


 Cite this: *RSC Adv.*, 2023, 13, 881

Electronic and optical properties of thiogermanate AgGaGeS₄: theory and experiment

 Tuan V. Vu,^{a,b} Vo D. Dat,^{a,b} A. A. Lavrentyev,^c B. V. Gabrelian,^d Nguyen N. Hieu,^{e,f} G. L. Myronchuk^{g,h} and O. Y. Khyzhun^{i,*g,h}

The electronic and optical properties of an AgGaGeS₄ crystal were studied by first-principles calculations, where the full-potential augmented plane-wave plus local orbital (APW+lo) method was used together with exchange–correlation pseudopotential described by PBE, PBE+*U*, and TB-mBJ+*U* approaches. To verify the correctness of the present theoretical calculations, we have measured for the AgGaGeS₄ crystal the XPS valence-band spectrum and the X-ray emission bands representing the energy distribution of the electronic states with the biggest contributions in the valence-band region and compared them on a general energy scale with the theoretical results. Such a comparison indicates that, the calculations within the TB-mBJ+*U* approach reproduce the electron-band structure peculiarities (density of states – DOS) of the AgGaGeS₄ crystal which are in fairly good agreement with the experimental data based on measurements of XPS and appropriate X-ray emission spectra. In particular, the DOS of the AgGaGeS₄ crystal is characterized by the existence of well-separated peaks/features in the vicinity of –18.6 eV (Ga-d states) and around –12.5 eV and –7.5 eV, which are mainly composed by hybridized Ge(Ga)-s/p and S-p state. We gained good agreement between the experimental and theoretical data with respect to the main peculiarities of the energy distribution of the electronic S 3p, Ag 4d, Ga 4p and Ge 4p states, the main contributors to the valence band of AgGaGeS₄. The bottom of the conduction band is mostly donated by unoccupied Ge-s states, with smaller contributions of unoccupied Ga-s, Ag-s and S-p states, too. The AgGaGeS₄ crystal is almost transparent for visible light, but it strongly absorbs ultra-violet light where the significant polarization also occurs.

 Received 1st December 2022
 Accepted 20th December 2022

DOI: 10.1039/d2ra07639j

rsc.li/rsc-advances

1 Introduction

Laser sources working in the UV, near-infrared, and mid-infrared ranges have shown their important role in many practical applications such as atmospheric and solar guidance, free-space communication, detectors, and others.^{1–7} Therefore, many nonlinear optical crystals have intensively been studied

and developed to produce laser sources of these ranges.^{3,8–12} Among these materials, special attention has been paid to those which can effectively convert the 1.060–1.064 μm emission from commercially available laser medium sources such as Nd:YAG, Nd:Cr:YAG, Nd:YLF, Nd:YVO₄, and Nd:YCO.^{13–16} However, only a few of the explored nonlinear optical materials, namely AgGaS₂, AgGaSe₂, and ZnGeP₂,^{17–21} are suitable for practical uses. At the same time, practical applications of these NLO crystals are also limited when pumping within the 1–9 μm range due to the opaqueness to wavelengths less than 2 μm and the multi-phonon absorption at wavelengths more than 9 μm. Laser damage threshold causes another practical restriction of these crystals because the anisotropic thermal expansion lowers their resistance to thermal shock.^{19,21–24}

With many advantages over the drawbacks of the aforementioned compounds, a thiogermanate AgGaGeS₄ crystal is currently considered among best NLO materials. The AgGaGeS₄ crystal possesses a wider transparency range (0.5–11.5 μm)²⁵ in comparison with ZnGeP₂. Its residual absorption around 1.064 μm wavelengths is as low as 0.05 cm^{–1} making the titled crystal to be suitable for Nd:YAG laser pumping.²⁶ The laser damage threshold of AgGaGeS₄ was revealed to be 203 MW cm^{–2} at 2.05 μm,²⁷ and 230 MW cm^{–2} at 9.55 μm,²⁸ allowing applications of

^aLaboratory for Computational Physics, Institute for Computational Science and Artificial Intelligence, Van Lang University, Ho Chi Minh City, Vietnam. E-mail: tuan.vu@vlu.edu.vn

^bFaculty of Mechanical – Electrical and Computer Engineering, Van Lang University, Ho Chi Minh City, Vietnam

^cDepartment of Electrical Engineering and Electronics, Don State Technical University, 1 Gagarin Square, 344010 Rostov-on-Don, Russian Federation

^dDepartment of Computational Technique and Automated System Software, Don State Technical University, 1 Gagarin Square, 344010 Rostov-on-Don, Russian Federation

^eInstitute of Research and Development, Duy Tan University, Da Nang 550000, Vietnam

^fFaculty of Natural Sciences, Duy Tan University, Da Nang 550000, Vietnam

^gDepartment of Experimental Physics and Information-Measuring Technology, Lesya Ukrainka Volyn National University, 13 Voli Avenue, 43025 Lutsk, Ukraine

^hFrantsevych Institute for Problems of Materials Science, National Academy of Sciences of Ukraine, 3 Krzhynhanivsky Street, 03142 Kyiv, Ukraine. E-mail: khyzhun@ipms.kiev.ua



operating conditions at higher current powers than AgGaS₂ or AgGaSe₂ crystals allow.^{27,28} Moreover, AgGaGeS₄ crystal reveals a wide range of phase-matching schemes, allowing the crystal to be effectively used for various frequency conversions.^{29,30} Finally, the growth/synthesis conditions were successfully explored aiming to obtain high-pure and large-size AgGaGeS₄ crystals (centimeter dimensions) that can be effectively used to convert over wide range of wavelengths including 1.064 μm, 4 μm, and 11 μm.

