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Multi-year, high-time resolution aerosol chemical composition and mass measurements from Fairbanks, Alaska†

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Fairbanks-North Star Borough, Alaska (FNSB) regularly experiences some of the worst wintertime air quality in the United States. Exceedances of the EPA's 24 h fine particulate matter (PM_{2.5}) rule are common, and can last for weeks-long periods. Here we present sub-hourly measurements of chemically-specified aerosol measurements over a 25 month span from an Aerosol Chemical Speciation Monitor (ACSM). This dataset includes measurements from all four seasons and over three separate winters (2020, 2021, 2022). It spans a long enough duration to provide an overview of typical seasonal and diurnal variations in aerosol concentrations, composition, and sources in Fairbanks. We observe consistent high PM_{2.5} concentrations in wintertime, which is dominated by organic aerosol (OA) and, to a lesser extent, sulfate (SO₄). We perform factor analysis of the OA using Positive Matrix Factorization (PMF), which reveals three factors, two of which are attributable to primary sources. These primary OA factors are highest in concentration and fractional contribution during wintertime. We show that high concentration periods are correlated with cold temperatures, and enriched in those organic aerosol components related to primary emissions. High concentration periods are also enriched in SO₄, though we show that some of the "SO₄" measured by the ACSM is very likely organosulfur compounds, which are more prevalent at high concentrations. We also show that within winter, there are significantly different diurnal patterns in PM components depending on meteorological parameters. This analysis is important for understanding air quality patterns in Fairbanks, and as context for the 2022 ALPACA measurement campaign.

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Environmental significance

Multi-year, high-time resolution aerosol chemical composition and mass measurements from Fairbanks, Alaska: Fairbanks, AK has some of the worst wintertime air quality in the US due to strong local sources and frequent periods of low dispersion. We present a unique dataset spanning three years (2020–2022), including three winters, that illustrates how particulate matter (PM) emissions sources and meteorology interact and lead to polluted conditions. We show how PM chemical composition changes across season, temperature, and pollution level. These findings are highly relevant to the population of Fairbanks and other similar areas by shedding light on what aspects of the pollution problem in Fairbanks may be addressable at the source. This also establishes important context for interpreting results from the 6-week intensive "ALPACA" field campaign during 2022.

1 Introduction

Fairbanks-North Star Borough (FNSB), Alaska suffers from perennially poor air quality during wintertime. The area

regularly exceeds the health-based PM_{2.5} threshold put forth by the U.S. Environmental Protection Agency (EPA) in the National Ambient Air Quality Standard (NAAQS) of 35 μg m⁻³ average over 24 hours, sometimes for multiple days at a time.¹ FNSB has been a designated EPA non-attainment area for the 24 hours PM_{2.5} NAAQS since 2009, and was reclassified as a "serious" non-attainment area in 2017. The area regularly makes the American Lung Association's list of cities with the worst air quality, including as recently as 2021.^{1,2}

Poor air quality can be a common feature for urban areas at high latitude in winter, where temperatures are very low and there is very little sunlight; Fairbanks is an extreme example.³ The cause of the pollution is typically local emissions that

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become trapped near the surface under low-dispersion conditions. Winters in Fairbanks are characterized by extreme cold (reaching as low as $-40\text{ }^{\circ}\text{C}$), very light winds, and strong surface-based inversions (frequently $0.5\text{ }^{\circ}\text{C m}^{-1}$ or more in the first 10 m above ground level). Fairbanks has a mix of local sources, including industry and power generation, transportation, and residential heating.⁴ Oil and biomass combustion are the primary fuel sources for residential heating, with relatively little use of natural gas or electric.⁵ Per capita demand for heating is very high given the extreme cold.

Previous studies of FNSB's air quality problems show that organic aerosol (OA) dominates wintertime PM mass (55–75%).⁴ Source apportionment studies have found domestic wood burning to be the largest single source of PM pollution, representing between 40–80% of wintertime PM in Fairbanks.^{6–8} Sulfate (SO_4) is the second-most prominent aerosol component, representing 15–20% of wintertime PM in the above studies.^{4,9,10} More recent PMF analysis from Ye and Wang showed decreasing woodsmoke and increasing SO_4 as a fraction of total PM over time. Multiple studies^{12,13} have investigated the sources and mechanisms underlying the observed high SO_4 levels, but have been unable to reproduce the high $\text{PM}_{2.5}$ SO_4 concentrations ($>10\text{ }\mu\text{g m}^{-3}$) through modeling. Recent work from Moon *et al.*¹⁰ shows that a large fraction of wintertime SO_4 are primary emissions, suggesting that the gap between measurements and models may be attributable to emission inventory inaccuracy for sulfur-containing fuels.

Despite the focus that air quality issues in FNSB have received, significant questions remain. Most of the PM speciation work in Fairbanks has been done using filter-sampling, which may have significant artifacts related to partitioning when sampling at ultra-cold temperatures. Additionally, filter sampling, typically done over the course of many hours or days, lacks the temporal resolution required to understand intraday changes and diurnal patterns of different aerosol species. Much of the above literature was published using data from ~ 2010 or earlier, and so these studies may not accurately reflect the current air quality problem in FNSB, as PM levels have declined over the past decade, per long-running monitoring data from the Alaska Department of Environmental Conservation (ADEC).¹⁴ Despite these decreases, FNSB is still a non-attainment area with respect to the EPA thresholds.

Here we present sub-hourly aerosol concentration and composition data from an Aerosol Chemical Speciation Monitor (ACSM) stationed in downtown Fairbanks from January 2020 through February 2022. The long duration of the deployment, the high time resolution, and the chemical composition information make this a novel dataset. Insights on air quality in this unique region may also apply to other urban areas that experience extreme cold and limited photochemistry associated with wintertime in high latitudes. These types of conditions have been under-studied.

In this paper we present our time series of speciated aerosol measurements, including source-resolved OA factors; we explore diurnal trends of aerosol species across seasons; and we shed light on the possible underlying reasons for the unexplained high particulate SO_4 levels previously observed. Our

findings are relevant to the air quality literature on northern latitude urban areas, of interest to the population of FNSB, and important for understanding the context for the 2022 winter intensive ALPACA field campaign (Alaskan Layered Pollution and Chemical Analysis).¹⁵

2 Methods

2.1 Measurement location & spatial context

The primary focus of this paper is data collected from an ACSM instrument, stationed in downtown Fairbanks, AK. Our measurement campaign took place from mid-January 2020 to the end of February 2022. Due to instrument downtime and repairs, we have a few noteworthy gaps in our measurements: during July 2020, and between April and October of 2021. Despite these missing data periods, we have coverage across all four seasons, and three winters. Winter is the main focus of this paper as Fairbanks' local air quality problems are largely a wintertime phenomenon.^{8,16,17} Fairbanks also can be subject to pollution episodes from regionally- and long-range transported wildfire smoke¹¹ during summer, but the scope of that problem is not easily addressable through local action. No measurements were made during the above-mentioned gaps, but the dataset is otherwise continuous.

The ACSM was installed at an existing air quality monitoring site operated by ADEC. The site is part of the US EPA's National Core (NCore) monitoring network, and is situated in downtown Fairbanks along the banks of the Chena River. Fig. 1a shows the location of the sampling site within the domain, and its proximity to major roadways and population density. We refer to the measurement site as the "NCore" site from herein.

Downtown Fairbanks has a history of impacts from traffic emissions.¹⁸ FNSB was designated as a non-attainment area for carbon monoxide (CO), largely due to automobile emissions,

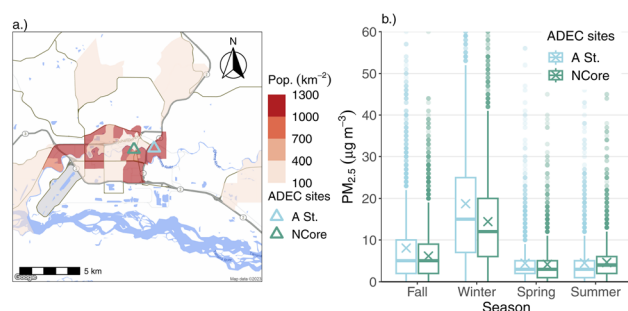


Fig. 1 Land use context for the ACSM measurements, which was stationed at the Alaska Department of Environmental Conservation (ADEC) NCore field site. (a) Overview map of Fairbanks that shows reference monitor location sites relative to population density and major roads. Population density is shown at the level of census block groups per the 2020 U.S. Census. (b) Seasonal comparison of hourly $\text{PM}_{2.5}$ concentration data at the ADEC NCore and A Street monitoring sites for the duration of the ACSM measurement time period. Mean concentrations are marked with an "x." Note: the Y-axis is cut off at $60\text{ }\mu\text{g m}^{-3}$ to emphasize differences between median and mean concentrations between the two sites, which does exclude a few outlier points above this cut-off.



and the NCore site was established in 2011 on account of these legacy reasons. Another ADEC PM_{2.5} monitoring station was established in 2019 (“A Street”), about 1.7 km east of the NCore site, due to concerns over high PM_{2.5} levels in residential areas from woodsmoke emissions that weren't fully captured by measurements at NCore. The A Street monitoring site was created as part of the State Implementation Plan (SIP) established to address the perennial non-attainment of the area. Recent mobile monitoring work suggests that the downtown site is relatively more impacted by local traffic emissions and less by local biomass-burning emissions during winter in Fairbanks compared to nearby residential areas (e.g., the neighborhood containing the A Street site).¹⁹ The highest PM_{2.5} concentrations are typically found in North Pole, a neighboring town approximately 20.6 km southeast of downtown Fairbanks. Wintertime PM_{2.5} concentrations in North Pole are out of attainment, and have a larger relative contribution from woodsmoke to total PM.⁷ The focus of this paper, however, is air quality in downtown Fairbanks.

2.2 Aerosol chemical speciation monitor

2.2.1 Measurements. The ACSM instrument has been described in detail elsewhere;^{20,21} it is a scaled-down version of the Aerosol Mass Spectrometer (AMS). In brief, the instrument allows for real-time, on-line chemical speciation of non-refractory aerosol components. Non-refractory (“NR”) is operationally defined by the ability of a component to rapidly vaporize upon impaction onto the 600 °C tungsten vaporizer near the ACSM's ionization region. The main NR aerosol species measured by the ACSM are organic aerosol (OA), SO₄, NO₃, NH₄, and chloride (Chl).

