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## Reassessing the role and lifetime of $Q_x$ in the energy transfer dynamics of chlorophyll $a$ †

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Chlorophylls are photoactive molecular building blocks essential to most photosynthetic systems. They have comparatively simple optical spectra defined by states with near-orthogonal transition dipole moments, referred to as  $B_x$  and  $B_y$  in the blue/green spectral region, and  $Q_x$  and  $Q_y$  in the red. Underlying these spectra is a surprisingly complex electronic structure, where strong electronic-vibrational interactions are crucial to the description of state characters. Following photoexcitation, energy-relaxation between these states is extremely fast and connected to only modest changes in spectral shapes. This has pushed conventional theoretical and experimental methods to their limits and left the energy transfer pathway under debate. In this work, we address the electronic structure and photodynamics of chlorophyll  $a$  using polarization-controlled static – and ultrafast – optical spectroscopies. We support the experimental data analysis with quantum dynamical simulations and effective heat dissipation models. We find clear evidence for  $B \rightarrow Q$  transfer on a timescale of  $\sim 100$  fs and identify  $Q_x$  signatures within fluorescence excitation and transient spectra. However,  $Q_x$  is populated only fleetingly, with a lifetime well below our  $\sim 30$  fs experimental time resolution. Outside of these timescales, the kinetics are determined by vibrational relaxation and cooling. Despite its ultrashort lifetime, our theoretical analysis suggests that  $Q_x$  plays a crucial role as a bridging state in  $B \rightarrow Q$  energy transfer. In summary, our findings present a unified and consistent picture of chlorophyll relaxation dynamics based on ultrafast and polarization-resolved spectroscopic techniques supported by extensive theoretical models; they clarify the role of  $Q_x$  in the energy deactivation network of chlorophyll  $a$ .

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## Introduction

Photosynthetic systems efficiently harness sunlight by absorbing it and transferring the excitation energy to a reaction center where charge separation occurs.<sup>1,2</sup> Globally, the vast majority of light-harvesting and charge-separation functionality relies on chlorophylls (Chls) or bacteriochlorophylls (BChls). They play essential roles in energy transfer and charge separation and serve as the principal cofactors in the early steps of photosynthesis. For this reason, much effort has been devoted to understanding their electronic structure and ultrafast

relaxation dynamics using various experimental and theoretical methods.<sup>3–8</sup>

Despite their structural diversity, all Chls share two prominent spectral bands: the lowest energy Q-band (550–720 nm) and the Soret- or B-band (350–480 nm). Historically, these bands have been characterized using the Gouterman four-orbital model of the  $\pi$ - $\pi^*$  transitions in porphyrins.<sup>9,10</sup> In this model, the Q-band consists of two distinct electronic states, which we name  $Q_{x,el}$  and  $Q_{y,el}$ . Their indices relate to the direction of the respective transition dipole moment within the plane of the macrocycle (Fig. 1a). Later, increasingly sophisticated quantum chemical calculations and experiments aimed to refine the assignment of the observed absorption bands to the two electronic states. However, the energetic position of the  $Q_{x,el}$  state remained ambiguous. In 2013, Reimers *et al.*<sup>3</sup> proposed a new band assignment based on vibronic coupling, mixing the states within the Q-band. We refer to the states after incorporating such coupling effects as  $Q_x$  and  $Q_y$ . This coupling strongly influences the spectral properties of Chl  $a$ . For example, the angle between the  $Q_x$  and  $Q_y$  transition dipole moments would naively be expected to be close to  $90^\circ$ , but the experimentally measured value for Chl  $a$  is  $\sim 70$ – $78^\circ$ .<sup>11,12</sup> In

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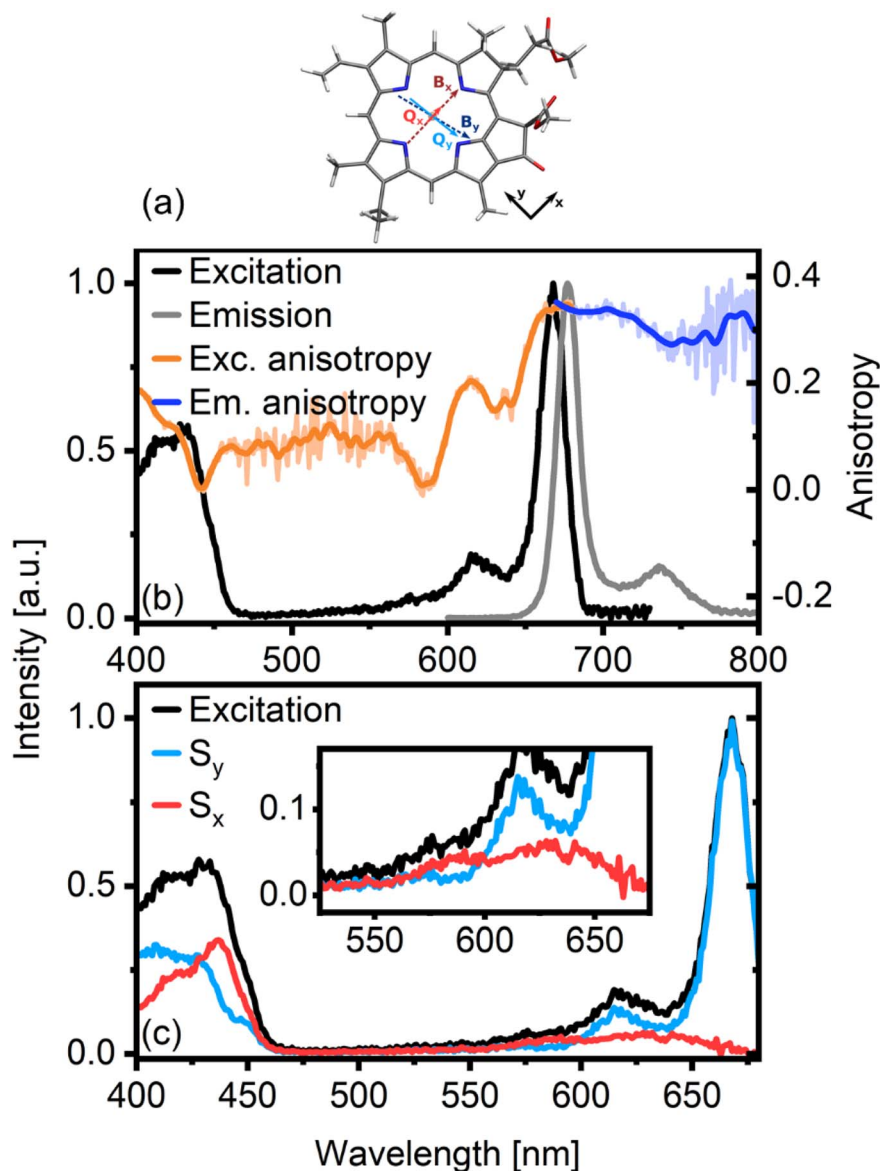
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**Fig. 1** Molecular structure of Chl *a* including the relative orientations of the main transition dipole moments along the porphyrin rings (a). There are two sets of optical transitions (B and Q), each comprising two almost perpendicular TDMs (subscripts  $x$  and  $y$ ). Panel (b) shows the excitation and emission anisotropy of Chl *a* in an isopropanol glass. The emission anisotropy ( $r_{em}$ , blue line) was measured for  $\lambda_{exc} = 660$  nm, while the excitation anisotropy ( $r_{exc}$ , orange line) was measured for  $\lambda_{em} = 690$  nm. The excitation anisotropy clearly shows the presence of differently oriented transitions. We therefore decompose it into  $x$ -polarized ( $S_x$ ) and  $y$ -polarized ( $S_y$ ) excitation spectra (c) for  $\beta = 17^\circ$ .

several subsequent two-dimensional electronic spectroscopy (2DES) studies, vibronic coupling was discussed as the basis of fast signal oscillations,<sup>13,14</sup> the vibronic origin of which was proven by polarization-controlled studies.<sup>7</sup> A similar conclusion was reached from theoretical studies, where it was shown that the  $Q_x$  and  $Q_y$  potential energy surfaces do not cross in an energetically accessible region, and non-adiabatic coupling facilitates ultrafast population transfer across the potential energy surfaces.<sup>5</sup> Previous to 2DES studies, results from transient absorption (TA) spectroscopy were interpreted so that  $Q_x \rightarrow Q_y$  transfer shows a strong and unusual solvent dependence, with  $Q_x$ -lifetimes ranging from 100–250 fs in different solvents.<sup>8,15</sup>

In this work, we reinterpret the internal conversion dynamics of Chl *a*, as determined by TA spectroscopy with  $\sim 20$  fs excitation and ultra-broadband probing. To offer a comprehensive picture of the relevant deactivation pathways in Chl *a*, we perform TA experiments exciting the B and the Q-band selectively and compare our results with theoretical models. We can unambiguously determine a B  $\rightarrow$  Q transfer time of  $\sim 100$  fs from our experimental data and corroborate our findings using theoretical models. We also show that coupling effects within the Q bands lead to almost instantaneous intra-band relaxation.

