# ChemComm

# **Accepted Manuscript**





This is an *Accepted Manuscript*, which has been through the RSC Publishing peer review process and has been accepted for publication.

*Accepted Manuscripts* are published online shortly after acceptance, which is prior to technical editing, formatting and proof reading. This free service from RSC Publishing allows authors to make their results available to the community, in citable form, before publication of the edited article. This *Accepted Manuscript* will be replaced by the edited and formatted *Advance Article* as soon as this is available.

To cite this manuscript please use its permanent Digital Object Identifier (DOI®), which is identical for all formats of publication.

More information about *Accepted Manuscripts* can be found in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics contained in the manuscript submitted by the author(s) which may alter content, and that the standard **Terms & Conditions** and the **ethical guidelines** that apply to the journal are still applicable. In no event shall the RSC be held responsible for any errors or omissions in these *Accepted Manuscript* manuscripts or any consequences arising from the use of any information contained in them.

## **RSC**Publishing

www.rsc.org/chemcomm

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxx

## **ARTICLE TYPE**

### Catalytic asymmetric synthesis of 1,3-enyne scaffolds: design and synthesis of conjugated nitro dienynes as novel Michael acceptors and development of new synthetic methodology

Xiang Li, Fangzhi Peng,\* Mingtao Zhou, Mingjie Mo, Ruirui Zhao, and Zhihui Shao\*

s Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX DOI: 10.1039/b000000x

A novel catalytic enantioselective synthesis of functionalized 1,3-enynes have been developed, featuring the design and synthesis of conjugated nitro dienynes as a useful new class of <sup>10</sup> Michael acceptors. Moreover, a simple, yet flexible catalytic cascade approach to functionalized enantioenriched acvelic

α,β-enones and cyclic dienones have also been developed.

1,3-Enynes are important structural motifs present in many naturally occurring and drugs<sup>1</sup> such as terbinafine,<sup>2</sup> a potent drug <sup>15</sup> for superficial fungal infections, and calichemicin  $\gamma 1$ ,<sup>3</sup> an effective antitumor antibiotic. 1,3-Enynes are also versatile building blocks in organic chemistry<sup>4</sup> and material science.<sup>5</sup> Therefore, considerable efforts have been devoted to develop new methods to synthesize 1,3-enynes.<sup>6,7</sup> Whereas several methods

- <sup>20</sup> have been developed for their racemic synthesis,<sup>6</sup> the asymmetric synthesis, especially catalytic asymmetric synthesis, of 1,3enynes still remains a challenging task. Recently, Krische group reported elegant Rh-catalyzed enantioselective reductive coupling of 1,3-diynes to  $\alpha$ -keto aldehydes and  $\alpha$ -hydroxy esters.<sup>7</sup>
- <sup>25</sup> However, these procedures are only applicable to the synthesis of enantioenriched 1,3-enynes with an allylic stereogenic center. To date, a general, highly enantioselective method for the synthesis of 1,3-enynes with a propargylic stereogenic center has not been achieved.
- <sup>30</sup> As part of our continuing interest in the catalytic asymmetric syntheses of chiral unsaturated structural motifs and building blocks,<sup>8</sup> herein we report a distinct approach toward the catalytic enantioselective synthesis of 1,3-enynes. This approach featuring for the first time the design and synthesis of conjugated nitro
- $_{35}$  dienynes as a useful new class of Michael acceptors, provides previously inaccessible enantioenriched 1,3-enynes with a propargylic stereogenic center in good yields with high enantioselectivity. Notably, We have further developed a simple, yet flexible novel catalytic cascade approach to functionalized 40 enantioenriched acyclic  $\alpha$ , $\beta$ -enones and cyclic dienones, two





conjugated nitro dienynes



Michael acceptors.

