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Developments and applications of α -bromonitrostyrenes in organic syntheses

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The presence of the bromo and nitro groups in the structure of α -bromonitrostyrene makes them highly reactive and versatile reagents in organic syntheses. α -Bromonitrostyrenes act as an effective dielectrophile in the reaction with various nucleophiles. In these reactions, the bromo and nitro groups behave as good leaving groups for the assembly of a diverse range of heterocyclic compounds, such as dihydrofurans, dihydropyranes, furans, pyrroles, pyrazoles, isooxazolines, spiropyrrolidines, etc. In the current review, we have focused on the transformations of α -bromonitrostyrenes under organocatalysis, metal catalysis, and base-catalysis systems as well as catalyst-free conditions, since 2010.

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1. Introduction

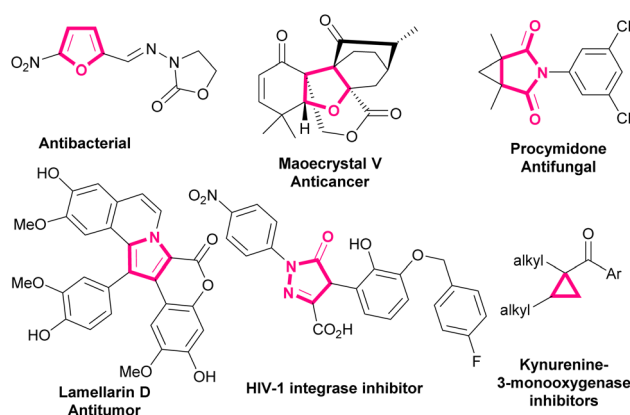
Bromonitroalkenes, especially bromonitrostyrenes are versatile building blocks for the construction of biological and pharmaceutical molecules and optically active compounds, such as dihydrofurans,^{1,2} pyrroles,^{3–6} and pyrazoles.^{7–9} Selected examples of biologically active molecules containing pyrrole, furan, dihydrofuran, or cyclopropane cores are shown in Scheme 1.

Due to the presence of both nitro and bromo groups, α -bromonitrostyrenes exhibit peculiar properties in comparison with nitrostyrenes and bromostyrenes.^{10–14} These substituents can behave as good leaving groups in the nucleophilic substitution reactions. Because of this feature, α -bromonitrostyrenes can easily be used in Michael addition reactions as a dipolarophile to make various functionalized dihydrofurans, pyrroles, pyrazoles, bicyclic oximes, triazoles, imidazoles, cyclopropanes, etc.¹⁵ In such reactions, α -bromonitroalkenes act as a 1,2-dielectrophile in a diverse range of cascade reactions with binucleophiles, such as 1,3-dicarbonyl compounds, enamines, aldehydes, alkenes, phenols, diazo compounds and other miscellaneous reagents.

Michael addition reactions of α -bromonitroalkenes can be performed in the presence of organocatalysts, metal catalysts, and bases, as well as catalyst-free systems. Among them, advances in enantioselective organocatalytic Michael reactions have provided various enantiomerically enriched compounds and biologically active molecules.^{16–19} On the other hand, domino Michael/Henry reactions are known as powerful

methods for rendering complex molecular architectures through constructing the carbon–carbon bonds in a one-pot procedure without isolating intermediates.^{20,21} Recently, progress in asymmetric domino reactions utilizing chiral organocatalysts has been well revealed and interesting achievements has been gained.^{22–24}

Recently, various research teams, such as Jakubec,²⁵ Gao,²⁶ Lee,²⁷ Halimehjani,²⁸ Shen,²⁹ and Zhang³⁰ reported the organic transformations of nitrostyrenes. In this review, we have specifically discussed the use of α -bromonitrostyrenes in various types of catalytic systems, such as organocatalysis, metal catalysis and base catalysis as well as catalyst-free reactions. We have also highlighted important features of the reactions and mechanisms to familiarize the readers with the reactivity and behavior of α -bromonitrostyrenes in organic syntheses.



Scheme 1 Selected examples of biologically active molecules containing pyrrole, furan, dihydrofuran, or cyclopropane rings.

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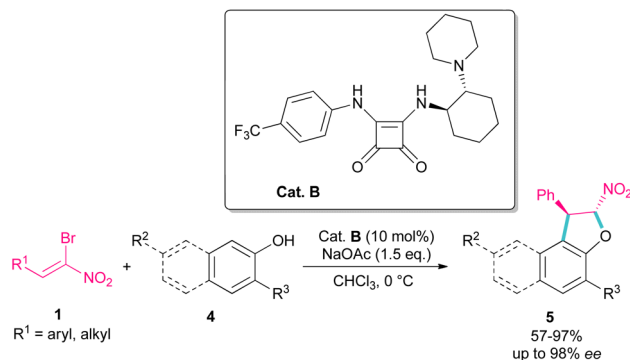
2. Transformations of bromonitrostyrenes

2.1. Organocatalyst-catalyzed transformations of bromonitrostyrenes

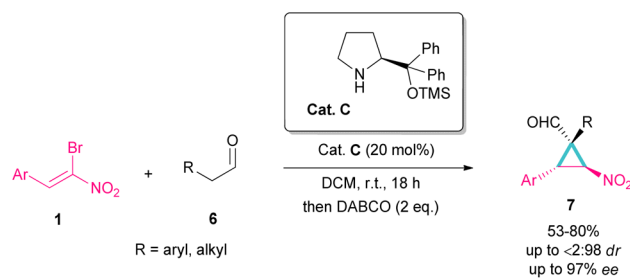
In 2010, Rueping *et al.* used (*E*)- β -bromonitrostyrenes **1** for the synthesis of dihydrofurans **3** from 1,3-dicarbonyl compounds **2** (Scheme 2).³¹ In this reaction, the chiral thiourea catalyst **A** can close two substrates together by hydrogen-bonding interaction and facilitate Michael addition of diketone **2** as a dinucleophile to bromonitrostyrene **1** as an electrophile to generate intermediate **II**. Then, an intramolecular nucleophilic addition delivered product **3** by the release of HBr. The same organocatalyst was used in the reaction of bromonitrostyrene, as a dielectrophile and 1,3-indandione as a dinucleophile to make spirocyclopropane in 57% yield with 35% ee.³²

Parra and co-workers used an organocatalysis system to synthesize enantioselective dihydroarylfuran derivatives **5** from the cyclization reaction between (*Z*)-bromonitroalkenes **1** and naphthol **4** (Scheme 3).³³ The hydrogen bonding interaction of chiral organocatalyst **B** with both substrates was responsible for the suitable orientation for Michael–Friedel–Crafts reaction and subsequent S_N2 reaction on the carbon attached bromide. It should be noted that HBr resulting from the reaction can be neutralized with a stoichiometric amount of a base.

