Chemical Science

EDGE ARTICLE



Cite this: Chem. Sci., 2023, 14, 2107

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 12th December 2022 Accepted 27th January 2023

DOI: 10.1039/d2sc06842g

rsc.li/chemical-science

Introduction

Desymmetrization of prochiral reagents is an ambitious but rewarding strategy, as the chiral products feature one or more stereocenters, after a single-step reaction in which the catalyst binds in the vicinity of the enantiotopic groups for an effective transfer of the chiral information.¹ Applied to cyclic 2,2disubstituted-1,3-diketones with enantiotopic carbonyl groups, this strategy was illustrated by Hajos and Parrish with the organocatalyzed enantioselective isomerization reaction of triketone **1** into ketol **2** (Scheme 1).² Even though the



Scheme 1 Enantioselective isomerization of 2,2-disusbtituted-1,3-cyclohexanediones.

COBRA, Normandie University, 76000 Rouen, France. E-mail: Michael.depaolis@ univ-rouen.fr

Organocatalyzed enantio- and diastereoselective isomerization of prochiral 1,3-cyclohexanediones into nonalactones bearing distant stereocenters[†]

Antoine d'Aleman, Oscar Gayraud, Catherine Fressigné, Emilie Petit, Laetitia Bailly, Jacques Maddaluno D and Michaël De Paolis *

The lactonization of 2-(2-nitrophenyl)-1,3-cyclohexanediones containing an alcohol side chain and up to three distant prochiral elements is reported by isomerization under the mediation of simple organocatalysts such as quinidine. Through a process of ring expansion, strained nonalactones and decalactone are produced with up to three stereocenters in high er and dr (up to 99:1). Distant groups, including alkyl, aryl, carboxylate and carboxamide moieties, were examined.

desymmetrization of 2,2-disubstituted-1,3-diones has been richly investigated since then,³ few reports actually showcase an enantioselective isomerization of these substrates. For example, Scheidt reported the conversion of enal **3**, tethered to 2-methyl-1,3-cyclohexanedione scaffold, to β -lactone **4** in the presence of chiral *N*-heterocyclic carbene (NHC).⁴ Through noncovalent interactions, enone **5** was isomerized to bicyclic diketones **6** by exposure to chiral phosphoric acid (CPA), as described by Lam.⁵ Later, Dong demonstrated the enantioselective transformation of 2-acetaldehyde-1,3-cyclohexanedione **7** to bicyclic γ -lactone **8** by Rh-catalyzed ketone hydroacylation reactions.⁶ Note that the enantioselective isomerization of cyclic 1,3-diones by a process effecting a ring enlargement is unknown.

Nine-membered ring compounds have important applications in medicinal chemistry,⁷ but are challenging to synthesise due to ring strain,⁸ and their tendency to undergo transannular reactions to form bicyclic products.⁹

Amidst the enantioselective ones leading to medium-sized lactones from racemic or prochiral materials, a general strategy presented itself based on the *in situ* generation of electrophiles, binding to transition metal-chiral ligands or organocatalysts, reacting with enolates or olefins.¹⁰ To our knowledge, though, a single example of enantioselective synthesis of nonalactones from prochiral reagents was reported,^{10a} and it is hitherto unknown with distant stereocenters. Herein is reported the distal enantio- and diastereoselective ring expansion of 2-aryl-1,3-cyclohexanediones, into nonalactones with up to three distal stereocenters through an organocatalyzed isomerization.

Results and discussion

Our interest in the topic stems from our previous report of a domino sequence starting from aldehyde 9 (R = H), containing the motif 2-(2-nitrophenyl)-1,3-cyclohexanedione, which



View Article Online

View Journal | View Issue

[†] Electronic supplementary information (ESI) available. CCDC 2162117. For ESI and crystallographic data in CIF or other electronic format see DOI: https://doi.org/10.1039/d2sc06842g

was converted into nonalactones when treated with carbanions (Scheme 1).¹¹ The resulting alcoholate initiated a series of transformations, eventually affording racemic nonalactones.

In a further step, the reactivity of the alcoholate holds promise for a enantioselective access to nonalactones 12 from alcohol 10 bearing two prochiral groups ($R \neq H$). In the presence of a chiral bidentate organocatalyst, we envisaged that the hydroxyl and the carbonyl moieties of 10 would be engaged in a chiral environment allowing the enantioselective formation of the fused bicyclic lactol 11 preceding the ring expansion by retro-Claisen condensation into 12.¹² Notably, our blueprint implies a spatial discrimination of prochiral elements through noncovalent interactions.

