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A novel platform for detection of protooncogene based on Au nanoclusters enhanced fluorescence

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ABSTRACT

For the first time, gold nanoclusters was found to exhibit high fluorescence enhancement ability based on metal-enhanced fluorescence (MEF) effect, which can effectively enhance the fluorescence of fluorescein isothiocyanate (FITC). By means of this phenomenon, Au nanoclusters have been successfully used in the construction of a fluorescence-enhanced sensing platform for the detection of protooncogene.

Keywords: Au nanoclusters (AuNCs); fluorescein isothiocyanate (FITC); fluorescence enhancement; protooncogene

1. Introduction

Fluorescence-based detection technology, in combination with nanotechnology, has been widely used in biological fields.¹⁻⁴ A number of fluorescent nanomaterials, including quantum dots,¹ up conversion nanoparticles² and dye-doped nanoparticles,³ are of considerable interest as sensors and indicators for biological detections due to their high fluorescence intensity, stability, and easy modification. To further increase fluorescence intensities of nanomaterial-based probes, metal nanostructures and dye molecules have been combined to fabricate metal-dye nanocomposites. It has been proven that fluorescence can be enhanced by several folds when the metal nanostructures and dye molecules are kept within a certain distance, but weakened or quenched otherwise.⁵⁻⁷ Such a fluorescence enhancement phenomenon is well-known as the metal-enhanced fluorescence (MEF) effect. The mechanisms of MEF have been extensively studied and several pathways of metal enhancement have been proposed.⁸⁻¹⁴ First of all, the occurrence of surface plasmon resonance (SPR) leads to a strongly enhanced absorption of the incident light.¹⁴⁻¹⁶ In fact, surface plasmon resonance has been widely used in explaining some interesting phenomenon such as enhancement of the photoluminescence,^{15,16} plasmon-controlled Förster Resonance Energy Transfer (FRET),¹⁷ enhanced energy transfer between quantum dots (QDs) and nanoparticles,¹⁸⁻²⁰ metal-enhanced surface plasmon-coupled phosphorescence,²¹ surface-enhanced Raman scattering (SERS)²² and plasmon-enhanced surface catalysis.²³ Considering the fluorescence enhancement, when the surface plasmon resonance band of a metal nanostructure overlaps the excitation of the fluorophore, the energy is transferred from the metal to the fluorophores so that the possibility of excitation of the dye molecules is increased. Secondly, the metal nanostructure can change the radiative deactivation rate of the fluorophores. Thus, the fluorescence lifetime and the quantum yield are changed.²⁴ Thirdly, the scattering of the metallic nanostructures affects the coupling efficiency of the fluorescence emission to the far field.^{25,26} By regulating the plasmon resonance band to the fluorophore emission wavelength, a fluorescence enhancement can be obtained. Also, the rapid development of nanotechnology has produced various new types of metallic nanostructures, which generates a number of new effects of metallic nanomaterials on dye molecular properties, such as the dimensional effects of metallic nanostructures on fluorescence intensity.²⁷

The fluorescence of metal nanoclusters has drawn continuous research interest in the fields of chemistry, biology and materials.^{4, 28-32} There has been a great deal of research work on the fluorescence of metal nanoclusters, especially gold and silver.^{29, 31, 33, 34} Owing to their ultrasmall size, biocompatibility and highly fluorescent properties, fluorescent gold nanoclusters (AuNCs) have become an attractive field of study. Up to now, the applications of fluorescent AuNCs in analysis are mainly based on the fluorescence quenching effect^{33, 35} of AuNCs through the interaction between gold and the limited analytes like Fe³⁺,³⁶ Hg²⁺,^{37, 38} Cu²⁺,^{39, 40} and cysteine.⁴¹ Also, AuNCs is as novel optical probes for in vitro and in vivo fluorescence imaging.^{29, 42} As we know, the fluorescence of AuNCs correlates not only with the metal quantization effect but also with the surface ligands or scaffolds.⁴³ How to rationally design fluorescent AuNCs with functional ligands or scaffolds to give them broad applications remains to be explored.

According to the mechanisms of metal enhancement proposed above, for the first time, we report a simple and homogeneous assay format for protooncogene by using BSA-protected gold nanoclusters-based enhanced fluorogenic nanoprobe. Except high selectivity and sensitivity, this mix-and-detect assay format is simple. Importantly, the assay is homogeneous because it occurs exclusively in the liquid phase, which makes it easy to automate or suitable for in situ detection.