Another organocatalyst was used in the reaction of bromonitroalkenes **1** and alkyl aldehydes **6** (Scheme 4).³⁴ A series of enantioenriched cyclopropanes **7** with quaternary carbon centers were prepared in the presence of a chiral pyrrolidine catalyst **C**. As illustrated in Scheme 5, the reaction started with the condensation of pyrrolidine with aldehyde to form imine **I**, followed by a 1,3-H shift to enamine **II**. The attack of bromonitrostyrene to the alkene moiety produced intermediate **III**. By release of the catalyst and subsequent intramolecular S_N2



Scheme 3 Reaction of (*Z*)-bromonitroalkenes and naphthols catalyzed by a chiral organocatalyst.

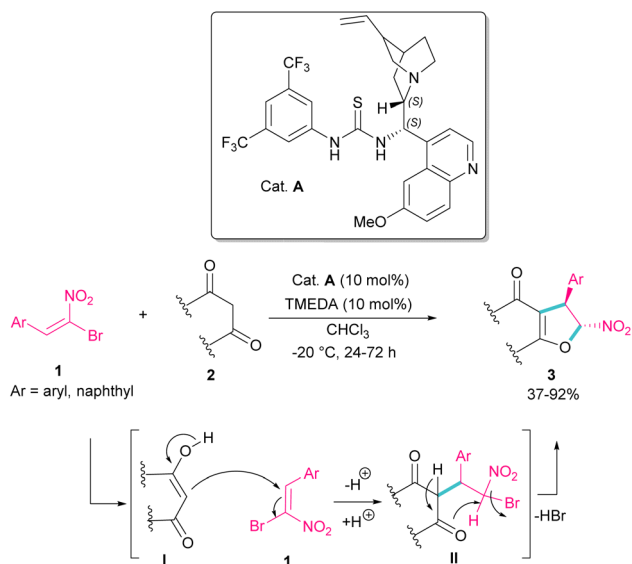


Scheme 4 Reaction of bromonitroalkenes and alkyl aldehydes catalyzed by chiral organocatalyst.

reaction, cyclopropane **7** was provided along with the elimination of bromide. It seems that the formation of two diastereomers **7** and **7'** from two configurations **IV** and **V** is possible. The configuration at C2 is generated from the intramolecular attack of the anion with DABCO, while the configuration at C4 is from the acidic proton at this C-atom, providing the thermodynamically stable product. In this case, the NO₂ group is at the *trans* position to the aryl ring at C3.

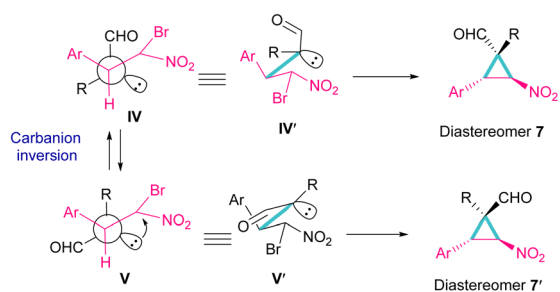
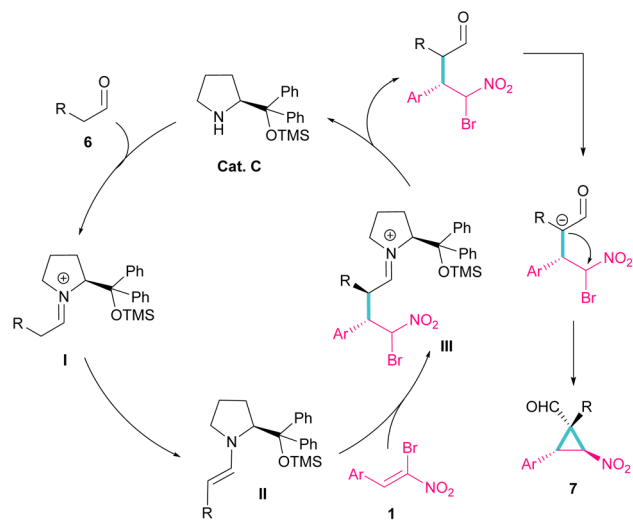
A domino Michael-alkylation reaction was reported by Feng and co-workers for the synthesis of dihydrofurans **9** from bromonitrostyrene **1** (Scheme 6).³⁵ For this purpose, they treated bromonitrostyrene **1** and cyclohexane-1,3-diones **8** in the presence of the organocatalyst **D**. For dimedone, two stereocenters of the bicyclic 2,3-dihydrofurans were obtained with high diastereo- and enantioselectivity, while three stereocenters of the bicyclic 2,3-dihydrofurans were prepared from prochiral 5-monosubstituted cyclohexane-1,3-dione products *via* the desymmetrization reaction.

In 2016, Feng *et al.* introduced a chiral organocatalyst **E** for the asymmetric synthesis of dihydrofurans **11** from (*Z*)-bromonitrostyrenes **1** and α -substituted cyano ketones **10** (Scheme 7).³⁶ In this regard, a domino Michael addition-alkylation between bromonitrostyrene and α -substituted cyano ketone in the presence of chiral *N,N'*-dioxide **E** was conducted under very mild reaction conditions. The authors found that the use of proton sponge, added in six equal parts, could enhance enantioselectivity and also the performance of the reaction in a very low temperature prevent the formation of the cyclopropane

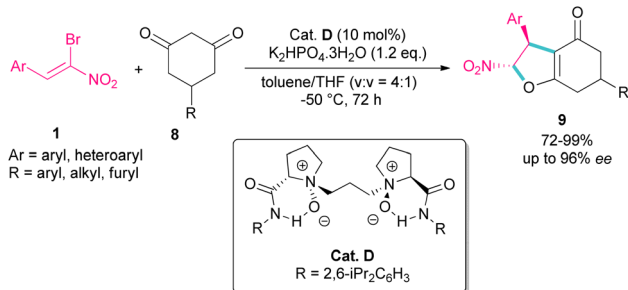


Scheme 2 Organocatalysis reaction of (*E*)- β -bromonitrostyrenes with 1,3-dicarbonyl compounds.





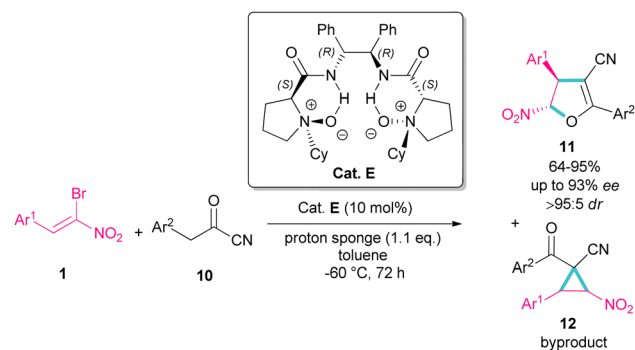
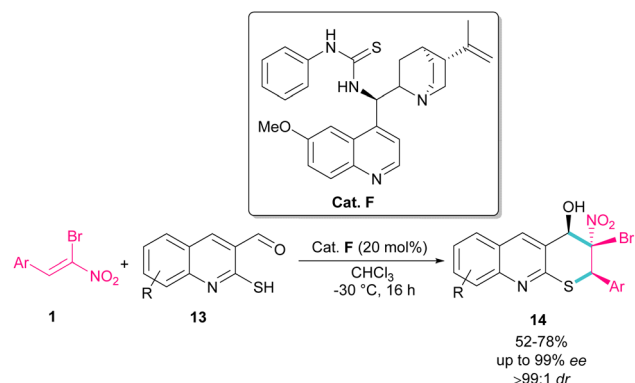
Scheme 5 Putative mechanism for reaction of bromonitroalkenes and alkyl aldehydes.



