ORGANIC CHEMISTRY

FRONTIERS

RESEARCH ARTICLE

Cite this: Org. Chem. Front., 2023, 10, 1441

Received 6th January 2023, Accepted 28th January 2023 DOI: 10.1039/d3qo00018d

rsc.li/frontiers-organic

Introduction

When specific C–H bonds of organic molecules can be directly functionalized via catalytic reactions, it often provides simple synthetic methods.¹ Metal-catalyzed C–H bond activation/ functionalization of heteroarenes, reported by Ohta et al. 1990, now represents one of the most effective methods for the preparation of heteroarylated arenes. $²$ For such coupling reac-</sup> tions, initially in most cases, only the "most reactive" C–H bond of (hetero)arenes could be functionalized. $3-7$ Nevertheless, the C–H bond functionalization methodology will only be really synthetically useful when it will be possible to activate a specific C–H bond on the molecules. Therefore, the discovery of conditions allowing the regioselective functionalization of several bonds of the same molecule, also called regiodivergent functionalization, is a very important aspect of the current research on metal-catalyzed reactions.

So far, the Pd-catalyzed reactions such as Suzuki coupling of heteroarylboronic acids with thiophenes bearing a 2-halobenzene substituent selectively lead to 1,2-diheteroarylated benzene derivatives (Scheme 1a). 8 On the other hand, the C3-

This journal is © the Partner Organisations 2023 **Org. Chem. Front.**, 2023, **10**, 1441-1455 | **1441**

Double C–H bond functionalization for C–C coupling at the β-position of thiophenes using palladium-catalyzed 1,4-migration associated with direct arylation†

Linhao Liu, **D** Marie Cordier, Thierry Roisnel **D** and Henri Doucet **D** *

The functionalization of C–H bonds at β-positions of 5-membered ring heteroarenes such as thiophene is generally more challenging than at α-positions. By using Pd-catalyzed 1,4-migration associated with direct arylation, under appropriate conditions, the functionalization of such thienyl β-positions of 2-arylthiophenes is possible. The oxidative addition of 2-(2-bromoaryl)thiophenes to palladium followed by Pd 1,4-migration activates these β-positions. Then, Pd-catalyzed direct coupling with heteroarenes provides β-heteroarylated 2-arylthiophene derivatives. In the course of this coupling reaction, a new C–C bond arises from the functionalization of two C–H bonds. Conversely, the Suzuki reaction using such 2-(2-bromoaryl)thiophenes provides 1,2-diheteroaryl-substituted benzene derivatives. These regiodivergent heteroarylations tolerate a range of substituents on the benzene ring and also several heteroarenes and provide a new route to π -extended polycyclic heteroaromatics. Moreover, these procedures employ easily available air-stable catalysts and inexpensive bases. RESEARCH ARTICLE
 (0) Checkforupdates
 (0) Checkforupdates
 COUDIC C – **H** bond functionalization for C – C
 COUDIC C – **H** bond functionalization for C – C
 **COUDIC and the B-position of thiophenes using

cause o**

heteroarylation of 2-aryl-3-halothiophenes for the access to the corresponding heteroaryl-substituted thiophene derivatives via metal-catalyzed reactions has been rarely described (Scheme 1b). $9,10$ This might be due to the tedious multi-step route to many 3-halothiophene derivatives.¹¹ To the best of our knowledge, the synthesis of 2-aryl-3-heteroarylthiophenes from 2-(2-bromoaryl)thiophenes has not been reported so far (Scheme 1c, bottom). We have recently reported that in the course of the heteroarylation of 1,2-dihalobenzenes by thiophenes, a partial Pd 1,4-migration occurred 12,13 in some cases, leading to a mixture of 1,2-diheteroarylated benzenes and also to aryl-substituted biheteroarenes such as 2′-aryl-2,3′-bithiophenes.¹⁴ This Pd 1,4-migration on thiophene derivatives provides an appealing method for functionalizing the C–H bonds at the β-position of thiophenes which are often more difficult to activate than those at the α -position.¹⁵ In addition, so far, the preparation of 2′-aryl-2,3′-bithiophene derivatives requires multi-step synthesis and the substrate scope is very limited.¹⁶ Therefore, the potential of Pd-catalyzed 1,4-migration associated with direct arylation for the access to 2-aryl-3-heteroarylthiophenes via double C–H bond functionalization needed to be explored. Herein, we report (i) on the regioselectivity of the Pd-catalyzed direct arylations vs. Suzuki couplings of 2-(2 bromoaryl)thiophenes and on the scope of the Pd-catalyzed 1,4-migration associated with direct arylation for the formation of 2-aryl-3-heteroarylthiophenes via double C–H bond functionalization (Scheme 1c).

Univ. Rennes, 35042 Rennes, France. E-mail: henri.doucet@univ-rennes1.fr; Tel: +33 (0)2 23 23 63 84

[†]Electronic supplementary information (ESI) available. CCDC 2210064 and 2217431. For ESI and crystallographic data in CIF or other electronic format see DOI: <https://doi.org/10.1039/d3qo00018d>

Scheme 1 Pd-catalyzed direct arylation vs. Suzuki coupling of 2-(2bromoaryl)thiophenes and related compounds.

Table 1 Influence of the reaction conditions on the Pd-catalyzed coupling of 2-(2-bromophenyl)-5-methylthiophene 1a with 2-acetylthiophene[®]

Entry	Catalyst	Solvent	Base	Conv. (%)	Ratio 2a:2b	Yield in 2b(%)
$\mathbf{1}$	$Pd(OAc)_{2}$	DMA	KOAc	100	19:81	64
2	Pd(OAc) ₂	DMA	Cs_2CO_3	5	56:44	nd
3	Pd(OAc) ₂	DMA	K_2CO_2	100	46:54	nd
4	$Pd(OAc)_{2}$	DMA	NaOAc	47	19:81	nd
5	Pd(OAc) ₂	DMA	CsOAc	84	40:60	nd
6	Pd(OAc) ₂	DMA	KOPiv	100	33:67	nd
7	$PdCl(C_3H_5)(dppb)$	DMA	KOAc	100	24:76	61
8	$Pd(OAc)_{2}$	DMF	KOAc	100	26:74	50
9	Pd(OAc) ₂	NMP	KOAc	100	24:76	52
10	Pd(OAc) ₂	Xylene	KOAc	Ω		Ω
11	$Pd(OAc)_{2}$	DEC	KOAc	Ω		Ω
12	$Pd(OAc)_{2}$	DMA	KOAc	86	20:80	53^b

 a [Pd] (0.02 equiv.), 2-acetylthiophene (2 equiv.), 2-(2-bromophenyl)-5methylthiophene 1a (1 equiv.), base (3 equiv.), 150 °C, 16 h, 2a : 2b ratios, determined by ¹H NMR and GC/MS analyses of the crude mix-

Results and discussion

First, we prepared a set of 2-(2-bromoaryl)thiophenes via Suzuki coupling or base deprotonation (Scheme 2). All the desired products 1a–1g were obtained in good yields.

Then, we evaluated the selectivity for the Pd-catalyzed coupling of 2-(2-bromophenyl)-5-methylthiophene 1a with 2-acetylthiophene (Table 1). The best yield of the desired isomer 2b arising from the Pd-catalyzed 1,4-migration associated with direct arylation was obtained using 2 mol% $Pd(OAc)_2$ catalyst and the KOAc base in DMA at 150 °C (Table 1, entry 1). Under these conditions, a mixture of 2a and 2b was obtained in a 19 : 81 ratio and 2b was isolated in 64% yield. The use of other bases such as Cs_2CO_3 , K_2CO_3 , NaOAc, CsOAc or KOPiv led to lower yields of 2b either due to lower conversions or poor selectivities (Table 1, entries 2–6). In the presence of the PdCl $(C_3H_5)(\text{dppb})^{17}$ catalyst, a quite similar 2a : 2b selectivity and

yield of 2b were obtained (Table 1, entry 7). The influence of a few solvents was also examined but both DMF and NMP led to lower selectivities in 2b, whereas xylene and diethyl carbonate (DEC) were ineffective (Table 1, entries 8–11). Finally, a lower reaction temperature of 130 \degree C instead of 150 \degree C gave 2b in only 53% yield due to partial conversion of 2-(2-bromophenyl)- 5-methylthiophene 1a (Table 1, entry 12). It should be mentioned that this methodology is limited to the use of 2-(2-bromoaryl)thiophenes, as from 3-(2-bromoaryl)thiophenes, complex mixtures of products were obtained.¹⁸

Next, the scope of this reaction using several heteroarenes and 2-(2-bromoaryl)thiophenes was investigated (Scheme 3). Using 2-isopropylthiazole and $1a$, a similar $a:b$ ratio was obtained and product 3b arising from the Pd 1,4-migration was isolated in 65% yield. 2-(2-Bromophenyl)-5-hexylthiophene 1b in the presence of 2-acetylthiophene or cyclopropyl-2-thienylketone afforded products 4b and 5b in 57% and 61% yields, respectively. No side-reaction on the cyclopropyl unit was detected in the course of this reaction. Then, the β-heteroarylation of the 2-chloro-substituted thiophene derivative 1c was studied. This reaction is particularly interesting as the access to 4-bromo-2-chlorothiophenes that could provide an alternative synthetic scheme for the preparation of similar products is very challenging.¹⁹ In the presence of 2-methylthiophene or 2-methyl-2-(thiophen-2-yl)-1,3-dioxolane, very high selectivities in favor of the formation of products b were observed with the formation of 6b and 7b in 69% and 71% yields, respectively. No cleavage of the thienyl C–Cl bond was observed allowing further transformations. In the presence of Scheme 2 Preparation of 2-(2-bromoaryl)thiophenes 1a–1g. selenophene, 1c gave a lower yield of 52% of 8b due to the for-

^a: 130 °C

Me Ratio 15a:15b 6:94 Ratio 16a:16b 12:88 15b 81% 16b 72% Me Me Ratio 17a:17b 5:95 Ratio 18a: 18b 13:87 b: PdCl(C_3H_5)(dppb) 5 mol% 17b 68% 18b 57% Scheme 3 Heteroarylations of thienyl rings via Pd-catalyzed 1,4 migration associated with direct arylation.

mation of unidentified side-products. 2-Isopropyl-4-methylthiazole, imidazo[1,2-a]pyrazine and 3,5-dimethylisoxazole in the presence of 1c afforded the target Pd 1,4-migration products 9b–11b in 35–65% yields. These three reactions were performed at a lower temperature of 130 °C due to a partial C–Cl bond cleavage at 150 °C. The reaction also tolerated the presence of two methyl substituents on the benzene ring.

However, with this more electron-rich 2-arylthiophene derivative 1e, 5 mol% $PdCl(C_3H_5)(dppb)$ catalyst had to be employed to reach a high conversion. With this catalyst, the target products 12b and 13b were obtained in 74% and 59% yields, respectively using 2-butylfuran and imidazo $[1,2-a]$ pyrazine as the reaction partners. The reaction also tolerated a 2-acetyl substituent on the thienyl ring. From 2-acetyl-5-arylthiophene 1h and thiophene, furan and thiazole derivatives as reaction partners, very high selectivities (88–94%) of isomers b were observed and the target products 14b–16b were obtained in 72–81% yields. Finally, the reaction of the 2-formyl-5-arylthiophene derivative 1g and 2-methylthiophene or 2-isobutylthiazole gave products 17b and 18b with 95% and 87% selectivity and in 68% and 57% yields, respectively.

The use of Pd-catalyzed 1,4-migration associated with direct arylation should allow the introduction of an heteroarene regioselectively at one of the two β-positions of the thienyl ring of non-symmetrical 2,5-diarylthiophenes. The heteroarylation should occur at the β-C–H bond on the same side of the thienyl ring bearing the 2-bromoaryl substituent. The starting materials 19–23 were prepared in 74–86% yields by Pd-catalyzed direct C5-arylation from 2-(2-bromoaryl)thiophene 1d and a set of aryl bromides using the $Pd(OAc)_2$ catalyst with the KOAc base in DMA (Scheme 4). Under these conditions – slight excess (1.3 equiv.) of an electron-poor aryl halide as the reaction partner – no cleavage of the C–Br bond of the bromotoluene unit of 1d was observed.

Then, the regioselectivity of the direct heteroarylation of 2,5-diarylthiophenes 19–23 was studied (Scheme 5). From 4-(5- (2-bromophenyl)thiophen-2-yl)benzonitrile 19 and 2-methylthiophene, the C4-heteroarylated product 24b was regioselectively obtained. The regioselectivity of the coupling reaction was determined by X-ray analysis, which confirms that there was no further Pd-migration after the initial Pd 1,4 migration. The reaction also tolerated furan, selenophene and thiazole derivatives as the reaction partners, affording products 25b–28b in 58–67% yields. From chloro- and propionylsubstituted 2,5-diarylthiophenes 20 and 21, the expected products 29b and 30b were also obtained in high yields. Again, under these conditions, no cleavage of the aryl C–Cl bond was observed allowing further transformations. No significant influence of the position of the nitrile substituent on the benzene ring was observed. Both *meta*- and *ortho-nitrile*substituted 2,5-diarylthiophenes 22 and 23 led to the desired

Scheme 4 Pd-catalyzed direct C5-arylation of the thienyl ring of 2-(2 bromoaryl)thiophene 1d.

