Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/foodfunction

1	
2	Carbonyl trapping and antiglycative activities of olive oil mill wastewater
3	Marta Navarro <sup>a</sup> · Alberto Fiore <sup>b,c</sup> · Vincenzo Fogliano <sup>d</sup> · Francisco J. Morales <sup>a</sup>
4	
5	
6	a Institute of Food Science, Technology and Nutrition, ICTAN-CSIC, E-28040 Madrid, Spain
	b School of Science, Engineering & Technology, Division of Food & Life Sciences, Abertay University, Dundee DD1 1HG, UK
	c Department of Agriculture and Food Science, University of Naples "Federico II", Naples, Italy
	d Food Quality and Design group, Wageningen University & Research Centre, PO Box 8129, 6700 EV Wageningen, The Netherlands
7	
8	
9	Correspondence to:
10	Francisco J. Morales (PhD)
11	Institute of Food Science, Technology and Nutrition (ICTAN-CSIC), E-28040 Madrid, Spain
12	Email: <u>fjmorales@ictan.csic.es</u>
13	Phone: +34 91 549 2300

#### 15 Abstract

16 The use of natural compounds as antiglycative agent to reduce the load of advanced glycation 17 end products from diet is very promising. Olive mill wastewater is a by-product of the olive oil 18 extraction processes with high content of hydroxytyrosol, hydroxytyrosol derivatives and 19 molecules containing o-dihydroxyl functions such as the verbascoside. Two powders were 20 obtained after ultrafiltration and nanofiltration of olive mill wastewater, and successive spray 21 drying with maltodextrin and acacia fiber. Samples were characterized by phenolic 22 composition and antioxidant capacity. Antiglycative capacity was evaluated by in vitro BSA-23 glucose and BSA-methylglyoxal assays, formation of Amadori products and direct trapping of 24 reactive dicarbonyls (methylglyoxal and glyoxal). Both ultrafiltrated and nanofiltrated olive mill 25 wastewater powders had an activity comparable to quercetin and hydroxytyrosol against inhibition of protein glycation (IC<sub>50</sub> = 0.3 mg mL<sup>-1</sup>). The antiglycative activity of the powder was 26 further investigated after separation by reverse phase solid extraction. Fractions extracted 27 28 with methanol content higher than 40% and rich in hydroxytyrosol and verbascoside, exerted 29 the highest reactivity against dicarbonyls. Data confirmed that direct trapping of dicarbonyl 30 compounds is a main route explaining the antiglycative action rather than of the already 31 known antioxidant capacity. Results support further investigations to evaluate the 32 technological feasibility to use olive mill wastewater powders as antiglycative ingredient in 33 foods or in pharmacological preparations in future.

34

35 Keywords Antiglycative activity; advanced glycation end-products (AGE); carbonyl trapping;

36 olive oil mill wastewater (OMW); hydroxytyrosol.

#### 38 1. Introduction

39 The health benefits of the Mediterranean diet have been firmly established, including the lowering of degenerative pathologies<sup>1</sup>, cardiovascular diseases<sup>2</sup> and cancer<sup>3-4</sup>. In this respect, 40 41 olive oil's effects, as important component of the Mediterranean diet, are related not only to 42 the peculiar fatty acid profile but also to the presence of a specific category of phenolic 43 compounds named secoiridoids. The main phenolic compound in olive plant (Olea europaea L.) 44 is the oleuropein, which is hydrolyzed into hydroxytyrosol (3,4-dihydroxyphenylethanol, HT) and tyrosol during extraction of olive oil<sup>5</sup>. Several epidemiological studies suggest that HT is an 45 efficient scavenger of peroxyl radicals<sup>6</sup> and prevents the damages of oxidative stress. In vivo 46 47 and in vitro studies, HT has shown various biological activities with potential beneficial health 48 effects. HT has been associated with anti-inflammatory, hypocholesterolemic, neuroprotective<sup>7</sup> and cardio protective activity as well as hypoglycemic effect which could 49 prevent glucose auto-oxidation reaction, process related to the formation of AGEs<sup>8</sup>. 50

51 Advanced glycation endproducts (AGEs) are the outcome of the non-enzymatic glycation 52 reaction between the carbonyl group of a reducing sugar and free amino group of a protein. 53 Several AGEs have been identified to date and many of them are fluorescent and yield protein crosslinking that alters protein structure and their biochemical properties<sup>9</sup>. AGEs are significant 54 55 contributor to classical features of ageing and common chronic pathologies such as diabetes and its complications and Alzheimer's disease<sup>10</sup>. Many researchers have linked the oxidative 56 57 reactions with the accelerating the rate of AGEs formation, since radicals and  $\alpha$ -dicarbonyls, 58 reactive intermediates highly reactive such as methylglyoxal (MGO), glyoxal (GO) or 3-59 deoxyglucosone, are generated. Previous studies have suggested that antioxidants protect 60 against glycation since they decreased the formation of reactive carbonyl species which is a 61 consequence of oxidative stress generated by formation of the reactive oxygen species (ROS)<sup>3</sup>. 62 AGEs can also be introduced in the systemic system of living bodies by exogenous sources like 63 the diet. Basically, AGEs, also named Maillard reaction products, are formed during thermal 64 processing of foods. The AGEs content in processed foods has increased in the past 50 years and dietary AGEs contributed to the total human body AGE pool<sup>11</sup>. 65

Due to the AGEs involvement in the development of several health disorders, the search for AGE formation inhibitors in biological systems and in foods has been the target of many investigations. The use of synthetic compounds was discontinued because of the possible side effects and the use of aminoguanidine was also proposed. However the strategy of using natural compounds as antiglycative alternative is very appealing. Some bioactive phytochemicals naturally occurring in plants and possessing redox properties have been

related to the inhibition of glycation processes<sup>12</sup>. Studies of HT *in vivo* and *in vitro* have displayed it to possess cardio protective, anticancer and antidiabetic effects related to its ability to inhibit oxidative stress. Considering that the relationship between HT, ROS and tissue injury has been involved<sup>13</sup> with the pathogenesis of insulin resistance, the antiglycative activity of HT is plausible and noteworthy to be investigated. Recently, Troise et al., (2014)<sup>14</sup> concluded that phenol compounds from olive mill wastewater (OMW) are effective inhibitors of the extent of the Maillard reaction products during thermal processing of UHT milk.

