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COMMUNICATION

A polymer-based ratiometric intracellular glucose sensor

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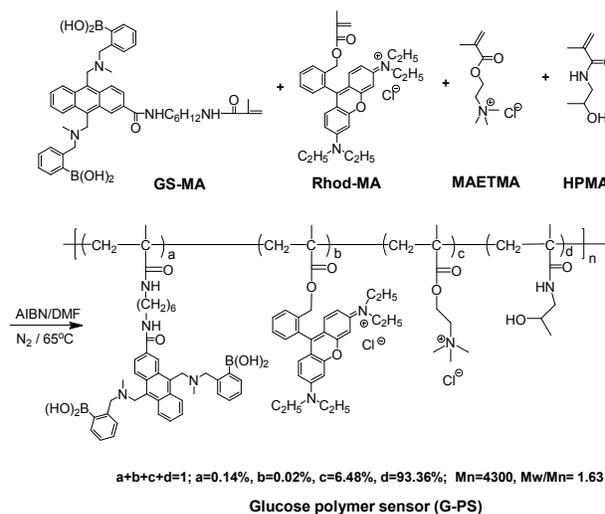
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Glucose metabolism level reflects cell proliferative status. A polymeric glucose ratiometric sensor comprising poly(N-(2-hydroxypropyl)methacrylamide) (PHPMA) and poly[2-(methacryloyloxy)ethyl]trimethylammonium chloride] (PMAETMA) was synthesized. Cellular internalization and glucose response of the polymer within HeLa cells was investigated.

Glucose is the major carbon donor and energy source for the maintenance of cell homeostasis and cell proliferation¹⁻². Radioactive labeled glucose and unmetabolized glucose analog (2-deoxyglucose) are commonly used for monitoring glucose uptake³⁻⁶; however, these analogs are still different from glucose, with some of them inducing cell death⁷⁻⁸. In order to monitor glucose uptakes and/or the changes of glucose concentrations, many different glucose sensors have been developed⁹. Most of these sensors were designed to be able to detect the concentration changes of glucose in the environment of cells, i.e. extracellular glucose. A few kinds of glucose sensors were reported to be capable of real-time measurements of intracellular glucose concentrations, which include enzyme-based sensors¹⁰, genetically encoded protein-based sensors¹¹⁻¹³, and the probes encapsulated by biologically localized embedding nanoparticles (PEBBLE)¹⁴.

Our center has been working on developing fluorescent biosensors for understanding cell metabolism at bulk and single cell levels¹⁵⁻²⁰. Very recently, we have synthesized a polymer film-based extracellular dual sensor for glucose and oxygen²¹. Herein, we report a polymeric sensor for intracellular glucose monitoring. The use of cell permeable polymer platform as intracellular sensor will enable the abundance of sensor materials and broaden the design flexibility of ratiometric sensors, which has not yet been explored. We use poly(N-(2-hydroxypropyl)methacrylamide) (PHPMA) (Scheme 1) as a biocompatible polymer, which has been widely used for drug delivery²². The polymer possesses a small fraction of poly[2-(methacryloyloxy)ethyl]-trimethylammonium chloride] (PMAETMA), which is a polymer with positive charges for ensuring its cell permeability²⁰. The glucose probe is a monomer, which can

be easily copolymerized with HPMA and MAETMA. To alleviate the influence of complex cellular environment on the accuracy of the measurement, a rhodamine-based glucose non-responsive probe (Rhod-MA) was integrated into the polymeric sensor for obtaining a ratiometric sensor.



Scheme 1: Synthesis of intracellular glucose sensor: G-PS. The ratios were determined by ¹H NMR (Figure S1 of supporting information) and UV-Vis (Figure S2) spectra.

The sensor was synthesized using the traditional radical polymerization approach (Scheme 1) with a co-polymerization of Rhod-MA²¹, GS-MA²¹, HPMA²⁰ and MAETMA in DMF. The polymeric glucose sensor (G-PS) was purified through precipitation from the DMF solution into acetone to remove any potential non-polymerized monomers and further by dialysis against water.

The sensor was characterized using gel permeation chromatography, and ¹H NMR. Zeta potential measurement indicated the positive charge of the polymer with a zeta potential of 18.5 mV.