Scheme 5 Heteroarylations of thienyl rings via Pd-catalyzed 1,4 migration associated with direct arylation.

products 31b and 32b in similar yields. It should be mentioned that for all these heteroarylations of 2,5-diarylthiophenes 19–23, complete regioselectivities in favor of the formation of products b arising from Pd 1,4-migration were observed. This selectivity could be due to the conformation of 2,5-diarylthiophenes 19–23 which would favor the Pd 1,4 migration.

The synthesis of 2,5-diaryl-4-heteroarylthiophene without the isolation of the first arylation compound was also attempted (Scheme 6). From 2-(2-bromo-4-methylphenyl)thiophene 1d, 3-bromopyridine and 2 mol% $Pd(OAc)_2$ catalyst, the 2,5-diarylated thiophene derivative was obtained in an impure form via filtration on a plug of silica. Then, 2-isopropyl-4 methylthiazole, 2 mol% $Pd(OAc)_2$ and KOAc were added to the crude mixture and heated again at 150 °C for 16 h, affording the desired product 33b in 58% yield.

Scheme 6 Preparation of a tri(hetero)arylated thiophene derivative via successive direct heteroarylations.

In order to compare the selectivity of the direct arylation reaction with Suzuki coupling, a few arylations of 2-(2-bromophenyl)-5-methylthiophene 1a and 2-(2-bromophenyl)-5-chlorothiophene 1c using arylboronic acids as the reaction partners were performed (Scheme 7). Again using 2 mol% $Pd(OAc)_2$ catalyst and DMA as the solvent, but the K_2CO_3 base instead of a carboxylate base, the formation of the expected 1,2-diheteroaryl-substituted benzene derivatives 2a, 34a and 35a was observed with complete selectivity. It should be mentioned that in the presence of the KOPiv base instead of K_2CO_3 , isomer 2a was obtained with 73% selectivity (27% of 2b) and in 52% yield indicating that the presence of a carboxylate base favors the Pd 1,4 migration. The reaction of 1c with 2-thienylboronic acid and the K_2CO_3 base also selectively led to the formation of the expected chloro-substituted 1,2-di(thiophen-2-yl) benzene 36a. Using 1c and (3,5-dimethylisoxazol-4-yl)boronic acid as the reaction partner, product 11a was also obtained with complete selectivity. These results demonstrate that the combination of Pd-catalyzed 1,4 migration reactions associated with direct arylation and Suzuki coupling can be employed as a tool to control regiodivergent heteroarylations.

π-Extended polycyclic heteroaromatics²⁰ have found several applications in the preparation of optoelectronic devices. The topology of the ring fusion and the substituents at their periphery are essential to tune their electronic properties. Therefore, we also attempted to apply the Pd-catalyzed 1,4-

Scheme 7 Heteroarylations of benzene rings via Suzuki coupling.

migration associated with direct heteroarylation to the preparation of such products. First, we employed a 4-bromothiophene derivative as a reaction partner with 1a (Scheme 8a). The desired naphtho[1,2-b:3,4-b′]dithiophene 37 was obtained in a moderate yield using an extended reaction time (48 h). Intermediate 38 was also isolated in a low yield in an impure form. We also studied the reaction outcome of 2-(2-bromophenyl)-5-methylthiophene 1a in the absence of the heteroaryl coupling partner (Scheme 8b). The formation of the desired fused polyheteroaromatic naphtho[1,2-b:3,4-c′]dithiophene 39 in 65% yield was observed. The structure of 39 was confirmed by X-ray analysis. The formation of 39 likely proceeded via the initial formation of 2-(2-bromophenyl)-2′-phenyl-3,3′-bithiophene. Then, an intramolecular Pd-catalyzed direct arylation gives naphtho[1,2 b:3,4-c′]dithiophene 39. The use of 1e also provides the corresponding naphtho[1,2-b:3,4-c′]dithiophene 40 in a moderate yield. This one-pot multi-step synthesis of π -extended polycyclic heteroaromatics is very attractive, as so far very few methods allow the preparation of such fused polyheteroaromatics.²¹ Organic Chemistry Frontiers

migration associated with direct herecognization to the properties are completed under the article. This we entire the minimal plane is determined under the common of θ and the strength onl

The access to 1,2-bis(heteroaryl)benzenes a proceeds certainly via a classical Suzuki coupling²² mechanism, whereas the access to 2′-aryl-2,3′-biheteroarenes b occurs likely via a Pd 1,4-migration followed by a direct arylation²³ as described in Scheme 9. In both cases, the first step of the catalytic cycle involves the oxidative addition of 2-(2-bromoaryl)thiophene to give intermediate A. For Suzuki coupling with arylboronic acids, classical transmetalation followed by reductive elimination gives 1,2-bis(heteroaryl)benzenes a. By contrast, with simple heteroarenes using the KOAc base, a Pd $1,4$ -migration¹²

Scheme 8 Application of the Pd-catalyzed 1,4-migration associated with direct arylation methodology for the preparation of π -extended polycyclic heteroaromatics.

Scheme 9 Proposed catalytic cycle for the access to b products.

occurs to give intermediate B. Then, after a concerted metallation deprotonation $(CMD)^{23}$ of the heteroarene coupling partner, intermediate C is obtained. Finally, reductive elimination provides 2′-aryl-2,3′-biheteroarenes b with the regeneration of the catalyst.

Conclusions

In summary, the use of either Pd-catalyzed 1,4-migration/ direct arylation or Suzuki coupling allowed the regiodivergent heteroarylation of 2-(2-bromoaryl)thiophenes. Using Suzuki coupling, 1,2-diheteroarylbenzenes were obtained. Conversely, the use of C–H heteroarylation conditions led to 2-aryl-3-heteroarylthiophenes due to a Pd 1,4-migration between the aryl and β-position of the thienyl units before heteroarylation. In the course of this coupling reaction, a new C–C bond arises from the functionalization of two C–H bonds. In addition to the advantage of being able to synthesize two classes of products from the same molecular building block, heteroarylation at the β-position of thiophenes by this method is synthetically very attractive because very few β-bromothiophenes, which could allow access to the same products by a classical direct arylation, are commercially available at an affordable cost. The reaction tolerates a variety of heteroarenes and substituents at the C5-position on 2-(2-bromoaryl)thiophenes and provides a new route to π-extended polycyclic heteroaromatics. Moreover, these reactions employ a low loading of air stable palladium sources associated with inexpensive bases. Indeed, we believe that the development of this methodology using other coupling partners for regiodivergent couplings will lead to a powerful tool for organic synthesis.

Experimental

General

DMA (99+%) extra pure and KOAc (99%) were purchased from ACROS and used without purification. All reagents were

Research Article Organic Chemistry Frontiers

weighed and handled in air. All reactions were carried out under an inert atmosphere with standard Schlenk techniques. 1 H, 19 F and 13 C NMR spectra were recorded on a Bruker Avance III 400 MHz spectrometer. High-resolution mass spectra were measured on a Bruker MaXis 4G spectrometer. Melting points were determined with a Kofler hot bench system. Chromatography purifications were performed using a CombiFlash NextGen 300 with Buchi FlashPure cartridges containing 40 μm irregular silica.

Preparation of the PdCl(C_3H_5)(dppb) catalyst¹⁷

An oven-dried 40 mL Schlenk tube equipped with a magnetic stirring bar under an argon atmosphere was charged with [Pd $(C_3H_5)Cl_2$ (182 mg, 0.5 mmol) and dppb (426 mg, 1 mmol). 10 mL of anhydrous dichloromethane was added, and then the solution was stirred at room temperature for twenty minutes. The solvent was removed under vacuum. The yellow powder was used without purification. ³¹P NMR (162 MHz, CDCl₃) δ = 19.3 (s).

General procedures for the preparation of 2-(2-bromophenyl) thiophenes 1a–1g

Procedure a: As a typical experiment, a mixture of the 1,2-dihalobenzene derivative (3 mmol), the thiophen-2-ylboronic acid derivative (4.5 mmol), 2.0 M K₂CO₃ aq. (3.6 mL), Pd(OAc)₂ (33.6 mg, 0.15 mmol) and dppf (83.1 mg, 0.15 mmol) in THF (18 mL) was stirred at 80 °C for 16 h. After being allowed to cool to room temperature, the resulting mixture was extracted with Et₂O (3×10 mL). The combined organic phase was dried over MgSO4, filtered, and concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to give 2-(2-bromoaryl)thiophenes. Research Article

weigher and random signification si

Procedure b: As a typical experiment, to a solution of the thiophene derivative (12.0 mmol) in THF (25 mL), nBuLi was added dropwise (1.55 M in hexane, 7.70 mL, 12.0 mmol) at 0 °C. After being stirred for 1 h, the 1,2-dihalobenzene derivative (10.0 mmol) was added to the solution at the same temperature. Then, the mixture was warmed to room temperature and stirred for 2 h. The residue was purified by column chromatography on silica gel to afford 2-(2-bromoaryl)thiophenes.

2-(2-Bromophenyl)-5-methylthiophene (1a). Following procedure b, from 1,2-dibromobenzene (2.36 g, 10 mmol) and 2-methylthiophene (1.18 g, 12 mmol), 1a was isolated in 67% (1.69 g) yield as a colorless oil. Eluent pentane.

¹H NMR (400 MHz, CDCl₃): δ 7.66 (dd, J = 8.0, 1.3 Hz, 1H), 7.46 $(dd, J = 7.8, 1.7 Hz, 1H$, 7.31 $(td, J = 7.5, 1.3 Hz, 1H)$, 7.16 (td, $J = 7.5$, 1.7 Hz, 1H), 7.11 (d, $J = 3.5$ Hz, 1H), 6.77 (d, $J =$ 3.5 Hz, 1H), 2.54 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 140.9, 139.4, 135.7, 133.8, 131.9, 128.8, 127.9, 127.5, 125.4, 122.8, 15.4.

LRMS calcd for $[M]^+$ C₁₁H₉BrS 252, found: 252.

2-(2-Bromophenyl)-5-hexylthiophene (1b). Following procedure b, from 1,2-dibromobenzene (2.36 g, 10 mmol) and 2-hexylthiophene (2.02 g, 12 mmol), 1b was isolated in 69% (2.23 g) yield as a colorless oil. Eluent pentane.

¹H NMR (400 MHz, CDCl₃): δ 7.68 (d, J = 8.0 Hz, 1H), 7.50 $(d, J = 7.8 \text{ Hz}, 1\text{H})$, 7.32 $(t, J = 7.5 \text{ Hz}, 1\text{H})$, 7.21–7.12 $(m, 2\text{H})$, 6.81 (d, $J = 3.5$ Hz, 1H), 2.88 (t, $J = 7.4$ Hz, 2H), 1.84-1.69 (m, 2H), $1.52-1.29$ (m, 6H), 0.95 (t, $J = 7.4$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 147.1, 139.1, 135.8, 133.8, 131.8, 128.7, 127.6, 127.4, 124.1, 123.7, 31.7 (m), 30.3, 29.0, 22.7, 14.2.

LRMS calcd for $[M]^+$ C₁₆H₁₉BrS 324, found: 324.

2-(2-Bromophenyl)-5-chlorothiophene (1c). Following procedure b, from 1,2-dibromobenzene (2.36 g, 10 mmol) and 2-chlorothiophene (1.42 g, 12 mmol), 1c was isolated in 66% (1.80 g) yield as a yellow oil. Eluent pentane.

¹H NMR (400 MHz, CDCl₃): δ 7.67 (dd, J = 8.0, 1.3 Hz, 1H), 7.43 $(dd, J = 7.8, 1.7 Hz, 1H$, 7.33 $(dd, J = 7.6, 1.3 Hz, 1H$, 7.19 (td, $J = 7.7$, 1.7 Hz, 1H), 7.06 (d, $J = 3.8$ Hz, 1H), 6.93 (d, $J =$ 3.8 Hz, 1H).

 13 C NMR (100 MHz, CDCl₃): δ 140.4, 134.6, 133.9, 131.8, 130.7, 129.5, 127.7, 127.2, 126.2, 122.8.

LRMS calcd for $[M]^{\dagger}$ C₁₀H₆BrClS 274, found: 274.

2-(2-Bromo-4-methylphenyl)thiophene (1d). Following procedure a, from 2-bromo-1-iodo-4-methylbenzene (0.891 g, 3 mmol) and 2-thienylboronic acid (0.576 g, 4.5 mmol), 1d was isolated in 62% (0.471 g) yield as a colorless oil. Eluent pentane.

¹H NMR (400 MHz, CDCl₃): δ 7.51 (s, 1H), 7.39-7.34 (m, 2H), 7.27 (dd, $J = 5.6$, 1.2 Hz, 1H), 7.13 (d, $J = 7.7$ Hz, 1H), 7.10 $(dd, J = 5.2, 3.6 Hz, 1H), 2.36 (s, 3H).$

¹³C NMR (100 MHz, CDCl₃): δ 142.0, 139.5, 134.2, 132.5, 131.8, 128.4, 127.6, 127.0, 125.9, 122.7, 20.9.

LRMS calcd for $[M]^+$ C₁₁H₉BrS 252, found: 252.

2-(2-Bromo-4,5-dimethylphenyl)-5-methylthiophene (1e). Following procedure b, from 1,2-dibromo-4,5-dimethylbenzene $(2.64 \text{ g}, 10 \text{ mmol})$ and 2-methylthiophene $(1.18 \text{ g}, 12 \text{ mmol})$, 1e was isolated in 63% (1.77 g) yield as a colorless oil. Eluent pentane.