79 In this framework, a food ingredient made from spray-dried OMW was selected as 80 promising candidate for the antiglycative action. OMW is a by-product of the olive oil extraction processes with high content of HT, HT derivatives and molecules containing o-81 dihydroxyl functions such as the verbascoside<sup>15</sup>. The aim of the present study was assess the 82 83 inhibitory capacity of different OMW powders on the formation of AGE by using different in 84 vitro models of MGO and glucose as inducers of the glycation reaction. Further, the ability of 85 the OMW ingredient for trapping of dicarbonyl compounds was also assessed. Antioxidant 86 activity was examined in order to study their possible relationship with AGE-inhibitory activity.

87

#### 88 2. Materials and methods

#### 89 2.1. Materials

90 D (+)-Glucose (GLC), bovine serum albumin (BSA), methylglyoxal (MGO, 40% aqueous 91 solution), glyoxal (GO, 40% aqueous solution), aminoguanidine (AG), quercetin (QE), 92 verbascoside (VB), 5-methylguinoxaline (5-MQ), nitrotetrazolium blue chloride (NBT), o-93 phenyldiamine (OPD), trolox (6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid), 2,4,6-94 tris(2-piridil)s-tirazina 2-2'-azinobis(3-ethylbenzothiazoline-6-sulphonic (TPTZ), acid) 95 diammonium salt (ABTS) and phenolic acids standards were provided by Sigma (St Louis, MO, 96 USA). Folin-Ciocalteu reagent, iron (III) chloride was purchased from Panreac (Madrid, Spain). 97 Hydroxytyrosol standard (HT, purity > 99 %) was acquired from Seprox Biotech (Madrid, Spain). 98 Pyridoxamine (PM) was acquired from Fluka Chemical (Madrid, Spain). Sodium dihydrogen 99 phosphate monohydrate, sodium carbonate anhydrous, formic acid, glacial acetic acid and 100 high-performance liquid chromatography (HPLC)-grade methanol were purchased from Merck 101 (Darmstadt, Germany). C18 cartridges Sep-Pak Plus were supplied by Waters (Ireland). The 102 Milli-Q water was obtained by an Elix3 water purification system coupled to a Milli-Q Advance 103 10 module (Millipore, Molsheim, France). All other chemicals and reagents were of analytical 104 grade.

#### 105 *2.2. Equipments*

Synergy<sup>™</sup> HT-multimode microplate reader with an automatic reagent dispense and
 temperature control from Biotek Instruments (VT, USA). HPLC Shimadzu (Kyoto, Japan)
 equipped with a LC-20AD pump, a SIL-20AHT autosampler, a CTO-10ASVP oven, and a DAD
 (SPD-M20A).

110 2.3. Preparation of olive mill wastewater powder

111 Olive oil polyphenols powders were obtained at Azienda Agricola Fangiano (Calabria, Italy) starting from the water fraction resulting from virgin-olive oil production (Carolea variety) as 112 113 follows. Olive water were separated from the oil and the olive paste by centrifugation; they 114 were treated with pectinases for 2 hours at 37° C then fractionated by a filtration plant made 115 up with three membranes having different cut off. Olive water was forced to pass through 116 microfiltration (cut off 25 kDa), ultrafiltration (cut off 8 kDa), and nanofiltration (cut off 0.3 kDa) membranes. At each stage a retentate containing the compounds not passing the 117 118 membrane pores was collected while the permeate move on to the next filtration step. For 119 this investigation, the ultrafiltration and nanofiltration retentates were concentrated by 120 inverse osmosis (cut off 0.1 kDa) up to 20 % dry weight and spray dried adding maltodextrin 121 and acacia fiber in a ratio 2:1 with the water mill dry weight (66,6% olive mill water 33,3% 122 maltodextrin and acacia fiber 1:1). Finally, a fine pale yellow of ultrafiltrated (UOMW) and 123 nanofiltrated (NOMW) powder with moisture content below 4% were obtained and used in 124 this study.

## 125 2.4. Analysis of phenolic compounds in OMW powder

126 The analysis of the phenolic component was carried out as described by Kokkinidou & 127 Peterson<sup>16</sup> with some modifications. Briefly, the OMW powder was dissolved in distilled water, in order to prepare a solution at concentration of 20 mg mL<sup>-1</sup>. To 1 mL of this solution, 10  $\mu$ L of 128 a 5 mg L<sup>-1</sup> solution of butyl -4-hydroxybenzoate as internal standard were added. The phenolic 129 130 fraction was extracted through the use of SPE cartridges Strata C18-E, and dried under a 131 nitrogen stream. Thereafter the precipitate was recovered in 500 µL of a solution of water-132 methanol/95-5 (v/v) ready to HPLC analysis. The instrument used for the chromatographic 133 analysis was a LC-20AD HPLC with UV-Vis detector SPD20A, set at 279 nm, and SCL-20A 134 controller (Shimadzu, Japan). The mobile phases were H<sub>2</sub>O 0.1% formic acid (A) and methanol (B). The flow was 0.8 mL min<sup>-1</sup>. A Prodigy ODS3 column was used (250 x 4.60 mm, 5 micron, 135 136 100 A, Phenomenex, USA). The sample (20  $\mu$ L) were separated with next gradient as follow; 0 137 min (5% B); 4 min (5% B); 40 min (98% B); 43 min (98% B);46 min (5% B); 49 min (5% B). Hydroxytyrosol, tyrosol and verbascoside were quantified by external calibration with
standards. Peaks are identified by retention time, DAD spectra and spiking with sample with
pure standards.

#### 141 2.5. Determination of total phenolic content

Total phenolic content (TPC) was estimated by the Folin-Ciocalteu method as described by Contini et al.<sup>17</sup>. Absorbance readings were recorded using a microplate spectrophotometer and quantified using gallic acid as a standard. Results were expressed as mg gallic acid equivalent (GAE)  $g^{-1}$  sample and all measurements were carried out in triplicate. The limit of quantification was set at 0.5mg GAE  $g^{-1}$  sample.

147 2.6. Determination of antioxidant capacity according to ABTS assay

Antioxidant activity was estimated in terms of radicals scavenging capacity of the extracts
 as described by *Mesías et al.*<sup>18</sup>. Aqueous solutions of Trolox at different concentrations were
 used for calibration. The limit of quantification was set at 1.1 μmol TEAC g<sup>-1</sup> sample. All
 measurements were performed in triplicate.