2. Materials and methods

2.1. Materials

All oligonucleotides (P1: 5'-FITC-CGCTCCAGAGGCAGTAACCAGAGCGTTTTTTTTTTT-(CH₂)₆-SH-3'; T1: 5'-TGGTTACCGCCTCTG-3'; ssDNA1: 5'-TGGTTACTGCCTCTG-3'; ssDNA2: 5'-CCAACCTGTCTTTCCTACG-3'; ssDNA3: 5'-AACCTGTCTTTCCTACG-3'; ssDNA4: 5'-CCTGTCTTTCCTACG-3') were purchased from Sangon Biotech (Shanghai, China) Co. Ltd. and purified by HPLC. The deionized water was purified using a Millipore filtration system (18.2 MΩ resistivity) and used in all experiments. All the experiments were carried out in phosphate buffer saline (PBS) buffer (0.1 mM PBS, 0.1 M NaCl, pH 8.00).

2.2. Instrumentation

The size distribution and structure of the AuNCs were probed by high-resolution transmission electron microscopy (HRTEM) using a JEM-2100 (HR) operated at an acceleration voltage of 200KV. Fluorescence spectrometer (F-4600, Hitachi Co. Ltd., Japan) with a Xenon lamp excitation source was employed to record fluorescence spectra. The excitation was set at 490 nm and the emission was monitored at 520 nm.

2.3. Synthesis of red fluorescent AuNCs

According to literature,⁴⁴ all glassware was washed with aqua regia (HCl:HNO₃ volume ratio=3:1) and rinsed with ethanol and ultrapure water (Caution: Aqua Regia is a very corrosive oxidizing agent, which should be handled with great care). In atypical experiment, aqueous HAuCl₄ solution (5 mL, 10 mM, 37°C) was added to BSA solution (5 mL, 50 mg/mL, 37 °C) under vigorous stirring. NaOH solution (0.5 mL, 1 M) was introduced 2 min later and the reaction was allowed to proceed under vigorous stirring at 37°C for 12 h. The obtained suspension was centrifuged and suspended with water before further characterizations and applications.

2.4. Fluorescence spectra measurements

The fluorescent probe P1 (10 nM) was hybridized with different amounts of T1 for 30 min in a PBS buffer solution of 195 μL, prior to the addition of the AuNCs solution (5 μL). The final

concentration of T1 ranged from 0 to 60 nM. For kinetic study of fluorescence enhancement, fluorescence spectra were recorded immediately after addition of AuNCs into P1 solution. For DNA assays, the fluorescence measurements were performed after incubated with AuNCs for 30 min. Fluorescence spectrometer (F-4600, Hitachi Co. Ltd., Japan) with a Xenon lamp excitation source was employed to record fluorescence spectra. The excitation was set at 490 nm and the emission was monitored at 520 nm.

3. Results and discussion

3.1. Fluorescence enhancement between FITC-tagged ssDNA and Au nanoclusters

The proposed mix-and-detect strategy is depicted in Scheme 1. In our process, two important factors, the adhesion layer and the distance,^{45,47} between the fluorophores and AuNCs, which exert great influence on fluorescence enhancement have been examined. The crucial role of the adhesion layer in plasmonic fluorescence enhancement has been reported by Heykel Aouani's group.⁴⁶ In Rizia Bardhan's report,⁴⁷ Au nanoshells (NSs) and Au nanorods wrapped in human serum albumin (HSA) were used as substrates, which more effectively enhanced the fluorescence compared with pure fluorophores. Like HSA and bovine serum albumin (BSA) were widely used in most studies as stabilizer and reducer to form fluorescent AuNCs. Thus, BSA was chosen in our study as the adhesion layer to protect AuNCs.⁴⁴ As to the influence of the distance between the fluorophores and AuNCs on fluorescence enhancement, the ideal distance for maximum enhancement reported was 5-11 nm.⁴⁷ In our study, DNA techniques were applied to adjust the distance between AuNCs and fluorophores (FITC).