Scheme 6 Reaction of cyclohexane-1,3-dione and bromonitrostyrenes catalyzed by organocatalyst.

compound as a byproduct and thus increasing the yield of dihydrofuran **11**. In addition to excellent enantioselectivity, high diastereo- and regioselectivity were also observed in this reaction.

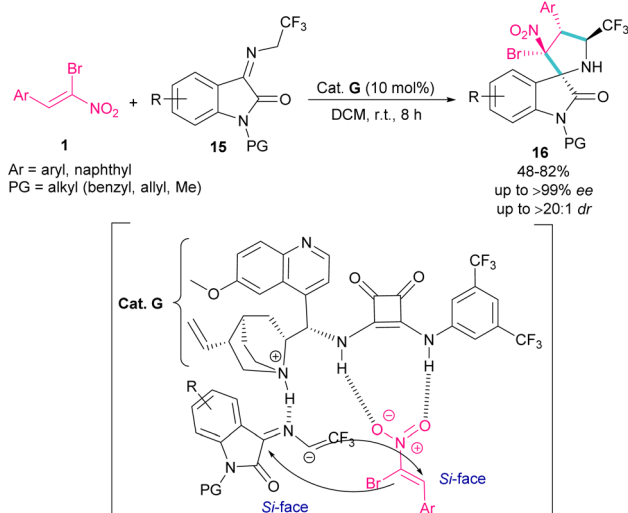
The use of α -bromonitroalkenes **1** in the synthesis of 3,4-dihydro-2*H*-thiopyrano[2,3-*b*]quinolines **14** was reported by Xie and co-workers in 2018 (Scheme 8).³⁷ An organocatalysis system was proposed for this synthetic method, which started with the deprotonation of 2-mercaptoquinoline-3-carbaldehyde **13** by tertiary amine of catalyst **F**, and the activation of α -bromonitroalkene **1** by the thiourea moiety of **F** through the formation of two hydrogen bonds. The subsequent domino Michael/Henry

Scheme 7 Organocatalysis reaction of (*Z*)-bromonitrostyrenes and α -substituted cyano ketones.Scheme 8 Organocatalyst-catalyzed reaction of 2-mercaptoquinoline-3-carbaldehyde and α -bromonitroalkene.

reaction resulted in product **14**. This method had the advantages of excellent enantio- and diastereoselectivity, mild reaction conditions, and the ability of 3,4-dihydro-2*H*-thiopyrano [2,3-*b*]quinolines to undergo further organic transformations. In addition, salicylaldehyde can also act as a coupling partner in the cyclization with bromonitroalkene.

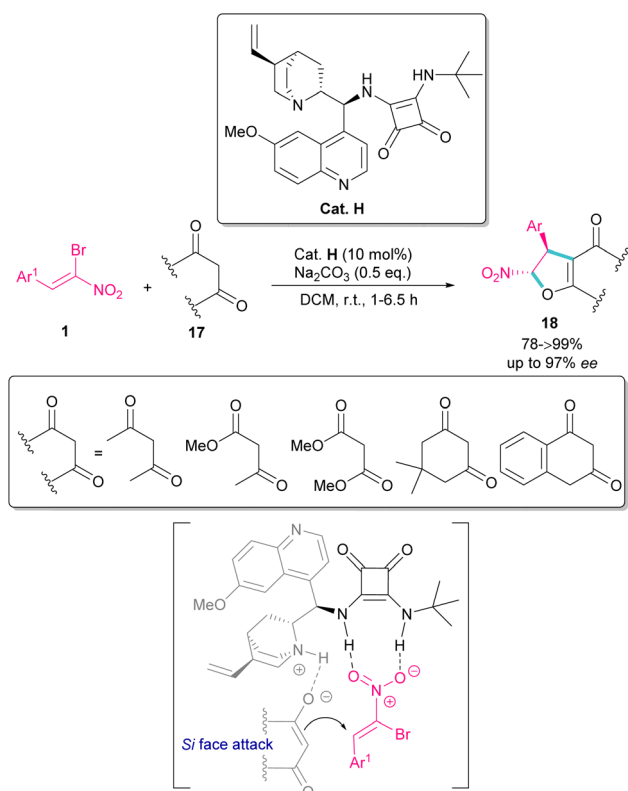
In 2020, an organocatalysis system was developed to prepare pyrrolidinyl spiro-oxindoles bearing quaternary carbon centers **16** (Scheme 9).³⁸ Han and co-workers designed a (3 + 2)-cycloaddition reaction between isatin-derived ketimine **15** and (*Z*)- α -bromonitroalkene **1** in the presence of cinchonidine-derived squaramide **G**. This bifunctional organocatalyst can control the stereoselectivity by the hydrogen bonding interactions with both substrates blocking one side of each substrate, allowing nucleophilic attack from the *Si*-face.





Scheme 9 Reaction of isatin-derived ketimine and (Z)- α -bromonitroalkene catalyzed by cinchonidine-derived squaramide.

In 2022, Tanyeli and co-workers extended a new organo-catalysis system for the synthesis of chiral dihydrofurans **18** from bromonitroalkenes **1** and 1,3-dicarbonyl compounds **17** (Scheme 10).³⁹ Their reaction proceeded through domino type Michael-S_N2 reactions in the presence of a quinine-derived sterically encumbered squaramide **H**. The main advantages of this method that differentiates it from previous methods was



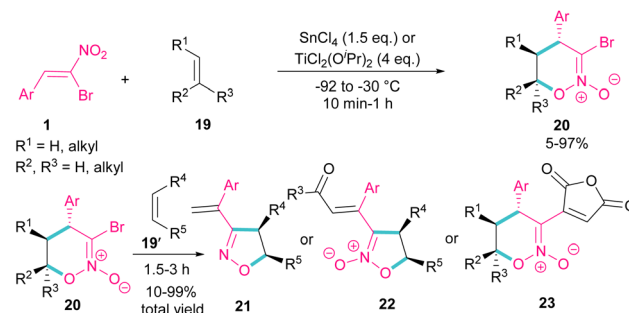
Scheme 10 Squaramide-catalyzed reaction of bromonitroalkenes and 1,3-dicarbonyl compounds.

the performance of the reaction in room temperature at short reaction times, while other similar reactions required cryogenic conditions and longer times to proceed. As shown in transition state, the quinoline moiety of the catalyst **H** activates the nucleophile **17** by hydrogen bonding. At the same time, the electrophile **1** is also activated by double hydrogen bonding with the squaramide group of **H**. Consequently, the nucleophilic attack of **17** from the *Si*-face of **1** afforded product with excellent enantioselectivity.