#### 152 2.7. Determination of antioxidant capacity according to FRAP assay

153 The formation of Fe<sup>2+</sup>-TPTZ complex from Fe<sup>3+</sup>-TPTZ complex was determined as described 154 by *Morales et al.*<sup>19</sup>. Results were expressed as  $\mu$ mol Trolox equivalent antioxidant capacity 155 (TEAC) g<sup>-1</sup> sample.

#### 156 2.8. Fractionation of the olive mill wastewater powdery ingredient

157 The ultrafiltrate olive mill wastewater (UOMW) powder was further fractionated by 158 reversed phased solid phase extraction (C18 SepPak cartridge) with sequential extraction in methanol. A solution of UOMW powder (1 mg mL<sup>-1</sup>) was loaded onto a pre-activated cartridge. 159 160 The first fraction was collected as eluted and termed as fraction UOMW-A. Then 1 mL of formic 161 acid (0.1%) with 5% of methanol was passed through cartridge and the resulting fraction 162 (UOMW-B) was collected. UOMW-C is the fraction that eluted with 1 mL of formic acid with 163 40% of methanol. At last, UOMW-D fraction was eluted with 1 mL of methanol was gone 164 through cartridge. Chromatographic separations of the whole extract and fractions were 165 performed on a C18 Kinetex column (75 mm x 4.6 mm, 2.6 µm, Phenomenex, USA). The 166 temperature was maintained at 40°C and the flow rate was 0.6 mL min<sup>-1</sup>. The mobile phase 167 was 0.5 % acetic acid in water (solvent A) and methanol (solvent B) and the total running time 168 was of 40 min. The gradient changed as follows: 5 %B was held for 5 min, then increased to 169 100 %B in 25 min, held for 1 min, and decreased to 5 %B in 9 min.

170 2.9. Assessment of in vitro glycation of BSA by methylglyoxal and glucose

171 The BSA-MGO assay and BSA-GLC were carried out as described by Mesías et al.<sup>18</sup>. 172 Aminoguanidine, hydroxytyrosol and quercetin were used as standards at a concentration of 0.1 mg mL<sup>-1</sup>, being AG (0.57 mg mL<sup>-1</sup>) the positive control of the assay. The systems containing 173 174 samples/control/blank were incubated at 37°C for 14 days (BSA-MGO assay) or 21 days (BSA-175 GLC assay). Reference was prepared without incubation and stored at -80 °C until analysis. In parallel, samples were dissolved in phosphate buffer (0.1 mol L<sup>-1</sup>, pH 7.4) and incubated at 37 176 °C for 14 days (BSA-MGO assay) or 21 days (BSA-GLC assay) in order to measure their intrinsic 177 178 fluorescence. The fluorescence intensity of all systems (sample/control/blank) was measured 179 in a microplate spectrophotometer. AGE formation was characterized by measuring 180 fluorescence with excitation and emission maxima at 340 and 420 nm for the BSA-MGO assay 181 and 360 and 420 nm for the BSA-GLC assay. The intrinsic fluorescence of the samples 182 incubated under the same conditions was subtracted to the overall fluorescence in each BSA-183 MGO or BSA-GLC system.

184 The percentage inhibition of AGEs formation was calculated according to following 185 equation: inhibition (%) =  $\{1 - [(fluorescence of solution with inhibitor – intrinsic fluorescence$ 186 of sample)/fluorescence of solution without inhibitor] x 100.

The IC<sub>50</sub> (the concentration required to inhibit glycation by 50%) were calculated from the
 dose-response curve using Microsoft-Excel computer software.

189 2.10. Fructosamine adduct assay

190 The fructosamine formation was tested following the procedure described by *Baker et*  $al.^{20}$  with slight modifications. The fructosamine has the ability to reduce NBT to tetrazinolyl 191 192 radical (NBT<sup>+</sup>) in alkaline solution forming a coloured compound called monoformazan (MF<sup>+</sup>). 193 Thereby, 50  $\mu$ L of samples for the BSA-GLC assay were mixed with 450  $\mu$ L of NBT and the 194 mixture was incubated in darkness at 37 °C for 60 min. Then, in the microplate was added 100  $\mu$ L of mixture and 100  $\mu$ L of sodium carbonate buffer (0.1 mol L<sup>-1</sup>, pH 10.35), in guadruplicate, 195 196 except the blank that samples was replaced by sodium carbonate buffer. The presence of fructosamine was characterized by absorbance of  $MF^{+}$  at 350 nm using a microplate 197 198 spectrophotometer.

199 The percentage of inhibition of the fructosamine formation was calculated using the 200 following equation: 201 Inhibition (%) = 100 - [(fluorescence of solution with inhibitor - intrinsic fluorescence of sample - fluorescence of blank) x 100/(fluorescence control positive - intrinsic fluorescence of blank)

203 control positive – fluorescence of blank)].

204 2.11. Direct MGO and GO trapping capacity

205 Direct MGO or GO trapping capacity of samples was determined as described by Mesías et al.<sup>18</sup> with modifications. The mixture of MGO (0.04 mg mL<sup>-1</sup>) or GO (0.03 mg mL<sup>-1</sup>) with 206 207 different concentration of the extracts was incubated at 37 °C for 168 h. Pyridoxamine (PM,  $0.1 \text{ mg mL}^{-1}$ ) instead samples were used as positive control. OPD was added in order to 208 209 derivatize of unreacted MGO or GO into 2-methylquinoxaline (2-MQ) or 1-quinoxaline (1-Q), 210 respectively. 5-quinoxaline (5-MQ) was used as internal standard. Chromatograms were 211 recorded at 315 nm being the retention times of 2-MQ, 1-Q and 5-MQ, 4.1, 3.4 and 5.6 min 212 respectively. The amount of unreacted MGO and GO was calculated from the ratio of 2-MQ or 213 1-Q and 5-MQ to the control at time zero after preparation.

214 The MGO trapping capacity was calculated from the following equation:

215 MGO decrease (%) = [(amount of MGO in control – amount of MGO in sample with tested 216 standard or PM solution)/amount of MGO in control] x 100%.  $IC_{50}$  value (mg mL<sup>-1</sup>) was 217 obtained from the dose-response curves of each experiment.

