

# RSC Advances

## At the heart of open access for the global chemistry community

#### **Editor-in-chief**

Russell J Cox Leibniz Universität Hannover, Germany

#### We stand for:



**Breadth** We publish work in all areas of chemistry and reach a global readership



**Quality** Research to advance the chemical sciences undergoes rigorous peer review for a trusted, society-run journal

### ŞĘĘ





**Community** Led by active researchers, we publish quality work from scientists at every career stage, and all countries

### Submit your work now

rsc.li/rsc-advances



Showcasing research from the Ångström Advanced Battery Centre, Department of Chemistry – Ångström Laboratory, Uppsala university, Sweden.

Inherent limitations of the hydrogen-bonding UPy motif as self-healing functionality for polymer electrolytes

The inclusion of hydrogen-bonding ureido pyrimidinone (UPy) groups is an effective means of introducing dynamically cross-linking and self-healing capabilities in polymer materials. These properties are highly desirable also for next-generation electrolyte materials for energy storage applications. However, we demonstrate that the addition of a high concentration of ions causes the hydrogen-bonding network to be disrupted by interactions with the ions, thereby cancelling out the effect of the UPy groups on the mechanical properties of the material and rendering the material mechanically unsuitable for electrolyte use.



As featured in:

See Jonas Mindemark *et al., RSC Appl. Polym.,* 2024, **2**, 374.

rsc.li/RSCApplPolym

