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Design and performance evaluation of all-inorganic AgTaS₃ perovskite solar cells

Tanvir Ahmed,^a Md. Choyon Islam,^a Md. Alamin Hossain Pappu,^a Md. Islahur Rahman Ebon, 🕩 a Sheikh Noman Shiddigue, a Mainul Hossain 🕩 and Jaker Hossain 匝 *a

Narrow bandgap AgTaS₃ perovskite can offer highly efficient thin film solar cells (SCs) and become Si counterparts that are leading in the market. Herein, we study the response of a n-CdS/p-AgTaS₃/p⁺-

Al_{0.8}Ga_{0.2}Sb device according to the variation of thickness, doping concentration, and defect densities in each layer using a solar cell capacitance simulator (SCAPS-1D). The optimized cell shows a $V_{\rm OC}$ of

0.78 V, PCE of 27.89% accompanied by a J_{SC} of 46.37 mA cm⁻², and a fill factor of 77.06%, paving the way for novel double heterojunction perovskite photovoltaic (PV) cells with remarkable performance.

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1. Introduction

The photovoltaic market is led by Si-based PV devices owing to their high availability and excellent performance in converting light to electricity.^{1,2} These PV cells have multiple drawbacks including high cost, weather dependence, space requirements, pollution concerns, and rigidity in the case of solar energy space.³ Contemporary studies highlight that perovskite solar cells (PSCs) have attracted much attention reaching PCEs as high as 25.6%.⁴ The substantial growth in research on perovskite solar cells (PSCs) can be attributed to their cost-effective solution processing method and appealing optoelectronic properties. These properties encompass a tunable bandgap, high absorption coefficient, minimal recombination rate and high mobility of charge carriers. Perovskites have earned extensive recognition in photovoltaics owing to their exceptional characteristics in light absorption, separation of excitons and facilitation of charge transport. Top of Form^{3,4} However, organic perovskite solar cell constancy is still a great concern. Humidity, heat, UV light, and constant exposure to sunlight degrade organic PSC performance.^{5,6} The study of degradation mechanisms in organic-inorganic hybrid perovskites has attracted considerable attention due to their potential in photovoltaic applications and the critical impact of stability on their commercial viability. These investigations are crucial for understanding factors that lead to performance degradation over time, such as moisture, oxygen exposure, thermal instability, light-induced effects, and ion migration. Transitioning to inorganic or

chalcogenide perovskites presents a significant advancement in overcoming the stability challenges faced by organic perovskites. One prominent aspect involves the MA⁺ acidic cation in MAPbI₃. Consequently, under conditions involving sunlight and oxygen, the deterioration of perovskites begins with the reaction between superoxide and the methylammonium group of atoms. Due to these challenges with organic perovskites, the inorganic or chalcogenides perovskites are introduced. This type of perovskites offers an excellent alternative to organic perovskites. Using inorganic perovskites, which lack organic elements, can boost the durability of perovskite crystals against exposure to oxygen and light. However, it is noteworthy that despite this improvement, inorganic perovskites have been found to lack sufficient stability when subjected to external factors of the environment. The instability in the phase of organic perovskite primarily arises from transitions occurring in the middle of preferred photoactive materials such as perovskites that have the black phase and the less desirable yellow phase, which is a non-perovskite material. In spite of the fact that the transition occurred in the yellow phase is reversible, it is accredited to the unfavorable factor relating to tolerance, contributing to the overall instability of the material.7

AgTaS₃ is a direct and narrow bandgap (0.97 eV) chalcogenide perovskite belonging to the I-V-VI₃ group that is suitable for both single and multi-junction solar cells.⁸ The high carrier mobility, high absorption coefficient, and stable crystal structure make AgTaS₃ an excellent candidate for highly efficient solar cells.^{9,10} Moreover, AgTaS₃ is environmentally friendly, unlike lead-based perovskite materials.

Here, the numerical investigation on the functionality of an n-CdS/p-AgTaS₃/p⁺-Al_{0.8}Ga_{0.2}Sb double heterostructure PV device with AgTaS₃ as the main absorber layer has been performed.

