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HFIP promoted carbo-lactonisation as a new route to functionalised lactones†

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We have been able to induce a series of lactonisation reactions of unsaturated acids using a carbocationic intermediate formed *in situ*. The transformation generates five and six membered lactone rings AND forms a new exocyclic C–C bond at the same time, thus removing the need for further derivatisation of typical lactonisation products (such as from iodo lactonisation). The reaction is operationally straightforward, does not employ any metals and utilises readily accessible alcohol and unsaturated acid substrates.

γ -Lactones are a privileged scaffold in natural products and biologically active compounds.¹ Their wide range of biological properties, such as antimicrobial,^{2a} anti-inflammatory^{2b} and cytotoxic activities,^{2c} have established 5- and 6-membered lactones as a common structural motif in drug design. Lactones possessing a quaternary centre adjacent to the oxygen atom, are widespread in natural products and have been extensively studied in medicinal chemistry.³

The use of heteroatom containing electrophiles to induce the lactonisation of an unsaturated carboxylic acid is commonplace in organic synthesis.⁴ Typically, an electrophile such as iodine promotes cyclisation by initial reaction with an alkene followed by cyclisation *via* nucleophilic attack of the carboxylic acid (Scheme 1A).⁴ One disadvantage of this approach is that it is often necessary to further transform the newly formed C-heteroatom bond in the product, especially if further C–C bond forming sequences are desired. Lactonisation reactions that are triggered by the formation of a new C–C bond are rare,⁵ and often rely on transition-metal catalysis.⁶ In this regard, we wondered if carbocations⁷ might play a role in initiating

lactonisation reactions. The reaction of an alkene with a carbocation initiator and a carboxylic acid should allow the formation of ring systems (here lactones) and form a new exocyclic C–C bond in the same reaction (Scheme 1B). This mode of lactonisation can be facilitated by the reaction of allylic or benzylic alcohols with Lewis or Brønsted acid in HFIP solvent.⁸ With this approach, a set of highly functionalised lactones could be prepared rapidly from readily available unsaturated acids and activated alcohols. The role of HFIP is likely essential in facilitating ionisation of an alcohol without the need to activate the hydroxyl group separately.⁹ Furthermore, the new method reported here enables an unconventional transformation to take place wherein combining a carboxylic acid partner and an alcohol starting material in the presence of a Brønsted acid (with no requirement for a Lewis acid additive) leads to the formation of a functionalised lactone, instead of the more traditional Fischer esterification product.¹⁰

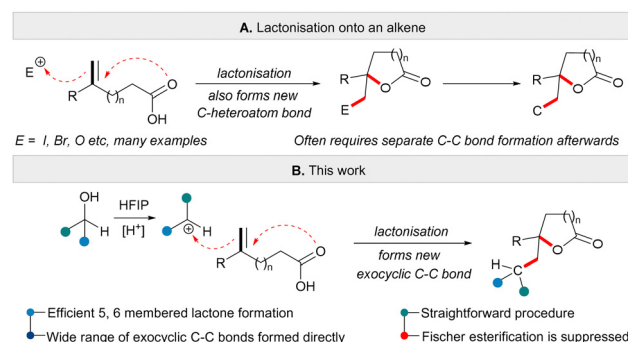
First, an extensive reaction optimisation was carried out to identify the optimum system for the formation of **2a** (see Scheme 2 and ESI† for full details). With the optimised metal-free conditions available (TFA in HFIP at RT gave **2a** in 94% yield), the scope of the method was then studied with regards to both the alcohol and the alkene component (Scheme 2). A range of activated benzylic and allylic alcohols