218 Similarly, the GO trapping capacity was calculated.

219 2.12. Statistical analysis

Data were analysed by using the software Statgraphics Centurion XV Statistical program (Herndon, VA). Homogeneity of variances was assessed via Levene's test. Analysis of variance (ANOVA) was followed by Bonferroni's or Tamhane's test when the variances were homogeneous or non-homogeneous, respectively. Differences were considered to be significant at P < 0.05. Data from at least three independent assays were used to obtain a mean value and its standard deviation.

226

## 227 3. Results

A polyphenol-rich powder was obtained from olive mill wastewater through ultrafiltration/nanofiltration and spray dried, and it was investigated as a promising functional ingredient to decrease the non-enzymatic protein glycation. Two types of powders were obtained, one after ultrafiltration (UOMW) of the olive mill wastewater extract and a second from the retentate after nanofiltration (NOMW). Both extracts were spray dried adding

233 maltodextrins and acacia fiber, and the powders were characterized by total polyphenols 234 content and antioxidant capacity (FRAP and ABTS) as summarised in table 1. The content in 235 polyphenolic compounds and the antioxidant capacity, regardless the method used, were 236 significantly (P< 0.05) higher in the UOMW powder as compared with NOMW powder. It is known that certain bioactive compounds present in plants possess antioxidant capacity as well 237 as ability to inhibit protein glycation as flavonoids<sup>21</sup>. During the spray drying of the extracts, 238 the yield of nanofiltration powder was lower than that of ultrafiltration, particularly when the 239 240 concentration of added dietary fibre was reduced to 33.3%. The average content in hydroxytyrosol, tyrosol and verbascoside were 33.0, 2.1 and 2.8  $\mu$ g g<sup>-1</sup> for the UOMW powder 241 and 13.8, 0.2, and 1.3  $\mu$ g g<sup>-1</sup> for the NOMW powder, respectively. The higher concentration of 242 243 polyphenols in the UOMW powder was due to the presence of bound polyphenols link to the 244 polysaccharide moiety (pectins and arabinans) that remains in the retentate of the extract<sup>22</sup>.

245 The antiglycative properties of OMW powders were evaluated by BSA-MGO and BSA-246 GLC assays. Figure 1a showed the formation of fluorescent AGEs in the BSA-MGO assay 247 (37°C/14 days). UOMW and NOMW powders behave in a dose-dependent manner and both 248 reduced significantly the formation of AGEs, with an  $IC_{50}$  of 0.886 and 1.029 mg mL<sup>-1</sup>, 249 respectively for the ultrafiltrated and nanofiltrated powder (table 2). But ultrafiltrated powder 250 was significantly more effective for antiglycative activity mediated by MGO. AG and QE were 251 used as positive controls due to their well-known effectiveness against the glycation process. 252 Suppression of dicarbonyls formation and direct reaction with dicarbonyls are the main mechanism of action of AG and QE<sup>23-24</sup>. The antiglycative activities of AG, QE, HT, and VB 253 254 solutions (0.14 mg mL<sup>-1</sup>) in the BSA-MGO assay were 68, 86, 70 %, and 84 %, respectively. The  $IC_{50}$  in the BSA-MGO assay were 0.090, 0.053, 0.086, and 0.053 mg mL<sup>-1</sup> for AG, QE, HT, and VB, 255 256 respectively (table 2). Control for maltodextrin and acacia fiber (MD+AF) shown a residual 257 3.5% antiglycative activity for BSA-MGO assay.

258 The figure 1b showed the formation of fluorescent AGEs for BSA-GLC assay (37°C for 21 259 days). Ultrafiltration and nanofiltration powders presented a dose-dependent inhibitory activity with not significant differences between them ( $IC_{50}$  of 0.457 and 0.429 mg mL<sup>-1</sup>, 260 261 respectively) as described in table 2. The IC<sub>50</sub> for the BSA-GLC assay were 0.100, 0.101, 0.400, 0.243 mg mL<sup>-1</sup> for AG, QE, HT, and VB, respectively (table 2). HT and VB are also effective in the 262 263 BSA-GLC system where the autoxidation of glucose is a side reaction to form reactive 264 dicarbonyl compounds that will promote the formation of AGEs. Control for maltodextrin and acacia fiber did not show antiglycative activity for BSA-GLC assay. 265

266 On the other hand, methylglyoxal (MGO) and glyoxal (GO) are key promoters of the 267 glycation process where the formation of AGEs and irreversible modification of proteins are 268 boosted. For this reason, the direct MGO and GO trapping capacity of UOMW and NOMW 269 extracts were investigated. Figures 2a and 2b depicted the MGO and GO trapping ability of 270 ultrafiltration and nanofiltration powders in the range from 0.1 to 2.5 mg mL<sup>-1</sup>. Both extracts 271 showed similar dose-dependent behavior and reached the highest MGO trapping abilities at 1 272 mg mL<sup>-1</sup>. It is noteworthy that control sample (maltodextrin + acacia fiber) showed a minor 273 dicarbonyl trapping capacity, representing nearly the 20% of the overall response. Results 274 shown that efficacy of OMW powder for trapping GO was lower as compared with the MGO trapping ability. The IC<sub>50</sub> in the GO-trapping assay was calculated to be 0.562 mg mL<sup>-1</sup> and 275 276 0.811 mg mL<sup>-1</sup> for UOMW and NOMW powders, respectively. In contrast, both powders were more effective for trapping MGO with  $IC_{50}$  values of 0.294 and 0.473 mg mL<sup>-1</sup>, respectively. The 277 278 ultrafiltrated powder was significantly (P<0.05) more reactive against MGO and GO than the 279 nanofiltrated powder for concentration higher than 0.1 mg mL<sup>-1</sup>.

