Polymer Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/polymers

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxx

ARTICLE TYPE

Synthesis of Hyperbranched Polymers and their Applications in Analytical Chemistry

Fengxia Sun,^{*a,b*} Xiaoling Luo,^{*a,b*} Lichao Kang,^{*a,b*} Xiayu Peng^{* c} and Chunxia Lu^{* *a,b*} DOI: 10.1039/b000000x

⁵ Hyperbranched polymers (HBPs) are increasingly attracting the attention of scientists due to their unique physical and chemical properties, as well as their potential applications in different fields. Many methods for the preparation of HBPs have hitherto been reported. This review focuses primarily on several novel synthesis methods for various HBPs. We present and discuss the advantages and disadvantages as well as recent progress relating to these synthesis methods, namely the slow monomer addition method, click chemistry, and the enzyme-catalyzed polymerization method. Besides, the applications of HBPs in analytical chemistry fields such as sample ¹⁰ pretreatment and immunosensors are also briefly summarized.

1. Introduction

Over the past thirty years, highly branched polymers (especially dendritic polymers) have received considerable attention due to their remarkable properties and potential 15 applications. Dendritic polymers, including dendrimers and hyperbranched polymers (HBPs), are the fourth major class of polymers after linear, branched, and cross-linked architectures. Dendrimers are three-dimensional monodisperse polymers with a perfectly tree-like globular ordered structure, and consist of

- ²⁰ branched units and terminal units. The synthesis of dendrimers requires multi-step approaches, including protection, deprotection, and complex purification steps. HBPs are three-dimensional polydisperse systems with an elliptic randomly branched structure, and consist of linear units, branched units, and terminal
- ²⁵ units. Compared to their dendrimeric analogues, HBPs have irregular shapes and are not perfectly symmetrical, but display comparable properties and can be easily synthesized through "one-pot" reactions.¹⁻⁵
- Although the theoretical possibility of the synthesis of HBPs ³⁰ was predicted by Flory in the middle of the 20th century, the first HBP (hyperbranched polyphenylene) was only intentionally synthesized by Kim and Webster in 1988. Hyperbranched polymers (HBPs) have since attracted much attention due to their unique physical and chemical properties, such as higher solubility,
- ³⁵ lower viscosity, non-/low entanglement, and large numbers of terminal functional groups. To date, numerous HBPs have been prepared by different methods, including the polycondensation of AB_n monomers, self-condensing vinyl polymerization, ring-opening multibranching polymerization, and so on.^{6–11}
- ⁴⁰ Most early studies on HBPs were focused on the synthetic methods, characterization, and industrial applications, such as in the fields of coatings, cross-linking or adhesive agents, rheology modifiers and processing aids, encapsulation nanomaterials, or the fabrication of organic–inorganic hybrids.^{12,13} More recently,
- ⁴⁵ many delicate supramolecular structures, such as nano- or microscale vesicles, micelles, fibers, films, and tubules, have been prepared through the self-assembly of HBPs.¹⁴⁻¹⁹ These

characteristics make HBPs very useful and demonstrated excellent potential applications in biomedical fields, such as 50 bioimaging, sensors, drug delivery, antifouling materials, tissue engineering, and cytomimetics, which has attracted considerable research interest and thereby promoted their rapid development. Details of these applications are provided in several recent papers

and reviews.²⁰⁻²³ There have been some review papers on the synthesis and applications of HBPs. For example, in 2004, Gao and Yates wrote excellent reviews on these topics, respectively.^{2,24} In 2012, Jin *et al.* wrote a tutorial review focused on the self-assembly of HBPs and their cytomimetic applications.²⁵ However, new ⁶⁰ reports appear quite frequently in this fast-moving field. In this review, we focus on recent developments concerning novel polymerization methods for HBPs, and their applications in analytical chemistry are also briefly summarized.

2. Synthesis of hyperbranched polymers

65 2.1 Slow monomer addition method

Since the first HBPs were synthesized from AB₂-type monomers, numerous HBPs have been prepared by various methods. The conventional approaches to HBPs, including the polycondensation of AB_n monomers, self-condensing vinyl ⁷⁰ polymerization, proton-transfer polymerization (PTP), and the so-called "A₂+B₃" methodology, each have certain merits, such as facile preparation or commercial availability of the monomers. However, these methods are still associated with some crucial problems, such as a low degree of branching (DB), very broad ⁷⁵ polydispersity (PDI), and uncontrollable gelation, which greatly limit the advanced application of HBPs.^{26,27}

Hence, researchers have been searching for improved methods and a new strategy is eagerly anticipated. Polymerization under slow monomer addition (SMA) conditions offers a versatile and favorable method for the preparation of HBPs with controlled molecular weight, high DB, and narrow PDI. Thus, the SMA method has received significant attention. This synthetic method may be divided into two major categories

Monomer	Condition	Types of polymers	Mw	PDI	Ref
	Pd ₂ (dba) ₃ ,CuI, PPh ₃ , piperidine, rt	phenylacetylene	8000-90,000	1.3-8.5	[28]
	1,1,1-Tris(hydroxymethyl)propane, 95°C over 12h, under argon atmosphere	polyglycerols	1250-6500	1.13-1.47	[29]
о	BTSSbF ₆ , 100°C in bulk	polyethers	594-5336	1.25-1.58	[30]
OF	trimethylolpropane (TMP), 95°C for 17h in dioxane	polyglycerols	4000-1475,000	1.1-1.4	[33]
OF	80°C for 12h in THF	polyglycerol	300-24,000	1.3-1.8	[34]
	TMP, 170°C over 12h	aliphatic polyether	830-1020	1.2-1.3	[35]
	BF ₃ ·OEt ₂ , 10°C for 24h in dry CH ₂ Cl ₂	glycopolymer	7400-122,400	1.70-1.45	[36]

Table 1. The monomers used in SMA and the polymerization results

35

according to the monomer used: step-growth polycondensation by SMA and ring-opening multibranching polymerization by SMA. The former route was first reported by Bharathi and Moore, who used 3,5-diiodophenylacetylene as an AB₂-type monomer, 1-

- 40 (3,5-diiodophenyl)-3,3-diethyltriazene as a B₂ core molecule, and Pd₂(dibenzylideneacetone)₃ as a catalyst; a hyperbranched polyphenylacetylene of controlled molecular weight was thereby synthesized.²⁸ Sunder et al. first synthesized hyperbranched polyglycerols by anionic ring-opening multibranching
 45 polymerization of glycidol under SMA conditions.²⁹ In contrast, Magnusson and co-workers synthesized hyperbranched polyethers by cationic ring-opening multibranching
- polyethers by cationic ring-opening multibranching polymerization under SMA conditions.^{30,31} To date, numerous HBPs have been synthesized by the SMA
- ⁵⁰ method, such as hyperbranched poly(siloxysilane),³² hyperbranched polyglycerol,³³⁻³⁵ and hyperbranched glycopolymer.³⁶ Monomers used in SAM and polymerization results such as molecular weight (Mw), and polydisperity index (PDI, Mw/Mn) are summarized in Table 1.
- ⁵⁵ In addition, Zhou and co-workers have studied the kinetics of the SMA method in the synthesis of hyperbranched polymers.³⁷ Both theory and experiment showed the SMA technique to be effective for increasing the molecular weight, reducing the degree

of intramolecular cyclization, and controlling the polydispersity ⁹⁵ index of the final hyperbranched polymers at less than 2. Thus, well-defined HBPs synthesized by the SMA method should be a prospective future field of great interest.