To test our hypothesis of near-instantaneous  $Q_x \rightarrow Q_y$  transfer, we go beyond global analysis of TA data and search for





$Q_x$ -signals using polarization-selective spectroscopy, followed by isolation of polarization-associated spectra.<sup>16</sup> Given reported  $Q_x$  lifetimes of up to several hundreds of femtoseconds,<sup>15</sup> we should observe  $x$ -polarized stimulated emission (SE) from  $Q_x$ , along with  $Q_x$  ground-state bleach (GSB) and excited-state absorption (ESA). While we successfully isolate the GSB signature of  $Q_x$ , we find no sign of  $x$ -polarized SE or ESA features to indicate a transiently populated  $Q_x$  state. We attribute this to vibronic mixing and rule out the conventional interpretation of  $Q_x$  as a state with up to 300 fs lifetime.<sup>15</sup> Instead, we offer an interpretation of the kinetic components found in TA based on intra-molecular vibrational redistribution and subsequent vibrational cooling, as described by an effective model.

The paper is structured as follows: to identify GSB signatures of  $Q_x$ , we first analyze polarization-associated spectra derived from cryogenic fluorescence excitation spectra (“Assignment of  $Q_x$  and  $Q_y$  transitions”). Direct excitation of  $Q_x$  is a non-ideal starting ground for studying dynamics, as the absorption features of  $Q_x$  and  $Q_y$  overlap strongly. Instead, we excite the B-band and try to isolate transient  $Q_x$  features. We compare these results with TA data after direct  $Q_x/Q_y$  excitation (“Ultrafast relaxation dynamics of Chl *a*”).

Similarly to the dissection of linear spectra, we attempt to isolate  $Q_x$ - and  $Q_y$ -related excited state absorption features in transient absorption anisotropy (TAA) (“Isolating  $Q_x$ -features by polarization control”) and compare our measured results to theoretical predictions (“System and system-bath relaxation dynamics at different timescales”). We show that neither TA experiments with sub 20 fs pump pulses nor polarization-controlled experiments can establish a timescale for  $Q_x \rightarrow Q_y$  transfer. Instead, our work offers a new perspective on the energy transfer dynamics in Chl *a*. We relate the experimentally observed ultrafast dynamics to solvent relaxation processes and obtain a robust and consistent picture supporting strong mixing between Q-states.

## Results

### Assignment of $Q_x$ and $Q_y$ transitions

Fig. 1b shows the normalized excitation (black line) and emission (grey line) spectra of Chl *a* in an isopropanol glass. The small Stokes shift of the fluorescence spectrum indicates a minor displacement of the excited state potential surface, as also discussed in earlier work.<sup>14,17</sup> Accordingly, the emission and absorption spectra are similar in shape, although not exactly symmetric. The differences are mainly in the sidebands, where the breakdown in mirror-image symmetry has been connected to vibronic coupling interactions.<sup>17,18</sup> The excitation spectrum in the Q band region has one dominant peak at *ca.* 670 nm and two shoulders at *ca.* 620 nm and 570 nm. Based on previous work, the main peak can be assigned to the  $Q_y$  transition.<sup>19</sup> However, assigning the shoulders – particularly determining the position of the  $Q_x$  transition – is not straightforward.<sup>6,17</sup> The locations of the  $Q_x$  and  $Q_y$  transitions in Chl *a* vary with the solvent and the coordination of the molecule. In isopropanol, where the system is penta-coordinated,<sup>20</sup> the  $Q_x$  band overlaps with  $Q_y$  vibrational bands,<sup>17</sup> making

a definite assignment of the shoulder bands challenging. Several groups have investigated the transition energy of  $Q_x$  and its vibronic sidebands, either by calculation<sup>4,5</sup> or experiment.<sup>3,17</sup> In this work, we employ polarized excitation and emission spectroscopy of Chl *a* in an isopropanol glass.

Alongside the excitation and emission spectra, Fig. 1b also shows fluorescence anisotropy traces. The emission anisotropy ( $r_{em}$ , blue line) was measured for  $\lambda_{exc} = 660$  nm, while the fluorescence excitation anisotropy ( $r_{exc}$ , orange line) was measured for  $\lambda_{Em} = 690$  nm. In the Q band region,  $r_{exc}$  of the main transition centered at 670 nm is almost flat. It reaches a value of 0.35, indicating near-parallel emission and absorption dipoles. As emission occurs from  $Q_y$ , this band must be  $y$ -polarized. The same high value is obtained from  $r_{em}$  upon excitation at 660 nm, proving the measurements to be consistent. While the main transition at 670 nm belongs to  $Q_y$ ,  $r_{exc}$  shows clearly that the high-energy side of the Q band region contains contributions from electronic states with different polarizations, as anisotropy values here are as low as zero.

Several approaches have been developed to decompose optical spectra by leveraging the information on transition dipole moment (TDM) directions available from anisotropies.<sup>21–23</sup> Most commonly, in the absence of a macroscopically oriented sample, these methods “project” the isotropic spectrum into contributions either parallel or orthogonal to a reference TDM. For excitation anisotropy measurements, this reference is the TDM of the emissive transition, while for emission anisotropy, the reference TDM is that of the pumped transition. In many cases, however, the TDMs of a given molecule are not exactly orthogonal, and the decomposition of the spectra in orthogonal components is not ideal. As earlier work has shown that the Chl *a*  $Q_x$  and  $Q_y$  TDMs are not orthogonal,<sup>11</sup> this type of spectral decomposition will not result in optimal separation of the contributions. In a recent study, we developed a slightly modified decomposition approach, which is suitable for systems where TDMs are not orthogonal.<sup>16</sup> In this approach, we can cleanly separate spectral contributions not only when the underlying TDMs are at an angle  $\theta = 90^\circ$ , but at any angle  $\theta = 90^\circ - \beta$ . The parameter angle  $\beta$  introduced here can be thought of as the rotation angle of an orthogonal molecular coordinate system defined by the reference TDM. For details, we refer to the corresponding publications.<sup>16,24</sup>

In Fig. 1c we show the decomposition of the Chl *a* excitation spectrum into components polarized parallel (here:  $S_y$ , blue in Fig. 1c) and orthogonal (here:  $S_x$ , red) to the emissive transition ( $Q_y$  fluorescence) according to the procedure outlined above. We note that besides  $Q_x$  features,  $S_x$  also contains signals with a TDM parallel to  $B_x$ , while  $S_y$  contains both  $Q_y$  and  $B_y$  contributions. If we assume orthogonal  $Q_x$  and  $Q_y$  transitions (corresponding to  $\beta = 0$ ), we clearly achieve sub-optimal separation of the spectrum with noticeable contamination of the  $S_x$  spectrum by features associated with the  $y$ -polarized transitions (see Fig. S1†). Increasing the value of  $\beta$  to  $17^\circ$ , however, maximizes the 670 nm peak in the  $S_y$  spectrum and simultaneously achieves optimal separation of the spectrum into “pure” but overlapping  $Q_x$  and  $Q_y$  contributions, each with their respective



vibronic progressions. Note that both the  $S_x$  spectral maximum at approximately 640 nm and the relative angle between  $Q_x$  and  $Q_y$  of  $\theta = 90 - \beta = 73^\circ$  are in good agreement with previously reported results.<sup>4</sup> In essence, polarization-associated analysis of Chl *a* excitation spectra is a straightforward and effective way of isolating overlapping  $Q_x$  and  $Q_y$  ground state transitions.

### Ultrafast relaxation dynamics of Chl *a*

Although Chl *a* has been studied extensively, open questions remain concerning the exact mechanism of energy transfer dynamics within this molecule. The most common scheme employed to explain Chl *a* energy relaxation is a sequential one, *i.e.*,  $B_x \rightarrow Q_x \rightarrow Q_y$ .  $B \rightarrow Q$  transfer seems to occur within 100–150 fs, as inferred by time-resolved fluorescence depletion

spectroscopy<sup>15</sup> or UV pump, NIR probe TA.<sup>25</sup> A time-constant in the 100–250 fs range with notable strong and unusual solvent dependence has been attributed to the subsequent  $Q_x \rightarrow Q_y$  transfer.<sup>15</sup>

To better elucidate the mechanism of energy transfer dynamics in Chl *a*, we perform TA experiments under different excitation conditions (B- and Q-band excitation) and in various solvents (acetone, ethanol (EtOH), and benzonitrile (BN)). In particular, we expect to observe direct evidence of  $B \rightarrow Q$  transfer and to understand the origin of the reported solvent dependence of the  $Q_x \rightarrow Q_y$  energy transfer step. In the following, we show a representative set of data for Chl *a* in acetone. The corresponding plots for the other two solvents are shown in ESI Fig S2 and S3.† Absorption and pump pulse spectra are shown in Fig. 2a (Fig. S2†). Chirp-corrected TA data

