. 10% (w/v) Fe₃O₄ and 10% (w/v) VC were collected to PVA solution. After this, the three above-mentioned solutions were added dropwise to the polydimethylsiloxane (PDMS) microneedle mold, respectively, followed by centrifugation (3500 rpm) and drying at 37 °C overnight to obtained PVA-MN, Fe₃O₄-MN and Fe₃O₄ + VC-MN.

2.3 Characterization

The surface morphology was characterized by scanning electron microscopy (SEM) (SIGMA HD). The optical images were captured using a portable electron microscope. To verify the successful loading of Fe_3O_4 and VC, Fe_3O_4 + VC-MN was analyzed using energy dispersive spectroscopy (EDS) (OXFORD X-MAS) and Fourier transform infrared (FTIR) (Nicolet5700, USA) spectroscopy. Additionally, the mechanical properties of microneedles were evaluated using a microcomputer-controlled electronic universal testing machine (CMT4503, China).

2.4 Solubility of the microneedle patches

3.7 g of agar powder was placed in 150 mL of pure water, and the mixture was heated to 90–100 °C with stirring to dissolve the powder. After the solution is solidified, it was cut into cubes with a thickness of 2 cm to simulate skin tissue. The dissolution of Fe_3O_4 + VC-MNs at different time points was analyzed after being inserted into agar blocks and nude mouse skin tissue. In order to detect the drug release capacity of Fe_3O_4 + VC-MNs after dissolution, FITC-containing microneedles were placed in PBS solution and artificial saliva (SBF). The variation in fluorescence intensity of the immersion solution was assessed by using a microplate reader (VICTOR Nivo, Finland).

2.5 Cytocompatibility of the microneedle patches

L929 cells, HUVECs and HN30 cells were cultured in DMEM with 10% FBS at 37 °C under 5% CO₂ in a saturated humidity atmosphere. All the cells were inoculated in 96-well plates (2000 cells per well) and divided into the PVA-MN group, Fe₃O₄-MN group, and Fe₃O₄ + VC-MN group after 24 hours of cultivation with different impregnating solutions. Then, the original medium was replaced by 100 μ L of CCK-8 working solution (10% by volume), and the samples were incubated for two hours before measuring the optical density (OD) values of each well at 450 nm. Cell viability was calculated using the following equation:

 $\begin{aligned} \text{Cell viability} = (\text{OD}_{\text{control group}} - \text{OD}_{\text{experimental group}}) / \text{OD}_{\text{control}} \\ \\ & \text{group} \times 100\% \end{aligned}$

Next, we conducted live/dead cell staining assays. Specifically, L929 cells, HUVECs, and HN30 cells were inoculated into 24-well plates at a density of 2×10^4 cells per well and cultured for 24 hours. Each well was stained with Calcein-AM and propidium iodide (PI) and observed by using a fluorescence microscope (Sunny ICX41 inverted biological microscope, HK).

2.6 Cell migration assay

L929 cells, HUVECs and HN30 cells were inoculated in 6-well plates at a density of 5×10^5 cells per well for 24 hours. After scratching, the culture medium of each well was replaced by different impregnating solutions. Cell migration was observed by using an inverted fluorescence microscope.

2.7 Hemolysis experiment

Fresh blood from nude mice was washed and centrifuged to obtain red blood cells (RBCs). The RBCs were diluted to 10% solution by using PBS and incubated with pure water (positive control), PBS (negative control), and extracts of three kinds of microneedle patches at 37° for 1.5 h. After that, 100 μ L supernatant in each group was taken into a 96-well plate, and the OD value was measured at 540 nm. Hemolysis (%) was calculated using the following equation:

Hemolysis (%) =
$$(A - A_0)/(A_1 - A_0) \times 100\%$$

where A is the measured OD value of the experimental groups, A_1 is the OD value of positive group, and A_0 is the OD value of the negative group.

2.8 Cell invasion assay

A 200 μ L serum free HN30 cell suspension with a concentration of 5 × 10⁵ cells was seeded in the upper chamber. 750 μ L of FBS-containing impregnation solution was added to the lower chamber. After 24 hours, the cells were stained with 0.1% crystal violet in a dark environment for 15 minutes, and the cell invasion was detected under an inverted fluorescence microscope.

2.9 Cell cloning experiment

HN30 cells were placed in a 6-well plate at a density of approximately 100 cells per well and cultivated with different impregnation solutions for seven days. After that, 0.1% crystal violet dye solution was added to each well in the dark for 15 minutes. Finally, the cells was observed and photographed under an inverted fluorescence microscope.