¹H NMR (400 MHz, CDCl₃): δ 7.47 (s, 1H), 7.27 (s, 1H), 7.11 $(d, J = 3.5 \text{ Hz}, 1\text{H})$, 6.79 $(d, J = 3.5 \text{ Hz}, 1\text{H})$, 2.57 (s, 3H), 2.29 (s, 3H), 2.26 (s, 3H).

 13 C NMR (100 MHz, CDCl₃): δ 140.3, 139.6, 137.8, 136.1, 134.4, 132.8, 132.7, 127.4, 125.3, 119.2, 19.3, 19.2, 15.4.

LRMS calcd for $[M]^+$ C₁₃H₁₃BrS 280, found: 280.

1-(5-(2-Bromo-4-methylphenyl)thiophen-2-yl)ethan-1-one (1f). Following procedure a, from 2-bromo-1-iodo-4-methylbenzene (0.891 g, 3 mmol) and 5-acetyl-2-thienylboronic acid (0.765 g, 4.5 mmol), 1f was isolated in 60% (0.531 g) yield as a yellow oil. Eluent pentane/EtOAc 95/5.

¹H NMR (400 MHz, CDCl₃): δ 7.65 (d, J = 3.5 Hz, 1H), 7.51 $(s, 1H)$, 7.36 $(d, J = 7.8 \text{ Hz}, 1H)$, 7.27 $(d, J = 3.5 \text{ Hz}, 1H)$, 7.15 $(d,$ $J = 7.8$ Hz, 1H), 2.57 (s, 3H), 2.36 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 190.8, 150.4, 144.2, 140.7, 134.5, 132.3, 131.5, 131.4, 128.7, 128.6, 122.3, 26.8, 20.9.

LRMS calcd for $[M]^+$ C₁₃H₁₁BrOS 296, found: 296.

5-(2-Bromo-4-methylphenyl)thiophene-2-carbaldehyde (1g). Following procedure a, from 2-bromo-1-iodo-4-methylbenzene (0.891 g, 3 mmol) and 5-formyl-2-thienylboronic acid (0.702 g, 4.5 mmol), 1g was isolated in 64% (0.539 g) yield as a colorless oil. Eluent pentane/EtOAc 95/5.

 1 H NMR (400 MHz, CDCl₃): δ 9.91 (s, 1H), 7.73 (d, *J* = 3.5 Hz, 1H), 7.52 (s, 1H), 7.39-7.34 (m, 2H), 7.16 (d, $J = 7.8$ Hz, 1H), 2.36 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 183.0, 151.9, 143.6, 141.0, 136.2, 134.5, 131.5, 131.2, 128.8, 128.7, 122.3, 20.9.

LRMS calcd for $[M]^+$ C₁₂H₉BrOS 280, found: 280.

General procedure for the preparation of 2′-aryl-2,3′-biheteroarenes 2b–18b

As a typical experiment, the reaction of the 2-(2-bromoaryl) thiophene derivatives 1a–1g (1 mmol), heteroarene (2 mmol) and KOAc (0.294 g, 3 mmol) at 150 °C for 16 h in DMA (4 mL) in the presence of $Pd(OAc)₂$ (4.5 mg, 0.02 mmol) under argon afforded the coupling product after evaporation of the solvent and purification on silica gel. The a : b ratios were determined by ${}^{1}\mathrm{H}$ NMR and GC/MS analyses of the crude mixtures.

1-(5′-Methyl-2′-phenyl-[2,3′-bithiophen]-5-yl)ethan-1-one (2b). From 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol) and 2-acetylthiophene (0.252 g, 2 mmol), a mixture of 2a and 2b was obtained in a 19 : 81 ratio and 2b was isolated in 64% (0.191 g) yield as a white solid: mp 92-94 \degree C. Eluent pentane, R_f 2a = 0.30, R_f 2b = 0.33.

¹H NMR (400 MHz, CDCl₃): δ 7.46 (d, J = 3.9 Hz, 1H), 7.40–7.32 (m, 5H), 6.91 (s, 1H), 6.79 (d, $J = 3.9$ Hz, 1H), 2.51 (s, 3H), 2.49 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 190.6, 147.5, 142.6, 139.4, 138.7, 133.8, 133.0, 129.8, 129.7, 128.8, 128.4, 127.5, 126.5, 26.7, 15.3.

HRMS calcd for $[M + H]^+$ C₁₇H₁₅OS₂ 299.0559, found: 299.0557.

2-Isobutyl-5-(5-methyl-2-phenylthiophen-3-yl)thiazole (3b). From 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol) and 2-isobutylthiazole (0.282 g, 2 mmol), a mixture of 3a and 3b was obtained in a 20 : 80 ratio and 3b was isolated in 65% (0.203 g) yield as a colorless oil. Eluent pentane, R_f 3a $= 0.20$, R_f 3**b** = 0.23.

¹H NMR (400 MHz, CDCl₃): δ 7.47 (s, 1H), 7.39–7.30 (m, 5H), 6.86 (s, 1H), 2.77 (d, J = 7.6 Hz, 2H), 2.50 (s, 3H), 2.09-1.99 (m, 1H), 0.95 (d, $J = 7.6$ Hz, 6H).

 13 C NMR (100 MHz, CDCl₃): δ 169.5, 139.9, 139.1, 137.5, 133.9, 132.5, 129.9, 128.7, 128.2, 127.6, 127.5, 42.4, 29.8, 22.4, 15.3.

HRMS calcd for $[M + H]^{+}$ C₁₈H₂₀NS₂ 314.1032, found: 314.1029.

1-(5′-Hexyl-2′-phenyl-[2,3′-bithiophen]-5-yl)ethan-1-one (4b). From 2-(2-bromophenyl)-5-hexylthiophene 1b (0.323 g, 1 mmol) and 2-acetylthiophene (0.252 g, 2 mmol), a mixture of 4a and 4b was obtained in a 22 : 78 ratio and 4b was isolated in 57% (0.210 g) yield as a yellow oil. Eluent pentane, R_f 4a = 0.27, R_f 4**b** = 0.30.

¹H NMR (400 MHz, CDCl₃): δ 7.46 (d, J = 4.0 Hz, 1H), 7.41–7.32 (m, 5H), 6.92 (s, 1H), 6.80 (d, $J = 4.0$ Hz, 1H), 2.81 (t, $J = 7.6$ Hz, 2H), 2.49 (s, 3H), 1.72 (quint., $J = 7.6$ Hz, 2H), 1.49–1.39 (m, 2H), 1.39–1.30 (m, 4H), 0.91 (t, $J = 7.6$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 190.6, 147.6, 145.6, 142.5, 138.4, 133.9, 133.0, 129.8, 129.5, 128.8, 128.3, 126.5, 126.3, 31.7, 31.5, 30.1, 28.9, 26.7, 22.7, 14.2.

HRMS calcd for $[M + H]^+$ C₂₂H₂₅OS₂ 369.1341, found: 369.1341.

Cyclopropyl(5′-hexyl-2′-phenyl-[2,3′-bithiophen]-5-yl)methanone (5b). From 2-(2-bromophenyl)-5-hexylthiophene 1b (0.323 g, 1 mmol) and cyclopropyl(thiophen-2-yl)methanone (0.304 g, 2 mmol), a mixture of 5a and 5b was obtained in a $20:80$ ratio and 5b was isolated in 61% (0.240 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, R_f 5a = 0.29, R_f 5b = 0.33. Organic Chemistry Frontiers

¹R NAR (200 MHz, CDCs)², 3 99.1 (a, 11), 7.53 (d, $j = 2$ HBAS column 19:28:28:00 HBA) reduced by the hydrodysis article. Published on 19:28:00 HBA) reduced by the second units are commonly

¹H NMR (400 MHz, CDCl₃): δ 7.59 (d, J = 4.0 Hz, 1H), 7.42–7.32 (m, 5H), 6.93 (s, 1H), 6.83 (d, $J = 4.0$ Hz, 1H), 2.81 (t, $J = 7.6$ Hz, 2H), 2.50–2.39 (m, 1H), 1.72 (quint., $J = 7.6$ Hz, 2H), 1.49–1.38 (m, 2H), 1.38–1.28 (m, 4H), 1.25–1.18 (m, 2H), 1.02–0.95 (m, 2H), 0.91 (t, $J = 7.6$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 192.8, 147.0, 145.5, 143.0, 138.3, 134.0, 131.9, 129.8, 129.6, 128.8, 128.3, 126.6, 126.4, 31.7, 31.6, 30.1, 29.0, 22.7, 17.9, 14.2, 11.3.

HRMS calcd for $[M + H]^+$ C₂₄H₂₇OS₂ 395.1498, found: 395.1503.

5′-Chloro-5-methyl-2′-phenyl-2,3′-bithiophene (6b). From 2- (2-bromophenyl)-5-chlorothiophene 1c (0.273 g, 1 mmol) and 2-methylthiophene $(0.196 \text{ g}, 2 \text{ mmol})$, a mixture of 6a and 6b was obtained in a 9 : 91 ratio and 6b was isolated in 69% (0.201 g) yield as a colorless oil. Eluent pentane, R_f 6a = 0.30, R_f 6**b** = 0.33.

¹H NMR (400 MHz, CDCl₃): δ 7.42–7.31 (m, 5H), 7.01 (s, 1H), 6.66 (d, $J = 3.5$ Hz, 1H), 6.55 (d, $J = 3.5$ Hz, 1H), 2.41 (s,

3H).
¹³C NMR (100 MHz, CDCl₃): δ 139.9, 136.2, 135.0, 133.3, 130.8, 130.0, 128.8, 128.7, 128.5, 128.4, 126.1, 125.5, 15.4.

HRMS calcd for $[M + H]^+$ C₁₅H₁₂ClS₂ 291.0064, found: 291.0064.

2-(5′-Chloro-2′-phenyl-[2,3′-bithiophen]-5-yl)-2-methyl-1,3-dioxolane (7b). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.273 g, 1 mmol) and 2-methyl-2-(thiophen-2-yl)-1,3-dioxolane (0.340 g, 2 mmol), a mixture of 7a and 7b was obtained in a 9:91 ratio and 7**b** was isolated in 71% (0.257 g) yield as a yellow solid: mp 128-130 °C. Eluent pentane/EtOAc 80/20, R_f $7a = 0.23$, $R_f 7b = 0.27$.

¹H NMR (400 MHz, CDCl₃): δ 7.40–7.31 (m, 5H), 7.02 (s, 1H), 6.82 (d, $J = 3.7$ Hz, 1H), 6.67 (d, $J = 3.7$ Hz, 1H), 4.05-3.98 (m, 2H), 3.98–3.91 (m, 2H), 1.74 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 147.0, 137.0, 136.9, 133.1, 130.3, 129.9, 129.0, 128.8, 128.6, 128.4, 125.9, 124.4, 107.2, 65.1, 27.5.

HRMS calcd for $[M + H]^+$ C₁₈H₁₆ClO₂S₂ 363.0275, found: 363.0277.

5-Chloro-2-phenyl-3-(selenophen-2-yl)thiophene (8b). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.273 g, 1 mmol) and selenophene (0.262 g, 2 mmol), a mixture of 8a and 8b was obtained in a 20 : 80 ratio and 11b was isolated in 52% (0.168 g) yield as a colorless oil. Eluent pentane, R_f 8a = 0.29, R_f 8**b** = 0.33.

¹H NMR (400 MHz, CDCl₃): δ 7.85 (dd, J = 5.6, 1.2 Hz, 1H), 7.43–7.36 (m, 5H), 7.15 (dd, $J = 5.6$, 3.8 Hz, 1H), 7.11 (dd, $J =$ 3.8, 1.2 Hz, 1H), 7.06 (s, 1H).

¹³C NMR (100 MHz, CDCl₃): δ 142.7, 136.6, 133.0, 132.8, 131.0, 130.3, 129.6, 128.9, 128.8, 128.5, 128.3.

HRMS calcd for $[M + H]^+$ C₁₄H₁₀ClSSe 324.9351, found: 324.9350.

5-(5-Chloro-2-phenylthiophen-3-yl)-2-isopropyl-4-methylthiazole (9b). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.273 g) , 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), a mixture of 9a and 9b was obtained in a 15 : 85 ratio and 9b was isolated in 59% (0.197 g) yield as a yellow solid: mp 91–93 °C. Eluent pentane/EtOAc 80/20, R_f 9a = 0.26, R_f 9b = 0.30. Open Access Article. Published on 31 January 2023. Downloaded on 4/27/2025 8:28:09 PM. This article is licensed under a [Creative Commons Attribution-NonCommercial 3.0 Unported Licence.](http://creativecommons.org/licenses/by-nc/3.0/) **[View Article Online](https://doi.org/10.1039/d3qo00018d)**

¹H NMR (400 MHz, CDCl₃): δ 7.31-7.15 (m, 5H), 6.89 (s, 1H), 3.23 (sept., $J = 7.6$ Hz, 1H), 2.01 (s, 3H), 1.37 (d, $J = 7.6$ Hz,

6H).
¹³C NMR (100 MHz, CDCl₃): δ 176.8, 149.0, 140.2, 133.2, 130.1, 128.9, 128.8, 128.5, 128.2, 127.3, 123.8, 33.5, 23.3, 15.8.

HRMS calcd for $[M + H]^+$ C₁₇H₁₇ClNS₂ 334.0486, found: 334.0489.