The desymmetrization of meso cyclic acid anhydrides well illustrates the concept of activation of carbonyl groups for the addition of a nucleophile, effecting thereby the discrimination of vicinal enantiotopic groups. To that end, Oda¹³ and Aitken¹⁴ independently identified Cinchona alkaloids as efficient organocatalysts. If it was thus conceivable to promote the enantioselective isomerization of **10** with these alkaloids, whether the process of desymmetrization could establish distant stereocenters was uncertain. For that matter, different strategies were developed requiring catalysts/ligands with large structures.¹⁵

To test our hypothesis, the alcohols *trans*- and *cis*-10a were prepared from the corresponding isomers of the olefins 15a after a sequence of reactions, without purification, including ozonolysis into aldehydes 16a followed by a careful reduction (Scheme 2A). Starting from 5-methyl-1,3-cyclohexanedione 14a, the construction of 15a began with one-pot *C*-arylation and *O*-allylation steps, preceding a Claisen rearrangement for which the stereochemical outcome, the ratio *trans/cis*-15a (60/40), is rectifiable under Tsuji-Trost conditions.¹⁶ This is an important aspect as the configuration of alcohol 10a had an interesting



Scheme 2 (A and B) Preparation and enantioselective isomerization of *trans*-10a and *cis*-10a. (a) dr was measured by ¹H NMR spectroscopy of the crude reaction mixture and HPLC analysis, er was determined by HPLC analysis.

bearing on the selectivity of the process (see the ESI for the optimization^{\dagger}).¹⁷

When exposed to quinidine (QD (1 equiv.), CHCl₃, 17 h, -40 to 0 °C), *cis*-**10a** was converted into *trans*-**12a** (79:21 diastereoisomers ratio (dr)), the relative configuration being determined by NOESY experiments, with enantioselectivity (96:4 enantiomers ratio (er), Scheme 2B) and with a yield of 50% yield over 3step amounting to a theoretical 85% yield for each step. Note that operating instead with 0.5 equivalent of QD gave *trans*-**12a** with close selectivity (86:14 dr, 94:6 er) after a longer reaction time (40 h, 38% yield). On the other hand, *trans*-**10a** was lactonized into **12a** (56% yield, 58:42 dr, *trans/cis*) upon exposure to QD (0.8 equiv.) in CHCl₃ at room temperature (16 h) and while *trans*-**12a** was formed without enantioselectivity, high enantiopurity (98:2 er) was measured for *cis*-**12a**.

Moreover, the treatment of *cis*-**10a** with quinine QN (1 equiv., -40 to 0 °C)—a natural pseudo enantiomer of QD—led to *trans*-**12a** (67% yield, 73 : 27 dr) having the attendant opposite enantioselectivity (3 : 97 er). Similarly, exposure of *trans*-**10a** to QN (0.5 equiv., rt) led to **12a** (70% yield, 55 : 45 dr) and, while the formation of *trans*-**12a** occurred without enantioselectivity, the enantiopurity of *cis*-**12a** was excellent (3 : 97 er).

To sum up, both alcohols *trans*-**10a** and *cis*-**10a** were simply lactonized into **12a** by an enantioselective ring expansion process upon exposure to readily available organocatalysts, that are easily recovered.

We then sought to identify the catalyst functions influencing the selectivity and, among them, the secondary benzylic alcohol appeared crucial for the enantioselectivity (Fig. 1). Bearing instead an alkyl ether or the acidic 2-nitroaniline, the corresponding catalysts induced the formation of *rac*-12a or were inactive.¹⁸

While no improvement was noted with cupreidine **17a** (*trans*-**12a**: 75 : 25 er), increasing the steric hindrance of the quinolin-6-ol scaffold was rewarding. Simply obtained by *O*-alkylation of **17a** with *i*-butyl bromide (47% yield), **17b** isomerized *cis*-**10a** to *trans*-**12a** (70 h, 51% yield) with better selectivity (90 : 10 dr, 96 : 4 er). Prepared by *O*-alkylation of **17a** with bromocyclohexane (15% yield), **17c** induced the formation of *trans*-**12a** (120 h, 63% yield) with high values of 95 : 5 dr and 96 : 4 er. For a slight modification of the ether appendage, **17d** was synthesized by *O*alkylation of **17a** with (bromomethyl)cyclohexane in excellent



Fig. 1 Tuning the catalyst. (a) Reaction conditions: *cis*-10a, QD or 17a-17e in CHCl₃, 40 h, -40 °C, then 0 °C over 6 h; (b) dr was measured by ¹H NMR spectroscopy of the crude reaction mixture and HPLC analysis, er was determined by HPLC analysis.

yield (92%). Gratifyingly, the lactonization of *cis*-**10a** catalyzed by **17d** was completed within a shorter timeframe (96 h, 60% yield) and with excellent selectivity (93 : 7 dr, 96 : 4 er). As noted with **17e**, a bulkier group at this position was detrimental to *trans*-**12a** enantiopurity (90 : 10 er).