In our strategy, quantitative readout of the target DNA can be realized through the following principle and process. Based on the mechanisms of metal enhancement about surface plasmon resonance enhanced energy transfer, BSA-protected Au nanoclusters can be used as the donor⁴⁸ and FITC is the acceptor. When the distance between AuNCs and FITC controlled by DNA is perfect, the fluorescence of FITC is enhanced strongly. We expect that AuNCs can adsorb dye-labeled single-stranded DNA (ssDNA, the hairpin structure) probe via the Au-S force between SH and the AuNCs and then enhance the fluorescence of the dye.⁴¹ In contrast, when a ssDNA probe is hybridized with its complementary target DNA, because the nucleobases are buried between the densely negatively charged helical phosphate backbones, the dye-labeled probe is away from the surface of AuNCs, resulting in getting rid of the fluorescence enhancement of the probe. As a result, the fluorescence of the probe is expected to provide a quantitative readout of the target DNA. In this work, protooncogene acts as the target DNA.

In our experiments, the AuNCs was synthesized through the reduction of chloroauric acid with bovine serum albumin (BSA) at physiological temperature using recently developed method⁴⁴ (see in Figure S1). The fluorescence enhancement ability of AuNCs toward the FITC-labeled ssDNA was evaluated via measurements upon mixing the fluorescent probe (FITC) and the prepared AuNCs. The FITC-labeled ssDNA probe (P1) used here is for a protooncogene aptamer. In the presence of AuNCs, the fluorescence of P1 was obviously enhanced (see the curve of P1+AuNCs in Figure 1). The enhancement kinetics was very fast, with up to more than 2 times fluorescence intensity obtained within 4 min after P1 was mixed with the AuNCs solution (see the curve of P1+AuNCs in the inset of Figure 1), which suggested that the interaction between FITC and AuNCs is quite strong and the AuNCs possesses a high fluorescence enhancement ability. In

131 addition, temperature from 20°C to 35°C has little effect on fluorescence enhancement of FITC
132 (see in Figure 2).

133 Considering that the fluorescence stability is highly pH dependent, we studied the effects of pH
134 on the fluorescence enhancement. According to the Figure 3, enormous changes of fluorescence of
135 FITC with different pH from 5.5 to 10.0 have happened. According to previous reports, this is
136 caused by the alternation of fluorescein structure between lactone and open loop tautomerism with
137 different pH. Under alkaline conditions, lactone is the main form and two hydroxide radicals will
138 have a certain degree of take off (with weak fluorescence). Under acidic conditions, the main form
139 is open loop (neutral molecules or proton, and weak fluorescence). Thus, when the pH value
140 increases, the fluorescence of FITC gets enhanced. Taking into account the effect of different pH
141 on fluorescence intensity (above) and Hoogsteen base pairing, 8 is the best suitable pH value.

142 When the distance between FITC-labeled ssDNA and AuNCs is too wide, the fluorescence
143 would decrease. It is known that the fluorescence can be significantly enhanced when P1 is
144 mixed with AuNCs. However, when P1 was hybridized with an equal amount of the
145 complementary target DNA T1 to form dsDNA, the P1/T1 duplex, the fluorescence largely
146 decreased (see the curve of P1/T1+AuNCs in Figure 1). This was because the distance between
147 FITC-labeled ssDNA and AuNCs was widened when T1 was added. The result confirmed the
148 influence of the distance between FITC and AuNCs on the fluorescence enhancement intensity.

149 On the basis of the aforementioned findings, by adjusting the distance between FITC and
150 AuNCs and keeping the pH value at the ideal level, a sensing platform for quantitative DNA assay
151 can be built using AuNCs with high fluorescence enhancement ability.

152 **3.2. Assay for protooncogene in aqueous buffer**

153 In the experiment to determine the linear range of our DNA sensor, 10 nM of P1 was hybridized
154 with T1 at various concentrations at room temperature for 30 min, and then the mixture was
155 incubated with an aliquot of AuNCs solution. As the concentration of T1 increased, the percentage
156 of P1 hybridized with T1 to form duplex also increased. As a result, the enhanced fluorescence of
157 P1 decreased (Figure 4a). Note that the fluorescence could still decrease when the concentration of
158 T1 exceeded that of P1. That might result from the absorption of T1 by AuNCs, which stopped
159 part of T1 from being formed into the P1/T1 duplex. On the basis of the Figure 4b, this DNA
160 sensor shows a linear range between 8 and 40 nM ($R^2=0.992$), with a detection limit of 4 nM (3
161 times the standard deviation rule), which is simple and homogeneous.