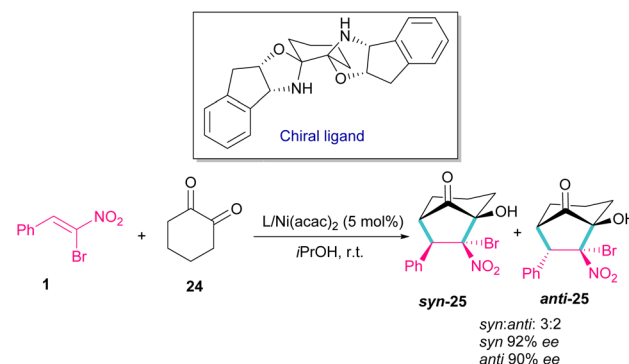
2.2. Metal-catalyzed transformations of bromonitrostyrenes

Bromonitrostyrenes can be applied for the construction of 3-bromo-5,6-dihydro-4*H*-1,2-oxazine *N*-oxides (Scheme 11).⁴⁰ An alkene was used as a coupling partner and the cyclization reaction was performed in the presence of SnCl₄ or TiCl₂(O*i*Pr)₂ as a Lewis acid catalyst. (4 + 2)-Cycloaddition of alkene to bromonitroalkene to form 3-bromo-5,6-dihydro-4*H*-1,2-oxazine *N*-oxide **20** as byproduct. In the next stage, (3 + 2)-cycloaddition of the obtained 3-bromo-5,6-dihydro-4*H*-1,2-oxazine *N*-oxides **20** with another alkene reactant **19'** was performed and three products, including 3-vinyloxazole **21**, isoxazole *N*-oxide **22**, and 3-functionalized 1,2-oxazine *N*-oxide **23** were obtained in different ratios.

In 2014, a chiral bisoxazolidine in the combination with Ni(acac)₂ catalyst was used for the preparation of optically active

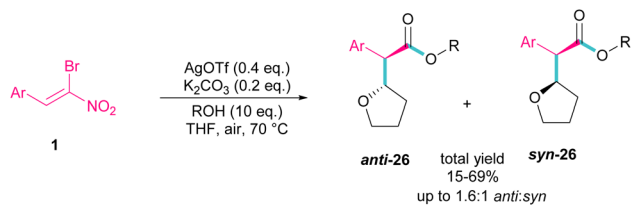
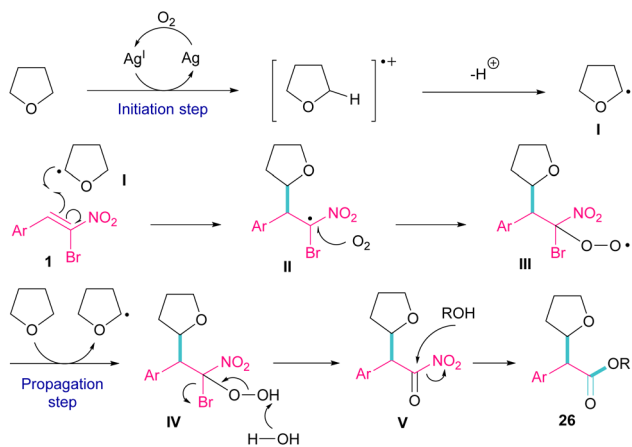


Scheme 11 Reaction of bromonitrostyrenes with alkenes in the presence of SnCl₄.



Scheme 12 Ni-catalyzed reaction of bromonitrostyrenes with 1,2-cyclohexadione.



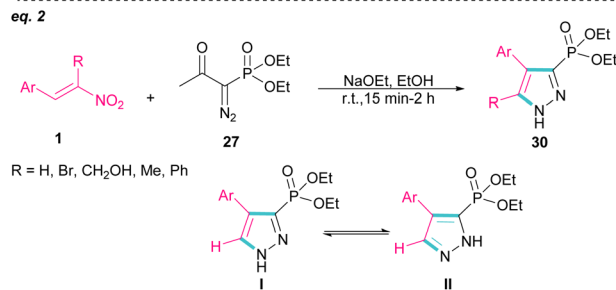
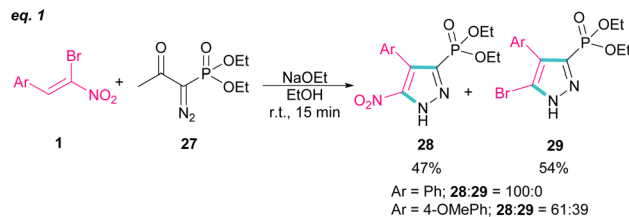
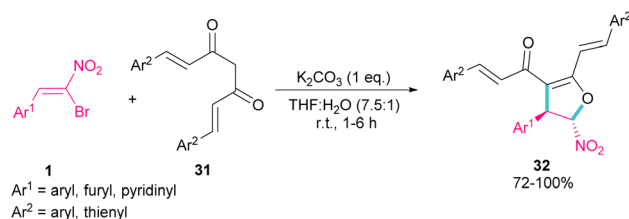
Scheme 13 Ag-catalyzed addition of THF radicals to β -bromonitrostyrenes.Scheme 14 Plausible mechanism for Ag-catalyzed addition of THF radicals to β -bromonitrostyrenes.

bicycle[3,2,1] octane of 1,2-diones **25** from nitroalkenes and **1** 1,2-cyclohexadione **24** (Scheme 12).⁴¹ α -Bromonitroalkene **1** in a domino Michael–Henry reaction with diketone **24** can furnish product **25** with two *syn* and *anti* isomers. In this reaction, excellent enantioselectivity was observed (98% ee), but diastereoselectivity was relatively lower (2 : 1 dr) in comparison with other nitroalkenes as coupling reactants (up to 99% ee, up to 7 : 1 dr).

Silver triflate can catalyze the three-component reaction of tetrahydrofuran, β -bromonitrostyrene and alcohol *via* a radical pathway (Scheme 13).⁴² The reaction was initiated by the formation of radical **I** from the interaction of Ag(I) with THF, which attacked β -bromonitrostyrene **1** to form intermediate **II**. Further attack of O₂ on **II** led to the peroxy radical **III**. Then, **III** abstracted a hydrogen from THF to give hydroperoxide **IV**. The cleavage of the O–O bond by H₂O resulted in intermediate **V** with the release of bromide. In this step, the nucleophilic attack of alcohol to the carbonyl group along with the removal of the second leaving group (NO₂) led to product **26**. It is noteworthy that only less hindered alcohols can attack intermediate **V** (Scheme 14).