2.2 Click chemistry

The concept of click chemistry was first proposed by ¹⁰⁰ Sharpless and co-workers in 2001. Since then, this technique has been extensively used in polymer science due to its remarkable features, such as fast reaction rates, mild reaction conditions, regioselectivity, and simple product isolation procedures.³⁸⁻⁴² Click chemistry may be divided into two main types, that is, the ¹⁰⁵ azide–alkyne click reaction and the thiol–ene/yne click reaction.

2.2.1 Azide-alkyne click polymerization

Copper-catalyzed azide–alkyne cycloaddition (CuAAC) is the most widely employed type of click chemistry in the synthesis and post-polymerization functionalization of HBPs. Since Voit 110 and co-workers first synthesized hyperbranched poly([1,2,3]triazole)s from AB₂ monomers by the CuAAC method, many polymer scientists have also carried out such experiments.⁴³ Table 2 lists the monomers used to synthesize

Monomer	Condition	Mw	PDI	Ref
	SMA; CuSO4, SA; under nitrogen for 24h in DMF	5600-6800	1.27-1.34	[48]
	Cp*Ru(PPh ₃) ₂ Cl, THF, 60°C, 30-50min;	5400-9400	2.4-2.7	[49]
$N \equiv N^+ - N^ (CH_2)_m - O - (CH_2)_m - N^ N^+ \equiv N$	[Cp*RuCl ₂] <i>n</i> , THF, 40°C, 2h	5000-7000	1.8-2.2	
m = 4 (2a), 6 (2b)	Cu(PPh ₃) ₃ Br, 60°C for 5-7h in DMF	9800-12,400	4.32-4.28	[50]
$N \equiv N^* - N - R - \bigcirc \bigcirc$	metal-free, in DMF/toluene (1:1 volume ratio), 100 °C for 6 h	16,300-25,200	2.0-2.1	[54]
	metal-free, Monomers mol ratio 1:1; in DMF/toluene, 100 °C for 6 h	15,100-15,400	1.4-1.5	[55]
$R = -CH_{2^{-}}(2a), -O(CH_{2})_{6^{-}}(2b)$ $= -CH_{2^{-}}(2a), -O(CH_{2})_{6^{-}}(2b)$	metal-free, in DMF/toluene mixture (1:1, v/v) at 100 °C for 24 h; under nitrogen	10,500-18,700	1.27-1.53	[56]
$ \begin{array}{c} R^{a} \\ \hline \\ 1 \\ 2a \\ 2b \\ 2c \\ 2c \\ R^{a} \\ \hline $	under nitrogen metal-free, in DMF, 60°C for 5h without protection from oxygen and moisture	4400-7200	2.21-2.65	[57]
$N_3 = (CH_2)_6 = O - R^* - O - (CH_2)_6 = N_3$ $R^* = N_3 = N_3$	CuI@A-21, in THF, 60°C for 12h	17,900-69,600	1.86-3.25	[58]
$R^{R^{-}}$ R^{R	conditions of the Cu solvents, and so on, a solubilities. ^{46,47} For e	AAC in terms nd obtained a ser xample, Xie and azole by an A ₂ +	of the cataly ies of HBPs co-workers	vtic sy with h synthe

Table 2. The monomers used in synthesis polytriazoles by click polymerizations and the results.

conditions of the CuAAC in terms of the catalytic system, solvents, and so on, and obtained a series of HBPs with higher ¹¹⁰ solubilities.^{46,47} For example, Xie and co-workers synthesized hyperbranched polytriazole by an A₂+B₃ route based on Cucatalyzed click reaction and subsequently characterized the product. The obtained hyperbranched polytriazole showed high solubility in common organic solvents such as cyclopentanone, ¹¹⁵ cyclohexanone, dimethyl formamide, and dimethyl sulfoxide, and also displayed nonlinear optical properties.⁴⁸ Qin and co-workers

HBPs showed poor solubility and were difficult to characterize.

synthesized HBPs were characterized by IR, ¹H NMR, and size-

workers tried to prepare HBPs from A₂+B₃-type monomers under

CuSO₄/SA catalysis.⁴⁵ Unfortunately, however, the obtained

55 exclusion liquid chromatography.44 In 2008, Katritzky and co-

successfully prepared a series of functional hyperbranched polytriazoles with high molecular weight from A₂+B₃ monomer pairs by Cu-catalyzed click polymerization.^{49,50} The obtained functional hyperbranched polytriazoles proved to be soluble in 5 common organic solvents. In addition, Wu and co-workers

- reported the synthesis of a new series of hyperbranched polytriazoles containing perfluoroaromatic rings by click chemistry under copper catalysis.⁵¹ By virtue of its great feasibility, the CuAAC reaction has proven effective for various
- ¹⁰ material construction applications, involving small molecules, linear polymers, HBPs, and other nanoscale or macroscopic materials. The resulting products have been widely applied in different areas.⁵² However, the copper catalyst could be used only once, and the removal of the copper residues from the final HBPs
- ¹⁵ is very difficult. This limits the application of the products in optoelectronic devices and biomedical fields because even very small amounts of catalyst residues are toxic and may be detrimental to the photophysical properties or have fatal consequences for organisms. Thus, new metal catalyst systems as that could greatly decrease the conper residues in the resulting
- 20 that could greatly decrease the copper residues in the resulting HBPs, as well as metal-free click polymerization, need to be further explored.

Indeed, metal-free click polymerization has attracted wide interest due to its excellent features, such as environmental ²⁵ friendliness and lower toxicity. For example, Tang's group has carried out extensive work, and developed an efficient metal-free click polymerization that was successfully applied for the preparation of functional poly(aroyltriazole)s.^{53,54} Furthermore, they extended the range of alkyne monomers used for metal-free

- ³⁰ click polymerization from aroylacetylenes to propiolates.⁵⁵ The results showed that the metal-free click polymerization method proceeded readily under ambient atmosphere without protection from oxygen and moisture. Subsequently, these authors also efficiently synthesized a kind of polyester with high molecular
- ³⁵ weight and regioregularity from diazide monomers by the metalfree click polymerization.⁵⁶

Scheme 1. Synthesis of hyperbranched polytriazoles via metal-free click polymerization (reproduced with permission of the American Chemical ⁴⁰ Society from ref. 57).