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^a Solar Energy Laboratory, Department of Electrical and Electronic Engineering, University of Rajshahi, Rajshahi 6205, Bangladesh. E-mail: jak_apee@ru.ac.bd

^b Department of Electrical and Electronic Engineering, University of Dhaka, Dhaka 1000, Bangladesh

All simulations have been conducted using SCAPS-1D under standard illumination, corresponding to 100 mW cm⁻² sun power at room temperature.

2. AgTaS₃ perovskite, cell build-up and computational framework

AgTaS₃ represents a unique category of sulfur-based single perovskites. Within the broader family of perovskites, sulfur variants are noted for their ability to reduce trap-related losses, increase grain size, and improve phase stability in perovskite photovoltaic (PV) cells.^{8,11-13} Moreover, chalcogenide perovskites, to which sulfur perovskites belong, are renowned for their exceptional light absorption, high refractive index, adjustable bandgap, pronounced optical nonlinearity, and advantageous defect chemistry. These properties are significantly influenced by the material's morphology, dimensions, and surface traits, contributing to their distinctiveness in the field of photovoltaics. Single crystal perovskites, known for their superior thermal stability relative to polycrystalline thin films, are seen as prime materials for creating stable and highly efficient SCs. This advantage enables single PSCs to serve as a good model for in-depth studies of surface phenomena and grain boundaries within perovskite materials.¹⁴⁻¹⁸ They significantly contribute to advancing technological development and enhancing the commercial prospects of perovskite solar cells. These characteristics further facilitate enhanced carrier mobility and extraction, leading to improved device performance.

Since the discovery of the inorganic perovskite mineral CaTiO₃, perovskite structures have been typified by such a configuration. The ideal single perovskite formula is ABX₃, wherein the larger 'A' cation contrasts with the smaller 'B' cation, while the X anion is positioned at the corners. In the depicted sulfide-based perovskite, AgTaS₃, a three-dimensional lattice is formed where the silver ion (Ag¹⁺) is linked with the cation tantalum (Ta⁵⁺), and the structure is stabilized by cornersharing sulfur anions (S²⁻) according to Fig. 1(a). This sulfur substitution for the typically found oxide or halide anion (X) in perovskites enhances thermal stability due to the chalcogenide nature of sulfur.^{14,15,18}

Fig. 1(a) delineates an orthorhombic crystallization of AgTaS₃, belonging to the *Cmc2*₁ space group featuring lattice constants 3.3755 Å, 14.0608 Å, and 7.7486 Å for *a*, *b*, and *c*, consequently with a unit cell volume, V = 367.77 Å³. The angles alpha, beta, and gamma are each positioned at 90°, indicating an orthogonal phase relationship.¹⁹ For the formation of a perovskite structure, charge neutrality must be maintained among the constituent ions. This implies that n(A) + n(B) = 3n(X), where *n* denotes the valence of the ions A, B, and X.¹⁷ This equation ensures that the overall electrical charge within the perovskite structure is balanced, a fundamental criterion for the stability and formation of these materials. Here, $n(Ag) + n(Ta) = 1 + 5 = 6 = 3 \times 2 = 3n(S)$ shows the charge neutrality of the AgTaS₃ structure.

Fig. 1(b) exhibits the conceptual diagram of the proposed device. CdS is highly stable and is used as the window layer because of its adjustable, wider, direct bandgap, and superior carrier mobility with excellent light transmission properties.²⁰⁻²² The back surface field (BSF) layer reduces carrier recombination and is composed of Al_{0.8}Ga_{0.2}Sb. Recombination of electrons and holes can reduce the $V_{\rm OC}$ and degrade photo conversion performance.²³ AgTaS₃ and a highly doped BSF layer create a $p-p^+$ interface and prevent the electrons at the surface from recombining with holes by developing an electric field, thereby increasing the V_{OC} and PCE.²⁴ Previous studies on Si solar cells, with Al_{0.8}Ga_{0.2}Sb BSF layer, have achieved a PCE as high as 38.3%.25 Highly doped p+-Al_{0.8}Ga_{0.2}Sb BSF layer generates an interface with the absorber layer, $AgTaS_3$ as p/p^+ . $AgTaS_3$ possesses an ionization potential of 4.97 eV with an affinity (E_A) of 4 eV. CdS, on the other hand, has an E_A of 4.2 eV with an ionization energy of 6.6 eV.