Most of studies of AgGaGeS₄ crystals depend essentially on experimental conditions and quality of the crystals. During the growth of AgGaGeS₄ crystals, many defects may occur forming additional electronic states in the band gap, making its value varying from 2.25 till 2.80 eV.^{31,32} In particular, vacancies on special crystallography sites have been established to form during annealing AgGaGeS₄ crystals and their presence affects the band gap value being within the energy range of 2.787–2.805 eV in such a case.²⁵ Moreover, the band gap of AgGaGeS₄ was experimentally discovered to be inversely proportional to temperature value,³³ and at temperatures lower than 100 K, the band gap of AgGaGeS₄ is bigger than 3 eV. It is well-known that the electronic structure plays a crucial role in determining the optical properties of semiconductors. For the case of AgGaGeS₄ crystal considering a promising laser convertor, the value of the band gap is very important as it determines the transparency of semiconducting materials. In this study, a detailed knowledge about the electronic structure of the AgGaGeS₄ crystal as well as its optical properties is provided based on first-principles calculations. It is worth mentioning that, to the best of our knowledge, the electronic structure of AgGaGeS₄ has not been studied theoretically yet. Furthermore, the theoretical densities of states of AgGaGeS₄ are verified in the present work by using measurements of experimental XPS and X-ray emission spectra.

2 Calculation method

The unit-cell of the AgGaGeS₄ crystal was constructed using its space group *Fdd2* and the lattice parameters $a = 12.019 \text{ \AA}$, $b = 22.924 \text{ \AA}$, $c = 6.8837 \text{ \AA}$ as established in ref. 34 in which each Ga or Ge atom belongs to a tetrahedron formed by S atoms, while Ag atoms occupy the space between these tetrahedra as presented in Fig. 1. For structural optimization as well as electronic properties calculations, a $1 \times 1 \times 1$ supercell is used, and the Brillouin zone is sampled with a 1000-*k*-point mesh to satisfy requirements of density functional theory (DFT) methods as implemented in the WIEN2k package, where the Kohn–Sham equations are solved by the APW+lo method.³⁵ In the APW+lo method, for the case of the AgGaGeS₄ crystal, the unit cell is partitioned into atomic spheres for Ag, Ga, Ge, and S atoms with their radii R_{MT} of 2.5, 2.2, 2.26, 1.91 a. u., respectively, where the basis functions are expressed as the sum of atomic partial waves to the maximum angular momentum $l_{\text{max}} = 10$, and the electron density ρ is expanded as Fourier series up to $G_{\text{max}} = 14$ a. u.⁻¹. Meanwhile, in the interstitial space of the unit cell, the plane waves are summed up to $l_{\text{max}} = 4$, and ρ is presented as the sum of the product of lattice harmonics and radial functions. Because thio germanate AgGaGeS₄ is a semiconductor

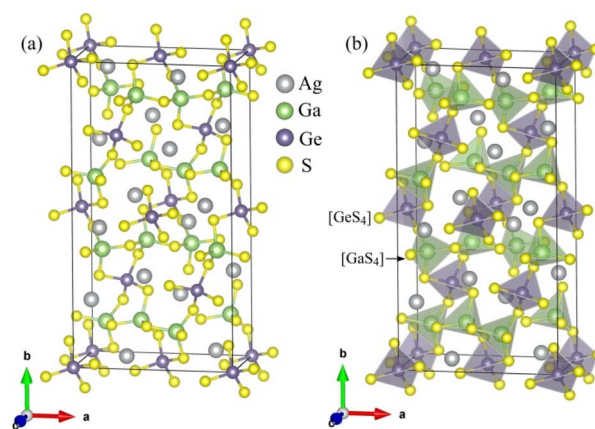


Fig. 1 (a) Positions of the atoms and (b) stacking the [GaS₄] and [GeS₄] tetrahedra in the AgGaGeS₄ unit cell that is outlined, for clarity.

consisting of atoms involving d orbitals, either Perdew–Burke–Ernzerhof generalized gradient approximation (PBE)³⁶ or Tran and Blaha modified Becke–Johnson (TB-mBJ) potential³⁷ are known to produce reasonable results being used for the exchange–correlation potential in such a case and the Hubbard parameter $U^{38,39}$ was also included to treat the Coulomb interactions between local orbitals.^{40,41} The parameter $R_{\text{MT}}^{\text{min}} \times k_{\text{max}} = 8.0$ is used as convergence criterion of the scf-calculations, while the iterations are suggested to reach the convergence when the difference in charge density for two subsequent iterations is less than 10^{-4} Ry. After reaching the convergence, an equilibrium structure of AgGaGeS₄ is achieved together with its wave functions $|kn\sigma\rangle$ at wave vector k and spin σ , and the corresponding eigenvalues of energy E_{kn} . This information can be used to calculate the imaginary part $\epsilon_2(\omega)$ of the dielectric function $\epsilon(\omega) = \epsilon_1(\omega) + i\epsilon_2(\omega)$ at the angular frequency ω of the electromagnetic radiation by applying the momentum operator p on the wave function $|kn\sigma\rangle$ as follows:⁴²

$$\epsilon_2^j(\omega) = \frac{4\pi^2 e^2}{Vm^2\omega^2} \times \sum_{m'\sigma} \langle kn\sigma | p_j | kn'\sigma \rangle \langle kn'\sigma | p_j | kn\sigma \rangle \times f_{kn} (1 - f_{kn'}) \delta(E_{kn'} - E_{kn} - \hbar\omega) \quad (1)$$

where f_{kn} is the Fermi distribution function, V is the volume of AgGaGeS₄ unit cell, while the electron's mass and charge are m and e , respectively. Then, the Kramers–Kronig relations⁴³ were used to calculate the real part $\epsilon_1(\omega)$ of the dielectric function. Finally, the obtained matrix of $\epsilon_2^j(\omega)$ was used in the “joint density of states”⁴⁴ to calculate optical properties of the AgGaGeS₄ crystal.