Metal-free click polymerization has become a versatile and ⁵⁵ powerful tool for the preparation of functional polytriazoles. However, hyperbranched polytriazoles have rarely been synthesized in the above-mentioned reports. On the basis of previous work, hyperbranched polytriazoles with high molecular weights and high regioregularities were synthesized from tripropiolates and tetraphenylethene-containing diazides by metal-free click polymerization under the optimized reaction conditions without protection from oxygen and moisture. ⁵⁷ This synthesis is shown in Scheme 1. The obtained HBPs were soluble in common organic solvents and showed high thermal stability, ⁶⁵ and also showed a unique propensity for aggregation-induced emission, making them potentially useful as fluorescent chemosensors for the superamplified detection of explosives. Furthermore, the resultant polymers can be readily photo-crosslinked, yielding two-dimensional fluorescent patterns with high 70 resolution.

Fig.1 Synthetic routes to polytriazoles by click polymerizations using recyclable and reusable supported Cu(I) catalyst (reproduced with permission of the Nature Publishing Group from ref. 58).



Besides using the approach of metal-free click 95 polymerization, more recently, Tang's group developed a new alternative strategy to solve these difficulties.⁵⁸ In their study, hyperbranched polytriazoles were synthesized through click polymerizations using a recyclable and reusable supported Cu(I) catalyst (SCC) designated as CuI@A-21, in which the catalyst 100 was immobilized on polystyrene spheres. This strategy is shown in Figure 1. Moreover, the advantage of CuI@A-21 over other copper catalysts was also studied. The results indicated that CuI@A-21efficiently catalyzed azide-alkyne click polymerizations and could be reused over at least four cycles, and ¹⁰⁵ the final polytriazoles had copper residues of less than 116 ppm, which was 24 times lower than for other Cu(I) catalysts. More importantly, this method should broaden the application of azide-alkyne click polymerizations in optoelectronic and biomedical fields due to the lower copper residues, and also 110 provide guidelines for other polymerizations with metal catalysts in polymer science.

In summary, with further research on new metal catalyst systems and metal-free click polymerization, the azide–alkyne click polymerization is expected to be at the forefront in the ¹¹⁵ preparation and application of HBPs.

2.2.2 Thiol-ene/yne reaction

During the course of the development of azide–alkyne click polymerizations, the thiol–ene/yne reaction was also successfully developed and may be included in the catalogue of click ⁵ polymerizations.⁵⁹⁻⁶¹ The thiol–ene click reaction has been used industrially since the 1970s, but has recently been used to prepare many types of dendrimers and hyperbranched polymers because of the rapid, robust, and efficient nature of the reaction.⁶²⁻⁶⁴

For instance, Ramakrishnan and co-workers reported a novel ¹⁰ route for the synthesis of peripherally "clickable" hyperbranched polyethers by click reaction.⁶⁵ The method was based on selfcondensation of a specifically designed AB₂-type monomer, bearing two propargylbenzyl ether groups and one hydroxyl group, and yielded a hyperbranched polyether bearing numerous

¹⁵ peripheral allyl groups. It is noteworthy that these polymers can be readily transformed into interesting functional systems by clicking with a variety of thiols, and thus can be used as nanodimensional clickable hyperscaffolds in various areas.

In 2013, Xue and co-workers successfully prepared a novel ²⁰ silicon-containing hyperbranched polymer by step-growth thiol–ene click polymerization under conditions of UV irradiation, using mercaptopropylmethyldiallylsilane as an AB₂-type monomer and mercaptopropyltriallylsilane as AB₃-type monomer, respectively.⁶⁶ The obtained Si-HBPs were characterized by

²⁵ NMR spectrometry, Fourier-transform infrared spectroscopy, gelpermeation chromatography, and matrix-assisted laser desorption/ionization time-of-flight mass spectrometry. The results showed that the obtained Si-HBPs could be employed as promising candidate materials for heavy metal adsorption due to ³⁰ the thioether bond.

Foix and co-workers synthesized a new multifunctional HBP with thioether and ester moieties in the structure and terminal hydroxyl groups by an iterative synthetic route, which consisted of esterification and thiol–ene click reaction.⁶⁷ The obtained HBP

- ³⁵ was used as a macroinitiator in the curing process of a commercially available cycloaliphatic epoxy resin, leading to a dual latent curing system promoted by UV irradiation. The study provided a latent curing agent, which showed no activity under normal conditions but became active in response to external
- ⁴⁰ stimuli, thus simplifying the curing process. However, the main drawback of this route was the tedious synthetic process for obtaining the thioether–ester HBP structure, which required a five-step iterative methodology.

To circumvent this drawback, the same research group 45 sought to improve the method. Recently, they devised a new twostage dual photoinitiated-thermal curing system for cycloaliphatic epoxy resins by the addition of an allyl-terminated hyperbranched polyester and a trithiol compound to the formulation. The allylterminated HBP obtained was used as a reactive modifier to 50 achieve high glass transition temperature (T_p) thermosets and to

- ⁵⁰ achieve high grass transition temperature (T_g) thermosets and to enhance the mechanical properties.⁶⁸ Moreover, Imbesi and coworkers reported a heterogeneous, amphiphilic fluoropolymer bearing pendant alkene functionalities that were crosslinked to produce a polymeric network by photo-initiated thiol–ene click
- ⁵⁵ reaction to generate a benchmark, anti-biofouling surface.⁶⁹ More importantly, this study provided a more robust thiol-ene crosslinking chemistry to address the mechanical and processing drawbacks of previous systems, such as high coating deposition

temperatures, long reaction times, and so on. To date, numerous ⁶⁰ multi-functional HBPs have been prepared by thiol-ene click reaction in the presence of different initiators, including hyperbranched poly(ester-amine), hyperbranched glycopolymers, and so forth.^{70,71}

Similar to the thiol-ene click reaction, the thiol-yne reaction ⁶⁵ is also an important type of click reaction. In fact, thiol-yne click polymerization has been widely applied to synthesize functional polymers with novel structures, which have found application in various areas, such as photovoltaic materials, high refractive index optical materials, drug-delivery vehicles, and ⁷⁰ biomaterials.⁷²⁻⁷⁵

Scheme 2. Synthetic routes to hyperbranched polymers from small molecules by the thiol-yne click polymerizations (reproduced with permission of the American Chemical Society from ref. 76)



For example, Perrier and co-workers reported a versatile approach to the synthesis of functional hyperbranched polymers 95 by photoinitiated thiol-yne click polymerization, employing 2,2dimethoxy-2-phenylacetophenone as photoinitiator.⁷⁶ This strategy for synthesized process is shown in Scheme 2. In this study, two sulfur-containing HBPs were synthesized from small molecules and macromolecules, namely prop-2-ynyl-3-100 mercaptopropanoate (PYMP) and alkyne-terminated polystyrene thiol (ATPST), respectively. Compared to conventional polymerization methods, the degree of branching of the obtained HBPs was much higher by this method. The results indicated that various functional groups could be incorporated into the HBPs 105 from various small molecular and macromonomers by the thiol-yne reaction. Alternating-block hyperbranched polymers were subsequently synthesized using the highly versatile thiol-yne reaction by the same authors.⁷⁷ Two linear block macromonomers were first designed and prepared from dimethyl 110 acrylamide/t-butyl acrylate and styrene as monomers, respectively, by reversible addition-fragmentation chain-transfer polymerization. The prepared macromonomers were easily polymerized to form alternating-block hyperbranched polymers. The same authors also studied the self-assembly properties of the 115 alternating-block hyperbranched polymers at various pH. The results showed that these products could potentially be used in

drug delivery and other applications.