Selecting catalysts **17c** and **17d**, the scope of the process was first examined with prochiral 5-alkyl-2-(2-nitrophenyl)-1,3cyclohexanediones (Scheme 3). Substituted with *i*-propyl group, *trans*-**12b** was isolated (62% yield—over three steps as every other given yields) with 92:8 dr and 99:1 er after exposure of *cis*-**10b** to **17d** (results obtained with **17c** are provided in the ESI†). Substituted with the *n*-propyl group, *cis*-**10c** was lactonized with excellent selectivity (*trans*-**12c**: 94:6 dr, 98:2 er). The *i*-butyl derivative *cis*-**10d** was transformed into *trans*-**12d** with 94:6 dr and 91:9 er, values obtained at -10 °C due to a solubility issue.

We then investigated the desymmetrization of 5-aryl-2-(2nitrophenyl)-1,3-cyclohexanediones. Starting with the lactonization of phenyl derivative *cis*-**10e**, *trans*-**12e** (71% yield, 93:7 dr) was isolated with moderate enantiopurity (86:14 er). A suitable crystal for X-ray crystallography was obtained with enhanced values (99:1 dr, 97:3 er) allowing the determination of the absolute (*S*, *S*)-configuration of the major isomer. The isomerization of the 4-methoxyphenyl derivative *cis*-**10f** led to *trans*-12f (49% yield) with high selectivity (95:5 dr, 90:10 er). Lactonization of the 4-fluorophenyl derivative *cis*-10g into 12g (64% yield, 40 h) occurred with an intriguing switch of diastereoselectivity. While the *cis*-lactone was so far formed in traces, *cis*-12g became slightly predominant (38:62 dr, *trans/cis*), the relative configuration being determined by NOESY experiments, and although the enantiopurity of *cis*-12g was low (67:33 er), *trans*-12g was enantioenriched (98:2 er).

With derivatives of 5-alkylcarboxylate-2-(2-nitrophenyl)-1,3cyclohexanedione such as **10h**, the selectivity of the process was worth examining. Unlike the previous pattern, *cis*-**12h** was isolated enantiopure (20% yield, 99 : 1 er) while *trans*-**12h** was obtained with lower enantiopurity (57% yield, 61 : 39 er), the relative configuration being determined by NOESY experiments. As noted at the beginning of the study, QD induced the lactonization of *cis*-**10a** with high enantioselectivity but with moderate diastereoselectivity. This proved to be advantageous, though, in this context as the treatment of *cis*-**10h** with QD led to an equal amount of *trans*- and *cis*-**12h**, the latter being isolated with an exquisite enantiopurity (43% yield, 98.5 : 1.5 er), values that remain consistent on a larger scale experiment (2.2 mmol, 41% yield, 98.5 : 1.5 er).¹⁹

Decreasing the steric hindrance of the ester, as with *cis*-**10i**, was slightly beneficial since **17d** catalyzed the formation of *cis*-



Scheme 3 Enantioselective lactonization of prochiral 2-(2-hydroxyethyl)-2-(2-nitrophenyl)-1,3-cyclohexanediones. (a) Reaction conditions: 17c, 17d or QD (1 equiv.) in CHCl₃ at -40 °C then 0 °C over 6 h, C = 0.12 M, 96 h. (b) dr was measured by both ¹H NMR spectroscopy of the crude reaction mixture and HPLC analysis, er was determined by HPLC analysis. (c) -10 °C then 0 °C over 6 h. (d) -20 °C then 0 °C over 6 h. (e) CH₂Cl₂ as co-solvant (1 : 1, v/v). (f) -40 °C to 0 °C.

12i in 26% yield and 99:1 er, *trans*-**12i** (40% yield) being isolated with 65:35 er values. The treatment of the hindered **10j** (unknown relative configuration), bearing a spiro γ -lactone moiety, afforded **12j** (34% yield) with modest selectivity (77:23 dr, 60:40 er).