2.3. Base-mediated transformations of bromonitrostyrenes

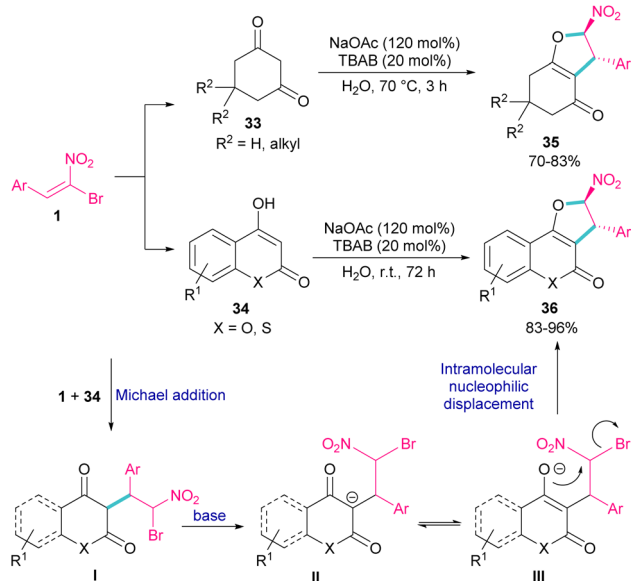
In 2007, Namboothiri and co-workers reported the synthesis of phosphonylpyrazoles **28** and **29** from diethyl 1-diazo-2-oxopropylphosphonate **27** with α -bromonitroalkenes **1** under basic conditions (Scheme 15, eqn (1)).⁴³ The elimination of the leaving group in the final product depended on the substituent at the aryl ring of bromonitroalkene substrate. When Ar = Ph,

Scheme 15 Reaction of diethyl 1-diazo-2-oxopropylphosphonate and α -bromonitroalkenes.Scheme 16 Base-mediated reaction of α -bromonitroalkenes with curcumins.

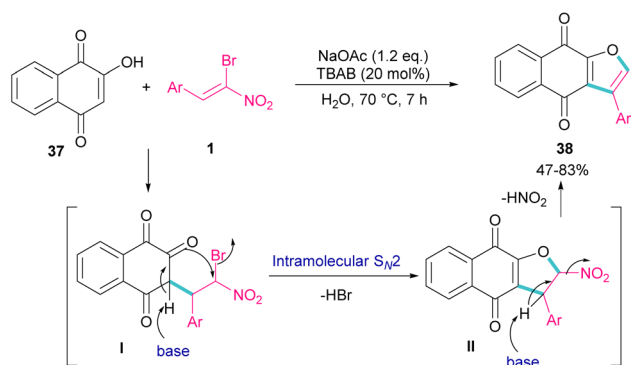
product **28** was the major product, while products **28** and **29** were obtained in a ratio of 61 : 39 if bromonitrostyrene contained a 4-OMePh moiety. In a similar work, by dynamic NMR investigation, the Namboothiri group showed that two tautomers **I** and **II** are presented in such a reaction mixture that have a small energy difference but a high barrier to interconversion (Scheme 15, eqn (2)).⁴⁴ Also, they could extend the substrate scope of bromonitrostyrenes in this work. Namboothiri *et al.* also reported the reaction of α -bromonitroalkenes **1** with curcumins **31** in the presence of a base (Scheme 16).⁴⁵ Single diastereomer of dihydrofurans **32** was obtained in a Michael addition–alkenylation cascade reaction.

In 2011, Xie *et al.* developed a method for the assembly of dihydrofurans from domino reaction of bromonitrostyrenes with 4-hydroxycoumarins or 1,3-carbonyl compounds (Scheme 17).⁴⁶ In this work, tricyclic 2,3-dihydrofurans **36** and bicyclic 2,3-dihydrofurans **35** were obtained in high yields under basic conditions in an aqueous solution. In all cases, only the *trans* isomer of products was isolated, showing excellent diastereoselectivity of this method. Regarding the mechanism, Michael addition of bromonitroalkene **1** to 1,3-dicarbonyl compounds **33** gave intermediate **I**, which was converted to the enolate **II** under basic conditions. Finally, product **36** was formed *via* an intramolecular nucleophilic displacement in **III**. Similar reaction conditions were used by Jianwu and co-workers





Scheme 17 Reaction of bromonitrostyrenes with 4-hydroxycoumarins mediated by base.

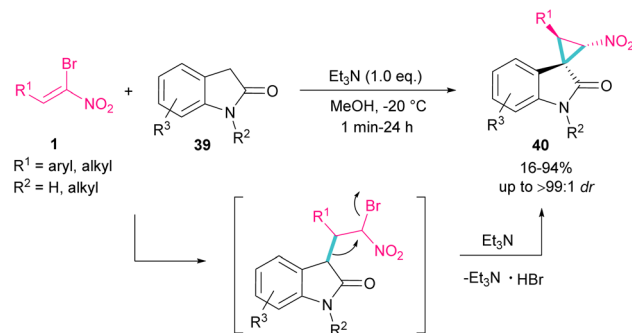


Scheme 18 Reaction of α -bromonitroalkenes and 2-hydroxynaphthalene-1,4-diones under basic conditions.

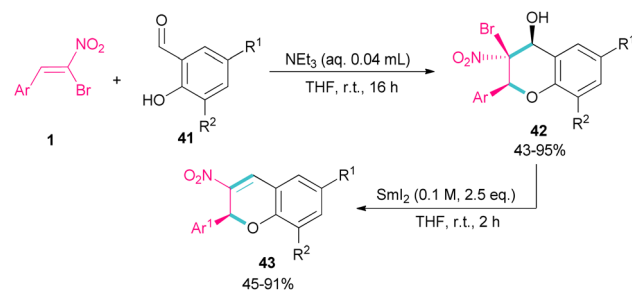
for the synthesis of furan structures (Scheme 18).⁴⁷ In their reaction, α -bromonitroalkenes **1** and 2-hydroxynaphthalene-1,4-diones **37** were applied as starting materials to form 3-phenylnaphtho[2,3-*b*]furan-4,9-diones **38** via Michael addition and subsequent intramolecular S_N2 reaction. At last, the absorption properties of the obtained products were determined by UV-Vis spectra and fluorescence spectroscopy.

Spirocyclopropyl oxindole frameworks **40** were constructed in high yield with excellent diastereoselectivity from the reaction of bromonitroalkene **1** and *N*-protected indolin-2-ones **39** (Scheme 19).⁴⁸ Due to high reactivity of indolin-2-one under ambient temperature and consequently low diastereoselectivity of the obtained product, a very low temperature was necessary for this reaction. The formation of cyclopropane was accomplished by the abstraction of a proton by a base and the liberation of bromide. Not only *N*-protected indolin-2-one but also *N*-H indolin-2-one was included in this transformation.