For this study we used an ACSM featuring a quadrupole mass spectrometer with unit mass resolution, a PM_{2.5} aerodynamic lens, and a “capture vaporizer” (CV).²² The use of the PM_{2.5} lens allows for transmission of large particles unlike the standard AMS aerodynamic lens, to facilitate direct comparison with regulatory PM_{2.5} measurements.²³ Similarly, the CV is meant to eliminate the artifact of particle bounce²⁴ attributable to some particle phase states and compositions, which can result in undercounting of PM mass for AMS and ACSM instruments compared to other PM measurements using the standard vaporizer (SV). We assume collection efficiency (CE) is equal to 1 for all components measured by the ACSM because of the CV.²⁵

We operate the ACSM at 30 minutes temporal resolution. Sub-hourly measurements of chemical composition, along with the long duration of these measurements, represent the most novel aspects of this study, as they take a large step forward in understanding the behavior of aerosol species, compared to one-in-three day, 24 h average speciated filter data from the NCore site. This allows us to explore diurnal patterns of aerosol constituents, including source-attributed PMF factors for OA, in a way that has not been done before in Fairbanks or other sub-arctic urban areas, especially across seasons. Typical ACSM operation requires alternating between measuring aerosols and gases *versus* gases-only, in order to correct the ion signal for contributions from background gases. For each 30 minutes run,

the ACSM would sample aerosol particles and gases for 15 minutes followed by 15 minutes of sampling air through a HEPA filter to remove aerosol signal. The difference in mass spectral signal is then attributed to aerosol particles.²⁶

2.2.2 Inlet. The instrument was housed in a small trailer located next to the ADEC NCore sampling trailer (internal temperature roughly 20 °C, though we did not measure this). The instrument sampled air from roughly a similar height to the NCore instruments, approximately 4 m above ground level and 1 m above the trailer roof. The inlet was downward facing and consisted of a 3 liters per minute, 2.5 μm cut cyclone (URG Corp., URG-2000-30ED). Sample air was transported to the ACSM *via* 1/2 inch O.D. (0.46 I.D.) stainless steel tubing. After entering the trailer the air passed through a Nafion dryer to maintain relative humidities below roughly 30%. Because the drying occurred inside the heated trailer with the ACSM, the sample air was heated. The resulting temperature rise (which was not measured) could result in the loss of some semi-volatile aerosol particle mass in the roughly 1 to 2 seconds residence time in the heated section, prior to measurement, but we make no corrections for this.

2.2.3 Data treatment. Due to operational constraints that included restrictions due to COVID, we were unable to apply the standard response factor (RF) NO₃ calibration to the instrument over the course of the campaign. Instead, we base our aerosol quantification on a “data calibration” correction, as follows: We use a comparison of ACSM-derived “SO₄” and measurements of total sulfur (S) from 24 h EPA NCore speciated filter measurements, where S is determined through X-ray fluorescence (XRF). We convert the total S measurement to “SO₄”, using the assumption that all S is in the form of SO₄, similar to the treatment of SO_x⁺ ions by the default ACSM fragmentation table. Using orthogonal distance regression, we compare ACSM SO₄ and XRF SO₄ (per above) to pick an airbeam (AB) reference value for our measurements that gives us a slope of 1. We then use that AB reference value to correct for any mass spectrometer signal drift over the long course of our measurements, which applies to all aerosol species measured by the ACSM. This is the same approach taken by Campbell *et al.*,²⁷ which used a portion of the ACSM SO₄ measurements we present here. We choose to perform this data calibration using SO₄ for the following three reasons: First, because of the CV for this instrument, we assume that CE ~1 for all measured aerosol components, including SO₄. Any uncertainty in SO₄ CE (due to e.g., aerosol phase state^{24,28}), which is typical in SV instruments, would make this approach much more challenging. Second, SO₄ is non-volatile and so should not suffer from any volatilization that could occur in the sampling line when going from outdoor temperature to room temperature; in wintertime in Fairbanks especially, it is possible that there may be evaporation of aerosol components upon transport indoors in the sampling line. Third, SO₄ mass spectral signals are limited to several key ions so the attribution of mass is more straightforward than with organic species.

Using the default relative ionization efficiency (RIE) for SO₄ (RIE = 1.2) and the RF value from the previous instrument calibration made prior to deployment (RF = 3.95 × 10⁻¹¹ amps



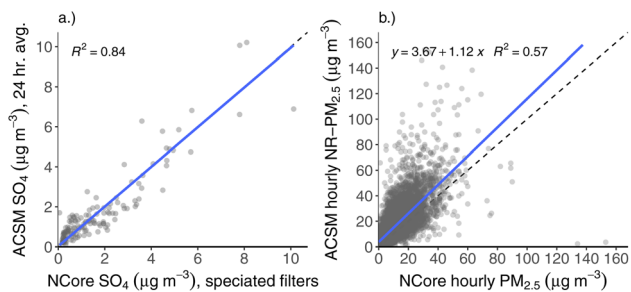


Fig. 2 Correlation plots between ACSM and ADEC NCore data. (a) Comparison of ACSM SO_4 , averaged over the 24 h period of filter collection for the speciated, one-in-three-day filter data from NCore. This plot is used to pick a single airbeam reference value such that the slope between the two SO_4 measurements is 1. (b) Comparison of ACSM total NR- $\text{PM}_{2.5}$ with hourly $\text{PM}_{2.5}$ from NCore, after choosing the optimized airbeam reference value based on the SO_4 comparison. ACSM data are averaged to the same timeframe as the NCore data for both comparisons.

$\mu\text{g}^{-1} \text{m}^{-3}$), we choose a AB reference value that results in the best agreement between ACSM and NCore filter SO_4 . We define best agreement with be when the AB-corrected ACSM SO_4 measurements the NCore filter SO_4 have a slope equal to one in comparison with one another, as shown in Fig. 2a. We then apply the AB reference value determined from this SO_4 -only comparison to the other aerosol species measured by the ACSM, as is typical.

2.3 PMF analysis

We conducted PMF analysis in order to perform source-apportionment for the OA fraction of NR- $\text{PM}_{2.5}$. PMF has been described in detail elsewhere,²⁹ including specifically for the application to AMS data.³⁰ In brief, PMF is an algorithmic technique that deconvolves the OA matrix into a linear combination of static mass spectral “factors.” We followed the common approach for applying PMF to AMS and ACSM datasets of exploring the solution space for multiple factors (1- through 6-factor solutions) and rotations (f_{peak} values from -1 to 1).³⁰ None of the factor mass spectra were constrained *a priori*, as is common with some approaches to PMF (e.g., SoFi³¹). In addition, we ran PMF on various seasonal subsets of the data, to compare solutions across seasons (e.g., summer-only vs. winter-only vs. full dataset). We used the ACSM analysis panel (v.1.6.1.6) to prepare both the OA MS and error matrices. We performed PMF using the PMF2.exe algorithm with the PMF Evaluation Tool (PET v.2.08) with unit mass ions up to m/z 149. For organic matrix and error matrix preparation we applied the following through the PET analysis panel: (1) removed NaNs and 0 columns, (2) removed “Bad” m/z s using a signal-to-noise ratio (SNR) <0.2 and downweighted “weak” m/z s with SNR <2 .

2.4 Diurnal pattern calculations

For diurnal pattern calculations, we have segmented our data in a few important ways. First, each quantity (e.g., aerosol species concentration, temperature, etc.) that we present diurnal patterns for we calculate the mean value within each hour of the

day. For example, all data between 9:00 am and 10:00 am would be used to calculate the mean value for the 9 o'clock hour, within a given categorical filter (e.g., Spring + weekend). Weekend vs. weekday data filtering was done based on Monday–Friday being weekdays, and Saturday–Sunday being weekends, and all distinctions between days used a 12:00 am cut time. Additionally, we split our data into four seasons, which we operationally defined in the following way: “Winter” was all data from December, January, and February; “Spring” was all data from March, April, and May; “Summer” was all data from June, July, and August; and “Fall” was all data from September, October, and November. These same operationally-defined seasons are used by Ye and Wang in a previous study of Fairbanks’ air quality. We include the previous year’s December with January and February when considering the winter season (e.g., December 2020 and January–February 2021 comprise “Winter 2021”).

2.5 Ancillary data sources

There were multiple data sources important for this study beyond the ACSM dataset. We gathered hourly meteorological data (wind direction, wind speed, and ground-level air temperature) collected at Fairbanks International Airport using the *worldMet* package for R, which scrapes the NOAA ISD (<https://www.ncdc.noaa.gov/isd>). These meteorological data are imperfect for use in this context, given that the airport is removed by roughly 4 miles from the NCore site downtown, and may reflect somewhat different wind patterns. Nonetheless, we use the airport meteorological data in the absence of other data and hope that they reflect wind and temperature behavior in downtown Fairbanks. Additionally, data from this weather station that have winds below 1.5 m s^{-1} are considered “calm”, and are given the value of 0 m s^{-1} . Observations from this weather station are not always made at even intervals, however, and so we compute hourly means in order to standardize the time series, which results in some hourly-average wind speeds below 1.5 m s^{-1} . For this reason, we display both median and mean wind speed values in diurnal plots (e.g., Fig. 7), where median wind speeds of zero indicate that winds are “calm” for the majority of the time.

We also downloaded the one-in-three-day, 24 h speciated $\text{PM}_{2.5}$ filter data from the EPA website (https://aqs.epa.gov/aqsweb/airdata/download_files.html), which was used for the “data calibration” procedure with ACSM SO_4 described above (shown in Fig. 2a). These speciated filter data run through the duration of the ACSM dataset. We downloaded hourly BAM $\text{PM}_{2.5}$ data measured at both aforementioned ADEC sites (A Street and NCore), which is used for Fig. 1b and 2b. Hourly ADEC $\text{PM}_{2.5}$ data were retrieved from the EPA website for measurements taken prior to Jan. 1, 2022 (data up to Jan. 1, 2022 are considered finalized by ADEC). Hourly $\text{PM}_{2.5}$ data from Jan. 1, 2022 to the end of the campaign (Feb. 25, 2022) were downloaded directly from the ADEC website (<https://dec.alaska.gov/air/air-monitoring/alaska-air-quality-real-time-data>). ADEC declares these data as preliminary, but we still use



them for the comparison here and find no meaningful difference of the preliminary data with respect to the ACSM compared with the finalized data. Lastly, we downloaded gas criteria pollutant data (ozone (O₃), sulfur dioxide (SO₂), and CO) made at the NCore site from the EPA website.