3-(5-Chloro-2-phenylthiophen-3-yl)imidazo[1,2-a]pyrazine (10b). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.273 g, 1 mmol) and imidazo[1,2-a]pyrazine (0.238 g, 2 mmol), a mixture of 10a and 10b was obtained in a 20 : 80 ratio and 10b was isolated in 65% (0.203 g) yield as a white solid: mp 140–142 °C. Eluent pentane/EtOAc 60/40, R_f 10a = 0.21, R_f 10b $= 0.23.$

 1 H NMR (400 MHz, CDCl₃): δ 9.08 (d, J = 1.5 Hz, 1H), 7.78 $(s, 1H)$, 7.64 $(d, J = 4.6 Hz, 1H)$, 7.49 $(dd, J = 4.6, 1.5 Hz, 1H)$, 7.27–7.17 (m, 3H), 7.09 (d, $J = 8.2$ Hz, 2H), 7.03 (s, 1H).

¹³C NMR (100 MHz, CDCl₃): δ 144.2, 141.0, 140.9, 135.7, 132.6, 130.6, 129.7, 129.4, 129.0, 128.9, 127.5, 122.6, 121.2, 116.9.

HRMS calcd for $[M + H]^+$ C₁₆H₁₁ClN₃S 312.0357, found: 312.0354.

4-(5-Chloro-2-phenylthiophen-3-yl)-3,5-dimethylisoxazole (11b). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.273 g, 1 mmol) and 3,5-dimethylisoxazole (0.194 g, 2 mmol), a mixture of 11a and 11b was obtained in a 20 : 80 ratio and 11b was isolated in 35% (0.101 g) yield as a white solid: mp 86–88 °C. Eluent pentane/EtOAc 80/20, R_f 11a = 0.30, R_f 11b = 0.33.

¹H NMR (400 MHz, CDCl₃): δ 7.33-7.27 (m, 3H), 7.23-7.19 (m, 2H), 6.79 (s, 1H), 2.06 (s, 3H), 1.98 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 166.2, 159.2, 140.1, 133.2, 129.2, 129.1, 129.0, 128.2, 128.0, 125.0, 111.2, 11.4, 10.5.

HRMS calcd for $[M + H]^{+}$ C₁₅H₁₃ClNOS 290.0401, found: 290.0402.

2-Butyl-5-(2-(3,4-dimethylphenyl)-5-methylthiophen-3-yl)furan (12b). From 2-(2-bromo-4,5-dimethylphenyl)-5-methylthiophene 1e (0.281 g, 1 mmol) and 2-butylfuran (0.248 g, 2 mmol), a mixture of 12a and 12b was obtained in a 7 : 93 ratio and 12b was isolated in 74% (0.240 g) yield as a yellow oil. Eluent pentane, R_f 12a = 0.14, R_f 12b = 0.17.

¹H NMR (400 MHz, CDCl₃): δ 7.23 (s, 1H), 7.19 (d, J = 7.7 Hz, 1H), 7.11 (d, $J = 7.7$ Hz, 1H), 6.99 (q, $J = 1.1$ Hz, 1H), 6.01 $(d, J = 3.2 \text{ Hz}, 1\text{H})$, 5.88 $(d, J = 3.2 \text{ Hz}, 1\text{H})$, 2.57 $(t, J = 7.6 \text{ Hz},$ 2H), 2.49 (d, J = 1.1 Hz, 3H), 2.29 (s, 3H), 2.27 (s, 3H), 1.57

(quint., $I = 7.6$ Hz, 2H), 1.35 (sext., $I = 7.6$ Hz, 2H), 0.91 (t, $I =$ 7.6 Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 155.0, 149.0, 138.3, 136.5, 136.2, 135.1, 132.3, 130.8, 129.7, 127.6, 127.1, 125.6, 107.0, 106.3, 30.4, 27.9, 22.4, 19.9, 19.7, 15.3, 14.0.

HRMS calcd for $[M + H]^+$ C₂₁H₂₅OS 325.1621, found: 325.1625.

3-(2-(3,4-Dimethylphenyl)-5-methylthiophen-3-yl)imidazo [1,2-a]pyrazine (13b). From 2-(2-bromo-4,5-dimethylphenyl)-5 methylthiophene 1e (0.281 g, 1 mmol) and imidazo[1,2-a]pyrazine (0.238 g, 2 mmol), a mixture of 13a and 13b was obtained in a 19:81 ratio and 13b was isolated in 59% (0.188 g) yield as a colorless oil. Eluent pentane/EtOAc 60/40, R_f 13a = 0.20, R_f $13b = 0.23$.

¹H NMR (400 MHz, CDCl₃): δ 9.08 (s, 1H), 7.78 (s, 1H), 7.63 $(d, J = 4.6 \text{ Hz}, 1H), 7.46-7.43 \text{ (m, 2H)}, 7.21 \text{ (s, 1H)}, 6.41-6.35 \text{ m}$ (m, 2H), 2.38 (s, 3H), 2.33 (s, 3H), 2.29 (s, 3H).

 13 C NMR (100 MHz, CDCl₃): δ 143.9, 140.9, 140.8, 139.1, 139.0, 136.7, 135.6, 133.6, 132.6, 131.4, 129.3, 126.5, 125.8, 125.5, 123.3, 116.9, 19.8, 19.5, 15.3.

HRMS calcd for $[M + H]^+$ C₁₉H₁₈N₃S 320.1216, found: 320.1214.

1-(5-Methyl-2′-(p-tolyl)-[2,3′-bithiophen]-5′-yl)ethan-1-one (14b). From 1-(5-(2-bromo-4-methylphenyl)thiophen-2-yl)ethan-1-one 1f $(0.295 \text{ g}, 1 \text{ mmol})$ and 2-methylthiophene $(0.196 \text{ g},$ 2 mmol), a mixture of 14a and 14b was obtained in an 8 : 92 ratio and 14b was isolated in 78% (0.243 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, R_f 14a = 0.14, R_f 14b = 0.17.

¹H NMR (400 MHz, CDCl₃): δ 7.71 (s, 1H), 7.32 (d, J = 8.0 Hz, 2H), 7.17 (d, $J = 8.0$ Hz, 2H), 6.71 (d, $J = 3.5$ Hz, 1H), 6.59 (d, $J = 3.5$ Hz, 1H), 2.57 (s, 3H), 2.73 (s, 3H), 2.38 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 190.6, 146.8, 141.7, 140.1, 139.1, 134.8, 134.6, 132.2, 130.3, 129.5, 129.4, 126.3, 125.5, 26.6, 21.4, 15.3.

HRMS calcd for $[M + H]^+$ C₁₈H₁₇OS₂ 313.0715, found: 313.0715.

1-(4-(5-Butylfuran-2-yl)-5-(p-tolyl)thiophen-2-yl)ethan-1-one (15b). From 1-(5-(2-bromo-4-methylphenyl)thiophen-2-yl) ethan-1-one 1f $(0.295 \text{ g}, 1 \text{ mmol})$ and 2-butylfuran $(0.248 \text{ g},$ 2 mmol), a mixture of 15a and 15b was obtained in a 6 : 94 ratio and 15b was isolated in 81% (0.274 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, R_f 15a = 0.33, R_f 15b = 0.37.

¹H NMR (400 MHz, CDCl₃): δ 7.88 (s, 1H), 7.36 (d, J = 8.0 Hz, 2H), 7.21 (d, $J = 8.0$ Hz, 2H), 6.03 (d, $J = 3.5$ Hz, 1H), 5.91 (d, $J = 3.5$ Hz, 1H), 2.64–2.55 (m, 5H), 2.40 (s, 3H), 1.58 (quint., $J = 7.6$ Hz, 2H), 1.35 (sext., $J = 7.6$ Hz, 2H), 0.92 (t, $J =$ 7.6 Hz, 3H).

 13 C NMR (100 MHz, CDCl₃): δ 190.8, 155.9, 147.4, 145.4, 142.3, 139.1, 132.3, 130.7, 129.5, 129.4, 129.3, 108.2, 106.6, 30.3, 27.8, 26.7, 22.4, 21.5, 13.9.

LRMS calcd for $[M]^{+}$ C₂₁H₂₂O₂S 338, found: 338.

HRMS calcd for $[M + H]^{+}$ C₂₁H₂₃O₂S 339.1413, found: 339.1414.

1-(4-(2-Isobutylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl)ethan-1 one (16b). From 1-(5-(2-bromo-4-methylphenyl)thiophen-2-yl) ethan-1-one 1f (0.295 g, 1 mmol) and 2-isobutylthiazole (0.282 g, 2 mmol), a mixture of 16a and 16b was obtained in a $12:88$ ratio and 16b was isolated in 58% (0.256 g) yield as a yellow solid: mp 87–89 °C. Eluent pentane/EtOAc 80/20, R_f 16a $= 0.20$, R_f **16b** = 0.23.

¹H NMR (400 MHz, CDCl₃): δ 7.71 (s, 1H), 7.51 (s, 1H), 7.27 $(d, J = 8.0$ Hz, 2H), 7.17 $(d, J = 8.0$ Hz, 2H), 2.80 $(d, J = 7.6$ Hz, 2H), 2.57 (s, 3H), 2.38 (s, 3H), 2.10–2.02 (m, 1H), 0.96 (d, $J =$ 7.6 Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 190.5, 170.4, 148.4, 142.3, 140.6, 139.6, 134.2, 131.1, 129.7, 129.6, 129.4, 128.7, 42.4, 29.8, 26.7, 22.3, 21.5.

HRMS calcd for $[M + H]^+$ C₂₀H₂₂NOS₂ 356.1137, found: 356.1136.

5-Methyl-2′-(p-tolyl)-[2,3′-bithiophene]-5′-carbaldehyde (17b). From 5-(2-bromo-4-methylphenyl)thiophene-2-carbaldehyde 1g $(0.281 \text{ g}, 1 \text{ mmol})$ and 2-methylthiophene $(0.196 \text{ g}, 2 \text{ mmol})$, a mixture of 17a and 17b was obtained in a 5 : 95 ratio and 17b was isolated in 68% (0.203 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, R_f 17^a = 0.23, R_f 17**b** = 0.27.

¹H NMR (400 MHz, CDCl₃): δ 9.88 (s, 1H), 7.79 (s, 1H), 7.34 $(d, J = 8.0 \text{ Hz}, 2\text{H}), 7.18 (d, J = 8.0 \text{ Hz}, 2\text{H}), 6.72 (d, J = 3.5 \text{ Hz},$ 1H), 6.60 (d, $J = 3.5$ Hz, 1H), 2.44 (s, 3H), 2.40 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 182.9, 148.2, 141.2, 140.4, 139.5, 138.4, 134.4, 132.6, 130.2, 129.6, 129.4, 126.5, 125.6, 21.5, 15.4.

HRMS calcd for $[M + H]^{+}$ C₁₇H₁₅OS₂ 299.0559, found: 299.0557.

4-(2-Isobutylthiazol-5-yl)-5-(p-tolyl)thiophene-2-carbaldehyde (18b). From 5-(2-bromo-4-methylphenyl)thiophene-2-carbaldehyde 1g (0.281 g, 1 mmol) and 2-isobutylthiazole (0.282 g, 2 mmol), a mixture of 18a and 18b was obtained in a 13 : 87 ratio and 18b was isolated in 57% (0.194 g) yield as a yellow solid: mp 90–92 °C. Eluent pentane/EtOAc 80/20, R_f 18a = 0.10, R_f 18**b** = 0.12. Organic Chemistry Frontiers

(1928 g. 2 mmol) at mixing of 16a and 16b was obtained in 3²% 14d as a 92% 140 yield as sychov 50.1 Article is licensed under the sychology and commons Commons Access Article is licensed und

¹H NMR (400 MHz, CDCl₃): δ 9.90 (s, 1H), 7.80 (s, 1H), 7.53 $(s, 1H)$, 7.28 $(d, J = 8.0$ Hz, 2H), 7.19 $(d, J = 8.0$ Hz, 2H), 2.80 $(d,$ $J = 7.6$ Hz, 2H), 2.39 (s, 3H), 2.10-2.02 (m, 1H), 0.97 (d, $J =$ 7.6 Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 182.7, 170.6, 149.8, 141.6, 140.8, 140.0, 138.0, 130.8, 129.7, 129.5, 129.4, 129.1, 42.4, 29.8, 22.3, 21.5.

HRMS calcd for $[M + H]^+$ C₁₉H₂₀NOS₂ 342.0981, found: 342.0981.

General procedure for the preparation of 2,5-diarylthiophenes 19–23

As a typical experiment, the reaction of 2-(2-bromo-4-methylphenyl)thiophene 1d (0.253 g, 1 mmol), aryl bromide (1.3 mmol) and KOAc (0.196 g, 2 mmol) at 150 °C for 16 h in DMA (4 mL) in the presence of Pd $(OAc)_2$ $(4.5 \text{ mg}, 0.02 \text{ mmol})$ under argon affords the coupling product after evaporation of the solvent and purification on silica gel.

4-(5-(2-Bromophenyl)thiophen-2-yl)benzonitrile (19). From 2-(2-bromo-4-methylphenyl)thiophene 1d (0.253 g, 1 mmol) and 4-bromobenzonitrile (0.237 g, 1.3 mmol), 19 was isolated

in 85% (0.301 g) yield as a yellow oil. Eluent pentane/EtOAc 95/5.