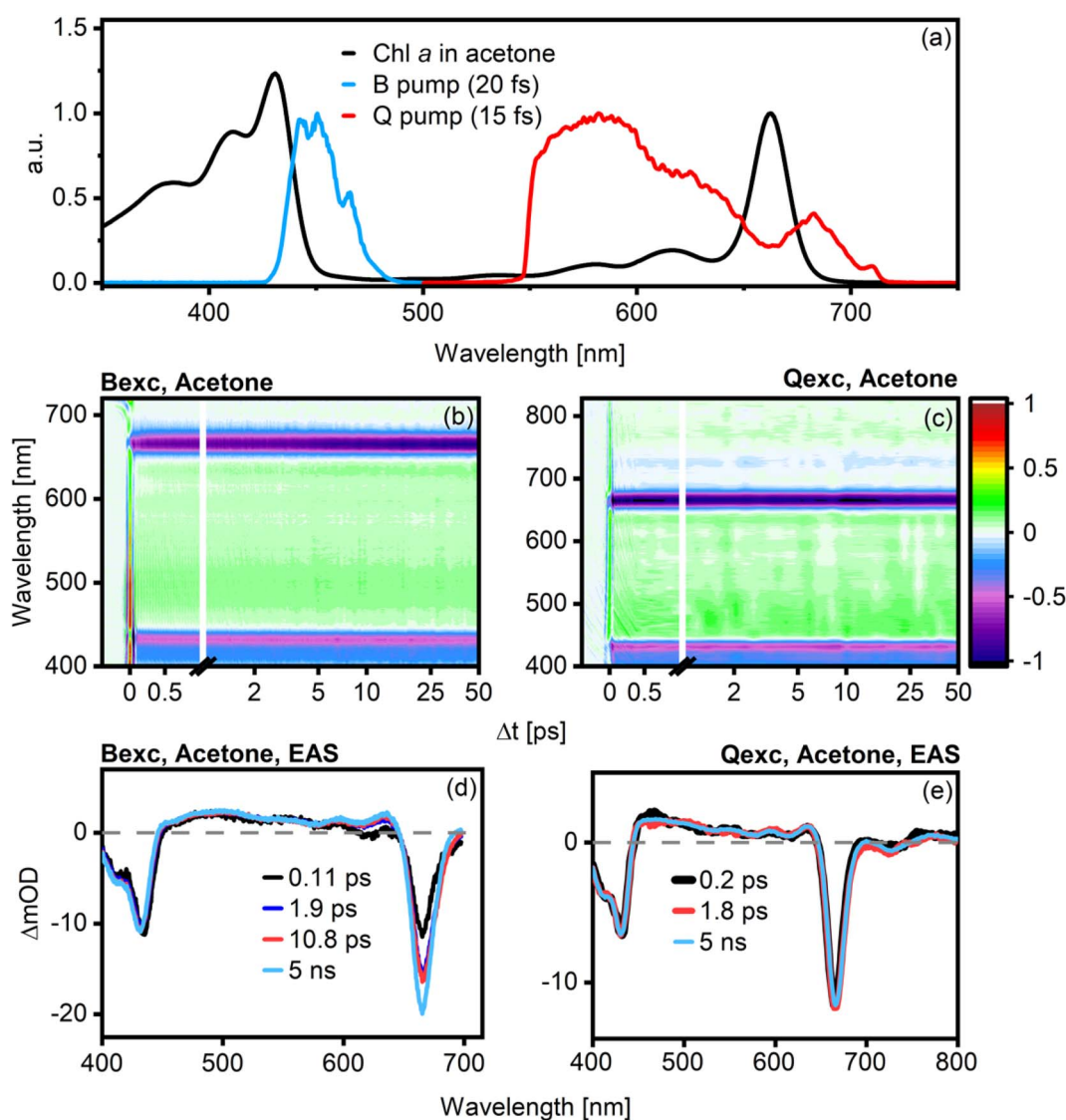


Fig. 2 Absorption spectrum of Chl *a* in acetone plotted against the pump spectra for B-band (blue line) and Q-band (red line) excitation (a). Chirp-corrected TA data for both excitation conditions are shown in (b) and (c). Visible changes in the dynamics are present after B excitation, with a clear increase of the Q-band GSB signal over time, but this is not true after Q excitation. The EAS and lifetimes extracted from global analysis of Chl *a* in acetone are shown in (d) after excitation in the B-band and in (e) after excitation in the Q band. The corresponding plots for Chl *a* in EtOH and benzonitrile are shown in Fig. S2.†



in magic angle (MA) configuration after B- and Q-band are shown in Fig. 2b and c, and the global analysis results are shown in Fig. 2d and e (Fig. S3†). The lifetimes corresponding to the individual spectral components are reported in each panel. Only the evolution-associated spectra (EAS) are shown for simplicity, while the decay-associated spectra for all solvents are shown in ESI Fig. S4.†

In the case of B-band excitation (Fig. 2b and d), the best fit is achieved with four kinetic components. The shortest component has a lifetime of  $\sim 110$  fs (black line), with negligible solvent dependence. It is characterized by a decrease and blue shift of the signal in the B-band region (430 nm) and a concomitant increase of the main Q-band signal (660 nm). This behavior is consistent with B  $\rightarrow$  Q transfer, and the B  $\rightarrow$  Q transfer time agrees with previously reported values.<sup>8</sup> The second component (dark blue line) shows minimal spectral evolution; the only noticeable feature is a slight broadening of the  $Q_y$  band. From the global fits, it appears to be strongly solvent-dependent, varying from *ca.* 400 fs (BN, *cf.* Fig. S3†) to 2 ps (acetone). Notably, the amplitude of this component is near-negligible, and it is not associated with any spectral evolution typical of state-to-state energy transfer. As changes to small-amplitude components do not substantially affect the goodness-of-fit, the lifetime associated with this component is inevitably highly uncertain. As such, although this component appears to exhibit some solvent dependence, we do not make strong quantitative claims about this behavior. We emphasize, however, that this component does not contain significant changes to stimulated emission or excited state absorption, implying that it is not related to population transfer. The third kinetic component (red line), with a lifetime of about 7–11 ps, correlates with a narrowing of the  $Q_y$  band and an increase in its amplitude. Finally, the last kinetic component (light blue line, fixed to 5 ns) is readily assigned to the excited state lifetime of Chl *a* as known from fluorescence lifetime measurements.<sup>26</sup>

In the case of Q-band excitation (Fig. 2c and e), only three components are necessary to obtain a satisfactory data fit. We again observe an ultrafast component of 100–300 fs (black line) that was previously assigned to  $Q_x \rightarrow Q_y$  transfer.<sup>8,15</sup> If this component was associated with population transfer, we would expect a signal decrease in the 520–630 nm  $Q_x$ -region concomitant with an increase in the  $Q_y$  ground-state bleach (GSB)/stimulated emission (SE) amplitude around 660 nm. As in the B-band excitation case, however, we do not observe any such features, leaving this component inconsistent with a population transfer process. An intermediate 1.5–15 ps component (red line) with extremely small amplitude similarly does not relate to significant spectral changes, suggesting a small structural reorganization process. The longest component is again the excited state lifetime of Chl *a*, fixed at 5 ns (light blue line).

### Isolating $Q_x$ features by polarization control

General problems with the analysis of the  $Q_x$  features of Chl *a* are the weakness of the transition and its strong spectral overlap with the  $Q_y$ -related signals (*cf.* Fig. 1). It could then be conceivable that the lack of  $Q_x \rightarrow Q_y$  population transfer

features in MA-TA stems from their smaller amplitudes being covered by the larger  $Q_y$  signals.

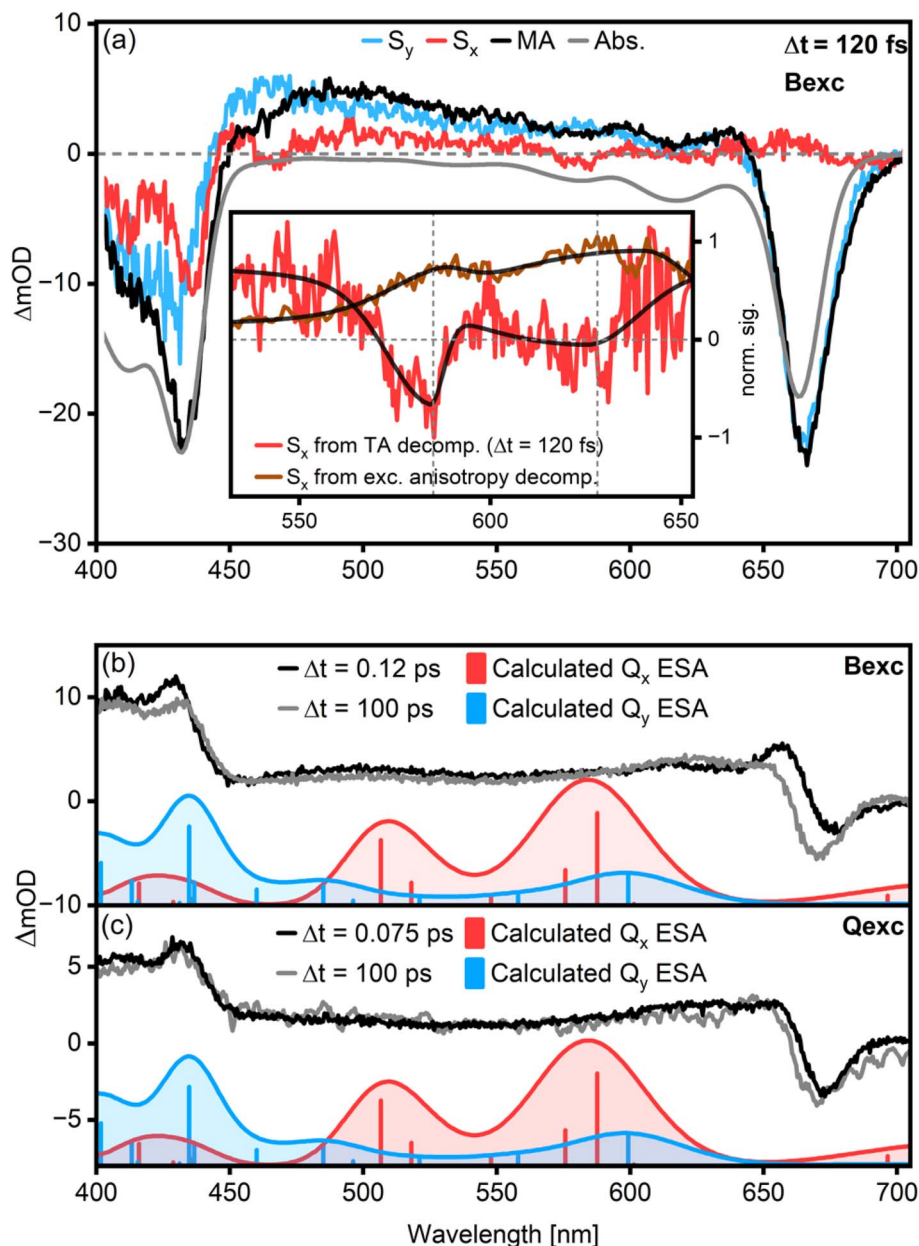
To isolate  $Q_x$ -signatures and to determine their transient behavior, we have measured transient absorption anisotropy (TAA) data for Chl *a* in acetone after B-band excitation. With these data, we can decompose the TA signals into polarized components in an analogous procedure to that described for excitation anisotropy.<sup>16,24</sup> In this case, a value of  $\beta = 32^\circ$  is necessary to fully suppress the  $Q_y$  peak. In the employed representation,  $S_x$  contains signals with a TDM parallel to that of  $B_x$ , including all  $Q_x$  features, while  $S_y$  contains the  $Q_y$  and  $B_y$  contributions. Representative spectra are shown for a time delay  $\Delta t = 120$  fs in Fig. 3a. At this delay time, effects due to pump-and-probe pulse overlap are negligible, and there is already population in the Q-band resulting from B  $\rightarrow$  Q transfer (*cf.* Fig. 2b and d).