2.6 Data analysis software used

We used multiple data analysis software tools for this study. We used the ACSM Local (v1.6.1.6) data processing package written for Igor Pro (v8) to process raw ACSM data files and to apply the ion transmission correction. To perform PMF analysis, we used the PMF Evaluation Tool (PET, v2.08A) for Igor Pro. Both ACSM Local and PET are created and maintained by Aerodyne Research, Inc. We performed all other data analysis and visualizations using R software (v3.6.3). In R, we used the *worldMet*, *tidyverse*, *ggmap*, *scales*, and *lubridate* libraries.

3 Results & discussion

3.1 PM_{2.5} comparisons

3.1.1 ADEC NCore and A Street PM_{2.5} sites comparison.

Fig. 1b shows the seasonal variations of regulatory PM_{2.5} measurements made by ADEC at the NCore and A Street monitoring sites. Median wintertime PM_{2.5} concentration for the full duration of ACSM deployment at A Street was ~1.25 times higher than NCore (15 and 12 μg m⁻³, respectively). Median summertime PM_{2.5} concentrations were higher at NCore compared to A Street, however. The higher wintertime and lower summertime concentrations at A Street compared to NCore are consistent with downtown being more impacted by traffic emissions relative to the near-downtown residential areas, which are more impacted by residential home heating emissions. An analysis of PM composition differences between the two sites would make the above case more definitively, but these results are also consistent with the relative differences shown by mobile monitoring in Robinson *et al.*¹⁹

3.1.2 ACSM NR-PM_{2.5} vs. ADEC NCore PM_{2.5}. Fig. 2b shows the comparison between total NR-PM_{2.5} from the ACSM with ADEC hourly BAM PM_{2.5}. As discussed in the previous section, we only used the SO₄ comparison to select an AB reference value, and then applied that AB reference value to all other aerosol species measured by the ACSM to calculate total NR-PM_{2.5}. The slope of linear regression fit line is 1.12, with an R² value of 0.57. This is good agreement given the typical uncertainty envelope for AMS measurements of ±30%, lending confidence in the “data calibration” procedure we performed using SO₄. However, we expect the slope of this comparison to be less than 1, given that we do not have hourly measurements of black or elemental carbon, which we would either add to the ACSM NR-PM_{2.5} (or subtract from the NCore PM_{2.5}) for a more precise comparison between the two instruments. Previous studies show that wintertime BC comprises up to between 5–10% of the PM in Fairbanks on average.^{3,19} Given that the NCore PM_{2.5} measurement captures BC PM, we see that these fit results are slightly biased in the direction opposite of what is expected, though still within the AMS uncertainty envelope. We

speculate that some of this difference may be due to evaporation of semi-volatile aerosol components from the filter tape of the BAM instrument over the roughly hour-long timeframe that they deposit for.

3.2 PMF solution

3.2.1 PMF factor descriptions. We identified three distinct OA mass spectral factor profiles that best describe the variability in the OA matrix using PMF. We label these profiles as biomass-burning OA (BBOA), hydrocarbon-like OA (HOA), and oxygenated OA (OOA), which are shown in Fig. 3 and discussed in detail below. We named each factor based on (1) similarity to previously-published factor profiles, which we downloaded from the AMS spectral database (<https://cires1.colorado.edu/jimenez-group/AMSsd/>), and (2) expected diurnal behavior, as discussed below in the subsequent subsection.

The HOA factor profile is highly similar to other previously published HOA profiles in the literature, and generally represents non-biomass fuel combustion emissions and/or lubricating oil emissions. In the Fairbanks context, this could possibly include OA from gasoline vehicle exhaust, diesel vehicle exhaust, and/or furnace heating oil emissions. The average cosine similarity of our HOA factor to three other published HOA factors was 0.920 (Zhang *et al.*:³² 0.921, Ng *et al.*:³³ 0.925, and Crippa *et al.*:³⁴ 0.913), which is high in the context of comparing PMF factors across studies.³⁰ It exhibits the common “picket fence” fragmentation pattern of reduced hydrocarbon molecules, and has prominent peaks at *m/z* 43, *m/z* 55, and *m/z* 57, which typically contain substantial contributions from reduced fragments C₃H₇⁺, C₄H₇⁺, and C₄H₉⁺, respectively. Without high-resolution mass spectra, we cannot definitively conclude that all (or nearly all) of the signal in the HOA factor derives from non-oxygenated C_xH_y fragments, but ratios of *m/z* 57 : *m/z* 55 and *m/z* 43 : *m/z* 41 above 1 are both typical of canonical HOA mass spectra, and is true for our HOA factor as well. Based on the diurnal pattern (discussed in more depth below) it seems likely that our HOA factor is largely from vehicle emissions. We also see that HOA correlates most

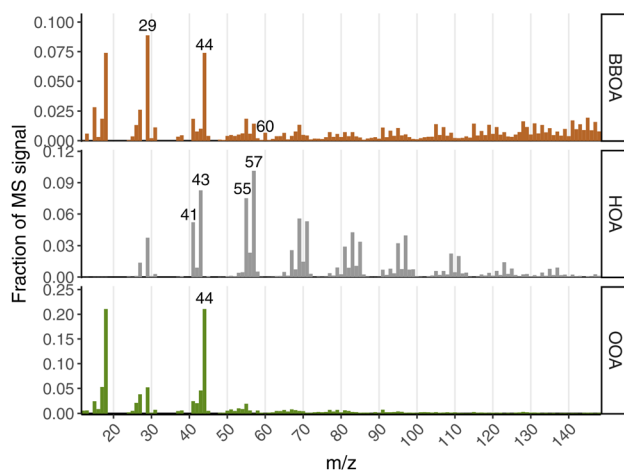


Fig. 3 Normalized mass spectra of each PMF factor for the 3-factor solution.



strongly ($R^2 = 0.5$) with CO during wintertime compared to all other aerosol species and OA factors (see Fig. S2†).

Our BBOA factor likely represents fresh wood-burning PM emissions. It contains almost all (92%) of the signal at m/z 60, which is typically uniquely comprised of the $C_2H_4O_2^+$ ion. This is the main unique ion fragment in AMS mass spectra of levoglucosan,³⁵ which is a marker of fresh biomass burning aerosol emissions.³⁶ The BBOA factor also contains a high fraction of mass at larger m/z (e.g., $m/z > 70$) compared to the OOA factor, which is a common feature of BBOA factor profiles in comparison to OOA from previous studies.^{33,34} Compared to three other BBOA factors reported in the literature, we computed an average cosine similarity of 0.767 (Ng *et al.*:³³ 0.734, Crippa *et al.*:³⁴ 0.770, and Crippa *et al.*:³⁴ 0.797). This similarity value is lower than that of our HOA factor compared against other HOA factors from the literature, but is expected to be lower given that biomass-burning tends to have more-variable composition,³⁵ and the CV tends to fragment molecules to a greater degree compared to the library mass spectra from SV instruments. Indeed, we see this with m/z 60 in this instrument: despite our BBOA containing the large majority of the m/z 60 signal, and there being no shortage of actual biomass-burning PM in Fairbanks, we see that the fraction of this marker ion (f_{60}) is substantially lower ($f_{60} = 0.0065$) than that of BBOA profiles from previously published studies using the SV ($f_{60} \sim 0.15$ – 0.04). Similarly, the fraction of m/z 73 (another marker for BBOA emissions³⁵) to total OA ($f_{73} = 0.002$) in the BBOA factor is also lower than in SV instruments, though we do see that m/z 60 and m/z 73 are relatively well-correlated ($R^2 = 0.85$ during winter). Despite these differences with the reference spectra, interpreting this PMF factor as “BBOA” is reasonable based on its spectral characteristics in the context of CV vs. SV differences, as well as its diurnal pattern, which we discuss in a subsequent section.

OOA is the least straightforward factor of the three to interpret, despite being very spectrally similar to other previously-published OOA factor profiles. The average cosine similarity between our OOA profile and three other studies was 0.953 (Zhang *et al.*:³² 0.918, Lanz *et al.*:³⁷ 0.987, and Ng *et al.*:³³ 0.954). The most prominent peak in the OOA MS is at m/z 44, which is dominated by the CO_2^+ ion, and comes from oxygen-containing groups such as e.g., carboxylic acids, alcohols, carbonyls. From mid-latitude studies using AMS PMF, OOA factors are often attributed to secondary OA (SOA). In Fairbanks, it is not clear that there should be much photochemically-formed SOA during wintertime, given the low solar insolation and low O_3 .¹⁷ Nonetheless, OOA is the dominant PMF factor, making up the majority of OA across all seasons. It comprises 53% of the OA in winter, and is even higher in all other seasons.

Despite the spectral similarity, we do not interpret our OOA factor as directly comparable to other OOA factors from the literature that represent SOA. It likely represents a mix of real biomass-burning OA and some kind of oxidized OA, which could also be biomass-burning OA just with more atmospheric aging. The three-factor solution identifies a “fresh” BBOA factor (discussed above), which contains all of m/z 60; this helps us establish a lower-bound on what the contribution of primary

biomass-burning emissions is to total OA (~31%) during wintertime. There are a few reasons why this is likely just a lower-bound, and that some real biomass-burning PM gets apportioned into the OOA factor, however. First, as discussed in more depth in Section 3.5, the diurnal patterns of OOA and BBOA are very similar, indicating that there is likely not any hard boundary between the two factors with respect to sources. Second, there is strong temporal correlation between OOA and BBOA in wintertime ($R^2 = 0.81$), which also may indicate similar source(s) (see Fig. S1†). Third, we know that the limitations of our ACSM—due to the unit-mass resolution mass spectrometer and the CV—will tend to decrease the complexity of the mass spectrum, and thus homogenize spectral differences between factor profiles. Specifically, the CV tends to lead to more fragmentation,³⁸ and Zheng *et al.*³⁹ show that a CV-ACSM apportioned more mass to SOA versus POA using PMF for a side-by-side comparison with a SV-AMS. Fourth and lastly, we know from previous AMS studies that BBOA spectra tend to be more variable than other primary OA types e.g., HOA, which reflects how variable biomass burning emissions themselves are. It is reasonable to expect variable composition across the ensemble of biomass burning emissions in Fairbanks, given the likely variability in stove types, stove operation, stages of fire (e.g., smoldering vs. flaming), and fuels used,^{40–42} and thus a single PMF factor may not fully capture all biomass-burning emissions in this context. So, we label the PMF factor as “OOA” given its spectral similarity to previous results, but think it likely represents a mix of aerosol types that likely includes some biomass-burning PM during winter. Indeed, a previous study from Kotchenruther identified an “aged woodsmoke factor” in their source-attribution for PM in Fairbanks,⁶ which our OOA factor may reflect during wintertime.