3 Experimental

For the present X-ray spectroscopy measurements, we use the AgGaGeS₄ crystal synthesized by the two-zone Bridgman method in ref. 34. In the above article, it was proved that the crystal under consideration is of high optical quality and without any additional phase admixtures. Therefore, in the present work we report on measurements of X-ray emission Ag



$L\beta_{2,15}$ ($N_{IV,V} \rightarrow L_{III}$ transition) and Ga(Ge) $K\beta_2$ ($N_{II,III} \rightarrow K_I$ transition) bands, which represent the energy distribution of the valence Ag d and Ga(Ge) p states, respectively. The X-ray emission Ag $L\beta_{2,15}$ and Ga(Ge) $K\beta_2$ bands were derived using DRS-2M spectrographs equipped with X-ray BKhV-7 tubes (gold anodes) using quartz crystals with the (10 $\bar{1}$ 0) and (1340) reflecting planes, respectively. The both crystals were focused by the Rowland circle geometry following Johann's technique.⁴⁵ The details of the present X-ray emission measurements are similar to those described in details in ref. 46 and 47 in the case of the Ag $L\beta_{2,15}$ band and the Ga(Ge) $K\beta_2$ bands, respectively. The XPS valence-band spectrum of the AgGaGeS₄ crystal was measured using the UHV-analysis system following the technique;³⁴ however, to gain better resolution, in the present work we employ a Mg $K\alpha$ source ($E = 1253.6$ eV) and the spectrum was recorded at constant pass energy of 20 eV. The XES S $K\beta_{1,3}$ band was measured as reported in ref. 34.

4 Results and discussion

4.1 Calculated electronic structure and XPS and XES data for the AgGaGeS₄ crystal

The major distinction between PBE, PBE+*U*, and TB-mBJ+*U* methods is that, how these methods describe exchange–correlation potential, which affects the theoretical electronic structure of a solid, in particular AgGaGeS₄. In order to determine the best method for evaluating the electronic band structure of the AgGaGeS₄ crystal, both theoretical total density of state and the experimental XPS valence-band spectrum are shown in Fig. 2, where the Fermi level has been shifted to 0 eV corresponding to the top of the highest occupied valence band in the compound under consideration. It is well known that, the XPS technique induces a higher energy position of the Fermi level than the theoretical valence highest bands⁴⁸ (in our case, zero of the energy scale of the experimental XPS spectrum corresponds to the Fermi level of the energy analyzer used in the UHV-analysis system, in fact, a PHOIBOS 150 analyzer). Therefore, presented in Fig. 2 the XPS valence band spectrum measured for the AgGaGeS₄ crystal was shifted by 0.9 eV toward the direction of the E_F position to gain correspondence between most prominent peaks/peculiarities detected for the experimental and theoretical results as recommended to do in the case of semi-conducting materials.⁴⁷

Fig. 2 shows a noticeable variation in the energy band gaps of the AgGaGeS₄ crystal as it is calculated by the PBE, PBE+*U*, and TB-mBJ+*U* methods, which are found to be equal to 1.575 eV, 2.012 eV, and 3.414 eV, respectively (note that, the values of $U_{Ga} = 0.65$ Ry and $U_{Ag} = 0.5$ Ry are used in the present DFT calculations). It is worth mentioning that, we treated different *U* parameters for finding the best correspondence of the shapes of theoretical DOS curves with the experimental XPS spectrum, energy positions of their most prominent fine-structure peculiarities and E_g values. In the present work we present data only for the case of the best agreement between the theoretical and experimental data. Available experimental data reveal the band gap of about 2.84 eV for AgGaGeS₄ at room temperature.^{25,31,49} Meanwhile, Davydyuk *et al.*³³ have shown experimental data

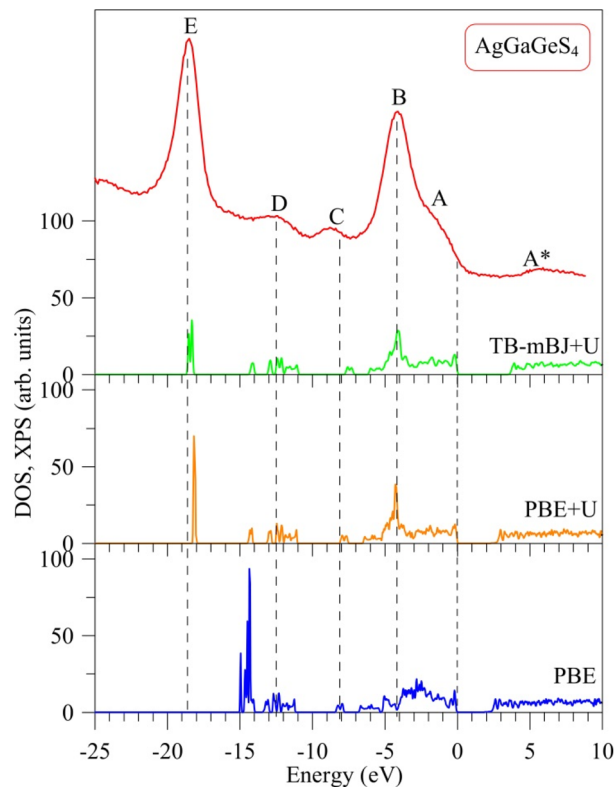


Fig. 2 Experimental electronic structure as determined in the present work by measurements of the XPS valence band spectrum versus theoretical total density of states revealed by DFT calculations employing the PBE, PBE+*U* and TB-mBJ+*U* approaches for AgGaGeS₄.

