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1. Introduction

Integrating renewable electricity and feedstocks into synthetic fuels is considered a key technology for sustainable mobility. With fuel design, new promising components or blends for synthetic fuels can be identified that exhibit *a priori* defined properties. These fuels often contain unconventional components. Most fuel design studies focus on screening molecules or mixtures that lie in a pre-defined range of physico-chemical and combustion-related properties with the aim of identifying a feasible fuel.¹⁻⁸ Many of these studies target fuel properties that allow for increased engine efficiency,¹ *e.g.*, through a high octane number,^{2,6-8} enthalpy of vaporization,^{6,8} laminar burning velocity,^{6,8} and compression ratio,³ which in turn can lead to a reduced global warming impact (GWI). In contrast, only a few studies consider the impact of fuel chemistry on emission

Identifying key environmental objectives for integrated process and fuel design[†]

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Integrated process and fuel design enables tailoring renewable fuels for optimal production while simultaneously fulfilling desired fuel specifications. In this work, we extend the integrated process and fuel design framework from [A. König *et al., Comput. Chem. Eng.,* 2020, **134**, 106712] towards multiple environmental impact categories of the life cycle assessment methodology as additional objectives to "production cost" and "global warming impact". We then apply the technique of [G. Guillén-Gosálbez, *Comput. Chem. Eng.,* 2011, **35**(8), 1469] to reduce the high dimensionality of the objective vector while still covering the major trade-offs of the optimization problem. Commonly, the input data required for this technique are normalized. We analyze the influence of normalization variants on the identification of the key environmental objectives. For the specific case of designing advanced spark-ignition engine fuels, our findings suggest that "land use" and "resource use of minerals and metals" represent key environmental objective optimization with these key objectives hold for both current and future technologies for feedstock and utility supply as well as different normalization variants. Our subsequent multi-objective optimization with these key objectives demonstrates that the obtained renewable bio-hybrid fuels, produced from biomass and renewable electricity, can balance the benefits and drawbacks of both biomass- and electricity-based fuels.

formation 1,4,5,8 or the eco-toxicity and human toxicity of the fuel. $^{8\text{--}10}$

Additionally, several studies focus solely on designing a fuel blend, neglecting other phases of the fuel's life cycle. For instance, designing blends with a minimal amount of fossil fuel and maximal amount of alternative fuel aims to reduce both fossil resource depletion and GWI.9-12 However, to estimate the environmental impact of a fuel over its whole life cycle, fuel production needs to be considered as well. Estimating the environmental impact of production processes for unconventional fuel components is challenging, since the processes typically have not been realized outside of laboratories. Identifying optimal synthetic fuels or blends thereof is thus a complex task that can be supported by mathematical optimization, e.g., with methods that combine process and product design in an integrated manner. To this end, Marvin et al. combined a reaction network generator with linear fuel property models to design a blend of biomass-based components with fossil gasoline that complies with ASTM standards, targeting different objectives such as energy loss, catalyst requirement and heat duty of reactions.13

Dahmen and Marquardt built on the method by Marvin *et al.* and combined mass-based screening by reaction network flux analysis (RNFA)¹⁴ with nonlinear fuel property models to design fuel blends for high efficiency engines with minimal resource

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consumption.¹⁵ RNFA uses experimental data like stoichiometry and yields for production processes and was further developed towards process network flux analysis (PNFA). PNFA also considers estimated energy demands of separation steps based on shortcut models by Bausa *et al.* for distillation columns.¹⁶⁻¹⁸ Combining PNFA with predictive fuel property models enabled the integrated early-stage design of production processes and fuels, minimizing the objectives production cost and GWI.¹⁹ The application of integrated process and fuel design in a case study for spark-ignition engine fuels with 47 fuel species identified a ketone-ester-alcohol-alkane (KEAA) blend. This KEAA blend represented a promising Pareto-optimal process and fuel design that compromises the conflicting objectives production cost and GWI.^{20,21}

While a lot of studies address the environmental impacts of the fuel, so far, the approach of König et al. is the only design approach that considers the well-to-wheel life cycle of the fuel.¹⁹ It uses a bi-objective optimization, minimizing production cost and GWI as the only environmental impact category, which may overlook a potential burden shift to other currently neglected impact categories. Electricity-based fuels (e-fuels), for instance, can reduce GWI substantially compared to fossil fuels if produced from renewable electricity and carbon sources, at the cost of increases in other impact categories, e.g., resource use of minerals and metals.^{22,23} In contrast, biomass-based fuels (biofuels), e.g., ethanol, are known to increase agricultural land occupation substantially compared to their fossil-based counterparts.²⁴ Therefore, a holistic design approach should cover a wide range of impact categories as additional objectives to quantify and avoid potential burden-shifting.25 With these additional objectives, the optimization formulation for integrated process and fuel design is a multi-objective optimization problem (MOP). However, MOPs are challenging for two reasons: the computational effort of approximating a wellresolved Pareto front increases exponentially with the number of objectives.²⁶ Additionally, human decision-making is difficult as visualizing, interpreting, and weighing many different objectives is a complex and subjective task.

In MOPs, the number of actual trade-offs among all objectives is often fewer than expected *a priori* since at least some objectives are highly correlated.²⁷ These correlated objectives may be less important or even redundant for optimization. According to Gal and Leberling, an objective is redundant if its removal does not change the original Pareto front.²⁸ Consequently, both challenges of MOPs may be resolved by considering only those objectives as key objectives for optimization that are most conflicting and thus covering the major trade-offs of the original MOP. However, identifying the key objectives of an optimization problem is challenging.

In the literature, several methods have been proposed to identify a key objective subset that adequately represents the trade-offs of the original MOP; see the review article by Li *et al.* for an extensive overview.²⁶ These methods are commonly referred to as dimensionality or objective reduction methods since they reduce the dimensionality of the objective space, *i.e.*, the number of objectives.^{29,30} For the sake of brevity, we will use the term "objective reduction" whenever we refer to reducing

the number of objectives in the objective space. Methods for objective reduction can be roughly categorized based on the timing of objective reduction and the expression of the key objective subset: objective reduction may take place during (online) or after (offline) the generation of a Pareto-optimal solution set,²⁶ and the key objectives are either expressed as a subset (selection) or linear combination (extraction) of the full objective vector.³¹ Most of these methods are correlation-based or aim to preserve the original MOP's Pareto dominance structure.³² Note that some approaches are also based on feature similarity measures,³³ the analytic hierarchy process,³⁴ objective clustering,^{35,36} and representative and extreme criteria.³⁷

Deb and Saxena developed correlation-based methods for both linear and nonlinear objective reduction, using principal component analysis (PCA) to evaluate the degree of correlation among objectives.^{31,38,39} PCA transforms a set of correlated objectives linearly into a smaller set of uncorrelated objectives, *i.e.*, the so-called principal components, capturing the maximum variance in the underlying data.⁴⁰ Correlation-based objective reduction with PCA has, however, two major drawbacks:⁴¹ transformed objectives can be difficult to interpret for decision makers since they no longer represent distinct objectives but linear combinations thereof. Additionally, existing PCA-based methods do not take the dominance relation among solutions into account. Thus, preserving the dominance structure cannot be guaranteed.

In contrast, the dominance-based method by Brockhoff and Zitzler evaluates changes in the Pareto dominance structure induced by objective reduction.⁴² These changes in the dominance structure are quantified by the so-called δ -error. The δ -error can be interpreted as the maximum distance between the Pareto fronts of the original and the reduced objective spaces. Following this approach, objectives are considered redundant if omitting them does not change the dominance structure, resulting in a δ -error of zero. Consequently, a minimum key objective subset without changes to the dominance structure is identified if omitting further objectives would increase the δ -error above zero. The approach also allows the number of objectives to be reduced even further, accepting changes in the dominance structure in favor of a smaller objective subset.

Guillén-Gosálbez *et al.* introduced a mixed-integer linear programming (MILP) formulation for this approach to minimize the error of omitting objectives with common branch and bound algorithms.^{43,44} The advantage of this dominance-based method is twofold: identified key objectives represent a true subset instead of transformations of the full objective space. Additionally, the Pareto dominance structure can be preserved, as reported by Guillén-Gosálbez *et al.*^{43,44} It should be noted though that identified key objective subsets are not guaranteed to preserve the Pareto dominance structure. The number of input datasets, the approach to generate them, *e.g.*, heuristics, and their normalization may affect Pareto dominance preservation.

The aforementioned objective reduction methods have been already applied in several case studies with environmental objectives to identify key objective subsets (see Note S1⁺). So far, for integrated process and fuel design, key environmental objectives have not yet been identified. However, their identification is crucial to enable model-based fuel design with a justified selection of the most relevant environmental objectives. With these key environmental objectives identified, potential burden shifts in early-stage fuel design can be captured with manageable computational cost. An *a priori* selection of key environmental objectives without the use of optimization-based techniques is not feasible, due to the problem size of integrated process and fuel design.

In this work, we close this gap by identifying the key environmental objectives of integrated process and fuel design for a case study on spark-ignition engine fuels. For this purpose, we extend the existing design framework, which considers the objectives production cost and GWI, by all 16 environmental impact metrics recommended for life cycle assessment (LCA) by the European Commission. Next, the dominance-based MILP approach of Guillén-Gosálbez et al.43,44 is applied for objective reduction to identify those objectives that are key to cover the major trade-offs of the optimization problem. Key objectives are derived for two scenarios to consider potential changes in energy and feedstock supply in the next decades. With these key objectives, we generate Pareto-optimal process and fuel designs that we evaluate regarding a potential burden shift. Additionally, these designs are benchmarked against both the KEAA blend of previous studies and fossil gasoline.

The article is structured as follows. First, the problem statement is presented. Next, the methodology of life cycle assessment is explained, followed by an overview of the general solution procedure. Methods used in the solution procedure are presented subsequently: the integrated process and fuel design framework, the objective reduction algorithm, and correlation between variables. Then, the case study on spark-ignition engine fuels is briefly described. After that, the key objectives are identified and used to generate Pareto-optimal process and fuel designs. Lastly, conclusions and recommendations are drawn.