After examining hydrogen bond acceptors, the isomerization of scaffolds bearing also hydrogen bond donors was evaluated with the secondary benzylcarboxamide derivative cis-10k. In this setting, 17c performed best the lactonization into trans-12k (120 h, 61% yield) and, in contrast with the carboxylate esters, the trans-isomer was prevalent (94:6 dr) and enantioenriched (88:12 er). On the other hand, 17d enabled the formation of trans-12k in shorter reaction time (72 h, 60% yield, 93 : 7 dr) but with lower selectivity (78:22 er). Embedding a more acidic N-H bond, the anilinecarboxamide cis-10l was isomerized into trans-12l (120 h, 66% yield, 95:5 dr) by 17c without enhancement of the enantiopurity (85:15 er). But this structural modification had more consequences when exposed to 17d, as the reaction of cis-10l (72 h, trans-12l: 67% yield, 90:10 dr) proceeded with higher enantioselectivity (84:16 er) than with cis-10k (78:22 er). When comparing the influence of alkyl carboxylate and secondary carboxamide functions on the selectivity, the latter steers the reaction pathway toward the enantioselective production of the trans-isomer.

The formation of a trisubstituted lactone was next studied from a meso reagent. Advantageously synthesized as a single isomer, *cis*-10m was converted by 17c into the trisubstituted lactone 12m (40% yield), isolated as a single diastereoisomer (99:1 dr) and with appreciable enantioselectivity (85:15 er). In contrast, exposure of *cis*-10m to 17d gave 12m (99:1 dr, 55:45 er) with low enantioselectivity, demonstrating thus the superior affinity of 17c for the topography of meso compounds. A step further, we sought to prepare the persubstituted lactone 12n from *cis/trans*-10n (50:50 dr). In this complex scenario however, 12n (53% yield, 58:32:10 dr) was obtained without enantioselectivity (56:44 er).

Subsequent to the isomerization of substrates with at least two prochiral elements, the strategy was applied to **100** bearing a single prochiral group (Scheme 4). Note that it was not part of our initial plan devised in Scheme 1 since it was unclear



Scheme 4 Enantioselective formation of 12o-q via the corresponding enolate. Reaction conditions: (a) 17d or QD (1 equiv.) in CHCl₃ at -40° C, 40 h then 0 °C over 6 h. The yields are given for three steps.

whether the chiral information of lactol **110** would be preserved during the ring expansion into enolate (**120**)⁻.

We were therefore surprised to measure the high enantioselectivity of **12o** (75% yield, 94:6 er) obtained by treatment of **10o** with **17d**, which dropped to 88:12 er when QD was instead employed. Indicative that a quaternary carbon within the **1**,3cyclohexanedione ring remains moderately compatible with the chemistry, **10p** was converted into **12p** (33% yield) with 72:28 er value. Pleasingly, bearing a longer alcohol appendage, **10q** was converted into decalactone **12q** with significant enantioselectivity (60% yield, 88:12 er).²⁰

Amidst all the structural modulations examined within the study, the aromatic substitution of the prochiral scaffold was left to explore. To that end, alcohols *cis*-**10r**,**s** were synthesized, each one embedding a different halo-substituted 2-nitrophenyl ring (Fig. 2).

Compared to the case of *cis*-**10a**, 4-bromo-2-nitrophenyl derivative *cis*-**10r** was isomerized with less diastereoselectivity (**12r**: 78% yield, 75:25 dr) but enhanced enantioselectivity (98.5:1.5 er) when exposed to **17d**. With a different pattern of substitution, 5-chloro-2-nitrophenyl derivative *cis*-**10s** was converted by **17d** into **12s** (65% yield, 72:28 dr) with excellent enantioselectivity (97.5:2.5 er).

To modulate the position of the nitro group, we set about preparing 2-(3-nitrophenyl)-1,3-cyclohexanedione derivative **10t**. But the instability of the precursor aldehyde **9t** thwarted our plan and a similar outcome was noted with the 4-nitrophenyl analogue **9u**. We surmised that a steric effect, induced by the ortho substitution of the aromatic ring, could prevent the degradation of the aldehyde. Lending credence to this hypothesis, 2-(2-bromo-4-nitrophenyl) derivative **9v** was successfully obtained and converted into alcohol *cis*-**10v**. In the presence of



Fig. 2 Enantioselective isomerization of *cis*-**10***r*,**s**,**v** into nonalactones *trans*-**12***r*,**s**,**v**. Reaction conditions: (a) **17d** or **QD** (1 equiv.) in CHCl₃ at -40 °C, then 0 °C over 6 h. (b) dr was measured by ¹H NMR spectroscopy of the crude reaction mixture and HPLC analysis, er was determined by HPLC analysis. (c) -10 °C. The yields are given for three steps.