where the band gap of the AgGaGeS₄ is from 2.84 eV to above 3 eV. Therefore, in our opinion, the TB-mBJ+*U* method, which features the energy band gap of 3.414 eV, looks to be the most suitable for calculating the electronic properties of the AgGaGeS₄ crystal. By Pauli principles, electrons with the same spin avoid each other causing the exchange effect; meanwhile electrons with opposite spin interact with each other due to the correlation effect. Both effects are not correctly introduced in the PBE method providing the interaction of an electron with itself. Due to the ground-state nature of the DFT approach, the valence band values are overestimated under such circumstances. This overestimation does not happen to the unoccupied conduction bands; therefore, the energy band gap occurring between the valence and conduction bands is narrowed. As it is shown in Fig. 2, the band gap revealed by the PBE method is by about 0.5 eV smaller than that derived by employing the PBE+*U* method. The Hubbard parameter *U* was introduced to the PBE+*U* method in the present DFT calculations to provide some correction to the correlation effect of the core electrons. This correlation correction effect, as it is evidenced from Fig. 2, improves the theoretical energy band gap value of the AgGaGeS₄ crystal and provides better valence band structure regarding the shape of the experimental XPS valence-band spectrum. The main peak occurring at around -3 eV in the PBE-derived spectrum shifts down to -4 eV in the PBE+*U*



like spectrum. In the latter case, the main valence band peak coincides well with the energy position of the peak B of the XPS curve. One of the most important improvements in the present DFT calculations brought by the U correction technique is the rearrangement of the electron energy levels. The PBE+U method recognizes more electrons for the electronic states positioned near -3 eV, and features another high intensity peak to be located near -18.1 eV instead of -14.9 eV as derived by the PBE method. These electronic peculiarities are in good agreement with the XPS data. It is worth mentioning that, the exchange correction is introduced in the TB-mBJ approach by a short-range potential model.³⁷ As can be seen from Fig. 2, when employing the U correction technique using TB-mBJ potential³⁷ (the TB-mBJ+U method), we observe better correspondence of the theoretical total DOS curve to the experimental electronic structure of the AgGaGeS₄ crystal; in such a case, the main peak near -18.6 eV is located exactly around the energy position of the peak E detected in the XPS valence-band spectrum and with an expected relative intensity.

Curves of partial DOS of AgGaGeS₄ are presented in Fig. 3. The highest peak E of the XPS spectrum corresponds to the energy level around -18.6 eV in the theoretical DOS curve, where the electrons originating from Ga-d orbitals form the

densest electronic state sub-band. The Ge(Ga)-s/p and S-s orbitals mainly donate into broader energy regions, however less dense sub-bands, ranging from -14.3 to -11 eV, and -8 to -7 eV. These bands correspond to the D and C peaks in the XPS spectrum, respectively. In the highest energy levels of the valence band of AgGaGeS₄ ranging from -6.1 to 0 eV, the Ag-d orbitals donate major parts to the peak B at -4 eV. As can be seen from Fig. 3, Ga-s states bring the principal contributions to the bottom of the valence band, Ga-p and S-p states produce the formation of the near-Fermi feature A of the XPS spectrum, while S-p states dominate at the top of the valence band of the quaternary sulfide under consideration. In addition to the above-mentioned contributors, Fig. 3 presents that Ge-p states also substantially contribute to the peak B of the XPS valence band spectrum of AgGaGeS₄.

To verify the above predictions made on the basis of the present DFT calculations, we have performed XES measurements of the appropriate bands giving information on the partial densities of states for atoms composing AgGaGeS₄. Our estimations employing the XPS technique bring the following atomic percentage of the constituting chemical elements in the AgGaGeS₄ crystal under study: 14.9 at% Ag, 12.2 at% Ga, 14.6 at% Ge, 58.3 at% S. The nominal atomic percentage in the ideal AgGaGeS₄ crystal should be as follows: 14.3 at% Ag, 14.3 at% Ga, 14.3 at% Ge, 57.1 at% S. Our XPS data indicate that the composition of the constituting chemical elements in the AgGaGeS₄ crystal used for the present experiments is close to the ideal composition for all the atoms except of Ga. In particular, literature data indicates⁵⁰ that some deficit of Ga in the AgGaGeS₄ crystals looks to be a problem for such kinds of compounds. In particular, partial lack of Ga atoms in comparison to the expected theoretically atomic ratios was detected previously for a series of AgGaGeS₄·nGeS₂ crystals possessing the crystal structure belonging to the SG *Fdd2*, with *n* equal to 0, 2, 3, and 4. We believe that this small lack of Ga in the crystal being investigated does not influence the present experimental results. In the present experiments we do not detect XPS or XES spectral peculiarities associated with vacancy formation. It should be noted that our XPS measurements of the XPS core-level Ga 2p_{3/2}, Ag 3d_{5/2}, and Ge 3d spectra in the AgGaGeS₄ crystal³⁴ indicate that their binding energies are equal to $1117.35(\pm 0.05)$, $367.45(\pm 0.05)$, and $30.86(\pm 0.05)$ eV, respectively, being in excellent agreement with the XPS data by Huang *et al.*⁵¹ for AgGaGe_nS_{2(n+1)} (*n* = 2, 3, 4, and 5) crystals. For the latter crystals, the XPS measurements⁵¹ give the binding energies of 1117.35 – 1117.03 eV, 367.45 – 367.38 eV, and 30.88 – 30.87 eV for the XPS core-level Ga 2p_{3/2}, Ag 3d_{5/2}, and Ge 3d spectra, respectively. Interestingly, the authors⁵¹ have detected some tendency of decreasing binding energies for the XPS core-level Ga 2p_{3/2} and Ag 3d_{5/2} spectra in the AgGaGe_nS_{2(n+1)} crystals with increasing the *n* value, while the binding energy of the Ge 3d spectra remains unchanged in such a case.