2. Problem statement

The problem we address within this study can be formally stated as follows. Given are the feedstock, electricity, heat, cooling, and refrigeration demands as well as the associated costs and environmental impacts of an array of possible fuel production pathways. Additionally, fuel requirements are given to ensure that targeted fuel standards are fulfilled. The goal of the analysis is to design fuel blends for spark-ignition engines that simultaneously minimize the production cost and environmental impacts. For this goal, we use the integrated process and fuel design framework by König *et al.*¹⁹ In this framework, decisions to be made comprise the mole fluxes of the fuel production network and the composition of the resulting fuel blend.

3. Methods

The life cycle assessment methodology is described first. Second, the overall solution procedure is outlined to identify key objectives for integrated process and fuel design. The framework for integrated process and fuel design is presented afterwards. Next, normalization of solutions is explained, followed by the applied objective reduction method. Subsequently, correlation between variables is described.

3.1. Life cycle assessment

Environmental impacts of production systems and products can be comprehensively evaluated along the entire life cycle by applying the methodology of life cycle assessment (LCA). Throughout the product system's life cycle, all material and energy flows that the product system exchanges with its environment are quantified. These exchanges are consumed natural resources, e.g., freshwater, and emitted pollutants, e.g., nitrous oxide, which are characterized according to their contribution to environmental impact categories such as water use and climate change. Owing to LCA's holistic approach, potential burden shifting between life cycle stages and impact categories can be identified and quantified. In the 1990s, LCA was standardized using the ISO 14040 and 14044 standards.45,46 According to these standards and recent guidelines for LCA practitioners, LCAs are commonly conducted in four phases that we describe briefly in the following sections.47

Goal and scope definition. In the first phase, the goal and scope of the study are defined. The goal describes the reasons for evaluating the product system under study as well as the intended application and audience. Based on the goal, the scope of the study is defined. The scope encompasses a description of the product system, its functions, and the system boundary. The so-called functional unit quantifies the functions of the product system and serves as a relative basis for a consistent comparison among different product systems such as synthetic fuels. Altering the physical unit, e.g., from gigajoule to kilojoule, of the functional unit would equally affect all results and as such not result in any qualitative difference. The system boundary includes all processes and life cycle stages that are required to fulfill the product system functions and to reach the goal of the LCA study. For instance, a so-called well-to-wheel system boundary comprises the entire life cycle of a fuel. Additionally, the evaluated environmental impact categories and their corresponding life cycle impact assessment (LCIA) method are outlined, e.g., the 16 environmental impact categories of the LCIA method "Environmental Footprint 3.0" (EF 3.0) as listed in Table 1.48

Life cycle inventory analysis. The life cycle inventory is gathered in the second phase. In this phase, material and energy balances are formulated for all processes within the system boundary, covering exchanges between processes and with the environment. Among LCA practitioners, it is common practice to divide the system into a foreground and a background system.⁴⁹ While the foreground system describes processes that the LCA practitioner can control, the background system comprises processes that cannot be directly influenced. The background system is typically modeled with aggregated datasets from LCA databases, *e.g.*, the ecoinvent database.⁵⁰

Life cycle impact assessment. In phase three, the LCIA, all exchanges with the environment are characterized regarding

Table 1 List of the considered environmental impact categories with their specified units as determined in the "Environmental Footprint 3.0" (EF 3.0) framework recommended by the Joint Research Centre (JRC) of the European Commission.⁴⁸ We adhere to the prescribed units outlined in the EF 3.0 framework for all impact categories. Altering units, *e.g.*, from megajoule to gigajoule, is possible but would affect all results related to the adjusted impact category equally.

Abbreviation	Impact category	Unit	Level of recommendation
GWI	Climate change	kg CO_2 eq.	Ι
OD	Ozone depletion	kg CFC-11 eq.	Ι
PM	Particulate matter	Disease incidence	Ι
A	Acidification	mol H^+ eq.	II
E _{fw}	Eutrophication, freshwater	kg P eq.	II
Em	Eutrophication, marine	kg N eq.	II
Et	Eutrophication, terrestrial	mol N eq.	II
IR	Ionizing radiation, human health	kBq U ²³⁵ eq.	II
POF	Photochemical ozone formation	NMVOC eq.	II
ET	Ecotoxicity, freshwater	CTUe	III
HT _c	Human toxicity, carcinogenic	CTUh	III
HTnc	Human toxicity, non-carcinogenic	CTUh	III
LU	Land use	Points	III
RUe	Resource use, energy carriers	MJ	III
RUm	Resource use, minerals and metals	kg Sb eq.	III
WU	Water use	m^3 world eq.	III

their impact on various environmental impact categories, *e.g.*, climate change and land use. The overall impact of the product system in a specific impact category is calculated by summing up these characterized exchanges. The set of environmental impact categories under study is defined in the first phase, the goal and scope definition, and should cover a wide range to identify potential burden shifting between impact categories. Noteworthily, environmental impact categories have been classified with respect to their quality level into three levels (Table 1): "Level I" is recommended and satisfactory, "Level II" is recommended but needs some improvements, and "Level III" is recommended but has to be applied with caution.⁵¹

Interpretation. In the fourth phase, the results are reviewed and interpreted regarding their ability to fulfill the previously defined goal of the LCA study. This phase includes identifying so-called hot-spots of the product system, *e.g.*, a specific process that contributes most to the environmental impacts. To reveal these hot-spots, LCA practitioners typically conduct contribution analyses. Furthermore, the results should be critically analyzed regarding their limitations and robustness, *i.e.*, assumptions made while conducting the LCA, consistency of underlying process data, and sensitivity and uncertainty of calculated results. Finally, sound conclusions should be derived to give recommendations for decision-makers. As LCA is an iterative approach, changes in the other three phases may be required, *e.g.*, refining the goal and scope of the study or including additional environmental impact categories.

3.2. Solution procedure to identify key objectives

The applied solution procedure is based on the work of Guillén-Gosálbez *et al.*^{43,44} and divided into nine steps (Fig. 1). First, an initial Pareto-optimal solution set is generated by applying the integrated process and fuel design method in the full objective space F (step 1). To generate this solution set, we minimize each single objective separately and subsequently determine the

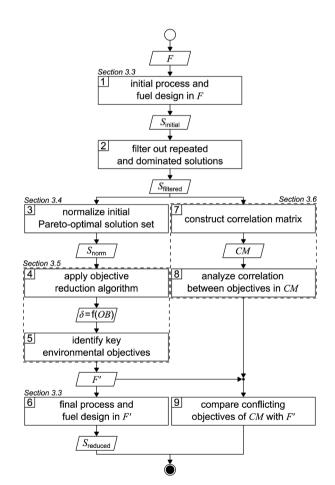


Fig. 1 Flowsheet of the applied solution procedure based on the work of Guillén-Gosálbez *et al.*^{43,44} *F*: full objective space, *F*': reduced objective space, *S*: solution set, δ : delta-error, OB: number of omitted objectives, and CM: correlation matrix.

range of each objective. With these ranges, we next apply biobjective optimization on every combination of objective pairs, using the ε -constraint method with solely four partitions in favor of manageable computational effort.^{52,53} From the 17 objectives considered in this work, there are 136 possible objective pairs with four partitions per pair, resulting in 544 solutions $S_{initial}$.

After repeated and dominated solutions have been sorted out (step 2), we obtain the filtered solutions S_{filtered} . Note that the aforementioned procedure represents a heuristic approach to limit the computational cost: the generated initial solutions are weakly Pareto-optimal since, by bi-objective optimization, we solely optimize two-dimensional projections of the 17dimensional objective space. Thus, the remaining dimensions of the objective space are not optimized. The filtered solutions S_{filtered} are used for the objective reduction algorithm and to construct a correlation matrix (left and right branches, Fig. 1).

In the left branch of Fig. 1, we proceed with normalizing the filtered solutions to obtain a normalized solution set S_{norm} (step 3). These normalized solutions are used as the input for the objective reduction method applied in step 4. By applying the objective reduction method, we calculate the δ -error, which quantifies the change in the Pareto dominance structure, as a function of the number of omitted objectives (OB). In step 5, we can identify the key objectives representing an acceptable compromise of omitted objectives and the induced δ -error. Lastly, in step 6, we generate a set of Pareto-optimal process and fuel designs $S_{reduced}$ in a reduced objective space F'. We apply multi-objective optimization on all key objectives, again using the ε -constraint method.^{52,53} In contrast to the first step, we increase the number of partitions from four to 32 in favor of a higher resolution of the Pareto front. Additionally, we refrain from using heuristics to generate these Pareto-optimal solutions, accepting increased computational cost that is partly offset by the smaller number of objectives considered.

In the right branch of Fig. 1, we use the filtered solutions S_{filtered} to construct a correlation matrix (CM) (step 7). Next, the degree of correlation among the objectives is analyzed to gain a first notion about redundancy and conflict (step 8). Lastly, in step 9, we compare the conflicting objectives from the correlation matrix with the key objectives identified by objective reduction. Based on this comparison, we derive conclusions about the suitability of correlation matrices to identify the key objectives covering an optimization problem's major trade-offs.

3.3. Integrated process and fuel design

Here, we concisely describe the integrated process and fuel design method by König *et al.*¹⁹ An in-depth description is given in the original work. Integrated process and fuel design combines a pathway model based on PNFA^{17,18} with the fuel property model of Dahmen and Marquardt.¹⁵ The most important characteristics of the pathway and fuel property models are described in the following section. Subsequently, the resulting optimization problem is explained.