OOA is the dominant PMF factor during summertime, comprising roughly 82% of OA. We do not see a strong correlation between OOA (or any other aerosol species or OA factor) and O_3 during summertime (see Fig. S2†). This may be again due to the fact that our OOA factor represents a mix sources, which is summer may include biogenic SOA, aged woodsmoke from long-range transport, and/or other aged OA. Nonetheless, as mentioned by Campbell *et al.*,²⁷ the large amount of OA, which is dominated by OOA and, to a lesser extent, BBOA, likely provides a large, mildly hygroscopic volume for aqueous-phase reactions to occur even in these cold and dark conditions.

3.2.2 Three-factor PMF solution vs. other solutions. We think that the three-factor PMF solution best describes the OA in Fairbanks, as measured by this ACSM. Solutions with four or more factors showed clear signs of “factor-splitting”, where two or more factors would have very similar diurnal and mass spectral characteristics, with the exception of a few major ion fragments being represented solely within only one of these factors. For example, higher-order (≥ 4 factors) PMF solutions would have two factors with the mass spectral attributes of HOA and diurnal patterns reflective of typical driving patterns, but one factor would have m/z 57 with no m/z 55, and the other would have m/z 55 with no m/z 57 (see Fig. S3†). It is very unlikely that these two factors could represent different aerosol types in the atmosphere, as these two unit-mass ions almost



always have contributions from molecular fragments that originate from the same parent molecules (*e.g.*, C_4H_7 at m/z 55 and C_4H_9 at m/z 57). We see similar splitting with the factor that we have designated as “BBOA” for higher-order PMF solutions as well. Thus we do not present any four-or-higher factor solutions as being valid for describing OA in Fairbanks for this dataset. The three-factor solution additionally had a lower Q/Q_{exp} (~ 20) compared to higher-order solutions.

The two-factor solution yields an HOA factor that is very spectrally similar to the HOA factor from the three-factor solution (cosine similarity of 0.999), as well as a non-HOA factor that has clear signatures of both biomass burning PM (*e.g.*, contains all m/z 60) and oxidized aerosol (*e.g.*, large peaks at m/z 44 and m/z 18). Cosine similarity of this non-HOA factor compared to the BBOA and OOA factors from the three-factor solution was high (0.920 and 0.989, respectively), which also indicates shared chemical characteristics. This mixed factor accounts for roughly 80% of the OA mass during wintertime, which is roughly the same as the sum of the OOA and BBOA factors (81%) from the three-factor solution. Thus, both the two- and three-factor solutions indicate that HOA contributes roughly 20% of the

wintertime OA mass, with the other roughly 80% being some mix of fresh biomass-burning emissions and more-oxidized OA. The three-factor solution, however, offers more specificity and is more quantitative with respect to biomass burning emissions. For these reasons, we present the three-factor PMF solution as that which best describes the mix of OA in Fairbanks, as measured by our CV-ACSM.

3.3 Time series and seasonal trends

Fig. 4 shows the full ACSM time series for the duration of deployment. Panels show, from top to bottom, (a) the relative contribution of each aerosol species and OA factor to the total NR-PM_{2.5} mass (monthly average), (b) the monthly mean concentrations of each aerosol species with OA separated into component PMF factors, (c) ambient ground-level temperature, and (d) the raw ACSM time series of the major aerosol species, which is at 30 minutes temporal resolution.

Multiple aspects of the time series panel in Fig. 4 are notable: first and most obviously, concentrations of all aerosol species are higher during the cold winter months compared to spring,

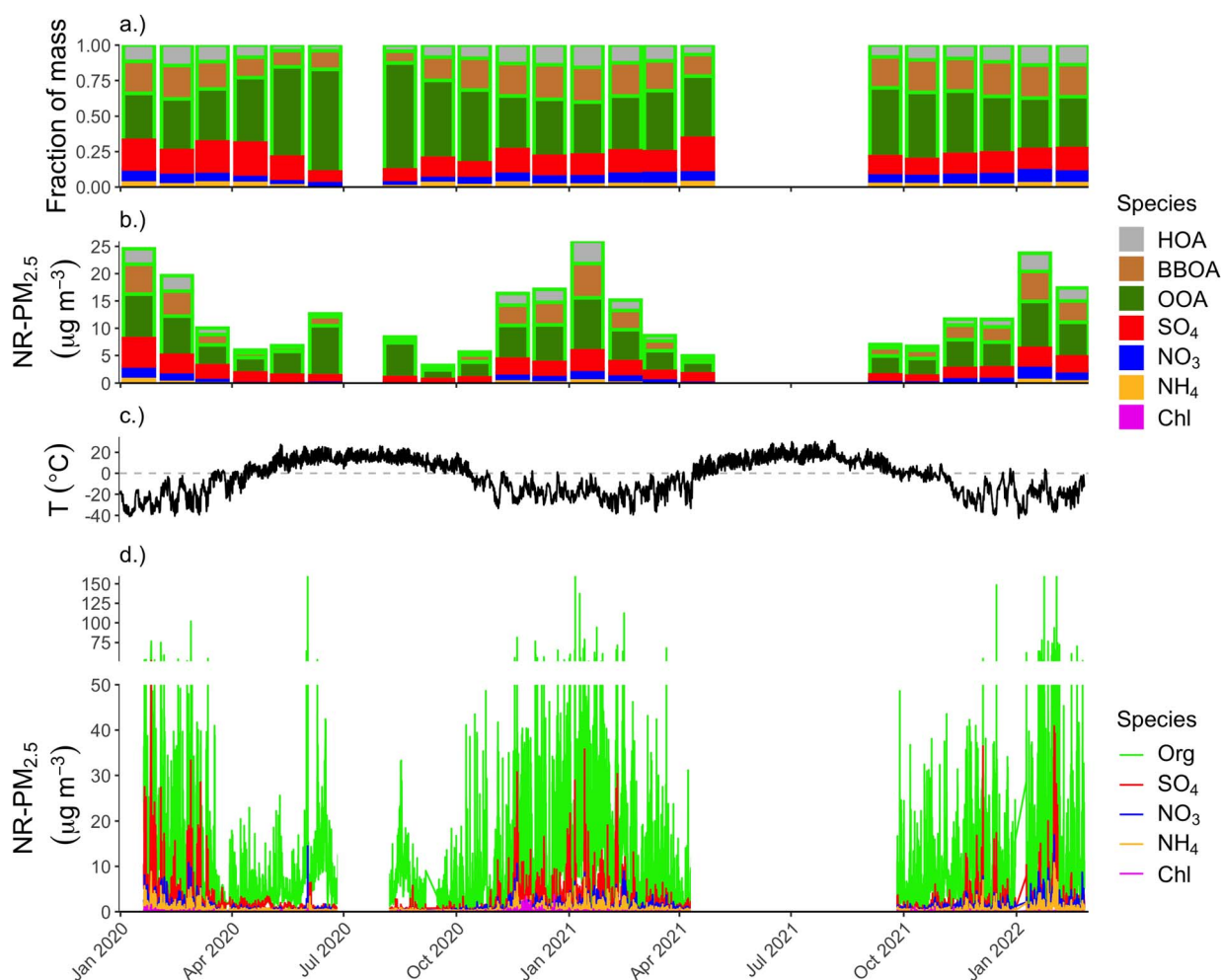


Fig. 4 Time series plot of the full dataset (a) monthly-average fractional contribution of each aerosol species and OA PMF factor to total NR-PM_{2.5}. (b) monthly-average mean concentration of all aerosol species, where the sum of all bars is the total NR-PM_{2.5} concentration. (c) Ambient temperature, measured at the Fairbanks International Airport. (d) Raw time series of 30 minutes average ACSM data for all major aerosol species.



summer, and fall months. During winter months, roughly 13% of our half-hourly measurements were above the 24 h PM_{2.5} NAAQS of 35 $\mu\text{g m}^{-3}$, and roughly 61% of the measurements were higher than the annual average PM_{2.5} NAAQS of 12 $\mu\text{g m}^{-3}$. Most of the high-concentration spikes in concentration above the 24 h NAAQS are driven by sharp increases in OA, but there are many instances in the winter months of SO₄ “spikes” exceeding 10 $\mu\text{g m}^{-3}$, and going as high as 52 $\mu\text{g m}^{-3}$. This SO₄ behavior is starkly different from mid-latitude urban areas where SO₄ aerosol tends to be regional in spatial scale,⁴³ and thus concentrations typically do not vary on the order of $\sim 10 \mu\text{g m}^{-3}$ on short, sub-hour-long timescales that we see in the winter in Fairbanks. This behavior in the SO₄ time series implies either rapidly-changing boundary layer conditions and/or local emissions of SO₄ or SO₄-forming precursors that convert to SO₄ on fast (sub-hour) timescales.

In the top two panels of Fig. 4 we summarize the seasonal patterns of NR-PM_{2.5} concentration and composition by showing month-long averages. The second-from-top panel (b) shows monthly average concentrations for each aerosol species, with the height of each bar representing species mass concentration and the sum of all bars representing the total NR-PM_{2.5} mass concentration. These data show that NR-PM_{2.5} mass in winter 2022 during the ALPACA campaign is roughly similar to the previous two winters. Average total NR-PM_{2.5} concentrations for January for 2020, 2021, and 2022 were 24.8, 26.0, and 23.8 $\mu\text{g m}^{-3}$, respectively. Across the full winter (Dec.–Feb.), average concentrations for 2020, 2021, and 2022 were 19.3, 17.7, and 20.7 $\mu\text{g m}^{-3}$, respectively.

Average concentrations during spring, summer, and fall periods were much lower than winter, at 7.5, 10.5, and 8.82 $\mu\text{g m}^{-3}$, respectively. Composition varied between winter and these other seasons as well. OA had the highest fractional contribution in the summer (89%) and the lowest in the spring (70%), though OA dominates NR-PM_{2.5} mass across all seasons. Roughly 76% of NR-PM_{2.5} is OA in the wintertime.

Summer air quality in Fairbanks can be heavily impacted by transported wildfire smoke. Our measurements only provide information on one summer (summer 2020), and so may not be indicative of summer air quality in Fairbanks more generally. Total PM_{2.5} measured by ADEC at both A Street and NCore sites showed summer PM_{2.5} concentrations during summer 2021 being roughly double that of summer 2020. According to the Alaska Interagency Coordination Center,⁴⁴ within Alaska there were roughly 180 000 acres burned during 2020, the large majority of which happens during summer. In 2019 and 2022, by contrast, roughly 2.7 and 3.2 million acres were burned in Alaska, respectively. Obviously comparing total acres burned across Alaska does not translate directly to air quality impacts in Fairbanks, specifically, but we use this as a proxy that suggests that the summer data we report here are likely on the low end of wildfire-related air quality impacts compared to other summers.