17d, the formation of *trans*-**12v** occurred slowly at $-10 \text{ }^{\circ}\text{C}$ (120 h, 60% yield) with modest selectivity (69 : 31 dr, 68 : 32 er). A slight enhancement was noted upon the catalysis of QD (40 h, $-10 \text{ }^{\circ}\text{C}$, 38% yield), *trans*-**12v** being formed with higher enantiopurity (78 : 22 er) but at the expense of the diastereoselectivity (47 : 53 dr). The nitro group in the ortho-position of the aromatic ring seems therefore favorable to achieve high stereoselectivity.

However, as illustrated in Scheme 5, this structural requirement advantageously paves the way to the synthesis of enantioriched alkaloids and derivatives. For instance, the indole 18a was prepared from 12a by reduction of the nitro group (Zn, AcOH, >95% yield) and the regioselective halogenation occurred in 6-position (19, 72% yield), offering a synthetic handle for the decoration of the aromatic ring. Underscoring the strain of the indolononalactone, the spontaneous air oxidation of the aromatic ring of 18a into hydroxylindolenine was noted (\sim 20% after two months of storage at rt, see the ESI for the details[†]). For that matter, we purposely performed the consecutive ring-expansion reaction of 18a by oxidative cleavage $(NaIO_4)^{21a}$ delivering the 12-membered ring 20 (56% yield). Not only this demonstrated a short route to enantioenriched 12membered ring from 6-membered ring 10a via 12a,^{21b} but 20 provides a platform for further ring expansion, as illustrated by Unsworth, to access macrocycles.^{21c}

Palladium-catalyzed alkylative dearomatization of the indole **18a** led to *trans*-allylic indolenine **21** (53% yield, non-optimized) after the distal diastereoselective formation of a quaternary carbon (97 : 3 dr).²² Bearing an additional reactive function, the aldehyde **22** was smoothly obtained after oxidative cleavage of the allylic appendage (>95% yield). Prepared from **12h** by hydrogenation (>95% yield), the indolic ethyl ester **18h** was treated with NaOH (1 equiv., 1,4-dioxane, H₂O) to give the carboxylic acid lactone **18ha** (56% yield, 68% brsm), demonstrating thus the robustness of the strained but sterically hindered ring.²³ To access new scaffolds, **12a** (80 : 20 dr) was deconstructed with LiAlH₄ into triol **23** (42% yield, 75 : 25 dr) and next converted into δ -lactone **24** (PhI(OAc)₂/TEMPO, 59%



Scheme 5 Synthetic manipulations.

yield, relative configuration determined by NOESY experiments).²⁴

Conclusions

While they are among the most difficult cyclic scaffolds to prepare for kinetic and thermodynamic reasons,^{8b} nonalactones and decalactone with up to three non-vicinal stereocenters were prepared by ring expansion of prochiral alcohols (21 examples). Modulation of the configuration and substitution patterns of the prochiral material allowed the exploration of various steric and electronic scenarios (alkyl, aryl, carboxylate and carboxamide), picturing the perimeter and potential of the strategy with high values of dr and er (up to 99:1) owing to the 2nitrophenyl function. Whether other electron-withdrawing groups may enable the enantioselective desymmetrization of the corresponding alcohol remains to be determined. However, this current limitation is counterbalanced by the enantio- and diastereoselective access to indoles, indolenines and derivatives thus provided, structural motifs notably encountered in many alkaloids (natural products or pharmaceuticals). From this investigation emerges a tool box in which readily available organocatalysts-QD or QN, 17c, 17d-effectively synthesized isomers of lactones 12 from cis-10 or trans-10. Bearing a guinolin-6-ol substituted with a methylenecyclohexyl ether, which increased the volume of the appendage, 17d offered a good balance between reactivity and stereoselectivity in most cases. Quinidine and quinine being commercially available, the two other derivatives are prepared in one step from cupreidine and are easily recovered after reaction, offsetting the catalytic load used to perform the energy-demanding ring expansion step. For that matter, the process of ring expansion of trans-10a into cis-12a required higher temperature to occur, without inducing lower enantioselectivity.

Futures studies will be focused on some features and implications of this complex and yet simple to implement process, such as the inversion of selectivity observed with fluorophenyl and carboxylate ester derivatives, or the origins of the stereoselectivity induced by the alkyl ether of the quinolin-6-ol appendage of the catalyst.