280 In the early phase of the glycation reaction, the reversible Schiff base leads a 281 rearrangement where a more stable Amadori product is formed. In an advanced phase of 282 glycation, irreversible AGEs are formed. Therefore, inhibition of fructosamine adducts could 283 reduce AGEs formation in foods and hence potentially ameliorate the diabetic complications in 284 vivo by a restriction of the dietary AGE intake and reduction of the circulating AGEs. The 285 Amadori products were quantified principally on the basis of the reduction of NBT by Amadori 286 products to monoformazan dye, which bears strong absorbance at 530 nm. Our results 287 pointed out that antiglycative activity and dicarbonyl trapping ability of ultrafiltrated and 288 nanofiltrated powders are relevant to the potential mitigation of diet-derived and 289 endogenously formed AGEs. As shows in the **table 3**, both extracts performed their inhibitory 290 effects also at an early stage of the glycation since a significant reduction in the formation of fructosamine was observed at all concentrations (0.36-3.57mg mL<sup>-1</sup>). The IC<sub>50</sub> for the 291 fructosamine assay was calculated to 2.34 and 2.48 mg mL<sup>-1</sup> for the ultrafiltrated and 292 293 nanofiltrated powder, respectively.

Since the antiglycative activity of both ultrafiltrated and nanofiltrated powders were in the same order of magnitude, the ultrafiltrated sample was selected for a more detailed investigation to elucidate the mechanisms of the antiglycative action considering its higher yield during the production process and consequently the potential economic advantages in the scale up process.

299 Sequential fractionation of the UOMW powder by reverse phase extraction was applied to 300 further investigation of the antiglycative capacity. Figure 3 described the chromatographic 301 profile of the whole extract and relative fractions A, B, C and D, eluted with in a gradient with 302 methanol. In the chromatograms were identified the peaks of HT, tyrosol and verbascoside. A 303 hump corresponding to the high molecular weight polysaccharides present in the olive pulp 304 was clearly detectable. Since UOMW powder is particularly rich in HT (42% of the total 305 polyphenols content), further investigations were focused in the HT content. The highest 306 concentration of HT (representing approx. 80% of the initial HT content) was collected in the 307 fraction C which is eluted with 40% MetOH. Fraction D (corresponding to fraction eluted with 308 100% MetOH) contains the lowest lower amount of HT corresponding to approximatively the 309 20% of the initial content. However verbascoside is eluted at fraction D together a major 310 proportion of the polysaccharides rich in polyphenols.

311 The antiglycative activity of each fraction obtained by solid phase extraction was 312 estimated by the BSA-MGO (Figure 4a) and BSA-GLC (Figure 4b) assays. In both BSA-MGO and 313 BSA-GLC assay, fractions A and B did not exert protective effect on the formation of 314 fluorescent AGEs, while C and D fractions shown an important antiglycative capacity. Fractions 315 C and D were able to inhibit from 30 to 49% of the formation of fluorescent AGEs in the BSA-316 MGO assay, and 31.3 to 55.2% in the BSA-GLC assay. Antiglycative activity of fraction eluted 317 with methanol was significantly higher than fraction eluted with 40% methanol. The partial 318 contribution of the antiglycative capacity of UOMW-C and UOMW-D fractions could explain 319 the activity of whole UOMW extract that involved the 76.3 and 88.4% of inhibition, BSA-MGO, 320 and BSA-GLC, respectively.

321 Figure 5 shows the kinetic of the MGO trapping ability of the ultrafiltrated powder and 322 fractions over time. UOMW-C and UOMW-D fractions were highly efficient for trapping MGO 323 as compared with UOMW-A and UOMW-B fractions. In fact, UOMW-C and UOMW-D fractions 324 reached a trapping capacity of 80.2 and 71.4% during incubation with MGO up to 168h 325 respectively. Both C and D fractions follow the same trend and a significant MGO-trapping 326 capacity is recorded in the first 24h of incubation. The MGO trapping capacity of whole powder 327 (1 mg mL<sup>-1</sup>) peaked at 97.6% after 168 h of incubation, but system was almost saturated 328 (percentage of inhibition higher than 85%) at 72h of reaction. In contrast, fractions A and B just 329 accounted for the 11 and 14.5 % of the MGO trapping ability after 168h, respectively. It was 330 selected the reaction time at 48h to compare the relative reactivity of the fraction with the 331 whole UOMW powder. At 48h, UOMW powder reached a MGO-trapping capacity of 78.5% and 332 the sum of fractions accounted to the 99.3%.

333 Several assays have established a relationship between antioxidant effects of certain plant extracts with the inhibition of protein modifications in the glycation process<sup>25</sup>. Jemai et al.<sup>26</sup> 334 suggested that the antidiabetic effect of olive leaves might be due to antioxidant activity of HT 335 and oleuropein. In this regard, total antioxidant capacity by FRAP and ABTS were evaluated in 336 the samples. Total phenolic content of the powder and its fraction was assessed since phenolic 337 compounds have been proposed as major contributors to antiglycative activity<sup>27</sup>. Table 1 338 339 summarized the results of TPC, FRAP and ABTS for the UOMW powders and its respective fractions A, B, C, and D. The reducing ability of the UOMW powder (780.4  $\mu$ mol TEAC g<sup>-1</sup>) was 340 higher than the sum of the reducing activity of its UOMW-C and UOMW-D fractions (199.2 and 341 241.7 µmol TEAC g<sup>-1</sup> respectively). ABTS results showed that UOMW powder had a TEAC value 342 of 605.6 µmol g<sup>-1</sup> and the free radical scavenging activity of UOMW was due to its UOMW-C 343 and UOMW-D fractions. The total phenolic content of UOMW powder was 61.8 mg GAE g<sup>-1</sup> and 344 345 UOMW-C and UOMW-D fractions accounted practically the total phenol content of the whole 346 powder. Data suggested that the antiglycative action was related to the phenolic moiety of the 347 ingredients, separating the polysaccharides that were added in the formulation steps, the 348 biological activity is linearly correlated with the antioxidant ability.

349

### 350 4. Discussion

Olive mill wastewater is a by-product of the olive oil extraction industry posing a serious environmental problem due to its highly organic load. However, OMW is also an important source of polyphenol, such as hydroxytyrosol, tyrosol verbascoside and oleuropein, with potential health benefits. Consequently, during recent years, new technologies for the recovery of phenols compounds from OMW have been studied<sup>28</sup>.