The response of the G-PS to glucose in phosphate buffer saline solution (PBS) was given in Figure 1. The blue emission with a maximum at 445 nm increased with increasing glucose concentration. The fluorescence intensity at 445 nm is found to increase by about 4.5 folds with 50 mM of glucose as compared to that without glucose. This is due to the effect of photo-induced electron transfer (PET)²³ (Detailed mechanism of the PET effect on glucose sensing is given in supporting information, Figure S3). The orange emission with a maximum of 580 nm from the rhodamine group did not show a response to glucose, indicating its suitability as an internal built-in reference probe for the glucose sensor.

Figure 1B plotted the intensity ratio changes with respect to glucose concentration. The sensor has excellent sensitivity to glucose concentration lower than 10 mM, especially in a concentration range from 0.1 to 5 mM. The sensor has linear response to glucose from 0.1 mM to 1 mM (Figure S4). Noting normal intracellular glucose concentration may vary from 0.1 to 5 mM depending on cell lines and status²⁴⁻²⁵, we believe that this sensor is capable for monitoring intracellular glucose concentration.

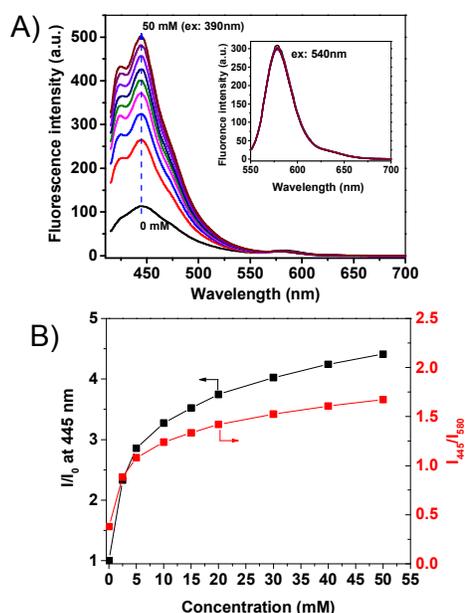


Figure 1. Typical responses of G-PS to glucose in PBS buffer. The insert figure in A shows the magnified peak at 580 nm under an excitation at 540 nm. B shows the glucose concentration dependent fluorescence intensity ratio changes. I_0 is the fluorescence intensity at 445 nm before the interaction with glucose. I is the fluorescence intensity after interaction with glucose.

The saccharide specificity of the sensor was compared among glucose, fructose, galactose and mannose. G-PS has responses to other saccharides and is most sensitive to fructose (Figure S5). This is common for many other amino-boronic-containing glucose sensors²⁶. Considering that there are few other saccharides except glucose used for cell culture, this specificity will not affect the sensor's application for the detection of glucose in cell metabolism research.

The sensor was internalized with human cervical cancer HeLa cell line. We found that the sensor at a concentration of 0.05 mg/mL in cell culture medium could stain cells after 3 hour of cellular internalization. To get better cellular images, the sensor concentration at 0.1 mg/mL and internalization time of 16 hours were usually used for cell staining. Results showed that the sensor is

cell permeable, and localizes in the cytoplasm area. Figure 2 shows the cellular distribution of the sensor in HeLa cells. The sensor is also cell permeable to other cell lines, like metastatic epithelial CPA cells, glioblastoma U87-MG cells, and mouse macrophage J774. A1 cells (Figure S6). The blue color (Figure 2A) represents the glucose probe, and the red color (Figure 2B) represents internal built-in probe. The pink color is the exact overlay of the images of Figures 2A and 2B. It is worthy to note that the ratio between intensity of blue and red fluorescence of G-PS does not overlap well in some area of cells (Figure 2C) which might be attributed to non-uniform subcellular distribution of glucose. The sensor's subcellular colocalization was further investigated with mitochondria-specific MitoTracker[®] Green and lysosome-specific LysoTracker[®] Green, respectively (Figures S7 and S8). Results showed no specific colocalizations of the sensor in the two important organelles. The possible influence of intracellular cellular pH on the sensor's responses to glucose was studied. Intracellular pH value was homogenized using a commercially available Intracellular pH Calibration Buffer Kit from pH 5.5 to 7.5 (Life Technology catalog number P35379) with valinomycin and nigericin, which help equilibrate the pH inside and outside of cells. We did not found significant fluorescence changes from cellular pH 5.5 to 7.5. (Figure S9)

