


 Cite this: *RSC Adv.*, 2023, **13**, 28288

Novel epoxy-terminated macromonomers and their polymerization for synthesis of bottle-brush type amphiphilic block copolymers†

 Muhammad Imran Malik *

Architecture of polymers has vital implications for their physical properties and applications. In this study, synthesis of a series of novel epoxy-terminated macromonomers namely Ep-DEGMME, Ep-TEGMME, Ep-EGMEE, Ep-EGMBE, and Ep-EGMHE is reported. The synthesized macromonomers vary in number of ethylene oxide units and length of the alkyl group. These macromonomers are first homopolymerized by anionic ring-opening polymerization for synthesis of homopolymers of a molar mass range. Subsequently, these macromonomers with different lengths of two segments (alkyl group and ethylene oxide units) are copolymerized with other monomers for synthesis of bottle-brush type architectures. In the first case, di- and tri-block copolymers of Ep-EGMBE are synthesized while using MeO-PEG or PEG as a macroinitiator; the resulting block copolymers have hydrophilic handle and hydrophobic brush. On the same lines, block copolymers of Ep-TEGMME with ϵ -caprolactone have hydrophobic handle and hydrophilic brush. The synthesized block copolymers are comprehensively characterized by SEC and liquid chromatography at critical conditions. The analysis reveals the successful synthesis of block copolymers while providing information on relative total molar mass, and individual block lengths of the block copolymers, along with amount of unwanted homopolymers in the sample.

 Received 30th August 2023
 Accepted 21st September 2023

DOI: 10.1039/d3ra05912j

rsc.li/rsc-advances

1. Introduction

Polymers can have different architectures, such as linear, branched, grafts, star, cross-linked, brushes, and so forth.¹ Among various reported architectures of polymers, brushes are a promising class with special properties, for instance, rapid conformational changes under influence of external stimuli, such as pH, temperature, ionic strength, heat or light. Owing to low hydrodynamic radii of the branched polymers, their physical properties, both bulk and solution, are very different from similar molar masses of their linear analogues.^{2–4} Recently, an increased interest in polymer brushes has emerged in various fields such as tissue engineering, drug delivery, sensing, lubrication, *etc.*^{5–10} The performance properties of any material not only depend upon the parent constituents but also on their surfaces. Physical properties that are directly associated with the surfaces of materials include friction, wettability, adhesion, colloidal stability, lubrication, and biocompatibility.^{9,11–13}

Polymer brushes are usually synthesized by means of physisorption or covalent attachment. Physisorption methodology

often suffers from the solvent and thermal instability unlike covalent attachment. Covalent attachment is mainly accomplished through “grafting from” or “grafting to” methods.^{14,15} In “grafting to” approach, polymer bearing reactive end-functionality is synthesized that subsequently reacts with a complementary surface reactive group of the backbone. The major disadvantage of this strategy is low grafting density of the synthesized polymer brushes. The “grafting from” technique has developed as a versatile approach for the synthesis of high grafting density polymer brushes. This method involves an *in situ* polymerization of surface immobilized initiator layer with a monomer, and provides reliability, versatility and controllability. It can be employed with a variety of polymerization methods such as reversible addition fragmentation chain transfer (RAFT) polymerization, ring-opening polymerization (ROP), atom transfer radical polymerization (ATRP), and nitroxide-mediated polymerization (NMP).^{8,13,16–19} In this regard, ROP is a favorable method for synthesis of polymer brushes by using epoxides as a functional monomer.²⁰ Epoxides have several advantages compared to other monomers, such as ease of accessibility, comparatively simple synthesis of monomers, possibility of different functional groups on the epoxy ring, and high ring-strain of the three membered cyclic ether.^{21,22}

Glycidyl ethers have an epoxide ring that may be attached with different groups having distinct polar character. Their ROP subsequently produce polymer brushes with different pendant groups that lead to peculiar physical and chemical properties.²³

Third World Center for Science and Technology, H. E. J. Research Institute of Chemistry, International Center for Chemical and Biological Sciences (ICCBS), University of Karachi, Karachi 75270, Pakistan. E-mail: mimran.malik@iccs.edu; m.imran.malik1@gmail.com

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3ra05912j>



Homopolymers of glycidyl ethers with different compositions of the pendant groups should have quite different thermal and morphological properties. Block copolymers of these epoxy-terminated macromonomers with other monomers such as ethylene oxide, lactones, higher epoxide *etc.* shall induce amphiphilicity resulting in synthesis of amphiphilic block copolymers. Controlled delivery of hydrophobic drugs by their improved solubilization and stabilization through amphiphilic block copolymers is one of the hot research domains of modern day science.²⁴ A well-defined nano-sized, core/shell type architectures called micelles are formed by amphiphilic block copolymers in solvents having preferential solvation for one block compared to the other block. For instance, in water the hydrophilic segment self-assembles to make protective shell over the core of the hydrophobic block. This hydrophilic corona protected hydrophobic core of the micelles serves as a reservoir for transport of hydrophobic drugs to the target site.^{25,26} There are several variations that can directly impact the performance of amphiphilic block copolymers, such as overall molar mass, relative block length, and chemical composition.^{27–32}

Anionic ROP of cyclic monomers is a feasible method of polymerization of monomers having an epoxide ring, such as glycidyl ethers, by employing metal salts of potassium, sodium, and cesium that generate alkoxide species by deprotonation of a hydroxy bearing compound. The generated alkoxide group ultimately attacks carbon atom of oxirane to effectively open the ring and creates another oxyanion for propagation of polymerization.^{33–38}

Industrial surfactants and emulsifiers are generally low molar mass polymers synthesized by ring-opening polymerization of cyclic monomers. Ring-opening of epoxide is, by far, the most widely used approach in this context for their synthesis. Some of the important industrial products in this context are fatty acid ethoxylates, polysorbate ethoxylates, and poloxamers (EO–PO block copolymers) *etc.*³⁹

In this study, we report synthesis of novel epoxy-terminated macromonomers containing different segments of varying polarity such as alkyl group of varying length and different number of ethylene oxide (EO) units. These groups are introduced by reaction of different glycol ethers namely DEGMME, TEGMME, EGME, EGMEE, EGMBE, and EGMHE with epichlorohydrin. The synthesized macromonomers are named as epoxy-terminated DEGMME, epoxy-terminated TEGMME, epoxy-terminated EGME, epoxy-terminated EGMEE, epoxy-terminated EGMBE, and epoxy-terminated EGMHE. The success of the synthesis of macromonomers is evaluated by ¹H-NMR. The synthesized macromonomers are polymerized by anionic ROP using diethylene glycol (DEG) as an initiator and sodium hydride as a catalyst. Homopolymers are characterized by size exclusion chromatography, and ¹H-NMR. Finally, di- and tri-block copolymers are synthesized by using PEG and MeO-PEG as a macro-initiator while using Ep-EGMEE as a macromonomer. These block copolymers have a bottle-brush type structure having hydrophilic handle with hydrophobic brush. Tri-block copolymer synthesized by using PEG as macroinitiator have a dumbbell type structure with hydrophilic shaft in the center and bulky side groups. On the same lines, sequential anionic

polymerization of Ep-TEGMME and ϵ -caprolactone resulted in bottle-brush type architecture with hydrophobic handle and hydrophilic brush, Fig. 1. As typical examples, we focus on TEGMME and EGMEE for synthesis of block copolymers. Size exclusion chromatography is employed for relative molar mass analysis of the block copolymers. Moreover, liquid chromatography at critical conditions is employed at chromatographic critical point (CCP) of PCL and P(Ep-TEGMME) for determination of unwanted critical homopolymers in the sample along with individual block length of the non-critical block in the block copolymer.