Soengas *et al.* established a two-step procedure for the preparation of 2-*C*-glycosyl-3-nitrochromenes **43** starting from



Scheme 19 Reaction of bromonitroalkene with *N*-protected indolin-2-ones.



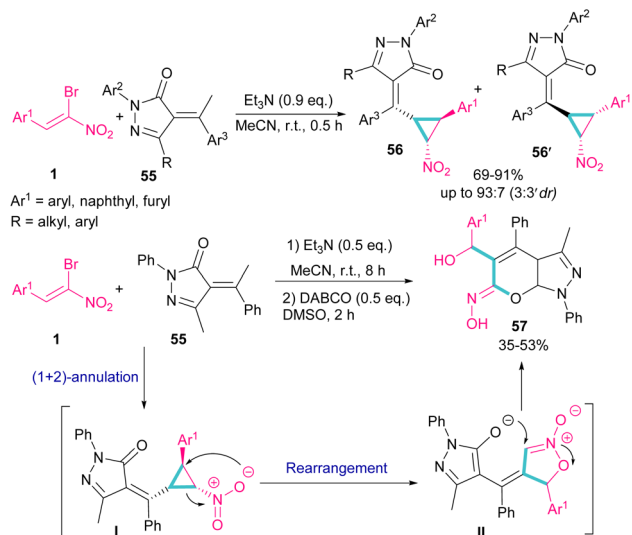
Scheme 20 NEt_3 -catalyzed reaction of bromonitroalkene with *ortho*-hydroxybenzaldehydes.

bromonitrostyrenes **1** and *ortho*-hydroxybenzaldehydes **41** (Scheme 20).⁴⁹ For this purpose, first, they treated bromonitrostyrenes **1** with *ortho*-hydroxybenzaldehydes **41** under basic conditions to achieve (2*S*,3*S*,4*S*)-3-bromo-3,4-dihydro-4-hydroxy-3-nitro-2*H*-1-benzopyrans **42** in moderate to excellent yields. In the next stage, the reaction was conducted in the presence of 0.1 M solution of SmI_2 in THF for couple hours to form 3-nitrochromenes **43**. SmI_2 can promote the β -elimination in benzopyran **42** with complete stereoselectivity. All steps were performed under mild conditions and all products were obtained with enantiomeric purity.

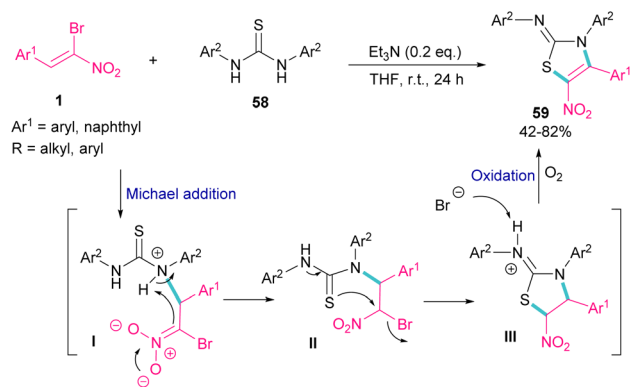
In 2015, again the Namboothiri research team synthesized imidazole scaffolds **45** through the reaction of amidine hydrochloride **44** with α -bromonitroalkenes **1** (Scheme 21).⁵⁰ For the assembly of 2,5-disubstituted imidazoles from α -bromonitroalkenes, 3.0 equivalents of Cs_2CO_3 were required to promote all steps of this imidazole synthesis. First, Cs_2CO_3 promoted the formation of amidine **I** by the elimination of HX from **44**. Michael addition of **1** to **I** gave intermediate **II**, which underwent an intramolecular S_N2 process via a 5-*exo-tet* fashion to provide nitroimidazoline **III** along with the removal of HBr in the presence of Cs_2CO_3 . Subsequent elimination of HNO_2 by Cs_2CO_3 led to product **45**. Moreover, the authors reported potential anti-parasitic activity of the resulting imidazoles.

In 2018, Namboothiri and co-workers used α -bromonitroalkene **1** in the reaction with lawsone **46** and 2-aminonaphthoquinone **47** to produce furan and pyrrole fused quinonoid compounds **48** and **49** using two different basic



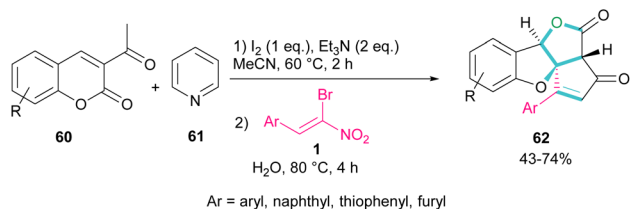


Scheme 25 Reaction of bromonitrostyrenes with α -alkylidene pyrazolones mediated by Et₃N.

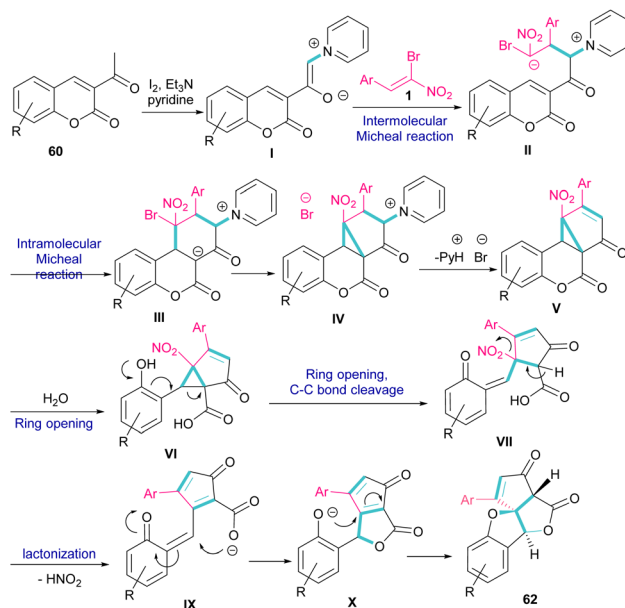


Scheme 26 Et₃N-promoted reaction of 1,3-disubstituted thioureas with 1-bromo-1-nitroalkenes.

ones **60**, and pyridine **61** was carried out through three steps intermolecular and intramolecular Michael additions. Initially, the C–H bond functionalization of 3-acetyl-2*H*-chromen-2-one **60** with pyridine **61** in the presence of I₂ and Et₃N furnished pyridinium ylide **I**. Then, sequentially intermolecular Michael addition of **1** with **I**, and intramolecular Michael reaction gave intermediate **II**, which was converted to cyclopropane **III**. Intermediate **IV** was obtained *via* the elimination of pyridinium



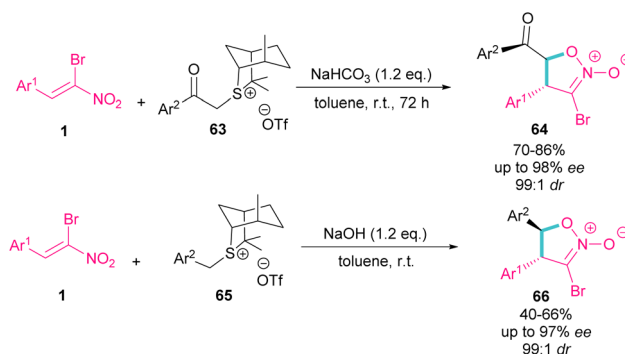
Scheme 27 Three-component reaction of β -bromo- β -nitrostyrenes, 3-acetyl-2*H*-chromen-2-ones, and pyridine.