Data availability

All experimental and characterization data including HPLC traces and NMR spectra are available in the ESI.[†] Crystallographic data for compound **12e** has been deposited in the Cambridge Crystallographic Data Centre under accession number CCDC 2162117.

Author contributions

A. A. and O. G. conducted the investigation and prepared the ESI.[†] C. F. conducted the DFT calculations. L. B. and E. P. contributed to the formal analysis of the results. J. M. reviewed the manuscript. M. D. P. conceptualized and supervised the research, wrote the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

AA thanks the Ministere de la Recherche et de l'Education Nationale for a PhD fellowship. This work has been partially supported by University of Rouen Normandy, INSA Rouen Normandy, the Centre National de la Recherche Scientifique (CNRS), European Regional Development Fund (ERDF), Labex SynOrg (ANR-11-LABX-0029), Carnot Institute I2C, the graduate school for research XL-Chem (ANR-18-EURE-0020 XL CHEM), and by Région Normandie. AA thanks Prof. Adam Daïch (University of Le Havre) for suggesting the oxidative ring expansion of compound **18a**. We thank Dr Jean-Francois Brière and Dr Julien Legros for helpful input, Dr Baptiste Picard for preliminary experiments and Jean-François Lohier (LCMT) for X-ray analysis. The CRIANN is gratefully acknowledged for providing computation resources.

Notes and references

- 1 C. Nájera, F. Foubelo, J. M. Sansano and M. Yus, *Tetrahedron*, 2022, **106**, 132629.
- 2 Z. G. Hajos and D. R. Parrish, *J. Org. Chem.*, 1974, **39**, 1615–1621.
- 3 (a) X.-P. Zeng, Z.-Y. Cao, Y.-H. Wang, F. Zhou and J. Zhou, Chem. Rev., 2016, 116, 7330-7396; (b) R. Chegondi, S. M. Patel, S. Maurya and A. Donthoju, Asian J. Org. Chem., 2021, 10, 1267-1281; (c) Y.-S. Zhao, X.-Q. Tang, J.-C. Tao, P. Tian and G.-Q. Lin, Org. Biomol. Chem., 2016, 14, 4400-4404; (d) F. Leonelli, A. Trombetta, A. La Bella, G. Lucarelli, N. Demitri, D. Lamba, L. M. Migneco and R. Marini Bettolo, Eur. J. Org. Chem., 2019, 2019, 1594-1599; (e) T. Yoshimura, Y. Enami and J. Matsuo, Synthesis, 2020, 52, 3667-3674; (f) P. Zhou and T. Xu, Chem. Commun., 2020, 56, 8194-8197; (g) X. Han, L. Shan, J. Zhu, C. Zhang, X. Zhang, F. Zhang, H. Wang, Y. Tu, M. Yang and W. Zhang, Angew. Chem., Int. Ed., 2021, 60, 22688-22692; (h) X.-L. Qin, A. Li and F.-S. Han, J. Am. Chem. Soc., 2021, 143, 2994-3002; (i) B. Yang, J. Dai, Y. Luo, K. K. Lau, Y. Lan, Z. Shao and Y. Zhao, J. Am. Chem. Soc., 2021, 143, 4179–4186; (j) L. Zhang and M. Oestreich, ACS Catal., 2021, 11, 3516–3522; (k) J. Li, J. Sun, W. Ren, J. Lei, R. Shen and Y. Huang, Org. Lett., 2022, 24, 2420-2424; (l) Z. Zhou, D. Xu, W. Jiang, J. chen, Y. Zhen, J. Huo, J. Yan, J. Gao and W. Xie, Org. Lett., 2022, 24, 9017-9022; (m) M. Dajek, A. Pruszczyńska, K. A. Konieczny and R. Kowalczyk, Adv. Synth. Catal., 2020, 362, 3613-3620; (n) Y. Kawamoto, D. Ozone, T. Kobayashi and H. Ito, Eur. J. Org. Chem., 2020, 4050-4058.
- 4 (a) M. Wadamoto, E. M. Phillips, T. E. Reynolds and K. A. Scheidt, *J. Am. Chem. Soc.*, 2007, **129**, 10098–10099;
 (b) T. Ema, K. Akihara, R. Obayashi and T. Sakai, *Adv. Synth. Catal.*, 2012, **354**, 3283–3290.