Pathway model. The pathway model based on PNFA was developed as an early-stage screening tool that evaluates

reaction pathways and subsequent processing options. The model comprises an array of possible production pathways that can be depicted as a reaction network. The reaction network links feedstocks to products via multiple biomass and hydrogen conversion routes and intermediates. Product and side-product flows are calculated based on stoichiometry and yield data from the literature. Thus, direct emissions that might occur during fuel production in industrial facilities are not modeled. The production cost and environmental impacts of side-products are allocated to the main product, *i.e.*, the fuel. Moreover, the model takes into account energy consumption of reactions and separation sequences, encompassing electricity, process heat, and cooling demands. Based on this reaction network and energy consumption, economic and environmental metrics are estimated. Previous studies focused on estimating production cost and GWI although other metrics can be implemented as well. These estimations comprise the prices and environmental impacts of feedstocks, auxiliaries, and waste disposal as well as investment costs. More details on the pathway model are presented in König et al. and Ulonska et al. 17,18

Fuel property model. Fuel property models are used to ensure that properties of designed fuels are within specifications to fulfill requirements of targeted fuel standards. These fuel properties are predicted based on prior determined data that stem from either experiments or model-based approaches, using mixture property models relying on pure-component properties. For more details on the fuel property model, see Dahmen and Marquardt¹⁵ and König *et al.*^{18,20}

Optimization problem. The continuous nonlinear program (NLP) of integrated process and product design is briefly summarized in the following section. In previous studies, integrated process and product design was formulated as a biobjective optimization problem with the objectives production cost and GWI per functional unit, *i.e.*, \$ per GJ_{fuel} and kg CO₂ eq. per GJ_{fuel}, respectively.^{19,20} In this study, we reformulate the optimization problem as an MOP with the objectives production cost (*C*) in \in and 16 environmental impacts (EI) per functional unit. See Table 1 for a list of the considered environmental impact categories.

$$\min \begin{bmatrix} C \\ EI_1 \\ \vdots \\ EI_{16} \end{bmatrix}$$
(1)

s.t. *pathway model:* mole balances for products and byproducts, utility requirements of reactions and separation steps, costs (feedstocks, utilities, disposal, and investment), EI estimation of feedstocks and processes, fixed annual fuel production α . *property model:* mole and mass fractions of fuel, fuel property models and mixing rules, fuel specifications.

The production cost is evaluated by dividing the total annualized production cost C_{total} by a fixed annual fuel production of $\alpha = 2.77 \times 10^{12} \frac{\text{kJ}}{\text{a}}$. This annual fuel production is related to the energy content of 100 000 tons of ethanol per year, in line with previous studies.^{17,19,20}

$$C = \frac{C_{\text{total}}}{\alpha} \tag{2}$$

The total annualized production $\cot C_{\text{total}}$ is determined by summing up the annualized costs due to investments C_{invest} , utilities C_{util} , feedstocks $C_{\text{feedstock}}$, and wastes C_{waste} . The considered utilities are electricity, process heat, cooling, and refrigeration.

$$C_{\text{total}} = C_{\text{invest}} + C_{\text{util}} + C_{\text{feedstock}} + C_{\text{waste}}$$
(3)

The *m*th environmental impact per functional unit EI_m is calculated by dividing the total environmental impact $\text{EI}_{\text{total},m}$ by the fixed annual fuel production α .

$$\mathrm{EI}_{m} = \frac{\mathrm{EI}_{\mathrm{total},m}}{\alpha} \tag{4}$$

The total environmental impact $EI_{total,m}$ comprises the impacts due to the supply of utilities $EI_{util,m}$ and feedstocks $EI_{feedstock,m}$.

$$EI_{total,m} = EI_{util,m} + EI_{feedstock,m}$$
(5)

In this MOP, design variables are the resulting fuel composition and the mole fluxes of the reaction network. Trade-offs among objectives can be evaluated by using the ε -constraint method.^{52,53} See Note S2† and König *et al.* for more details regarding the economic and environmental objectives as well as the original optimization problem including all constraints.^{19,20}

3.4. Normalization of solutions

There are good reasons to normalize solutions prior to objective reduction: the herein considered objectives differ regarding their physical units, e.g., GWI in kg CO_2 eq. and water use in m³ eq. Thus, without normalization, the δ -error's unit would also differ depending on the omitted objective since the error is directly calculated from the objective function values (see eqn (10) and (11) below). Moreover, the scales among objectives can vary by several orders of magnitude, depending on the objectives' units. Therefore, without normalization, objectives with a small scale would be omitted preferentially, as the δ -error of omitting these objectives would be also small. However, solutions can be normalized in different ways, rendering the choice of the applied normalization variant arbitrary. We emphasize, therefore, that normalization is a critical step that induces ambiguity in the identification of key objective sets.54 In fact, the identified key objective subsets and the corresponding approximation errors might change if different normalization variants are used.

Solutions are usually normalized with mathematical or environmental reference values. As mathematical reference values, the objective-specific minimum and maximum values of all solutions are commonly used to normalize the range of each objective. Environmental reference values are typically the environmental impacts of a reference system, *e.g.*, a country's environmental impacts or planetary boundaries of the Earth. These environmental reference values can be optionally weighted, e.g., based on expert judgment, to rank the importance among objectives. Both mathematical and environmental reference values have advantages and drawbacks. While mathematical normalization attributes arbitrarily equal relevance to all objectives, mathematical reference values can be objectively calculated. In contrast, environmental normalization derives the relevance of objectives by comparison with the reference system. However, the reference system can be chosen arbitrarily and derived reference values are often uncertain, e.g., global environmental impacts or the Earth's planetary boundaries. Additionally, optional weighting based on expert judgements is highly subjective. For energy system optimization, Postels et al. analyzed the influence of normalization variants based on mathematical and environmental reference values on the identified key objective subset.55 The authors show that, in their case study, mathematical normalization variants lead to similar results while the identified key objective subsets differed substantially for the environmental reference values.

In this work, we focus on comparing four mathematical normalization variants regarding the key objective subsets identified by subsequent objective reduction. The variants (N_1-N_4) normalize each objective value $OF_{s,i}$ of solution s regarding objective *i*. For this purpose, the objective-specific minimum OF_i and maximum $\overline{OF_i}$ of all solutions are used as reference values, following Marler and Arora.⁵⁶

$$N_1: \operatorname{OF}_{s,i}^{\operatorname{norm}, N_1} = \frac{\operatorname{OF}_{s,i} - \underline{\operatorname{OF}}_i}{\overline{\operatorname{OF}}_i - \underline{\operatorname{OF}}_i}$$
(6)

$$N_2: OF_{s,i}^{norm,N_2} = \frac{OF_{s,i} - \underline{OF_i}}{\underline{OF_i}}$$
(7)

$$N_3: \operatorname{OF}_{\mathrm{s},i}^{\mathrm{norm},\mathrm{N}_3} = \frac{\operatorname{OF}_{\mathrm{s},i}}{\operatorname{OF}_i}$$
(8)

$$N_4: \operatorname{OF}_{s,i}^{\operatorname{norm}, N_4} = \frac{\operatorname{OF}_{s,i}}{\overline{\operatorname{OF}_i}}$$
(9)

3.5. Objective reduction

To identify key objectives of integrated process and fuel design, we apply the dominance-based MILP approach for objective reduction by Guillén-Gosálbez *et al.*^{43,44} This approach reduces the number of objectives in the objective vector using mathematical optimization such that changes in the Pareto dominance structure are minimized, *i.e.*, the major trade-offs of the original problem are covered. Before objective reduction is applied, it is required to normalize Pareto-optimal solutions used as the input for objective reduction (see Section 3.4). In the following section, we briefly explain the formulation of the optimization problem for objective reduction. See Note S3[†] for an exemplary minimization problem describing the underlying idea of this approach.

Here, the optimization problem for objective reduction is briefly summarized for completeness. For a more extensive description of the MILP formulation, see the work of Guillén,

Gosálbez *et al.*^{43,44} This MILP formulation is based on the problem of computing a "minimum objective subset of size *k* with minimum error" (*k*-EMOSS) by Brockhoff and Zitzler.⁴² For *q* objectives in the full objective space *F*, the optimization problem seeks to minimize the maximum δ -error induced by omitting objectives:

$$\min \max_{\mathbf{s},\mathbf{s}',i} \left(\delta_{\mathbf{s},\mathbf{s}',i} \right) \tag{10}$$

s.t. constraints 11–18.

All solutions of the minimization problem must satisfy a set of constraints that are described in the following section. The δ -error is the difference between solution s and solution s' regarding the *i*th objective (eqn (11)). Under this constraint, OF_{s,*i*} represents the normalized value of the *i*th objective of Pareto solution s, whereas ZO_i and ZD_{s,s'} are two binary variables. ZO_i is 1 if the *i*th objective is omitted while ZD_{s,s'} is 1 if solution s dominates solution s' in the reduced objective space.

$$\delta_{\mathbf{s},\mathbf{s}',i} = (\mathbf{OF}_{\mathbf{s}',i} - \mathbf{OF}_{\mathbf{s},i})\mathbf{ZO}_i\mathbf{ZD}_{\mathbf{s},\mathbf{s}'} \ \forall \ \mathbf{s} \neq \mathbf{s}',i \tag{11}$$

In eqn (12) and (13), $ZD_{s,s'}$ is calculated using a third binary variable $YP_{s',s,i}$, which is 1 if solution s is better than solution s' in the *i*th objective.

$$\left(q - \sum_{i} \text{ZO}_{i}\right) - q\left(1 - \text{ZD}_{s,s'}\right) \leq \sum_{i} \text{YP}_{s',s,i}(1 - \text{ZO}_{i}) \leq \left(q - \sum_{i} \text{ZO}_{i}\right) + q\left(1 - \text{ZD}_{s,s'}\right) \forall s \neq s'$$
(12)

$$\sum_{i} \operatorname{YP}_{\mathbf{s}',\mathbf{s},i}(1 - \operatorname{ZO}_{i}) \leq \left(q - \sum_{i} \operatorname{ZO}_{i}\right) - 1 + q \operatorname{ZD}_{\mathbf{s},\mathbf{s}'} \forall \mathbf{s} \neq \mathbf{s}'$$
(13)

The following four equations are used to linearize the product of the binary variables ZO_i and $ZD_{s,s'}$ in eqn (11):

$$(OF_{s',i} - OF_{s,i})ZOD_{s,s',i} = \delta_{s,s',i} \forall s \neq s', i$$
(14)

$$\text{ZOD}_{s,s',i} \le \text{ZO}_i \ \forall \ s \neq s', i \tag{15}$$

$$ZOD_{s,s',i} \le ZD_{s,s'} \forall s \neq s', i$$
(16)

$$\operatorname{ZOD}_{s,s',i} \ge \operatorname{ZO}_i + \operatorname{ZD}_{s,s'} - 1 \ \forall \ s \neq s', i \tag{17}$$

Lastly, the number of objectives to be omitted (OB) is prescribed:

$$\sum_{i} ZO_{i} = OB.$$
(18)

Note that alternative objective subsets may exist which exhibit equal δ -errors. To find those alternatives for a prescribed number of omitted objectives, we iteratively exclude previously found objective subsets from the solution space by adding integer cut constraints.