162 **3.3. Selectivity assays**

163 In addition, control experiments were conducted to confirm that the decreased fluorescence was
164 due to the specific DNA structural switching. Four other types of mismatch ssDNA were
165 systemically studied with the same assay protocol. However, none of the four analogues could
166 induce the distinct fluorescence decrease, even at a very high concentration (10 μ M), as compared
167 to the P1 sample (Figure 5). This result has proven that the described mix-and-detect assay is
168 highly selective toward protooncogene.

169 **4. Conclusion**

170 In conclusion, for the first time, we have revealed that when kept within a certain distance with
171 FITC and when the pH value is suitable, AuNCs possesses high fluorescence enhancement

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2
3 172 efficiency. Inspired by these findings, we employed AuNCs in the construction of a sensing
4 173 platform for the quantitative detection of specific DNA. This mix-and-detect assay format is
5 174 simple. Importantly, the assay is homogeneous because it occurs exclusively in the liquid phase,
6 175 which makes it easy to automate or suitable for in situ detection. In addition, AuNCs can be
7 176 readily synthesized on a large scale and used as efficient fluorescence enhancement nanomaterials
8 177 without further processing. What's worth mentioning is that aptamers in vitro selected nucleic acid
9 178 molecules with high specificity and affinity toward a wide spectrum of targets⁴⁹ are widely
10 179 recognized as promising candidates for biosensing due to their intrinsic advantages.^{50, 51} Besides
11 180 the DNA hybridization, specific aptamer-target recognition can induce dramatic structural
12 181 switching of the DNA probe.⁵² Combined with the use of assorted aptamers, the ability of AuNCs
13 182 to discriminate specific DNA could offer a new approach to detect a broad range of analytes.
14 183 What's not neglectable is the high fluorescence of the AuNCs itself which allows it to be widely
15 184 used in the emerging multiplex fluorescence imaging in biological applications in the near future.
16 185 With these remarkable advantages, we believe this work provides opportunities to develop simple,
17 186 rapid, and low-cost nanoprobes for molecular diagnostics.

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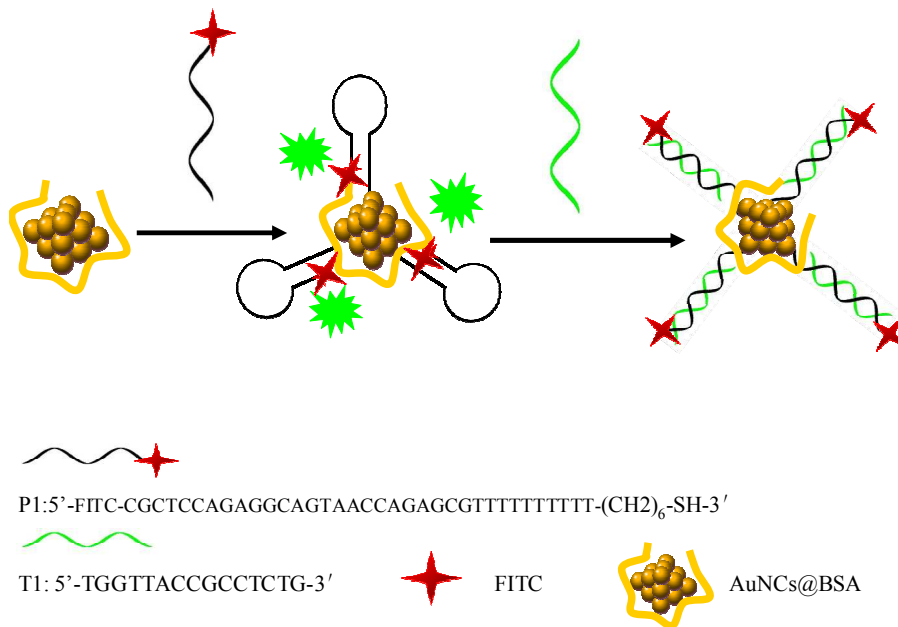
192 **References**