¹H NMR (400 MHz, CDCl₃): δ 7.68 (d, J = 8.3 Hz, 2H), 7.62 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.52 \text{ (s, 1H)}, 7.40-7.36 \text{ (m, 2H)}, 7.28 \text{ (d, } J =$ 3.7 Hz, 1H), 7.15 $(d, J = 8.1$ Hz, 1H), 2.36 $(s, 3H)$.

¹³C NMR (100 MHz, CDCl₃): δ 143.6, 142.1, 140.0, 138.5, 134.4, 132.8, 131.6, 131.4, 129.0, 128.5, 125.9, 125.1, 122.3, 118.9, 110.6, 20.8.

LRMS calcd for $[M]^+$ C₁₈H₁₂BrNS 353, found: 353.

2-(2-Bromo-4-methylphenyl)-5-(4-chlorophenyl)thiophene (20). From 2-(2-bromo-4-methylphenyl)thiophene 1d (0.253 g, 1 mmol) and 1-bromo-4-chlorobenzene (0.249 g, 1.3 mmol), 20 was isolated in 74% (0.269 g) yield as a yellow oil. Eluent pentane/EtOAc 95/5.

¹H NMR (400 MHz, CDCl₃): δ 7.55 (d, J = 8.3 Hz, 2H), 7.53 $(s, 1H)$, 7.40 $(d, J = 7.9$ Hz, 1H), 7.35 $(d, J = 8.3$ Hz, 2H), 7.29–7.23 (m, 2H), 7.15 (d, $J = 8.1$ Hz, 1H), 2.38 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 143.4, 141.7, 139.6, 134.3, 133.4, 132.9, 132.1, 131.5, 129.2, 128.7, 128.5, 127.0, 123.4, 122.4, 20.8.

LRMS calcd for $[M]^+$ C₁₇H₁₂BrClS 364, found: 364.

1-(4-(5-(2-Bromo-4-methylphenyl)thiophen-2-yl)phenyl)propan-1-one (21). From 2-(2-bromo-4-methylphenyl)thiophene 1d $(0.253 \text{ g}, 1 \text{ mmol})$ and 4-bromopropiophenone $(0.277 \text{ g},$ 1.3 mmol), 21 was isolated in 82% (0.316 g) yield as a colorless oil. Eluent pentane/EtOAc 95/5.

¹H NMR (400 MHz, CDCl₃): δ 7.98 (d, J = 8.3 Hz, 2H), 7.70 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.52 \text{ (s, 1H)}, 7.43-7.38 \text{ (m, 2H)}, 7.28 \text{ (d, } J =$ 3.7 Hz, 1H), 7.15 (d, $J = 8.1$ Hz, 1H), 3.00 (q, $J = 7.6$ Hz, 2H), 2.37 (s, 3H), 1.25 (t, $J = 7.6$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 200.1, 143.3, 143.0, 139.9, 138.6, 135.7, 134.4, 132.0, 131.5, 129.0, 128.9, 128.5, 125.6, 124.5, 122.4, 31.9, 20.9, 8.4.

LRMS calcd for $[M]^+$ C₂₀H₁₇BrOS 384, found: 384.

3-(5-(2-Bromo-4-methylphenyl)thiophen-2-yl)benzonitrile (22). From 2-(2-bromo-4-methylphenyl)thiophene 1d (0.253 g, 1 mmol) and 3-bromobenzonitrile (0.237 g, 1.3 mmol), 22 was isolated in 84% (0.297 g) yield as a yellow oil. Eluent pentane/ EtOAc 95/5.

 1 H NMR (400 MHz, CDCl₃): δ 7.89 (s, 1H), 7.82 (d, J = 8.3 Hz, 1H), 7.57-7.51 (m, 2H), 7.48 (t, $J = 7.8$ Hz, 1H), 7.39 (d, $J = 8.3$ Hz, 1H), 7.33 (d, $J = 3.7$ Hz, 1H), 7.27 (d, $J = 3.7$ Hz, 1H), 7.16 (d, $J = 8.1$ Hz, 1H), 2.37 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 143.0, 141.8, 140.0, 135.7, 134.4, 131.8, 131.5, 130.7, 129.9, 129.8, 129.1, 128.9, 128.5, 124.4, 122.4, 118.7, 113.3, 20.9.

LRMS calcd for $[M]^+$ C₁₈H₁₂BrNS 353, found: 353.

2-(5-(2-Bromo-4-methylphenyl)thiophen-2-yl)benzonitrile (23). From 2-(2-bromo-4-methylphenyl)thiophene 1d (0.253 g, 1 mmol) and 2-bromobenzonitrile (0.237 g, 1.3 mmol), 23 was isolated in 86% (0.304 g) yield as a yellow oil. Eluent pentane/ EtOAc 95/5.

¹H NMR (400 MHz, CDCl₃): δ 7.74 (d, J = 8.3 Hz, 1H), 7.68–7.63 (m, 2H), 7.59 (t, $J = 7.8$ Hz, 1H), 7.52 (s, 1H), 7.41 (d, $J = 8.3$ Hz, 1H), 7.37 (t, $J = 7.8$ Hz, 1H), 7.31 (d, $J = 3.7$ Hz, 1H), 7.15 $(d, J = 8.1 \text{ Hz}, 1H), 2.37 \text{ (s, 3H)}.$

Research Article Organic Chemistry Frontiers

¹³C NMR (100 MHz, CDCl₃): δ 143.9, 139.9, 139.7, 137.4, 134.5, 134.4, 133.1, 131.7, 131.6, 129.5, 128.7, 128.5, 127.6, 127.5, 122.3, 119.0, 109.8, 20.8.

LRMS calcd for $[M]^{\dagger}$ C₁₈H₁₂BrNS 353, found: 353.

General procedure for the preparation of heteroarylated 2,3 diarylthiophenes 24b–32b

As a typical experiment, the reaction of the 2-(2-bromophenyl)- 5-arylthiophene derivative 19–23 (1 mmol), heteroarene (2 mmol) and KOAc (0.294 g, 3 mmol) at 150 \degree C for 16 h in DMA (4 mL) in the presence of Pd $(OAc)_2$ $(4.5 \text{ mg}, 0.02 \text{ mmol})$ under argon affords the coupling product after evaporation of the solvent and purification on silica gel. Research Article

¹C EMR (100 MHz, CDCs): δ 1353, 1395, 1387, 1374, 44{4D-Repress¹2-mentionfilm (201). From $\frac{1}{2}$ (128, 128, 1314, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 1315, 13

4-(5-Methyl-2′-(p-tolyl)-[2,3′-bithiophen]-5′-yl)benzonitrile (24b). From 4-(5-(2-bromophenyl)thiophen-2-yl)benzonitrile 19 (0.340 g, 1 mmol) and 2-methylthiophene (0.196 g, 2 mmol), 24b was isolated in 71% (0.263 g) yield as a yellow solid: mp 144–146 °C. Eluent pentane/EtOAc 80/20, $R_f = 0.30$.

¹H NMR (400 MHz, CDCl₃): δ 7.69 (d, J = 8.3 Hz, 2H), 7.66 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.45 \text{ (s, 1H)}, 7.35 \text{ (d, } J = 8.1 \text{ Hz}, 2\text{H}), 7.18 \text{ (d, }$ $J = 8.1$ Hz, 2H), 6.74 (d, $J = 3.5$ Hz, 1H), 6.60 (d, $J = 3.5$ Hz, 1H), 2.45 (s, 3H), 2.40 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 139.9, 139.8, 139.7, 138.6, 138.4, 135.4, 132.9, 132.5, 130.6, 129.6, 129.5, 127.6, 126.1, 125.8, 125.6, 119.0, 110.8, 21.5, 15.4.

>HRMS calcd for $[M + H]^+$ C₂₃H₁₈NS₂ 372.0875, found: 372.0874.

4-(4-(5-Butylfuran-2-yl)-5-(p-tolyl)thiophen-2-yl)benzonitrile (25b). From 4-(5-(2-bromophenyl)thiophen-2-yl)benzonitrile 19 $(0.340 \text{ g}, 1 \text{ mmol})$ and 2-butylfuran $(0.248 \text{ g}, 2 \text{ mmol})$, $25b$ was isolated in 67% (0.266 g) yield as a yellow solid: mp 82–84 \degree C. Eluent pentane/EtOAc 80/20, $R_f = 0.27$.

¹H NMR (400 MHz, CDCl₃): δ 7.70 (d, J = 8.3 Hz, 2H), 7.65 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.63 \text{ (s, 1H)}, 7.40 \text{ (d, } J = 8.1 \text{ Hz}, 2\text{H}), 7.22 \text{ (d, }$ $J = 8.1$ Hz, 2H), 6.08 (d, $J = 3.2$ Hz, 1H), 5.93 (d, $J = 3.2$ Hz, 1H), 2.62 (t, $J = 7.6$ Hz, 2H), 2.42 (s, 3H), 1.59 (quint., $J = 7.6$ Hz, 2H), 1.37 (sext., $J = 7.6$ Hz, 2H), 0.94 (t, $J = 7.6$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 155.7, 148.0, 140.0, 138.7, 138.5, 138.4, 132.8, 131.0, 129.6, 129.4, 129.3, 125.8, 125.3, 119.0, 110.7, 107.9, 106.6, 30.3, 27.9, 22.4, 21.4, 13.9.

HRMS calcd for $[M + H]^{+}$ C₂₆H₂₄NOS 398.1573, found: 398.1574.

4-(4-(Selenophen-2-yl)-5-(p-tolyl)thiophen-2-yl)benzonitrile (26b). From 4-(5-(2-bromophenyl)thiophen-2-yl)benzonitrile 19 (0.340 g, 1 mmol) and selenophene (0.262 g, 2 mmol), 26b was isolated in 58% (0.235 g) yield as a white solid: mp 162–164 °C. Eluent pentane/EtOAc 80/20, $R_f = 0.27$.

¹H NMR (400 MHz, CDCl₃): δ 7.90 (dd, J = 4.8, 1.8 Hz, 1H), 7.71 (d, $J = 8.3$ Hz, 2H), 7.67 (d, $J = 8.3$ Hz, 2H), 7.50 (s, 1H), 7.35 (d, $J = 8.1$ Hz, 2H), 7.22–7.16 (m, 4H), 2.40 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 143.3, 140.3, 139.7, 138.9, 138.3, 134.5, 132.9, 130.9, 130.3, 129.9, 129.7, 129.6, 128.5, 127.6, 125.9, 118.9, 110.9, 21.5.

HRMS calcd for $[M + H]^+$ C₂₂H₁₆NSSe 406.0163, found: 406.0164.

4-(4-(2-Isopropyl-4-methylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl) benzonitrile (27b). From 4-(5-(2-bromophenyl)thiophen-2-yl) benzonitrile 19 (0.340 g, 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), 27**b** was isolated in 63% (0.261 g) yield as a white solid: mp 144–146 °C. Eluent pentane/EtOAc $80/20$, $R_f = 0.17$.

¹H NMR (400 MHz, CDCl₃): δ 7.70 (d, J = 8.3 Hz, 2H), 7.66 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.35 \text{ (s, 1H)}, 7.22 \text{ (d, } J = 8.1 \text{ Hz}, 2\text{H}), 7.12 \text{ (d, }$ $J = 8.1$ Hz, 2H), 3.26 (sept., $J = 7.6$ Hz, 1H), 2.35 (s, 3H), 2.04 (s, 3H), 1.40 $(d, J = 7.6$ Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 176.7, 148.9, 143.4, 139.6, 138.4, 138.2, 132.9, 130.6, 129.7, 129.3, 128.9, 128.2, 125.8, 124.2, 118.9, 110.9, 33.5, 23.3, 21.4, 15.9.

HRMS calcd for $[M + H]^+$ C₂₅H₂₃N₂S₂ 415.1297, found: 415.1296.

4-(4-(2-Isobutylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl)benzonitrile (28b). From 4-(5-(2-bromophenyl)thiophen-2-yl)benzonitrile 19 (0.340 g, 1 mmol) and 2-isobutylthiazole (0.282 g, 2 mmol), $28b$ was isolated in 66% (0.274 g) yield as a yellow solid: mp 140–142 °C. Eluent pentane/EtOAc 80/20, $R_f = 0.23$.

¹H NMR (400 MHz, CDCl₃): δ 7.71 (d, J = 8.3 Hz, 2H), 7.67 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.55 \text{ (s, 1H)}, 7.47 \text{ (s, 1H)}, 7.30 \text{ (d, } J = 8.1 \text{ Hz},$ 2H), 7.19 (d, $J = 8.1$ Hz, 2H), 2.80 (d, $J = 7.6$ Hz, 2H), 2.40 (s, 3H), 2.11–2.03 (m, 1H), 0.97 (d, $J = 7.6$ Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 170.2, 141.7, 140.4, 140.3, 139.1, 138.1, 133.0, 131.6, 130.0, 129.6, 129.5, 129.0, 127.3, 125.9, 118.9, 111.1, 42.5, 29.9, 22.4, 21.5.

HRMS calcd for $[M + H]^+$ C₂₅H₂₃N₂S₂ 415.1297, found: 415.1295.

5-(5-(4-Chlorophenyl)-2-(p-tolyl)thiophen-3-yl)-2-isopropyl-4 methylthiazole (29b). From 2-(2-bromo-4-methylphenyl)-5-(4 chlorophenyl)thiophene 20 (0.350 g, 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), 29b was isolated in 74% (0.314 g) yield as a yellow solid: mp 90-92 °C. Eluent pentane/ EtOAc 80/20, $R_f = 0.20$.