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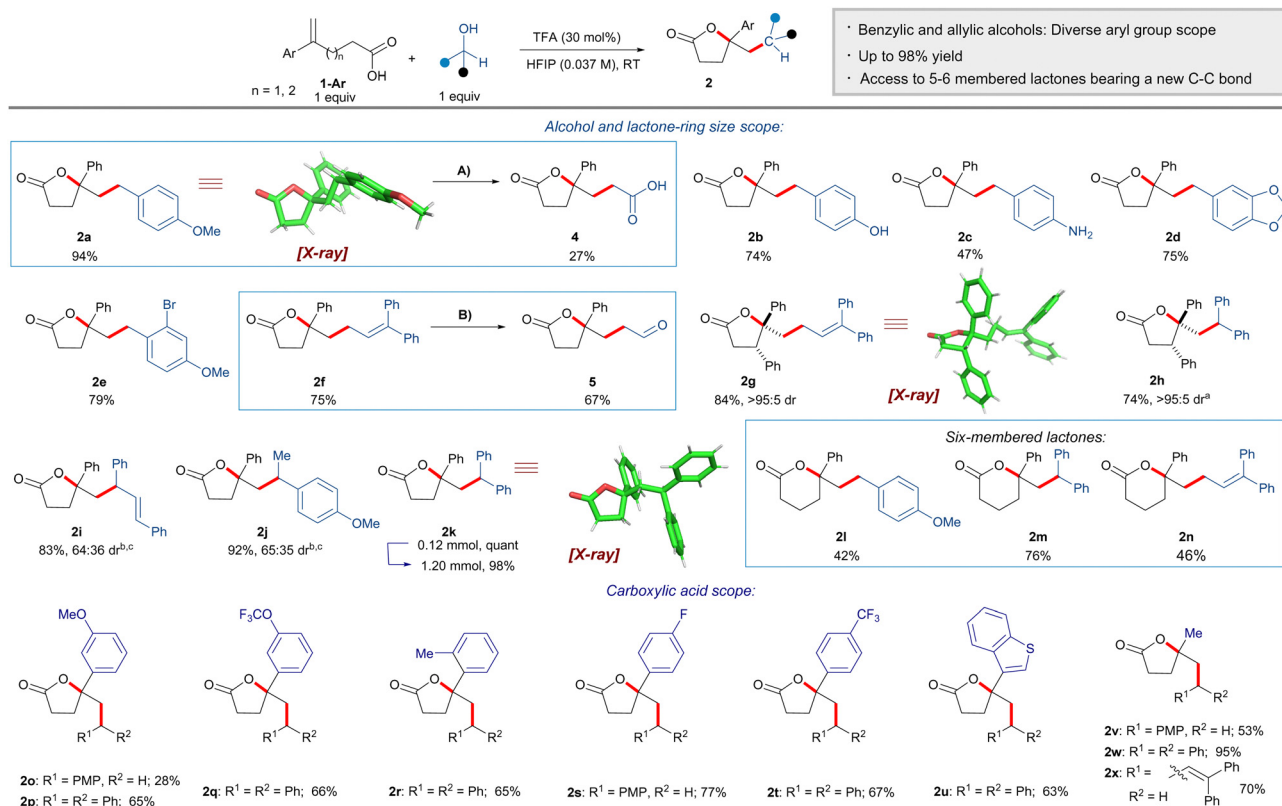
† Electronic supplementary information (ESI) available: Copies of all experimental procedures, ¹H/¹³C NMR spectroscopic characterisation and X-ray data. CCDC 2349117 (**2a**), 2349118 (**2g**), 2349119 (**2k**), 2349120 (**2t**), 2349121 (**4**), 2349122 (**9a**). For ESI and crystallographic data in CIF or other electronic format see DOI: <https://doi.org/10.1039/d4cc02472a>

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Scheme 1 Direct carbo-lactonisation of unsaturated acids.





Scheme 2 Alcohol and aryl group reaction scope. (a) Relative stereochemistry of **2h** is assigned based on the X-ray crystal structure of **2g**.¹⁴ (b) dr was calculated based on NMR analysis of the crude reaction mixture; (c) relative stereochemistry at the exocyclic position could not be determined. Reaction conditions: (A) RuCl₃ (5 mol%), NaIO₄ (20 equiv.), CCl₄/CH₃CN/pH 7 buffer; 2 : 2:3, 0 °C to rt, overnight; (B) O₂/O₃, CH₂Cl₂, -78 °C, 5 min then DMS (excess), -78 °C to rt, 3 h.