Fig. 1(c) exhibits the illumined band structure of the proposed structure. The BSF layer aligns with the absorber layer as a result of E_A , 3.69 eV, and an ionization potential of 5.09 eV of Al_{0.8}Ga_{0.2}Sb. Here, the quasi-Fermi level (QFL) for n-type material, $E_{\rm Fn}$, is just underneath the conduction band, $E_{\rm C}$, while the QFL for p-type, $E_{\rm Fp}$, is just above the valence band, $E_{\rm V}$. Sunlight generates electrons at the conduction band of the p–n and p–p⁺ junction. An aluminium (Al) metal grid acts as a cathode to accumulate electrons. Electrons, moving in the opposite direction, are blocked by the BSF layer and accumulated by Al. Holes are generated in the valence band and acquired by the back contact molybdenum, Mo.

SCAPS-1D employs the Poisson equation, connecting charge to electron and hole continuity equations with the result of the PSC attribute. The Poisson equation, Hole continuity equation, and electron continuity are defined by eqn (1), (2), and (3), respectively:²⁰

$$\frac{\mathrm{d}^2\psi}{\mathrm{d}x^2} = \frac{q}{\varepsilon_0\varepsilon_\mathrm{r}}(p(x) - n(x) + N_\mathrm{D} - N_\mathrm{A} + \rho_\mathrm{P} - \rho_\mathrm{N}) \tag{1}$$

$$\frac{1}{q}\frac{\partial J_p}{\partial_x} = G_{\rm op} - R(x) \tag{2}$$

$$\frac{1}{q}\frac{\partial J_n}{\partial_x} = -G_{\rm op} + R(x) \tag{3}$$

here, x = distance; $\psi =$ electrostatic potential; $\varepsilon_r =$ relative permittivity; $\varepsilon_0 =$ free space permittivity; p defines volume concentration of holes; q means electron charge and n defines volume concentration of electrons, subsequently, while $N_{\rm D}$ and $N_{\rm A}$ are used to describe the donor and acceptor charges; and the symbols $\rho_{\rm P}$ and $\rho_{\rm N}$ represent the charge densities of electrons and holes, correspondingly. The drift-diffusion equation using electron and hole mobility helps to find out the transmission attribute of carriers:²⁰

$$J_p = \frac{\mu_p p}{q} \frac{\partial E_{\mathbf{F}_p}}{\partial_x} \tag{4}$$

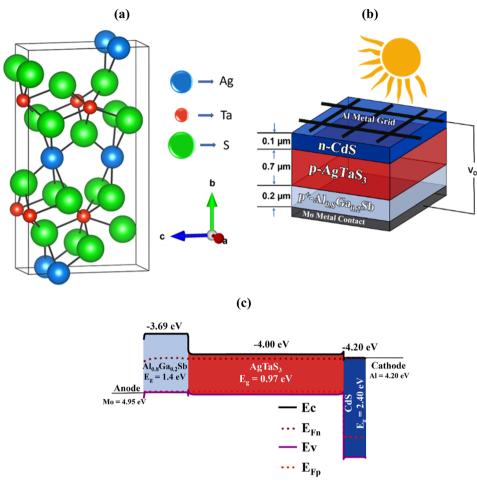


Fig. 1 (a) Crystal structure of AgTaS₃ perovskite, (b) conceptual diagram, and (c) energy band structure of the AgTaS₃-based perovskite SC.

$$J_n = \frac{\mu_n n}{q} \frac{\partial E_{\mathrm{F}_n}}{\partial_x} \tag{5}$$

where, symbols μ_p = hole mobility; μ_n = electron mobility.