¹H NMR (400 MHz, CDCl₃): δ 7.54 (d, J = 8.3 Hz, 2H), 7.36 $(d, J = 8.3 \text{ Hz}, 2\text{H}), 7.22 \text{ (s, 1H)}, 7.21 \text{ (d, } J = 8.1 \text{ Hz}, 2\text{H}), 7.10 \text{ (d, }$ $J = 8.1$ Hz, 2H), 3.26 (sept., $J = 7.6$ Hz, 1H), 2.34 (s, 3H), 2.04 (s, $3H$, 1.40 (d, $J = 7.6$ Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 176.5, 148.7, 141.5, 140.8, 138.0, 133.7, 132.5, 131.0, 129.6, 129.3, 128.4, 128.2, 127.6, 126.9, 124.7, 33.5, 23.3, 21.4, 15.9.

HRMS calcd for $[M + H]^+$ C₂₄H₂₃ClNS₂ 424.0955, found: 424.0952.

1-(4-(4-(2-Isopropyl-4-methylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl)phenyl)propan-1-one (30b). From 1-(4-(5-(2-bromo-4 methylphenyl)thiophen-2-yl)phenyl)propan-1-one 21 (0.371 g, 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), 30b was isolated in 73% (0.325 g) yield as a yellow solid: mp 120–122 °C. Eluent pentane/EtOAc 80/20, $R_f = 0.27$.

¹H NMR (400 MHz, CDCl₃): δ 7.99 (d, J = 8.3 Hz, 2H), 7.70 $(d, J = 8.3 \text{ Hz}, 2H), 7.36 \text{ (s, 1H)}, 7.23 \text{ (d, } J = 8.1 \text{ Hz}, 2H), 7.12 \text{ (d, }$ $J = 8.1$ Hz, 2H), 3.27 (sept., $J = 7.6$ Hz, 1H), 3.02 (q, $J = 7.6$ Hz, 2H), 2.35 (s, 3H), 2.05 (s, 3H), 1.40 (d, $J = 7.6$ Hz, 6H), 1.25 (t, J $= 7.6$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 200.1, 176.6, 148.9, 142.7, 140.6, 138.2, 138.1, 135.9, 130.9, 129.6, 129.0, 128.7, 128.3, 125.5, 124.5, 33.6, 31.9, 23.3, 21.4, 15.9, 8.5.

HRMS calcd for $[M + H]^{+}$ C₂₇H₂₈NOS₂ 446.1607, found: 446.1608.

3-(4-(2-Isopropyl-4-methylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl) benzonitrile (31b). From 3-(5-(2-bromophenyl)thiophen-2-yl) benzonitrile 22 (0.340 g, 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), 31b was isolated in 73% (0.302 g) yield as a yellow solid: mp 128–130 °C. Eluent pentane/EtOAc $80/20$, $R_f = 0.23$. Open Access Article. Published on 31 January 2023. Downloaded on 4/27/2025 8:28:09 PM. This article is licensed under a [Creative Commons Attribution-NonCommercial 3.0 Unported Licence.](http://creativecommons.org/licenses/by-nc/3.0/) **[View Article Online](https://doi.org/10.1039/d3qo00018d)**

¹H NMR (400 MHz, CDCl₃): δ 7.88 (s, 1H), 7.82 (d, J = 8.3 Hz, 1H), 7.57 (d, $J = 8.3$ Hz, 1H), 7.50 (t, $J = 7.8$ Hz, 1H), 7.30 (s, 1H), 7.21 (d, $J = 8.1$ Hz, 2H), 7.12 (d, $J = 8.1$ Hz, 2H), 3.27 (sept., $J = 7.6$ Hz, 1H), 2.35 (s, 3H), 2.04 (s, 3H), 1.40 (d, $J =$ 7.6 Hz, 6H).

 13 C NMR (100 MHz, CDCl₃): δ 176.7, 148.9, 142.7, 139.2, 138.3, 135.3, 130.9, 130.6, 130.0, 129.7, 129.6, 129.0, 128.7, 128.6, 128.3, 124.3, 118.6, 113.5, 33.5, 23.3, 21.4, 15.9.

HRMS calcd for $[M + H]^+$ C₂₅H₂₃N₂S₂ 415.1297, found: 415.1296.

2-(4-(2-Isopropyl-4-methylthiazol-5-yl)-5-(p-tolyl)thiophen-2-yl) benzonitrile (32b). From 2-(5-(2-bromophenyl)thiophen-2-yl) benzonitrile 23 (0.340 g, 1 mmol) and 2-isopropyl-4-methylthiazole (0.282 g, 2 mmol), $32b$ was isolated in 76% (0.315 g) yield as a white solid: mp 130–132 °C. Eluent pentane/EtOAc 80/20, $R_f = 0.30$.

¹H NMR (400 MHz, CDCl₃): δ 7.75 (d, J = 8.3 Hz, 1H), 7.66 $(d, J = 8.3 \text{ Hz}, 1\text{H})$, 7.61 $(t, J = 7.8 \text{ Hz}, 1\text{H})$, 7.39 $(t, J = 7.8 \text{ Hz},$ 1H), 7.59 (s, 1H), 7.23 (d, $J = 8.1$ Hz, 2H), 7.12 (d, $J = 8.1$ Hz, 2H), 3.26 (sept., $J = 7.6$ Hz, 1H), 2.34 (s, 3H), 2.12 (s, 3H), 1.39 $(d, J = 7.6 \text{ Hz}, 6H)$.

¹³C NMR (100 MHz, CDCl₃): δ 176.6, 149.0, 143.6, 138.3, 137.2, 137.1, 134.6, 133.2, 131.5, 130.5, 129.6, 129.4, 128.6, 128.5, 127.8, 124.2, 118.9, 109.9, 33.5, 23.3, 21.4, 15.9.

HRMS calcd for $[M + H]^{+}$ C₂₅H₂₃N₂S₂ 415.1297, found: 415.1302.

2-Isopropyl-4-methyl-5-(5-(pyridin-3-yl)-2-(p-tolyl)thiophen-3 yl)thiazole (33b). The reaction of 2-(2-bromo-4-methylphenyl) thiophene 1d (0.253 g, 1 mmol), 3-bromopyridine (0.174 g, 1.1 mmol) and KOAc (0.196 g, 2 mmol) at 150 °C for 16 h in DMA (4 mL) in the presence of $Pd(OAc)₂ (4.5 mg, 0.02 mmol)$ under argon affords the corresponding 2,5-diarylated product after evaporation of the solvent and filtration on a silica plug. Then, the reaction of this crude mixture, 2-isopropyl-4-methylthiazole $(0.282 \text{ g}, 2 \text{ mmol})$, and KOAc $(0.294 \text{ g}, 3 \text{ mmol})$ at 150 °C for 16 h in DMA (4 mL) in the presence of $Pd(OAc)₂$ (4.5 mg, 0.02 mmol) under argon affords the coupling product 33b after evaporation of the solvent and purification on silica gel in 58% (0.226 g) yield as a colorless oil. Eluent pentane/ EtOAc $60/40$, $R_f = 0.23$.

¹H NMR (400 MHz, CDCl₃): δ 8.90 (d, J = 2.4 Hz, 1H), 7.53 $(dd, J = 4.8, 1.6 Hz, 1H$, 7.88 $(dt, J = 8.0, 1.5 Hz, 1H)$, 7.33 $(ddd, J = 8.0, 4.8, 0.8 Hz, 1H$, 7.30 (s, 1H), 7.22 (d, $J = 8.1 Hz$, 2H), 7.12 (d, $J = 8.1$ Hz, 2H), 3.26 (sept., $J = 7.6$ Hz, 1H), 2.35 (s, 3H), 2.05 (s, 3H), 1.40 (d, $J = 7.6$ Hz, 6H).

¹³C NMR (100 MHz, CDCl₃): δ 176.6, 148.8, 146.8, 142.4, 138.2, 138.1, 132.8, 130.8, 130.1, 129.6, 128.6, 128.4, 128.3, 124.5, 123.8, 33.5, 23.3, 21.4, 15.9.

HRMS calcd for $[M + H]^+$ C₂₃H₂₃N₂S₂ 391.1297, found: 391.1298.

General procedure for the preparation of 1,2-di(heteroaryl) benzenes 2a, 11a and 34a–36a

As a typical experiment, the reaction of the 2-(2-bromophenyl) thiophene derivative 1 (1 mmol), heteroarylboronic acid (2 mmol) and K_2CO_3 (0.276 g, 2 mmol) at 150 °C for 16 h in DMA (4 mL) in the presence of Pd $(OAc)_2$ (11.3 mg, 0.05 mmol) under argon affords the coupling product after evaporation of the solvent and purification on silica gel.

1-(5-(2-(5-Methylthiophen-2-yl)phenyl)thiophen-2-yl)ethan-1 one (2a). From 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol) and (5-acetylthiophen-2-yl)boronic acid (0.340 g, 2 mmol), 2a was obtained in 61% (0.182 g) yield as a yellow oil. Eluent pentane, $R_f = 0.30$.

¹H NMR (400 MHz, CDCl₃): δ 7.54 (d, J = 3.9 Hz, 1H), 7.50–7.46 (m, 2H), 7.42–7.32 (m, 2H), 6.88 (d, $J = 3.9$ Hz, 1H), 6.64 (d, $J = 3.5$ Hz, 1H), 6.61 (d, $J = 3.5$ Hz, 1H), 2.53 (s, 3H), 2.45 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 190.8, 152.0, 144.2, 141.1, 139.6, 134.4, 132.8, 132.6, 131.4, 131.0, 129.0, 128.3, 127.9, 127.4, 125.6, 26.8, 15.5.

HRMS calcd for $[M + H]^+$ C₁₇H₁₅OS₂ 299.0559, found: 299.0558.

2-Methyl-5-(2-(thiophen-2-yl)phenyl)thiophene (34a). From 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol) and 2-thienylboronic acid (0.256 g, 2 mmol), 34a was obtained in 80% (0.205 g) yield as a colorless oil. Eluent pentane, $R_f =$ 0.33.

¹H NMR (400 MHz, CDCl₃): δ 7.54-7.45 (m, 2H), 7.39-7.31 $(m, 2H)$, 7.28 $(dd, J = 5.1, 1.2 Hz, 1H)$, 6.98 $(dd, J = 5.1, 3.5 Hz$, 1H), 6.93 (dd, $J = 3.5$, 1.2 Hz, 1H), 6.67 (d, $J = 3.8$ Hz, 1H), 6.61 $(d, J = 3.8 \text{ Hz}, 1\text{H}), 2.46 \text{ (s, 3H)}.$

¹³C NMR (100 MHz, CDCl₃): δ 143.0, 140.7, 140.4, 134.4, 133.7, 131.2, 131.0, 128.0, 127.7, 127.2, 127.1, 127.0, 126.0, 125.3, 15.4.

HRMS calcd for $[M + H]^+$ C₁₅H₁₃S₂ 257.0453, found: 257.0453.

5-(2-(5-Methylthiophen-2-yl)phenyl)thiophene-2-carbaldehyde (35a). From 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol) and (5-formylthiophen-2-yl)boronic acid (0.312 g, 2 mmol), 35a was obtained in 41% (0.116 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, $R_f = 0.23$.

¹H NMR (400 MHz, CDCl₃): δ 9.86 (s, 1H), 7.62 (d, J = 3.9 Hz, 1H), 7.52-7.47 (m, 2H), 7.44-7.34 (m, 2H), 6.98 (d, $J =$ 3.9 Hz, 1H), 6.64 (d, $J = 3.5$ Hz, 1H), 6.61 (d, $J = 3.5$ Hz, 1H), 2.45 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 183.1, 153.6, 143.7, 141.3, 139.4, 136.4, 134.5, 132.5, 131.5, 130.9, 129.3, 128.5, 128.0, 127.5, 125.6, 15.4.

HRMS calcd for $[M + H]^{+}$ C₁₆H₁₃OS₂ 285.0402, found: 285.0400.

2-Chloro-5-(2-(thiophen-2-yl)phenyl)thiophene (36a). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.271 g, 1 mmol) and 2-thienylboronic acid (0.256 g, 2 mmol), 36a was obtained in 91% (0.252 g) yield as a yellow oil. Eluent pentane, $R_f = 0.33$.

¹H NMR (400 MHz, CDCl₃): δ 7.53-7.43 (m, 2H), 7.40-7.33 $(m, 2H)$, 7.31 (dd, $J = 5.1$, 1.2 Hz, 1H), 7.00 (dd, $J = 5.1$, 3.5 Hz, 1H), 6.93 (dd, $J = 3.5$, 1.2 Hz, 1H), 6.77 (d, $J = 3.8$ Hz, 1H), 6.68 $(d, J = 3.8 \text{ Hz}, 1H)$.

¹³C NMR (100 MHz, CDCl₃): δ 142.2, 141.5, 133.8, 133.2, 131.4, 130.9, 130.4, 128.4, 128.2, 127.5, 127.2, 126.4, 126.3, 126.2.

HRMS calcd for $[M + H]^+$ C₁₄H₁₀ClS₂ 276.9907, found: 276.9909.