3.6. Correlation between two variables

Calculating the correlation coefficient is a common means to get a notion of the trade-off between variables. Often, the Pearson correlation coefficient is used to evaluate the degree of correlation between two variables for a given sample. Applied to our work, the variables are two objectives *i* and *i'*, *e.g.*, production cost and GWI. The sample comprises *n* solutions of the integrated process and fuel design. Regarding the *i*th objective, $OF_{s,i}$ denotes the objective function value of the *s*th solution, whereas OF_i^m is the mean value of objectives *i* and *i'*, the Pearson correlation between two objectives and the ratio between the covariance of both objectives and the product of their standard deviations:

$$r_{i,i'} = \frac{\sum_{s=1}^{n} (OF_{s,i} - OF_{i}^{m}) (OF_{s,i'} - OF_{i'}^{m})}{\sqrt{\sum_{s=1}^{n} (OF_{s,i} - OF_{i}^{m})^{2}} \sqrt{\sum_{s=1}^{n} (OF_{s,i'} - OF_{i'}^{m})^{2}}}.$$
 (19)

Noteworthily, the Pearson correlation coefficient ranges between -1 and 1: perfectly negative or positive correlation is implied by values of -1 or 1, whereas a value of zero implies that no linear correlation exists. For *q* objectives, a correlation matrix CM of size $q \times q$ can be constructed whose (i,i') entry is

$$CM_{i,i'} = r_{i,i'}.$$
(20)

In such a correlation matrix, all diagonal entries equal 1 since each objective is perfectly correlated with itself (i = i'). Negative or positive correlation indicates a conflicting or redundant pair of objectives, respectively.

4. Integrated process and fuel design for spark-ignition engine fuels

We apply the above described solution procedure in a case study on integrated process and fuel design for spark-ignition engine fuels by König et al.20 In the following, we briefly review the main features of this case study and describe which aspects we adapt in this work. The case study considers 47 pre-screened fuel species as suitable components for multi-component gasoline-type fuels to be used in spark-ignition engines (see Table S2[†]). For these fuel species, reaction routes for biomass and hydrogen conversion have been collected from the literature and the authors' previous studies for the pathway model. Energy demands for downstream processing have been calculated with thermodynamically sound intermediate-fidelity methods for distillation columns.16 As possible feedstocks, biomass, CO₂, and renewable hydrogen are considered for fuel production. To distinguish the renewable fuels by feedstock type, we use the terms 'bio-fuel' if biomass is used, 'e-fuel' if CO_2 and renewable hydrogen are used, and 'bio-hybrid fuel' if all three feedstocks are used.

For the fuel property model, pure-component data have been incorporated for all fuel species from databases, the literature,

and property prediction models. Fuel requirements for the socalled ultra-high efficiency engine (UHEE) fuels have been adapted from previous studies.¹⁹ These UHEE fuel requirements have been derived for highly boosted spark-ignition engines with a high compression ratio, targeting high engine efficiency and low combustion-induced pollutant emissions. The considered fuel properties are the research octane number, density, oxygen content, olefin and aromatic content, surface tension, kinematic viscosity, enthalpy of vaporization, bubble point pressure, and distillation curve.

4.1. Goal and scope definition

The overall aim of this case study is to identify those environmental metrics that are key for the integrated process and fuel design of renewable spark-ignition engine fuels. For this purpose, this LCA's goal is to analyze the environmental impacts of production-optimal fuels with tailor-made properties for spark-ignition engines within a German fuel production setting. These environmental impacts are benchmarked with fossil gasoline and the KEAA blend of previous studies.^{20,21} For a consistent comparison among different fuels, we define the functional unit as "the provision of 1 GJ of enthalpy of combustion", as recommended by Müller et al.57 and by following previous work by König et al.20 The fuel production model comprises a well-to-wheel system boundary that considers the supply of feedstocks and utilities, the pathway model for fuel production, and fuel combustion with air (Fig. 2). In contrast to previous studies, we do not allow fossil gasoline as a blending component since our aim is to design fully renewable fuels.

We characterize all material and energy flows exchanged with the environment according to the LCIA method EF 3.0, which is recommended by the Joint Research Centre of the European Commission.⁴⁸ We, therefore, extend the objective space by 15 environmental impact categories of the EF 3.0 method, in addition to the objectives production cost and GWI considered in previous case studies.^{20,21} An overview of the included impact categories with the corresponding abbreviations, units, and recommendation levels is presented in Table 1. Consequently, the resulting objective vector contains in total one economic and 16 environmental objectives. Of these 16 environmental objectives, we subsequently identify those objectives that cover the key trade-offs of integrated process and fuel design by applying the objective reduction approach. Note that, in this work, solely environmental objectives can be omitted by objective reduction to keep the only economic objective, production cost, in the key objective subset.

4.2. Life cycle inventory

For the case study, we distinguish between a 'today' and a 'future' scenario to consider potential technology changes in feedstock and energy supply (Table 2). While the 'today' scenario represents incumbent technologies, the 'future' scenario portrays a likely, fully renewable energy system. As our 'today' scenario slightly adjusts LCA datasets and prices compared to the work of König *et al.*,^{19,20} we analyze the implications of these adjustments in Note S4.† Overall, our adjustments result in a similar Pareto front for GWI and production cost, with slight increases in GWI scores due to our changes regarding the modeled steam and refrigeration supply. In the following, we briefly describe the inventory data used for the LCA as well as assumed prices for each scenario. A detailed overview of the LCA datasets is presented in Note S5.†

The feedstock CO_2 is either supplied by carbon capture at a steel plant or direct air capture (DAC). Carbon capture at a steel plant requires 0.87 MJ of electricity and 0.95 MJ of process heat per kg of captured CO_2 , according to von der Assen *et al.*⁵⁸ For DAC, we consider the predicted energy requirements

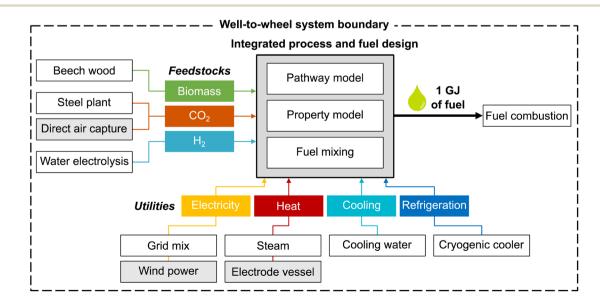


Fig. 2 Well-to-wheel system boundary of fuels optimized by integrated process and fuel design. The functional unit is "the provision of 1 GJ of enthalpy of combustion". Light grey boxes denote technologies that are used in the 'future' scenario, *e.g.*, electricity from wind power instead of the power grid.

Table 2 Scenario-specific technologies of the life cycle inventory

	Scenario		
Inputs	Today	Future	
Feedstocks			
Carbon dioxide	Steel plant	Direct air capture	
Biomass	Beech wood	Beech wood	
Hydrogen	Water electrolysis ^a	Water electrolysis ^a	
Utilities			
Electricity ^a	Grid mix	Wind power	
Process heat	Steam	Electrode vessel	
Cooling	Cooling water	Cooling water	
Refrigeration	Cryogenic cooler	Cryogenic cooler	

regardless of the scenario.

of a temperature-swing adsorption system, following Deutz *et al.*⁵⁹ Each kg of captured CO_2 *via* DAC requires 1.80 MJ of electricity and 5.40 MJ of process heat. The process heat for DAC is supplied by heat pumps with a modeled coefficient of performance of 3.28.⁶⁰ Beech wood is assumed to be a representative lignocellulosic biomass, using process data from the LCA database ecoinvent.⁵⁰ Note that no carbon credit, *i.e.*, avoided burden, is given for CO_2 that is removed from the atmosphere by either biomass growth or DAC since this CO_2 is released again to the atmosphere at the fuel's end-of-life.²⁰

Designing renewable fuels necessitates renewable hydrogen production in both scenarios. Therefore, hydrogen is assumed to be supplied renewably *via* polymer electrolyte membrane (PEM) water electrolysis in both scenarios, using electricity from wind power. Using electricity from wind power for electrolysis also in the 'today' scenario reflects a realistic scenario of importing renewable energy *via* hydrogen. Already today, socalled green hydrogen hubs are developed by several large ports, *e.g.*, those of Antwerp-Bruges, Rotterdam and Hamburg, to import and produce green hydrogen. These green hydrogen hubs are scheduled to start operating by 2025 ("Shell Holland 1", "HyoffWind", and "Plugpower") and 2026 ("Hamburg Green Hydrogen Hub"). The modelled PEM electrolysis requires 8.94 kg of water and 47.6 kW h of electricity from wind power per kg of hydrogen, using process data by Reuß *et al.*⁶¹

Regarding utilities, we consider the provision of electricity, process heat, cooling, and refrigeration. We assume that electricity for processes other than water electrolysis is mainly supplied by the German grid in the 'today' scenario and by clean electricity from wind power in the 'future' scenario, using process data from ecoinvent.⁵⁰ Process heat is provided *via* steam or an electrode vessel with a power-to-heat efficiency of 95%.⁵⁷ For cooling and refrigeration, the use of cooling water⁵⁰ and cryogenic coolers is assumed, respectively. In this early-stage design, we assume a constant refrigeration temperature of -100 °C as a rough estimation and model cryogenic coolers with an energy efficiency ratio of 0.25, based on curve fit functions of Ladner *et al.*⁶²

Use phase pollutants from non-ideal fuel combustion are not covered in integrated process and fuel design. Soot emissions for hydrocarbon fuels can be predicted using the particulate matter index by Aikawa et al.,63 which predicts engine-out emissions based on the fuel's volatility and its chemical tendency to form soot. As a measure for the latter, the number of double bonds is used. However, soot chemistry becomes more complicated for oxygenated hydrocarbons since different oxygen functionalities reduce the formation of soot precursors to different degrees.⁶⁴ While this effect is captured by the yield sooting index by Das et al.,65 the yield sooting index does however not predict engine-out emissions. Further pollutant emissions like unburned hydrocarbons, carbon monoxide, and nitrogen oxides depend strongly on the engine operation and can therefore not be predicted at an early design stage. Additionally, engine efficiencies are disregarded for simplicity, i.e., it is assumed that different fuels could be used with equal efficiency in spark-ignition engines.20

For price estimations, we adopt prices from previous studies on integrated process and fuel design by König et al. where possible, maintaining comparability with preceding studies as much as possible.19,20 Due to the German scope of this study, the following prices are converted from United States dollar (\$) to euro (\in), using 2022's average exchange rate of 1.05 \$ per \in . These adopted prices comprise 7.5 \$-ct per kW h of electricity from the grid,⁶⁶ 9.5 \$ per ton of steam,⁶⁶ 6.5 \$-ct per m³ of cooling water,66 5.78 \$ per kg of hydrogen from water electrolysis using electricity from wind power,⁶⁷ 50 \$ per ton of biomass,68 and 40 \$ per ton of CO2 captured at a steel plant.69 We further assume prices of 6.0 \$-ct per kW h of electricity from wind power,70 6.3 \$-ct per kW h of process heat from electrode vessels using electricity from wind power with a power-to-heat efficiency of 95%, and 200 \$ per ton of CO₂ captured from ambient air.71,72 Cryogenic refrigeration costs are modeled for the assumed refrigeration temperature of -100 °C and a twostage compression refrigeration system, according to Luyben:73 refrigeration costs result in 22.4 or 18.9 \$-ct per kW h of refrigeration if electricity from the grid or wind power is used, respectively.