2.10 Apoptosis experiment

HN30 cells were plated in a 6-well plate at a density of approximately 5×10^5 cells per well. After 48 hours of cultivation with different impregnation solutions, the cells were collected and resuspended with Annexin V binding solution. Then, Annexin V-FITC and PI staining solutions were sequentially added to the cell suspension. After incubation under dark conditions, the apoptosis of tumor cells was investigated using flow cytometry (CytoFLEXS, USA).

2.11 Cellular reactive oxygen species (ROS) experiment

HN30 cells were seeded in a 6-well plate (5 \times 10⁵ cells per well) in advance and co-cultured with different impregnation solutions. DCFH-DA working solution was obtained after being diluted by using serum-free DMEM at a ratio of 1:1000 (10 μ mol L⁻¹). After 24 hours, the cell culture was replaced with 1 μ L DCFH-DA working solution (Rosup as the positive control group). The cells were then incubated at 37 °C for 20 minutes and observed by using an inverted fluorescence microscope to determine the level of ROS in tumor cells. Besides, the cells were resuspended in PBS, and the fluorescence intensity of DCFH-DA in different groups was detected by flow cytometry to indirectly reflect the content of ROS.

2.12 Cellular lipid oxidative damage (MDA) assay

HN30 cells were collected in three centrifuge tubes with no less than 3×10^6 cells in each group. 0.5 mL of extraction solution was added to each tube and sonicated (90 W, 4 s per time, 2 s interval, and a total time of 10 minutes) to obtain the suspension. Then 0.1 mL of anhydrous ethanol, the standard sample, and the three samples to be tested, were mixed with 1 mL of working solution in a water bath at 100 °C for 40 minutes. After cooling to room temperature and centrifuging for 10 minutes, the OD value at 532 nm was measured on an enzyme-linked immunosorbent assay. The MDA value of each group was calculated according to the following formula, which corresponds to the oxidative damage in tumor cells.

MDA (nmol mg⁻¹ prot) =
$$\Delta A_1 / \Delta A_2 \times C \times f \div C_{pr}$$

where ΔA_1 is the measured OD value of the sample – the blank tube OD value; ΔA_2 is the OD value of the standard tube – the OD value of the blank tube; *C* is the concentration of the standard (10 nmol mL⁻¹); *f* is the dilution factor of the sample before it was added to the assay system; and $C_{\rm pr}$ is the protein concentration of the sample to be tested (mg prot per mL).

2.13 Cellular glutathione (GSH) detection experiment

HN30 cells were inoculated in 6-well plates, with a cell count of 1×10^6 . 500 µL of PBS (0.01 M, pH 7.4) added to each well for homogenization, and the supernatant was collected after being centrifuged ($10\,000 \times g$, 10 minutes) at 4 °C. The OD value was measured at 405 nm using an enzyme-linked immunosorbent assay.

The formula for calculating the GSH content in cells was

GSH content (µmol L⁻¹) = (
$$\Delta A_{405} - b$$
) ÷ $a \times 2 \times f$ ÷ C_{pr}

where *y* is the standard OD value – the standard blank well OD value; *x* is the concentration of the standard; *a* is the slope of the calibration curve; *b* is the intercept of the calibration curve; ΔA_{405} is the OD value of the sample well – the OD value of the assay blank; 2 is the dilution factor of supernatant preparation; *f* is the dilution factor of the sample before it was added to the assay system; and $C_{\rm pr}$ is the protein concentration of the sample to be tested (g prot L⁻¹).

2.14 In vivo application of the microneedle patch

4 week-old BALB/c-nu nude mice were purchased from Hubei Beente Biological Co., Ltd. After one week of adaptive feeding, a 150 μ L HN30 cell suspension (5 \times 10⁶ cells per mL) was injected into the right armpit of the mice. After tumorigenesis, all the mice were randomly divided into three groups (PVA-MN group, Fe₃O₄-MN group, and Fe₃O₄ + VC-MN group) according to different treatments. The body weight, tumor size, and tumor volume of nude mice were regularly observed at pre-set timepoints. The whole blood from all animals was collected for routine blood analysis, and the heart, liver, spleen, lungs, and kidneys were separated for haematoxylin and eosin (H&E) staining. All experiments were conducted in accordance with

2.15 Statistical analysis

The data are presented as the mean \pm standard deviation (SD) and were analysed using GraphPad (version 7.0) with student's *t*-tests and one-way analysis of variance (ANOVA). The threshold of statistical significance was p < 0.05.