2. Experimental section

2.1. Materials

Diethylene glycol (DEG, Scharlau), sodium hydride (NaH, Alfa Aesar), sodium hydroxide (NaOH, Merck), and epichlorohydrin (ECH) (Daejung) were used without further purification. Diethylene glycol monomethyl ether (DEGMME, Daejung), triethylene glycol monomethyl ether (TEGMME, Daejung), ethylene glycol monoethyl ether (EGMEE, Daejung), ethylene glycol monobutyl ether (EGMEE, Daejung), ethylene glycol monohexyl ether (EGMHE, Leap Chem) were used as received. Polyethylene glycol (PEG₄₀₀ Sigma-Aldrich, Germany), methoxy-polyethylene glycol (MeO-PEG₅₅₀, Sigma-Aldrich, Germany) were dried by azeotropic distillation over toluene prior to use. ϵ -Caprolactone (ϵ -CL, TCI, Japan) was dried over CaH₂ and distilled under reduced pressure. HPLC grade chloroform, acetone, and acetonitrile were purchased from Daejung, Korea.

2.2. Synthesis of epoxy-terminated macro-monomer

Glycol ethers (DEGMME, TEGMME, EGME, EGMEE, EGMBE, and EGMHE) and NaOH were placed in a three-neck round-bottom flask equipped with a bubbler, a thermometer, and a dropping funnel. The mixture was stirred to dissolve NaH. After a pre-decided time-interval, an excess amount of epichlorohydrin (ECH) was added drop-wise. The temperature in the reaction flask was kept below 40 °C by putting flask in cold water. After the addition of whole ECH stirring continued for two hours. Thereafter, the product mixture was dissolved in acetonitrile followed by filtration under vacuum to remove excess NaOH and NaCl. The solvent (CH₃CN) and unreacted ECH were removed on rotary evaporator under reduced pressure. The reaction of the synthesis of epoxy-terminated macromonomer is demonstrated in Scheme 1. The synthesized macromonomers vary in the length of EO block in a range of 1–3 units and the alkyl group (CH₃, C₂H₅, C₄H₉, and C₆H₁₃) depending upon the used glycol ether. Several different macromonomers are successfully synthesized namely epoxy-terminated DEGMME, epoxy-terminated TEGMME, epoxy-terminated EGME, epoxy-terminated EGMEE, epoxy-terminated EGMBE, and epoxy-terminated EGMHE.

2.3. Polymerizations

The synthesis of homopolymers by anionic ring-opening polymerization was conducted in a round bottom flask containing DEG as an initiator and sodium hydride as a catalyst along with



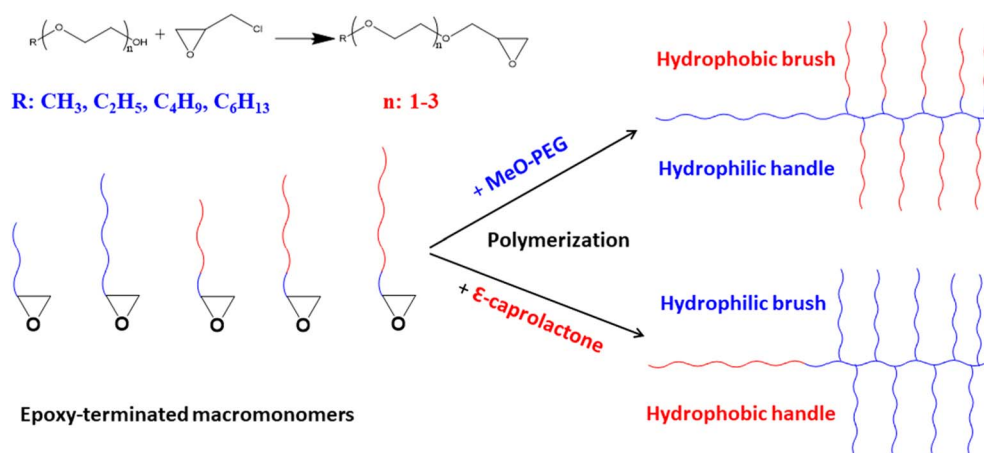
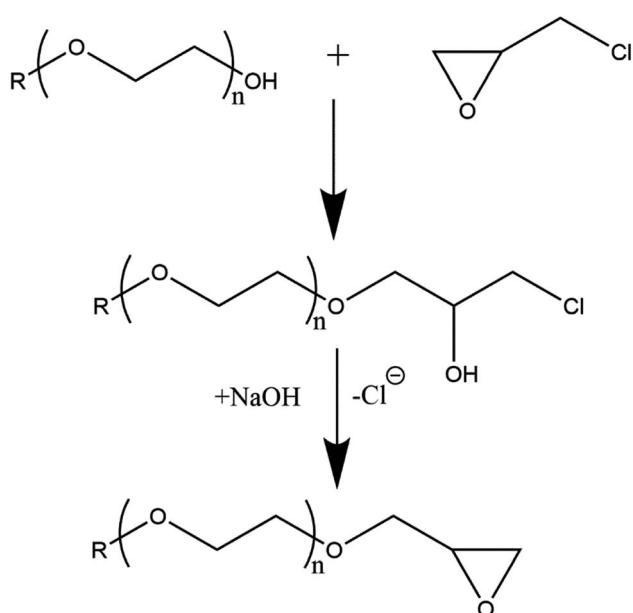


Fig. 1 Variety of epoxy-terminated macromonomers and their polymerization for synthesis of bottle-brush type architectures.



Scheme 1 Synthesis of epoxy terminated macromonomers; R: CH₃, C₂H₅, C₄H₉, C₆H₁₃, and n: 1–3.

respective epoxy-terminated macromonomer. For instance, calculated amounts of DEG, NaH, and Ep-TEGMME or Ep-EGMME were taken in flask that is subjected to three evacuation and nitrogen purging cycles followed by heating to 200 °C for synthesis of homopolymer brushes with different target molar masses. The amounts of the reagents used in the synthesis of different target molar masses of homopolymers are given in Table S1-ESI Material.† After polymerization, active chain-end was neutralized by adding equimolar amount of acetic acid.

First series of block copolymers was synthesized by anionic ROP of Ep-EGMME using MeO-PEG₅₅₀ and PEG₄₀₀ as a macroinitiator and NaH as a catalyst. Calculated amounts of the reagents were added in a round bottom flask that was subjected to three evacuation and nitrogen purging cycles. All the

polymerizations were carried out at 200 °C in an inert atmosphere. After, polymerization, active chain-end was neutralized by adding equimolar amount of acetic acid. The mass ratio of the macroinitiators and epoxy-terminated macromonomers was varied as per target length of the hydrophobic block while keeping same length of hydrophilic block in di- and tri-block copolymers, Table S2-ESI Material.†

For second series of block copolymers, different molar masses of the first block were prepared in a range of 500–3000 g mol⁻¹ by varying ratios of initiator and monomer as shown in Table S1-ESI Material.† In this case, TEGMME was used as initiator. P(Ep-TEGMME)_{2K} was selected for polymerization of ε-caprolactone. The synthesized P(Ep-TEGMME)_{2K} serves as a macroinitiator and initial ratios of the macroinitiator and ε-caprolactone (monomer) were varied in order to have different block length of the hydrophobic polycaprolactone while having same length of P(Ep-TEGMME), Table S3-ESI Material.† After completion of polymerization of Ep-TEGMME, the temperature of flask was reduced to 140 °C and calculated amount of ε-caprolactone was added. Three different block copolymers were synthesized having same length of the hydrophilic P(Ep-TEGMME) block (2000 g mol⁻¹) and different length of the PCL block. After, polymerization, active chain end was neutralized by adding equimolar amount of acetic acid. The products are named as P(Ep-TEGMME)_{2K}-b-PCL (3–1), P(Ep-TEGMME)_{2K}-b-PCL (1–1), P(Ep-TEGMME)_{2K}-b-PCL (1–3), where number in parenthesis represent the initial mass ratio of the macro-initiator [P(Epoxy-TEGMME)_{2K}] and the monomer (ε-caprolactone).