- 193 1. J. H. Warner, A. Hoshino, K. Yamamoto and R. Tilley, *Angew. Chem. Int. Ed.*, 2005, **117**,
194 4626-4630.
- 195 2. S. F. Lim, R. Riehn, C.-K. Tung, W. S. Ryu, R. Zhuo, J. Dalland and R. H. Austin, *Nanotechnology*,
196 2009, **20**, 405701.
- 197 3. S.-Y. Liu, L. Huang, J.-F. Li, C. Wang, Q. Li, H.-X. Xu, H.-L. Guo, Z.-M. Meng, Z. Shi and Z.-Y.
198 Li, *J. Phys. Chem. C*, 2013, **117**, 10636-10642.
- 199 4. Y. Bao, C. Zhong, D. M. Vu, J. P. Temirov, R. B. Dyer and J. S. Martinez, *J. Phys. Chem. C*, 2007,
200 **111**, 12194-12198.
- 201 5. B. Yang, N. Lu, D. Qi, R. Ma, Q. Wu, J. Hao, X. Liu, Y. Mu, V. Reboud and N. Kehagias, *Small*,
202 2010, **6**, 1038-1043.
- 203 6. K. Aslan, M. Wu, J. R. Lakowicz and C. D. Geddes, *J. Am. Chem. Soc.*, 2007, **129**, 1524-1525.
- 204 7. J. Zhang and J. R. Lakowicz, *Opt. Express*, 2007, **15**, 2598-2606.
- 205 8. P. K. Jain, W. Huang and M. A. El-Sayed, *Nano Lett.*, 2007, **7**, 2080-2088.
- 206 9. P. K. Jain and M. A. El-Sayed, *J. Phys. Chem. C*, 2008, **112**, 4954-4960.
- 207 10. W. Huang, W. Qian, P. K. Jain and M. A. El-Sayed, *Nano Lett.*, 2007, **7**, 3227-3234.
- 208 11. M. Mahmoud and M. El-Sayed, *J. Phys. Chem. C*, 2007, **111**, 17180-17183.
- 209 12. P. K. Jain, K. S. Lee, I. H. El-Sayed and M. A. El-Sayed, *J. Phys. Chem. B*, 2006, **110**, 7238-7248.
- 210 13. P. K. Jain, X. Huang, I. H. El-Sayed and M. A. El-Sayed, *Plasmonics*, 2007, **2**, 107-118.
- 211 14. Q.C. Sun, H. Munderoor, J. C. Ribot, V. Singh, I. I. Smalyukh and P. Nagpal, *Nano Lett.*, 2013, **14**,
212 101-106.