45 To target the synthesis of challenging 1,3-envnes and tackle an issue of broader significance in organic synthesis, conjugated nitro dienynes 1 (Fig. 1) were designed. Whereas nitroolefins as Michael acceptors have been extensively investigated,<sup>9</sup> the closely related conjugated nitro dienynes 1, have not been 50 designed and synthesized, despite their potential in synthetic chemistry. Thus we began our studies by developing a method for the synthesis of nitro dienvnes 1. As shown in Scheme 1, vinyl bromides 2 underwent Sonogashira coupling with propargylic alcohol, leading to the corresponding alcohols 3 in high yields. 55 Oxidation of the hydroxyl group of **3** with 2-iodoxybenzoic acid (IBX) afforded aldehydes 4. 4 underwent Henry reaction with LiAlH<sub>4</sub> as the base<sup>10</sup> followed by reaction with trifluoroacetic anhydride (TFAA) in the presence of Et<sub>3</sub>N produced the desired nitro dienynes 1. Notably, a variety of aromatic and aliphatic 60 nitro dienynes 1a-g could be easily prepared with complete



Scheme 1 Synthesis of conjugated nitro dienynes 1.

<sup>55</sup> With conjugated nitro dienynes **1** in hand, we focused our attention on developing a method for the catalytic enantioselective synthesis of 1,3-enynes, especially with a

This journal is  $\ensuremath{\mathbb{C}}$  The Royal Society of Chemistry [year]

propargylic stereogenic center. We envisioned that if nitro dienynes 1 could act as novel extended Michael acceptors,<sup>11</sup> a catalytic enantioselective 1,4-addition (not 1,6- or 1,8-addition)<sup>12</sup> to 1 would provide a route to previously inaccessible 1,3-enynes.

- <sup>5</sup> To explore the feasibility of this method, the model reaction between nitro dienyne **1a** and di-*tert*-butyl malonate **5a** was investigated. Screening of chiral catalysts indicated that bifunctional thioureas, which have been proved to be effective in the Michael addition of malonates to nitroolefins,<sup>13</sup> did not
- <sup>10</sup> promote the model reaction. We then turned our attention to chiral metal catalysts. Recently, Huang and co-workers introduced a novel chiral diamine and the corresponding Ni (II) complex. This complex effectively catalyzed the enantioselective Michael addition of di-*tert*-butyl malonate to nitroolefins.<sup>14</sup>
- <sup>15</sup> Therefore, we applied this complex to conjugated nitro dienyne substrates. Huang's 1:1 NiBr<sub>2</sub>–L1 complex<sup>14</sup> promoted the reaction to afford the desired product **6a** in moderate yield (56%) with moderate enantioselectivity (66% ee) (Table 1, entry 1). These results highlighted the difference in reactivity of our nitro
- <sup>20</sup> dienyne substrates relative to known nitroolefins. Pleasingly, Evans-type chiral Ni (II)-diamine complexes,<sup>15</sup> which have been used extensively in asymmetric catalysis,<sup>16</sup> efficiently catalyzed the model reaction to provide the product **6a** in good yields with high enantioselectivity (entries 2-5). The use of 1:2 NiBr<sub>2</sub>–L3 <sup>25</sup> complex afforded 90% yield and 93% ee (entry 3). Finally,
- screening of solvents showed that *m*-xylene was still optimal.<sup>17</sup>

 Table 1 Optimization of the reaction conditions<sup>a</sup>

Ph	1a	≫NO <sub>2</sub> + 〈	CO <sub>2</sub> t-Bu NiBr <sub>2</sub> + CO <sub>2</sub> t-Bu <i>m</i> -xyler 5a rt	t-BuO <sub>2</sub> C	CO <sub>2</sub> t-Bu
		Ph	Ph Ph Ph Ph Ph L2	Ar L3: Ar = 4-Br-C L4: Ar = 4-MeC L5: Ar = $C_6H_5$	Ar 6H4 D-C6H4
Entry	L	$NiBr_2$ -L	Time	$\operatorname{Yield}^{b}(\%)$	$ee^{c}$ (%)
$1^d$	L1	1:1	24	56	66
2	L2	1:2	15	86	-95
3	L3	1:2	40	90	93
4	L4	1:2	40	87	91
5	L5	1:2	40	87	92

<sup>30</sup> <sup>a</sup> All reactions were performed at the 0.2 mmol scale using 1.5 equiv of 5a with 5 mol % of catalyst. <sup>b</sup> Isolated yields. <sup>c</sup> Determined by chiral HPLC. <sup>d</sup> At 80 °C and 5 mol % PMP was used.