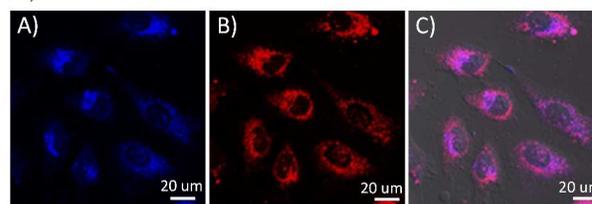


Figure 2. Cell images of G-PS in HeLa cells. A: blue channel for glucose probes excited at 405 nm; B: red channel for rhodamine internal reference excited at 561 nm; C: overlay of A and B with bright field image.

Cytotoxicity of the sensor to HeLa cell line was studied using 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyl tetrazolium bromide (MTT) assay (Figure S10). No significant cell cytotoxicity was observed at a sensor concentration of 0.05 mg/mL after internalization with cells for 24 hours

The fluorescent response of G-PS to intracellular glucose changes was tested with HeLa cells. According to a known protocol⁶, cells were treated by medium without serum for 16 hours before the glucose uptake experiments were performed in KRH buffer (50 mM of HEPES, 137 mM of NaCl, 4.7 mM of KCl, 1.85 mM of CaCl₂, 1.3 mM of MgSO₄ and 0.1% BSA). Intracellular glucose concentrations and its dynamic changes (Figure 3) were determined by referring the titration curve. To check the influence of extracellular glucose concentration on intracellular glucose concentration, we used two extracellular glucose concentrations, i.e. 10 mM and 25 mM, respectively. It was found that the intracellular glucose concentration of the starved HeLa cells was 0.12 mM²⁷.

After cells started to take up glucose from the KRH buffer, intracellular glucose concentration started to increase. At the high extracellular glucose concentration (25 mM), intracellular glucose reached equilibrium within 5 minutes. With an increase of incubation time, the glucose concentration did not change much. At the normal extracellular glucose (10 mM) condition, it took about 30 minutes to reach the equilibrium of intracellular glucose. The

intracellular glucose concentration after 30 minutes internalization with glucose was determined to be 0.20 mM using the sensor G-PS.

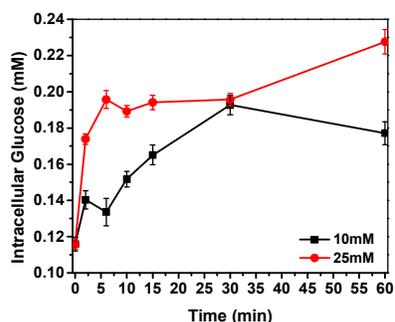


Figure 3. Intracellular glucose concentration detected by G-PS. 10 mM and 25 mM of extracellular glucose were applied to cell media after 60 min of glucose starvation.

Conclusions

In conclusion, we have developed a polymer-based ratiometric glucose sensor for monitoring the intracellular glucose concentration. It was demonstrated that the sensor is capable of measuring dynamic intracellular glucose concentration and the changes. Considering the flexibility of the sensor design using polymer technology, we believe the use of polymers as intracellular sensors will broaden the design of sensors with multifunctionality. This polymer based sensors will provide a new and flexible platform for intracellular glucose sensing, and can be extended to other sensors. Further investigation of the use of this G-PS for single live cell glucose analysis is in progress.

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Notes and references

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†Supplementary Data: The synthesis of sensor, sensor selectivity, and co-localizations.