It is well-known that the X-ray emission S Kβ_{1,3}, Ag Lβ_{2,15}, Ga Kβ₂ and Ge Kβ₂ bands plotted in Fig. 3 represent the energy distributions of the valence electronic S-p, Ag-d, Ga-p and Ge-p states, respectively.^{45,52,53} As shown in Fig. 3, the maximum position of the XES S Kβ_{1,3} band is located around the

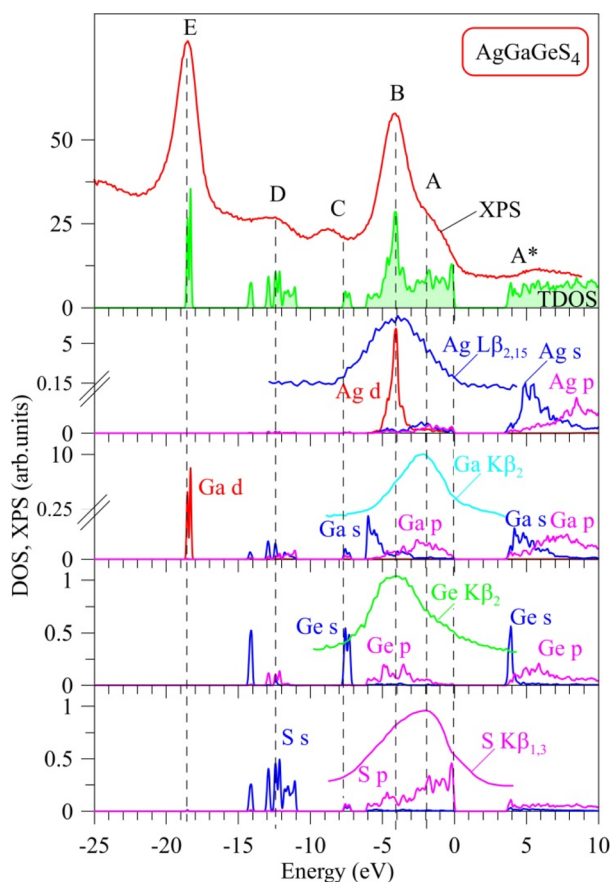


Fig. 3 Experimental electronic structure (XPS valence-band spectrum, XES S Kβ_{1,3}, Ag Lβ_{2,15}, Ga Kβ₂ and Ge Kβ₂ bands) versus total DOS and its main constituent orbitals calculated by TB-mBJ+U for AgGaGeS₄ crystal.



peculiarity A of the XPS spectrum indicating that the main contributions of S 3p states are expected to be positioned in the upper part of the AgGaGeS₄ valence band, with essential contributions of these electronic states throughout the whole band region ranging from -6.1 to 0 eV being in excellent predictions of the present DFT calculations. Furthermore, the maxima of the XES Ag L $\beta_{2,15}$ and Ge K β_2 bands are detected around the main maximum B of the XPS valence-band spectrum of the AgGaGeS₄ crystal, while the maximum of the Ga K β_2 band is detected in the vicinity of the shoulder A of the XPS spectrum. These experimental data indicate that the valence Ag-d, and Ge-p states contribute mainly in the energy region in the vicinity of the maximum B of the XPS spectrum, while Ga-p states around the peculiarity A, again supporting the present DFT band-structure calculations of the AgGaGeS₄ compound. Therefore, we gain good agreement between the experimental and theoretical data with respect to the main peculiarities of the energy distribution of the electronic S 3p, Ag 4d, Ga 4p and Ge 4p states, the main contributors to the valence band of AgGaGeS₄. Some peculiar hybridization of the mentioned electronic states is observed in the AgGaGeS₄ valence band suggesting the existence of covalent bonding nature (in addition to the ionic component) in the AgGaGeS₄ crystal. The similar peculiarities of the chemical bonding are characteristic for related quaternary silver- and gallium-bearing selenide, AgGaSiSe₄.⁵⁴ The bottom of the conduction band of AgGaGeS₄, as Fig. 3 presents, is mostly donated by unoccupied Ge-s states, with smaller contributions of unoccupied Ga-s, Ag-s and S-p states, too.

4.2 Optical properties of the AgGaGeS₄ crystal

The optical properties of a material relate to how this material interacts with the incident electromagnetic (EM) waves, which can be characterized by the dielectric function $\varepsilon(\omega) = \varepsilon_1(\omega) + i\varepsilon_2(\omega)$. The real part $\varepsilon_1(\omega)$ of the dielectric function indicates on how a material stores and re-emits energy of the EM waves, while the imaginary part $\varepsilon_2(\omega)$ presents the permanent energy absorption of this material. The calculated dielectric function $\varepsilon(\omega)$ of the AgGaGeS₄ crystal is depicted in Fig. 4. The spectrum of its real part $\varepsilon_1(\omega)$ is featured by the static value of about 5, and main peaks near 4 eV and 6 eV indicate values of the

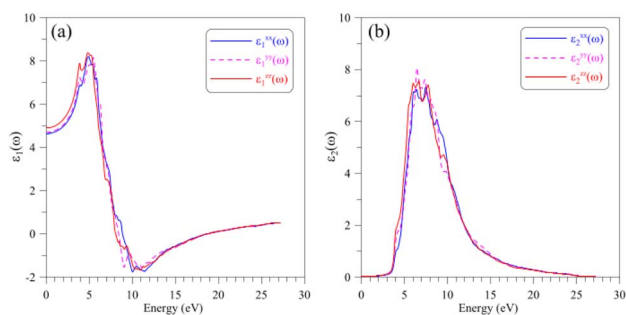


Fig. 4 Dielectric function including (a) real part $\varepsilon_1(\omega)$ and (b) imaginary part $\varepsilon_2(\omega)$ of the AgGaGeS₄ crystal obtained by DFT calculations within the TB-mBJ+*U* method.

corresponding energies/wavelengths which are mostly reflected by the AgGaGeS₄ crystal.