- 1
2
3 213 15. T. V. Shahbazyan, *Nano Lett.*, 2012, **13**, 194-198.
4 214 16. J. Yoo, X. Ma, W. Tang and G.-C. Yi, *Nano Lett.*, 2013, **13**, 2134-2140.
5 215 17. L. Zhao, T. Ming, L. Shao, H. Chen and J. Wang, *J. Phys. Chem. C*, 2012, **116**, 8287-8296.
6 216 18. M. Lunz, V. A. Gerard, Y. K. Gun'ko, V. Lesnyak, N. Gaponik, A. S. Susha, A. L. Rogach and A. L.
7 217 Bradley, *Nano Lett.*, 2011, **11**, 3341-3345.
8 218 19. C.C. Chang, Y. D. Sharma, Y.S. Kim, J. A. Bur, R. V. Shenoi, S. Krishna, D. Huang and S.-Y. Lin,
9 219 *Nano Lett.*, 2010, **10**, 1704-1709.
10 220 20. S. Jin, E. DeMarco, M. J. Pellin, O. K. Farha, G. P. Wiederrecht and J. T. Hupp, *J. Phys. Chem.*
11 221 *Lett.*, 2013, **4**, 3527-3533.
12 222 21. M. J. Previte, K. Aslan, Y. Zhang and C. D. Geddes, *J. Phys. Chem. C*, 2007, **111**, 6051-6059.
13 223 22. M. Banik, P. Z. El-Khoury, A. Nag, A. Rodriguez-Perez, N. Guarrott-xena, G. C. Bazan and V. A.
14 224 Apkarian, *ACS Nano*, 2012, **6**, 10343-10354.
15 225 23. L.-B. Zhao, M. Zhang, Y.-F. Huang, C. T. Williams, D.-Y. Wu, B. Ren and Z.-Q. Tian, *J. Phys.*
16 226 *Chem. Lett.*, 2014, **5**, 1259-1266.
17 227 24. W. M. Leevy, S. T. Gammon, H. Jiang, J. R. Johnson, D. J. Maxwell, E. N. Jackson, M. Marquez,
18 228 D. Piwnica-Worms and B. D. Smith, *J. Am. Chem. Soc.*, 2006, **128**, 16476-16477.
19 229 25. K. Aslan, J. R. Lakowicz and C. D. Geddes, *J. Phys. Chem. B*, 2005, **109**, 6247-6251.
20 230 26. K. Aslan, Z. Leonenko, J. R. Lakowicz and C. D. Geddes, *J. Phys. Chem. B*, 2005, **109**, 3157-3162.
21 231 27. J. Chen, Y. Jin, N. Fahrudin and J. X. Zhao, *Langmuir*, 2013, **29**, 1584-1591.
22 232 28. M. Ganguly, A. Pal, Y. Negishi and T. Pal, *Langmuir*, 2013, **29**, 2033-2043.
23 233 29. L. Shang and G. U. Nienhaus, *Biophys. Rev.*, 2012, **4**, 313-322.
24 234 30. Y. Wang, J. Chen and J. Irudayaraj, *Acs Nano*, 2011, **5**, 9718-9725.
25 235 31. H. Qian, M. Zhu, Z. Wu and R. Jin, *Acc. Chem. Res.*, 2012, **45**, 1470-1479.
26 236 32. J.M. Liu, J.T. Chen and X.P. Yan, *Anal. Chem.*, 2013, **85**, 3238-3245.
27 237 33. F. Wen, Y. Dong, L. Feng, S. Wang, S. Zhang and X. Zhang, *Anal. Chem.*, 2011, **83**, 1193-1196.
28 238 34. B. Paramanik and A. Patra, *J. Mater. Chem. C*, 2014, **2**, 3005-3012.
29 239 35. H. Dai, Y. Shi, Y. Wang, Y. Sun, J. Hu, P. Ni and Z. Li, *Biosens. Bioelectron.*, 2014, **53**, 76-81.
30 240 36. J.-a. Annie Ho, H.-C. Chang and W.-T. Su, *Ana. Chem.*, 2012, **84**, 3246-3253.
31 241 37. Y.-H. Lin and W.-L. Tseng, *Anal. Chem.*, 2010, **82**, 9194-9200.
32 242 38. D. Hu, Z. Sheng, P. Gong, P. Zhang and L. Cai, *Analyst*, 2010, **135**, 1411-1416.
33 243 39. M. Zhang and B.-C. Ye, *Analyst*, 2011, **136**, 5139-5142.
34 244 40. H. Liu, X. Zhang, X. Wu, L. Jiang, C. Burda and J.-J. Zhu, *Chem. Commun.*, 2011, **47**, 4237-4239.
35 245 41. M.-L. Cui, J.-M. Liu, X.-X. Wang, L.-P. Lin, L. Jiao, L.-H. Zhang, Z.-Y. Zheng and S.-Q. Lin,
36 246 *Analyst*, 2012, **137**, 5346-5351.
37 247 42. V. Venkatesh, A. Shukla, S. Sivakumar and S. Verma, *ACS Appl. Mater. Inter.*, 2014, **6**, 2185-2191.
38 248 43. Z. Wu and R. Jin, *Nano Lett.*, 2010, **10**, 2568-2573.
39 249 44. J. Xie, Y. Zheng and J. Y. Ying, *J. Am. Chem. Soc.*, 2009, **131**, 888-889.
40 250 45. X. Zhang, C. A. Marocico, M. Lunz, V. A. Gerard, Y. K. Gun'ko, V. Lesnyak, N. Gaponik, A. S.
41 251 Susha, A. L. Rogach and A. L. Bradley, *ACS Nano*, 2014, **8**, 1273-1283.
42 252 46. H. Aouani, J. Wenger, D. Gérard, H. Rigneault, E. Devaux, T. W. Ebbesen, F. Mahdavi, T. Xu and S.
43 253 Blair, *ACS Nano*, 2009, **3**, 2043-2048.
44 254 47. R. Bardhan, N. K. Grady, J. R. Cole, A. Joshi and N. J. Halas, *ACS Nano*, 2009, **3**, 744-752.
45 255 48. S. Raut, R. Rich, R. Fudala, S. Butler, R. Kokate, Z. Gryczynski, R. Luchowski and I. Gryczynski,
46 256 *Nanoscale*, 2014, **6**, 385-391.