In search for  $Q_x$ -features, we identify two weak negative peaks at 580 nm (*ca.*  $17\,200\text{ cm}^{-1}$ ) and 625 nm (*ca.*  $16\,000\text{ cm}^{-1}$ ), matching the  $Q_x$  features observed in the excitation anisotropy measurements (Fig. 3a, red line and inset). They show an energy spacing of about  $1100\text{ cm}^{-1}$  as calculated by fitting a bi-gaussian function to the peaks (Fig. 3a, inset, black lines), in excellent agreement with magnetic circular dichroism spectra for pentacoordinated Chl *a*.<sup>3</sup>

The lineshape of the  $S_x$  spectrum remains identical for time delays between 100 fs and several ps (Fig. S5†). This means that while we observe the  $Q_x$  GSB, no  $Q_x$  SE (or excited-state absorption (ESA)) features are discernible. We support this statement further by performing a global analysis of the polarization-decomposed TA maps shown in the ESI (Fig. S6†). Similar time constants are needed to fit  $S_x$  and  $S_y$ , and there is no evidence for a transient population of an excited  $Q_x$  state.

We further examine the possibility of  $Q_x$  ESA features by comparing the experimental excited state spectrum to calculations for both B-excitation (Fig. 3b) and Q-excitation (Fig. 3c). The experimental excited state spectrum is determined from the measured TA spectrum by subtracting the GSB.<sup>27,28</sup> The latter is approximated by a scaled absorption spectrum. The resulting spectra only contain ESA and SE contributions. We then compare the lineshapes of these spectra at early and late times with calculations for the  $Q_x$ - and  $Q_y$ -associated ESA, considering one acetone molecule coordinating to the central Mg of a Chl *a* molecule. Notably, the excited-state spectrum at early times after B-band excitation has a positive feature at 655 nm which is absent at later times and in the case of Q-band excitation. We attribute this to ESA from the B-band, which is cancelled by the Q-band GSB, explaining why we do not observe this feature in the magic angle TA data in Fig. 2d. While the calculated  $Q_y$  ESA-peaks correspond to measured features at both early and late times, the features related to  $Q_x$  ESA are missing. In particular, two intense  $Q_x$ -features are expected in the  $16\,000\text{--}21\,000\text{ cm}^{-1}$  region at early times. However, the measured spectra in this region at early and late times are effectively identical. While ESA from  $Q_y$  is clearly visible, the lack of excited-state signatures associated with  $Q_x$  means that  $Q_x$  is either not significantly involved in the energy relaxation network or its lifetime is shorter than the experimental time resolution.





**Fig. 3** Results of the Chl *a* TA data decomposition into polarized components at  $\Delta t = 120$  fs (a) for  $\beta = 32^\circ$ . Here,  $S_x$  is chosen to indicate the direction parallel to the pumped TDM and contains all contributions with this polarization. Accordingly, the  $Q_y$  transitions are suppressed and appear only in  $S_y$ . The  $Q_x$  GSB transitions extracted from polarized TA and steady-state anisotropy match well (a, inset). They show an energy spacing of about  $1100\text{ cm}^{-1}$  as calculated by fitting a bi-gaussian function to the peaks. Excited-state associated spectra of Chl *a* after excitation in the B-band (b) and the Q-band (c) were calculated for different time delays (grey and black lines) and are compared to theoretical  $Q_x$  and  $Q_y$  ESA band positions (blue and red histograms and curves, oscillator strength unscaled). The theoretical ESA band positions were calculated considering one acetone molecule coordinating to the central Mg of a Chl *a* molecule. Calculated ESA transitions from  $Q_y$  match the experimental spectrum well, while  $Q_x$ -ESA features are absent.

### System and system-bath relaxation dynamics at different timescales

We compare our experimental results with quantum dynamics calculations (details in Fig. S7, S8 and Table T1†). Our XMC-CASPT2 results predict that the excited-state potentials of Chl *a* are nested on top of each other with minimal curve displacement. This is illustrated in Fig. 4a with a cut through the potential energy surface (PES) along the mode with the

highest projection onto the non-adiabatic coupling (NAC) vector, with a vibrational frequency of  $1489\text{ cm}^{-1}$  (mode 171 in the frequency analysis). Cuts along modes with weaker coupling and different frequencies show the same trend (Fig. S8†). We did not observe any crossings in the central region of the PES where most of the population resides. The temporal evolution of the population of the adiabatic states is depicted in Fig. 4b. We note that while the dynamics simulations include the non-