The effective density of states (DOS) at the conduction band $N_{\rm c}$ and effective DOS at the valence band $N_{\rm v}$ can be determined from the following equations:²⁰

$$N_{\rm c} = 2 \left(\frac{m_{\rm e}^* KT}{2\pi\hbar^2}\right)^{\frac{3}{2}} \tag{6}$$

$$N_{\rm v} = 2 \left(\frac{m_{\rm h}^* KT}{2\pi\hbar^2}\right)^{\frac{3}{2}} \tag{7}$$

where m_e^* and m_h^* present the electron and hole effective mass for AgTaS₃, which have the values of 0.61 and 0.27, which in turn were used to calculate the N_c and N_v of the compound.⁸

In the simulation, a Gaussian energy distribution is applied to model the defect states in each layer. The defect characterization in the BSF layer is considered neutral, indicating no net charge contribution to the material's conductivity or recombination processes. Conversely, for the window and the absorber layers, defects are modeled as single acceptor types and single donor types, in turn implying that these defects can accept or donate an electron. This distinction in defect modeling is critical for accurately simulating the electronic properties of each layer and their impression on the device's overall efficiency. However, the SCAPS-1D simulator has proved itself to be a reliable one in the era of solar cells that can provides results similar to that obtained from the experimental study as documented in the literature.^{26–28} Table 1 presents the input parameters used for the numerical analysis.

3. Results and discussion

3.1 Varying the properties of the absorber layer

Fig. 2(a) shows the output characteristics of the proposed device for varying absorbing layer width, concentration of doping and defects for the proposed perovskite-based solar cell. As the width of AgTaS₃ ranges from 0.3 μ m to 1.1 μ m, V_{OC} remains constant at around 0.78 V. However, the insignificant fall in V_{OC} may be the result of a rising dark current with width that increases the possibility of recombination.²⁰ Contrary, the J_{SC} has increased because, in the longer wavelength regime, thicker AgTaS₃ absorbs more light and generates a large

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 Table 1
 Characteristic delineating features of each device layer of AgTaS₃ PSC

Parameters	n-CdS ^{20,21}	p-AgTaS ₃ ^{8,19}	p ⁺ -Al _{0.8} Ga _{0.2} Sb ^{29,3} 0.2	
Depth (µm)	0.1	0.7		
Energy gap (eV)	2.4	0.97	1.4	
Affinity of electron (eV)	4.2	4.0	3.69	
Dielectric permittivity	10	13.64	12.04	
Effective density of states at CB (cm^{-3})	$2.2 imes10^{18}$	$1.2 imes10^{19}$	$7.8 imes10^{17}$	
Effective density of states at VB (cm^{-3})	$1.8 imes10^{19}$	$3.52 imes10^{18}$	$1.8 imes 10^{19}$	
Mobility of electron (cm ² V ^{-1} s ^{-1})	100	12	200	
Mobility of hole (cm ² V ⁻¹ s ⁻¹)	25	1.2	420	
Donor concentration, $N_{\rm D}$ (cm ⁻³)	1.00×10^{18}	0	0	
Acceptor concentration, N_A (cm ⁻³)	0	1.00×10^{16}	$1.00 imes10^{19}$	
Defect concentration (cm^{-3})	1.00×10^{14}	$1.00 imes 10^{14}$	$1.00 imes 10^{14}$	

number of carriers.³¹ Fill factor (FF) drops sharply from 78.11% to 76.01% because of the shorter carrier lifetime relative to the layer size.^{32,33} PCE increases from 25.64% to 27.89% up to 0.7 μ m of AgTaS₃, beyond which recombination of the carriers prevents further increase in PCE.

Fig. 2(b) illustrates the impact of carrier densities in AgTaS₃ on the perovskite solar cell output characteristics. $V_{\rm OC}$ shows negligible change when the doping varies from 10^{14} cm⁻³ to 10^{18} cm⁻³. This is because, at lower doping levels, the recombination current decreases, thereby increasing $V_{\rm OC}$, whereas at higher doping levels, an increment of recombination leads to a reduction in $V_{\rm OC}$. Then, again, beyond 10^{18} cm⁻³, the $V_{\rm OC}$ enhances as recombination decreases. This may happen due to the decrease in the ideality factor, *n*, with the acceptor concentration.^{20,34} The fill factor grows proportionally with $V_{\rm OC}$, and it goes from 78.17% to 85.67%. A modest increase in $V_{\rm OC}$ indicates better FF, as the relation between those is defined by an empirical formula with the help of the diode parameter.²⁰ FF is higher when doping is 10^{20} cm⁻³, with the PCE reaching its maximum value at 33.53%.