REFERENCES

- M. G. V. Heiden, L. C. Cantley, C. B. Thompson, *Science* 2009, 324, 1029.
- R. J. DeBerardinis, C. B. Thompson, *Cell* 2012, 148, 1132.
- B. Teusink, J. A. Diderich, H. V. Westerhoff, K. van Dam, M. C. Walsh, *J. Bacteriol.* 1998, 180, 556.
- E. P. Brizinski, A. T. Phillips, R. F. Roberts, *Appl. Environ. Microb.* 2008, 74, 6941.
- M. G. Vander Heiden, D. R. Plas, J. C. Rathmell, C. J. Fox, M. H. Harris, C. B. Thompson, *Mol. Cell. Biol.* 2001, 21, 5899.
- N. Yamamoto, M. Ueda, T. Sato, K. Kawasaki, K. Sawada, K. Kawabata, H. Ashida, *Current protocols in pharmacology / editorial board*, S.J. Enna 2011, Chapter 12, Unit 12 14 1.
- S. F. Ramirez-Peinado, Alcazar-Limones, L. Lagares-Tena, N. El Mjiyad, A. Caro-Maldonado, O. M. Tirado, C. Munoz-Pinedo, *Cancer Res.* 2011, 71, 6796.
- O. Zagorodna, S. M. Martin, D. T. Rutkowski, T. Kuwana, D. R. Spitz, C. M. Knudson, *Oncogene* 2012, 31, 2738.
- M. S. Steiner, A. Duerkop, O. S. Wolfbeis, *Chem. Soc. Rev.* 2011, 40, 4805.
- M. H. Asif, S. M. U. Ali, O. Nur, M. Willander, C. Brannmark, P. Stralfors, U. H. Englund, F. Elinder, B. Danielsson, *Biosens. Bioelectron.* 2010, 25, 2205.
- B. H. Hou, H. Takanaga, G. Grossmann, L. Q. Chen, X. Q. Qu, A. M. Jones, S. Lalonde, O. Schweissgut, W. Wiechert, W. B. Frommer, *Nat. Protoc.* 2011, 6, 1818.
- C. X. Bittner, A. Loaiza, I. Ruminot, V. Larenas, T. Sotelo-Hitschfeld, R. Gutierrez, A. Cordova, R. Valdebenito, W. B. Frommer, L. F. Barros, *Front. Neuroenergetics.* 2010, 2, 26.
- J. V. Veetil, S. Jin, K. J. Ye, *Diabetes Sci. Technol.* 2012, 6, 1276.
- H. Xu, J. W. Aylott, R. Kopelman, *Analyst* 2002, 127, 1471.
- X. Zhou, F. Su, Y. Tian, C. Youngbull, R. H. Johnson, D. R. Meldrum, *J. Am. Chem. Soc.* 2011, 133, 18530.
- Y. Tian, B. R. Shumway, A. C. Youngbull, Y. Li, A. K. Jen, R. H. Johnson, D. R. Meldrum, *Sensor Actuat B-Chem* 2010, 147, 714.
- C.Y. Chen, T. H. Kim, W. Wu, C. Huang, H. Wei, C. W. Mount, Y. Tian, S. H. Jang, S. H. Pun, A. K. Jen, *Biomaterials* 2013, 34, 4501.
- F. Su, R. Alam, Q. Mei, Y. Tian, C. Youngbull, R. H. Johnson, D. R. Meldrum, *PLoS One* 2012, 7, e33390.
- L. Kelbauskas, S. P. Ashili, J. Houkal, D. Smith, A. Moham-madrea, K. B. Lee, J. Forrester, A. Kumar, Y. H. Anis, T. G. Paulson, C. A. Youngbull, Y. Tian, M. R. Holl, R. H. Johnson, D. R. Meldrum, *J. Biomed. Opt.* 2012, 17, 037008.
- H. Lu, F. Su, Q. Mei, Y. Tian, W. Tian, R. H. Johnson, D. R. Meldrum, *J. Mater. Chem.* 2012, 22, 9890.
- L. Zhang, F. Su, S. Buizer, H. Lu, W. Gao, Y. Tian, D. R. Meldrum, *Biomaterials* 2013, 34, 9779-9788.
- T. Lammers, K. Ulbrich, *Adv. Drug Del. Rev.* 2010, 62, 119.
- T. Kawanishi, M. A. Romey, P. C. Zhu, M. Z. Holody, S. Shinkai, *J. Fluoresc.* 2004, 14, 499.
- J. R. Garrett, X. X. Wu, S. Jin, K. M. Ye, *Biotechnol. Progr.* 2008, 24, 1085.
- G. W. Cline, K. F. Petersen, M. Krssak, J. Shen, R. S. Hundal, Z. Trajanoski, S. Inzucchi, A. Dresner, D. L. Rothman, G. I. Shulman, *New Engl. J. Med.* 1999, 341, 240.
- H. S. Mader, O. S. Wolfbeis, *Microchim. Acta* 2008, 162, 1.
- The initial glucose concentration in the starved Hela cells was determined by using a glucose assay kit (Biovision, Milpitas, CA, Cat-K606).