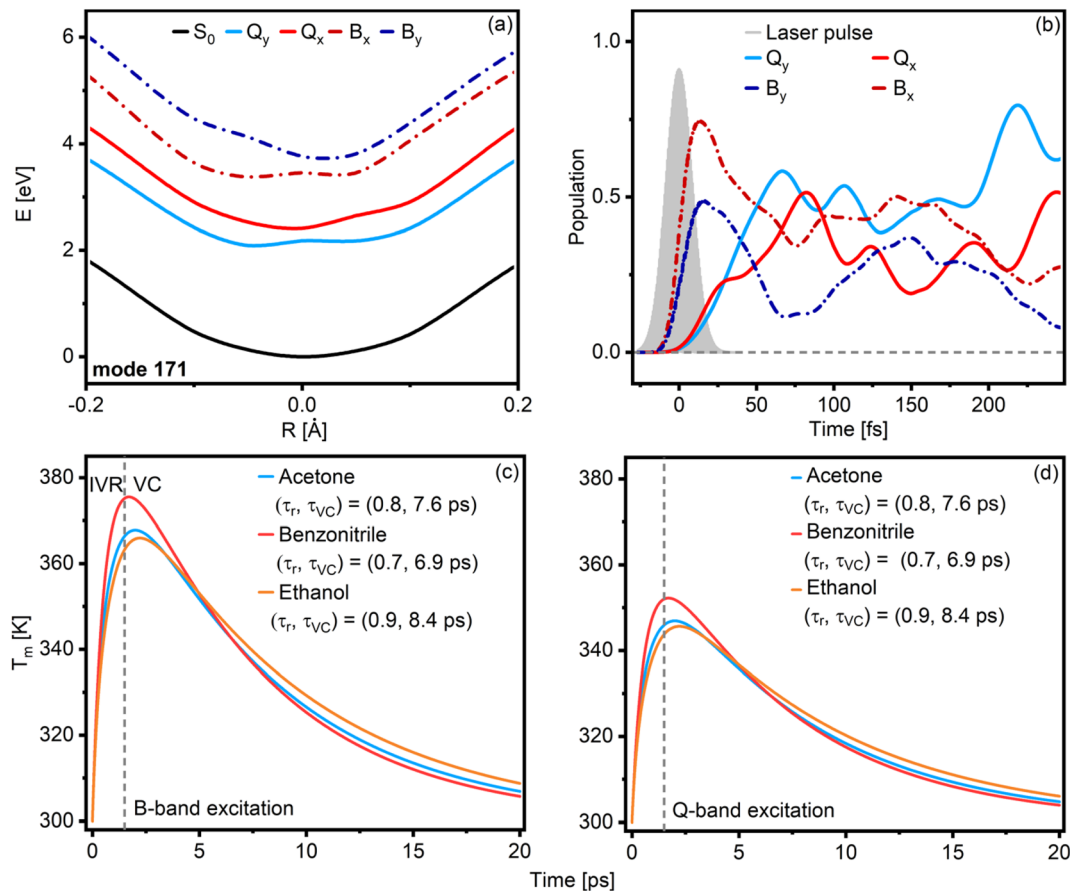


Fig. 4 Cuts through the potential energy surfaces of Chl *a* along the mode with the strongest projection on the non-adiabatic coupling vector (a). The curve displacement between the excited states is negligible, resulting in nested potentials with no evident curve crossings.  $B_x$  and  $B_y$ , as well as  $Q_x$  and  $Q_y$ , show a small energy gap and therefore large coupling, which explains the almost instantaneous population transfer between them. The comparatively large energy gap between  $B_x$  and  $Q_x$ , on the other hand, leads to predicted transfer times of  $\sim 100$  fs (b). At longer timescales, energy relaxation occurs via intra-molecular vibrational redistribution (IVR), followed by slower equilibration with the surrounding solvent (vibrational cooling VC). We capture the effects of biphasic energy relaxation in a simplified model for B-band (c) and Q-band (d) excitation scenarios and for different solvents. Since less energy is deposited into the  $Q_y$  vibrational manifold, the peak local temperature is less than for B-band excitation.

adiabatic coupling terms, the depicted populations correspond to those of the electronic states within the Born–Oppenheimer approximation and not the vibronic states observed in experiment. After excitation to the B-band with an explicitly simulated pump pulse, a population transfer from B to Q occurs on a moderately longer timescale than the intra-band transfer. This is explained by the significant energy difference between the B and Q bands (Fig. 4a). The exact timescale of the B  $\rightarrow$  Q transfer depends on the amount of initially excited vibrational levels contained in the wavepacket. On PES constructed with normal modes of lower overlap with the NAC vector and in different spectral ranges, the dynamics are generally slower than normal modes of high overlap (Fig. S8†). Nonetheless, all simulations have in common that the intra-band transfer in both the B and the Q band is almost instantaneous. In contrast, the B to Q transfer takes about 50–150 fs. This is in excellent agreement with our experimental data presented in the discussion of Fig. 2. Further, we find that the B  $\rightarrow$  Q transfer heavily depends on coupling the B band with  $Q_x$ . If this coupling is turned off, the B  $\rightarrow$  Q transfer slows down significantly. On

the contrary, if B/ $Q_y$  coupling is turned off, the population dynamics do not change much (Fig. S8†).

While we can readily explain ultrafast time constants with our quantum dynamics simulations, the dynamics at longer times are not practically accessible with this type of calculation. At these longer timescales, we expect energy redistribution and dissipation to the solvent to be the dominating contributions to the dynamics.<sup>28</sup> As such, the bath must be considered when trying to reproduce the longer kinetic components observed in TA.

We construct a simplified numerical model to test our hypothesis of cooling dynamics on multiple timescales. Excitation of Chl *a* with a visible laser pulse always means that a large amount of energy is deposited into the vibrational modes of the molecule. Energy deposited – directly or indirectly – into a high-lying state of  $Q_y$ , will relax in a non-linear sequence of intra-molecular vibrational redistribution (IVR) events, solute-to-solvent energy transfer, and solvent equilibration. Using an effective molecular temperature  $T_m$  (details and modeling parameters in the ESI text, Table T2, Fig. S9 and S10†)





that increases with the energy deposited into the system, we predict a molecule-to-solvation shell transfer time of  $\sim 1$  ps. This is consistent with observations from our previous work on carotenoids.<sup>29–31</sup> The dynamics of  $T_m$  are determined by two timescales:  $\tau_{\text{IVR}} \sim 1$  ps, which accounts for the rise in  $T_m$  due to thermalization of the  $Q_y$  state, and  $\tau_{\text{VC}} \sim 7–9$  ps, which is the relaxation of the solute due to vibrational cooling (VC) into the solvation shell. The exact values of the time constants vary with the solvent (Fig. 4c and d). The results indicate that the relaxation kinetics are almost independent of whether one excites  $Q_y$  directly or indirectly *via* the B-band. However, the maximum  $T_m$  reached should be affected by excitation wavelength, as this defines the amount of energy that thermalizes (Fig. 4c and d). As  $\tau_{\text{IVR}}$  is associated with heating and  $\tau_{\text{VC}}$  with cooling, we expect to observe thermal broadening effects of the  $Q_y$ -band over  $\sim 1$  ps as the maximum  $T_m$  is achieved, followed by spectral narrowing over  $\sim 10$  ps as the population of low-frequency vibrational modes drops while the solute and first solvation shell re-equilibrate with the bulk solvent. Careful inspection of the decay-associated and evolution-associated spectra (DAS and EAS, respectively) in the Q-band region (Fig. S11†) confirms these predictions for excitation in the B-band, although the observed lineshape changes are minimal. After Q-band excitation, IVR and cooling are much less pronounced. In particular, we do not observe a narrowing in  $\sim 10$  ps as expected. Instead, we observe a minimal but continuous redshift typical of Stokes shift dynamics (Fig. S11†).<sup>28</sup>

## Discussion

The results presented in this work clarify some essential aspects of the mechanism of energy relaxation in Chl *a*. When excited in the B-band, population transfer to the Q band occurs within 120 fs, as demonstrated by TA experiments (Fig. 2d) and quantum dynamical calculations (Fig. 4b). The Q band contains  $Q_x$  and  $Q_y$ , two distinct but strongly coupled states with overlapping but discernible absorption spectra (Fig. 1). However, polarized TA measurements show no evidence of x-polarized excited state features (Fig. 3). As such, we can conclude that

either  $Q_x$  is not involved in  $B \rightarrow Q$  energy relaxation in Chl *a*, or that the  $Q_x \rightarrow Q_y$  transfer is faster than our time resolution (*ca.* 30 fs). Our quantum chemical calculations strongly support the latter interpretation, as the  $B \rightarrow Q$  transfer was shown to slow significantly if the  $B/Q_x$  coupling is turned off (Fig. S8†).

An implication of these observations is that none of the time constants found in MA-TA (*cf.* Fig. 2) can be assigned to  $Q_x \rightarrow Q_y$  transfer. The small amplitudes and only subtle influence on the spectra of the kinetic components other than the one related to  $B \rightarrow Q$  transfer suggest instead an association with IVR and subsequent heat dissipation to the solvent.

In Fig. 5, we propose a comprehensive scheme for the internal conversion dynamics of Chl *a*. After the initial  $B \rightarrow Q$  transfer, the state we detect is y-polarized. Therefore, if  $Q_x$  is participating in the energy transfer as suggested by the quantum dynamics calculations, its relaxation towards a vibrationally hot, y-polarized state must be faster than our time resolution. The relaxation from there to the vibrationally relaxed  $Q_y$  occurs through IVR, leading to heat exchange with the first solvation shell. Further relaxation to bulk solvent (VC) then occurs on slower timescales. Overall, IVR and VC are sufficient to explain the timescales in the evolution of the Chl *a* TA spectra beyond the first few hundred fs after B-band excitation:  $\sim 1$  ps for IVR to a hot pseudo-thermal state,  $\sim 10$  ps for relaxation of the hot thermal state (Fig. 4c and *cf.* Fig. 2d). The effect of IVR and VC on the lineshape is much less pronounced after Q-band excitation. In this case, the changes in the lineshape can be explained by Stokes shift dynamics.<sup>28</sup>

In summary, our findings shed new light on the internal conversion dynamics in Chl *a*. Although the heating-cooling cycle on the  $Q_y$  surface of Chl *a* is highly complicated and requires further investigation from both experimental and theoretical sides, we reach a good agreement between model and experiment. We present experimental and theoretical evidence for  $B \rightarrow Q$  transfer on a  $\sim 120$  fs timescale. The relaxation dynamics in the Q band are dominated by the strong coupling of the  $Q_x$  and  $Q_y$  states and occur on a  $< 30$  fs timescale. This leads to the absence of x-polarized excited-state signatures in TAA experiments. We attribute this ultrafast  $Q_x$  lifetime to the

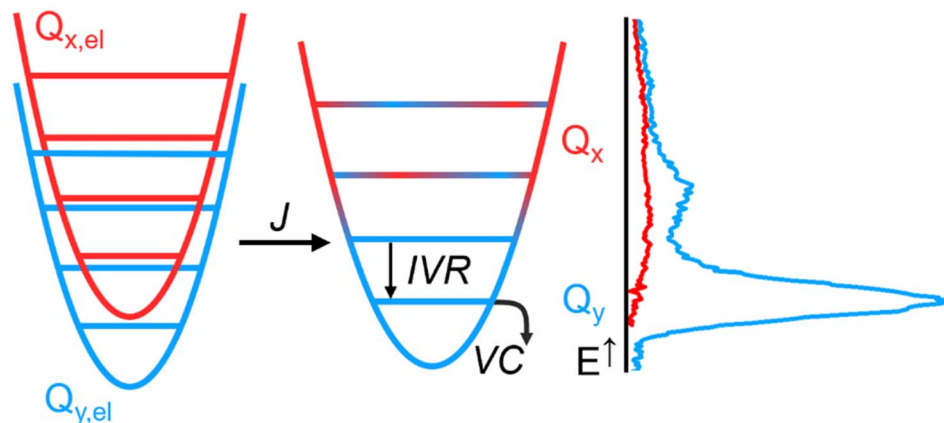


Fig. 5 Vibronic mixing ( $J$ ) in the Q band leads to strongly coupled states  $Q_x$  (red) and  $Q_y$  (blue). Ultrafast transfer *via*  $Q_x$  populates a vibrationally excited  $Q_y$ , which in turn relaxes *via* intra-molecular vibrational redistribution (IVR), followed by slower equilibration with the surrounding solvent (vibrational cooling VC). The right panel shows a sketch of the spectral profiles of  $Q_x$  and  $Q_y$ , as discussed in Fig. 1.