4-(2-(5-Chlorothiophen-2-yl)phenyl)-3,5-dimethylisoxazole (11a). From 2-(2-bromophenyl)-5-chlorothiophene 1c (0.271 g, 1 mmol) and (3,5-dimethylisoxazol-4-yl)boronic acid (0.282 g, 2 mmol), 11a was obtained in 58% (0.168 g) yield as a yellow oil. Eluent pentane/EtOAc 80/20, $R_f = 0.30$.

¹H NMR (400 MHz, CDCl₃): δ 7.56 (d, J = 7.8 Hz, 1H), 7.43 $(t, J = 7.6 \text{ Hz}, 1\text{H}), 7.37 (t, J = 7.6 \text{ Hz}, 1\text{H}), 7.20 (d, J = 7.8 \text{ Hz},$ 1H), 6.78 (d, $J = 3.8$ Hz, 1H), 6.69 (d, $J = 3.8$ Hz, 1H), 2.18 (s, 3H), 1.95 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 166.3, 159.5, 140.9, 134.3, 132.1, 130.8, 129.8, 129.0, 128.3, 128.0, 126.4, 125.5, 116.0, 11.4, 10.5.

HRMS calcd for $[M + H]^+$ C₁₅H₁₃ClNOS 290.0401, found: 290.0402.

2,5-Dimethylnaphtho[1,2-b:3,4-b′]dithiophene (37). The reaction of 2-(2-bromophenyl)-5-methylthiophene 1a (0.253 g, 1 mmol), 4-bromo-2-methylthiophene (0.354 g, 2 mmol), and KOAc (0.294 g, 3 mmol) at 150 °C for 48 h in DMA (4 mL) in the presence of $Pd(OAc)_2$ (4.5 mg, 0.02 mmol) under argon affords the coupling product after evaporation of the solvent and purification on silica gel 37 in 35% (0.094 g) yield as a white solid: mp 138–140 °C. Eluent pentane, R_f 37 = 0.20, R_f 38 $= 0.23.$ Research Article

2-this map (2-this map). Schwarf (1961), term is 0.27 pm and 0.462 mm/h date is under a proportion on the space of the complex are a setting product attention (1972), $\frac{1}{2}$ are the space of the space

¹H NMR (400 MHz, CDCl₃): δ 8.30–8.25 (m, d1H), 8.10–8.04 $(m, 1H)$, 7.65 $(q, J = 1.2$ Hz, 1H), 7.58–7.51 $(m, 2H)$, 7.23 $(q, J =$ 1.2 Hz, 1H), 2.73 $(d, J = 1.2$ Hz, 3H), 2.71 $(d, J = 1.2$ Hz, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 141.1, 138.4, 134.7, 133.1, 132.6, 132.0, 127.3, 126.8, 125.7, 125.5, 124.7, 124.2, 120.8, 120.7, 16.3, 16.2.

HRMS calcd for $[M + H]^{+}$ C₁₆H₁₃S₂ 269.0453, found: 296.0456.

The intermediate 3-bromo-5,5′-dimethyl-2′-phenyl-2,3′ bithiophene 38 was also isolated in a low yield in an impure form as a colorless oil:

 1 H NMR (400 MHz, CDCl₃): δ 7.33–7.21 (m, 5H), 6.82 (q, J = 1.2 Hz, 1H), 6.63 (q, $J = 1.2$ Hz, 1H), 2.52 (s, 3H), 2.41 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 140.2, 138.2, 134.4, 131.4, 129.3, 128.7, 128.6, 128.5, 128.2, 127.4, 108.9, 15.6, 15.5.

HRMS calcd for $[M + H]^+$ C₁₆H₁₄BrS₂ 348.9715, found: 348.9717.

2,6-Dimethyl-4-phenylnaphtho[1,2-b:3,4-c′]dithiophene (39). The reaction of 2-(2-bromophenyl)-5-methylthiophene 1a $(0.253 \text{ g}, 1 \text{ mmol})$ and KOAc $(0.294 \text{ g}, 3 \text{ mmol})$ at 150 °C for

16 h in DMA (4 mL) in the presence of $Pd(OAc)₂$ (4.5 mg, 0.02 mmol) under argon affords the coupling product after evaporation of the solvent and purification on silica gel 39 in 65% (0.112 g) yield as a white solid: mp 172-174 °C. Eluent pentane, $R_f = 0.13$.

¹H NMR (400 MHz, CDCl₃): δ 8.41 (d, J = 7.2 Hz, 1H), 7.86 $(d, J = 7.8, 2.0 \text{ Hz}, 1H), 7.61-7.53 \text{ (m, 2H)}, 7.53-7.40 \text{ (m, 5H)},$ 6.64 (q, $J = 1.3$ Hz, 1H), 3.06 (s, 3H), 2.42 (s, 3H).

 13 C NMR (100 MHz, CDCl₃): δ 138.7, 135.7, 135.6, 132.4, 132.3, 131.3, 130.7, 129.2, 128.8, 128.7, 128.6, 128.4, 126.4, 126.1, 125.5, 124.0, 123.3, 18.4, 16.2.

HRMS calcd for $[M + H]^+$ C₂₂H₁₇S₂ 345.0766, found: 345.0767.

4-(3,4-Dimethylphenyl)-2,6,8,9-tetramethylnaphtho[1,2-b:3,4 c' dithiophene (40). The reaction of 2-(2-bromo-4,5-dimethylphenyl)-5-methylthiophene 1e (0.281 g, 1 mmol) and KOAc $(0.294 \text{ g}, 3 \text{ mmol})$ at 150 °C for 16 h in DMA (4 mL) in the presence of $Pd(OAc)₂$ (4.5 mg, 0.02 mmol) under argon affords the coupling product after evaporation of the solvent and purification on silica gel 40 in 51% (0.102 g) yield as a white solid: mp 102–104 °C. Eluent pentane, $R_f = 0.17$.

¹H NMR (400 MHz, CDCl₃): δ 8. 15 (s, 1H), 7.60 (s, 1H), 7.34 $(s, 1H)$, 7.30 $(d, J = 7.0$ Hz, 1H), 7.22 $(d, J = 7.0$ Hz, 1H), 6.73 $(q,$ $J = 1.3$ Hz, 1H), 3.04 (s, 3H), 2.45 (s, 3H), 2.43 (s, 3H), 2.42 (s, 3H), 2.38 (s, 3H), 2.33 (s, 3H).

¹³C NMR (100 MHz, CDCl₃): δ 137.6, 136.7, 136.6, 135.3, 135.0, 134.7, 132.9, 132.2, 131.8, 130.8, 130.5, 130.4, 129.8, 129.1, 128.5, 126.9, 126.8, 126.1, 124.5, 123.3, 20.7, 20.0, 19.9, 19.8, 18.2, 16.1.

HRMS calcd for $[M + H]^{+}$ C₂₆H₂₅S₂ 401.1392, found: 401.1393.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We are grateful to the CSC for a grant to L. L. We thank the CNRS and "Rennes Metropole" for providing financial support.

References

1 For reviews on metal-catalyzed C–H bond functionalization: (a) L. Ackermann, Carboxylate-assisted transition-metalcatalyzed C–H bond functionalizations: mechanism and scope, *Chem. Rev.*, 2011, 111, 1315-1345; (b) P. B. Arockiam, C. Bruneau and P. H. Dixneuf, $Ruthenium(n)$ -catalyzed C-H bond activation and functionalization, Chem. Rev., 2012, 112, 5879–5918; (c) L. Theveau, C. Schneider, C. Fruit and C. Hoarau, Orthogonal palladium-catalyzed direct C–H bond arylation of heteroaromatics with aryl halides, ChemCatChem, 2016,

8, 3183–3194; (d) S. Ruiz, P. Villuendas and E. P. Urriolabeitia, Ru-catalysed C–H functionalisations as a tool for selective organic synthesis, Tetrahedron Lett., 2016, 57, 3413–3432; (e) S. Agasti, A. Dey and D. Maiti, Palladium-catalyzed benzofuran and indole synthesis by multiple C–H functionalizations, Chem. Commun., 2017, 53, 6544–6556; (f) T. Gensch, M. J. James, T. Dalton and F. Glorius, Increasing catalyst efficiency in C−H activation catalysis, Angew. Chem., Int. Ed., 2018, 57, 2296–2306; (g) J. Kalepu, P. Gandeepan, L. Ackermann and L. T. Pilarski, C4–H indole functionalisation: precedent and prospects, Chem. Sci., 2018, 9, 4203–4216; (h) K. Hirano and M. Miura, A lesson for site-selective C–H functionalization on 2-pyridones: radical, organometallic, directing group and steric controls, *Chem. Sci.*, 2018, 9, 22– 32; (i) A. M. Prendergast and G. P. McGlacken, Transition metal mediated C–H activation of 2-pyrones, 2-pyridones, 2-coumarins and 2-quinolones, Eur. J. Org. Chem., 2018, 6068–6082; (j) S. Mao, H. Li, X. Shi, J.-F. Soulé and H. Doucet, Environmentally benign arylations of 5-membered ring heteroarenes by Pd-catalyzed C−H bonds activations, ChemCatChem, 2019, 11, 269–286; (k) S. Rej, Y. Ano and N. Chatani, Bidentate directing groups: An efficient tool in C–H bond functionalization chemistry for the expedient construction of C–C bond, Chem. Rev., 2020, 120, 1788–1887; (l) A. Saha, M. Shankar, S. Sau and A. K. Sahoo, Multiple annulations of inert C(sp²)–H bonds with alkynes, Chem. Commun., 2022, 58, 4561–4587. Organic Chemistry Frontiers Waccess and Son 6, this
physics at low etably tooling, creat chemistric and the physical certification as 2006, the
 $325-142$ plane, it for
the same of the same of the same of the same of the s

- 2 (a) Y. Akita, A. Inoue, K. Yamamoto, A. Ohta, T. Kurihara and M. Shimizu, Palladium-catalyzed coupling reaction of chloropyrazines with indole, Heterocycles, 1985, 23, 2327– 2333; (b) A. Ohta, Y. Akita, T. Ohkuwa, M. Chiba, R. Fukunaga, A. Miyafuji, T. Nakata, N. Tani and Y. Aoyagi, Palladium-catalyzed arylation of furan, thiophene, benzo $[b]$ furan and benzo[b]thiophene, Heterocycles, 1990, 31, 1951– 1958.
- 3 (a) R. Rossi, F. Bellina, M. Lessi and C. Manzini, Crosscoupling of heteroarenes by C–H functionalization: recent progress towards direct arylation and heteroarylation reactions involving heteroarenes containing one heteroatom, Adv. Synth. Catal., 2014, 356, 17-117; (b) H.-Y. Huang, A. Benzai, X. Shi and H. Doucet, Effective tools for the metal-catalyzed regiodivergent direct arylations of (hetero) arenes, Chem. Rec., 2021, 21, 343–356.
- 4 For selected examples of direct arylations of thiophenes: (a) K. Masui, A. Mori, K. Okano, K. Takamura, M. Kinoshita and T. Ikeda, Syntheses and properties of donor−acceptor-type 2,5-diarylthiophene and 2,5-diarylthiazole, Org. Lett., 2004, 6, 2011–2014; (b) E. David, S. Pellet-Rostaing and M. Lemaire, Heck-like coupling and Pictet–Spengler reaction for the synthesis of benzothieno [3,2-c]quinolines, Tetrahedron, 2007, 63, 8999–9006; (c) H. A. Chiong and O. Daugulis, Palladium-catalyzed arylation of electron-rich heterocycles with aryl chlorides, Org. Lett., 2007, 9, 1449–1451; (d) J. Roger, F. Požgan and H. Doucet, Ligand-less palladium-catalyzed direct 5-aryla-

tion of thiophenes at low catalyst loadings, Green Chem., 2009, 11, 425–432; (e) B. Liégault, I. Petrov, S. I. Gorlesky and K. Fagnou, Modulating reactivity and diverting selectivity in palladium-catalyzed heteroaromatic direct arylation through the use of a chloride activating/blocking group, J. Org. Chem., 2010, 75, 1047–1060.