- 1
2
3 257 49. D. S. Wilson and J. W. Szostak, *Ann. Rev. Biochem.*, 1999, **68**, 611-647.
4 258 50. J. Liu, Z. Cao and Y. Lu, *Chem. Rev.*, 2009, **109**, 1948-1998.
5 259 51. S. Song, L. Wang, J. Li, C. Fan and J. Zhao, *Trends Anal. Chem.*, 2008, **27**, 108-117.
6 260 52. D. Li, S. Song and C. Fan, *Acc. Chem. Res.*, 2010, **43**, 631-641.
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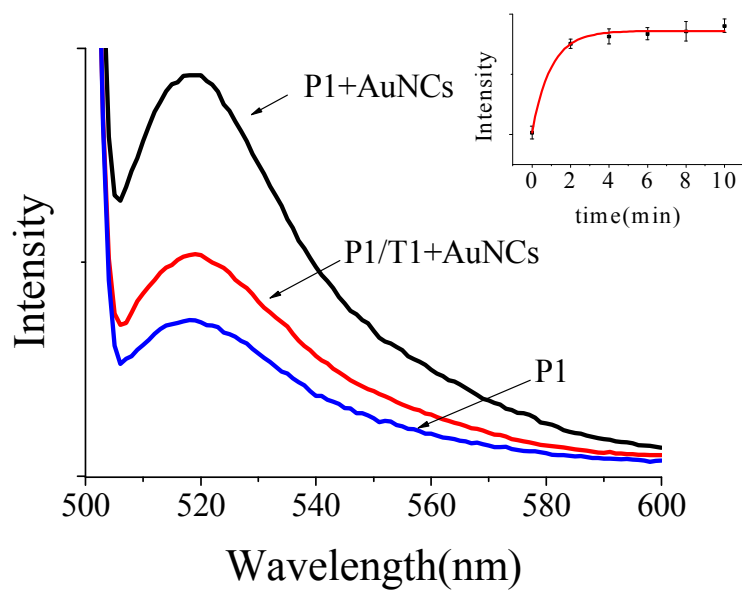


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264 **Scheme 1.** The novel platform for detection of protooncogene based on Au nanoclusters

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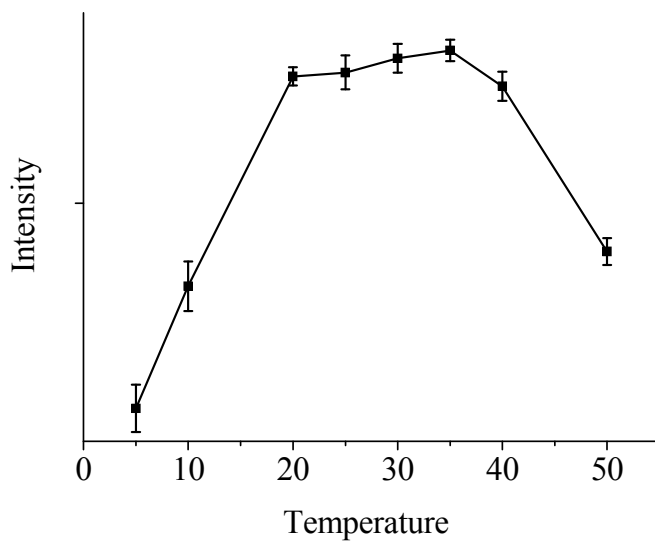
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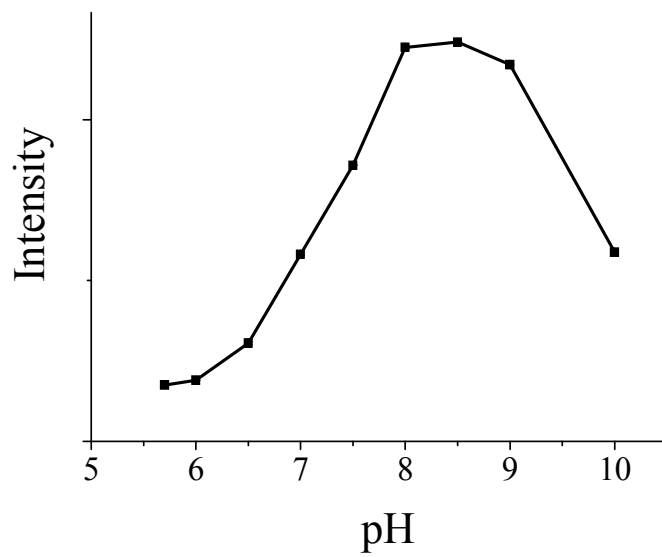
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268 **Figure 1.** Fluorescence spectra of P1+AuNCs (10 nM P1 with AuNCs), P1/T1 duplex +AuNCs (10 nM P1 and 50
269 nM T1 with AuNCs) and P1 (10 nM P1 without AuNCs). Inset: Kinetic study for the fluorescence change of P1 in
270 the presence of AuNCs. Excitation and emission wavelengths are 490 and 520 nm, respectively.