small and internal-coordinate-independent energy gap between  $Q_x$  and  $Q_y$ . Regardless of its short lifetime,  $Q_x$  seems important for efficient  $B \rightarrow Q$  relaxation, as the  $B \rightarrow Q$  transfer rate slows down if  $B/Q_x$  coupling is deactivated. Based on these results, we reinterpret the solvent-dependent, longer time constants found in TA measurements as a combination of IVR, VC, and possibly Stokes shift dynamics on  $Q_y$ . Overall, our unique combination of extensive theoretical modeling, numerical methods, and highly sensitive TA experiments allows us to paint a clear and consistent picture of the energy transfer in Chl *a* under the excitation conditions relevant to photosynthetic systems and to shed light on the previously controversial role of the  $Q_x$  state in the Chl *a* dynamics.

## Materials and methods

### Materials

HPLC and spectroscopy grade solvents acetonitrile (ACN), ethyl acetate (EtOAc), acetone, ethanol (EtOH), BN and methanol (MeOH) were obtained from VWR (Ismaning, Germany) and ultra-pure water (18 M $\Omega$  cm<sup>-1</sup>) from a Millipore S.A.S. Milli-Q Academic system (18.2 M $\Omega$  cm<sup>-1</sup>, Molsheim, France).

### Chromatography

**Analytical HPLC.** An Agilent 1260 Infinity II LC system was used with a 1260 Infinity Degasser, a 1260 Series quaternary pump and 1260 Series diode array detector; Merck LiChrospher® 100 RP-18 (5  $\mu$ m) LiChroCART® 125-4, with or without a Phenomenex ODS 4  $\times$  2 mm i.d. pre-column; injection volume: 100  $\mu$ L. Isocratic mobile phase: ACN/MeOH/EtOAc 6/2/2,<sup>32</sup> flow rate 0.5 mL min<sup>-1</sup>. Data were processed with OpenLab CDS Data Analysis 2.3.

**Semi-preparative HPLC.** Büchi Pure C-830 with prep HPLC pump 300 bar, fraction collector, and prep sample injection valve. Gynkotech LC-System with manual sampler, M480 pump, Phenomenex DG-301 online degasser, Gynkotech UVD 640 diode array detector and a Rheodyne injection valve with 5 mL loop. Column: Phenomenex Luna C8(2), 100 Å, 250  $\times$  100 mm with a Phenomenex pre-column C18 15  $\times$  21.2 mm; mobile phase ACN/MeOH/EtOAc 6/2/2, isocratic conditions, flow rate 4 mL min<sup>-1</sup>. Data were processed with Gynkosoftware 5.50 or Büchi Pure software 1.5.

### Spectroscopy

HR-MS were measured at the MS facility of the Department of Chemistry, University of Munich. Data were processed using Xcalibur.

UV-vis spectra were measured on a PerkinElmer Lambda 365 spectrometer. Emission spectra were measured on a Spectrofluorometer FS5 (Edinburgh Instruments). All spectra are background-corrected.

### Extraction procedure for Chl *a*

Frozen spinach was purchased at a local supermarket (Edeka junger Spinat). 8 portions of frozen spinach (approximately 15 g each) were thawed by soaking the spinach in water; water was

removed from the soaked spinach by squeezing the spinach with a table cloth. The spinach was then mixed with a blender in 100 mL of cold acetone/MeOH (9/1 v/v). The resulting slurry was centrifuged at 4 °C for 5 min at 1000 rpm. The supernatant was filtered; the filtrate was collected on ice and directly applied to semi-preparative HPLC. The Chl *a* fraction was collected on ice, and the solvents were evaporated *in vacuo*, yielding 9.7 mg of Chl *a*.

HR ESI-MS (positive ion mode):  $m/z$  (found) = 893.5426;  $m/z$  (calculated) = 893.5426,  $\Delta$  = 0.56 ppm.

### TA spectroscopy

Chl *a* samples were prepared by dissolving the pigment in acetone, EtOH or BN to an OD of 0.2–0.3 and then degassing the solution with nitrogen for about one minute.

TA experiments were performed for Chl *a* in acetone, ethanol (EtOH), and benzonitrile (BN). Chl *a* was excited in the B and Q bands in separate experiments for all solvents. The pulse duration of the two pump pulses was comparable, with the B-pump at 20 fs and the Q-pump at 15 fs as measured by SD-FROG and SHG-FROG, respectively. The measured and retrieved FROG traces, as well as the retrieved spectral and temporal phase profiles, are shown in the ESI (Fig. S12†). The TA setups used for B- and Q-band excitation have been described in detail before.<sup>33,34</sup> Briefly, a 5 kHz, 2.5 mJ, 25 fs amplified laser system (Coherent Legend Elite Duo HE+ Ti:Sapphire MOPA) is used to seed a 1 m long, commercial hollow-core fiber (HCF) with a core diameter of 250  $\mu$ m filled with 1 bar of Argon gas. The resulting supercontinuum is either frequency-doubled in an achromatic second-harmonic scheme to yield the B-band pump pulse<sup>33</sup> or spectrally filtered to obtain the Q-band pump pulse.<sup>34</sup> In the first case, the pump is compressed by a three-prism configuration, while in the second case, the compression is achieved by chirped mirrors. In both experiments, the probe pulse is given by a supercontinuum obtained by seeding a CaF<sub>2</sub> crystal with the NIR part of the HCF output. For B-band excitation, a double-chopping scheme was employed to reduce scatter, and the pump energy was 60 nJ per pulse with a focal spot diameter of 200  $\mu$ m; for Q-band excitation, the pump pulse was not chopped and its energy was 16 nJ per pulse with a focal spot of 140  $\mu$ m. The data was recorded at MA (54.7°) configuration between pump and probe for B- and Q-pumping. Additionally, datasets with parallel and perpendicular polarization of pump and probe were recorded for B-band excitation of Chl *a* in acetone in order to calculate the TAA.

Global analysis was performed with the open-source software Glotaran.<sup>35,36</sup>

### Quantum dynamics simulations

The quantum dynamics simulations were performed on 2D PES at the XMS-CASPT2 (ref. 37–39) level of theory, using the basis set ANO-RCC-VDZP<sup>40–42</sup> and a CAS(6/6) active space in the gas phase. As coordinates, we chose the normal modes exhibiting the highest overlap with the non-adiabatic coupling vector at the  $Q_y$  minimum geometry,<sup>5</sup> optimized at the CAM-B3LYP/6-311G\* level.<sup>43</sup> These modes are 171 and 198, lying at 1489 cm<sup>-1</sup> and 1639 cm<sup>-1</sup>, respectively. Other normal



modes (239 and 103) with smaller overlap with the NAC vector and in a different spectral range were also tested for PES construction. Geometries are available in the ESI.† The resulting quantum dynamics is similar for all modes, but the time-scales slow down with decreasing overlap with the NAC vector (Fig. S9†). At each point of the PES, a RASSCF calculation was performed with state averaging over four states, followed by one with state averaging over eight states (Fig. S8†). The ensuing XMS-CASPT2 calculation was performed for the states of interest, namely the  $S_0$ ,  $Q_y$ ,  $Q_x$ ,  $B_x$ , and  $B_y$ . An IPEA<sup>44</sup> and an imaginary shift<sup>45</sup> of 0.1 were applied, as in our previous study.<sup>5</sup> With this approach, an energy difference between  $Q_x$  and  $Q_y$  at the FC point of 0.225 eV is obtained, which corresponds well with our reference value of 0.23 eV.<sup>5</sup> The non-adiabatic coupling elements were computed at the XMS-CASPT2 level of theory. All PES, NAC, and TDM calculations were performed with the OpenMolcas 23.06 program package.<sup>46–48</sup> The QDng software package<sup>49</sup> was used for the subsequent quantum dynamic simulations. The propagation was initially performed by assuming a delta excitation into  $B_y$ , therefore populating this state's manifold of vibrational levels. Later, an explicitly simulated pump pulse with a maximum field strength of 0.0026 GV cm<sup>-1</sup>, an FWHM of 20 fs, and a central frequency of  $\omega_0 = 3.60$  eV was also applied to excite the population from the ground state into the B Band. The laser parameters were chosen to excite a similar population (about 5%) into the B-band as in our experimental studies.