Fig. 2(c) illustrates the influences of defects on the execution of the PSC. The defect improves the dark current with an

increase in Shockley read hall (SRH) recombination and diminishes the performance of the gadget.³⁵ For defects in the range of 10^{12} cm⁻³ to 10^{16} cm⁻³, V_{OC} changes from 0.81 V to 0.52 V. This significant change affects the FF and PCE and the J_{SC} changes marginally. PCE decreases from 29.91% to 16.62% on account of V_{OC} and FF.

Quantum efficiency (QE) defines the photon absorption efficiency of a perovskite-based SC. No change in QE occurs when acceptor density is varied, as displayed in Fig. 3(a). Enhancing the thickness increases QE proportionally. At 0.7 μ m thickness, this structure can absorb around 95% of the incident photons, as seen in the figure. The increment in current density with absorber width is illustrated in Fig. 3(b). The thicker layer can absorb more photons, which increases the photocurrent from 41.6 mA cm⁻² current at 0.3 μ m to 46.37 mA cm⁻² current at 0.7 μ m of the thickness. Beyond 0.7 μ m, the photocurrent of SC saturates.

3.2 Varying the properties of the window layer

Fig. 4 displays the change in output characteristics of the suggested PSC as a result of alteration in the width, carrier

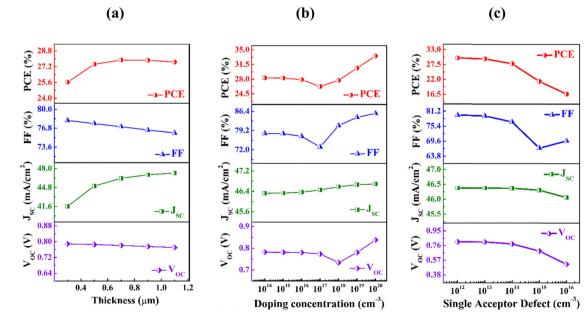


Fig. 2 The effects of changing the absorber's (a) breadth, (b) carrier, and (c) defects on the offered cell.

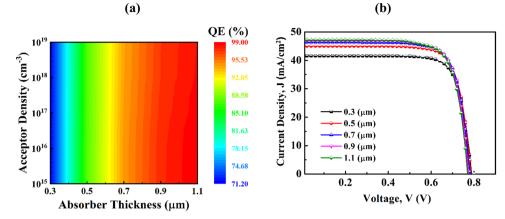


Fig. 3 (a) Variation of QE with acceptor density vs. thickness of absorber and (b) current density vs. voltage characteristics with thickness of the absorber.

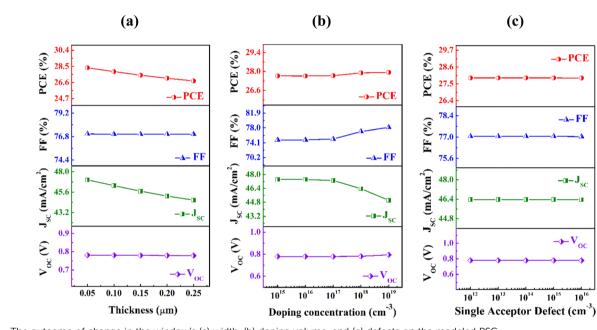


Fig. 4 The outcome of change in the window's (a) width, (b) doping volume, and (c) defects on the modeled PSC.

concentration and defect volume of the CdS window. Open circuit voltage shows continual behavior at the level of 0.78 V for the thickness of 0.05 μ m to 0.25 μ m. The FF shows a saturation value of 77%. The $J_{\rm SC}$ drops to 44.65 mA cm⁻² from 47.05 mA cm⁻² because of an increase in recombination with the breadth of the window.³⁶ The expansion of the breadth in the SC design results in a greater number of photons being absorbed outside the intended space charge region. Absorption mitigates the electron–hole pair generation and intensifies the flow of minority charge carriers, thereby increasing the rate of charge recombination, which negatively impacts the operational efficiency of the solar cell.²⁰ Owing to the increase in recombination, PCE drops to 26.8% at 0.25 μ m thickness, whereas it is 28.35% when the thickness is 0.05 μ m.