2.4. Size exclusion chromatography (SEC)

An Agilent SEC instrument (infinity 1200 series), equipped with an isocratic pump, a manual injector, a column oven and a RI detector, was used for SEC analysis. A set of three PLgel columns (300 × 7.5 mm, 5 μm, Mixed D) connected in front with a PLgel guard column (50 × 7.5 mm) were placed in a column oven maintained at 30 °C. Chloroform (HPLC grade) was used as an eluent at a flow rate of 1.0 mL min⁻¹. The SEC system was calibrated with narrow molar mass polystyrene standards



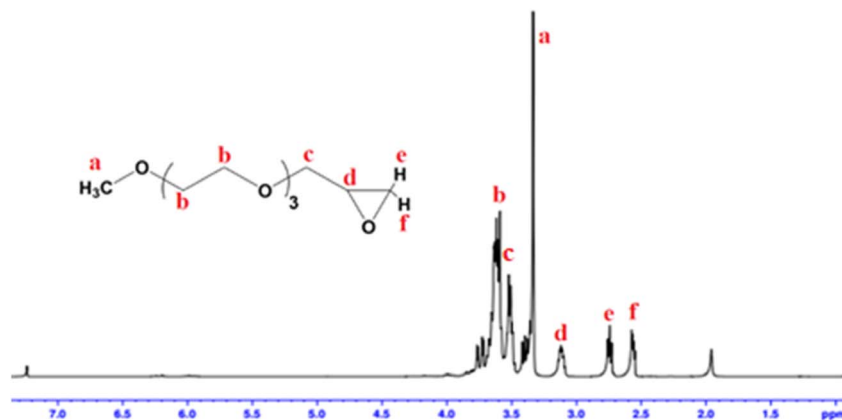


Fig. 2 ^1H -NMR spectrum of Ep-TEGMME.

(ranging from 165 g mol^{-1} to $3 \times 10^6 \text{ g mol}^{-1}$) procured from Polymer Standard Services (Germany). Therefore, all the molar mass values obtained by SEC analysis are relative to linear PS. The data was processed by Agilent OpenLAB ChemStation. Sample concentrations were kept between 3.0 and 5.0 mg mL^{-1} , and $100 \mu\text{L}$ of sample solution was injected.

2.5. Nuclear magnetic resonance spectroscopy (NMR)

^1H -NMR spectra of epoxy-terminated macromonomers and homopolymer were recorded using Bruker AVANCE AN-500 MHz spectrometer. Tetramethylsilane was used as an internal standard and analytes were dissolved in CDCl_3 . TopSpin 3.6.1 software was used for data processing.

2.6. Liquid chromatography at critical conditions (LCCC)

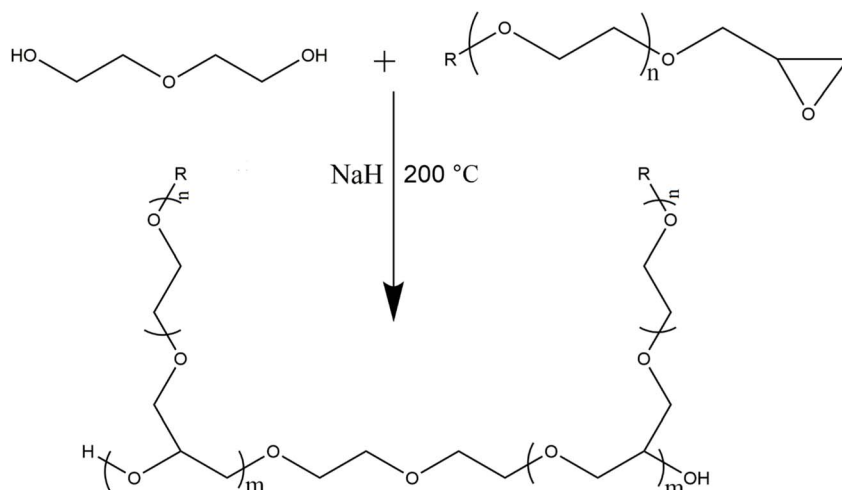
Liquid chromatographic analyses at critical conditions were performed at an Agilent technologies chromatograph (infinity 1200 series). The system is equipped with a binary pump, a column oven, an auto-sampler, and an evaporative light

scattering detector. The column oven was maintained at $30 \text{ }^\circ\text{C}$. The temperature of ELSD was set at $40 \text{ }^\circ\text{C}$ and nitrogen was used as carrier gas at a pressure of 50 psi . The mobile phase was delivered at 0.5 mL min^{-1} . Sample concentrations were kept between 3.0 and 10.0 mg mL^{-1} , and $20 \mu\text{L}$ of sample solution was injected. The data was processed by Agilent Open LAB ChemStation for LC system. Following columns were used in this study.

- ZORBAX ODS, pore size 70 \AA , particle size 5 mm , and column dimensions $250 \times 4.6 \text{ mm}$ (Phenomenex, USA).
 - Ultisil diol column, pore size 120 \AA , particle size 5 mm , and column dimensions $250 \times 4.6 \text{ mm}$ (Welch, China).
- HPLC grade acetone, acetonitrile, and double deionized water were used for preparation of mobile phases. Solvents were mixed by volume and vacuum degassed.

3. Results & discussion

Macromonomers with epoxide ring attached to different groups varying in number of EO repeat units and length of alkyl chains



Scheme 2 General reaction scheme of polymerization of epoxy-terminated macromonomers where R: CH_3 , C_2H_5 , C_4H_9 , C_6H_{13} , n : $1-3$, and m : degree of polymerization.



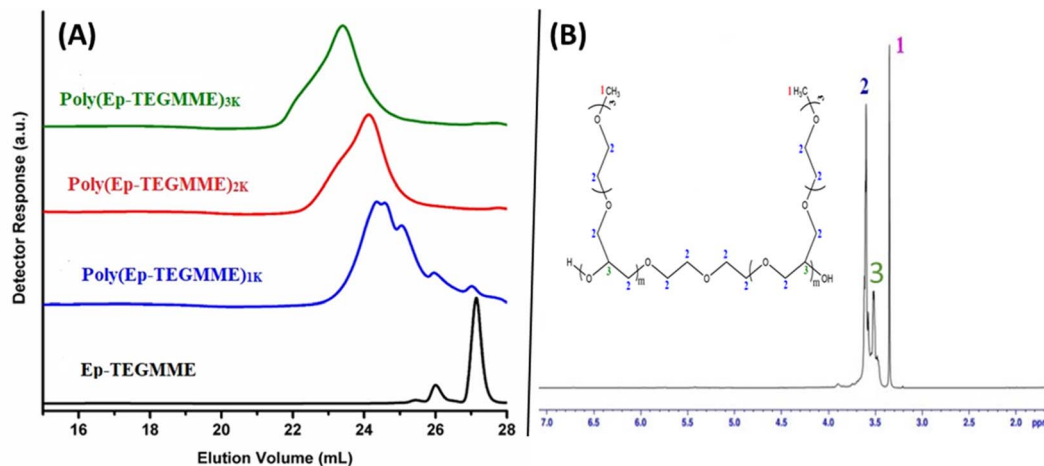


Fig. 3 P(Ep-TEGMME) (A) SEC chromatograms of homopolymers with different target molar masses on a set of three PLgel columns (300 × 7.5 mm, 5 μm, Mixed D) and PLgel guard column (50 × 7.5 mm) in CHCl_3 at flow rate of 1.0 mL min^{-1} ; (B) $^1\text{H-NMR}$ spectrum of P(Ep-TEGMME) $_{2\text{K}}$.