The OOA factor is the largest contributor to OA across all seasons, though varies considerably over the course of the year. In summer, OOA represents 82% of all OA (and 73% of total NR-PM_{2.5}). In winter, OOA is 53% of OA, while BBOA is 31% and HOA is 15%. If the HOA and BBOA factors represent the aerosol

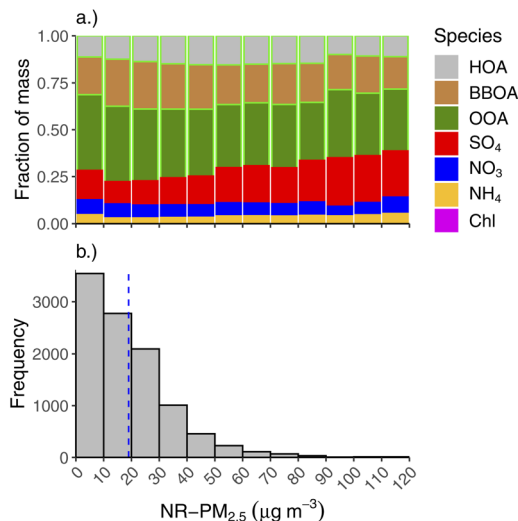


Fig. 5 Aerosol composition as a function of total 30 minutes NR-PM_{2.5} concentration during wintertime. (a) Top panel shows the fractional contribution of each aerosol species and OA PMF factor to the total mass for 10 $\mu\text{g m}^{-3}$ concentration bins. (b) Bottom panel is a histogram illustrating the frequency of NR-PM_{2.5} concentrations during winter. Mean wintertime concentration is marked with a blue line on the histogram.

components that can be reduced due to behavior changes and/or locally-implemented emissions controls, roughly half of the OA in the wintertime could be addressable through modifiable factors. However, as mentioned in the previous section, we think it is likely that biomass burning emissions are represented in PMF space by both the BBOA and OOA factors, and thus BBOA's fractional contribution to OA of 31% represents a lower-bound of how much of the OA is attributable to biomass burning. Inclusion of some fraction of the OOA would increase the fraction of total OA attributable to biomass-burning emissions sources.

The second highest contributor to NR-PM_{2.5} is SO₄, which makes up 13% of NR-PM_{2.5} during wintertime and only 6% during summer. SO₄ fraction is highest in the spring (19%), though this is when absolute concentrations of NR-PM_{2.5} were lowest. During winter months, SO₄ makes up fractionally more of the total NR-PM_{2.5} at high mass concentrations, as shown in Fig. 5a. Fig. 5b shows a histogram of wintertime total NR-PM_{2.5} concentration (bottom panel), binned by intervals of 10 $\mu\text{g m}^{-3}$. The top panel of 5 shows the average fractional contribution in each concentration bin for each aerosol species and OA factors to the total. Thus, SO₄ is a major driver of the highest concentrations we see in Fairbanks in the winter. Note, for each aerosol species, the absolute concentration also tends to increase for each concentration bin (*e.g.*, there is more BBOA at higher concentrations than lower concentration, but its fractional contribution might be lower due to increases in other constituents, like SO₄).

3.4 Aerosol concentrations and ambient temperature

There is a strong relationship between ambient temperature and PM concentration and composition in Fairbanks,



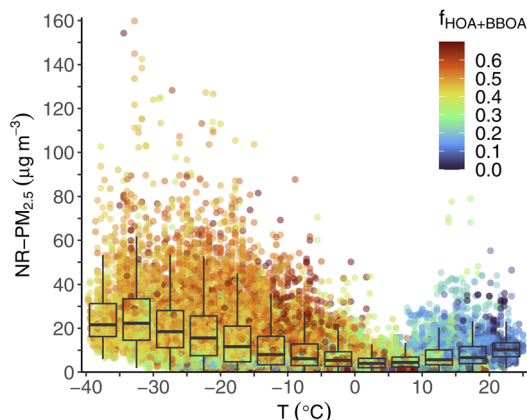


Fig. 6 Total 30 minutes NR-PM_{2.5} concentration as a function of ambient temperature. Points are colored by the fraction of OA from the two primary PMF factors (HOA & BBOA), labeled $f_{\text{HOA+BBOA}}$. Overlaid box plots indicate the distribution of NR-PM_{2.5} concentration within 5 °C temperature bins.

illustrated in Fig. 6. Obviously temperature is highly correlated with season, and we have already shown that concentrations are higher in winter compared to summer. Nonetheless, important behavior is revealed by looking at PM concentrations *vs.* temperature across the entire dataset. Fig. 6 shows all ACSM measurements of total NR-PM_{2.5} concentration plotted against ambient temperature, and colored by the fraction of the sum of HOA and BBOA PMF factors to total OA. We overlay box plots of NR-PM_{2.5} concentration within temperature bins of 5 °C increments.

The peak concentrations we measured were almost exclusively at low temperatures: for example, almost all (93%) NR-PM_{2.5} >60 µg m⁻³ were below -10 °C. Median concentrations within 5 °C bins also increased as temperatures decreased. The highest median 5 °C bin concentration was 22.3 µg m⁻³ at -35 to -30 °C. We observed similar patterns of increasing concentrations at colder temperature for other quantities (*e.g.*, 75th percentile, minimum concentration, *etc.*).

PM composition is also correlated with temperature, as cold periods tend to be more impacted by primary source contributions than warm periods. We illustrate this by coloring the points on Fig. 6 by the fraction of OA from one of the two primary OA factors, HOA or BBOA, $f_{\text{HOA+BBOA}}$. During warm periods, *e.g.*, 15 °C and above, $f_{\text{HOA+BBOA}}$ tends to be low (<0.20), as indicated by green or blue-colored points. During colder periods, especially when concentrations are highest, most of the points are yellow/orange/red, indicating higher contributions from primary sources ($f_{\text{HOA+BBOA}} > 0.45$). As others have documented,^{16,17,19} meteorological conditions (*e.g.*, surface temperature inversions that lead to low dispersion) play a key role in trapping pollution in Fairbanks, leading to high PM concentrations. During these cold temperature periods, primary emissions make up a higher relative contribution to PM concentrations. This is consistent with high demand for activities that drive primary emissions (*e.g.*, home heating using wood stoves, car idling for warming during startup) as well as

gas-particle partitioning that heavily favors the condensed phase at ultra-low temperatures.

3.5 Diurnal behavior of aerosol species

The diurnal behavior of aerosol species varied by season as well, which we show in Fig. 7. This plot shows the mean hourly concentrations of total NR-PM_{2.5}, each major aerosol species, and the three OA PMF factors. Each hourly mean species concentration is calculated for each season and day type (weekday *vs.* weekend). The panel groupings are meant to allow for quick visual assessments of: (1) the contribution of OA to total NR-PM_{2.5} as a function of hour of day across seasons, (2) comparisons of the concentrations and daily patterns between the three OA species, and (3) comparisons of the concentrations and daily patterns between each of the inorganic species.

Across all seasons, we see that total NR-PM_{2.5} and OA follow the same broad diurnal trend as one another. This is unsurprising given that OA makes up the majority of aerosol mass in Fairbanks in all seasons. We also see that NR-PM_{2.5} and OA have distinct diurnal patterns depending on the season.

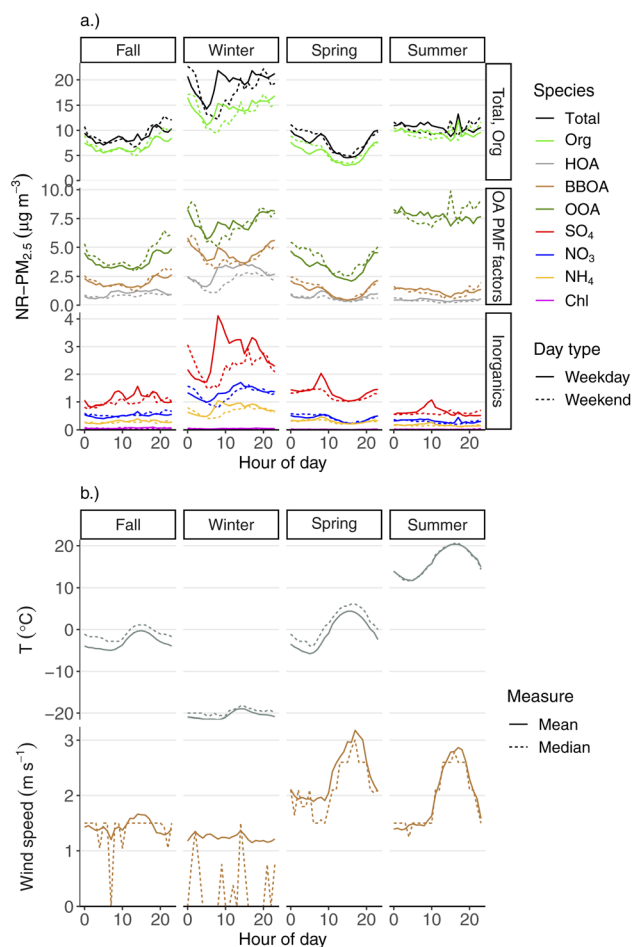


Fig. 7 Seasonal diurnal plots of hourly (a) mean aerosol species and OA PMF factor concentrations, and (b) mean and median temperature and wind speed. Each diurnal pattern is calculated for each season (Fall, Winter, Spring, and Summer) and type of day (weekend *vs.* weekday).



3.5.1 Diurnal patterns in wintertime. In winter, total NR-PM_{2.5} concentrations tend to be high throughout the course of the day and most of the overnight period. The main diurnal feature of the total NR-PM_{2.5} wintertime trace is that concentrations drop to their lowest point in the early part of morning (around 5 am) having decreased during the few hours prior. This is likely driven by a combination of relatively lower primary source emissions during the early morning (less driving, possibly less home heating) and relatively higher winds, causing increased dispersion. This part of the very early morning is one of the few periods of the day during winter where wind speeds are more often than not non-calm, as shown in Fig. 7b, and more starkly in Fig. 8b for the month of January. Fig. 7b shows the diurnal patterns of wind speed and temperature by season. We see that mean wind speeds are lowest in the winter (roughly 1 m s⁻¹), and that the mean value is relatively constant over the course of the day. However, the median windspeed for many hours of the day is below the “calm” threshold, meaning less than 1.5 m s⁻¹, which we display as 0 m s⁻¹ on the plot. The early morning (after midnight and before 5 am) is one of the few

times when the median wind speed is above the calm threshold, meaning that this early morning period has consistently higher dispersion relative to most of the rest of the day. This early morning feature of decreasing concentration that we see for NR-PM_{2.5} is present for every individual aerosol species to some extent as well.

