


 Cite this: *RSC Adv.*, 2024, 14, 22418

Correction: Analyzing the impact of the size of fluoro and chloro substituents on induced mesomorphism in hydrogen bonded liquid crystals

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DOI: 10.1039/d4ra90077d

rsc.li/rsc-advances

 Correction for 'Analyzing the impact of the size of fluoro and chloro substituents on induced mesomorphism in hydrogen bonded liquid crystals' by M. K. Sonali *et al.*, *RSC Adv.*, 2024, 14, 20398–20409, <https://doi.org/10.1039/D3RA08569D>

The authors regret that the names of the hydrogen bonded compounds in Tables 1–3 were not correctly given in the original article. The corrected versions of Tables 1–3 are shown herein.

Table 1 DSC phase transition temperature (°C) and the corresponding enthalpy (J g⁻¹) of P18 with fluorobenzoic acids^a

| Hydrogen bonded compound | Method | Transition | Transition temperatures (°C) (enthalpy J g ⁻¹) | (ΔT) _{LC} |
|--------------------------|---|---|--|------------------------------|
| P18:2FBA | DSC (h) | Cryst → SmA | 72.0 (4.57) | 13.24 |
| | DSC (c) | SmA → Iso | 89.45 (83.45) | |
| P18:3FBA | DSC (h) | Iso → SmA | 74.26 (1.50) | 12.79 |
| | | SmA → Cryst | 61.02 (68.67) | |
| | DSC (c) | Cryst → SmA | 74.06 (5.06) | |
| | | SmA → Iso | 101.41 (58.58) | |
| P18:4FBA | DSC (h) | Iso → SmA | 78.78 (1.72) | 28.02 |
| | | SmA → Cryst ¹ | 65.99 (34.15) | |
| | | Cryst ¹ → Cryst ² | 61.92 (6.04) | |
| | DSC (c) | Cryst ¹ → Cryst ² | 73.53 (20.22) | |
| | | Cryst ² → SmA | 89.58 (73.03) | |
| | | SmA → Iso | 107.36 (5.93) | |
| DSC (c) | Iso → SmA | 103.33 (1.86) | 28.02 | |
| | SmA → Cryst ¹ | 75.31(26.88) | | |
| | Cryst ¹ → Cryst ² | 67.03 (64.09) | | |

^a DSC (h) is heating cycle, DSC (c) is cooling cycle. (ΔT)_{LC} is the thermal range of mesomorphism obtained from the cooling cycle.

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Table 2 DSC phase transition temperature ($^{\circ}\text{C}$) and the corresponding enthalpy (J g^{-1}) of P18 with chlorobenzoic acids

| Hydrogen bonded compound | Method | Transition | Transition temperatures ($^{\circ}\text{C}$) (enthalpy J g^{-1}) | $(\Delta T)_{\text{LC}}$ |
|--------------------------|---------|-------------------------|--|--------------------------|
| P18:2ClBA | DSC (h) | Cryst \rightarrow SmA | 51.27 (13.88) | 9.02 |
| | | SmA \rightarrow Iso | 71.68 (84.68) | |
| | DSC (c) | Iso \rightarrow SmA | 58.38 (21.57) | |
| P18:3ClBA | DSC (h) | SmA \rightarrow Cryst | 49.36 (29.04) | 9.13 |
| | | Cryst \rightarrow SmA | 72.75 (18.40) | |
| | DSC (c) | Iso \rightarrow SmA | 68.69 (53.48) | |
| P18:4ClBA | DSC (h) | SmA \rightarrow Cryst | 59.56 (23.20) | 20.06 |
| | | Cryst \rightarrow SmA | 75.46 (36.78) | |
| | DSC (c) | Iso \rightarrow SmA | 82.16 (29.29) | |
| | | SmA \rightarrow Cryst | 62.10 (52.71) | |

Table 3 DSC phase transition temperature ($^{\circ}\text{C}$) and the corresponding enthalpy (J g^{-1}) of P8 with fluoro and chlorobenzoic acids

| Hydrogen bonded compound | Method | Transition | Transition temperatures ($^{\circ}\text{C}$) (enthalpy J g^{-1}) | $(\Delta T)_{\text{LC}}$ |
|--------------------------|---------|-------------------------|--|--------------------------|
| P8:2FBA | DSC (h) | Cryst \rightarrow Iso | 66.42 (50.09) | 25.64 |
| | | Iso \rightarrow SmA | 53.69 (1.61) | |
| | DSC (c) | SmA \rightarrow Cryst | 28.05 (26.61) | |
| P8:3FBA | DSC (h) | Cryst \rightarrow Iso | 91.52 (59.96) | 25.14 |
| | | Iso \rightarrow SmA | 69.08 (3.34) | |
| | DSC (c) | SmA \rightarrow Cryst | 43.94 (51.45) | |
| P8:4FBA | DSC (h) | Cryst \rightarrow SmA | 61.44 (40.49) | 56.57 |
| | | SmA \rightarrow Iso | ^a | |
| | DSC (c) | Iso \rightarrow SmA | 99.71 (6.09) | |
| P8:3ClBA | DSC (h) | SmA \rightarrow Cryst | 43.14(41.65) | 12.84 |
| | | Cryst \rightarrow Iso | 72.00(51.88) | |
| | DSC (c) | Iso \rightarrow SmA | 66.03 (5.86) | |
| P8:4ClBA | DSC (h) | SmA \rightarrow Cryst | 53.19(48.92) | 52.42 |
| | | Cryst \rightarrow SmA | 91.73 (35.04) | |
| | DSC (c) | Iso \rightarrow SmA | 117.26 (6.34) | |
| | | SmA \rightarrow Cryst | 109.57 (2.55) | |
| | | SmA \rightarrow Cryst | 57.15(20.28) | |

^a Not resolved.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