The present reaction serves as a general method for the enantioselective synthesis of functionalized 1,3-enynes. As <sup>35</sup> shown in Table 2, a range of electron-rich and electron-deficient aryl-substituted nitro dienynes **1b-e** proceeded with various 1,3-dicarbonyl compounds smoothly to provide the desired 1,3-enynes in good yields (88-95%) with high enantioselectivity (92-93% ee). Heteroaromatic substituted nitro dienyne **1f** also <sup>40</sup> provided good results (entries 6 and 16). Notably, alkyl-substituted nitro dienynes **1g** was also suitable for this reaction (entries 7 and 10). It is noteworthy that no 1,6- or 1,8-addition was observed in all cases.

 Table 2 Synthesis of various enantioenriched 1,3-enynes<sup>a</sup>

~	NO <sub>2</sub>	COR <sup>2</sup>	NiBr <sub>2</sub> (5 mol %) L <b>3</b> (10 mol %)	R <sup>2</sup> OC	COR3
R <sup>1</sup>	+	COR <sup>3</sup>	<i>m</i> -xylene, rt 40-60 h R	1 6a-p	NO <sub>2</sub>
Entry	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Yield <sup>b</sup> (%)	$ee^{c}$ (%)
1	Ph	Ot-Bu	Ot-Bu	<b>6a</b> , 90	93
2	4-Me-C <sub>6</sub> H <sub>4</sub>	Ot-Bu	Ot-Bu	<b>6b</b> , 93	93
3	3-Me-C <sub>6</sub> H <sub>4</sub>	Ot-Bu	Ot-Bu	6c, 90	93
4	4-MeO-C <sub>6</sub> H <sub>4</sub>	Ot-Bu	Ot-Bu	6d, 92	93
5	4-Cl-C <sub>6</sub> H <sub>4</sub>	Ot-Bu	Ot-Bu	6e, 88	92
6	2-thienyl	Ot-Bu	Ot-Bu	<b>6f</b> , 90	92
7	Me	Ot-Bu	Ot-Bu	6g, 85	94
8	Ph	Me	Me	<b>6h</b> , 90	90
9	Ph	OEt	OEt	6i, 92	93
10	Me	OEt	OEt	<b>6j</b> , 87	90
$11^{d}$	Ph	Me	Ot-Bu	<b>6k</b> , 94	93 (93) <sup>e</sup>
$12^{d}$	4-Me-C <sub>6</sub> H <sub>4</sub>	Me	Ot-Bu	<b>61</b> , 93	$93(93)^{e}$
$13^{d}$	3-Me-C <sub>6</sub> H <sub>4</sub>	Me	Ot-Bu	6m, 92	$92(92)^{e}$
$14^d$	4-MeO-C <sub>6</sub> H <sub>4</sub>	Me	Ot-Bu	<b>6n</b> , 95	$93 (93)^{e}$
$15^{d}$	4-Cl-C <sub>6</sub> H <sub>4</sub>	Me	Ot-Bu	<b>60</b> , 90	$92(92)^{e}$
$16^{d}$	2-thienyl	Me	Ot-Bu	<b>6p</b> , 90	$93 (93)^{e}$

<sup>*a*</sup> All reactions were performed at the 0.2 mmol scale using 1.5 equiv of **5**. <sup>*b*</sup> Isolated yields. <sup>*c*</sup> Determined by chiral HPLC. <sup>*d*</sup> 1:1 diastereomeric mixture; determined by <sup>1</sup>H-NMR. <sup>*e*</sup> Enantiomeric excess of the diastereomer.