Concentrations of the primary OA factors (HOA and BBOA) are at their lowest point during early mornings, with pronounced increases at other parts of the day. This also contributes to the overall wintertime diurnal pattern of total NR-PM_{2.5}, in addition to the meteorological influence discussed above. Both HOA and BBOA have morning increases (around 8 am), suggesting higher activity from primary emitters during this time, which is in line with expectations of both morning commuting and, possibly, higher wood stove use following morning wake-up. The morning increase of HOA is shifted towards later hours in the day during weekends compared to weekdays, which is likely reflective of different weekend vs. weekday driving patterns. During weekdays, the diurnal HOA concentration peaks around typical evening commute time, which is consistent with its source being primary vehicle emissions. HOA concentrations tend to decline starting in the mid-to-late evening. If our HOA factor had a significant contribution from heating-oil combustion particles from residential heating, we might expect to see an increase in the evening, similar to BBOA. These diurnal and weekday/weekend patterns, and the mass spectral profile (see Fig. 3), are what leads us to deduce that this HOA factor is largely vehicular in origin. However, it seems likely that heating-oil combustion exhaust would have a very similar mass spectrum in this instrument to lubricating oil or diesel exhaust, and so we cannot completely rule out a contribution from this source to our HOA factor.

The diurnal behavior of BBOA during wintertime evenings is the opposite of HOA. BBOA concentrations increase in the mid-to-late evening, likely reflecting increased wood stove use as people are in their houses preparing for sleep. Wood stoves also tend to have higher emission factors during startup compared to steady-state use (due to *e.g.*, cold fuels, cold catalyst, *etc.*),⁴⁵ which may also explain the BBOA increase in the early evening compared to later in the evening and early morning. Conversely, overnight wood stove use (while people are sleeping) may tend to have lower emissions compared to evening hours, which could also help explain the decrease of BBOA signal in the early mornings, in addition to increased dispersion as discussed above. OOA follows a similar diurnal pattern to BBOA, which is one of the reasons why we assume that OOA also represents real biomass-burning PM. Early afternoon is when we see the daily minimum for BBOA, which likely reflects relatively reduced wood stove use due to both warmer mid-day temperatures and/or people being in their homes less.

Interestingly, SO₄ shows similar patterns to HOA, both diurnally and with respect to weekday/weekend behavior. Like HOA, there is a strong morning increase, around typical morning work and school commute times. Similarly, it stays high throughout the day, and then decreases following evening commute times. The weekend/weekday differences for HOA and SO₄ are similar as well, with both the morning increase and

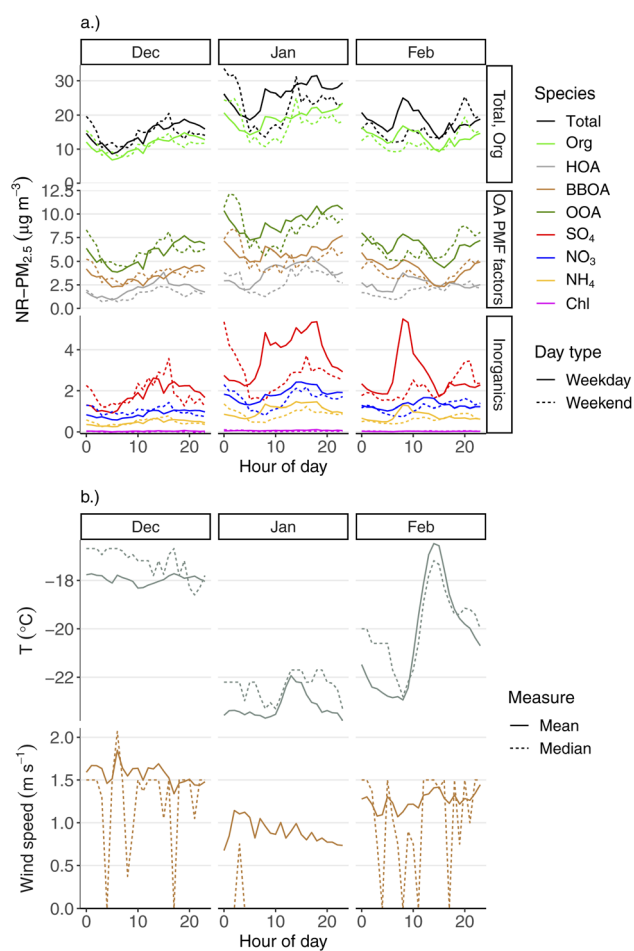


Fig. 8 Monthly diurnal plots of hourly (a) mean aerosol species and OA PMF factor concentrations, and (b) mean and median temperature and wind speed. Each diurnal pattern is calculated for each wintertime month (December, January, and February) and type of day (weekend vs. weekday).



evening decrease being shifted later on weekends. The diurnal similarity to HOA suggests that some meaningful fraction of the sulfate could be linked to vehicle emissions, either through direct SO_4 emission or precursor emissions followed by fast secondary chemistry.

Within the winter season, there are distinct diurnal patterns for aerosol species, wind speed, and temperatures for each month as well. These patterns are shown in the same way as the previous plot, but grouped by winter month (December, January, February) in Fig. 8. Most notably, the diurnal temperature pattern is quite different in February compared to December or January (Fig. 8b). December and January have relatively flat temperature profiles over the course of the day. February, however, shows a 5 °C temperature swing between early-morning minimum temperature and afternoon maximum. This reflects the lack of solar heating in the darkest part of the year (late December) compared to February. All aerosol species have an inverse concentration pattern in February compared to temperature, where early-morning concentrations tend to be at a maximum, but fall to an afternoon minimum, coincident with the temperature increase. This very likely reflects greater dispersion during this afternoon period, as a result of *e.g.*, increased boundary layer height due to solar heating. Within each winter month, we still see very different diurnal behavior between PM constituents, including the primary pollutants (BBOA, HOA, SO_4).

Fig. S4† shows the wintertime diurnal pattern for all aerosol species and wind speed and temperature for each of 2020, 2021, and 2022. These plots only use January and February data, as we do not have December 2019 data for the ACSM. We see similar patterns for each species by year, *e.g.*, the morning increase in SO_4 , highest levels of BBOA factor being overnight, *etc.*

3.5.2 Diurnal patterns in other (non-winter) seasons. The diurnal patterns of all species for non-winter seasons differ from those in winter. For example, in spring the daily minimum in HOA concentration occurs during the mid-afternoon, which is the opposite of wintertime. This is likely driven by different meteorology (and potentially lower emissions strength), where there are much higher afternoon wind speeds. However, across spring, summer, and winter, we still do see the morning “bump” of HOA, as well as a weekday/weekend pattern suggestive of reduced or shifted weekend driving volumes, especially in the morning. We see similar morning and weekday/weekend behavior in non-winter seasons for SO_4 as well, which again lends support for the idea that some SO_4 is linked to vehicle emissions.

We also see an afternoon “clean out” in spring for all species, which reflects higher dispersion driven by temperature increases and an afternoon maximum in wind speed. We do not see a similar afternoon clean-out during summer for most aerosol species, despite wind speeds having a similar diurnal pattern across the two seasons. For example, there is little diurnal variation at all for NR- $\text{PM}_{2.5}$, OA, or OOA during the summer. We speculate that this may be due to a combination of nearly round-the-clock daylight (providing opportunity for photochemical PM production) and the prevalence of more

regional-scale PM sources (*e.g.*, wildfire smoke and/or photochemical PM).

3.6 SO_4 fragmentation pattern

Previous studies have shown that the SO_4 signal in the AMS can contain contributions from organosulfur (OS) compounds. Those studies looking at AMS fragmentation patterns of OS compounds have included sodium hydroxymethanesulfonate (Na-HMS);⁴⁶ organosulfates, sulfonates, and sulfonic acids;⁴⁷ and sulfones.⁴⁸ With the exception of methanesulfonic acid (MSA), these OS compounds contain very little HSO_3^+ or H_2SO_4^+ in their mass spectra, and reduced contribution from SO_3^+ compared to the $(\text{NH}_4)_2\text{SO}_4$ mass spectrum; MSA has some HSO_3^+ , but no H_2SO_4^+ or SO_3^+ . Ammonium sulfate ($(\text{NH}_4)_2\text{SO}_4$), by contrast, has contributions from each of SO_3^+ , HSO_3^+ , and H_2SO_4^+ in its mass spectra.

We probe the idea that some of the “ SO_4 ” signal in our dataset may come from OS compounds by examining the wintertime mass spectrum fragmentation pattern of SO_4 as a function of total concentration, using the same binning approach from Fig. 5. Fig. 9 shows the ratio of each H_ySO_x^+ fragment relative to SO^+ (m/z 48), which is one of the two largest SO_4 ion signals in the AMS, the other being SO_2^+ (m/z 64). We choose to report these $\text{H}_y\text{SO}_x^+/\text{SO}^+$ ratios, following the analysis of Hu *et al.*⁴⁸ Since this is a unit-mass resolution instrument, our attribution of the SO_4 -related signal at m/z 48 as “ SO^+ ” is not measured directly, but rather estimated based on the default AMS frag table, but we use the chemical ion fragment names for clarity.

$\text{SO}_2^+/\text{SO}^+$ is relatively flat across all concentration bins in Fig. 9, with a total average value (marked with a dotted horizontal line) of 1.73. However, we see that each of the other H_ySO_x^+ ions decreases relative to SO^+ with higher total NR- $\text{PM}_{2.5}$ concentrations. This trend is consistent with OS compounds contributing relatively more to total SO_4 signal in the AMS at higher concentrations.