356 It is known that under hyperglycaemic conditions, the glycation process leads to the formation and accumulation of AGEs, compounds able to modify function and structure 357 proteins of tissues and organs inducing some metabolic diseases. Ramkissoon et al.<sup>21</sup> reported 358 359 that many constituents in plants possess a high TPC and antioxidant potential as well as high 360 ability to inhibit protein glycation. The high incidence of chronic diseases such as diabetes, 361 Alzheimer or cardiovascular diseases and the key role of glycation process in the production of 362 these had been a high relevance in several reports that have appeared in the literature following such benefit<sup>9-29</sup>. In the organism the total pool of circulating AGEs is the sum of those 363 364 with dietary origin, including AGEs absorbed as such products as well as those formed in vivo from the absorbed dicarbonyl compounds, and those endogenously formed<sup>30</sup>. There is a direct 365

366 correlation between circulating AGE level, of which two-thirds remain in the body for 72 hours, and those consumed<sup>11</sup>. Consequently, the dietary AGE restriction is an efficient tool to 367 368 ameliorate the pathological consequences of the glycation process in vivo until effective and safe drugs become available<sup>11</sup>. Our investigation point out that the by-products from olive oil 369 370 must be considered as an efficient inhibitor of the formation of AGEs, and in extension very 371 likely will act as effective inhibitors of the glycation process. OMW extracts spray dried with 372 maltodextrin and acacia fiber interfered with the formation of Amadori product and also 373 reduced the formation of fluorescent AGE by trapping reactive dicarbonyls, such as MGO and 374 GO. After partial fractionation of the OMW powder, hydroxytyrosol and verbascoside revealed 375 as the most active compound exerting the antiglycative and dicarbonyl trapping activity. 376 Despite the promising data obtained in vitro, in vivo studies are needed in a further step as 377 well as metabolic studies to clarify its mechanism of action.

378

### 379 5. Conclusion

380 Results suggest that OMW powder would also act at the early stage of the glycation 381 process being effective inhibitors of the fructosamine formation. The inhibition of dicarbonyl 382 compounds formation by OMW powder was in the same magnitude than that of 383 pyridoxamine, hydroxytyrosol, and verbascoside. On the other hand, it is plausible that the 384 antiglycative capacity and antioxidant properties of extract are due to its phenolic content, 385 especially hydroxytyrosol and verbascoside, but also by the presence of other high molecular 386 compounds. The high molecular weight compounds present in the olive as cellulose, arabinans and pectins are present as well in the OMW powder and a significant part of the phenolic 387 compounds are bound to the polysaccharide moiety<sup>31</sup> as shown by the fact that they are 388 389 present in the ultrafiltration fraction. Olive mill wastewater powder could be considered as a 390 potent antiglycative ingredient which was found to inhibit AGEs formation at different stages 391 of the pathway. In addition, direct trapping of dicarbonyl compounds have been confirmed as 392 a novel mechanism of action apart of the already known antioxidant capacity. Nevertheless, 393 further research is ongoing in food models to confirm the mitigation of AGE formation and the 394 mechanism of action before to conclude firmly that OMW powder is a diet AGE inhibitor.

395

#### 396 Acknowledgements

This work was funded by project AVANSECAL (S2013/ABI-3028, Madrid Community, ES), CSIC 201370E027, program PON01\_ 02863 (Italian Ministry of University and Scientific Research,

- 399 MIUR), and APQ project (Regione Calabria, Italy). M. Navarro thanks additional funding by the
- 400 JAE program (Spanish National Research Council).

402	Ref	erences
403	1.	C. Manna, P. Galletti, G. Maisto, V. Cucciolla, S. D'Angelo, and V. Zappia, Transport
404		mechanism and metabolism of olive oil hydroxytyrosol in Caco-2 cells, FEBS letters, 2000,
405		<b>470</b> , 341-344.
406	2.	E. Miro-Casas, M. I. Covas, M. Farre, M. Fito, J. Ortuño, T. Weinbrenner et al.,
407		Hydroxytyrosol Disposition in Humans, Clin. Chem., 2003, 49, 945-952.
408	3.	E. Escrich, R. Moral, and M. Solanas, Olive oil, An essential component of the
409		Mediterranean diet, and breast cancer, Public Health Nutr., 2011, 14, 2323-2332.
410	4.	T. Psaltopoulou, R. I. Kosti, D. Haidopoulos, M. Dimopoulos, and D. B. Panagiotakos, Olive
411		oil intake is inversely related to cancer prevalence: a systematic review and a
412		metaanalysis of 13800 patients and 23340 controls in 19 observational studies, Lipids
413		Health Dis., 2011, <b>10</b> , 127.
414	5.	M. De Bock, E. B. Thorstensen, J. G. B. Derraik, H. V. Henderson, P. L. Hofman, and W. S.
415		Cutfield, Human absorption and metabolism of oleuropein and hydroxytyrosol ingested as
416		olive (Olea europaea L.) leaf extract, Mol. Nutr. Food Res., 2013, 57, 2079-2085.
417	6.	S. D'Angelo, C. Manna, V. Migliardi, O. Mazzoni, P. Morrica, G. Capasso G et al.,
418		Pharmacokinetics and metabolism of hydroxytyrosol, a natural antioxidant from olive oil,
419		Drug Metabolism and Disposition, 2001, <b>29</b> , 1492-1498.
420	7.	S. Martín-Peláez, M. I. Covas, M. Fitó, A. Kusar, I. Pravst, Health effects of olive oil
421		polyphenols: recent advances and possibilities for the use of health claims, Mol. Nutr.
422		Food Res., 2013, <b>57</b> , 760-771.
423	8.	K. Hamden, N. Allouche, M. Damak, and A. Elfeki, Hypoglycemic and antioxidant effects of
424		phenolic extracts and purified hydroxytyrosol from olive mill waste in vitro and in rats,
425		Chem-Biol. Interact., 2009, <b>180</b> , 421-432.
426	9.	M. W. Poulsen, R. V. Hedegaard, J. M. Anderson, B. De Courten, S. Bügel, J. Nielsen et al.,
427		Advanced glycation endproducts in food and their effects on health. Food Chem. Toxicol.,
428		2013, <b>60</b> , 10-37.
429	10.	A. Meeprom, W. Sompong, C. B. Chan, and S. Adisakwattana, Isoferulic Acid, a new anti-
430		glycation agent, inhibits fructose-and glucose-mediated protein glycation in vitro,
431		Molecules, 2013, <b>18</b> , 6439-6454.
432	11.	H. Vlassara, and J. Uribarri, Advanced Glycation End Products (AGE) and Diabetes: Cause,
433		Effect, or Both? Curr. Diabetes Rep., 2014, 14, 1-10.