5. Results

Both the continuous NLP of integrated process and fuel design and the MILP of the objective reduction approach are implemented in GAMS v35.2.0 and solved on a computer with an Intel E5-1660 v4 3.2 GHz processor and 128 GB RAM. The continuous NLP is solved with the deterministic global solver BARON v21.1.13, using one thread, a branching priority of 20 on the molar fuel fraction, a time limit of 1000 seconds, and a relative optimality tolerance of 0.1% (single objective optimization) or 1% (bi- and multi-objective optimization). For the MILP, we use the solver CPLEX v20.1.0.1 and four parallel threads. Any other solver settings are left at default values. Similar to König *et al.*,²⁰ some solutions of the continuous NLP are globally optimal while other runs do not converge to the specified optimality even though we increased the time limit drastically. Similar to previous work, the lower bound convergence stagnates for these runs, and we follow the assumption of König *et al.* that the found solutions are globally optimal.

After filtering out the initial 544 solutions S_{initial}, 156 and 158 unique solutions Stoday and Sfiltered remain for the 'today' and 'future' scenario, respectively (see Tables S6 and S7[†]). With these unique Pareto-optimal solutions, we construct correlation matrices for both scenarios to gain a first notion about conflicting and redundant objective pairs (see Tables S10 and S11⁺). The correlation matrices indicate that the most conflicting objectives are land use in the 'today' scenario as well as land use and freshwater eutrophication in the 'future' scenario. Furthermore, weakly correlated objectives are production cost in both scenarios and freshwater eutrophication in the 'today' scenario. All other objective pairs show large coefficients in the correlation matrices, indicating redundancy among these objectives. We next normalize the scenario-specific filtered solutions with each normalization variant (see Section 3.4), yielding four normalized solution sets per scenario, e.g., $S_{\text{norm},N_1}^{\text{today}}$, $S_{\text{norm},N_2}^{\text{today}}$ and so on.

5.1. Identifying key environmental objectives

After generating the normalized solution sets, we identify key environmental objectives and redundant ones for each scenario. To analyze how the problem's dominance structure may change if we omit redundant objectives, we apply the dominance-based objective reduction approach (see Section 3.5). Thus, we quantify this change, *i.e.*, the δ -error, as function of the number of omitted objectives for each normalization variant (Fig. 3). Note that the normalization variants lead to different ranges of values regarding the normalized solutions and, consequently, resulting δ -errors. To enable comparability between δ -errors calculated with different normalization variants, we use the relative δ -error δ_{rel} introduced by Postels *et al.*:⁵⁵ For each normalization variant, the δ -error is referenced to its maximum value that occurs if all but one, *i.e.*, 16 out of 17, objectives are omitted. Additionally, we specify a threshold δ^* of 0.1 to identify small key objective subsets with a still acceptable δ -error (Fig. 3, dotted horizontal line).

Regarding the determined threshold δ^* of 0.1, two numbers of omitted objectives are of greater interest: the maximum number of omitted objectives for which all (square, Fig. 3) or at least one (circle, Fig. 3) normalization variant is below the

threshold. For both scenarios, many objectives can be omitted without inducing a δ -error, *i.e.*, a change in the dominance structure of integrated process and fuel design. Without inducing a δ -error, up to nine and ten objectives can be omitted in the 'today' and 'future' scenario (square, Fig. 3), respectively. In the 'today' scenario, all normalization variants result in the same minimal reduced objective subset without a δ -error of $F_{OB=9}^{today} = (C, LU, RU_m, RU_e, WU, E_{fw}, HT_c, PM)^T$. In contrast, in the 'future' scenario, two minimal reduced objective subsets without δ -error are identified by all normalization variants: $F_{\text{OB}=10}^{\text{future,1}} = (\text{C, LU, RU}_{\text{m}}, \text{RU}_{\text{e}}, \text{WU, ET, } \text{HT}_{\text{nc}})^T$ and $F_{\text{OB}=10}^{\text{future,2}} =$ $(C, LU, RU_m, RU_e, WU, ET, IR)^T$. Hence, the objectives production cost (C), land use (LU), resource use of minerals and metals (RU_m) and energy carriers (RU_e), and water use (WU) are present in the reduced objective subsets without δ -error in both scenarios.

With normalization variant N_2 , further objectives can be omitted while still maintaining a δ -error below the threshold δ^* . In both scenarios, up to 14 objectives can be omitted (circle, Fig. 3), resulting in equally reduced objective subsets of $F_{OB=14}^{N_2,today}$ = $F_{\text{OB}=14}^{N_2,\text{future}} = (\text{C, LU, RU}_{\text{m}})^T$ with δ -errors of 0.09 and 0.06, respectively. Note that increasing the number of Pareto points generated as inputs for objective reduction has negligible effect on the δ -error while the identified reduced objective subset $F_{OB=14}^{N_2,today}$ is entirely unaffected (see Note S6^{\dagger}). The other normalization variants, N_1 and N_4 , result in the same reduced objective subsets but with much greater δ -errors. In fact, normalization variant N_1 reaches the maximal δ -error of one for 14 omitted objectives in the 'today' scenario, rendering the choice of the three objectives arbitrary. Notable, alternative objective subsets with equivalent approximation error and three objectives were not found. However, in nearoptimal objective subsets, other sets of three objectives can be seen (e.g., land use (LU) and water use (WU)).

Overall, we find that the normalization variants considered herein do not affect the minimal reduced objective subsets without δ -error identified in both scenarios (square, Fig. 3). If even smaller reduced objective subsets are targeted, normalization affects, however, both the identified reduced objective subsets and the induced δ -error. Nonetheless, land use (LU) and resource use of minerals and metals (RU_m) stand out against the other objectives since they are present in all reduced objective subsets for up to 14 objectives omitted, irrespective of the

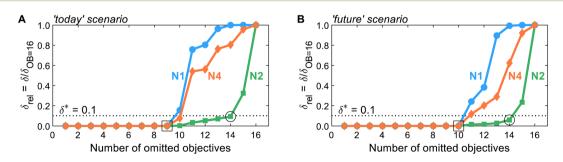


Fig. 3 The δ -error δ_{rel} as a function of the number of omitted objectives for normalization variants N_1 , N_2 , and N_4 in the (A) 'today' or (B) 'future' scenario. Note that normalization variant N_3 is not shown since it yields identical results to N_2 , as both variants differ only by a constant offset of 1. A threshold δ^* of 0.1 is introduced to identify small objective subsets with an acceptable δ -error. For numerical results and the corresponding identified objective subsets, see Tables S8 and S9.†

scenario and normalization variant (see Tables S8 and S9†). We, therefore, suggest using land use (LU) and resource use of minerals and metals (RU_m) as key environmental objectives for integrated process and fuel design of spark-ignition engine fuels.

When we compare these findings with the constructed correlation matrices (see Note S7†), we find, as *a priori* expected, that correlation matrices alone provide insufficient information

to select reduced objective subsets. Land use is the only objective that is conflicting with any other objective in the 'today' scenario while, in the 'future' scenario, land use and freshwater eutrophication are both in conflict with the other objectives. On the one hand, this degree of conflict matches our findings from objective reduction that land use is a key objective. On the other hand, the correlation matrices indicate that resource use of minerals and metals is correlated with the other objectives

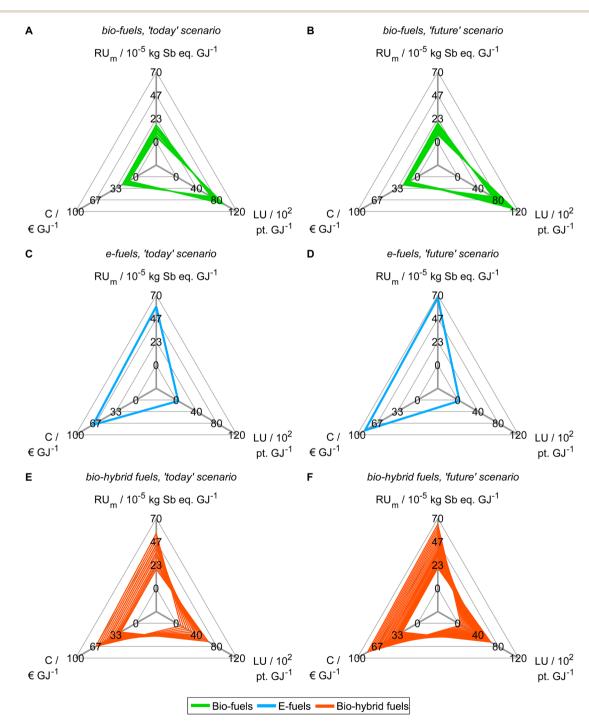


Fig. 4 Pareto-optimal process and fuel designs of bio-, e-, and bio-hybrid fuels regarding the reduced objective space of the 'today' (A, C and E) and 'future' (B, D and F) scenarios. In Note S8,† we present an alternative figure that shows the single-objective minima of the Pareto-optimal solutions generated with the reduced objective subsets. For contribution analyses of these solutions, see Note S9.†

although it is also a key objective from objective reduction. Therefore, correlation matrices alone appear unsuitable for choosing key objectives since the error inherent to omitting objectives cannot be quantified.