A disadvantage of this method was the very tedious procedure required for the synthesis of monomers containing both thiol and ethynyl groups. Recently, Gao and co-workers tried to s solve this problem concerning the preparation of the monomers and hyperbranched polythioether-ynes by a new strategy, namely a combination of thiol–yne click reaction and thiol–halogen click-like reaction.⁷⁸ Two series of hyperbranched polythioether-ynes with high molecular weights and high degrees of branching

¹⁰ were readily achieved. The main advantages of this strategy include its simplicity, rapidity, flexibility, and high efficiency for the preparation of multifunctional HBPs.

polymerization macroinitiators.^{82,83} This approach avoids the ⁶⁰ direct manipulation of these reactive and noxious thiol chemicals, and highly limits the potential of oxidative disulfide formation. Furthermore, the author also adopted this methodology to synthesize materials with site specific functionalization for potential drug delivery applications.

65 2.3 Enzyme-catalyzed polymerization

In recent years, the application of enzyme-catalyzed polymerization for the synthesis of HBPs has received increasing attention because of its unique advantages. Polymer synthesis by

Scheme 3. Synthesis of hyperbranched poly(amine-ester)s with ester terminals (reproduced with permission of the Royal Society of Chemistry

from ref. 91).

15



35

Compared to the developed azide–alkyne and thiol–ene click polymerizations, however, the application of the thiol–yne click polymerization to synthesize functional polymers is still ⁴⁰ limited.⁷⁹ Considering its advantages of readily available monomers and simple reaction conditions, it is anticipated that

- the thiol–yne click polymerization will be further developed for the preparation of functional polymers and find extensive applications in diverse areas.
- ⁴⁵ In addition, thiol-Michael addition chemistry has also been studied as an example of click chemistry for the preparation of functional materials.^{80,81} Although efficient and highly useful, thiol-based chemistry has their intrinsic drawbacks mainly derived from the manipulation of thiol functionalized compounds
- ⁵⁰ which are pungent, sensitive to oxidation and commercially limited to a relatively small range of simple molecules. Recently, in order to overcome these drawbacks, Rannard's group has carried out extensive research, and developed an efficient one-pot deprotection/functionalization thiol-Michael addition strategies in
- ss which xnthates have been utilized as protection groups of thiols that was successfully applied for the preparation of xanthatefunctional linear-dendritic polymers hybrids and four generations of xanthate-functionalized dendron atom transfer radical

enzyme-catalyzed polymerization was first reported in the mid-1980s, but only became established in 1999, after a breakthrough ⁹⁵ was achieved with the use of immobilized lipase B (Novozyme 435) as a catalyst to give polymers.⁸⁴ Various lipases from different sources have been used in enzyme-catalyzed polymer syntheses. Among these, lipase B, a hydrolysis enzyme from *C. antarctica* (CALB) immobilized on Lewatit by physical ¹⁰⁰ adsorption, has been the most widely employed enzyme for polymer synthesis due to its excellent activity towards many monomers.

Application of enzymatic methods for the synthesis of hyperbranched polymers was first reported by Frey and co-¹⁰⁵ workers.⁸⁵ In their study, ε-caprolactone was used as a cyclic AB monomer and 2,2-bis(hydroxymethyl)butyric acid served as an AB₂ co-monomer, and the reaction was catalyzed by immobilized lipase B under mild conditions. A series of hyperbranched copolyesters with different DB was successfully prepared by the ¹¹⁰ enzymatic route based on a novel combination of ring-opening AB polymerization and AB₂ polycondensation. The results showed this enzymatic route to be general and highly efficient, giving access to various hyperbranched aliphatic polyesters.

Utilization of enzyme-catalyzed polymerization for the ¹¹⁵ synthesis of HBPs has since attracted more and more attention. In

2010, López-Luna and co-workers reported the CALB-catalyzed synthesis of hyperbranched polylactones.⁸⁶ In their work, 2,2-bis(hydroxymethyl)butyric acid was also used as an AB₂ co-monomer and 1,1,1,2-tetrafluoroethane was used as a green, non-

- ⁵ toxic, relatively polar, and hydrophobic solvent. The main drawback was that the obtained hyperbranched polylactones had a low degree of branching, which may have been due to the limited solubility of 2,2-bis(hydroxymethyl)butyric acid in 1,1,1,2-tetrafluoroethane.
- ¹⁰ Subsequently, the same authors made improvements to the reaction medium.⁸⁷ In their later study, hyperbranched poly-llactides were synthesized by a novel CALB-catalyzed process in an ionic liquid, using bis(hydroxymethyl)butyric acid as the AB₂ core co-monomer and 1-butyl-3-methylimidazolium
- ¹⁵ hexafluorophosphate [C₄MIM][PF₆] as the medium. The results were consistent with those of other reported theoretical studies, showing imidazolium-based ionic liquids to be suitable for sustaining lipase activities with a high solvating ability for the substrates and products with high molecular weights.⁸⁸⁻⁹⁰
- $_{20}$ However, the above studies were restricted to co-polymerization of ϵ -caprolactone or δ -valerolactone with 2,2-bis(hydroxymethyl)butyric acid as an AB_2-type core to prepare HBPs.
- More recently, a series of novel multifunctional poly ²⁵ (amine-ester)-type HBPs has been successfully prepared by CALB-catalyzed polycondensation of triethanolamine (TEOA) and diesters by Xu and co-workers.⁹¹ This synthesis is shown in Scheme 3. The reaction conditions were optimized, and the precise structure of the products, including the degree of
- ³⁰ branching, the percentages of the components, and outer surface groups, were confirmed by FTIR, NMR, GPC, and so on. Moreover, the biological properties of the multifunctional poly(amine-ester) HBPs were systematically investigated, including their biodegradability, biocompatibility, cytotoxicity,
- ³⁵ and amphiphilicity. The results indicated that the synthesis of multifunctional poly(amine-ester) HBPs can provide materials with potential applications in biomedical fields, such as drug delivery and therapeutics.
- In addition, significant advances have been made by 40 employing novel enzymes in polymer synthesis. For example, a new lipase from the yeast *Yarrowia lipolytica* was used in the enzyme-catalyzed ring-opening polymerization of lactones by Barrera-Rivera and co-workers.^{92,93} The results showed that this enzyme could efficiently catalyze the ring-opening 45 polymerization of ε -caprolactone in both organic solvents and various ionic liquids. The main advantages were that pre-culture
- steps were not needed, fast rate of the chemical reactions, and a reduction in the total reaction time.
- Moreover, a novel cutinase from *Humicola insolens* showing ⁵⁰ activity for polyester synthesis was explored by Hunsen and coworkers.⁹⁴ In their study, the cutinase was immobilized on Lewatit and used as a catalyst for ring-opening polymerizations with ε -caprolactone and ω -pentadecalactone. The kinetics and mechanism of the reaction with the immobilized cutinase were
- ss subsequently explored by the same authors,⁹⁵ and the results indicated that the *Humicola insolens* cutinase was an excellent new biocatalyst with great promise for the synthesis of further HBPs.