434 12. S. Rahbar, and J.L. Figarola, Inhibitors and breakers of advanced glycation endproducts
435 (AGEs): a review, Current Medicinal Chemistry-Immunology, *Endrocr. Metab. Agents*,
436 2002, **27**, 135-161.

- 437 13. M.I. Fernández-Mar, R. Mateos, M.C. García-Parrilla, B. Puertas, and E. Cantos-Villar,
  438 Bioactive compounds in wine: resveratrol, hydroxytyrosol and melatonin: a review, *Food*439 *Chem.*, 2012, **130**, 797-813.
- 440 14. A.D. Troise, A. Fiore, A. Colantuono, S. Kokkinidou, D.G. Peterson and V. Fogliano, Effect of
  441 Olive Mill Wastewater Phenol Compounds on Reactive Carbonyl Species and Maillard
  442 Reaction End-Products in Ultrahigh-Temperature-Treated Milk. *J. Agric. Food Chem.* 2014,
  443 62, 10092-10100.
- K. Hamden, N. Allouche, B. Jouadi, S. El-Fazaa, N. Gharbi, S. Carreau et al., Inhibitory
  action of purified hydroxytyrosol from stored olive mill waste on intestinal disaccharidases
  and lipase activities and pancreatic toxicity in diabetic rats, *Food Sci. Biotechnol.*, 2010, 19,
  439-447.
- 448 16. S. Kokkinidou, and D.G. Peterson, Response surface methodology as optimization strategy
  449 for reduction of reactive carbonyl species in foods by means of phenolic chemistry, *Food*450 *Funct.*, 2013, **4**, 1093-1104.
- 451 17. M. Contini, S. Baccelloni, R. Massantini, G. Anelli, Extraction of natural antioxidants from
  452 hazelnut (Corylus avellana L.) shell and skin wastes by long maceration at room
  453 temperature, *Food Chem.*, 2008, **110**, 659-669.
- M. Mesías, M. Navarro, V. Gökmen, and F. J. Morales, Antiglycative effect of fruit and
  vegetable seed extracts: inhibition of AGE formation and carbonyl trapping abilities, *J. Sci. Food Agric.*, 2013, **3**, 2037–2044.
- 457 19. F. J. Morales, S. Martín, O. C. Açar, G. Arribas-Lorenzo, and V. Gökmen, Antioxidant
  458 activity of cookies and its relationship with heat-processing contaminants: a risk/benefit
  459 approach. *Eur. Food Res. Technol.*, 2009, **228**, 345–354.
- 460 20. J. R. Baker, D. V. Zyzak, S. R. Thorpe, and J. W. Baynes (1994) Chemistry of the
  461 fructosamine assay: D-Glucosone is the product of oxidation of Amadori compounds,
  462 *Clinical Chem.*, 1994, **40**, 1950-1955.
- 463 21. J. S. Ramkissoon, M. F. Mahomoodally, N. Ahmed, and A. H. Subratty, Antioxidant and
  464 anti-glycation activities correlates with phenolic composition of tropical medicinal herbs,
  465 Asian Pac. J. Trop. Med., 2013, 6, 561-569.
- 466 22. M. A. Coimbra, S. M. Cardoso, and J. A. Lopes-da-Silva, Olive Pomace, a Source for
  467 Valuable Arabinan-Rich Pectic Polysaccharides. Ed. Rauter, AP; Vogel, P; Queneau, Y.

Page 17 of 26

# Food & Function

468		Carbohydrates in sustainable development in: renewable resources for chemistry and
469		biotechnology Book Series: Topics in Current Chemistry 2010, <b>294</b> , 129-141.
470	23.	N. Ahmed, Advanced glycation endproducts - role in pathology of diabetic complications,
471		Diabetes Res. Clin. Pr., 2005, <b>67</b> , 3-21.
472	24.	J-W Wu, C-L Hsieh, H-Y Wang, and H-Y Chen, Inhibitory effects of guava (Psidium guajava
473		L.) leaf extracts and its active compounds on the glycation process of protein, Food Chem.,
474		2009, <b>113</b> , 78-84.
475	25.	C. H. Wu, and G. C. Ten, Inhibitory effect of naturally occurring flavonoids on the
476		formation of advanced glycation endproducts, J. Agric. Food Chem., 2005, 53, 3167–3173.
477	26.	H. Jemai, A. El Feki, and S. Sayadi, Antidiabetic and Antioxidant Effects of Hydroxytyrosol
478		and Oleuropein from Olive Leaves in Alloxan-Diabetic Rats, J. Agric. Food Chem., 2009, 57,
479		8798-8804.
480	27.	X. Peng, K. W. Cheng, J. Ma, B. Chen, C. T. Ho, and C. Lo et al, Cinnamon bark
481		proanthocyanidins as reactive carbonyl scavengers to prevent the formation of advanced
482		glycation endproducts, J. Agric. Food Chem., 2008, 56, 1907–1911.
483	28.	A. M. Giuffrè, V. Sicari, A. Piscopo, and L. Lovadj, Antioxidant activity of olive oil mill
484		wastewater obtained from different thermal treatments, Grasas y aceites, 2012, 63, 209-
485		213.
486	29.	C. Delgado-Andrade, Maillard reaction products: some considerations on their health
487		effects, Clin. Chem. Lab. Med., 2014, 52, 53-60.
488	30.	J. Uribarri, W. Cai, O. Sandu, M. Peppa, T. Goldberg, and H. Vlassara, Diet-derived
489		advanced glycation end products are major contributors to the body 's AGE pool and
490		induce inflammation in healthy subjects, Ann. NY Acad. Sci., 2005, <b>1043</b> , 461–466.
491	31.	V. Gökmen, A. Serpen, and V. Fogliano, Direct measurement of the total antioxidant
492		capacity of foods: the `QUENCHER' approach. <i>Trends Food Sci. Tech.</i> , 2009, <b>20</b> , 278-288.
493		

494 **Table 1.** Antioxidant activity and total phenol content of nanofiltrated olive mill wastewater 495 (NOMW), ultrafiltrated olive mill wastewater (UOMW) powders and its fractions obtained by 496 reverse-phase extraction. Results are expressed as mean  $\pm$  SD for n = 4. Different letters in the 497 same column denote significant differences P < 0.05. LoQ < 0.5 GAE mg g<sup>-1</sup> or < 1.1 TEAC µmol 498 g<sup>-1</sup>.