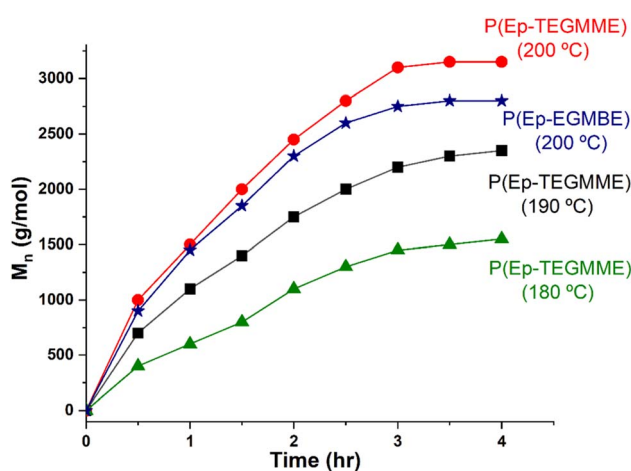


Fig. 4 Propagation of molar mass as a function of time at different polymerization temperatures.

induce peculiar properties to their homo- and block copolymers. First target in this study was the development of a method for synthesis of macromonomers that allows to tailor their structure for different lengths of the variable segments.

3.1. Epoxidation of glycol ethers

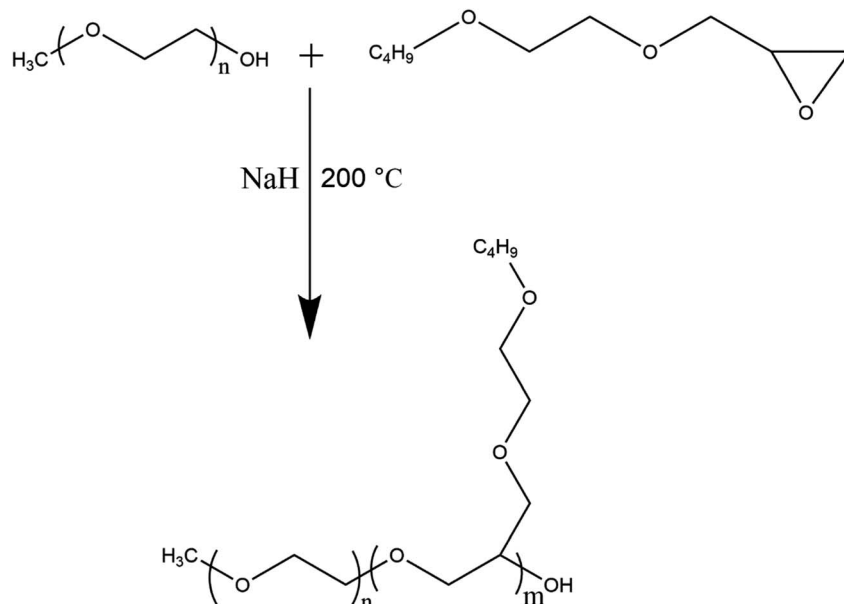
Epoxide ring is introduced to a series of glycol ethers namely DEGME, TEGMME, EGME, EGMME, and EGMHE through their reaction with epichlorohydrin in order to synthesize a novel library of macromonomers, Scheme 1. $^1\text{H-NMR}$ spectroscopy is employed for the confirmation of formation of epoxy-terminated macromonomer. As a typical example, $^1\text{H-NMR}$ spectrum of Ep-TEGMME is presented in Fig. 2. The peaks in a range of 2.5–3.2 ppm (peaks d, e, and f) correspond to the protons of the epoxide ring. It is pertinent to mention that protons “e” and “f” have same chemical environment but they are showing different chemical shifts because they are diastereotopic protons and have different chiral environment. A peak

at 3.3 and 3.7 ppm corresponds to methyl proton (a) and protons of three EO repeat units (b), respectively. The low intensity peak at 3.6 ppm corresponds to the proton attached to the epoxide group. Hence, all the peaks are justified and indicate successful synthesis of Ep-terminated macromonomers. It is pertinent to mention here that yields in all cases were above 90%.

3.2. Homopolymers of epoxy-terminated macromonomers

Polymerization of the synthesized macromonomers is conducted to make homopolymers having peculiar properties. As mentioned earlier sodium ion generate alkoxide species by deprotonation terminal hydroxyl groups of DEG or any other compound containing terminal hydroxyl group. The generated alkoxide group ultimately attacks carbon atom of oxirane to effectively open the ring and creates another oxyanion for propagation of polymerization.^{33–38} A general scheme of polymerization of the synthesized epoxy-terminated macromonomers is presented in Scheme 2. In the scheme R and n correspond to the alkyl group and number of ethylene oxide units in the macromonomer while m refers to the degree of polymerization. Generally, ROP is employed for synthesis of low molar mass polymers and achieving very high molar masses is practically not possible. Size exclusion chromatography (SEC), the primary technique for analysis of molar mass and its distributions, is employed for analysis of the synthesized polymers.^{40,41} Different target molar masses of homopolymers are synthesized by varying the molar feed ratio of monomer-to-initiator (M/I). SEC elograms of three products of P(Ep-TEGMME) of different target molar masses, *i.e.*, 1000 g mol^{-1} (1 K), 2000 g mol^{-1} (2 K), and 3000 g mol^{-1} (3 K) are shown in Fig. 3A. Separation in SEC is based on the hydrodynamic size of the polymer molecules in dilute solution, the elution volumes are subsequently compared with polymer standards of known molar mass in order to obtain relative molar mass of the analyzed polymers. Retention volumes of the synthesized polymers decrease with increase in the molar mass as per





Scheme 3 Reaction scheme for synthesis of MeO-PEG₅₅₀-*b*-P(Ep-EGMBE).

expectation based on the initial monomer-to-initiator ratio. Molar mass data of the homopolymers as obtained by SEC using PS calibration is given in Table S4, ESI Material.† Moreover, ¹H-NMR spectroscopy confirmed the synthesis of homopolymers through disappearance of signals around 2.6, 2.8, and 3.2 ppm that are attributed to the protons of the oxirane ring, Fig. 3B.

Different experimental parameters such as amount of catalyst, monomer-to-initiator ratio, and polymerization temperature are optimized. At the optimized conditions, the augmentation in the molar mass as a function of time at various temperatures for P(Ep-TEGMME) is presented in Fig. 4. The molar mass of the P(Ep-TEGMME) increased with polymerization times at different rates while polymerizing at different temperatures. A reasonable propagation in the molar mass is achieved for reaction at 200 °C. Same polymerization temperature is found to be rational for synthesis of P(Ep-EGMBE).

3.3. Block copolymers of epoxy-terminated macromonomers

Sequential polymerization of epoxy-terminated macromonomers with other monomers produces block copolymers with unique architectures. Herein, we synthesized two types of block copolymers rendering two unique bottle-brush type architectures. The first type has hydrophilic handle and hydrophobic brush while the second type has hydrophobic handle and hydrophilic brush.

3.3.1. Block copolymers having hydrophilic handle and hydrophobic brush. In the first series of block copolymers, PEG₄₀₀ and MeO-PEG₅₅₀ are used as macroinitiator while polymerizing Ep-EGMBE. Di- and tri-block copolymers are synthesized having a bottle-brush like architecture with hydrophilic handle and hydrophobic brush. The reaction scheme for synthesis of diblock copolymer initiated with MeO-PEG₅₅₀ is presented in Scheme 3. Triblock copolymers synthesized by