Previous work in Fairbanks also shows that OS compounds make up a significant fraction of particulate sulfur especially at high concentrations. Campbell *et al.*²⁷ estimated that 26–41% of particulate sulfur in Fairbanks is in the form of S(IV) during polluted episodes (where $\text{PM}_{2.5} > 35 \mu\text{g m}^{-3}$), and that a majority of this S(IV) is expected to be hydroxymethanesulfonate (HMS). Moon *et al.*¹⁰ find that HMS contributes significantly (26%) to total PM sulfur during the most-polluted portion of the ALPACA field campaign. HMS forms *via* the aqueous phase reaction of SO_2 and formaldehyde (HCHO). This formation pathway for particulate sulfur does not require high oxidant levels reliant on the availability of sunlight, unlike *e.g.*, the gas phase oxidation of SO_2 by OH or the aqueous phase oxidation of SO_2 by H_2O_2 and O_3 .⁴⁹ While HCHO can be formed through photochemical reactions with a variety of VOCs,⁵⁰ HCHO is also a primary emission of vehicular combustion and biomass-burning,⁵¹ which likely dominate over photochemical sources in the cold, dark winter conditions of Fairbanks.⁵² Our data do not independently confirm that some of the AMS SO_4 signal is HMS specifically, but the change we see in SO_4



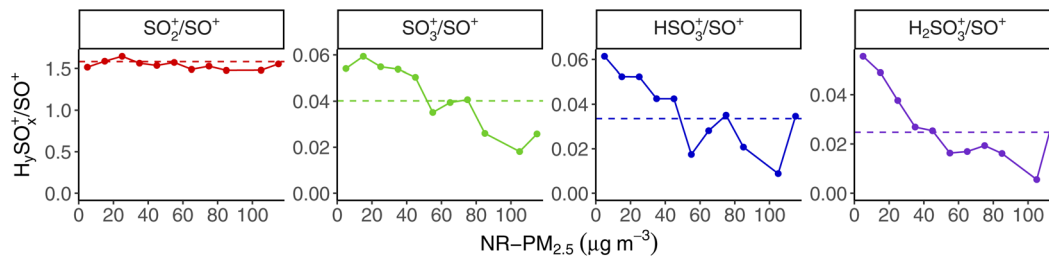


Fig. 9 CV ACSM wintertime ratios of the major sulfur-containing fragment ions (SO_2^+ , SO_3^+ , HSO_3^+ , and H_2SO_4^+) to SO^+ that make up AMS SO_4 as a function of total NR- $\text{PM}_{2.5}$ concentration. Ratios are determined from the slope of the regression fit for all data within each $10 \mu\text{g m}^{-3}$ NR- $\text{PM}_{2.5}$ concentration bin. Average for all wintertime data is shown by the horizontal dotted line in each panel.

fragmentation patterns at higher NR- $\text{PM}_{2.5}$ concentrations is consistent with a greater contribution from OS compounds like HMS.

Some of the ACSM SO_4 signal being OS compounds formed from vehicular HCHO emissions could help explain the similarity between HOA and SO_4 diurnal patterns. It also is consistent with the trend we see of ACSM SO_4 contributing a larger fraction to total NR- $\text{PM}_{2.5}$ at higher mass concentrations, where HOA and BBOA make up a larger fraction of total OA than at lower mass concentration. OS formation from HCHO would also imply that these combustion-related HCHO emissions are rate-limiting much of the time, as opposed to SO_2 , in the aqueous phase reaction between HCHO and SO_2 that forms OS compounds (like HMS).

Because of the significantly reduced signal at SO_3^+ , HSO_3^+ , and H_2SO_4^+ in CV instruments compared to SV instruments due to greater fragmentation, even when measuring $(\text{NH}_4)_2\text{SO}_4$, there are much higher signal-to-noise issues when trying to quantify OS compounds. Preliminary analysis of ALPACA data from a high-resolution time-of-flight SV AMS (HR-ToF-AMS) independently confirms these same trends in the SO_4 fragmentation pattern, however, which eliminates the concern that this result could be an artifact of the unit-mass fragmentation table. These SV HR-ToF-AMS data will be presented in a future publication.

4 Conclusions

We present high time resolution observations of PM mass and composition from downtown Fairbanks made over the course of three years and across all four seasons. These measurements are important for better understanding air quality issues in a city that has dealt with problematic levels of air pollution over the course of decades.

We show how different aerosol species vary over the course of the day, between different types of day (weekday vs. weekend), and between and within seasons. Identifying the diurnal pattern of different aerosol components sheds light on their sources, and is a major novel aspect of this study. We also deconvolve the dominant non-refractory aerosol species-OA into different factors, which reflect source contributions. We estimate that, at a minimum, half of the OA comes from primary emissions sources during winter, and thus may be

addressable through various mechanisms (emissions controls, behavior changes, *etc.*). Some fraction of the second-most abundant aerosol species by mass- SO_4 -may also be addressable through similar mechanisms.

Because Fairbanks has such perennial air quality problems during winter, focusing on modifiable emissions sources makes sense. However, we show, as others have before, that pollutant concentrations are also strongly controlled by meteorological conditions (*e.g.*, temperature, wind speed) as well. In Fairbanks, it truly takes both strong local emissions and low dispersion levels to create the air quality issues typical of wintertime. This interplay between dispersion and emissions creates a potential strategy for mitigating the most-polluted periods by reducing emissions when dispersion is lowest, which has led FNSB to institute a tiered “burn ban” system limiting wood stove use during forecasted temperature inversions.

Lastly, this study helps put the ALPACA campaign¹⁵ into context, which will provide many more-detailed insights on air quality issues in Fairbanks from winter 2022. We show here that wintertime concentrations in 2022 are similar to the two prior years, and that chemical composition and diurnal patterns are broadly similar as well. Thus, insights from ALPACA should be broadly applicable to wintertime in Fairbanks, as well as cold, dark urban places more generally.

Author contributions

R. J. W. and J. M. planned experimental design. M. B., J. R. C., J. M., and R. J. W. set up ACSM for data collection in Fairbanks. M. B., J. R. C., J. M., and E. S. R. monitored the ACSM during the course of data collection. E. S. R. and P. F. D. analyzed data and wrote the manuscript, with significant input from all other co-authors. J. M. and W. S. organized ALPACA 2022.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 Alaska Department of Environmental Conservation, *State Implementation Plan*, 2016, <https://dec.alaska.gov/air/anpms/communities/fbks-pm2-5-moderate-sip/>.
- 2 American Lung Association, *State of the Air Report*, 2021.
- 3 J. Schmale, S. R. Arnold, K. S. Law, T. Thorp, S. Anenberg, W. R. Simpson, J. Mao and K. A. Pratt, *Earth's Future*, 2018, **6**, 1385–1412.
- 4 W. Simpson, *et al.*, *ALPACA White Paper*, 2019, <https://alpaca.community.uaf.edu/wp-content/uploads/sites/758/2019/05/ALPACA-whitepaper.pdf>.
- 5 U.S. Department of Agriculture, *Wood Energy for Residential Heating in Alaska: Current Conditions, Attitudes, and Expected Use*, 2010.
- 6 R. A. Kotchenruther, *Atmos. Environ.*, 2016, **142**, 210–219.
- 7 T. Ward, B. Trost, J. Conner, J. Flanagan and R. K. M. Jayanty, *Aerosol Air Qual. Res.*, 2012, **12**, 536–543.
- 8 Y. Wang and P. K. Hopke, *Aerosol Air Qual. Res.*, 2014, **14**, 1875–1882.
- 9 R. Davey, *Characterizing Wintertime Aerosol Composition and Sulfate Formation in Fairbanks, Alaska*, 2020, <http://hdl.handle.net/11122/11261>.
- 10 A. Moon, U. Jongebloed, K. K. Dingilian, A. J. Schauer, Y.-C. Chan, M. Cesler-Maloney, W. R. Simpson, R. J. Weber, L. Tsiang, F. Yazbeck, S. Zhai, A. Wedum, A. J. Turner, S. Albertin, S. Bekki, J. Savarino, K. Gribanov, K. A. Pratt, E. J. Costa, C. Anastasio, M. O. Sunday, L. M. D. Heinlein, J. Mao and B. Alexander, *ACS ES&T Air*, 2024, **1**, 139–149.
- 11 L. Ye and Y. Wang, *Atmosphere*, 2020, **11**, 1203.
- 12 K. M. Shakya and R. E. Peltier, *Environ. Sci. Technol.*, 2013, **47**, 9332–9338.
- 13 P. L. Joyce, R. v. Glasow and W. R. Simpson, *Atmos. Chem. Phys.*, 2014, **14**, 7601–7616.
- 14 Alaska Department of Environmental Conservation, *Design Values FNSB*, 2021, <https://dec.alaska.gov/air/air-monitoring/community-data/fnsb-summary-pm25/>.
- 15 W. R. Simpson, J. Mao, G. J. Fochesatto, K. S. Law, P. F. DeCarlo, J. Schmale, K. A. Pratt, S. R. Arnold, J. Stutz, J. E. Dibb, J. M. Creamean, R. J. Weber, B. J. Williams, B. Alexander, L. Hu, R. J. Yokelson, M. Shiraiwa, S. Decesari, C. Anastasio, B. D'Anna, R. C. Gilliam, A. Nenes, J. M. S. Clair, B. Trost, J. H. Flynn, J. Savarino, L. D. Conner, N. Kettle, K. M. Heeringa, S. Albertin, A. Baccharini, B. Barret, M. A. Battaglia, S. Bekki, T. Brado, N. Brett, D. Brus, J. R. Campbell, M. Cesler-Maloney, S. Cooperdock, K. C. d. Carvalho, H. Delbarre, P. J. DeMott, C. J. Dennehy, E. Dieudonne, K. K. Dingilian, A. Donato, K. M. Douglgeris, K. C. Edwards, K. Fahey, T. Fang, F. Guo, L. M. D. Heinlein, A. L. Holen, D. Huff, A. Ijaz, S. Johnson, S. Kapur, D. T. Ketcherside, E. Levin, E. Lill, A. R. Moon, T. Onishi, G. Pappaccogli, R. Perkins, R. Pohorsky, J.-C. Raut, F. Ravetta, T. Roberts, E. S. Robinson, F. Scotto, V. Selimovic, M. O. Sunday, B. Temime-Roussel, X. Tian, J. Wu and Y. Yang, *ACS ES&T Air*, 2024, **1**, 200–222.
- 16 H. N. Tran and N. Mölders, *Atmos. Res.*, 2011, **99**, 39–49.
- 17 M. Cesler Maloney, W. R. Simpson, T. Miles, J. Mao, K. S. Law and T. J. Roberts, *J. Geophys. Res.: Atmos.*, 2022, **127**, e2021JD036215.
- 18 N. R. Council, *The Ongoing Challenge of Managing Carbon Monoxide Pollution in Fairbanks, Alaska*, The national academies press technical report, 2002.
- 19 E. S. Robinson, M. Cesler-Maloney, X. Tan, J. Mao, W. Simpson and P. F. DeCarlo, *Environ. Sci.: Atmos.*, 2023, **3**, 568–580.
- 20 N. L. Ng, S. C. Herndon, A. Trimborn, M. R. Canagaratna, P. L. Croteau, T. B. Onasch, D. Sueper, D. R. Worsnop, Q. Zhang, Y. L. Sun and J. T. Jayne, *Aerosol Sci. Technol.*, 2011, **45**, 780–794.
- 21 V. Crenn, J. Sciare, P. L. Croteau, S. Verlhac, R. Fröhlich, C. A. Belis, W. Aas, M. Äijälä, A. Alastuey, B. Artiñano, D. Baisnée, N. Bonnaire, M. Bressi, M. Canagaratna, F. Canonaco, C. Carbone, F. Cavalli, E. Coz, M. J. Cubison, J. K. Esser-Gietl, D. C. Green, V. Gros, L. Heikkinen, H. Herrmann, C. Lunder, M. C. Minguiñón, G. Monik, C. D. O'Dowd, J. Ovadnevaite, J.-E. Petit, E. Petralia, L. Poulain, M. Priestman, V. Riffault, A. Ripoll, R. Sarda-Estève, J. G. Slowik, A. Setyan, A. Wiedensohler, U. Baltensperger, A. S. H. Prévôt, J. T. Jayne and O. Favez, *Atmos. Meas. Tech.*, 2015, **8**, 5063–5087.
- 22 T. Joo, Y. Chen, W. Xu, P. Croteau, M. R. Canagaratna, D. Gao, H. Guo, G. Saavedra, S. S. Kim, Y. Sun, R. Weber, J. Jayne and N. L. Ng, *ACS Earth Space Chem.*, 2021, **5**, 2565–2576.
- 23 J. Peck, L. A. Gonzalez, L. R. Williams, W. Xu, P. L. Croteau, M. T. Timko, J. T. Jayne, D. R. Worsnop, R. C. Mlake-Lye and K. A. Smith, *Aerosol Sci. Technol.*, 2016, **50**(8), 781–789.
- 24 A. M. Middlebrook, R. Bahreini, J. L. Jimenez and M. R. Canagaratna, *Aerosol Sci. Technol.*, 2011, **46**, 258–271.
- 25 W. Hu, P. Campuzano-Jost, D. A. Day, P. Croteau, M. R. Canagaratna, J. T. Jayne, D. R. Worsnop and J. L. Jimenez, *Aerosol Sci. Technol.*, 2017, **51**(6), 735–754.
- 26 N. L. Ng, S. C. Herndon, A. Trimborn, M. R. Canagaratna, P. L. Croteau, T. B. Onasch, D. Sueper, D. R. Worsnop, Q. Zhang, Y. L. Sun and J. T. Jayne, *Aerosol Sci. Technol.*, 2011, **45**, 780–794.