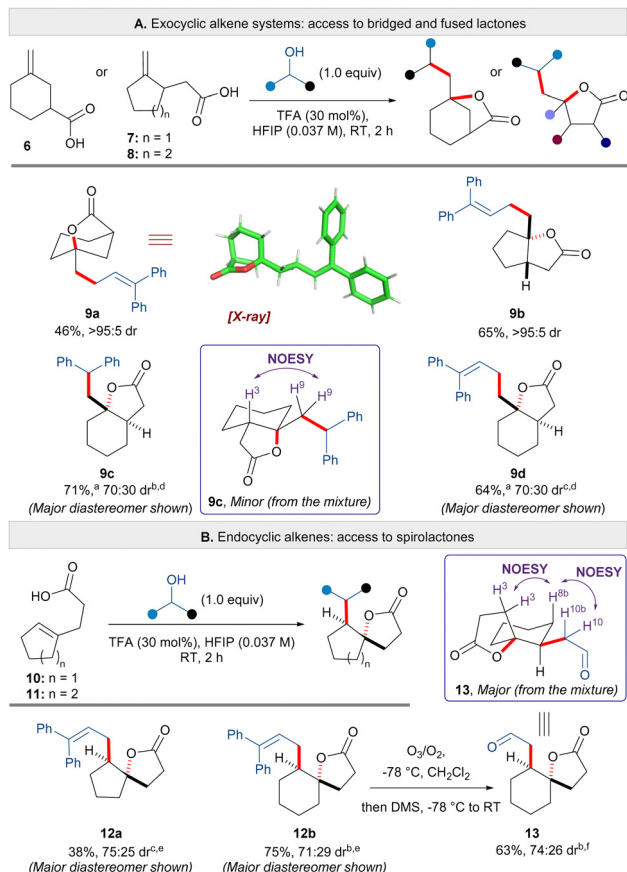
were shown to be viable in this reaction, forming the corresponding 5-membered lactones **2a–k** in good to excellent yields. Free hydroxyl (**2b**), amine (**2c**), halogen (**2e**) and acetal (**2d**) groups on the benzylic alcohols were well tolerated under the reaction conditions. Activated allylic alcohols could also be engaged, furnishing products **2f**, **2g** and **2i** bearing a versatile exocyclic alkene functionality in high yields. The presence of an aryl substituent on the allylic alcohol was essential, as unsatisfactory results were obtained with alkyl substituted allylic alcohols (eg prenyl alcohol). Asymmetric secondary alcohols furnished products **2i–j** in excellent yields but as a mixture of diastereomers. In contrast, excellent diastereoselectivity was observed for carboxylic acid substrates possessing a Ph substituent adjacent to the alkene; γ -lactones **2g** and **2h** were both isolated in good yields and as single diastereomers. The reliability of this transformation was demonstrated by a scale-up experiment: carboxylic acid **1-Ph** and benzhydrol afforded excellent yields of lactone **2k** when performed on both a 0.12 mmol and 1.2 mmol scale. Notably, the products of this transformation could be further derivatised to install versatile functional groups. For example, RuCl₃ mediated oxidative cleavage of the electron-rich PMP group in lactone **2a** furnished carboxylic acid **4** in 27% yield.¹¹ Alternatively, lactones formed from reaction with allylic alcohols were elaborated at the exocyclic position using ozonolysis;¹² lactone **2f** was subjected to ozonolysis to afford aldehyde **5** in 67% yield.

The developed carbo-lactonisation methodology could be extended to 6-membered lactones. Products **2l–n** were isolated in moderate to good yields, Scheme 2. Here, cyclisation yields to access δ -lactones were found to be somewhat lower compared to γ -lactones, which is consistent with previous results for other HFIP-promoted heterocyclisations.⁸

Next, we evaluated the scope of this method with regard to the terminal alkene substituent. We were pleased to observe that inductively electron-donating (**2r**) and mesomerically electron-donating (**2o–q**) substituents in the arene *ortho*- and *meta*-positions were well tolerated, furnishing γ -lactones in up to 66% yield. Alkenes conjugated with aryl groups bearing inductively electron-withdrawing groups could also be engaged, as demonstrated with **2s** and **2t** which were isolated in 77% and 67% yield respectively. We were also pleased to find that a trifluoromethyl functional group,^{13a} was well tolerated. Furthermore, a carboxylic acid bearing a benzothiophene^{13b} could be engaged to furnish γ -lactone **2u** in 63% using benzhydrol as the electrophile. Notably, the reaction is not limited to aryl substituted terminal alkenes. Aliphatic carboxylic acids could also be efficiently engaged furnishing lactones **2v–x** in good to excellent yields using benzylic and allylic alcohols.

The limitation of this methodology is represented by carboxylic acid substrates bearing aryl groups with strong mesomerically electron-withdrawing groups in the *para* and *meta*-position





Scheme 3 Synthesis of bridged, fused and spirolactones. (a) Reaction performed at 0 °C; (b) dr calculated after column chromatography; (c) dr calculated based on the crude reaction mixture; (d) relative stereochemistry of the diastereomers is assigned based on NOESY analysis (see ESI†); (e) relative stereochemistry is assigned based on aldehyde **13**; (f) relative stereochemistry of the major diastereomer is assigned based on NOESY analysis of the mixture.

(eg. CN), as well as substrates possessing electron-rich five-membered rings (eg thiophene, see ESI† for details).