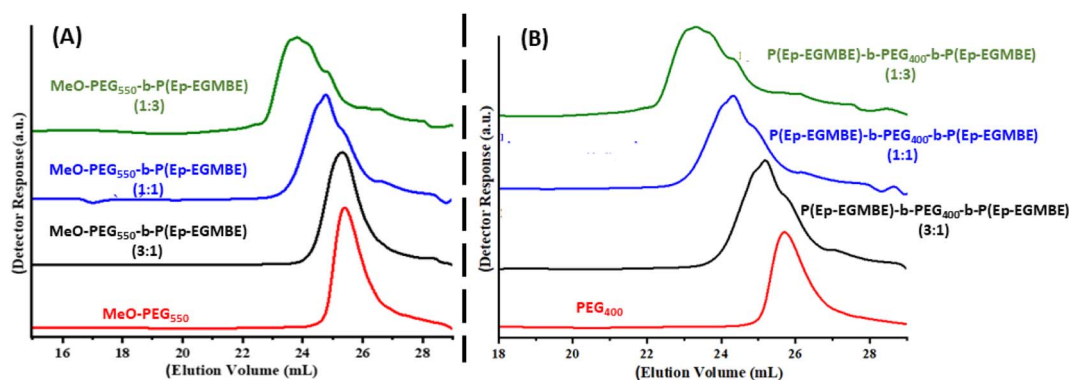
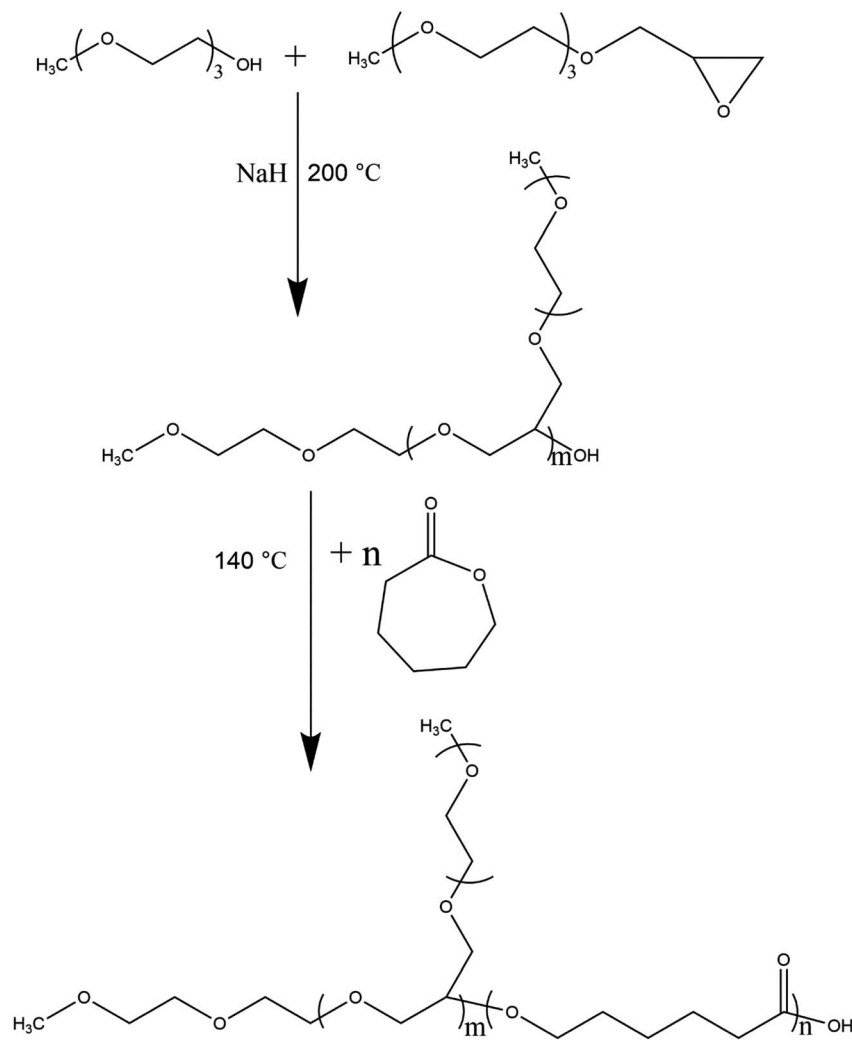


Fig. 5 Size exclusion chromatography on a set of three PLgel columns (300 × 7.5 mm, 5 μm, Mixed D) and PLgel guard column (50 × 7.5 mm) in chloroform (CHCl₃) at flow rate of 1.0 mL min⁻¹, (A) MeO-PEG₅₅₀-*b*-P(Ep-EGMBE), (B) P(Ep-EGMBE)-*b*-PEG₄₀₀-*b*-P(Ep-EGMBE); numbers in parenthesis represent initial feed ratios of macro-initiator and monomer in mass.





Scheme 4 Reaction scheme for synthesis of $P(\text{Ep-TEGMME})_{2K}\text{-b-PCL}$ by sequential addition of monomers.

using PEG_{400} would have similar reaction scheme as presented for homopolymer (Scheme 2). The only difference would be higher number of ethylene oxide units in the center. The synthesized triblock copolymers have a dumbbell type architecture. Synthesized di- and tri-block copolymers of Ep-EGMBE are characterized by size exclusion chromatography (SEC) for augmentation in molar mass. SEC elugrams of di- and tri-block copolymers having different ratios of monomer-to-macroinitiator are demonstrated in Fig. 5. As can be noticed, elution volume of the block copolymers decreased with increase in the monomer (EGMBE) content in the initial feed. This is an expected behavior and indicates the formation of the block copolymers. Molar mass averages and dispersity (D) of the synthesized di- and tri-block copolymers using PS calibration curve are listed in Tables S5 and S6-ESI Material.†

3.3.2. Block copolymers having hydrophobic handle and hydrophilic brush. The second series of block copolymer with bottle-brush like architecture have hydrophobic handle and hydrophilic brush, synthesized through sequential addition of monomers. Epoxy-terminated triethylene glycol monomethyl ether

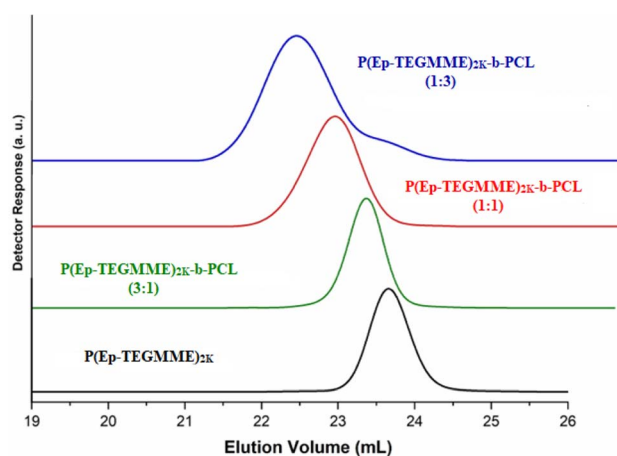


Fig. 6 Size exclusion chromatography of $P(\text{Ep-TEGMME})_{2K}\text{-b-PCL}$ on a set of three PLgel columns ($300 \times 7.5 \text{ mm}$, $5 \mu\text{m}$, Mixed D) and PLgel guard column ($50 \times 7.5 \text{ mm}$) in chloroform (CHCl_3) at flow rate of 1.0 mL min^{-1} ; numbers in parenthesis represent initial feed ratios of macro-initiator and monomer in mass.



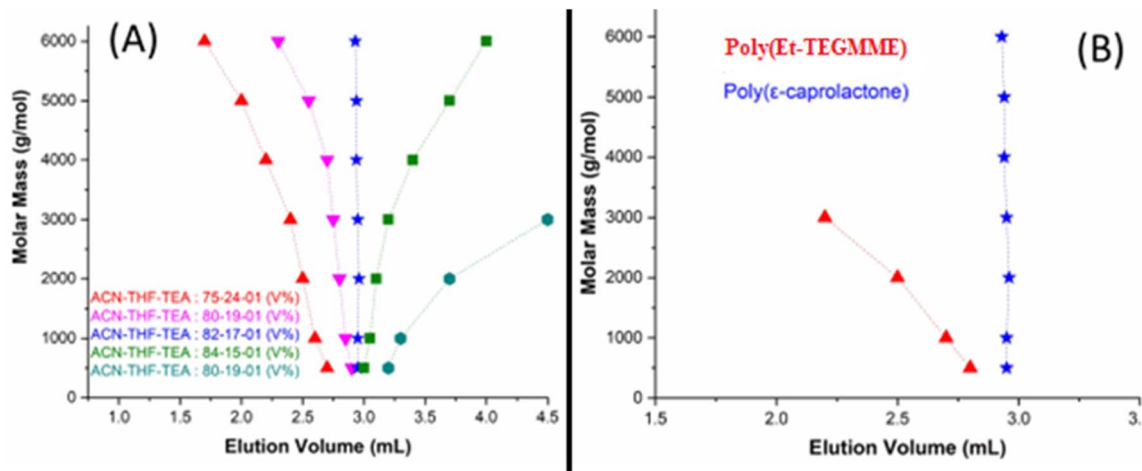


Fig. 7 (A) Elution behavior of different molar masses of PCL as a function of the MP composition (ACN/THF/TEA); (B) elution behavior of P(Ep-TEGMME) at the CCP of PCL: mobile phase: ACN/THF/TEA: 82/17/01(V%), column: ZORBAX ODS, pore size 70 Å, particle size 5 mm, and column dimensions 250 × 4.6 mm (Phenomenex, USA), detector: ELS, column temperature: 30 °C.