To determine the absolute configuration, the 1,3-enyne **6j** was <sup>50</sup> converted to the compound **7** (Eq. 1). The (*S*) configuration was established by comparison of the optical rotation of **7** with previously reported value of this compound.<sup>18</sup>

 $\begin{array}{c} EtO_2C \\ \hline \\ \hline \\ Me \end{array} \begin{array}{c} CO_2Et \\ \hline \\ \hline \\ OD_2 \\ \hline \\ OD_2 \\ \hline \\ EtOAc, rt \\ (86\%) \end{array} \begin{array}{c} EtO_2C \\ \hline \\ Me \\ \hline \\ OD_2 \\ \hline \hline \\ OD_2 \\ \hline \\$ 

To further demonstrate the utility of enantioenriched 1,3ss enynes **6**, these compounds were treated with TsOH for the hydrolysis/decarboxylation reaction. Interestingly, reaction of 1,3-enyne **6a** and **6k** with a catalytic amount of TsOH (20 mol %) did not afford the hydrolysis/decarboxylation products **8a'** and **10a'** but instead directly led to acyclic  $\alpha,\beta$ -enone **8a** (Eq. 2) and cocyclic dienone **10a** (Eq. 3) in good yields almost without loss of enantiomeric purity. The enantioselectivity of **8a** was determined via transformation into the corresponding ester **9a**.



We were excited with these unexpected results as the obtained acyclic  $\alpha,\beta$ -enones and cyclic dienones are synthetically useful but inaccessible compounds. On the other hand, the incorporation of 1,3-enynes leading to the completely regioselective formation of synthetically useful enones is also noteworthy, as there are no reports of TsOH-catalyzed hydration reactions of 1,3-enynes.<sup>19</sup> <sup>70</sup> Only activated aryl alkynes have been previously shown to undergo TsOH-catalyzed hydration to afford the corresponding ketones (non-activated aryl alkynes: 0–9% yields).<sup>20</sup>

In view of the importance of acyclic  $\alpha,\beta$ -enones and cyclic

2

50

dienones, we investigated the scope of the cascade reactions. As shown in Table 3, the scope of the reactions was remarkable. Various functionalized acyclic  $\alpha,\beta$ -enones **9a-f** and cyclic dienones **10a-f** were obtained in good yields with high enantioselectivity. While several methods have been developed

- for the syntheses of acyclic  $\alpha,\beta$ -enones<sup>21</sup> and cyclic dienones,<sup>22</sup> respectively, these reported protocols are not applicable to the synthesis of our acyclic  $\alpha,\beta$ -enones 9 and cyclic dienones 10. Functionalized acyclic  $\alpha,\beta$ -enones 9 represent a novel class of
- <sup>10</sup> multifunctional chiral synthons as they contains enone, ester and nitro groups. Due to the vinylogous reactivity, cyclic dienones have been demonstrated as a useful class of substrates in the enantioselective transition-metal catalyzed<sup>23</sup> and organocatalyzed<sup>24</sup> reactions. However, the reported asymmetric
- 15 reactions used unfunctionalized racemic cyclic dienone substrates.

**Table 3** Synthesis of various functionalized enantioenriched acyclic  $\alpha,\beta$ -enones 9 and cyclic dienones 10<sup>*a*</sup>



<sup>*a*</sup> All reactions were performed with **6** (0.2 mmol) and 20 mol % TsOH under <sup>20</sup> reflux. <sup>*b*</sup> Isolated yield of **8**. <sup>*c*</sup> Isolated yield of **9**. <sup>*d*</sup> Determined by chiral HPLC.

#### Conclusions

In conclusion, we have developed a novel catalytic enantioselective synthesis of previously inaccessible 1,3-enynes in good yields with high enantioselectivity. This approach <sup>25</sup> features for the first time the design and synthesis of conjugated nitro dienynes as a useful new class of Michael acceptors. Moreover, we have further developed a simple, yet flexible catalytic cascade approach to synthetically useful functionalized enantioenriched acyclic  $\alpha$ , $\beta$ -enones and cyclic dienones.