However, the magnitudes of these peaks vary slightly depending on the direction of the incident waves due to the anisotropy of the AgGaGeS₄ crystal. In the energy interval 3–5 eV, the real dielectric component along the *y*-axis, $\varepsilon_1^{yy}(\omega)$, is the lowest by intensity; therefore, the reflectivity of AgGaGeS₄ along the *y*-direction is supposed to be the lower than in the *x*- and *z*-directions. The shifts of the $\varepsilon_1(\omega)$ peaks are observed mainly in the energy ranges 3–6 eV and 8–12 eV. Therefore, the intensity of the reflected waves by AgGaGeS₄ not only depends on the direction but also on the energy of the EM waves. The polarization of the imaginary part $\varepsilon_2(\omega)$ of the dielectric function occurs at energy intervals 3–4.5 eV and 5.5–16.5 eV corresponding to the highest permanent absorption of the EM waves whose wavelengths are in ranges 257–413 nm, and 75–225 nm, respectively. One of the consequences of permanent absorbing EM waves is to excite the electrons from the valence band to the conduction band. Therefore, the main peaks/features of the $\varepsilon_2(\omega)$ function located near 3.5 eV, 6 eV, 7 eV and 9 eV originate due to the transition of electrons from S-p orbitals in the valence band to the Ge-s orbitals in the conduction band.

Regarding to the optical properties of the AgGaGeS₄ crystal, it is well-known that the square root of the dielectric function $\varepsilon(\omega)$ gives the refraction index in the form of complex numbers, where real part $n(\omega)$ is related to how the EM waves are slowed down and inclined in materials, and the imaginary part $k(\omega)$ describes how fast the energy of EM waves vanishes in materials. As shown in Fig. 5(a and b), the shapes of spectra of the

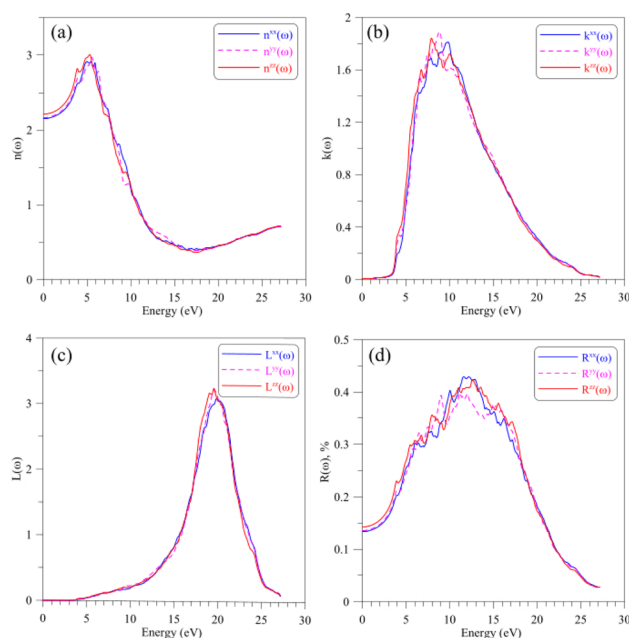


Fig. 5 Optical parameters such as (a) refractive index $n(\omega)$, (b) extinction coefficient $k(\omega)$, (c) electron energy-loss spectrum $L(\omega)$, and (d) optical reflectivity $R(\omega)$ of the AgGaGeS₄ crystal obtained by DFT calculations within the TB-mBJ+*U* method.



refractive index and the extinction coefficient, $n(\omega)$ and $k(\omega)$, are similar to those of the $\varepsilon_1(\omega)$ and $\varepsilon_2(\omega)$ spectra, respectively (cf. Fig. 4(a and b)). The spectrum of the refractive index $n(\omega)$ as presented in Fig. 5(a) reaches its highest peaks in the energy range of 3–7 eV signifying the range of EM waves which are essentially inclined and reflected by the AgGaGeS₄ crystal. The EM waves at slightly higher energy levels, 7–12 eV, according to Fig. 5(b), are likely to vanish in the AgGaGeS₄ crystal. At higher energy levels, 17–23 eV, most energy of the incident EM waves will be lost. The electron energy-loss spectrum $L(\omega)$ presented in Fig. 5(c) reveals the existence of a maximum centered at around 19.5 eV. This value corresponds to plasma frequency of the AgGaGeS₄ crystal. The reflectivity spectrum $R(\omega)$, shown in Fig. 5(d), reveals high values in the energy range of 5–7 eV as the result of the absorption and re-emit effects governed by the $n(\omega)$ and $k(\omega)$ parameters. The direction-dependence of the $R(\omega)$ function is also obvious in this energy range, where the reflectivity along the z-direction is the highest followed by reflectivity along the x- and y-directions. At energy levels higher than 17 eV, the reflectivity of AgGaGeS₄ decreases abruptly due to the sharp increase of the energy-loss (Fig. 5(c)).

The absorption coefficient $\alpha(\omega)$ of AgGaGeS₄, as shown in Fig. 6, is neglectable for the EM waves whose energies are smaller than 2.5 eV. Therefore, AgGaGeS₄ is transparent for the most part of the visible light. Starting from 2.5 eV, the absorption increases slightly and reaches its fast increase in the vicinity of 3.4 eV, which is determined as the theoretical band gap of the AgGaGeS₄ crystal. The polarization of the $\alpha(\omega)$ spectrum is minor up to the energy range where the absorption coefficient $\alpha(\omega)$ reaches the highest values (above 10^6 cm^{-1}). These highest values of the $\alpha(\omega)$ spectrum are detected in the energy range of 7–12 eV. This energy range is located inside the energy range of 5.5–16.5 eV, where the $\varepsilon_2(\omega)$ function possesses its highest values. Consequently, it is expected that the AgGaGeS₄ crystal strongly absorbs the EM waves with the wavelengths being in range of 103–177 nm. At these wavelengths, the $\alpha(\omega)$ function is strongly polarized: the $\alpha^{zz}(\omega)$ component reveals higher intensity as compared to those of the other two remaining directions. The EM waves shorter than 61 nm are weakly absorbed by AgGaGeS₄ and the crystal is almost transparent up to ultra-violet lights.