(Ep-TEGMME) is first polymerized to achieve different molar masses in a range of 500–3000 g mol⁻¹ at the optimized polymerization conditions using TEGMME as an initiator. Different molar masses of P(Ep-TEGMME) are used for establishment of the chromatographic critical point (CCP) of P(Ep-TEGMME). P(Ep-TEGMME)_{2K} is selected as a macroinitiator for synthesis of its block copolymer with ε-caprolactone, Scheme 4. The mass ratios of macroinitiator and ε-caprolactone are varied systematically to have different block lengths of the hydrophobic PCL block attached to the same length of the hydrophilic block [P(Ep-TEGMME)_{2K}]. SEC

elution volume of the block copolymers decreased systematically compared to macroinitiator with an increase in the ε-caprolactone content of the initial feed, Fig. 6. SEC elution order was P(Ep-TEGMME)_{2K}-b-PCL (1-3), P(Ep-TEGMME)_{2K}-b-PCL (1-1), and P(Ep-TEGMME)_{2K}-b-PCL (3-1) followed by P(Ep-TEGMME)_{2K} (the macroinitiator), as expected. Molar mass averages and dispersity (*D*) of the synthesized P(Ep-TEGMME)_{2K}-b-PCL using PS calibration curve are listed in Table S7-ESI Material.†

The next relevant question for any block copolymer is the connectivity of two blocks to each other and the individual

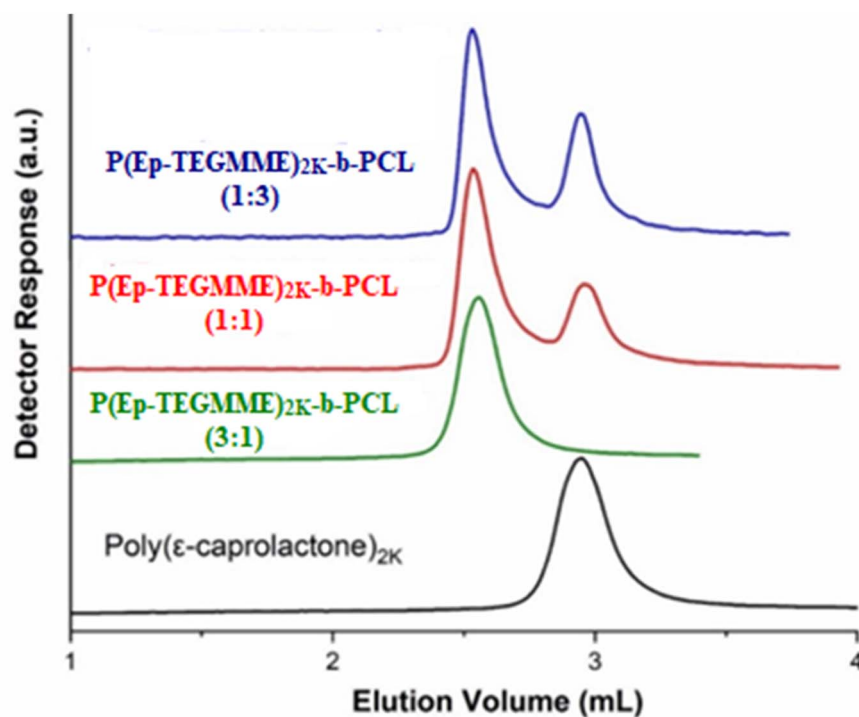


Fig. 8 P(Ep-TEGMME)_{2K}-b-PCL at LCCC-PCL, mobile phase ACN/THF/TEA: 82/17/01 (V%), column: ZORBAX ODS, pore size 70 Å, particle size 5 mm, and column dimensions 250 × 4.6 mm (Phenomenex, USA).

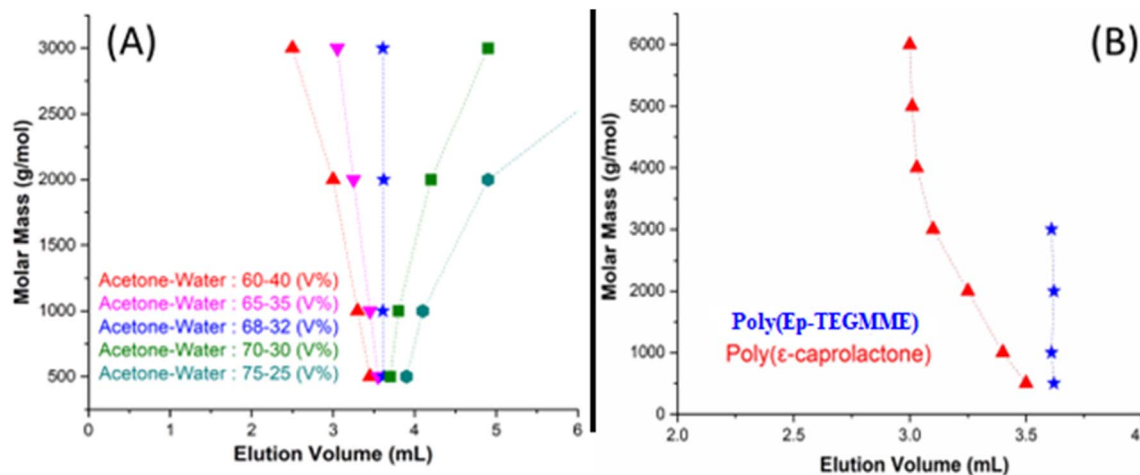


Fig. 9 (A) Elution behavior of different molar masses of P(Ep-TEGMME) as a function of the MP composition (acetone/water); (B) elution behavior of PCL at the CCP of P(Ep-TEGMME), MP: acetone/water: 68/32 (V%), column: Ultisil diol column, 120 Å, 5 μm, 250 × 4.6 mm (Welch, China), detector: ELS, column temperature: 30 °C.

block lengths. Moreover, determination of presence/absence of any unwanted homopolymers in the sample is imperative which can impede the final application properties. Liquid chromatography at critical conditions (LCCC) is literally the only technique that can reveal this important information.^{36,42–51} The approach requires establishment of chromatographic critical point (CCP) of each block that allows exclusion of the non-critical block.^{52–54}