Thermophilic enzymes have also been examined as potential ⁶⁰ catalysts in polymer synthesis due to their outstanding stability towards high temperatures, organic solvents, and chemical denaturants. For example, Yang and co-workers investigated the catalytic activities of thermophilic esterase from *Archaeoglobus fulgidus* and thermophilic lipase from *Fervidobacterium nodosum* ⁶⁵ in the ring-opening polymerization of ε -caprolactone.^{96,97} The results showed that these thermophilic enzymes had stronger affinity for ε -caprolactone than the widely used CALB, which was confirmed by kinetic analysis and molecular docking studies.

70 Scheme 4. Synthesis protocol of hyperbranched polymers and its interaction with aflatoxins B1 (AFB1) (reproduced with permission of Elsevier Science Ltd from ref. 99).



To sum up, enzyme-catalyzed polymerization has undergone rapid development during the last decade by virtue of its high catalytic activity, good enantioselectivity, mild reaction 110 conditions, and the biocompatibility and biodegradability of the catalytic products. It shows great potential for more environmentally friendly synthesis of polymers, making it particularly suitable for the synthesis of biomedical materials, drug intermediates, and so on.⁹⁸ Although this research has 115 started only recently, with improvement of the reaction conditions and exploration of novel enzyme catalysts and reaction media, enzyme-catalyzed polymerization looks set to play an important role in future polymer science.

3. Application of HBPs in analytical chemistry

HBPs have been extensively applied in analytical chemistry s areas such as immunosensors, extraction, heavy metal detection, early diagnosis, biotoxin detection, and so on. These applications rely on their large numbers of terminal functional groups, modification of which may provide a means of controlling the solubility, reactivity, and electrochemical and luminescence properties of HBPs.

3.1 Applications of HBPs in sample pretreatment

Sample pretreatment is of major importance in various analytical fields. The main objective is to achieve lower limits of detection by removing matrix constituents that may affect the 15 detection or enrichment of analytes. Many methods, such as solid-phase microextraction, miniature clean-up columns with various extraction sorbents, and antibody-based clean-up columns, have hitherto been employed. However, because of the complexity of the matrices, a simpler, more rapid, and reliable 20 sample preparation procedure still needs to be devised. polycyclic aromatic hydrocarbons (PAHs).¹⁰⁰ Compared to conventional adsorbent materials, the prepared porous polymer monolith exhibited higher enrichment efficiency and selectivity for PAHs.

60 3.2 Applications of HBPs in immunosensors

Ge and co-workers synthesized a HBP by polycondensation between *p*-phenylenediamine and trimesic acid, and then conjugated it with lipase using carbodiimide as a coupling reagent.¹⁰¹ The results showed that by using the HBP as an 65 immobilization matrix the natural activity of the lipase was maintained, and that the conjugated lipase exhibited significantly enhanced stability at high temperature or in the presence of organic solvents compared to its native counterpart.

Following on from this experiment, there have been many ro researchers starting to investigate the applications of HBPs as carriers to immobilize bioactive compounds such as enzymes, antigens, antibodies, *etc.*, on different activated substrates as sensitive or amplifying elements for the determination of various targets. Wherein, immunosensors based on HBPs as amplifying rs elements has received considerable attention in the fields of polymer science and chemical analysis. The assay principle of this immunosensor is shown in Scheme 5.

Scheme 5. Schematics illustration of immunosensor design based on HBPs as signal amplifying element.



Liu and co-workers established a novel chemical clean-up procedure coupled to high-performance liquid chromatography ⁴⁵ for the determination of aflatoxins in cereal powder samples.⁹⁹ In their work, a new well-defined HBP containing hydrogenbonding donor groups and π electrons was prepared and used as a solid-phase extraction sorbent to extract and enrich aflatoxins

based on non-covalent interactions (Scheme 4). The prepared 50 HBP showed higher adsorption efficiencies compared to a traditional C18 sorbent, indicating potential application as a sorbent in the rapid enrichment of trace analytes in real matrices. In addition, the same research group also synthesized a novel HBP containing stereocontortedcores for use as an on-line solid-55 phase microextraction adsorbent for highly efficient adsorption of ¹⁰⁰ For example, Shen and co-workers reported the use of a novel aromatic–aliphatic hyperbranched polyamide as a support for antibody immobilization in piezoelectric immunosensors for the detection of α -fetoprotein (AFP), a widely used tumor marker for the diagnosis of patients with germ cell tumors and ¹⁰⁵ hepatocellular carcinoma.¹⁰² In this work, they first synthesized the HBP through direct polycondensation of the monomer 5-[3-(4-aminophenyl)propionylamino]isophthalic acid, then activated it with EDC/NHS, and subsequently covalently linked it to electrodes that had been modified with cystamine. The results ¹¹⁰ showed that the aromatic–aliphatic hyperbranched polyamide could increase the amount of antibody immobilized and thus increased the sensitivity of the whole piezoelectric immunosensor. The main disadvantage was the covalent attachment between the antibody and the HBP, which may have lowered the antibody activity.

- Subsequently, the authors improved the immobilization of 5 the antibody by the combined use of HBPs and protein A.¹⁰³ In this strategy, protein A was used as a binding material to immobilize antibodies onto an HBP that had been synthesized from *p*-phenylenediamine and trimesic acid and had already been covalently linked to the sensor. More importantly, the method
- ¹⁰ provided a new strategy for highly sensitive clinical detection of target analytes and should also be applicable to the construction of other kinds of immunosensors.

Moreover, an ultra-sensitive, magnetic nanospheres-based electrochemical immunoassay amplified method using HBP

- ¹⁵ conjugates to detect hepatitis B antigen (HBsAg) was developed by Ding and co-workers.¹⁰⁴ In their work, hepatitis B surface antibodies were covalently immobilized on the surface of magnetic nanospheres, while hepatitis B secondary antibody labeled with horseradish peroxidase (HRP) was covalently linked
- $_{20}$ to the HBP. A sandwich-type immuno-complex was formed when HBsAg was present. The concentration of HBsAg was then determined by means of differential pulse voltammetry. The detection limit of this method was found to be 0.008 $\mu g/L$, which is lower than that of conventional ELISA. The results indicated
- 25 that by using the HBP as an immobilization substrate, the activity of the immobilized bioactive compound was maintained, demonstrating good prospects for practical clinical analysis.
- Other researchers have carried out a great deal of excellent work on the applications of HBPs in different types of sensors ³⁰ and have obtained many promising results.^{105–110} These detection methods provide powerful candidates for sensor technologies, showing high sensitivity and rapid detection of various targets, and hence may find widespread applications.
- Recently, in our lab, a successful attempt has been made in ³⁵ the synthesis of HBPs via polycondensation of AB₂ type monomer. The resulting HBPs contain functional end groups in which one sulfhydryl group and large number amino groups. Based on HBPs as signal amplifying element, we have successfully developed a simple, rapid and ultrasensitive
- ⁴⁰ immunochromatographic strip assay for the determination of bisphenol A (BPA) which is an estrogen mimic. The details were:
 1) synthesis of HBPs; 2) preparation the hyperbranched immune complexes: the conjugation of antibody and gold nanoparticles with HBPs; 3) immunochromatographic strip assay. The signal
- ⁴⁵ enhancement method enabled us to achieve BPA detection with the visual sensitivity of 1 ng/mL by naked eyes, which is at least 10-fold improvement of the sensitivity of traditional lateral flow strip methods. The experiment demonstrated that the signal amplified method based on HBPs could be adopted for all lateral
- ⁵⁰ flow strip detection methods to analyze various environmental pollutants.