3 Results and discussion

3.1 Synthesis and characterization of Fe₃O₄ + VC-MNs

The geometries of the individual microneedle and the entire patch affect the performance of the microneedle patches, which are typically 10–1000 μ m long.²² When applied quasistatically (manually) to the skin/mucosa, longer microneedles increase the reliability of insertion to overcome the deformation caused by shorter microneedles simply bending the surface without penetration.^{23,24} The optical pictures of PVA-MN showed that the patch is a square with a 1 cm side length, and the microneedle bodies are neatly arranged into a 10 \times 10 array, with a transparent appearance (Fig. 2a).

Biosafety is a prerequisite for *in vivo* applications of the materials. To guarantee that the effective concentration of the drug loaded against the tumor was harmless to normal cells, L929 cells, HUVEC cells and HN30 cells were treated with different concentrations of the Fe₃O₄ and VC mixture. Based on the data in Fig. S3,† a drug concentration of 500 µg mL⁻¹ was selected for microneedle synthesis. As displayed in Fig. 2b, the morphology of Fe₃O₄ + VC-MN was consistent with that of PVA-MN, while it possessed a black appearance due to the Fe₃O₄ present. SEM images of Fe₃O₄ + VC-MN (Fig. 2c) demonstrated that the microneedle body presented a pyramid shape, with a height of about 900 µm, which was sufficient to ensure penetration of the skin or mucosal tissue.²⁵⁻²⁷



Fig. 2 (a) Light microscopy images of PVA-MNs. (b) Light microscopy images of $Fe_3O_4 + VC$ -MNs. (c) SEM images of $Fe_3O_4 + VC$ -MNs. (d) EDS image of $Fe_3O_4 + VC$ -MN. (e) FTIR spectra of VC, Fe, PVA and $Fe_3O_4 + VC$ -MN. (f) Mechanical compression curves of MN patches.

0.6

Displacement (mm)

Furthermore, the EDS pattern shown in Fig. 2d demonstrates that the Fe element is densely distributed at the microneedle and bottom of the patch, verifying the successful loading of Fe₃O₄. In addition, FTIR spectra were acquired to elucidate the chemical composition of Fe₃O₄ + VC-MN. As presented in Fig. 2e, the characteristic absorption peaks of Fe–O (558 cm^{-1}), C=O (1321 cm^{-1}) and C-OH (1093 cm^{-1}) appeared, indicating the successful loading of Fe₃O₄ and VC.²⁸⁻³¹ Furthermore, in order to verify the penetration ability of microtargeted tissues, the mechanical properties of microneedles were evaluated. It can be observed from Fig. 2f that at a displacement of 400 µm, the mean compression force of PVA-MN, Fe₃O₄-MN, and Fe₃O₄ + VC-MN was 0.189 N, 0.372 N, and 0.315 N per needle, respectively, without buckling, which could meet the minimum force value (0.058 N) required for skin penetration according to previous studies.32,33

3.2 Solubility and drug release of Fe₃O₄ + VC-MNs

To test the *in vitro* solubility and puncture ability of Fe₃O₄ + VC-MN, we first utilized agar blocks to simulate human tissue and recorded the morphological changes in the microneedles after insertion. It can be observed that the tip of the Fe_3O_4 + VC-MN dissolved rapidly with a significant decrease in height as time progressed (Fig. 3a and S1⁺), and all the microneedles were mostly dissolved within 30 s, including the bottom surface. Simultaneously, to ensure the penetrating force of $Fe_3O_4 + VC_2$ MN, it was punctured on the mouse skin surface (Fig. S1[†]). As shown in Fig. 3b, with the extension of time, the tip of the microneedle dissolved significantly at 3 min and basically dissolved at 10 min. However, the dissolution rate of the microneedles inserted into the skin tissue was significantly slower than that after insertion into the agar block, possibly ascribed to the higher water content and softer texture of agar. Fig. 3c displays the H&E staining image of skin tissue after being punctured with PVA-MN. The formation of pinholes on the skin surface corresponding to the microneedle array can be observed. Furthermore, the H&E staining image of skin tissue after being punctured by Fe₃O₄ + VC-MN at the puncture site shows the conical channel formed, as well as the retention of black Fe₃O₄ particles, which proved the excellent puncture



Fig. 3 (a) Dissolution of the $Fe_3O_4 + VC-MNs$ on the surface of agar blocks at different time points. (b) Dissolution of the $Fe_3O_4 + VC-MNs$ on the surface of nude mouse skin tissue at different time points. (c) H&E-stained image of the nude mouse skin surface after PVA-MN puncture. (d) H&E-stained image of the nude mouse skin surface after $Fe_3O_4 + VC-MN$ puncture. (e) Dissolution of the FITC-PVA-MNs in PBS. (f) Dissolution of the FITC-PVA-MNs in SBF.