In this context, CCP of PCL is established on a RP ODS column in a mobile phase (MP) constituting ACN and THF. In the ACN/THF MP, 1% of triethylamine (TEA) is added in order to avoid unwanted polar interactions of the polar groups in the polymer chains with free silanol groups of the stationary phase (SP).⁴⁶ The eluotropic strength of THF is higher on RP columns compared to ACN. In a MP composition of ACN/THF/TEA: 75/25/01 (V%), PCL eluted in order of decreasing molar mass, a typical SEC regime (Fig. 7A). A decrease of THF content to 19% resulted in a shift of elution volumes to higher value, however, elution order remained the same. An additional decrease in the THF content of MP to 15% not only shifted elution volume further but also reversed the elution order, PCL eluted in order of increasing molar mass in a typical IC regime. There must be a MP composition between two opposing elution orders where all PCL should elute as a sharp peak near the void volume of the column. This is indeed the case in a MP composition of ACN/THF/TEA: 82/17/01 (V%), so-called CCP of PCL on RP column under current conditions. Under these conditions, P(Ep-TEGMME) eluted in order of decreasing molar mass earlier than the void volume of the column, Fig. 7B. Hence, the elution volume of P(Ep-TEGMME)_{2K}-b-PCL represents the block length of P(Ep-TEGMME) with minimal effect of the PCL block. Furthermore, PCL homopolymers get separated from the rest of the sample at its CCP, Fig. 8. Same macroinitiator P(Ep-TEGMME)_{2K} is used for synthesis of block copolymers, therefore, elution of block copolymer under conditions of LCCC-PCL is not affected and all three block copolymers under study have the similar elution volume indicating the same length of P(Ep-

TEGMME) block. However, the content of unwanted PCL homopolymers has increased with the increase in the ε-caprolactone content of the initial feed. The increase in the initial monomer content (ε-caprolactone) will effectively increase the length of PCL block of block copolymer whose effect is nullified at its CCP. Hence, analysis at LCCC-PCL reveal similar length of P(Ep-TEGMME) block and an increase in the PCL homopolymer content of the products with higher monomer ratio in the initial feed. It is pertinent to mention that free P(Ep-TEGMME) coelute with the block copolymers and analysis at LCCC-P(Ep-TEGMME) is required for its determination.

The analysis of P(Ep-TEGMME)_{2K}-b-PCL samples by size exclusion chromatography showed an expected increase in the hydrodynamic volume that matches well with the amount of initial ε-caprolactone content. Moreover, analysis at LCCC-PCL on RP column indicate the similar length of P(Ep-TEGMME) block along with the PCL homopolymer content of the sample. It is however not possible to unequivocally decide about presence of free macroinitiator, P(Ep-TEGMME)_{2K}, in the sample and the individual lengths of the PCL block. LCCC at CCP of P(Ep-TEGMME) is required to obtain this information. P(Ep-TEGMME) is a polar polymer compared to PCL, therefore, at its CCP on a NP column PCL should be excluded from the pores of the SP.⁴⁵

On a NP SP, more polar component of the MP has higher eluotropic strength compared to less polar component. In this case, a binary mixture of acetone and water is used.⁵⁵ Obviously, the eluotropic strength of water is higher compared to acetone. In a MP composed of 75% acetone in water, P(Ep-TEGMME) have stronger interaction with the SP and eluted in order of increasing molar mass after the void volume of the column, Fig. 9A. The interaction strength decreased considerably while using 70% acetone but the elution order remained the same, higher molar masses eluted later than lower molar masses. Further decrease of acetone content to 65% resulted in reversal of the elution order, higher molar masses eluted earlier than the lower molar masses in a typical SEC order. In between these two



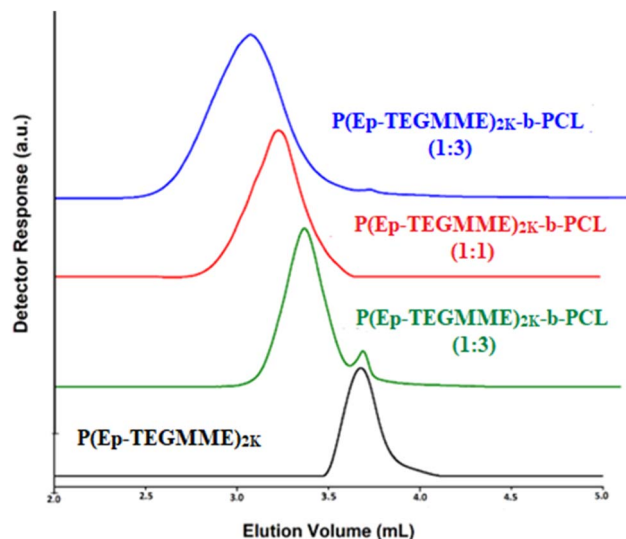


Fig. 10 P(Ep-TEGMME)_{2K}-b-PCL at LCCC-P(Ep-TEGMME), MP: acetone/water: 68/32 (V%), column: Ultisil diol column, 120 Å, 5 μm, 250 × 4.6 mm (Welch, China), detector: ELS, column temperature: 30 °C.

opposing elution orders, there is a point at 68% acetone where P(Ep-TEGMME) of a molar mass range 500–3000 eluted at the same elution volume irrespective of its molar mass. This is the CCP of P(Ep-TEGMME) under current conditions. At the CCP of P(Ep-TEGMME), PCL eluted in a typical SEC regime earlier than the void volume of the column, Fig. 9B. These conditions can be used for estimation of the individual block length of PCL in P(Ep-TEGMME)-b-PCL along with separation of free P(Ep-TEGMME) from the rest of the sample. The analysis of the synthesized P(Ep-TEGMME)-b-PCL at LCCC of P(Ep-TEGMME) is demonstrated in Fig. 10. The elution volume of block copolymers decreased with the increase in the initial ε-caprolactone content which indicates an increase in the individual PCL block length. The elution volumes correspond well with the calibration curve of PCL under same conditions. It is important to note that generally not much free P(Ep-TEGMME)_{2K} are present in the block copolymer samples. A small amount of free P(Ep-TEGMME)_{2K} is noticed in P(Ep-TEGMME)_{2K}-b-PCL (3-1) which may be attributed to excess of P(Ep-TEGMME)_{2K} compared to ε-caprolactone in the initial feed.

Hence, the analysis of P(Ep-TEGMME)_{2K}-b-PCL (3-1) by SEC, LCCC-PCL, LCCC-P(Ep-TEGMME) revealed total relative molar mass of macroinitiator and the block copolymers, length of P(Ep-TEGMME) and PCL blocks in the block copolymers along with the content of free homopolymers of both types in the sample. The physical properties and applications of these fascinating architectures of amphiphilic block copolymers will be topic of our next publication of this series.

4. Conclusion

In this study, a facile and convenient method of tailoring of epoxy-terminated macromonomers is reported. ¹H-NMR analysis of macromonomers confirm the successful synthesis

of the macromonomers. Size exclusion chromatography of the synthesized homopolymers of novel epoxy-terminated macromonomers indicate the success of polymerization by increase in the hydrodynamic volume. Moreover, absence of protons of oxirane ring in ¹H-NMR spectrum of the polymers confirm the ring-opening polymerization. Amphiphilic block copolymer brushes with hydrophilic handle and hydrophobic brush as well as with hydrophobic handle and hydrophilic brush are synthesized by copolymerization of epoxy-terminated macromonomers with other appropriate monomers. The analysis of bottle-brush like architectures of amphiphilic block copolymers by liquid chromatography at critical conditions revealed individual block lengths and homopolymer content of the samples.

Data availability

Data will be made available on request.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The research is supported by Higher Education Commission of Pakistan under “National Research Program for Universities” against project # 5735.