Scheme 28 Possible mechanism for reaction of β -bromo- β -nitrostyrenes, 3-acetyl-2*H*-chromen-2-ones, and pyridine.

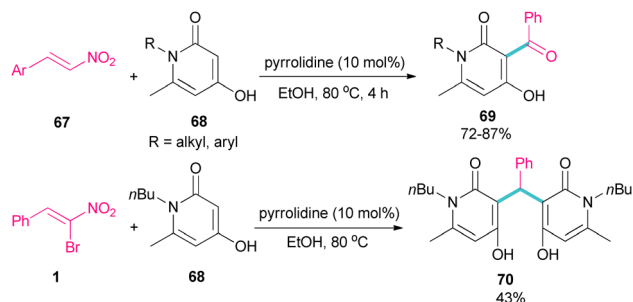
bromide in **III**, followed by two steps cyclopropane ring-opening and rearrangement to produce **VI**. The release of HNO₂ gave **VII**, which underwent an intramolecular lactonization to render **IX**. Another intramolecular Michael addition of the phenoxide ion with the C=C bond gave product **62** (Scheme 28).

In the same year, the Xie group designed a (4 + 1) ylide annulation between chiral sulfonium salts **63** and **65** with α -bromonitroalkenes **1** access to enantioenriched isoxazoline *N*-oxides **64** and **66** (Scheme 29).⁵⁸ Two types of chiral sulfonium salts, such as α -benzoyl sulfonium triflate salts **63** and α -benzyl sulfonium triflate salts **65** could smoothly generate the sulfonium ylide intermediates, which served as C1 synthon in (4 + 1)-annulation with bromonitroalkene as a 4-atomic synthon. All products were obtained in high yields with excellent enantio- and diastereoselectivity. Also, changing the anions of sulfonium salts from OTf⁻ to Br⁻, ClO₄⁻ and BF₄⁻, showed that the type of anion has a great effect on the activity, and enantioselectivity in this asymmetric annulation.



Scheme 29 Reaction of sulfonium salts and substituted α -bromonitroalkenes.





Scheme 30 Pyrrolidine-catalyzed reaction of 1-butyl-4-hydroxy-6-methylpyridin-2(1H)-one and nitroalkenes.

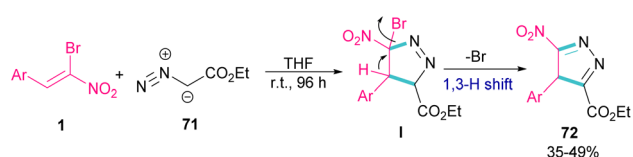
In 2023, Li *et al.* investigated the application of pyrrolidine as a catalyst in the reaction of 1-butyl-4-hydroxy-6-methylpyridin-2(1H)-one 68 and nitroalkenes 67 (Scheme 30).⁵⁹ They achieved an open ring product 69, when α -bromonitrostyrene 1 was used as a coupling reactant, while the reaction of other nitroalkenes led to 4-hydroxy-3-benzoylpyridin-2(1H)-ones 70 as the main product. This may be because of the high steric hindrance in α -bromonitrostyrene, which inhibits the formation of product 69.

2.4. Catalyst-free transformations of bromonitrostyrenes

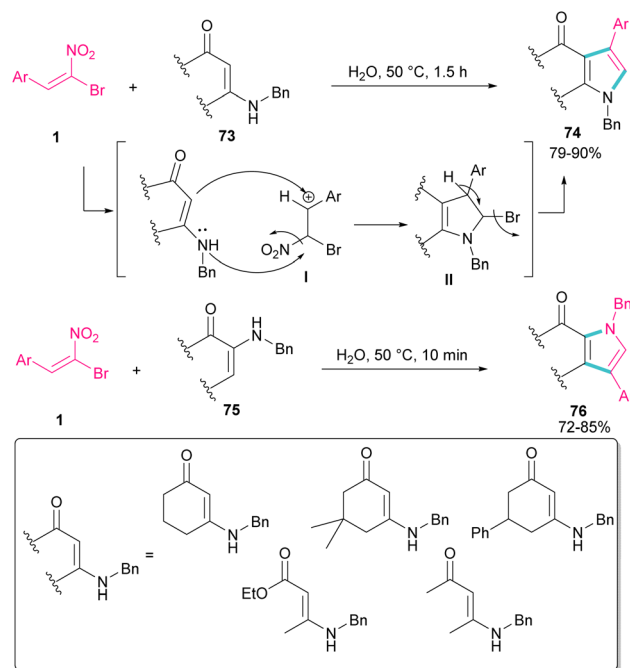
In 2009, Xie and co-workers reported the synthesis of pyrazoles 72 from 1,3-cycloaddition of α -bromo- α -nitroalkenes 1 with ethyl diazoacetate 71 under catalyst-free conditions (Scheme 31).⁶⁰ In general, the reaction involved the nucleophilic addition of 71 to 1 to form intermediate I, followed by the elimination of bromide along with the 1,3-H shift to furnish the final pyrazole product 72.

In 2010, Rueping and Parra treated (*E*)- β -bromonitrostyrenes 1 with enaminones 73 to synthesize pyrrole derivatives 74 (Scheme 32).⁶¹ The main advantages of their reaction were the performance of the reaction in water as a green solvent, very short reaction time, and excellent product yields. In this reaction, bromonitrostyrenes 1 acted as a trifunctional synthon and reacted well with both enaminone 73 and *N*-benzylenaminone 75 as binucleophilic synthons. First, the nucleophilic attack of nitrogen to the carbon atom in I concomitant with removal of a leaving group produced II. Next, the deprotonation and the release of another leaving group afforded the desired product 74.