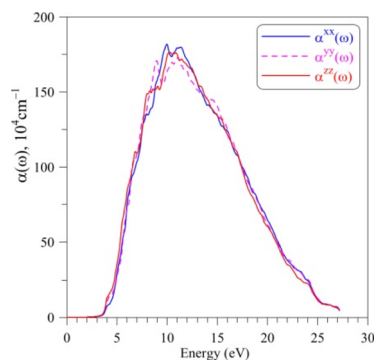


Fig. 6 Absorption coefficient $\alpha(\omega)$ AgGaGeS₄ crystal calculated by the TB-mBJ+*U* method.

5 Conclusions

The first-principles calculations using PBE, PBE+*U*, and TB-mBJ+*U* methods were performed to study the electronic and optical properties of the AgGaGeS₄ crystal. Results of the present DFT calculations within the TB-mBJ+*U* method are found to be in the best agreement with the experiments based on XPS and XES measurements. In particular, in accordance with the present first-principles band-structure calculations, S 3p states are located mainly at the top and in the upper part of the AgGaGeS₄ valence band, with essential contributions of these electronic states throughout the whole band region ranging from −6.1 to 0 eV, Ag 4d and Ge 4p states contribute mainly in the central part of the band, while Ga 4p states in its upper portion. The present theoretical predictions have been supported by measurements of the X-ray emission bands giving information on peculiarities of the energy distribution of the electronic S 3p, Ag 4d, Ga 4p and Ge 4p states, the main contributors to the valence band of AgGaGeS₄. Some peculiar hybridization of the mentioned electronic states is observed in the AgGaGeS₄ valence band suggesting the existence of essential covalent constituent (in addition to the ionic one) of the chemical bonding in the AgGaGeS₄ crystal. The bottom of the conduction band of AgGaGeS₄, in accordance with our DFT calculations, is mainly formed by unoccupied Ge-s states, with smaller contributions of unoccupied Ga-s, Ag-s and S-p states, too. Plasma frequency of the AgGaGeS₄ crystal is equal to about 19.5 eV. The electromagnetic waves shorter than 61 nm are weakly absorbed by AgGaGeS₄ and the crystal is almost transparent up to ultra-violet light.

Conflicts of interest

There are no conflicts to declare.

References

- 1 J. Šulc and H. Jelínková, *Solid-state lasers for medical applications, Lasers for medical applications*, Elsevier 2013, pp. 127–176.
- 2 D. Kochiev, A. Prokhorov, A. Lukashev, I. Hcherbakov and S. Vartapetov, *Surgical solid-state lasers and their clinical applications, Handbook of Solid-State Lasers*, Elsevier 2013, pp. 572–597.
- 3 V. Petrov, *Opt. Mater.*, 2012, **34**, 536–554.
- 4 H. Hügel, *Opt. Lasers Eng.*, 2000, **34**, 213–229.
- 5 K. Pribil and J. Flemmig, *Solid state laser communications in space (SOLACOS) high data rate satellite communication system verification program, Space Optics 1994, Space Instrumentation and Spacecraft Optics, SPIE, 1994*, pp. 39–48.
- 6 G. Edwards, R. Logan, M. Copeland, L. Reinisch, J. Davidson, B. Johnson, R. Maciunas, M. Mendenhall, R. Ossoff and J. Tribble, *Nature*, 1994, **371**, 416–419.
- 7 H. R. Aldag, *Solid state dye laser for medical applications, Visible and UV Lasers, SPIE 1994*, pp. 184–189.