References

- 1 L. Y. Qiu and Y. H. Bae, *Pharm. Res.*, 2006, **23**, 1–30.
- 2 M. W. Neiser, S. Muth, U. Kolb, J. R. Harris, J. Okuda and M. Schmidt, *Angew. Chem., Int. Ed.*, 2004, **43**, 3192–3195.
- 3 H.-i. Lee, J. Pietrasik, S. S. Sheiko and K. Matyjaszewski, *Prog. Polym. Sci.*, 2010, **35**, 24–44.
- 4 A. L. Liberman-Martin, C. K. Chu and R. H. Grubbs, *Macromol. Rapid Commun.*, 2017, **38**, 1700058.
- 5 C. Feng and X. Huang, *Acc. Chem. Res.*, 2018, **51**, 2314–2323.
- 6 T. Chen, I. Amin and R. Jordan, *Chem. Soc. Rev.*, 2012, **41**, 3280–3296.
- 7 S. Peng and B. Bhushan, *RSC Adv.*, 2012, **2**, 8557–8578.
- 8 N. D. Ogbonna, M. Dearman, C.-T. Cho, B. Bharti, A. J. Peters and J. Lawrence, *JACS Au*, 2022, **2**, 898–905.
- 9 R. Verduzco, X. Li, S. L. Pesek and G. E. Stein, *Chem. Soc. Rev.*, 2015, **44**, 2405–2420.
- 10 A. Chremos and P. E. Theodorakis, *ACS Macro Lett.*, 2014, **3**, 1096–1100.
- 11 W. J. Brittain and S. Minko, *J. Polym. Sci., Part A: Polym. Chem.*, 2007, **45**, 3505–3512.
- 12 O. Azzaroni, *J. Polym. Sci., Part A: Polym. Chem.*, 2012, **50**, 3225–3258.
- 13 W.-L. Chen, R. Cordero, H. Tran and C. K. Ober, *Macromolecules*, 2017, **50**, 4089–4113.
- 14 J. A. Johnson, Y. Y. Lu, A. O. Burts, Y. Xia, A. C. Durrell, D. A. Tirrell and R. H. Grubbs, *Macromolecules*, 2010, **43**, 10326–10335.



- 15 J. A. Johnson, Y. Y. Lu, A. O. Burts, Y.-H. Lim, M. G. Finn, J. T. Koberstein, N. J. Turro, D. A. Tirrell and R. H. Grubbs, *J. Am. Chem. Soc.*, 2011, **133**, 559–566.
- 16 T. Hirai, M. Sato, T. Kato, H. Shimamoto, K. Uno, N. Ootzawa and A. Takahara, *Polym. J.*, 2018, **50**, 931–939.
- 17 R. Ducker, A. Garcia, J. Zhang, T. Chen and S. Zauscher, *Soft Matter*, 2008, **4**, 1774–1786.
- 18 S. S. Sheiko, B. S. Sumerlin and K. Matyjaszewski, *Prog. Polym. Sci.*, 2008, **33**, 759–785.
- 19 Y. L. Kamble, D. J. Walsh and D. Guironnet, *Macromolecules*, 2022, **55**, 10255–10263.
- 20 C. G. Rodriguez, R. C. Ferrier, A. Helenic and N. A. Lynd, *Macromolecules*, 2017, **50**, 3121–3130.
- 21 Y. Xia and J. Zhao, *Polymer*, 2018, **143**, 343–361.
- 22 M. I. Childers, J. M. Longo, N. J. Van Zee, A. M. LaPointe and G. W. Coates, *Chem. Rev.*, 2014, **114**, 8129–8152.
- 23 P. Verkoyen, T. Johann, J. Blankenburg, C. Czych and H. Frey, *Polym. Chem.*, 2018, **9**, 5327–5338.
- 24 P. Verkoyen and H. Frey, *Macromol. Rapid Commun.*, 2020, **41**, 2000225.
- 25 G. Gaucher, M.-H. Dufresne, V. P. Sant, N. Kang, D. Maysinger and J.-C. Leroux, *J. Controlled Release*, 2005, **109**, 169–188.
- 26 K. Kataoka, A. Harada and Y. Nagasaki, *Adv. Drug Delivery Rev.*, 2001, **47**, 113–131.
- 27 S. B. Darling, *Prog. Polym. Sci.*, 2007, **32**, 1152–1204.
- 28 I. W. Hamley, *Block Copolymers in Solution: Fundamentals and Applications*, John Wiley & Sons, Ltd, Chichester, England, 2005.
- 29 S. Zhou, X. Deng and H. Yang, *Biomaterials*, 2003, **24**, 3563–3570.
- 30 A. S. Mikhail and C. Allen, *Biomacromolecules*, 2010, **11**, 1273–1280.
- 31 X. Shuai, H. Ai, N. Nasongkla, S. Kim and J. Gao, *J. Controlled Release*, 2004, **98**, 415–426.
- 32 M. L. Adams, A. Lavasanifar and G. S. Kwon, *J. Pharm. Sci.*, 2003, **92**, 1343–1355.
- 33 S. Paul, Y. Zhu, C. Romain, R. Brooks, P. K. Saini and C. K. Williams, *Chem. Commun.*, 2015, **51**, 6459–6479.
- 34 A.-L. Brocas, C. Mantzaridis, D. Tunc and S. Carlotti, *Prog. Polym. Sci.*, 2013, **38**, 845–873.
- 35 M. I. Malik, B. Trathnigg and C. O. Kappe, *Eur. Polym. J.*, 2009, **45**, 899–910.
- 36 M. I. Malik, B. Trathnigg and C. O. Kappe, *J. Chromatogr. A*, 2009, **1216**, 1167–1173.
- 37 M. I. Malik, B. Trathnigg and C. O. Kappe, *Eur. Polym. J.*, 2008, **44**, 144–154.
- 38 M. I. Malik, B. Trathnigg and C. O. Kappe, *Macromol. Chem. Phys.*, 2007, **208**, 2510–2524.
- 39 J. Herzberger, K. Niederer, H. Pohlit, J. Seiwert, M. Worm, F. R. Wurm and H. Frey, *Chem. Rev.*, 2016, **116**, 2170–2243.
- 40 A. M. Striegel, *Multiple detection in size-exclusion chromatography*, ACS Publications, 2004.
- 41 A. Lederer and J. Brandt, in *Molecular Characterization of Polymers*, ed. M. I. Malik, J. Mays and M. R. Shah, Elsevier, 2021, pp. 61–96, DOI: [10.1016/B978-0-12-819768-4.00012-9](https://doi.org/10.1016/B978-0-12-819768-4.00012-9).
- 42 M. I. Malik and H. Pasch, in *Molecular Characterization of Polymers*, ed. M. I. Malik, J. Mays and M. R. Shah, Elsevier, 2021, pp. 1–59, DOI: [10.1016/B978-0-12-819768-4.00007-5](https://doi.org/10.1016/B978-0-12-819768-4.00007-5).
- 43 M. I. Malik, *Chromatographia*, 2021, **84**, 1089–1094.
- 44 T. Ahmad, S. S. Durrani, F. J. Iftikhar, S. G. Musharraf and M. I. Malik, *Eur. Polym. J.*, 2021, **142**, 110161.
- 45 M. I. Malik, *J. Chromatogr. A*, 2020, **1609**, 460440.
- 46 R. Khatoon, S. Rahim, R. Abdul-Karim, S. G. Musharraf and M. I. Malik, *Macromolecules*, 2019, **52**, 7688–7695.
- 47 M. I. Malik and H. Pasch, *Prog. Polym. Sci.*, 2014, **39**, 87–123.
- 48 M. I. Malik, G. W. Harding and H. Pasch, *Anal. Bioanal. Chem.*, 2012, **403**, 601–611.
- 49 M. I. Malik, G. W. Harding, M. E. Grabowsky and H. Pasch, *J. Chromatogr. A*, 2012, **1244**, 77–87.
- 50 M. I. Malik, B. Trathnigg, K. Bartl and R. Saf, *Anal. Chim. Acta*, 2010, **658**, 217–224.
- 51 M. I. Malik, B. Trathnigg and R. Saf, *J. Chromatogr. A*, 2009, **1216**, 6627–6635.
- 52 M. A. Bashir and W. Radke, *J. Chromatogr. A*, 2012, **1225**, 107–112.
- 53 M. A. Bashir and W. Radke, *J. Chromatogr. A*, 2007, **1163**, 86–95.
- 54 M. A. Bashir and W. Radke, *J. Chromatogr. A*, 2006, **1131**, 130–141.
- 55 H. Ahmed, B. Trathnigg, C. O. Kappe and R. Saf, *Eur. Polym. J.*, 2009, **45**, 2338–2347.