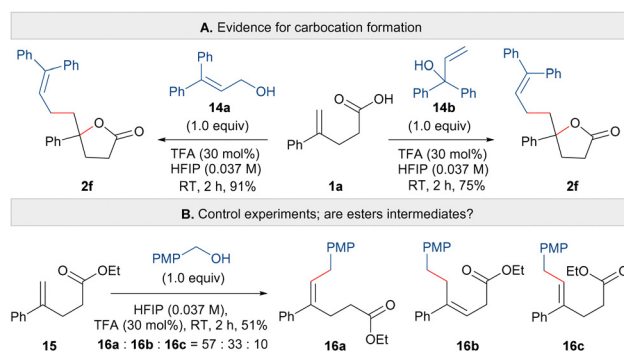
Given the successful formation of alkyl substituted lactones **2v–x**, and to further explore the breadth of this transformation, we applied the carbo-lactonisation methodology to the synthesis of bridged, fused and spirolactones (Scheme 3). To our delight, aliphatic carboxylic acid substrates possessing exocyclic alkenes furnished products **9a–d** in moderate to good yields (Scheme 3A). The reaction between carboxylic acid substrate **6** and 1,1-diphenylprop-2-en-1-ol afforded bridged lactone **9a** in 46% yield (structure confirmed by X-ray). Carbo-lactonisation of five-membered carboxylic acid **7** proceeded with excellent diastereoselectivity to furnish the *cis*-fused [5,5] bicyclic lactone **9b** in 65% yield. The levels of diastereoselectivity dropped with six-membered substrate **8**; products **9c** and **9d** were isolated in 71% and 64% yield respectively as a 70:30 mixture of diastereomers. NOESY analysis indicated *trans*-stereochemistry at the ring fusion for the major diastereomer (see ESI† for details). Next, carbo-lactonisation of carboxylic acids bearing aliphatic endocyclic alkenes were explored (Scheme 3B). Substrate **10** and 1,1-diphenylprop-2-en-1-ol furnished [5,5] spirolactone **12a**

in 38% yield with moderate diastereoselectivity. [6,5] Spirolactone **12b** could also be accessed from cyclisation of acid **11** in 75% yield and in a similar diastereomeric ratio. The use of an allylic alcohol for cyclisation of substrates **12a** and **12b** allowed for the installation of an exocyclic alkene that could be used for further transformations. Thus, we performed ozonolysis of an exocyclic alkene to afford aldehyde **13** in 63% yield isolated as a 74:26 mixture of diastereomers. This material was used to determine the relative stereochemistry of products **12a** and **12b** based on NOESY analysis of the mixture of diastereomers **13**. The observed *trans*-configuration for the major isomer is consistent with a predominate *trans*-addition of a carbocation across the double bond of the nucleophile.⁸

Finally, we explored the mechanism of the carbo-lactonisation reaction. Firstly, to show that the reaction indeed does proceed *via* carbocation formation, the synthesis of lactone **2f** was carried out using transposed isomers of allylic alcohol **14a** and **14b** (Scheme 4A). In both cases, the same regioisomer of lactone **2f** was observed and isolated, which is consistent with the formation of a delocalised carbocation. Next, to shed light on whether the C–C or C–O bond is formed in a concerted or stepwise manner, and also to probe the possible intermediacy of esters,¹⁵ the reaction was attempted using unsaturated ester **15** and 4-methoxybenzyl alcohol (Scheme 4B). Compound **15**, bearing a terminal alkene but lacking a carboxylic acid functional group, yielded a mixture of unsaturated esters **16a**, **16b** and **16c** in 51% yield. Here it appears that a benzylic cation has undergone reaction with the alkene but that cyclisation has not occurred from an oxygen nucleophile.

These results support the hypothesis that the carbo-lactonisation proceeds in a stepwise fashion in which the carbocation generated from the alcohol reacts with the terminal alkene, forming a new C–C bond and another stabilised cation; this is followed by intramolecular attack of the carboxylic acid group. This mechanism could also explain the moderate diastereoselectivities observed in the synthesis of spirolactones (**12a** and **12b**) where a concerted (*trans*) addition of a heteroatom and a cation across the substrate alkene would be expected to form a single diastereoisomer.

In conclusion, this work details the development of a carbo-lactonisation methodology to form a diverse array of substituted γ - and δ -lactones. Activated benzylic and allylic alcohols



Scheme 4 Control experiments.



were engaged in the cyclisation to form new C–C and C–O bonds and to set-up tetrasubstituted stereogenic centres. The reaction has a broad substrate scope with yields of up to 98%.

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Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

There are no conflicts of interest to declare.

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