### Calculated ESA spectra

Geometries of the  $Q_y$  and  $Q_x$  states of Chl *a* with one axially coordinated acetone molecule were optimized using Gaussian 16C.02 (ref. 50) with the CAM-B3LYP density functional<sup>51</sup> and the 6-311G\* basis set.<sup>43,52,53</sup> Bulk solvent effects were described with the polarizable continuum model (PCM).<sup>54,55</sup> Five roots were included in the optimization. Optimized geometries were verified as energy minima by the absence of imaginary vibrational frequencies.

Excited state absorption spectra for the optimized geometries were simulated at the double-hybrid TDDFT level using the ORCA 5.0.3 software package.<sup>56–58</sup> The SCS- $\omega$ PBEP86 density functional<sup>59</sup> was used in the Tamm–Dancoff approximation<sup>60</sup> along with the def2-TZVPD<sup>61,62</sup> basis. The calculation was accelerated by employing the RIJCOSX approximation<sup>63,64</sup> with the def2/J<sup>65</sup> and def2-TZVPD/C<sup>66</sup> auxiliary basis sets. A tighter-than-usual integration grid (ORCA keyword DefGrid3) and convergence criteria (ORCA keywords VeryTightSCF TightPNO) were employed. The bulk solvent was described implicitly *via* the linear-response conductor-like polarizable continuum model (LR-CPCM),<sup>67,68</sup> assuming equilibrium solvation in the excited state.

### Data availability

Data for this article, including absorption spectra, pump pulse spectra, transient absorption and transient absorption anisotropy maps are available at <https://zenodo.org/records/14033283>.

### Author contributions

EK and AK performed the experiments; SM provided the sample; SR, LB, CD, and RdVR performed calculations; EK, AK, ET, SR, LB, RdVR, CDPD, and JH analyzed and interpreted the data; EK and JH prepared the plots; EK, AK, ET and JH wrote the main text. All authors reviewed the manuscript before submission.

### Conflicts of interest

The authors declare no competing interests.

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### References

- 1 R. E. Blankenship, *Molecular Mechanisms of Photosynthesis*, Wiley/Blackwell, Chichester, West Sussex, 2nd edn, 2014.
- 2 R. Croce and H. van Amerongen, Natural strategies for photosynthetic light harvesting, *Nat. Chem. Biol.*, 2014, **10**, 492–501.
- 3 J. R. Reimers, Z.-L. Cai, R. Kobayashi, M. Rätsep and E. Krausz, Assignment of the Q-Bands of the Chlorophylls: Coherence Loss via  $Q_x$ - $Q_y$  Mixing, *Sci. Rep.*, 2013, **3**, 2761.
- 4 C. Zahn, T. Stensitzki and K. Heyne, Femtosecond anisotropy excitation spectroscopy to disentangle the  $Q_x$  and  $Q_y$  absorption in chlorophyll *a*, *Chem. Sci.*, 2022, **13**, 12426–12432.
- 5 S. Reiter, L. Bäuml, J. Hauer and R. de Vivie-Riedle, Q-Band relaxation in chlorophyll: new insights from multireference quantum dynamics, *Phys. Chem. Chem. Phys.*, 2022, **24**, 27212–27223.
- 6 Y. Song, A. Schubert, E. Maret, R. Burdick, B. Dunietz, E. Geva and J. Ogilvie, Vibronic structure of photosynthetic pigments probed by polarized two-dimensional electronic spectroscopy and *ab initio* calculations, *Chem. Sci.*, 2019, **10**, 8143–8153.
- 7 E. Bukarté, A. Haufe, D. Paleček, C. Büchel and D. Zigmantas, Revealing vibronic coupling in chlorophyll *c1* by polarization-controlled 2D electronic spectroscopy, *Chem. Phys.*, 2020, **530**, 110643.
- 8 W. P. Bricker, P. Shenai, A. Ghosh, Z. Liu, M. Enriquez, P. Lambrev, H.-S. Tan, C. Lo, S. Tretiak, S. Fernandez-Alberti and Y. Zhao, Non-radiative relaxation of photoexcited chlorophylls: Theoretical and experimental study, *Sci. Rep.*, 2015, **5**, 13625.
- 9 M. Gouterman, Spectra of porphyrins, *J. Mol. Spectrosc.*, 1961, **6**, 138–163.





- 10 M. Gouterman, G. H. Wagnière and L. C. Snyder, Spectra of Porphyrins Part II. Four Orbital Model, *J. Mol. Spectrosc.*, 1963, **11**, 108–127.
- 11 M. Linke, A. Lauer, T. Von Haimberger, A. Zacarias and K. Heyne, Three-dimensional orientation of the Q<sub>y</sub> electronic transition dipole moment within the chlorophyll *a* molecule determined by femtosecond polarization resolved VIS pump-IR probe spectroscopy, *J. Am. Chem. Soc.*, 2008, **130**, 14904–14905.
- 12 R. Simonetto, M. Crimi, D. Sandonà, R. Croce, G. Cinque, J. Breton and R. Bassi, Orientation of chlorophyll transition moments in the higher-plant light-harvesting complex CP29, *Biochemistry*, 1999, **38**, 12974–12983.
- 13 E. Meneghin, C. Leonardo, A. Volpato, L. Bolzonello and E. Collini, Mechanistic insight into internal conversion process within Q-bands of chlorophyll *a*, *Sci. Rep.*, 2017, **7**, 1–7.
- 14 V. Tiwari, W. K. Peters and D. M. Jonas, Electronic resonance with anticorrelated pigment vibrations drives photosynthetic energy transfer outside the adiabatic framework, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 1203–1208.
- 15 Y. Shi, J. Y. Liu and K. L. Han, Investigation of the internal conversion time of the chlorophyll *a* from S<sub>3</sub>, S<sub>2</sub> to S<sub>1</sub>, *Chem. Phys. Lett.*, 2005, **410**, 260–263.
- 16 Y. Xu, L. Mewes, E. Thyraug, V. Sláma, F. Šanda, H. Langhals and J. Hauer, Isolating Pure Donor and Acceptor Signals by Polarization-Controlled Transient Absorption Spectroscopy, *J. Phys. Chem. Lett.*, 2023, **14**, 5390–5396.
- 17 M. Rätsep, J. Linnanto and A. Freiberg, Mirror symmetry and vibrational structure in optical spectra of chlorophyll *a*, *J. Chem. Phys.*, 2009, **130**, 194501.
- 18 J. R. Reimers, M. Rätsep, J. M. Linnanto and A. Freiberg, Chlorophyll Spectroscopy: Conceptual Basis, Modern High-Resolution Approaches, and Current Challenges, *Proc. Est. Acad. Sci.*, 2022, **71**, 127.
- 19 C. Houssier and K. Sauer, Circular dichroism and magnetic circular dichroism of the chlorophyll and protochlorophyll pigments, *J. Am. Chem. Soc.*, 1970, **92**, 779–791.
- 20 S. Krawczyk, The effects of hydrogen bonding and coordination interaction in visible absorption and vibrational spectra of chlorophyll *a*, *Biochim. Biophys. Acta, Bioenerg.*, 1989, **976**(2–3), 140–149.
- 21 E. W. Thulstrup, J. Michl and J. H. Eggers, Polarization spectra in stretched polymer sheets, II.1 Separation of  $\pi$ - $\pi^*$  absorption of symmetrical molecules into components, *J. Phys. Chem.*, 1970, **74**, 3868–3878.
- 22 A. C. Albrecht, Polarizations and assignments of transitions: The method of photoselection, *J. Mol. Spectrosc.*, 1961, **6**, 84–108.
- 23 E. Thyraug, T. J. Sørensen, I. Gryczynski, Z. Gryczynski and B. W. Laursen, Polarization and symmetry of electronic transitions in long fluorescence lifetime triangulenium dyes, *J. Phys. Chem. A*, 2013, **117**, 2160–2168.
- 24 Y. Xu, M. Peschel, M. Jänchen, R. Foja, G. Storch, E. Thyraug, R. de Vivie-Riedle and J. Hauer, Determining Excited-State Absorption Properties of a Quinoid Flavin by Polarization-Resolved Transient Spectroscopy, *J. Phys. Chem. A*, 2024, **128**, 3830–3839.
- 25 P. M. Shenai, S. Fernandez-Alberti, W. P. Bricker, S. Tretiak and Y. Zhao, Internal Conversion and Vibrational Energy Redistribution in Chlorophyll *a*, *J. Phys. Chem. B*, 2016, **120**, 49–58.
- 26 J. S. Connolly, F. A. Janzen and E. B. Samuel, Fluorescence lifetimes of chlorophyll *a* : solvent, concentration and oxygen dependence, *Photochem. Photobiol.*, 1982, **36**(5), 559–563.
- 27 D. A. Cherepanov, F. Gostev, I. Shelaev, A. Aibush, A. Semenov, M. Mamedov, V. Shuvalov and V. Nadochenko, Visible and Near Infrared Absorption Spectrum of the Excited Singlet State of Chlorophyll *a*, *High Energy Chem.*, 2020, **54**, 145–147.
- 28 P. Martinsson, J. Oksanen, M. Hilgendorff, P. Hynninen, V. Sundström and E. Akesson, Dynamics of ground and excited state chlorophyll *a* molecules in pyridine solution probed by femtosecond transient absorption spectroscopy, *Chem. Phys. Lett.*, 1999, **309**(5–6), 386–394.
- 29 V. Šebelík, C. D. P. Duffy, E. Keil, T. Polívka and J. Hauer, Understanding Carotenoid Dynamics via the Vibronic Energy Relaxation Approach, *J. Phys. Chem. B*, 2022, **126**(22), 3985–3994.
- 30 V. Balevičius, A. Pour, J. Savolainen, C. Lincoln, V. Lukeš, E. Riedle, L. Valkunas, D. Abramavicius and J. Hauer, Vibronic energy relaxation approach highlighting deactivation pathways in carotenoids, *Phys. Chem. Chem. Phys.*, 2015, **17**, 19491–19499.
- 31 V. Balevičius, C. Lincoln, D. Viola, G. Cerullo, J. Hauer and D. Abramavicius, Effects of tunable excitation in carotenoids explained by the vibrational energy relaxation approach, *Photosynth. Res.*, 2018, **135**(1–3), 55–64.
- 32 S. M. Petrovic, J. B. Zvezdanović and D. Z. Marković, The identification of chlorophyll and its derivatives in the pigment mixtures: HPLC-chromatography, visible and mass spectroscopy studies, *Adv. Technol.*, 2012, **1**, 16–24.
- 33 E. Keil, P. Malevich and J. Hauer, Achromatic frequency doubling of supercontinuum pulses for transient absorption spectroscopy, *Opt. Express*, 2021, **29**, 39042.
- 34 A. Kumar, P. Malevich, L. Mewes, S. Wu, J. Barham and J. Hauer, Transient absorption spectroscopy based on uncompressed hollow core fiber white light proves pre-association between a radical ion photocatalyst and substrate, *J. Chem. Phys.*, 2023, **158**, 144201.
- 35 J. J. Snellenburg, S. P. Laptinok, R. Seger, K. M. Mullen and I. H. M. Van Stokkum, Glotaran: A Java-Based Graphical User Interface for the R Package TIMP, *J. Stat. Software*, 2012, **49**(3), 1–22.
- 36 I. H. M. Van Stokkum, D. S. Larsen and R. Van Grondelle, Global and target analysis of time-resolved spectra, *Biochim. Biophys. Acta, Bioenerg.*, 2004, **1657**(2–3), 82–104.
- 37 T. Shiozaki, W. Györfy, P. Celani and H.-J. Werner, Communication: Extended multi-state complete active space second-order perturbation theory: Energy and nuclear gradients, *J. Chem. Phys.*, 2011, **135**, 081106.