4. Conclusions and outlook

This short review has summarized some of the notable advances in the synthesis methods for HBPs, as well as the ⁵⁵ application of HBPs in analytical chemistry. These different synthesis methods each have certain advantages and disadvantages. The synthesis of HBPs with both high molecular weight and a high degree of branching by the slow monomer addition method of is a versatile and favorable method. It will continue to be widely

used for the synthesis of HBPs, and it will certainly be combined with other techniques.

Click chemistry is an important and ideal method for the synthesis of HBPs. Although research on click polymerization 65 has led to great progress being made, its further development is still full of challenges that await exploration. Opportunities mainly exist in areas such as the design of new monomers, the development of catalytic systems, exploration of applications, and so on. It is anticipated that through the collective efforts of 70 researchers, click polymerization will be further developed to become a versatile, efficient, reliable, and powerful tool for the synthesis of various novel functional hyperbranched polymers. Enzyme-catalyzed polymerization shows significant promise for the synthesis of HBPs. However, its research is still in the 75 exploratory stage. Further research is required into problems such as the control of reaction conditions, exploration of novel reaction media and novel enzyme catalysts, and so forth. Besides, the application of HBPs demonstrated in this review only opens a narrow window.

In summary, rapid progress has been made in the application of different HBPs. However, the application of HBPs in analytical chemistry fields is still in its infancy. With deeper and broader research, HBPs are anticipated not only to make a great contribution to the development of polymer science, but also to ss be functionalized for practical applications.

Acknowledgment

The authors thank the National Natural Science Foundation of China (NNSFC) Project (Grant No. 31260412) for financial support.

90 Notes and references

- ^a Analysis and Testing Center, Xinjiang Academy of Agriculture and Reclamation Science, Wuyi Road No. 221, Shihezi, 832000, P.R.China
 ^b Supervision and Testing Center Food Quality, Ministry of Agriculture (Shihezi), Wuyi Road No. 221, Shihezi, 832000, P.R.China
- ⁹⁵ ^c Key Laboratories of Sheep Breeding and Reproduce, Xinjiang Academy of Agriculture and Reclamation Science, Wuyi Road No. 221, Shihezi, 832000, P.R.China

* Corresponding author. *E-mail:* shzpxy@163.com

- 100 1 Tomalia, D. A.; Fréchet, J. M. J. Polym. Sci. Part A: Polym. Chem. 2002, 40, 2719-2728.
 - 2 Gao, C.; Yan, D.Y. Prog. Polym. Sci. 2004, 29, 183-275.
 - 3 Carlmark, A.; Hawker, C.; Hult, A.; Malkoch, M. Chem. Soc. Rev. 2009, 38, 352-362.
- ¹⁰⁵ 4 Voit, B. I.; Lederer, A. Chem. Rev. 2009, 109, 5924-5973.
 - 5 Satoh, T. Int. J. Polym. Sci. 2012. doi:10.1155/2012/816163.
 - 6 Satoh, T.; Tamaki, M.; Taguchi, T.; Misaka, H.; Hoai, N. T.; Sakai, R.; Kakuchi, T. J. Polym. Sci. Part A: Polym. Chem. 2011, 49, 2353-2365.
- 110 7 Wang, N.; Ding, L. J. Polym. Res. 2012, 19, 1-8.
 - 8 Liang, H.; Li, X.; Xiong, L.; Huang, S. Synth. React. Inorg. M. 2013, 43, 1034-1039.
 - 9 Xiao, L.; Liao, L.; Guo, X.; Liu, L. Macromol. Chem. Phys. 2013, 214, 2500-2506.