- 27 J. R. Campbell, M. Battaglia, K. Dingilian, M. Cesler-Maloney, J. M. S. Clair, T. F. Hanisco, E. Robinson, P. DeCarlo, W. Simpson, A. Nenes, R. J. Weber and J. Mao, *Environ. Sci. Technol.*, 2022, **56**, 7657–7667.
- 28 E. S. Robinson, T. B. Onasch, D. Worsnop and N. M. Donahue, *Atmos. Meas. Tech.*, 2017, **10**, 1139–1154.
- 29 P. Paatero and U. Tapper, *Environmetrics*, 1994, **5**, 111–126.
- 30 I. M. Ulbrich, M. R. Canagaratna, Q. Zhang, D. R. Worsnop and J. L. Jimenez, *Atmos. Chem. Phys.*, 2009, **9**, 2891–2918.
- 31 F. Canonaco, M. Crippa, J. G. Slowik, U. Baltensperger and A. S. H. Prévôt, *Atmos. Meas. Tech.*, 2013, **6**, 3649–3661.
- 32 Q. Zhang, M. R. Alfarra, D. R. Worsnop, J. D. Allan, H. Coe, M. R. Canagaratna and J. L. Jimenez, *Environ. Sci. Technol.*, 2005, **39**, 4938–4952.
- 33 N. L. Ng, M. R. Canagaratna, J. L. Jimenez, Q. Zhang, I. M. Ulbrich and D. R. Worsnop, *Environ. Sci. Technol.*, 2011, **45**, 910–916.
- 34 M. Crippa, P. F. DeCarlo, J. G. Slowik, C. Mohr, M. F. Heringa, R. Chirico, L. Poulain, F. Freutel, J. Sciare, J. Cozic, C. F. D. Marco, M. Elsassler, J. B. Nicolas, N. Marchand, E. Abidi, A. Wiedensohler, F. Drewnick, J. Schneider, S. Borrmann, E. Nemitz, R. Zimmermann, J.-L. Jaffrezo, A. S. H. Prévôt and U. Baltensperger, *Atmos. Chem. Phys.*, 2013, **13**, 961–981.
- 35 M. J. Cubison, A. M. Ortega, P. L. Hayes, D. K. Farmer, D. Day, M. J. Lechner, W. H. Brune, E. Apel, G. S. Diskin, J. A. Fisher, H. E. Fuelberg, A. Hecobian, D. J. Knapp, T. Mikoviny, D. Riemer, G. W. Sachse, W. Sessions, R. J. Weber, A. J. Weinheimer, A. Wisthaler and J. L. Jimenez, *Atmos. Chem. Phys.*, 2011, **11**, 12049–12064.
- 36 C. J. Hennigan, A. P. Sullivan, J. L. Collett and A. L. Robinson, *Geophys. Res. Lett.*, 2010, **37**, L09806.
- 37 V. A. Lanz, M. R. Alfarra, U. Baltensperger, B. Buchmann, C. Hueglin and A. S. H. Prévôt, *Atmos. Chem. Phys.*, 2007, **7**, 1503–1522.
- 38 W. Hu, D. A. Day, P. Campuzano-Jost, B. A. Nault, T. Park, T. Lee, P. Croteau, M. R. Canagaratna, J. T. Jayne, D. R. Worsnop and J. L. Jimenez, *ACS Earth Space Chem.*, 2018, **2**, 410–421.
- 39 Y. Zheng, X. Cheng, K. Liao, Y. Li, Y. J. Li, R.-J. Huang, W. Hu, Y. Liu, T. Zhu, S. Chen, L. Zeng, D. R. Worsnop and Q. Chen, *Atmos. Meas. Tech.*, 2020, **13**, 2457–2472.
- 40 D. Bhattu, P. Zotter, J. Zhou, G. Stefanelli, F. Klein, A. Bertrand, B. Temime-Roussel, N. Marchand, J. G. Slowik, U. Baltensperger, A. S. H. Prevot, T. Nussbaumer, I. E. Haddad and J. Dommen, *Environ. Sci. Technol.*, 2019, **53**, 2209–2219.
- 41 A. Trubetskaya, C. Lin, J. Ovadnevaite, D. Ceburnis, C. O'Dowd, J. J. Leahy, R. F. D. Monaghan, R. Johnson, P. Layden and W. Smith, *Energy Fuels*, 2021, **35**, 4966–4978.
- 42 F. Fachinger, F. Drewnick, R. Giere and S. Borrmann, *Atmos. Environ.*, 2017, **158**, 216–226.
- 43 W. Aas, A. Mortier, V. Bowersox, R. Cherian, G. Faluvegi, H. Fagerli, J. Hand, Z. Klimont, C. Galy-Lacaux, C. M. B. Lehmann, C. L. Myhre, G. Myhre, D. Olivie, K. Sato, J. Quaas, P. S. P. Rao, M. Schulz, D. Shindell, R. B. Skeie, A. Stein, T. Takemura, S. Tsyro, R. Vet and X. Xu, *Sci. Rep.*, 2019, **9**, 953.
- 44 Alaska Interagency Coordination Center, 2023, <https://fire.ak.blm.gov/predsvcs/intel.php>.
- 45 U.S. Environmental Protection Agency, *Wood Stove Emissions: Particle Size and Chemical Composition.*, U.S. environmental protection agency technical report, 2000.
- 46 E. Dovrou, C. Y. Lim, M. R. Canagaratna, J. H. Kroll, D. R. Worsnop and F. N. Keutsch, *Atmos. Meas. Tech.*, 2019, **12**, 5303–5315.
- 47 Y. Chen, L. Xu, T. Humphry, A. P. S. Hettiyadura, J. Ovadnevaite, S. Huang, L. Poulain, J. C. Schroder, P. Campuzano-Jost, J. L. Jimenez, H. Herrmann, C. O'Dowd, E. A. Stone and N. L. Ng, *Environ. Sci. Technol.*, 2019, **53**, 5176–5186.
- 48 W. Hu, P. Campuzano-Jost, D. A. Day, P. Croteau, M. R. Canagaratna, J. T. Jayne, D. R. Worsnop and J. L. Jimenez, *Aerosol Sci. Technol.*, 2017, **51**, 735–754.
- 49 J. M. Moch, E. Dovrou, L. J. Mickley, F. N. Keutsch, Z. Liu, Y. Wang, T. L. Dombek, M. Kuwata, S. H. Budisulistiorini, L. Yang, S. Decesari, M. Paglione, B. Alexander, J. Shao, J. W. Munger and D. J. Jacob, *J. Geophys. Res.: Atmos.*, 2020, **125**, e2020JD032706.
- 50 G. M. Wolfe, J. Kaiser, T. F. Hanisco, F. N. Keutsch, J. A. d. Gouw, J. B. Gilman, M. Graus, C. D. Hatch, J. Holloway, L. W. Horowitz, B. H. Lee, B. M. Lerner, F. Lopez-Hilifiker, J. Mao, M. R. Marvin, J. Peischl, I. B. Pollack, J. M. Roberts, T. B. Ryerson, J. A. Thornton, P. R. Veres and C. Warneke, *Atmos. Chem. Phys.*, 2016, **16**, 2597–2610.
- 51 A. Fortems-Cheiney, F. Chevallier, I. Pison, P. Bousquet, M. Saunois, S. Szopa, C. Cressot, T. P. Kurosu, K. Chance and A. Fried, *Atmos. Chem. Phys.*, 2012, **12**, 6699–6721.
- 52 Z. Rao, Z. Chen, H. Liang, L. Huang and D. Huang, *Atmos. Environ.*, 2016, **124**, 207–216.