499

	TFC	FRAP	ABTS
	GAE mg g <sup>-1</sup>	TEAC μmol g⁻¹	TEAC μmol g⁻¹
Sample			
NOMW	49.9 ± 5.5 <sup>b</sup>	439.4 ± 15.5 <sup>b</sup>	333.7 ± 34.9 <sup>b</sup>
UOMW	$61.8 \pm 3.4$ <sup>a</sup>	$780.4 \pm 23.0$ <sup>a</sup>	605.6 ± 28.8 <sup>a</sup>
Fractions			
UOMW-A	< LoQ	< LoQ	$2.0 \pm 0.4$ <sup>c</sup>
UOMW-B	< LoQ	$5.0 \pm 0.1$ <sup>c</sup>	$5.0 \pm 0.3$ <sup>c</sup>
UOMW-C	$24.6 \pm 0.2$ <sup>c</sup>	199.2 ± 2.6 <sup>d</sup>	249.3 ± 29.6 <sup>d</sup>
UOMW-D	$30.0 \pm 0.2$ <sup>c</sup>	$241.7 \pm 17.1$ <sup>d</sup>	$379.8 \pm 34.0$ <sup>b</sup>

500

Table 2. Concentration (mg mL<sup>-1</sup>) of aminoguanidine, pyridoxamine, quercetin, hydroxytyrosol,
tyrosol, verbascoside, and ultrafiltrated (UOMW) and nanofiltrated (NOMW) olive mill
wastewater powders exerting the 50% inhibition of BSA glycation. Different letters in the same
column denote significant differences P < 0.05 in the same column. IC<sub>50</sub> higher than 1 mg/mL
(n.a.)

507

	IC <sub>50</sub> - Antiglycative activity		
	BSA-GLC assay	BSA-MGO assay	
Aminoguanidine	0.100 <sup>a</sup>	0.090 <sup>ª</sup>	
Pyridoxamine	0.283 <sup>b</sup>	0.046 <sup>b</sup>	
Quercetin	0.101 <sup>ª</sup>	0.053 <sup>b</sup>	
Hydroxytyrosol	0.400 <sup>c</sup>	0.086ª	
Tyrosol	n.a.	n.a.	
Verbascoside	0.243 <sup>b</sup>	0.053 <sup>b</sup>	
UOMW powder	0.457 <sup>c</sup>	0.886 <sup>c</sup>	
NOMW powder	0.429 <sup>c</sup>	1.029 <sup>d</sup>	

508

510 **Table 3.** Percentage of inhibition of the formation of fructosamine by ultrafiltrated olive mill

511 wastewater (UOMW) and nanofiltrated olive mill wastewater (NOMW) powders in the range

512 from 0.36 - 3.57 mg mL<sup>-1</sup>. Results are expressed as mean  $\pm$  SD for n = 4. Different letters in the

513 same column denote significant differences P < 0.05.

514

mg mL <sup>-1</sup>	UOMW	NOMW
0.36	10.02 ± 5.45 <sup>a</sup>	$10.54 \pm 2.65$ <sup>a</sup>
0.71	25.48 ± 1.90 <sup>b</sup>	$33.52 \pm 1.61$ <sup>b</sup>
1.43	44.67 ± 2.92 <sup>c</sup>	$43.67 \pm 0.94$ <sup>c</sup>
3.57	$57.31 \pm 1.90$ <sup>d</sup>	$56.75 \pm 6.25$ <sup>d</sup>

515

517 Figure 1. Antiglycative activity of ultrafiltrated (solid bar) and nanofiltrated (empty bar) olive mill wastewater OMW extracts in the range from 0.14 to 1.43 mg mL<sup>-1</sup> on the formation of 518 fluorescent AGEs in (A) BSA-MGO and (B) BSA-GLC assays. Aminoguanidine (AG), quercetin 519 (QE), hydroxytyrosol (HT), and verbascoside (VB) at 0.14 mg mL<sup>-1</sup> were used as reference and 520 showed an antiglycative activity of 67.7%, 86.3%, 69.8%, and 84.4% respectively in BSA-MGO 521 522 and 64.3%, 63.1%, 26.5% and 54.4% respectively in BSA-GLC. AG was used as control of the 523 assay. Maltodextrin and acacia fiber (MD+AF, 1.43 mg/mL) was used as control of the powder. 524 Results are expressed as mean  $\pm$  SD for n = 4. Different letters mean significant differences (P < 525 0.05) between samples for a same concentration.

Figure 2. MGO-trapping capacity (A) and GO-trapping capacity (B) of ultrafiltrated olive mill wastewater (UOMW,●) and nanofiltrated olive mill wastewater (NOMW, ○) powders (0.1 - 2.5 mg mL<sup>-1</sup>) after incubation (168h). Results are expressed as mean ± SD for n = 4. PM (0.1 mg mL<sup>-1</sup>) was used as control positive and presented and MGO and GO trapping capacity of 99.6% and 14.5% respectively. Control for maltodextrin + acacia fiber depicted as dotted line.

Figure 3. Chromatographic profile of the ultrafiltrated olive mill wastewater (UOMW) powder
and fractions (UOMW-A, OUMW-B, UOMW-C and UOMW-D) fractions. See materials and
methods for details. Butyl -4-hydroxybenzoate (IS).

Figure 4. Antiglycative activity of ultrafiltrated olive mill wastewater (UOMW) powder and fractions in the BSA-MGO (A) and in BSA-GLC (B) assay. UOMW was used as a reference presenting an antiglycative activity of 76.3% and 88.4% for BSA-MGO and BSA-GLC, respectively. Control for maltodextrin and acacia fiber has not activity as described in figure 1. Results are expressed as mean  $\pm$  SD for n = 4. Different letters denote significant differences P <0.05.

Figure 5. Methylglyoxal trapping activity of ultrafiltrate olive mill wastewater powder and
fractions over time. Ultrafiltrated olive mill wastewater (UOMW) powder (◆) and UOMW-A
UOMW-B (□), UOMW-C (●), UOMW-D (○) fraction





Food & Function Accepted Manuscript