We acknowledge financial support from the NSFC (21162034, 21372193, 21362040), Doctoral Fund of Ministry of Education of China (20135301110002), and the Government of Yunnan Province (2012FB114, 2013FA026).

#### Notes and references

- <sup>35</sup> <sup>a</sup> Key Laboratory of Medicinal Chemistry for Natural Resource, Ministry of Education, School of Chemical Science and Technology, Yunnan University Kunning 650091, China. Fax: (+86)871-65035538; E-mail: pengfangzhi@ynu.edu.cn (F. Peng); zhihui\_shao@hotmail.com (Z. Shao) † Electronic Supplementary Information (ESI) available: Representative experimental procedure, compound characterization data, and copies of
- spectra. See DOI: 10.1039/b000000x/ 1 (a) T. Akiyama, K. Takada, T. Oikawa, N. Matsuura, Y. Ise, S.
- (a) T. Ariyana, K. Takada, T. Okawa, N. Malsudia, T. Ise, S. Okada and S. Matsunaga, *Tetrahedron*, 2013, **69**, 6560; (b) N. El-Jaber, A. Estevez-Braun, A. G. Ravelo, O. Munoz-Munoz, A. Rodriguez-Afonso and J. R. Murguia, *J. Nat. Prod.*, 2003, **66**, 722;

(c) L. Garlaschelli, E. Magistrali, G. Vidari and O. Zuffardi, *Tetrahedron Lett.*, 1995, **36**, 5633.