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272
273 **Figure 2.** Temperature dependences of the FITC fluorescence intensity (enhanced fluorescence spectra of 10nM
274 P1 in the presence of AuNCs at 5, 10, 20, 25, 30, 35, 40 and 50°C).
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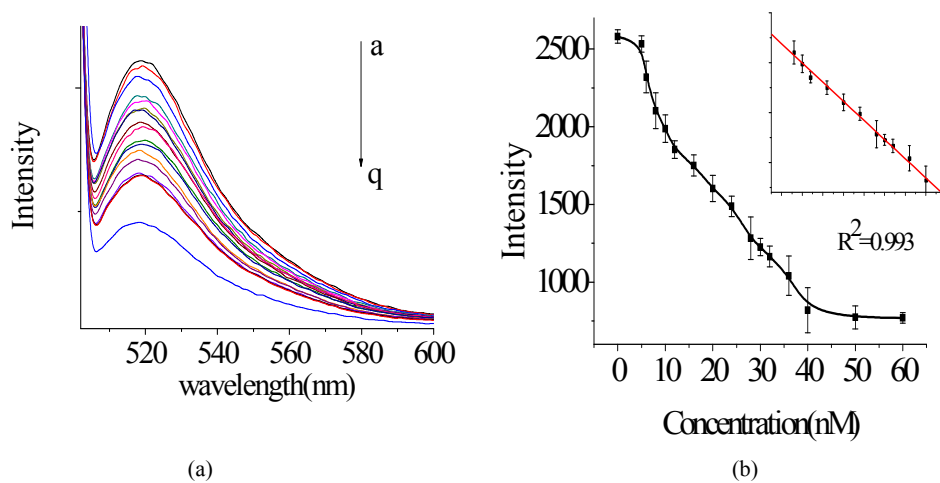


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278 **Figure 3.** The pH (pH 5.5, 6.0, 7.0, 7.5, 8.0, 8.5, 9.0 and 10) dependences of the FITC fluorescence intensity
279 (fluorescence spectra of 10 nM P1 in the presence of AuNCs)

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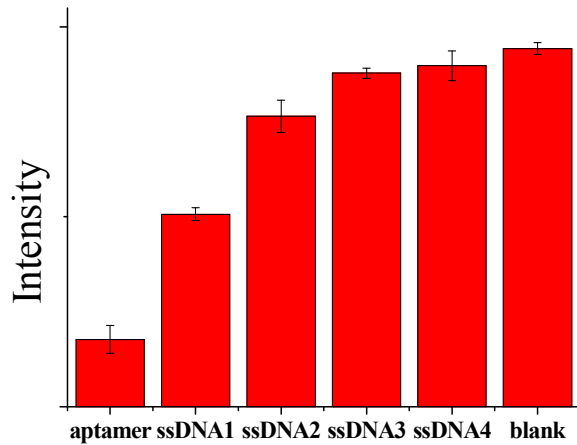
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284 **Figure 4.** (a) The calibration curve for DNA detection (a-p: 0, 5, 6, 8, 10, 12, 16, 20, 24, 28, 30, 32, 36, 40, 50 and
285 60 nM. q: 10 nM P1 in PBS buffer without AuNCs). (b) Fluorescence spectra of P1 (10 nM) with AuNCs in the
286 presence of different concentrations of T1 (0, 5, 6, 8, 10, 12, 16, 20, 24, 28, 30, 32, 36, 40, 50 and 60 nM). Inset:
287 amplification of the linear concentration range of the calibration curve. Excitation and emission wavelengths are
288 490 and 520 nm, respectively.

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291

292 **Figure 5.** Selectivity of the AuNCs-based protooncogene sensor over ssDNA1, ssDNA2, ssDNA3, ssDNA 4 and
293 blank sample (each 10 μ M). Excitation and emission wavelengths are 490 and 520 nm, respectively.

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