- 5 For selected examples of direct arylations of furans: (a) M. Parisien, D. Valette and K. Fagnou, Direct arylation reactions catalyzed by $Pd(OH)₂/C$: Evidence for a soluble palladium catalyst, J. Org. Chem., 2005, 70, 7578–7584; (b) B. Liégaut, D. Lapointe, L. Caron, A. Vlassova and K. Fagnou, Establishment of broadly applicable reaction conditions for the palladium-catalyzed direct arylation of heteroatom-containing aromatic compounds, *J. Org. Chem.*, 2009, 74, 1826–1834; (c) J. J. Dong, J. Roger, F. Požgan and H. Doucet, Low catalyst loading ligand-free palladium-catalyzed direct arylation of furans: an economically and environmentally attractive access to 5-arylfurans, Green Chem., 2009, 11, 1832–1846.
- 6 For selected examples of direct arylations of pyrroles or indoles: (a) F. Bellina, S. Cauteruccio and R. Rossi, Palladium- and copper-mediated direct C-2 arylation of azoles—including free (NH)-imidazole, -benzimidazole and -indole—under base-free and ligandless conditions, Eur. J. Org. Chem., 2006, 1379–1382; (b) X. Wang, D. V. Gribkov and D. Sames, Phosphine-free palladiumcatalyzed C–H bond arylation of free (N–H)-indoles and pyrroles, J. Org. Chem., 2007, 72, 1476–1479; (c) J. Roger and H. Doucet, Regioselective C-2 or C-5 direct arylation of pyrroles with aryl bromides using a ligand-free palladium catalyst, Adv. Synth. Catal., 2009, 351, 1977–1990.
- 7 For selected examples of direct arylations of thiazoles: (a) A. Mori, A. Sekiguchi, K. Masui, T. Shimada, M. Horie, K. Osakada, M. Kawamoto and T. Ikeda, Facile synthesis of 2,5-diarylthiazoles via palladium-catalyzed tandem C−H substitutions. Design of tunable light emission and liquid crystalline characteristics, J. Am. Chem. Soc., 2003, 125, 1700–1701; (b) A. Yokooji, T. Okazawa, T. Satoh, M. Miura and M. Nomura, Palladium-catalyzed direct arylation of thiazoles with aryl bromides, Tetrahedron, 2003, 59, 5685– 5689; (c) G. L. Turner, J. A. Morris and M. F. Greaney, Direct arylation of thiazoles on water, Angew. Chem., Int. Ed., 2007, 46, 7996–8000; (d) L.-C. Campeau, M. Bertrand-Laperle, J.-P. Leclerc, E. Villemure, S. Gorelsky and K. Fagnou, C2, C5, and C4 azole N-oxide direct arylation including roomtemperature reactions, J. Am. Chem. Soc., 2008, 130, 3276-3277; (e) J. J. Dong, J. Roger, C. Verrier, T. Martin, R. Le Goff, C. Hoarau and H. Doucet, Carbonates: eco-friendly solvents for palladium-catalysed direct arylation of heteroaromatics, Green Chem., 2010, 12, 2053–2063.
- 8 For examples of Suzuki couplings with 2-(2-bromoaryl)thiophenes using heteroarylboronic acids: (a) M. Daniels, P. de Jong, T. Vandermeeren, L. Van Meervelt, M. Van der Auweraer and W. Dehaen, Bay-substituted thiaza[5]helicenes: Synthesis and implications on structural and spectroscopic properties, J. Org. Chem., 2019, 84, 13528–13539;

(b) C. Maeda, S. Nomoto, K. Akiyama, T. Tanaka and T. Ema, Facile synthesis of azahelicenes and diaza[8]circulenes through the intramolecular Scholl reaction, Chem. – Eur. J., 2021, 27, 15699–15705; (c) F. Chen, J. Kim, Y. Matsuo, Y. Hong, D. Kim, T. Tanaka and A. Osuka, ortho-Phenylene-bridged hybrid nanorings of 2,5-pyrrolylenes and 2,5-thienylenes, Asian J. Org. Chem., 2019, 8, 994–1000.

- 9 For examples of C3-heteroarylations of thiophenes using Suzuki or Stille couplings with 2-aryl-3-halothiophenes: (a) R. Sanz, V. Guilarte, E. Hernando and A. M. Sanjuan, Synthesis of regioselectively functionalized benzo[b]thiophenes by combined ortho-lithiation–halocyclization strategies, J. Org. Chem., 2010, 75, 7443–7446; (b) G. Balaji, T. S. Kale, A. Keerthi, A. M. Della Pelle, S. Thayumanavan and S. Valiyaveettil, Low band gap thiophene–perylene diimide systems with tunable charge transport properties, Org. Lett., 2011, 13, 18–21; (c) D. Tian, Y. Zhou, Z. Li, S. Liu, J. Shao, X. Yang and J. Shao, Thieno[3,2-b]thiophene-based discotic liquid crystal mesogens: Rational synthesis, physical properties and self-assembly, ChemistrySelect, 2017, 2, 8137–8145.
- 10 For a rare example of C3-heteroarylation of thiophene via Suzuki couplings with 2-aryl-3-thiophenesboronic acid using a heteroaryl halide: R. Asato, C. J. Martin, J. P. Calupitan, R. Mizutsu, T. Nakashima, G. Okada, N. Kawaguchi, T. Yanagida and T. Kawai, Photosynergetic amplification of radiation input: from efficient UV induced cycloreversion to sensitive X-ray detection, Chem. Sci., 2020, 11, 2504–2510.
- 11 For the preparation of 3-bromothiophenes: (a) P. Blanchard, P. Verlhac, L. Michaux, P. Frere and J. Roncali, Fine tuning of the electronic properties of linear π -conjugated oligomers by covalent bridging, *Chem.* – *Eur.* J., 2006, 12, 1244–1255; (b) M. He and F. Zhang, Synthesis and structure of alkyl-substituted fused thiophenes containing up to seven rings, J. Org. Chem., 2007, 72, 442–451; (c) J. I. Tietz, A. J. Seed and P. Sampson, Preparation of brominated 2-alkoxythiophenes via oxidation and etherification of 2-thienyltrifluoroborate salts, Org. Lett., 2012, 14, 5058–5061.
- 12 For reviews on 1,4-migration of Pd in catalytic organometallic reactions: (a) S. Ma and Z. Gu, 1,4-Migration of rhodium and palladium in catalytic organometallic reactions, Angew. Chem., Int. Ed., 2005, 44, 7512–7517; (b) Y. Yasui, Intramolecular palladium migration, J. Synth. Org. Chem., Jpn., 2008, 66, 251–252; (c) X. Dong, H. Wang, H. Liu and F. Wang, Recent advances in transition metal migration involving reactions, Org. Chem. Front., 2020, 7, 3530–3556; (d) M.-Y. Li, D. Wei, C.-G. Feng and G.-Q. Lin, Tandem Reactions involving 1,4-Palladium Migrations, Chem. – Asian J., 2022, 17, e202200456.
- 13 For selected examples of Pd-catalyzed reactions involving Pd-migration: (a) R. C. Larock, Y. D. Lu, A. C. Bain and C. E. Russell, Palladium-catalyzed coupling of aryl iodides, nonconjugated dienes and carbon nucleophiles by palladium migration, J. Org. Chem., 1991, 56(15), 4589–4590;

(b) Q. Huang, A. Fazio, G. Dai, M. A. Campo and R. C. Larock, Pd-catalyzed alkyl to aryl migration and cyclization: An efficient synthesis of fused polycycles via multiple C−H activation, J. Am. Chem. Soc., 2004, 126, 7460– 7461; (c) J. Zhao, D. Yue, M. A. Campo and R. C. Larock, An aryl to imidoyl palladium migration process involving intramolecular C−H activation, J. Am. Chem. Soc., 2007, 129, 5288–5295; (d) M. A. Campo, H. Zhang, T. Yao, A. Ibdah, R. D. McCulla, Q. Huang, J. Zhao, W. S. Jenks and R. C. Larock, Aryl to aryl palladium migration in the Heck and Suzuki coupling of o-halobiaryls, *J. Am. Chem.* Soc., 2007, 129, 6298-6307.

- 14 (a) L. Liu and H. Doucet, One pot access to 2′-aryl-2,3′ bithiophenes via twofold palladium-catalyzed C−X/C−H coupling associated to a Pd-1,4-migration, Adv. Synth. Catal., 2022, 364, 2783-2795; (b) L. Liu and H. Doucet, Scope and Limitations of the Palladium-Catalyzed Direct 1,2-Diheteroarylation of 1,2-Dihalobenzene Derivatives, Synthesis, DOI: [10.1055/a-1981-3213](https://doi.org/10.1055/a-�1981-3213).
- 15 For examples of Pd-catalyzed direct C3-arylations of thiophene derivatives: (a) T. Okazawa, T. Satoh, M. Miura and M. Nomura, Palladium-catalyzed multiple arylation of thiophenes, J. Am. Chem. Soc., 2002, 124, 5286–5287; (b) J. J. Dong, J. Roger and H. Doucet, Palladium-catalysed direct 3- or 4-arylation of thiophene derivatives using aryl bromides, Tetrahedron Lett., 2009, 50, 2778–2781; (c) K. Ueda, S. Yanagisawa, J. Yamaguchi and K. Itami, A general catalyst for the β-selective C–H bond arylation of thiophenes with iodoarenes, Angew. Chem., Int. Ed., 2010, 49, 8946–8949; (d) D.-T. D. Tang, K. D. Collins, J. B. Ernst and F. Glorius, Pd/C as a catalyst for completely regioselective C–H functionalization of thiophenes under mild conditions, Angew. Chem., Int. Ed., 2014, 53, 1809–1813; (e) K. Yuan and H. Doucet, Benzenesulfonyl chlorides: new reagents for access to alternative regioisomers in palladium-catalysed direct arylations of thiophenes, Chem. Sci., 2014, 5, 392–396. Research Article

(b) C. Macda, S. Nonzon, K. Akiyama, T. Tanaka and (d) Q. Hung, A. Paris, D. Dai, M. A. Campo and

T. Ema, Backers Article yards and alleaded phenomenon and performation-of-performation-of-performation-o
	- 16 For rare examples of synthesis of 2′-aryl-2,3′-bithiophene derivatives: (a) S. Varello and S. T. Handy, Double couplings of dibromothiophenes using boronic acids and boronates, Synthesis, 2009, 138–142; (b) A. Acharya, G. Parameshwarappa, B. Saraiah and H. Ila, Sequential one-pot synthesis of tri- and tetrasubstituted thiophenes and fluorescent push–pull thiophene acrylates involving (het)aryl dithioesters as thiocarbonyl precursors, J. Org. Chem., 2015, 80, 414–427; (c) Y. P. Zhong, J. M. Gao, P. Liu and W. J. Deng, Synthesis and electrochromic properties of star-shaped conjugated oligomers, Mater. Res. Innovations, 2015, 19, s51–s54; (d) J.-H. Li, Q. Huang, W. Rao, S.-Y. Wang and S.-J. Ji, A trisulfur radical anion $(S_3(^{-})$ involved sulfur insertion reaction of 1,3-enynes: Sulfide sources control chemoselective synthesis of 2,3,5-trisubstituted thiophenes and 3-thienyl disulfides, Chem. Commun., 2019, 55, 7808–7811.
	- 17 T. Cantat, E. Génin, C. Giroud, G. Meyer and A. Jutand, Structural and kinetic effects of chloride ions in the palla-

dium-catalyzed allylic substitutions, J. Organomet. Chem., 2003, 687, 365–376.

- 18 From a mixture of 4-(2-bromophenyl)-2-methylthiophene and 2-methylthiophene or 2-acetylthiophene, under the same conditions, complex mixtures of products were obtained.
- 19 For rare examples of the synthesis of 4-bromo-2-chlorothiophenes: (a) S. Dapperheld, M. Feldhues, H. Litterer, F. Sistig and P. Wegener, Selektive debromierung von thiophen-derivaten durch elektrochemische reduction, Synthesis, 1990, 403–405; (b) L. N. Lucas, J. Van Esch, R. M. Kellogg and B. L. Feringa, A new synthetic route to symmetrical photochromic diarylperfluorocyclopentenes, Tetrahedron Lett., 1999, 40, 1775–1778; (c) C. Oberdorf, D. Schepmann, J. M. Vela, H. Buschmann, J. Holenz and B. Wunsch, Thiophene bioisosteres of spirocyclic σ receptor ligands: Relationships between substitution pattern and σ receptor affinity, *J. Med. Chem.*, 2012, 55, 5350-5360.
- 20 W. Hagui, H. Doucet and J.-F. Soulé, Application of palladium-catalyzed $C(sp^2)$ -H bond arylation to the synthesis of polycyclic (hetero)aromatics, Chem, 2019, 5, 2006–2078.
- 21 Y. Koga, T. Kaneda, Y. Saito, K. Murakami and K. Itami, Synthesis of partially and fully fused polyaromatics by annulative chlorophenylene dimerization, Science, 2018, 359, 435–439.
- 22 N. Miyaura and A. Suzuki, Palladium-catalyzed cross-coupling reactions of organoboron compounds, Chem. Rev., 1995, 95, 2457–2483.
- 23 For a concerted metalation deprotonation mechanism: (a) D. L. Davies, S. M. A. Donald and S. A. Macgregor, Computational study of the mechanism of cyclometalation by palladium acetate, J. Am. Chem. Soc., 2005, 127, 13754– 13755; (b) M. Lafrance and K. Fagnou, Palladium-catalyzed benzene arylation: Incorporation of catalytic pivalic acid as a proton shuttle and a key element in catalyst design, J. Am. Chem. Soc., 2006, 128, 16496–16497; (c) S. I. Gorelsky, D. Lapointe and K. Fagnou, Analysis of the concerted metalation-deprotonation mechanism in palladium-catalyzed direct arylation across a broad range of aromatic substrates, J. Am. Chem. Soc., 2008, 130, 10848–10849; (d) S. I. Gorelsky, Origins of regioselectivity of the palladium-catalyzed (aromatic)C–H bond metalation–deprotonation, Coord. Chem. Rev., 2013, 257, 153–164; (e) D. L. Davies, S. A. Macgregor and C. L. McMullin, Computational studies of carboxylate-assisted C–H activation and functionalization at group 8–10 transition metal centers, Chem. Rev., 2017, 117, 8649–8709; (f) R. A. Alharis, C. L. McMullin, D. L. Davies, K. Singh and S. A. Macgregor, The importance of kinetic and thermodynamic control when assessing mechanisms of carboxylate-assisted C–H activation, J. Am. Chem. Soc., 2019, 141, 8896–8906. Organic Chemistry Frontiers

dium-catagod silyie anbetituding, J. Organomet. Chem, 12 2 M. Miyama and A. Surallished on 4/27/2023. Diversed are commons are commons are commons are commons are commons are commons are commo