5.2. Integrated process and fuel design with key environmental objectives

Based on our findings from objective reduction, we proceed with generating Pareto-optimal process and fuel designs for sparkignition engine fuels in a reduced objective space (Fig. 4). It would be sensible to perform the integrated fuel design with the minimal objective subset that does not induce a δ -error irrespective of the normalization variant (see the square, Fig. 3). However, multi-objective optimization with eight objectives comes at the cost of high computational effort for a well-resolved Pareto front. Here, we thus choose the smaller reduced objective subset $F_{\text{OB}=14}^{N_3,\text{totary}} = F_{\text{OB}=14}^{N_3,\text{tuture}} = (\text{C, LU, RU}_m)^T$ containing both key

 Table 3
 Number of unique Pareto-optimal solutions based on the fuel type

	Scenario	
	Today	Future
Total	43	46
Bio-fuels	19	18
E-fuels	1	1
Bio-hybrid fuels	23	27

environmental objectives land use (LU) and resource use of minerals and metals (RU_m) to gain a well-resolved Pareto front with manageable computational effort. Note though that this reduced objective subset induces a δ -error that is small for normalization variant N_2 but large for N_1 and N_4 (see Section 5.1). We obtain 43 and 46 unique Pareto-optimal solutions for the 'today' and the 'future' scenario, respectively (see Tables S12 and S13†). The majority of Pareto-optimal solutions are biohybrid fuels, followed by bio-fuels and only one e-fuel (Table 3). For contribution analyses of these solutions, see Note S9.†

In both scenarios, the bio-fuels yield the lowest production cost (C) and resource use of minerals and metals (RU_m) but the highest land use (LU) (Fig. 4A and B). In contrast, the e-fuel has minimal land use but the highest scores in both other objectives (Fig. 4C and D). Notably, in the 'future' scenario, e-fuels exhibit slightly higher production cost and resource use of minerals and metals: CO₂ supply by DAC is expected to be much more expensive than CO₂ capture from industrial point sources, e.g., capture at steel plants, resulting in increased production cost (see Fig. S6C and D⁺). Additionally, the demand for minerals and metals increases as electrification with renewable energy is increasing in all life cycle phases (see Fig. S7C and D⁺). The bio-hybrid fuels can balance all three objectives (Fig. 4E and F), leveling the advantages and drawbacks of bio- and e-fuels: lower land use scores than those of bio-fuels are possible but at the cost of increases in the other objectives and vice versa.

Based on the generated Pareto-optimal solutions, we next analyze trends in the full objective space and benchmark the generated solutions with gasoline and the KEAA blend, which

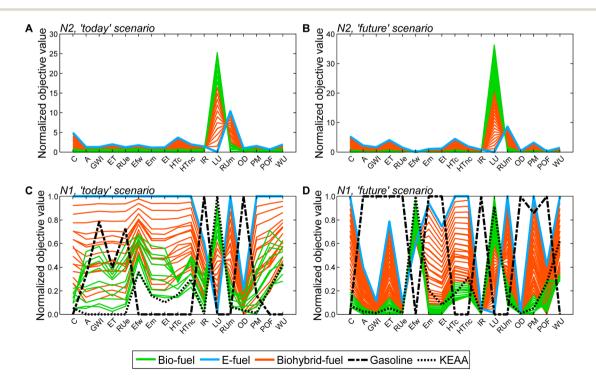


Fig. 5 Pareto-optimal process and fuel designs of bio-, e-, and bio-hybrid fuels evaluated in the full objective space for the 'today' (A and C) and 'future' (B and D) scenarios. In (A) and (B), each objective is normalized according to normalization variant N_2 to analyze trends among the generated solutions. In (C) and (D), objectives are normalized with variant N_1 to facilitate the comparison with both benchmarks gasoline and the bio-based KEAA blend. For consistency, the KEAA blend of previous studies is recalculated with our 'today' and 'future' scenarios.

was designed in a previous study considering only production cost and GWI.²⁰ For the analysis of the generated solutions, we normalize the solutions with variant N_2 (Fig. 5A and B). For the comparison with both benchmarks shown in Fig. 5C and D, we normalize with variant N_1 instead. With variant N_1 , all objectives are scaled to the range [0,1], facilitating comparison especially for objectives with small variance among all generated solutions, *e.g.*, acidification (A).

When we evaluate the Pareto-optimal solutions in the full objective space (Fig. 5A and B), we see that land use (LU) and resource use of minerals and metals (RU_m) represent, as expected, the most apparent trade-off in both scenarios and for all fuels, *i.e.*, the higher the land use the lower the resource use of minerals and metals and vice versa. Besides land use (LU) and resource use of minerals and metals (RU_m), there is also larger variance among solutions regarding production cost (C) and human toxicity (HT) for the 'today' scenario as well as additionally ecotoxicity (ET) and particulate matter (PM) for the 'future' scenario. Overall, the e-fuel performs worst in all objectives except for land use in the 'today' scenario and additionally in freshwater eutrophication (E_{fw}) in the 'future' scenario. Conversely, the bio-fuels rank worst for these objectives and best for all others. As already indicated by the results from the reduced objective space (see Fig. 4), bio-hybrid fuels level the impacts of bio- and e-fuels in most objectives.

Comparing the generated solutions with the benchmark gasoline reveals that the designed fuels can achieve better or at least similar scores compared to gasoline in many objectives (Fig. 5C and D). However, in both scenarios, gasoline has the lowest production cost (C), freshwater eutrophication (E_{fw}), human toxicity (HT), land use (LU), resource use of minerals and metals (RU_m), and water use (WU). For human toxicity (HT), electricity consumption is the main driver for the generated solutions (see Fig. S19 and S20†). Depending on whether electricity from wind power or the grid is consumed, the major contributors are the production of either metals or lignite and hard coal, respectively, and associated waste treatment processes. Conversely, gasoline performs worse, if not the worst, in ionizing radiation (IR) and ozone depletion (OD), acidification (A), GWI, ecotoxicity (ET), and resource use of fossil energy carriers (RU_e).

Benchmarking the generated solutions with the KEAA blend from previous fuel design studies^{20,21} shows that the KEAA blend is still an exceptionally promising synthetic fuel, except for land use (LU). Apart from land use, the KEAA blend yields the lowest scores among all synthetic fuels for almost every objective in the 'today' scenario. Moreover, the KEAA blend is also among the best synthetic fuels in our 'future' scenario although the blend was originally designed for a scenario considering solely today's technologies.^{20,21} Nonetheless, we can identify bio-fuel designs among our generated solutions that yield better scores than the KEAA blend in all objectives for the 'future' scenario (see Fig. S5†).

6. Discussion

Our findings indicate that up to nine or ten of 17 objectives can be omitted in early-stage fuel design for our 'today' or 'future' scenario, respectively. For these objective subset sizes, no error is induced regarding the approximated Pareto front, irrespective of the applied normalization variant. If even smaller objective subsets are desired for fuel design, one has to weigh the benefit of a decreasing computational effort against the drawback of an increasing approximation error. Notably, the approximation error differs depending on the normalization variant applied. Normalization is thus a critical, arbitrary step since various normalization variants exist.

By reducing the objective space even further, we find that "land use" and "resource use of minerals and metals" are present in all reduced objective subsets irrespective of the scenario, normalization variant, and objective subset size. In our view, these results underline the relevance of both objectives to cover the major trade-offs in the design of bio-, e-, and bio-hybrid fuels for spark-ignition engines. We, therefore, consider "land use" and "resource use of minerals and metals" as key environmental objectives and suggest the inclusion of both, in addition to "production cost," for fuel design.

Our integrated process and fuel design with these key environmental objectives and "production cost" demonstrates that bio-fuels yield among the best scores in most objectives but shift burdens to large increases in land use due to biomass cultivation. Conversely, the e-fuel has the lowest land use but the highest scores regarding all other objectives. We show that the generated bio-hybrid fuels can step in as balanced fuel designs: by using both biomass and renewable electricity as feedstocks, they can balance the benefits and drawbacks of bioand e-fuels. Compared to the KEAA blend of previous studies,20,21 most blend components of the generated bio-hybrid fuels have similar shares, e.g., ethanol and ethyl acetate (see Note S10[†]). However, some blend components differ to a larger extent: the bio-hybrid fuels contain more than twice as much methyl acetate (45%) as the KEAA blend, whereas methyl isopropyl ketone, one of KEAA's main components, is completely absent.

In the following, we discuss the generalizability of our findings towards a universal assessment of renewable fuels in terms of the considered scenario, the product, the production scale, data gaps in the process model, and the methodological maturity of the environmental impact categories. In general, it cannot be guaranteed that our findings hold generically if the scenarios are altered or other products are to be designed. Thus, we underline that the herein identified key environmental objectives are case-specific for both the considered scenarios and the design of advanced spark-ignition engine fuels. Similar findings are, however, likely if alternative utilities and feedstocks are considered whose costs and environmental impacts differ only marginally from those of the technologies considered herein.

Notably, our 'future' scenario does not reflect potential changes in the background system although grid decarbonization will progress and less carbon-intensive technologies, *e.g.*, low-carbon steel, will become more established in the future. This influence of background system changes on environmental impacts was analyzed for a national energy system optimization by Reinert *et al.*⁷⁴ The authors found that while accounting for changes does indeed reduce most

environmental impacts, the general trends are adequately captured with a static background system as applied in our work. Overall, we therefore do not expect our findings to change substantially if changes in the background system would be considered.