- 10 Sato, E.; Uehara, I.; Horibe, H.; Matsumoto, A. Macromolecules. 2014, 47, 937-943.
- 11 Bai, L. B.; Zhao, K.; Wu, Y. G.; Li, W. L.; Wang, S. J.; Wang, H. J.; Zhao, H. C. Chinese. J. Polvm. Sci. 2014, 32, 385-394.
- 5 12 Giardiello, M.; McDonald, T. O.; Martin, P.; Owen, A. and Rannard, S. P. J. Mater. Chem., 2012, 22, 24744-24752.
- 13 Harbron, R. L.; McDonald, T. O.; Rannard, S. P.; Findlay, P. H. and Weaver, J. V. M. Chem. Commun., 2012, 48, 1592-1594.
- 14 Yan, D. Y.; Zhou, Y. F. and Hou, J. Science, 2004, 303, 65-67.
- 10 15 Dong, W. Y.; Zhou, Y. F.; Yan, D. Y.; Li, H.Q. and Liu, Y. Phys. Chem. Chem. Phys., 2007, 9, 1255-1262.
 - 16 Zhou, Y. F. and Yan, D. Y. Chem. Commun., 2009, 1172-1188.
 - 17 Sun, X. Y.; Huang, W.; Zhou, Y. F. and Yan, D. Y. Phys. Chem. Chem. Phys., 2010, 12, 11948-11953.
- 15 18 Liu, J. Y.; Pang, Y.; Huang, W.; Zhu, X. Y.; Zhou, Y. F. and Yan, D. Y. Biomaterials, 2010, 31, 1334-1341.
 - 19 Hatton, F. L.; Chambon, P.; McDonald, T. O.; Owen, A. and Rannard, S. P. Chem. Sci., 2014, 5, 1844–1853.
- 20 Zhou, Y.; Huang, W.; Liu, J.; Zhu, X.; Yan, D. Adv. Mater. 2010, 22, 4567-4590.
- 21 Wilms, D.; Stiriba, S. E.; Frey, H. Accounts. Chem. Res. 2009, 43, 129-141
- 22 Calderón, M.; Quadir, M. A.; Sharma, S. K.; Haag, R. Adv. Mater. 2010, 22, 190-218.
- 25 23 Yan, D.; Gao, C.; Frey, H. (Eds.). John Wiley & Sons, 2011, Vol. 8. 24 Yates, C. R.; Hayes, W. Eur. Polym. J. 2004, 40, 1257-1281.
 - Jin, H.; Huang, W.; Zhu, X.; Zhou, Y.; Yan, D. Chem. Soc. Rev. 2012, 25 41, 5986-5997
- 26 Müller, A. H.; Yan, D.; Wulkow, M. Macromolecules. 1997, 30, 7015-7023
- 27 Yan, D.; Müller, A. H.; Matyjaszewski, K. Macromolecules. 1997, 30, 7024-7033.
- Bharathi, P.; Moore, J. S. Macromolecules. 2000, 33, 3212-3218. 28
- 29 Sunder, A.; Hanselmann, R.; Frey, H.; Mülhaupt, R. Macromolecules. 1999, 32, 4240-4246.
- 30 Magnusson, H.; Malmström, E.; Hult, A. Macromolecules. 2001, 34, 5786-5791.
- Rahm, M.; Westlund, R.; Eldsäter, C.; Malmström, E. J. Polym. Sci. 31 Part A: Polym. Chem. 2009, 47, 6191-6200.
- 40 32 Gong, C.; Miravet, J.; Fréchet, J. M. J. Polym. Sci. Part A: Polym. Chem. 1999, 37, 3193-3201.
- 33 Kainthan, R. K.; Muliawan, E. B.; Hatzikiriakos, S. G.; Brooks, D. E. Macromolecules. 2006, 39, 7708-7717.
- 34 Wilms, D.; Wurm, F.; Nieberle, J.; Böhm, P.; Kemmer-Jonas, U.; Frey, H. Macromolecules. 2009, 42, 3230-3236.
- Rokicki, G.; Rakoczy, P.; Parzuchowski, P.; Sobiecki, M. Green. 35 Chem. 2005, 7, 529-539.
- Tamaki, M.; Taguchi, T.; Nakabayashi, S.; Mori, K.; Kitajyo, Y.; 36 Sakai, R.; Satoh, T. Polym. Chem. 2010, 1, 82-92.
- 50 37 Zhou, Z.; Jia, Z.; Yan, D. Sci. China. Chem. 2010, 53, 891-897.
- 38 Kolb, H. C.; Finn, M. G.; Sharpless, K. B. Angew. Chem. Int. Edit. 2001, 40, 2004-2021.
- Chu, C.; Liu, R. Chem. Soc. Rev. 2011, 40, 2177-2188. 39
- 40 Tasdelen, M. A.; Kahveci, M. U.; Yagci, Y. Prog. Polym. Sci. 2011, 36.455-567.
- 41 Lau, Y. H.; Rutledge, P. J.; Watkinson, M.; Todd, M. H. Chem. Soc. Rev. 2011, 40, 2848-2866.
- Fu, R.; Fu, G. D. Polym. Chem. 2011, 2, 465-475. 42
- Scheel, A. J.; Komber, H.; Voit, B. I. Macromol. Rapid. Comm. 2004, 43 25, 1175-1180
- 44 Malkov, G. V.; Lodygina, V. P.; Shastin, A. V.; Estrin, Y. I.; Badamshina, E. R.; Mikhailov, Y. M. Polym. Sci. Ser. B. 2007, 49, 301-304.
- 45 Katritzky, A. R.; Meher, N. K.; Hanci, S.; Gyanda, R.; Tala, S. R.; Mathai, S.; Ciaramitaro, D. A. J. Polym. Sci. Part A: Polym. Chem. 2008, 46, 238-256.
- Yadav, S. K.; Mahapatra, S. S.; Cho, J. W. Polymer. 2012, 53, 2023-46 2031
- 47 Shenoi, R. A.; Lai, B. F.; Kizhakkedathu, J. N. Biomacromolecules. 2012, 13, 3018-3030.

- 48 Xie, J.; Hu, L.; Shi, W.; Deng, X.; Cao, Z.; Shen, Q. Polym. Int. 2008, 57, 965-974.
- 49 Qin, A.J.; Lam, J.W.Y.; Jim, C. K. W.; Zhang, L.; Yan, J. J.; Häussler, M.; Liu, J. Z.; Dong, Y. Q.; Liang, D. H.; Chen, E. Q.; Jia, G. C.; Tang, B. Z. Macromolecules. 2008, 41, 3808-3822. 75
- Wang, J.; Mei, J.; Yuan, W. Z.; Lu, P.; Qin, A. J.; Sun, J. Z.; Ma, Y. 50 G.; Tang, B. Z. J. Mater. Chem. 2011, 21, 4056-4059.
- Wu, W.; Fu, Y.; Wang, C.; Ye, C.; Qin, J.; Li, Z. Chem-Asian. J. 51 2011, 6, 2787.
- 80 52 Johnson, J. A.; Finn, M. G.; Koberstein, J. T.; Turro, N. J. Macromol. Rapid. Comm. 2008, 29, 1052-1072.
- 53 Qin, A.; Lam, J. W.; Dong, H.; Lu, W.; Jim, C. K.; Dong, Y.; Tang, B. Z. Macromolecules. 2007, 40, 4879-4886.
- 54 Qin, A.; Tang, L.; Lam, J. W.; Jim, C. K.; Yu, Y.; Zhao, H.; Tang, B. Z. Adv. Funct. Mater. 2009, 19, 1891-1900.
- 55 Li, H.; Mei, J.; Wang, J.; Zhang, S.; Zhao, Q.; Wei, Q.; Tang, B. Z. Sci. China. Chem. 2011, 54, 611-616.
- 56 Li, H.; Wang, J.; Sun, J. Z.; Hu, R.; Qin, A.; Tang, B. Z. Polym. Chem. 2012, 3, 1075-1083.
- 90 57 Li, H.; Wu, H.; Zhao, E.; Li, J.; Sun, J. Z.; Qin, A.; Tang, B. Z. Macromolecules. 2013, 46, 3907-3914.
 - 58 Wu, H.; Li, H.; Kwok, R. T.; Zhao, E.; Sun, J. Z.; Qin, A.; Tang, B. Z. Sci. Rep. 2014, 4.
 - Wang, C.; Ren, P. F.; Huang, X. J.; Wu, J.; Xu, Z. K. Chem. Commun. 59 2011, 47, 3930-3932.
 - Altintas, O.; Vogt, A. P.; Barner-Kowollik, C.; Tunca, U. Polym. 61 Chem. 2012, 3, 34-45.
- 130, 5062-5064.
- 64 Rissing, C.; Son, D. Y. Organometallics. 2009, 28, 3167-3172.
- 2011, 49, 1735-1744.
- Xue, L.; Yang, Z.; Wang, D.; Wang, Y.; Zhang, J.; Feng, S. J. 105 66 Organomet. Chem. 2013, 732, 1-7.
- 67 Foix, D.; Ramis, X.; Serra, A.; Sangermano, M. Polymer. 2011, 52, 3269-3276.
- 68 Flores, M.; Tomuta, A. M.; Fernandez-Francos, X.; Ramis, X.; Sangermano, M.; Serra, A. Polymer. 2013, 54, 5473-5481.
- 69 Imbesi, P. M.; Raymond, J. E.; Tucker, B. S.; Wooley, K. L. J. Mater. Chem. 2012, 22, 19462-19473.
- 70 Zhang, L.; Zhu, J.; Zhou, W.; Wang, J.; Wang, Y. Thermochim. Acta. 2011, 524, 128-134.
- 115 71 Alkan, C.; Günther, E.; Hiebler, S.; Ensari, Ö. F.; Kahraman, D. Sol. Energy. 2012, 86, 1761-1769.
 - 72 Liu, J.; Lam, J. W.; Jim, C. K.; Ng, J. C.; Shi, J.; Su, H.; Tang, B. Z. Macromolecules. 2010, 44, 68-79.
- 73 Fairbanks, B. D.; Sims, E. A.; Anseth, K. S.; Bowman, C. N. Macromolecules. 2010, 43, 4113-4119. 120
 - 74 Türünç, O.; Meier, M. A. J. Polym. Sci. Part A: Polym. Chem. 2012, 50, 1689-1695.
 - 75 Lomba, M.; Oriol, L.; Alcalá, R.; Sánchez, C.; Moros, M.; Grazú, V.; Dela Fuente, J. M. Macromol. Biosci. 2011, 11, 1505-1514.
- Konkolewicz, D.; Gray-Weale, A.; Perrier, S. J. Am. Chem. Soc.. 125 76 2009, 131, 18075-18077.
 - 77 Konkolewicz, D.; Poon, C. K.; Gray-Weale, A.; Perrier, S. Chem. Commun. 2011, 47, 239-241.
 - 78 Han, J.; Zhao, B.; Tang, A.; Gao, Y.; Gao, C. Polym. Chem. 2012, 3, 1918-1925.
 - 79 Yao, B.; Sun, J.; Qin, A.; Tang, B. Z. Chinese. Sci. Bull. 2013, 58, 2711-2718.
 - 80 Lowe, A. B. Polym. Chem., 2010, 1, 17-36.
 - 81 Nair, D. P.; Podgórski, M.; Chatani, S.; Gong, T.; Xi, W.; Fenoli, C. R. and Bowman, C. N. Chem. Mater., 2014, 26, 724-744.
 - 82 Auty, S. E. R.; Andrén, O. C. J.; Hern, F. Y.; Malkoch, M. and Rannard, S. P. Polym. Chem., DOI: 10.1039/c4py01253d.
 - Auty, S. E. R.; Andrén, O.; Malkoch, M. and Rannard, S. P. Chem. 83 Commun., 2014, 50, 6574-6577.
- 140 84 Deng, F.; Bisht, K. S.; Gross, R. A.; Kaplan, D. L. Macromolecules. 1999, 32, 5159-5161.