- 5 A. R. Burns, A. G. E. Madec, D. W. Low, I. D. Roy and H. W. Lam, *Chem. Sci.*, 2015, **6**, 3550–3555.
- 6 X. Wu, Z. Chen, Y.-B. Bai and V. M. Dong, J. Am. Chem. Soc., 2016, 138, 12013–12016.
- 7 (a) Heterocyclic Chemistry III, eds. Katritzky, A. R., Ramsden, C. A., Scriven, E. F. V. and Taylor, R. J. K., Elsevier, Amsterdam, 2008, vol. 14, pp. 547-611; (b) I. Shiina, Chem. Rev., 2007, 107, 239-273; (c) T. Huber, R. E. Wildermuth and T. Magauer, Chem. - Eur. J., 2018, 24, 12107-12120; (d) J. S. Clark, R. Berger, S. T. Hayes, L. H. Thomas, A. J. Morrison and L. Gobbi, Angew. Chem., Int. Ed., 2010, 49, 9867-9870; (e) Z. Meng and A. Fürstner, J. Am. Chem. Soc., 2019, 141, 805-809; (f) S. Shiomi, K. Wilailak, W. Soutome, H. Takayama, M. Kitajima and H. Ishikawa, J. Org. Chem., 2022, 87, 3730-3735; (g) C. Liu, Y. Li, W.-Y. Shi, Y.-N. Ding, N. Zheng, H.-C. Liu and Y.-M. Liang, Org. Lett., 2021, 23, 4311-4316; (h) A. Lawer, J. A. Rossi-Ashton, T. C. Stephens, B. J. Challis, R. G. Epton, J. M. Lynam and W. P. Unsworth, Angew. Chem., Int. Ed., 2019, 58, 13942-13947; (i) G. Yang, Y. Ke and Y. Zhao, Angew. Chem., Int. Ed., 2021, 60, 12775-12780; (j) J. R. Donald and W. P. Unsworth, Chem. - Eur. J., 2017, 23, 8780-8799; (k) A. K. Clarke and W. P. Unsworth, Chem. Sci., 2020, 11, 2876-2881.
- 8 (a) J. D. Dunitz and V. Prelog, Angew. Chem., 1960, 72, 896–902; (b) C. Galli, G. Illuminati, L. Mandolini and P. Tamborra, J. Am. Chem. Soc., 1977, 99, 2591–2597.
- 9 (a) Y. Ban, K. Yoshida, J. Goto and T. Oishi, J. Am. Chem. Soc., 1981, 103, 6990-6992; (b) I. Shiina, Y. Takasuna, R. Suzuki, H. Oshiumi, Y. Komiyama, S. Hitomi and H. Fukui, Org. Lett., 2006, 8, 5279-5282; (c) M. Mewald, J. W. Medley and M. Movassaghi, Angew. Chem., Int. Ed., 2014, 53, 11634-11639; (d) K. L. White and M. Movassaghi, J. Am. Chem. Soc., 2016, 138, 11383-11389; (e) For a review: E. Reyes, L. Prieto, L. Carrillo, U. Uria and J. L. Vicario, Synthesis, 2022, 54, 4167-4183.
- 10 (a) L. Ding, H. Song, C. Zheng and S.-L. You, J. Am. Chem. Soc., 2022, 144, 4770–4775; (b) J. L. Payne, Z. Deng, A. L. Flach and J. N. Johnston, Chem. Sci., 2022, 13, 7318–7324; (c) X. Jiang, X. Xu, W. Xu, P. Yu and Y.-Y. Yeung, Org. Lett., 2021, 23, 6316–6320.
- 11 (a) D. Reyes Loya, A. Jean, M. Cormier, C. Fressigné, S. Nejrotti, J. Blanchet, J. Maddaluno and M. De Paolis, *Chem. – Eur. J.*, 2018, 24, 2080–2084; (b) D. Reyes Loya and M. De Paolis, *Chem. – Eur. J.*, 2019, 25, 1842–1847.
- 12 For an example of enantioselective four-atom ringexpansion, see: Y. Zhou, M. Y.-L. Wei, J. Rodriguez and Y. Coquerel, *Angew. Chem., Int. Ed.*, 2018, **58**, 456–460.
- 13 (a) J. Hiratake, Y. Yamamoto and J. Oda, *J. Chem. Soc., Chem. Commun.*, 1985, 1717; (b) J. Hiratake, M. Inagaki, Y. Yamamoto and J. Oda, *J. Chem. Soc., Perkin Trans.*, 1987, 1, 1053.
- 14 (a) R. A. Aitken, J. Gopal and J. A. Hirst, J. Chem. Soc., Chem. Commun., 1988, 632; (b) R. A. Aitken and J. Gopal, Tetrahedron: asymmetry, 1990, 1, 517–520.
- 15 (a) C.-C. Hsiao, H.-H. Liao and M. Rueping, Angew. Chem., Int. Ed., 2014, 53, 13258–13263; (b) A. J. Metrano and