Fig. 4(b) expresses that the photocurrent decreases when the dopant concentration is above 10^{17} cm⁻³. The current density

remains at 47 mA cm⁻² for carriers altering from 10¹⁵ to 10¹⁷ cm⁻³. At 10¹⁹ cm⁻³ concentration, it falls to 45 mA cm⁻² from 47 mA cm⁻². A higher order of doping may lead to parasitic free carrier absorption, which intensifies the recombination level and causes $J_{\rm SC}$ to decrease by 2 mA cm⁻².^{37,38} The $V_{\rm OC}$ does not show any significant change. The fill factor shows a minor change from 74.87% to 78.00% as per the equation below:³⁹

$$FF = \frac{P_{\rm MP}}{V_{\rm OC}J_{\rm SC}} \tag{8}$$

where $P_{\rm MP}$ states the topmost power of the solar cell, $V_{\rm OC}$ is open circuit voltage and $J_{\rm SC}$ means short circuit current. PCE has no notable change and remains fairly constant; hence, FF escalates.

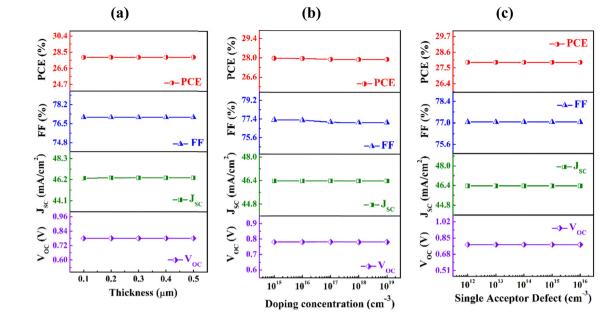


Fig. 5 The alteration of cell operation with (a) width, (b) carrier volume and (c) defects of BSF of the $n-CdS/p-AgTaS_3/p^+-Al_{0.8}Ga_{0.2}Sb$ device.

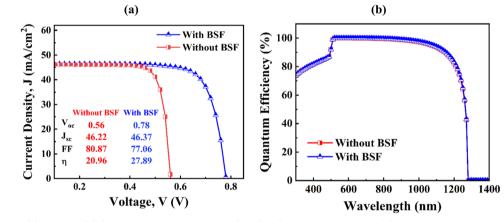


Fig. 6 Presentation of (a) J-V, and (b) QE with and without the Al_{0.8}Ga_{0.2}Sb BSF layer of the urged PSC.

Fig. 4(c) represents that the defects in the CdS window have no influence on the characteristics of this projected SC. Even as the volume of defects varies within the range of 10^{12} cm⁻³ to 10^{16} cm⁻³, the shrink in carrier diffusion length as a sequel of diminishing the carrier lifetime is not significant enough with respect to the window breadth.⁴⁰ Consequently, the performance of the cell is largely consistent.

3.3 Varying the properties of the BSF layer

Fig. 5 demonstrates the consequences of varying the properties of $Al_{0.8}Ga_{0.2}Sb$ BSF on the output characteristics of $AgTaS_3$ -based PSC. Fig. 5(a) examines changes in the BSF layer's thickness from 0.1 µm to 0.5 µm, observing its effect on the solar cell's performance. Notably, the V_{OC} remains largely unaffected by this variation, while the J_{SC} shows a marginal increase from 46.36 to 46.38 mA cm⁻². This improvement in J_{SC}

may be attributed to a stronger electric field at the interface, which effectively reduces the surface recombination velocity. Therefore, the insertion of the BSF layer elevates the cell's efficiency from 20.96% to 27.9%.^{20,41}