- S. L. Iverson and J. P. Uetrecht, Chem. Res. Toxicol., 2001, 14, 175.
- 3 N. Zein, A. M. Sinha, W. J. McGahren and G. A. Ellestad, *Science*, 1988, **240**, 1198.
- 4 (a) W. Zhang, S. Zheng, N. Liu, J. B. Werness, I. A. Guzei and W. Tang, J. Am. Chem. Soc., 2010, **132**, 3664; (b) J. B. Werness and W. Tang, Org. Lett., 2011, **13**, 3664; (c) F. Liu, D. Qian, L. Li, X. Zhao and J. Zhang, Angew. Chem. Int. Ed., 2010, **49**, 6669.
- 55 5 K. Campbell, C. J. Kuehl, M. J. Ferguson, P. J. Stang and R. R. Tykwinski, J. Am. Chem. Soc., 2002, 124, 7266.
- 6 Selected examples: (a) D. Cheng, F. Ling, Z. Li, W. Yao and C. Ma, *Org. Lett.*, 2012, 14, 3146; (b) B. Alcaide, P. Almendros and T. M. del Campo, *Org. Biomol. Chem.*, 2012, 10, 7603; (c) W. Zhang, M. Nishiura and Z. Hou, *Angew. Chem. Int. Ed.*, 2008, 47, 9700; (d) I.
  - Nishiura and Z. Hou, Angew. Chem. Int. Ed., 2008, 47, 9700; (d) I.
     M. Lyapkalo and M. A. K. Vogel, Angew. Chem. Int. Ed., 2006, 45, 4019.
- 7 (a) R. R. Huddleston, H.-Y. Jang and M. J. Krische, *J. Am. Chem. Soc.*, 2003, **125**, 11488; (b) C.-W. Cho and M. J. Krische, *Org. Lett.*,
   <sup>65</sup> 2006, **8**, 3873.
- 8 (a) Z. Li, S. Zhang, S. Wu, X. Shen, L. Zou, F. Wang, X. Li, F. Peng, H. Zhang and Z. Shao, *Angew. Chem. Int. Ed.*, 2013, **52**, 4117; (b) B. Fan, X. Li, F. Peng, H. Zhang, A. S. C. Chan and Z. Shao, *Org. Lett.*, 2010, **12**, 304; (c) X. Li, X. Li, F. Peng and Z. Shao, *Adv. Synth. Catal.* 2012, **354**, 2873.
- 9 For a review: O. M. Berner, L. Tedeschi and D. Enders, *Eur. J. Org. Chem.*, 2002, 1877.
- 10 S. Belot, A. Massaro, A. Tenti, A. Mordini and A. Alexakis, Org. Lett., 2008, 10, 4557.
- 75 11 For a review on extended Michael acceptors: A. G. Csákÿ, G. de la Herrán and M. C. Murcia, *Chem. Soc. Rev.*, 2010, **39**, 4080.
- 12 For a seminal discussion on the regioselectivity of conjugate addition to extended Michael acceptors: N. Krause and S. Tho-rand, *Inorg. Chim. Acta*, 1999, **296**, 1.
- 80 13 (a) T. Okino, Y. Hoashi, T. Furukawa, X. Xu and Y. Takemoto, J. Am. Chem. Soc., 2005, **127**, 119; (b) J. Ye, D. J. Dixon and P. S. Hynes, Chem. Commun., 2005, 4481.
  - 14 Q. Zhu, H. Huang, D. Shi, Z. Shen and C. Xia, Org. Lett., 2009, 11, 4536.
- 85 15 D. A. Evans, S. Mito and D. Seidel, J. Am. Chem. Soc., 2005, 127, 9958.
- Selected examples: (a) A. Nakamura, S. Lectard, D. Hashizume, Y. Hamashima and M. Sodeoka, J. Am. Chem. Soc., 2010, 132, 4036;
  (b) C. Fallan and H. W. Lam, Chem. Eur. J., 2012, 18, 11214; (c) J. Chen, J. Chen, Y. Xie and H. Zhang, Angew. Chem. Int. Ed., 2012, 51, 1024.
  - 17 DCM: 86% yield, 86% ee; Toluene: 75% yield, 78% ee.
  - 18 D. M. Barnes, J. Ji, M. G. Fickes, M. A. Fitzgerald, S. A. King, H. E. Morton, F. A. Plagge, M. Preskill, S. H. Wagaw, S. J. Wittenberger and J. Zhang, J. Am. Chem. Soc., 2002, **124**, 13097.
- For a review on catalytic hydration of alkynes: L. Hintermann and A. Labonne, *Synthesis*, 2007, 1121.
- (a) N. Olivi, E. Thomas, J. Peyrat, M. Alami and J. Brion, *Synlett*, 2004, 2175; (b) G. Le Bras, L. Provot, J. Peyrat, M. Alami and J.
   Brion, *Tetrahedron Lett.*, 2006, 47, 5497.
  - 21 (a) R. G. Kelleher, M. A. McKervey and P. Vibuljan, J. Chem. Soc., Chem. Commun., 1980, 486; (b) K. Tanaka, T. Shoji and M. Hirano, Eur. J. Org. Chem., 2007, 2687; (c) M. Lemhadri, Y. Fall, M. Santelli and H. Doucet, Synthesis, 2009, 1021.
- 105 22 (a) E. Wenkert and M. K. Schorp, *J. Org. Chem.*, 1994, **59**, 1943; (b) X. Fu, S. Zhang, J. Yin, T. L. McAllister, S. A. Jiang, C.-H. Tann, T. K. Thiruvengadam and F. Zhang, *Tetrahedron Lett.*, 2002, **43**, 573; (b) M. Tissot, D. Poggiali, H. Henon, D. Mueller, A. Alexakis, L. Guenee and M. Mauduit, *Chem. Eur. J.*, 2012, **18**, 8731.
- 110 23 (a) H. Hénon, M. Mauduit and A. Alexakis, *Angew. Chem. Int. Ed.*, 2008, **47**, 9122; (b) K.-s. Lee and A. H. Hoveyda, *J. Am. Chem. Soc.*, 2010, **132**, 2898.
- (a) X. Tian, Y.-K. Liu and P. Melchiorre, *Angew. Chem. Int. Ed.*, 2012, **51**, 6439; (b) X. Tian and P. Melchiorre, *Angew. Chem. Int. Ed.*, 2013, **52**, 5360.

This journal is © The Royal Society of Chemistry [year]