S. J. Miller, Acc. Chem. Res., 2019, 52, 199-215; (c) W. Luo, L. Lin, Y. Zhang, X. Liu and X. Feng, Org. Lett., 2017, 19, 3374-3377; (d) J. T. Payne, P. H. Butkovich, Y. Gu, K. N. Kunze, H. J. Park, D.-S. Wang and J. C. Lewis, J. Am. Chem. Soc., 2018, 140, 546-549; (e) Y. Lou, J. Wei, M. Li and Y. Zhu, J. Am. Chem. Soc., 2022, 144, 123-129.

- 16 Along this study, the ratio trans/cis-15 varied with the substitution of the 1,3-cyclohexanedione ring (see ESI for the details[†]). An isomerization step was devised under Tsuji-Trost conditions (Pd(Ph₃)₄ cat., THF, rt) to convert pure trans-15 into an isomeric mixture from which cis-15 is isolated. The method was illustrated from trans-15b vielding a mixture of isomers (68:32, trans/cis) in spite of the steric hindrance of the *i*-Pr group (see the ESI for the details[†]).
- 17 In the absence of a catalyst, no formation of 12a was noted at a low temperature. Likely by acidic activation, the slow conversion of cis-10a into rac-12a occurred at room temperature in CDCl₃ (after 1 h) or on silica gel, making mandatory а full conversion to achieve high enantioselectivity. Otherwise, oxidation of the unreacted alcohol into aldehyde was performed by addition of Dess-Martin periodinane.
- 18 Several bifunctional organocatalysts were tested among the cinchona alkaloids including stronger H-bonding appendage as thiourea but lower conversion was noted (see ESI doi details[†]). Further, the Takemoto catalyst performed the isomerization with lower er values (see ESI⁺).
- 19 We sought to confirm the absolute configuration of 12h,i, to no avail. While the absolute configuration of 12e was used as a basis of extrapolation to the other cases, the peculiar selectivity noted with 12h,i suggests a different mechanism with potential implication on this point. Moreover, performing the experiment at room temperature (4 h, 70%

yield) revealed that the enantiopurity of cis-12h (18% yield) only marginally decreased (96:4)er). the was stereoselectivity being more impacted (75:25 dr) with trans-12h isolated in 70:30 er. Incidentally, trans-12h was measured with reversed and higher, though still modest, enantiopurity when the reaction was conducted at room temperature than at -40 °C (45:55 er), probably due to the epimerization of enantioenriched cis-12h into trans-12h occurring at room temperature.

- 20 Since E-enolates potentially exhibit planar chirality when embedded in medium-sized rings, it may also be the case for enolate (12)⁻. See: K. Tomooka, T. Ezawa, H. Inoue, K. Uehara and K. Igawa, J. Am. Chem. Soc., 2011, 133, 1754-1756Whether chirality of the transient enolate (120- $(\mathbf{q})^{-}$, deprived of stereogenic carbons, could account for the enantioselective isomerization of 10o-q or be the result of an enantioselective protonation of $(120-q)^{-}$ by the catalyst was not investigated.
- 21 (a) L. J. Dolby and D. L. Booth, J. Am. Chem. Soc., 1966, 88, 1049-1051; (b) T. C. Stephens and W. P. Unsworth, Synlett, 2020, 31, 133-146; (c) Z. Yang, C. R. B. Swanson and W. P. Unsworth, Synlett, 2022, DOI: 10.1055/a-1932-9717.
- 22 N. Kagawa, J. P. Malerich and V. H. Rawal, Org. Lett., 2008, 10. 2381-2384.
- 23 Owing to a restriction of conformation, the formation of indolononalactone 12h incidentally revealed a point-toplanar chirality transfer, which was not observed, at room temperature, with the less hindered 12a (see the ESI for the details[†]). For a review: R. Lopez and C. Palomo, Angew. Chem., Int. Ed., 2022, 61, e202113504.
- 24 T. M. Hansen, G. J. Florence, P. Lugo-Mas, P. Chen, J. N. Abrams and C. J. Forsyth, Tetrahedron Lett., 2003, 44, 57-59.