Fig. 5(b) explores the impact of doping concentrations ranging from 10^{15} cm⁻³ to 10^{18} cm⁻³. Here, the device current remains almost unchanged. Nonetheless, there is a slight

Table 2 The performance factors of the AgTaS_3-based PV cell without and with the Al_0.8Ga_0.2Sb layer

	Device structure			
PV parameter	CdS/AgTaS ₃	CdS/AgTaS ₃ /Al _{0.8} Ga _{0.2} Sb		
$V_{\rm OC}$ (V)	0.56	0.78		
$J_{\rm SC}$ (mÅ cm ⁻²)	46.22	46.37		
FF (%)	80.87	77.06		
PCE (%)	20.96	27.89		

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Table 3 Comparative analysis of perovskite-based solar cells

Solar cells	Туре	$V_{\rm OC}$ (V)	$J_{ m SC}~({ m mA~cm^{-2}})$	FF (%)	PCE (%)	Ref.
ZnO/CdS/CH ₃ NH ₃ SnI ₃ /GaAs	Numerical	0.96	33.86	73.2	23.8	21
bl-Y:TiO ₂ /MAPbI _{3-x} Cl _x (solution)/spiro-MeOTAD	Exp.	1.13	22.75	75	19.3	45
FTO/c-TiO ₂ /mTiO ₂ /PbI ₂ :CH ₃ NH ₃ I/spiro-OMeTAD/Au	Exp.	1.09	22.4	80	19.1	46
FTO/SnO ₂ /PCBM/perovskite/spiro-OMeTAD/Ag	Exp.	1.19	19.4	79	17.9	47
mp-TiO ₂ /FAPbI ₃ /MAPbI ₃ (ETL)/spiro-MeOTAD	Exp.	1.03	20.97	74	16.01	48
TO/SnO ₂ /MBM/perovskite/spiro-OMeTAD/Ag	Exp.	1.17	24.16	81.86	24.16	49
TO/AMD PTAA/FA _{0.9} Cs _{0.1} PbI ₃ /PCBM/BCP	Exp.	1.148	24.97	83.84	23.89	50
TO/P3CT-N/CsPbI ₃ /PCBM/C60/BCP/Ag	Exp.	1.238	20.52	80.89	20.17	51
FTO/c-TiO ₂ /ChPbI ₃ -CsPbI ₃ /Carbon	Exp.	1.11	20.13	80.79	18.05	52
n-CdS/p-AgTaS ₃ /p ⁺ -Al _{0.8} Ga _{0.2} Sb	Numerical	0.78	46.37	77.06	27.89	This we

decrease in the FF value to 0.27%, attributed to a lower ratio of generation to recombination.⁴² This leads to a decrease in the overall PCE from 27.97% to 27.89%.

In Fig. 5(c), the analysis shows that variations in defect density, spanning from 10^{12} cm⁻³ to 10^{16} cm⁻³, do not significantly affect the performance of the cell. The measurements across this span of bulk defects indicate that they maintain a constant value, suggesting that the defect density in this interval does not substantially influence cell operation. This is a frequent observation in recent studies.^{20,37,40,42} As carrier diffusion length exceeds the breadth of the BSF layer for these defects, no deterioration is observed in the study.^{40,42}

3.4 Ultimate outcome of AgTaS₃ PSC

The exhibition of the contribution of the $Al_{0.8}Ga_{0.2}Sb$ BSF layer is showcased in Fig. 6. The current *vs.* voltage (*J*-*V*) curve is shown in Fig. 6(a). Without the $Al_{0.8}Ga_{0.2}Sb$ layer, the voltage is 0.56 V, photocurrent is 46.22 mA cm⁻² and the efficiency is 20.96%. But, with the inclusion of BSF in this PSC structure, the voltage has an incremental value of 0.78 V. The built-in potential created in the p-AgTaS₃/p⁺-Al_{0.8}Ga_{0.2}Sb junction helps to increase the voltage.⁴¹ The current of the PSC remains fairly constant. The PCE of the device increased to 27.89%. However, the FF slightly diminishes from 80.87% to 77.06%, which may happen as a consequence of the change in diode properties.²⁰

Fig. 6(b) reveals the absorption power at different wavelengths of light with and without the BSF layer. From the figure, it is visualized that QE is lower in the wavelength range of 400–500 nm. CdS absorbs a large number of shorter wavelength photons. This absorption results in window gain, which leads to lower QE.^{43,44} In near 600 nm wavelength, QE increases beyond 95% as the absorber layer absorbs photons of all wavelengths. The BSF layer helps marginally increase quantum efficiency.