Deng *et al.* showed that depending on the type of nitrostyrenes and the reaction solvent, two different pyrazole products could be obtained under acidic conditions (Scheme 33).⁶² When β -chloro- β -nitrostyrenes 77 were used as reactants in the



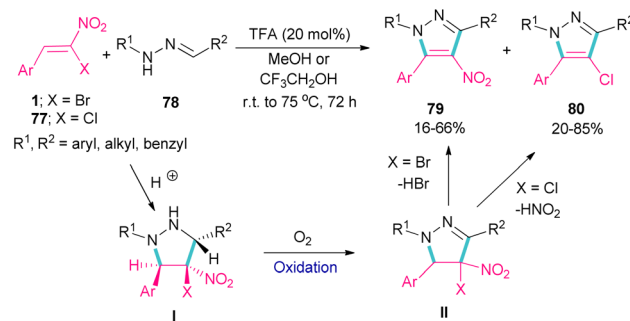
Scheme 31 Reaction of α -bromo- α -nitroalkenes with ethyl diazoacetate under catalyst-free conditions.



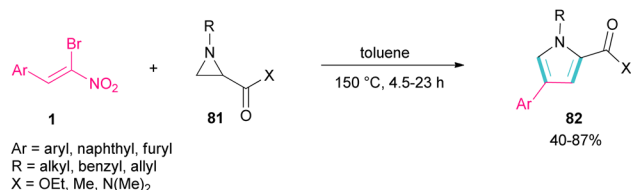
Scheme 32 Catalyst-free reaction of (*E*)- β -bromonitrostyrenes with enaminones.

reaction with hydrazones 78, the elimination of HNO₂ occurred to produce 4-chloro-tetrasubstituted pyrazoles 80 in MeOH as a solvent. Whereas, 4-nitro-tetrasubstituted pyrazoles 79 were obtained from β -bromo- β -nitrostyrenes 1 and hydrazones 78 as starting materials in MeOH. In this case, the removal of HBr was favored over HNO₂, which could be due to the easy cleavage of the C-Br bond compared to the C-Cl bond. In fact, the leaving-group abilities of functional groups can be classified as Br > NO₂ > Cl. In general, the reactions involved pyrazolidine intermediate I, which was subjected to oxidation and elimination steps. It should be noted that when a more acidic alcoholic solvent like CF₃CH₂OH was utilized, the formation of 4-bromo-tetrasubstituted pyrazoles was also observed in the reaction mixture.

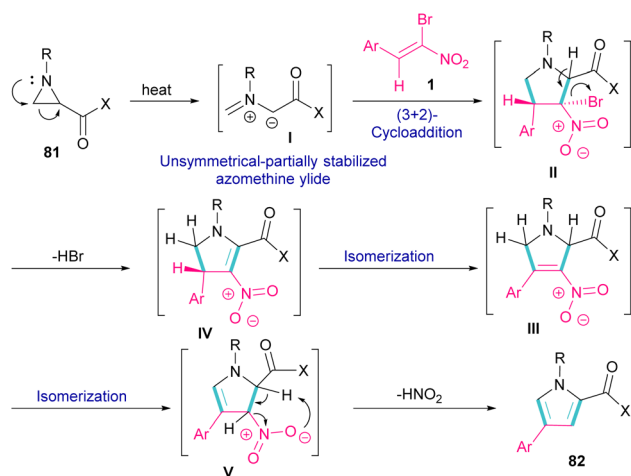
Another domino reaction was carried out by the Khan research team to construct 1,2,4-trisubstituted pyrrole



Scheme 33 TFA-catalyzed reaction of bromonitrostyrenes with hydrazones.

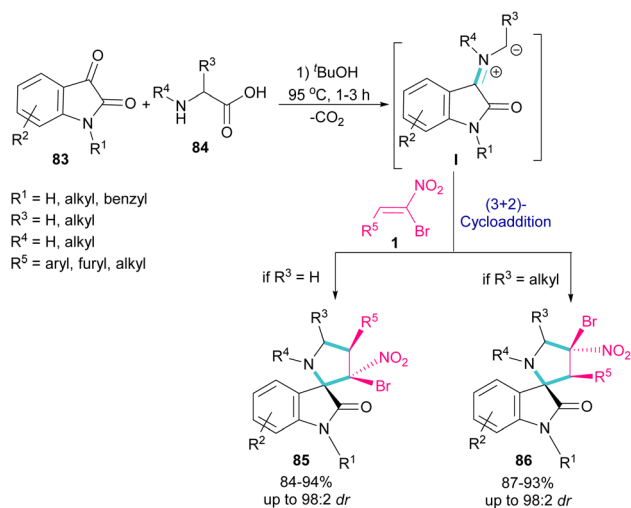


Scheme 34 Catalyst-free reaction of unactivated aziridines with β -bromo- β -nitrostyrenes.



Scheme 35 Plausible mechanism for reaction of unactivated aziridines with β -bromo- β -nitrostyrenes.

derivatives **82** from 1,3-dipolar cycloaddition of unactivated aziridines **81** with β -bromo- β -nitrostyrenes **1** (Scheme 34).⁶³ This reaction involved *in situ* generated unsymmetrical azomethine ylide from aziridine, followed by a cascade elimination and aromatization step. As shown in Scheme 35, the mechanism started with simultaneous cleavage of the C–C bond of



Scheme 36 Reaction of oxindole, bromonitrostyrenes and α -amino acids.

aziridine **81** under heat to afford azomethine ylide **I**. The interaction of **I** with β -bromo- β -nitrostyrene **1** resulted in the unstable cycloadduct **II**, which readily underwent E2 elimination of HBr to render **III**. In this step, **III** was isomerized to **IV** and then **V**, followed by the elimination of HNO₂ to form product **82**.

In 2019, Ganesh and co-workers developed a multi-component reaction, including oxindoles **83**, bromonitrostyrenes **1** and α -amino acids **84** to generate tetra-substituted α -spiro pyrrolidine structures **85** and **86** (Scheme 36).⁶⁴ In the first step, azomethine ylide **I** was formed from the condensation of oxindole and α -amino acid. Next, (3 + 2)-cycloaddition of bromonitrostyrene with **I** provided spiro pyrrolidine. Two diastereomers could be obtained depending on the substituent at the α -position of amino acid. If R³ = H, diastereomer **85** was the major product, while spiro pyrrolidine **86** was obtained as a major diastereomer when R³ = alkyl. The diastereoselectivity may be due to the steric effects in transition states of azomethine ylides.

3. Conclusions

In this review, we have described the advances in the synthesis of carbocyclic and heterocyclic compounds from the reaction of α -bromonitrostyrenes with various coupling reactants in straightforward and atom economical manners. Organocatalytic domino Michael reactions offer a direct and sustainable route for the synthesis of diastereo- and enantioselective bioactive products under mild reaction conditions. Although it seems that the use of Lewis acid metal catalysts and bases as promoters in the reactions of α -bromonitrostyrene as a reactive dipolarophile with good leaving groups (bromo and nitro), the use of this synthon in the cross-coupling reactions, as well as regio- and stereoselective syntheses can provide useful insights for further researches in this field. In our opinion, α -bromonitrostyrenes can be considered as a simple and accessible building block for the construction of more complex organic molecules.

Conflicts of interest

There are no conflicts to declare.

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