- 8 L.-H. Li, J.-Q. Li and L.-M. Wu, *J. Solid State Chem.*, 2008, **181**, 2462–2468.
- 9 A. Godard, *C. R. Phys.*, 2007, **8**, 1100–1128.
- 10 J. Kim and T. Hughbanks, *Inorg. Chem.*, 2000, **39**, 3092–3097.
- 11 V. Dmitriviev, G. Gurzadyan and D. Nikogosyan, *Handbook of Nonlinear Crystals*, Springer, Berlin, 1999.
- 12 B. Levine, *Phys. Rev. B: Solid State*, 1973, **7**, 2600.
- 13 J. Sulc, H. Jelinkova, J. K. Jabczynski, W. Zendzian, J. Kwiatkowski, K. Nejezchleb and V. Skoda, *Comparison of diode-side-pumped triangular Nd: YAG and Nd: YAP laser, Solid State Lasers XIV: Technology and Devices*, SPIE 2005, pp. 325–334.
- 14 O. Guillot-Noël, B. Bellamy, B. Viana and D. Vivien, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **60**, 1668–1677.
- 15 T. Pollak, W. Wing, R. Grasso, E. Chicklis and H. Jenssen, *IEEE J. Quantum Electron.*, 1982, **18**, 159–163.
- 16 H. Rast, H. Caspers and S. Müller, *Phys. Rev.*, 1968, **169**, 705.
- 17 K. Regelskis, J. Želudevičius, N. Gavrilin and G. Račiukaitis, *Opt. Express*, 2012, **20**, 28544–28556.
- 18 X. Zhao, S. Zhu, B. Zhao, B. Chen, Z. He, R. Wang, H. Yang, Y. Sun and J. Cheng, *J. Cryst. Growth*, 2008, **311**, 190–193.
- 19 P. Canarelli, Z. Benko, A. Hielscher, R. Curl and F. Tittel, *IEEE J. Quantum Electron.*, 1992, **28**, 52–55.
- 20 G. Bhar and R. Smith, *IEEE J. Quantum Electron.*, 1974, **10**, 546–550.
- 21 G. Boyd, E. Buehler and F. Storz, *Appl. Phys. Lett.*, 1971, **18**, 301–304.
- 22 J. Petit, S. A. S. Hassani, P. Galtier, A. Godard, M. Raybaut and M. Lefebvre, *Progress in ZnGeP2 and AgGaS2 crystal growth: first results on difference-frequency generation and optical parametric oscillation, Optical Materials in Defence Systems Technology VI*, SPIE, 2009, pp. 132–139.
- 23 N.-H. Kim, D. H. Shin and R. S. Feigelson, *Mater. Sci. Eng., B*, 1996, **38**, 229–235.
- 24 R. Eckardt, Y. Fan, R. Byer, R. Route, R. Feigelson and J. Van der Laan, *Appl. Phys. Lett.*, 1985, **47**, 786–788.
- 25 W. Huang, Z. He, B. Zhao, S. Zhu and B. Chen, *J. Alloys Compd.*, 2019, **796**, 138–145.
- 26 W. Huang, Z. He, S. Zhu, B. Zhao, B. Chen and S. Zhu, *Inorg. Chem.*, 2019, **58**, 5865–5874.
- 27 P. G. Schunemann, K. T. Zawilski and T. M. Pollak, *J. Cryst. Growth*, 2006, **287**, 248–251.
- 28 Y. Andreev, P. Geiko, V. Badikov, G. Bhar, S. Das and A. Chaudhury, *MCLC S&T, Sect. B: Nonlinear Opt.*, 2002, **29**, 19–27.
- 29 K. Miyata, V. Petrov and K. Kato, *Appl. Opt.*, 2007, **46**, 5728–5731.
- 30 V. Petrov, V. Badikov, G. Shevyrdyaeva, V. Panyutin and V. Chizhikov, *Opt. Mater.*, 2004, **26**, 217–222.
- 31 M. Shevchuk, V. Atuchin, A. Kityk, A. Fedorchuk, Y. Romanyuk, S. Calus, O. Yurchenko and O. Parasyuk, *J. Cryst. Growth*, 2011, **318**, 708–712.
- 32 I. Olekseyuk, A. Gulyak, L. Sysa, G. Gorgut and A. Lomzin, *J. Alloys Compd.*, 1996, **241**, 187–190.
- 33 G. Y. Davydyuk, G. Myronchuk, G. Lakshminarayana, O. Yakymchuk, A. Reshak, A. Wojciechowski, P. Rakus, N. AlZayed, M. Chmiel and I. Kityk, *J. Phys. Chem. Solids*, 2012, **73**, 439–443.
- 34 O. Y. Khyzhun, O. V. Parasyuk and A. O. Fedorchuk, *Adv. Alloys Comp.*, 2013, **1**, 15–29.
- 35 P. Blaha, K. Schwarz, F. Tran, R. Laskowski, G. K. H. Madsen and L. D. Marks, *J. Chem. Phys.*, 2020, **152**, 074101.
- 36 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1997, **78**, 1396.
- 37 D. Koller, F. Tran and P. Blaha, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 155109.
- 38 V. I. Anisimov, I. V. Solovyev, M. A. Korotin, M. T. Czyżyk and G. A. Sawatzky, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1993, **48**, 16929–16934.
- 39 P. Novák, F. Boucher, P. Gressier, P. Blaha and K. Schwarz, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2001, **63**, 235114.
- 40 A. B. Shick, A. I. Liechtenstein and W. E. Pickett, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1999, **60**, 10763–10769.
- 41 E. R. Ylvisaker, W. E. Pickett and K. Koepernik, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **79**, 035103.
- 42 C. Ambrosch-Draxl and J. O. Sofo, *Comput. Phys. Commun.*, 2006, **175**, 1–14.
- 43 F. Wooten, *Am. J. Phys.*, 1973, **41**, 939–940.
- 44 C. Ambrosch-Draxl, J. A. Majewski, P. Vogl and G. Leising, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1995, **51**, 9668–9676.
- 45 A. Meisel, G. Leonhardt and R. Szargan, *X-Ray Spectra and Chemical Binding*, Springer-Verlag, Berlin/Heidelberg, 1989.
- 46 O. Y. Khyzhun, Y. V. Zaulychny and E. A. Zhurakovsky, *J. Alloys Compd.*, 1996, **244**, 107–112.
- 47 V. L. Bekenev, V. V. Bozhko, O. V. Parasyuk, G. E. Davydyuk, L. V. Bulatetska, A. O. Fedorchuk, I. V. Kityk and O. Y. Khyzhun, *J. Electron Spectrosc. Relat. Phenom.*, 2012, **185**, 559–566.
- 48 P. D. C. King, T. D. Veal, C. F. McConville, F. Fuchs, J. Furthmüller, F. Bechstedt, J. Schörmann, D. J. As, K. Lischka, H. Lu and W. J. Schaff, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2008, **77**, 115213.
- 49 I. D. Olekseyuk, A. V. Gulyak, L. V. Sysa, G. P. Gorgut and A. F. Lomzin, *J. Alloys Compd.*, 1996, **241**, 187–190.
- 50 R. E. Nikolaev and R. G. Vasilyeva, *J. Solid State Chem.*, 2013, **203**, 340–344.
- 51 W. Huang, Z. He, S. Zhu, B. Zhao, B. Chen and S. Zhu, *Inorg. Chem.*, 2019, **58**, 5865–5874.
- 52 M. Kavčič, J. C. Dousse, J. Szlachetko and W. Cao, *Nucl. Instrum. Methods Phys. Res. B*, 2007, **260**, 642–646.
- 53 U. Eiichirou, K. Jun and M. Uda, *Nucl. Instrum. Methods Phys. Res. B*, 1993, **75**, 24–27.
- 54 A. S. Krymus, I. V. Kityk, P. Demchenko, O. V. Parasyuk, G. L. Myronchuk, O. Y. Khyzhun and M. Piasecki, *Mater. Res. Bull.*, 2017, **95**, 177–184.