- 95 60 Tasdelen, M. A. Polym. Chem. 2011, 2, 2133-2145. 62 Franc, G.; Kakkar, A. K. Chem. Soc. Rev. 2010, 39, 1536-1544.
 - 100 63 Killops, K. L.; Campos, L. M.; Hawker, C. J. J. Am. Chem. Soc. 2008,

 - 65 Roy, R. K.; Ramakrishnan, S. J. Polym. Sci. Part A: Polym. Chem.

- 85 Skaria, S.; Smet, M.; Frey, H. Macromol. Rapid. Comm. 2002, 23, 292-296.
- 86 García-Arrazola, R.; López-Guerrero, D. A.; Gimeno, M.; Bárzana, E. J. Supercrit. Fluid. 2009, 51, 197-201.
- 5 87 Mena, M.; López-Luna, A.; Shirai, K.; Tecante, A.; Gimeno, M.; Bárzana, E. Bioproc. Biosyst. Eng. 2013, 36, 383-387.
- 88 Yoshizawa-Fujita, M.; Saito, C.; Takeoka, Y.; Rikukawa, M. Polym. Advan. Technol. 2008, 19, 1396-1400.
- 89 Chanfreau, S.; Mena, M.; Porras-Domínguez, J. R.; Ramírez-Gilly, M.;
 ¹⁰ Gimeno, M.; Roquero, P.; Bárzana, E. *Bioproc. Biosyst. Eng.* 2010, *33*, 629-638.
- 90 Mena, M.; Chanfreau, S.; Gimeno, M.; Bárzana, E. Bioproc. Biosyst. Eng. 2010, 33, 1095-1101.
- 91 Xu, F.; Zhong, J.; Qian, X.; Li, Y.; Lin, X.; Wu, Q. Polym. Chem. 15 2013, 4, 3480-3490.
- 92 Barrera-Rivera, K. A.; Flores-Carreón, A.; & Martínez-Richa, A. J. Appl. Polym. Sci. 2008, 109, 708-719.
- 93 Barrera-Rivera, K. A.; Marcos-Fernández, A.; Vera-Graziano, R.; Martínez-Richa, A. J. Polym. Sci. Part A: Polym. Chem. 2009, 47, 5792-5805.
- 94 Hunsen, M.; Azim, A.; Mang, H.; Wallner, S. R.; Ronkvist, A.; Xie, W.; Gross, R. A. *Macromolecules*. 2007, 40, 148-150.
- 95 Hunsen, M.; Abul, A.; Xie, W.; Gross, R. Biomacromolecules. 2008, 9, 518-522.
- 25 96 Levisson, M.; van der Oost, J.; Kengen, S. W. Extremophiles. 2009, 13, 567-581.
 - 97 Ma, J.; Li, Q.; Song, B.; Liu, D.; Zheng, B.; Zhang, Z.; Feng, Y. J. Mol. Catal. B-Enzym. 2009, 56, 151-157.
- 98 Yang, Y.; Yu, Y.; Zhang, Y.; Liu, C.; Shi, W.; Li, Q. Process. Biochem. 2011, 46, 1900-1908.
- 99 Liu, X.; Li, H.; Xu, Z.; Peng, J.; Zhu, S.; Zhang, H. Anal. Chim. Acta. 2013, 797, 40-49.
- 100 Guan, X.; Zhao, C.; Liu, X.; Zhang, H. J. Chromatogr. A. 2013, 1302, 28-33.
- 35 101 Ge, J.; Yan, M.; Lu, D.; Zhang, M.; Liu, Z. Biochem. Eng. J. 2007, 36, 93-99.
 - 102 Shen, G.; Liu, M.; Cai, X.; Lu, J. Anal. Chim. Acta. 2008, 630, 75-81.
 - 103 Shen, G.; Cai, C.; Wang, K.; Lu, J. Anal. Biochem. 2011, 409, 22-27.
- 40 104 Ding, F. L.; Cheng, Q.; Li, J. F.; Jiang, F. E. Chinese. J. Anal. Chem. 2012, 40, 1514-1518.
 - 105 Jiménez, A.; Armada, M.; Losada, J.; Villena, C.; Alonso, B.; Casado, C. M. Sensor. Actuat. B-Chem. 2014, 190, 111-119.
- 106 Wang, C.; Lin, M.; Liu, Y.; Lei, H. *Electrochim. Acta.* 2011, *56*, 1988-1994.
- 107 Tang, D.; Tang, J.; Su, B.; Chen, G. Biosens. Bioelectron. 2011, 26, 2090-2096.
- 108 Erdem, A.; Karadeniz, H.; Caliskan, A. Analyst. 2011, 136, 1041-1045.
- 50 109 Wu, X. R.; Li, J. R. Acta. Chim. Sinica. 2011, 69, 1211-1216.
- 110 Fu, C.; Li, J. P. Chinese. J. Anal. Chem. 2013, 41, 1762-1772.