Furthermore, we are confident that the key environmental objectives also hold for other liquid carbon-based energy carriers produced from renewable feedstocks, e.g., diesel-type fuels, since the involved processing steps are similar among different synthetic fuels. The two identified key environmental objectives "land use" and "resource use of minerals and metals" account for the fact that the production of renewable carbonbased energy carriers involves a step to chemically reduce CO₂, which is present in the atmosphere. In the case of biomass, CO₂ reduction through photosynthesis requires land for biomass cultivation. For e-based energy carriers, CO₂ is reduced by hydrogen from renewable electricity via water electrolysis, which requires minerals and metals for the construction of wind parks and photovoltaic plants. These findings suggest that both key environmental objectives also cover the trade-offs of GWI, which is strongly interconnected with CO₂.

The scale of fuel production modelled might change the identified subset of key environmental objectives. While we assume constant prices and environmental impacts of feedstocks and auxiliaries as input parameters for our model, these parameters can be expected to vary for different production scales. With increasing production scale, prices and environmental impacts of feedstocks and auxiliaries might change due to market effects such as the price elasticity of demand and land-use change emissions induced by the development of new land for biomass cultivation or wind parks. Although such scaling effects could be modeled, in principle, by incorporating integrated assessment models into our framework, solving these models would further increase computational cost drastically.

Regarding data gaps in the process model, our findings might change if more detailed models of transportation, production, and use phase are developed for the fuel design framework. Currently, the cost and environmental impacts of feedstock transportation are neglected. However, the supply of feedstocks, in particular renewable hydrogen, may require their production or cultivation in regions with high renewable energy potential and subsequent transportation around the globe. During fuel production, potential direct emissions are not covered since the fuel design framework was developed as an early-stage screening tool based on stoichiometry and yield data. In the use phase, combustion-induced pollutants, *e.g.*, soot, nitrogen oxides, and carbon monoxide, are neglected since, so far, they cannot be predicted accurately at an early design stage.

Noteworthily, the environmental impact categories considered herein vary regarding their methodological maturity (Table 1), which might be an additional source of uncertainty for the identified key environmental objectives. Only three of 16 impact categories are classified as recommended and satisfactory. Although also recommended, the remaining 13 impact categories are still classified as in need of some improvement or to be applied with caution (see Section 3.1). For instance, human toxicity (HT) and the identified key environmental objectives land use (LU) and resource use of minerals and metals (RU_m) are recommended but should be applied with caution. Regarding human toxicity, it is questionable whether gasoline is actually less toxic to humans over the entire life cycle than all renewable fuels generated herein (see Fig. 5C and D). This finding might hint at data gaps in the underlying LCIA methods because gasoline is well-known for its acute toxicity.

As a consequence, we recommend the verification our findings with the approach described herein if the case study is altered or another product or production scale is studied. Likewise, the identified key environmental objectives should be verified if the fuel design framework is enhanced, *e.g.*, by a sophisticated engine model, or if environmental impact categories that are currently immature are improved.

7. Conclusions

This work suggests key environmental objectives for the integrated process and fuel design of spark-ignition engine fuels. For this goal, we add 16 common environmental impact metrics of the life cycle assessment methodology as additional objectives to the integrated process and fuel design framework. We analyze a 'today' and a 'future' scenario to consider technology changes in feedstock and utility supply. To identify a reduced objective subset covering the problem's key trade-offs, we use a dominance-based optimization approach to reduce the number of objectives in the objective vector with minimal changes to the problem's dominance structure. As it is required to normalize input data for this objective reduction approach, we evaluate the normalization's influence on the identified reduced objective subset and the induced change in the problem's dominance structure. Lastly, Pareto-optimal process and fuel designs are generated for the identified reduced objective subset.

Overall, we consider "land use" as well as "resource use of minerals and metals" as key environmental objectives, covering the major trade-offs of integrated process and fuel design for advanced spark-ignition engine fuels. Therefore, we recommend the inclusion of both as additional objectives to production cost in future fuel design studies. As demonstrated herein, we suggest the evaluation of the identified key objectives regarding their robustness towards different normalization variants since normalization of input data for objective reduction is arbitrary. Using these three objectives for integrated process and fuel design, we show that the obtained bio-hybrid fuels for spark-ignition engines can balance the benefits and drawbacks of pure bio- and e-fuels.

As a complement to this study's objective reduction approach, future studies could assess the absolute sustainability of the Pareto-optimal process and fuel designs generated herein. This way, burden-shifting observed among the generated designs could be put into perspective by comparison with the Earth's planetary boundaries. However, LCA-based planetary boundaries would introduce a new source of uncertainty for two reasons.⁷⁵ First, planetary boundaries are derived from complex ecological mechanisms that are difficult to quantify and to attribute to LCIA models. Second, downscaling of the socalled safe operating space would be required since synthetic fuels would account only for a fraction of all human activities. As different allocation principles could be applied for downscaling, this step is reported as subject to the greatest uncertainty when applying planetary boundaries.⁷⁶

Lastly, we stress that our recommendations on key environmental objectives refer explicitly to early-stage process and fuel design. In this context, the emphasis is on using as few objectives as possible to reduce the computational effort without neglecting major trade-offs of the optimization problem. Therefore, our recommendations by no means imply that objectives identified as redundant for fuel design are negligible for overall decision-making. For instance, global warming impact is surprisingly not suggested as a key environmental objective for early-stage fuel design, indicating that its trade-offs with the other objectives can be adequately captured by the reduced objective subsets identified in this work. Nonetheless, global warming impact represents one of the most important metrics in decision-making to evaluate the climate change mitigation potential of new technologies.

Author contributions

Simon Voelker: conceptualization, methodology, software, investigation, visualization, writing – original draft, project administration. Philipp Ackermann: conceptualization, methodology, software, investigation, writing – review & editing. Marcel Granderath: software, investigation, visualization, writing – review & editing. Clemens Kortmann: software, investigation, writing – review & editing. Joern Viell: conceptualization, writing – review & editing. Alexander Mitsos: conceptualization, supervision, writing – review & editing, funding acquisition. Niklas von der Assen: conceptualization, supervision, writing – review & editing, funding acquisition.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 M. Dahmen and W. Marquardt, *Energy Fuels*, 2016, **30**(2), 1109.
- 2 R. L. McCormick, G. Fioroni, L. Fouts, E. Christensen, J. Yanowitz, E. Polikarpov, K. Albrecht, D. J. Gaspar, J. Gladden and A. George, *SAE Int. J. Fuels Lubr.*, 2017, **10**(2), 442.

- 3 D. Gschwend, P. Soltic, A. Wokaun and F. Vogel, *Energy Fuels*, 2019, **33**(3), 2186.
- 4 G. Fioroni, L. Fouts, J. Luecke, D. Vardon, N. Huq,
 E. Christensen, X. Huo, T. Alleman, R. McCormick,
 M. Kass, E. Polikarpov, G. Kukkadapu and
 R. A. Whitesides, *SAE Int. J. Adv. Curr. Pract. Mobil.*, 2019, 1(3), 1117.
- 5 X. Huo, N. A. Huq, J. Stunkel, N. S. Cleveland, A. K. Starace,
 A. E. Settle, A. M. York, R. S. Nelson, D. G. Brandner, L. Fouts,
 P. C. St. John, E. D. Christensen, J. Luecke, J. H. Mack,
 C. S. McEnally, P. A. Cherry, L. D. Pfefferle,
 T. J. Strathmann, D. Salvachúa, S. Kim, R. L. McCormick,
 G. T. Beckham and D. R. Vardon, *Green Chem.*, 2019,
 21(21), 5813.
- 6 F. vom Lehn, L. Cai, R. Tripathi, R. Broda and H. Pitsch, *Appl. Energy Combust. Sci.*, 2021, 5, 100018.
- 7 J. G. Rittig, M. Ritzert, A. M. Schweidtmann, S. Winkler, J. M. Weber, P. Morsch, K. A. Heufer, M. Grohe, A. Mitsos and M. Dahmen, *AIChE J.*, 2023, **69**(4), e17971.
- 8 L. Fleitmann, P. Ackermann, J. Schilling, J. Kleinekorte, J. G. Rittig, F. vom Lehn, A. M. Schweidtmann, H. Pitsch, K. Leonhard, A. Mitsos, A. Bardow and M. Dahmen, *Energy Fuels*, 2023, **37**(3), 2213.
- 9 N. A. Yunus, K. V. Gernaey, J. M. Woodley and R. Gani, Comput. Chem. Eng., 2014, 66, 201.
- 10 L. Zhang, S. Kalakul, L. Liu, N. O. Elbashir, J. Du and R. Gani, *Ind. Eng. Chem. Res.*, 2018, **57**(20), 7008.
- 11 H. Hashim, M. Narayanasamy, N. A. Yunus, L. J. Shiun,
 Z. A. Muis and W. S. Ho, *J. Cleaner Prod.*, 2017, 146(2), 208.
- 12 S. Kalakul, L. Zhang, Z. Fang, H. A. Choudhury, S. Intikhab, N. Elbashir, M. R. Eden and R. Gani, *Comput. Chem. Eng.*, 2018, **116**(1), 37.
- 13 W. A. Marvin, S. Rangarajan and P. Daoutidis, *Energy Fuels*, 2013, 27(6), 3585.
- 14 A. Voll and W. Marquardt, AIChE J., 2012, 58(6), 1788.
- 15 M. Dahmen and W. Marquardt, *Energy Fuels*, 2017, **31**(4), 4096.
- 16 J. Bausa, R. v. Watzdorf and W. Marquardt, *AIChE J.*, 1998, 44(10), 2181.
- 17 K. Ulonska, M. Skiborowski, A. Mitsos and J. Viell, *AIChE J.*, 2016, **62**(9), 3096.
- 18 A. König, K. Ulonska, A. Mitsos and J. Viell, *Energy Fuels*, 2019, **33**(2), 1659.
- 19 A. König, L. Neidhardt, J. Viell, A. Mitsos and M. Dahmen, *Comput. Chem. Eng.*, 2020, **134**, 106712.
- 20 A. König, M. Siska, A. M. Schweidtmann, J. G. Rittig, J. Viell,
 A. Mitsos and M. Dahmen, *Chem. Eng. Sci.*, 2021, 237, 116562.
- 21 P. Ackermann, K. E. Braun, P. Burkardt, S. Heger, A. König, P. Morsch, B. Lehrheuer, M. Surger, S. Völker, L. M. Blank, M. Du, K. A. Heufer, M. Roß-Nickoll, J. Viell, N. von der Aßen, A. Mitsos, S. Pischinger and M. Dahmen, *ChemSusChem*, 2021, 14(23), 5254.
- 22 C. Hank, L. Lazar, F. K. Mantei, M. Ouda, R. J. White, T. Smolinka, A. Schaadt, C. Hebling and H.-M. Henning, *Sustainable Energy Fuels*, 2019, 3(11), 3219.