Wavenumbers (cm⁻¹)

strength and *in situ* drug release properties of $Fe_3O_4 + VC-MN$ (Fig. 3d).

Next, FITC was used as a simulation drug loaded on microneedles (FITC-MN) to test the drug release capacity.³⁴ After being placed in PBS and SBF solution, FITC-MN was completely dissolved in roughly 10 minutes (Fig. 3e and f). For visually presenting drug release after dissolution, FITC-MN was punctured into an agar block. According to the area of yellow-green color at different time points, the solubility of the microneedle and FITC release were indirectly reflected (Fig. S2†). All these results indicated that $Fe_3O_4 + VC$ -MN possessed admirable solubilization, drug loading and release capacities, which assured the reliability of its *in vivo* application.

3.3 Anti-tumor effects of Fe₃O₄ + VC-MNs in vitro

The prognosis of tumors is closely related to the ability of tumors to proliferate, invade, migrate, and undergo apoptosis.^{35,36} For assessing the antitumor effect of microneedles, PVA-MNs (control group), Fe_3O_4 -MNs and Fe_3O_4 + VC-MNs (experimental groups) were prepared for the following cellular experiments. Preliminarily, we conducted live and dead cell staining to evaluate cell survival after microneedle treatment. As shown in Fig. 4a, a large number of HN30 cells died



Fig. 4 (a) Images of live and dead HN30 cells in different groups. (b) Migration of HN30 cells after scratching in different groups. (c) Proportions of HN30 cell scratches that had healed in different groups. (d) Migration of HN30 cells in the different groups. (e) Number of migrated HN30 cells in different groups. (f) Cloning of the HN30 cells in different groups. (g) Number of HN30 cell clones in different groups. (h) Apoptosis of HN30 cells in different groups. (i) Relative levels of HN30 cell apoptosis in different groups. Scale bar = $200 \,\mu$ m. *p < 0.05, **p < 0.01, ***p < 0.001, ***p < 0.001.

after 24 hour coculture with Fe₃O₄-MNs and Fe₃O₄ + VC-MNs, compared with that of PVA-MNs, while the L929 cells and HUVECs showed no obvious cell death (Fig. S4†). Then, the migration of HN30 cells under the actions of the three types of microneedles was assayed. As can be seen from Fig. 4b and c, after scratching, the migration of HN30 cells was significantly inhibited in Fe₃O₄-MN and Fe₃O₄ + VC-MN groups, where in the degree of inhibition was more remarkable in the Fe₃O₄ + VC-MNs group, while there was no such inhibitory effect in L929 cells and HUVECs (Fig. S5†), which further confirmed the selective killing effect of Fe₃O₄ + VC-MNs on tumor cells. Besides, we assayed the hemolysis of different microneedle patches. As shown in Fig. S6,† the hemolysis rates in experimental groups are all lower than 5%, demonstrating the admirable compatibility of Fe₃O₄ + VC-MN.

Afterwards, the influence of microneedles on cell invasion ability was evaluated. Fig. 4d clearly shows that the invasion of HN30 cells is restrained in the Fe₃O₄-MN and Fe₃O₄ + VC-MN groups compared with in the PVA-MN group. According to the quantitative statistical data in Fig. 4e, the amount of cells invaded in the Fe₃O₄ + VC-MN group was significantly lower than that in the Fe₃O₄ + VC-MN group. The results of further cell cloning experiments showed the same trend, and the number of cloning clusters in the Fe₃O₄ + VC-MN group was less than that in the other two groups (Fig. 4f and g). Subsequently, flow cytometry was used to detect apoptosis after the application of different microneedles, and the results show that the number of apoptotic cells in the Fe₃O₄ + VC-MN group was the highest among the three groups (Fig. 4h and i).

3.4 Cellular mechanisms of Fe₃O₄ + VC-MNs

Reactive oxygen species (ROS), hydroxyl radical, play a mediating role in the process of cell life activity, stress and death. Once the levels of ROS (in excess) and antioxidant components (present at low levels) cannot be maintained in balance, intracellular oxidative stress can result, thereby affecting the cell state.37 To verify the effects of Fe3O4 and VC on cellular oxidative stress in the tumor microenvironment, we first examined the performance of intracellular ROS in HN30 cells under the action of three microneedles. Fig. 5a shows the levels of intracellular ROS of HN30 cells (green fluorescence) after Fe₃O₄ + VC-MN treatment was highest, which was close to that in the positive control group (Rosup). In addition, Fig. 5b shows the results of ROS detection by flow cytometry. After stimulation with different microneedle extracts, the cells in the Fe₃O₄ + VC-MN group produced more ROS than that in PVA-MN and Fe₃O₄-MN groups, indicating that the co-existence of Fe_3O_4 and VC promoted the production of more ROS. Furthermore, the result in Fig. 5c showed that the MDA content of HN30 cells was the highest after Fe₃O₄ + VC-MN treatment, meaning the most severe intracellular oxidative damage.