- 38 A. A. Granovsky, Extended multi-configuration quasi-degenerate perturbation theory: The new approach to multi-state multi-reference perturbation theory, *J. Chem. Phys.*, 2011, **134**, 214113.
- 39 J. Finley, P.-Å. Malmqvist, B. O. Roos and L. Serrano-Andrés, The multi-state CASPT2 method, *Chem. Phys. Lett.*, 1998, **288**(2–4), 299–306.
- 40 B. O. Roos, V. Veryazov and P.-O. Widmark, Relativistic atomic natural orbital type basis sets for the alkaline and alkaline-earth atoms applied to the ground-state potentials for the corresponding dimers, *Theor. Chem. Acc.*, 2004, **111**, 345–351.
- 41 B. O. Roos, R. Lindh, P.-Å. Malmqvist, V. Veryazov and P.-O. Widmark, Main Group Atoms and Dimers Studied with a New Relativistic ANO Basis Set, *J. Phys. Chem. A*, 2004, **108**(15), 2851–2858.
- 42 P.-O. Widmark, P.-Å. Malmqvist and B. O. Roos, Density matrix averaged atomic natural orbital (ANO) basis sets for correlated molecular wave functions, *Theor. Chim. Acta*, 1990, **77**, 291–306.
- 43 T. Yanai, D. P. Tew and N. C. Handy, A new hybrid exchange–correlation functional using the Coulomb-attenuating method (CAM-B3LYP), *Chem. Phys. Lett.*, 2004, **393**, 51–57.
- 44 G. Ghigo, B. O. Roos and P.-Å. Malmqvist, A modified definition of the zeroth-order Hamiltonian in multiconfigurational perturbation theory (CASPT2), *Chem. Phys. Lett.*, 2004, **396**, 142–149.
- 45 N. Forsberg and P.-Å. Malmqvist, Multiconfiguration perturbation theory with imaginary level shift, *Chem. Phys. Lett.*, 1997, **274**, 196–204.
- 46 G. Li Manni, *et al.*, The OpenMolcas Web: A Community-Driven Approach to Advancing Computational Chemistry, *J. Chem. Theory Comput.*, 2023, **19**, 6933–6991.
- 47 F. Aquilante, *et al.*, Modern quantum chemistry with [Open] Molcas, *J. Chem. Phys.*, 2020, **152**, 214117.
- 48 I. Fernandez-Galván, *et al.*, OpenMolcas: From Source Code to Insight, *J. Chem. Theory Comput.*, 2019, **15**(11), 5925–5964.
- 49 M. Kowalewski and R. de Vivie-Riedle, *QDng: A Grid Based Molecular Quantum Dynamics Package (1.0.0)*, Deposited 8 April 2024, Zenodo, DOI: [10.5281/zenodo.10944497](https://doi.org/10.5281/zenodo.10944497).
- 50 M. J. Frisch, G. W. Trucks, H. B. Schlegel *et al.*, *Gaussian 16 Rev. C.02*, Wallingford, CT, 2016.
- 51 R. Krishnan, J. S. Binkley, R. Seeger and J. A. Pople, Self-consistent molecular orbital methods. XX. A basis set for correlated wave functions, *J. Chem. Phys.*, 1980, **72**, 650–654.
- 52 A. D. McLean and G. S. Chandler, Contracted Gaussian basis sets for molecular calculations. I. Second row atoms,  $Z = 11$ –18, *J. Chem. Phys.*, 1980, **72**, 5639–5648.
- 53 M. M. Francl, W. J. Pietro, W. J. Hehre, S. Binkley, M. S. Gordon, D. J. DeFrees and J. A. Pople, Self-consistent molecular orbital methods. XXIII. A polarization-type basis set for second-row elements, *J. Chem. Phys.*, 1982, **77**, 3654–3665.
- 54 J. Tomasi, B. Mennucci and R. Cammi, Quantum Mechanical Continuum Solvation Models, *Chem. Rev.*, 2005, **105**, 2999–3094.
- 55 G. Scalmani and M. J. Frisch, Continuous surface charge polarizable continuum models of solvation. I. General formalism, *J. Chem. Phys.*, 2010, **132**, 114110.
- 56 F. Neese, The ORCA program system, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2012, **2**, 73–78.
- 57 F. Neese, F. Wennmohs, U. Becker and C. Riplinger, The ORCA quantum chemistry program package, *J. Chem. Phys.*, 2020, **152**, 224108.
- 58 F. Neese, Software update: The ORCA program system—Version 5.0, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2022, **12**, e1606.
- 59 M. Casanova-Páez, M. B. Dardis and L. Goerigk,  $\omega$ B2PLYP and  $\omega$ B2GPPLYP: The First Two Double-Hybrid Density Functionals with Long-Range Correction Optimized for Excitation Energies, *J. Chem. Theory Comput.*, 2021, **17**, 5165–5186.
- 60 S. Hirata and M. Head-Gordon, Time-dependent density functional theory within the Tamm-Dancoff approximation, *Chem. Phys. Lett.*, 1999, **314**, 291–299.
- 61 F. Weigend and R. Ahlrichs, Balanced basis sets of split valence, triple zeta valence and quadruple zeta valence quality for H to Rn: Design and assessment of accuracy, *Phys. Chem. Chem. Phys.*, 2005, **7**, 3297–3305.
- 62 D. Rappoport and F. Furche, Property-optimized Gaussian basis sets for molecular response calculations, *J. Chem. Phys.*, 2010, **133**, 134105.
- 63 F. Neese, F. Wennmohs, A. Hansen and U. Becker, Efficient, approximate and parallel Hartree-Fock and hybrid DFT calculations. A ‘chain-of-spheres’ algorithm for the Hartree-Fock exchange, *Chem. Phys.*, 2009, **356**, 98–109.
- 64 R. Izsák and F. Neese, An overlap fitted chain of spheres exchange method, *J. Chem. Phys.*, 2011, **135**, 144105.
- 65 F. Weigend, Accurate Coulomb-fitting basis sets for H to Rn, *Phys. Chem. Chem. Phys.*, 2006, **8**, 1057–1065.
- 66 A. Hellweg, C. Hättig, S. Höfener and W. Klopper, Optimized accurate auxiliary basis sets for RI-MP2 and RI-CC2 calculations for the atoms Rb to Rn, *Theor. Chem. Acc.*, 2007, **117**, 587–597.
- 67 R. Cammi, B. Mennucci and J. Tomasi, Fast Evaluation of Geometries and Properties of Excited Molecules in Solution: A Tamm-Dancoff Model with Application to 4-Dimethylaminobenzonitrile, *J. Phys. Chem. A*, 2000, **104**, 5631–5637.
- 68 M. Cossi, N. Rega, G. Scalmani and V. Barone, Energies, structures, and electronic properties of molecules in solution with the C-PCM solvation model, *J. Comput. Chem.*, 2003, **24**, 669–681.