- 23 J. D. Medrano-García, M. A. Charalambous and G. Guillén-Gosálbez, ACS Sustainable Chem. Eng., 2022, 987.
- 24 I. Muñoz, K. Flury, N. Jungbluth, G. Rigarlsford,
 L. M. i. Canals and H. King, *Int. J. Life Cycle Assess.*, 2014, 19(1), 109.
- 25 J. Kleinekorte, L. Fleitmann, M. Bachmann, A. Kätelhön,
 A. Barbosa-Póvoa, N. von der Assen and A. Bardow, *Annu. Rev. Chem. Biomol. Eng.*, 2020, 11, 203.
- 26 B. Li, J. Li, K. Tang and X. Yao, ACM Comput. Surv., 2015, 48(1), 1.
- 27 Z. J. N. Steinmann, A. M. Schipper, M. Hauck and M. A. J. Huijbregts, *Environ. Sci. Technol.*, 2016, 50(7), 3913.
- 28 T. Gal and H. Leberling, Eur. J. Oper. Res., 1977, 1(3), 176.
- 29 P. Vaskan, G. Guillén-Gosálbez, M. Turkay and L. Jiménez, *Ind. Eng. Chem. Res.*, 2014, 53(50), 19559.
- 30 D. Vázquez, R. Ruiz-Femenia, L. Jiménez and J. A. Caballero, Ind. Eng. Chem. Res., 2018, 57(20), 6992.
- 31 D. K. Saxena, J. A. Duro, A. Tiwari, K. Deb and Q. Zhang, *IEEE Trans. Evol. Comput.*, 2013, 17(1), 77.
- 32 H. K. Singh, A. Isaacs and T. Ray, *IEEE Trans. Evol. Comput.*, 2011, 15(4), 539.
- 33 A. López Jaimes, C. A. Coello Coello and D. Chakraborty, in Proceedings of the 10th Annual Conference on Genetic and Evolutionary Computation, ed. C. Ryan, ACM, New York, NY, 2008, p. 673.
- 34 J. Wheeler, J. A. Caballero, R. Ruiz-Femenia, G. Guillén-Gosálbez and F. D. Mele, *Comput. Chem. Eng.*, 2017, 102(4), 64.
- 35 M. Pal, S. Saha and S. Bandyopadhyay, *Inf. Sci.*, 2018, **423**(1), 200.
- 36 D. G. Oliva, G. Guillén-Gosálbez, J. M. Mateo-Sanz and L. Jiménez-Esteller, *Comput. Chem. Eng.*, 2013, **56**, 202.
- 37 N. V. Thoai, J. Glob. Optim., 2012, 52(3), 499.
- 38 K. Deb and D. K. Saxena, in *Proceedings of the World Congress* on Computational Intelligence (WCCI-2006), 2006, p. 3352.
- 39 D. K. Saxena and K. Deb, in Evolutionary Multi-Criterion Optimization: 4th International Conference, EMO 2007, Matsushima, Japan, March 5–8, 2007, Proceedings, ed. S. Obayashi, Springer Berlin/Heidelberg, Berlin, Heidelberg, 2007, p. 772.
- 40 K. Deb and D. K. Saxena, On Finding Pareto-Optimal Solutions Through Dimensionality Reduction for Certain Large-Dimensional Multi-Objective Optimization Problems, 2005, 1.
- 41 D. Brockhoff and E. Zitzler, *Evol. Comput.*, 2009, **17**(2), 135. 42 D. Brockhoff and E. Zitzler, in *Parallel Problem Solving from*
- Nature PPSN IX: 9th International Conference, Reykjavik, Iceland, September 9–13, 2006, Proceedings, ed. D. Hutchison, T. Kanade, J. Kittler, J. M. Kleinberg, F. Mattern, J. C. Mitchell, M. Naor, O. Nierstrasz, C. Pandu Rangan, B. Steffen, M. Sudan, D. Terzopoulos, D. Tygar, M. Y. Vardi, G. Weikum, T. P. Runarsson, H.-G. Beyer, E. Burke, J. J. Merelo-Guervós, L. D. Whitley, X. Yao and E. K. Burke, Springer Berlin Heidelberg, Berlin, Heidelberg, 2006, p. 533.
- 43 G. Guillén-Gosálbez, Comput. Chem. Eng., 2011, 35(8), 1469.
- 44 A. Kostin, G. Guillén-Gosálbez, F. D. Mele and L. Jiménez, Ind. Eng. Chem. Res., 2012, 51(14), 5282.

- 45 ISO, ISO 14040: Environmental Management: Life Cycle Assessment: Principles and Framework, 2006.
- 46 ISO, ISO 14044: Environmental Management: Life Cycle Assessment: Requirements and Guidelines, 2006.
- 47 J. Sadhukhan, K. S. Ng and E. M. Hernandez, *Biorefineries* and Chemical Processes, Wiley, 2014.
- 48 European Commission, Supporting information to the Characterisation Factors of Recommended EF Life Cycle Impact Assessment Methods: Version 2, from ILCD to EF 3.0, 2018.
- 49 R. Frischknecht, Life Cycle Inventory Analysis for Decision-Making: Scope-dependent Inventory System Models and Context-specific Joint Product Allocation, PhD thesis, ETH Zurich, Zurich, 1998.
- 50 G. Wernet, C. Bauer, B. Steubing, J. Reinhard, E. Moreno-Ruiz and B. Weidema, *Int. J. Life Cycle Assess.*, 2016, **21**(9), 1218.
- 51 European Commission, International Reference Life Cycle Data System (ILCD) Handbook: General Guide for Life Cycle Assessment: Provisions and Action Steps, Publications Office, Luxembourg, 2011.
- 52 Y. Y. Haimes, L. S. Lasdon and D. A. Wismer, *IEEE Trans. Syst. Man. Cybern.*, 1971, **SMC-1**(3), 296.
- 53 G. Mavrotas, Appl. Math. Comput., 2009, 213(2), 455.
- 54 D. Vázquez, M. J. Fernández-Torres, R. Ruiz-Femenia, L. Jiménez and J. A. Caballero, *Comput. Chem. Eng.*, 2018, 108, 382.
- 55 S. Postels, N. von der Assen, P. Voll and A. Bardow, in 28th International Conference on Efficiency, Cost, Optimization, Simulation and Environmental Impact of Energy Systems (ECOS 2015): Pau, France, 30 June-3 July 2015, Curran Associates Inc., Red Hook, NY, 2015, p. 787.
- 56 R. T. Marler and J. S. Arora, *Struct. Multidiscip. Optim.*, 2004, **26**(6), 369.
- 57 L. J. Müller, A. Kätelhön, M. Bachmann, A. Zimmermann, A. Sternberg and A. Bardow, *Front. Energy Res.*, 2020, **8**, 1.
- 58 N. von der Assen, L. J. Müller, A. Steingrube, P. Voll and A. Bardow, *Environ. Sci. Technol.*, 2016, 50(3), 1093.
- 59 S. Deutz and A. Bardow, Nat. Energy, 2021, 6, 203.
- 60 A. David, B. V. Mathiesen, H. Averfalk, S. Werner and H. Lund, *Energies*, 2017, **10**(4), 578.
- 61 M. Reuß, T. Grube, M. Robinius, P. Preuster, P. Wasserscheid and D. Stolten, *Appl. Energy*, 2017, **200**, 290.
- 62 D. R. Ladner, in Cryocoolers 16: Proceedings of the 16th International Cryocooler Conference, ICC Press, 2011.
- 63 K. Aikawa, T. Sakurai and J. J. Jetter, *SAE Int. J. Fuels Lubr.*, 2010, **3**(2), 610.
- 64 C. K. Westbrook, W. J. Pitz and H. J. Curran, *J. Phys. Chem. A*, 2006, **110**(21), 6912.
- 65 D. D. Das, P. C. St. John, C. S. McEnally, S. Kim and L. D. Pfefferle, *Combust. Flame*, 2018, **190**, 349.
- 66 M. M. El-Halwagi, in *Sustainable Design through Process Integration*, Elsevier, 2012, p. 15.
- 67 T. Grube and B. Höhlein, in *Wasserstoff und Brennstoffzelle: Technologien und Marktperspektiven*, ed. J. Lehmann and J. Töpler, Springer Vieweg, Berlin, Heidelberg, 2014, p. 225.

- 68 M. Ruth, Hydrogen Production Cost Estimate Using Biomass Gasification: Independent Review: Technical Report DE-AC36-08GO28308, National Renewable Energy Laboratory (NREL), 2011.
- 69 M. A. Quader, S. Ahmed, R. A. Raja Ghazilla, S. Ahmed and M. Dahari, *J. Cleaner Prod.*, 2016, **120**(3), 207.
- 70 F. Ueckerdt, L. Hirth, G. Luderer and O. Edenhofer, *Energy*, 2013, **63**(1), 61.
- 71 F. Sabatino, A. Grimm, F. Gallucci, M. van Sint Annaland, G. J. Kramer and M. Gazzani, *Joule*, 2021, 5(8), 2047.
- 72 N. McQueen, K. V. Gomes, C. McCormick, K. Blumanthal, M. Pisciotta and J. Wilcox, *Prog. Energy*, 2021, **3**(3), 32001.
- 73 W. L. Luyben, Comput. Chem. Eng., 2017, 103, 144.
- 74 C. Reinert, S. Deutz, H. Minten, L. Dörpinghaus, S. von Pfingsten, N. Baumgärtner and A. Bardow, *Comput. Chem. Eng.*, 2021, 153(2), 107406.
- 75 S. Sala, E. Crenna, M. Secchi and E. Sanyé-Mengual, J. Environ. Manage., 2020, 269, 110686.
- 76 M. W. Ryberg, M. Owsianiak, J. Clavreul, C. Mueller, S. Sim, H. King and M. Z. Hauschild, *Sci. Total Environ.*, 2018, 634, 1406.