GSH, as an important antioxidant, is capable of effectively trapping and eliminating free radicals, and protects the sulfhydryl groups of basic enzymes and proteins from oxidation or inactivation.³⁸ Stable GSH levels are key for maintaining normal cellular biological functions, while the excessive reduced GSH Paper



Fig. 5 (a and b) ROS levels in HN30 cells. (c) MDA concentrations in HN30 cells in different groups. (d) GSH levels in HN30 cells in different groups. Scale bar = $200 \ \mu m$. *p < 0.05, **p < 0.01, ***p < 0.001, ****p < 0.001.

will weaken the CDT efficacy of ROS against tumor cells.³⁹ Fig. 5d shows the intracellular GSH contents in HN30 cells. The results indicated that the concentration of GSH in the tumor cells in the Fe₃O₄ + VC-MN group was significantly less than that in the PVA-MN group, demonstrating that the presence of Fe₃O₄ and VC inhibited the production of GSH. All these experimental results confirmed that the targeted release of Fe₃O₄ and VC at the tumor site could induce *in situ* production of sufficient H₂O₂ by consumption intracellular reduced GSH, thereby promoting the Fenton reaction to inhibit tumors.

3.5 In vivo application of Fe₃O₄ + VC-MNs

Studies have shown that microneedles can induce more apoptosis and inhibit cancer cell proliferation in vivo and enhance anti-tumor effects by delivering drugs.40 To verify the efficacy and biosafety of Fe₃O₄ + VC-MNs, we constructed subcutaneous tumor models in nude mice and applied the three kinds of microneedles to a local site (Fig. 6a). At the present time point, we regularly photographed and observed the gross and local tumors of mice, meanwhile comparing the volume variations of tumors in different groups. From the optical images in Fig. 6b and S7,† we found that compared with the PVA-MNs, the application of Fe₃O₄-MN and Fe₃O₄ + VC-MN groups resulted in smaller tumors. According to quantitative analysis of tumor volume (Fig. 6c), it could be determined that the tumor volume in the Fe₃O₄ + VC-MN group was the smallest. Ultimately, we evaluated the in vivo biocompatibility of microneedles. Based on the monitoring the results of the weight of nude mice, it was found that the weight of all the mice steadily increased over time, with no significant difference among the three groups (Fig. 6d). Meanwhile, Fig. 6e presents H&E staining images of the heart, liver, spleen, lungs, and kidneys of mice after sacrifice. Moreover, Table. S1[†] shows the results of routine blood tests after death on the tenth day after microneedle treatment in different groups. The results showed that after microneedle treatments, there was no significant damage to the visceral organs. All the above results demonstrated the effectiveness and biosafety of Fe₃O₄ + VC-MNs when applied to OSCC treatment.



Fig. 6 (a) Schematic diagram of the *in vivo* experimental process. (b) Pictures of tumors from three groups at different time points. (c) Tumor volume changes. (d) Weight variation in the nude mice. (e) H&E staining images of the heart, liver, spleen, lung, and kidney. Scale bar = $200 \ \mu\text{m}$. *p < 0.05, **p < 0.01, ***p < 0.001, ****p < 0.001.

4 Conclusion

The treatment of OSCC still suffers from limited efficacy, severe toxic side effects and difficulty in eradication. In this study, a local TME-responsive functional microneedle patch was successfully constructed using the solubility and drug delivery capability of a PVA base, achieving targeted drug delivery at the tumor site for the synergistic treatment of OSCC. This microneedle patch provides a new therapeutic idea of TME responsiveness, which is expected to be further applied in the clinic and provides a new idea for other tumor-related treatment.

Author contributions

S. Z. and Y. L. contributed equally to this work. S. Z.: conceptualization, methodology, data curation, software, schematic drawing, writing – original draft, writing—review and editing,

Nanoscale Advances

supervision, and project administration. Y. L.: conceptualization, methodology, data curation, software, schematic drawing, writing – original draft, writing—review and editing, supervision, and project administration. B. C.: conceptualization and methodology.

Conflicts of interest

There are no conflicts to declare.

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Paper

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