Table 2 shows the performance parameters of AgTaS₃ PV cells with and without the Al_{0.8}Ga_{0.2}Sb layer. The PCE ascends from 20.96% to 27.89% when the BSF layer is used. The single-junction n-CdS/p-AgTaS₃ model has a $V_{\rm OC}$ of 46.22 V and a $J_{\rm SC}$ of 46.22 mA cm⁻² with 80.87% FF. The double heterojunction n-CdS/p-AgTaS₃/p⁺-Al_{0.8}Ga_{0.2}Sb has 0.78 V of $V_{\rm OC}$, 46.37 mA cm⁻² of $J_{\rm SC}$ and 77.06% FF. Elevation of the PCE for this double heterojunction solar cell conforms to the detailed balance limit unveiled by Shockley–Queisser (SQ).

There are many previous reports of various perovskitesbased solar cells. Table 3 shows the relative studies on different perovskite solar cells. Organic perovskites have shown impressive performance, but the instability of this type of perovskite makes it a very disadvantageous material. Also, the use of lead halides in SC industries leads to toxicity in the environment. The efficiency and short circuit current density of some previous literature works are much lower than those in this work. Also, the thermal stability of AgTaS₃ is superior to that of organic perovskites. These disadvantageous issues make it a preferable material in this era.

4. Conclusion

AgTaS₃-based n-CdS/p-AgTaS₃/p⁺-Al_{0.8}Ga_{0.2}Sb PSC system has been successfully designed and optimized with the help of the SCAPS-1D simulator. Window layer CdS frames a p-n junction with the AgTaS₃ absorber layer and this layer produces a $p-p^+$ junction with the Al_{0.8}Ga_{0.2}Sb BSF layer. The depth of the different PSC panels is optimized to 0.7 µm for the absorber layer, 0.1 µm for the window layer, and 0.2 µm for the BSF layer. Doping concentration has been taken as 1.0×10^{18} , 1.0×10^{16} and 1.0×10^{19} cm⁻³ for the respective layer. Defect density for Cds, AgTaS₃, and Al_{0.8}Ga_{0.2}Sb has been optimized at the value of 1.0×10^{14} cm⁻³. After optimizing the physical parameters of each layer, the optimal PCE has been up to 27.89%. J_{SC} , V_{OC} , and FF increased to 46.37 mA cm⁻², 0.78 V and 77.06%, respectively. V_{OC} has shown an increment from 0.56 V to 0.78 V after adding a thin BSF layer. J_{SC} rises to a small extent after adding the BSF layer from 46.22 mA cm^{-2} to 46.37 mA cm⁻² as a consequence of the reduction in carrier recombination. The p-AgTaS₃/p⁺-Al_{0.8}Ga_{0.2}Sb junction produces a high potential barrier and raises the V_{OC} as well as the PCE. This result paves the way for a highly efficient AgTaS₃ inorganic perovskite solar cell.

Author contributions

Tanvir Ahmed: data curation; formal analysis; investigation; validation; writing – original draft (equal). Md. Choyon Islam: formal analysis; validation; writing – original draft. Md. Alamin Hossain Pappu: data curation; formal analysis; validation; writing – original draft. Md. Islahur Rahman Ebon: formal

analysis; investigation; validation; writing – original draft. Sheikh Noman Shiddique: formal analysis; investigation; validation; writing – original draft. Mainul Hossain: formal analysis; investigation; validation; writing – original draft. Jaker Hossain: conceptualization; formal analysis; methodology; supervision; validation; writing – original draft; writing – review and editing.

Data availability

Data will be available from the corresponding author upon reasonable request.

Conflicts of interest

There are no conflicts to declare.

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